



1 **Measurement report: Hygroscopicity of Size-Selective Aerosol**  
2 **Particles at Heavily Polluted Urban Atmosphere of Delhi:**  
3 **Impacts of Chloride Aerosol**

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15 **Abstract.** Recent studies reveal that wintertime chloride emission in the Delhi region is crucial in governing  
16 enhancement to theoretically calculated aerosol hygroscopicity and aerosol-bound liquid water to trigger Delhi's  
17 fog episodes. Here, we reported that the high volume fractional contribution of ammonium chloride into aerosol  
18 governs the high aerosol hygroscopicity and associated liquid water content based on the experimental data first  
19 time in Delh. The episodically high chlorides bonded with excess ammonia present in Delhi's atmosphere, which  
20 could lead to haze and fog formation under high relative humidity in the region. Therefore, our study suggests  
21 that controlling the plastic-contained waste, open burning, and e-waste industrial chloride emission could  
22 significantly minimize Delhi's heavily polluted haze/fog events. The high chloride (H-Cl) period was observed  
23 significantly ( $p < 0.05$ ) higher hygroscopicity ( $0.35 \pm 0.06$ ) compared to high biomass burning (H-BB) ( $0.18 \pm$   
24  $0.04$ ), high hydrocarbon-like organic aerosol (H-HOA) ( $0.17 \pm 0.05$ ), and relatively cleaner period ( $0.27 \pm 0.07$ ).

25 In this study, we present the measurement results of bulk aerosol composition of non-refractory PM1 from ACSM  
26 and size-resolved (Nucleation, Aitken, and Accumulated mode particles) hygroscopic growth factor and  
27 associated hygroscopicity parameter at 90% relative humidity (RH) measured using H-TDMA at Delhi Aerosol  
28 Supersite (DAS) first time. The hygroscopic parameter ( $\kappa_{\text{H-TDMA}_90\%}$ ) was significantly ( $p < 0.05$ ) enhanced with  
29 the size of the particles. The observed  $\kappa_{\text{H-TDMA}_90\%}$  ranged from .00 to 0.11 ( $0.03 \pm 0.02$ ), 0.05 to 0.22 ( $0.11 \pm$   
30  $0.03$ ), 0.05 to 0.30 ( $0.14 \pm 0.04$ ), 0.05 to 0.41 ( $0.18 \pm 0.06$ ), and 0.05 to 0.56 ( $0.22 \pm 0.07$ ) for 20, 50, 100, 150,  
31 and 200 nm aerosol particles, respectively. The Inorganic-to-organic aerosol ratio in aerosol modulated mainly  
32 the aerosol hygroscopicity. In addition, the accumulation mode particle's hygroscopicity was regulated potentially



33 by the volume fraction of  $\text{NH}_4\text{Cl}$  and OA in aerosol particles. Interestingly, our results reveal that the daytime  
34 flattening pattern of accumulation aerosol particles in diurnal variation is potentially due to counter the effect of  
35 increment of  $(\text{NH}_4)_2\text{SO}_4$  and  $\text{NH}_4\text{NO}_3$  and decrement of  $\text{NH}_4\text{Cl}$  and OA in aerosol particles.

## 36 **1. Introduction**

37 The Intergovernmental Panel on Climate Change (IPCC) reported that aerosol-cloud interaction is still not fully  
38 understood and has significant uncertainties in quantifying global radiative budgets. Aerosol hygroscopicity plays  
39 a pivotal role in overcoming and explaining these uncertainties. Hygroscopicity is crucial to understand how the  
40 aerosol particles act as cloud condensation nuclei (CCN) and forms fog droplets/haze at sub-saturated/nearly  
41 saturation and cloud droplets at atmospheric supersaturation levels (McFiggans et al., 2006; Topping and  
42 McFiggans, 2012). Its understanding is crucial to predict the aerosol size distribution and scattering properties  
43 better in global models under varying atmospheric humidity (RH) conditions (Randall et al., 2007).  
44 Hygroscopicity at higher RH atmospheric conditions leads to an enhanced aerosol cross-sectional area, resulting  
45 in efficient light scattering by the aerosol particles (Tang and Munkelwitz, 1994). It mainly depend on particle  
46 size and chemical composition. Generally, the inorganic salts such as ammonium salts of sulfate, nitrate, and  
47 chloride, are highly hygroscopic (Hu et al., 2011; Petters and Kreidenweis, 2007), organic aerosol (OA) are  
48 comparatively less hygroscopic (Jimenez et al., 2009; Kroll et al., 2011), while dust particles and black/elemental  
49 carbon particles are stated as hydrophobic (Seinfeld and Pandis, 2006). Further, the elevated atmospheric RH  
50 during winter and monsoon favour the formation of more oxidized secondary organic aerosol (SOA) via aqueous-  
51 phase (Ervens et al., 2011) and heterogeneous reactions (McNeill, 2015), leads to enhancement in organic aerosol  
52 hygroscopicity (Jimenez et al., 2009; Mei et al., 2013) which adversely impact on the local visibility (Li et al.,  
53 2016; Liu et al., 2012). However, aerosol loading inversely affects aerosol hygroscopicity (Mandariya et al.,  
54 2020a). Apart from it, aerosol loading is also a critical factor in deciding the lifetime of cloud, which affects the  
55 region's rain quantitatively (Albrecht, 1989; Lohmann and Feichter, 2005).

56 Over the past decades, aerosol hygroscopicity has been intensively measured using hygroscopic tandem  
57 differential mobility analyzer (H-TDMA) (Massling et al., 2005; Gysel et al., 2007; Mandariya et al., 2020;  
58 Swietlicki et al., 2008; Yeung et al., 2014; Kecorius et al., 2019) and CCN (Bhattu and Tripathi, 2015; Gunthe et  
59 al., 2011; Massoli et al., 2010; Ogawa et al., 2016) counter under sub- and supersaturation levels, respectively.  
60 Petters and Kreidenweis (2007) introduced a hygroscopicity parameter, kappa ( $\kappa$ ), to associate aerosol  
61 hygroscopicity with its chemical composition. Furthermore, hygroscopicity associated with OA potentially varies



62 with OA chemical properties like solubility, the extent of dissociation in aerosol water, and surface activity  
63 (Hallquist et al., 2009; Jimenez et al., 2009), leads to difficulty in the quantification of OA hygroscopicity, result  
64 in introducing more discrepancies in predicted and measured aerosol hygroscopicity. Hence, there is a need to  
65 explore measurement-based aerosol hygroscopicity for Delhi's atmosphere to understand the frequent haze/cloud  
66 formations better.

67 In past decades, fast economic growth and industrialization in the IGP led to severe air quality during wintertime  
68 (Wester et al., 2019). Delhi is potentially affected by local and regional air pollution problems in wintertime (Arub  
69 et al., 2020; Bhandari et al., 2020; Gani et al., 2019; Prakash et al., 2018). Recent studies (Gani et al., 2019; Rai  
70 et al., 2020) have shown chloride is one of the predominant factors to degrade the air quality in the Delhi region  
71 and significantly favour the haze/fog formation during winter (Gunthe et al., 2021). Trash and biomass burning  
72 for heating and waste degradation are among the main contributors to chloride in Delhi (Rai et al., 2020). A recent  
73 study conducted in Delhi reported that frequent high chloride events promotes high aerosol liquid water content  
74 under elevated humid condition leads to haze and poor visibility in the city (Chen et al., 2022). In addition, Gunthe  
75 et al. (2021) showed higher chloride also facilitates enhancement in aerosol hygroscopicity, however, this study  
76 was based on theoretical hygroscopicity. Therefore, it is essential to investigate the impacts of chloride on aerosol  
77 hygroscopicity and its potential to enhance aerosol-bound liquid water based on field measurements. Moreover,  
78 the hygroscopicity of the aerosol particles in the heavily polluted urban atmosphere, which confines to highly  
79 complex composition, is very limited, like Delhi, situated at Indo Gangetic Plain (IGP), India, where air quality  
80 severely degrades during haze/fog-dominated. To the author's best knowledge, the current study is first in Delhi,  
81 India, exploring a complex atmosphere of IGP using H-TDMA-measured aerosol hygroscopicity. Hence, it is  
82 essential to measure size-resolved aerosol hygroscopicity in Delhi's atmosphere and investigate its role in the  
83 context of high chloride.

## 84 **2. Experimental Methods**

### 85 **2.1 Aerosol Measurements**

86 Real-time atmospheric aerosol measurements were conducted simultaneously using Hygroscopic-Tandem  
87 Differential Mobility Analyzer (H-TDMA) and Aerodyne Aerosol Chemical Speciation Monitor (ACSM,  
88 Aerodyne Research, Billerica MA) during winter (1<sup>st</sup> February 2020 to 16<sup>th</sup> March 2020) at the Indian Institute of  
89 Technology (IIT) Delhi in Block 5, at the height of nearly 15 m. Details on the sampling site can be found in Arub  
90 et al. (2020). In this study, the HTDMA system was used to investigate the hygroscopic growth of size-resolved



91 particles at 90 % RH. The HTDMA system has been previously used in many field campaigns (Massling et al.,  
92 2007; Wu et al., 2013b; Zhang et al., 2016). The HTDMA system (TROPOS, Germany) is comprised of two  
93 Differential Mobility Analyzers (DMAs, type Hauke-median, TROPOS, Germany), a Condensation Particle  
94 Counter (CPC, Model 3772, TSI Inc., USA) along with a humidifier system located between the two DMAs. The  
95 role of first DMA is to select the quasi-monodisperse particles at a dry diameter ( $D_p$ , dry) with 30% RH. After  
96 that, the size-selected particles pass through a humidity conditioner, which can be adjusted from 30% to 90% RH  
97 by regulating the aerosol and sheath air flow by mixing dry air with RH<5% and humid air with ~95% RH  
98 (Maßling et al., 2003). The uncertainties associated with RH measurement at 90% RH is 1.0%. The particle  
99 hygroscopic growth distribution at dry size ( $D_p$ , dry) at a certain humidity can be easily determined with CPC.  
100 There are two humidity sensors (Vaisala) in the system for aerosol and sheath respectively. The humidity sensors  
101 positioned in the second DMA were calibrated automatically with 100 nm ammonium sulfate ((NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>)  
102 particles every 30 min at 90% RH to analyze the stability at high RH. The measurement error of the HTDMA  
103 mainly depends on the uncertainty in measuring and controlling the RH within the system (Su et al., 2010).  
104 Therefore, all RH sensors were calibrated using the Vaisala salt kit comprising LiCl, NaCl, KCl etc. prior the  
105 measurement campaign. Both the DMAs were size calibrated by applying the Latex particles with the standard  
106 size of 200 nm before the start of the measurement. The number concentration peak occurred at 203 nm, referring  
107 to accuracy of DMAs size selection at 1.5%. HTDMA system was operated at 90% RH to measure the hygroscopic  
108 growth factors (HGFs) for particles with  $D_p$ , dry of five different sizes i.e. 20, 50, 100, 150 and 200 nm. The time  
109 resolution of the full scan covering the five sizes was about 30 min.

110 ACSM was operated at nearly 0.1 lpm at 1 min time resolution in a temperature-controlled laboratory. ACSM  
111 was set to run to measure mass-to-charge ratio ( $m/z$ )  $m/z$  10 to  $m/z$  140. The ACSM measures non-refractory  
112 particulate matter less than 1  $\mu\text{m}$  (NR-PM<sub>1</sub>). The concentrate PM<sub>1</sub> aerosol beam was impacted on the vaporizer at  
113 600 °C and flash-vaporized compounds were subsequently ionized through impact ionization at 70 eV electron  
114 and detected with a quadrupole mass spectrometer (Ng et al., 2011). The 200 ms amu<sup>-1</sup> scan speed and pause  
115 setting at 125 for a sampling time (64 s) were set to acquire aerosol mass spectra in ACSM. Detailed operational  
116 procedures for the ACSM are explained elsewhere in Gani et al. (2019).

## 117 **2.2 Data Analysis**

### 118 **2.2.1 ACSM**



119 Details on ACSM calibration and data processing are in Patel et al., 2021. We conducted Positive matrix  
120 factorization (PMF) on the data and found a four-factor solution (hydrocarbon-like OA, “HOA”; biomass burning  
121 OA, “BBOA”; less-oxidized OA, “LO-OOA”; more-oxidized OA “MO-OOA) to best represent the data set.  
122 Further details about PMF analysis are in section S.1 of the SI.

123 The windrose plot was plotted by openair in R package (<http://www.r-project.org>, <http://www.openair-project.org>). The 48-hour back trajectory of air masses reaching Delhi super site (DSL) at 500 m above the ground  
124 at every hour for the entire study period was estimated by an offline based Hybrid Single-Particle Lagrangian  
125 Integrated Trajectory (HYSPLIT4) model developed by NOAA/Air Resources Laboratory (ARL) (Draxler and  
126 Hess, 1997). The input meteorological data for back trajectories were taken from the Global Data Assimilation  
127 System (GDAS 0.5 degree) archive maintained by ARL (<http://ready.arl.noaa.gov/archives.php>). Further,  
128 utilizing these estimated back trajectories as input combined with the measured mass fraction of chemical species  
129 of bulk aerosol, Potential Source Contribution Function (PSCF) analysis was carried out with the help of a tool  
130 called Zefir (V 3.7) written in Igor Pro (WaveMetrics). Detail description regarding Zefir tool can be found  
131 elsewhere (Petit et al., 2017). In addition, box plots reported in the subsequent section were also plotted with the  
132 help of this tool. The aerosol liquid water content (ALWC) as a function of inorganic species mass concentration,  
133 ambient temperature (T), and ambient relative humidity (RH), was calculated by ISORROPIA-II model  
134 (Fountoukis and Nenes, 2007).  
135

### 136 2.2.2 H-TDMA

137 The humidity sensor of DMA2 was automatically calibrated with 100 nm ammonium sulfate particles after each  
138 scan cycle. Overall, we recorded 1483 H-TDMA scans cycles. Afterward, the % difference between measured  
139 and theoretical growth factors (say  $\Delta q$ ) was calculated after each scan cycle for 100 nm ammonium sulfate  
140 particles. Those scan cycles came between  $\Delta q \leq \pm 5\%$ , were only carried out for further data treatment, and the  
141 rest scans cycles were discarded (Kecorius et al., 2019). Thus, we had 1102 H-TDMA scan cycles following this  
142 data quality check. Regarding good scan cycles, we had 1449, 1431, 1438, 1470, and 1420 good H-TDMA scans  
143 for 20, 50, 100, 150, and 200 nm particles, respectively, to further analyze. Afterward that, a piecewise linear  
144 TDMAinv algorithm, namely TDMAinv Toolkit, written in IgorPro and developed by Gysel et al. (2009), was  
145 used to do post-data treatment on the raw HGF. Because the measured distribution function is a skewed and  
146 smoothed integral transform of the actual growth factor probability density functions (GF-PDFs). A detailed  
147 description of the raw data processing in the TDMAinv toolkit to measure real HGFs is described in Gysel et al.



148 (2009). The TDMAinv toolkit was successfully used in various studies around the globe (Gysel et al., 2007; Liu  
149 et al., 2012; Sjogren et al., 2007; Wang et al., 2018a) and at Kanpur, India (Mandariya et al., 2020a). Besides, the  
150 RH in the DMA2 generally achieved the set value of 90% and remained stable within  $\pm 1\%$ , although occasionally,  
151 it faced a more considerable drift. All growth factors measured between 88 and 92% RH were corrected to a target  
152 value of 90% (HGF<sub>90%</sub>) (Gysel et al., 2007) using the kappa-model suggested in the TDMAinv toolkit (Gysel  
153 et al., 2009) to minimize this DMA2 RH drifts. After it, 979, 957, 972, 969, and 966 scans were found corrected  
154 at target RH for 20, 50, 100, 150, and 200 nm aerosol particles, respectively, which further averaged for 60 min  
155 time resolution and finally, these numbers reached to 425, 429, 419, 424, and 417, respectively.

156 Further, size-resolved hygroscopicity factors ( $\kappa$ , say  $\kappa_{H-TDMA_{90\%}}$ ) were calculated from the respective size-  
157 resolved target RH corrected HGFs using equation (1) kappa-Köhler theory (Mandariya et al., 2020a; Petters and  
158 Kreidenweis, 2007).

$$159 \quad \kappa_{H-TDMA_{90\%}} = (HGF_{90\%}^3 - 1) \left[ \frac{1}{RH} \exp \exp \left( \frac{4\sigma M_w}{RT\rho_w D_o HGF_{90\%}} \right) - 1 \right] \quad (1)$$

160 Where,  $\kappa_{H-TDMA_{90\%}}$  is the hygroscopicity factor at 90% RH, HGF<sub>90%</sub> is the size-resolve HGF at 90% RH, RH  
161 is the atmospheric relative humidity in fraction,  $\sigma$  is the surface tension of the aerosol liquid droplet-air interface  
162 at the droplet surface in N/m and can be assumed nearly to pure water, R is the universal gas constant in J K<sup>-1</sup> mol<sup>-1</sup>,  
163 M<sub>w</sub> is the molecular mass of water, T is the ambient temperature in Kelvin (K),  $\rho_w$  is the density of water in  
164 kg/m<sup>3</sup>, and D<sub>o</sub> is the dry mobility diameter of the particle in m.

#### 165 2.2.4 Derived Secondary Inorganic Salts

166 The ACSM mainly measures OA, NO<sub>3</sub>, SO<sub>4</sub>, NH<sub>4</sub>, and Cl. Therefore, we adopted a simplified ion-pairing scheme  
167 reported by Gysel et al. (2007). However, Gysel et al. (2007) did not include NH<sub>4</sub>Cl in their ion-pairing scheme;  
168 therefore, we elaborated this scheme and made some modifications in this scheme to include ACI in the  
169 calculation. Hence, our modified ion-pairing scheme includes ACI, AN, AS, ABS, and SA are shown below:

170 **Case-1  $R_{SO_4}(NH_4 \text{ to } SO_4) \leq 1$**

$$171 \quad SA = 98.0795 \times \max(0, (n_S - n_A))$$

$$172 \quad ABS = 115.11 \times n_A$$

$$173 \quad AS = 0$$



$$174 \quad AN = \left( \min \left( \left( n_A - \left( \frac{ABS}{115.11} \right) - \left( \frac{2 \times AS}{132.1405} \right) \right), n_N \right) \right) \times 80.0434$$

$$175 \quad ACl = \left( \min \left( n_C, \left( n_A - \left( \frac{ABS}{115.11} \right) - \left( \frac{2 \times AS}{132.1405} \right) - \left( \frac{AN}{80.0434} \right) \right) \right) \right) \times 53.54$$

176 **Case-2**  $1 < R_{SO_4} < 2$

$$177 \quad SA = 0$$

$$178 \quad ABS = 115.11 \times (2 \times n_S - n_A)$$

$$179 \quad AS = 132.1405 \times (n_S - n_A)$$

$$180 \quad AN = \left( \min \left( \left( n_A - \left( \frac{ABS}{115.11} \right) - \left( \frac{2 \times AS}{132.1405} \right) \right), n_N \right) \right) \times 80.0434$$

$$181 \quad ACl = \left( \min \left( n_C, \left( n_A - \left( \frac{ABS}{115.11} \right) - \left( \frac{2 \times AS}{132.1405} \right) - \left( \frac{AN}{80.0434} \right) \right) \right) \right) \times 53.54$$

182 **Case-3**  $R_{SO_4} \geq 2$

$$183 \quad SA = 0$$

$$184 \quad ABS = 0$$

$$185 \quad AS = 132.1405 \times n_S$$

$$186 \quad AN = \left( \min \left( \left( n_A - \left( \frac{ABS}{115.11} \right) - \left( \frac{2 \times AS}{132.1405} \right) \right), n_N \right) \right) \times 80.0434$$

$$187 \quad ACl = \left( \min \left( n_C, \left( n_A - \left( \frac{ABS}{115.11} \right) - \left( \frac{2 \times AS}{132.1405} \right) - \left( \frac{AN}{80.0434} \right) \right) \right) \right) \times 53.54$$



188 Here,  $n$  denotes the number of moles, whereas A, N, S, and C denotes the  $\text{NH}_4$ ,  $\text{NO}_3$ ,  $\text{SO}_4$ , and Cl species. We  
189 also predicted these inorganic salts concentrations from the ISORROPIA v2.1 model using  $\text{NH}_4$ ,  $\text{SO}_4$ ,  $\text{NO}_3$ , and  
190 Cl. We found a strong correlation and nearly unit slope (0.9999) between the calculated and modelled inorganic  
191 salts as presented in figure S1, which strongly justifies the new ion-pairing scheme adopted in this study.

### 192 3. Result and Discussions

#### 193 3.1 Overview of meteorology, trace gases, and aerosol characterization

194 Meteorological parameters provided a quick summary of the local weather conditions at the sampling site. Figure  
195 1 shows the hourly resolved temporal variability of meteorological parameters (RH, T, WD, and WS), particle  
196 number size distribution (PNSD), particle volume size distribution (PVSD), principal components in non-  
197 refractory  $\text{PM}_{10}$  and OA and their fractional mass contribution. In addition, the temporal variability in the  
198 atmospheric gases ( $\text{NO}_x$ , CO, and  $\text{SO}_2$ ) have shown in figure S5. Delhi's winter climate is mainly influenced by  
199 a depression created by the Western Disturbances caused by cold waves in the region. The ambient relative  
200 humidity (RH) and temperature (T) variability between 24.2–96.6% and 9.0–28.5°C with an average ( $\pm 1$  STD) of  
201  $56.0 \pm 18.2\%$  and  $18.7 \pm 4.2^\circ\text{C}$ , respectively, showed the Delhi's atmosphere was varied from wet and cold to dry  
202 and relatively warm from February to March, respectively. The nighttime was somewhat cold and humid  
203 compared to the daytime throughout the sampling. The ambient RH showed a diurnal pattern with a peak during  
204 early morning 06:00-07:00 hr and the valley during noontime 13:00-15:00 hr, while ambient temperature showed  
205 an opposite trend with a rise during noontime could be correlated with noontime higher solar radiation. This  
206 comparatively higher ambient temperature and  $\text{O}_3$  peak concentration during noontime indicate the daytime  
207 photooxidation process.

208 The wind speed (WS) and wind direction (WD) varied from 0.0 to 5.6 ( $1.0 \pm 1.0$ ) m/s and 4.0 to 345.7 ( $197.1 \pm$   
209  $84.4$ ) degrees from the North, respectively, as shown in figure S6. Predominant wind directions were WNW-  
210 WSW and E-ESE. It indicates that the atmosphere remained stagnant during the study period, and measured  
211 aerosol potentially represents Delhi's emissions and local aerosol chemistry.

212 Furthermore, ambient trace gases  $\text{NO}_x$  and CO showed substantial variability during the sampling period, with a  
213 peak in local burning activities. During intensely biomass burning activities, ambient  $\text{NO}_x$  ambient level reached  
214 a maximum of 421.2 ppb ( $58.4 \pm 61.9$ ). Moreover, the CO showed peak level at similar periods as  $\text{NO}_x$ , and its  
215 concentration varied from 0.0 to 7.66 ppm ( $0.58 \pm 0.79$ ), as shown in figure S5. The diurnal variation of trace  
216 gases is shown in figure 2 (f, g, h, and i). The CO and  $\text{NO}_x$  concentrations showed two peaks in days (06:00-





217 08:00 and 17:00-20:00), attributed to the morning local biomass/trash burning emissions, night-time traffic rush  
218 hours, and regional biomass burning activities. Besides, SO<sub>2</sub> showed a different trend from CO and NO<sub>x</sub>. SO<sub>2</sub>  
219 varied dynamically from 0.46 to 9.55 ppb ( $4.41 \pm 1.20$ ) and showed a peak during the morning (09:00-12:00 hr)  
220 and midnight (21:00 to 02:00 hr).

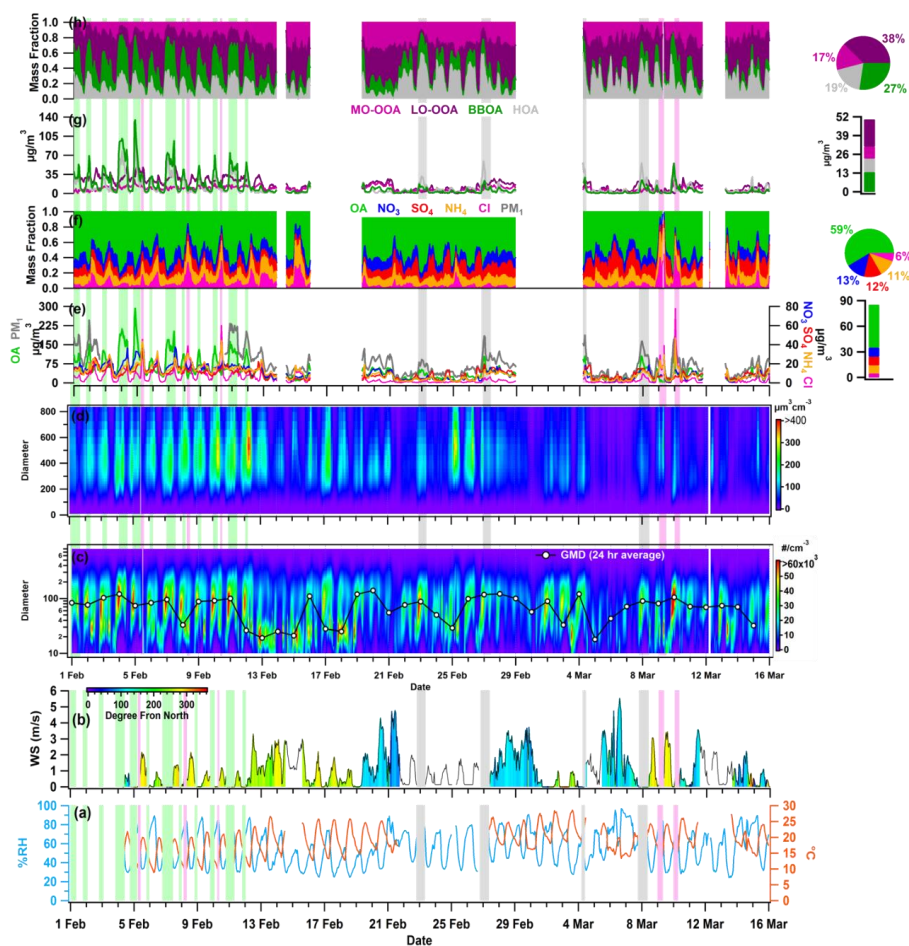
221 PM<sub>1</sub> particle number concentration varied from 408 to 29845 /cm<sup>3</sup> ( $11319 \pm 5552$ ). High particle number  
222 concentrations generally observed were associated with local burning events. The particle concentration increased  
223 in the evening (at 18:00 hr) and reached its maximum value at midnight, as shown in figure 2(e). This time  
224 generally indicates the resumption of residential burning activity and traffic emissions. These activities probably  
225 caused the lower geometric mean diameter (GMD) ( $\approx 47$  nm) of the particle number size distribution (PNSD),  
226 which further increased nearly up to 87 nm, as shown in figure 2(t), pointing out the night-time aging of organic  
227 aerosol. The hourly averaged mean diurnal GMD of PVSD varied from nearly 274 to 324 nm, as indicated in  
228 figure 2(y). However, during the sampling period, its variation was observed to vary between 221.3 and 418.4 nm  
229 with the mean value of  $309.1 \pm 33.1$  nm, which is close to this study's higher-end particle size of 200 nm  
230 hygroscopicity measurement. Therefore, ACSM bulk aerosol composition could be the best choice for discussing  
231 the hygroscopicity of 200 nm aerosol particles.

232 The hourly time-resolved NR-PM<sub>1</sub> (say hereafter PM<sub>1</sub>) concentration varied from 9.0 to 357.9  $\mu\text{g}/\text{m}^3$ , averaging  
233  $81.2 \pm 56.6$   $\mu\text{g}/\text{m}^3$ . This observation lies well within the boundary of 12.7-392  $\mu\text{g}/\text{m}^3$  (NR-PM<sub>1</sub>), reported by Gani  
234 et al. (2019) for the current site. Prakash et al. (2018) said that PM<sub>1</sub> mass concentration is 83% of PM<sub>2.5</sub>,  
235 representing the dominance of combustion-based particles. Further, we observed that ACSM measured NR-PM<sub>1</sub>  
236 (PM<sub>1</sub>) was highly correlated ( $r^2 = 0.83$ ,  $p < 0.05$ ) with MPSS measured PM<sub>1</sub>, assuming an effective aerosol density  
237 1.6 g/cm<sup>3</sup> (figure S2). It means that a substantial mass of PM<sub>1</sub> was composed with non-refractory material and  
238 other refractory material like black carbon, metals and crustal material, which was not measured in the current  
239 study, constituted less than 5% of PM<sub>1</sub> (Prakash et al., 2018). OA was the predominant fraction of PM<sub>1</sub> with an  
240 average mass concentration of  $46.5 \pm 39.6$   $\mu\text{g}/\text{m}^3$ , consistent with 112  $\mu\text{g}/\text{m}^3$  observed during winter (December-  
241 February) at the present site (Gani et al., 2019). However, lower OA concentration could be explained by the  
242 measuring period of February-March, as aerosol loading starts decreasing in February after reaching its peak in  
243 December-January (Gupta and Mandariya, 2013). Campaign average fractional contribution OA to PM<sub>1</sub> was 56%,  
244 ranging from 1 to 84%. This high OA contribution in PM<sub>1</sub> is consistent with other studies conducted in IGP  
245 (Chakraborty et al., 2016a; Gani et al., 2019; Mandariya et al., 2019) and worldwide (Jimenez et al., 2009; Zhang  
246 et al., 2007). Peaked OA mass concentrations were noted between 21:00-23:00 hr (figure 2(k)), consistent with



247 previous studies conducted at the current site (Gani et al., 2019; Rai et al., 2020). Campaign average mass  
248 concentration of  $\text{NO}_3$  was  $10.1 \pm 7.0 \mu\text{g}/\text{m}^3$  and showed diurnal variation with a peak in the morning and midnight  
249 (figure 2(l)). Besides,  $\text{SO}_4$  showed slight enhancement at 08:00 hr and remained nearly constant from noon to  
250 17:00 hr (figure 2(m)). However, Cl varied between 0.13 to  $77.83 \mu\text{g}/\text{m}^3$ , and higher concentrations of Cl were  
251 found episodic throughout the campaign. The Cl- concentration was found consistent with Gani et al., 2019's  
252 previously reported value of  $0.1\text{-}66.6 \mu\text{g}/\text{m}^3$  at the current site. Figure 1 shows the temporal variation of all  
253 possible OA factors. BBOA mass concentration varied between 0.0 to  $134.7 \mu\text{g}/\text{m}^3$ . The two peaks during the  
254 night (21:00-23:00 hr) and morning (07:00-08:00 hr) featured the BBOA diurnal variation (figure 2(r)). Besides,  
255 LO-OOA diurnal variation peaked in the morning (10:00 hr) and remained nearly flattered at noontime. It could  
256 indicate its steady formation rate. However, the diurnal variation of MO-OOA showed a slight bump during  
257 noontime, indicating its formation via daytime photooxidation (Mandariya et al., 2019; Sun et al., 2016). Overall,  
258 oxygenated organic aerosol (OOA) was the dominant fraction of OA during the sampling period.

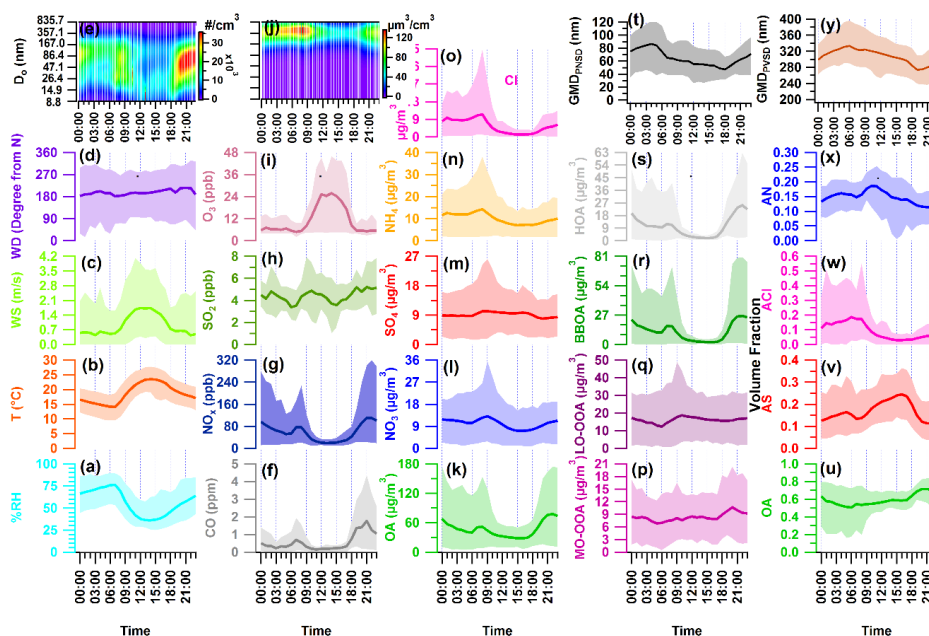
259 Furthermore, based on the variability in aerosol chemical composition, three different events were observed: first,  
260 high-residential or local biomass burning (H-BB), second, high-hydrocarbon like OA (H-HOA), and third, high-  
261 chloride (H-Cl) period, characterized by high BBOA, HOA, and Cl, respectively. In addition, we also classified  
262 a "Clean Period" where  $\text{PM}_{10}$  loading was less than 25 percentile ( $\leq 38.7 \mu\text{g}/\text{m}^3$ ) of the sampling period. H-BB  
263 events showed a dynamic variation in BBOA mass concentration from 16.3 to  $134.7 (50.7 \pm 24.0) \mu\text{g}/\text{m}^3$ .  
264 Although, a higher mass concentration of HOA ( $9.6 - 109.4 \mu\text{g}/\text{m}^3$ ) was also observed in these events. It indicates  
265 that HOA also has probably similar sources during this event. Nevertheless, a higher HOA concentration ( $4.8\text{-}$   
266  $58.9 \mu\text{g}/\text{m}^3$ ) was noted during H-HOA events, although these concentrations were significantly lower than those  
267 observed during H-BB events. Nevertheless, fractional mass contribution of HOA to OA was largest among all  
268 OA species. Besides, H-Cl events observed a higher concentration of both primary organic aerosol HOA and  
269 BBOA. BBOA contributed nearly 40.0, 21.1, 32.5, and 13.1% to OA during H-BB, H-HOA, H-Cl, and relatively  
270 clean events, respectively, indicating different BBOA sources. Besides, the HOA's average contribution of 41.6%  
271 was observed in the H-HOA event was the highest among all events. In addition, Cl's fractional mass contribution  
272 in  $\text{PM}_{10}$  reached up to 44.9% in the H-Cl event compared to 21.2% (H-BB) and 7.3% (H-HOA) events.



273

274 **Figure 1:** Temporal variability of ambient (a) relative humidity (RH), temperature (T), (b) wind speed (WS), wind  
 275 direction (WD), (c) particle number-size distribution (PNSD), 24-average geometric mean diameter (GMD), (d) particle  
 276 volume-size distribution (PVSD), (e) particulate matter (PM<sub>1</sub>), organic aerosol (OA), nitrate (NO<sub>3</sub>), sulfate (SO<sub>4</sub>),  
 277 ammonium (NH<sub>4</sub>), chloride (Cl), (f) fractional contribution of OA, NO<sub>3</sub>, SO<sub>4</sub>, NH<sub>4</sub>, and Cl in PM<sub>1</sub>, (g) more oxidized-  
 278 oxygenated OA (MO-OOA), less oxidized-oxygenated OA (LO-OOA), biomass burning OA (BBOA), hydrocarbon like-  
 279 OA (HOA), and (h) fractional contribution of MO-OOA, LO-OOA, BBOA, and HOA in OA. The pie chart sub-plot  
 280 represents the overall average contribution of species, and the bar sub-plot represents the overall campaign average  
 281 value of different species. All other species are represented with specific color coding mentioned in legends.

282



283

284 **Figure 2: Diurnal variation of ambient meteorological parameters (a) % ambient relative humidity (RH), (b)**  
 285 **temperature (T), (c) wind speed (WS), (d) wind direction (WD), and (e) particle number size distribution (PNSD), mass**  
 286 **concentration of ambient trace gases (f) carbon mono-oxide (CO), (g) nitrogen oxides (NO<sub>x</sub>), (h) sulfur dioxide (SO<sub>2</sub>),**  
 287 **and (i) ozone (O<sub>3</sub>), (j) particle volume size distribution (PVSD), mass concentration of aerosol constituents (k) organic**  
 288 **aerosol (OA), (l) nitrate (NO<sub>3</sub>), (m) sulfate (SO<sub>4</sub>), (n) ammonia (NH<sub>4</sub>), and (o) chloride (Cl), mass concentration of**  
 289 **organic aerosol species (p) more oxidized-oxygenated OA (MO-OOA), (q) less oxidized-oxygenated OA (LO-OOA), (r)**  
 290 **biomass burning OA (BBOA), and (s) hydrocarbon like-OA (HOA), (t) geometric mean diameter of particle number**  
 291 **size distribution (GMD<sub>PNSD</sub>) and volume fractional contribution of (u) organic aerosol (OA), (v) ammonium sulfate**  
 292 **(AS), (w) ammonium chloride (ACI), and (x) ammonium nitrate (AN) in PM<sub>1</sub>, and (y) geometric mean diameter of**  
 293 **particle volume size distribution (GMD<sub>PVSD</sub>). Upper and lower boundary of shaded area represents the 95<sup>th</sup> and 5<sup>th</sup>**  
 294 **percentile values of respective species.**

295 **3.2 Hygroscopicity of Nucleation, Aitken, and Accumulation Mode Particles**

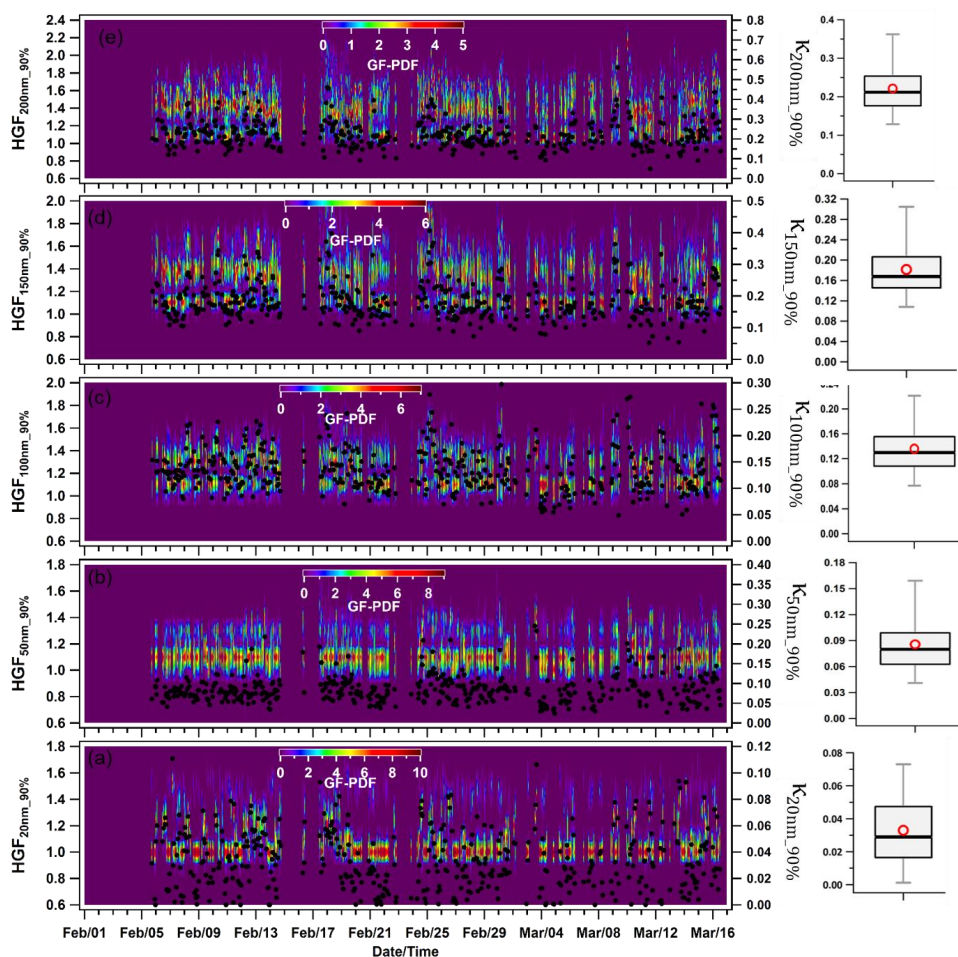
296 **3.2.1 Temporal variability**

297 Figure 3 shows the dynamic variability in the hourly averaged HGF<sub>90%</sub> and hygroscopicity parameter ( $\kappa_{H-TDMA\_90\%}$ ,  
 298 kappa) of Nucleation, Aitken, and Accumulation mode aerosol particles at 90% ambient relative humidity. The  
 299 hygroscopic growth factors of 20 (HGF<sub>90%\_20nm</sub>), 50 (HGF<sub>90%\_50nm</sub>), 100 (HGF<sub>90%\_100nm</sub>), 150 (HGF<sub>90%\_150nm</sub>), and



300 200 nm ( $HGF_{90\%,200nm}$ ) size particles varied between 1.00-1.41, 1.05-1.39, 1.11-1.49, 1.12-1.63, and 1.12-1.179  
301 with an average of  $1.14 \pm 0.09$  (average  $\pm$  standard deviation),  $1.16 \pm 0.06$ ,  $1.27 \pm 0.07$ ,  $1.35 \pm 0.09$ , and  $1.41 \pm$   
302  $0.09$ , respectively. These mean hygroscopic growth factors were noted as statistically ( $p < 0.05$ ) different from each  
303 other. In addition, the hygroscopicity ( $\kappa_{20nm,90\%}$  and  $\kappa_{50nm,90\%}$ ) of 20 and 50 nm aerosol particles varied between  
304 0.00-0.11 and 0.02-0.25, with an average of  $0.03 \pm 0.02$  and  $0.09 \pm 0.03$ , respectively. Nucleation mode particles  
305 were observed, mainly monomodal GF-PDF (figure 3(a)), comprising nearly  $74 \pm 24\%$  nearly hydrophobic  
306 particles ( $HGF < 1.2$ ). However, this contribution was raised to 100%, which was observed to have a good  
307 association with night-time local burning activities, as shown in the figure 3(a). The nucleation mode particles  
308 ( $\kappa_{20nm,90\%}$ ) showed significantly ( $p < 0.05$ ) lower hygroscopicity than Aitken mode particles ( $\kappa_{50nm,90\%}$ ). Hong et  
309 al. (2015) reported that nucleation mode particles are more sensitive to condensable vapors like fresh VOCs,  
310  $H_2SO_4$  and HCl. However, the present study did not measure these species. The  $\kappa$  of Aitken size particles were  
311 comparable with  $0.24 \pm 0.08$  of  $52.6 \pm 6.9$  size particles reported by Gunthe et al. (2011) for Beijing. Beijing is  
312 also one of the most polluted urban locations like Delhi, which could justify the comparison. However, Gunthe et  
313 al. (2011) performed this study using CCN on supersaturation levels. The campaign average hygroscopicity  
314 parameter ( $\kappa_{90\%}$ ) increased significantly ( $p < 0.05$ ) with particle size, which can be attributed to the kelvin  
315 effect (Wang et al., 2018a). In the accumulation size range (100, 150, and 200 nm),  $\kappa_{90\%}$  increased to  $\sim 0.56$ . The  
316 overall sampling average values of  $\kappa_{100nm,90\%}$ ,  $\kappa_{150nm,90\%}$ , and  $\kappa_{200nm,90\%}$  were  $0.14 \pm 0.04$ ,  $0.18 \pm 0.06$ , and  $0.22 \pm$   
317  $0.07$ , respectively. The  $\kappa_{200nm,90\%}$  varied between 0.05 and 0.56. The similar kind of variation in  $\kappa$  with particle  
318 size has been demonstrated in Kanpur, situated at the center of IGP, India (Mandariya et al., 2020a) and worldwide  
319 studies (Cerully et al., 2015; Enroth et al., 2018; Fan et al., 2020; Kawana et al., 2016; Kim et al., 2020; Kitamori  
320 et al., 2009; Ogawa et al., 2016; Sjogren et al., 2012; Wang et al., 2018a). Moreover, this was attributed to the  
321 predominant increment in inorganic to OA fraction in particles with increment in size. Furthermore,  $\kappa_{H-TDMA,90\%}$   
322 was found approximately in the 0.13-0.77, reported by Arub et al. (2020) at Delhi for  $PM_{10}$  without considering  
323 BC. Although, Arub et al. (2020) theoretically predicted particles' hygroscopicity by considering a particle's  
324 chemical composition. They found a decrease in  $\kappa$  calculation by 10% when BC was considered in aerosol  
325 chemical composition. Also,  $\kappa_{H-TDMA,90\%}$  measured in the current study were found in line with the global average  
326 value of  $0.27 \pm 0.21$  for continental aerosols (Petters and Kreidenweis, 2007; Pringle et al., 2010). Further, to  
327 understand the impact of a particle's chemical composition, local meteorology, and air mass trajectories on  $\kappa_{H-}$   
328  $TDMA,90\%$  for accumulation mode particle discussed in subsequent sections.

329



330

331 Figure 3: Temporal variability in hygroscopic parameter kappa ( $\kappa$ ) of nucleation mode particles (a) 20 nm ( $\kappa_{20nm\_90\%}$ ),  
 332 Aitken mode particles (b) 50 nm ( $\kappa_{50nm\_90\%}$ ), and Accumulation mode particles (c) 100 nm ( $\kappa_{100nm\_90\%}$ ), (d) ( $\kappa_{150nm\_90\%}$ ),  
 333 and (e) 200 nm ( $\kappa_{200nm\_90\%}$ ). The box plots represent the variability in the hygroscopicity of respective sizes of particles  
 334 in which low and high whisker traces represent the 5 and 95 percentile, respectively. The red marker indicates the  
 335 average of the data, whereas the upper and lower sides of the boxes indicate the 75 and 25 percentile of the data,  
 336 respectively.

### 337 3.2.2 Diurnal variability

338 The diurnal variability in  $\kappa_{H-TDMA\_90\%}$  was found different for nucleation ( $\kappa_{20nm\_90\%}$ ), Aitken ( $\kappa_{50nm\_90\%}$ ), and  
 339 Accumulation ( $\kappa_{100nm\_90\%}$ ,  $\kappa_{150nm\_90\%}$ , and  $\kappa_{200nm\_90\%}$ ) mode particles. Figure 4 displayed a diel variation of an  
 340 average of hourly-resolved  $\kappa$  for each size. The bigger size particles exhibited higher values of  $\kappa$  than smaller size

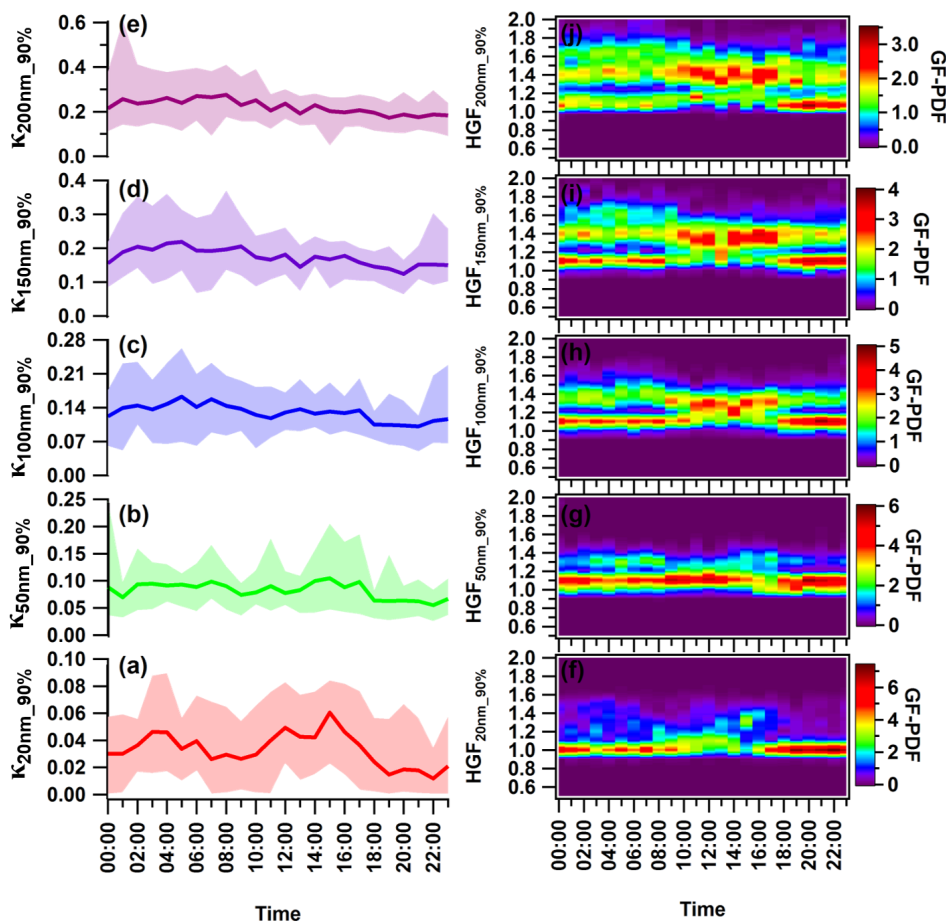


341 particles, which is a similar trend reported at Kanpur, India (Mandariya et al., 2020a) and other worldwide  
342 locations (Fan et al., 2020; Hong et al., 2015). In general, it was observed that all size particles exhibited late-  
343 night hump (02:00-05:00 hr) in  $\kappa_{H-TDMA\_90\%}$ . Besides, only  $\kappa_{20nm\_90\%}$  demonstrated a clear diurnal variability with  
344 two peaks, one late night (02:00-04:00 hr) and the other in noontime (14:00-16:00 hr), and two valleys during the  
345 morning (07:00-10:00 hr) and night (19:00-22:00 hr). These valleys reflect the strong impacts of local burning  
346 and traffic activities (Pringle et al., 2010). In addition, nucleation size particles were potentially contributed by  
347 nearly hydrophobic particles ( $HGF < 1.2$ ) from evening to midnight. They showed mono-modal GF-PDF around  
348 unit hygroscopic growth factor, possibly indicating local emission generated particles. The 20 nm particles are  
349 small enough and lie on the boundary of nucleation mode particles. Achtert et al. (2009) reported a similar diurnal  
350 trend of Nucleation and Aitken mode particles, attributed the lower values to the emission of hydrophobic aerosol  
351 particles during the local burning emissions. Daytime hump is attributed to the intense photochemical oxidation  
352 process, which causes the enhancement of more oxidized species on the aerosol particle. Furthermore, their  
353 chemical composition is dominantly controlled by the gaseous condensation of  $H_2SO_4$ ,  $HNO_3$ , and VOCs (Hong  
354 et al., 2015). The aerosol's chemical composition can address this variability of  $\kappa_{H-TDMA\_90\%}$ . However,  $\kappa_{50nm\_90\%}$   
355 also follows a similar diurnal variability as  $\kappa_{20nm\_90\%}$ , although it showed less variability. Further, as the dry size  
356 of the aerosol particles increased to accumulated mode, diurnal variation shifted toward nearly steady for the rest  
357 of the day. Hong et al. (2018) also observed no obvious diurnal pattern for 100 and 150 nm particles of organic-  
358 dominated aerosols over the Pearl River Delta region in China.

359 Furthermore, the diurnal cycles of aerosol physicochemical properties also reflect the dynamic diurnal variation  
360 in the planetary boundary layer (PBL) that leads to the accumulation of particles during night-time. Although this  
361 study did not quantify size-resolved chemical composition, so, this study used bulk-aerosol composition to address  
362 the trend variability only. However, daily average aerosol PNSD varied between 18.0-140.0 nm with a mean of  
363  $73.1 \pm 33.8$  nm. And, the mode of PVSD changed approximately around 300-600 nm. Therefore, it could be an  
364 excellent approximation to discuss  $\kappa_{200nm\_90\%}$  variability with aerosol's bulk chemical properties. The midnight to  
365 early morning hump in hygroscopicity of accumulation mode particles can be attributed to the high rise in the  
366 ratio of inorganic volume fraction to OA volume fraction (Fan et al., 2020), as illustrated in figure 2 (r, s, t, u, and  
367 v). Moreover, during mid-night and early morning in the winter, water-soluble organic and inorganic gases are  
368 partitioned and/or coagulated/condensed on the surface of the pre-existing particles. Further, in the presence of  
369 high RH and lower temperature, primary and secondary less oxidized organic aerosol participated in the aging  
370 process, which leads to enhancement their oxidation via aqueous/heterogeneous reaction, according to it increase



371 the particle's hygroscopicity (Jimenez et al., 2009; Wu et al., 2016). Similar results were observed by Fan et al.  
 372 (2020) during winter in urban Beijing, and they attributed it with the enhancement of more hygroscopic particles  
 373 due to the aqueous-oxidation and/or condensation process on the pre-existing particles. In general, higher  
 374 noontime solar radiation favors more intense photooxidation processes. It supports the partitioning of relatively  
 375 more oxidized and less volatile organics on the particulate surface, enhancing the hygroscopicity of accumulation  
 376 mode particles (Duplissy et al., 2011; Massoli et al., 2010; Tritscher et al., 2011). However, interestingly, we  
 377 observed a noontime flatten pattern of  $\kappa_{H-TDMA\_90\%}$ , and it could be attributed to the mix of the positive and negative  
 378 impact of an enhancement in the volume fraction of OA and more hygroscopic ammonium sulfate and decrement  
 379 in  $ACl$ , and  $AN$ 's volume fraction. Lower volume fractional contribution of highly volatile  $ACl$  could be the  
 380 potential factor that modulates accumulation mode particle's hygroscopicity. This can be supported by the strong  
 381 correlation of  $\kappa_{H-TDMA\_90\%}$  and volume fraction of  $ACl$  in that size particles ( $\epsilon_{ACl}$ ).







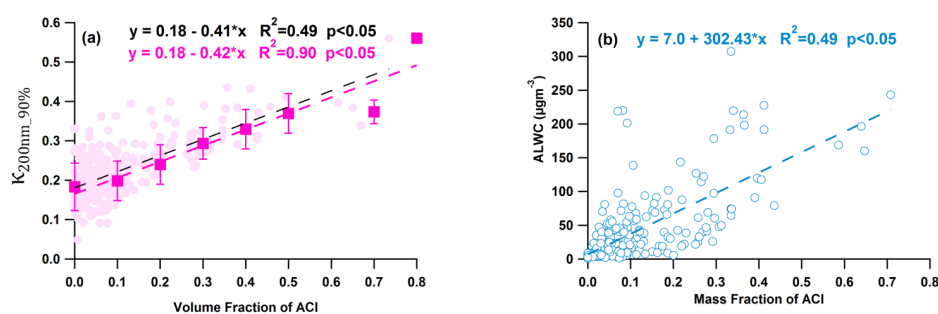
383 **Figure 4: Diurnal variance in the hygroscopic parameter kappa ( $\kappa$ ) of nucleation mode particles (a) 20 nm ( $\kappa_{20nm\_90\%}$ ),**  
384 **Aitken mode particles (b) 50 nm ( $\kappa_{50nm\_90\%}$ ), and Accumulation mode particles (c) 100 nm ( $\kappa_{100nm\_90\%}$ ), (d) 150 nm**  
385 **( $\kappa_{150nm\_90\%}$ ), and (e) 200 nm ( $\kappa_{200nm\_90\%}$ ) and hygroscopic growth factor of (f) 20 nm (HGF<sub>20nm\_90%</sub>), (g) 50 nm**  
386 **(HGF<sub>50nm\_90%</sub>), (h) 100 nm (HGF<sub>100nm\_90%</sub>), (i) 150 nm (HGF<sub>150nm\_90%</sub>), and 200 nm ( $\kappa_{200nm\_90\%}$ ) aerosol particles. The**  
387 **solid line represents diurnal average values, and the upper and lower shaded area represents 95 and 5 percentile values**  
388 **of corresponding average values. Different color coding has been used to represent various size-specific kappa values.**  
389 **The color scale represents the growth factor probability density function of hygroscopic growth factor.**

### 390 3.2.3 Driving Factor of Hygroscopicity

391 A correlation analysis was carried out between measured chemical species and aerosol to explore the factors  
392 governing aerosol hygroscopicity, as shown in the figure 5. Organic aerosol was observed negatively impact  $\kappa$ ,  
393 explained by a negative correlation (figure S7(a)). This negative correlation of OA with  $\kappa$  is also observed in India  
394 (Bhattu et al., 2016; Mandariya et al., 2020b) and worldwide (Enroth et al., 2018; Hong et al., 2014; Kawana et  
395 al., 2016; Kitamori et al., 2009; Wang et al., 2018a; Wu et al., 2013a). This result indicates that primary  
396 constituents dominated the OA during high loading, considered nearly hydrophobic or less hygroscopic. In  
397 addition, the current study observed that an enhancement of 10% of OA by volume in 200 nm aerosol particles  
398 would be responsible for a 4% decrement in its hygroscopicity (figure S7(a)). Interestingly, ammonium sulfate  
399 and nitrate showed a positive but poor correlation with hygroscopicity. It could be due to sulfate and nitrate aerosol  
400 dominating the bigger particles (>200nm). However, a 10% enhancement of AS by volume was found to be  
401 responsible for the enhancement of hygroscopicity only by 1.6%. Besides, figure 5(a) shown an increasing volume  
402 fraction of ACI in PM<sub>1</sub> with an increase in aerosol hygroscopicity, and this strong positive correlation is  
403 responsible for an enhancement in kappa by 4.2% over the increment of 10% ACI by volume, which was the  
404 highest among all chemical species. Further, ammonium chloride has a more significant water uptake potential  
405 (Chen et al., 2022; Zhao et al., 2020), which can be justified by the solid correlation of ALWC with a mass fraction  
406 of ACI in PM<sub>1</sub> as shown in figure 5(b). This indicates that particles with a more considerable ammonium chloride  
407 fraction uptake more water vapor, leading to higher hygroscopic aerosol particles. It is clear that the increases in  
408 ammonium chloride fraction enhanced aerosol liquid water content and led to higher hygroscopicity of aerosol  
409 particles. A Recent study in Delhi by Chen et al. (2022) unveils that ammonium chloride fraction in PM<sub>1</sub> aerosol  
410 enormously enhances during the higher relative humidity conditions during the winter season due to the co-  
411 condensation of semivolatile ammonium chloride with water vapor on the particles and leads to enhance water  
412 uptake and lead severe winter haze in Delhi. The very high volume fractions (>30%) of ACI in atmospheric NR-



413 PM<sub>1</sub> were observed episodic, suggesting a high fraction of Cl in the particle phase is strongly dependent on excess  
 414 ammonia in the atmosphere. These results indicate that ammonia is the controlling factor for chloride partitioning  
 415 in the particle phase, resulting in high aerosol water content under high RH and lower temperature conditions. As  
 416 the ACL is strongly dependent on the RH and temperature.



417

418 **Figure 5: Correlation plot for (a)  $\kappa_{200nm,90\%}$  vs volume fraction of ammonium chloride aerosol ( $VF_{ACI}$ ) and (b) aerosol**  
 419 **liquid water content (ALWC) vs mass fraction of ammonium chloride ( $MF_{ACI}$ ).**

420 **3.2.4 Hygroscopicity during high biomass burning (H-BB), high-hydrocarbon like OA (H-HOA), high-Cl**  
 421 **(H-Cl), and Clean Periods**

422 Delhi's atmosphere is a complex array of chloride and organic aerosol sources like combustion (crop residue,  
 423 agriculture waste, medical waste, municipal waste, plastic, etc., burning) and industrial sources. Therefore, all  
 424 episodic events were classified into three to investigate the impact of chloride and OA on aerosol hygroscopicity.  
 425 First, high biomass burning (H-BB) event; second, high-hydrocarbon like OA (H-HOA) event; and third, high-Cl  
 426 (H-Cl) event according to the substantial fractional contribution of their markers in respective periods. In addition,  
 427 we also classified a "Clean Period" where PM<sub>1</sub> loading was less than 25 percentile ( $\leq 38.7 \mu\text{gm}^{-3}$ ) of the sampling  
 428 period. Further, aerosol chemical composition data were filtered according to hygroscopic parameter data for  
 429 further analysis. By performing so, data information that is characteristic of the local emission and atmospheric  
 430 chemistry in question and the effects of various potential transported air mass types can be retrieved. It is valuable  
 431 to extract any possible information about aerosol sources and transformation process evaluation to interpret its  
 432 influence on the aerosol's hygroscopicity.

433 **3.2.4.1 Relatively Clean Period**



434 The relatively Clean period was predominantly dominated by E, S-E winds; however, pollution was associated  
435 with calm winds, as illustrated in figure S9. All BBOA, HOA, and ACI were observed to be associated with  
436 similar sources and found an excellent association with ambient relative humidity. The mean concentration of  
437 organic aerosol, ACI, AN, and AS was observed at  $11.0 \pm 6.4$ ,  $1.4 \pm 1.1$ ,  $3.0 \pm 1.5$ , and  $4.4 \pm 2.2 \mu\text{gm}^{-3}$ ,  
438 respectively. These mass concentrations were significantly lower than in other specified periods. However, OA  
439 was still the dominant species, with 56% by volume in the  $\text{PM}_{10}$ , as indicated in figure 8. Among all OA factors,  
440 HOA was predominantly dominated in OA with 33%, although secondary organic aerosol confined the overall  
441 54.4% of OA. Secondary OA is defined with relatively higher oxidized OA, and the oxidation state of OA  
442 positively impacts OA hygroscopicity (Kim et al., 2017; Richard et al., 2011; Wu et al., 2013a). The Clean period's  
443 mean hygroscopicity of 20, 50, 100, 150, and 200 nm particles were observed at  $0.03 \pm 0.02$ ,  $0.09 \pm 0.04$ ,  $0.14 \pm$   
444  $0.06$ ,  $0.22 \pm 0.09$ , and  $0.27 \pm 0.07$ , respectively, significantly ( $p < 0.05$ ) different to each other. However, the  
445 accumulation particle's (200 nm) hygroscopicity was not significantly ( $p > 0.05$ ) higher than the 150 nm particles.  
446 The hygroscopicity increment with size from 20 to 200 nm can also be explained by the fractional increment of  
447 more hygroscopic ( $\text{GF} > 1.2$ ) particles relative to nearly hydrophobic or less hygroscopic particles ( $\text{GF} < 1.2$ ).  
448 Nucleation particles, 20 nm was dominated mostly by less hygroscopic particles ( $76.8 \pm 21.7\%$ ), indicates  
449 influence by fresh emission sources, whereas, Aitken (50 nm) and Accumulation (200 nm) size aerosol were  
450 confined with  $69.3 \pm 14.7$  and  $25.4 \pm 10.8$  less hygroscopic particles, respectively. These results point out that  
451 accumulation-size aerosols dominated secondary aerosols, which can also supports their GF-PDF as shown in  
452 figure 7(a). Nucleation size aerosol particles (20nm) showed nearly mono modal GF-PDF with the mode of unit  
453 growth factor. In contrast, the mode shifted towards the higher end as aerosol size increased and GF-PDF shifted  
454 from unit to multi-mode.

#### 455 3.2.4.2 High-Cl (H-Cl) events

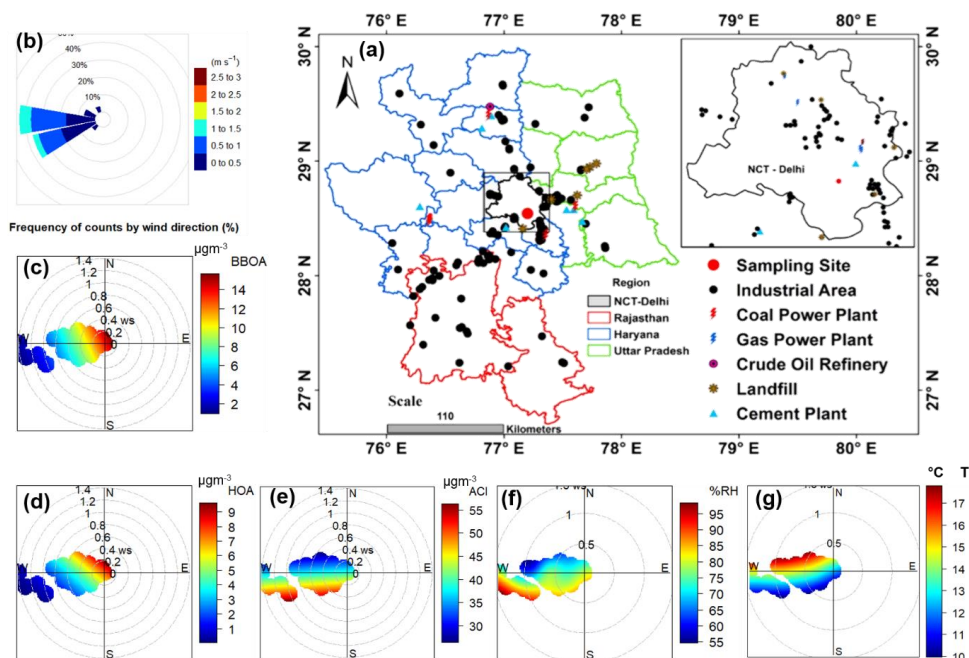
456 H-Cl events representing the substantial loading of ACI on the receptor site, as shown in the pie chart in figure 8,  
457 were chosen mainly due to the significant jump ( $>20\%$ ) in fractional volume contribution  $\text{NH}_4\text{Cl}$  ( $\epsilon_{\text{ACI}}$ ) in the  $\text{PM}_{10}$   
458 aerosol. This period observed apparent surface wind from W-direction, although WNW, WSW, and SE winds  
459 also influence the site, as shown in figure 6(b). BBOA and HOA are potentially contributed from the WNW and  
460 SE directions, as explained in the bipolar plot figure 6(c & d), and seem to come from a similar local source.  
461 Among inorganic species, ACI observed excellent association with ambient RH, as shown in figure 6(e) and (f),  
462 indicating the atmospheric gaseous HCl neutralized with  $\text{NH}_3$  gas in the presence of atmospheric water content.  
463 HCl sources could be coal power plants, trash burnings in solid waste dumping sites, and other industries located



464 in the W-WSW direction (Gani et al., 2019), as shown in the map in figure 6(a). Atmospheric high CI events are  
465 potentially dominated by trash burning in Delhi during winter (Shukla et al., 2021; Tobler et al., 2020). Moreover,  
466 bipolar plots (figure 6(e and f)) suggest that ACI formation under high RH conditions associated with a relatively  
467 calm atmosphere trigger particles' hygroscopicity. This hypothesis can be supported with a good association of  
468 aerosol liquid water content (ALWC) as discussed in previous section. Furthermore, GF-PDF of all size particle  
469 marked relatively more fractionally contribution of secondary mode particles as showed in figure 7(d). Overall  
470 more hygroscopic ( $HGF_{90\%} > 1.2$ ) particles were marked by 42, 47, 50, 74, and 83% contributions in the 20, 50,  
471 100, 150, and 200 nm size particles, respectively. Hence, ACI is a critical factor to enhance aerosol hygroscopicity  
472 to trigger fog/haze formation under higher RH and colder atmospheric conditions as discussed in the previous  
473 section.

474 Similarly, Gunthe et al. (2021) observed that high local emission of hydrochloric acid in Delhi during February-  
475 March gets partitions into aerosol liquid water under high humid conditions, enhancing the water uptake capacity  
476 of aerosol sustain particle's hygroscopic growth, result in fog/haze formation. Moreover, worldwide studies on  
477 size-resolved hygroscopicity observed CI less than 1%, so they omitted ACI as an aerosol constituent into the  
478 discussion. In addition, the current study did not find any strong correlation of  $\kappa$  with AS and AN. It could be due  
479 to their association with larger particle sizes. Besides, ACI could be associated with comparatively lower size  
480 particles ( $\leq 200$  nm). Furthermore, in context to look influence of air mass trajectories, we further mapped  
481 aerosol's constituents in the association of air mass back trajectories in PSCF to see the potential area source  
482 contribution influencing the aerosol evaluation processes, ultimately aerosol's hygroscopicity. However, we did  
483 not find any back trajectory influencing the receptor site, as all trajectory endpoints were observed above the  
484 planetary boundary layer height.

485



486

487 **Figure 6:** Map of (a) Delhi showing various types of industries located in the region and nearby locations, (b) wind rose  
 488 diagram and conditional bi-polar plots showing variation in mass concentration of (c) biomass burning OA (BBOA),  
 489 (d) hydrocarbon like OA (HOA), (e) ammonium chloride (ACl), (f) % ambient relative humidity (RH), and (g) ambient  
 490 temperature (T), with wind direction (WD) and wind speed (WS) during H-Cl events. A background map showing  
 491 various industrial locations was adapted from Rai et al. (2020).

492 **3.2.4.3 High biomass burning (H-BB) Events**

493 High BB periods were noted during the initial period (1-12 February) of the field campaign. The predominant  
 494 surface wind circulations were from W, W-WNW, and W-WSW directions (figure S8(b)). The aerosol was  
 495 dominated by local emissions, as aerosol constituents are mainly associated with slower wind circulations from  
 496 landfill sites, industrial areas, and coal power plants, as shown in figure S8(a). Further, it could justify the potential  
 497 source contribution function (PSCF) analysis considering 48-hr air mass back trajectories, as shown in figure S10.  
 498 Therefore, BBOA possibly contributed from the open local biomass burning activities at landfill sites or others.  
 499 Biomass burning organic aerosol confined the most considerable fraction, 39%, of organic aerosol, following  
 500 HOA, 28%. Figure S5(b, c, and d) clearly shows that BBOA and HOA have similar source profiles but differ from  
 501 the ACl source. Moreover, ACl was not found to have a good association with ambient RH and but was associated  
 502 with nearby coal power plant's emissions. However, 48 h air mass back trajectories indicated that the current city



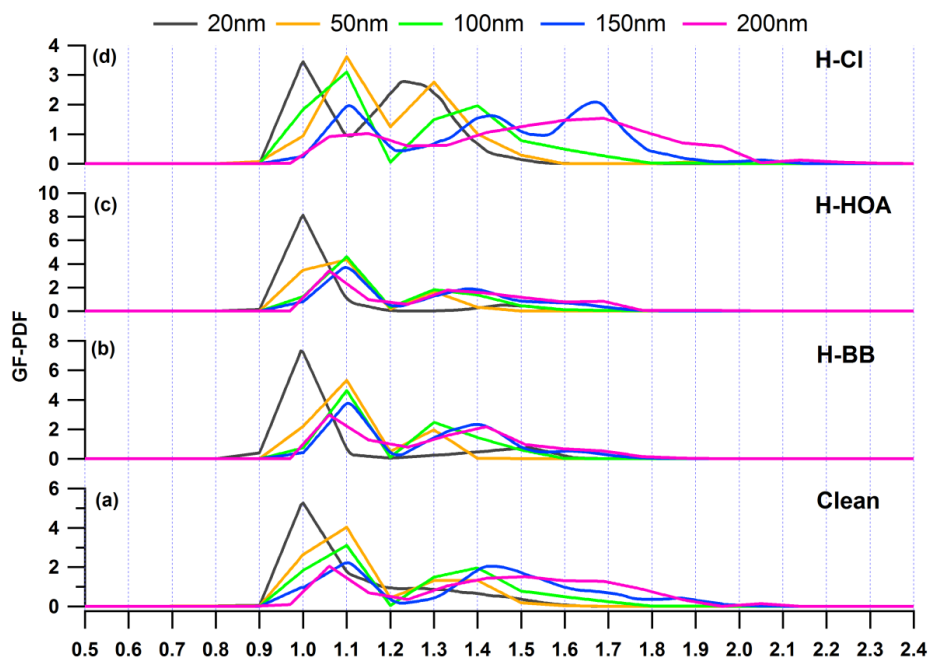
503 was also influenced by air mass from some parts of Uttar Pradesh, Punjab, and Haryana. These states are the  
504 potential hub of crop residue burning, industrial activities and brick kilns. These cities have a substantial fraction  
505 of OA in PM<sub>1</sub> and OA mainly affected by biomass activities during winter. The H-BB event captured a  
506 considerable volume fraction, 71% of OA in PM<sub>1</sub> and BBOA contributed almost 39%, as illustrated in the figure  
507 8. So, lower inorganic to OA ratio was a potential factor in decreasing the aerosol hygroscopicity in H-BB events.  
508 Further, a primary organic aerosol contribution was enhanced during this event and, on average, raised to 67%.  
509 OA loading inversely affects the aerosol's hygroscopicity. Mandariya et al. (2020) reported a similar observation  
510 in Kanpur, and the authors suggested that the contribution of primary biomass burning (BBOA) and hydrocarbon-  
511 like OA adversely affects aerosol hygroscopicity. BBOA showed a good negative correlation with the  
512 hygroscopicity of 200 nm particles, supporting the following conclusion. Apart from this, the Nucleation size  
513 particle (20 nm) showed  $0.02 \pm 0.02$  hygroscopicity parameter with mono mode GF-PDF with the unit mode  
514 (figure 7(b)) and confined  $83.7 \pm 18.6$  % nearly hydrophobic particles. Furthermore, as aerosol size increased,  
515 hygroscopicity parameter ( $\kappa_{H-TDMA_{90\%}}$ ) enhanced significantly ( $p < 0.05$ ) as the contribution of relatively  
516 secondary aerosol particles ( $GF > 1.2$ ) increased with aerosol size. Accumulation size aerosol, 100 nm contributed  
517 approximately 54% by nearly hydrophobic ( $GF < 1.2$ ) and 46% by more hygroscopic ( $GF > 1.2$ ) particles.

#### 518 3.2.4.4 High-HOA (H-HOA) Events

519 H-HOA events were identified based on the considerable mass concentration and fraction of HOA in the organic  
520 aerosol. These periods were noted generally 19:00 hr to Morning 09:00 hr during 22-23 and 26-27 February and  
521 4, and 7-8 March as indicated in figure 1. PSCF explore the probability of impacts of long-range transported  
522 aerosol. Interestingly, it was observed that air masses over Delhi, Haryana, and Uttar Pradesh were potentially  
523 associated with hydrocarbon-like OA (figure S10). BBOA also followed a similar path as HOA. However, the  
524 potential area source of ACI was the nearby region of Delhi and Haryana. The HOA loading was significantly  
525 ( $p < 0.05$ ) higher than in H-BB, H-Cl and Clean periods. However, emission sources were different during both H-  
526 HOA and H-BB periods. As HOA was the potential contributor to OA, it is likely the critical constituent to  
527 modulate aerosol hygroscopicity in the region during these events. HOA is mainly considered hydrophobic  
528 (Duplissy et al., 2011). Therefore, elevated HOA contribution (41%) in OA could be responsible for lower  $\kappa$  in  
529 these events. The overall hygroscopicity of 20, 50, 100, 150, and 200 nm size particles was recorded as  $0.01 \pm$   
530  $0.01$ ,  $0.06 \pm 0.03$ ,  $0.11 \pm 0.03$ ,  $0.14 \pm 0.04$ , and  $0.17 \pm 0.05$ , respectively. The predominant fractional contribution  
531 of primary aerosol particles ( $GF < 1.2$ ) seems to be a reason for this lower hygroscopicity of particles, as shown in  
532 figure 7(c). Overall, OA predominantly constitutes the fraction in the PM<sub>1</sub>, and primary OA contributed



533 approximately 60% in OA. However, relative increment in the contribution of other more hygroscopic constituents  
 534 like secondary organic aerosol (LO-OOA and MO-OOA), ACI and ammonium sulfate (AS) in the aerosol possible  
 535 tried to balance the negative impact high-HOA on which limited  $\kappa$ .



536

537 **Figure 7: Growth Factor Probability Density Function (GF-PDF) of 20, 50, 100, 150, and 200 nm aerosol particles for**  
 538 **the (a) clean, (b) H-BB, (c) H-HOA, and H-Cl periods.**

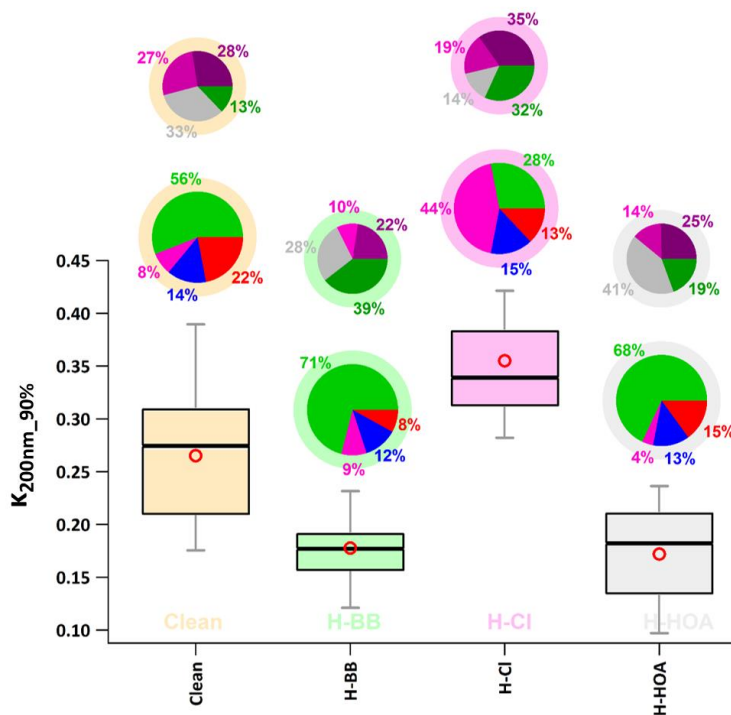
#### 539 3.2.4.4 Comparison of $\kappa$ of different events

540 We choose a 200 nm accumulation particle size particle representing the bulk aerosol chemical composition to  
 541 compare the aerosol hygroscopicity among various periods. Further, in the present study, the mode of particle-  
 542 volume size distribution varied from 400 nm to 600 nm particle dry mobility diameter. Therefore, 200 nm size  
 543 accumulation particles are the best choice to compare hygroscopicity parameters among different periods  
 544 considering bulk aerosol composition in various mentioned periods. In addition, a good Pearson's  $r$  value, 0.76,  
 545 was found among  $\kappa_{200nm\_90\%}$  and  $\kappa_{chem\_90\%}$ , derived from the dry  $PM_{10}$  particle's chemical composition measured  
 546 from the ACSM based on the ZSR mixing rule (Stokes and Robinson, 1966), which justifies our choice.



547 The H-Cl event noted the highest value ( $0.36 \pm 0.06$ ) of  $\kappa_{200\text{nm},90\%}$  against H-BB ( $0.18 \pm 0.04$ ), H-HOA ( $0.17 \pm$   
548  $0.05$ ), and Clean ( $0.27 \pm 0.07$ ) events, as illustrated in figure 8. The H-Cl event observed that the average  $\kappa_{200\text{nm},90\%}$   
549 value was significantly ( $p < 0.05$ ) higher than those observed in other events. It means that a substantial increment  
550 in Cl emission in the Delhi region could significantly enhance the aerosol liquid water content leading to higher  
551 aerosol hygroscopicity, which can further strengthen cloud condensation nuclei formation, possibly triggering  
552 haze/fog events in Delhi NCR (Gunthe et al., 2021). These results suggest that controlling the open trash/waste  
553 burning in the region could help control Cl emission, which leads to minimizing the haze/fog formation possibility  
554 during high atmospheric conditions. However, the difference in  $\kappa_{200\text{nm},90\%}$  values between H-BB and H-HOA  
555 events was not observed significantly ( $p > 0.05$ ), possibly due to the relative changes in primary, secondary OA,  
556 and inorganic species. In the H-HOA events, the negative effect of a significantly higher fractional (41%)  
557 contribution of HOA to OA possibly balances with a positive impact of a 7% increment in secondary OA relative  
558 to H-BB. Worldwide studies (Jimenez et al., 2009; Mandariya et al., 2019; Sun et al., 2013) reported secondary  
559 organic aerosol associated with a higher O/C ratio, and several studies reported that the O/C ratio positively  
560 correlated to  $\kappa$  (Jimenez et al., 2009; Kim et al., 2020) as described in the earlier text. Furthermore, impacts of  
561 5% decrement in ACl during H-HOA event concerning H-BB event possibly managed by 7% increment in AS  
562 fraction. Overall, these relative changes in aerosol constituents worked to insignificant changes in  $\kappa$  during H-BB  
563 and H-HOA periods. Nevertheless, H-BB and H-HOA events witnessed significant ( $p < 0.05$ ) lower hygroscopicity  
564 compared to a relatively cleaner atmosphere. The aerosol associated with relatively cleaner events was with a  
565 higher inorganic-to-organic ratio. In addition, the aerosol in clean periods comprised a significantly higher fraction  
566 of secondary organic aerosol, which could be the reason for the higher hygroscopicity associated with organic  
567 aerosol compared to other events. Worldwide (Aiken et al., 2008; Cerully et al., 2015b; Chakraborty et al., 2016b;  
568 Mandariya et al., 2019) studies have reported that organic aerosol loading inversely impacts the oxidation/aging  
569 process of OA. Overall, all these were responsible for higher hygroscopicity in relatively cleaner periods.





570

571 **Figure 8:** Box plot showing variation in H-TDMA measured hygroscopic parameter of 200 nm size particles  $\kappa_{H-TDMA}$   
 572 ( $\kappa_{200nm,90\%}$ ) in high biomass burning (H-BB), high-chloride (H-Cl), and high-hydrocarbon like organic aerosol (H-  
 573 HOA) events. Different colors represent respective events in the plot. A bigger pie chart represents the overall average  
 574 volume fractional contribution of various aerosol species indicated by color-coding. In addition, minor pie charts  
 575 described the event average mass fractional contribution of different OA species in OA. Diffused ring color of the pie  
 576 chart displays the respective event.

577 **4. Conclusions**

578 The current study explored the temporal variation of Nucleation (20 nm), Aitken (50 nm), and Accumulation (150  
 579 and 200 nm) mode particle hygroscopicity in Delhi during the winter period, February-March 2020. In addition,  
 580 the present study highlighted the hygroscopicity variation in relatively higher chloride, biomass burning, and  
 581 hydrocarbon-like organic aerosol. Because, Delhi highlighted as one of the most polluted cities, faces high  
 582 chloride pollution episodes during winter haze/fog. Therefore, we reported temporal variation in the size-resolved  
 583 hygroscopic parameter ( $\kappa_{H-TDMA,90\%}$ ) at the sub-saturated level (90% RH) for the first time in Delhi. However, we  
 584 reported hygroscopicity of nucleation and Aitken mode particles for the first time in India.



585 The observed  $\kappa_{H-TDMA\_90\%}$  ranged from .00 to 0.11 ( $0.03 \pm 0.02$ ), 0.05 to 0.22 ( $0.11 \pm 0.03$ ), 0.05 to 0.30 ( $0.14 \pm$   
586  $0.04$ ), 0.05 to 0.41 ( $0.18 \pm 0.06$ ), and 0.05 to 0.56 ( $0.22 \pm 0.07$ ) for 20, 50, 100, 150, and 200 nm aerosol particles,  
587 respectively. The study period's mean value of the hygroscopicity parameter was significantly ( $p < 0.05$ ) enhanced  
588 with the size of the particles.  $\kappa_{20nm\_90\%}$  and  $\kappa_{50nm\_90\%}$  were observed to show dynamic diurnal variation. In contrast,  
589 as particle size increased in accumulation mode particles, noontime variation became flatten, which found an  
590 attribute with neutralizing positive and negative impacts of increment and decrement in volume fraction of  $\text{NH}_4\text{Cl}$   
591 and OA, respectively. Interestingly, the variation in  $\kappa_{200nm\_90\%}$  was observed potentially due to variation in  $\text{NH}_4\text{Cl}$   
592 and OA instead  $(\text{NH}_4)_2\text{SO}_4$ .

593 Furthermore, the pollution episodes were generally associated with local biomass burning and industrial and  
594 waste-burning emissions in Delhi and nearby regions. We mainly focused on emphasizing the impacts of high  
595 biomass burning (H-BB), high hydrocarbons like OA (H-HOA), and high chloride emission (H-Cl) on aerosol  
596 hygroscopicity and their relative comparison with the cleaner period. The H-Cl period was observed significantly  
597 ( $p < 0.05$ ) higher hygroscopicity ( $0.35 \pm 0.06$ ) compared to H-BB ( $0.18 \pm 0.04$ ), H-HOA ( $0.17 \pm 0.05$ ), and  
598 relatively cleaner period ( $0.27 \pm 0.07$ ). However, H-BB and H-HOA showed no significant difference in  
599 hygroscopicity. However, they witnessed lower aerosol hygroscopicity concerning a relatively cleaner  
600 atmosphere. It could be attributed to the lower organic aerosol loading and higher inorganic-to-organic aerosol  
601 ratio in the aerosol. High atmospheric chloride aerosol (ammonium chloride) was observed most affectionately to  
602 the atmospheric water, leading to higher aerosol liquid water content at high chloride events. In addition, a high  
603 volume fraction of ammonium chloride in aerosol enhanced the aerosol hygroscopic nature. Furthermore, our  
604 results based on episodic interpretation revealed that chloride emission is a significant concern in Delhi, which  
605 enhances the aerosol's hygroscopicity, helps to act as CCN to form fog droplets during winter colder days, and  
606 leads to fog/haze formation in Delhi. In addition, high chloride present in aerosol overcomes the negative impact  
607 of high OA loading on CCN activity. Hence, our results suggest controlling waste materials' open burning to  
608 decrease the haze/fog events in Delhi during winter.

#### 609 **Supporting Information**

610 Supplementary pieces of information are mentioned in the supplementary file.

611 **Data availability.** Data can be accessed at the following repository:

612 <https://web.iitd.ac.in/~gazala/publications.html> (Mandariya et al., 2023).



613 **Author contributions**

614 AH, MMVH, NAB, and GH operated aerosol instrumentation and collection of data on-board in Delhi. KP  
615 analysed the ACSM data. AKM, AH, and GH conceptualized the structure of the manuscript. AKM analysed,  
616 evaluated H-TDMA data, and wrote the manuscript. AH analysed MPSS data. AKM, AH, KP, JSA, LHR, AW,  
617 and GH internally reviewed the manuscript and helped to write the manuscript.

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620 **Competing interests**

621 The authors declare that they have no conflict of interest.

622

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