



# 1 Distribution, chemical and molecular composition of high and low-

# 2 molecular-weight humic-like substances in ambient aerosols

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#### 17 Abstract

18 Humic-like Substances (HULIS) encompass a continuum of molecular weight (MW) ranges, yet our understanding of how HULIS characteristics vary with MW is 19 still limited and not well-established. In this study, a combination of ultrafiltration and 20 21 solid-phase extraction protocols was employed to fractionate the high MW (HMW, >1 kDa) and low MW (LMW, < 1kDa) HULIS fractions from ambient aerosols collected 22 23 during summer and winter at a rural site. Subsequently, comprehensive characterization 24 by using total organic carbon, high-performance size exclusion chromatography 25 (HPSEC), UV-vis and fluorescence spectroscopy, Fourier-transform infrared spectroscopy (FTIR), negative electrospray ionization high resolution mass 26 spectrometry (ESI- HRMS) were conducted. The results revealed that HMW HULIS 27 were dominated by larger-sized chromophores, substantially constituting a higher 28 fraction of total organic carbon and UV absorption at 254 nm than LMW HULIS. While 29 both HMW and LMW HULIS shared similar fluorophore types and functional groups, 30 the former exhibited higher levels of humification and a greater presence of polar 31 functional groups (e.g., -COOH, >C=O). HRMS analysis further unveiled that 32 molecular formulas within HMW HULIS generally featured smaller sizes but higher 33 degrees of unsaturation and aromaticity compared to those within LMW HULIS 34 fractions. This observation suggests the possibility of small molecules assembling to 35 36 form the HMW HULIS through intermolecular weak forces. Moreover, HMW HULIS contained a higher proportion of CHON but fewer CHO compounds than LMW HULIS. 37 In both HMW and LMW HULIS, the unique molecular formulas were primarily 38





39	characterized by lignin-like species, yet the former displayed a prevalence of N-
40	enriched and highly aromatic species. Additionally, HMW HULIS contained more
41	unique lipids-like compounds, while LMW HULIS exhibited a distinct presence of
42	tannin-like compounds. These findings provide valuable insights into the distribution,
43	optical properties, and molecular-level characteristics of HULIS in atmospheric
44	aerosols, thereby advancing our understanding of their sources, composition, and
45	environmental implications.
46	
47	Keywords: Humic-Like Substances, molecular weight fractionation, optical properties,

48 high-performance size exclusion chromatography, negative electrospray ionization-

- 49 high resolution mass spectrometry
- 50





#### 51 1. Introduction

52 HUmic-Like Substances (HULIS) are complex and heterogeneous mixtures of water-soluble organic matters (WSOM) that are of great importance in the atmospheric 53 environment. They usually share similar physicochemical properties (e.g., acidity, 54 55 absorption, fluorescence, functional groups) with naturally occurring humic substances (Graber and Rudich, 2006; Zheng et al., 2013) and are prevalent in fog, clouds, 56 57 rainwater and ambient aerosols (Birdwell and Valsaraj, 2010; Fan et al., 2016a; Santos 58 et al., 2012). With substantial hygroscopic and surface-active properties, HULIS 59 enhance the hygroscopic growth of particles, thereby contributing to the formation of the cloud condensation nuclei and ice nuclei (Chen et al., 2021a; Dinar et al., 2007). 60 Moreover, acting as an important component of brown carbon (BrC), HULIS 61 62 effectively absorb near-ultraviolet and visible light, thus influencing the global radiative balance and atmospheric chemistry processes (Bao et al., 2022; Zhang et al., 2020). 63 Furthermore, HULIS have the potential to catalyze the formation of reactive oxygen 64 species, leading to potential adverse health effects (Ma et al., 2019; Zhang et al., 2022b). 65 The chemical composition of atmospheric HULIS exhibit significant 66 heterogeneity and typically comprises macromolecular compounds containing aromatic 67 rings with highly conjugated structures, as well as long-chain hydrocarbon with polar 68 groups (e.g., -OH, -COOH, -NO2) (Fan et al., 2013; Huo et al., 2021). To unravel the 69 70 structural characteristics and properties of HULIS, a range of analytical techniques, including absorption and fluorescence spectroscopy, Fourier transform infrared 71 spectroscopy (FTIR), nuclear magnetic resonance spectroscopy (<sup>1</sup>H NMR), have been 72





utilized (Huo et al., 2021; Qin et al., 2022; Zou et al., 2020). These studies have 73 74 provided insights into the overall structural characteristics of complex HULIS, including their abundances, chemical and optical characteristics (Huo et al., 2021; 75 Mukherjee et al., 2020; Win et al., 2018; Zhang et al., 2022b; Zheng et al., 2013). In 76 77 recent years, high-resolution mass spectrometry (HRMS) techniques, such as Fourier transform ion cyclotron resonance mass spectrometry (FT-ICR-MS) and orbitrap 78 79 HRMS, in combination with electrospray ionization (ESI), have emerged as powerful 80 tools for elucidating the molecular-level characteristics of HULIS (Lin et al., 2012; Sun 81 et al., 2021; Wang et al., 2019; Zou et al., 2023). By utilizing HRMS, researchers have 82 gained deeper insights into the complexity and chemical heterogeneity of HULIS at the molecular level. 83

Operationally, HULIS are defined as the hydrophobic fraction of water-soluble 84 85 organic matter (WSOM) typically extracted through solid-phase extraction (SPE) protocol (Fan et al., 2012; Zou et al., 2020). Thus, the abundance and characteristics of 86 HULIS are contingent upon the chemical composition of WSOM. Previous studies have 87 88 shown that aerosol WSOM, also known as brown carbon (BrC), are comprised of a continuum of molecular weight (MW) species, as revealed by high-performance 89 exclusion chromatography (HPSEC) analysis (Di Lorenzo et al., 2017; Fan et al., 2023; 90 Wong et al., 2019). These studies have highlighted that BrC typically consist of both 91 92 high-MW (HMW) and low-MW (LMW) chromophores in various aerosols. For example, BrC emitted from fresh biomass burning (BB) are dominated by low MW 93 chromophores (Di Lorenzo et al., 2017; Wong et al., 2019). However, BrC derived from 94





95	aged BB aerosols and ambient aerosols tend to possess more HMW chromophores that
96	are highly chemically resistant (Di Lorenzo et al., 2017; Fan et al., 2023; Wong et al.,
97	2019). Further characterizations of different MW BrC can be conducted using an
98	ultrafiltration (UF) protocol (Fan et al., 2021). This approach enabled researchers to
99	obtain the distributions of content, chromophores and fluorophores within various MW
100	BrC fractions. Despite these advancements, the chemical structures and molecular
101	composition of different MW HULIS fractions remain poorly understood.
102	Consequently, a combination of UF and SPE protocols for the fractionation and
103	characterization of MW-separated HULIS is crucial, as it not only provides insights into
104	MW distributions but also illuminates the chemical heterogeneities of aerosols HULIS.
105	In this study, a combination of UF-SPE isolation protocol was developed to
106	fractionate and characterize the MW HULIS fractions. Two distinct sets of ambient
107	PM <sub>2.5</sub> samples collected during summer and winter periods were utilized to facilitate a
108	comparative analysis of MW HULIS. Initially, the WSOM were fractionated into high-
109	MW (HMW, >1 kDa) and low-LMW (LMW, <1 kDa) species using the UF protocol.
110	Subsequently, the resulting MW MSOM fractions underwent SPE to obtain different
111	MW HULIS fractions. The obtained HMW and LMW HULIS fractions were
112	comprehensively characterized using advanced analytical techniques, including total
113	organic carbon analysis, UV-vis and fluorescence spectroscopy, HPSEC, and HRMS to
114	unveil their abundances, absorption and fluorescence properties, and molecular
115	characteristics. The findings of this study hold great significance in advancing our
116	understanding of the definition and molecular profiles of HULIS, as well as facilitating





117 further investigations into their potential impacts on the atmospheric environment.

118

## 119 2. Materials and methods

120 2.1. Atmospheric fine particles sampling

121 Atmospheric PM<sub>2.5</sub> were sampled on the rooftop of a building within the campus of Anhui Science and Technology University (32.21°N, 118.72°E), around 20 m above 122 123 ground level. Detailed information regarding the sampling site can be found in our 124 previous studies (Cao et al., 2022; Fan et al., 2021). The PM<sub>2.5</sub> samples were collected 125 using a high-volume PM2.5 sampler (JCH-1000, Juchuang Ltd., Qingdao) onto prebaked quartz fiber filters ( $8 \times 10$  inches, Whatman). Sampling took place from July 126 25 to August 12, 2021, during summer, and from December 19, 2021 to January 6, 2022, 127 during winter. Blank filters were also collected as control samples. All aerosol PM2.5 128 filter samples were stored at -20 °C in a freezer prior to analysis. The atmospheric 129 pollutant data (NO<sub>2</sub>, SO<sub>2</sub> and O<sub>3</sub>) near sampling site during sampling period were 130 obtained from the website (https://www.aqistudy.cn) and are summarized in Table S1. 131 132 Additionally, the fire spots were investigated using data from the website (https://firms.modaps.eosdis.nasa.gov/map/) and are visualized in Fig. S1. 133

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135 2.2. Application of UF-SPE for isolating MW HULIS fractions

Punches of summer and winter aerosol PM<sub>2.5</sub> filter samples were taken and combined for the extraction of water-soluble organic matter (WSOM), respectively. The filters were immersed in 300 mL of ultrapure water and subjected to ultrasonication for

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139	30 min. The resulting suspensions were then filtered through a 0.22 $\mu m$ membrane ( $\Phi$
140	47 mm, Jinteng, China) to obtain bulk WSOM samples. These bulk filtrates were further
141	subjected to UF and SPE in tandem to obtain different MW HULIS fractions. Please
142	refer to Fig. S2 for a schematic representation of the fractionation steps.
143	Before UF, the bulk WSOM were diluted to DOC concentration below 30 mg/L to
144	minimize the concentration effects and prevent the accumulation of organic matters at
145	the membrane surface during UF. The detailed UF procedure followed the profile
146	described in our previous study (Fan et al., 2021). Briefly, each bulk WSOM solution
147	was passed through a pre-cleaned 1 kDa cut-off membrane in a stirred UF cell (Amicon
148	8200, Millipore, USA), with a pressure at 0.2 MPa applied by ultrapure $N_{\rm 2}.$ The
149	concentration factor was ~10. The resulting retentate was considered as HMW (>1 kDa)
150	WSOM, while the permeate solutions represented the LMW (<1 kDa) WSOM. Finally,
151	each MW fraction was diluted to the initial volume for further treatment and analysis.
152	Mass balances of WSOM during one-step UF process generally ranged from 92% to
153	99%, as determined by total organic carbon (TOC) and UV absorption at 254 nm
154	(UV <sub>254</sub> ), indicating good performance of UF without substantial loss or organic
155	contamination.
156	Subsequently, SPE was applied to isolate the so-called HULIS fractions from the
157	bulk and each MW fraction of WSOM, following the protocol proposed in our previous

- passed through pre-activated HLB columns (Waters Oasis, 500 mg/6 mL, USA). The
- 160 fractions retained on the resins (referred to as HULIS) were eluted with pure methanol

studies (Fan et al., 2013; Zou et al., 2020). Briefly, the acidified aqueous samples were





161	and dried using a gentle stream of pure $N_2$ . Finally, the bulk, HMW and LMW HULIS
162	fractions were obtained. A blank filter control was performed using the same procedure
163	described above, and the analysis signals of samples were corrected by blank control.
164	
165	2.3. HPSEC analysis
166	The apparent MW distributions of MW HULIS fractions were analyzed using a
167	high-performance liquid chromatography (HPLC) system (LC-20AT, Shimadzu, Japan)
168	equipped with a refractive index detector (RID-10A, Shimadzu) and a diode array
169	detector (SPD-M20A, Shimadzu). The wavelength of the diode array detector was set
170	at 254 nm. Separation was performed using an aqueous gel filtration column (Polysep-
171	GFC-P 3000, Phenomenex) preceded by a guard column (Polysep-GFC-P,
172	Phenomenex). The mobile phase consisted of a mixture of water and methanol (90:10
173	v/v) containing 25 mM ammonium acetate (Di Lorenzo et al., 2017; Wong et al., 2019).
174	The sample injection volume was 100 $\mu L,$ and the flow rate was maintained at 1 mL
175	min <sup>-1</sup> . The HPSEC calibration was performed using a series of polyethylene glycol
176	(PEG) standards (Kawasaki et al., 2011; Zhang et al., 2022c). The chromatographic
177	peak areas were integrated to represent the abundances of corresponding MW species.
178	It should be noted that the MW values estimated here are nominal rather than absolute.
179	The weight-average MW (Mw), number-average MW (Mn) and polydispersivity
180	( $\rho$ ), were determined using the following equations (Song et al., 2010):

$$Mw = \frac{\sum_{i=1}^{n} (h_i M W_i)}{\sum_{i=1}^{n} h_i}$$
(1)

$$Mn = \frac{\sum_{i=1}^{n} h_i}{\sum_{i=1}^{n} (h_i / MW_i)}$$
(2)





$$\rho = \frac{Mw}{Mn} \tag{3}$$

	Mn
181	where $h_i$ and $MW_i$ are the absorption intensity of the chromatogram and the MW of
182	molecules corresponding to the <i>i</i> th retention time, respectively.
183	
184	2.4. Measurements of WSOC content and optical properties
185	The concentration of water-soluble organic carbon (WSOC) in HMW and LMW
186	HULIS was measured using a Shimadzu TOC analyzer (TOC-VCPN, Japan) following
187	the non-purgeable organic carbon protocol.
188	The UV-vis spectra were recorded using a UV-vis spectrophotometer (UV 2600,
189	Shimadzu, Japan) over a wavelength range of 200-700 nm with 1 nm increments.
190	Excitation-emission matrix (EEM) spectra were determined using a fluorescence
191	spectrophotometer (F4600, Hitachi, Japan). The scanning ranges for excitation (Ex) and
192	emission (Em) wavelengths were 200-400 and 290-520 nm, respectively, with a
193	scanning speed was 12,000 nm/min.
194	To characterize the chemical and optical properties of MW HULIS fractions,
195	several commonly used spectra parameters were calculated, including the specific UV
196	absorbance at 254 nm (SUVA $_{254}$ ), the UV absorbance ratio between 250 and 365 nm

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 $(E_2/E_3)$ , spectra slope ratios  $(S_R)$ , the absorption Angstrom exponent (AAE), and mass

absorption efficiency (MAE<sub>365</sub>), fluorescence indices (FI), biological index (BIX), and

humification degree (HIX) (Fan et al., 2021; Li et al., 2022; Wu et al., 2021). Further

details can be found in Text S1 of the Supporting Information (SI).





#### 202 2.5. FTIR spectrometry

203	The functional groups in HMW and LMW HULIS were characterized using a
204	Nicolet iS50 FTIR spectrometer (Thermal Fisher, USA). Before analysis, the freeze-
205	dried MW HULIS and pure KBr were thoroughly mixed, finely ground, and pressed
206	into pellets under dry conditions. Then, the FTIR spectra of samples were recorded
207	within the wavenumbers ranging from 4000 to 400 cm <sup>-1</sup> , with a resolution of 4 cm <sup>-1</sup> . To
208	ensure accuracy, each spectrum was baseline-corrected using the pure KBr spectrum.
209	
210	2.6. HRMS analysis and data processing

The MW HULIS fractions were analyzed using a Q-Exactive mass spectrometer (Thermo Scientific, Germany) equipped with a heated electrospray ionization (ESI) source. The system operated in negative ESI mode with a resolution of 140,000 at m/z = 200. The detection mass range was set from 60 to 900 m/z. To ensure accurate mass measurements, mass calibration was carried out using a commercial standard mixture of ESI-L Low Concentration Tuning Mix (G1969-85000, Agilent, USA).

The acquired mass spectra were processed using Xcalibur software (V2.2, Thermo Scientific). The mathematically possible formulas for all ions were calculated with a signal-to-noise ratio  $(s/n) \ge 5$  using a mass tolerance of 5 ppm. The assigned molecular formulas followed specific constraints, with limitations on the following elements: C  $\le$ 50, H  $\le$  100, O  $\le$  20, N  $\le$  3, S  $\le$  2. Additionally, the elemental ratios of H/C, O/C, N/ C, and S/C were constrained to the ranges of 0.3–3.0, 0–3.0, 0–0.5, and 0–2.0, respectively. The double-bond equivalents (DBE) and modified aromaticity index





- 224 (AI<sub>mod</sub>) values of the assigned neutral assigned formula ( $C_cH_hO_oN_nS_s$ ) were calculated
- using the equations (4-5) (He et al., 2023; Song et al., 2022):

$$DBE = 1 + \frac{1}{2}(2c - h + n)$$
(4)

$$AI_{mod} = \frac{1 + c - 0.5o - 0.5n - 0.5h}{c - 0.5o - n}$$
(5)

226 The intensity-weighted molecular parameters  $(X_w)$  of MW, H/C, O/C, DBE, and

AI values were calculated according to the equation (6) (He et al., 2023; Zhang et al.,

228 2021; Zou et al., 2023):

$$X_W = \frac{\sum (I_i \cdot X_i)}{\sum I_i} \tag{6}$$

- 229 where X represents the aforementioned parameters, and  $I_i$  denote the intensity for each
- assigned formula *i*.

231

# 232 3. Results and discussion

- 233 3.1. Size and distribution of MW HULIS fractions
- 234 3.1.1. Molecular size of HMW and LMW HULIS

235 Fig. 1 shows the HPSEC chromatograms of MW HULIS. Both HMW and LMW

236 HULIS exhibit MW continuum distributions ranging from 100 to 20,000 Da, which is

237 consistent with the reported distributions of BrC in BB-derived and various ambient

- aerosol in previous studies (Di Lorenzo et al., 2017; Fan et al., 2023; Wong et al., 2017).
- 239 However, the chromatographic patterns for HMW HULIS clearly differ from those

240 observed for LMW HULIS in both aerosol samples. As seen in Fig. 1, HMW HULIS

- 241 display an additional and stronger absorption peak at around 4000 Da (peak *iii*), along
- 242 with a more pronounced peak at 2200 Da (peak *ii*) and a similar magnitude peak at 570





243 Da (peak i) compared to LMW HULIS. This suggests that HMW HULIS contain the



244 majority of larger molecular size chromophores within the bulk WSOM.

245

Fig. 1. Average HPSEC chromatograms of bulk, HMW and LMW HULIS fractions in
(a) summer and (b) winter aerosol, respectively. The yellow and cyan shadows represent
MW size regions of >1 kDa and <1 kDa, respectively.</li>

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Moreover, the molecular size of MW HULIS can be further reflected by the 250 differences in Mw and Mn. As listed in Table 1, the average Mw and Mn of HMW 251 HULIS are 2233-2315 and 654-707 Da, respectively, greatly lager than that of LMW 252 HULIS (989-1071 and 293-394 Da, respectively). These differences indicate that the 253 254 sources and formation processes of HMW HULIS may differ from LMW HULIS. Many 255 previous studies have demonstrated that complex atmospheric aging processes significantly enhance the formation of large molecular size chromophores, while 256 concurrently leading to the bleaching of small size ones (Di Lorenzo et al., 2018; Di 257 Lorenzo et al., 2017; Wong et al., 2017; Wong et al., 2019). Therefore, the higher 258 259 proportions of large-size chromophores and resulting larger apparent molecular size of HMW HULIS may indicate their possible secondary formation nature. 260





261

262 Table 1. The summary of typical quantity and quality parameters of each MW HULIS

	263	fraction	from BB	and	ambient	aerosols
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			Summer			Winter	
		Bulk	HMW	LMW	Bulk	HMW	LMW
HPSEC-derived	Mw	1975±13	2315±38	1071±24	1918±56	2233±42	989±67
parameters	Mn	591±53	707±48	394±13	525±57	654±17	293±32
	ρ	3.4±0.3	3.3±0.2	2.7±0.2	3.7±0.3	3.4±0.2	3.4±0.2
HULIS/WSOM	TOC	65±1	68±1	51±2	63±2	67±2	41±1
(%) <sup>a</sup>	UV254	66±5	65±2	55±4	67±1	65±1	61±2
Optical	$E_2/E_3$	12.02±0.54	11.72±0.31	$14.98 \pm 0.98$	6.30±0.24	6.54±0.16	7.24±0.43
parameters	MAE <sub>365</sub>	$0.21 \pm 0.02$	0.23±0.01	$0.20{\pm}0.01$	$1.04 \pm 0.02$	$1.06 \pm 0.01$	$0.88 {\pm} 0.00$
	AAE	7.11±0.32	$7.59{\pm}0.00$	8.25±0.23	6.66±0.06	6.25±0.06	7.28±0.03
	FI	$2.00{\pm}0.04$	$1.99{\pm}0.03$	$2.04{\pm}0.05$	2.06±0.01	$1.97{\pm}0.03$	$2.25 \pm 0.02$
	BIX	0.95±0.01	0.86±0.07	$1.02{\pm}0.01$	0.96±0.01	0.81±0.01	$1.07 \pm 0.02$
	HIX	2.42±0.06	2.43±0.04	2.40±0.05	3.13±0.25	5.64±0.34	1.94±0.16

<sup>264</sup> <sup>a</sup> The ratios of contents of SPE-isolated HULIS fractions to that of corresponding

265 WSOM fractions determined by TOC and/or absorbance at 254 nm (UV<sub>254</sub>).

266

### 267 3.1.2. Relative abundances of HMW and LMW HULIS

268 The contribution of MW-HULIS fractions to their corresponding MW-WSOM fractions, quantified in terms of TOC and UV absorption at 254 nm for both summer 269 270 and winter aerosols are summarized in Table 1. In general, the ratios of HULIS/WSOM 271 of HMW fractions (in terms of TOC and UV254) (65-68%) were higher than the ratios (41-61%) observed for LMW fractions. This finding suggests that the higher presence 272 273 of hydrophobic and conjugated aromatic structures in HMW WSOM, but more hydrophilic OC and non-aromatic species (e.g., aliphatic dicarboxylic acid) in the 274 LMW WSOM (Fan et al., 2012; Zou et al., 2020). 275

Fig. 2 illustrates the distribution of distinct MW fractions within reconstructed





277 WSOM, wherein "non-HULIS" refers to the content differences between the MW WSOM and its HULIS fractions. The HMW HULIS fraction contributed 39-41% of 278 TOC and 40-47% of UV254 to the bulk WSOM. In contrast, the LWM HULIS fraction 279 only make up a smaller proportion, accounting for 16-20% of TOC and 17-21% of 280 281 UV254 within the bulk WSOM. Specifically, the ratios between HMW HULIS and LMW HULIS (H/L) ranged from 1.88 to 2.75 for both summer and winter aerosols in 282 283 terms of either TOC or UV254. These findings emphasize that HMW HULIS 284 significantly dominate the bulk aerosol HULIS fractions. Notably, the H/L ratio for 285 winter aerosols was higher than that for summer aerosols, suggesting that larger-sized 286 HULIS contributed more to the bulk HULIS fractions in winter aerosols.



- 288 Fig. 2. Relative proportions of different MW fractions in summer and winter aerosols
- determined by TOC and UV254.
- 290

287

- 291 The non-HULIS fractions are also important constituents within aerosol WSOM,
- 292 but exhibit some differences between HMW and LMW fractions. The contributions of





293	HMW non-HULIS to bulk WSOM were ${\sim}19\%$ as determined by TOC and 21-25%
294	measured by UV254. In case of LMW non-HULIS, the contributions were higher in
295	terms of TOC (20-26%) but lower in terms of UV254 (11-18%). These results indicate
296	that the LMW WSOM contain a larger proportion of hydrophilic organic species with
297	weak or no light absorption.
298	
299	3.2. Optical characteristics of MW HULIS fractions
300	3.2.1. Light absorption characteristics
301	The absorption spectra of MW HULIS fractions in ambient aerosols are shown in
302	Fig. S3. These spectra exhibit a featureless shape with a general decrease in absorbance
303	as the wavelength increases, which is a typical characteristic of HULIS found in
304	rainwater, biomass burning (BB), and ambient aerosols (Huo et al., 2021; Santos et al.,
305	2009; Zhang et al., 2022b). The E2/E3 ratio, commonly used as an indicator of the
306	chemical characteristics of organic species, is inversely correlated with higher
307	aromaticity and larger molecular weight (Fan et al., 2021; Li et al., 2022; Sun et al.,
308	2021). As listed in Table 1, the $E_2/E_3$ of HMW HULIS fractions generally were lower
309	than that of LMW HULIS in both ambient aerosols. This is consistent with the
310	expectation that larger-sized HULIS generally possess more polyconjugated and
311	polymeric structures (Fan et al., 2021; Zhang et al., 2022c), leading to greater
312	aromaticity and larger molecular size.

MAE<sub>365</sub> and AAE are commonly used to characterize the light absorption capacity
 and the spectral dependence of light absorption by aerosol chromophores, respectively





315	(Bao et al., 2022; Fan et al., 2016b; Kumar et al., 2017; Yuan et al., 2021; Zou et al.,
316	2020). As listed in Table 1, the average $MAE_{365}$ values of HMW HULIS are 0.23 and
317	$1.06 \text{ m}^2 \text{ g}^{-1}$ in summer and winter aerosol, respectively. These values are higher than
318	the corresponding values of 0.20 and 0.88 $\mbox{m}^2\mbox{ g}^{\mbox{-1}},$ respectively, for LMW HULIS. In
319	addition, HMW HULIS presented lower AAE values, being 7.59 and 6.25 in summer
320	and winter aerosol, respectively, than the corresponding values of 8.25 and 7.28,
321	respectively, for LMW HULIS (Table 1). As shown in Fig. S4, the LMW HULIS exhibit
322	lower $MAE_{365}$ and higher AAE values, falling within the left-upper range of the values
323	previously reported for various ambient aerosol-derived and BB-derived HULIS (Bao
324	et al., 2022; Fan et al., 2018; Fan et al., 2016b; Hong et al., 2022; Huo et al., 2018; Liu
325	et al., 2018; Ma et al., 2019; Sun et al., 2021; Tang et al., 2020; Wu et al., 2018; Wu et
326	al., 2020; Yuan et al., 2021; Zhang et al., 2022a). It has been widely reported that
327	pronounced photooxidation and photobleaching processes of BrC can lead to a
328	reduction in their absorption capacity (Wu et al., 2018; Wu et al., 2020; Zhang et al.,
329	2022a), but an enhancement of their spectra dependence on wavelength (Chen et al.,
330	2021b; Sun et al., 2021). Therefore, it can be speculated that LMW HULIS are more
331	susceptible to enrich the by-products resulting from the degradation and oxidation of
332	BrC during processes like photooxidation and photobleaching.

333

3.2.2. Fluorescence characteristics 334

The EEM contours of MW HULIS fractions from both summer and winter 335 aerosols are presented in Fig. S5. These HULIS fractions from both seasons exhibit 336





337	similar EEM spectra features, with a predominance of humic-like fluorophores (Ex/Em
338	= 210-235/395-410 nm). This observation suggest that humic-like fluorophores are
339	fundamental constituents of both HMW and LMW HULIS, which are consistent with
340	previous findings for aerosols MW WSOM (Fan et al., 2021) and bulk HULIS in BB-
341	derived and ambient aerosols (Fan et al., 2020; Qin et al., 2018). In this study, the
342	fluorescence regional integration (FRI) method was applied to characterize the
343	fluorescent composition of MW HULIS. Using FRI, EEM spectra were divided five
344	fluorescence regions (labeled as I to V) (Fig. S5), which were successively assigned to
345	simple aromatic proteins (I and II), fulvic acid-like (III), soluble microbial byproduct-
346	like (IV), and humic acid-like (V) substances, respectively, as established in previous
347	studies (Chen et al., 2003; Qin et al., 2018; Wang et al., 2021b). As shown in Fig. S6,
348	the large-size aromatic proteins (II) and fulvic acid-like substances (III) dominated the
349	fluorophores within MW HULIS in both summer and winter aerosols, comprising
350	approximately 62-64% of the total fluorescence intensity. This finding is consistent
351	with previous reports on bulk HULIS in summer and winter aerosols from industrial
352	and urban cities (Qin et al., 2018; Wang et al., 2021b). In comparison, the HMW HULIS
353	in both summer and winter aerosols generally exhibited a higher proportion of humic
354	acid-like substances (V), while having a lower abundance of small-size aromatic
355	proteins I compared to LMW HULIS. These differences are particularly pronounced in
356	winter aerosols, with the humic acid-like substances accounting for 23% in HMW
357	HULIS compared to 13% in LMW HULIS, and small-size aromatic proteins I
358	comprising 9% in HMW HULIS compared to 17% in LMW HULIS (Fig. S6).





359	Furthermore, the higher HIX values of HMW HULIS (5.64) in comparison to LMW
360	HULIS (1.94) further support these differences (Table 1). The pronounced BB
361	emissions and potential NO2-related oxidation of OA, as evidenced by the presence of
362	more hotspots (Fig. S1) and higher concentration of NO <sub>2</sub> (Table S1), are likely driving
363	these marked distinctions between HMW and LMW HULIS in winter aerosols. In
364	general, these findings imply that the HMW HULIS have a stronger level of
365	humification and oxidation, while the LMW HULIS appear to be of a simpler nature
366	and are more likely associated with fresh emissions (e.g., BB).

367

# 368 3.2.3 Functional groups of MW HULIS

Fig. 3 depicts the FTIR spectra of HMW and LMW HULIS in both summer and 369 370 winter aerosols. In general, both HMW and LMW HULIS present similar absorption peaks, including pronounced peaks at 3434 cm<sup>-1</sup> (O-H stretching of phenols and 371 carboxylic acids), 1721 cm<sup>-1</sup> (mainly C=O stretching of carboxylic acids), 1636 cm<sup>-1</sup> 372 373 (mainly C=C stretching of aromatic rings and C=O stretching of conjugated carbonyl groups) and 1390 cm<sup>-1</sup> (O-H deformation and C-O stretching of phenolic groups) were 374 observed (Fan et al., 2020; Fan et al., 2016b; Mukherjee et al., 2020; Wang et al., 2021a). 375 376 Additionally, weak peaks at 2929-2980 cm<sup>-1</sup> and 1045-1281 cm<sup>-1</sup>, attributed to C-H 377 stretching of aliphatic -CH2 and -CH3, and C-O stretching of esters and ethers, respectively, were also observed (Fan et al., 2016b; Wang et al., 2021a; Zhang et al., 378 2021). These observations indicate that both HMW and LMW HULIS contain complex 379 multi-component mixtures of compounds, encompassing aliphatic and aromatic species, 380







### 381 as well as carboxyl and phenolic functional groups.

382

# **Fig. 3.** FTIR spectra of HMW and LMW HULIS in (a) summer and (b) winter aerosols.

384

385 As shown in Fig. 3, more intense peaks at 1721 and 1636 cm<sup>-1</sup> were observed in HULISs in summer aerosols compared to those in winter aerosols. In addition, the peaks 386 at 1045-1281 cm<sup>-1</sup> in summer HULISs appear to be more complex and overlapping than 387 388 those in winter HULISs. These findings imply higher abundances of aromatic carboxyl acids and other O-containing groups (i.e., -OH, C=O and C-O) in summer HULISs than 389 in winter ones, possibly attributed to complex oxidation reactions prevailing in summer 390 season (Fan et al., 2020; Qin et al., 2022). This could be partly associated with the 391 enhanced oxidation processes driven by the higher concentration of O3 in summer 392 (Table S1). Our previous study has proved that the O3 oxidation of BB BrC lead to the 393 generation of more intense peaks at approximately 1725 cm<sup>-1</sup> (Fan et al., 2020). 394 Moreover, distinct differences in relative peak intensity between HMW and LMW 395 396 HULIS fractions were observed. HMW HULIS generally exhibit more intense at 1721





397	$\rm cm^{-1}$ compared to LMW HULIS in both seasonal aerosols (Fig. 3). This finding suggests
398	that HMW HULIS contain a higher abundance of C=O groups, likely associated with
399	the oxidation of the unsaturated structures with addition of polar functional groups (e.g.,
400	-COOH, >C=O) during SOA processes (Fan et al., 2020; Pillar-Little and Guzman,
401	2018).
402	
403	3.3 Molecular-level insights into MW HULIS
404	3.3.1. Seasonal variations in the molecular composition of MW HULIS
405	The molecular-level characteristics of MW HULIS were examined using negative
406	ESI- HR-MS analysis. Fig. 4 displays the reconstructed mass spectra of all HULIS
407	fraction in both summer and winter aerosols. Hundreds of peaks can be observed in the
408	spectra ranging from m/z 100 to 450 for all samples, with most ions being abundant
409	within the $m/z$ 150-350 range. These spectrum characteristics are similar to those
410	previously reported for HULIS in ambient aerosols and BB emissions (He et al., 2023;
411	Song et al., 2022; Sun et al., 2021; Wang et al., 2019; Zhang et al., 2021; Zou et al.,
412	2023).







Fig. 4. Mass spectra of bulk and MW HULIS in (a) summer and (b) winter aerosols.
The pie charts represent the intensity distributions of four compound categories (CHO,
CHON, CHOS, and CHONS).

417

413

As listed in Table 2, the number of assigned formulas within MW HULIS in 418 summer aerosols were 655-672, which was higher than the range of 470-506 observed 419 in winter aerosols. This suggests that the MW HULIS in summer aerosols exhibited 420 greater diversity than those in winter aerosols, mainly due to the stronger SOA 421 formation that enhanced the heterogeneity of HULIS fractions in the summer. The 422 identified formulas were then classified into four groups (i.e., CHO, CHON, CHOS and 423 424 CHONS) according to their elemental composition. As depicted in pie charts in Fig. 4, 425 summer HULISs are predominantly composed of CHO (54-71%), while winter





426	HULISs feature a high concentration of both CHON (30-58%) and CHO (21-38%). The
427	notably higher content of CHO in summer HULISs are likely due to a wide distribution
428	of biogenic VOC-derived SOAs during the summer season (Li et al., 2022; Sun et al.,
429	2023). CHON content in winter HULISs is generally higher than in summer ones,
430	potentially due to more significant contributions from direct BB, as well as secondary
431	nitrogen-related chemical processes during the winter season (He et al., 2023; Song et
432	al., 2022; Zhang et al., 2021; Zou et al., 2023). This finding is supported by the greater
433	number of fire spots (Fig. S1) and higher concentrations of $NO_2$ (Table S1) during
434	winter. The higher proportions of CHON compounds in aerosol HULIS typically lead
435	to enhanced light absorption capabilities (He et al., 2023; Song et al., 2022; Zeng et al.,
436	2021). This provides an strong explanation for why winter HULIS exhibit higher
437	MAE <sub>365</sub> values compared to summer HULIS. Additionally, CHOS is more abundant in
438	winter HULISs (17-25%) than in summer aerosols (13-18%). Previous studies have
439	demonstrated that both coal combustion and the oxidation initiated by $SO_2$ can lead to
440	the generation of larger amounts of S-containing compounds (Song et al., 2018; Song
441	et al., 2022; Zou et al., 2023). This finding suggested that the increased levels of coal
442	combustion and SO <sub>2</sub> -related SOAs, as evidenced by higher concentration of SO <sub>2</sub> (Table
443	S1), are significant contributors to the presence of BrC in winter compared to in summer.





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24





	CHON	244	232	1.29	0.53	0.16		3.89		2.01	4.87	0.55	0.23	0.44	-0.23
	CHOS	59	292	2.08	0.40		0.12		4.40	2.04	0.46	0.06	-4.06	0.02	-1.27
	CHONS	29	236	1.74	0.48	0.22	0.20	2.82	3.02	2.57	2.88	0.38	-0.57	0.27	-0.77
	Total	470	242	1.46	0.50	0.10	0.03	2.37	0.85	1.98	4.04	0.43	-0.57	0.33	-0.46
ΜM	CHO	176	249	1.23	0.54					1.82	5.25	0.48	-0.65	0.30	-0.15
	CHON	195	239	1.34	0.49	0.16		3.99		1.95	4.88	0.52	0.29	0.45	-0.36
	CHOS	107	280	1.94	0.54		0.10		5.42	2.17	1.20	0.13	-4.25	0.02	-0.85
	CHONS	28	272	1.67	0.69	0.16	0.14	5.89	6.03	2.63	2.99	0.37	-3.15	0.13	-0.29
	Total	506	256	1.47	0.54	0.06	0.04	1.61	1.78	2.01	3.96	0.39	-1.45	0.26	-0.40

25





447	Table 2 summarizes the intensity-weighted molecular parameters for MW HULIS
448	in both summer and winter aerosols. Evidently, the MWw of summer HULISs are 258-
449	278, which are higher than the corresponding values of 242-256 for winter HULISs.
450	This observation indicates that summer HULISs exhibit larger sizes, consistent with
451	their higher HPSEC-derived Mw and Mn compared to winter HULISs. Moreover,
452	summer HULISs exhibit higher O/Cw ranging from 0.60 to 0.66, as well as $OS_{C,w}$
453	ranging from -0.29 to -0.10, which exceed the respective values of 0.50 to 0.54 and -
454	0.46 to -0.37 observed in winter HULISs. Conversely, winter HULISs display higher
455	$AI_{mod,w}$ values (0.26-0.33) than those (0.13-0.16) for summer ones. These findings
456	suggest that summer HULISs are characterized by a high degree of oxidation, while
457	winter HULISs exhibit stronger aromaticity.

458

459 3.3.2. Comparison on molecular composition of HMW and LMW HULIS

CHO compounds. The CHO compounds are prominent constituents within 460 HULIS fractions, accounting for 54% and 21% in summer and winter HMW HULIS, 461 respectively, whereas these proportions increase to 71% and 38% in LMW HULIS (Fig. 462 4). It is worth noting that CHO compounds that undergo deprotonation in ESI- mode 463 are likely associated with the presence of carboxyl, carbonyl, alcohol and ester (Lin et 464 al., 2012; Wang et al., 2018). Moreover, CHO compounds in LMW HULIS exhibit a 465 higher oxygenation level compared to HMW HULIS, as evidenced by the higher O/Cw 466 and  $OS_{C,w}$  values. As shown in Table 2, the O/C<sub>w</sub> for CHO in LMW HULIS are 0.55-467 0.64, which are higher than 0.47-0.54 observed in HMW HULIS. In contrast, the H/Cw 468





469	for CHO in HMW HULIS were consistently higher than those in LMW HULIS, with
470	values of 1.41 vs. 1.34 in summer and 1.36 vs. 1.23 in winter (Table 2). This disparity
471	strongly suggests a higher saturation level of CHO compounds within HMW HULIS.
472	This conclusion is further corroborated by the lower $\text{DBE}_w$ and $\text{AI}_{\text{mod},w}$ observed for
473	CHO in HMW HULIS compared to LMW HULIS (Table 2). It is known that these
474	values serve as estimations of C=C density and aromatic and condensed aromatic
475	structures (Song et al., 2022; Zhang et al., 2021). Taken together, the CHO compounds
476	within HMW HULIS exhibit a more aliphatic nature but lower aromaticity and
477	oxidation levels when compared to those within LMW HULIS.

CHON compounds. HMW HULIS fractions consist of a higher proportion of 478 CHON compounds compared to LMW HULIS, with proportions of 27% vs. 8% in 479 summer and 58% vs. 30% in winter (Fig. 4). This observation suggests that HMW 480 HULIS contain a higher content of N-containing components. It is noted that the LMW 481 HULIS are generally characterized by higher O/Nw values of 6.60 in summer and 3.99 482 in winter compared to 5.23 in summer and 3.89 in winter for HMW HULIS. This 483 484 indicates that the CHON compounds within LMW HULIS are more highly oxidized than those within HMW HULIS. In general, compounds with  $O/N \ge 3$  are indicative of 485 oxidized N groups such as nitro (-NO2) or nitrooxy (-ONO2), while compounds with 486 O/N < 3 may denote the reduced N-containing functional groups (i.e., amines) (He et 487 488 al., 2023; Song et al., 2022; Zeng et al., 2021). In this study, a majority of the CHON compounds, comprising 73-85% in summer and 59-64% in winter, exhibited  $O/N \ge 3$ 489 in both MW HULIS fractions. This suggests that high concentrations of nitro 490





491	compounds or organonitrates dominate the CHON compounds (Sun et al., 2023; Wang
492	et al., 2018; Zeng et al., 2021), especially in summer samples, primarily due to the
493	hydroxyl radical oxidation of biogenic or anthropogenic VOC precursors, as well as
494	BB emissions (Song et al., 2022; Sun et al., 2021; Zhang et al., 2021; Zou et al., 2023).
495	Furthermore, the CHON compounds exhibiting $O/N \ge 3$ were more abundant in LMW
496	HULIS compared to HMW HULIS, accounting for 85% vs. 73% in summer and 64%
497	vs. 59% in winter. In contrast, HMW HULIS contained more CHON compounds with
498	O/N < 3 compared to LMW HULIS. These findings collectively indicate that the CHON
499	within HMW HULIS possess lower content of nitro compounds or organonitrates than
500	LMW HULIS. Based on FTIR analysis, it is known that HMW HULIS contain more
501	carboxylic groups than LMW HULIS, which indicate a higher likelihood of HMW
502	HULIS containing more amino acids.

CHOS and CHONS compounds. In this study, we observed that CHOS 503 accounted for proportions of 13% to 25% in all MW HULIS fractions, while CHONS 504 had a lower proportion of 3% to 7% (Fig. 4). Notably, the distribution of CHOS differed 505 between HMW and LMW HULIS in both season samples. As depicted in Fig. 4, HMW 506 HULIS contained fewer CHOS compounds compared to LMW HULIS, with 507 proportions of 13% vs. 17% in summer and 17% vs. 25% in winter. This finding 508 suggests that a greater number of CHOS compounds are incorporated into the LMW 509 HULIS fractions, which potentially leading to a reduction in the light absorption of 510 LMW HULIS (Zeng et al., 2021; Zhang et al., 2021). Furthermore, as indicated in Table 511 2, both the CHOS and CHONS within LMW HULIS exhibited higher O/Sw values than 512





513	HMW HULIS in both seasonal samples. Consequently, the S-containing compounds
514	within LMW HULIS were characterized by a higher degree of oxidation, primarily
515	attributed to SO <sub>2</sub> -related chemical oxidation process, in comparison to those in HMW
516	HULIS. Moreover, it was observed that 61% to 92% of CHOS compounds exhibited
517	O/S > 4, and 3% to 43% of CHONS compounds with $O/S > 7$ for all MW HULIS
518	fractions. Among them, HMW HULIS own lower proportions of CHOS with $O/S > 4$
519	and CHONS with $O/S > 7$ than LMW HULIS, suggesting a reduced presence of
520	potential organosulfates and nitrooxyorganosulfates within HMW HULIS (Sun et al.,
521	2023; Wang et al., 2018; Zeng et al., 2021; Zou et al., 2023).

522

3.3.3. Comparative analysis of unique molecular formulas in HMW and LMW HULIS 523 524 In this study, particular emphasis was placed on the unique molecular formulas within the HMW or LMW HULIS fractions. Fig. 5a, b illustrates the Van Krevelen (VK) 525 diagram depicting the distribution of unique molecular formulas within HMW and 526 LMW HULIS in summer and winter samples. It is evident that a majority of unique 527 formulas within LMW HULIS are concentrated around the origin with O/C > 0.5, 528 accounting for 83% in summer and 64% in winter. In contrast, most formulas within 529 HMW HULIS exhibited O/C < 0.5, representing about 58% for both seasonal samples. 530 These findings indicate that the unique molecules within LMW HULIS consist of more 531 polar O-containing organic compounds than those within HMW HULIS. 532

533







Fig. 5. Van Krevelen diagrams for the unique molecular formulas within HMW and
LMW HULIS from (a) summer and (b) winter aerosols. (c) The contributions of major
substances classes in unique formulas.

538

534

539 The molecular formulas are further categorized into seven groups based on previous studies, including lignin-like species, protein/amino sugars, condensed 540 aromatics, tannin-like species, carbohydrate-like species, unsaturated hydrocarbons, 541 and lipid-like species (He et al., 2023; Sun et al., 2021; Sun et al., 2023). The 542 classification rules for these formulas can be found in Table S2. Fig. 5c provides an 543 overview of the relative contributions of the number of unique formulas from each of 544 the seven groups for HMW and LMW HULIS. The results indicate that the dominant 545 546 substance class in the unique formulas within both MW HULIS are lignin-like species, accounting for proportions of 31-40%. This finding indicates that lignin derivatives are 547 fundamental components in both HMW and LMW HULIS either in summer or winter 548 aerosols. Additionally, there are notable differences in the molecular characteristics of 549 550 lignin-like species within HMW and LMW HULIS. As listed in Table S3, lignin-like species within HMW HULIS exhibit lower MWw and O/Cw, but higher N/Cw and 551 Almod,w values than those within LMW HULIS in both seasonal samples. These 552





553	observations suggest that the unique lignin-like substances in HMW HULIS likely
554	contain more N-enriched and highly aromatic species, while those in LMW HULIS
555	tend to concentrate more aliphatic O-containing compounds. These distinctions in
556	composition and characteristics between HMW and LMW HULIS fractions provide
557	valuable insights into their origins and transformations in the atmosphere.

Moreover, there are notable variations in the contributions of lipids-like, 558 559 protein/amino sugars, carbohydrates, condensed aromatics, and tannins species 560 between HMW and LMW HULIS. In general, HMW HULIS have a higher proportion 561 of lipids-like species, carbohydrates and condensed aromatics than LMW HULIS in both summer and winter aerosols. Among these, the most remarkable difference in 562 composition between HMW HULIS and LMW HULIS is seen in lipids-like species, 563 accounting for 15% versus 1% in summer and 20% versus 3% in winter (Fig. 5). As 564 reported in previous studies, lipids-like species primarily originate from biogenic 565 emissions (He et al., 2023; Li et al., 2022; Sun et al., 2021). This suggests that there is 566 a stronger contribution from biogenic emissions to HMW HULIS. Additionally, these 567 568 species in HMW HULIS were usually characterized by lower DBEw and slightly lower OS<sub>C,w</sub> when compared to LMW HULIS (Table S3), indicating they present stronger 569 saturation and fewer oxidized substituents. On the other hand, tannins species 570 contribute a higher proportion to LMW HULIS, constituting 30% in summer and 11% 571 in winter, while comprising only 5%-6% in HMW HULIS in both season aerosols. 572 Tannin-like species are known to consist of various polyphenolic groups containing 573 hydroxyl and carboxylic functional groups (He et al., 2023; Li et al., 2022; Ning et al., 574





575	2019; Sun et al., 2021). The slightly lower DBEw but much higher DBE-Ow for unique
576	tannin-like species within HMW HULIS were observed compared to LMW HULIS
577	(Table S3), suggesting that the former ones are enriched in more unsaturated O-
578	containing functional groups, particularly carboxylic functional groups.
579	
580	3.4. Atmospheric implications
581	This study provides comprehensive comparison between HMW and LMW HULIS
582	regarding their distributions, chemical structures, molecular sizes and compositions.
583	HMW HULIS appear to be larger than LMW HULIS, as evidenced by both
584	ultrafiltration natures and the MW distributions of chromophores analyzed by HPSEC.
585	However, HRMS analysis revealed that the average MWw of identified formulas within
586	HMW HULIS were lower than those of LMW HULIS (Table 2). This discrepancy can
587	likely be attributed to the "assembled structures" that construct the aerosol HULIS, as
588	suggested in many previous studies focusing on HULIS and BrC characterization (Fan
589	et al., 2021; Fan et al., 2023; Phillips et al., 2017; Qin et al., 2022). In fact, the results
590	from EEM-FRI and FTIR analysis support the notion that HMW and LMW HULIS
591	likely consist of potential structures assembled by similar basic fluorophores and
592	functional groups. Based on this theory, HMW HULIS may consist of macromolecular
593	species primarily assembled from small molecules through weak forces (i.e., $\pi$ – $\pi$ , van
594	der Waals, hydrophobic, or hydrogen bonds) and/or charge-transfer interactions (Fan et
595	al., 2021; Phillips et al., 2017), which can potentially disassemble during ESI ionization
596	and form low MW molecules.





597	Based on the molecular-level characterization, significant distinctions in
598	properties between HMW HULIS and LMW HULIS become evident. HMW HULIS
599	generally exhibit stronger aromaticity but lower oxidation degree when compared to
600	LMW HULIS. In terms of molecular composition, HMW HULIS contain higher
601	quantities of CHON species but lower quantities of CHO compounds than LMW
602	HULIS. Furthermore, more lipids-like species were identified as unique molecules in
603	HMW HULIS, while more tannin-like species with abundant carboxylic groups were
604	observed as unique molecules in LMW HULIS. Given these pronounced differences
605	between HMW and LMW HULIS, it can be speculated that the higher levels of aromatic
606	structures, greater presence of CHON molecules and the presence of lipids-like species
607	may serve as driving factors in the formation of potential assembled structures in HMW
608	HULIS. Additionally, it is well-established that CHON can enhance the light absorption
609	of organic aerosols (OA), while CHO species may have the opposite effect, weakening
610	light absorption (He et al., 2023; Song et al., 2022; Wang et al., 2019; Zeng et al., 2021).
611	Therefore, it is reasonable to conclude that HMW HULIS possess stronger light
612	absorbing capability, which is consistent with their larger $MAE_{365}$ values.
613	Importantly, HMW HULIS contain amounts of carboxylic functional groups,

reduced nitrogen species (e.g., amines) and aromatic species than LMW HULIS. These functional groups have strong complexation abilities with transition metals (Wang et al., 2021a; Wang et al., 2021b), thus influencing the transformation and chemical behavior of metals. Moreover, the OA-metals complex can potentially enhance the catalytic generation of reactive oxygen species (ROS) in organic aerosols (Win et al.,





619	2018; Zhang et al., 2022a), thereby playing significant roles in adverse health effects of
620	OA. These results reinforce the significance of HMW HULIS in light absorption, metal
621	complexation, and the potential ROS generation ability of aerosol BrC.

622

# 623 4. Conclusions

This study successfully isolated and characterized HMW and LMW HULIS in 624 625 atmospheric aerosols using the UF-SPE technique, yielding insights into their 626 distribution, optical properties and molecular-level characteristics. Both HMW and 627 LMW HULIS exhibited a continuum of MW distributions ranging from 100 to 20,000 Da. However, HMW HULIS displayed more extensive and intricate MW distributions, 628 suggesting differences in their sources and formation processes compared to LMW 629 630 HULIS. In general, HMW HULIS constituted a higher percentage of TOC and UV254 in aerosols compared to LMW HULIS, indicating the prevalence of hydrophobic and 631 conjugated aromatic structures in the former. Moreover, HMW HULIS exhibited higher 632 aromaticity, stronger light absorption abilities, weaker spectra dependence, and stronger 633 634 humification and conjugation, compared to LMW HULIS. Interestingly, HRMS analysis revealed slightly lower MWw values for HMW HULIS than LMW HULIS, 635 which contradicted the HPSEC results and the nature of UF fractionation. This finding 636 strongly suggests the possibility of small molecules assembling to form 637 638 macromolecules in HMW HULIS. Regarding molecular composition, HMW HULIS contained a higer proportion of CHON compounds, constituting 27% vs. 8% in summer 639 and 58% vs. 30% in winter, while LMW HULIS were primarily composed of CHO 640





641	compounds, accounting for 71% vs. 54%% in summer and 38% vs. 21% in winter. Both
642	HMW and LMW HULIS featured lignin-like substances as major unique molecular
643	formulas, but HMW HULIS exhibited more N-enriched and highly aromatic species,
644	whereas LMW HULIS contained a higher proportion of polar O-containing functional
645	groups. Additionally, HMW HULIS included a greater number of unique lipids-like
646	compounds, while LMW HULIS tend to concentrate more tannin-like compounds.
647	These observations shed light on the complex nature of MW HULIS, and their diverse
648	sources and transformations. Future research should expand the geographical and
649	seasonal coverage to gain a more comprehensive understanding of the molecular-level
650	characteristics of MW HULIS in various atmospheric environments. Furthermore,
651	exploring additional physicochemical properties of MW HULIS will provide valuable
652	insights into their potential health and environmental implications. Overall, this study
653	offers valuable insights into the molecular-level characteristics of aerosol HULIS,
654	enhancing our understanding of their evolution, sources and potential environmental
655	effects.

### 656 Author contribution

Kingjun Fan: Methodology, Supervision, Funding acquisition, Writing-review &
editing. Ao Cheng: Sampling, Data curation. Xufang Yu: Writing-review & editing.
Tao Cao: Sampling, Investigation. Dan Chen: Investigation, Data curation. Wenchao
Ji: Formal analysis. Yongbing Cai: Writing-review & editing. Fande Meng: Writingreview & editing. Jianzhong Song: Methodology, Writing-review & editing. Pingan
Peng: Writing-review & editing.





# 663 Declaration of Competing Interest

- 664 The authors declare that they have no known competing financial interests or personal
- relationships that could have appeared to influence the work reported in this paper.

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- 672

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