

1 Soil-biodegradable plastic films do not decompose in a lake sediment over 9 months  
2 of incubation

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18 **Abstract**

19 Agriculture relies heavily on the use of plastic mulch films, which increase crop yields  
20 and can lower water demands. In recent years, soil-biodegradable mulch films have  
21 been marketed to replace the non-biodegradable, conventional polyethylene-based  
22 mulch films. These biodegradable mulch films are designed to be ploughed into the  
23 soil after use to biodegrade *in situ* by soil microorganisms. However, research has  
24 shown that part of the mulch film material may be transported from the fields to  
25 neighboring environments, including aquatic ecosystems. Research on potential  
26 biodegradation of soil-biodegradable plastics in freshwater habitats is lacking. Here,  
27 we investigated the mineralization of soil-biodegradable agricultural mulch films in  
28 freshwater lake sediments of Lake Lucerne, Switzerland. Two types of commercial  
29 soil-biodegradable mulch films were incubated within lake sediment cores, along with  
30 traditional PE plastic, and the production of CO<sub>2</sub> and CH<sub>4</sub> was followed over time  
31 relative to non-plastic containing control sediments. After the 40-week incubation  
32 period, films were visually intact and showed no signs of mineralization. Gas  
33 analyses showed no additional production of either CO<sub>2</sub> or CH<sub>4</sub> in the degradable  
34 mulch film incubations, compared to control or PE plastic incubations. We conclude  
35 that these two used soil biodegradable mulch films have a low biodegradability in  
36 lake sediments, likely reflecting that the microbial community structure in the lake  
37 sediment lacks active microbial degraders. Our results highlight the importance of  
38 preventing transport of soil-biodegradable mulch films from agricultural soils to  
39 surrounding aquatic environments.

## 42 Introduction

43

44 Part of the solution to overcoming plastic pollution in the environment is to use  
45 biodegradable plastic instead of conventional plastics in applications in which  
46 plastics are used in the open environment and for which complete recollection of the  
47 plastic after use is impossible. A picture-case example are thin agricultural mulch  
48 films. Modern agriculture relies heavily on the use of plastic mulch films to elevate  
49 soil temperature and maintain soil moisture, to provide protection and to prevent  
50 weed growth. Conventional mulch films are composed of polyethylene (PE). If these  
51 films are thin, they cannot readily be reused nor recycled after application, which  
52 means they need to be incinerated or landfilled after single use (Serrano-Ruiz et al.,  
53 2021). Such a short life-cycle of plastic products is undesirable, and both incineration  
54 and landfilling are considered undesirable end of life treatment options. Another  
55 disadvantage of conventional thin (<25  $\mu\text{m}$ ) mulch films is that they cannot easily be  
56 re-collected entirely after their application, leaving PE plastic residues in the soil.  
57 Because PE films persist, these residuals building up in soils over time, eventually  
58 harming agricultural productivity (Gao et al., 2019). Furthermore, the films may  
59 slowly decompose into microplastics. Both ~~both~~-macro- and microplastics have been  
60 shown to be transported by wind and water, away from the initial application site,  
61 towards waterways, lakes and coastal seas (Egessa et al., 2020; Ren et al., 2021;  
62 Yang et al., 2022), where they have been shown to cause harm to the ecosystem  
63 (Galloway et al., 2017; Hale et al., 2020).

64 In an attempt to limit soil plastic pollution, the agricultural industry has developed  
65 soil-biodegradable mulch films as substitutes of conventional films. These soil-  
66 biodegradable films can be composed of different materials, for example  
67 biologically derived polymers, forms of polysaccharides, polypeptides, but also lipid  
68 based products, and are designed to be degradable by soil microorganisms under  
69 the moisture and temperature conditions found in the soil environment (Yang et al.,  
70 2020). A standard has been set by the EU (2018 EU standard EN 17033) for the  
71 biodegradability of mulch films named as such. In general, the biodegradation of the  
72 soil-biodegradable mulch films (hereafter referred to as BD-films) involves two  
73 stages: i. Breakdown of the large pieces into small organic molecules by abiotic or  
74 biological (enzymatic) processes. II. Uptake and metabolic use of these organic  
75 molecules by microorganisms under conversion into CO<sub>2</sub> and microbial biomass  
76 (Zumstein et al., 2018).

77 Soil-biodegradable mulch films are designed to be biodegraded in soil. It is, however,  
78 possible that fragments of the of BD-films may be transported away from agricultural  
79 soils before being completely biodegraded. While we are not aware of studies  
80 specifically on the transport of BD- films, transport of conventional mulch films by  
81 wind or water has been observed (Dris et al., 2015; Ren et al., 2021; Yang et al.,  
82 2022).

83 Earlier studies have shown successful biodegradation of BD-films in soil  
84 environments (Li et al., 2014; Sintim et al., 2020), with complete biodegradation  
85 being predicted to take several years. Research on the aquatic biodegradation of  
86 BD-plastics is predominantly focused on the marine realm. There, several studies  
87 have shown microbial colonization on BD-plastics (Eich et al., 2015; Nauendorf et  
88 al., 2016; Morohoshi et al., 2018), and some showed a disintegration of the plastics  
89 over time (Briassoulis et al., 2020; Lott et al., 2020; Wei et al., 2021), although these  
90 studies were not performed on mulch films, but on other BD-plastic products, such  
91 as bags. The transferability of studies on different BD-plastics is limited, but as there

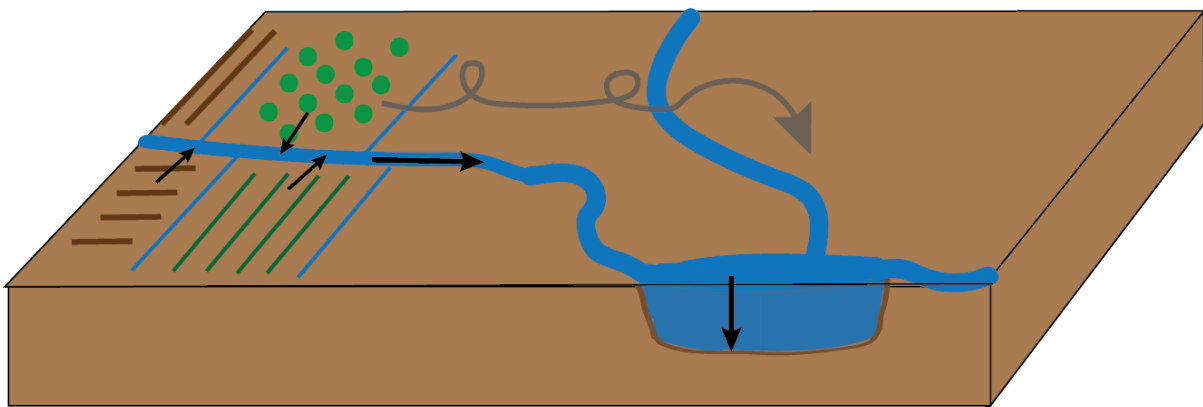
92 are currently no studies on BD mulch films in the aquatic environment available, we  
93 still introduce these studies on other BD plastic products here. All the above  
94 presented studies are disintegration studies, which generally cannot distinguish  
95 whether BD-plastics merely physically degraded into smaller particles (i.e.,  
96 fragmentation), or whether they have been microbially biodegraded, resulting in  
97 microbial biomass and CO<sub>2</sub> and/or CH<sub>4</sub> production, which makes an assessment of  
98 the mineralization of the BD plastics difficult. A recent study by Lott et al. 2021 (Lott  
99 et al., 2021) measured marine biodegradation of BD plastic components (PHB,  
100 PBSe) by CO<sub>2</sub> measurements, in experiments with marine sediments. There, they  
101 found a 116 and 222 days in incubation experiments under benthic and eulittoral  
102 marine conditions.

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104 Here, we determine the biodegradability of BD- films in freshwater lake sediments by  
105 performing sediment incubation studies with two types of soil-biodegradable, and  
106 one traditional PE plastic mulch films. Our results show that none of the tested films  
107 biodegraded over the 40 weeks incubation period: we detected no excess production  
108 of CO<sub>2</sub> or methane in the incubation experiments with BD-plastics as compared to  
109 PE-films and plastic-free control sediments. Furthermore, there were no signs of any  
110 physical degradation of any of the films after the 40 weeks incubation. Our findings  
111 suggest that the conditions in the tested sediment, likely including abiotic factors  
112 (unfavorable temperature and water chemistry) and biotic factors (low abundance  
113 and activity of BD-MF degrading microorganisms) impaired biodegradation in the  
114 lake. Our findings suggest significant higher stability of soil-biodegradable mulch  
115 films in lake sediments than soils, highlighting the need for measures to control off-  
116 field transport of soil BD-films.

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120 Fig. 1. Transport of plastics from agricultural sources to lake sediments, either via  
121 water ways or via wind transport.

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## Methods

### Sediment collection

Sediment cores were taken from Lake Lucerne (Switzerland) at 46°59'38"N 8°20'57"E in June 2021, [at an approximate water depth of 40 m](#). Multiple deployments with a multicorer device were used to collect the required number of sediment cores, which were later randomly distributed over the treatments. Sediments were collected into 10 cm diameter, transparent butyrate plastic core liners. All sediment cores were brought into a climate room of 10°C (to mimic the temperature of Lake Lucerne bottom waters, see (Fiskal et al., 2019)) within 2 hours after core collection.

### Gas analysis incubation experiment

To determine the production of CO<sub>2</sub> and CH<sub>4</sub> from the sediments, 16 cores were setup for incubation studies in the 10°C climate room. An overview of the experimental setup is presented in Fig. 2. Four cores did not receive any treatment (film-free control group), four cores were supplied with PE films, and eight with biodegradable mulch films purchased from two Swiss agricultural vendors: four cores received BD mulch films purchased from Sansonnens FG Frères SA (Rueyres-les-Prés, Switzerland), called MF-S from here on, and four with biodegradable plastics purchased from Gvz-rossat AG (Otelfingen, Switzerland), called MF-R from here on. These plastics are classified as 'OK biodegradable soil' by TÜV Austria, and are composed mainly of PBAT and polylactic acid (PLA)(Nelson et al., 2019). The plastics (all kinds) were cut and weighted. Each core received 300 milligrams of plastics, folded multiple times to form a rectangular stack. To increase the surface area for degradation and to make the stacks of plastic film loose, rather than squeezed together, the plastics were cut into comb shaped pieces (see Fig. S1 for plastics weights and shape), which allowed sediment between the folded layers. Part of the water was removed from each core, to allow for an air headspace above the overlying water. The oxygen concentration in the headspace was not tracked, but results from earlier experiments with the same sediments showed oxygen was not completely consumed over one year's time. ~~(van Grinsven et al., in prep. Data can be made available to reviewers).~~

For a part of the cores, the stopper at the bottom was pushed upwards slightly (5-15 cm) to decrease the water and headspace volumes above the sediment surface, to create more equal headspace volumes in each core. This type of cores was distributed over the treatments (2 per treatment). The procedure was not repeated for all used cores, because not all stoppers turned out to be tight enough to be pushed up and remain in position. All cores, pushed up or not, had a 7 cm height gas headspace ( $\pm 550 \text{ cm}^3$ ). Water headspace volumes varied between cores. One of the PE cores leaked water over the course of the experiment and was therefore discarded from the results. Films (biodegradable/PE) were pushed into the sediment surface (1-2 cm). Each of the cores was closed off with an adapted stopper, that contained two sampling ports, as specified in van Grinsven et al. (in prep). These stoppers allowed for non-invasive gas sampling over the course of the experiment. The cores were incubated in the dark at 10°C for 40 weeks. Gas samples (10 mL) were taken monthly. The removed gas was resupplied with N<sub>2</sub> gas to prevent underpressure in the cores. The volume of gas exchanged on a monthly basis was less than 2% of the gas headspace. Gas samples were pushed into 60 mL serum

173 bottles containing N<sub>2</sub> gas, which were analyzed for CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O  
174 concentrations. Gas samples were analyzed by gas chromatography (GC; Agilent  
175 6890N, Agilent Technologies) using a Carboxen 1010 column (30 m x 0.53 mm,  
176 Supelco), a flame ionization detector and an auto-sampler (Valco Instruments Co.  
177 Inc.)

178 The change in headspace CO<sub>2</sub> and CH<sub>4</sub> concentration over time was taken as a  
179 measure of biodegradation. The headspace CO<sub>2</sub> and CH<sub>4</sub> concentrations of all  
180 cores of each treatment were averaged, ~~as shown in Fig. 3~~. The data of each  
181 individual core is shown in the supplemental material (Fig. S2; S3).

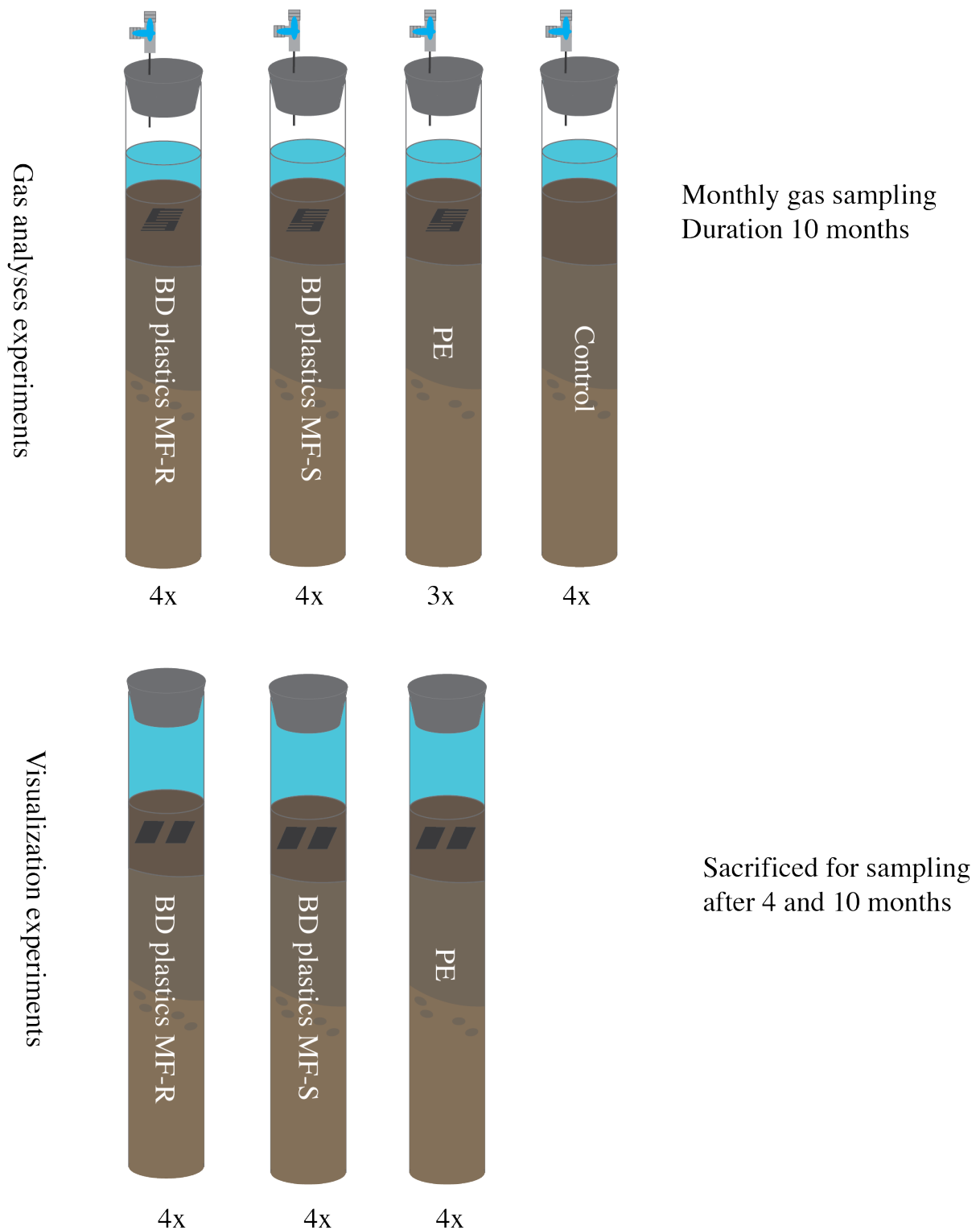
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183 Visualization incubation experiment

184 Identical cores were used for the gas analysis and visualization experiments. 12  
185 cores were used for the visualization experiment: four cores received PE plastics,  
186 four cores MF-S BD-plastics, and four cores MF-R BD-plastics, as shown in Fig. 2.  
187 No control cores were included, as these experiments were purely used for  
188 visualization of the extracted plastics, and control experiments would therefore not  
189 result in any results, as nothing could have been visualized. As a smaller quantity of  
190 plastics could be added to the visualization experiments, the plastics did not need to  
191 be folded, and therefore, they were not cut either (Fig. S1). Two unfolded pieces of  
192 plastics were added to each core. No cores were pushed up and no headspace was  
193 created, the cores were filled with water up to the stopper, to cause minimal  
194 disturbance and to keep conditions close to *in situ* conditions. Bottom water  
195 contained around 400 μM of oxygen at the start of the experiment. Earlier  
196 experiments (~~van Grinsven et al., in prep, data available to reviewers on request~~)  
197 showed limited oxygen consumption over the course of one year. Non-adapted  
198 stoppers were used, as these cores were not subsampled over the course of the  
199 experiment. Incubation occurred in parallel to the gas analysis experiment, at 10°C in  
200 the dark.

201 After 18 weeks, two cores per treatment were sampled and sacrificed. The plastics  
202 were removed from the sediments with tweezers, pulled through the overlying water,  
203 and put into a 4% paraformaldehyde solution overnight at 4°C to fixate the samples.  
204 The samples were then rinsed twice with PBS 1x to remove paraformaldehyde, and  
205 then stored at -20°C until further usage. This was repeated after 40 weeks for the  
206 remaining sediment cores.

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208 (Biodegradable) plastic samples were used for microscopy after all samples were  
209 retrieved at the end of the experiments. Samples were stained with DAPI (1 mg mL<sup>-1</sup>)  
210 and placed on microscope slides with a mounting and anti-fading medium. The  
211 samples were viewed under a Leica fluorescence microscope with 100x and 400x  
212 magnification, to look for biofilm growth and visible degradation of the plastics.

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 216 **Fig. 2.** Experimental setup. The gas analysis experiments contained a headspace to  
 217 allow for gas sampling, via a sampling port. The visualization experiments had no  
 218 gas headspace, the entire space above the sediment was filled with *in situ* bottom  
 219 water. No control experiment was included in the visualization experiments, as these  
 220 experiments were solely used for microscopy of the retrieved plastics (see methods).  
 221 All experiments were performed at *in situ* (Lake Lucerne) temperature, in the dark.  
 222 BD – Biodegradable; PE – polyethylene; MF-R and MF-S – Biodegradable mulch  
 223 films of two Swiss companies (see methods).

## 224 Results

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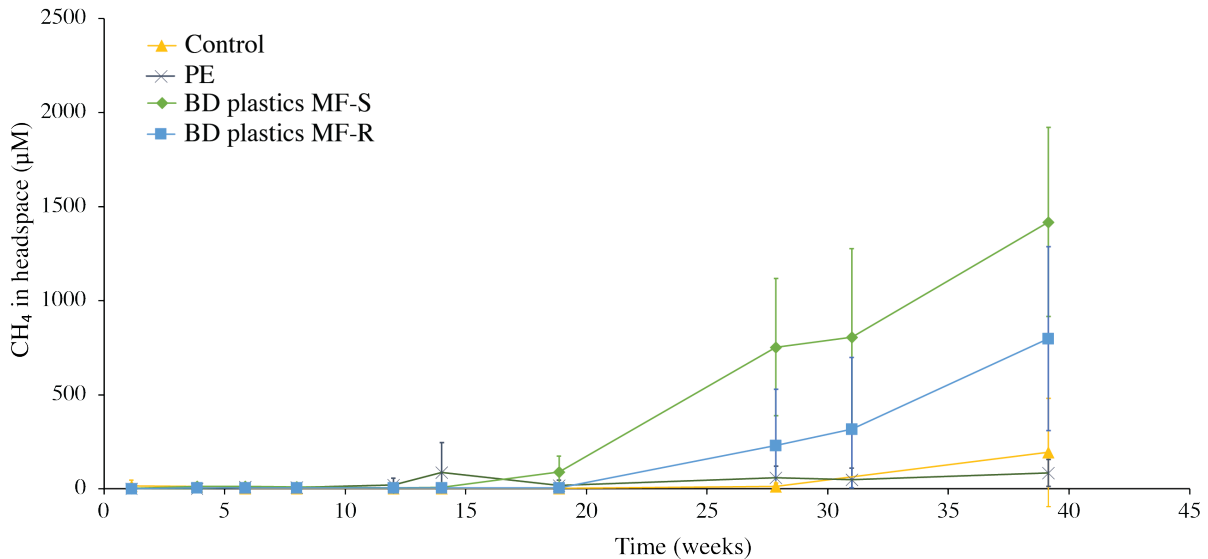
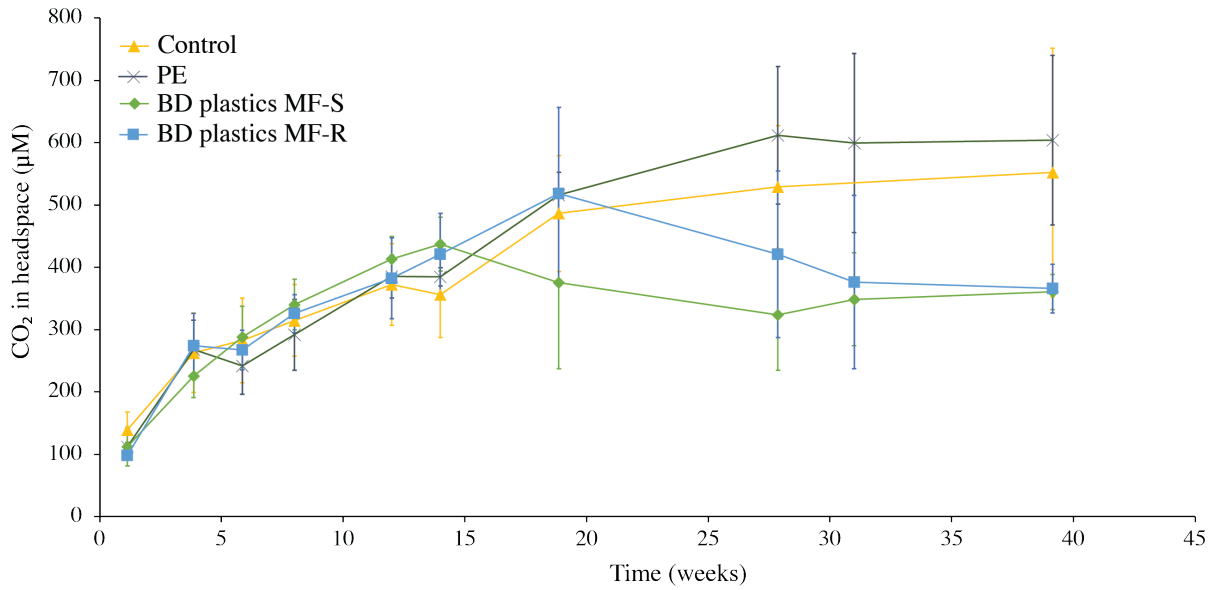
226 Concentrations of CO<sub>2</sub> and methane in the gas headspace of the sediment cores  
227 were followed over the course of the experiment, to assess plastic biodegradation.  
228 Given the organic composition of each of the plastics types, being composed of  
229 organic polymers, CO<sub>2</sub> and possibly CH<sub>4</sub> were expected to be key biodegradation  
230 products, alongside with microbial biomass. ~~In aquatic environments, research on  
231 BD plastics has predominantly been focused on assessing degradation by visual  
232 assessments (photography, microscopy) or by repeated measurements of remaining  
233 plastic mass over time (Briassoulis et al., 2020; Lott et al., 2020). A downside of  
234 these approaches is that fragmented plastics, which have only been partially  
235 degraded and then form micro- and nanoparticles (shown to form out of both PE and  
236 BD plastics (Wei et al., 2021)), can easily be missed, and their mass is incorrectly  
237 counted as being completely degraded.~~

238 Overall, the change in CO<sub>2</sub> concentration over time did not differ significantly  
239 between any of the treatments, including the control (pairwise ANOVA with post-hoc  
240 analyses with a Bonferroni adjustment revealed that all the pairwise differences  
241 between groups were not statistically significantly different,  $p > 0.05$ ). During the first  
242 20 weeks, the four treatments behave similarly, with a steady increase in headspace  
243 CO<sub>2</sub> concentration, with an exception of MF-S, which started to differ from 15 weeks  
244 onwards (Fig. 3). From 20 weeks (15 for MF-S) to the termination at 40 weeks, the  
245 concentration change differed slightly between treatments, but variation within  
246 treatments was large, and the difference between groups not statistically significant,  
247 as described above. In the incubations with PE-films, the CO<sub>2</sub> concentration  
248 increased until week 30 and then stabilized. In the control experiment, this  
249 stabilization was observed from week 20 until week 40. In both the BD-plastics  
250 treatments, a decrease in the CO<sub>2</sub> concentration relative to the preceding weeks was  
251 observed in the middle part of the incubation period, from either week 15 or 20 (MF-  
252 S and MF-R, respectively) to week 30, after which the concentrations stabilized.  
253 Individual cores, however, showed variations within the treatments, and the  
254 difference between treatments were not statistically significant, as stated above.  
255 Overall, the results show that the BD-plastics treatments did not produce larger  
256 amounts of CO<sub>2</sub> than the control (no addition) treatment.

257 Methane production from all sediments, including the control treatment, remained  
258 around zero from week 0 to week 15. After week 15, the methane concentration in  
259 the PE treatments increased slightly (Fig. 3; more detailed view available in the  
260 supplemental information, Fig. S4), and in the BD-plastics treatment MF-S, more  
261 pronounced. In the control experiment, methane concentrations dipped slightly below  
262 the starting concentrations, but then increased from week 28 onwards, albeit with  
263 large variations between individual sediment cores. Both BD-plastics treatments  
264 showed large increases in relative amounts of methane from week 20 onwards,  
265 although the large variations between individual cores within the treatments lead to  
266 large, overlapping error bars. The stabilization of the CO<sub>2</sub> concentration and the  
267 increase in CH<sub>4</sub> concentration in the BD-plastics experiments matched timewise,  
268 both occurring from week 15 - 20 onwards, although the increase in CH<sub>4</sub> continued  
269 after week 28, when the CO<sub>2</sub> concentration in the BD-plastics experiments had  
270 stabilized.

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275 Fig. 3. Headspace CO<sub>2</sub> and methane concentrations, in µM. An excerpt of the CH<sub>4</sub> graph,  
276 with a smaller y-axis, is available in Fig. S4.

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278 Visualization of the plastic film surfaces after retrieval from the sediments after 40  
279 weeks did not show any clear signs of degradation or biofilm formation, on any of the  
280 film types (PE, MF-S, MF-R; Fig. S5). DAPI staining showed a low number of  
281 microbes on the plastics' surfaces. Both regular and microscopic visualization, of all  
282 types of used plastics, showed no variation in the appearance between the end to  
283 the start of the experiment.

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287 **Discussion**

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289 The biodegradability of BD-plastic films depends largely on the match between their  
290 composition and the environment. In soils, BD mulch films have been shown to  
291 biodegrade successfully (Li et al., 2014; Sintim et al., 2020). However, due to this  
292 required match between the properties of the material and the in situ conditions for  
293 biodegradation, transport of BD mulch films to aquatic environments could result in a  
294 buildup of material, as is suggested by our results. The biodegradability of the mulch  
295 films investigated in this research is in strong contrast to the biodegradability of these  
296 films under previously investigated soil conditions. Whereas within several soils,  
297 substantial degradation of the incorporated BD-plastic films was observed (11-98%  
298 after 12 months (determined by 'percentage of area remaining', (Li et al., 2014); 12-  
299 23% degradation of powdered BD-plastics after 28 days (determined by weight loss;  
300 Adhikari et al., 2016)), no biodegradation could be observed in our sediment  
301 experiments, neither based on produced CO<sub>2</sub> nor from visual analysis by  
302 microscopy. Similar conclusions were drawn based on a 100-day marine, coastal  
303 zone sediment experiment (Nauendorf et al., 2016), even though biofilm growth on  
304 biodegradable plastic bags (composed of biodegradable polyester and corn starch)  
305 was observed in that study. Freshwater BD-plastic degradation studies are more  
306 rare, and are focused on biofilm formation, rather than on biodegradation rates  
307 (Morohoshi et al., 2018). As observed by Nauendorf et al. (2016), biofilm formation  
308 and degradation are not necessarily related. Most marine sediment studies combine  
309 a biofilm assessment with visual assessments of degradation, but don't quantify the  
310 degradation, neither respiration nor fragmentation. Quantification of degradation  
311 percentages based on assessment of the remaining plastics is difficult, as  
312 microparticles, that form as a result of physical breakdown, are difficult to recover  
313 and quantify (Wei et al., 2021). Furthermore, weight assessments are often  
314 unreliable, as microbial biomass, attached to the plastics is difficult to remove  
315 completely without disturbing the plastic mass itself (Nauendorf et al., 2016). Our  
316 results show a large variation between individual incubations, within treatments (Fig.  
317 S2). Sediment cores were all taken from the same location, but by several multicorer  
318 casts. Potentially, variations between the cores existed from the start, for example in  
319 the amount of carbon present in each sediment core. By randomly distributing the  
320 cores over the treatments, we have guaranteed that this did not cause a biased  
321 result in the treatment comparisons. It may, however, have resulted in larger  
322 variations within treatments. Furthermore, it is possible certain cores were more  
323 disturbed than others while being transported to the lab, resulting in slightly changed  
324 oxygen conditions within the upper cm of the cores. Overall, the effect of the  
325 treatments did not exceed the variation between incubations within each treatment  
326 (Fig. 3), indicating there was no strong treatment effect.

327

328 The BD mulch films used in this study were designed to degrade in agricultural soils.  
329 A major difference between aquatic sediment conditions and soils may be the  
330 oxygen availability: due to the low diffusion speed of oxygen through water, oxygen  
331 is limiting in lake sediments, despite oxygen-rich overlying water. Lake Lucerne,  
332 where the sediments used in this study were collected, is an oligotrophic lake, with a  
333 high oxygen availability in the bottom waters (which was also remained during the  
334 experiment duration). In Lake Lucerne sediments, oxygen has been shown to be  
335 depleted from 1.0 cm downwards (Fiskal et al., 2019). The plastics were buried  
336 shallowly in the sediments, and will, at least partly, have been surrounded by

337 sediments containing dissolved O<sub>2</sub> (DO) in our experiments. Although exact DO  
338 penetration depths can only be determined with micro-sensors, we can assume  
339 initial oxic conditions here, based on the natural oxygen penetration depth of these  
340 sediments (Fiskal et al., 2019), the oxygen concentration in the overlying water  
341 ( $\pm 400 \mu\text{M}$ , Fiskal et al., 2019), and the disturbance caused by the pushing of the  
342 plastics into the sediments, bringing oxygen-rich water in contact with the sediment  
343 surrounding the plastic film. After the initial oxic conditions, DO may have become  
344 depleted within the originally anoxic sediment depths ( $> 1 \text{ cm}$ ), or shallower.  
345 Although under oxic conditions, microbial growth and substrate turnover is expected  
346 to occur more rapidly, earlier research showed similar BD-plastic biodegradation  
347 rates under oxic and anoxic conditions, despite higher microbial biomass formation  
348 on BD-plastics under oxic incubation conditions (Nauendorf et al., 2016). The onset  
349 of methane production in our incubations, from week 15 onwards, paired to the  
350 stabilization of the CO<sub>2</sub> concentration, suggests anoxic conditions established after  
351 15 weeks into the experiments. Although the increase in methane concentration in  
352 our MF-S and MF-R experiments could suggest enhanced breakdown of the BD-  
353 plastics, it's timely match to the decrease in CO<sub>2</sub> production, rather suggests that the  
354 respiration of sedimentary carbon, rather than the BD-plastics carbon, is being  
355 degraded, similarly to the control and PE experiment. It is, however, interesting to  
356 note that the BD-plastics experiments experienced this oxic-anoxic conversion,  
357 whereas the control and PE plastics didn't.

358  
359 Within soil experiments, several biotic and abiotic factors affect the extent of  
360 biodegradation. As the used BD-mulch films are designed to biodegrade in soils, we  
361 also consider these factors to compare the biodegradation in the aquatic realm to  
362 that of the soils. A first factor would be abiotic fragmentation by to photodegradation,  
363 which requires sunlight to reach the plastics. In soils, photodegradation affects the  
364 plastics, decreasing their quality and molecular weight (Zhang et al., 2021). In  
365 shallow surface waters, photodegradation may also affect plastic degradation. Our  
366 experiments, with sediment cores of 40 m water depth, were performed in the dark,  
367 as photodegradation is not relevant at such water depths. For future studies, pre-  
368 degradation of plastics by photodegradation, prior to incorporation into the sediment,  
369 would be interesting, as further discussed below. Besides photodegradation, Li et al.  
370 (2014) identified soil pH, soil temperature, and fungi abundance as potential factors  
371 for differences between degradation efficiency between sites. (Sander, 2019)  
372 suggested nitrogen limitations in soils may affect and limit BD-plastic biodegradation,  
373 as the N-poor BD-plastics solely supply the colonizing microorganisms with carbon,  
374 hydrogen and oxygen-rich compounds, and not with nitrogen. Organisms will  
375 therefore have to acquire nitrogen from different sources, which, if not sufficiently  
376 available, may limit growth and activity of the BD-plastic-colonizing and degrading  
377 organisms. (Lopardo et al., 2019) showed that the addition of BD-plastics to  
378 aquaculture eluent filters can improve nitrogen removal efficiency, by providing a N-  
379 limited carbon source. Lake Lucerne sediments have been shown to contain  
380 relatively high amounts of nitrate, up to around 10 cm depth ( $2 \mu\text{M}$  at 2.5 cm,  $33 \mu\text{M}$   
381 at 0.5 cm depth, (Fiskal et al., 2019). Nitrogen limitation is therefore not expected to  
382 have limited microbial colonization or growth in the experiments, although more  
383 research would be needed to determine the limiting factors for microbial growth. The  
384 here-used mulch films were new, and not deployed on agricultural fields prior to  
385 incubation. Therefore, their degradation status might differ from a part of the mulch  
386 films that enter aquatic environments from agricultural fields. It is, however, unknown

387 what is the general state of plastics when they are transported towards the aquatic  
388 ecosystem. Newer BD mulch foils are more likely to still contain the most labile,  
389 easily degradable compounds, which may be degraded after deployments in fields,  
390 either by chemical or biological degradation, On the other hand, used mulch films are  
391 more likely to contain a larger surface area, and are more likely to already contain a  
392 biofilm. Further research on a potential priming effect of soil pre-incubation, is  
393 necessary. Furthermore, it would be interesting to test other lake environments, such  
394 as eutrophic lakes, which differ in microbial and chemical composition from the  
395 currently studied lake Lucerne (van Grinsven et al., 2022).

396  
397 The limited mobility of soil bacteria has earlier been suggested to potentially limit  
398 colonization of BD-plastics in soils, as direct contact is required between the soil and  
399 plastics to allow for colonization (Sander, 2019). Such problems may, however, be  
400 easier overcome within aquatic environment, as the water-saturation allows for a  
401 higher mobility of microbes and an easier exchange of compounds via diffusion and  
402 flow. However, the community of microorganisms may differ strongly between soil  
403 and sediment environments, mostly due to different redox and nutrient conditions  
404 between the two environments. As the genera of microbes that are capable of BD-  
405 plastic degradation are still being investigated and discovered, especially in  
406 freshwater environments, it is not possible to simply scan the microbial community  
407 for BD-plastic degraders, and make an assessment on the potential for  
408 biodegradation based on the microbial community composition. We did not  
409 determine the microbial community composition in these experiments, as the results  
410 would most likely not have been able to provide us information on the plastic  
411 degradation: the sedimentary microbial community is highly diverse, differs strongly  
412 with depth, and contains many organisms with unknown functions (Han et al., 2020).  
413 The lack of suitable BD-plastic degrading microorganisms is, however, the most  
414 likely explanation for the large difference in BD mulch film decomposition between  
415 our sediment experiment, and previously performed soil experiments. The sediments  
416 likely lack microorganisms that produce hydrolytic enzymes (esterases) that can  
417 cleave PBAT and PLA, the main polyesters in the used BD-films. When specific  
418 genes known to be associated to BD plastic degradation are identified in the future, it  
419 would be interesting to screen our incubation experiment sediments for those genes,  
420 to confirm such a hypothesis.

## 421 422 **Conclusions**

423 Two types of biodegradable plastic mulch films showed no signs of degradation over  
424 a 10-month period within lake surface sediments, suggesting very limited  
425 biodegradability in freshwater sediment environments. These findings stress the  
426 importance of limiting the transport of plastics from agricultural soils to waterways –  
427 not only for traditional plastics, but also for soil biodegradable plastics. Given that  
428 these plastics are designed to remain in the soil rather than to be removed after use,  
429 they remain present in the natural environment until complete degradation. The  
430 extent of the transport of BD mulch films from soils to waterways and lakes is up to  
431 this point unknown. Our results show it is of high importance to investigate how  
432 much of the applied biodegradable mulch films are transported to aquatic  
433 environments, to be able to assess the potential pollution of lakes and coastal areas  
434 by the current and future applications of biodegradable plastic mulch films.

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437

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442

443 **Competing interests**

444

445 The contact author has declared that neither they nor their co-authors have any  
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448 **References**

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