

## **Supplementary:Material**

### **Rate coefficients for the reactions of OH radical with C3-C11 alkanes determined by the relative rate technique**

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**This file contains, Supplementary Tables: 1-2 and Supplementary Figures:1-15**

**Table S1.** Initial conditions of the various species injected into the reactor.

Alkanes	H <sub>2</sub> O <sub>2</sub>	N <sub>2</sub> /Air	PAMs
10 ppb	100 ul	250 L	2.5 L

**Table S2.** Rate constants for OH + Alkanes in N<sub>2</sub> and Air at 273-323 K.

T (K)	$K_{OH+n\text{-Hexane}}$ (cm <sup>3</sup> molecule <sup>-1</sup> s <sup>-1</sup> )	Slope <sup>a</sup>	$K_{OH}^b$ (cm <sup>3</sup> molecule <sup>-1</sup> s <sup>-1</sup> )	Bath gas
2,3-Dimethylbutane				
273	4.17×10 <sup>-12</sup>	1.224±0.005	(5.10±0.05)×10 <sup>-12</sup>	Air
		1.220	5.09×10 <sup>-12</sup>	N <sub>2</sub>
283	4.44×10 <sup>-12</sup>	1.201±0.008	(5.33±0.07)×10 <sup>-12</sup>	Air
		1.187	5.27×10 <sup>-12</sup>	N <sub>2</sub>
293	4.70×10 <sup>-12</sup>	1.181±0.009	(5.55±0.09)×10 <sup>-12</sup>	Air
		1.146	5.39×10 <sup>-12</sup>	N <sub>2</sub>
303	4.97×10 <sup>-12</sup>	1.126±0.004	(5.60±0.04)×10 <sup>-12</sup>	Air
		1.134	5.64×10 <sup>-12</sup>	N <sub>2</sub>
313	5.22×10 <sup>-12</sup>	1.091±0.001	(5.70±0.01)×10 <sup>-12</sup>	Air
		1.088	5.68×10 <sup>-12</sup>	N <sub>2</sub>
323	5.48×10 <sup>-12</sup>	1.061±0.001	(5.82±0.01)×10 <sup>-12</sup>	Air
		1.052	5.76×10 <sup>-12</sup>	N <sub>2</sub>
Methylcyclopentane				
273	4.17×10 <sup>-12</sup>	1.544±0.006	(6.44±0.03)×10 <sup>-12</sup>	Air
		1.515	6.32×10 <sup>-12</sup>	N <sub>2</sub>
283	4.44×10 <sup>-12</sup>	1.447±0.040	(6.43±0.18)×10 <sup>-12</sup>	Air
		1.478	6.56×10 <sup>-12</sup>	N <sub>2</sub>
293	4.70×10 <sup>-12</sup>	1.417±0.016	(6.66±0.08)×10 <sup>-12</sup>	Air
		1.448	6.80×10 <sup>-12</sup>	N <sub>2</sub>
303	4.97×10 <sup>-12</sup>	1.398±0.014	(6.95±0.07)×10 <sup>-12</sup>	Air
		1.430	7.11×10 <sup>-12</sup>	N <sub>2</sub>
313	5.22×10 <sup>-12</sup>	1.388±0.002	(7.24±0.01)×10 <sup>-12</sup>	Air
		1.359	7.09×10 <sup>-12</sup>	N <sub>2</sub>
323	5.48×10 <sup>-12</sup>	1.337±0.010	(7.33±0.05)×10 <sup>-12</sup>	Air
		1.334	7.31×10 <sup>-12</sup>	N <sub>2</sub>
n-Heptane				
273	4.17×10 <sup>-12</sup>	1.287±0.040	(5.37±0.17)×10 <sup>-12</sup>	Air
		1.288	5.37×10 <sup>-12</sup>	N <sub>2</sub>

283	4.44×10 <sup>-12</sup>	1.308±0.020	(5.81±0.09)×10 <sup>-12</sup>	Air
		1.340	5.95×10 <sup>-12</sup>	N <sub>2</sub>
293	4.70×10 <sup>-12</sup>	1.311±0.049	(6.16±0.23)×10 <sup>-12</sup>	Air
		1.318	6.19×10 <sup>-12</sup>	N <sub>2</sub>
303	4.97×10 <sup>-12</sup>	1.323±0.011	(6.58±0.06)×10 <sup>-12</sup>	Air
		1.312	6.52×10 <sup>-12</sup>	N <sub>2</sub>
313	5.22×10 <sup>-12</sup>	1.359±0.005	(7.09±0.01)×10 <sup>-12</sup>	Air
		1.275	6.66×10 <sup>-12</sup>	N <sub>2</sub>
323	5.48×10 <sup>-12</sup>	1.318±0.005	(7.22±0.03)×10 <sup>-12</sup>	Air
		1.264	6.93×10 <sup>-12</sup>	N <sub>2</sub>
3-Methylheptane				
273	4.17×10 <sup>-12</sup>	1.584±0.018	(6.61±0.07)×10 <sup>-12</sup>	Air
		1.616	6.74×10 <sup>-12</sup>	N <sub>2</sub>
283	4.44×10 <sup>-12</sup>	1.591±0.003	(7.06±0.02)×10 <sup>-12</sup>	Air
		1.643	7.29×10 <sup>-12</sup>	N <sub>2</sub>
293	4.70×10 <sup>-12</sup>	1.612±0.076	(7.58±0.36)×10 <sup>-12</sup>	Air
		1.630	7.66×10 <sup>-12</sup>	N <sub>2</sub>
303	4.97×10 <sup>-12</sup>	1.595±0.047	(7.93±0.24)×10 <sup>-12</sup>	Air
		1.639	8.14×10 <sup>-12</sup>	N <sub>2</sub>
313	5.22×10 <sup>-12</sup>	1.598±0.072	(8.34±0.38)×10 <sup>-12</sup>	Air
		1.558	8.13×10 <sup>-12</sup>	N <sub>2</sub>
323	5.48×10 <sup>-12</sup>	1.550±0.001	(8.49±0.01)×10 <sup>-12</sup>	Air
		1.526	8.36×10 <sup>-12</sup>	N <sub>2</sub>
n-Octane				
273	4.17×10 <sup>-12</sup>	1.637±0.075	(6.83±0.31)×10 <sup>-12</sup>	Air
		1.607	6.70×10 <sup>-12</sup>	N <sub>2</sub>
283	4.44×10 <sup>-12</sup>	1.642±0.007	(7.29±0.03)×10 <sup>-12</sup>	Air
		1.716	7.62×10 <sup>-12</sup>	N <sub>2</sub>
293	4.70×10 <sup>-12</sup>	1.617±0.020	7.60±0.10)×10 <sup>-12</sup>	Air
		1.655	7.78×10 <sup>-12</sup>	N <sub>2</sub>
303	4.97×10 <sup>-12</sup>	1.643±0.009	(8.17±0.04)×10 <sup>-12</sup>	Air
		1.687	8.38×10 <sup>-12</sup>	N <sub>2</sub>
313	5.22×10 <sup>-12</sup>	1.689±0.041	(8.82±0.22)×10 <sup>-12</sup>	Air
		1.666	8.70×10 <sup>-12</sup>	N <sub>2</sub>
323	5.48×10 <sup>-12</sup>	1.625±0.053	(5.90±0.29)×10 <sup>-12</sup>	Air
		1.633	8.95×10 <sup>-12</sup>	N <sub>2</sub>
Propane				
273	4.17×10 <sup>-12</sup>	0.170	7.09×10 <sup>-13</sup>	Air
		0.167	6.97×10 <sup>-13</sup>	N <sub>2</sub>
283	4.44×10 <sup>-12</sup>	0.181±0.041	8.97×10 <sup>-13</sup>	Air
		0.197	8.75×10 <sup>-13</sup>	N <sub>2</sub>
293	4.70×10 <sup>-12</sup>	0.215±0.027	(1.01±0.13)×10 <sup>-12</sup>	Air

		0.195	$9.14 \times 10^{-13}$	N <sub>2</sub>
303	$4.97 \times 10^{-12}$	0.199±0.013	$(9.91 \pm 0.63) \times 10^{-13}$	Air
		0.208	$1.04 \times 10^{-12}$	N <sub>2</sub>
313	$5.22 \times 10^{-12}$	0.220±0.010	$(1.15 \pm 0.05) \times 10^{-12}$	Air
		0.214	$1.12 \times 10^{-12}$	N <sub>2</sub>
323	$5.48 \times 10^{-12}$	0.223±0.051	$(1.22 \pm 0.28) \times 10^{-12}$	Air
		0.224	$1.23 \times 10^{-12}$	N <sub>2</sub>
		Isobutane		
273	$4.17 \times 10^{-12}$	0.350±0.033	$(1.46 \pm 0.14) \times 10^{-12}$	Air
		0.338	$1.41 \times 10^{-12}$	N <sub>2</sub>
283	$4.44 \times 10^{-12}$	0.400±0.020	$(1.78 \pm 0.09) \times 10^{-12}$	Air
		0.378	$1.68 \times 10^{-12}$	N <sub>2</sub>
293	$4.70 \times 10^{-12}$	0.389±0.024	$(1.83 \pm 0.11) \times 10^{-12}$	Air
		0.394	$1.85 \times 10^{-12}$	N <sub>2</sub>
303	$4.97 \times 10^{-12}$	0.416±0.054	$(2.07 \pm 0.27) \times 10^{-12}$	Air
		0.421	$2.09 \times 10^{-12}$	N <sub>2</sub>
313	$5.22 \times 10^{-12}$	0.398±0.025	$(2.08 \pm 0.13) \times 10^{-12}$	Air
		0.415	$2.17 \times 10^{-12}$	N <sub>2</sub>
323	$5.48 \times 10^{-12}$	0.424±0.009	$(2.32 \pm 0.05) \times 10^{-12}$	Air
		0.428	$2.34 \times 10^{-12}$	N <sub>2</sub>
		n-Butane		
273	$4.17 \times 10^{-12}$	0.316±0.140	$(1.32 \pm 0.58) \times 10^{-12}$	Air
		0.224	$9.34 \times 10^{-13}$	N <sub>2</sub>
283	$4.44 \times 10^{-12}$	0.389±0.068	$(1.73 \pm 0.30) \times 10^{-12}$	Air
		0.418	$1.86 \times 10^{-12}$	N <sub>2</sub>
293	$4.70 \times 10^{-12}$	0.408±0.004	$(1.92 \pm 0.02) \times 10^{-12}$	Air
		0.436	$2.05 \times 10^{-12}$	N <sub>2</sub>
303	$4.97 \times 10^{-12}$	0.449±0.031	$(2.23 \pm 0.15) \times 10^{-12}$	Air
		0.470	$2.34 \times 10^{-12}$	N <sub>2</sub>
313	$5.22 \times 10^{-12}$	0.455±0.027	$(2.37 \pm 0.15) \times 10^{-12}$	Air
		0.481	$2.51 \times 10^{-12}$	N <sub>2</sub>
323	$5.48 \times 10^{-12}$	0.468±0.024	$(2.57 \pm 0.13) \times 10^{-12}$	Air
		0.489	$2.68 \times 10^{-12}$	N <sub>2</sub>
		Isopentane		
273	$4.17 \times 10^{-12}$	0.676±0.006	$(2.82 \pm 0.03) \times 10^{-12}$	Air
		0.686	$2.86 \times 10^{-12}$	N <sub>2</sub>
283	$4.44 \times 10^{-12}$	0.693±0.058	$(3.08 \pm 0.26) \times 10^{-12}$	Air
		0.679	$3.01 \times 10^{-12}$	N <sub>2</sub>
293	$4.70 \times 10^{-12}$	0.694±0.057	$(3.26 \pm 0.27) \times 10^{-12}$	Air
		0.707	$3.32 \times 10^{-12}$	N <sub>2</sub>
303	$4.97 \times 10^{-12}$	0.683±0.086	$(3.39 \pm 0.43) \times 10^{-12}$	Air
		0.656	$3.26 \times 10^{-12}$	N <sub>2</sub>
313	$5.22 \times 10^{-12}$	0.681±0.016	$(3.56 \pm 0.08) \times 10^{-12}$	Air

		0.664	$3.47 \times 10^{-12}$	N <sub>2</sub>
323	$5.48 \times 10^{-12}$	0.664±0.050	$(3.64 \pm 0.27) \times 10^{-12}$	Air
		0.654	$3.58 \times 10^{-12}$	N <sub>2</sub>
		n-pentane		
273	$4.17 \times 10^{-12}$	0.687±0.020	$(2.86 \pm 0.08) \times 10^{-12}$	Air
		0.651	$2.71 \times 10^{-12}$	N <sub>2</sub>
283	$4.44 \times 10^{-12}$	0.681±0.001	$(3.02 \pm 0.03) \times 10^{-12}$	Air
		0.658	$2.92 \times 10^{-12}$	N <sub>2</sub>
293	$4.70 \times 10^{-12}$	0.664±0.051	$(3.12 \pm 0.24) \times 10^{-12}$	Air
		0.670	$3.15 \times 10^{-12}$	N <sub>2</sub>
303	$4.97 \times 10^{-12}$	0.655±0.078	$(3.25 \pm 0.39) \times 10^{-12}$	Air
		0.675	$3.36 \times 10^{-12}$	N <sub>2</sub>
313	$5.22 \times 10^{-12}$	0.641±0.058	$(3.35 \pm 0.30) \times 10^{-12}$	Air
		0.668	$3.49 \times 10^{-12}$	N <sub>2</sub>
323	$5.48 \times 10^{-12}$	0.623±0.017	$(3.41 \pm 0.09) \times 10^{-12}$	Air
		0.653	$3.58 \times 10^{-12}$	N <sub>2</sub>
		Cyclopentane		
273	$4.17 \times 10^{-12}$	0.898±0.049	$(3.75 \pm 0.20) \times 10^{-12}$	Air
		0.878	$3.66 \times 10^{-12}$	N <sub>2</sub>
283	$4.44 \times 10^{-12}$	0.918±0.006	$(4.08 \pm 0.03) \times 10^{-12}$	Air
		0.933	$4.14 \times 10^{-12}$	N <sub>2</sub>
293	$4.70 \times 10^{-12}$	0.972±0.021	$(4.57 \pm 0.10) \times 10^{-12}$	Air
		0.948	$4.45 \times 10^{-12}$	N <sub>2</sub>
303	$4.97 \times 10^{-12}$	0.949±0.001	$(4.72 \pm 0.01) \times 10^{-12}$	Air
		0.950	$4.72 \times 10^{-12}$	N <sub>2</sub>
313	$5.22 \times 10^{-12}$	0.993±0.007	$(5.18 \pm 0.04) \times 10^{-12}$	Air
		0.943	$4.92 \times 10^{-12}$	N <sub>2</sub>
323	$5.48 \times 10^{-12}$	0.959±0.003	$(5.26 \pm 0.02) \times 10^{-12}$	Air
		0.961	$5.27 \times 10^{-12}$	N <sub>2</sub>
		2,2-Dimethylbutane		
273	$4.17 \times 10^{-12}$	0.299±0.010	$(1.25 \pm 0.01) \times 10^{-12}$	Air
		0.299	$1.25 \times 10^{-12}$	N <sub>2</sub>
283	$4.44 \times 10^{-12}$	0.329±0.023	$(1.46 \pm 0.10) \times 10^{-12}$	Air
		0.337	$1.49 \times 10^{-12}$	N <sub>2</sub>
293	$4.70 \times 10^{-12}$	0.371±0.004	$(1.74 \pm 0.02) \times 10^{-12}$	Air
		0.346	$1.63 \times 10^{-12}$	N <sub>2</sub>
303	$4.97 \times 10^{-12}$	0.372±0.028	$(1.85 \pm 0.14) \times 10^{-12}$	Air
		0.387	$1.92 \times 10^{-12}$	N <sub>2</sub>
313	$5.22 \times 10^{-12}$	0.389±0.009	$(2.03 \pm 0.05) \times 10^{-12}$	Air
		0.394	$2.06 \times 10^{-12}$	N <sub>2</sub>
323	$5.48 \times 10^{-12}$	0.376±0.009	$(2.06 \pm 0.05) \times 10^{-12}$	Air
		0.38826	$2.17 \times 10^{-12}$	N <sub>2</sub>
		2-Methylpentane		

273	4.17×10 <sup>-12</sup>	0.938±0.002	(3.91±0.01)×10 <sup>-12</sup>	Air
		0.924	3.85×10 <sup>-12</sup>	N <sub>2</sub>
283	4.44×10 <sup>-12</sup>	0.962±0.031	(4.17±0.14)×10 <sup>-12</sup>	Air
		0.951	4.22×10 <sup>-12</sup>	N <sub>2</sub>
293	4.70×10 <sup>-12</sup>	0.982±0.002	(4.62±0.01)×10 <sup>-12</sup>	Air
		0.961	4.52×10 <sup>-12</sup>	N <sub>2</sub>
303	4.97×10 <sup>-12</sup>	0.954±0.002	(4.74±0.01)×10 <sup>-12</sup>	Air
		0.964	4.79×10 <sup>-12</sup>	N <sub>2</sub>
313	5.22×10 <sup>-12</sup>	0.957±0.028	(5.00±0.15)×10 <sup>-12</sup>	Air
		0.929	4.85×10 <sup>-12</sup>	N <sub>2</sub>
323	5.48×10 <sup>-12</sup>	0.943±0.001	(5.17±0.01)×10 <sup>-12</sup>	Air
		0.932	5.11×10 <sup>-12</sup>	N <sub>2</sub>
3-Methylpentane				
273	4.17×10 <sup>-12</sup>	0.882±0.016	(3.68±0.07)×10 <sup>-12</sup>	Air
		0.908	3.79×10 <sup>-12</sup>	N <sub>2</sub>
283	4.44×10 <sup>-12</sup>	0.908±0.017	(4.03±0.08)×10 <sup>-12</sup>	Air
		0.967	4.30×10 <sup>-12</sup>	N <sub>2</sub>
293	4.70×10 <sup>-12</sup>	0.921±0.056	(4.33±0.26)×10 <sup>-12</sup>	Air
		0.951	4.47×10 <sup>-12</sup>	N <sub>2</sub>
303	4.97×10 <sup>-12</sup>	0.928±0.035	(4.61±0.17)×10 <sup>-12</sup>	Air
		0.983	4.88×10 <sup>-12</sup>	N <sub>2</sub>
313	5.22×10 <sup>-12</sup>	0.886±0.073	(4.63±0.38)×10 <sup>-12</sup>	Air
		0.940	4.91×10 <sup>-12</sup>	N <sub>2</sub>
323	5.48×10 <sup>-12</sup>	0.914±0.010	(5.01±0.06)×10 <sup>-12</sup>	Air
		0.927	5.08×10 <sup>-12</sup>	N <sub>2</sub>
2,4-Dimethylpentane				
273	4.17×10 <sup>-12</sup>	0.921±0.008	(3.84±0.04)×10 <sup>-12</sup>	Air
		0.922	3.84×10 <sup>-12</sup>	N <sub>2</sub>
283	4.44×10 <sup>-12</sup>	0.928±0.032	(4.12±0.14)×10 <sup>-12</sup>	Air
		0.943	4.19×10 <sup>-12</sup>	N <sub>2</sub>
293	4.70×10 <sup>-12</sup>	0.932±0.028	(4.38±0.13)×10 <sup>-12</sup>	Air
		0.946	4.45×10 <sup>-12</sup>	N <sub>2</sub>
303	4.97×10 <sup>-12</sup>	0.924±0.011	(4.59±0.06)×10 <sup>-12</sup>	Air
		0.929	4.62×10 <sup>-12</sup>	N <sub>2</sub>
313	5.22×10 <sup>-12</sup>	0.927±0.024	(4.84±0.12)×10 <sup>-12</sup>	Air
		0.898	4.69×10 <sup>-12</sup>	N <sub>2</sub>
323	5.48×10 <sup>-12</sup>	0.901±0.011	(4.94±0.06)×10 <sup>-12</sup>	Air
		0.878	4.81×10 <sup>-12</sup>	N <sub>2</sub>
Cyclohexane				
273	4.17×10 <sup>-12</sup>	1.252±0.044	(5.22±0.18)×10 <sup>-12</sup>	Air
		1.260	5.25×10 <sup>-12</sup>	N <sub>2</sub>
283	4.44×10 <sup>-12</sup>	1.310±0.033	(5.82±0.15)×10 <sup>-12</sup>	Air
		1.379	6.12×10 <sup>-12</sup>	N <sub>2</sub>

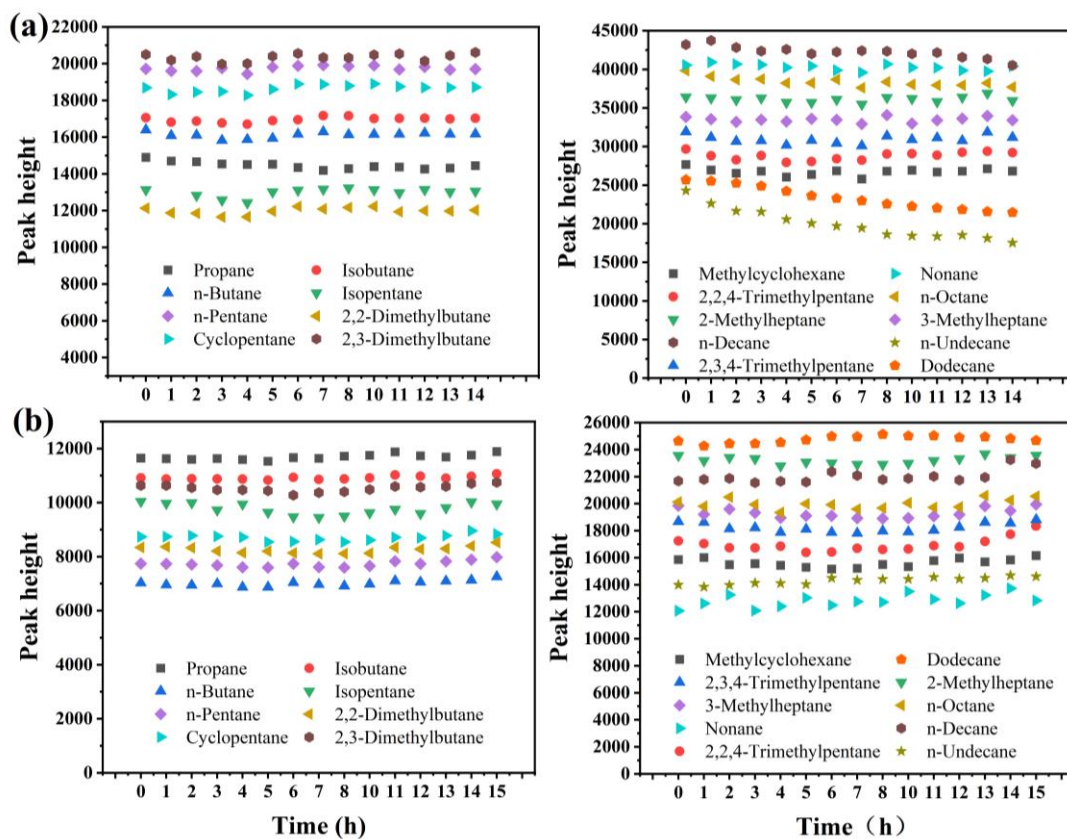
293	$4.70 \times 10^{-12}$	1.307±0.036	$(6.14 \pm 0.17) \times 10^{-12}$	Air
		1.354	$6.36 \times 10^{-12}$	N <sub>2</sub>
303	$4.97 \times 10^{-12}$	1.310±0.050	$(6.51 \pm 0.25) \times 10^{-12}$	Air
		1.261	$6.21 \times 10^{-12}$	N <sub>2</sub>
313	$5.22 \times 10^{-12}$	1.332±0.047	$(6.96 \pm 0.25) \times 10^{-12}$	Air
		1.243	$6.49 \times 10^{-12}$	N <sub>2</sub>
323	$5.48 \times 10^{-12}$	1.276±0.008	$(6.99 \pm 0.05) \times 10^{-12}$	Air
		1.118	$6.13 \times 10^{-12}$	N <sub>2</sub>
		2-Methylhexane		
273	$4.17 \times 10^{-12}$	1.365±0.001	$(5.69 \pm 0.01) \times 10^{-12}$	Air
		1.372	$5.72 \times 10^{-12}$	N <sub>2</sub>
283	$4.44 \times 10^{-12}$	1.347±0.046	$(5.98 \pm 0.20) \times 10^{-12}$	Air
		1.327	$5.89 \times 10^{-12}$	N <sub>2</sub>
293	$4.70 \times 10^{-12}$	1.303±0.035	$(6.12 \pm 0.16) \times 10^{-12}$	Air
		1.280	$6.02 \times 10^{-12}$	N <sub>2</sub>
303	$4.97 \times 10^{-12}$	1.263±0.012	$(6.28 \pm 0.06) \times 10^{-12}$	Air
		1.250	$6.21 \times 10^{-12}$	N <sub>2</sub>
313	$5.22 \times 10^{-12}$	1.222±0.030	$(6.38 \pm 0.16) \times 10^{-12}$	Air
		1.212	$6.33 \times 10^{-12}$	N <sub>2</sub>
323	$5.48 \times 10^{-12}$	1.184±0.029	$(6.49 \pm 0.16) \times 10^{-12}$	Air
		1.170	$6.41 \times 10^{-12}$	N <sub>2</sub>
		3-Methylhexane		
273	$4.17 \times 10^{-12}$	0.746±0.028	$(3.11 \pm 0.12) \times 10^{-12}$	Air
		0.875	$3.65 \times 10^{-12}$	N <sub>2</sub>
283	$4.44 \times 10^{-12}$	0.811±0.172	$(3.60 \pm 0.76) \times 10^{-12}$	Air
		0.868	$3.85 \times 10^{-12}$	N <sub>2</sub>
293	$4.70 \times 10^{-12}$	0.702±0.014	$(3.30 \pm 0.07) \times 10^{-12}$	Air
		0.722	$3.39 \times 10^{-12}$	N <sub>2</sub>
303	$4.97 \times 10^{-12}$	0.701±0.153	$(3.48 \pm 0.76) \times 10^{-12}$	Air
		0.715	$3.143.55 \times 10^{-12}$	N <sub>2</sub>
313	$5.22 \times 10^{-12}$	0.787±0.160	$(4.11 \pm 0.08) \times 10^{-12}$	Air
		0.726	$4.79 \times 10^{-12}$	N <sub>2</sub>
323	$5.48 \times 10^{-12}$	0.824±0.107	$(4.52 \pm 0.06) \times 10^{-12}$	Air
		0.738	$4.05 \times 10^{-12}$	N <sub>2</sub>
		2,2,4-Trimethylpentane		
273	$4.17 \times 10^{-12}$	0.611±0.001	$(2.55 \pm 0.01) \times 10^{-12}$	Air
		0.629	$2.62 \times 10^{-12}$	N <sub>2</sub>
283	$4.44 \times 10^{-12}$	0.634±0.022	$(2.81 \pm 0.10) \times 10^{-12}$	Air
		0.633	$2.81 \times 10^{-12}$	N <sub>2</sub>
293	$4.70 \times 10^{-12}$	0.625±0.006	$(2.94 \pm 0.03) \times 10^{-12}$	Air
		0.632	$2.97 \times 10^{-12}$	N <sub>2</sub>
303	$4.97 \times 10^{-12}$	0.638±0.024	$(3.17 \pm 0.12) \times 10^{-12}$	Air
		0.632	$3.14 \times 10^{-12}$	N <sub>2</sub>

313	$5.22 \times 10^{-12}$	0.633±0.030 0.614	$(3.30 \pm 0.16) \times 10^{-12}$ $3.20 \times 10^{-12}$	Air N <sub>2</sub>
323	$5.48 \times 10^{-12}$	0.616±0.008 0.608	$(3.38 \pm 0.04) \times 10^{-12}$ $3.33 \times 10^{-12}$	Air N <sub>2</sub>
Methylcyclohexane				
273	$4.17 \times 10^{-12}$	1.806±0.005 1.860	$(7.53 \pm 0.26) \times 10^{-12}$ $7.76 \times 10^{-12}$	Air N <sub>2</sub>
283	$4.44 \times 10^{-12}$	1.873±0.018 1.883	$(8.31 \pm 0.08) \times 10^{-12}$ $8.36 \times 10^{-12}$	Air N <sub>2</sub>
293	$4.70 \times 10^{-12}$	1.869±0.017 1.835	$(8.77 \pm 0.06) \times 10^{-12}$ $8.62 \times 10^{-12}$	Air N <sub>2</sub>
303	$4.97 \times 10^{-12}$	1.848±0.001 1.833	$(9.18 \pm 0.01) \times 10^{-12}$ $9.11 \times 10^{-12}$	Air N <sub>2</sub>
313	$5.22 \times 10^{-12}$	1.859±0.019 1.793	$(9.70 \pm 0.10) \times 10^{-12}$ $9.36 \times 10^{-12}$	Air N <sub>2</sub>
323	$5.48 \times 10^{-12}$	1.804±0.022 1.742	$(9.89 \pm 0.12) \times 10^{-12}$ $9.55 \times 10^{-12}$	Air N <sub>2</sub>
2,3,4-Trimethylpentane				
273	$4.17 \times 10^{-12}$	1.522±0.023 1.527	$(6.35 \pm 0.09) \times 10^{-12}$ $6.37 \times 10^{-12}$	Air N <sub>2</sub>
283	$4.44 \times 10^{-12}$	1.477±0.034 1.502	$(6.56 \pm 0.15) \times 10^{-12}$ $6.67 \times 10^{-12}$	Air N <sub>2</sub>
293	$4.70 \times 10^{-12}$	1.435±0.022 1.439	$(6.74 \pm 0.11) \times 10^{-12}$ $6.76 \times 10^{-12}$	Air N <sub>2</sub>
303	$4.97 \times 10^{-12}$	1.377±0.014 1.379	$(6.84 \pm 0.07) \times 10^{-12}$ $6.85 \times 10^{-12}$	Air N <sub>2</sub>
313	$5.22 \times 10^{-12}$	1.355±0.017 1.333	$(7.07 \pm 0.09) \times 10^{-12}$ $6.96 \times 10^{-12}$	Air N <sub>2</sub>
323	$5.48 \times 10^{-12}$	1.296±0.019 1.297	$(7.10 \pm 0.10) \times 10^{-12}$ $7.11 \times 10^{-12}$	Air N <sub>2</sub>
2-Methylheptane				
273	$4.17 \times 10^{-12}$	1.586±0.188 1.654	$(6.61 \pm 0.78) \times 10^{-12}$ $6.90 \times 10^{-12}$	Air N <sub>2</sub>
283	$4.44 \times 10^{-12}$	1.499±0.016 1.563	$(6.66 \pm 0.07) \times 10^{-12}$ $6.94 \times 10^{-12}$	Air N <sub>2</sub>
293	$4.70 \times 10^{-12}$	1.332±0.038 1.409	$(6.26 \pm 0.18) \times 10^{-12}$ $6.62 \times 10^{-12}$	Air N <sub>2</sub>
303	$4.97 \times 10^{-12}$	1.344±0.092 1.335	$(6.68 \pm 0.46) \times 10^{-12}$ $6.64 \times 10^{-12}$	Air N <sub>2</sub>
313	$5.22 \times 10^{-12}$	1.396±0.017 1.318	$(7.29 \pm 0.09) \times 10^{-12}$ $6.88 \times 10^{-12}$	Air N <sub>2</sub>
323	$5.48 \times 10^{-12}$	1.339±0.035 1.312	$(7.34 \pm 0.19) \times 10^{-12}$ $7.19 \times 10^{-12}$	Air N <sub>2</sub>

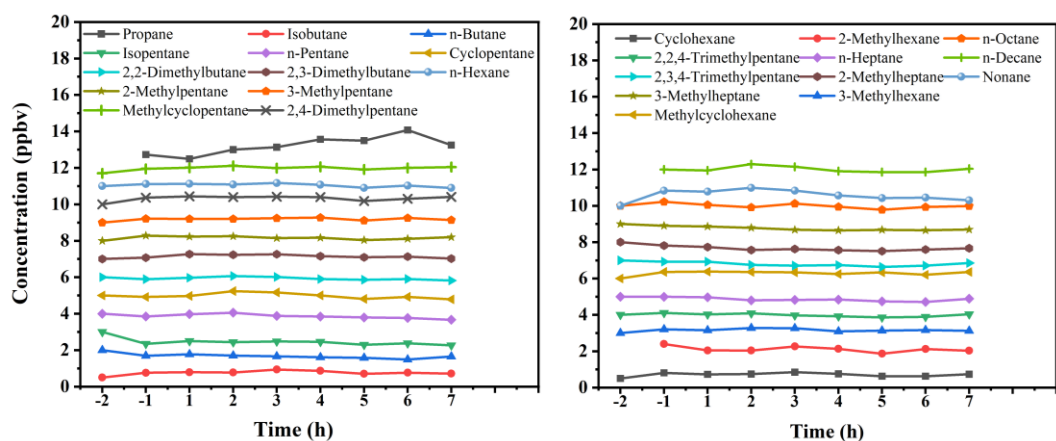


Nonane				
273	$4.17 \times 10^{-12}$	1.882±0.008	$(7.85 \pm 0.04) \times 10^{-12}$	Air
		2.009	$8.38 \times 10^{-12}$	N <sub>2</sub>
283	$4.44 \times 10^{-12}$	1.924±0.003	$(8.54 \pm 0.02) \times 10^{-12}$	Air
		1.982	$8.80 \times 10^{-12}$	N <sub>2</sub>
293	$4.70 \times 10^{-12}$	1.858±0.039	$(8.73 \pm 0.18) \times 10^{-12}$	Air
		1.888	$8.87 \times 10^{-12}$	N <sub>2</sub>
303	$4.97 \times 10^{-12}$	1.909±0.016	$(9.49 \pm 0.08) \times 10^{-12}$	Air
		1.917	$9.53 \times 10^{-12}$	N <sub>2</sub>
313	$5.22 \times 10^{-12}$	1.950±0.167	$(1.02 \pm 0.09) \times 10^{-12}$	Air
		1.858	$9.70 \times 10^{-12}$	N <sub>2</sub>
323	$5.48 \times 10^{-12}$	1.917±0.040	$(1.05 \pm 0.02) \times 10^{-11}$	Air
		1.842	$1.01 \times 10^{-11}$	N <sub>2</sub>
n-Decane				
273	$4.17 \times 10^{-12}$	2.483±0.608	$(1.04 \pm 0.25) \times 10^{-11}$	Air
		2.391	$9.97 \times 10^{-12}$	N <sub>2</sub>
283	$4.44 \times 10^{-12}$	2.237±0.008	$(9.93 \pm 0.04) \times 10^{-12}$	Air
		2.309	$1.03 \times 10^{-11}$	N <sub>2</sub>
293	$4.70 \times 10^{-12}$	2.225±0.181	$(1.05 \pm 0.08) \times 10^{-11}$	Air
		2.232	$1.05 \times 10^{-11}$	N <sub>2</sub>
303	$4.97 \times 10^{-12}$	2.206±0.095	$(1.10 \pm 0.05) \times 10^{-11}$	Air
		2.280	$1.2132 \times 10^{-11}$	N <sub>2</sub>
313	$5.22 \times 10^{-12}$	2.278±0.050	$(1.19 \pm 0.03) \times 10^{-11}$	Air
		2.240	$1.17 \times 10^{-11}$	N <sub>2</sub>
323	$5.48 \times 10^{-12}$	2.261±0.083	$(1.24 \pm 0.05) \times 10^{-11}$	Air
		2.193	$1.20 \times 10^{-11}$	N <sub>2</sub>
n-Undecane				
273	$4.17 \times 10^{-12}$	2.426	$1.01 \times 10^{-11}$	Air
		2.598	$1.08 \times 10^{-11}$	N <sub>2</sub>
283	$4.44 \times 10^{-12}$	2.425±0.032	$(1.08 \pm 0.02) \times 10^{-11}$	Air
		2.488	$1.10 \times 10^{-11}$	N <sub>2</sub>
293	$4.70 \times 10^{-12}$	2.287±0.008	$(1.07 \pm 0.01) \times 10^{-11}$	Air
		2.374	$1.12 \times 10^{-11}$	N <sub>2</sub>
303	$4.97 \times 10^{-12}$	2.288±0.261	$(1.14 \pm 0.13) \times 10^{-11}$	Air
		2.396	$1.19 \times 10^{-11}$	N <sub>2</sub>
313	$5.22 \times 10^{-12}$	2.456±0.086	$(1.28 \pm 0.05) \times 10^{-11}$	Air
		2.337	$1.22 \times 10^{-11}$	N <sub>2</sub>
323	$5.48 \times 10^{-12}$	2.446±0.065	$(1.34 \pm 0.04) \times 10^{-11}$	Air
		2.350	$1.29 \times 10^{-11}$	N <sub>2</sub>

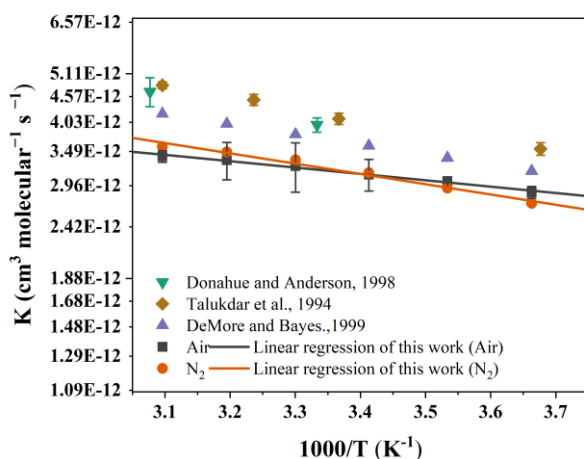
<sup>a</sup><sup>b</sup>The error bar was taken as  $2\sigma$ .



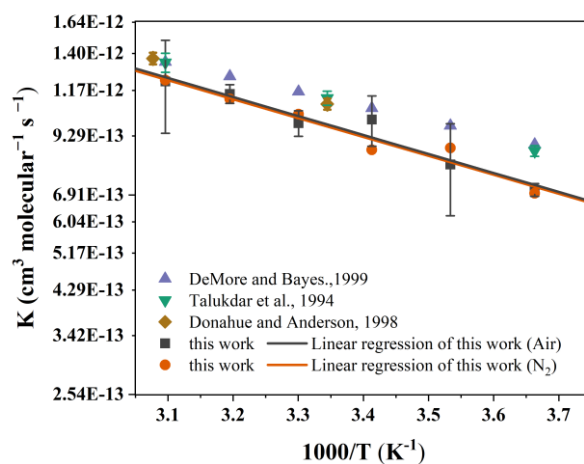
**Figure S1.** (a) Peak height of C<sub>3</sub>-C<sub>11</sub> alkanes in Alkanes+N<sub>2</sub>+dark at 0-14 h. (b) Peak height of C<sub>3</sub>-C<sub>11</sub> alkanes in Alkanes+N<sub>2</sub>+H<sub>2</sub>O<sub>2</sub>+dark at 0-15h. The following data have been displaced for reasons of clarity: a) Propane, Isobutane, n-Butane, Isopentane, 2,2-Dimethylbutane, Methylcyclohexane, 2,2,4-Trimethylpentane, 2,3,4-Trimethylpentane, 2-Methylheptane, 3-Methylheptane, n-Undecane, Ducedane vertically displaced by 8000, 2000, -3000, -6000, -8000, -4000, -4000, -2500, -1000, -4000, -12000, 8000 units, respectively; b) Propane, Isobutane, n-Butane, n-pentane, Cyclopentane, 2,2-Dimethylbutane, 2,3,4-Trimethylpentane, 2-Methylheptane, Nonane, n-Decane, n-Undecane, Ducedane vertically displaced by 8000, 3000, -3000, -2000, -2500, -1000, 1000, 4000, -8000,1000,-1500,17000 units, respectively.



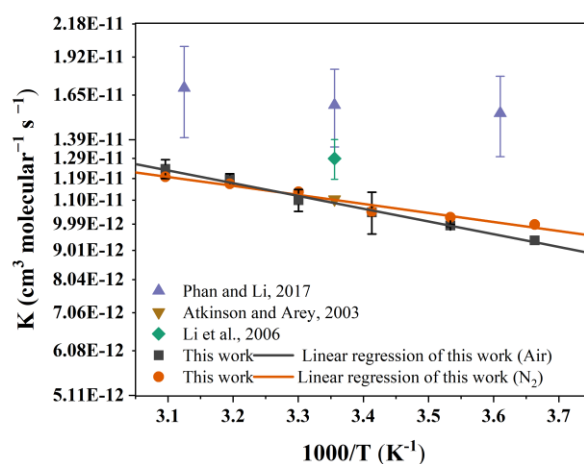
**Figure S2.** Concentration of C3-C11 alkanes in Alkanes+N<sub>2</sub>+UV light at different time. The following data have been displaced for reasons of clarity: Propane, Isobutane, n-Butane, Isopentane, n-pentane, Cyclopentane, 2,2-Dimethylbutane, 2,3-Dimethylbutane, 2-Methylpentane, 3-Methylpentane, n-Hexane, Methylcyclopentane, Cyclohexane, 2-Methylhexane, 3-Methylhexane, 2,2,4-Trimethylpentane, n-Heptane, Methylcyclohexane, 2,3,4-Trimethylpentane, 2-Methylheptane, 3-Methylheptane, n-Decane, vertically displaced by -3, -9.5, -8, -7, -6, -5, -4, -3, -2, -1, 1, 1.7, 9.5, -19, -7, -6, -5, -4, -3, -2, -1, 2 units, respectively. The -2 and -1 of abscissa refer to 2 and 1 hours before turning on the light respectively. 1-7 refers to 1-7 hours after turning on the light respectively.



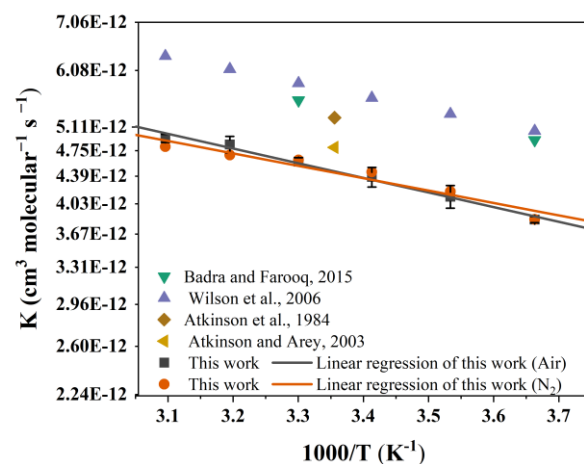
**Figure S3.** Arrhenius plot for the reaction of n-pentane with OH at 273-323 K along with available literature data. The error bar was taken as  $2\sigma$ .



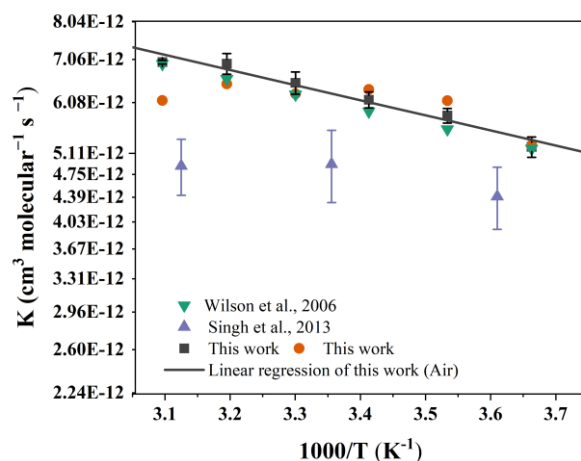
**Figure S4.** Arrhenius plot for the reaction of propane with OH at 273-323 K along with available literature data. The error bar was taken as  $2\sigma$ .



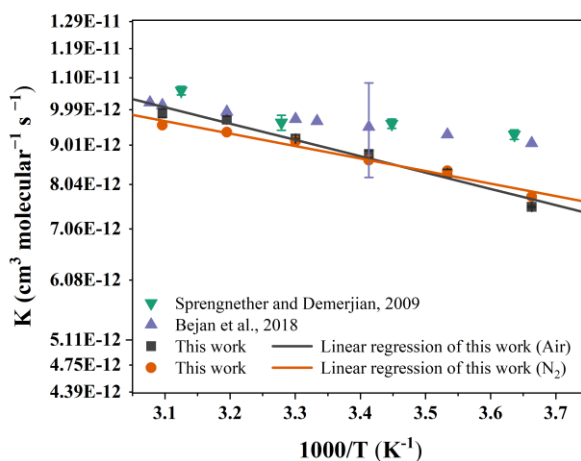
**Figure S5.** Arrhenius plot for the reaction of methylcyclopentane with OH at 273-323 K along with available literature data. The error bar was taken as  $2\sigma$ .



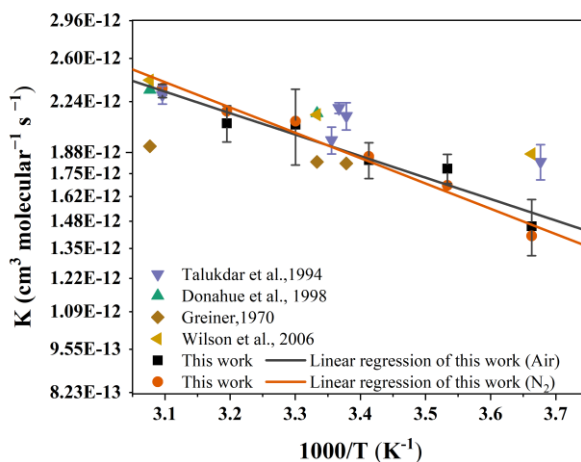
**Figure S6.** Arrhenius plot for the reaction of 2,4-Dimethylpentane with OH at 273-323 K along with available literature data. The error bar was taken as  $2\sigma$ .



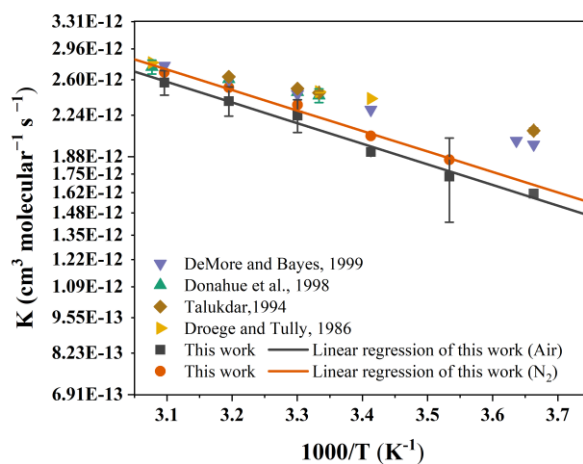
**Figure S7.** Arrhenius plot for the reaction of cyclohexane with OH at 273-323 K along with available literature data. The error bar was taken as  $2\sigma$ .



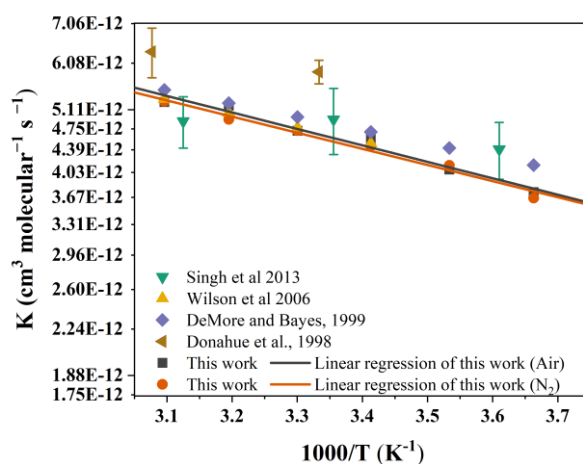
**Figure S8.** Arrhenius plot for the reaction of methylcyclohexane with OH at 273-323 K along with available literature data. The error bar was taken as  $2\sigma$ .



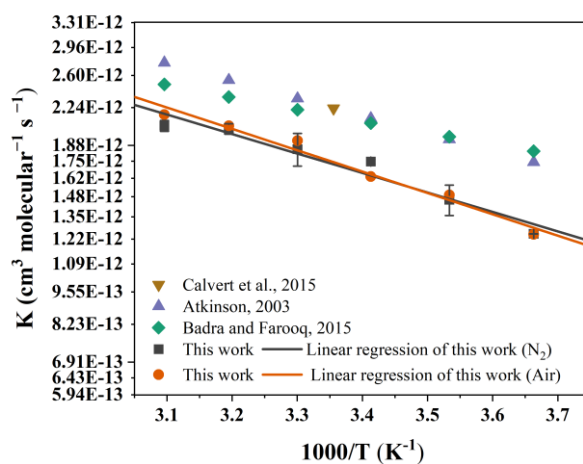
**Figure S9.** Arrhenius plot for the reaction of Isobutane with OH at 273-323 K along with available literature data. The error bar was taken as  $2\sigma$ .



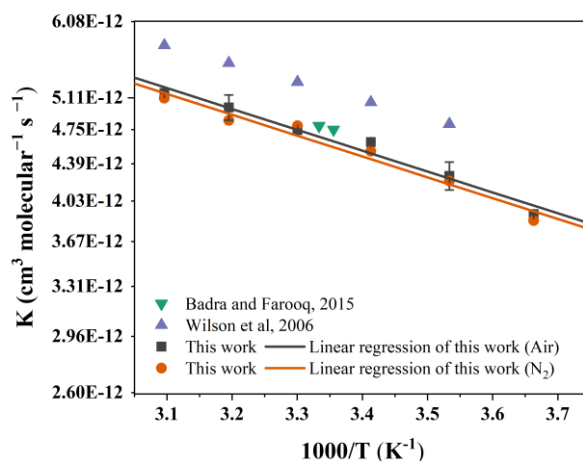
**Figure S10.** Arrhenius plot for the reaction of n-butane with OH at 273-323 K along with available literature data. The error bar was taken as  $2\sigma$ .



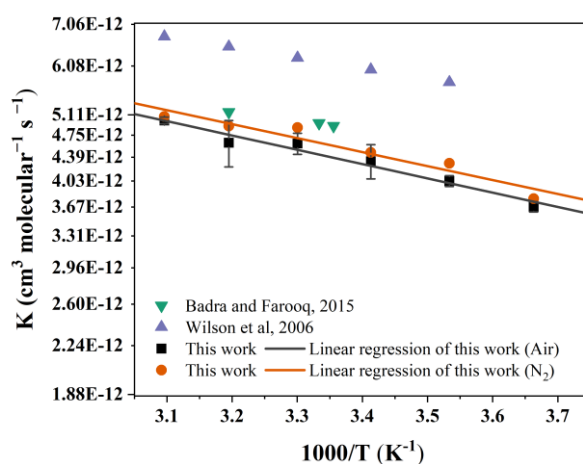
**Figure S11.** Arrhenius plot for the reaction of Cyclopentane with OH at 273-323 K along with available literature data. The error bar was taken as  $2\sigma$ .



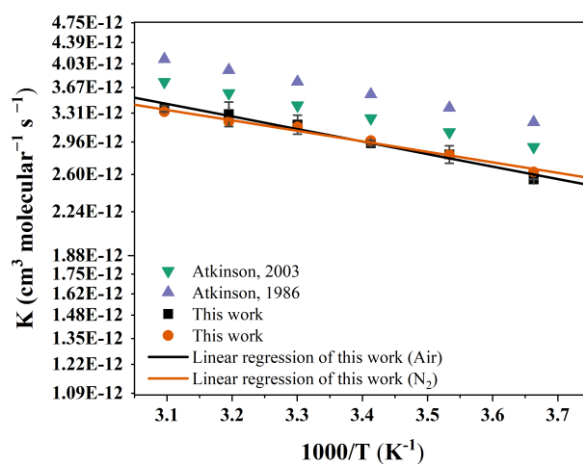
**Figure S12.** Arrhenius plot for the reaction of 2,2-Dimethylbutane with OH at 273-323 K along with available literature data. The error bar was taken as  $2\sigma$ .



**Figure S13.** Arrhenius plot for the reaction of 2-Methylpentane with OH at 273-323 K along with available literature data. The error bar was taken as  $2\sigma$ .



**Figure S14.** Arrhenius plot for the reaction of 3-Methylpentane with OH at 273-323 K along with available literature data. The error bar was taken as  $2\sigma$ .



**Figure S15.** Arrhenius plot for the reaction of 2,2,4-Trimethylpentane with OH at 273-323 K along with available literature data. The error bar was taken as  $2\sigma$ .