

1 **Role of atmospheric aerosols in severe winter fog over Indo Gangetic Plains of India: a**  
2 **case study**

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10  
11 **Abstract**

12 Winter fog and severe aerosol loading in the boundary layer over North India, particularly in the Indo-  
13 Gangetic Plain (IGP), disrupt daily life of millions of people in the region. To understand better the  
14 role of aerosol-radiation feedback on the occurrence, spatial extent, and persistence of winter fog; and  
15 the associated aqueous chemistry in fog in the IGP, several model simulations have been performed  
16 using the Weather Research and Forecasting model coupled with chemistry (WRF-Chem). While  
17 WRF-Chem was able to represent the fog formation for the December 23-24, 2017 fog event over the  
18 central IGP in comparison to station and satellite observations, the model underestimated PM<sub>2.5</sub>  
19 concentrations compared to the Central Pollution Control Board of India monitoring network. While  
20 evaluating aerosol composition for fog events in IGP, we found that the WRF-Chem aerosol  
21 composition was quite different from measurements obtained during the Winter Fog Experiment in  
22 Delhi, with secondary aerosols, particularly chloride aerosol fraction being strongly underpredicted  
23 (~66.6%). Missing emission sources (e.g., industry and residential burning of cow dung and trash) and  
24 aerosol and chemistry processes need to be investigated to improve model-observation agreement. By  
25 investigating a fog event on December 23-24, 2017 over central IGP, we found that the aerosol-  
26 radiation feedback weakens turbulence, lowers the boundary layer height, and increases PM<sub>2.5</sub>  
27 concentrations and RH within the boundary layer. Factors affecting the feedback include loss of  
28 aerosols through deposition of cloud droplets and internal mixing of absorbing and scattering aerosols.  
29 Aqueous-phase chemistry increases the PM<sub>2.5</sub> concentrations, which subsequently affects the aerosol-  
30 radiation feedback by both increased mass concentrations and aerosol sizes. With aerosol-radiation  
31 interaction and aqueous phase chemistry, fog formation began 1-2 hours earlier and caused a longer  
32 fog duration than when these processes were not included in the WRF-Chem simulation. The increase  
33 in RH in both the experiments is found to be important for fog formation as it promoted the growth of  
34 aerosol size through water uptake, increasing the fog water content over IGP. The results from this

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35 study suggest that the aerosol-radiation feedback and secondary aerosol formation play an important  
36 role in the air quality and the intensity and lifetime of fog over IGP, yet other feedbacks, such as  
37 aerosol-cloud interactions, need to be quantified.

38

## 39 **1 Introduction**

40 The Indo-Gangetic Plain (IGP; 21°35'-32°28'N latitude. and 73°50'-89°49'E longitude) in the  
41 northern part of the Indian subcontinent is one of the most densely populated and heavily polluted  
42 regions in South Asia. The rapid population and economic growth in the IGP region over the last  
43 decade have increased air pollution over this region. This is evident from the increasing trend in AOD  
44 and NO<sub>2</sub> column concentration over India reported in recent studies (Dey and Di Girolamo, 2011;  
45 Ghude et al., 2013; Krishna Moorthy et al., 2013), which has slowed and reversed only recently  
46 (Sarkar et al., 2019). The high concentration of aerosols along the IGP and their adverse effects on  
47 human health and the environment are increasing (Ghude et al., 2016). Consequently, more than 500  
48 million people living in the IGP breathe air that exceeds the National Ambient Air Quality Standards  
49 (NAAQS), which has reduced the life expectancy of the people (Debnath et al., 2022; Lelieveld et al.,  
50 2015). Lelieveld et al., (2015) estimated a very high number of premature deaths (0.716 million per  
51 year) linked to aerosols (PM<sub>2.5</sub>), thus making Southeast Asia one of the largest regions affected by  
52 premature mortality globally.

53 One of the major environmental concerns in the IGP is the urban air quality during winter,  
54 especially over the mega-cities, e.g., Delhi, located in the north-western part of IGP (Ghude et al.,  
55 2020; Jena et al., 2021; Sengupta et al., 2022). Several urban air pollution hotspots along the IGP  
56 extend from northwest to east with monthly average PM<sub>2.5</sub> greater than 200 μg m<sup>-3</sup> (NAAQS=60μg m<sup>-3</sup>  
57 <sup>3</sup>, 24 hr average) in the winter season (Bharali et al., 2019; Krishna et al., 2019). IGP is dominated  
58 mainly by fine mode particulates, especially over central to eastern IGP, during post-monsoon and  
59 winter (Kumar et al., 2018). Biomass burning (agricultural waste burning, domestic heating, etc.) is an  
60 important contributor to the observed high PM<sub>2.5</sub> loading over IGP during these seasons (Kulkarni et  
61 al., 2020; Pant et al., 2015; Pawar and Sinha, 2022; Sharma et al., 2010; Yadav et al., 2020). Delhi is  
62 affected substantially by the emissions from agricultural waste burning in the north-western states of  
63 Punjab and Haryana during the post-monsoon (October-November) season (Badarinath et al., 2009;  
64 Jethva et al., 2018; Kumar et al., 2021). Studies showed that PM<sub>2.5</sub> increased from ~50 μg m<sup>-3</sup> to as  
65 high as 300 μg m<sup>-3</sup> (Ojha et al., 2020), and AOD reached 0.98 with the presence of absorbing aerosols  
66 (Singh et al., 2018) during the peak biomass burning in post-monsoon.

67 IGP experiences fog (both radiation and advection fog) every winter after the passage of the  
68 synoptic wind system called the “Western Disturbances”. The majority of fog events in the IGP during  
69 December-January are radiation fog (Deshpande et al., 2023; Ghude et al., 2023), formed due to  
70 radiative cooling of the surface. The number of low visibility days due to haze/fog formation has been

71 increasing significantly (Ghude et al., 2017; Jenamani, 2007; Singh and Dey, 2012), impacting socio-  
72 economic activities, e.g., aviation (Kulkarni et al., 2019). The increase in the intensity and regional  
73 extent of fog over IGP is consistent with the increasing trend in aerosol concentration due to  
74 increasing anthropogenic emissions (Sarkar et al., 2006; Syed et al., 2012).

75 Several factors control the formation and persistence of fog in the IGP, e.g., stable boundary  
76 layer, low temperature, availability of moisture (supplied by the Western Disturbances and irrigation  
77 activities), and the aerosol number and composition (Acharja et al., 2022; Dhangar et al., 2021). It has  
78 also been suggested that the atmospheric rivers (moisture incursion from Arabian Sea) act as a source  
79 of water vapor over IGP, which fuels the intensification of fog and haze (Verma et al., 2022) during  
80 winter. The high aerosol concentration in the boundary layer influences fog formation (Gautam et al.,  
81 2007; Safai et al., 2019) over the IGP by providing the needed cloud condensation nuclei (CCN) for  
82 activation into fog droplets. In addition, aerosols induce surface cooling by reducing solar radiation at  
83 the surface while warming the lower troposphere by absorption (Ding et al., 2016; Yu et al., 2002). A  
84 reduction in surface-reaching solar radiation by ~19% has been reported during winter over Kanpur in  
85 the IGP (Dey and Tripathi, 2007). The reduced solar flux affects the boundary layer stability and  
86 depth by suppressing the thermals and thus further increasing the surface aerosol concentration via  
87 aerosol-radiation feedback, which is very strong over the IGP (Bharali et al., 2019). Kumar et al.,  
88 (2020) have shown that aerosol-radiation feedback significantly improves the accuracy of PM<sub>2.5</sub> and  
89 temperature forecasts in Delhi. Srivastava et al., (2018) reported that the direct aerosol forcing over  
90 polluted regions is very large with values up to  $-80.0 \pm 7.2 \text{ W m}^{-2}$  over the IGP in the winter season.

91 Aerosol-radiation interaction determines that the aerosol distribution is critical for the evolution  
92 of fog (Bodaballa et al., 2022; Steeneveld et al., 2015), while microphysics is important for fog  
93 formation and dispersal (Boutle et al., 2018; Maalick et al., 2016). Although the relationship between  
94 the aerosol chemical composition and aerosol activation to CCN has not been fully understood yet,  
95 studies have found that the chemical composition and mixing state of aerosols affect the  
96 hygroscopicity ( $\kappa$ ) of aerosols (Bodaballa et al., 2022; Ma et al., 2013; Moore et al., 2012; Zhang et  
97 al., 2014a). Fog processes involve a complex interplay between local meteorology, radiation,  
98 microphysics, and aerosol chemistry, making it difficult to understand the fog lifecycle (Acharja et al.,  
99 2022; Maalick et al., 2016; Zhang et al., 2014b). There is considerable heterogeneity in the spatial and  
100 temporal aerosol properties over IGP and the poor estimates of their mixing state. Therefore,  
101 prediction of fog by weather models is still challenging with biases in fog's onset and dispersal  
102 timings.

103 Previous studies have focussed on the impacts of meteorological conditions, topography, or  
104 anthropogenic emissions on the poor air quality and intensification of fog during winter over IGP (e.g.  
105 Hakkim et al., 2019). However, studies on the effect of feedback induced by the aerosols on the  
106 meteorological conditions and thus on aerosol concentration are very limited over this region, except  
107 for a few above-mentioned studies which discuss how the aerosol-radiation feedback favors haze and

108 fog during winter. Moreover, fog can provide a medium for aqueous-phase reactions. While several  
109 earlier studies have reported an increase in secondary aerosols during fog over IGP, a sensitivity study  
110 examining the impact of fog on aqueous phase chemistry has not yet been done over IGP.

111 In the present work, we aim to find the suitable chemistry/physics as well as the meteorology  
112 initial/boundary conditions that lead to improved simulations of fog events in the Weather Research  
113 and Forecasting model coupled with chemistry (WRF-Chem; (Fast et al., 2006; Grell et al., 2005;  
114 Powers et al., 2017). We also explore the role of aerosol-radiation feedback on fog properties as the  
115 high aerosol loadings in northern India can impact the heating rates, temperature inversions, and  
116 boundary layer height. The role of aqueous chemistry on fog properties and vice-versa is also  
117 investigated.

118

## 119 2 Methodology

120 Fog formed due to radiative cooling, based on the onset time of fog and low wind speeds (Deshpande  
121 et al., 2023 and references therein), at the surface on both 23<sup>rd</sup> and 24<sup>th</sup> December 2017 over a  
122 widespread region of the IGP (Fig. 1a, b). The fog region is located over an area with high PM<sub>2.5</sub>  
123 anthropogenic emissions (Fig. 1c). The IGP is a large region with varying meteorology and aerosol  
124 characteristics, therefore, it is divided into three areas, northwest (NWIGP: latitude-longitude range,  
125 27°N-32°N, 75°E-79°E), central (CIGP: latitude-longitude range, 25°N-28°N, 79°E-83°E), and east  
126 (EIGP: latitude-longitude range, 24°N-27°N, 83°E-87°E) which are marked by the black rectangles in  
127 Fig. 1c. Although biomass burning and anthropogenic emissions dominate throughout the IGP during  
128 post-monsoon and winter season, the north-westerly wind system results in the gradient distribution of  
129 AOD over this region. The downwind regions, CIGP and EIGP are influenced by the long-range  
130 transport from the NWIGP, resulting in high AOD with dominant fine particulates over CIGP and  
131 EIGP, especially during post-monsoon and winter (Kedia et al., 2014; Kumar et al., 2018). Therefore,  
132 representative stations from each region listed in section 2.2 are considered for the sensitivity  
133 analyses.

134

### 135 2.1 Modeling

136 The WRF-Chem model version 4.0.3 is used for this study. Earlier studies have successfully  
137 used WRF-Chem to predict fog (Pithani et al., 2019) and in the study of aerosol-radiation feedback on  
138 air quality (Kumar et al., 2020; Bharali et al., 2019) and fog (Shao et al., 2023). The model domain is  
139 centered at Delhi (77.1°E, 28.7°N) with 300 grid points in the east-west, 170 grid points in the south-  
140 north direction (Fig. 1c), and 50 vertical eta levels with the model top at 50 hPa. The horizontal grid  
141 spacing of the domain is 10 km, while the vertical grid spacing varies from higher resolution (~200 m)  
142 in the boundary layer to coarser resolution (~1200 m) near the model top. We conduct three model  
143 configurations (Table 1) for December 20-24, 2017 to identify the best configuration for  
144 meteorological simulations. The three experiments have been designed with different combinations of

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145 meteorological initial/lateral boundary conditions and planetary boundary layer (PBL) physics.  
146 Experiment 1 (EXP1) uses the National Centers for Environmental Predictions (NCEP) Final Analysis  
147 (GFS-FNL; 1° x 1°, 6 hourly) meteorology data for initial and boundary conditions and the YSU  
148 (Yonsei University; (Hong et al., 2006) PBL scheme. Experiments 2 and 3 (EXP2, EXP3) use ERA-  
149 Interim Project (1.125° x 0.703°, 6 hourly) for meteorology initial and boundary conditions. EXP2  
150 uses the YSU PBL scheme while EXP3 uses the ACM2 (Asymmetric Convective Model version 2)  
151 PBL scheme. ACM2, is a hybrid of the original nonlocal closure (Pleim and Chang, 1992) and a local  
152 closure eddy diffusion scheme (Pleim, 2007a, 2007b). The YSU PBL option was coupled with the  
153 Noah LSM while ACM2 was coupled with Pleim-Xiu LSM. While YSU permits investigations of  
154 both aerosol-radiation (AR) and aerosol-cloud interactions (ACI), ACI is not possible when using the  
155 ACM2 PBL scheme because in WRF the ACM2 PBL scheme does not provide the exchange  
156 coefficient for heat, which is required to calculate the maximum supersaturations and therefore the  
157 activation fraction for aerosol mass and number for each bin/mode, which is based on the Abdul-  
158 Razzak and Ghan, (2002) scheme. However, ACM2 has been shown to perform well for air quality in  
159 the polluted regions (Mohan and Gupta, 2018; Gunwani and Mohan, 2017; Xie et al., 2012, Mohan  
160 and Bhati, 2011). Mohan and Gupta (2018) tested the YSU and ACM2 schemes during the summer  
161 time (1-15 June 2010) and focused on the evaluation of temperature, wind speed, PBL height, ozone,  
162 and PM<sub>10</sub>. Although these studies recommend using the nonlocal ACM2 PBL scheme for air quality  
163 prediction for IGP, there are still seasonal, day-night, and regional biases in the PBL schemes.  
164 Gunwani and Mohan (2017) showed that ACM2, QNSE (Quasi Normal Scale Elimination), and MYJ  
165 (Mellor-Yamada-Janjić) schemes work well in predicting temperature, humidity, and wind speed in  
166 different regions of India. Mohan and Bhati (2011) found that using the ACM2 PBL scheme with  
167 Pleim Xiu surface physics improved wintertime meteorology estimates in Delhi indicating its potential  
168 in fog predictions, whereas Pithani et al., (2019) recommend using the local PBL scheme MYNN2.5  
169 (Mellor-Yamada-Nakanishi-Niino level 2.5). Shin and Hong (2011) found that a non-local (e.g.,  
170 ACM2, YSU) scheme is favorable in unstable conditions and a local scheme (e.g., MYJ, Boulac) in  
171 stable conditions. All these studies suggest the need for careful consideration of the above-mentioned  
172 biases while selecting a PBL scheme. Therefore, to ensure that WRF captures all the relevant  
173 meteorological parameters including relative humidity reasonably well during fog events in winter, we  
174 designed EXP1, EXP2 and EXP3.

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175 The advantage of Pleim-Xiu LSM (PX-LSM) is that it allows nudging of soil moisture and  
176 temperature to improve the prediction of meteorology near the surface (Pleim and Gilliam, 2009;  
177 Pleim and Xiu, 2003; Xiu and Pleim, 2001) which Noah LSM does not include. The PX-LSM  
178 includes two-layer soil (0–1 and 1–100 cm) model, canopy moisture, and aerodynamic and stomatal  
179 resistance. Ground surface (1 cm) temperature is calculated from the surface energy balance using a  
180 force-restore algorithm for heat exchange within the soil. Although the two-layer approach in PX-  
181 LSM is less detailed than the multilayer soil models such as the Noah LSM (four soil layers; Chen and

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182 Dudhia 2001), it performs well with realistic initialization for soil moisture and through dynamic  
183 adjustment in the model simulation where soil moisture is indirectly nudged according to differences  
184 in 2-m temperature (T2) and 2-m relative humidity (RH) between the model and observation (Pleim  
185 and Xiu, 2003). Soil moisture nudging adjusts the surface evaporation (direct soil surface evaporation,  
186 vegetative evapotranspiration, and evaporation from wet canopies) which then affects the partitioning  
187 of available surface energy into latent and sensible heat flux and thus reduces errors in T2 and 2-m  
188 RH.

189 For EXP2, meteorological initial conditions were refreshed every 24 hours, while EXP3 was a  
190 continuous run but soil moisture was nudged to the Era-Interim dataset to improve the prediction of  
191 surface fluxes. All other physics and chemistry options are the same for all the experiments except the  
192 surface physics option, which changes with the PBL scheme used. The deposition of cloud droplets is  
193 an important moisture and aerosol sink during fog events. For all these simulations, the deposition  
194 velocity of cloud droplets was reduced to  $0.01 \text{ m s}^{-1}$  based on Stoke's Law and previous studies  
195 (Katata et al., 2015; Tav et al., 2018) because its default value ( $0.1 \text{ m s}^{-1}$ ), is large.

196 To examine the radiative effects of aerosols and aqueous phase chemistry additional  
197 simulations have been done using the meteorological configuration in EXP3, with aerosol-radiation  
198 (wFB) feedback plus aqueous chemistry (wAq.chem), without aerosol-radiation feedback (nFB) but  
199 with aqueous chemistry, and without aqueous chemistry (noAq.chem) but with aerosol-radiation  
200 feedback. The analysis has been done for the fog events on 23<sup>rd</sup> and 24<sup>th</sup> December 2017 as

$$201 \quad F_{\text{ARF}} = P(\text{wFB-nFB})$$

$$202 \quad F_{\text{Aq.chem}} = P(\text{wAq.chem-noAq.chem})$$

203 where  $F_{\text{ARF}}$  is the impact due to aerosol radiation feedback on the meteorological parameters/chemical  
204 species (P) and  $F_{\text{Aq.chem}}$  is the impact due to the inclusion of aqueous phase chemistry

205  
206 Emissions used in the WRF-Chem simulations are from the EDGAR-HTAP v2 (*Emissions  
207 Database for Global Atmospheric Research- Hemispheric Transport of Air Pollution;  $0.1^\circ \times 0.1^\circ$* )  
208 inventory for anthropogenic emissions and FINN v2.2 (*Fire INventory from NCAR; 1 km x 1 km*) fire  
209 emission inventory (Wiedinmyer et al., 2011). Trash-burning emissions (Chaudhary et al., 2021) are  
210 also included in the simulations. The model calculates the biogenic emissions online using MEGAN  
211 v2.04 (*Model of Emissions of Gases and Aerosols from Nature*) (Guenther et al., 2006). The initial and  
212 lateral boundary conditions for chemical constituents are from the global chemistry transport model  
213 CAM-Chem (*Community Atmosphere Model with Chemistry*) (Emmons et al., 2020).

214 The MOZART (Model for Ozone and Related chemical Tracers) chemical mechanism  
215 (Emmons et al., 2010) is used for gas-phase chemistry, which includes 85 gas-phase species, 39  
216 photolysis, and 157 gas-phase reactions. It has been updated to include an explicit treatment of  
217 aromatic compounds, HONO,  $\text{C}_2\text{H}_2$ , and isoprene oxidation scheme (Knote et al., 2014). The lumped  
218 toluene used by Emmons et al., (2010) has been speciated into benzene, toluene, and lumped isomers

219 of xylenes (Knote et al., 2014). For this study, HCl emissions, transport, dry, and wet deposition are  
220 represented. However, HCl gas-phase reaction is not included in MOZART.

221 The Model for Simulating Aerosol Interactions and Chemistry (MOSAIC) with four size bins  
222 (0.039–0.156, 0.156–0.625, 0.625–2.500, and 2.5–10.0  $\mu\text{m}$  dry diameters) coupled with MOZART  
223 gas-phase chemistry is used (Fast et al., 2006; Zaveri et al., 2008). The bin sizes are defined by their  
224 lower and upper dry particle diameters, so there is no transfer of particles between bins during water  
225 uptake or loss. It is assumed that aerosols in each bin are internally mixed with the same chemical  
226 composition while they are externally mixed in different bins.

227 The aerosol composition includes sulfate ( $\text{SO}_4^{2-}$ ), ammonium ( $\text{NH}_4^+$ ), nitrate ( $\text{NO}_3^-$ ), aerosol  
228 water, sea salt ( $\text{Na}^+$ ,  $\text{Cl}^-$ ), methanesulfonate ( $\text{CH}_3\text{SO}_3^-$ ), carbonate ( $\text{CO}_3^{2-}$ ), calcium ( $\text{Ca}^{2+}$ ), black carbon  
229 (BC), organic mass (OC), and unspecified inorganic species such as silica, inert minerals, and trace  
230 metals lumped together as other inorganic mass (OIN). For OC, primary OC and secondary OC are  
231 represented separately, where the latter is simulated using the volatility basis set (VBS) approach.  
232 Reactive inorganic species such as potassium ( $\text{K}^+$ ) and magnesium ( $\text{Mg}^{2+}$ ) are usually present in much  
233 smaller amounts and are equivalent to  $\text{Na}^+$  since their sulfate, nitrate, and chloride salts are similar in  
234 terms of their solubility in water.

235 MOSAIC treats condensation and evaporation of trace gases to/from particles, nucleation  
236 (new particle formation), and coagulation. Aerosol coagulation (Brownian) is based on (Jacobson et  
237 al., 1994) and nucleation is based on (Wexler et al., 1994) parameterization of  $\text{H}_2\text{SO}_4\text{-H}_2\text{O}$   
238 homogeneous nucleation. Sulfate, nitrate, chloride, and ammonium aerosols are mainly formed  
239 through oxidation and neutralization/condensation of gas precursors. Gas-phase sulfuric acid ( $\text{H}_2\text{SO}_4$ )  
240 is produced by the gas-phase oxidation of  $\text{SO}_2$  by OH and nitric acid ( $\text{HNO}_3$ ) formation is via the  
241 oxidation of  $\text{NO}_2$  by OH. HCl is a primary emission product. The neutralization/condensation of  
242  $\text{H}_2\text{SO}_4$ , HCl, and  $\text{HNO}_3$  with  $\text{NH}_3$  produces ammonium such as ammonium sulfate ( $(\text{NH}_4)_2\text{SO}_4$ ),  
243 ammonium bisulfate ( $\text{NH}_4\text{HSO}_4$ ), ammonium chloride ( $\text{NH}_4\text{Cl}$ ) and ammonium nitrate ( $\text{NH}_4\text{NO}_3$ ),  
244 respectively. The thermodynamic modules in MOSAIC for the dynamic gas-particle partitioning of  
245 aerosols MTEM (Multicomponent Taylor Expansion Method) and MESA (Multicomponent  
246 Equilibrium Solver for Aerosols) calculate the activity coefficient in aqueous phase aerosols and  
247 compute the intraparticle solid-liquid phase equilibrium respectively (Zaveri et al., 2005, 2008). The  
248 Adaptive Step Time-split Euler Method (ASTEM) coupled with MESA-MTEM dynamically  
249 integrates the mass transfer equations.

250 Aqueous-phase chemistry uses a bulk water approach employing the Fahey and Pandis (2001)  
251 mechanism. It calculates sulfate formation, formaldehyde oxidation, and non-reactive uptake of nitric  
252 acid, hydrochloric acid, ammonia, and other trace gases (Chapman et al., 2009; Pye et al., 2020).  
253 Aqueous-phase sulfate is produced via oxidation of  $\text{SO}_2$  by  $\text{H}_2\text{O}_2$ ,  $\text{O}_3$ , TMI (Transition metal Ion:  
254 Fe(III), Mn(II)) catalyzed  $\text{O}_2$  and  $\text{NO}_2$ . TMI concentrations are prescribed in the model to  $0.01 \mu\text{g m}^{-3}$   
255 for Fe(III) and  $0.005 \mu\text{g m}^{-3}$  for Mn(II) (Martin and Good, 1991). The Fe(III) values are within the

256 range of water soluble iron in winter time aerosol reported in India (Kumar and Sarin, 2010). Wet  
257 removal (scavenging), is represented by the (Neu and Prather, 2012) scheme for trace gases and Easter  
258 et al., (2004) for aerosols.

259

## 260 2.2 Observations

261 To evaluate the model output, observations of aerosols and meteorology have been obtained  
262 from several satellites as well as ground-based measurement platforms. To examine the aerosol  
263 loading and spatial and temporal distribution, daily Level 2 Aerosol Optical Depth (AOD) retrievals  
264 from the Moderate Resolution Imaging Spectroradiometer (MODIS) aboard Terra and Aqua satellites  
265 are obtained at the spatial resolution of 10 km x 10 km (at nadir) pixel array. It provides aerosol  
266 properties from the Dark Target (DT) algorithm applied over the ocean and dark land (e.g.,  
267 vegetation) and Deep Blue (DB) algorithms over the entire land areas, including both dark and bright  
268 surfaces. Each MOD04\_L2 (Terra) / MYD04\_L2 (Aqua) products are available at a 5-minute time  
269 interval with an output grid of 135 pixels in width by 203 pixels in length.

270 The Indian National Satellites (INSAT-3D) in the geostationary orbit at inclinations of 82°  
271 longitude provide an imager fog product (3DIMG\_L2C\_FOG) with a spatial resolution of 4 km every  
272 30 min ([www.mosdac.gov.in](http://www.mosdac.gov.in)). For daytime, the visible channel observation is used to detect fog,  
273 whereas thermal infrared is used to reduce false alarms such as medium/high clouds and snow areas.  
274 INSAT 3D's 'day microphysics' data component analyzes solar reflectance at three wavelengths: 0.5  
275 μm (visible), 1.6 μm (shortwave infrared), and 10.8 μm (thermal infrared). Night-time fog is derived  
276 from TIR-1 (12.0 μm and 10.0 μm) and MIR (10.8 μm and 3.9 μm) channel brightness temperature  
277 over the Indian region. INSAT-3D provides fog intensity varying from 1 to 4 indicating SHALLOW  
278 for visibility > 600 m; MODERATE, DENSE, and VERY\_DENSE, respectively for visibility varying  
279 from 0 to 500 m (Banerjee and Padmakumari, 2020). If the visibility is greater than 700 m it indicates  
280 no fog while visibility > 1000 m represents very clear skies. Validation of INSAT-3D fog products  
281 over the IGP shows a 66%-68% probability of detection and a 10% false alarm rate. It also captures  
282 the entire life cycle of fog from formation to dissipation. However, detecting fog during multilayer  
283 clouds is still challenging with INSAT-3D (Arun et al., 2018; Chaurasia and Gohil, 2015; Chaurasia  
284 and Jenamani, 2017).

285 Ground-based monitoring sites provide hourly data of relative humidity, surface temperature,  
286 and wind speed measured by the Central Pollution Control Board, CPCB (<http://cpcb.nic.in>). Given  
287 the data availability from CPCB stations, nine stations have been considered representing each region  
288 of IGP, which include, Amritsar, IGI Airport (Indira Gandhi International Airport, Delhi), IHBAS  
289 (Institute of Human Behaviour and Allied Sciences, Delhi), Dwarka (Delhi), RKP (Ramakrishna  
290 Puram, Delhi) in the North-West IGP; Kanpur, Lucknow in Central IGP and Patna, Muzaffarpur in  
291 East IGP.

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292 In addition, measurements of several aerosols, trace gases, and meteorology at Delhi (IGI  
293 Airport) from the Winter Fog Experiment (WiFEX) for the period December 10-31, 2017, have also  
294 been used to validate the model output. The WiFEX, an initiative of the Ministry of Earth Sciences  
295 (MoES), India, is a ground-based measurement campaign at the IGI Airport Delhi to understand fog's  
296 physical and chemical features. Additional details of the WiFEX project and related publications can  
297 be found in Ghude et al., (2017).

298

### 299 **3 Meteorology Evaluation**

300 Previous studies simulating fog highlight the importance of high model vertical resolution  
301 (Pithani et al., 2019; Van Der Velde et al., 2010) for representing the fog formation and the growth of  
302 the fog layer, model initialization (Yadav et al., 2022), initial relative humidity (Bergot and Guedalia,  
303 1994; Pithani et al., 2020), and PBL schemes (Chen et al., 2020; Pithani et al., 2019). In the present  
304 study, 2-m relative humidity (RH2), 2-m temperature (T2), and 10-m wind speed (WS) from WRF-  
305 Chem have been evaluated using ground-based measurements from CPCB monitoring network and  
306 WIFEX campaign for nine stations across the IGP. The comparison of WRF-Chem results with  
307 observations shows that RH2 and T2 are sensitive to the choice of the meteorological initial and  
308 boundary conditions as illustrated by six stations in major cities (Fig. S1). WRF-Chem compares  
309 better with the observations for simulations driven by the ERA-Interim reanalysis than with GFS-FNL  
310 reanalysis since ERA-Interim provides more realistic RH2 than GFS-FNL (Figs. S2 a-f). For example,  
311 RH2 from EXP1 (GFS) varies from 10 to 50%, while RH2 from EXP2 and EXP3 varies from 30 to  
312 100%, which is closer to observation, especially for NWIGP and CIGP. For EIGP, RH2 from EXP1  
313 (GFS) compares better than ERA-Interim, which overestimates the observed RH2. ERA-Interim and  
314 YSU PBL scheme showed damping of RH2 continuously increasing the bias in RH2 with time (not  
315 shown), which was corrected in EXP2 by refreshing meteorology every day at 00h UT during the  
316 model simulation. In addition, maps of surface RH2 and T2 (Figs. S2 g-j) show that the GFS-FNL  
317 dataset has lower relative humidity throughout the domain as compared to ERA-Interim. There are  
318 differences in simulated 2-m temperature between these two datasets which are of smaller relative  
319 magnitude compared to the RH2.

320 The GFS-FNL driven meteorology EXP1 has a warm bias in NWIGP and CIGP, especially during  
321 night-time, while over EIGP, the model prediction agrees well with observations. EXP2 with the  
322 ERA-Interim driven meteorology and YSU PBL scheme also shows good agreement between  
323 modeled and observed T2 in EIGP. The ERA-Interim driven meteorology with the ACM2 PBL  
324 scheme in EXP3 has a cold bias of up to 7°C over EIGP during daytime from 22<sup>nd</sup> to 24<sup>th</sup> December.  
325 The wind speed evaluation shows that WRF-Chem is over-predicting wind speed. However, it is also  
326 possible that some CPCB stations (e.g., Amritsar and RK Puram) have a wind speed low bias due to  
327 the low measurement height and obstructions such as tall trees near the monitoring station as shown in  
328 FigS3. WRF-chem in general overestimates wind speed and several earlier studies have reported this

329 bias in wind speed (e.g., Mohan and Gupta 2018; Pithani et al.,2019). Moreover, WRF-Chem does not  
 330 have the capability to represent building meteorology and parameterizes the effects of urban areas on  
 331 meteorology through roughness length, which likely leads to overestimation of wind speed. Note that  
 332 at other sites (e.g., over IGI-Delhi and Kanpur) the model measurement agreement is better.

333 The WRF-Chem performance has been statistically assessed against observation using the Taylor  
 334 Diagram (Taylor, 2001), which provides a statistical summary of how well the model output agrees  
 335 with the observation in terms of the Pearson correlation, their centered root-mean-square error  
 336 (RMSE) difference, and the ratios of their variances (Fig. 2). The centered RMS difference is  
 337 proportional to the distance to point “OBS” in the x-axis which measures the extent to which the  
 338 simulated and observed datasets match. The centered RMS difference ( $E'$ ), the correlation ( $R$ ), and the  
 339 standard deviations,  $\sigma_m^2$  (simulated) and  $\sigma_o^2$  (observed) are calculated as:

$$340 \quad R = \frac{\frac{1}{N} \sum (M_n - \bar{M})(O_n - \bar{O})}{\sigma_m \sigma_o} \quad (1)$$

$$341 \quad E'^2 = \frac{1}{N} \sum [(M_n - \bar{M}) - (O_n - \bar{O})]^2 \quad (2)$$

$$342 \quad \sigma_m^2 = \frac{1}{N} \sum (M_n - \bar{M})^2 \quad (3)$$

$$343 \quad \sigma_o^2 = \frac{1}{N} \sum (O_n - \bar{O})^2 \quad (4)$$

344 where the overall mean of a field is indicated by an overbar.

345 Each point in the two-dimensional space of the Taylor diagram represents the above mentioned three  
 346 different statistical metrics simultaneously, as they are related by the follow equation

$$347 \quad E'^2 = \sigma_o^2 + \sigma_m^2 - 2\sigma_o\sigma_m R \quad (5)$$

348 The diagram is constructed based on the similarity of the above equation and the Law of Cosines:

$$349 \quad c^2 = a^2 + b^2 - 2ab\cos\phi$$

350 The percentage bias has also been included to further evaluate the WRF-Chem results. In Fig.  
 351 2, better agreement of WRF-Chem results with observations is shown by the marker's proximity to the  
 352 “OBS” dashed black line. The WRF-Chem RH has a good correlation for all three experiments with  $r$   
 353  $> 0.75$  at all the locations in IGP for all the experiments. However, the RMSE (shown by red dashed  
 354 contours) and the standard deviations are larger for EXP1 compared to EXP2 and EXP3 which lie  
 355 closer to the dashed black line indicating that the simulated RH variations are similar to observations.

356 The mean bias is also large ( $>20\%$ ) for EXP1 (GFS-FNL) for all the stations, marked by red triangles.  
 357 For example, simulated RH at Dwarka (4) and Lucknow (7) for EXP2, and IGI Airport (2), IHBAS  
 358 (3), Lucknow (7), and Patna (8) for EXP3 show good agreement with observation, with  $r > 0.7$ ,  
 359 standard deviation within  $\pm 0.25$  and mean bias within 10%. Among these stations, the model performs  
 360 better for Dwarka (4) and Lucknow (7) for EXP2, IGI Airport (2), and IHBAS (3) for EXP3 with a  
 361 smaller centered RMSE ( $< 0.75$ ).

362 For all the experiments, WRF-Chem T2 agrees well with observations with a correlation  
 363 between 0.8 and 0.95. The points are concentrated near the dashed line showing a low RMSE and

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364 standard deviation for T2, signifying a good agreement of simulated T2 with observation in terms of  
365 temporal variation but the T2 relative bias is large for EXP1 (>20%). The RMSE and relative bias for  
366 EXP1 are larger for several of the stations. For example, simulated T2 agrees best with observation at  
367 IHBAS (3) for EXP1 and IGI Airport (2) for EXP2, with smaller centered RMSE and standard  
368 deviation, and bias <5%. The temporal variability of T2 and RH is predicted well for all the  
369 combinations of inputs (Fig. S1), however, the accuracy of simulated T2 and RH is sensitive to the  
370 choice of meteorological initial/boundary conditions. WRF-Chem predicted RH and T2 agree better  
371 with observations when initialized with ERA-Interim meteorology than with GFS-FNL.

372 The WRF-Chem runs driven by ERA-Interim with YSU (EXP2) and ACM2 PBL (EXP3)  
373 schemes predicted the surface meteorology better over the IGP than the WRF-Chem run driven by  
374 GFS (EXP1). By examining the modeled cloud water content (averaged over all the grids in the  
375 analysis region) in the lowest model level with the INSAT-3D satellite fog intensity for the 23<sup>rd</sup> and  
376 24<sup>th</sup> December 2017 (Fig. 3), it is apparent that WRF-Chem with the ACM2 PBL scheme compared  
377 qualitatively well with observations obtained from INSAT-3D satellite in terms of fog coverage over  
378 CIGP, while the WRF-Chem run with the YSU PBL scheme did not produce widespread fog.  
379 However, there is also fog over EIGP in WRF-Chem with the ACM2 PBL scheme although it is not  
380 observed by the satellite. This is because the model has a cold bias in T2 and a high surface RH over  
381 East IGP with ACM2 PBL and Pleim-Xiu surface scheme as discussed earlier, which favors the  
382 formation of fog in this region. The time series in Fig. 4 shows that EXP3 is capable of predicting the  
383 duration of fog on 23<sup>rd</sup> and 24<sup>th</sup> December. There is a data gap from INSAT 3D observations because  
384 it is unable to capture fog during daytime in the presence of mid and high-level clouds. Although fog  
385 LWC data is available for Delhi from the WiFEx campaign, the WRF-Chem simulation does not  
386 produce fog in NWIGP, therefore, the WiFEx observations are not included in this evaluation.

387 In conclusion, EXP3 is the best configuration for predicting fog formation where the ERA-  
388 Interim meteorology, the ACM2 PBL and surface schemes, and soil moisture nudging is used in the  
389 WRF-Chem simulation. Therefore, the evaluation of predicting AOD, surface aerosol concentrations,  
390 and aerosol composition as well as analysis of the impact of aerosols on fog formation uses the EXP3  
391 configuration.

392

#### 393 4 Aerosol Evaluation

394 Aerosol is an important factor in correct prediction of fog (Maalick et al., 2016; Stolaki et al.,  
395 2015) as the number of fog droplets depends on the aerosol size distribution and concentration.  
396 Aerosols as CCN can affect the liquid water content in fog and therefore an increase in aerosol  
397 concentration can significantly affect fog lifetime (Stolaki et al., 2015; Zhang et al., 2014b). AOD  
398 retrievals from the MODIS satellite have been used to validate the modeled AOD (Fig. 5). It is  
399 observed that the model captures several important features of the MODIS retrieved AOD spatial  
400 distribution but at the same time somewhat struggles to reproduce the observed AOD magnitude in

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401 some parts of the domain. One possible reason for the underestimation would be the EDGAR-HTAP  
402 emission inventory, which has a low bias for residential sector PM<sub>2.5</sub> emissions in India (Sharma et al.,  
403 2022). For instance, the model successfully predicts high aerosol loading seen by MODIS on 20 and  
404 21 December over CIGP and EIGP. This is the region with dense fog both in model and observation.  
405 Higher AOD (>0.5) over CIGP and EIGP can be attributed to the accumulation of aerosols that are  
406 transported by north-westerly winds to these regions from NWIGP (Dey and Di Girolamo, 2011; Jain  
407 et al., 2020; Jethva et al., 2018; Kumar et al., 2018; Yadav et al., 2020). However, WRF-Chem  
408 underestimates AOD over the NWIGP (AOD<0.3) throughout the simulation period and during 23-24  
409 December over CIGP and EIGP where the latter may be related to enhanced scavenging of aerosols by  
410 fog droplets.

411 The west to east gradient in aerosol loading over IGP is consistent with surface PM<sub>2.5</sub>  
412 distribution (Fig. 6a). Surface PM<sub>2.5</sub> concentration is highest in EIGP (>100  $\mu\text{g m}^{-3}$ ) and it decreases  
413 gradually towards NWIGP (~60-80  $\mu\text{g m}^{-3}$ ). The time series of PM<sub>2.5</sub> from CPCB measurements and  
414 the model at stations representative of each region in IGP shows that simulated PM<sub>2.5</sub> compares well  
415 with observation in terms of day-to-day variation over most of the locations in the IGP (Fig. 6 b-e).  
416 The comparison is good over Amritsar (an NWIGP location), where PM<sub>2.5</sub> is mostly primary aerosols  
417 from local emissions e.g., residential heating related biomass burning. Agricultural waste burning is at  
418 its peak during post monsoon months (Oct-Nov), whereas during winter burning for residential  
419 heating increases and the stable boundary layer confines these emissions near the surface (Kumar et  
420 al., 2021; Pawar and Sinha, 2022). PM<sub>2.5</sub> at Amritsar shows a bimodal distribution with morning and  
421 evening peaks whereas it is absent in the model likely due to the absence of diurnal variations in the  
422 WRF-Chem anthropogenic emissions.

423 At Delhi, the daily variations are predicted well although WRF-Chem underestimates PM<sub>2.5</sub>  
424 observations during the first 4 days. Delhi experiences severe air pollution and haze with high PM  
425 loading (> 500  $\mu\text{g m}^{-3}$ ) (Bharali et al., 2019). The model is successful in predicting the high PM<sub>2.5</sub>  
426 episode on the 24<sup>th</sup> of December, but WRF-Chem underpredicts the SO<sub>4</sub><sup>2-</sup>, NH<sub>4</sub><sup>+</sup>, NO<sub>3</sub><sup>-</sup> and Cl<sup>-</sup>  
427 concentrations (Fig. 7). Although simulated SO<sub>2</sub> and NH<sub>3</sub> are comparable with observation, sulfate,  
428 and ammonium are underestimated in the model. SO<sub>4</sub><sup>2-</sup> is underestimated by ~ 9  $\mu\text{g m}^{-3}$ , while NH<sub>4</sub><sup>+</sup>,  
429 NO<sub>3</sub><sup>-</sup> and Cl<sup>-</sup> are underestimated by ~30  $\mu\text{g m}^{-3}$ , ~19  $\mu\text{g m}^{-3}$  and ~40  $\mu\text{g m}^{-3}$  on average, respectively.  
430 In addition, the WRF-Chem model results show that a large percentage of PM<sub>2.5</sub> is classified as “other  
431 inorganics”, which is usually dominated by PM<sub>2.5</sub> other than BC and OC. The WiFEx observations  
432 show that secondary aerosols contribute ~50% of the PM<sub>2.5</sub> concentration, which is in line with other  
433 studies that found secondary aerosols contribute 15-50% to PM<sub>2.5</sub> mass (Sharma and Mandal, 2017;  
434 Behera and Sharma, 2010; Nagar et al., 2017) and NO<sub>3</sub><sup>-</sup> constituted 9-13% of PM<sub>2.5</sub> mass (Lalchandani  
435 et al., 2021; Sharma and Mandal, 2017). This leads to the underestimation of PM<sub>2.5</sub> over Delhi.  
436 Studies report very high chloride over the IGP with values exceeding 100  $\mu\text{g m}^{-3}$  (Lalchandani et al.,  
437 2021) during winter emitted from increased trash burning and industrial emissions (Pant et al., 2015;

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438 Patil et al., 2013). WRF-Chem incorporates trash-burning emissions which include HCl emissions  
439 from Chaudhury et al.,(2021) for this study however, the inventory contains annual emissions and fails  
440 to resolve the seasonality of trash-burning emissions as identified by Nagpure et al., (2015). They  
441 suggested almost all the waste-burning emissions in neighbourhoods with higher socioeconomic status  
442 in Delhi occur due to the use of waste as cheap heating fuel by individuals such as night watchmen  
443 and pavement dwellers. Chaudhary et al., (2021) considers waste burning that occurs due to lack of  
444 collection infrastructure, and at landfills and, therefore, shows a concentration of waste burning  
445 emissions around the periphery of Delhi but low waste burning emissions in the relatively prosperous  
446 city centre. In addition, emissions from other sources (e.g., industries) are unaccounted for in the  
447 model which likely leads to the underestimation in modeled chloride.

448 Over the CIGP and EIGP, the underestimation in PM<sub>2.5</sub> is mostly observed during the dense  
449 fog. It is well known that the hygroscopic aerosols grow in size and are deposited to the surface during  
450 fog (Gupta and Mandariya, 2013; Kaul et al., 2011). PM<sub>2.5</sub> shows an increase initially with the onset  
451 of fog and then it decreases as the aerosols grow and get deposited through fog droplets. A two order  
452 higher deposition rate (Fig. 6 f, g) during fog compared to the deposition rate of dry aerosol results in  
453 the lower PM<sub>2.5</sub> over CIGP and EICP during fog events.

454 A statistical analysis (Table S1) shows a minimum normalized mean bias (NMB) for PM<sub>2.5</sub> at  
455 Amritsar (2.2%), while in other stations, the normalized mean bias ranges from 48 to 53%, similar to  
456 the reported range of model bias (underestimated by 40–60%) in winter over IGP by earlier studies  
457 (Bran and Srivastava, 2017; Ojha et al., 2020). This was accomplished by incorporating trash-burning  
458 emissions in the model simulation, which improved the PM<sub>2.5</sub> prediction, increasing NMB by ~4-8%  
459 in IGP. RMSE values range from 41 to 138 µg m<sup>-3</sup> (normalized RMSE~0.4 to 0.7) comparable to the  
460 reported values by these studies. The Pearson correlation coefficient (r) for the simulated and observed  
461 day-to-day variation in PM<sub>2.5</sub> lies between 0.4 and 0.7 for all the stations in Fig. 6 except at Patna and  
462 Muzaffarpur which lies within the range in these studies. The poor correlation at Patna and  
463 Muzaffarpur is due to the low modeled PM<sub>2.5</sub> concentrations which are caused by increased dry  
464 deposition of aerosol particles activated as fog droplets during fog periods, as discussed earlier.  
465 Furthermore, the fog events in WRF and observations have somewhat different time periods causing  
466 WRF-predicted and the observed PM<sub>2.5</sub> concentrations to decrease at different times.

467 Previous studies have reported that models tend to underestimate the AOD observation (David  
468 et al., 2018; Pan et al., 2015) during the post-monsoon and winter when agricultural waste burning and  
469 anthropogenic emissions dominate. While anthropogenic emissions include a contribution from the  
470 residential sector, the emissions from small-scale burning for residential heating over IGP especially  
471 during winter are likely underestimated in the current emission inventory (Sharma et al., 2022). This  
472 leads to an underestimation of aerosol concentration in the model. Other possible causes for the  
473 underestimation are the biases in the simulated meteorology (Govardhan et al., 2015; Kumar et al.,  
474 2015; Pan et al., 2015) which affects the aerosol concentration. We corrected some of the biases in

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475 meteorology as discussed earlier however there are still residual biases in the simulated meteorology  
476 e.g., overestimation of wind speed by WRF-Chem. We also observe underestimation of secondary  
477 aerosols over NWIGP which contribute significantly to the aerosol loading over IGP. Secondary  
478 aerosol formation is substantial over CIGP and EIGP in the model compared to NWIGP which will be  
479 discussed in a later section. The underestimation of  $PM_{2.5}$  could also be linked to the uncertainty in the  
480 model's chemistry scheme to simulate the secondary aerosols due to missing chemical processes or  
481 due to underestimation of sulfur oxidation at different RH levels (Acharja et al., 2022; Pawar et al.,  
482 2023; Ruan et al., 2022). Moreover, several modeling studies have shown significant improvements in  
483 forecasting surface  $PM_{2.5}$  by assimilation of satellite AOD and  $PM_{2.5}$  (Ghude et al., 2020; Jena et al.,  
484 2020; Kumar et al., 2020) suggesting the importance of correct initialization of the model in  
485 simulating aerosols over IGP.

486

## 487 **5 Effect of Aerosol Radiation feedback**

488 Interactions of aerosols with radiation affects temperature and surface heat fluxes, thereby  
489 weakening the turbulence in the PBL and stabilizing the boundary layer height (Fig. 8b) compared to  
490 the clean environment (Fig. 8a). In the presence of well mixed aerosols within the PBL, the radiative  
491 effect of aerosols lowers the noontime PBL height (Fig. 8b). However, the presence of absorbing  
492 aerosols in the PBL warms the air and changes the thermodynamics. Three cases are shown in Fig.  
493 8(c-e) where increases of scattering aerosol concentrations at the top of PBL (Fig. 8c) increases  
494 scattering of radiation by the aerosol layer and reduces the surface reaching solar radiation similar to  
495 Fig. 8b. Higher concentrations of absorbing aerosols at the top of PBL (Fig. 8d) warms the air above  
496 the boundary layer and strengthens the capping inversion stabilizing the PBL and suppressing its  
497 growth. The shallow PBL and weakened daytime vertical mixing confines aerosols and water vapor  
498 near the surface and worsens the air quality of a region. The aerosols trapped in the stagnant PBL  
499 further affects the radiation flux at the surface and creates a positive feedback loop wherein the PBL is  
500 continually suppressed until interrupted by some synoptic weather phenomenon, such as the western  
501 disturbances in the IGP. On the other hand, higher concentration of absorbing aerosols within the PBL  
502 (Fig. 8e) warms the air in the PBL and this results in the higher PBL height. The raised PBL decreases  
503 the aerosol concentration near the surface which is termed as a negative feedback effect.

504 The aerosol radiation feedback can affect shortwave heating rates (SWHR). The high aerosol  
505 loading over the IGP (Fig. 6 and Fig. 7) allows the AR feedback to reduce the PBL height by more  
506 than 140 m throughout the IGP compared to the surrounding region with AR feedback (Fig. 9a). The  
507 difference in PBL height with and without aerosol radiation feedback is largest during noontime. The  
508 suppressed PBL is due to the decrease in the surface heating flux and the consequent weakening of  
509 turbulence in the PBL. The surface solar radiation flux (SWF) decreases by 5-35 % while the surface  
510 latent heat (LH) and sensible heat (HFX) fluxes decrease by 5-35 % and 10-60 %, respectively (Fig.  
511 S3). The stable, shallow PBL reduces the vertical mixing of aerosols and moisture and confines them

512 near the surface, resulting in increased  $PM_{2.5}$  concentrations and RH near the surface with AR  
513 feedback (Fig. 9). Although T2 should decrease with the reduction in surface SWF, T2 shows mixed  
514 signals with both cooling and warming over IGP. While surface cooling is observed over NWIGP and  
515 EIGP, T2 increases with AR feedback over most of CIGP. The response of AR feedback to T2 varies  
516 in these three regions probably due to differences in the distribution and types of aerosols and the  
517 presence of fog. Increase in surface concentration of  $PM_{2.5}$  occurs more over NWIGP and EIGP with  
518 increase in BC and OIN over NWIGP, and sulfate aerosol over EIGP which results in the surface  
519 cooling due to positive AR feedback in these two regions. Over the CIGP, the AR feedback causes a  
520 depletion of surface  $PM_{2.5}$  (Fig. 9d), which is likely due to their hygroscopic growth, and then dry  
521 deposition (average  $PM_{2.5}$  dry deposition flux of  $331 \mu g m^{-2} hr^{-1}$  with AR feedback and  $282 \mu g m^{-2} hr^{-1}$   
522 without AR feedback) in dense fog. The increase in RH with AR feedback favours the growth of  
523 aerosols in size by the uptake of water

524 Examining further, the time variation of the changes in PBL height, T2, and RH between the  
525 simulations with and without aerosol-radiation feedback (Fig. 9g) shows an increase in T2 while the  
526 surface fluxes, sensible heat flux, latent heat flux, and downward shortwave radiation flux decrease  
527 over CIGP (Fig. 9h). AR feedback affects mostly the lower atmosphere at multiple levels; however,  
528 our finding suggests that the decreased shortwave radiation flux decreases the surface fluxes and thus  
529 the turbulence in the boundary layer resulting in a reduced PBL height on both days. Figure 9 g and h  
530 clearly show a decrease in HFX and LH following the decrease in SWF. Moreover, we observe that  
531 the PBL height is sensitive to latent heat flux likely due to its strong dependence on moisture  
532 availability (Xiu and Pleim, 2001; Zhang and Anthes, 1982).

533 The impact of AR feedback on T2 depends on factors such as the presence of absorbing  
534 aerosols and their vertical distribution via heating or increased SWF (as observed in CIGP, Fig. S4).  
535 Absorbing aerosols in WRF-Chem include BC and OIN (other inorganic aerosols), which both  
536 increase near the surface (Fig. 9e, Fig. S5) due to their confinement in the stable PBL. Some areas in  
537 the fog-affected region show a decrease in BC as well as  $SO_4^{2-}$  likely due to increased dry deposition  
538 in fog water as discussed earlier in this section for  $PM_{2.5}$ . As a result, AR feedback changes the  
539 absorbing to scattering ratio of aerosols over IGP indicated by the decrease in SSA (Single Scattering  
540 Albedo; Fig. S6). In EIGP, sulfate concentrations are larger with AR feedback than without AR  
541 feedback with time periods where the difference is  $>1 \mu g m^{-3}$  (Fig.11). The BC concentration changes  
542 are small ( $<0.5 \mu g m^{-3}$ ) in the EIGP, resulting in a higher SSA near the surface with AR feedback in  
543 EIGP. In the CIGP, BC concentrations increase while sulfate aerosols decrease within the PBL with  
544 AR feedback (Fig.11) compared to the simulation without AR feedback. A decrease in SSA is seen for  
545 the CIGP throughout the boundary layer while in EIGP the decrease occurs near the top of the PBL;  
546 difference in SSA due to AR feedback is negligible in NWIGP. Also contributing to the higher SSA in  
547 EIGP is the increase in RH (Fig. 9) due to AR feedback favoring the growth of aerosols in size by  
548 uptake of water and the production of secondary aerosols such as  $SO_4^{2-}$  and  $NH_4^+$ .

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549 A similar observation has been made by Ramachandran et al., (2020) where SSA decreases  
550 with increasing altitude due to absorbing carbonaceous aerosols at higher elevations which contributes  
551  $\geq 75\%$  to the aerosol absorption over IGP. Increased shortwave heating (Fig. 10) is probably caused by  
552 the increased absorbing aerosols near the surface which overwhelms the surface cooling due to  
553 reduced shortwave radiation at the surface.

554 The increase in 2-m RH is substantial over CIGP on 24<sup>th</sup> December (Fig. 9g) compared to the  
555 previous day following the decrease in PBL height which constrains the moisture near the surface.  
556 The decrease in RH by 2% or more when aerosol-radiation feedback is included compared to no  
557 aerosol-radiation feedback is likely due to increase in T2. However, the increase in RH in the  
558 afternoon associated with a decrease in LH and PBL height is important for the air to saturate which  
559 then favors the formation of fog in a polluted environment. Note that the increase in T2 with AR  
560 feedback is very small ( $< 0.5^\circ\text{C}$ ) which reduces further after noon ( $\sim 12:30$  pm IST) on both days.

561 Another important factor that can affect the extent of change in PBL height is the distribution  
562 of aerosols in the vertical (illustrated in Fig. 8). The pressure-time cross-sections of differences in T,  
563  $\text{PM}_{2.5}$ , BC, and  $\text{SO}_4^{2-}$  between aerosol radiation (AR) feedback (wFB) and no aerosol radiation  
564 feedback (nFB) for three regions, NWIGP, CIGP, and EIGP are shown in Fig. 11. The difference in  
565 the PBL height reaches a maximum with the AR feedback during midday (12:30-15:30 IST). Increase  
566 in temperature in the boundary layer is observed with AR feedback particularly at the upper PBL in all  
567 the regions of IGP. This induces a temperature inversion resulting in a stable and suppressed PBL. In  
568 all the regions the decrease in PBL height (100-200 m) is larger on 24<sup>th</sup> December compared to 23<sup>rd</sup>  
569 December. The difference in the PBL height on 23<sup>rd</sup> and 24<sup>th</sup> December with AR feedback on these  
570 days is possibly controlled by the aerosol distribution during the previous day or early morning on the  
571 same day. For example, in all the regions an increase in  $\text{PM}_{2.5}$  is observed the previous night (23:30  
572 onwards) till  $\sim 11:30$  of December 24, with increased BC over NWIGP and CIGP whereas both BC  
573 and  $\text{SO}_4^{2-}$  over EIGP. The increased  $\text{PM}_{2.5}$  concentrations suppress the development of the PBL after  
574 sunrise with AR feedback on December 24 compared to that on December 23, leading to the observed  
575 differences in  $\Delta\text{PBL}$  height on these two days. Increase in BC concentrations in NWIGP and CIGP are  
576 found above the PBL on 24<sup>th</sup> December whereas BC concentrations decrease within the PBL. This BC  
577 concentration gradient creates a temperature inversion, for example between 10:30-14:30 IST. The  
578 increase in BC warms the air in the PBL; however, the warming is not strong enough to cause  
579 negative feedback over CIGP. On 23<sup>rd</sup> December a small increase in BC is uniform throughout the  
580 PBL, while there is a decrease in  $\text{SO}_4^{2-}$  concentrations, resulting in a warmer PBL (Fig. 11) with AR  
581 feedback.

582 In EIGP, BC distribution is similar to that in CIGP with AR feedback while there is a  
583 substantial increase in sulfate aerosol in the PBL. This results in the strongest extinction in EIGP as  
584 evident from the largest difference in PBL height and surface cooling with AR feedback among the  
585 three regions. Although  $\Delta\text{PBL}$  is small on 23<sup>rd</sup> December, it still results in the accumulation of



586 aerosols during night-time (~23:30 pm onwards) which further strengthens the AR feedback effect the  
587 next day in NWIGP and CIGP. Thus, AR feedback stabilizes the PBL, increases PM<sub>2.5</sub> and RH in the  
588 PBL making conditions favourable for persistence of fog over IGP.

589

## 590 **6 Effect of Aqueous phase chemistry**

591 In this section we discuss the impact of aqueous phase chemistry on aerosol composition and  
592 its interaction with meteorology. There is a considerable difference in the surface concentration of  
593 PM<sub>2.5</sub> (>16 µg m<sup>-3</sup>) in the absence of aqueous chemistry over CIGP and EIGP where fog occurs (Fig.  
594 12a) while the difference is negligible over NWIGP where fog does not occur. This is due to the  
595 formation of secondary aerosols through aqueous phase chemistry and the hygroscopic growth of  
596 aerosols during fog in these regions with the inclusion of aqueous chemistry in the model. In the  
597 region between CIGP and EIGP (83E-84E; marked by the box in Fig. 12a), PM<sub>2.5</sub> concentration is less  
598 in the simulation with aqueous-phase chemistry than without aqueous-phase chemistry because  
599 deposition of fog water aerosols to the surface increases as the fog thickens (Fig. 13, Fig. S7). Figure  
600 13 shows the relation between formation of secondary aerosols, deposition flux of PM<sub>2.5</sub>, and fog with  
601 and without aqueous phase chemistry. During the fog event, the secondary aerosols (SO<sub>4</sub><sup>2-</sup>, NH<sub>4</sub><sup>+</sup>)  
602 increase significantly by 4-10 µg m<sup>-3</sup> due to aqueous phase chemistry adding to the PM<sub>2.5</sub> burden over  
603 IGP. The intensity of fog is high around midnight December 24-25 compared to that on 23<sup>rd</sup> and 24<sup>th</sup>  
604 (1:30-11:30 IST)) which increases the dry deposition flux of PM<sub>2.5</sub> causing a sharp drop in the PM<sub>2.5</sub>  
605 concentration on 24<sup>th</sup> December compared to the previous night's fog event. The observed change in  
606 PM<sub>2.5</sub> over a region is the net result of the formation of secondary aerosols and its deposition with fog  
607 droplets.

608 The composition distribution of PM<sub>2.5</sub> (Fig. 12b) has a similar distribution for the simulations  
609 with and without aqueous phase chemistry over NWIGP where fog did not occur. The primary  
610 aerosols are higher (BC > 9%, OC ~ 16-30%, OIN > 50%), than the secondary aerosols (<5%). While  
611 the model requires fog for accelerated formation of secondary inorganic aerosol, experimental data  
612 (Fig. 7) supports significant formation of secondary inorganic aerosol at elevated RH levels even in  
613 haze aerosol (Acharja et al., 2022). On the other hand, the central and east IGP stations are fog-  
614 covered and therefore, there is an increase in secondary aerosols especially SO<sub>4</sub><sup>2-</sup> and NH<sub>4</sub><sup>+</sup> when  
615 aqueous phase chemistry is included in the simulation. SO<sub>4</sub><sup>2-</sup> is chemically produced via aqueous  
616 phase chemistry in cloud water, hence the abrupt increase whereas NH<sub>4</sub><sup>+</sup> maintains a gas-aerosol and  
617 gas-cloud equilibrium with NH<sub>3</sub> and SO<sub>4</sub><sup>2-</sup> via neutralizing the drop or aerosol. NO<sub>3</sub><sup>-</sup> is high in the  
618 model compared to SO<sub>4</sub><sup>2-</sup> and NH<sub>4</sub><sup>+</sup> and it decreases by ~1-2 % with aqueous phase chemistry. We  
619 observe a small increase in NO<sub>3</sub><sup>-</sup> during fog, however it drops as fog intensifies, more rapidly than that  
620 without aqueous phase chemistry likely due to increase in dry deposition. This results in lower  
621 average NO<sub>3</sub><sup>-</sup> to PM<sub>2.5</sub> ratio with aqueous phase chemistry. Moreover, NO<sub>3</sub><sup>-</sup> is high over the fog  
622 covered CIGP and EIGP compared to NWIGP suggesting that transport and chemistry of NO<sub>x</sub> in

623 CIGP and EIGP produce more  $\text{HNO}_3$ . Aerosol  $\text{NO}_3^-$  is also in equilibrium with  $\text{HNO}_3$  and it is formed  
624 only if excess  $\text{NH}_3$  is available beyond the sulfate neutralization. Thus,  $\text{NH}_4^+$  and  $\text{NO}_3^-$  changes are  
625 likely due to changing the partitioning between gas and liquid based on the production of sulfate.

626  $\text{PM}_{2.5}$  is mostly composed of organic aerosols (OA) over CIGP and EIGP (Lalchandani et al.,  
627 2021; Srinivas and Sarin, 2014) whereas  $\text{PM}_{2.5}$  is OIN (dust) and OA over NWIGP (Ram et al.,  
628 2012a; Sharma and Mandal 2023). Although observational studies report  $\text{Cl}^-$  as one of the largest  
629 contributors (12-17%) to  $\text{PM}_{2.5}$  after the organics (Lalchandani et al., 2021; Pant et al., 2015) during  
630 winter,  $\text{Cl}^-$  is largely underestimated by the model as discussed in section 4 and contributes only ~3%.  
631 A small increase (2-4%) in secondary organic aerosols (SOA) from glyoxal production in aerosols  
632 occurs for the simulation with aqueous phase chemistry included during intense fog, suggesting there  
633 are feedbacks between cloud chemistry (without glyoxal aqueous chemistry) and aerosol chemistry.  
634 However, similar to  $\text{NO}_3^-$ , average SOA (ASOA (anthropogenic)+BSOA (biogenic) + GlySOA)  
635 shows a decrease when aqueous phase chemistry is included. SOA contributes significantly to organic  
636 aerosol loading over IGP (Kaul et al., 2011; Mandariya et al., 2019).

637 The WRF-Chem results on aerosol composition during fog behave similarly to those of  
638 previous observational studies. For example, Ram et al., (2012a) reported an increase of EC, OC, and  
639 WSOC concentrations by ~30% during fog and haze events at Allahabad, a location in the Central  
640 IGP, and a marginal increase of these constituents at Hisar (NWIGP). Several studies report an  
641 increase in inorganic ions ( $\text{NH}_4^+$ ,  $\text{NO}_3^-$ , and  $\text{SO}_4^{2-}$ ) during fog over IGP and elsewhere (Gundel et al.,  
642 1994; Ram et al., 2012a). Recent studies suggest that a significant fraction of atmospheric particulate  
643 matter in the IGP is comprised of carbonaceous aerosol (~30–35% of the PM) and water-soluble  
644 inorganic species (~10–20% of the PM) during October–January when emissions from biomass  
645 burning (including residential heating) are dominant over IGP (Ram et al., 2012b; Rengarajan et al.,  
646 2007; Tare et al., 2006).

647 Both the simulations with and without aqueous-phase chemistry include the AR feedback. The  
648 aqueous chemistry increases the mass of  $\text{PM}_{2.5}$  and the size of the aerosols, both of which contribute to  
649 AR feedback, thus increasing RH and PBL stability. The increase in RH also saturates the air,  
650 promotes aerosol growth by water uptake, and thus favors fog formation. Since the secondary  
651 inorganic aerosols are scattering aerosols, the increased scattering of radiation further reduces the  
652 solar radiation reaching the surface (Fig. 14a). Over CIGP the presence of higher aerosol loading  
653 reduces the T2 during daytime, particularly on the 24<sup>th</sup> of December which then reduces the PBL  
654 height and increases RH near the surface (Fig. 14b). These conditions favor fog formation over the  
655 CIGP. Further, the fog water content with aqueous-phase chemistry is higher than that without  
656 aqueous-phase chemistry on 24<sup>th</sup> December post-midnight (Fig. 13b). This is likely due to saturation  
657 of air due to increase in RH and lower T2, induced by the AR feedback caused by the increase in  
658  $\text{PM}_{2.5}$ . Although the difference in T2 is small (<0.4), favourable conditions mentioned above are  
659 conducive to fog formation. Because aqueous chemistry increases sulfate concentrations, the size of

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660 the aerosols also increase. The increased aerosol size, which can grow further by water uptake, also  
661 impacts the solar radiation reaching the surface, affecting fog formation and dissipation.

662

### 663 **7 Effect of AR feedback and aqueous chemistry on the duration of fog**

664 Aerosol and its radiative effects impact fog characteristics, including the fog liquid water content  
665 (LWC), the fog lifetime over a region and hence its spatial and temporal distribution. Variations of fog  
666 LWC in WRF-Chem contrast the fog in the CIGP and EIGP (Figure 15) as well as among the three  
667 experiments (with aqueous chemistry plus AR feedback, with aqueous chemistry without AR  
668 feedback, and without aqueous chemistry but with AR feedback). WRF-Chem does not simulate fog  
669 over NWIGP in the model for the study period. In Figure 15, only foggy grid points are considered for  
670 the first fog event on 23-24 December. The LWC is 5-15% higher with AR feedback than without AR  
671 feedback and without aqueous phase chemistry for both CIGP and EIGP. The interquartile range is  
672 larger for the simulation with and without AR feedback than without aqueous phase chemistry in  
673 CIGP showing large variability in the LWC. On the other hand, in EIGP the variability in LWC is  
674 greater in the simulation with AR feedback compared to the other two experiments.

675         The formation and dissipation times of the two fog events for the three experiments are listed  
676 in Tables 2 and 3 for CIGP and EIGP. The 23-24 December fog starts forming two hours earlier and  
677 the 24-25 December fog forms one hour earlier in both CIGP and EIGP with AR feedback than  
678 without AR feedback. In the simulation without aqueous phase chemistry, fog formation is delayed by  
679 an hour or two compared to the simulation with aqueous chemistry plus AR feedback in CIGP. In  
680 EIGP the 23-24 December fog forms at the same time with AR feedback and without aqueous phase  
681 chemistry while the 24-25 December fog is delayed by an hour without aqueous phase chemistry. Fog  
682 dissipation usually occurs after sunrise when the shortwave radiative warming at the surface warms  
683 the air, which results in PBL mixing. In addition, absorbing aerosols like BC affect fog dissipation by  
684 increasing the radiative heating in and above the fog. We find an increase in BC and shortwave  
685 heating in the PBL with AR feedback (Fig. 10,11) and warming over CIGP with AR feedback. Fog  
686 intensity starts to decrease after 01:00 UTC (06:30 IST), however, in our study, we find that the fog  
687 dissipates completely in the afternoon (~10:00 UTC or 15:30 IST) for both the simulations with AR  
688 feedback and no aqueous chemistry while an hour later without AR feedback in CIGP. Fog dissipation  
689 is delayed in EIGP with AR feedback compared to that without AR feedback and without aqueous  
690 phase chemistry. In both the regions, fog lifetime increases with AR feedback. All the stations,  
691 however do not show the same pattern, for example, the 23-24 December fog in Lucknow forms and  
692 dissipates at the same time for simulations with AR feedback and without aqueous phase chemistry,  
693 and the 24-25 December fog forms later with AR feedback than without AR feedback. Patna shows no  
694 difference in the 24-25 December fog formation in all the three experiments. To gain better insights on  
695 the fog timing, we recommend that simulations at higher spatial and temporal resolutions be  
696 performed to represent better the fog dynamics at point locations. Furthermore, there are other

697 important factors to consider, e.g., improved emissions, better simulations of aerosol chemical  
698 composition, and evaluation of aerosol deposition.

699 The AR feedback and aqueous-phase chemistry have the potential to impact aerosol-fog  
700 interactions. We can learn about the effect of the aerosol-radiation interactions on CCN concentrations  
701 because the WRF-Chem model calculates the CCN concentrations at different supersaturations as a  
702 diagnostic output. We compare CCN at 0.02% supersaturations, a value typical of fog, among the  
703 three experiments. For the 23-24 December fog in IGP, hourly CCN concentrations are ~10% higher  
704 for the simulations with AR feedback with or without aqueous chemistry than with no AR feedback  
705 (Figure S8) during the first 8 hours of the fog event (16:00-24:00 IST 23 December). Surprisingly, the  
706 simulation with no aqueous chemistry has higher CCN concentrations than the simulation with  
707 aqueous chemistry, as more CCN are expected with aqueous chemistry. because aqueous chemistry  
708 adds sulfate to the aerosol mass increasing the mass of PM<sub>2.5</sub>. Increased PM<sub>2.5</sub> further contributes to  
709 AR feedback, increasing RH which favours the growth of the aerosol size, categorizing more aerosols  
710 as CCN. However, the dry deposition flux (ddmass) also increases in dense fog which causes rapid  
711 loss in CCN and activated aerosols during fog events with the AR feedback (Fig. S7) and more so  
712 without aqueous-phase chemistry. Shao et al. (2023) examined aerosol-fog interactions for two  
713 consecutive fog events by comparing WRF-Chem results with current emissions strengths to those  
714 with low emission strengths. They show that the first fog event promotes formation of the second fog  
715 event leading to wider fog distribution, and longer fog lifetime favoured by multiple feedbacks  
716 including AR feedback i.e., low temperature, high humidity and high stability similar to our study.  
717 While Shao et al. (2023) observe a delay in dissipation of the first event and early formation of second  
718 fog event, we find an early dissipation and early formation of fog with AR feedback as discussed  
719 earlier in this section. In summary, aqueous phase chemistry together with AR feedback promotes  
720 early formation of fog while AR feedback alone promotes early dissipation of fog and plays a critical  
721 role in the formation and evolution of the fog over IGP.

722

## 723 8 Conclusions

724 The effects of aerosol-radiation (AR) feedback and aqueous chemistry in air quality and fog  
725 have been assessed over IGP. We carried out three experiments using WRF-Chem testing different  
726 combinations of PBL schemes and meteorology initial and boundary conditions. The best  
727 representation of surface meteorology for the IGP region for the case study (December 20-24, 2017)  
728 used ERA-Interim reanalysis to drive the meteorology and ACM2 PBL scheme with soil moisture  
729 nudging to ERA-Interim. With this meteorology configuration for WRF-Chem, evaluation of aerosol  
730 concentrations with measurements and the impact of aerosols on atmospheric processes during fog  
731 were examined. Further, we included trash-burning emissions to represent anthropogenic chloride  
732 aerosols in our configuration. Incorporation of trash burning emissions did improve the model  
733 simulations of PM<sub>2.5</sub> and better captured the day-to-day variability of PM<sub>2.5</sub> in IGP, yet underestimated

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734 its magnitude compared to CPCB observations. Moreover, secondary aerosols particularly, chloride  
735 aerosols are underestimated in the model. This underestimation is likely caused by a low bias in the  
736 residential burning emission inventory and a failure of the emission inventory to represent residential  
737 sector emissions from the use of trash as cheap heating fuel properly. AOD regional distribution is  
738 predicted well by the model for most of the IGP. However, AOD is underestimated over NWIGP  
739 likely due to an underestimation of fugitive emissions during wintertime cold spells.

740 The AR interactions showed a significant impact on meteorology and air quality over IGP. A  
741 WRF-chem simulation with AR interactions resulted in a lower PBL height by ~50-270 m compared  
742 to a simulation without AR interactions leading to accumulation of aerosols and moisture near the  
743 surface. Reduced surface shortwave radiation flux and the surface sensible and latent heat fluxes due  
744 to aerosol radiative effect suppressed the turbulence resulting in a stable PBL. The shallow PBL  
745 further increased surface  $PM_{2.5}$  ( $> 8 \mu g m^{-3}$ ) and RH (2-8%) over IGP and this positive feedback  
746 mechanism promoted thickening of fog over IGP. However, an increase in absorbing aerosols in the  
747 PBL gave negative feedback, increasing the shortwave heating and temperature particularly over  
748 CIGP. Fog forms when air is saturated which occurs when the surface temperature is reduced or the  
749 moisture content increases causing saturation of air. This study suggests that increase in RH saturated  
750 the air and the increase in aerosols favoured fog formation as depicted by the thickening of fog  
751 intensity. Aqueous phase chemistry on the other hand contributed significantly to secondary aerosols  
752 in the fog, especially sulfate aerosols, indicating substantial formation of secondary aerosols in the  
753 cloud. The underpredicted secondary aerosols over NWIGP where no fog occurred implies  
754 underestimation of formation of aerosols through gas and aerosol chemistry in the model. This  
755 underestimation could also be linked to an underestimation of pH in the default MOSAIC scheme  
756 (Ruan et al., 2022) which slows the secondary aerosol formation, or an underestimation of the aqueous  
757 sulfur oxidation in haze aerosol at  $> 80\%$  RH before the onset of fog (Acharja et al., 2022), or missing  
758 multiphase oxidation processes (Wang et al., 2022). Nevertheless, we find that the model successfully  
759 simulates the same changes in the inorganic composition during fog in IGP as reported by  
760 observational studies referred earlier in section 6. We also observed that AR feedback with aqueous  
761 chemistry initiated the fog formation 1-2 hours earlier than the initiation time in the simulation  
762 without AR feedback and without aqueous phase chemistry whereas AR feedback alone led to early  
763 dissipation of fog. In addition, fog acted as an important sink of aerosols in a polluted environment  
764 with increased dry deposition with cloud water. Thus, AR feedback and aqueous chemistry play a  
765 significant role in modulating the distribution and concentration of aerosols and evolution of fog in the  
766 PBL. [Aerosol-cloud interactions were not investigated in this study due to the limitation of the ACM2](#)  
767 [PBL scheme in providing necessary information with other modules in WRF. Previous studies of](#)  
768 [aerosol-fog interactions have found that ACI also promotes early onset of fog formation and increases](#)  
769 [fog duration \(Maalick et al., 2016; Yan et al.,2021\). While these previous studies were applied to](#)

770 midlatitude fog events, it is likely that ACI also plays a dominant role in IGP fogs, suggesting that  
771 future studies are needed to fully understand aerosol effects on IGP fog events.

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772 The large emission of aerosols and trace gases in the IGP makes the atmospheric dynamics as  
773 well as chemistry complex, suggesting the need for more studies using both models and ground-based  
774 measurements to better understand the processes. While all aerosol types interact with solar radiation  
775 and reduce the surface reaching flux, presence of absorbing aerosols in the boundary layer and its  
776 vertical distribution plays an important role in modulating the meteorology over IGP. It is therefore  
777 crucial to improve the simulation of absorbing aerosols e.g., BC in the vertical as well as at the surface  
778 to increase the accuracy in predicting formation as well as the dissipation of fog in this region.  
779 Emissions from burning for residential heating are an important source of aerosols in IGP during post-  
780 monsoon and winter and the inclusion of these sources in the emission inventory would improve the  
781 prediction of wintertime aerosols. For example, the underestimation of chloride aerosol in the model  
782 indicates unaccounted emission sources over IGP and the need for more work on better quantifying  
783 trash burning emissions, which may not only improve particulate chloride in the model but also  
784 improve simulations of other aerosol chemical components through aerosol thermodynamics.  
785 Additionally, more detailed modeling studies are required to understand the missing chemical  
786 processes if any in the model which leads to biases in sulfate, nitrate and ammonium partitioning  
787 between gas and aerosol phases. We find that the change in PBL height with AR feedback is sensitive  
788 to changes in LH, signifying the role of soil moisture in PBL dynamics. Several studies have reported  
789 cooling over IGP due to an increase in irrigation (Kumar et al., 2017; Mishra et al., 2020). Further  
790 investigations into the role of irrigation in the increasing fog events over NWIGP would help in better  
791 understanding the formation and persistence of fog over this region. It can be concluded that fog  
792 forecasting is a complex process due to the multiple factors involved and this work suggests that AR  
793 feedback is important in fog forecasting while aqueous phase chemistry plays an important role in  
794 defining the composition of aerosols over IGP.

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811  
812 **Data availability:** All the model simulations are archived on the NCAR campaign storage  
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814 corresponding author. WIFEX data can be made available by contacting Dr S.D. Ghude. Trash  
815 Burning emission data is available on Mendeley data (doi- <http://dx.doi.org/10.17632/t2tn4t9473.1>).  
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817 aerosol data from the Central Pollution Control Board (CPCB) is available at <http://cpcb.nic.in>.

818  
819 **Author contributions:**

820 CB: Conceptualization, Formal Analysis, Writing  
821 MB: Conceptualization, Supervision, Writing-review and editing, Funding acquisition  
822 RK: Conceptualization, Supervision, Writing-review and editing  
823 SDG: provided ground-based observation data, Writing-review and editing  
824 VS and BS: provided trash burning emission data, Writing-review and editing

825  
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827  
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1270 Table 1 Experiment set-up for the study. Numbers in parentheses for the physics options denote the  
 1271 namelist settings of the WRF-Chem model.

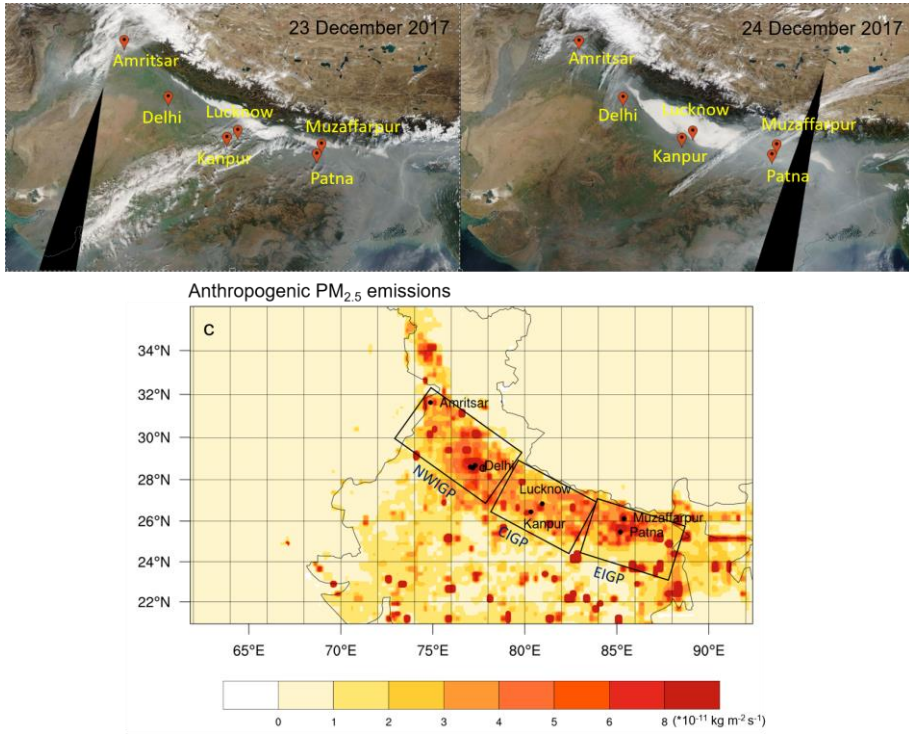
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	EXP 1	EXP 2	EXP 3
Meteorology Initial /lateral boundary Condition:	<i>NCEP Final Analysis (GFS-FNL), 1° x 1°, 6 hourly</i>	<i>ERA-Interim Project, 1.125° x 0.703°, 6 hourly</i>	<i>ERA-Interim Project, 1.125° x 0.703°, 6 hourly</i>
Physics Options			
Cloud Physics	<i>Morrison 2-mom (10)</i>	<i>Morrison 2- mom (10)</i>	<i>Morrison 2- mom (10)</i>
Longwave Radiation	<i>RRTMG scheme (4)</i>	<i>RRTMG scheme (4)</i>	<i>RRTMG scheme (4)</i>
Shortwave Radiation	<i>Goddard shortwave (2)</i>	<i>RRTMG scheme (4)</i>	<i>RRTMG scheme (4)</i>
Surface Layer Physics	<i>Revised MM5 Monin-Obukhov scheme (1)</i>	<i>Revised MM5 Monin-Obukhov scheme (1)</i>	<i>Pleim-Xiu (7)</i>
Surface Model	<i>unified Noah land-surface model (2)</i>	<i>NoahMP (4)</i>	<i>Pleim-Xiu (7)</i>
PBL Scheme	<i>YSU scheme (1)</i>	<i>YSU (1)</i>	<i>ACM2 (7)</i>
Convective Parameterization	<i>Grell-Freitas (3)</i>	<i>Grell-Freitas (3)</i>	<i>Grell-Freitas (3)</i>
	<i>Continuous simulation</i>	<i>*Meteorology refreshed every 24 hr</i>	<i>**Continuous simulation: Soil nudging included</i>

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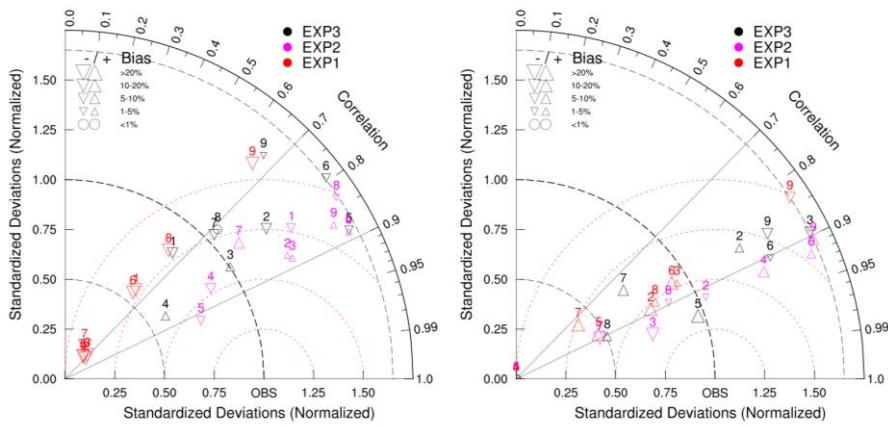
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1277 **Figure 1** The MODIS reflectance (true color) map representing low cloud over Indo Gangetic Plains,  
 1278 India (study region) indicative of likely fog and haze on 23<sup>rd</sup> December (a) and 24<sup>th</sup> December  
 1279 (b)2017. (c) Anthropogenic emission of PM<sub>2.5</sub> over IGP for December 2017 obtained from EDGAR-  
 1280 HTAP. The boxes represent the regions Northwest IGP (NWIGP), Central IGP (CIGP), and East IGP  
 1281 (EIGP).

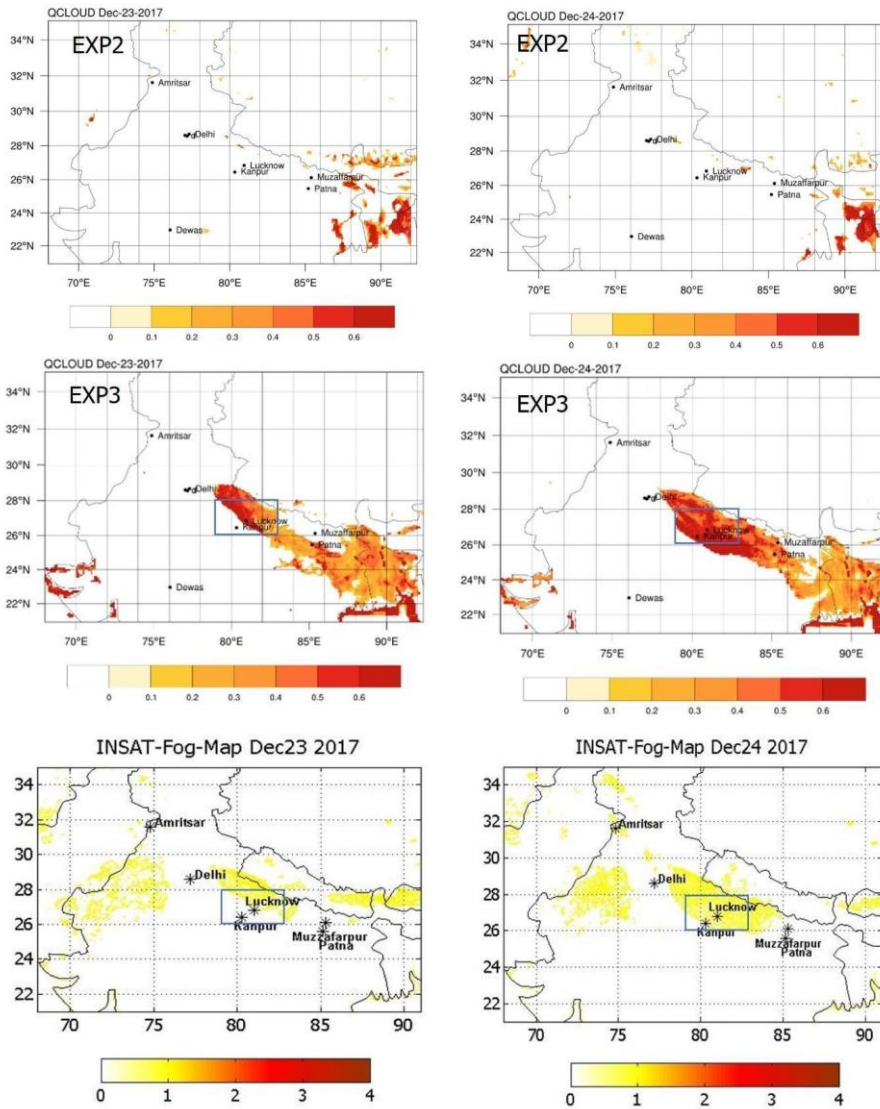
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**Commented [CB19]:** Figure modified in response to Referee report 2, comment 9



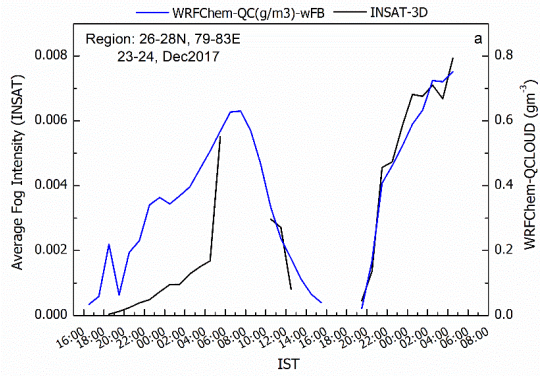
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 1285 Figure 2 Taylor Diagram of simulated (WRF-Chem) and observed (CPCB) relative humidity (left) and  
 1286 2-m temperature (right) over IGP. The colors indicate the experiments. The red dotted contours  
 1287 represent RMS values. The marker (triangles) size varies with a mean bias between the experiments  
 1288 and observation. Upside-down triangles represent positive bias (exp-obs) and vice versa. The stations  
 1289 over IGP are denoted by number 1. Amritsar, 2. IGI Airport (Delhi), 3. IHBAS (Delhi), 4. Dwarka  
 1290 (Delhi), 5. RKP (Delhi), 6. Kanpur, 7. Lucknow, 8. Patna, 9. Muzaffarpur. The locations are marked  
 1291 in Fig.1a.  
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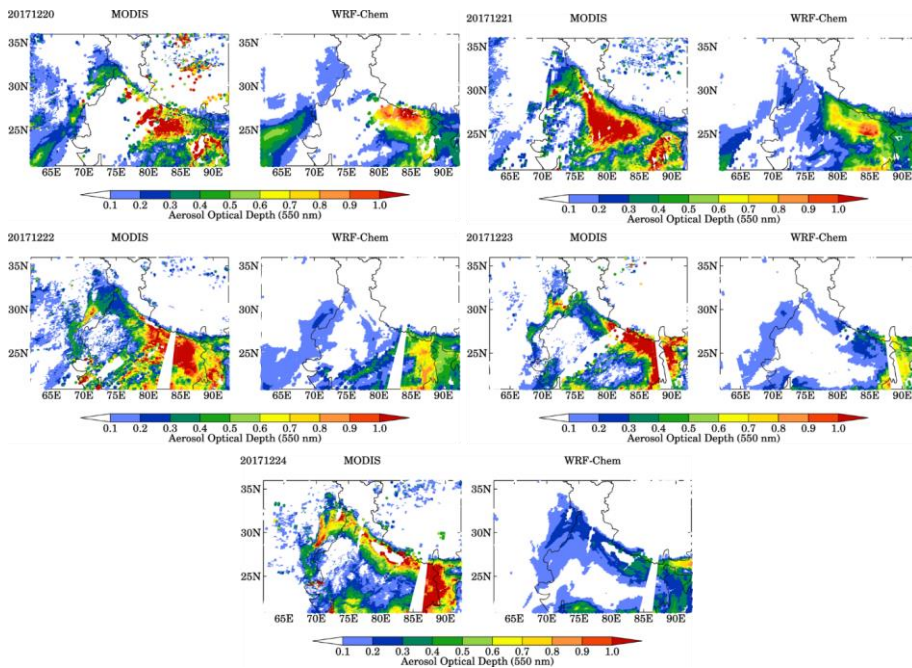
1294 Figure 3 Average WRF-Chem surface layer cloud water mixing ratios ( $\text{g m}^{-3}$ ) for EXP2 and EXP3  
 1295 (top four panels) and INSAT-3D satellite fog intensity (bottom two panels) on 23 and 24 December  
 1296 2017. INSAT-3D satellite fog intensity varies from 0 to 4 indicating SHALLOW, MODERATE,  
 1297 DENSE, and VERY\_DENSE, respectively. The rectangle in central IGP is the region for the time  
 1298 series analysis.

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 1301 Figure 4 Average Hourly variation of fog on 23 and 24 December 2017 from WRF-Chem EXP3  
 1302 simulation and INSAT-3D satellite between 26°N-28°N,79°E-83°E (region shown in Fig 3). The time  
 1303 is in IST (Indian Standard Time; IST is 5.5 hours ahead of Universal Time Coordinate (UTC)).

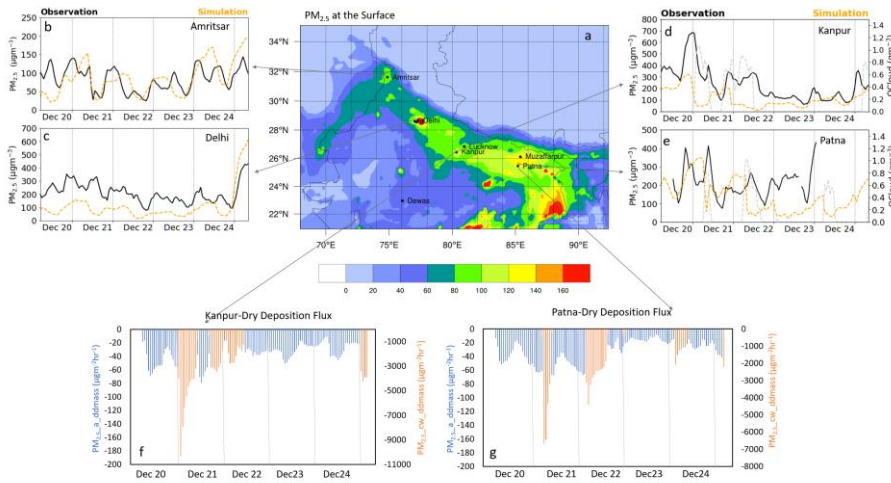
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 1307 Figure 5 Comparison of WRF-Chem AOD with MODIS observation over the model domain on 20,  
 1308 21, 22, 23, and 24 December 2017.

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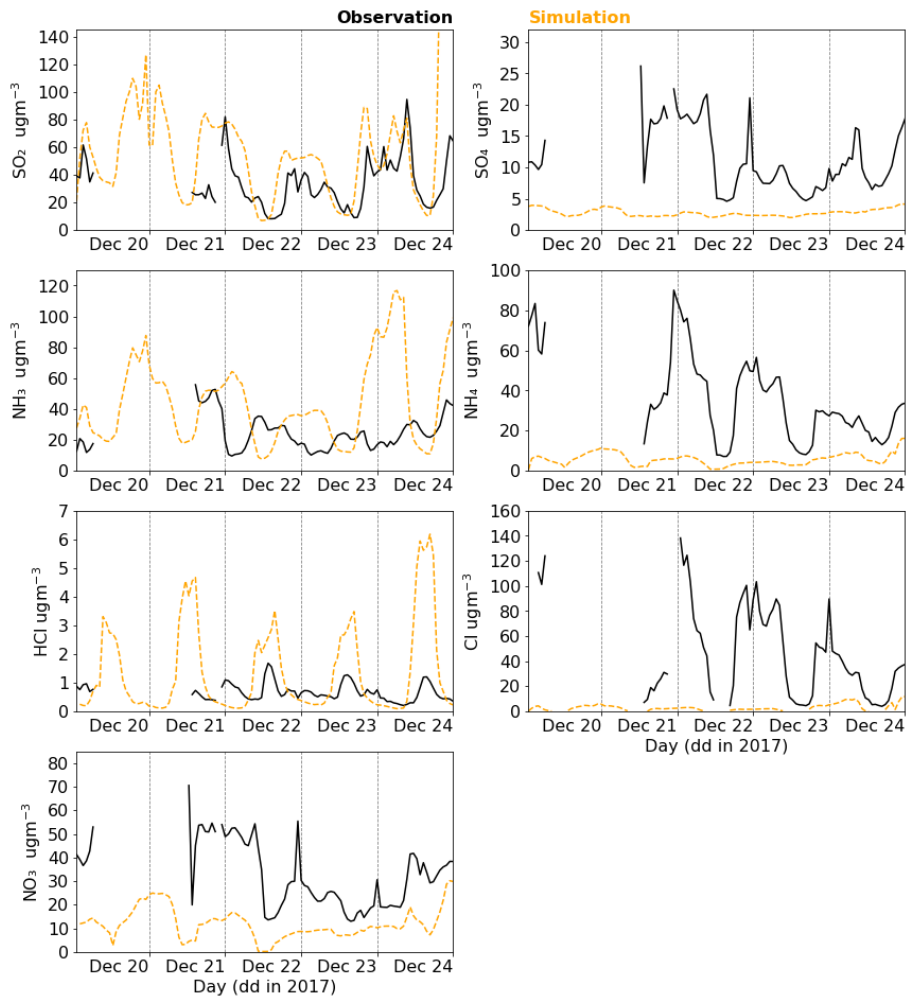
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1313 Figure 6 WRF-Chem simulated surface PM<sub>2.5</sub> map over IGP (a); comparison of WRF-Chem PM<sub>2.5</sub>  
1314 with CPCB observation for the period 20-24 Dec 2017 for (b) Amritsar, (c) Delhi, (d) Kanpur and (e)  
1315 Patna. Dry Deposition rate of PM<sub>2.5</sub> for (f) Kanpur and (g) Patna. The grey dotted line in (d) Kanpur  
1316 and (e) Patna is fog (QCloud) present during the study period.

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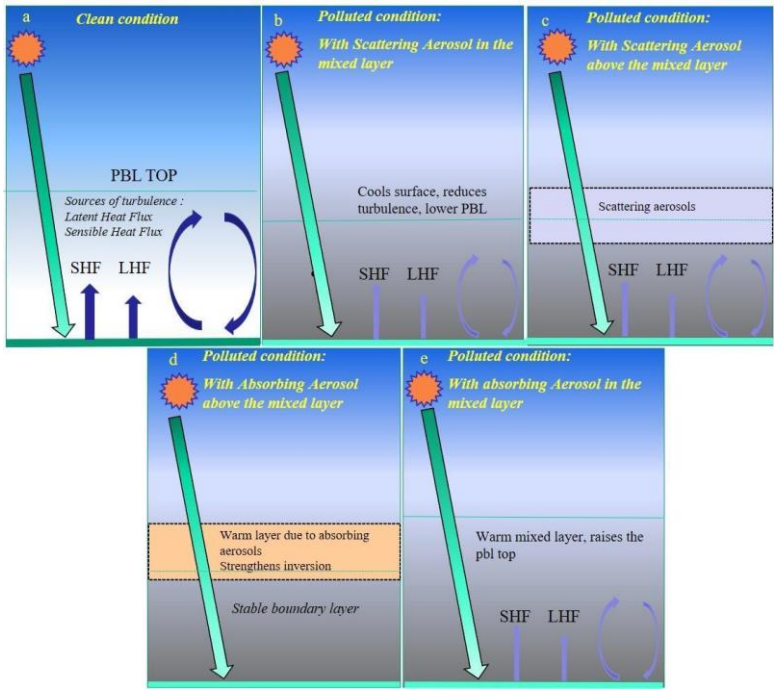
1318 Figure 7 Comparison of WRF-Chem simulated ions ( $\text{SO}_4^{2-}$ ,  $\text{NH}_4^+$ ,  $\text{NO}_3^-$ ,  $\text{Cl}^-$ ) and trace gases ( $\text{SO}_2$ ,  
 1319  $\text{NH}_3$  &  $\text{HCl}$ ) with the observation from WIFEX campaign at Delhi.  
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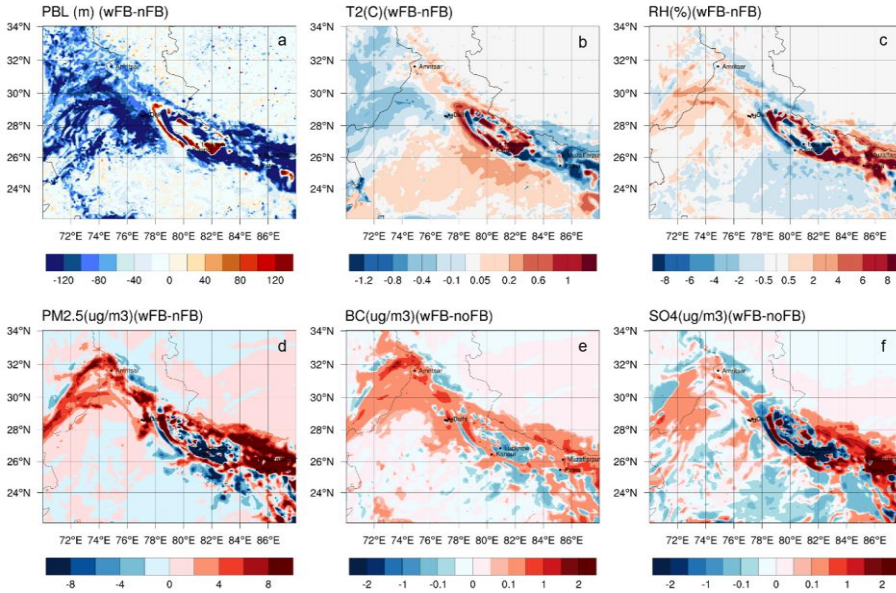
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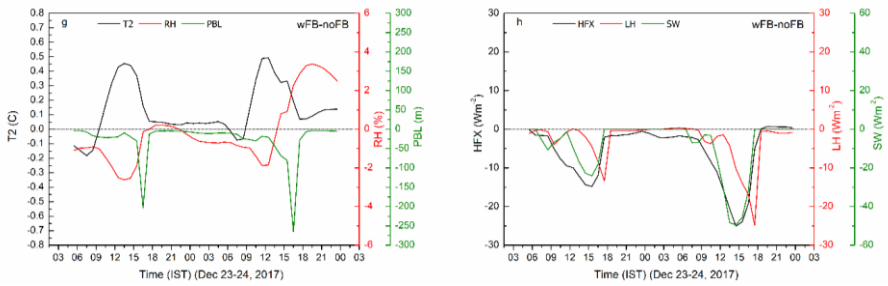


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Figure 8 Schematic diagram of Aerosol Radiation Feedback.



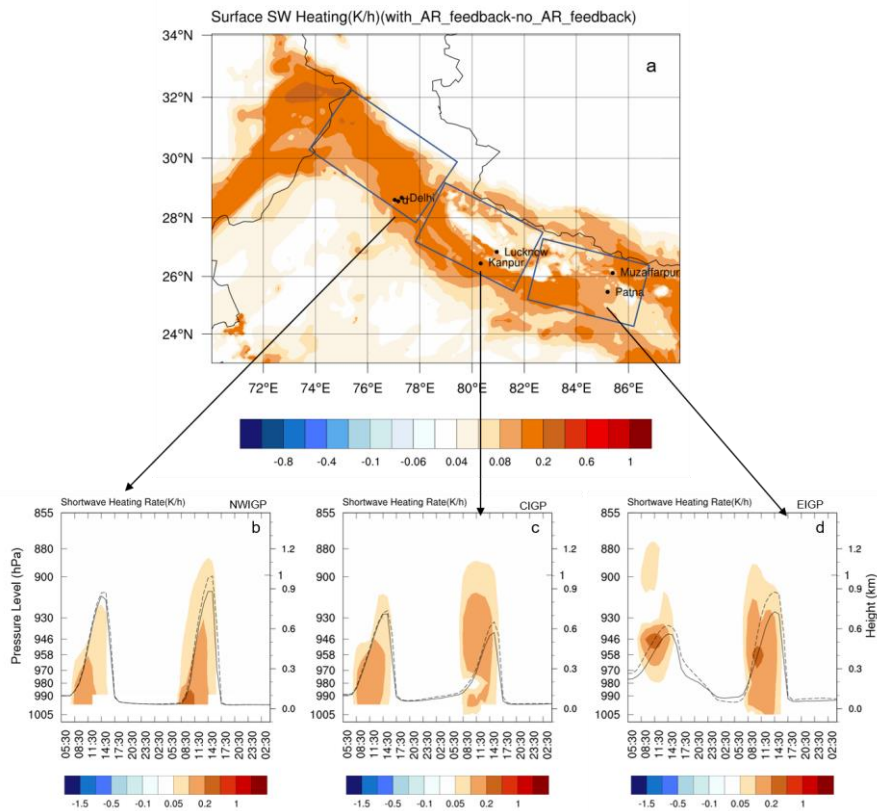
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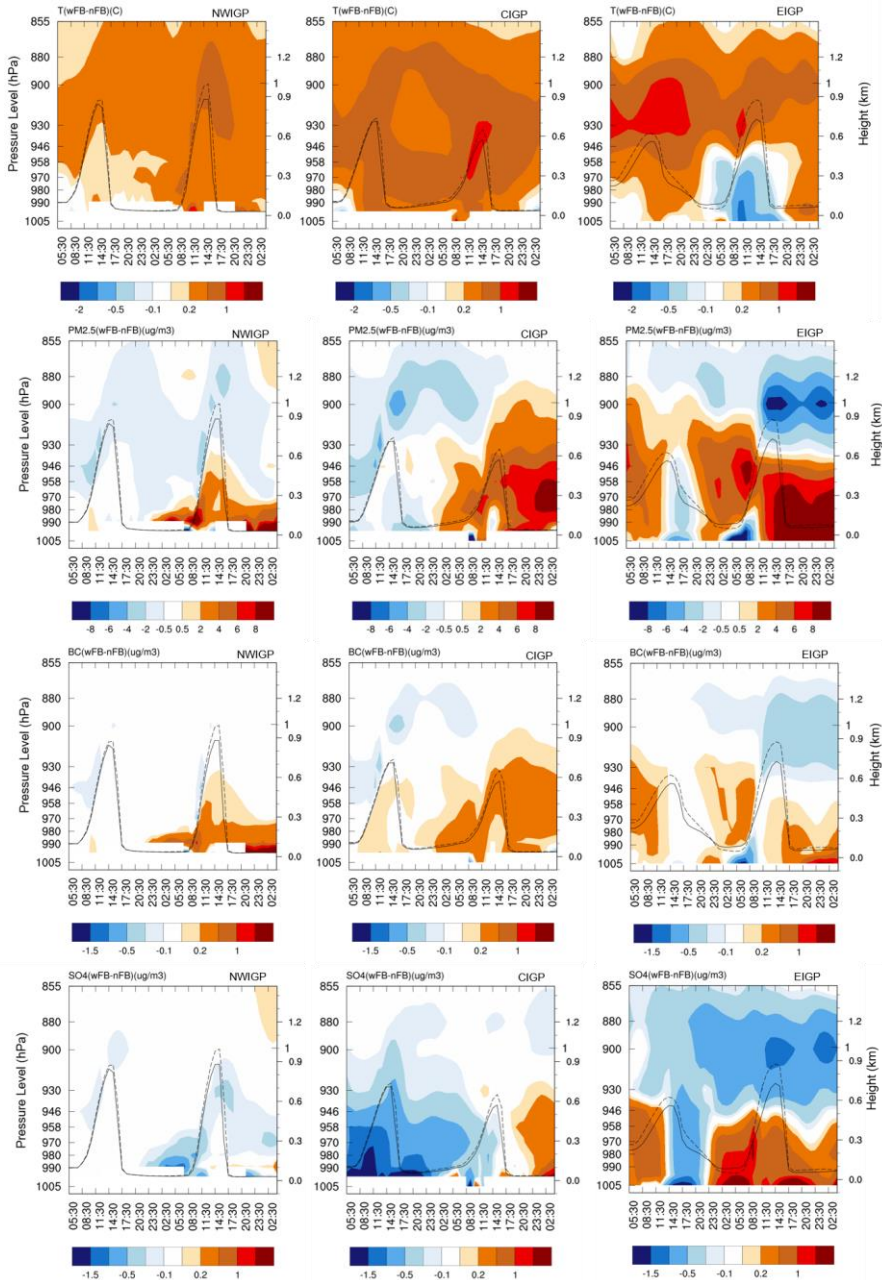
1330 Figure 9 Effect of Aerosol Radiation feedback (wFB-nFB) on (a) PBL height, (b) 2-m temperature, (c)  
 1331 2-m relative humidity, (d) surface  $PM_{2.5}$ , (e) surface BC and (f) surface  $SO_4$  for December 24 at local  
 1332 noon (13:30-15:30 IST). (g) The time series of  $\Delta PBL$ ,  $\Delta T_2$ , and  $\Delta RH$ ; (h)  $\Delta HFX$  (sensible heat flux),  
 1333  $\Delta LH$  (latent heat flux), and  $\Delta SWF$  (downward shortwave flux) over CIGP for December 23 and 24.  $\Delta$   
 1334 denotes the difference between with and without AR feedback (wFB-nFB).

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1338 Figure 10 Differences in shortwave heating rates ( $K h^{-1}$ ) between simulations with and without aerosol  
 1339 radiation feedback (a) at the surface, and for pressure-time cross-sections over (b) NWIGP, (c) CIGP.  
 1340 And (d) EIGP for December 23 and 24. The solid and dashed lines are the PBL height with and  
 1341 without AR feedback respectively. The time is in IST.

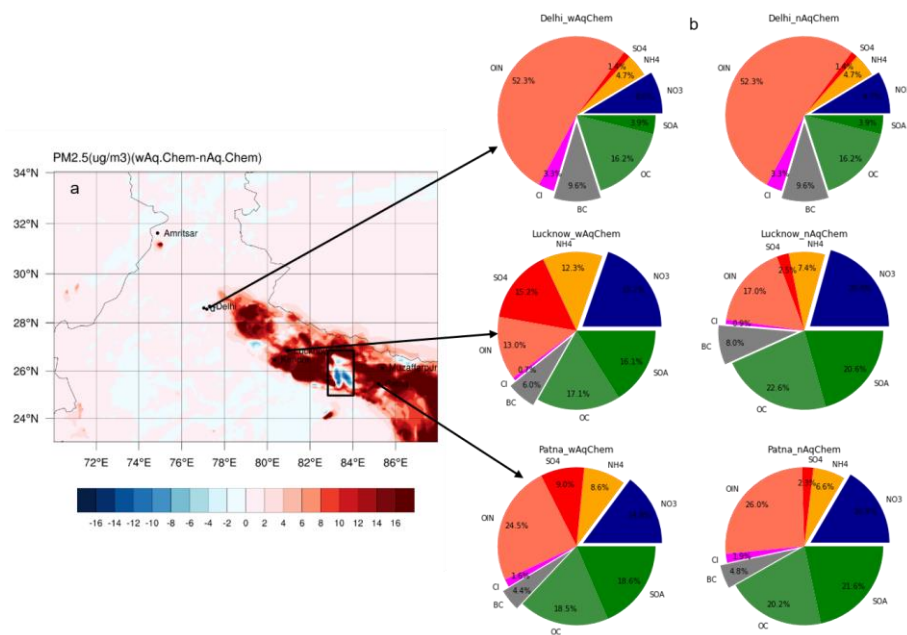


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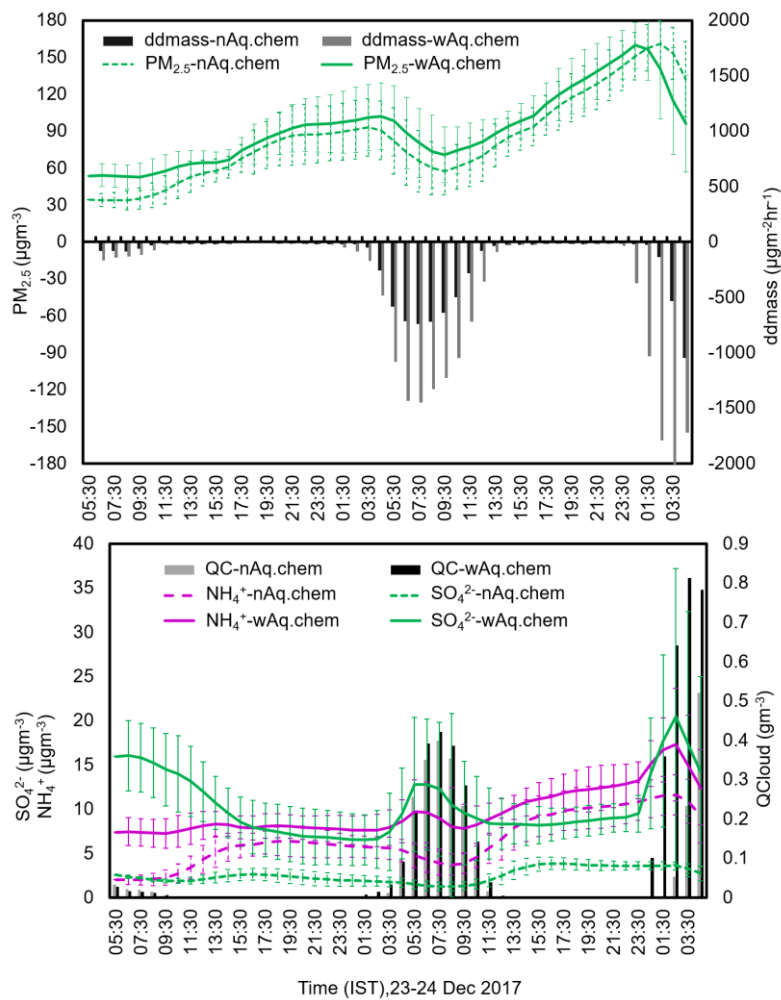
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1345 Figure 11 Pressure-time cross-section of the differences in T, PM<sub>2.5</sub>, BC and SO<sub>4</sub><sup>2-</sup> between  
 1346 simulations with and without the AR feedback for December 23 and 24. The solid and dashed lines  
 1347 are the PBL height with and without AR feedback respectively. The time is in IST.  
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 1351 Figure 12 (a) Surface ΔPM<sub>2.5</sub> (wAq.chem-noAq.chem) and (b) pie charts of PM<sub>2.5</sub> composition  
 1352 distribution for the two cases, with and without Aqueous phase Chemistry for 24 Dec 2017. The  
 1353 stations Delhi, Lucknow (LKN), and Patna are representative of NWIGP, CIGP, and EIGP regions  
 1354 respectively.  
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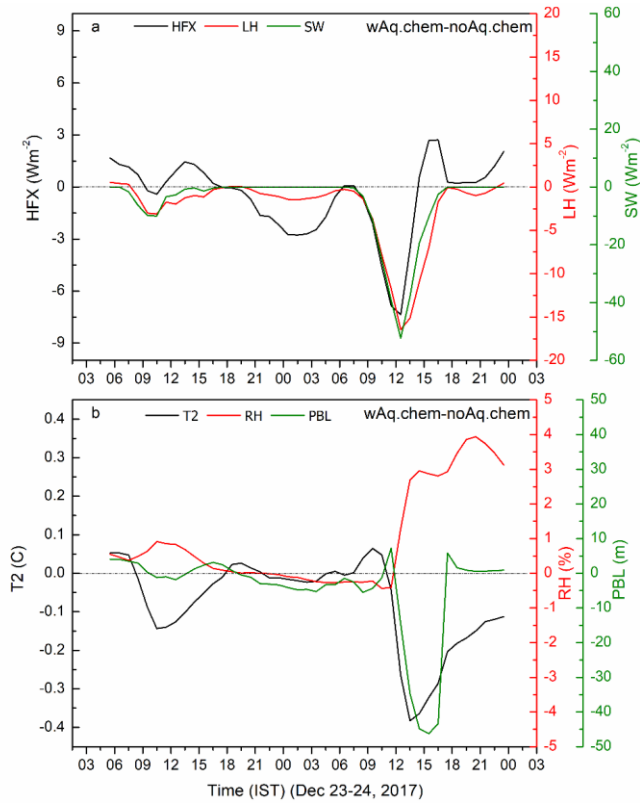


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1359 Figure 13 Time series of (a)  $PM_{2.5}$  and its dry deposition ( $ddmass$ ) flux change, (b)  $SO_4^{2-}$ ,  $NH_4^+$  and  
 1360 LWC (QCloud) with and without aqueous phase chemistry included in the model, averaged over the  
 1361 region bounded by a black rectangle in Fig. 12, for 23 and 24 December, 2017.

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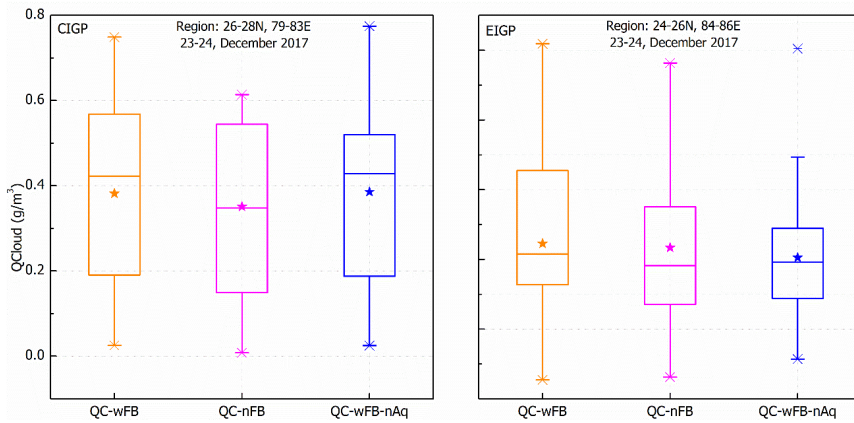
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1366 Figure 14 Time series of (a)  $\Delta$ HFX (sensible heat flux),  $\Delta$ LH (latent heat flux), and  $\Delta$ SWF  
 1367 (downward shortwave flux); (b)  $\Delta$ T2,  $\Delta$ RH, and  $\Delta$ PBL over CIGP (79E-83E,26N-28N), for  
 1368 23 and 24 December, 2017.  $\Delta$  denotes the difference between with and without aqueous phase  
 1369 chemistry.

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1374 Figure 15 Averages (stars), medians (horizontal lines), quartiles (boxes), maxima, and minima for  
1375 LWC (QCloud) averaged over CIGP (left panel) and EIGP (right panel) for the fog event on 23-24  
1376 December 2017. Gold is for the simulation with AR feedback and aqueous chemistry, magenta for the  
1377 simulation with no AR feedback but includes aqueous chemistry, and blue for the simulation with AR  
1378 feedback but no aqueous chemistry. WRF-Chem does not produce fog in the NWIGP during the study  
1379 period.

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Table 2: Table showing the start and end time of fog1 on 23-24 December 2017 with LWC for the sensitivity experiments, with AR feedback, no AR feedback and no Aqueous phase chemistry

Fog 1 (December 23-24, 2017)											
	EXP-wFB			EXP-nFB			EXP-nAq.Chem			Duration of Fog	Duration
	Start time (IST)	End time (IST)	Duration of Fog	Start time (IST)	End time (IST)	Duration of Fog	Start time (IST)	End time (IST)	Duration of Fog		
<b>CIGP</b>	16:30	15:30	23h	18:30	17:30	23h	18:30	15:30	21h		
LWC ( $\text{g m}^{-3}$ )	0.034	0.036 $\pm$ 0.032		0.141 $\pm$ 0.154	0.068 $\pm$ 0.005		0.184 $\pm$ 0.138	0.034 $\pm$ 0.021			
Kanpur	05:30	13:30	8h	05:30	12:30	7h	05:30	12:30	7h		
LWC ( $\text{g m}^{-3}$ )	0.334 $\pm$ 0.487	0.017		0.458 $\pm$ 0.357	0.173 $\pm$ 0.071		0.533	0.025 $\pm$ 0.0123			
Lucknow	23:30	14:30	15h	00:30	14:30	14h	23:30	14:30	15h		
LWC ( $\text{g m}^{-3}$ )	0.269 $\pm$ 0.145	0.087 $\pm$ 0.040		0.232 $\pm$ 0.132	0.029 $\pm$ 0.024		0.139 $\pm$ 0.084	0.025 $\pm$ 0.012			
<b>EIGP</b>	21:30	12:30	15h	23:30	10:30	11h	21:30	10:30	13h		
LWC ( $\text{g m}^{-3}$ )	0.099 $\pm$ 0.092	0.007		0.198 $\pm$ 0.188	0.084 $\pm$ 0.060		0.026 $\pm$ 0.008	0.153 $\pm$ 0.119			
Patna	00:30	12:30	12h	04:30	10:30	6h	02:30	10:30	8h		
LWC ( $\text{g m}^{-3}$ )	0.100 $\pm$ 0.090	0.007		0.009 $\pm$ 0.005	0.038 $\pm$ 0.041		0.196 $\pm$ 0.198	0.166 $\pm$ 0.130			
Muzzafarpur	05:30	11:30	6h	06:30	10:30	4h	06:30	09:30	3h		
LWC ( $\text{g m}^{-3}$ )	0.112 $\pm$ 0.146	0.043 $\pm$ 0.057		0.051 $\pm$ 0.041	0.003		0.142 $\pm$ 0.151	0.157 $\pm$ 0.064			

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1398 Table 2: Table showing the start time of fog 2 on 24 December 2017 with LWC for the sensitivity  
 1399 experiments, with AR feedback, no AR feedback and no Aqueous phase chemistry. Fog2 end time  
 1400 could not be noted as simulation ended on 25 December 2017, 00UT (5:30 IST) before fog2  
 1401 dissipates.

<b>Fog 2 (December 24, 2017)</b>			
<b>Start time</b>			
<b>(IST)</b>			
	<b>EXP-wFB</b>	<b>EXP-nFB</b>	<b>EXP-nAq.Chem</b>
<b>CIGP</b>	<b>19:30</b>	<b>20:30</b>	<b>21:30</b>
LWC ( $\text{g m}^{-3}$ )	0.025	0.008±0.007	0.025
Kanpur	21:30	22:30	23:30
LWC ( $\text{g m}^{-3}$ )	0.041±0.007	0.298±0.218	0.482±0.398
Lucknow	21:30	20:30	00:30
LWC ( $\text{g m}^{-3}$ )	0.203±0.165	0.005	0.229±0.209
<b>EIGP</b>	<b>00:30</b>	<b>01:30</b>	<b>01:30</b>
LWC ( $\text{g m}^{-3}$ )	0.024±0.030	0.072±0.088	0.014±0.009
Patna	03:30	03:30	03:30
LWC ( $\text{g m}^{-3}$ )	0.030± 0.046	0.018	0.060
Muzzafarpur	04:30	No fog	No fog
LWC ( $\text{g m}^{-3}$ )	0.159±0.038		

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