

1 **Sea salt reactivity over the northwest Atlantic: An in-depth look**  
2 **using the airborne ACTIVATE dataset**

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4 Eva-Lou Edwards<sup>1</sup>, Yonghoon Choi<sup>2,3</sup>, Ewan C. Crosbie<sup>2,3</sup>, Joshua P. DiGangi<sup>2</sup>, Glenn S.  
5 Diskin<sup>2</sup>, Claire E. Robinson<sup>2,3,†</sup>, Michael A. Shook<sup>2</sup>, Edward L. Winstead<sup>2,3</sup>, Luke D. Ziemba<sup>2</sup>,  
6 and Armin Sorooshian<sup>1,4</sup>

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9 <sup>1</sup>Department of Chemical and Environmental Engineering, University of Arizona, Tucson, AZ,  
10 85721, USA

11 <sup>2</sup>NASA Langley Research Center, Hampton, VA, 23681, USA

12 <sup>3</sup>Analytical Mechanics Associates, Inc., Hampton, VA, 23666, USA

13 <sup>4</sup>Department of Hydrology and Atmospheric Sciences, University of Arizona, Tucson, AZ,  
14 85721, USA

15  
16 †Deceased

17  
18 \*Corresponding author: [armin@arizona.edu](mailto:armin@arizona.edu)

19 **Abstract**

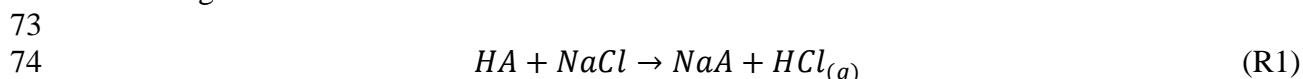
20 Chloride (Cl<sup>-</sup>) displacement from sea salt particles is an extensively studied phenomenon with  
21 implications on human health, visibility, and the global radiation budget. Past works have  
22 investigated Cl<sup>-</sup> depletion over the northwest Atlantic (NWA), yet an updated, multiseasonal, and  
23 geographically expanded account of sea salt reactivity over the region is needed. This study uses  
24 chemically resolved mass concentrations and meteorological data from the airborne Aerosol Cloud  
25 meTeorology Interactions oVer the western ATlantic Experiment (ACTIVATE) to quantify  
26 seasonal, spatial, and meteorological trends in Cl<sup>-</sup> depletion and to explore the importance of  
27 quantifying (1) non-sea salt sources of Na<sup>+</sup> and (2) mass concentrations of lost Cl<sup>-</sup> instead of  
28 relative amounts displaced. Lost Cl<sup>-</sup> mass concentrations are lowest in December-February and  
29 March, moderate around Bermuda in June, and highest in May (median losses of 0.04, 0.04, 0.66,  
30 and 1.76  $\mu\text{g m}^{-3}$ , respectively), with losses in May high enough to potentially accelerate  
31 tropospheric oxidation rates. Inorganic acidic species can account for all Cl<sup>-</sup> depletion in  
32 December-February, March, and June near Bermuda, yet none of the lost Cl<sup>-</sup> in May, suggesting  
33 organic acids may be of importance for Cl<sup>-</sup> displacement in certain months. Contributions of dust  
34 to Na<sup>+</sup> are not important seasonally but may cause relevant overestimates of lost Cl<sup>-</sup> in smoke and  
35 dust plumes. Higher percentages of Cl<sup>-</sup> depletion often do not correspond to larger mass  
36 concentrations of lost Cl<sup>-</sup>, so it is highly recommended to quantify the latter to place depletion  
37 reactions in context with their role in atmospheric oxidation and radiative forcing.

## 38 1. Introduction

39 Chlorine (Cl) is a common constituent of trace gases and aerosol particles found in Earth's  
40 atmosphere. Chlorine-containing species play a critical role in the global radiation budget for many  
41 reasons, including their ability to produce highly reactive Cl radicals. These radicals can perturb  
42 atmospheric chemical processes by inducing reactions that would otherwise be less likely to occur  
43 and/or accelerating the rates of certain reactions. For example, Cl radicals in the stratosphere can  
44 incite reactions that destroy ozone (O<sub>3</sub>; Molina and Rowland, 1974; Solomon et al., 2023),  
45 therefore allowing increased amounts of shortwave radiation to reach the surface and harmfully  
46 affect living beings.

47 Cl radicals typically react faster with volatile organic compounds (VOCs) compared to  
48 hydroxyl radicals (OH; Roberts et al., 2008; Thornton et al., 2010; Young et al., 2014), which has  
49 particular importance in the troposphere. Cl radicals oxidize methane ~16 times faster than OH  
50 (Faxon and Allen, 2013 and references therein), thus reducing the lifetime of this important  
51 greenhouse gas. Accelerated oxidation of methane and other VOCs can result in increased O<sub>3</sub>  
52 production near the surface (Knipping and Dabdub, 2003; Pechtl and von Glasow, 2007; Tanaka  
53 et al., 2003), which can have deleterious effects on animals (e.g., respiratory problems, increased  
54 mortality; Lippmann, 1989; Nuvolone et al., 2018) and plants (e.g., decreased growth and  
55 photosynthesis; Wittig et al., 2009). Cl radicals may be responsible for 0.8% of the global oxidation  
56 of methane, 14% of ethane, 8% for propane, and 7% for longer-chain alkanes (Wang et al., 2021)  
57 and can play an exceptionally critical role in governing atmospheric composition in the early  
58 morning when OH radicals are less abundant (Young et al., 2013; Riedel et al., 2014; Osthoff et  
59 al., 2008). Due to their significant impacts on radiative forcing, rates of chemical cycling, and the  
60 health of living organisms, it is critical to quantify and understand sources of atmospheric Cl  
61 radicals.

62 Sea salt aerosol particles are the largest reservoir of reactive atmospheric Cl. Keene et al.  
63 (1999) estimated that at any given time there are ~22 Tg of reactive Cl in the troposphere, and that  
64 68% of this mass is found in particulate form, primarily sea salt. More recently, Wang et al. (2021)  
65 suggested there are 2.44 Tg of reactive tropospheric Cl with 90% in particulate form as sea salt.  
66 The fact that estimates for the reactive tropospheric Cl budget have decreased by an order of  
67 magnitude over the past two decades motivates continued research on tropospheric halogen  
68 chemistry and its impacts. Although the Cl in sea salt will not directly photolyze to produce Cl  
69 radicals, it can be displaced by acidic species (e.g., sulfate [SO<sub>4</sub><sup>2-</sup>], nitrate [NO<sub>3</sub><sup>-</sup>], organic acids)  
70 and released in a reactive gaseous form (e.g., ClNO<sub>2</sub>, HCl, Cl<sub>2</sub>) that has the potential to produce  
71 Cl radicals. This phenomenon is called chloride (Cl<sup>-</sup>) depletion and can be generalized with the  
72 following reaction:



75  
76 where A is one of the acidic species mentioned above. Note that most of the generated HCl is  
77 removed by deposition, but a fraction (~16% globally; Wang et al., 2021) reacts with OH to  
78 produce Cl radicals, which initiates rapid cycling between these radicals and their inorganic non-  
79 radical reservoirs. In addition to producing reactive chlorine-containing gases, Cl<sup>-</sup> depletion can  
80 alter the acidity (e.g., Keene and Savoie, 1998), hygroscopicity (e.g., Drozd et al., 2014; Ghorai et  
81 al., 2014; Randles et al., 2004), and optical properties (Finlayson-Pitts and Pitts, 2000; Tang et al.,  
82 1997) of sea salt particles. Such changes affect partitioning of other chemicals (e.g., water vapor,  
83 ammonia [NH<sub>3</sub>], SO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup>) between the gas and particle phases (Chen et al., 2021), the rates

84 and types of reactions occurring within sea salt particles (Chameides and Stelson, 1993), the  
85 activity of these particles as cloud condensation nuclei (e.g., Chatterjee et al., 2020), and their  
86 interactions with solar radiation, all of which can have implications for visibility, air quality,  
87 biogeochemical cycles, and Earth's radiation budget.

88 Many factors dictate the extent to which  $\text{Cl}^-$  depletion occurs in an air mass including  
89 meteorology (e.g., wind speed, temperature, relative humidity [RH], available solar radiation), the  
90 size distribution and mixing state of sea salt particles, and the availability and length of exposure  
91 to surrounding acidic species (Su et al., 2022 and references therein). Regarding the latter,  $\text{Cl}^-$   
92 depletion is therefore typically observed where marine particles and acidic species are both  
93 present, such as where emissions from biomass burning (BB) advect over a marine location (Braun  
94 et al., 2017; Maudlin et al., 2015; Li et al., 2003; Yokelson et al., 2009; Akagi et al., 2013; Dang  
95 et al., 2022; Crosbie et al., 2022), in regions with active phytoplankton and marine bacteria that  
96 emit dimethyl sulfide (DMS), which can oxidize to form sulfuric acid ( $\text{H}_2\text{SO}_4$ ; Seinfeld and Pandis,  
97 2016; Tang et al., 2019; Yan et al., 2020), and/or in and around urban coastal environments (e.g.,  
98 Kong et al., 2014; Chatterjee et al., 2020; AzadiAghdam et al., 2019; Nolte et al., 2008) where  
99 anthropogenic emissions serve as precursors for various acidic species.

100 For this reason, the northwest Atlantic (NWA) is an opportune region for observing and  
101 studying  $\text{Cl}^-$  depletion. Cities extending along the East Coast of North America consistently emit  
102 sulfur dioxide ( $\text{SO}_2$ ), nitrogen oxides ( $\text{NO}_x$ ), and VOCs, which can oxidize to form  $\text{H}_2\text{SO}_4$ , nitric  
103 acid ( $\text{HNO}_3$ ), and organic acids, respectively, while sea salt particles are ubiquitous over the region  
104 due to wave breaking (Reid et al., 2001; Ferrare et al., 2023). Occasional long-range transport from  
105 BB in Alaska, Canada, and the western United States (U.S.; Fehsenfeld et al., 2006; Mardi et al.,  
106 2021), agricultural fires throughout the eastern and southeastern U.S. (Jaffe et al., 2020; McCarty  
107 et al., 2007), wintertime wood burning for residential heating (Corral et al., 2021; Sullivan et al.,  
108 2019), and seasonally varying emissions from vegetation and ocean biological activity (Savoie et  
109 al., 2002; Corral et al., 2022) can also introduce acidic species to this region.

110  $\text{Cl}^-$  depletion has been observed over the NWA for decades (Table 1). Previous datasets  
111 typically span 2 – 3 months, and most are reflective of conditions during the boreal summer,  
112 although there are a handful of studies extending outside of this period (i.e., Keene et al., 1990;  
113 Yao and Zhang, 2012; Zhao and Gao, 2008; Haskins et al., 2018). Combining results from these  
114 works to build seasonal and temporal statistics is challenged by the fact that each dataset is specific  
115 to a certain altitude (or range of altitudes), location(s), time period, sampling method, and size  
116 range of sampled particles. In addition to these logistical constraints, there is an overall shortage  
117 of  $\text{Cl}^-$  depletion data for the spring, fall and winter, which is of concern as depletion processes are  
118 sensitive to several properties that fluctuate seasonally over the NWA (e.g., temperature, solar  
119 radiation, RH).

120 Most past works over the NWA report on  $\text{Cl}^-$  depletion along the United States East Coast  
121 (USEC) and/or at Bermuda. To our knowledge, there is an absence of discussion about the gradient  
122 in  $\text{Cl}^-$  depletion moving from the USEC to the open ocean environment closer to Bermuda. Corral  
123 et al. (2021) showed strong gradients in aerosol optical depth along this direction for several  
124 particle types including sea salt and  $\text{SO}_4^{2-}$ , suggesting there may be a gradient in  $\text{Cl}^-$  depletion as  
125 well. Furthermore,  $\text{Cl}^-$  depletion results from previous studies typically reflect conditions near the  
126 surface, yet Shinozuka et al. (2004) showed that the vertical scattering profile of sea salt in the  
127 lower 1 km of the atmosphere becomes increasingly less uniform with increasing wind speed. Also  
128 of note is that most datasets referenced in Table 1 are now several decades old. Mass  
129 concentrations of  $\text{SO}_2$ ,  $\text{NO}_x$ ,  $\text{SO}_4^{2-}$ , and  $\text{NO}_3^-$  over the eastern U.S. and Canada have steadily

130 decreased since 1990 due, in part, to the Clean Air Act of 1963 and its subsequent amendments  
131 (Feng et al., 2020; Kuklinska et al., 2015). Such reductions warrant an updated analysis of Cl<sup>-</sup>  
132 depletion over the NWA.

133 We note that Cl<sup>-</sup> depletion results from the Wintertime Investigation of Transport, Emissions  
134 and Reactivity (WINTER) aircraft campaign (Haskins et al., 2018) are an exception to many of  
135 the points raised above. As an airborne campaign from February – March 2015, WINTER provides  
136 data relevant to halogen chemistry at altitudes throughout the boundary layer, at a time of year that  
137 had previously not been studied, and in a year recent enough to capture the aforementioned  
138 reductions in anthropogenically sourced acidic species. However, WINTER flights specifically  
139 sampled over and downwind of various pollution sources in the eastern and southeastern U.S.,  
140 meaning Cl<sup>-</sup> depletion results may be disproportionately reflective of highly polluted, coastally  
141 influenced air masses as compared to other air mass types observed over the NWA during winter  
142 and spring (e.g., those (i) occurring after synoptically forced frontal systems have moved through,  
143 (ii) associated with cold air outbreaks (CAOs), and (iii) occurring when southerly winds advect  
144 maritime air masses northward along the East Coast).

145 It is common for Cl<sup>-</sup> depletion studies to base their calculations on the assumption that sea salt  
146 particles are the only source of atmospheric sodium (Na<sup>+</sup>; i.e., Na<sup>+</sup> is used as the reference species  
147 for determining the extent of Cl<sup>-</sup> depletion observed), including nearly all the works listed in Table  
148 1. The validity of this assumption is dependent on several factors, including the proximity to urban  
149 emissions, if dust particles are present, and the size range of particles sampled. Ooki et al. (2002)  
150 found Na<sup>+</sup> to be highly correlated with potassium (K<sup>+</sup>) in particles < 1.1 μm in urban air masses,  
151 implying that these two species have the same source in fine, anthropogenically sourced particles.  
152 K<sup>+</sup> is thought to come mainly from BB (Echalar et al., 1995; Andreae et al., 1998; Andreae and  
153 Merlet, 2001) and anthropogenic activities (Ooki et al., 2002 and references therein), suggesting  
154 that marine air masses heavily influenced by BB or urban emissions may have nonnegligible  
155 contributions from non-sea salt sources to total Na<sup>+</sup>, especially if submicron particles contribute  
156 significantly to total mass concentrations (which would depend on the size range of particles  
157 sampled). Na<sup>+</sup> can also be found in mineral dust (Seinfeld and Pandis, 2016), which has motivated  
158 a handful of studies to discern between the amounts of Na<sup>+</sup> coming from dust and sea salt using a  
159 system of equations (e.g., Boreddy and Kawamura, 2015; AzadiAghdam et al., 2019). The NWA  
160 is known to be periodically influenced by Asian, African, and North American dust (e.g., Aldhaif  
161 et al., 2020) and emissions from BB (Fehsenfeld et al., 2006; Schroder et al., 2018; Sullivan et al.,  
162 2019; Mardi et al., 2021), and is consistently influenced by anthropogenic activities throughout  
163 the year. Several works shown in Table 1 have acknowledged that these additional sources of Na<sup>+</sup>  
164 may influence estimates of Cl<sup>-</sup> depletion over the NWA, but none have quantitatively explored this  
165 possibility.

166 Finally, most Cl<sup>-</sup> depletion studies report the percentage of Cl<sup>-</sup> in unreacted sea salt particles  
167 that has been displaced by acidic species, an approach useful for quantifying the extent of Cl<sup>-</sup>  
168 depletion processes independently of the sea salt mass concentrations present, which can vary  
169 seasonally, temporally, and geographically. However, reporting Cl<sup>-</sup> depletion as a percentage can  
170 make it more difficult to conceptualize and quantify the degree to which depletion reactions may  
171 be affecting atmospheric oxidation potential. Several past works focusing on the NWA have  
172 reported the magnitude of Cl<sup>-</sup> displaced from sea salt particles, either in units of nmol m<sup>-3</sup> (e.g.,  
173 Keene and Savoie, 1998; Keene et al., 1990) or pptv (Keene et al., 2007; Haskins et al., 2018),  
174 which we find useful for comprehensive interpretation considering that Singh and Kasting (1988)  
175 suggested ppbv concentrations of HCl have the potential to produce enough Cl radicals to oxidize

176 20 – 40% of nonmethane alkanes in the marine troposphere. Thus, reporting Cl<sup>-</sup> depletion both as  
177 a percentage and as a mass concentration benefits the atmospheric chemistry community as results  
178 can be used either comparatively or to improve quantification of Cl radical budgets and the  
179 atmospheric oxidation capacity in a given region. Although a few past works in the NWA have  
180 reported mass concentrations of displaced Cl<sup>-</sup>, there is still a need for results reflecting current  
181 conditions across a range of seasons as we have discussed above.

182 In summary, there is a demand for an updated, multi-seasonal, spatially resolved dataset  
183 reflecting Cl<sup>-</sup> depletion processes in the NWA boundary layer across a variety of meteorological  
184 conditions and air mass types. There is also interest in (i) exploring the sensitivity of Cl<sup>-</sup> depletion  
185 results to accounting for non-sea salt sources of Na<sup>+</sup>, especially in seasons and/or air masses  
186 influenced by dust and BB emissions, as well as (ii) quantifying both the percentage and magnitude  
187 of Cl<sup>-</sup> displaced from sea salt particles for straightforward comparisons to other works and to link  
188 results more easily to boundary layer Cl radical budgets and their potential influence on  
189 atmospheric oxidation rates. This study seeks to address these points by using data from the NASA  
190 Aerosol Cloud meTeorology Interactions oVer the western ATlantic Experiment (ACTIVATE)  
191 airborne field campaign (Sorooshian et al., 2019). The statistical approach, large number of flights  
192 spanning a range of seasons and meteorological conditions, and type of instruments deployed on  
193 this campaign make the ACTIVATE dataset well-suited to address several of the outstanding  
194 uncertainties and unknowns regarding Cl<sup>-</sup> depletion over the NWA.

195 **Table 1.** Relevant information from previous works, sorted chronologically, documenting Cl<sup>-</sup>  
 196 depletion over the Northwest Atlantic (NWA). “USEC” stands for United States East Coast, and  
 197 “U.S.” stands for United States.

Reference(s)	Dates	Location	Platform(s)	Reference species to determine Cl <sup>-</sup> depletion	Discusses possibility of non-sea salt sources of Na <sup>+</sup> and/or Cl <sup>-</sup>
Keene et al. (1990)	Jul – Sep 1988	USEC and near Bermuda	Ship and aircraft	Na <sup>+</sup>	No
Keene and Savoie (1998)	Apr – May 1996	Bermuda	Surface station	Na <sup>+</sup>	No
Nolte et al. (2008)	May – Jun 2002	Tampa, Florida (U.S.)	Surface stations	Na <sup>+</sup>	Yes
Yao and Zhang (2012)	Jun – Jul 2002, Oct – Nov 2002	Kejimikujik, Nova Scotia	Surface station	Na <sup>+</sup>	No
Keene et al. (2004)	Jul – Aug 2002	USEC	Ship	Mg <sup>2+</sup> , <sup>2</sup>	No
Quinn and Bates (2005)	Jul – Aug 2002	USEC	Ship	Na <sup>+</sup>	No
Keene et al. (2007)	Jul – Aug 2004	Appledore Island, Maine (U.S.)	Surface station	Na <sup>+</sup> and Mg <sup>2+</sup>	Yes
Zhao and Gao (2008)	Jul – Sep 2006	Newark, New Jersey (U.S.)	Surface station	Na <sup>+</sup>	Yes
Bondy et al. (2017)	Jun – Jul 2011	Centreville, Alabama (U.S.)	Surface station	Na <sup>+</sup> and Mg <sup>2+</sup>	Yes
Haskins et al. (2018)	Feb – Mar 2015	USEC and over land around major pollution sources across the eastern U.S. <sup>1</sup>	Aircraft	Na <sup>+</sup>	Yes

198 <sup>1</sup>The Wintertime Investigation of Transport, Emissions, and Reactivity (WINTER) airborne field  
 199 campaign focused on three regions over the U.S.: i) the northeast metropolitan corridor  
 200 (encompassing major cities from Boston to Washington D.C.), ii) the Ohio River Valley, and iii)  
 201 the Southeast. Research flights also extended over coastal waters to sample polluted air masses  
 202 downwind from their sources.

203 <sup>2</sup>Magnesium (Mg<sup>2+</sup>) was chosen as the reference species for sea salt in Keene et al. (2004) as Na<sup>+</sup>  
 204 had a relatively higher and more variable background in the quartz-fiber sampling media used.

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## **2. Data and methods**

### **2.1 ACTIVATE campaign description**

The ACTIVATE field campaign focused on characterizing relationships between aerosol particles, meteorology, and marine boundary layer clouds over the NWA using two research aircraft flying in coordination. Operations were based out of NASA Langley Research Center (LaRC), although a multitude of other sites supported various aspects of the project. The high-flying King Air usually flew steadily at ~9 km releasing dropsondes and using a suite of remote sensors to retrieve particle and cloud properties below the aircraft. The low-flying HU-25 Falcon (hereafter referred to as the “Falcon”) made in situ measurements of trace gases, aerosol particle properties, cloud and precipitation properties (if present), and meteorological conditions in and around boundary layer clouds or in clear conditions usually below 3 km.

ACTIVATE placed a high priority on building statistics to fulfill its objectives and address current uncertainties regarding aerosol-cloud interactions and remote sensing capabilities over the NWA. To acquire such statistics, the Falcon and King Air achieved 174 and 168 flights with 574 and 592 total flight hours, respectively, from 2020 – 2022 (note that 162 of these were “joint” flights where the aircraft flew in coordination; Sorooshian et al., 2023). The campaign included multiple seasons, with each aircraft adhering to an intentional and consistent flight strategy throughout, to better constrain the multitude of variables affecting a given clear or cloudy scene. As mentioned above, the King Air flew fixedly at ~9 km regardless of the amount of cloud coverage below. In the presence of low-level (<3 km) clouds, the Falcon conducted “cloud ensembles” by flying 3-minute legs at the following key vertical positions: near the ocean surface (MinAlt; ~150 m), below cloud base, above cloud base, below cloud top, and above cloud top. In the absence of low-level clouds, the Falcon switched to “clear ensembles,” which involved 3-minute legs at MinAlt, ~230 m (an altitude useful for remote sensing validation), and at altitudes falling slightly below and above the boundary layer height (see Fig. 2 in Sorooshian et al. [2023] for an illustration of these ensembles). The campaign was executed over six deployments, which are referred to as Winter 2020 (February – March 2020), Summer 2020 (August – September 2020), Winter 2021 (January – April 2021), Summer 2021 (May – June 2021), Winter 2022 (November 2021 – March 2022), and Summer 2022 (May – June 2022) as recommended in Sorooshian et al. (2023). Note that Winter 2022 includes two months in 2021 but is referred to as “Winter 2022” for simplicity.

### **2.2 Falcon data**

The main instrument providing data for this study is a particle into liquid sampler (PILS; Brechtel Manufacturing Inc. [BMI]) that was operated downstream from an isokinetic Clarke-style shrouded solid double-diffuser inlet (BMI; McNaughton et al., 2007) onboard the Falcon. The PILS grows aerosol particles with diameters of 50 - 5000 nm at ambient RH into droplets large enough to be collected via inertial impaction (Sorooshian et al., 2006; Crosbie et al., 2020). Droplets striking the impaction plate are pumped into vials that are analyzed offline using ion chromatography (IC) to quantify air equivalent mass concentrations of Na<sup>+</sup>, ammonium (NH<sub>4</sub><sup>+</sup>), K<sup>+</sup>, magnesium (Mg<sup>2+</sup>), calcium (Ca<sup>2+</sup>), Cl<sup>-</sup>, NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup>, and oxalate. PILS data are critical to this study due to the instrument’s ability to capture particles containing sea salt, dust, and other refractory species that are largely omitted by the aerosol mass spectrometer (AMS). PILS flowrates were set such that it took 300 - 420 s (5 – 7 minutes) to fill each vial, the minimum duration for collecting enough particle mass to be above speciated detection limits while also meeting injection



251 volume requirements for IC analysis. Note that the time spent collecting one PILS sample is greater  
252 than the duration of the individual level legs (~3 minutes) comprising clear and cloudy ensembles.  
253 The possibility that each PILS sample could represent atmospheric properties sampled during  
254 multiple level legs and/or periods of ascent or descent between level legs impacted our analysis in  
255 two ways. First, PILS measurements must be considered as a representation of water-soluble ionic  
256 composition throughout the lower 3 km of the atmosphere, meaning they cannot provide vertically  
257 resolved information. Second, we exclude PILS data collected during cloudy ensembles to  
258 eliminate possible cloud contamination. During cloudy ensembles, it is likely that the Falcon  
259 intercepted a cloud within any interval of 5 – 7 minutes, and in doing so, shattered droplets and  
260 other cloud artifacts were collected in the awaiting sample vial. Additionally, while flying through  
261 clouds, large droplets and ice particles can impact onto the walls within the isokinetic inlet where  
262 they may resuspend and, therefore, cause delayed sampling, of larger particles previously caught  
263 on these walls.

264 The PILS was operated without upstream acid and base denuders since (1) the removal  
265 efficiency for specific relevant gases is not well quantified, (2) it is not known how the removal of  
266 gases affects the particle-phase equilibrium for semi-volatile species (e.g.,  $\text{NO}_3^-$ ), and (3) the  
267 addition of denuders decreases the transmission efficiency of coarse-mode sea salt particles into  
268 the PILS. While there could be a small positive artifact from certain gases (e.g.,  $\text{SO}_2$ ,  $\text{HNO}_3$ ), the  
269 PILS should be much less sensitive to this issue than filter collection methods with offline analysis.  
270 However, the absence of a base denuder opened the possibility for  $\text{NH}_3$ , a highly soluble trace gas,  
271 to contribute to particulate  $\text{NH}_4^+$  mass concentrations. During quality control analyses, PILS  $\text{NH}_4^+$   
272 mass concentrations were unjustifiably high in many samples, prompting us to omit this species  
273 from this study's analysis. As  $\text{NH}_4^+$  is a critical species for deriving parameters relevant to  $\text{Cl}^-$   
274 depletion, we alternatively use  $\text{NH}_4^+$  mass concentrations from a high-resolution time-of-flight  
275 aerosol mass spectrometer (HR-ToF-AMS; Aerodyne; DeCarlo et al., 2008; hereafter referred to  
276 as an "AMS"), which provided non-refractory mass concentrations of  $\text{NH}_4^+$  (among other species)  
277 for particles 60 – 600 nm in diameter at a 30-s time resolution. The AMS additionally provided  
278 mass concentrations of spectral markers for organic components, of which we use the tracers for  
279 oxygenated organics,  $m/z$  44, and methanesulfonic acid (MSA),  $m/z$  79. The AMS collection  
280 efficiency was set to unity as there was not compelling evidence to lower this value when  
281 comparing AMS and PILS  $\text{SO}_4^{2-}$  mass concentrations. AMS data were filtered to isolate those  
282 from clear ensembles and then averaged over the 5- to 7-minute interval for each PILS sample.  
283 Due to differences in the size range of the PILS and AMS,  $\text{NH}_4^+$  mass concentrations from the  
284 AMS represent a lower limit in this analysis.

285 Horizontal wind speed and static air temperature data were obtained using the Turbulent Air  
286 Motion Measurement System (TAMMS; Thornhill et al., 2003) operating at 20 Hz time resolution,  
287 while the diode laser hygrometer (DLH; Diskin et al., 2002) supplied water vapor mixing ratios  
288 and values of RH at 1 Hz time resolution. A commercial cavity ringdown spectrometer (G2401-  
289 m; PICARRO, Inc.) provided carbon monoxide (CO) measurements at 0.4 Hz resolution (DiGangi  
290 et al., 2021), which are used to qualitatively compare the extent to which certain seasons were  
291 influenced by anthropogenic emissions (Panagi et al., 2020; Naeher et al., 2001; Saide et al., 2011).  
292 Data are only considered from clear ensembles for each of the parameters described in this  
293 paragraph.

294 The Falcon occasionally intercepted clouds during clear ensembles. During these cloud passes,  
295 certain instruments (e.g., the AMS) sampled downstream of a counterflow virtual impactor (CVI;

296 BMI; Shingler et al., 2012) for droplet residual characterization. We removed data collected during  
297 periods with active CVI sampling from our analysis for all variables mentioned above.  
298

### 299 **2.3 Deployment selection and season/category classifications**

300 This analysis focuses on data collected during the Winter 2022 and Summer 2022  
301 deployments as they cover the largest geographical range over the NWA, thus presenting the best  
302 opportunity for studying spatial gradients in  $\text{Cl}^-$  depletion. During Winter 2022, sampling was  
303 extended northward on flights when the Falcon flew to Quonset State Airport in Rhode Island,  
304 refueled, and returned to LaRC, an option that was unavailable during the first four deployments  
305 due to challenges associated with the COVID-19 pandemic. Summer 2022 is the only deployment  
306 to (i) execute “transit flights” (i.e., flights where the Falcon flew to Bermuda, refueled, and flew  
307 back to LaRC on the same day) and (ii) include a set of out-and-back flights based in Bermuda.  
308 Additionally, Winter 2022 and Summer 2022 supply the largest and most continuous dataset  
309 compared to the first two years of the campaign. Nearly half of the total Falcon flights occurred  
310 within these two deployments, and sampling occurred consistently from 31 November 2021 to 18  
311 June 2022 with a brief break from 30 March – 02 May 2022. The high frequency of flights over a  
312 ~7-month period allows us to explore the seasonal evolution of properties relevant to  $\text{Cl}^-$  depletion,  
313 while also observing their fluctuations on daily to multiday time scales.

314 To capture both seasonal and spatial trends, Winter 2022 and Summer 2022 data are distributed  
315 among the following categories by season/month and/or by the geographical area sampled:  
316 December-February (30 November 2021 – 26 February 2022), March (02 – 29 March 2022), May  
317 (03 – 20 May 2022), March transit (22 March 2022), May transit (18, 21, and 31 May 2022), and  
318 June Bermuda (02 – 13 June 2022). Note that some flights from the Winter 2022 and Summer  
319 2022 deployments are omitted from this study because they are either composed entirely of cloudy  
320 ensembles and/or PILS data are unavailable during the clear ensembles. To explore relationships  
321 between (i) speciated mass concentrations and  $\text{Cl}^-$  depletion, and (ii) phenomena occurring on finer  
322 time scales (e.g., the passage of weather fronts, transport events of African dust plumes),  
323 meteorological conditions and/or notable influence from distinct aerosol types are documented for  
324 each research flight (RF). We also select RFs sampling various airstreams associated with passing  
325 frontal systems and dust-influenced air masses to further illustrate relationships between these  
326 phenomenon and properties relevant to  $\text{Cl}^-$  depletion.  
327

### 328 **2.4 Calculations relevant to $\text{Cl}^-$ depletion**

329 The following section describes how various properties associated with  $\text{Cl}^-$  depletion were  
330 derived using PILS and AMS bulk speciated mass concentrations and literature-based ratios for  
331 ions in sea salt, dust, and emissions from various combustion processes. Identifying the amount of  
332  $\text{Cl}^-$  displaced from sea salt particle begins with quantifying the original amount of  $\text{Cl}^-$ , which we  
333 derive from  $\text{Na}^+$  in sea salt ( $\text{ssNa}^+$ ) as this species has a relatively high mass fraction and is  
334 chemically inert in sea salt particles. We use Eqs. 1 – 5 to resolve contributions of sea salt and dust  
335 to bulk PILS mass concentrations of  $\text{Na}^+$  and  $\text{Ca}^{2+}$  (see Sect. S1 in the Supplement for additional  
336 information about these equations, Table S1 for variable nomenclature, and Table S2 for values of  
337 constant parameters [e.g., mass ratios]).  
338

$$Na_{bulk}^+ = ssNa^+ + Na_{dust}^+ \quad 1$$

$$Ca_{bulk}^{2+} = ssCa^{2+} + Ca_{dust}^{2+} \quad 2$$

$$ssCa^{2+} = ssNa^+ \cdot \left(\frac{Ca^{2+}}{Na^+}\right)_{ss} \quad 3$$

$$Ca_{dust}^{2+} = Na_{dust}^+ \cdot \left(\frac{Ca^{2+}}{Na^+}\right)_{dust} \quad 4$$

$$ssNa^+ = \frac{Ca_{bulk}^{2+} - Na_{bulk}^+ \cdot \left(\frac{Ca^{2+}}{Na^+}\right)_{dust}}{\left(\frac{Ca^{2+}}{Na^+}\right)_{ss} - \left(\frac{Ca^{2+}}{Na^+}\right)_{dust}} \quad 5$$

339 We then use an analogous set of equations (Eqs. 6 – 14) to explore if various combustion  
 340 processes contribute nonnegligible amounts of  $Na^+$  to bulk PILS  $Na^+$  mass concentrations (see  
 341 Sect. S2 for more information).

$$Na_{bulk}^+ = ssNa^+ + Na_{dust}^+ + Na_{comb}^+ \quad 6$$

$$Ca_{bulk}^{2+} = ssCa^{2+} + Ca_{dust}^{2+} \quad 7$$

$$K_{bulk}^+ = ssK^+ + K_{dust}^+ + K_{comb}^+ \quad 8$$

$$ssCa^{2+} = ssNa^+ \cdot \left(\frac{Ca^{2+}}{Na^+}\right)_{ss} \quad 9$$

$$Ca_{dust}^{2+} = Na_{dust}^+ \cdot \left(\frac{Ca^{2+}}{Na^+}\right)_{dust} \quad 10$$

$$ssK^+ = ssNa^+ \cdot \left(\frac{K^+}{Na^+}\right)_{ss} \quad 11$$

$$K_{dust}^+ = Ca_{dust}^{2+} \cdot \left(\frac{K^+}{Ca^{2+}}\right)_{dust} \quad 12$$

$$Na_{comb}^+ = K_{comb}^+ \cdot \left(\frac{Na^+}{K^+}\right)_{comb} \quad 13$$

$$ssNa^+ = \frac{Na_{bulk}^+ - K_{bulk}^+ \cdot \left(\frac{Na^+}{K^+}\right)_{comb} + Ca_{bulk}^{2+} \cdot \left[\left(\frac{K^+}{Ca^{2+}}\right)_{dust} \cdot \left(\frac{Na^+}{K^+}\right)_{comb} - \left(\frac{Na^+}{Ca^{2+}}\right)_{dust}\right]}{1 - \left[\left(\frac{Ca^{2+}}{Na^+}\right)_{ss} \cdot \left(\frac{K^+}{Ca^{2+}}\right)_{dust} \cdot \left(\frac{Na^+}{K^+}\right)_{comb}\right] - \left[\left(\frac{K^+}{Na^+}\right)_{ss} \cdot \left(\frac{Na^+}{K^+}\right)_{comb}\right] - \left[\left(\frac{Ca^{2+}}{Na^+}\right)_{ss} \cdot \left(\frac{Na^+}{Ca^{2+}}\right)_{dust}\right]} \quad 14$$

342

343 Combustion-generated particles over the NWA can stem from a range of seasonal and  
 344 perennial processes, each with a different  $Na^+$  and  $K^+$  emission factor. We use empirical, literature-  
 345 based values of  $\left(\frac{Na^+}{K^+}\right)_{comb}$  for particles emitted from the following combustion-related  
 346 activities/phenomena: agricultural burning, forest fires, industrial operations, sauna stove wood  
 347 burning for residential heating, car driving, and coal burning for electricity generation (Table S3).

348 Note that only one value at a time can be used for  $\left(\frac{Na^+}{K^+}\right)_{comb}$  in Eqs. 13 and 14, which forces the  
349 assumption that all combustion-generated particles collected in PILS samples are from the same  
350 source and/or have the same  $\left(\frac{Na^+}{K^+}\right)_{comb}$  value.

351 Mass concentrations of  $ssNa^+$  determined either by Eqs. 1 – 5 or Eqs. 6 – 14 are then used to  
352 determine sea salt mass concentrations (Eq. 15) as well as quantities relevant to  $Cl^-$  depletion (Eqs.  
353 16 – 26).

$$Sea\ salt = ssNa^+ \cdot \left( \frac{total\ mass}{Na^+} \right)_{ss} \quad 15$$

$$\%Cl^- \text{ depletion} = 100 \cdot \frac{ssNa^+ \cdot \left( \frac{Cl^-}{Na^+} \right)_{ss} - Cl^-_{bulk}}{ssNa^+ \cdot \left( \frac{Cl^-}{Na^+} \right)_{ss}} \quad 16$$

$$Lost\ Cl^- = ssNa^+ \cdot \left( \frac{Cl^-}{Na^+} \right)_{ss} - Cl^-_{bulk} \quad 17$$

$$Lost\ Cl^-_{bulk} = Na^+_{bulk} \cdot \left( \frac{Cl^-}{Na^+} \right)_{ss} - Cl^-_{bulk} \quad 18$$

$$Lost\ Cl^-_{diff} = Lost\ Cl^-_{bulk} - Lost\ Cl^- \quad 19$$

$$nssSO_4^{2-} = SO_{4,bulk}^{2-} - ssNa^+ \cdot \left( \frac{SO_4^{2-}}{Na^+} \right)_{ss} \quad 20$$

$$ExSO_4^{2-} = nssSO_4^{2-} - \frac{MW_{SO_4^{2-}}}{MW_{NH_4^+}} \cdot \frac{NH_{4,bulk}^+}{y_{SO_4^{2-}}} \quad 21$$

$$ExNH_4^+ = NH_{4,bulk}^+ - \frac{MW_{NH_4^+}}{MW_{SO_4^{2-}}} \cdot y_{SO_4^{2-}} \cdot nssSO_4^{2-} \quad 22$$

$$ExNO_3^- = NO_{3,bulk}^- - \frac{MW_{NO_3^-}}{MW_{NH_4^+}} \cdot \frac{ExNH_4^+}{y_{NO_3^-}} \quad 23$$

$$Excess\ acidic\ species = ExSO_4^{2-} + ExNO_3^- + oxalate_{bulk} \quad 24$$

$$Lost\ Cl^- \text{ attr. to } A = [A] \cdot y_A \cdot \frac{MW_{Cl^-}}{MW_A} \quad 25$$

$$Lost\ Cl^- \text{ attr. to excess acidic species} = \sum_{A=ExSO_4^{2-}, ExNO_3^-, oxalate_{bulk}} Lost\ Cl^- \text{ attr. to } A \quad 26$$

354

355 We first calculate the percentage of  $Cl^-$  originally in sea salt particles that has been displaced  
 356 by acidic species ( $\%Cl^-$  depletion; Eq. 16) to facilitate comparisons between our results and other  
 357 studies. Subsequently, mass concentrations of displaced  $Cl^-$  are calculated using two approaches  
 358 to explore the effects of accounting for non-sea salt sources of  $Na^+$ : Approach 1 quantifies  
 359 displaced  $Cl^-$  using derived mass concentrations of  $ssNa^+$  (lost  $Cl^-$ ; Eq. 17), while Approach 2  
 360 determines displaced  $Cl^-$  using bulk PILS  $Na^+$  mass concentrations (lost  $Cl^-_{bulk}$ ; Eq. 18), thus  
 361 assuming sea salt is the only source of  $Na^+$ . Mass concentrations of lost  $Cl^-_{bulk}$  will always be  
 362 greater than corresponding values of lost  $Cl^-$ , and differences between the two (lost  $Cl^-_{diff}$ ; Eq. 19)  
 363 are used to assess the significance in accounting for non-sea salt sources of  $Na^+$  when evaluating  
 364 the extent of  $Cl^-$  depletion processes and their potential effects on atmospheric chemistry.

365 As mentioned above, acidic species are responsible for displacing  $\text{Cl}^-$  from sea salt particles.  
366 However, only a subset of the bulk PILS mass concentrations of  $\text{SO}_4^{2-}$  and  $\text{NO}_3^-$  are available for  
367  $\text{Cl}^-$  depletion reactions, as (i)  $\text{SO}_4^{2-}$  is a naturally occurring component of sea salt and (ii) available  
368  $\text{NH}_4^+$  will neutralize certain amounts of  $\text{SO}_4^{2-}$  and potentially  $\text{NO}_3^-$ , leaving them relatively  
369 unreactive. Equations 20 – 23 determine mass concentrations of non-sea salt, unneutralized  $\text{SO}_4^{2-}$   
370 , and  $\text{NO}_3^-$ , which are added to bulk PILS mass concentrations of oxalate to quantify the amount  
371 of excess acidic species (Eq. 24) available for displacing  $\text{Cl}^-$  from sea salt particles. Note that we  
372 use oxalate here as a proxy variable to represent organic acids in general as it is typically the most  
373 abundant organic acid in tropospheric aerosol particles (e.g., Hilario et al., 2021; Ziemba et al.,  
374 2011; Cruz et al., 2019). We calculate the theoretical amount of lost  $\text{Cl}^-$  attributable to each excess  
375 acidic species (Eq. 25) as well as the total amount attributed to all measured excess acidic species  
376 (Eq. 26). Results from Eq. 26 can be compared to values from Eq. 17 to identify the amount of lost  
377  $\text{Cl}^-$  explained by the measured excess acidic species, and discrepancies in these values may indicate  
378 there are additional species contributing to  $\text{Cl}^-$  depletion (e.g., weak organic acids [Laskin et al.,  
379 2012]; reactions initiated by  $\text{O}_3$  [Keene et al., 1990]).

## 380 **2.5 MERRA-2 and NAAPS reanalysis products**

381 Wind speed and wind direction at 950 hPa were obtained from the Modern-Era Retrospective  
382 Analysis for Research and Application, Version 2 (MERRA-2; Gelaro et al., 2017) to provide  
383 context for large-scale boundary layer wind patterns over the region during each season/category  
384 and/or flights of interest. Monthly averages were attained for December 2021 and January,  
385 February, March, May, and June 2022 at  $0.5^\circ \times 0.625^\circ$  spatial resolution, while 3-hour averages  
386 were acquired for periods pertinent to each transit flight as well as the case study flights discussed  
387 in Sects. 3.2 and 3.7.1. Monthly averages for December 2021, January 2022, and February 2022  
388 were combined and averaged to produce a single wind vector field representative of the December-  
389 February category, while averages for March, May, and June 2022 are used to portray conditions  
390 for the March, May, and June Bermuda categories, respectively. The 950 hPa pressure layer was  
391 selected as this is the Falcon’s median pressure altitude during the Winter 2022 and Summer 2022  
392 deployments.

393 We relied on the Navy Aerosol Analysis and Prediction System (NAAPS) to identify the  
394 presence of surface-level dust and smoke over the region on selected days using images from the  
395 Aerosol Modeling archive (<https://www.nrlmry.navy.mil/aerosol/>) for the “Eastern United States”  
396 and “Tropical Atlantic.” We selected images at 1800Z for each day as this time is most relevant to  
397 flights during the Winter 2022 and Summer 2022 deployments. NAAPS surface dust and smoke  
398 mass concentrations are gridded reanalysis products available at  $1^\circ \times 1^\circ$  spatial resolution and 6-  
399 hourly temporal resolution, where simulations of dust depend on surface erodible fraction and  
400 surface friction velocity (Lynch et al., 2016), and those of smoke depend on size and duration of  
401 satellite-detected hotspots (Reid et al., 2009; Hyer et al., 2013). Modeled atmospheric transport of  
402 dust and smoke particles is then governed by the Navy Global Environmental Model (NAVGEM;  
403 Hogan et al., 2014). These products are used to explore how influence from dust and smoke plumes  
404 may affect calculations of  $\text{Cl}^-$  depletion for case studies presented in Sect. 3.7.1.

405

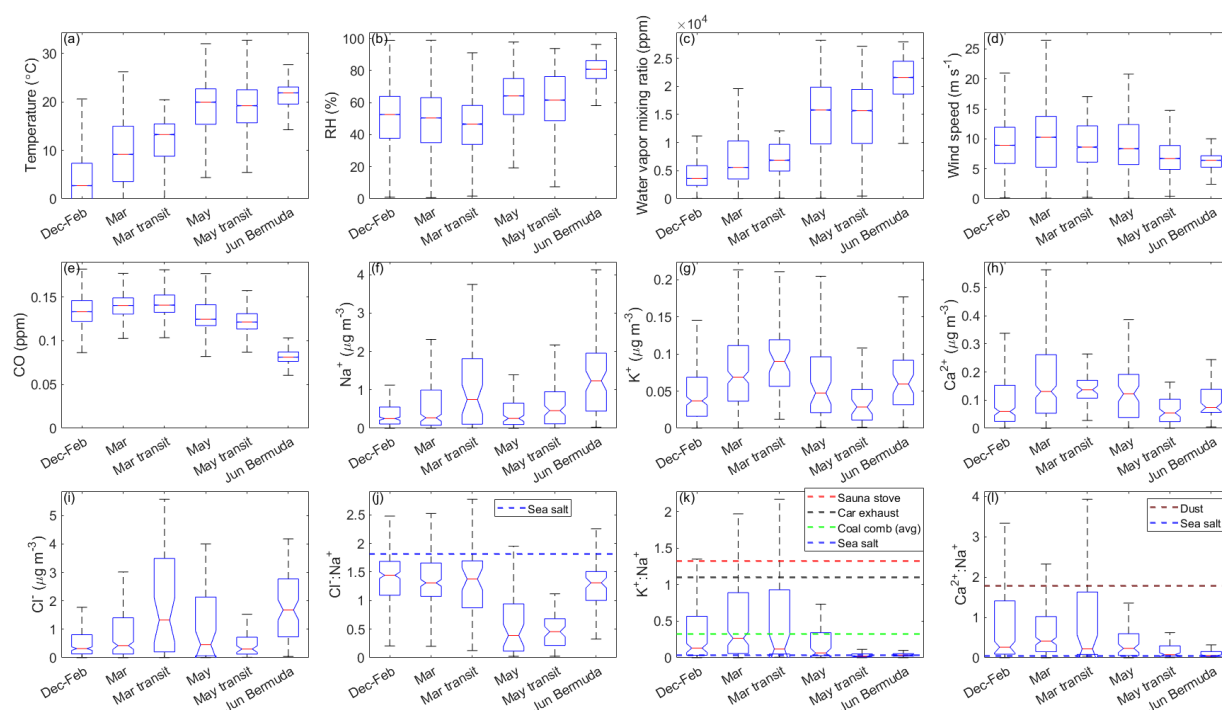
## 406 **3. Results and discussion**

### 407 **3.1 Meteorological context**

408 Meteorological conditions during the Winter 2022 and Summer 2022 deployments are mostly  
409 consistent with climatological characteristics reported for the NWA in Sorooshian et al. (2020)  
410 and Painemal et al. (2021). Median temperatures are lowest in December-February ( $2.7^{\circ}\text{C}$ )  
411 followed by March ( $9.2^{\circ}\text{C}$ ), March transit ( $13.3^{\circ}\text{C}$ ; recall the March transit flights are in late  
412 March), May transit ( $19.2^{\circ}\text{C}$ ), May ( $19.9^{\circ}\text{C}$ ), and June Bermuda ( $21.9^{\circ}\text{C}$ ; Fig. 1). Median water  
413 vapor mixing ratios and RH follow the same trend with the exception that RH slightly decreases  
414 from December-February (53%) to March (50%) and March transit (47%). Median wind speeds  
415 are highest for March ( $10.3\text{ m s}^{-1}$ ), similar for December-February, March transit, and May (8.9,  
416 8.6, and  $8.4\text{ m s}^{-1}$ , respectively), and lowest for May transit and June Bermuda ( $6.7$  and  $6.4\text{ m s}^{-1}$ ,  
417 respectively). MERRA-2 wind fields at 950 hPa (e.g., Fig. 2) show westerly flow along the USEC  
418 for December-February that transitions to southwesterly flow for March and March transit, which  
419 is a typical progression as the Bermuda High begins to strengthen (Davis et al., 1997). For May  
420 and May transit, zonal flow returns north of  $34^{\circ}\text{N}$  while relatively weak southwesterly flow  
421 persists to the south. Southwesterly winds dominate for June Bermuda, and large-scale flow  
422 patterns across the NWA appear conventional for a fully developed summertime Bermuda High.

423 Median CO volume mixing ratios are highest for categories sampling solely along the USEC  
424 (i.e., December-February [133 ppb], March [141 ppb], and May [124 ppb]) compared to June  
425 Bermuda (81 ppb), affirming sampled coastal air masses were most influenced by anthropogenic  
426 emissions. We refrain from using CO to compare levels of anthropogenic influence between  
427 categories focused on the USEC as CO exhibits seasonal dependence over the NWA (Buchholz et  
428 al., 2021). Specifically, peak values are typically observed in early spring due to wintertime  
429 accumulation caused by reduced destruction by OH, while increased rates of oxidation by OH over  
430 summer lead to minimum concentrations in late summer.

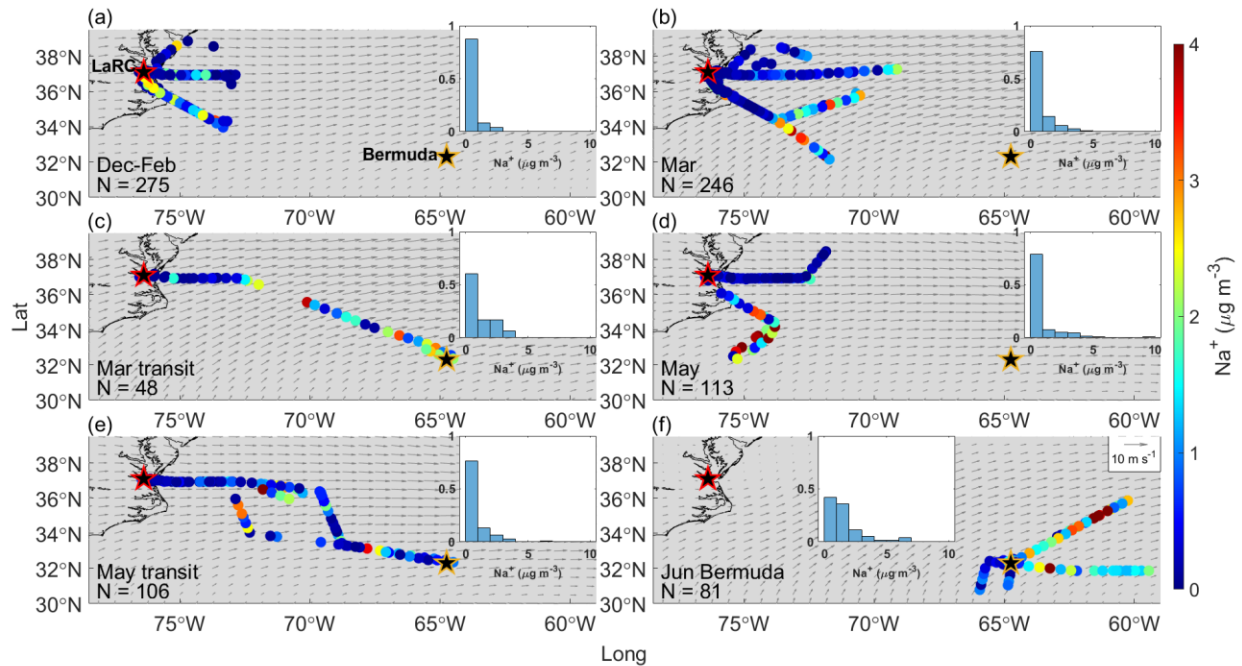
431 Precipitation is considered in this work as (i) wet scavenging processes remove sea salt  
432 particles more efficiently than several other particle types (Galloway et al., 1993), and (ii) strong  
433 winds associated with precipitation events can enhance sea salt emissions and offset scavenging  
434 losses (Dadashazar et al., 2021; Grandey et al., 2011), both of which can influence the amount of  
435  $\text{Cl}^{-}$  available for depletion reactions on shorter time scales than the seasonal factors discussed  
436 above. The NWA receives the most rainfall from December – February followed by June – August,  
437 with precipitation rates peaking along the Gulf Stream in all seasons (Painemal et al., 2021).  
438 Hawcroft et al. (2012) showed that 65 – 80% and 50 – 70% of the rainfall over the NWA in  
439 December – February and June – August, respectively, is associated with midlatitude cyclones  
440 (MLC)s, a common year-round weather phenomenon for the region (e.g., Braun et al., 2021;  
441 Eichler and Higgins, 2006) largely dictating the eastward transport of trace gases and particulates  
442 from North America to the adjacent marine environment (Keim et al., 2005; Cooper et al., 2002,  
443 2001). Despite their frequency and known effects on other aerosol properties (e.g., aerosol optical  
444 depth and size distribution; Grandey et al., 2011), there is uncertainty in how frontal passages  
445 influence parameters relevant to  $\text{Cl}^{-}$  depletion over the NWA. During this study, meteorological  
446 conditions were often driven by MLCs, with synoptic conditions changing every few days (Table  
447 2). We discuss key variables in the context of prefrontal and postfrontal airstreams associated with  
448 MLCs to explore the influence of midlatitude weather disturbances on depletion reactions and Cl  
449 radical budgets over the NWA. Finally, note that clear-ensemble data for December-February do  
450 not extend eastward of  $\sim 73^{\circ}\text{W}$  due to frequent cloud cover below 3 km over the ocean. This should  
451 be taken into consideration when comparing results for December-February to other categories,  
452 especially for continentally sourced properties and/or those that depend on wind fetch.



453

454 **Figure 1.** Notched box plots showing seasonal/categorical differences in (a) temperature, (b)  
 455 relative humidity (RH), (c) water vapor mixing ratio, (d) wind speed, (e) carbon monoxide (CO)  
 456 mixing ratios, bulk mass concentrations from a particle into liquid sampler (PILS) of (f) chloride  
 457 (Cl<sup>-</sup>), (g) sodium (Na<sup>+</sup>), (h) potassium (K<sup>+</sup>), and (i) calcium (Ca<sup>2+</sup>), as well as ratios of these mass  
 458 concentrations for (j) Cl<sup>-</sup>:Na<sup>+</sup>, (k) K<sup>+</sup>:Na<sup>+</sup>, and (l) Ca<sup>2+</sup>:Na<sup>+</sup>. Data are from clear ensembles only.  
 459 Typical ratios for particular ions in sea salt and/or dust are marked with dashed lines in j, k, and l.  
 460 In k, we use additional lines to indicate ratios of K<sup>+</sup>:Na<sup>+</sup> reported in the literature for inefficient  
 461 batch combustion in a sauna stove (1.33; Lamberg et al., 2011), car exhaust (1.1; Huang et al.,  
 462 1994), and coal combustion (0.33; Ondov et al., 1989). The solid red line in the center of each box  
 463 indicates the median, box edges represent the 25<sup>th</sup> and 75<sup>th</sup> percentiles, and the lower and upper  
 464 whiskers indicate the lower limit (first quartile - 1.5 × interquartile range) and upper limit (third  
 465 quartile + 1.5 × interquartile range), respectively. The notches span the 95<sup>th</sup> confidence interval  
 466 for the median.





467

468 **Figure 2.** Bulk PILS  $\text{Na}^+$  mass concentrations from clear ensembles during (a) December 2021-  
 469 February 2022, (b) March 2022, (c) March 2022 transit flights between NASA Langley Research  
 470 Center (LaRC; marked with a red-edged star) and Bermuda (marked with a golden-edged star), (d)  
 471 May 2022, (e) May 2022 transit flights between LaRC and Bermuda, and (f) the Bermuda field  
 472 campaign in June 2022. Normalized histograms in each panel show the distribution of bulk PILS  
 473  $\text{Na}^+$  mass concentrations for that specific category since overlap among the colored dots can hide  
 474 some from view. Grey arrows indicate the average magnitude and direction of winds at 950 hPa  
 475 from MERRA-2 for the period relevant to each category.

476 **Table 2.** Dates, sample quantities, meteorological conditions, and aerosol particle properties  
 477 relevant to Cl<sup>-</sup> depletion for research flights (RFs) considered in each category. Median values of  
 478 Na<sup>+</sup><sub>bulk</sub> and Ca<sup>2+</sup><sub>bulk</sub> are based on bulk PILS data while values of lost Cl<sup>-</sup>, Cl<sup>-</sup> depletion, and excess  
 479 acidic species are derived using Eqs. 1 – 5, 16, 17, and 20 - 24. "N PILS samples" refers to the  
 480 total number of PILS samples collected during clear ensembles on the date indicated, while "N<sub>PILS</sub>"  
 481 refers to the number of these samples providing enough information to determine a given property.  
 482 "N<sub>PILS&AMS</sub>" refers to the number of coinciding mass concentrations from the PILS and aerosol  
 483 mass spectrometer (AMS) necessary for calculating excess acidic species mass concentrations.

Category	Date	RF(s)	N PILS samples	Meteorological conditions and/or relevant notes	Na <sup>+</sup> <sub>bulk</sub>		Ca <sup>2+</sup> <sub>bulk</sub>		Lost Cl <sup>-</sup>		Cl <sup>-</sup> depletion		Excess acidic species	
					Median (μg m <sup>-3</sup> )	N <sub>PILS</sub>	Median (μg m <sup>-3</sup> )	N <sub>PILS</sub>	Median (μg m <sup>-3</sup> /pptv)	N <sub>PILS</sub>	Median (%)	N <sub>PILS</sub>	Median (μg m <sup>-3</sup> )	N <sub>PILS &amp; AMS</sub>
Dec-Feb	30 November 2021	94	7	Remains of post-frontal conditions	0.14	7	0.31	7	-0.17/NA <sup>1</sup>	7	0	7	0.29	13
	01 December 2021	95	16	Prefrontal, high pressure; smoke in boundary layer near coast	0.30	16	0.49	16	-0.16/NA <sup>1</sup>	16	0	16	0.59	136
	07 December 2021	96	5	Postfrontal, cold high pressure behind a strong cold front	0.19	5	0.20	5	-0.12/NA <sup>1</sup>	5	0	5	0.03	22
	11 January 2022	100, 101	6	Cold high pressure, cold air outbreak (CAO) conditions	0.34	4	0.05	6	0.12/80	4	20	4	0.49	21
	12 January 2022	102, 103	33	Cold high pressure	0.21	29	0.06	21	0.01/7	15	4	15	0.20	109
	15 January 2022	104	3	Postfrontal	0.63	3	0.05	2	0.01/7	2	4	2	0.35	20
	18 January 2022	105	11	Low pressure moves offshore, sets up CAO conditions	0.22	2	0.06	2	NaN	0	NaN	0	0.01	10
	19 January 2022	107, 108	26	Short-lived high pressure	0.24	14	0.06	10	-0.05/NA <sup>1</sup>	6	0	6	0.14	66
	24 January 2022	109, 110	26	Postfrontal, weak high pressure	0.07	15	0.03	13	-0.04/NA <sup>1</sup>	8	0	8	0.02	86
	26 January 2022	111, 112	20	Postfrontal	0.12	12	0.03	10	0.00/0	7	0	7	0.01	83
	27 January 2022	113, 114	18	Cold high pressure	0.25	16	0.01	5	0.06/40	5	21	5	0.36	41
	01 February 2022	115	8	High pressure	0.90	6	0.05	7	0.41/273	5	21	5	1.00	37
	02 February 2022	116	17	High pressure	0.73	16	0.03	6	0.18/120	6	12	6	0.41	44
	03 February 2022	117, 118	15	High pressure	1.03	14	0.03	5	0.04/27	5	2	5	0.00	10
	15 February 2022	120, 121	34	Postfrontal conditions, cold high pressure	0.25	27	0.03	24	0.08/53	21	17	21	0.56	69
16 February 2022	122, 123	21	Cold high pressure	0.20	18	0.08	20	0.10/67	16	27	16	0.53	105	
19 February 2022	124, 125	38	Weak postfrontal	0.12	30	0.06	37	0.06/40	23	24	23	0.24	186	

	22 February 2022	126, 127	25	Prefrontal, high pressure	1.41	25	0.12	24	0.45/300	24	17	24	0.64	184
	26 February 2022	128, 129	16	Postfrontal	0.13	16	0.06	15	-0.02/NA <sup>1</sup>	15	0	15	0.27	130
	<b>Overall</b>		<b>345</b>		<b>0.25</b>	<b>275</b>	<b>0.06</b>	<b>235</b>	<b>0.04/27</b>	<b>190</b>	<b>6</b>	<b>190</b>	<b>0.30</b>	<b>1372</b>
Mar	02 March 2022	130	39	Postfrontal, high pressure	0.30	36.00	0.16	39	0.04/27	33	8	33	1.20	298
	03 March 2022	131, 132	71	Weak prefrontal	0.91	57.00	0.27	71	0.10/67	57	9	57	1.19	537
	04 March 2022	133, 134	42	Cold high pressure	1.56	40.00	0.12	39	0.42/280	36	14	36	1.02	242
	13 March 2022	138	8	Postfrontal, CAO conditions	0.12	6.00	0.06	7	-0.12/NA <sup>1</sup>	6	0	6	0.02	22
	14 March 2022	139, 140	38	Late postfrontal, cold high pressure; smoke plume sampled from a woodland fire	0.16	37.00	0.06	37	0.03/20	35	13	35	0.22	305
	18 March 2022	141	14	Weak postfrontal	0.18	14.00	0.04	12	0.05/33	12	35	12	0.33	98
	26 March 2022	144, 145	29	Postfrontal; sampled dust, smoke, and potentially pollen	0.05	22.00	0.04	22	-0.02/NA <sup>1</sup>	13	0	13	0.00	147
	28 March 2022	146	17	Postfrontal	0.07	17.00	0.05	12	-0.01/NA <sup>1</sup>	10	0	10	0.13	98
	29 March 2022	147, 148	19	Postfrontal, high pressure, CAO conditions	0.21	17.00	0.05	5	0.02/13	4	34	4	0.00	43
	<b>Overall</b>		<b>277</b>		<b>0.27</b>	<b>246</b>	<b>0.13</b>	<b>244</b>	<b>0.04/27</b>	<b>206</b>	<b>10</b>	<b>206</b>	<b>0.57</b>	<b>1790</b>
May	03 May 2022	149	15	Weak prefrontal; presence of smoke potentially from New Mexico	0.42	15	0.14	12	0.89/594	7	85	7	0.03	92
	05 May 2022	150, 151	18	Postfrontal	0.05	14	0.04	14	0.42/280	2	89	2	0.02	91
	16 May 2022	153, 154	39	Prefrontal to an approaching cold front yet also postfrontal to a departing band of precipitation	0.26	39	0.26	7	0.65/434	1	73	1	0.05	85
	17 May 2022	155	37	Postfrontal	0.08	17	0.01	13	1.53/1020	2	73	2	0.05	52
	20 May 2022	158	28	Warm high pressure, southerly flow due to Bermuda high <sup>2</sup> ; haze with potential sampling of bioaerosol	1.75	28	0.17	27	1.91/1274	21	48	21	0.97	148
	<b>Overall</b>		<b>137</b>		<b>0.26</b>	<b>113</b>	<b>0.12</b>	<b>73</b>	<b>1.76/1174</b>	<b>33</b>	<b>64</b>	<b>33</b>	<b>0.05</b>	<b>468</b>
Mar transit	22 March 2022	142, 143	48	High pressure, two days after a cold front and two days before another cold front	0.75	48	0.14	48	0.11/73	43	9	43	0.36	423
May transit	18 May 2022	156, 157	67	Postfrontal along East Coast, aircraft passed across the cold front on the way to Bermuda	0.51	58	0.05	50	1.37/914	31	74	31	0.27	216

	21 May 2022	159, 160	42	Warm high pressure, anticyclonic flow around Bermuda high	0.50	37	0.08	26	1.67/1114	17	75	17	1.87	137
	31 May 2022	161	11	Postfrontal	0.18	11	0.02	5	0.22/147	5	67	5	0.02	20
	<b>Overall</b>		<b>120</b>		<b>0.46</b>	<b>106</b>	<b>0.05</b>	<b>81</b>	<b>1.33/887</b>	<b>53</b>	<b>74</b>	<b>53</b>	<b>0.44</b>	<b>373</b>
Jun Bermuda	02 June 2022	162, 163	4	Prefrontal	0.64	4	0.03	3	0.71/474	2	44	2	2.62	12
	03 June 2022	164	1	Prefrontal, tropical system approaching from the southwest	0.30	1	NaN	0	NaN	0	NaN	0	0.02	1
	05 June 2022	165	29	Could only fly in the morning due to approaching tropical cyclone (TC), TC departs 06 June 2022.	1.76	29	0.08	26	1.35/900	26	36	26	1.97	213
	07 June 2022	167	1	High behind departing TC	2.21	1	NaN	0	NaN	0	NaN	0	0.02	1
	08 June 2022	168, 169	2	High pressure behind TC, African dust known to be in domain	4.28	2	1.07	1	1.12/747	1	11	1	0.04	9
	10 June 2022	170	1	High pressure, isolated thunderstorms, African dust known to be in domain	2.28	1	0.06	1	0.68/454	1	17	1	1.19	9
	11 June 2022	172, 173	20	High pressure, African dust known to be in domain	0.33	20	0.21	12	0.15/100	11	11	11	1.12	71
	13 June 2022	174	25	High pressure, African dust known to be in domain but sampled away from dust for contrast	1.34	23	0.06	24	0.48/320	23	17	23	1.89	170
	<b>Overall</b>		<b>83</b>		<b>1.24</b>	<b>81</b>	<b>0.07</b>	<b>67</b>	<b>0.66/440</b>	<b>64</b>	<b>25</b>	<b>64</b>	<b>1.82</b>	<b>486</b>

484 <sup>1</sup>Negative mass concentrations in  $\mu\text{g m}^{-3}$  are reported for lost  $\text{Cl}^-$  and can be conceptualized as the  
485 amount of measured particulate  $\text{Cl}^-$  in excess of what would be in unreacted sea salt particles based  
486 on Eqs. 1 – 4. Negative values may suggest there are additional non-sea salt sources of particulate  
487  $\text{Cl}^-$  within the sampled air mass. In these cases, we do not provide corresponding gas phase  
488 concentrations of lost  $\text{Cl}^-$  in pptv as these are only meaningful when  $\text{Cl}^-$  is displaced from sea salt  
489 particles.

490 <sup>2</sup>Davis et al. (1997)

### 491 **3.2 Seasonal, spatial, and frontal trends in Na<sup>+</sup>**

492 Cl<sup>-</sup> depletion studies are motivated by the fact that radicals produced via depletion reactions  
493 can influence atmospheric chemistry, the extent to which largely depends on the quantity of  
494 radicals generated. Therefore, the amount of Cl<sup>-</sup> in sea salt available to depletion reactions is  
495 critical to quantify, which is why a large portion of our initial discussion is about trends in bulk  
496 Na<sup>+</sup> mass concentrations as they are a reliable indicator of sea salt mass concentrations. Bulk PILS  
497 Na<sup>+</sup> mass concentrations are remarkably similar for December-February, March, and May (median  
498 mass concentrations of 0.25, 0.27, and 0.26  $\mu\text{g m}^{-3}$ , respectively), higher for March transit and  
499 May transit (0.75 and 0.46  $\mu\text{g m}^{-3}$ , respectively), and highest in and around Bermuda (1.24  $\mu\text{g m}^{-3}$ ).  
500 In general, past works also typically report higher sea salt mass concentrations in open-ocean  
501 environments compared to coastal locations (Table S4), which is intuitive considering that wind  
502 fetch is one important factor governing atmospheric sea salt mass concentrations. However, if Na<sup>+</sup>  
503 mass concentrations were dictated chiefly by wind fetch over the NWA, values would mostly  
504 increase moving eastward, which is not always the case (e.g., Fig. 2e). In fact, there does not appear  
505 to be any distinct spatial gradients in Na<sup>+</sup> mass concentrations for the seasons/categories presented,  
506 yet (i) overlap of flight tracks makes it difficult to view all mass concentrations at once, and (ii)  
507 we do not have enough data to state that this is always true for the region.

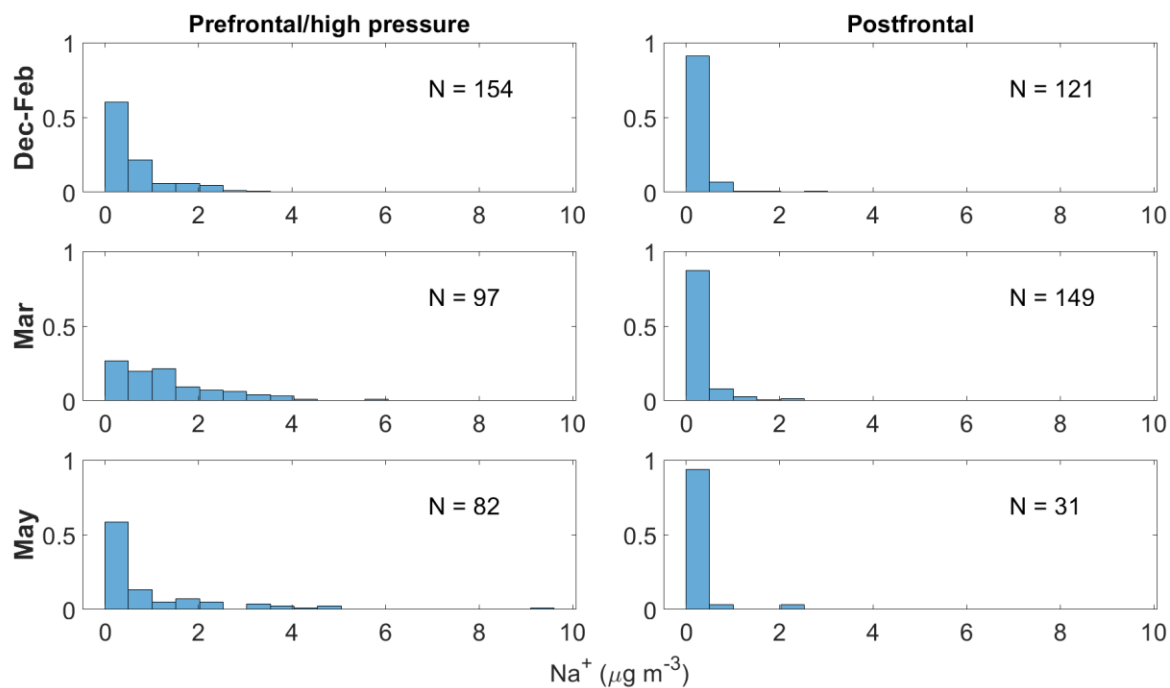
508 Aside from wind fetch, removal via wet scavenging processes is another factor dictating sea  
509 salt mass concentrations over marine environments. We explore the effect of passing frontal  
510 systems on bulk Na<sup>+</sup> mass concentrations for December-February, March, and May as (i) bulk Na<sup>+</sup>  
511 appears seasonally independent among these categories and (ii) flights sampled the same general  
512 region, allowing us to remove coastal versus open-ocean sampling as a confounding variable.  
513 When applying the meteorological conditions identified for each day in Table 2, bulk Na<sup>+</sup> mass  
514 concentrations are generally higher during prefrontal/high pressure conditions compared to  
515 postfrontal scenes for each seasonal/monthly category (Fig. 3). It is not unusual for bulk Na<sup>+</sup> mass  
516 concentrations to exceed 3  $\mu\text{g m}^{-3}$  in prefrontal and/or high-pressure conditions, especially in  
517 March and May, yet values never exceed this threshold in postfrontal conditions. Although bulk  
518 statistics suggest frontal passages may reduce sea salt mass concentrations over the NWA, data  
519 from prefrontal and postfrontal conditions are not guaranteed to be linked, meaning samples  
520 quantifying bulk Na<sup>+</sup> before and after each frontal passage are not always available. Therefore, we  
521 isolate bulk Na<sup>+</sup> mass concentrations for flights straddling frontal passages to assess the  
522 relationship of sea salt mass concentrations and MLCs on a case-study level.

523 Postfrontal conditions on 19 February 2022 (RFs 124 and 125) are associated with bulk Na<sup>+</sup>  
524 mass concentrations mostly < 0.3  $\mu\text{g m}^{-3}$  and moderate westerly winds bringing continental air  
525 over the NWA (Fig. S1). Three days later (22 February 2022; RFs 126 and 127), prefrontal  
526 conditions show increased bulk Na<sup>+</sup> mass concentrations that are distributed evenly from 0.3 – 2.7  
527  $\mu\text{g m}^{-3}$  and southerly winds along the coast. Bulk Na<sup>+</sup> mass concentrations then swiftly decrease  
528 to values mostly below 0.3  $\mu\text{g m}^{-3}$  by 26 February 2022 (RFs 128 and 129) as another MLC moves  
529 through the region, although note clear-ensemble sampling was more restricted to the coastline on  
530 this day compared to 19 and 22 February 2022.

531 Unfortunately, samples straddling a frontal passage for March are unavailable, but we use  
532 consecutive flights from 02 – 04 March 2022 to depict the “recharge” of sea salt mass  
533 concentrations following a MLC (Fig. S2). Bulk Na<sup>+</sup> mass distributions gradually shift towards  
534 larger mass concentrations moving from postfrontal conditions with gentle westerly winds (02  
535 March 2022; RF 130), to weak prefrontal conditions with stronger northwesterly and southwesterly  
536 winds converging at 36 °N (03 March 2022; RFs 131 and 132), and, finally, to cold high-pressure

537 conditions with moderate anticyclonic flow around a high over the northeastern U.S. (04 March  
538 2022; RFs 133 and 134). Air masses sampled on 02 and 03 March 2022 appear more continentally  
539 influenced and may have been more recently affected by large-scale precipitation compared to the  
540 marine air mass sampled on 04 March 2022, which is a potential explanation for the differences in  
541 bulk  $\text{Na}^+$  mass concentrations.

542 Flights on 16 May 2022 (RFs 153 and 154) sampled an air mass recently impacted by a  
543 retreating band of precipitation yet also considered as prefrontal due to an approaching cold front  
544 (Fig. S3). As there was limited time for sea salt mass concentrations to recharge between the  
545 consecutive MLCs, it is unsurprising there is little difference in bulk  $\text{Na}^+$  mass concentrations  
546 between 16 May 2022 and the postfrontal conditions sampled on 17 May 2022 (RF 155). Frontal  
547 influence dissipated by 20 May 2022 (RF 158) with southwesterly flow returning along the  
548 coastline in association with the strengthening Bermuda High. This, and the absence of  
549 precipitation for several days, may help explain the increase in bulk  $\text{Na}^+$  mass concentrations from  
550 mostly below  $1 \mu\text{g m}^{-3}$  on 16 – 17 May 2022 to a mostly above this value on 20 May 2022. The  
551 three case studies presented are meant to illustrate how rapidly sea salt mass concentrations can  
552 change over the NWA due, in part, to fluctuations in synoptic-scale wind patterns and/or large-  
553 scale precipitation associated with MLCs. However, we acknowledge that there are many other  
554 confounding atmospheric variables influencing sea salt mass concentrations during these case  
555 studies and that flight tracks do not cover the exact same locations on each of these days. Although  
556 we do not have enough data to make definitive claims, bulk statistical and case study analyses  
557 suggest sea salt mass concentrations decrease behind passing MLCs over the NWA, which  
558 corresponds to reduced potential in the amount of reactive chlorine-containing gases that could be  
559 produced via depletion reactions compared to in prefrontal and high-pressure conditions.



560

561 **Figure 3.** Normalized histograms showing differences in bulk PILS  $\text{Na}^+$  mass concentrations from  
 562 clear ensembles occurring in prefrontal and/or high-pressure versus postfrontal conditions for  
 563 December-February (top row), March (middle row), and May (bottom row). These categories are  
 564 shown as they represent flights occurring in and around the East Coast, eliminating coastal versus  
 565 open-ocean sampling as a confounding variable.

### 566 3.3 Seasonal trends in $K^+$ , $Ca^{2+}$ , $Cl^-$ and ion mass ratios

567 As described above, the NWA receives BB emissions from continuous sources (e.g.,  
568 fossil fuel combustion for transportation and industrial efforts along the USEC), seasonal practices  
569 (e.g., agricultural waste burning in spring, wood burning in winter), and intermittent yet influential  
570 events (e.g., forest fires). Using  $K^+$  as a tracer for such activities, BB influence is greatest during  
571 March and March transit flights with median bulk  $K^+$  mass concentrations of 0.07 and 0.09  $\mu\text{g m}^{-3}$ ,  
572 respectively, compared to 0.04, 0.05, 0.03 and 0.06  $\mu\text{g m}^{-3}$  for the December-February, May,  
573 May transit, and June Bermuda categories, respectively. This agrees with previous findings where  
574 mass concentrations of organic carbon and particles with diameters 2.5 – 10  $\mu\text{m}$  ( $PM_{\text{coarse}}$ ) were  
575 much higher in March than in any other month at a coastal site in Florida (Edwards et al., 2021),  
576 and this was attributed mostly to the annual peak in prescribed burning across the southeastern  
577 U.S. (Jaffe et al., 2020; McCarty et al., 2007). Our bulk  $K^+$  mass concentrations are comparable to  
578 mean values reported at a receptor site for BB and urban emissions from East Asia (0.02 – 0.05  $\mu\text{g}$   
579  $\text{m}^{-3}$ ; Boreddy and Kawamura, 2015) as well as those in polluted air masses containing dust (0.03  
580  $\mu\text{g m}^{-3}$ ) and biogenically influenced air masses (0.03  $\mu\text{g m}^{-3}$ ) over the southeastern U.S. during the  
581 Study of Emissions and Atmospheric Composition, Clouds, and Climate Coupling by Regional  
582 Surveys (SEAC<sup>4</sup>RS; Kacenelenbogen et al., 2022). However, bulk  $K^+$  values are mostly lower than  
583 average  $K^+$  mass concentrations in air masses influenced by agricultural burning (0.10  $\mu\text{g m}^{-3}$ ) and  
584 wildfire emissions (0.09  $\mu\text{g m}^{-3}$ ) during SEAC<sup>4</sup>RS (Kacenelenbogen et al., 2022) and also lower  
585 than average mass concentrations (0.82  $\mu\text{g m}^{-3}$ ) measured during the Fire Influence on Regional to  
586 Global Environments and Air Quality (FIREX-AQ) airborne field campaign (Adachi et al., 2022)  
587 sampling BB plumes in the western and southeastern U.S. Thus, BB particles were consistently  
588 present during the Winter 2022 and Summer 2022 deployments, yet relatively dilute compared to  
589 their levels in air masses more heavily influenced by BB processes. This is an important point to  
590 consider when contemplating how BB emissions may affect estimates of  $Cl^-$  depletion, which is  
591 discussed in greater detail in Sect. 3.7.2.

592 We use bulk  $Ca^{2+}$  to identify influence from dust particles and see a similar trend as above  
593 where median bulk  $Ca^{2+}$  mass concentrations are higher in certain spring categories (0.13, 0.14,  
594 and 0.12  $\mu\text{g m}^{-3}$  for the March, March transit, and May categories, respectively) compared to  
595 December-February (0.06  $\mu\text{g m}^{-3}$ ) and June Bermuda (0.07  $\mu\text{g m}^{-3}$ ). Higher springtime bulk  $Ca^{2+}$   
596 mass concentrations are likely due to periodic influence from Asian dust plumes, which arrive  
597 most frequently over the region from March-May (Aldhaif et al., 2020), and/or to increased  
598 suspension of dust particles in BB plumes from agricultural fires across the eastern and  
599 southeastern U.S. due to turbulent mixing around flames and the burn front (e.g., Kavouras et al.,  
600 2012; Popovicheva et al., 2014; Maudlin et al., 2015; Schlosser et al., 2017; Palmer, 1981).  
601 Interestingly, bulk  $Ca^{2+}$  mass concentrations are lowest for May transit (0.05  $\mu\text{g m}^{-3}$ ), but this may  
602 be explained by the episodic nature of dust events over the NWA (e.g., Wu et al., 2015; Perry et  
603 al., 1997; Prospero, 1999) and the fact that this category is comprised of only three days. African  
604 dust plumes become more common over the NWA from June-August (Zuidema et al., 2019) with  
605 the strengthening of the Bermuda High, yet the Summer 2022 deployment ended just as these  
606 plumes were becoming evident over the region (see meteorological notes for 10, 11, and 13 June  
607 2022 in Table 2). There does not appear to be distinct spatial trends in bulk  $Ca^{2+}$  over the region  
608 for most categories (Fig. S4), presumably as fluctuations in bulk  $Ca^{2+}$  may be largely driven by  
609 periodic influence from long-range dust transport, smoke plumes from fires along the USEC  
610 advecting over the ocean, and midlatitude weather disturbances (Fig. S5). However, a gradient  
611 seems to exist along the March transit flights (RFs 142 and 143 on 22 March 2022) such that bulk



612  $\text{Ca}^{2+}$  mass concentrations are highest to the east of LaRC and then decrease to the southeast  
613 towards Bermuda. This potential sampling of a dust plume and its implications on calculations  
614 relevant to  $\text{Cl}^-$  depletion are explored further in Sect. 3.7.1.

615 Median  $\text{Cl}^-$  mass concentrations exhibit slightly different seasonal trends than bulk  $\text{Na}^+$ , with  
616 values lowest for May transit ( $0.31 \mu\text{g m}^{-3}$ ), slightly higher for December-February, March, and  
617 May ( $0.32$ ,  $0.43$ , and  $0.46 \mu\text{g m}^{-3}$ , respectively), and much higher for March transit and Bermuda  
618 ( $1.33$  and  $1.68 \mu\text{g m}^{-3}$ , respectively). The fact that May transit has the third highest median bulk  
619  $\text{Na}^+$  mass concentration yet the lowest  $\text{Cl}^-$  median is the main difference in seasonal trends between  
620 these species, which may seem to suggest  $\text{Cl}^-$  depletion processes are most active for May transit.  
621 However, the number of PILS samples providing (i) bulk  $\text{Na}^+$  and (ii)  $\text{Cl}^-$  mass concentrations are  
622 very different for May (113 and 43, respectively) and May transit (106 and 65, respectively), yet  
623 comparable for December-February, March, March transit, and June Bermuda (Table S5). Thus,  
624 it is best to avoid drawing conclusions about  $\text{Cl}^-$  depletion from individual trends in bulk  $\text{Na}^+$  and  
625  $\text{Cl}^-$ , and to instead focus on samples providing mass concentrations for both species. These samples  
626 were isolated to generate the statistics shown in Fig. 1j, which (i) can be considered as a precursory  
627 analysis for  $\text{Cl}^-$  depletion over the NWA where sea salt is assumed to be the only source of  $\text{Na}^+$ ,  
628 and (ii) are directly comparable to many past works making this assumption. Ratios of  $\text{Cl}^-:\text{Na}^+$  are  
629 below 1.81 for all categories, suggesting  $\text{Cl}^-$  depletion processes are consistently occurring over  
630 the region. However, median values are much lower for May (0.39) and May transit (0.46)  
631 compared to December-February (1.44), March (1.31), March transit (1.38), and June Bermuda  
632 (1.31), suggesting that depletion reactions are particularly prevalent in late spring. May and May  
633 transit ratios are comparable to those previously reported along the USEC (Quinn and Bates, 2005;  
634 Nolte et al., 2008; Zhao and Gao, 2008) in late spring and summer, especially for submicron sea  
635 salt particles.

636 As mentioned above,  $\text{Cl}^-:\text{Na}^+$  ratios are only an appropriate means to illustrate the extent of  $\text{Cl}^-$   
637 depletion if sea salt is the predominant source of each species. Ratios of bulk  $\text{K}^+:\text{Na}^+$  and  $\text{Ca}^{2+}:\text{Na}^+$   
638 are useful for indicating if other particle types may be contributing to bulk  $\text{Na}^+$  concentrations as  
639 these ions are present in distinctly different proportions in sea salt, emissions from various  
640 combustion processes, and dust particles. Combustion and/or BB activities do not appear to  
641 contribute meaningfully to bulk  $\text{Na}^+$  for May, May transit, and June Bermuda as  $\text{K}^+:\text{Na}^+$  ratios  
642 ( $0.065$ ,  $0.020$ , and  $0.037$ , respectively) are fairly similar to the reference value for sea salt ( $0.036$ ;  
643 Seinfeld and Pandis, 2016; Finlayson-Pitts and Pitts, 2000), whereas ratios exceeding this value  
644 are observed for December-February ( $0.132$ ), March ( $0.267$ ), and March transit ( $0.119$ ). Table 2  
645 indicates smoke was only directly sampled on four days of the Winter 2022 and Summer 2022  
646 deployments (01 December 2021, 14 March 2022, 26 March 2022, and 03 May 2022), suggesting  
647 increased  $\text{K}^+:\text{Na}^+$  ratios for December-February, March, and March transit may have been driven  
648 by increased background levels of BB particles over the NWA from widespread and continuous  
649 residential wood burning and prescribed agricultural burning in winter and early spring as opposed  
650 to acute BB events. All categories have median  $\text{Ca}^{2+}:\text{Na}^+$  ratios exceeding the reference value for  
651 sea salt ( $0.038$ ; Bowen, 1979; Finlayson-Pitts and Pitts, 2000), with values of  $0.412$ ,  $0.261$ ,  $0.233$ ,  
652  $0.219$ ,  $0.075$ , and  $0.050$  for March, December-February, May, March transit, May transit, and June  
653 Bermuda, respectively. These results nicely motivate an investigation into how estimates of  $\text{Cl}^-$   
654 depletion change when eliminating contributions of (i) dust and (ii) both dust and combustion  
655 emissions to bulk  $\text{Na}^+$  mass concentrations, which are the topics of Sects. 3.7.1 and 3.7.2,  
656 respectively.

657

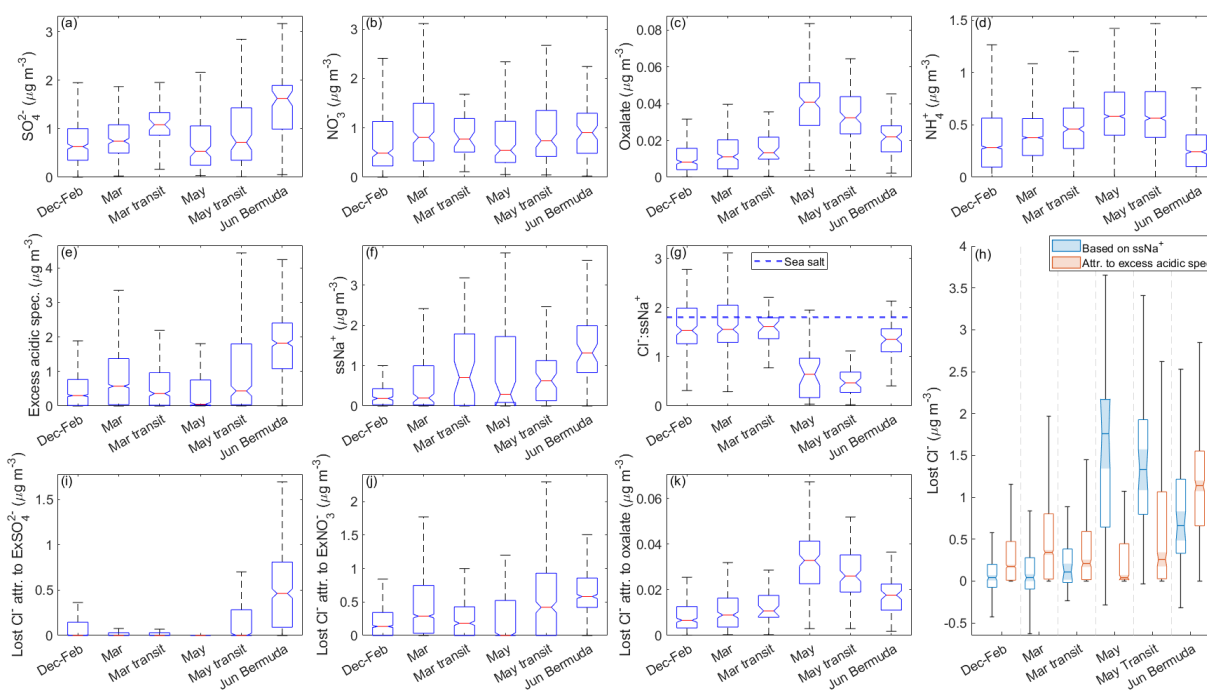
### 658 3.4 Seasonal, spatial, and frontal trends in acidic species

659 Sea salt mass concentrations alone control the maximum amount of reactive chlorine-  
660 containing gases that can be released via  $\text{Cl}^-$  depletion reactions, but available acidic species are  
661 an important factor in regulating the extent to which these reactive gases are actually released.  
662 Median mass concentrations of bulk  $\text{SO}_4^{2-}$  show that this acidic species is a common constituent  
663 of sampled air masses, especially for March transit and June Bermuda (Fig. 4; Table S6). Median  
664 bulk  $\text{NO}_3^-$  mass concentrations are of similar magnitude to bulk  $\text{SO}_4^{2-}$ , yet exhibit less variability  
665 among the categories, while oxalate is present in relatively low amounts for December-February,  
666 March, and March transit, increases sharply for May and May transit, and then decreases slightly  
667 for June Bermuda. In Sect. 2.4, we describe how  $\text{ssNa}^+$  mass concentrations and subsequently  
668 derived parameters can be calculated either by assuming (i) dust and sea salt or (ii) dust, sea salt,  
669 and combustion-sourced particles contribute to bulk  $\text{Na}^+$ . In this section and Sects. 3.5, 3.6, 3.7.1,  
670 and 3.7.3, we discuss values based on the first assumption, whereas those based on the second  
671 assumption are the topic of Sect. 3.7.2.

672 After accounting for contributions of sea salt to  $\text{SO}_4^{2-}$  and neutralization of non-sea salt  $\text{SO}_4^{2-}$   
673 and  $\text{NO}_3^-$  with  $\text{NH}_4^+$ , excess  $\text{SO}_4^{2-}$  ( $\text{ExSO}_4^{2-}$ ) is typically nonexistent for all categories except June  
674 Bermuda (median of  $0.63 \mu\text{g m}^{-3}$ ; Fig. S6), while a range of mass concentrations of excess  $\text{NO}_3^-$   
675 ( $\text{ExNO}_3^-$ ) remain for all categories except May (0.24, 0.51, 0.32, 0.74,  $1.02 \mu\text{g m}^{-3}$  for December-  
676 February, March, March transit, May transit, and June Bermuda, respectively). Thus, mass  
677 concentrations of measured acidic species available to participate in  $\text{Cl}^-$  depletion reactions are  
678 relatively low for May ( $0.05 \mu\text{g m}^{-3}$ ; contributed mostly by oxalate), moderate for December-  
679 February, March, March transit, and May transit ( $0.30, 0.57, 0.36, 0.44 \mu\text{g m}^{-3}$ , respectively), and  
680 relatively high for June Bermuda ( $1.82 \mu\text{g m}^{-3}$ ). However, recall that oxalate is used in this study  
681 as a proxy for general trends in organic acids, many of which have been shown to considerably  
682 displace  $\text{Cl}^-$  from sea salt particles (e.g., Laskin et al., 2012), including formate, acetate, MSA, and  
683 succinate (Kerminen et al., 1998; Braun et al., 2017); thus the results based on oxalate are a lower  
684 bound for the effects organic acids have on depletion reactions. Although lower than other aerosol  
685 constituents, oxalate mass concentrations are highest for May and May transit along with those of  
686  $m/z$  44, a marker of oxygenated organics that has been shown to correlate with organic acids  
687 (Zhang et al., 2005; Takegawa et al., 2007; Sorooshian et al., 2010), and  $m/z$  79, a marker for MSA  
688 (Zorn et al., 2008; Van Rooy et al., 2021). Median  $m/z$  44 mass concentrations especially suggest  
689 organic acids may play an important role in sea salt particle chemistry for May and May transit as  
690 values ( $0.46$  and  $0.41 \mu\text{g m}^{-3}$ , respectively) (i) are comparable to those of other dominant acidic  
691 species over the region, (ii) represent the mass only of the particle fragments (i.e., carboxylic acids)  
692 able to displace  $\text{Cl}^-$ , and (iii) reflect a lower limit of what is actually available for depletion  
693 reactions as AMS measurements are for particles 60 – 600 nm.

694 Like sea salt mass concentrations, excess acidic species do not display clear zonal or  
695 meridional trends over the NWA (Fig. S7) but do appear to decrease near the USEC following the  
696 passage of MLCs (Fig. S8). The reasons are uncertain for such high mass concentrations of excess  
697 acidic species for June Bermuda, but a probable cause may be emissions of DMS from marine  
698 organisms oxidizing to produce  $\text{H}_2\text{SO}_4$  (e.g., Luria et al., 1989; Andreae et al., 2003). Excess acidic  
699 species mass concentrations are not nearly as high near Bermuda for March transit and May transit  
700 compared to June Bermuda, suggesting the increased values in June may be (i) due to greater  
701 photochemical production of  $\text{SO}_4^{2-}$  with increased incident solar radiation (Parungo et al., 1987;  
702 Corral et al., 2021) or (ii) due to an episodic surge in local marine biological activity, which has  
703 been shown to occur around Bermuda when higher doses of solar radiation become available to

704 the upper mixed layer of the ocean (Vallina and Simó, 2007; Toole and Siegel, 2004). Level-3 (8-  
705 day average, 4 km resolution) sea surface chlorophyll a concentrations from MODIS-Aqua show  
706 consistent values around Bermuda for March transit, May transit, and June Bermuda. However,  
707 there is an important distinction between biomass and ocean biological activity such that steady  
708 biomass around Bermuda does not necessarily correspond to similar gaseous emission rates for  
709 these categories. Thus, additional research is needed to better understand the seasonal variations  
710 in excess acidic species around Bermuda.

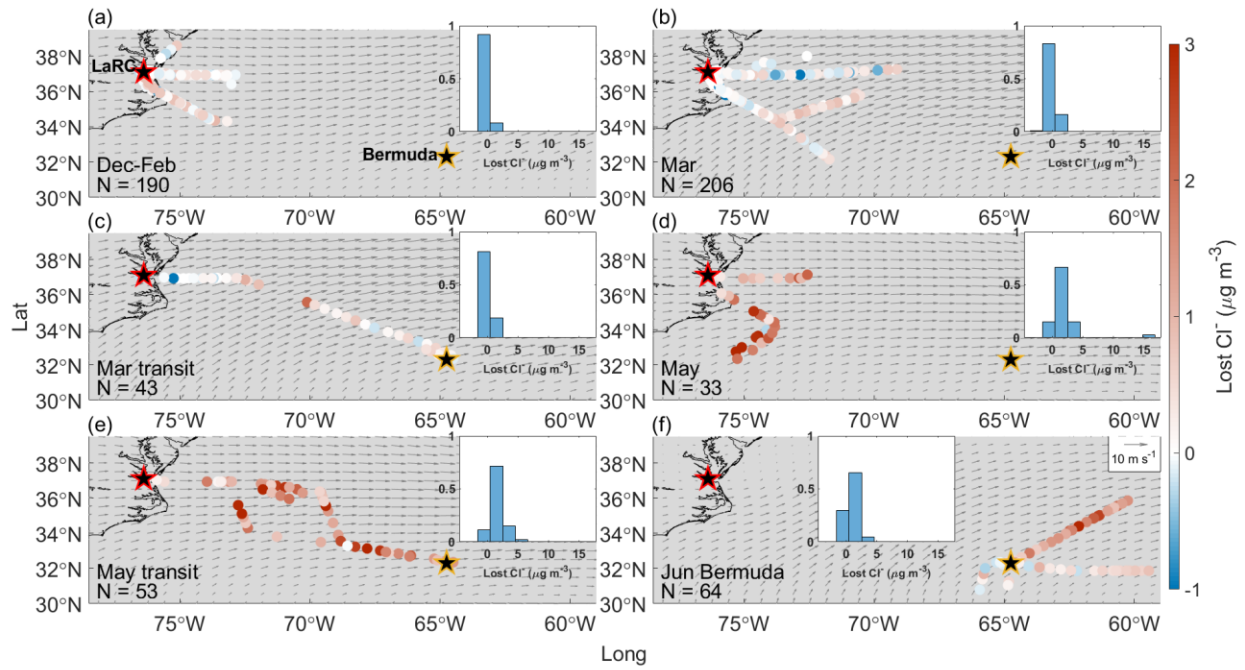


711  
 712 **Figure 4.** Notched box plots showing seasonal/categorical differences in observed mass  
 713 concentrations from clear ensembles of bulk PILS (a) sulfate ( $\text{SO}_4^{2-}$ ), (b) nitrate ( $\text{NO}_3^-$ ), and (c)  
 714 oxalate, as well as (d) AMS ammonium ( $\text{NH}_4^+$ ). Similar plots are shown for derived mass  
 715 concentrations of (e) total excess acidic species, (f) sea salt  $\text{Na}^+$  ( $\text{ssNa}^+$ ), the ratio of (g)  $\text{Cl}^-:\text{ssNa}^+$ ,  
 716 (h) mass concentrations of actual and theoretical lost  $\text{Cl}^-$ , as well as theoretical mass concentrations  
 717 of lost  $\text{Cl}^-$  attributable to (i) excess  $\text{SO}_4^{2-}$  ( $\text{ExSO}_4^{2-}$ ), (j) excess  $\text{NO}_3^-$  ( $\text{ExNO}_3^-$ ), and (k)  
 718 oxalate. The value of  $\text{Cl}^-:\text{Na}^+$  in sea salt (1.81; Seinfeld and Pandis, 2016) is indicated in g with a horizontal  
 719 dashed blue line. In h, light blue boxes represent the actual  $\text{Cl}^-$  displaced from sea salt particles  
 720 based on derived mass concentrations of  $\text{ssNa}^+$ , while light red boxes represent the theoretical  
 721 amount of  $\text{Cl}^-$  that could have been displaced by the derived mass concentrations of excess acidic  
 722 species. The properties of the boxes are the same as described in Fig. 1.

### 723 3.5 Seasonal, spatial, and frontal trends in Cl<sup>-</sup> depletion

724 Median ssNa<sup>+</sup> mass concentrations display similar trends to bulk Na<sup>+</sup> with comparable values  
725 among the December-February, March, and May categories (0.19, 0.20, and 0.29 μg m<sup>-3</sup>,  
726 respectively), higher mass concentrations for March transit and May transit (0.71 and 0.63 μg m<sup>-3</sup>,  
727 respectively), and highest values for June Bermuda (1.32 μg m<sup>-3</sup>). Median ratios of Cl<sup>-</sup>:ssNa<sup>+</sup>  
728 (1.54, 1.56, 1.62, 0.65, 0.47, and 1.35 for December-February, March, March transit, May, May  
729 transit, and June Bermuda, respectively) are higher than those of Cl<sup>-</sup>:Na<sup>+</sup> for each category, serving  
730 as a preliminary example of how neglecting contributions of dust to bulk Na<sup>+</sup> can lead to  
731 overestimates of Cl<sup>-</sup> depletion. Regardless of magnitude, Cl<sup>-</sup>:Na<sup>+</sup> and Cl<sup>-</sup>:ssNa<sup>+</sup> ratios both convey  
732 that the greatest fraction of available sea salt Cl<sup>-</sup> is converted to reactive chlorine-containing gas  
733 during the month of May (i.e., May and May transit categories) over the NWA. Lost Cl<sup>-</sup> mass  
734 concentrations are relatively low for December-February, March, and March transit (0.04, 0.04,  
735 and 0.11 μg m<sup>-3</sup>, respectively) then abruptly increase for May and May transit (1.76 and 1.33 μg  
736 m<sup>-3</sup>, respectively) followed by a moderate decrease for June Bermuda (0.66 μg m<sup>-3</sup>). These mass  
737 concentrations correspond to increases in atmospheric mixing ratios of reactive chlorine-  
738 containing gas of 27, 27, 73, 1174, 887, and 440 pptv, respectively, suggesting Cl<sup>-</sup> depletion  
739 processes have the potential to considerably alter rates of boundary layer VOC oxidation in May  
740 over the NWA; recall that Singh and Kasting [1998] reported ppbv levels of HCl can produce  
741 enough Cl radicals to oxidize 20 – 40% of tropospheric nonmethane alkanes. However, note our  
742 reported lost Cl<sup>-</sup> mass concentrations are for particles with diameters < 5 μm, so although May  
743 appears to be the only category where Cl<sup>-</sup> depletion is severe enough to potentially accelerate  
744 tropospheric VOC oxidation, lost Cl<sup>-</sup> mass concentrations may be higher in reality for other  
745 categories, depending on the extent of depletion reactions in larger sea salt particles.

746 There is not a clear spatial gradient in lost Cl<sup>-</sup> over the region (Fig. 5), but mass concentrations  
747 decrease near the USEC after passing frontal systems (Fig. S9), both of which are intuitive as bulk  
748 Na<sup>+</sup> and excess acidic species mass concentrations display the same trends. Although median lost  
749 Cl<sup>-</sup> mass concentrations are above 0 for all categories, negative lost Cl<sup>-</sup> mass concentrations are  
750 observed in 45, 42, 35, 3, 2, and 14% of the samples for December-February, March, March transit,  
751 May, May transit, and June Bermuda, respectively. Negative lost Cl<sup>-</sup> values can be interpreted as  
752 there being more Cl<sup>-</sup> in a sample than expected for unreacted sea salt particles based on derived  
753 mass concentrations of ssNa<sup>+</sup>. Such values may indicate influence from non-sea salt sources of Cl<sup>-</sup>  
754 , such as biomass burning (Jing et al., 2017; Park et al., 2013; Cao et al., 2016), mineral dust  
755 (Sullivan et al., 2007), and waste incineration (Moffet et al., 2008). Especially in December-  
756 February and March, negative mass concentrations of lost Cl<sup>-</sup> often occur in samples with  
757 relatively high mass concentrations of bulk Ca<sup>2+</sup> (Fig. S10) and K<sup>+</sup> (Fig. S11), which can be  
758 considered tracers for many of the non-sea salt sources of Cl<sup>-</sup> mentioned above. However, there  
759 are several exceptions to these relationships, and we leave a more thorough investigation into non-  
760 sea salt sources of particulate Cl<sup>-</sup> to future studies.



761

762 **Figure 5.** Same as Fig. 2, except for lost Cl<sup>-</sup>.

### 763 **3.6 Attributing lost Cl<sup>-</sup> to acidic species**

764 Median mass concentrations of excess acidic species have the potential to displace 0.17, 0.34,  
765 0.21, 0.04, 0.26, and 1.14  $\mu\text{g m}^{-3}$  (117, 228, 141, 27, 172, and 758 pptv, respectively) of Cl<sup>-</sup> from  
766 sea salt particles for December-February, March, March transit, May, May transit, and June  
767 Bermuda, respectively. These hypothetical losses exceed actual mass concentrations of lost Cl<sup>-</sup> for  
768 all categories except May and May transit, suggesting measured excess acidic species often did  
769 not react to their full potential with available particulate Cl<sup>-</sup>, considering median %Cl<sup>-</sup> depletion  
770 values are 6, 10, 9, and 64% for December-February, March, March transit, and June Bermuda,  
771 respectively. The extent of depletion reactions in December-February, March, March transit, and  
772 June Bermuda may have been limited by meteorological variables (e.g., temperature, RH) and/or  
773 restricted access of acidic species to particulate Cl<sup>-</sup> due to the size distribution and/or mixing state  
774 of sea salt particles (Su et al., 2022 and references therein).

775 Most lost Cl<sup>-</sup> can be attributed mostly to ExNO<sub>3</sub><sup>-</sup> in December-February, March, March transit,  
776 and May transit, which is consistent with findings from past works (e.g., Nolte et al., 2008; Yao  
777 and Zhang, 2012; Zhao and Gao, 2008). Excess SO<sub>4</sub><sup>2-</sup> and ExNO<sub>3</sub><sup>-</sup> have the potential to contribute  
778 equally to Cl<sup>-</sup> losses for June Bermuda, yet since actual lost Cl<sup>-</sup> was much lower than theoretical  
779 lost Cl<sup>-</sup>, the extent to which each species contributed is unknown. Oxalate has the potential to  
780 displace the least Cl<sup>-</sup> for all categories (0.01, 0.01, 0.01, 0.03, 0.03, and 0.02  $\mu\text{g m}^{-3}$  for December-  
781 February, March, March transit, May, May transit, and June Bermuda, respectively), although it  
782 is but one organic acid among thousands (Robinson et al., 2007). As mentioned above, there is  
783 convincing evidence that organic acids had considerable presence in sampled air masses,  
784 especially for Mar transit, May transit and May. This may be due to rising amounts of incident  
785 solar radiation accelerating photochemical oxidation of abundant biogenic and anthropogenic  
786 VOCs along the USEC to produce secondary organic aerosols (SOA), followed by further  
787 oxidation of these SOA to produce oxygenated organics, many of which can serve as weak acids  
788 in Cl<sup>-</sup> depletion reactions. It is possible that unmeasured organic acids are responsible for the lost  
789 Cl<sup>-</sup> that currently cannot be accounted for in May and May transit, although further research is  
790 necessary to explore this idea, specifically studies quantifying mass concentrations of additional  
791 organic acids in the context of Cl<sup>-</sup> depletion.

792

### 793 **3.7 Outcomes from quantifying Cl<sup>-</sup> depletion semi-unconventionally**

794 In the following subsections we examine the effects of accounting for (i) dust and (ii) dust and  
795 combustion emissions as a source of Na<sup>+</sup>, as well as focusing our discussions on mass  
796 concentrations of Cl<sup>-</sup> displaced from sea salt particles instead of either %Cl<sup>-</sup> depletion or Cl<sup>-</sup>:Na<sup>+</sup>  
797 ratios alone. We consider these to be “semi-unconventional” approaches as a handful of studies  
798 have employed at least one of these methods, but they are not commonly used in Cl<sup>-</sup> depletion  
799 studies (based on the 76 studies presented in Table S3 in Su et al., 2022). However, we  
800 acknowledge many works neglect non-sea salt sources of Na<sup>+</sup> after determining crustal  
801 contributions are unlikely (e.g., Rastogi et al., 2020; Bondy et al., 2017) or avoid calculating Cl<sup>-</sup>  
802 depletion for particles of a certain size range when anthropogenic sources seem to contribute to  
803 Na<sup>+</sup> and/or Cl<sup>-</sup> (e.g., Feng et al., 2017; Nolte et al., 2008). This work builds on past studies to  
804 provide an all-encompassing method for quantifying Cl<sup>-</sup> depletion in air masses influenced by dust  
805 and/or combustion emissions, as well as relating Cl<sup>-</sup> losses to their potential effects on atmospheric  
806 oxidation processes. We now discuss when, if ever, these methods are of importance for the NWA  
807 and provide a few lessons learned for future works interested in using these methods.

808

### 809 3.7.1 Significance of accounting for Na<sup>+</sup> in dust

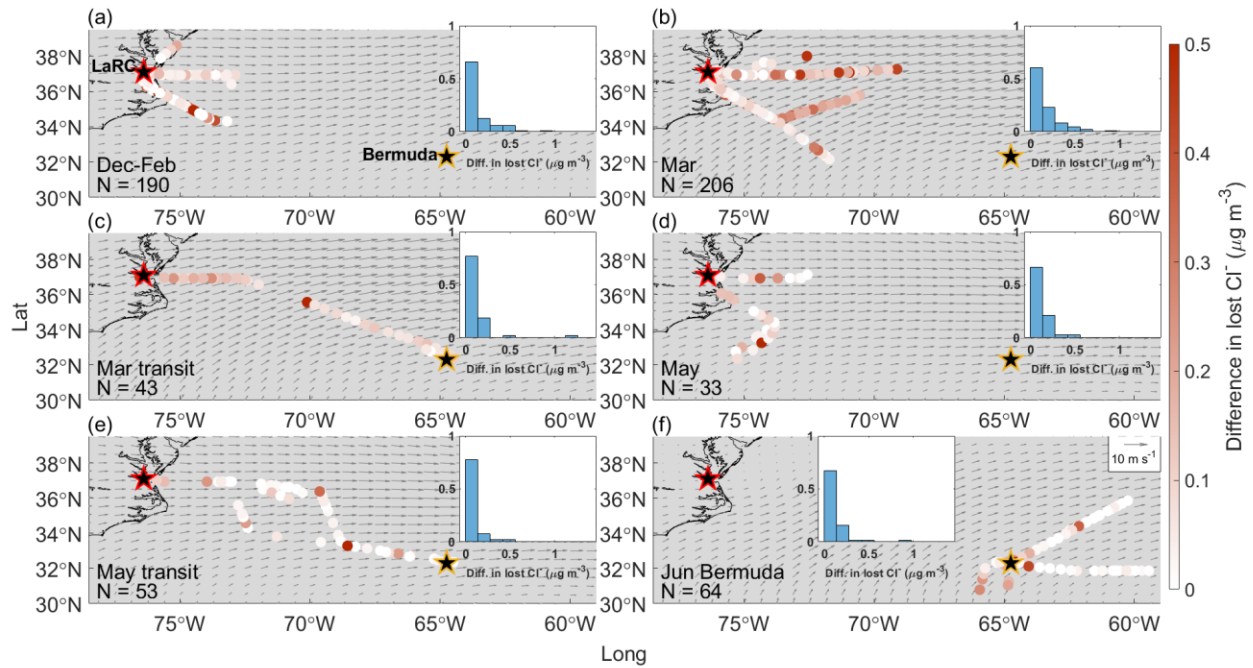
810 To facilitate understanding of the results below, recall mass concentrations of lost Cl<sub>diff</sub><sup>-</sup>  
811 quantify the difference in estimating Cl<sup>-</sup> depletion when dust is considered as a source of Na<sup>+</sup>  
812 (Approach 1) versus when Na<sup>+</sup> is attributed entirely to sea salt (Approach 2). Median lost Cl<sub>diff</sub><sup>-</sup>  
813 mass concentrations are 0.05, 0.1, 0.09, 0.05, 0.02, and 0.01 μg m<sup>-3</sup> (33, 64, 59, 34, 11, and 7 pptv,  
814 respectively) for December-February, March, March transit, May, May transit, and June Bermuda,  
815 respectively, meaning that Cl<sup>-</sup> losses are overestimated by a factor of 2.24, 3.38, 1.80, 1.03, 1.01,  
816 and 1.01, respectively, when using Approach 2 versus Approach 1. However, even though  
817 overestimates are proportionately large for December-February, March, and March transit, it may  
818 not be critical to account for dust as a source of Na<sup>+</sup> on a seasonal scale (Fig. 6). Specifically,  
819 lost Cl<sub>bulk</sub><sup>-</sup> mass concentrations for December-February, March, and March transit (58, 91, and  
820 133 pptv, respectively) are still well below the point where they would significantly accelerate  
821 VOC oxidation in the boundary layer. Similarly, Approaches 1 and 2 both lead to the conclusion  
822 that depletion reactions in May have the potential to accelerate tropospheric VOC oxidation, while  
823 lost Cl<sub>diff</sub><sup>-</sup> values are too small for May transit and June Bermuda to affect overarching conclusions  
824 regarding relationships between Cl<sup>-</sup> depletion and VOC oxidation rates. However, this study  
825 reports mass concentrations of lost Cl<sup>-</sup> and lost Cl<sub>diff</sub><sup>-</sup> for particles with ambient diameters < 5 μm,  
826 so it is possible that contributions of Na<sup>+</sup> from dust particles > 5 μm may be sufficiently high to  
827 lead to critical overestimates in Cl<sup>-</sup> depletion, especially considering that lost Cl<sup>-</sup> mass  
828 concentrations may increase when additionally accounting for depletion in larger sea salt particles.

829 Although not critically important on a seasonal scale, Approaches 1 and 2 produce  
830 considerably different estimates of lost Cl<sup>-</sup> for several flights sampling air masses more heavily  
831 influenced by dust. Median bulk Ca<sup>2+</sup> mass concentrations are 5.2 and 8.2 times higher on 30  
832 November and 01 December 2021 (RFs 94 and 95, respectively) than the December-February  
833 median without corresponding enhancements in bulk Na<sup>+</sup>, suggesting a higher presence of dust  
834 than usual. Using Approach 1, 100% and 88% (0.14 and 0.23 μg m<sup>-3</sup>, respectively) of median bulk  
835 Na<sup>+</sup> mass concentrations are attributed to dust for 30 November and 01 December (Table S7),  
836 respectively, which results in corrections of lost Cl<sup>-</sup> up to 0.63 μg m<sup>-3</sup> (420 pptv) compared to  
837 overestimates based on Approach 2 (Fig. 7). Dust particles sampled on these flights were likely  
838 lofted in smoke plumes extending over the NWA from fires in the eastern and southeastern U.S.  
839 On 03 March 2022 (RFS 131 and 132), median bulk Ca<sup>2+</sup> and Na<sup>+</sup> mass concentrations are 2.1 and  
840 3.4 times higher, respectively, than categorical medians, as it appears the NWA was heavily  
841 influenced by BB emissions from agricultural fires throughout the eastern U.S. Although only 15%  
842 of the median bulk Na<sup>+</sup> mass concentration is attributed to dust, lost Cl<sub>diff</sub><sup>-</sup> mass concentrations are  
843 as high as 1.05 μg m<sup>-3</sup> (700 pptv), with most between 0.11 and 0.32 μg m<sup>-3</sup> (73 - 213 pptv). As  
844 mentioned in Sect. 3.3, there is interest in exploring the spatial gradient in bulk Ca<sup>2+</sup> along March  
845 transit flights (RFs 142 and 143) to see how estimates of Cl<sup>-</sup> depletion are affected by the transition  
846 from a potentially dust-influenced air mass (directly east of LaRC) to one with less dust influence  
847 (to the southeast towards Bermuda). Although lost Cl<sub>diff</sub><sup>-</sup> mass concentrations are lower compared  
848 to those of previous case studies, Approach 2 overestimates Cl<sup>-</sup> depletion more for the air mass  
849 closest to the USEC compared to that closest to Bermuda. The air mass with higher bulk Ca<sup>2+</sup> mass  
850 concentrations appears to be composed of emissions from widespread springtime BB, and the  
851 shape of the plume is such over the NWA that the aircraft would fly in it near the USEC but not  
852 necessarily near Bermuda. The case studies above suggest that Cl<sup>-</sup> depletion can be considerably  
853 overestimated in smoke plumes when using Approach 2 as entrained dust particles can contribute  
854 meaningfully to bulk Na<sup>+</sup> mass concentrations, and that these overestimates may be of



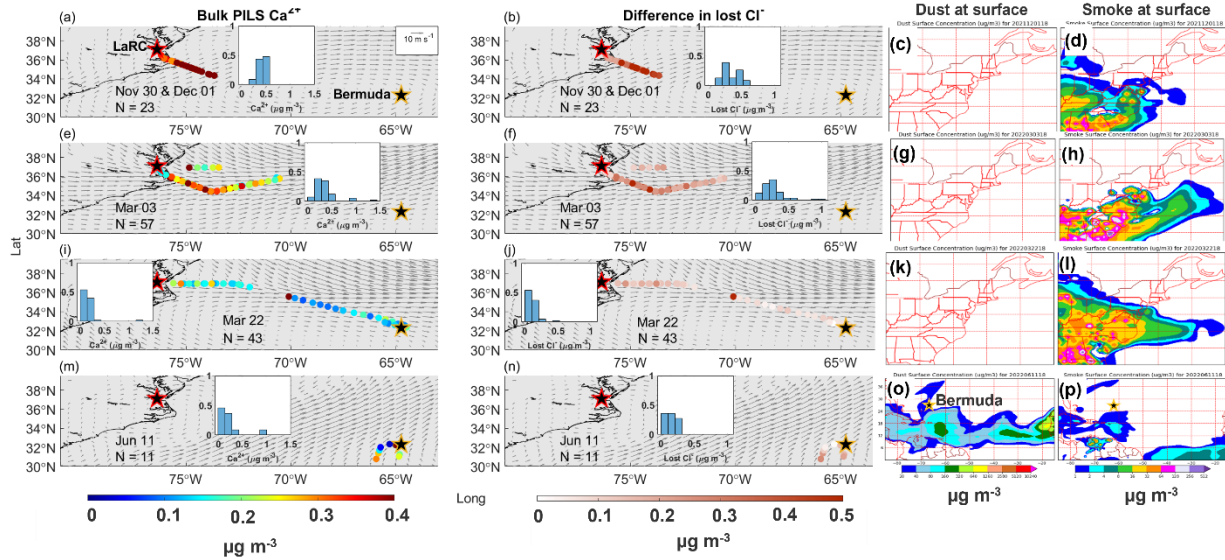
855 consequence when relating  $\text{Cl}^-$  depletion to potential increases in VOC oxidation over the region.  
856 Median  $\text{Ca}^{2+}$  mass concentrations are 3 times higher ( $0.21 \mu\text{g m}^{-3}$ ) than the June Bermuda median  
857 on 11 June 2022 (RFs 172 and 173) without similar enhancements in bulk  $\text{Na}^+$ , suggesting  
858 increases in bulk  $\text{Ca}^{2+}$  are likely due to African dust sampling (as opposed to increased sea salt  
859 mass concentrations). The arrival of African dust near Bermuda results in overestimates of lost  $\text{Cl}^-$   
860 up to  $0.315 \mu\text{g m}^{-3}$  (210 pptv) via Approach 2, which are not large enough to affect predictions for  
861 potential increases in rates of tropospheric VOC oxidation. Sampling ended near the beginning of  
862 the peak season for long-range transport of African dust to the NWA (e.g., Prospero, 1996;  
863 Zuidema et al., 2019), so we do not have many flights to choose from for studying effects of  
864 African dust plumes on  $\text{Cl}^-$  depletion calculations. Using 6300 ppm as a mass ratio of  $\text{Na}^+$  in dust  
865 particles (Seinfeld and Pandis, 2016), 131.54 and  $73.66 \mu\text{g m}^{-3}$  of dust would be necessary to cause  
866 critical overestimates of lost  $\text{Cl}^-$  (i.e., lost  $\text{Cl}_{\text{bulk}}^-$  values would reach  $1.5 \mu\text{g m}^{-3}$  using Approach 2)  
867 assuming 0 and  $0.66 \mu\text{g m}^{-3}$  of  $\text{Cl}^-$  were already being displaced from sea salt particles, respectively  
868 (note  $0.66 \mu\text{g m}^{-3}$  is the median lost  $\text{Cl}^-$  value for June Bermuda). Edwards et al. (2021) reported  
869 peak African dust mass concentrations of  $73.32 \mu\text{g m}^{-3}$  near Miami, Florida, so it may be possible  
870 for values to reach these levels over Bermuda, but it would take a relatively large plume. Therefore,  
871 it is typically not critical to use Approach 1 when quantifying  $\text{Cl}^-$  depletion near Bermuda, yet it  
872 may be important to use this approach during strong African dust events.

873 Furthermore, past works have demonstrated the uptake of precursors to acidic species (e.g.,  
874  $\text{NO}_x$ ,  $\text{SO}_2$ ; Grassian, 2002; Hanisch and Crowley, 2003; Ullerstam et al., 2002), inorganic acids  
875 (e.g.,  $\text{H}_2\text{SO}_4$ ,  $\text{HNO}_3$ ; Ooki and Uematsu, 2005; Sullivan et al., 2007), organic acids (Al-Hosney et  
876 al., 2005; Carlos-Cuellar et al., 2003), and HCl (Zhang and Iwasaka, 2001; Ooki and Uematsu,  
877 2005; Sullivan et al., 2007; Santschi and Rossi, 2006; Sorooshian et al., 2012) on dust particles.  
878 Thus, in addition to considering dust as a source of  $\text{Na}^+$ , it may also be important to account for its  
879 presence to avoid overestimating  $\text{Cl}^-$  depletion and its impacts on atmospheric oxidation as (i)  
880 uptake of acidic species and their precursors may reduce amounts available for depletion reactions,  
881 and (ii) deposition of HCl on dust particles may reduce the amount of Cl radicals produced  
882 following  $\text{Cl}^-$  displacement.



883

884 **Figure 6.** Same as Fig. 2, except for differences in lost  $\text{Cl}^-$  when sea salt is assumed to be the only  
 885 source of bulk  $\text{Na}^+$  versus when sea salt and dust are both considered to contribute to bulk  $\text{Na}^+$   
 886 mass concentrations.



887

888 **Figure 7.** Spatial relationships between mass concentrations of (a) bulk PILS  $\text{Ca}^{2+}$  and (b)  
 889 differences in lost  $\text{Cl}^-$ , as well as NAAPS reanalysis surface mass concentrations of (c) dust and  
 890 (d) smoke for the case study on 30 November – 01 December 2022 (RFs 94 and 95). The second,  
 891 third, and fourth rows correspond to case studies on 03 March (RFs 131 and 132), 22 March (RFs  
 892 142 and 143), and 11 June (RFs 172 and 173) 2022, respectively, where (e, f, g, h), (i, j, k, l), and  
 893 (m, n, o, p) display the same variables as (a, b, c, d), respectively. Normalized histograms for bulk  
 894 PILS  $\text{Ca}^{2+}$  and differences in lost  $\text{Cl}^-$  show the distribution of values for that specific case study  
 895 since overlap among the colored dots can hide some from view. Grey arrows indicate the average  
 896 magnitude and direction of MERRA-2 winds at 950 hPa for the month(s) relevant to each category.  
 897 NASA Langley Research Center (LaRC) and Bermuda are marked with red-edged and golden-  
 898 edged stars, respectively.

### 899 3.7.2 Significance of accounting for Na<sup>+</sup> in dust and combustion-sourced particles

900 As shown above, air masses influenced by BB frequently advect over the NWA, especially in  
901 March, occasionally increasing dust mass concentrations to levels capable of causing considerable  
902 overestimates in Cl<sup>-</sup> depletion. However, there is little to no effect on Cl<sup>-</sup> depletion calculations  
903 when accounting for contributions to Na<sup>+</sup> from combustion particles emitted via agricultural  
904 burning and forest fires as median Na<sub>comb</sub><sup>+</sup> mass concentrations are 0.00 μg m<sup>-3</sup> for all categories  
905 (Tables S8 and S9, respectively). Therefore, it may be more important to quantify contributions of  
906 dust as opposed to the combustion-sourced particles in smoke plumes over the NWA to avoid  
907 overestimates of Cl<sup>-</sup> depletion. However, recall median bulk K<sup>+</sup> mass concentrations for this study  
908 are 2 and 14 times lower than values measured in air masses more heavily influenced by (i)  
909 agricultural burning (Kacenenbogen et al., 2022) and (ii) wildfire smoke (Adachi et al., 2022),  
910 respectively. Thus, it is possible quantifying Na<sub>comb</sub><sup>+</sup> is important for accurate estimates of Cl<sup>-</sup>  
911 depletion in more concentrated BB plumes, yet we cannot explore this with the flights available  
912 and leave such an investigation to future studies. When combustion emissions are attributed to  
913 industrial operations, residential wood burning in sauna stoves, car driving, or coal burning at  
914 power plants, there is also no influence on Cl<sup>-</sup> depletion calculations for any category (i.e., all  
915 median Na<sub>comb</sub><sup>+</sup> values are 0.00 μg m<sup>-3</sup>; Tables S10 – S13). Thus, particles generated by the myriad  
916 of combustion processes occurring along the eastern U.S. may be too dilute over the NWA to affect  
917 calculations of Cl<sup>-</sup> depletion not only in air masses reaching Bermuda but also in those much closer  
918 to the USEC (e.g., Fig. S12).

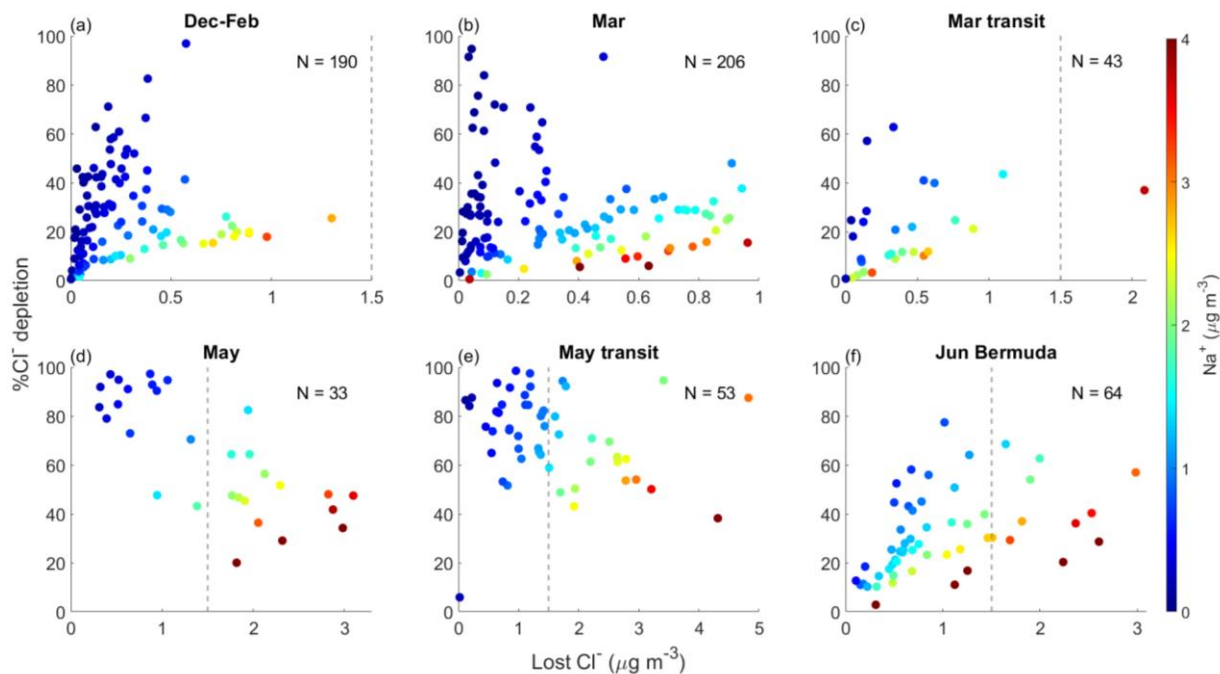
919 Since mass concentrations of Na<sub>comb</sub><sup>+</sup> are typically negligible, Eqs. 1 – 4 and 6 – 13 should  
920 provide the same median mass concentrations of ssNa<sup>+</sup> and Na<sub>dust</sub><sup>+</sup> for each category. However,  
921 many samples are excluded when using Eqs. 6 – 13 as their K<sup>+</sup> mass concentrations are below IC  
922 detection limits, causing adjustments in median ssNa<sup>+</sup> and Na<sub>dust</sub><sup>+</sup> values for several categories.  
923 Despite the advantages in accounting for non-sea salt sources of Na<sup>+</sup>, one disadvantage is potential  
924 dataset reduction. For example, 275, 246, 48, 113, 106, and 81 samples provide bulk Na<sup>+</sup> mass  
925 concentrations for December-February, March, March transit, May, May transit, and June  
926 Bermuda, respectively, yet only 202, 220, 48, 64, 75, and 66, respectively, can be used in Eqs. 1 –  
927 4, with even fewer available for use in Eqs. 6 – 13 where concurrent mass concentrations of bulk  
928 Na<sup>+</sup>, Ca<sup>2+</sup>, and K<sup>+</sup> are necessary. Thus, future studies may want to weigh the consequences of  
929 neglecting contributions of Na<sup>+</sup> from non-sea salt sources versus potential reductions to the number  
930 of samples included in statistical analyses.

931

### 932 3.7.3 Significance of focusing on lost Cl<sup>-</sup> instead of %Cl<sup>-</sup> depletion

933 Values of %Cl<sup>-</sup> depletion display similar trends to lost Cl<sup>-</sup> mass concentrations, where most  
934 percentages are (i) relatively low for December-February, March, and March transit (half are ≤  
935 10%, nearly all are ≤ 50%), (ii) relatively high for May and May transit (nearly all are > 40%),  
936 and (iii) relatively moderate for June Bermuda (values are distributed fairly evenly from 0 – 30%,  
937 and nearly all are ≤ 60%, Fig. S13). However, these %Cl<sup>-</sup> depletion values can only be used to  
938 show relative seasonal/categorical differences, and they cannot (i) inform when Cl<sup>-</sup> mass transfer  
939 is greatest from the particulate to gas phase or (ii) place such depletion reactions in the context of  
940 their potential influence on tropospheric VOC oxidation rates. For example, samples with higher  
941 %Cl<sup>-</sup> depletion values can easily be misinterpreted as having greater Cl<sup>-</sup> losses when in reality the

942 opposite may be true. Lost  $\text{Cl}^-$  and % $\text{Cl}^-$  depletion have a negative correlation for May and May  
943 transit, meaning that samples with the least (most) displaced  $\text{Cl}^-$  have the highest (lowest) % $\text{Cl}^-$   
944 depletion values (Fig. 8). The remaining categories have mostly positive correlations between lost  
945  $\text{Cl}^-$  and % $\text{Cl}^-$  depletion, although % $\text{Cl}^-$  depletion values are typically higher for samples with  
946 relatively low bulk  $\text{Na}^+$  mass concentrations at a fixed lost  $\text{Cl}^-$  value. This trend may be due to  
947 samples containing sea salt particles with varying size distributions (i.e., lower bulk  $\text{Na}^+$  mass  
948 concentrations may mean smaller sea salt particles were collected in a given sample) and  
949 considering that smaller sea salt particles are typically more susceptible to depletion reactions (e.g.,  
950 Su et al., 2022 and references therein). However, this behavior may also be an artifact of increased  
951 sensitivity of % $\text{Cl}^-$  depletion to samples with relatively small  $\text{ssNa}^+$  mass concentrations. Many of  
952 the samples with mass concentrations of lost  $\text{Cl}^-$  high enough to potentially influence VOC  
953 oxidation rates (i.e., lost  $\text{Cl}^- > 1.5 \mu\text{g m}^{-3}$ ) have % $\text{Cl}^-$  depletion values  $< 40\%$ , while nearly all  
954 samples with %  $\text{Cl}^-$  depletion values  $> 80\%$  do not have  $\text{Cl}^-$  losses capable of affecting such rates.  
955 Thus, we highly recommend future studies quantify mass concentrations of lost  $\text{Cl}^-$  to make results  
956 from depletion studies more suitable for understanding mass exchange between sea salt particles  
957 and the surrounding atmosphere and the consequences this can have on rates of tropospheric  
958 chemistry and radiative forcing.



959

960 **Figure 8.** Relationships between mass concentrations of lost  $\text{Cl}^-$  and  $\% \text{Cl}^-$  depletion for (a)  
 961 December-February, (b) March, (c) March transit, (d) May, (e) May transit, and (f) June Bermuda.  
 962 Markers are colored by bulk PILS  $\text{Na}^+$  mass concentrations, and the vertical dashed gray line in  
 963 some panels denotes where mass concentrations of lost  $\text{Cl}^-$  may begin to have considerable  
 964 influence on tropospheric VOC oxidation rates.

#### 965 4. Conclusions

966 This study investigates  $\text{Cl}^-$  depletion in sea salt particles over the NWA from approximately  
967 December 2021 – June 2022 using an airborne dataset quantifying the chemical composition of  
968 particles  $< 5 \mu\text{m}$  among other parameters throughout the lower 3 km of the atmosphere. Trends in  
969 bulk PILS  $\text{Na}^+$  suggest sea salt mass concentrations (1) do not exhibit seasonal variation but are  
970 reduced following the passage of MLCs near the USEC, and (2) are higher in the open-ocean  
971 environment of Bermuda than along the USEC. Losses of  $\text{Cl}^-$  are greatest in May and least in  
972 December-February and March, with median lost  $\text{Cl}^-$  mass concentrations of 1.76, 0.04, and 0.04  
973  $\mu\text{g m}^{-3}$  (1174, 27, and 27 pptv), respectively. Mass concentrations of measured excess acidic  
974 species can account for all the  $\text{Cl}^-$  depletion observed in December-February, March, and June  
975 near Bermuda, yet none in May, suggesting unmeasured organic acids may be largely responsible  
976 for displacement in certain months. Accounting for dust as a source of  $\text{Na}^+$  is not critical for  
977 accurately predicting how  $\text{Cl}^-$  depletion reactions will influence rates of tropospheric VOC  
978 oxidation on a seasonal basis, yet this may be important for large smoke and dust plumes over the  
979 NWA. Combustion-sourced particles do not contribute enough  $\text{Na}^+$  to meaningfully affect  $\text{Cl}^-$   
980 depletion estimates in any season for the air masses sampled. Finally, quantifying  $\text{Cl}^-$  depletion as  
981 a percentage sufficiently captures seasonal trends in depletion processes but fails to convey the  
982 effects they may have on atmospheric oxidation rates.

983 These results help address several uncertainties regarding  $\text{Cl}^-$  depletion over the NWA and its  
984 influence on regional oxidation cycles. First, by identifying factors affecting regional sea salt mass  
985 concentrations, we help advance the scientific community towards better understanding and  
986 forecasting of regional fluctuations in this major reactive atmospheric Cl reservoir. Additionally,  
987 seasonally resolved mass concentrations of lost  $\text{Cl}^-$  reveal that depletion reactions correspond to  
988 increases in HCl capable of producing concentrations of Cl radicals sufficient to oxidize 20 – 40%  
989 of nonmethane alkanes in the marine troposphere in May, which can have numerous implications  
990 including potentially accelerating  $\text{O}_3$  production over this highly populated region. The possibility  
991 for dust to cause meaningful overestimates of  $\text{Cl}^-$  depletion is a regionally novel finding and should  
992 encourage future studies and modeling efforts to monitor and account for smoke and dust plumes  
993 advecting over the NWA when quantifying sea salt reactivity. Additionally, our results reveal the  
994 importance in quantifying absolute  $\text{Cl}^-$  losses as samples with the highest values of % $\text{Cl}^-$  depletion  
995 often have relatively low  $\text{Cl}^-$  losses, and lost  $\text{Cl}^-$  and % $\text{Cl}^-$  depletion are negatively correlated in  
996 May, which is critical to recognize as  $\text{Cl}^-$  depletion has the greatest potential effect on tropospheric  
997 VOC oxidation rates during this month compared to all other studied.

998 Lost  $\text{Cl}^-$  mass concentrations are similar between median values reported in this study and the  
999 mean presented in Keene et al. (1990) for summertime conditions around Bermuda (0.66 and 0.68  
1000  $\mu\text{g m}^{-3}$ , respectively), while our values also fall within the range observed over Bermuda in spring  
1001 (0.22 – 1.35  $\mu\text{g m}^{-3}$ ; Keene and Savoie, 1998). Keene et al. (1990) reported lower lost  $\text{Cl}^-$  mass  
1002 concentrations along the USEC from July-September than our findings in May (1.11 and 1.76  $\mu\text{g}$   
1003  $\text{m}^{-3}$ , respectively), while our median in May is above the range shared in Keene et al. (2007) for  
1004 July-August (0 – 1.31  $\mu\text{g m}^{-3}$ ). Haskins et al. (2018) quantified median lost  $\text{Cl}^-$  mass concentrations  
1005 of 0.30  $\mu\text{g m}^{-3}$  over the ocean from February – March, which is 7 times higher than our medians  
1006 for December-February and March (0.04 and 0.04  $\mu\text{g m}^{-3}$ , respectively), yet note their study  
1007 specifically targeted polluted winter air masses while ours did not. Many past works along the  
1008 North American east coast have been able to attribute  $\text{Cl}^-$  depletion largely to inorganic acids in  
1009 the summer and fall (Zhao and Gao, 2008; Keene et al., 2007; Nolte et al., 2008; Yao and Zhang,  
1010 2012), with Keene et al. (1990) reporting a lowest contribution of 38%. We can attribute all  $\text{Cl}^-$

1011 depletion to inorganic acids in December-February, March, and June, yet find inorganic acids do  
1012 not contribute at all to displacement reactions in May. Our study suggests depletion reactions are  
1013 still occurring to the extent they were in the 1990s and 2000s over the NWA except that organic  
1014 acids are possibly becoming increasingly responsible for  $\text{Cl}^-$  displacement, especially in May,  
1015 although further research is needed to verify this.

1016 Although the ACTIVATE dataset is well-equipped to explore seasonal and spatial trends in  
1017  $\text{Cl}^-$  depletion over the NWA, there are several caveats and limitations to be mindful of when  
1018 reviewing our results. Reported mass concentrations of sea salt and lost  $\text{Cl}^-$  should be interpreted  
1019 as a lower limit due to the size range of particles sampled ( $< 5 \mu\text{m}$ ). Additionally, calculations for  
1020 the neutralization of  $\text{SO}_4^{2-}$  and  $\text{NO}_3^-$  by  $\text{NH}_4^+$  combine speciated mass concentrations from two  
1021 separate instruments, each considering a different size range of particles, meaning mass  
1022 concentrations of excess acidic species should be considered as an upper limit for particles  $< 5$   
1023  $\mu\text{m}$ . We recommend accounting for non-sea salt sources of  $\text{Na}^+$  when appropriate but acknowledge  
1024 that it may limit statistical analyses as the procedure for disentangling contributions of various  
1025 sources to bulk  $\text{Na}^+$  requires synchronous mass concentrations of multiple species.

1026 Overall, this study presents an updated account of sea salt reactivity over the NWA while also  
1027 providing unprecedented statistics for (i) responses in parameters relevant to  $\text{Cl}^-$  depletion to  
1028 passing frontal systems, (ii) sea salt particle mass concentrations within the lower 3 km of the  
1029 atmosphere between the USEC and Bermuda, (iii) the extent of  $\text{Cl}^-$  depletion occurring in a variety  
1030 of air masses in winter, spring, and early summer as well as the importance of (iv) accounting for  
1031 smoke and dust plumes as a source of  $\text{Na}^+$  and (v) quantifying  $\text{Cl}^-$  depletion absolutely instead of  
1032 relatively. Our finding that depletion reactions are extensive enough to alter rates of VOC  
1033 oxidation along the USEC in May is impactful on multiple levels ranging from human health to  
1034 regional radiative forcing, while reporting that inorganic acidic species are not contributing to  
1035 these losses informs future works and the chemical modeling community that additional acidic  
1036 species are critical to first identify and then to monitor. Finally, this study reveals the limitations  
1037 in using traditional methods when quantifying  $\text{Cl}^-$  depletion and will hopefully motivate future  
1038 works to either be mindful of these limitations or choose alternative methods.

### 1039 **Data availability**

1040 The ACTIVATE dataset can be found at  
1041 <https://doi.org/10.5067/SUBORBITAL/ACTIVATE/DATA001> (ACTIVATE Science Team,  
1042 2020). Level-3 (8-day, 4 km resolution) sea surface chlorophyll a concentrations from MODIS-  
1043 Aqua can be found at <https://doi.org/10.5067/AQUA/MODIS/L3M/CHL/2022>.

### 1044 **Author contributions**

1045 YC, ECC, JPD, GSD, CER, MAS, ELW, and LDZ collected and/or prepared the data. ELE  
1046 conducted the data analysis. ELE, ECC, and AS conducted data interpretation. ELE and AS  
1047 prepared the manuscript with editing from YC, ECC, JPD, GSD, MAS, ELW, and LDZ.



1048 **Competing interests**

1049 At least one of the (co-)authors is a member of the editorial board of Atmospheric Chemistry and  
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1051 **Disclaimer**

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1054

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