

1 Observationally constrained analysis of sulfur cycle in the marine atmosphere with NASA 2 ATom measurements and AeroCom model simulations

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32 Abstract

33
34 The **atmospheric** sulfur cycle plays a key role in air quality, climate, and ecosystems, such as
35 pollution, radiative forcing, new particle formation, and acid rain. In this study, we compare the
36 spatially and temporally **resolved measurements from the NASA ATom mission with**
37 **simulations from five AeroCom-III models for** four sulfur **species**, dimethyl sulfide (DMS),
38 sulfur dioxide (SO₂), particulate methanesulfonate (MSA), and particulate sulfate (SO₄). **We**
39 **focus on** remote regions over the Pacific, Atlantic, and Southern Oceans from near the surface to
40 ~12-km altitude range **covering** all four seasons. **In general, the differences among model results**
41 **can be greater than one-order of magnitude. Comparing with observations, model-simulated SO₂**
42 **is generally low whereas, SO₄ is generally high. Simulated DMS concentrations near** the sea
43 surface **exceed** observed **levels by a factor of five in most cases, suggesting potential**
44 **overestimation of DMS emissions in all models. With GEOS model simulations of tagging**
45 **emission from anthropogenic, biomass burning, volcanic, and oceanic sources, we find that**
46 **anthropogenic emissions are the dominant source of sulfate aerosol (40-60% of the total amount)**
47 **in the ATom measurements at almost all altitudes, followed by volcanic emissions (18-32%) and**
48 **oceanic sources (16-32%). Similar source contributions can also be derived at broad ocean basins**
49 **and on monthly scales, indicating the representativeness of ATom measurements for global**
50 **ocean. Our work presents the first assessment of AeroCom sulfur study using ATom**

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105 1. Introduction

106 Atmospheric sulfur species have wide-ranging environmental and health impacts. About two-
107 third of sulfur emissions come from anthropogenic activities (Chin et al., 2000); therefore,
108 considerable efforts have been made to reduce these sulfur emissions. For example, acid rain
109 occurs when sulfur dioxide (SO₂) is oxidized to form sulfuric acid and particulate sulfate (SO₄),
110 which fall to the ground with the rain (Bian et al., 1993; Grennfelt et al., 2020) and can devastate
111 aquatic ecosystems (Josephson et al., 2014; McDonnell et al., 2021). Through the competing
112 neutralization reaction of SO₄ and nitrate with NH₃ and other alkaline species, SO₄ affects
113 strongly both particulate nitrate formation (Bian et al., 2017) and aerosol pH (Huang et al., 2020;
114 Nault et al., 2021). Sulfate is a key component of particulate matter (PM), which degrades air
115 quality (Dong et al., 2018; Tan et al., 2018) and directly reflects the [solar radiation](#) (Moch et al.,
116 2022; Myhre et al., 2013). Due to its highly hygroscopic nature, sulfate aerosols [act as efficient](#)
117 [cloud condensatin nucleus](#) (Boucher et al., 2013; Breen et al., 2021; Seinfeld et al., 2016) and
118 thus indirectly radiative forcing (Penner et al., 2016; Wang et al., 2021) through aerosol-cloud
119 interactions. The contribution of aerosols to atmospheric clouds and energy budget remains the
120 largest uncertainty in climate models (Gryspeerd et al., 2023; Jia et al., 2021, 2022; Klein et al.,
121 2013; Malavelle et al., 2017). Sulfate is important primarily because the atmospheric sulfate
122 component itself contributes to radiative forcing (RF) almost as much as all other major non-
123 natural aerosol components, as concluded from 16 AeroCom model [studies](#) (Myhre et al., 2013).
124 More importantly, uncertainty in sulfate simulations in current climate models is a major
125 contributor to biases in aerosol optical depth (AOD, Fig. 3 in Glib et al., 2021) and RF (Fig. 7 in
126 Myhre et al., 2013).

127
128 Unlike other major atmospheric aerosols, a significant fraction (i.e., roughly a quarter) of sulfate
129 in the atmosphere comes from marine biological emissions (Chin et al., 1996). The impact of
130 oceanic sulfate is particularly pronounced on marine shallow clouds, which are characterized by
131 low droplet number concentrations and weak updraft velocities (Rissman et al., 2004). Sulfur
132 research has also focused on the tropical upper troposphere (TUT), where the growth of new
133 aerosol particles and homogeneous nucleation involving sulfuric acid is at a maximum
134 (Williamson et al., 2019), and where deep convective transport allows a small portion of the
135 sources to reach the lower stratosphere. The resulting sulfate aerosols in the stratosphere can
136 persist for years (Holton et al., 1995). Unfortunately, the observations in the TUT region and
137 above are sparse. Acquiring atmospheric composition and its chemical/physical properties over
138 remote oceans is challenging, although satellites can often provide total column constraints of
139 aerosol optical depth.

140
141 The NASA Earth Venture Suborbital (EVS-2) Atmospheric Tomography (ATom) airborne
142 mission provided abundant measurements of gases and aerosols over the world's oceans (Hodzic
143 et al., 2020; Thompson et al., 2021). In particular, a suite of instruments integrated on the NASA
144 Douglas DC-8 jetliner (hereafter DC-8) made measurements of many important sulfur species
145 including dimethyl sulfide (DMS), SO₂, particulate methanesulfonate (MSA) and SO₄ over the
146 Pacific and Atlantic Oceans in both hemispheres and the Southern Ocean in all four seasons.
147 [These regions provide us with highly heterogeneous natural and anthropogenic source](#)

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152 [environments, which is not usually the case for traditional continental studies](#). The
153 comprehensive [ATom](#) sulfur dataset provides us with unprecedented opportunities to assess
154 sulfur source, transport, chemistry, deposition, and particle activation and growth represented in
155 the global aerosol models, and to estimate the extent of anthropogenic influence on remote
156 oceanic atmospheric composition and cloud properties.

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158 This study has two specific scientific goals. First, we explore the vertical and seasonal variation
159 of sulfur species (i.e., DMS, SO₂, MSA, and SO₄) using ATom measurements and simulations
160 from five global models that participated in the AeroCom-ATom model experiments. AeroCom
161 is an international initiative of scientists aiming at the advancement of the understanding of the
162 global aerosol and its impact on climate (<https://aerocom.met.no/>). Here we focus on remote
163 regions over the Pacific, Atlantic, and Southern Oceans, from near the surface to an altitude of
164 about 12 km, covering all four seasons. Second, we determine whether the produced SO₄
165 originated from anthropogenic or natural sources by using tagged tracers associated with
166 emission types.

168 Our work is the first study to use ATom measurements for comparison with the AeroCom
169 models, focusing on all sulfur species simulated in current aerosol climate models. This work
170 extends previous efforts using ATom measurements to evaluate the organic carbon (Hodzic et
171 al., 2020) and black carbon (Katich et al., 2018) of AeroCom models, as well as individual
172 models focusing on new particle formation in the tropics (Williamson et al., 2019), fine aerosol
173 lifetime (Gao et al., 2022), aerosol vertical transport (Yu et al., 2019), sea salt (Bian et al.,
174 2019), smoke (Schill et al., 2020), mineral dust (Froyd et al., 2022), and DMS chemistry (Fung
175 et al., 2022). Furthermore, to our knowledge, there are no studies that systematically investigate
176 the changes and sources of all major sulfur species [over the remote ocean](#). Our study aims not
177 only to reveal sulfur variability based on multiple measurements and model simulations, but also
178 to tease out the underlying processes behind the variability through a comprehensive analysis of
179 simulated sulfur species in aerosol climate models.

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181 The structure of this paper is as follows. Section 2 describes the ATom measurements and the
182 AeroCom models used in this study. Section 3 presents the ATom-AeroCom sulfur comparison
183 from different perspectives, namely the overall comparison in Sect. 3.1, the vertical profiles in
184 Sect. 3.2, and the regional and seasonal analysis in Sect. 3.3. The sulfur budget analysis is given
185 in Sect. 4. We further present investigations of source origins for aerosol SO₄ along flight tracks
186 and over oceans in Sect. 5. Finally, we summarize our findings in Sect. 6.

188 2. Data

189 2.1 ATom measurements

190 ATom was a NASA-funded Earth Venture Suborbital project designed to study the effects of air
191 pollution on chemically reactive gases, aerosols, and greenhouse gases in the remote atmosphere.
192 ATom deployed a large suite of gas and aerosol measurement instruments on the NASA DC-8
193 aircraft for systematic sampling, covering an extended region of the globe from 85°N to 85°S
194 over the Pacific and Atlantic Oceans, with vertical profiles from near-surface to near-tropopause
195 (i.e., 0.2-12 km, Thompson et al., 2021). Four ATom deployments (ATom-1 to -4) were
196 executed over each of the four seasons from 2016 to 2018, and their flight paths are shown in
197 Fig. 1. The extensive aerosol and gas measurements made during ATom include inorganic and

200 organic aerosols, precursor gases, particle size distributions and particle composition. Table 1
201 lists the instruments for ATom sulfur species observations used in this study including the
202 relevant sampling details needed for the model comparison.

203
204 We use SO₄ and MSA that had been measured by two instruments, the University of Colorado
205 Aerodyne high-resolution time-of-flight aerosol mass spectrometer (AMS, Canagaratna et al.,
206 2007; Guo et al., 2021), and the NOAA Particle Analysis by Laser Mass Spectrometry (PALMS,
207 Froyd et al., 2019). The latter makes *in situ* measurements of the chemical composition of
208 individual aerosol particles. Furthermore, AMS measured submicron aerosols while PALMS
209 provided mass mixing ratio and size distribution up to 3 μm in dry diameter (Brock et al., 2019).
210 It is worth noting that AMS data were independently processed and reported at both 1-s and 60-s
211 time resolutions by instrument PI (Jimenez et al., 2019). The detection limit varied with different
212 averaging time resolutions, and they were provided directly for each sampling point in AMS
213 datasets. Some negative measurements were also presented in AMS datasets, and this is normal
214 for measurements of very low concentrations in the presence of instrumental noise. The AMS
215 data at 60-s resolution is recommended owing to more robust peak fitting at low concentrations
216 (Hodzic et al., 2020). Given the complex data overlays (i.e., starting, ending, and frequency)
217 reported from multiple instruments, the ATom team also provide a 10-s merged dataset to
218 facilitate users' applications. In this study, we evaluate data reported in different time
219 resolutions, using AMS as an example, to ensure the quality of merged data that are exclusively
220 used as the primary dataset in this work.

221
222 Two instruments were used for SO₂ measurements: the California Institute of Technology
223 Chemical Ionization Mass Spectrometer (CIMS) and the NOAA Laser Induced Fluorescence
224 (LIF) (Table 1). The CIMS uses CF₃O⁻ as a reagent ion which reacts with SO₂ via fluoride ion
225 transfer chemistry. The product ion is detected by a compact time-of-flight mass spectrometer
226 (CToF). The precision of the CIMS SO₂ measurement decreases with increasing water vapor
227 concentration (Eger et al., 2019; Huey et al., 2004; Jurkat et al., 2016; Rickly et al., 2021),
228 making it challenging to measure SO₂ in remote ocean regions. In these regions, the ambient
229 water vapor may be sufficiently high that the CIMS SO₂ precision at 1-s resolution (~130 parts
230 per trillion by volume, pptv) is insufficient for measuring ambient SO₂ value there (<100 pptv).
231 To address this shortcoming, the ATom science team added a new instrument, the NOAA LIF, to
232 the ATom-4 payload. The NOAA LIF instrument uses red-shifted laser-induced fluorescence to
233 detect SO₂ at very low ppt levels (Rickly et al., 2021; Rollins et al., 2016). Both instruments
234 report negative values and the detection limit of the LIF instrument is about 2 pptv.

235
236 DMS was measured during ATom by two instruments, the University of California, Irvine
237 Whole Air Sampler (WAS), and the NCAR Trace Organic Gas Analyzer (TOGA). The WAS
238 reported DMS for all four ATom deployments, while the TOGA reported data for ATom-2 to -4
239 and not for ATom-1 due to possible issues associated with the TOGA inlet (the inlet was
240 changed for ATom-2 to -4). Both instruments have comparable detection limit (1 pptv) and
241 accuracy (~15%). However, the sampling time interval of WAS (variable but ~180s) was longer
242 than TOGA (~120s).

243
244 **2.2 AeroCom models**

245 Five global aerosol models participated in an AeroCom-ATom model experiment
246 (<https://wiki.met.no/aerocom/phase3-experiments>): CAM-ATRAS, E3SM, GEOS, IMPACT,
247 and OsloCTM3. The experiment required all participating models to (1) conduct three-year-
248 simulations of 2016-2018 (i.e., covering the whole ATom observation period); (2) use or nudge
249 meteorological data for the simulation period; and (3) use the same pre-defined emission fields
250 for precursor gases and aerosol tracers. The suggested emissions are the Coupled Model
251 Intercomparison Project Phase 6 Community Emissions Data System (CEDS, Hoesly et al.,
252 2018) for anthropogenic source, daily biomass burning emission (such as The Global Fire
253 Assimilation System, GFAS), a dataset based on satellite volcanic SO₂ observations from the
254 OMI instrument on the Aura satellite (Carn et al., 2016, 2017) for outgassing and eruptive
255 volcanic emission, and DMS concentration in sea surface from Lana et al. (2011). Wind-driven
256 emissions, such as dust and sea salt, are calculated online by each model. Table 2 summarizes
257 the detailed model characteristics and input datasets relevant to this study. It is worth noting that
258 CEDS specifies anthropogenic emissions from various sectors, including emissions from
259 shipping. The version of CEDS used in this work has emissions up to 2014 and all models use
260 2014 emission for ATom periods. Furthermore, unlike other models that use CEDS emissions,
261 the anthropogenic emissions of OsloCTM3 are obtained following Shared Socioeconomic
262 Pathways (SSP) under Representative Concentration Pathway (RCP) scenario with medium
263 radiative forcing by the end of the century (SSP245, Fricko et al., 2017), and the emissions are
264 interpolated to 2016 and 2017. Following the experimental protocol, all models provided results
265 for all ATom [periods](#) except for OsloCTM3 that omitted data in ATom-4. Unlike traditional
266 AeroCom experiments that used gridded daily/monthly averaged data, modelers are required to
267 interpolate model results along flight track every 10 s (see more discussion in Sect. 3.1) using
268 three-dimensional high frequency (e.g., hourly or even less depending on the models' time step)
269 data to facilitate the comparison. It is worth noting that the models do not have any actual
270 information at 10-s time resolution, given their time steps are at least 10[×] greater and their
271 spatial resolutions are coarse. However, the interpolation methodology suggested here provides
272 the best model information at their current configuration to compare with aircraft measurements.

273
274 The AeroCom-ATom experiment also designed three sensitivity simulations by tracking gas and
275 aerosol emissions to anthropogenic, biomass burning, and volcanic sources to attribute the origin
276 of sulfur sources on sulfur simulations over remote oceans. These experiments were conducted
277 with the Goddard Earth Observing System (GEOS) model. The setup of the GEOS model
278 followed the experiment protocol generally, but GEOS used its own daily biomass burning
279 emissions that were derived from the Quick Fire Emissions Dataset (QFED) developed based on
280 MODIS fire radiative power and calculated in near real-time at 0.1° resolution (Darmenov and
281 da Silva, 2015; Pan et al., 2020). Emissions from biogenic sources were calculated using the
282 Model for Emissions of Gases and Aerosols from Nature (MEGAN) embedded in the GEOS
283 model.

285 **2.3 Tag-tracer study in GEOS**

286 Tag trackers or tags are tied to sources of selected emission types and/or emission locations.
287 Such tag isolates plume from certain activities and is a powerful tool to help understand source
288 attribution or diagnose model performance at the process level. The mechanism behind this
289 technique is that each specific aerosol component in GEOS GOCART is modeled independently
290 of the other components, and the contribution of each emission type to the total aerosol mass is

291 not disturbed by the other emission types. Therefore, additional aerosol tracers can be easily
292 "tagged" to capture emission type (e.g., anthropogenic, biomass burning, etc.) and location
293 (local, regional or global scale). Tags can be multi-instantiated and computed simultaneously
294 with their baseline counterparts, thereby increasing the computational efficiency of [the aerosol](#)
295 [models](#).

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297 Tag-tracer technique in GEOS has been widely used in aerosol and gas studies (Bian et al., 2021;
298 Nielsen et al., 2017; Strode et al., 2018) and in supporting various aircraft field campaigns such
299 as Arctic Research of the Composition of the Troposphere from Aircraft and Satellites
300 (ARCTAS) and AToM. Such techniques are also adopted in other models such as GEOS-Chem
301 model (Fisher et al., 2017; Ikeda et al., 2017; Lin et al., 2020) and Community Earth System
302 Model (CESM, Butler et al., 2018).

303
304 Four tags linked to emission types of anthropogenic, biomass burning, volcanic, and marine
305 emissions were used in GEOS model to identify anthropogenic versus natural sources of sulfate,
306 and the results are discussed in Sect. 5.

308 3. AToM-AeroCom comparisons of sulfur species

309 This section presents a comparison of sulfur species between AToM measurements and
310 AeroCom model simulations. The consistency and diversity of data across remote regimes, both
311 horizontally and vertically, help us understand the effects of emissions, transport, and chemical
312 transformations, and shed light on improving the processes in models to best represent the AToM
313 observations.

315 3.1 Overall comparison

316 The overall performance of SO₄ PDF distribution observed from the AMS and PALMS
317 instruments and simulated by five AeroCom models for four AToM deployments is presented in
318 Fig. 2. Also shown in Fig. 2 are the corresponding various percentiles, namely, 0th (minimum),
319 25th, 50th (median), 75th, and 100th (maximum), and the mean for statistical analyses. [The median](#)
320 [and mean values are further given in Table S1](#). The AToM team provided a 10-s merged dataset
321 deliberately by integrating data from various instruments to a unified temporal resolution. We
322 use this 10-s merged data where observations above detection limit (DL) throughout the main
323 text unless otherwise stated. When multiple instruments measured the target field, only points
324 where all instrument measured above DL values were included in analysis, as AMS 10-s in red
325 and PALMS 10-s in grey in Fig. 2. All model results were sampled mimicking flight
326 observations (see Sect. 2.2), and only data with measurements available were used in
327 comparison. This approach ensures that model evaluation is based on high-quality
328 measurements. It is worth noting that the given statistical values in this method represent more
329 regions having high tracer concentration or mixing ratio. In the supplementary material, we
330 further give a model-observation comparison for all available measurement data including
331 negatives.

332
333 The mean of PALMS SO₄ is generally about 10-50% higher than AMS SO₄ across four AToM
334 deployments. This performance may be attributed, at least in part, to the fact that the sample size
335 range of PALMS (~3 μm) is larger than that of AMS (~0.75 μm), as mentioned in Sect. 2.1.
336 However, the difference between the two observations is much smaller than the difference

338 between observation and model. Clearly, the differences in simulated SO₄ among models are
339 high and can easily exceed several orders of magnitude. Most observed and simulated SO₄
340 exhibit highest probability density around SO₄ values of 10-100 ng sm⁻³. With the exception of
341 GEOS and CAM-ATRAS, the model SO₄ PDFs show higher tails beyond 100 ng sm⁻³, which
342 explain the higher median and mean SO₄ simulated by the models. Statistical analysis performed
343 on selected percentiles (box-and-whisker panels in Fig. 2) indicates that multi-model SO₄
344 medians are about 3.7 (ATom-1), 2.2 (ATom-2), 1.9 (ATom-3), and 1.2 (ATom-4) times higher
345 than observed. In general, nearly all measurements and models indicate that SO₄ concentrations
346 on a global ocean basis are highest during the Northern Hemisphere (NH) spring season (ATom-
347 4). Similar analysis was also performed on all (e.g., both positive and negative) measurement
348 data (Fig. S2), the median/mean values of observations are naturally smaller than those in Fig. 2
349 by 8-20%, but the PDF distributions are almost identical between the two treatments.
350

351 Figure 3 shows the PDF distribution and statistics for SO₂. All observed and simulated data were
352 reprocessed by including points above the detection limit (2 pptv) only. Both instruments (CIMS
353 and LIF) were deployed during ATom-4. Despite CIMS being less precise than LIF (Rollins et
354 al., 2016), both instruments agreed within 95% and CIMS measured SO₂ concentrations were
355 consistently 3-7% lower than LIF measurements. This difference is within the combined
356 uncertainties of the two measurements, but it suggests a systematic calibration difference that is
357 currently unresolved (Rickly et al., 2021). Meanwhile, the width of CIMS SO₂ PDF (measured at
358 half-height) is narrower in ATom-4 than ATom-3, because of improved measurement precision
359 in ATom-4. The CIMS resolution was improved in ATom-4, which enables a better separation of
360 SO₂ and formate-H₂O. The CIMS SO₂ PDF in ATom-4 is around 10 pptv and is more consistent
361 with LIF measurements and model simulations. In contrast, the distribution of SO₂ measured by
362 CIMS during ATom-1 to -3 is spread much wider than the models. Throughout ATom periods,
363 models, especially E3SM, GEOS, and OsloCTM3, show higher peak heights and narrower peak
364 widths. Statistics indicate lower model SO₂ medians than observed (box-and-whisker in Fig. 3),
365 especially during ATom-1. However, the model means are comparable or even higher than those
366 observed, indicating that the models simulate episode events [that were not reported in](#)
367 [measurements](#). Consequently, the simulated mean/median ratio is higher than the observed value.
368 Among the four ATom deployments, ATom-4 has much better model observation consistency.
369 Figure S3 presents the corresponding analysis, including the measured negative values.
370 Compared to Fig. 3, the observed median and mean values drop substantially (up to 50%),
371

372 Atmospheric DMS observations are scarce, especially on a global scale. Thus, DMS
373 measurements by the two instruments (WAS and TOGA) during the four ATom deployments
374 provide an unprecedented opportunity to investigate biological DMS over global remote oceans
375 and evaluate model DMS simulations on spatial and temporal distributions. By excluding points
376 with measured values below detection limit (i.e., 1 pptv), the overall DMS comparison in Fig. 4
377 indicates TOGA has higher data peaks and probability densities when DMS ranges from 3-10
378 pptv. However, this does not appear to be consistent with the lower median and mean values of
379 TOGA, indicating a higher tail in the WAS DMS PDF. Likewise, although the peak of WAS
380 DMS PDF is significantly higher than all models from 3-10 pptv (~5-20 pptv for ATom-3), the
381 median and mean of the WAS DMS are lower, suggesting an even higher tail in model DMS
382 PDF. Overall, there is a big gap between the WAS and TOGA DMS measurements, and both are
383 surprisingly low compared to the models. Statistical analysis performed on selected percentiles

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386 (the box-and-whisker) indicates that multi-model DMS medians are about 4.9 (ATom-1), 8.6
387 (ATom-2), 6.6 (ATom-3), and 7.7 (ATom-4) times higher than observed, while model GEOS has
388 a better performance (i.e., 1.2, 2.7, 2.3, and 2.8 correspondingly). [The model DMS median](#)
389 [values are mostly higher than the observed values. The model DMS mean values are even higher](#)
390 [than the observed means \(sometimes by more than a factor of 10\). This reflects a few very high](#)
391 [predicted DMS values.](#) Based on what we know about DMS sources and sinks, these very large
392 simulated DMS appear most commonly in the boundary layer (BL). Indeed it is confirmed in Fig
393 5 by looking at the ratios of DMS median values between model simulations and observations.
394 The analyses are performed on four vertical ranges (e.g., the entire vertical column, the BL 0-1.5
395 km, the low-middle free troposphere 1.5-6 km, and the upper troposphere 6-12 km). The last
396 column “MMM/MOM” refers to multi-model median to multi-observation median. The high
397 ratio stems mostly from the BL, above which the consistency is much better. Meanwhile, the
398 PDF distribution and statistics of the models agree better with the WAS measurement than with
399 the TOGA measurement. We should also acknowledge that this is a very limited set of
400 observations we used here, and that there are some longer-term DMS observations near the
401 surface that were used as input for the parameterization of DMS emissions. More DMS
402 observations near the ocean surface are needed to make a confident comparison.

403 3.2 Vertical profiles

404 Vertical profiles of ATom-1 to -4 for observed and modeled SO₄, SO₂, DMS, and MSA are
405 shown in Figs. 6-9, respectively, for five latitude bands (from the north to the south) and for both
406 the Pacific and Atlantic Ocean basins. Again, the profiles include equal amounts of data for each
407 measurement and model result. In other words, all comparisons show only available points
408 where the two observed values (i.e., AMS vs. PALMS for SO₄ and MSA, CIMS vs. LIF for
409 SO₂, and TOGA vs. WAS for DMS) are greater than their detection limits, and where the model
410 values are extracted.
411

412
413 The average and range of sulfur tracers for ATom-1 to -4 are shown in Figs. 6-9 and their
414 corresponding details in each ATom are further given in Figs. S5-8. As shown in Fig. 6, the SO₄
415 measured by the two instruments are close to each other and lie generally within the [range](#) of
416 modelled SO₄ throughout the ATom periods. The spread of modeled SO₄ concentrations is large,
417 [exceeding an order of magnitude, especially in the upper troposphere.](#) Despite the need for
418 improvements, the models are generally able to capture the shape of the SO₄ profile.
419 Specifically, CAM-ATRAS and GEOS have good SO₄ vertical gradients over the tropical and
420 NH oceans, but their SO₄ values are too low compared to measurements over the Southern
421 Hemisphere (SH) free troposphere. The SO₄ of IMPACT and OsloCTM3 decreases too slowly
422 with altitude, as shown by their overestimated SO₄ values at high altitudes globally. [The results of](#)
423 [E3SM are generally within the ranges as predicted by the other models.](#) However, the performance of
424 these models’ SO₄ vertical profiles cannot simply be explained by the way the oxidant is applied,
425 because among the five models, CAM-ATRAS, IMPACT, and OsloCTM3 used interactive
426 oxidant calculations, while E3SM and GEOS used archived oxidant data (Table 2). [Of the five](#)
427 [models, OsloCTM3 and GEOS participated in the multi-model OH assessment \(Nicely et al.,](#)
428 [2000\) and OsloCTM3 had a shorter methane lifetime \(relative to OH\) than GEOS.](#)

429
430 Figure 7 shows generally lower modeled SO₂ volume mixing ratios compared to the CIMS
431 observations for most altitudes and latitude bins. The spread among modeled SO₂ values exceeds

Deleted: Even though the model DMS median is mostly higher than the observed value, the degree of overestimation is not as serious as the mean value that can be more than tenfold, indicating a few points are simulated with extraordinarily high DMS values.

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443 an order of magnitude around the measured SO₂. SO₂ is better simulated by model IMPACT in
444 the NH [than other four AeroCom models](#) and by models CAM-ATRAS and OsloCTM3 in the
445 SH than other [three](#) AeroCom models. The tropical Pacific appears to be an interesting region,
446 with all models except GEOS failing to capture observed local SO₂ sources. Basically, the
447 observed SO₂ is high at the surface, falls rapidly in the BL, and then gradually decreases above
448 the BL, except for ATom-1, during which a second peak appears just above the BL (see Fig. S6
449 for the details of ATom-1 to -4 separately). These observations indicate a strong local source for
450 SO₂ in all seasons and a transport source in the low free-troposphere NH summer (ATom-1).
451 Like observations, the model GEOS [predicts](#) a local source for SO₂ at the surface, but it misses
452 the plume above the BL in ATom-1, and its vertical SO₂ convection is consistently too weak.
453 Since only one flight was in ATom-1, more observations are needed to confirm whether GEOS
454 has been failing to catch the plume there during the NH summer. All other models show lower
455 SO₂ at the surface than in the lower free troposphere, which is inconsistent with the observed
456 profiles. Figure S6 also shows an excellent agreement of SO₂ profiles measured by the CIMS and
457 LIF during ATom-4 and models agree with measurements better in ATom-4 as well.

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458 DMS measurements fill in another piece of the puzzle for the atmospheric sulfur budget. As
459 shown in Fig. 8, all five AeroCom models generally overestimate DMS in the BL, particularly
460 for models CAM-ATRAS and OsloCTM3. This large bias close to the surface requires us to
461 revisit the DMS emissions employed in our models. Of the five models, DMS emissions of
462 E3SM, and IMPACT are derived directly from climate emission inventories, while the DMS
463 emission of the other three models are parameterized using monthly climatological DMS
464 concentrations in sea water and surface meteorologies (e.g. surface wind and temperature, see
465 details in Table 3). Specifically, the parameterization used to convert DMS seawater
466 concentrations into DMS emission fluxes was using Nightingale et al. (2000) in CAM-ATRAS
467 and OsloCTM3 and Liss and Merlivat (1986) in GEOS. The three models used two inventories
468 of monthly DMS seawater concentrations, Lana et al. (2011) for CAM-ATRAS and GEOS, and
469 Kettle and Andreae (2000) for OsloCTM3. It is worth noting that even the latest climatological
470 database by Lana et al. (2011) was constructed by compiling measurements before 2000, so the
471 potential long-term change of DMS emission caused by environment change could be missed
472 (Barford, 2013). Also, although the data [set](#) used by Lana et al. (2011) is large (i.e., ~47,000
473 seawater concentration measurements), interpolation and extrapolation techniques were still
474 necessary in creating a global monthly climatological DMS emission. Gali et al. (2018) reported
475 updated oceanic DMS levels on a global scale using remote sensing satellite data. However,
476 much effort is still needed to accurately establish global rates of change in order to create global
477 DMS emissions for climate modeling. This parameterization of air-sea exchange is important
478 because CAM-ATRAS and OsloCTM3, using the same parameterization but different DMS
479 seawater concentrations, reported close emissions in [Table 4](#). On the other hand, the DMS
480 emissions of CAM-ATRAS are almost twice as high as those of GEOS. This difference in
481 emissions results from different parameterizations in the two models, since both models read the
482 same DMS seawater concentration.

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485 Meanwhile, the modeled DMS vertical gradient is generally steeper than the observed one (e.g.,
486 Fig. 8 A54N-90N), implying slower vertical transport or faster chemical conversion of DMS to
487 SO₂ in the model. The data [submitted by the AeroCom models](#) did not provide us with enough
488 information to obtain the determinants. Currently, GEOS and OsloCTM3 account for two

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492 products from the oxidation of DMS (i.e., SO₂ and MSA) but only GEOS output MSA results.
493 The other models consider DMS oxidation products only as SO₂. These chemical processes in
494 the model may also need to be revisited. Previous studies proposed other chemical reactions for
495 DMS loss in the atmosphere. For example, halogen chemistry represented 71% of the DMS loss
496 in the study of Hoffmann et al. (2016). Veres et al. (2020) estimated that about 30% of DMS in
497 the atmosphere was oxidized to hydroperoxymethyl thioformate (HPMTF), reported only in
498 ATom-4. To this end, the HPMTF serves as a new reservoir of oceanic sulfur and its life cycle in
499 the atmosphere is unknown. The new finding indicates that important components of Earth's
500 sulfur cycle are not yet been fully understood and urges us to reassess this fundamental marine
501 chemical cycle. However, including these chemical DMS losses further reduces DMS above the
502 surface, making DMS in the models even lower at high altitudes.

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504 The GEOS MSA matches observations (Fig. 9) in the lower troposphere. In the upper
505 troposphere (UT), the GEOS MSA tends to decrease slowly or even increase with altitude. These
506 patterns do not agree with observations, and this inconsistency can be explained at least partially
507 by the MSA gas-aerosol partitioning defined in the model and observations. AMS and PALMS
508 only measure the particle phase of MSA, but GEOS MSA is the total MSA and is not accurately
509 represented by observations, especially in UT. Yan et al. (2019) reported that the ratio of MSA to
510 SO₄ can be reduced by 30% when calculations do not consider methanesulfonic acid in the gas
511 phase (MSAg) at low temperatures.

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513 3.3 Regional and seasonal analysis

514 In order to analyze model performance on a regional and seasonal basis, Figs. 10-12 show
515 histograms of SO₄, SO₂, and DMS concentrations as a function of altitude (rows) and latitudinal
516 band (columns). Only multi-model median is shown here to highlight any common problems in
517 the models. Further details of each individual model are given in Figs. S9-11 and discussed in
518 supplementary material. Each model in this study has its bias at a specific time and location.
519 With the information provided by Figs. S9-11, modelers can further explore the simulation to
520 identify potential causes of model anomalies.

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521 High SO₄ concentration regions vary across seasons (Fig. 10). In the free troposphere (i.e., 1.5 –
522 12 km), these regions cover the tropics to mid-latitudes in summer and winter (i.e., ATom-1 and
523 ATom-2) and shift to mid- to high-latitudes in spring and autumn (i.e., ATom-3 and ATom-4).
524 The areas with the highest concentration appeared in the SH high-latitudes during ATom-3 (SH
525 spring) and the NH high-latitudes during ATom-4 (NH spring). In the BL, the tropical
526 atmospheric SO₄ concentration appears to be always elevated, and SO₄ concentration levels and
527 SO₄ interregional variation are more pronounced in ATom-1 (NH summer). Among all AToms,
528 the performance of the model SO₄ simulation is best for ATom-4 and worst for ATom-1 (NH
529 summer). Compared to observations, model tends to simulate higher SO₄ concentrations in the
530 free troposphere. Both observations and simulations show that the SO₄ over the Pacific is higher
531 than that over the Atlantic during the NH high-latitude autumn (ATom-3) and the NH mid-
532 latitude spring (ATom-4). The differences between observations and simulations are generally
533 larger in the Atlantic than in the Pacific, particularly in the SH. SO₄ concentration levels in
534 simulations and observations can differ significantly in certain areas of each ATom. Differences
535 may be caused by majority models or a few individual models. For example, in summer and
536 winter, the CAM-ATRAS model gave the highest estimates of atmospheric SO₄ in the oceanic

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553 BL, but the IMPACT and OsloCTM3 models gave the highest estimates of atmospheric SO₄ in
554 the free troposphere (Fig. S9). All models except the GEOS model generally overestimate SO₄ in
555 the atmosphere.

556
557 Atmospheric SO₂ (Fig. 11) is most abundant in the BL of NH mid-latitude Pacific Ocean during
558 ATom-1 (NH summer) and the tropical Pacific BL during ATom-3 (NH autumn), and this high
559 SO₂ region extends to the atmosphere above. Areas where free tropospheric SO₂ concentrations
560 are relatively large do not necessarily follow the example of the BL. For instance, free
561 troposphere appears to be more polluted than other regions in the NH Pacific during ATom-2
562 and in the SH mid-latitude Atlantic (A40S-20S) during ATom-4, but not in the BL, implying a
563 potential source of SO₂ by horizontal transport. The interregional variation of SO₂ in BL is much
564 larger than in the free troposphere, from which local oceanic sources of SO₂ can be inferred. In
565 terms of model-observation comparison, model simulated SO₂ in the free troposphere is
566 generally lower, which is opposite to the case of SO₄. A rapid SO₂ to SO₄ chemical conversion in
567 models could be one of reasons. Fig. S10 further shows individual model SO₂ simulation. For
568 example, the E3SM model gives significantly higher SO₂ compared with the measurements and
569 other models in BL (Fig. S10). Unlike the case of SO₄, all models tend to underestimate SO₂ in
570 the free troposphere, with some exceptions, such as the GEOS model for the mid- to high-
571 latitude North Pacific winter (ATom-2) and the CAM-ATRAS and IMPACT models for the mid-
572 latitude South Atlantic autumn (ATom-4).

573
574 Surface DMS (Fig. 12) is generally higher in the tropics when the ocean is warm and in mid-high
575 latitudes during springtime (e.g., ATom-3 SH spring and ATom-4 NH spring). A remarkable
576 pattern of high model DMS values in the BL is revealed throughout the ATom cycle. This
577 phenomenon also occurs in the free lower troposphere, but not necessarily in the upper
578 troposphere. The high model DMS in BL can be attributed to (1) too high DMS emission, (2) too
579 slow DMS chemical loss, and (3) too slow DMS vertical transport from BL to free troposphere.
580 Additional insight can be obtained by focusing on remote high-latitudes, for example SH high-
581 latitude (40°S-70°S) Pacific, where land source impacts are limited. Thus, the higher simulated
582 SO₂ there in the BL in SYom-4 ruled out a chemical cause due to low DMS loss. The extremely
583 high surface DMS is also not due to the slow vertical transport because simulated DMS is also
584 high in the layers above the BL. A large model DMS emission is likely responsible for the
585 simulated high surface DMS. The overestimation of surface DMS multi-model median in Fig. 12
586 is clearly attributable to the contribution of all models shown in Fig. S11, with the models CAM-
587 ATRAS and OsloCTM3 being more prominent.

588 589 **4. Sulfur budget from AeroCom models**

590 Budget analysis is a simple and basic method that has been widely used to document the
591 underlying performance of a model. This analysis allows us to evaluate the AeroCom-III sulfur
592 simulations against previous AeroCom-I and -II studies and serves as a record for future model
593 evaluations. Table 4 summarizes the global sulfur budgets for emissions, wet/dry deposition and
594 chemistry from the five models. Clearly, the largest source of sulfur (~70 TgS/yr) is SO₂ emitted
595 directly from anthropogenic (~78%), biomass burning (~2%), and volcanic sources (~20%).
596 Biogenic DMS (~15-30 TgS) produced and outgassed from decomposition of marine organic
597 molecules provides the largest natural source of sulfur to the atmosphere. A small amount of SO₄
598 (< 3%) is emitted directly from anthropogenic sources.

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600
601 DMS is oxidized in the atmosphere by OH and NO₃ radicals to form SO₂ and MSA. This
602 biological source of SO₂, along with SO₂ emitted directly from other sources, reacts with
603 hydroxyl radicals (OH) in the gas phase and hydrogen peroxide (H₂O₂) and ozone (O₃) in the
604 aqueous phase to produce sulfuric acid (H₂SO₄) and eventually sulfate particles, which play an
605 important role in the formation of clouds over the oceans.

606
607 In the five models, DMS predicts the shortest global average lifetime (0.6-2.0 days), followed by
608 SO₂ (1.1-1.8 days), and SO₄ the longest lifetime (3.1-5.6 days). Among them, GEOS has the
609 lowest global burden and shortest lifetime for all sulfur species. The magnitudes of global
610 burdens and lifetimes shown here support the model performance shown in Figs. 2-8. For
611 example, models CAM-ATRAS and OsloCTM3 predicts the highest DMS emission, which is
612 consistent with the highest DMS value (Fig. 4 and S11) and longest lifetime simulated by the
613 two models.

614
615 The key budget items include DMS emission, SO₂ emission, sulfate source or total deposition
616 (source and deposition are pretty much the same as expected), lifetime (inversely proportional to
617 the loss rate), and total atmospheric mass load. From the multi-model mean and standard
618 deviation, the diversity can be calculated. Figure 13 shows the global mean budget items in the
619 percentage deviation of each model from the multi-model mean, following the same concept
620 shown in Schulz et al. (2006) and Gliss et al. (2021). It reveals the processes causing model
621 differences. For example, E3SM and GEOS have approximately the same SO₂ emissions and
622 total sulfate sources, but the sulfate lifetime is much shorter in GEOS (implying faster removal
623 rates) thus smaller sulfate burden that is consistent with lower sulfate concentrations in GEOS
624 than in E3SM. At the same time, the lower total sulfate source in E3SM is compensated by
625 longer lifetime compared to CAM-ATRAS, resulting in a comparable global burden of SO₄ in
626 the two models.

627
628 It is worth pointing out that the much lower atmospheric SO₄ mass loading of the GEOS
629 simulations is not necessarily related to the poor performance of the GEOS SO₄ simulations, as
630 revealed by the model-measurement comparison in Figs 2, 6 and S9. Although the multi-model
631 mean (or median) often represents the best predictor in the modeling domain, common modeling
632 problems or too small model sample can compromise this effort.

633
634 To date, there have been no sulfur budget reports focusing on the vast ocean. However, previous
635 AeroCom studies have reported global sulfate atmospheric loading and its diversity across
636 multiple AeroCom models using monthly and global mean column loadings. Table 5 summarizes
637 these studies, including their reported global and annual sulfate multi-model mean (MMM) and
638 diversity (δ). δ is related to the standard deviation (std_dev) and is defined as $\delta = \text{std_dev} /$
639 $\text{MMM} * 100$ (%). The results of this work are lower than AeroCom-I but higher than AeroCom-
640 II, which may be related to the different target years involved in these studies. One point to note
641 is that the diversity δ of AeroCom-III models has not reduced since AeroCom-I, which was
642 studied nearly 20 years ago.

643 644 5. Source origins for aerosol SO₄ along flight track and Ocean basins

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651 In this section, we perform an analysis of source attribution by tagging the sulfur source types
652 using the GEOS model. This model is the only one that provides tagged data. Our goal is to
653 understand the sources (anthropogenic, biological, volcanic) of sulfate aerosols in remote regions
654 and how chemistry, transport, and removal processes determine the vertical distribution of
655 sulfate aerosols across seasons and ocean locations.

656
657 Figure 14a presents a quantitative summary of the source attribution of aerosol SO₄ sampled
658 along the ATom flight tracks. The analysis was performed over four seasons, spanning the
659 troposphere and three vertical layers (i.e., marine boundary layer, free troposphere and upper
660 troposphere). Overall, anthropogenic emissions were the dominant source (40–60% of the total)
661 of simulated tropospheric SO₄ along the ATom flight tracks for almost all altitudes and seasons,
662 followed by volcanic (18–32%) and oceanic sources (16–32%). Anthropogenic pollution
663 prevailed over remote oceans most in spring and autumn (ATom-3 and -4). The overall
664 contributions from volcanic and oceanic sources are comparable during the ATom periods.
665 Meanwhile, the ocean source contribution has an obvious seasonal variation which is most active
666 during the SH summer (ATom-2), when marine biochemical activity in the vast Southern Ocean
667 is the largest. Volcanos show the largest contribution in the NH summer 2016 (ATom-1) during
668 the four ATom deployments. Given the irregular character of eruptions, the volcanic contribution
669 deserves further discussion below.

670
671 In the vertical direction, SO₄ from anthropogenic emissions contributes more than 50% to the
672 free to upper troposphere. Even in the marine boundary layer, anthropogenic sources of SO₄ still
673 account for the largest fraction, except in the SH summer (ATom-2) when oceanic source
674 became dominant. The relative importance of volcanic and marine sources varies not only
675 seasonally but also vertically. Oceanic sources understandably make up a significant fraction
676 (26–42%) of SO₄ in the boundary layer. In the free troposphere, their contribution drops off
677 sharply, reflecting their local surface source characteristics. On the other hand, SO₄ from
678 anthropogenic emissions (including shipping emission) expands in the free troposphere,
679 suggesting that the source originated from distant continental areas. Volcanic SO₄ remains nearly
680 constant throughout the troposphere, making volcanoes the second largest source there.
681 Meanwhile, the contribution of others (OTH including biomass burning) to remote ocean SO₄ is
682 relatively small (< 3%) and will not be discussed further in this study.

683
684 The sources of SO₄ discussed above are deduced from the location and timing of the ATom
685 flight path. Conclusions about the total contribution of the ocean needs caution, as there may be
686 representativeness issues using such narrow-band and instantaneous sampling. There might be a
687 situation where, for example, volcanoes provide a very large signal but only account for a small
688 measured area, and in most regions, volcanoes play a very minor role. Whereas oceanic sources
689 in the marine boundary layer perhaps were the dominant source for a much wider region, the SO₄
690 concentration resulting from the DMS was overall a smaller amount compared to other sources
691 where near a volcanic or anthropogenic source. To address this representation issue, we perform
692 one more analysis with the model data averaged over a wider oceanic region (the shaded area in
693 Fig. 1) and over a longer period (i.e., monthly mean over ATom periods). Such source
694 attributions are given in Fig. 14b.

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701 Qualitative conclusions drawn from source attribution along the flight tracks generally apply to
702 the ocean basin source attribution, albeit to a slightly different extent. This confirms that
703 continental man-made sources dominate tropospheric SO₄ even over oceans. There is a clear
704 seasonal variation in oceanic contribution, which is largest in austral summer (ATom-2)
705 followed by boreal summer (ATom-1). Concerning volcanic sources, emissions from volcanoes
706 are of two types. One type is the volcanic degassing emissions that tend to remain nearly
707 constant throughout the year and are equivalent to about 20% of SO₂ global anthropogenic
708 emissions. This degassing emission ensures that volcanoes contribute more than 20% to SO₄
709 over the oceans. The other type consists in the volcanic eruptions. Due to the irregularity of
710 volcanic eruptions in terms of different eruption locations, magnitudes, and times, volcanic
711 eruptions can cause severe fluctuations in SO₄ in the atmosphere. Compared with the source
712 attribution along the flight trajectory, the volcanic contribution decreased over a larger spatial
713 and temporal domain (i.e., ocean basin and monthly mean) in the NH winter 2017 by 32%
714 (ATom-2) and increased in all other three seasons by 14-33%, especially in the NH spring 2018
715 (ATom-4), when the massive Kilauea eruption in Hawaii began on 3 May 2018. Contrarily, the
716 anthropogenic contribution increased in the NH winter (ATom-2) by 5% and decreased in other
717 seasons by 7-21%.

718 719 **6. Conclusions**

720 This study investigates sulfur [species](#) in remote tropospheric regions at global and seasonal
721 scales using airborne ATom measurements and AeroCom models. The goal is to understand the
722 atmospheric sulfur cycle over the remote oceans, each model's behavior and the spread of model
723 simulations, as well as the observation-model discrepancies. Such understanding and comparison
724 with real observations are crucial to narrow down the uncertainty in model sulfur simulation.
725 Even after decades of development, models are still struggling to accurately simulate sulfur
726 distributions, with differences between models often exceeding an order of magnitude. On the
727 other hand, the agreement between instruments is usually much better. Differences between
728 modeled SO₄ are particularly large in the tropical upper troposphere, where deep convective
729 transport allows a small portion of sulfur to reach the lower stratosphere where sulfate aerosols
730 can persist for many years. Compared with observations, simulated SO₂ is generally low while
731 SO₄ is high. Modeled DMS values are typically an order of magnitude higher than observed
732 DMS near the surface, pointing to a need to revisit the DMS emission inventories and/or the
733 biogeochemical modules used to predict DMS emissions. Our work also suggests investigating
734 three other potential corresponding processes [to improve sulfur simulation](#): whether the chemical
735 conversion from SO₂ to SO₄ is too rapid, whether DMS-generated free tropospheric SO₂ is too
736 low, and whether the vertical transport of DMS and SO₂ from BL to free troposphere is too low.
737 This further investigation requires atmospheric oxidant fields and the ability to track SO₂
738 production and loss using tagged tracers.

739
740 We investigate source attribution of SO₄ over remote oceans seasonally and vertically. Sampled
741 at the location and time of ATom measurements, anthropogenic emissions were the dominant
742 source (40–60% of the total) of simulated tropospheric SO₄ at almost all heights and seasons,
743 followed by volcanic (18–32%) and oceanic sources (16–32%). These contributions changed to
744 34–56%, 17–37%, and 19–37% when extended to the broad Pacific and Atlantic during the
745 months of ATom deployment. This survey confirms that anthropogenic sources dominate
746 tropospheric SO₄ even over oceans. Given that we find DMS source to be overestimated in the

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749 models, the anthropogenic sources overall are a larger portion of the budget, and biogenic is
750 likely smaller than volcanic. Volcanic degassing throughout the year contributes about 20%, and
751 this proportion is increased by explosive eruptions that vary in location and timing. The oceanic
752 contribution has obvious seasonal variation, the largest in the Southern Hemisphere summer,
753 followed by the Northern Hemisphere summer.

754
755 It is understood that anthropogenic sulfur emissions currently offset a significant portion of
756 greenhouse gas warming, but they are rapidly declining through emissions controls. As these
757 anthropogenic emissions decrease, natural sources of sulfur, particularly bio-derived sulfur
758 compounds discharged from the world's oceans, will increase their relative contribution.
759 Therefore, more efforts are needed to understand the sulfur cycle in remote environments. On the
760 other hand, our study is the first asserting that anthropogenic emissions remain a major source of
761 sulfate aerosols generated over remote oceans during the ATom deployment periods, suggesting
762 that any limitation of anthropogenic sulfur emissions would have modern global implications.

763
764 Even after two decades of development, the diversity of sulfate simulations from AeroCom-I to
765 AeroCom-III has not decreased. However, accurate sulfate simulation in current climate models
766 is crucial to reduce radiative forcing biases. More importantly, apart from the shortcomings of
767 individual models, all modelers [involved in this work](#) should focus on the calculation of the air-
768 sea exchange flux formula as it plays a key role in determining DMS emissions. [To our
769 knowledge, many other aerosol models employ similar formulas in air-sea flux calculations, so
770 the findings here are applicable to them as well.](#) Modelers also need to study DMS and SO₂
771 vertical transport as well as SO₄ wet deposition during long-distance transport, as model biases
772 are greatest at high altitudes. One suggestion to modelers is that the use of online oxidant fields
773 is insufficient to explain the model sulfate bias, as there was no systematic bias in the sulfate
774 simulations between the models using interactive oxidants and the models using archival
775 oxidants in this study. The complexity of chemistry deserves more attention.

776
777 *Code availability.* The GEOS Earth System Model source code and the instructions for model build are available
778 at <https://github.com/GEOS-ESM/GEOSgcm/> (Last accessed: 28 August 2023).

779
780 *Data availability.* The AeroCom model outputs needed to reproduce the results described in this paper are
781 publicly available for download at <https://acd-ext.gsfc.nasa.gov/anonftp/acd/tropo/bian/ATom-AeroCom-Sulfur/>.
782 The ATom data was obtained from their ESPO Data Archive: <https://espo.nasa.gov/atom/content/ATom>, last
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784
785 *Author contributions.*

786 BH and MC conceptualized ATom-AeroCom experiment. BH performed analysis and wrote the manuscript. BH,
787 PRC, MLi, MTL, RBS, HM, JEP, HW, KZ, and JZ provided AeroCom model results and ECA, KF, RSH, JJ, PCJ,
788 MLa, BAN, AWR, GS, and LX contributed to ATom measurements. All authors contributed to the editing of the
789 manuscript.

790
791 *Competing interests.*

792 At least one of the co-authors is a member of the editorial board of Atmospheric Chemistry and Physics.

793
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1306 Table 1. ATom sulfur measurements used in the study

Instrument	SO ₄		SO ₂		MSA		DMS	
	AMS ^a	PALMS ^b	CIMS ^c	LIF ^d	AMS	PALMS	TOGA ^e	WAS ^f
ATom deployment(s)	1 to 4	1 to 4	1 to 4	4	1 to 4	1 to 4	2 to 4	1 to 4
Frequency	60 s	180 s	1 s	1 s	1 s	180 s	120 s	Variable but ~180 s
Accuracy	±35% (2s)	±60% at 10 ng m ⁻³ ±20% at 1 µg m ⁻³	±25%	± 9% (1s)	±35% (2s)	±70%	15% or better	15%
precision			130pptv					10%
Detection limit	5-15 ng sm ⁻³	~10 ng sm ⁻³		2 pptv	2.5 ng sm ⁻³ (60 s)	~15 ng sm ⁻³	1 ppt	1 ppt
Cut-off size (dry diameter)	~0.75 µm	0.1-3 µm			~0.75 µm	0.1-3 µm		
Primary Investigator(s)	Jose Jimenez and Pedro Campuzano Jost	Karl Froyd and Gregory Schill	Paul Wennberg	Andrew Rollins	Jose Jimenez and Pedro Campuzano Jost	Karl Froyd and Gregory Schill	Eric Apel	Donald Blake
References	Guo et al., 2021; Schueneman et al., 2021	Froyd et al., 2019	Allen et al., 2022; Crouse et al., 2006	Rollins et al., 2016	Hodshire et al., 2019	Froyd et al., 2019	Apel et al., 2015	Simpson et al., 2001

1307 ^aAMS: Aerosol Mass Spectrometer

1308 ^bPALMS: Particle Analysis by Laser Mass Spectrometry

1309 ^cCIMS: Chemical Ionization Mass Spectrometer

1310 ^dLIF: Laser Induced Fluorescence

1311 ^eTOGA: NCAR Trace Organic Gas Analyzer

1312 ^fWAS: Whole Air Sampler

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1314 Table 2. AeroCom Models used in this study

Model Abbreviation	Model Version	Nominal Resolution	Vertical Levels	Meteorological Fields	Ocean Surface Temperature Data	Interactive Aerosol-Meteorology	Endogenous Oxidants	Endogenous DMS Emission	Aerosol Module	Anthropogenic Emission	Volcano Emission	Key References
CAM-ATRAS	CAM5-ATRAS2	1.9° × 2.5°	30	MERRA-2	HadSST	Yes	Yes	No	Microphysics, 12 sectional size bins, and internal mixing of aerosol constituents in each bin.	CEDS (Hoesly et al., 2018)	Degassing (Andres and Kaagoo, 1998), Eruption (Neely and Schmidt, 2016)	Liu and Matsui 2021; Matsui 2017; Matsui and Mahowald, 2017
E3SM	v1.0	1° × 1°	72	ERA-Interim	HadSST	Yes	No	No	Microphysics, MAM4, internal mixing within a mode, external mixing between modes	CEDS (Hoesly et al., 2018)	Continuous emission (Denener et al., 2006), No eruptive emissions.	Rasch et al., 2019; Wang et al., 2020; Zhang et al., 2022
GEOS	Icarus-3 3 p2	1° × 1°	72	MERRA-2	MERRA sst	Yes	No	Yes	GOCART, Bulk, external mixing	CEDS (Hoesly et al., 2018)	Carns et al., 2016, 2017	Bian 2017; Colarco et al., 2010; Chin et al., 2000
IMPACT		1.9° × 2.5°	30	Open IFS ECMWF	HadSST	No	Yes	no	Microphysics, internal mixing within a mode, external mixing between modes	CEDS (Hoesly et al., 2018)	AeroCom volcanic emissions	Zhu et al., 2017; Zhu et al., 2019
OsloCTM3	OsloCTM3v1.02	2.25° × 2.25°	60	Open IFS ECMWF	Open IFS ECMWF	No	Yes	Yes	Bulk, external mixing	SSP245 with linear interpolation for 2017	AeroCom volcanic emissions, continuous from Dentener (2006)	Lund et al., 018; Sowde et al., 2012

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1316 Table 3. DMS emission used/calculated by the five AeroCom models

Model abbreviation	Emission inventory	DMS concentration in sea water	DMS flux calculation	Meteorological fields
CAM-ATRAS	No	Lana et al. (2011)	Nightingale et al. 2000	Wind from ECMWF-IFS
E3SM	Yes			
GEOS	No	Lana et al. (2011)	Liss and Merlivat, (1986), Saltzman et al. (1993)	SST and wind from GEOS
IMPACT	Yes			
OsloCTM3	No	Kettle and Andreae (2000)	Nightingale et al. (2000)	Wind from ECMWF-IFS

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Table 4. Global sulfur budget in 2017

		Emission		SUPSO ₂ ¹	SUPMSA	SUPSO ₄	Dry	Wet	TotalSource	Burden	Lifetime
		TgS/yr	TgS/yr	TgS/yr	TgS/yr	TgS/yr	TgS/yr	TgS/yr	TgS/yr	TgS	days
CAM- ATRAS	DMS	26.05	-26.05	--	--	--	--	--	26.05	0.13	1.8
	SO ₂	68.67	26.05	--	-55.67	-39.05	--	--	94.72	0.445	1.7
	SO ₄	1.76	--	--	55.67	-4.72	-53.23	--	58.09	0.67	4.2
E3SM	DMS	19.43	-19.40	--	--	--	--	--	19.43	0.0658	1.24
	SO ₂	67.92	19.40	--	-38.56	-48.76	--	--	87.32	0.3825	1.60
	SO ₄	1.74	--	--	38.56	-6.95	-33.31	--	40.31	0.6183	5.60
GEOS	DMS	15.57	-14.84	-0.74	--	--	--	--	15.57	0.0252	0.59
	SO ₂	67.06	14.84	--	-37.49	-32.93	-11.39	--	81.90	0.3488	1.55
	SO ₄	1.68	--	--	37.49	-5.27	-33.90	--	39.17	0.3269	3.05
	MSA	--	--	0.74	--	-0.10	-0.64	--	-0.74	0.0063	3.11
IMPACT	DMS	18.22	-18.22	--	--	--	--	--	18.05	0.0369	0.75
	SO ₂	64.76	18.22	--	-51.44	-31.29	--	--	82.98	0.4134	1.82
	SO ₄	1.36	--	--	51.44	-3.48	-49.32	--	52.80	0.7502	5.19
OsloCTM3	DMS	26.93	-26.93	--	--	--	--	--	26.93	0.1496	2.03
	SO ₂	52.80	26.93	--	-49.23	-29.01	-1.49	--	79.73	0.2346	1.08
	SO ₄	1.053	--	--	55.49	-6.35	-50.29	--	56.54	0.8681	5.60

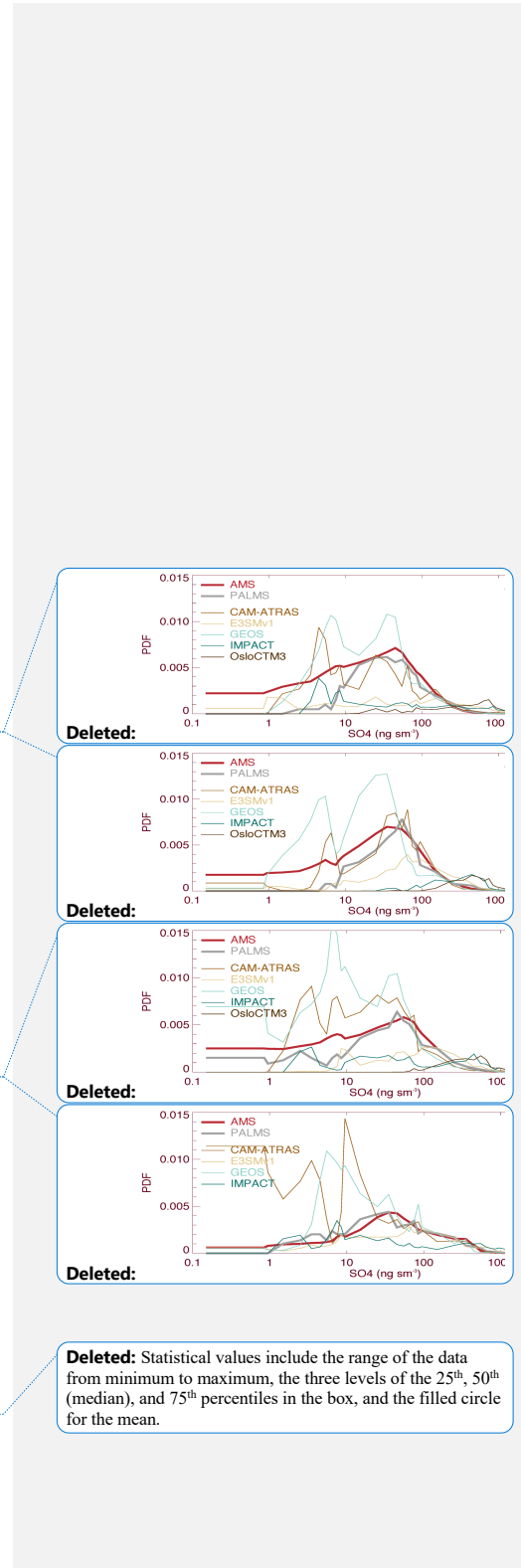
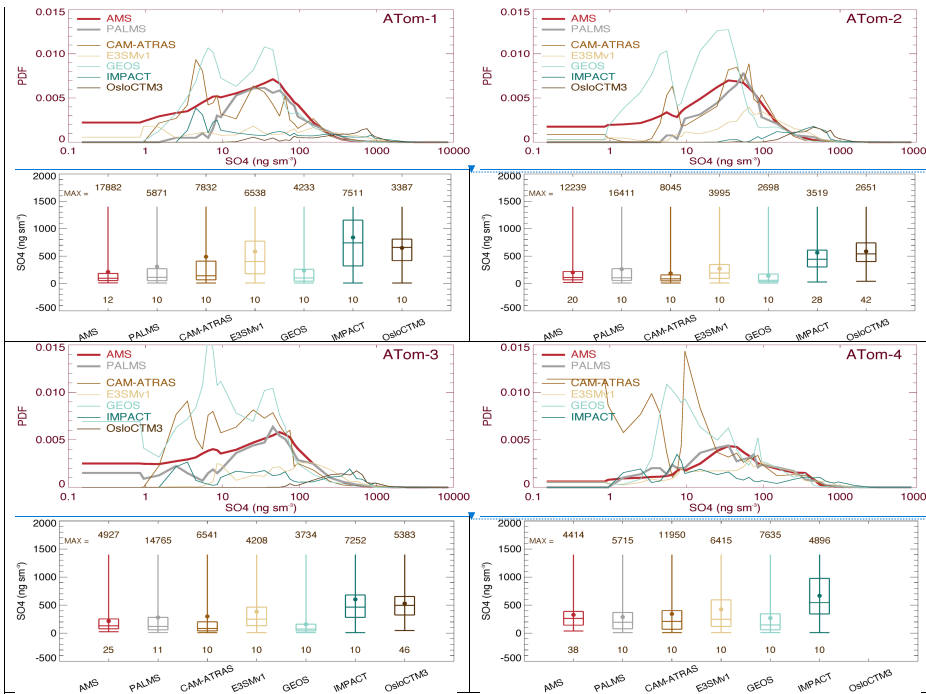
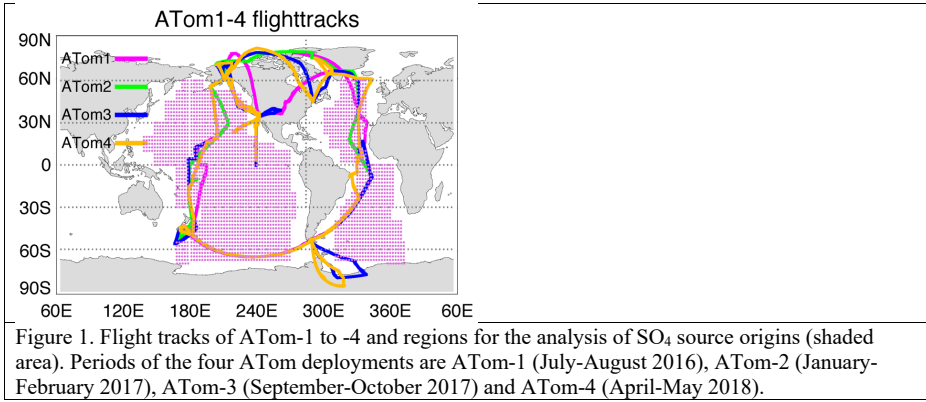
¹SUPSO₂: chemical production for SO₂

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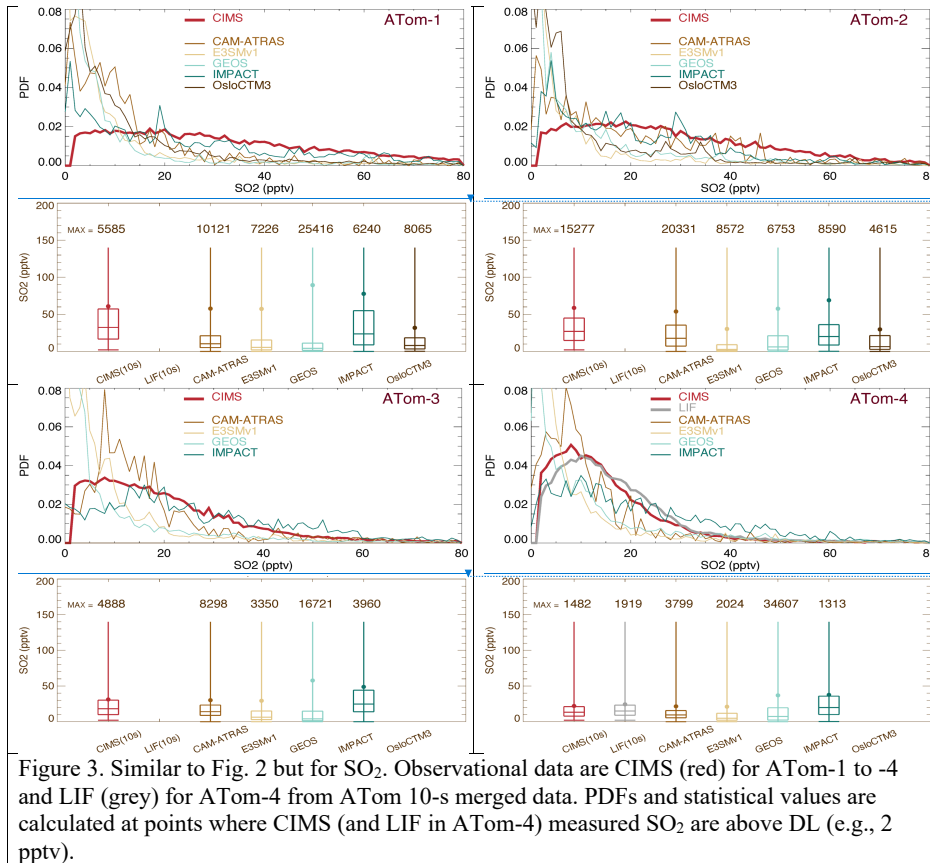
Table 5. Global and annual sulfate multimodel mean and diversity from three AeroCom phases

reference	AeroCom-I		AeroCom-II		AeroCom-III	
	Textor et al., 2006	Myher et al., 2013	Kipling et al., 2016	Gliß et al., 2021	This work	
Study year	2000	2006	2006	2010	2017	
# of models	16	16	18	14	5	
MMM (Tg)	2.0	1.05	1.48	1.87	1.94	
δ (%)	25.0	26.4	34.6	38.8	28.0	
observation	No	No	No	AC, AS, AE, and AOD from Ground station and AOD from MODIS	DMS, SO ₂ , SO ₄ and MSA from ATom	

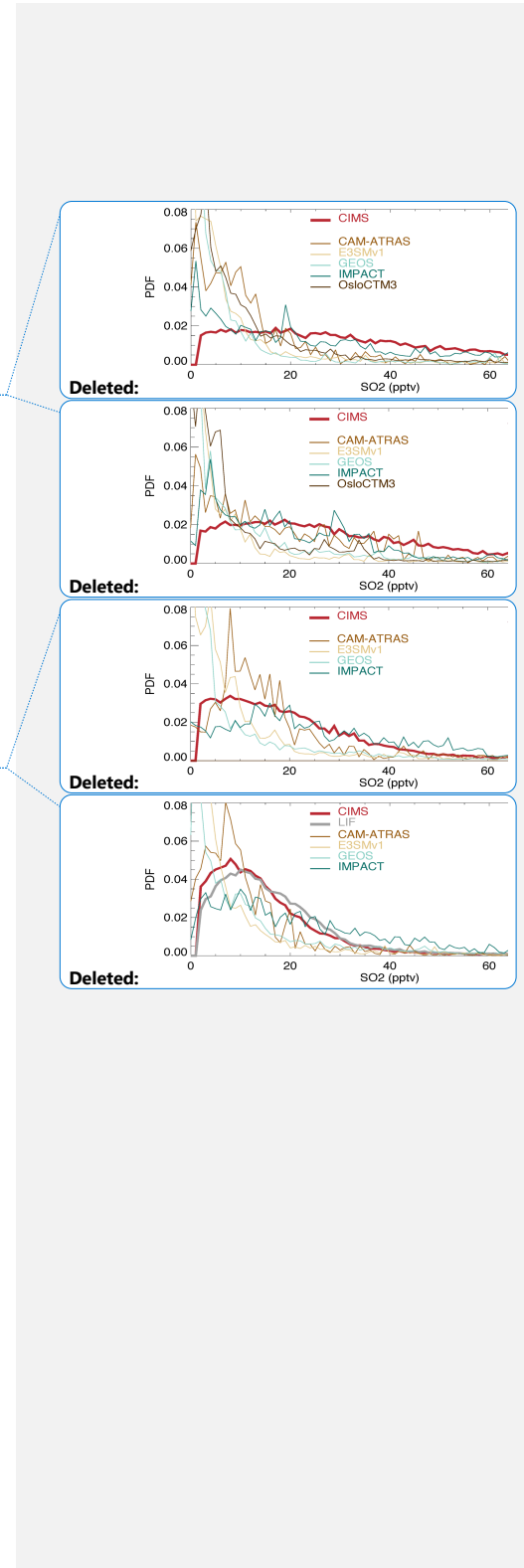
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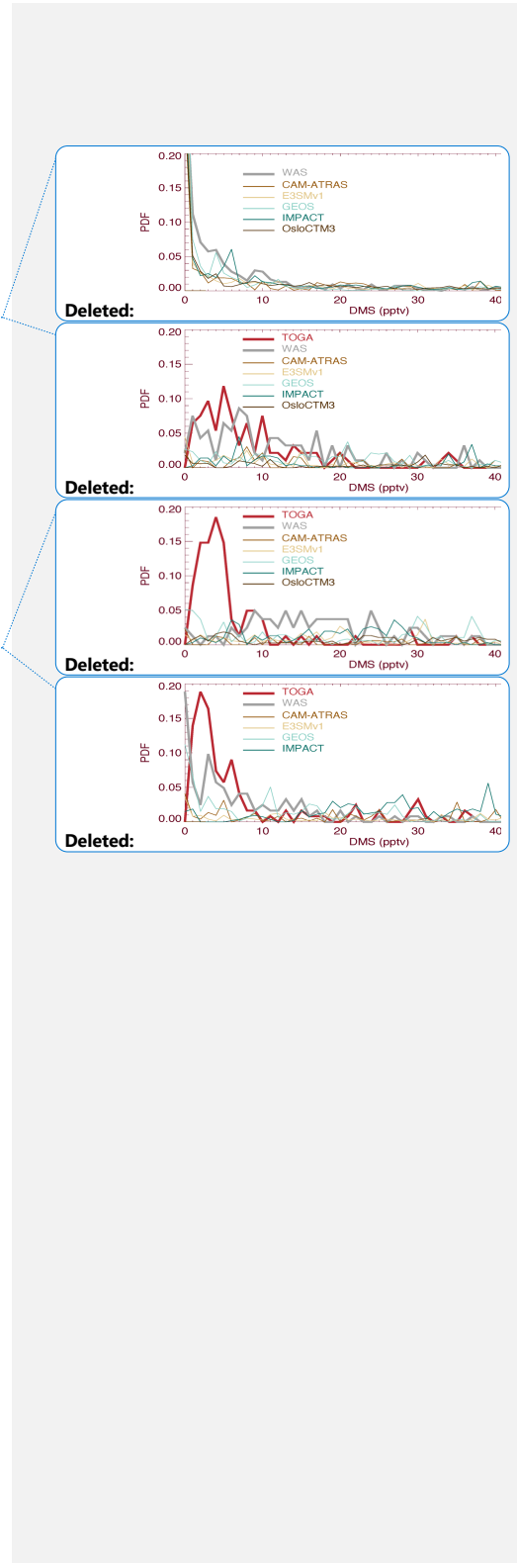
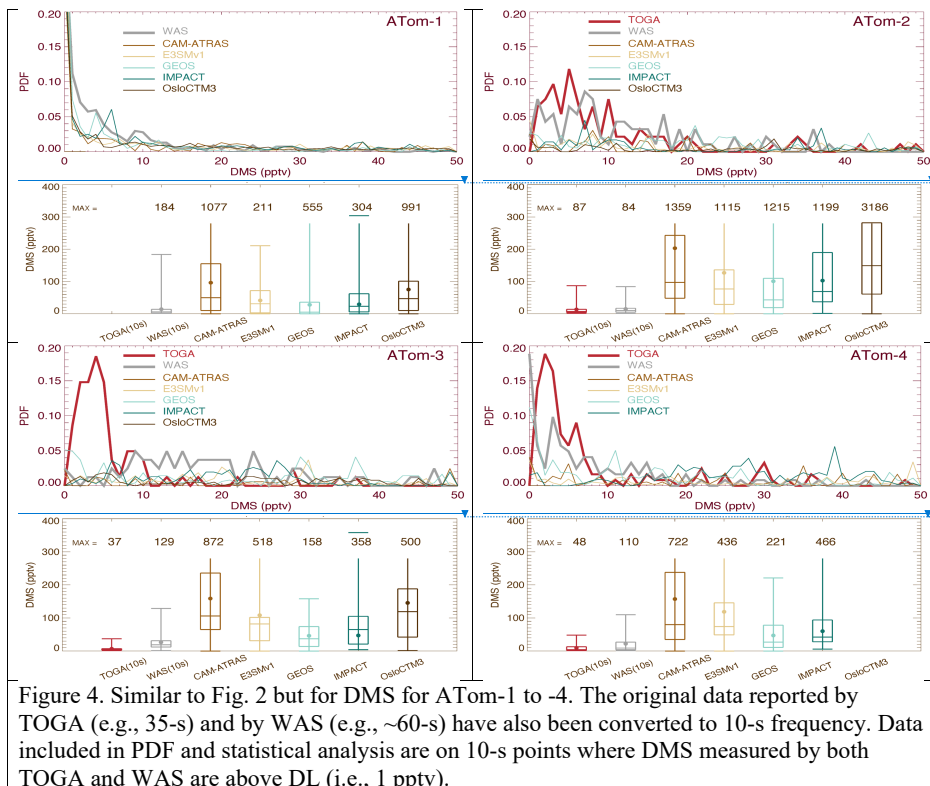


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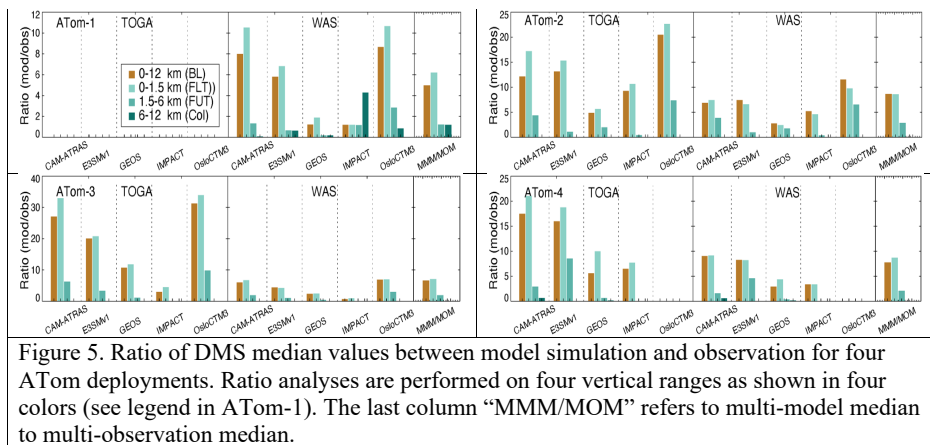


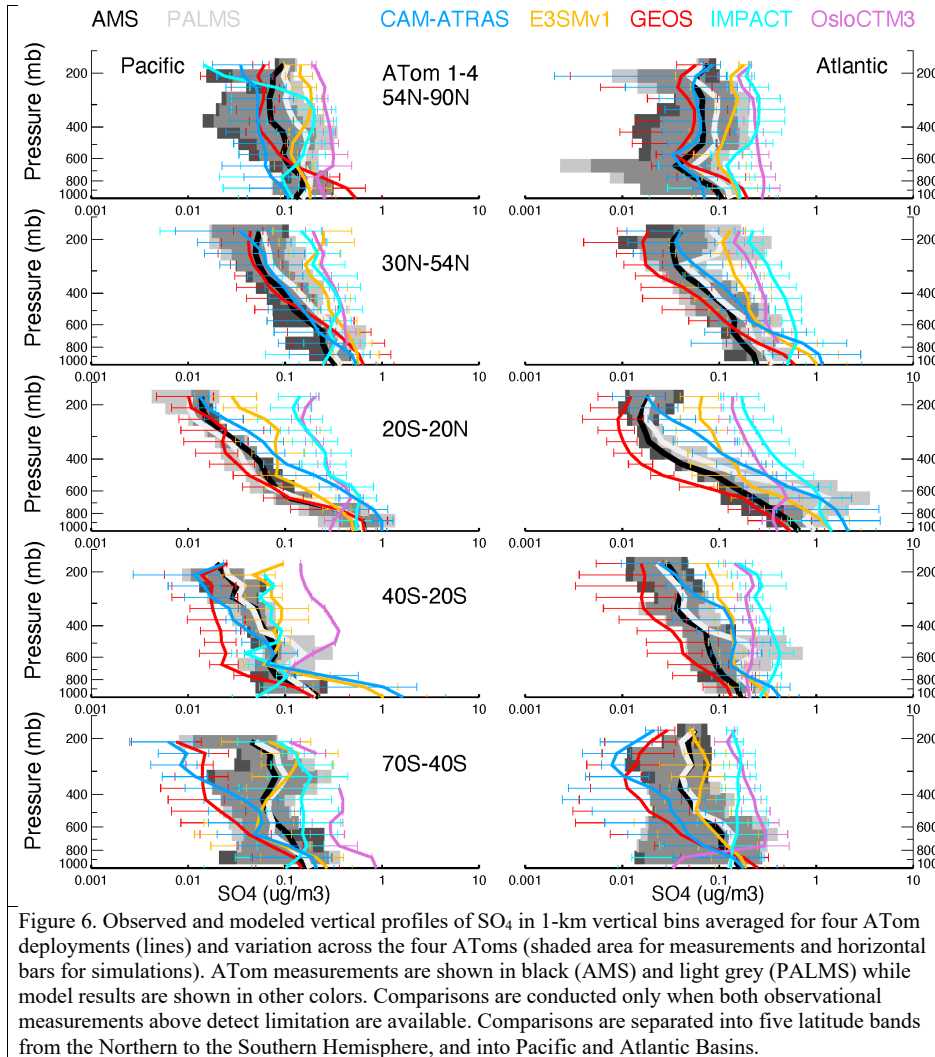
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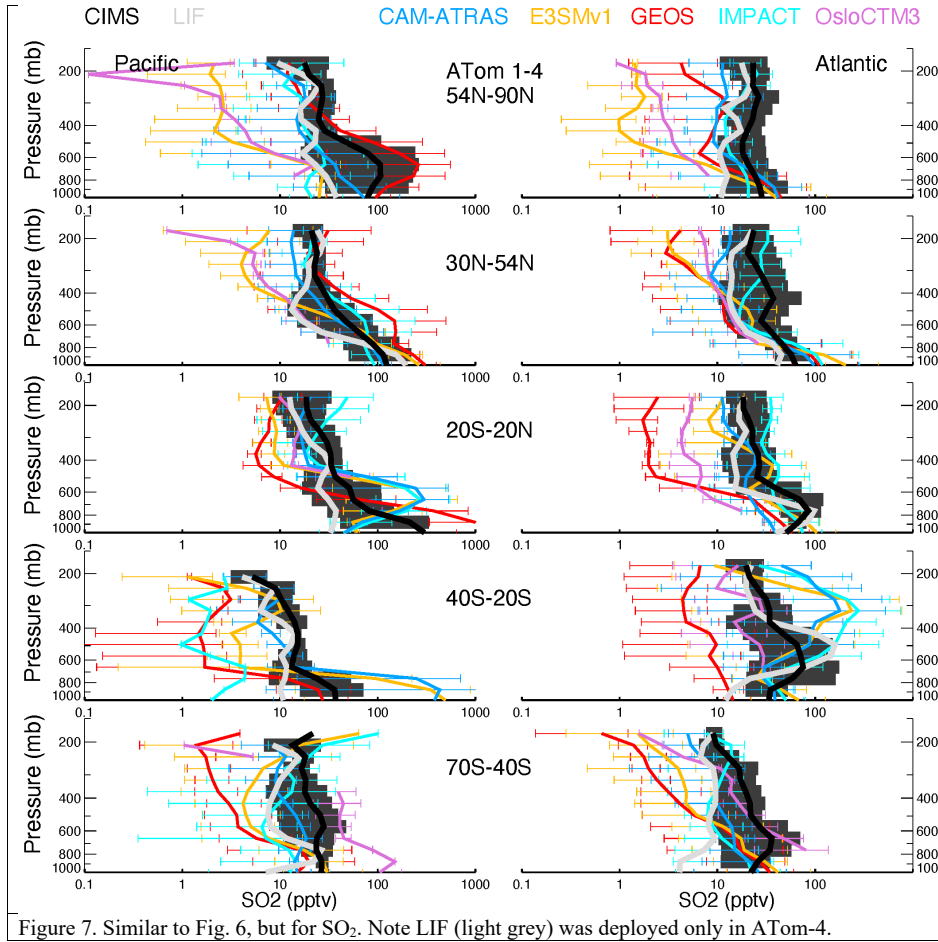


Figure 7. Similar to Fig. 6, but for SO₂. Note LIF (light grey) was deployed only in ATom-4.

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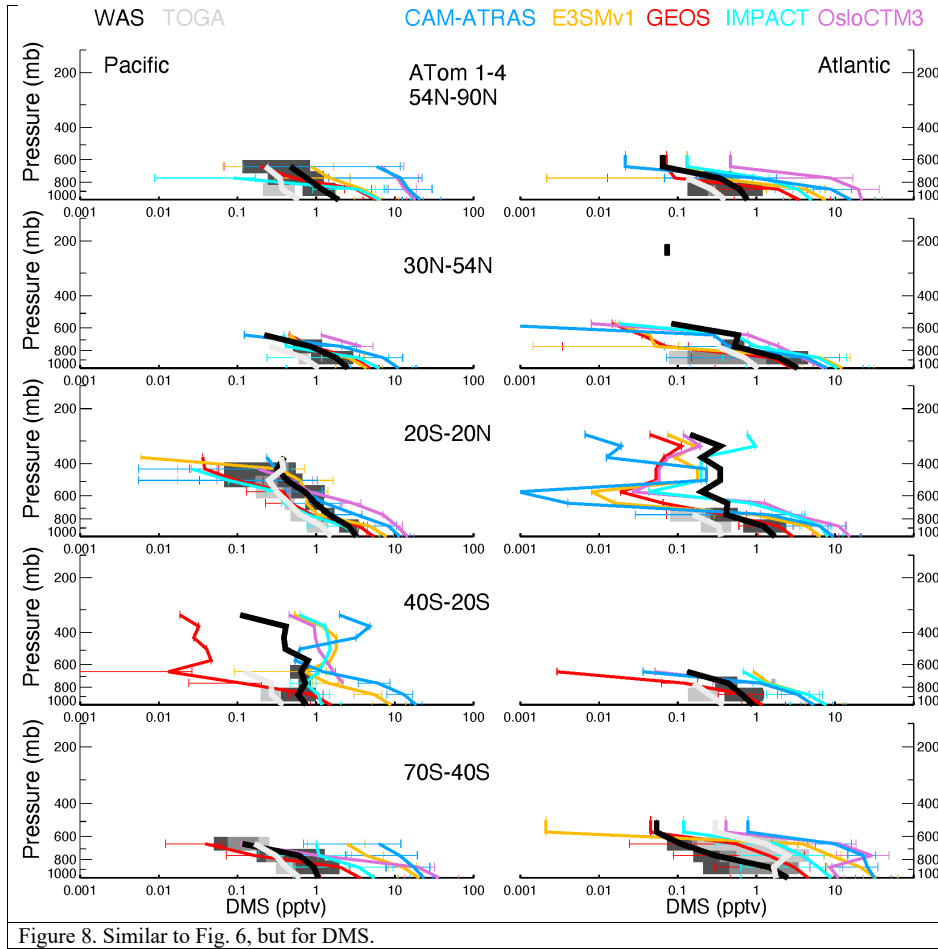
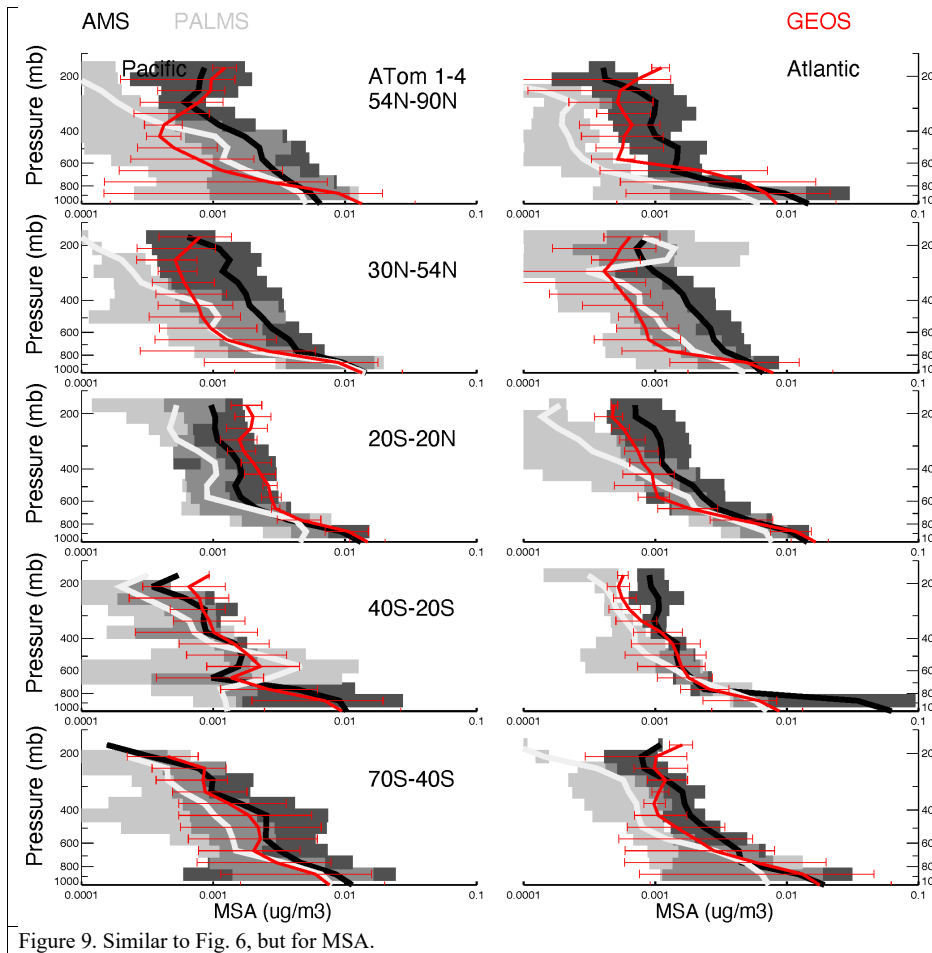


Figure 8. Similar to Fig. 6, but for DMS.

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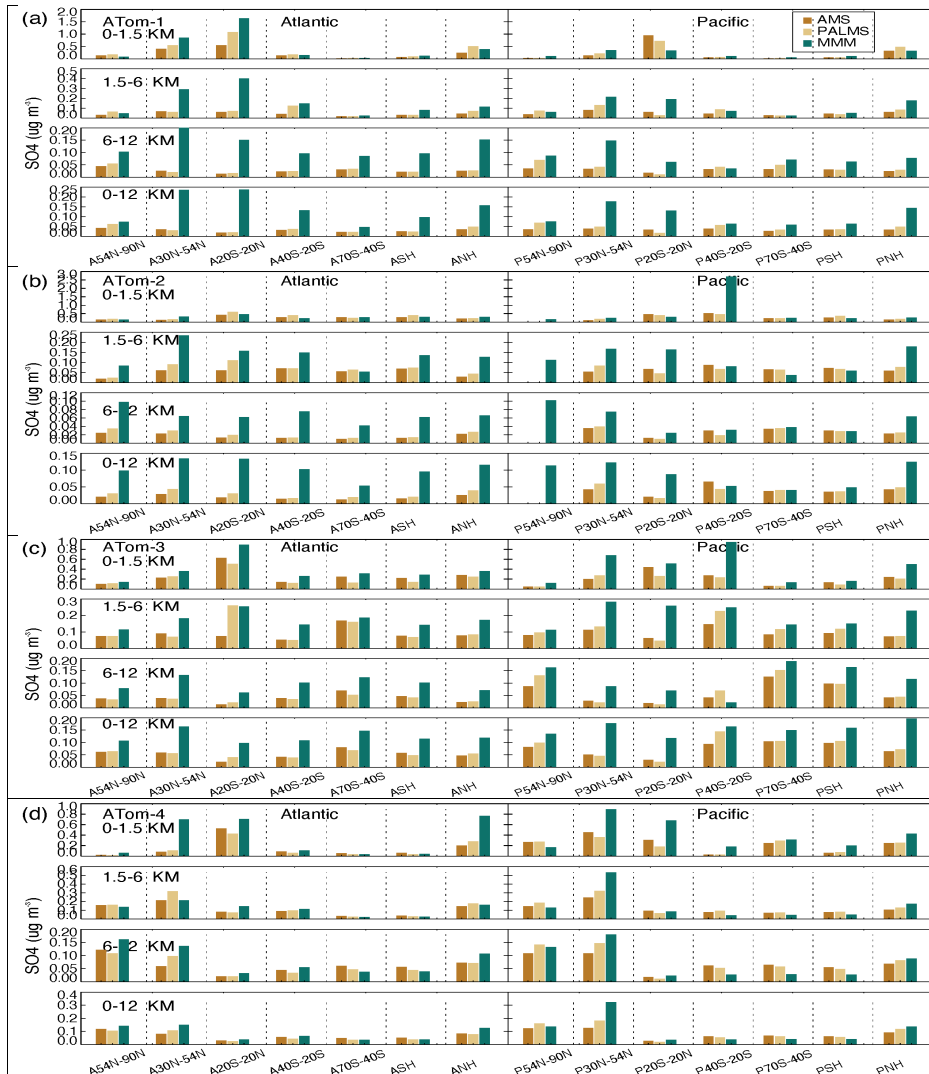


Figure 10. Median SO₄ concentrations from two measurements (AMS orange and PALMS yellow) and multi-model simulation (green) at seven latitudinal bands (including SH and NH) and four vertical layers (i.e., 0-1.5 km, 1.5-6 km, 6-12km, and 0-12 km) over Atlantic and Pacific oceans for four ATom deployments (a-d).

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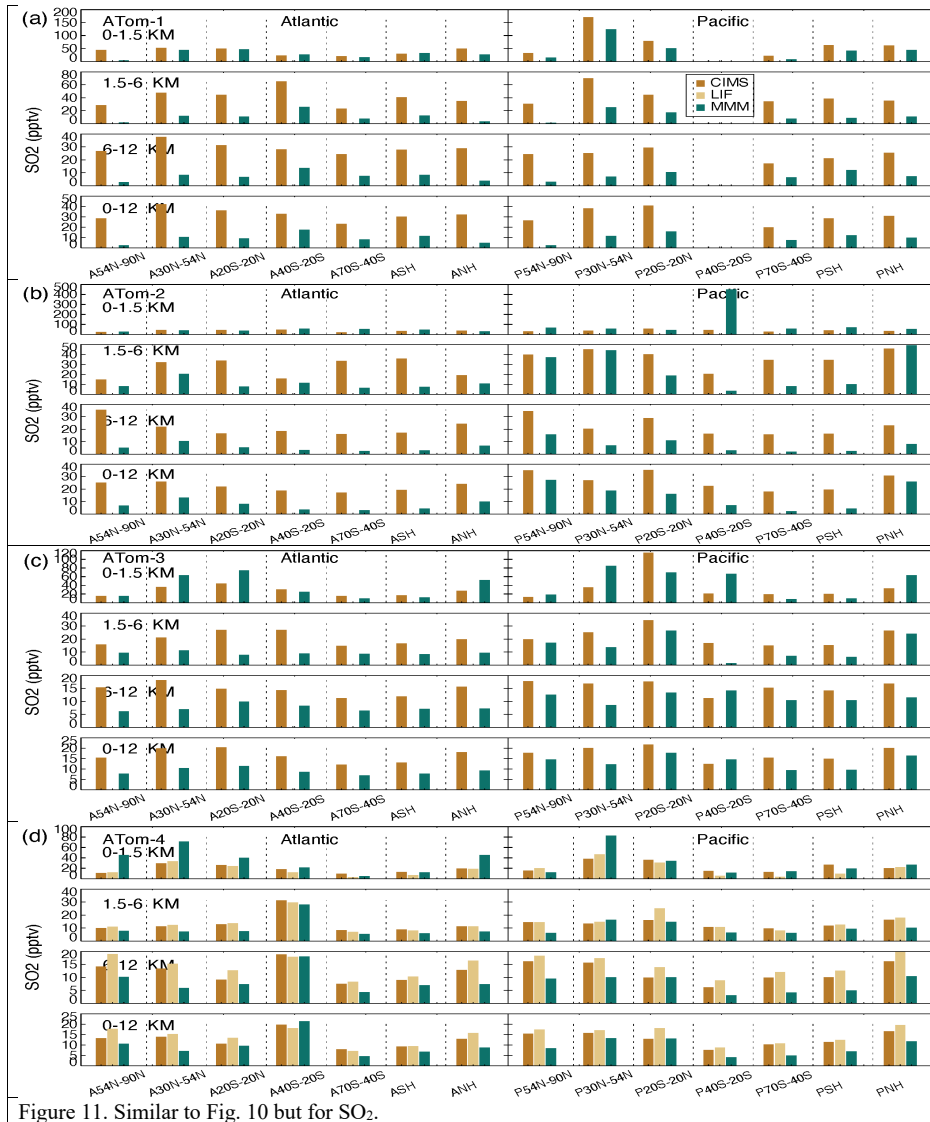


Figure 11. Similar to Fig. 10 but for SO₂.

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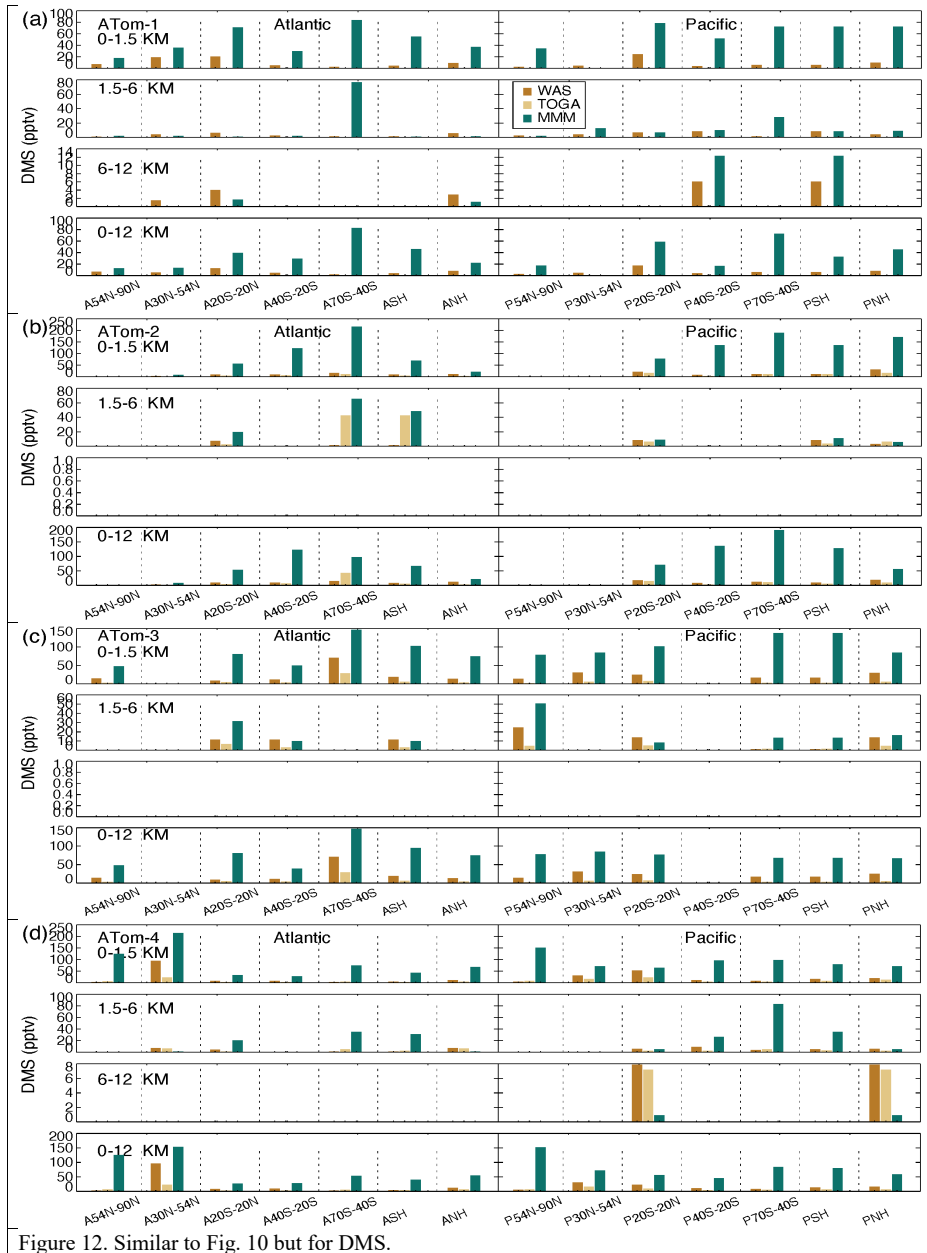


Figure 12. Similar to Fig. 10 but for DMS.

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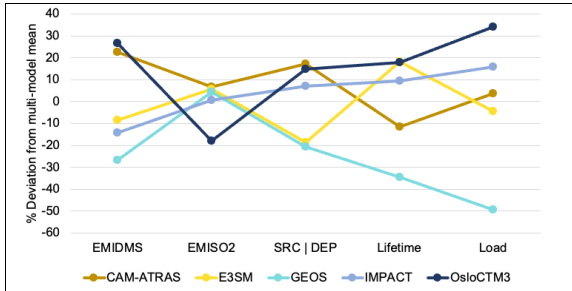


Figure 13. Deviation from multi-model mean for key budget items in sulfur study include DMS emission, SO₂ emission, sulfate source or total deposition, sulfate lifetime, and total sulfate atmospheric mass load.

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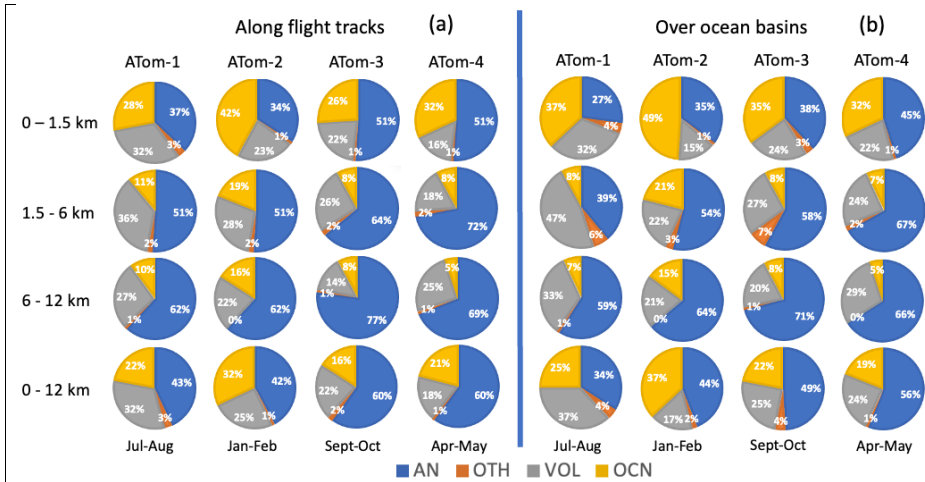


Figure 14. Source origins in percentage (%) for aerosol SO₄ along flight tracks (a) and for a wide oceanic area (b) based on the results from GEOS. Source origins are identified as anthropogenic (AN), volcanic (VOL), oceanic (OCN), and other sources (OTH). Ocean basins include shaded region shown in Fig. 1.

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