# Measurement report: <u>SecondaryImpact of cloud processes on</u> <u>secondary</u> organic aerosols at a forested mountain site in southeastern China

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**Abstract.** Aerosol particles play <u>erucialcritical</u> roles in <u>both-climate dynamics-</u>and human health. However, <u>there remains a</u> significant gap in our <u>understanding of</u> aerosol composition and evolution, particularly <u>regarding</u> secondary organic aerosols (SOA), and <u>theirits</u> interaction with clouds in high-altitude background areas in China-<u>remain less understood</u>. Here we conducted real-time measurements of submicron aerosols (PM<sub>1</sub>) using aerosol mass spectrometers at a forested mountain site

- 20 (1128 m a.s.l.) in southeastern China in November 2022. Our results revealed that organic The average ( $\pm 1\sigma$ ) PM<sub>1</sub> mass concentration was 4.45  $\pm$  6.51 µg m<sup>-3</sup>, which was ubiquitously lower than those at other mountain sites in China. Organic aerosol (OA) constituted a substantial portion the largest fraction of PM<sub>1</sub> (41.1 %), with the OA being primarily of 42.9 %) and was dominantly secondary origin, as evidenced indicated by athe high oxygen-to-carbon (O/C) ratio (0.85–0.96). and carbon oxidation state (0.21–0.49). Back trajectory analysis revealed that higher concentrations of PM<sub>1</sub> were mainly
- 25 associated with the transport from the western and southwestern regions. Notably, the remarkably enhanced PM<sub>1</sub> concentrations observed during daytime on cloudless days were identified to be likely produced from cloud evaporation. Positive matrix factorization resolved two distinct SOAOOA factors: i.e., less oxidized oxygenated OA (LO-OOA) and more oxidized OOA (MO-OOA). Interestingly, While MO-OOA was scavenged efficiently during cloud events, while cloud evaporation contributed significantly was found to release a significant amount of LO-OOA. The ratio from airmass
- 30 transported from polluted regions. The distinct increase of OA/ACO increased/∆CO (CO after subtracting the background level) with athe decrease in theof O/C ratio, suggestingduring the cloud evaporative period further demonstrates that OA remainingremained in cloud droplets are generally maintainedin a moderate oxidation state. Furthermore, our results indicated Moreover, organic nitrates were also estimated and showed a higher contribution of organic nitrates to the total

nitrate during the cloudy periods periods (27 %) compared to evaporative periods (3 %). Notably, a substantial contribution of

35 nitrate in PM<sub>1</sub> (20.9 %) was observed, particularly during high PM periods, implying that nitrate formed in polluted areas interacted with clouds and significantly impacted the regional background site. than the evaporative period (3 %). Overall, our study underscores results demonstrate the importance of understanding the dynamics of secondary organic aerosols SOA and the impacts influences of cloud processing processes in regional mountainous mountain areas in southeastern China.

## **1** Introduction

- 40 Aerosol particles play essential roles in regional and global climate (Ramanathan et al., 2001; Kanakidou et al., 2005), as well as air pollution (Huang et al., 2014) and public health (Kampa and Castanas, 2008). Submicron aerosol (PM<sub>1</sub>) from both natural and anthropogenic sources can be transported to the upper <u>layerlevel</u> of atmospheric boundary layer or even lower free troposphere through convection and frontal uplift (Monks et al., 2009; Huang et al., 2020; Carbone et al., 2014). At high altitudes, aerosol species can be transported over longer distances, depending on their particle sizes and compositions
- 45 (Pokorná et al., 2022; Tang et al., 2016; Zhong et al., 2022), and the aging processes during transport results in continuous changes in chemical and physical properties (Calvo et al., 2013; Hallquist et al., 2009). High-altitude aerosols-Aerosols can serve as cloud condensation nuclei or ice nuclei have an important impact on CCN properties at high altitudes where atmospheric conditions favour the formation of clouds, thereby affecting the lifetime and optical properties of clouds through aerosol-cloud interactions (Haywood and Boucher, 2000). They can also have potential impacts on ground-level air pollution
- 50 through downward mixing (Timonen; Asmi et al., 20132012; Rejano et al., 2021). On the other hand, aerosol-cloud interactions also modify the chemical composition of aerosols within cloud droplets. This modification, in turn, affects the characteristics of surrounding aerosols after the cloud has evaporated (Roth et al., 2016). As a result, measurements of high-altitude regional aerosols are of great importance for a better understanding of aerosol-cloud interactions and their impacts on air pollution and climate.
- 55 Mountain sites are crucial platforms for studying aerosol characteristics over regional scalescales and the influences of diverse emission sources (e.g., biomass burning, industrial process, and biogenic emissions) and atmospheric processes (e.g., in-cloud processing, new particle formation). Compared to balloon and aircraft observations, mountain sites have advantages in continuous and long-term observations- because of low cost, stable geographic location, and accessibility for instrument maintenance. In addition, the different meteorological conditions at mountain sites can have great impacts on aerosol formation, aging, and scavenging patterns. For example, Gao et al. (2023) showed that cloud processes can promote the formation of secondary organic aerosol (SOA) by multiphase oxidation. Li et al. (2013) found that high relative
- humidity (RH) at Mt. Hua can lead to a reduction in particle acidity, and thus reduce the formation of biogenic SOA by suppressing the acid-catalysis reaction. Chen et al. (2021) reported that fog scavenging was more efficient towards particles with aerodynamic diameter larger than 700 nm in Taiwan mountain regions, while smaller particles could continue to grow
- 65 viaremained growth through the process of gas-to-aqueous partition partitioning. Although several studies over mountain

sites have been conducted to characterize chemical composition and sources several mountain stations have been settledestablished in China to gain more information on the chemical composition and sources (Zhang et al., 2018; Zhang et al., 2019; Du et al., 2015; Zhang et al., 2014), optical properties (Wang et al., 2015), and hygroscopic properties of aerosols (Ding et al., 2021), most of themthese studies are mainly conducted on Qinghai-Tibetan Plateau and in Northern China Plain (NCP), while the studies in southeastern China is very emain limited.

- In this work, a high-resolution time-of-flight aerosol mass spectrometer (AMS hereafter) and a quadruple aerosol chemical speciation monitor (ACSM hereafter) were deployed at a forested mountain site in southeastern China. The chemical composition and properties of PM<sub>1</sub> are characterized, the elemental composition and oxidation state of OA are determined, and the potential transport pathways of PM<sub>1</sub> are also investigated using back trajectory analysis. Furthermore, the effect of
- 75 cloud processes on aerosol species is also-discussed. The sources of organic aerosol are investigated using positive matrix factorization (PMF) analysis of combined high-resolution organic and inorganic aerosol mass spectra. Particularly, particleparticulate organic nitrates (ONs) are determined and quantified based on PMF results. Finally, the potential transport pathways of PM<sub>1</sub> are investigated using backward trajectory analysis.



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**Figure 1.** Location of the sampling site- (Mt. Damaojian, red circle on the map). The mean concentration ( $in-\mu g$  m<sup>-3</sup>) and chemical composition of submicron aerosols (NR-PM<sub>1</sub>+ BC if it was available) measured at selected mountain sites in China are also shown. Note that BC and chloride are not accounted for Mt. Tai due to the lack of measurement or data in the relevant study. Detailed information of these sampling sites is presented in Table S1 in the Supplement.

# 2 Methods

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#### 2.1 Site and instrumentation

The campaign was carried out from 1 November to 30 November 2022 at Shanghuang Atmospheric Boundary Layer and Eco-EnvironmentEnvironmental Observatory of Chinese Academy of Sciences (SH-site) on the top of ), located at Mt.

90 Damaojian (119.51°E, 28.58°N, 1128 m a.s.l.) in Wuyi County, Zhejiang Province (Fig. 1). This site is a typical background site in southeastern China that is surrounded by mountains and forests, and there are no strong local anthropogenic sources nearby.

 $PM_1$  species were measured using a suite of real-time instruments with 1-520 min time resolution, including an AMS operated under the "V-mode" and a quadrupole ACSM for non-refractory (NR)-PM<sub>1</sub> composition, together with a seven-

95 wavelength Aethalometer (AE33, Magee Scientific Corp.) for equivalent black carbon (BC) mass concentration. The AMS measurements were only conducted during two periods (15–20 November and 24–28 November, respectively) in this study

due to the malfunction of the instrument. Briefly, aerosol particles were sampled into an air-conditioned room through stainless steel tube (O.D.: 1/4 inch), and the residence time was estimated as 5 s. -A nafion dryer was placed upstream of the ACSM and AMS to remove the moisture, after that, aerosol particles were sampled into AMS, ACSM, and AE33,

- respectively. Because there was no size cutoff in front of the sampling line, the AMS and ACSM may report slightly higher concentrations than expected because of measuring some cloud droplets larger than 2.5 µm. According to a previous study during three fog episodes at a rural site in North China Plain, the influence of fog droplets on submicron aerosol measurements was less than 20% (Kuang et al., 2024). Simultaneously, air pollutants including NO<sub>x</sub>, O<sub>3</sub>, PM<sub>2.5</sub>, and PM<sub>40</sub> were measured by a suite of gas analyzers (Thermo Scientific Inc., USA), PM<sub>2.5</sub> and PM<sub>10</sub> were measured using continuous ambient particulate monitors with PM<sub>2.5</sub> and PM<sub>10</sub> cutoff before the sampling inlet (Model 5014i, Thermo Scientific., USA), and CO was measured by a Picarro greenhouse gas analyzer (G2401, Picarro Inc., USA). In addition, meteorological parameters containing temperature (*T*), RH, wind speed (WS), wind direction (WD), and pressure (*P*) were measured at the same siteby an automatic weather station. In addition, (PDR). More details and descriptions of the instruments and data are provided in Table S1.
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#### 2.2 Data analysis

# 2.2.1 ACSM and AMS

ACSM data were analyzed using ACSM standard data analysis software (v2.5.13) and AMS data were analyzed using SQUIRREL v1.65F and PIKA v1.25F. A composition-dependent collection efficiency (CDCE) was applied to the ACSM/AMS data according to Middlebrook et al. (2012). Elemental analysis of high-resolution mass spectra (HRMS) was performed using the "Improved-Ambient" (I-A) method (Canagaratna et al., 2015). The default relative ionization efficiency (RIE) values of 1.1, 1.4, and 1.3 were applied for nitrate, organics, and chloride- (Canagaratna et al., 2007; Nault et al., 2023). According to the ion efficiency (IE) calibration results using ammonium sulfate, the RIE values of ammonium and sulfate were 5.05 and 0.73 for ACSM, and 5.26 and 1.28 for AMS, respectively. Detailed comparison of the concentrations of NR-PM1 species measured by AMS and ACSM are shown in Fig. S1. The ACSM data during the sampling period was corrected by using the regression coefficients between ACSM and AMS. As shown in Fig. S1g and h, after this adjustment, the concentrations of NR-PM1 tracked well with PM2.5 and PM40 measured by gas analyzers (r<sup>2</sup> = 0.7060 and slope = 0.66,48 for ACSM, r<sup>2</sup> = 0.93 and slope = 0.54 for AMS respectively) and PM10 ((r<sup>2</sup> = 0.53 and slope = 0.27 for ACSM, r<sup>2</sup> = 0.99 and slope = 0.39 for AMS respectively) measured by the particle monitor, suggesting that the AMS/ACSM quantification was

<sup>125</sup> reasonable.

PMF Evaluation Tool (PET v3.04) was employed to further deconvolve the HRMS <u>derived from AMS</u> into different source factors following the procedures reported by Ulbrich et al. (2009) and Zhang et al. (2011). In addition to organic fragment ions, the major fragment ions of inorganic species, i.e., SO<sup>+</sup> (m/z 48), SO<sub>2</sub><sup>+</sup> (m/z 64), SO<sub>3</sub><sup>+</sup> (m/z 80), HSO<sub>3</sub><sup>+</sup> (m/z 81), H<sub>2</sub>SO<sub>4</sub><sup>+</sup>

(*m/z* 98) for sulfate, NO<sup>+</sup> (*m/z* 30), NO<sub>2</sub><sup>+</sup> (*m/z* 46) for nitrate, NH<sup>+</sup> (*m/z* 15), NH<sub>2</sub><sup>+</sup> (*m/z* 16), NH<sub>3</sub><sup>+</sup> (*m/z* 17) for ammonium,
and Cl<sup>+</sup> (*m/z* 35), HCl<sup>+</sup> (*m/z* 36) for chloride were also included into the HR data and error matrices for PMF. A more detailed description of the procedures was given in Sun et al. (2012). After checking the key diagnostic plots (Fig. S2), mass spectra, and the correlations with related tracers, a four-factor solution was considered as the optimal solution in this study-(Text S1).

## 2.2.2 Estimation of organic nitrates

- 135 ONs were estimated from the PMF results (Zhang et al., 2011; Xu et al., 2015). Briefly, NO<sub>x</sub><sup>+</sup> (i.e., NO<sup>+</sup> and NO<sub>2</sub><sup>+</sup>) are major fragments of nitrate functionality (-ONO<sub>2</sub>), which can be referred to as the total nitrate measured by AMS. Combining inorganic with organic mass spectra in PMF, NO<sup>+</sup> and NO<sub>2</sub><sup>+</sup> can be separated into different organic aerosol (OA) factors and an inorganic nitrate aerosol factor (NIA). According to previous studies, the ratios of NO<sup>+</sup>/NO<sub>2</sub><sup>+</sup> for ONs are approximately 2.25–3.7 times higher than pure NH<sub>4</sub>NO<sub>3</sub> (Fry et al., 2013; Fry et al., 2009). Consistently, the PMF results in our study (Fig.
- 140 7) show that the average NO<sup>+</sup>/NO<sub>2</sub><sup>+</sup> ratios of LO-OOA and MO-OOA were 13.19 and 11.2, falling within the range of ONs. In contrast, a NO<sup>+</sup>/NO<sub>2</sub><sup>+</sup> ratio of 3.56 was observed for NIA, reflecting its characteristics of inorganic nitrates. Therefore, the mass concentration of ONs (NO<sub>3,org</sub>) wasTherefore, the PMF analysis of nitrate is reasonable, and based on this result, the mass concentration of ONs (NO<sub>3,org</sub>) can be calculated by summing these two ion signals distributed in all OA factors as follows:

$$145 \quad NO_{3,org} = NO_{org}^{+} + NO_{2,org}^{+}$$
(1)

$$NO_{org}^{+} = \sum ([OA \ factor]_i \times f_{NO^{+},i})$$
(2)

$$NO_{2,org}^{+} = \sum ([OA \ factor]_i \times f_{NO_2^+,i})$$
(3)

where [OA factor]<sub>*i*</sub> represents the mass concentration of OA factor *i* resolved by PMF,  $f_{NO^+,i}$  and  $f_{NO^+_2,i}$  are the mass fractions of NO<sup>+</sup> and NO<sub>2</sub><sup>+</sup> in OA factor *i*, respectively.

# 150 2.2.3. Backward trajectory analysis

The Hybrid Single-Particle Lagrangian Integrated Trajectories (HYSPLIT) model and meteorological data from the NOAA Global Data Assimilation System (GDAS) were used to calculate 72 h backward trajectories at the SH site. The trajectory <u>arrival</u> time was set from 0:00 to 23:00 at 1 h intervals, and the <u>arrival</u> height <u>at the site</u> was set as 1100 m. To further show the <u>aerosol</u> particle concentration levels in different regions, the map <u>was colored</u> contained the back trajectory (Fig. S5) was

155 <u>coloured</u> by the time-averaged organic carbon surface mass concentration (ENSEMBLE) from the <u>M2T1NXAER v5.12.4</u> dataset (0.5 × 0.625°, hourly). This dataset, part of the Modern-Era Retrospective analysis for Research and Applications version 2 (MERRA-2) model, which (Gelaro et al., 2017), was obtained sourced from the NASA Giovanni website (<u>http://giovanni.sei.gsfe.nasa.gov/giovanni).https://giovanni.gsfc.nasa.gov).</u>

## 160 3.1. General descriptions

# 3.1. Mass concentration and chemical composition of submicron aerosols

- Figure 2 shows the time series of meteorological parameters (T, RH, WD, and WS), air pollutants (PM<sub>2.5</sub> and CO) along with  $PM_1$  (NR-PM<sub>1</sub> + BC) species during the campaign. The total  $PM_1$  concentration varied dynamically from 0.3  $\mu$ g m<sup>-3</sup> to  $57.9539.8 \ \mu g \ m^{-3} \ during the whole sampling period, with an average (\pm 1\sigma) of <math>4.45 \pm 6.513 \pm 4.8 \ \mu g \ m^{-3}$ . As shown in Fig. 1, 165 organics heldaccounted for the largest contribution (42.9 %) to the total PM<sub>1</sub> during the sampling period (41.1 %), followed by nitrate (20.9 sulfate (16.7 %), ammonium (1715.0 %), sulfate (11.0 nitrate (14.7 %), BC (9.010.2 %), and chloride (4.0.5 %). In addition, BC was observed to correlate well with PM<sub>1</sub> and PM<sub>2.5</sub> ( $r^2 = 0.59$  and  $0-\frac{9}{20}$ .62, respectively), suggesting that the aerosol particles at this site may be largely influenced by the regional biomass burning plumes in the Yangtze River Delta (Zhang et al., 2015). The concentration and composition of  $PM_1$  are quite different from those observed at other Chinese mountain sites with similar altitudes in different seasons (Fig. 1), such as Mt. Wuzhi, where PM1 has a mean 170 mass concentration of 10.9 µg m<sup>-3</sup>, and sulfate makes up a significant part (30.9 %) of total PM<sub>4</sub> (Zhu et al., 2016). Note that the 1). For example, the average PM<sub>1</sub> concentration at SH site is even here is much lower compared to Mt. Wuzhi (10.9  $\pm$  7.8 µg m<sup>-3</sup>) at a similar altitude, and is also lower than those at mountain sites with at higher attitude (Fig. 1), altitude mountains such as Mt. Tai (34.3 Yulong (5.7  $\pm$  5.4  $\mu$ g m<sup>-3</sup>) in NCP (Zhang et al., 2014), and Mt. Yulong (5.4  $\mu$ g m<sup>-3</sup>) (Zheng et al., 2017), 175 and Mt. Waliguan (9.1  $\pm$  5.3 µg m<sup>-3</sup>). Although PM<sub>1</sub> concentration has a strong seasonal dependence, such as in Mt. Tai, with low concentration in autumn (18.1  $\mu$ g m<sup>-3</sup>) (Zhang et al., and high in summer (33.5  $\mu$ g m<sup>-3</sup>), the average PM<sub>1</sub> concentration at our site is still much lower than Mt. Daban (11.4 µg m<sup>-3</sup>) and Mt. Tai in the same season (autumn). 2019). Considering the occurrence of frequent cloud events during the sampling period, the low PM<sub>1</sub> concentration might be mainly associated with cloud scavenging (Kim, which will be further validated in subsequent sections. Notably, a relatively lower contribution of 180 sulfate to  $PM_1$  was observed (16.7 %) in this site compared to other mountain sites (14.0 %–38.1 %), which was likely attributed to the significant reduction of SO<sub>2</sub> emission in China during the past decade (Wen et al., 2023). The decrease in sulfate contribution was associated with an elevation-et al., 2019). Indeed, the unexpectedly high fraction of nitrate at the regional background site might indicate contribution. This shift suggests that most NH<sub>4</sub> will probably be in the form of ammonium nitrate formed from which is more volatile than ammonium sulfate and thus unlikely to transport over a large scale. Notably, at Mt. Tai, even though the contributions of sulfate were high, nitrate still accounted for considerable 185 fractions (14.9 %-29.0 %) that were comparable to this site (20.9 %). Taken together, PM<sub>1</sub> at this site is more likely influenced by anthropogenic-emitted NO<sub>x</sub>-can have a significant impact on emissions over a smaller regional scale-in the
- southeastern China. One explanation was that nitrate formed in polluted regions interacted with clouds and affected the regional nitrate level as cloud evaporates (Tao et al., 2018)., consistent with the high urban density in eastern China.
   190 Figure 33a shows the relative contribution of each PM<sub>1</sub> component as a function of PM<sub>1</sub> concentration together with the
- probability density of PM<sub>1</sub> mass loading. The highest two probabilities frequencies of PM<sub>1</sub> concentrations were distributed

within  $0-\underline{63} \ \mu g \ m^{-3}$  and  $\underline{12} \ \underline{185-8} \ \mu g \ m^{-3}$  (48.529.0 % and  $\underline{30.129.8} \$ %, respectively). The fraction of nitrate increased significantly with PM<sub>1</sub> concentration, and meanwhile, the fraction of organics and BC exhibited a decreasing trend. This result suggested that high levels of PM<sub>1</sub> at the SH site might be mainly attributed to the nitrate formation and transport. It is

- 195 also We noted that a nitrate-dominant peak of PM<sub>1</sub> mass loading was observed at the nighttime of 4 November (Fig. 2d), which was associated with the corresponding increase in CO, further emphasizing the contribution of transport to nitrate EP1, Fig. 2d), and the concentrations of organic and nitrate as a function of PM<sub>1</sub> mass during this event and the rest of the campaign are shown in Fig. S3. Almost all the data points with PM<sub>1</sub> concentrations above 20 µg m<sup>-3</sup> were from this event. During EP1, a steeper slope for nitrate relative to PM<sub>1</sub> was found than that for organics, which was contrary to slopes during
- 200 the rest period (Fig. S3). These distinct differences in slopes for nitrate and organics implied different mechanisms of PM<sub>1</sub> elevation during these two periods. Consequently, we also excluded EP1 from the statistic of Fig. 3a, and the result is presented in Fig. 3b. After removing EP1, organics became the dominant contributor (>40 %) across all PM<sub>1</sub> concentrations. Despite this, there was still an increasing trend for nitrate and a decreasing trend for sulfate along with increased PM<sub>1</sub>, yet the contribution of ammonium remained relatively constant. This further supported our previous hypothesis that there was a
- 205 conversion of ammonium sulfate to ammonium nitrate with the increase in  $PM_1$  concentration. Overall, these results suggested that high levels of  $PM_1$  at the SH site might be mainly attributed to the formation or transport of organics, meanwhile, nitrate also plays a nonnegligible role.





**Figure 2.** Time series of (a) *T* and RH; (b) WD coloured by WS; (c) mass concentration of PM<sub>2.5</sub> and mixing ratio of CO; (d) mass concentrations of NR-PM<sub>1</sub> species measured by ACSM together with BC measured by AE33; and (e, f) mass concentrations and contributions of NR-PM<sub>1</sub> species measured by AMS. The yellow and purple shaded areas represent six selected episodes (EP1–EP5 and C1, respectively).





**Figure 3.** Variations of aerosol composition as a function of  $PM_1$  mass concentration and the probability density of  $PM_1$  during (a) the entire <u>campaignstudy period and</u> (b) the period excluding EP1.

The diurnal cycles of PM<sub>1</sub> species, air pollutants, and meteorological parameters during the entire campaign are illustrated in Fig. S3. The mean and median values of organics and nitrate both showed distinct noon peaks at around 14:00, which could be attributed to the daytime photochemical production and the low wind speed (Tang et al., 2022; Xu et al., 2018b). In addition, these noon peaks may also be related to the development of atmospheric boundary layer, which facilitates the mixing of air masses transported from polluted areas. The high nighttime peak of nitrate was mainly due to the influences of the nitrate event on 4 November. Comparatively, sulfate, chloride, and BC showed relatively flat diurnal variations, suggesting the regional characteristics of these species (Zhang et al., 2015).

#### 3.2 Comparisons of two different periods

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2019). Figure S4 shows the variation of organics, nitrate, and sulfate mass concentrations as a function of RH during the entire campaign. Overall, the mass concentrations of organics decreased significantly with increasing RH, while nitrate and sulfate only showed slight decreases. Previous studies have shown that aerosol mass generally increases on foggy days (Chen et al., 2021). This phenomenon could be due to the cloud scavenging effect under high RH at this site. Another explanation is that submicron aerosols grow to larger sizes under high RH that AMS aerodynamic lens cannot transmit (Chakraborty et al., 2016). To further investigate aerosol characteristics under different meteorological Meteorological conditions, we selected two periods (Fig. 2, denoted as P1 and P2 hereafter) with largely different humidity and PM<sub>4</sub> concentrations. P1 was a typical cloud scavenging period with relative humidity remaining to be saturated (RH = 100 %), and meanwhile, PM<sub>4</sub>-stayed at a very low level, with an average value of 1.34 µg m<sup>-3</sup>. Whereas in P2, the mean RH decreased to 93.9 % with almost no cloud event, and a high PM<sub>4</sub>-event

RH plays a crucial role in determining the composition, formation, and evolution of PM<sub>1</sub> (Sun et al., 2013; Xu et al.,

happened with a maximum mass concentration of 20.65 μg m<sup>-3</sup> (on average 9.39 μg m<sup>-3</sup>). Then, the PM<sub>1</sub> concentration
 decreased rapidly due to the enhanced mountain-valley wind (WS > 4 m s<sup>-1</sup>) and increasing RH (71 % to 100 %)
 during the nighttime. Therefore, P2 was more likely to be a cloud evaporation period, where aerosol particles were
 released from the cloud droplets after water evaporation (Fanourgakis et al., 2019). and air mass origin

The entire study period was characterized by five PM<sub>1</sub> episodes (EP1–EP5), as marked in Fig. 2. Meanwhile, a clean period (C1) with low PM<sub>1</sub> levels was also selected for comparison. The relationship between the PM<sub>1</sub> species concentration, meteorology, and air mass transport during these six periods is shown in Fig. 4 and Fig. S5. EP1 exhibited a nitrate-dominant PM<sub>1</sub> peak, with nitrate concentration rapidly increased by ~18 µg m<sup>-3</sup> within an hour. In addition, a CO peak was also observed at that time, suggesting potential contributions of pollution transport to nitrate. Given the lack of anthropogenic activities near the sampling site, the enhanced nitrate concentrations likely originated from NO<sub>x</sub> transported from adjacent towns in the northeast, as indicated by the nitrate wind polar plot (Fig. S6) showing high levels of nitrate mainly associated

- 250 with northeastern wind directions. This was further supported by the back trajectory analysis, in which air masses on 4 Nov were also from the northeast. Moreover, the RH was stable at 100% during EP1, which could facilitate the heterogenous uptake of N<sub>2</sub>O<sub>5</sub> and subsequent nitrate formation in the aqueous phase at night (Brown et al., 2006; Li et al., 2020). In contrast to EP1, PM<sub>1</sub> in other episodes (EP2–5) were all dominated by organics. EP2 was associated with low
- concentrations of CO, indicating relatively slight pollution transport influence. In addition, clear and similar diurnal variation
- 255 patterns were found in aerosol species and RH, suggesting that PM<sub>1</sub> in this period was mainly affected by the evolution of the atmospheric boundary layer. EP3 was initiated by an organic-only increase under high temperature (~15–25 °C) and low RH (~40 %–80 %), while other aerosol species remained at very low levels. Considering the low CO levels in this period, the increasing trend of OA could be attributed to the local biogenic emissions. The concentrations of organic, together with other PM<sub>1</sub> species and CO, were further elevated after airmass sources shifted from southwest to northwest on 13 Nov,
- 260 which may bring a large amount of aerosols from megacities such as Zhengzhou and Hangzhou to the SH site (Fig. S5c). EP4 and EP5 were two similar episodes, where PM<sub>1</sub> concentrations were generally associated with the changes in RH and WD. During these two periods, air masses both originated from the west and southwest, but the trajectory distances during EP5 were shorter, suggesting that the air masses transported more slowly in EP5.

Overall, although these episodes had different mechanisms of PM<sub>1</sub> variation, they were mostly influenced by the transport of

- 265 western and southwestern air masses, indicating a substantial impact of anthropogenic emissions from big cities like Nanchang and Fuzhou. These areas are marked by higher pollution according to the organic carbon distribution in Fig. S5. However, we have also identified similar transport pathways and even higher CO concentrations during C1 (0.37 ppm on average) than EP2-5 (0.18-0.31 ppm on average), while the PM<sub>1</sub> concentration stayed at very low levels for the whole period. It is worth noting that PM<sub>1</sub> peaks in EP2-EP5 all occurred along with RH below 100 %. When the RH returned to
- 270 100 %, PM<sub>1</sub> concentrations gradually diminished to levels comparable to those observed during the clean period (C1). Furthermore, as shown in Fig. S7, there were notable reductions in the mean and median mass concentrations of all PM<sub>1</sub> species over the entire campaign, ranging between -2.6 %-44.4 % and 2.8 %-50.1 % when RH reached 100 % from conditions of lower RH, respectively. Considering the frequently occurring cloud events at this site, it was most likely that these variations of RH were affected by the cloud process, which may further play an important role in PM<sub>1</sub> concentrations.
- 275 Figure S8 shows the particle depolarization ratio measured by the particle lidar during EP5 and C1. During EP5, clouds were identified at  $\sim$ 1 km above the sampling site from 2:00 to 12:00 on 26 Nov (Fig. S8a). As expected, these clouds gradually disappeared afterward, possibly owing to strong solar radiation at noon on the mountaintop causing cloud droplet evaporation. Meanwhile, a large amount of aerosol particles (PDR =  $\sim$ 0.15) was released from the clouds, which exhibited a strong agreement with the timing of the PM<sub>1</sub> peak. These particles were then scavenged by the increased RH and strong wind
- 280 (WS > 4 m s<sup>-1</sup>). This cloud evaporation phenomenon was also found in urban Guangzhou, which caused remarkably enhanced nitrate mass concentration at noon (Tao et al., 2018). In contrast, no such evaporation process was observed during C1 (Fig. S8b). Instead, high PDR values were found generally at around 0 km, which means that clouds constantly existed near the sampling area during this period, coinciding with the constant 100% RH. Therefore, PM<sub>1</sub> transported to this site was likely to be scavenged by the cloud during C1. These findings were also evidenced by the photos taken at this site at local
- 285 time ~15:00 on 19 Nov and 26 Nov, where severe cloud cover was observed in C1 (Fig. S9a), yet the weather was generally sunny and cloudless in EP5 (Fig. S9b). As a result, besides regional transport, cloud processes can also have significant impacts on aerosol particles at this site.



Figure 4. Temporal variations of NR-PM<sub>1</sub> species measured by ACSM, as well as CO, RH, WD, and WS during six ease events.

# 3.3 Impacts of cloud processes on submicron aerosols

To further explore the impacts of cloud scavenging and cloud evaporation on aerosol characteristics, we select C1 and EP5, in which NR-PM<sub>1</sub> were also measured by AMS, for further investigation in this section.

## 3.23.1 Size distributions and composition

- 295 The average chemically resolved size distributions of NR-PM<sub>1</sub> during P4C1 and P2EP5 are shown in Fig. 45. Generally, all species were distributed in accumulation mode in both periods. However, a smaller peak size (300–500 nm) and broader size distribution were observed during P4C1 when compared to those of P2EP5 (~700 nm), probably owing to the wet removal of larger and hygroscopic particles in P4C1 (Ge et al., 2012). AmmoniumAlso, the complex and broad size distribution observed in C1 suggests that these smaller particles are likely externally mixed with organics, which may further imply the
- 300 potential formation of SOA from local biogenic sources. In contrast, the uniform size distribution across all particulate species in EP5 indicates their well-mixed and highly aged characteristics. Moreover, ammonium showed similar size distribution with sulfate and nitrate during P2EP5, while in P4C1, it was mixed mainly with sulfate and exhibited a quite different size distribution with nitrate, indicating the potential contribution of organic nitrate.

Clear differences were also found in aerosol composition during P4C1 and P2EP5. Organics were the dominant contributor 305 to total NR-PM1 mass during P4C1 (68 %), followed by nitrate (12 %), ammonium (11 %), and sulfate (8 %). In contrast, despite the decreased contribution of organics (51 %), the mass fractions of sulfate and nitrate increased considerably during P2 (17-EP5 (by 9.4% and 19-7.0%, respectively). These changes in the mass fraction of species were attributed to the lower mass scavenging efficiency of organics than inorganic species (Gilardoni et al., 2014). The average HRMS of OA during P1C1 and P2EP5 are presented in Fig. \$5\$10. The HRMS of OA were quite similar for the two periods, with a significant peak *m/z* 44 (mainly CO<sub>2</sub><sup>+</sup>). The OA was highly oxidized, with C<sub>x</sub>H<sub>y</sub>O<sub>1</sub><sup>+</sup> dominating the total OA in P1C1 and P2EP5 by 41 % and 40 %, followed by C<sub>x</sub>H<sub>y</sub><sup>+</sup> (31 % and 32 %), and C<sub>x</sub>H<sub>y</sub>O<sub>2</sub><sup>+</sup> (20 % and 19 %). The contributions of the two major oxygencontaining ion fragments (C<sub>x</sub>H<sub>y</sub>O<sub>1</sub><sup>+</sup> and C<sub>x</sub>H<sub>y</sub>O<sub>2</sub><sup>+</sup>) at the SH site were much higher than those at various urban or suburban sites in China, such as 37.4 % in urban Nanjing (Wang et al., 2016), and 52.9 % in suburban Lanzhou (Tang et al., 2022). Note that higher fraction of CO<sub>2</sub><sup>+</sup> (3 % higher) was found in P4C1 than P2EP5 (Fig. \$5\$e\$10c), indicating a higher oxidation degree of OA in P4C1 (Xu et al., 2014). This is consistent with the higher O/C and OSc in C1 (0.96 and 0.49) in P1-than

**P2EP5** (0.85 and 0.21).



**Figure 45.** Averaged size distributions and chemical composition of NR-PM<sub>1</sub> during (a) P4<u>C1</u> and (b) P2<u>EP5</u>. 320

## 3.3.2.2 Elemental ratios

Figure <u>56</u> shows the time series and box plots of N/C, H/C, OM/OC, O/C, and carbon oxidation state (OSc) of OA, as well as two ion ratios at the SH site. The hourly averaged ratios were only reported for periods with OA concentrations above 0.7  $\mu$ g m<sup>-3</sup>. The average N/C ratios were 0.018 for <u>P4C1</u> and 0.027 for <u>P2EP5</u>, consistent with the higher fraction of C<sub>x</sub>H<sub>y</sub>N<sub>p</sub><sup>+</sup> in

- 325 OA during P2EP5. OA had a slightly lower H/C ratio (1.44 vs. 1.48) and higher average ratios of O/C (0.96 vs. 0.85), OM/OC (2.42 vs. 2.29), and OSc (0.49 vs. 0.21) during P1C1 than those during P2EP5, indicating more oxidized OA during P1C1. The O/C ratios were overall within the range of  $0.94 \pm 0.18$  at regional background sites (Zhou et al., 2020), yet much higher than those observed at urban and suburban sites. These results suggest that OA at the SH site was relatively well-aged. The ratio of fragment ions NO<sup>+</sup> (m/z 30) to NO<sub>2</sub><sup>+</sup> (m/z 46) is a good indicator for identifying the presence of ONs (Farmer et
- al., 2010; Lin et al., 2021). The mean ratio of NO<sup>+</sup>/NO<sub>2</sub><sup>+</sup> in P4C1 was 4.9, which exceeds the value of 3.88 for pure ammonium nitrate (NH4NO<sub>3</sub>-(AN) obtained from the AMS IE calibration, indicating a potential contribution of ONs. Conversely, the mean NO<sup>+</sup>/NO<sub>2</sub><sup>+</sup> in P2EP5 (3.9) was almost identical to that of ANNH4NO<sub>3</sub>, implying the dominance of inorganic nitrates (INs). Additionally, the CH<sub>2</sub>SO<sub>2</sub><sup>+</sup> (*m/z* 79) and CH<sub>3</sub>SO<sub>2</sub><sup>+</sup>(*m/z* 80) ions were used as signature fragments of methanesulfonate, a typical organic sulfur species generated in marine and remote coastal regions (Chen et al., 2019).
  However, the mean ratios of CH<sub>3</sub>SO<sub>2</sub><sup>+</sup> and CH<sub>2</sub>SO<sub>2</sub><sup>+</sup> during P4C1 and P2EP5 (0.98 and 1.99, respectively) were lower than the value of 2.9 reported in previous studies for MSA, indicating the minimal contribution of MSA (Song et al., 2019). The low contributions of MSA further suggest that aerosols over this region are likely to have negligible oceanic influences. This is also consistent with the back trajectory results in section

3.2, which demonstrate that PM<sub>1</sub> levels at this site are dominantly influenced by continental sources in the west and southwest.





Figure 56. Time series and box plots of 1-hour averaged (a) N/C and H/C, (b) OM/OC, OSc, and O/C, and (c) NO<sup>+</sup>/NO<sub>2</sub><sup>+</sup> and CH<sub>3</sub>SO<sub>2</sub><sup>+</sup>/CH<sub>2</sub>SO<sub>2</sub><sup>+</sup>-<sup>+</sup> during the AMS sampling site (left) and their mean values in C1 and EP5 (right). Only ratios determined with good S/N (i.e., organics > 0.7 μg m<sup>-3</sup>) are shown.

<u>The horizontal black and blue dash line in the bottom right plot represent the  $CH_3SO_2^+/CH_2SO_2^+$  value for pure MSA (2.9) and the NO<sup>+</sup>/NO<sub>2</sub><sup>+</sup> value for pure NH<sub>4</sub>NO<sub>3</sub> (3.88), respectively.</u>

## 350 3.4 Source apportionment of OA and contribution of organic nitrates

Four factors were resolved by PMF, including three types of SOA and one inorganic factor: less oxidized oxygenated OA (LO-OOA), more oxidized oxygenated OA (MO-OOA), OA associated with sulfate ions (SO<sub>4</sub>-OA), and inorganic nitrate aerosol (NIA). These four factors together on average accounted for 87.5 % of the total NR-PM<sub>1</sub> mass. The mass spectra profiles and OA ion family composition of the four factors are shown in Figs. <u>67</u> and <u>78</u>.

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Figure 67. High-resolution mass spectral profiles (left) and time series (right) of four factors. The correlations of four factors with corresponding tracers are also shown.



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Figure 78. Mass fractional composition of OA ion families for the resolved four PMF factors.

LO-OOA and MO-OOA were identified by prominent peaks of  $CO^+$  and  $CO_2^+$  in the mass spectra. LO-OOA had a high fraction of  $C_2H_3O^+$  (*m/z* 43, 6.8 % of the total signal), while MO-OOA, corresponding to more oxidized and aged components, had a higher abundance of  $CO_2^+$  (17.6 % vs. 12.8 %),  $C_xH_yO_2^+$  (23 % vs. 19 %), and O/C ratio (1.09 vs. 0.76) 365 than LO-OOA. The average O/C ratios of these two OOA factors are similar to those at Mt. Bachelor Observatory, where the O/C ratios of SV-OOA and LV-OOA were 0.67 and 1.17, respectively (Zhou et al., 2019). LO-OOA correlated tightlycorrelate well with  $C_2H_3O^+NO_3$  (r<sup>2</sup> = 0.95), and 74), coinciding with their secondary nature. Meanwhile, MO-OOA showed exhibit a moderate correlation with sulfate BC ( $r^2 = 0.56$ ), 52), indicating that they were likely coated on BC particles 370 and underwent extensive aging processes during long-range transport. Overall, the LO-OOA and MO-OOA components account for 39.1 % and 29.4 % of the total measured OA mass, respectively (Fig. S6S11). The mass spectra of the SO4-OA factor had a large amount of  $NH_x^+$  and  $SO_x^+$ , together accounting for 48.5 % of the total mass of this factor. Meanwhile, organic components also made up a considerable fraction, yet on average contributing 51 %. It is also noticed that the O/C ratio of the organic fraction of this factor is 0.98, which is even higher than LO-OOA and comparable to MO-OOA, suggesting that this factor has experienced aging processes during the regional transport. Not surpriseAs expected, no 375 primary OA factor (e.g., hydrocarbon-like OA, biomass burning OA, etc.) was resolved during this study due to the negligible influences of local emissions, which was consistent with the PMF results at other background sites (Zhou et al., 2019; Zhu et al., 2016).

AAn inorganic nitrate aerosol factor was also separated from these OA factors, with nitrates in this factor accounting for 92.6 % of the total NO<sub>x</sub><sup>+</sup> ions. <u>DespiteApart from</u> NIA, NO<sub>x</sub><sup>+</sup> ions were more assigned in MO-OOA (6.0 % of the total NO<sub>x</sub><sup>+</sup>) than LO-OOA (1.4 % of the total NO<sub>x</sub><sup>+</sup>), suggesting that ONs were more associated with MO-OOA. <u>This is contrary to other</u> studies which reported that ONs were more correlated with less oxidized OA (Zhang et al., 2016; Yu et al., 2019). According to One possible reason is that ONs formed and mixed with MO-OOA components during the aqueous aging processes of MO-OOA-coated BC particles. This hypothesis is supported by Cao et al. (2022), which demonstrated that ONs exhibit 385 similar volatility to that of MO-OOA when coated on BC, previous studies, the ratios of  $NO^{+}/NO_{+}^{+}$ approximately 2.25-3.7 times higher than pure NH<sub>4</sub>NO<sub>2</sub> (Fry et al., 2013; Fry et al., 2009). Consistently, the average NO<sup>+</sup>/NO<sub>2</sub><sup>+</sup> ratios of LO-OOA and MO-OOA were 13.19 and 11.2, falling within the range of ONs. In contrast, a NO<sup>+</sup>/NO<sub>2</sub><sup>+</sup> f 3.56 was observed for NIA, reflecting its characteristics of inorganic nitrates. Based on the PMF results above, the mass concentrations of ONs during the AMS sampling period were estimated (Fig. 109). Considering ONs as part of 390 organics, we chose a RIE value of 1.4 for the estimated ONs, while a nitrate RIE value (1.1) was correspondingly applied for INs. The results showed that ONs made a greater contribution to total nitrate in P1 than in P2 (27 % vs. 3 %). This was further supported by the result of Huang et al. (2021), which found that organic nitrate can increase rapidly when RH > 70 %. The average mass concentration of ONs in P4C1 was  $30 \pm 22$  ng m<sup>-3</sup>, which was comparable to lower than that in P2EP5 (40  $\pm$  23 ng m<sup>-3</sup>), likely owing to which can also be explained by the evaporative release of ONs in EP5. However, considering the eloud seavenging slightly elevated values (10 ng m<sup>-3</sup>) between these two periods, we cannot rule out the possible 395 formation of ON from aqueous phase processes (Xian et al., 2023) and gas phase reaction initiated by NO<sub>3</sub> during nighttime (Ayres et al., 2015). Also, the low levels of ON might cause uncertainties in its estimation. Since ONs at this site waswere close to the value (40 ng m<sup>-3</sup>) obtained reported by Hao et al. (2014) at a forest-urban mixed site in Finland (Hao et al., 2014), we considered our quantification of ONs reasonable. However, significant discrepancies differences were observed between the INs mass concentrations during P4C1 and P2EP5 (0.08 µg m<sup>-3</sup> vs.1.47 µg m<sup>-3</sup>), revealing that). This could be attributed to 400that water-soluble HNO<sub>3</sub> or NH<sub>4</sub>NO<sub>3</sub> was scavenged by cloud droplets and removed by precipitation before transporting to this site during C1, while in EP5, these INs dominated the elevation of total nitrate during cloud evaporation-in P2. These differences in INs also led to a greater contribution of ONs to total nitrate in C1 than in EP5 (27 % vs. 3 %).



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Figure 82. Time series and relative contributions of inorganic nitrate and organic nitrate during the AMS sampling period.



410 Figure 10. (a) Scatter plot of OA/ΔCO as a function of O/C ratios and (b) mass concentrations and proportions of MO-OOA and LO-OOA during C1 and EP5.

## 3.45 Evolution of OA

The formation and evolution of OA can be investigated using the ratio of OA to ΔCO (CO minus background CO) to remove the atmospheric dilution effects (SunDecarlo et al., 20112010). In this study, 0.12 ppmvppm (average of the lowest 5 % concentration) was used as the background mixing ratio of CO (Fig. S7S12), which is close to the 0.1 ppm used in Hu et al. (2013) and Yuan et al. (2013). OA/ΔCO was 28.3 ± 26.3 µg m<sup>-3</sup> ppmv<sup>-1</sup> during the studingstudy period, which is comparable to (41.7 ± 23.0 µg m<sup>-3</sup> ppmv<sup>-1</sup>) in suburban Sichuan Basin (Hu et al., 2016) but much lower than the mean value (70 ± 20 µg m<sup>-3</sup> ppmv<sup>-1</sup>) in worldwide urban air (De Gouw and Jimenez, 2009). The scatter plot of OA/ΔCO as a function of O/C ratios during P4C1 and P2EP5 is shown in Fig. 9a10a. Interestingly, different OA/ΔCO variations were found with the increasing 420
O/C during P4C1 and P2EP5. During P4C1, OA/ΔCO trendedtended to increase with the increase of the O/C ratio,

indicating aging process produced SOA (Hu et al., 2017). In contrast, a remarkable decrease trend of OA/ $\Delta$ CO was observed with the increasing O/C during <u>P2EP5</u>, suggesting that less oxidized OA may contribute more significantly to the high OA concentration during this period. Moreover, <u>considering the cloud evaporation process in this period</u>, the negative correlation between OA/ $\Delta$ CO and O/C in <u>P2EP5</u> also implies that <u>more oxidized OA had almost been scavenged by clouds</u>, <u>while</u>the

- 425 less oxidized OA previously formed or seavengedincorporated into cloud droplets wascan be released during cloud evaporation. Another possible mechanism was that after the cloud was evaporated, SOA formation from biogenic VOCs could then be strengthened by stronger solar radiation, which may also contribute to the increase of less oxidized OA. Conversely, more oxidized OA in cloud droplets may have already undergone mass reduction through fragmentation reactions (Lee et al., 2012), which was far less likely to be reintroduced into the atmosphere. Similarly, significant increases
- 430 in the LO-OOA concentration (0.27 to 1.70 μg m<sup>-3</sup>) and fraction (40 % to 70 %) were observed from P4C1 to P2EP5, while the MO-OOA concentration (0.40 vs. 0.74 μg m<sup>-3</sup>) did not show large variation (Fig. 9b10b), further supporting our conclusion. In addition, these released LO OOA is likely composed of humic like substances (HULIS), which can account for 49 % of the water soluble organic matter at a nearby mountain site in southeastern China (Tao et al., 2023; Chen et al., 2016).

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Figure 9. (a) Scatter plot of OA/ $\Delta$ CO as a function of O/C ratios and (b) mass concentrations and proportions of MO-OOA and LO-OOA during P1 and P2.

- 440 Figure 10a11a shows the Van Krevelen diagrams of OA in this study-together with other mountain sites. The slope of H/C to O/C in the present study was -0.66 (r<sup>2</sup> = 0.64), suggesting the addition of carboxyl functional groups during OA evolution (Heald et al., 2010). It This slope is interestingslightly flatter than those (-0.7 to mention that\_1.0) of other remote/rural regions across the two OOA factors exhibited a strong alignment with world (Chen et al., 2015), indicating the fitting line, which suggested that the evolution processes of OA inat this site likely follows a transformation pathway from LO-
- 445 OOA to MO OOA.were more associated with fragmentation reactions. Consistent evolution trends are also shown in the  $f_{43}$ vs.  $f_{44}$  space (Fig. 10b11b). The SO<sub>4</sub>-OA and MO-OOA showed similar high oxidative properties oxidation degrees, with  $f_{43}$ and  $f_{44}$  located at the upper part of the triangular region because of the larger fractional contribution of CO<sub>2</sub><sup>+</sup> in the organic

mass spectrum. The LO-OOA was situated in the middle region of the triangle, while NIA resided near the bottom right. Moreover, <u>the mass spectra of LO-OOA and MO-OOA both overlapped withresembles to those of</u> the aged OAs observed at other elevated sites (Xu et al., 2018a; Zhou et al., 2019). These results together reveal that OA observed at SH site is representative of the background-aged SOA in the YRD region in China.

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Figure 1011. (a) Van Krevelen diagrams (H/C vs. O/C) and (b)  $f_{43}$  vs.  $f_{44}$  of OA (grey circles, hourly averaged) and four aerosol factors identified by the PMF analyses during the AMS sampling period.

#### **3.5 Backward trajectory analysis**

alculated for clusters were determined, including two clusters om the northeast ŧh one Among the four mace 460 contribution sur SH rents (Event 2 edominantly associated with results suggest that the high <del>of PM<sub>1</sub> at the SH</del> the regional transport of air mass from the west and southwest. site Additionally the nitrate

465 event on 4 November (Event 1, discussed in section 3.1) was primarily linked to C3, indicating a potential long distance transport contribution of nitrate particles from the NCP region.



Figure 11. (a) Average 72 h backward trajectory clusters calculated at 1 h intervals during the entire campaign and (b) temporal variation 470 of 1 hour averaged PM<sub>1</sub> concentration coloured by corresponding clusters.

#### 4 Conclusions

The chemical Chemical composition and sources of PM<sub>1</sub> were investigated at a forested mountain site in southeastern China in November 2022<sub>7</sub> were characterized by-using two different aerosol mass spectrometers. The average mass concentration of total PM<sub>1</sub> (4.45 ± 6.51 µg m<sup>-3</sup>) was notablyoverall lower than those observed at other mountain sites in China, yet <u>it was</u>
similarly dominated by OAorganics (41.1 %). Remarkably, nitrate (20.9 %) and ammonium (17.0 %)sulfate exhibited unexpectedly highlower contributions (16.7 %) to PM<sub>1</sub> in this forested compared to other mountain area,sites, in contrast to higher contributions of nitrate (14.7%), indicating the impactinfluences of anthropogenic emissions on clouds and subsequently influencing regional background areas through atmospherie transport. The OA-over a relatively small regional scale in southeastern China. Back trajectory analysis revealed that higher concentrations of submicron aerosols at this mountain site were associated with the transport from the western and southwestern regions. OA was primarily ofdominantly secondary origin and highly aged, as suggested by the high O/C (0.85–0.96) and OSc (0.21–0.49) ratios. PMF analysis of combined organic and inorganic spectra identified two types of SOA. Notably, OOA and an OA factor associated with SO<sub>4</sub>. Most importantly, we found that cloud scavenging and evaporation processes demonstrated differential impacts on influence LO-OOA and MO-OOA. Cloud differently. The cloud scavenging efficiently removed showed a greater efficiency in

- 485 removing MO-OOA, aligningconsistent with the smaller size distributions of interstitial particles observed-during eloud events. Conversely, cloud events. In contrast, the evaporation released a substantial of cloud tended to release a large amount of LO-OOA, emphasizinghighlighting that SOA retainedremained in cloud droplets generally maintainedwas mostly in a moderate oxidation state. Backward trajectory analysis indicated that higher concentrations of submicron aerosols at the mountain site were associated with transport from the western and southwestern regions. In conclusion<u>Overall</u>, this study
- 490 <u>underscoresillustrates</u> the <u>significanceimportance</u> of SOA in-a forested mountain site in southeastern China, where aerosol <u>concentrations</u>, composition, size distributions, and oxidation <u>statesstate</u> are <u>significantly influencedaffected</u> by aerosolcloud interactions-<u>substantially</u>. Moreover, our study provides valuable data and insights into understanding the aerosolcloud interactions in mountainous areas that are often under cloudy conditions, and cloud evaporation can be considered as a plausible mechanism to explain certain aerosol episodes in these regions.
- 495 **Data availability**. Data used in this study can be accessed at repository under: <u>https://doi.org/10.5281/zenodo.10312334</u> (Zhang et al., 2023).

Author contribution. WX and YS designed the research. WX, WZ, and HQ conducted the measurements. ZZ, YZ, WZ, XX, AD, YZ, XC and YS analysed the data. YK, XP, ZW, LL, QF, DRW, and YS reviewed and commented on the paper. ZZ and YS wrote the paper.

500 Competing interests. <u>DRW is the manufacturer of AMS and ACSM in this study.</u>

utilized in this study. The authors declare that they have no conflict of interest.

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