Supporting Information

Physicochemical and Temporal Characteristics of Individual Atmospheric Aerosol Particles in Urban Seoul during KORUS-AQ Campaign: Insights from Single-Particle Analysis

Hanjin Yoo^{1,2}, Li Wu³, Hong Geng^{4,+}, and Chul-Un Ro^{1,2,+}

¹ Department of Chemistry, Inha University, Incheon, 22201, Republic of Korea
 ² Particle Pollution Management Center, Inha University, Incheon, 21999, Republic of Korea
 ³ School of Earth Science System, Tianjin University, Tianjin, China

⁴ Institute of Environmental Science, Shanxi University, Taiyuan, China

Correspondence to: Chul-Un Ro (curo@inha.ac.kr) and Hong Geng (genghong@sxu.edu.cn)

A summary of the supporting information:

Figure S1

Section A. Classification of individual particle types

Table S1

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Figure S1. Google map of the sampling site. (Map Copyright © Google Earth)

Section A. Classification of individual particle types

The method for classifying the chemical species of individual aerosol particles is as follows: firstly, particles with an atomic fraction of a chemical species above 90% were considered as a single chemical species. Secondly, reacted and internally mixed particles were identified based on all chemical species and morphology. Thirdly, elements with a concentration of less than 1% were disregarded for the classification, except for N and S (0.5%) which indicate the presence of nitrate and sulfate, respectively. More detailed criteria for the classification based on their elemental compositions and morphology can be found in Table S1 (Geng et al., 2011a, 2014). In total, 8004 individual particles from 52 samples collected during 5/23-6/5, 2016 (excluding 5/24 due to rain) were classified into seven major types: genuine and reacted mineral dust; reacted sea-spray aerosols (SSAs) and a mixture of SSA and others; secondary aerosol particles (including secondary organic aerosols (SOAs) and secondary organic and inorganic aerosols (SOIAs)); Fe-rich particles; heavy metal-containing particles; particles from combustion events such as soot, tar balls, fly ash, and char particles; and biogenic and humic-like substances (HULIS) particles. The characteristics and possible sources of each particle type can be found in Table S1.

Particle types		Characteristic morphologies and chemical compositions	Possible sources
Genuine mineral dust		Irregular-shaped and bright on their secondary electron image (SEI), including	Soil, mining, and construction sites
		aluminosilicates, SiO ₂ , CaCO ₃ , CaMgCO ₃ , TiO ₂ , etc.	
Reacted mineral dust		Irregular-shaped particles often surrounded by liquid droplet shade	Reactions of mineral dust with airborne NO_x/SO_2 and/or
		Mineral dust particles with N or S.	(NH4)2SO4/NH4NO3
Reacted sea-spray aerosols (SSAs)		Liquid droplets or/with irregular shape solids containing N or S along with Na, Mg,	Reactions of SSAs with NOx/SO2 and/or (NH4)2SO4/
		and Cl.	NH4NO3
	Secondary organic	Dark liquid droplets, in which the sum of C and O is more than 90% and the contents o	Accumulation and condensation of semi-volatile organic
Secondary	aerosols (SOAs)	f C and O are comparable.	compounds.
aerosols	Secondary organic	Liquid droplets or solid particles in bright angular shape,	Mixing of (NH ₄) ₂ SO ₄ /NH ₄ NO ₃ with organic carbons
	and inorganic	mostly containing C, O, S and sometimes with N	
	aerosols (SOIAs)		
Fe-rich particles		Bright irregular particles with Fe content of more than 20%	Metallurgical industries, mining, etc.
Heavy metal-containing particles		Particles containing heavy metal elements, i.e., Zn, Ba, Cu, Mn, Pb, Co, V, etc.	Vehicle emissions, tire and brake pad, metallurgical
			industries, etc.
Particles	Soot aggregates	Fractal-like structure on SEI, with more than 90% of C and O	Soot: Combustions
from	Tar balls	Bright and round spherules with high contents of C and O	Tar ball: smoldering combustion, such as biomass burning
combustion	Fly ash	Bright and round spherules with high contents of Al, Si, and O	Fly ash: Thermal power and industrial plants
events	Char particles	Bright and irregular morphology with more than 90% of C and O (C is 3 times higher	Char particles: coal combustions
		than O in atomic concentration)	
Biogenic	Humic-like	Mainly containing comparable C and O, sometimes with S. Bright and irregular on SEI	from soil humic organics
and HULIS	substances (HULIS)		
	Biogenic	Unique morphology containing typically N and/or P	from ocean or forest emission or plant debris

 Table S1. Characteristic morphologies and chemical compositions and possible sources of individual particle types

Figure S2. Morphology, X-ray spectra, and elemental compositions of biogenic particles: (a) fungal spore, (b) microorganism, and (c) trichome or leaf fragment.

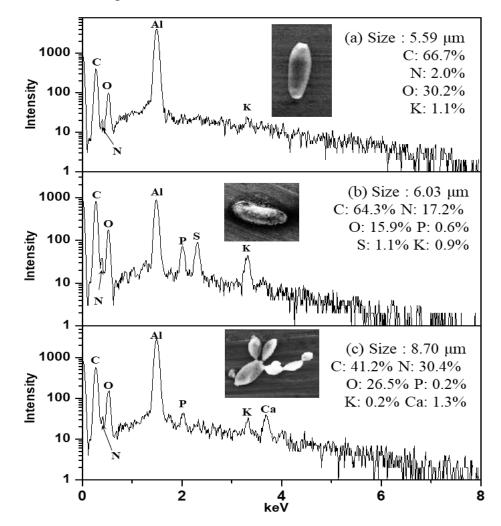


Figure S3. Hourly values of PM₁₀ and PM_{2.5} concentrations recorded in Olympic Park, Seoul, during 5/23-6/5, 2016. The blue dots are the sampling times for single-particle EPMA analysis. Black line: PM₁₀; red line: PM_{2.5}.

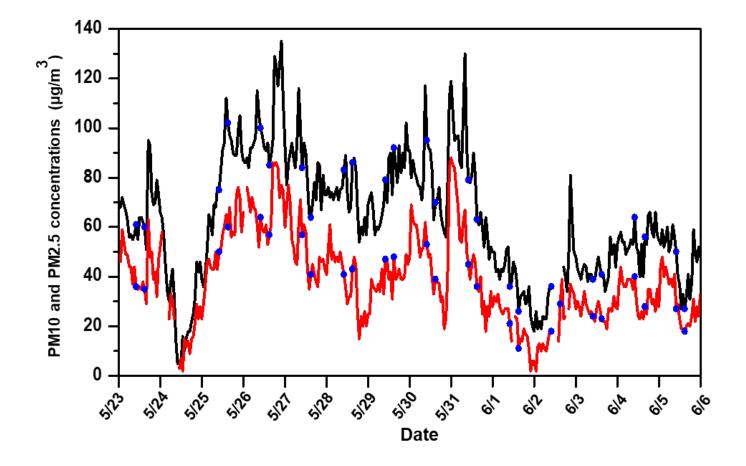


Figure S4. Typical 72-h backward trajectories at 3 different receptor heights (250, 500, and 1000 m above ground level) for (a) 5/23, (b) 5/25, (c) 5/29, (d) 5/30, (e) 6/1, and (f) 6/4.

