

## *Supplementary Material*

### **Enhancing characterization of organic nitrogen components in aerosols and droplets using high-resolution aerosol mass spectrometry**

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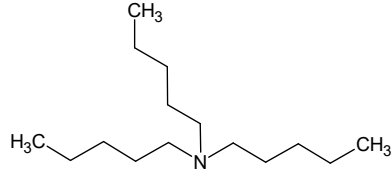
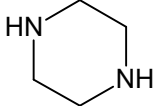
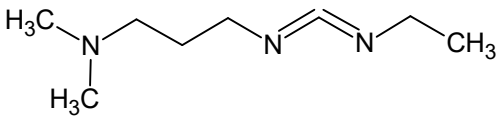
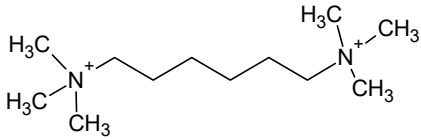
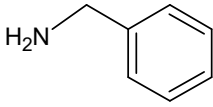
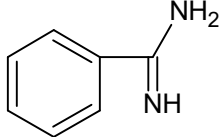
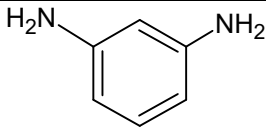
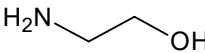
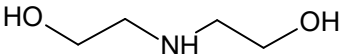
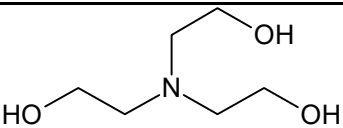
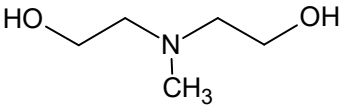
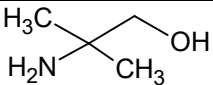
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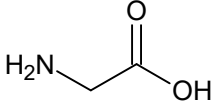
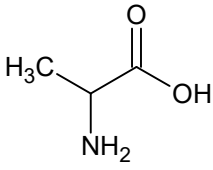
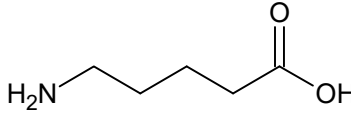
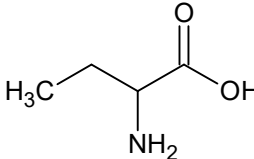
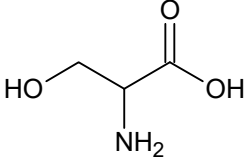
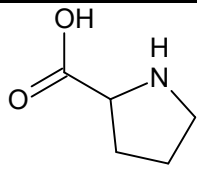
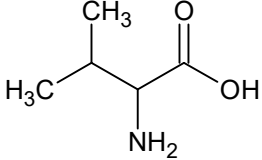
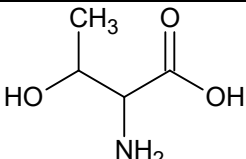
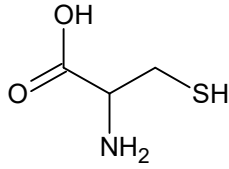
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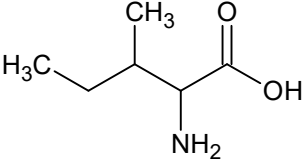
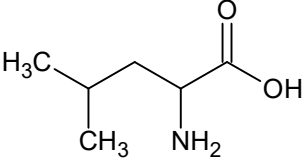
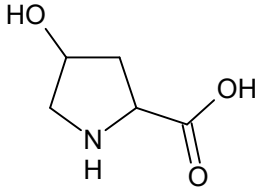
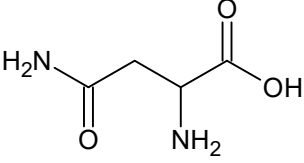
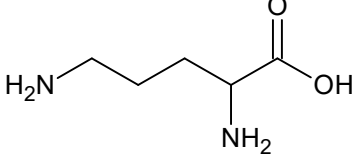
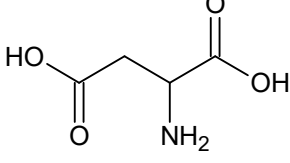
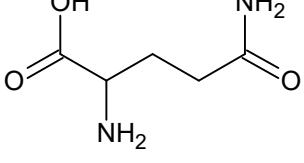
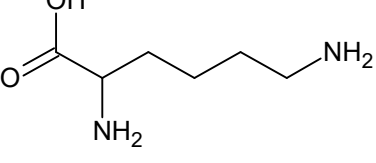
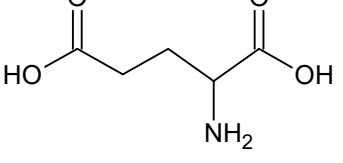
Table S1. List of N-containing organic (ON) standards analyzed in this study.

No.	Name	Abbr.	Functional Groups	Chemical Formula	Molecular weight	Structure
1	Methylamine (Hydrochloride)	MA	Amine	CH <sub>5</sub> N (HCl)	31.0571 (36.4610)	$\text{H}_3\text{C}-\text{NH}_2$
2	Dimethylamine (Hydrochloride)	DMA	Amine	C <sub>2</sub> H <sub>7</sub> N (HCl)	45.0837 (36.4610)	$\text{H}_3\text{C}-\text{NH}-\text{CH}_3$
3	Trimethylamine (Hydrochloride)	TMA	Amine	C <sub>3</sub> H <sub>9</sub> N (HCl)	59.1103 (36.4610)	$\text{H}_3\text{C}-\text{N}(\text{CH}_3)_2$
4	Ethylamine (Hydrochloride)	EA	Amine	C <sub>2</sub> H <sub>7</sub> N (HCl)	45.0837 (36.4610)	$\text{H}_3\text{C}-\text{CH}_2-\text{NH}_2$
5	Diethylamine (Hydrochloride)	DEA	Amine	C <sub>4</sub> H <sub>11</sub> N (HCl)	73.1368 (36.4610)	$\text{H}_3\text{C}-\text{CH}_2-\text{NH}-\text{CH}_2-\text{CH}_3$
6	Triethylamine (Hydrochloride)	TEA	Amine	C <sub>6</sub> H <sub>15</sub> N (HCl)	101.1900 (36.4610)	$\text{H}_3\text{C}-\text{CH}_2-\text{N}(\text{CH}_2\text{CH}_3)_2$
7	Isopropylamine (Hydrochloride)	iPPA	Amine	C <sub>3</sub> H <sub>9</sub> N (HCl)	59.11 (36.4610)	$\text{H}_3\text{C}-\text{CH}(\text{CH}_3)-\text{NH}_2$
8	Butylamine (Hydrochloride)	BTA	Amine	C <sub>4</sub> H <sub>11</sub> N (HCl)	73.14 (36.4610)	$\text{H}_3\text{C}-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{NH}_2$
9	Isobutylamine (Hydrochloride)	iBTA	Amine	C <sub>4</sub> H <sub>11</sub> N (HCl)	73.14 (36.4610)	$\text{H}_3\text{C}-\text{CH}(\text{CH}_3)-\text{CH}_2-\text{NH}_2$
10	1,4-Diaminobutane (Hydrochloride)	BDA	Amine	C <sub>4</sub> H <sub>12</sub> N <sub>2</sub> (2HCl)	88.15 (72.9220)	$\text{H}_2\text{N}-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{NH}_2$
11	Pentylamine (Hydrochloride)	PTA	Amine	C <sub>5</sub> H <sub>13</sub> N (HCl)	87.16 (36.4610)	$\text{H}_3\text{C}-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{NH}_2$
12	Tributylamine (Hydrochloride)	TBA	Amine	C <sub>12</sub> H <sub>27</sub> N (HCl)	185.3495 (36.4610)	$\text{H}_3\text{C}-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{N}(\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3)_2$

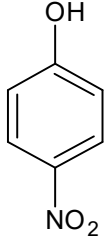
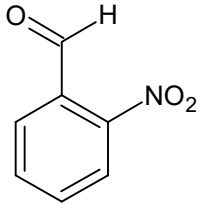
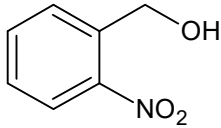
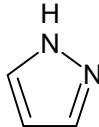
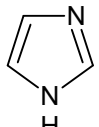
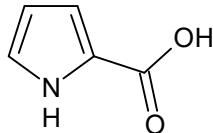
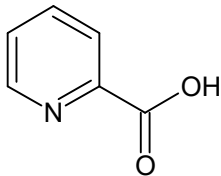
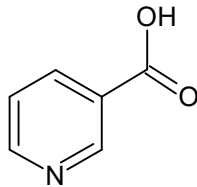
13	Tri-n-amyliamie (Hydrochloride)	TAA	Amine	$C_{15}H_{33}N$ (HCl)	227.4292 (36.4610)	
14	Piperazine (Dihydrochloride)	PIP	Amine	$C_4H_{10}N_2$ (2HCl)	86.1356 (72.9220)	
15	1-(3-Dimethylaminopropyl)-3-ethylcarbodiimide (Hydrochloride)	DMPED	Amine	$C_8H_{17}N_3$ (HCl)	155.2407 (36.4610)	
16	Hexamethonium chloride	HM	Amine	$C_{12}H_{30}N_2(Cl_2)$	202.3800 (70.9060)	
17	Benzylamine (Hydrochloride)	BZA	Amine	$C_7H_9N$ (HCl)	107.1531 (36.4610)	
18	Benzamidine (Hydrochloride)	BZD	Amine	$C_7H_8N_2$ (HCl)	120.1518 (36.4610)	
19	1,3-phenylenediamine (Dihydrochloride)	PHLDA	Amine	$C_6H_8N_2$ (2HCl)	108.1411 (72.9220)	
20	Ethanolamine Also tried MEA.HCl	EAOH	Amine	$C_2H_7NO$	61.0831	
21	Diethanolamine (Hydrochloride)	DEAOH	Amine	$C_4H_{11}NO_2$ (HCl)	105.1356 (36.4610)	
22	Triethanolamine	TEAOH	Amine	$C_6H_{15}NO_3$	149.1882	
23	Methyldiethanolamine (Hydrochloride)	MDEA	Amine	$C_5H_{13}NO_2$ (HCl)	119.1622 (36.4610)	
24	2-Amino-2-methyl-1-propanol (Hydrochloride)	AMP	Amine	$C_4H_{11}NO$ (HCl)	89.1362 (36.4610)	

25	Bis(2-hydroxypropyl)amine	BOHA	Amine	$C_6H_{15}NO_2$	133.1888	
26	4-Aminophenol (Hydrochloride)	APA	Amine	$C_6H_7NO$ (HCl)	109.1259 (36.4610)	
27	Ethylene diamine tetraacetic acid	EDTA	Amine	$C_{10}H_{16}N_2O_8$	292.2426	
28	Urea	UREA	Amide	$CH_4N_2O$	60.0553	
29	Benzamide	BZAM	Amide	$C_7H_7NO$	121.1366	
30	N,N'-Methylenebisacrylamide	NMAM	Amide	$C_7H_{10}N_2O_2$	154.1665	
31	Nicotinamide	NCAM	Amide	$C_6H_6N_2O$	122.1246	
32	Pyrazinecarboxamide	PRCAM	Amide	$C_5H_5N_3O$	123.1127	
33	Acetaminophenol	AAP	Amide	$C_8H_9NO_2$	151.1626	

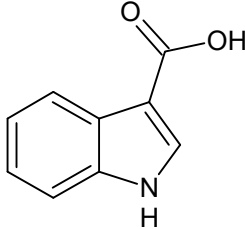
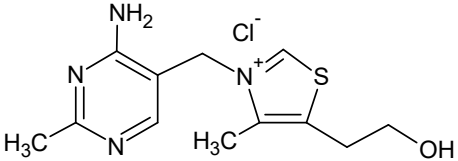
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35	L-Alanine	ALA	Amino Acid	$C_3H_7NO_2$	89.0932	
36	$\gamma$ -Aminobutyric acid	GABA	Amino Acid	$C_4H_9NO_2$	103.1198	
37	R-2-aminobutyric acid	RGABA	Amino Acid	$C_4H_9NO_2$	103.1198	
38	L-Serine	SER	Amino Acid	$C_3H_7NO_3$	105.0926	
39	L-Proline	PRO	Amino Acid	$C_5H_9NO_2$	115.1305	
40	L-Valine	VAL	Amino Acid	$C_5H_{11}NO_2$	117.1463	
41	L-Threonine	THR	Amino Acid	$C_4H_9NO_3$	119.1192	
42	L-Cysteine	CYS	Amino Acid	$C_3H_7NO_2S$	121.1580	

43	L-Isoleucine	ILE	Amino Acid	$C_6H_{13}NO_2$	131.1729	
44	L-Leucine	LEU	Amino Acid	$C_6H_{13}NO_2$	131.1729	
45	trans-4-hydroxyl-proline	PROOH	Amino Acid	$C_5H_9NO_3$	131.1299	
46	L-Asparagine	ASN	Amino Acid	$C_4H_8N_2O_3$	132.1179	
47	L-Ornithine (Hydrochloride)	ORN	Amino Acid	$C_5H_{12}N_2O_2$ (HCl)	132.1610 (34.4610)	
48	L-Aspartic Acid	ASP	Amino Acid	$C_4H_7NO_4$	133.1027	
49	L-Glutamine	GLN	Amino Acid	$C_5H_{10}N_2O_3$	146.1445	
50	L-Lysine	LYS	Amino Acid	$C_6H_{14}N_2O_2$	146.1876	
51	L-Glutamic Acid	GLU	Amino Acid	$C_5H_9NO_4$	147.1293	

52	L-Methionine	MET	Amino Acid	$C_5H_{11}NO_2S$	149.2110	
53	L-Histidine (Monochloride. Monohydrate)	HIS	Amino Acid	$C_6H_9N_3O_2$ (HCl·H <sub>2</sub> O)	155.1546 (36.4610+18.013)	
54	L-Methionine Sulfoxide	METSO	Amino Acid	$C_5H_{11}NO_3S$	165.2110	
55	L-Phenylalanine	PHE	Amino Acid	$C_9H_{11}NO_2$	165.1891	
56	L-Arginine (Hydrochloride)	ARG	Amino Acid	$C_6H_{14}N_4O_2$ (HCl)	174.2010 (36.4610)	
57	L-Tyrosine	TYR	Amino Acid	$C_9H_{11}NO_3$	181.1885	
58	L-Tryptophan	TRP	Amino Acid	$C_{11}H_{12}N_2O_2$	204.2252	
59	4-Aminobenzoic acid	ABA	Amino acid	$C_7H_7NO_2$	137.1360	
60	4-Nitroimidazole	NID	Nitro- compounds	$C_3H_3N_3O_2$	113.0748	

61	4-Nitrophenol	NPH	Nitro-compounds	$C_6H_5NO_3$	139.1088	
62	2-Nitrobenzaldehyde	NBAD	Nitro-compounds	$C_7H_5NO_3$	151.1195	
63	2-nitrobenzyl alcohol	NBAC	Nitro-compounds	$C_7H_7NO_3$	153.1354	
64	Pyrazole	PRZ	N-heterocycle	$C_3H_4N_2$	68.0773	
65	Imidazole	ID	N-heterocycle	$C_3H_4N_2$	68.0773	
66	Pyrrole-2-carboxylic acid	PRCA	N-heterocycle	$C_5H_5NO_2$	111.0987	
67	2-Picolinic acid	PCA	N-heterocycle	$C_6H_5NO_2$	123.1094	
68	Nicotinic acid	NCA	N-heterocycle	$C_6H_5NO_2$	123.1094	



69	Indole-3-carboxylic acid	IDCA	N-heterocycle	C <sub>9</sub> H <sub>7</sub> NO <sub>2</sub>	161.1574	
70	Thiamine (hydrochloride)	THA	N-heterocycle	C <sub>12</sub> H <sub>17</sub> ClN <sub>4</sub> OS (HCl)	300.8076 (36.4610)	
71	Amino acid standard	AS	Amino Acid	-	-	Physiological acidics and neutrals
72	Protein	PROTEIN	Protein	-	-	Cytochrome C from horse heart
73	Elliott Soil	HA	Humic substance	-	-	Standard HA from IHSS*, No. 1S102H
74	Waskish Peat	FA	Humic substance	-	-	Reference HA from IHSS, No. 1R107F
75	Suwannee River	NOM	Humic substance	-	-	Aquatic NOM (Natural Organic Matter) from IHSS, No. 1R101N

\*International Humic Substances Society (<http://www.ihss.gatech.edu/>).

Table S2. List of the N-containing organic (ON) standards used to make the 1:1 (mass ratio) mixtures with sulfate (in ammonium sulfate).

No.	ID. in Table S1	Name	No.	ID. in Table S1	Name
<b>1</b>	1	Methylamine (Hydrochloride)	<b>10</b>	34	Glycine
<b>2</b>	2	Dimethylamine (Hydrochloride)	<b>11</b>	35	L-Alanine
<b>3</b>	3	Trimethylamine (Hydrochloride)	<b>12</b>	39	L-Proline
<b>4</b>	4	Ethylamine (Hydrochloride)	<b>13</b>	41	L-Threonine
<b>5</b>	5	Diethylamine (Hydrochloride)	<b>14</b>	42	L-Cysteine (Hydrochloride)
<b>6</b>	6	Triethylamine (Hydrochloride)	<b>15</b>	59	4-Aminobenzonic acid
<b>7</b>	20	Ethanolamine (Hydrochloride)	<b>16</b>	68	Nicotinic acid
<b>8</b>	19	1,3-phenylenediamine (Dihydrochloride)	<b>17</b>	23	Methyldiethanolamine (Hydrochloride)
<b>9</b>	33	Acetaminophenol	<b>18</b>	14	Piperazine (Dihydrochloride)

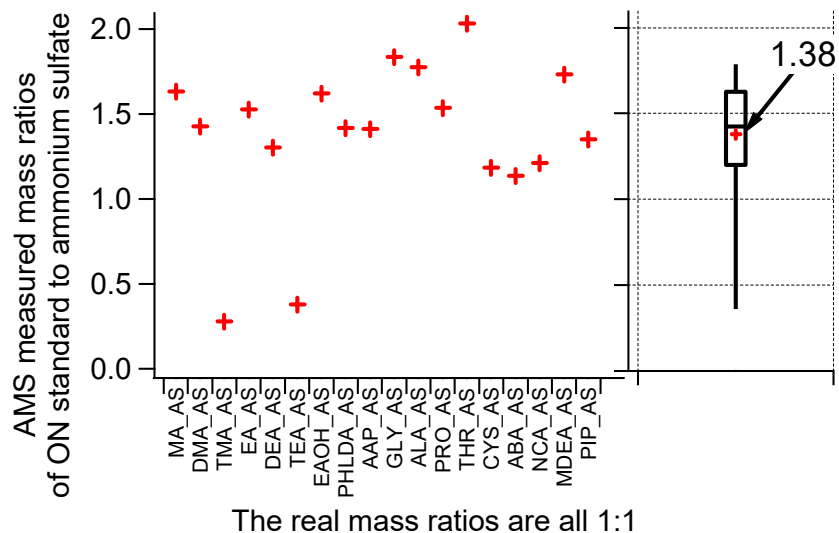


Figure S1. AMS measured mass ratios of the mixtures of different ON standards with ammonium sulfate. All mass ratios of the mixtures are prepared in 1:1 (the whiskers above and below the boxes indicate the 90<sup>th</sup> and 10<sup>th</sup> percentiles, the upper and lower boundaries of the boxes indicate the 75<sup>th</sup> and 25<sup>th</sup> percentiles, and the lines in the boxes indicate the median values and the cross symbols indicate the mean values).

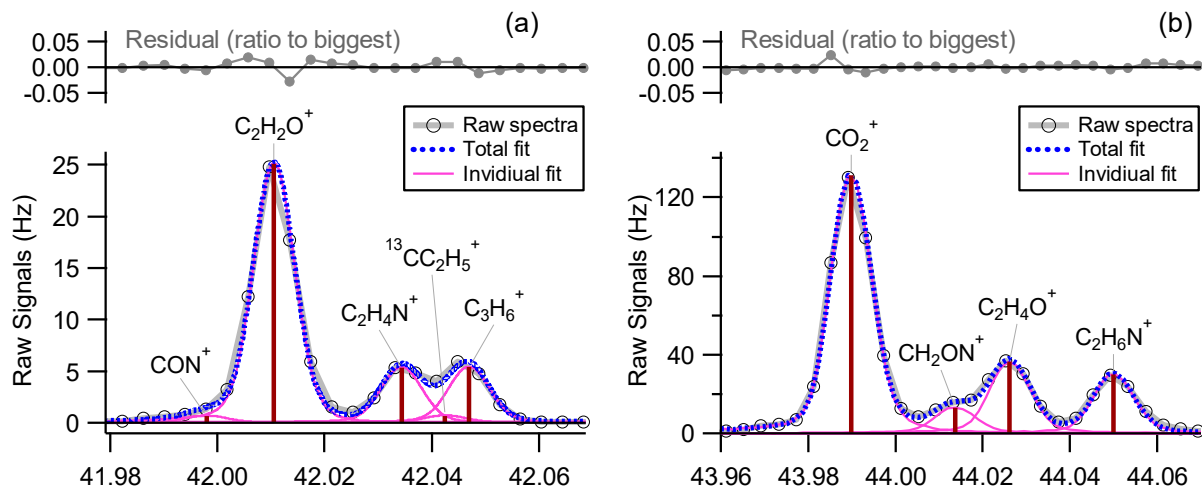
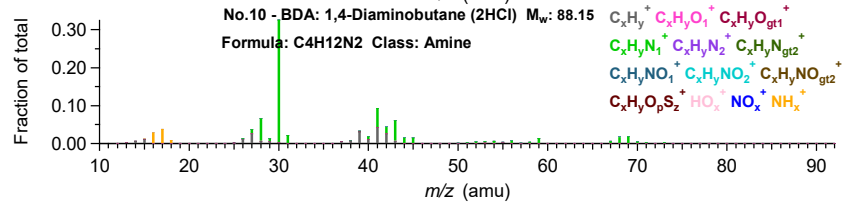
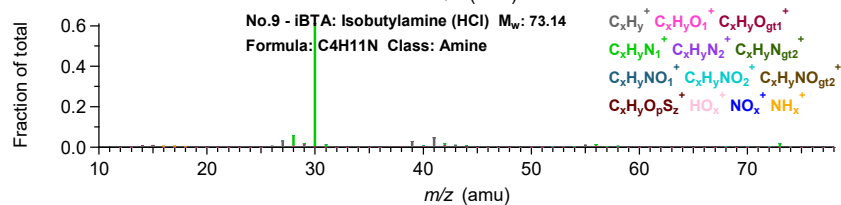
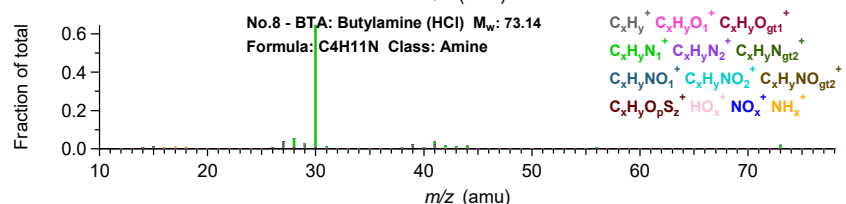
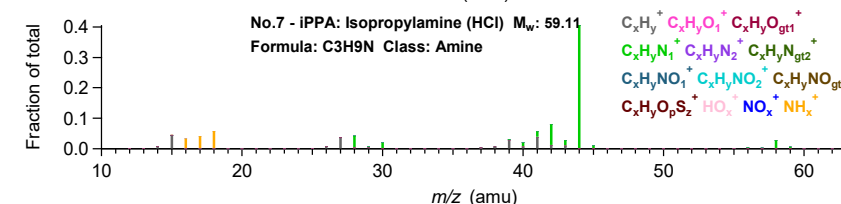
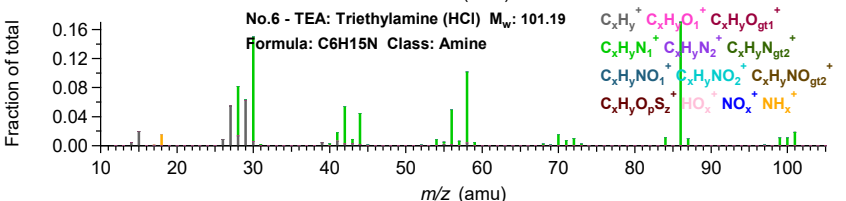
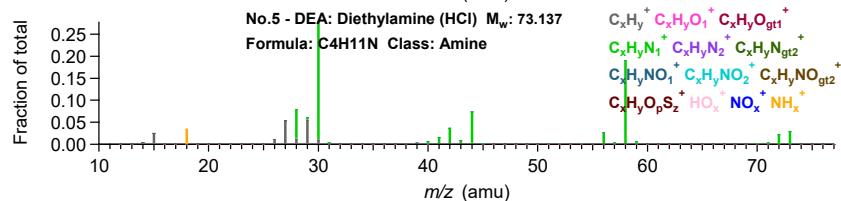
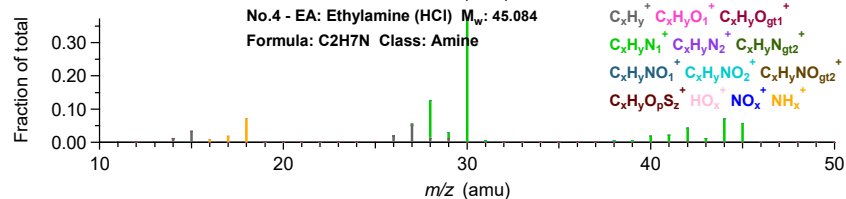
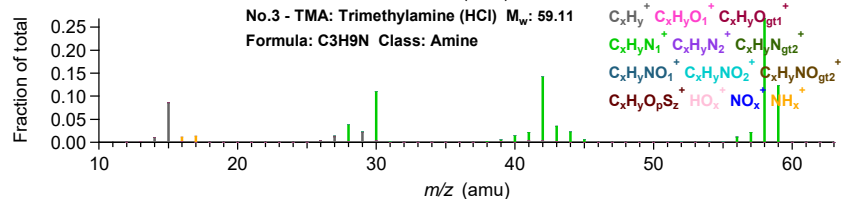
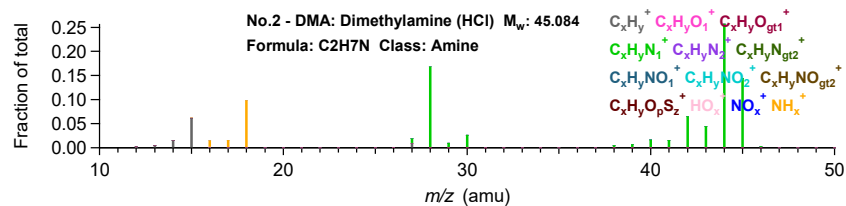
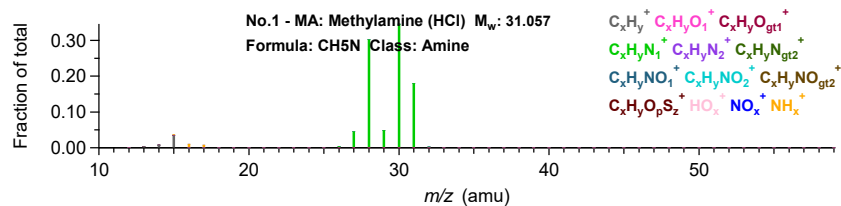
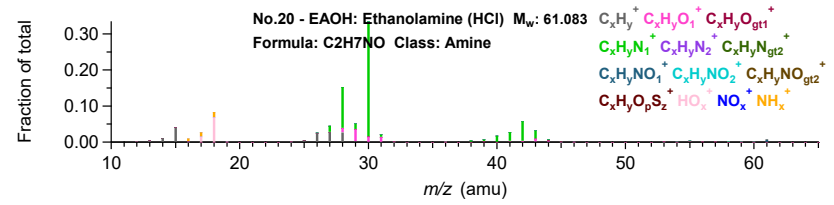
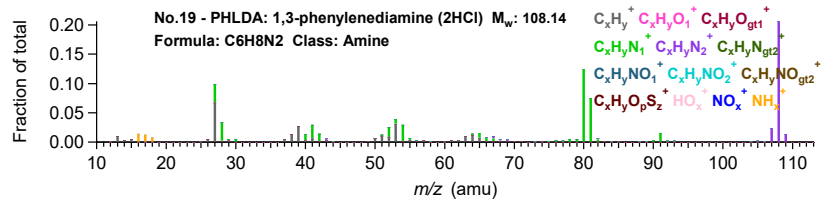
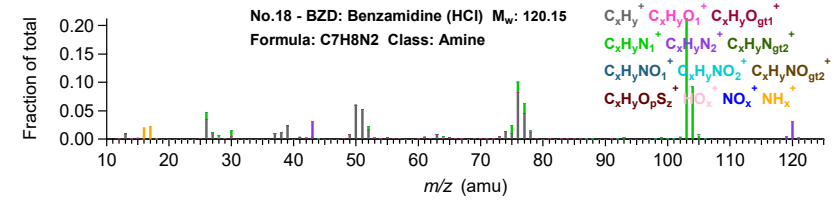
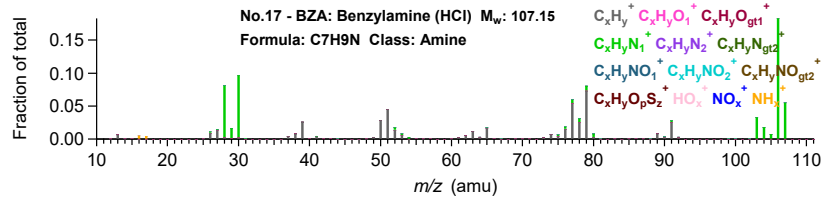
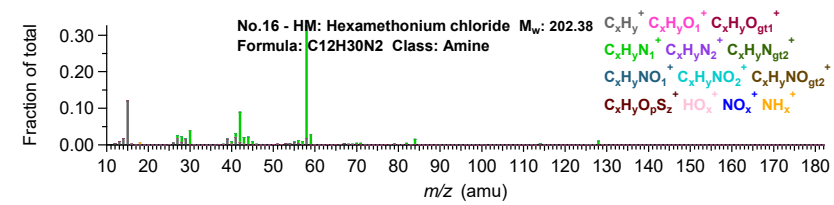
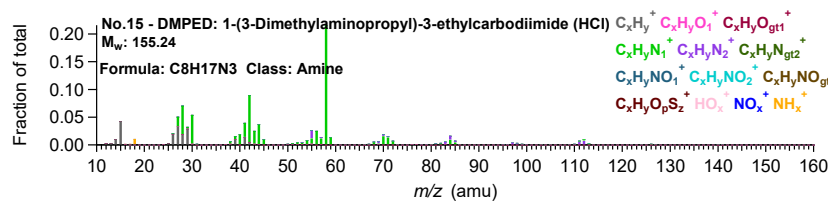
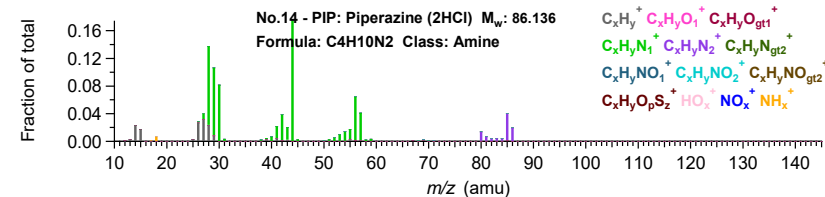
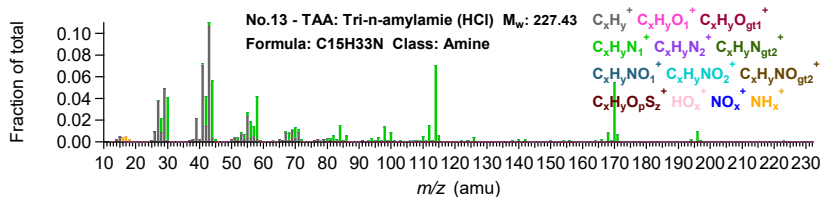
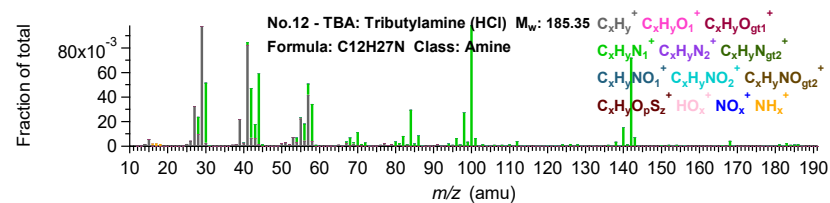
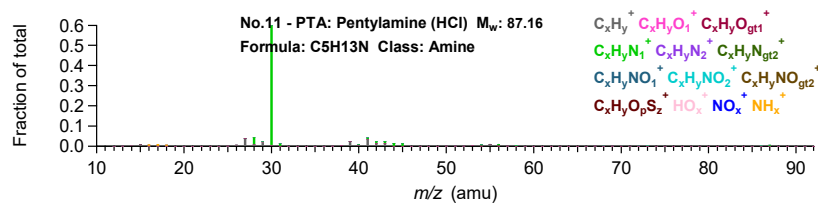
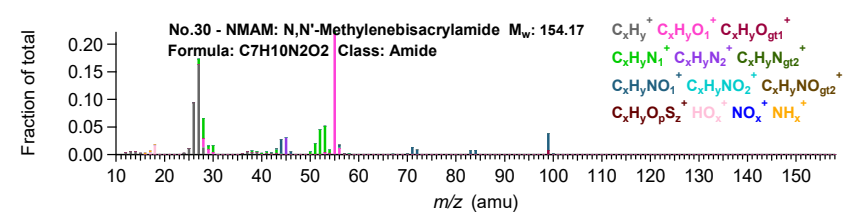
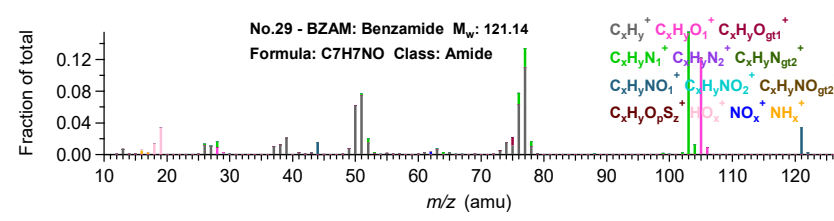
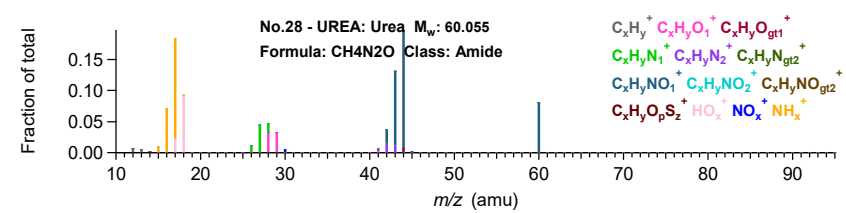
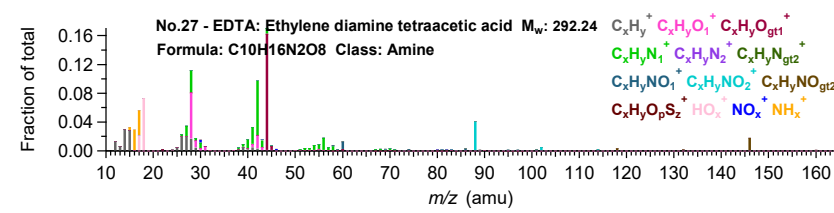
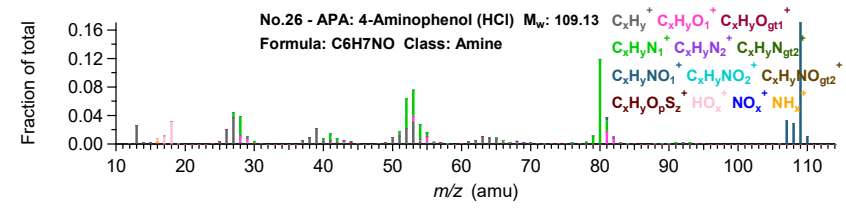
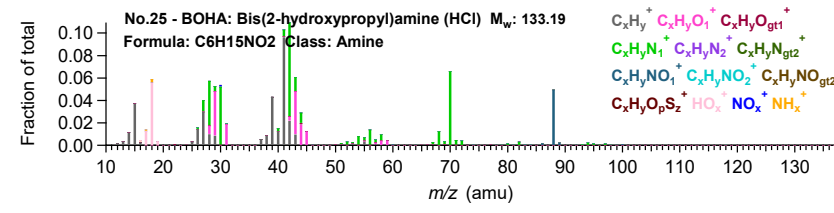
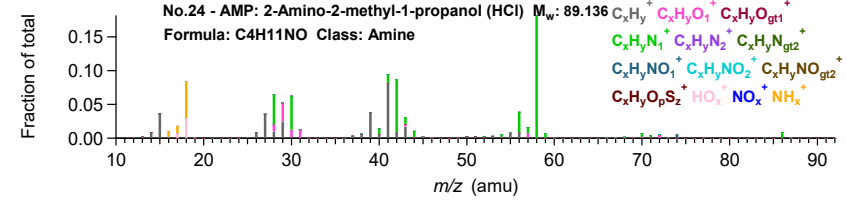
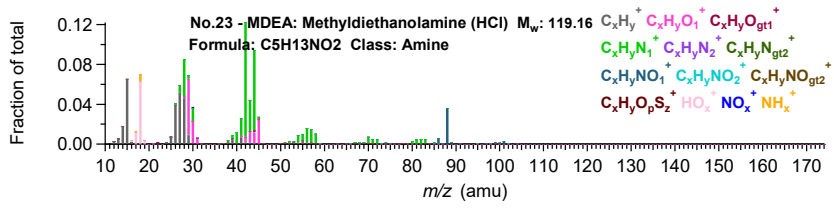
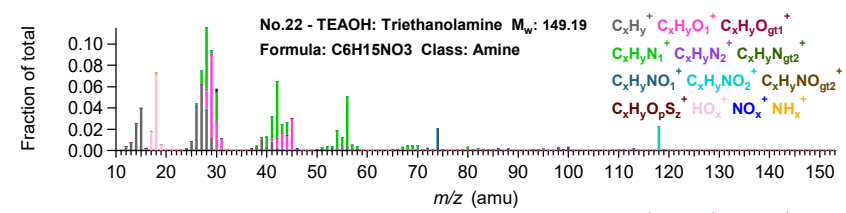
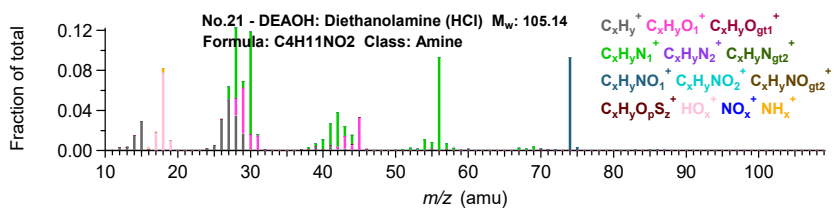
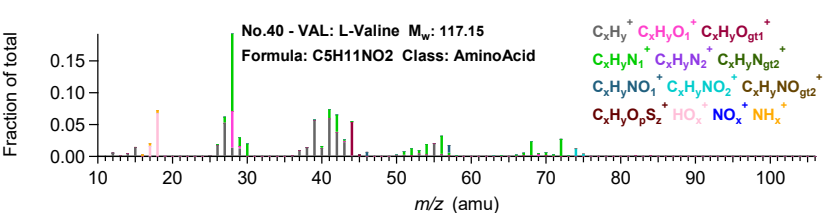
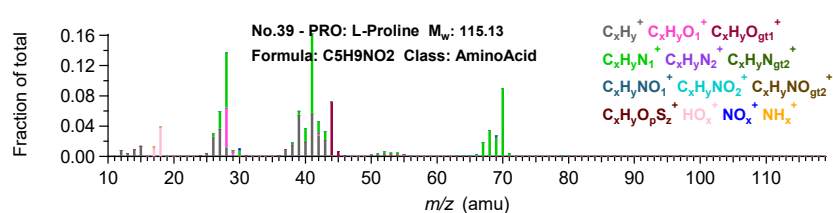
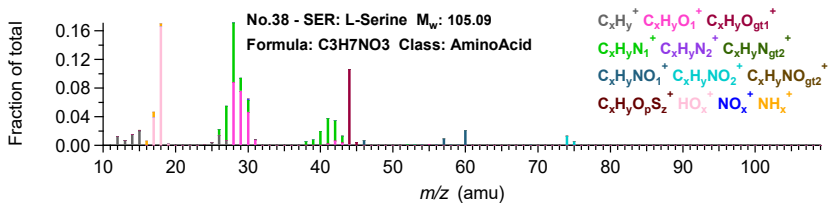
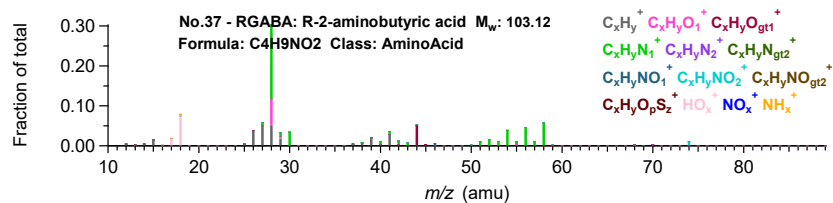
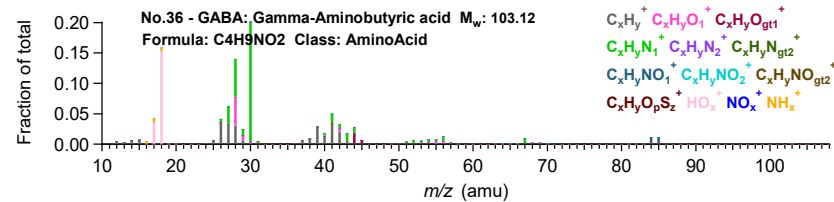
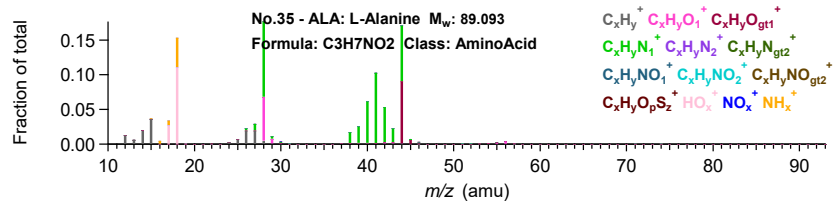
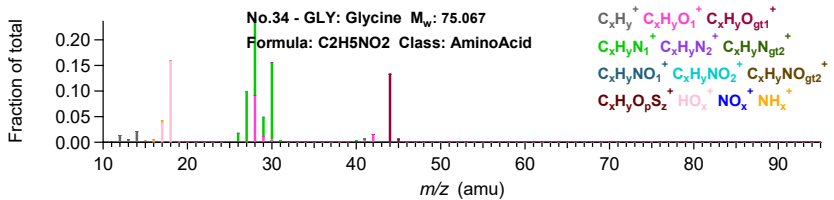
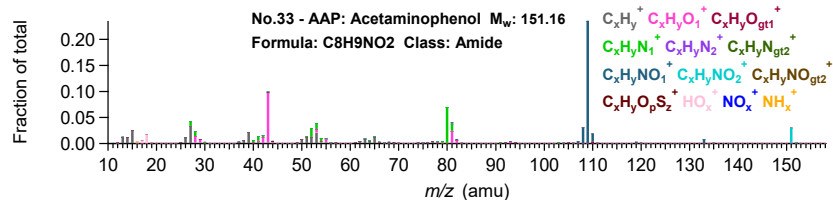
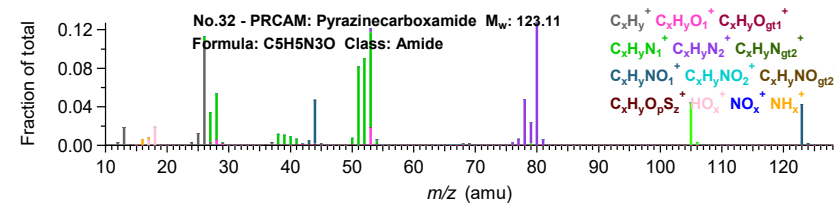
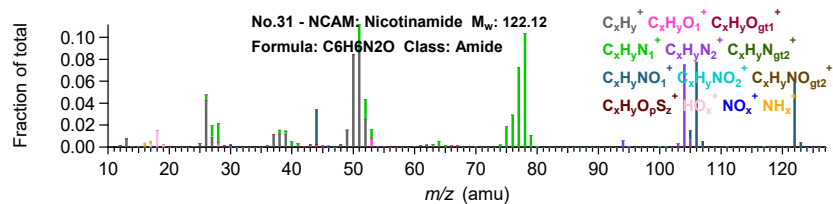


Figure S2. Examples of high resolution mass spectral analysis of (a)  $m/z$  42 and (b)  $m/z$  44.

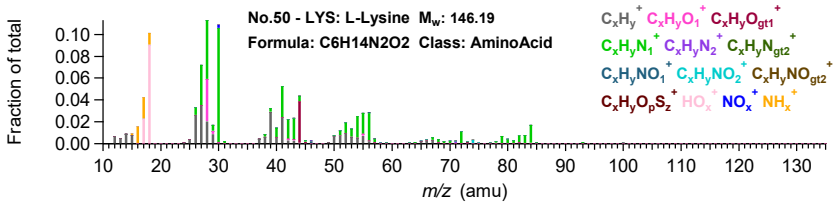
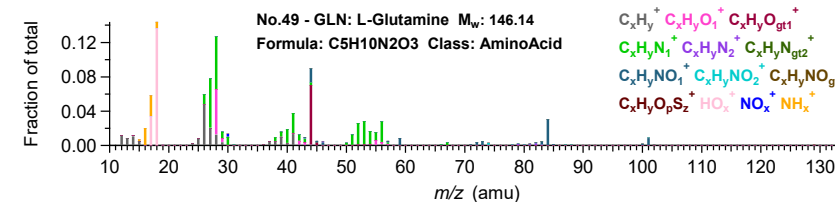
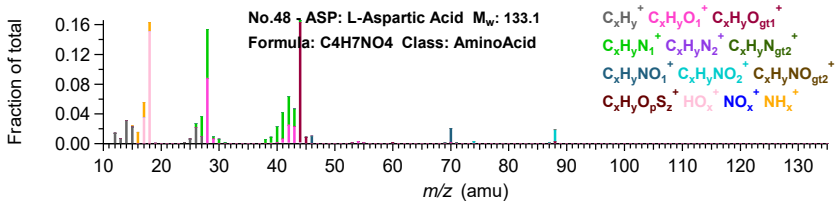
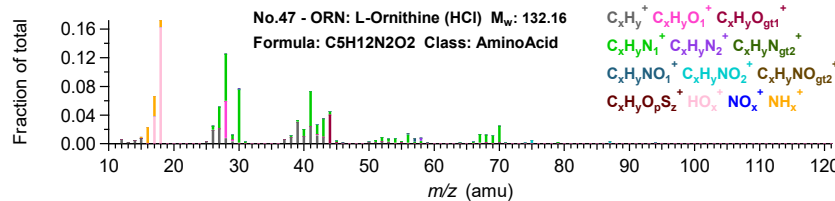
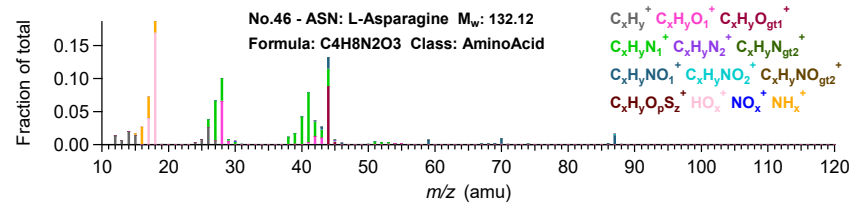
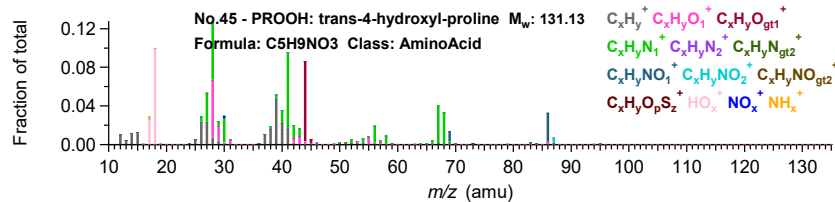
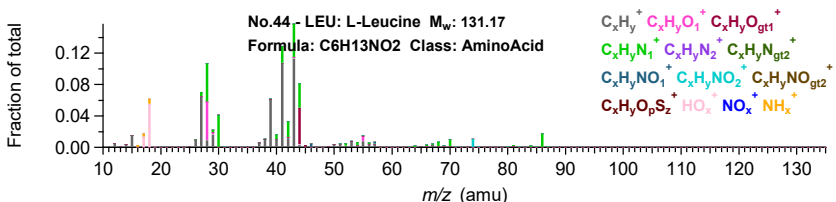
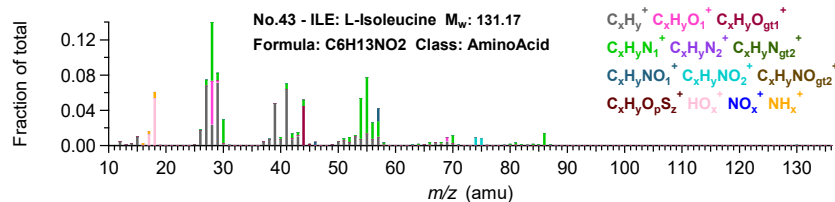
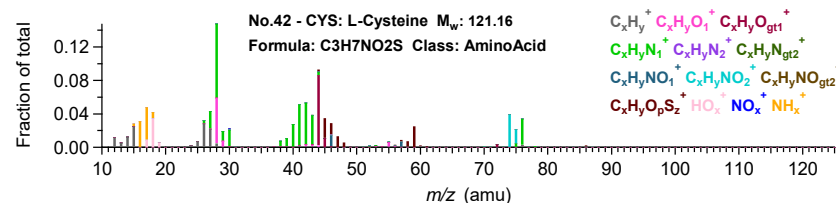
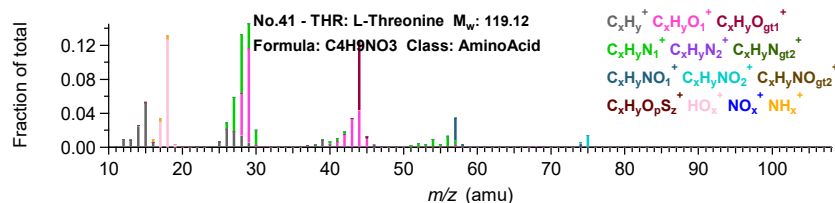




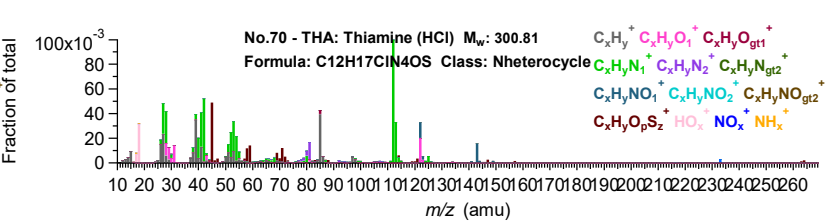
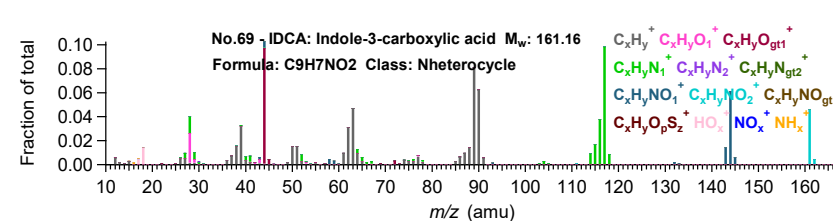
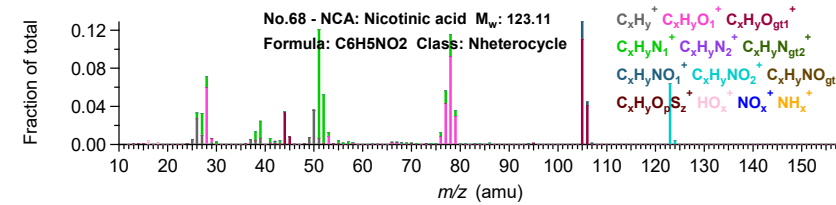
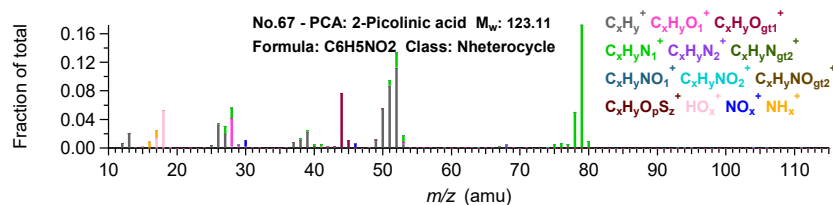
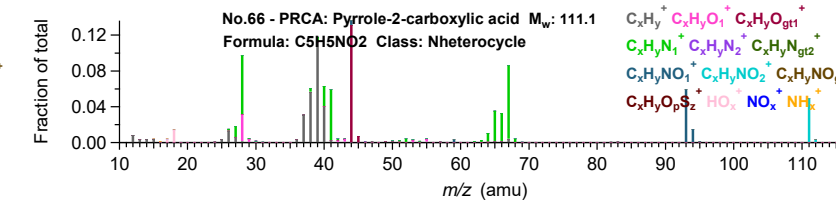
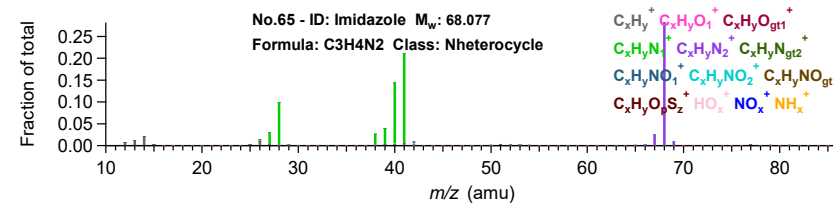
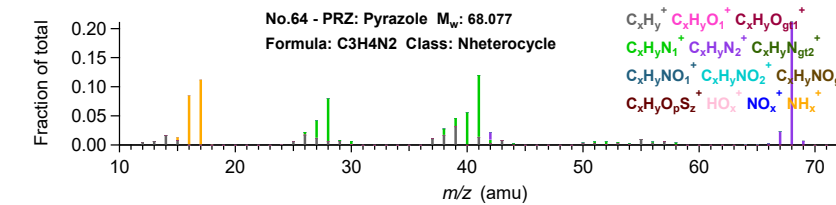
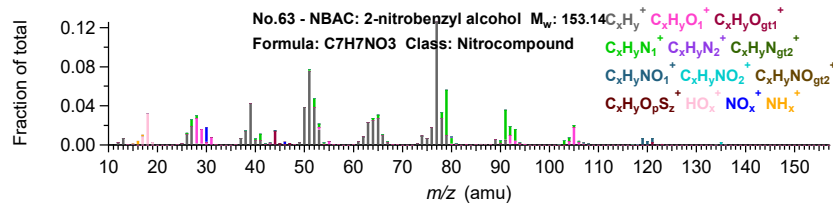
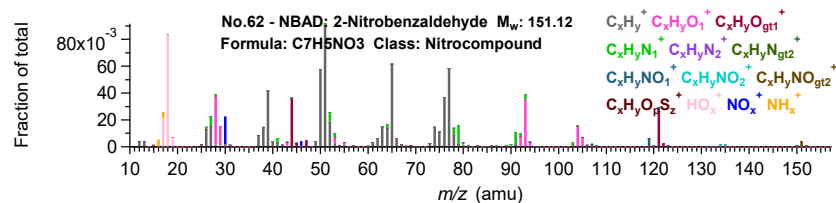
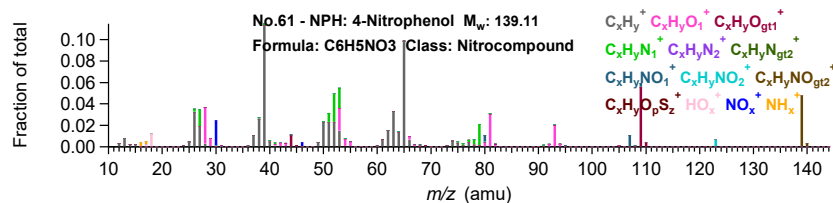












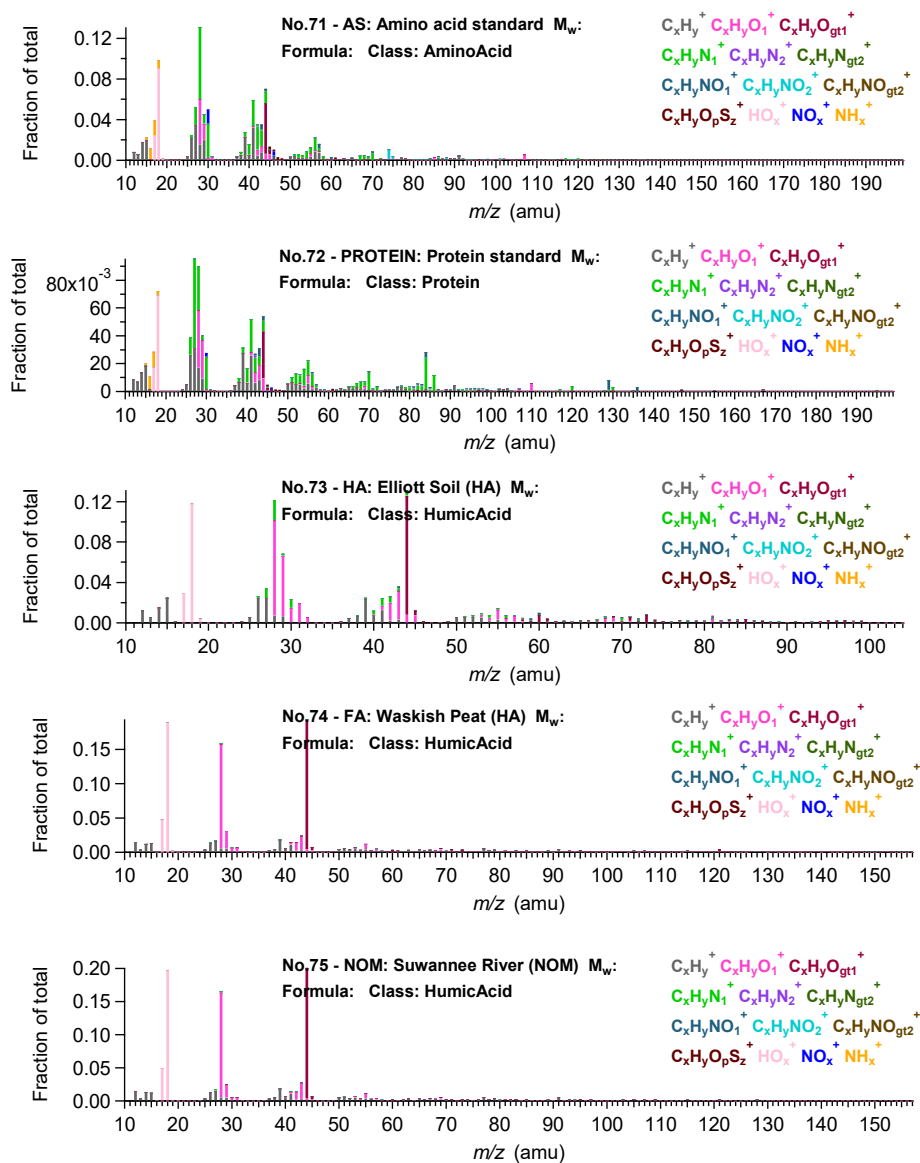


Figure S3. High resolution mass spectra of the 75 ON standards determined in this study.

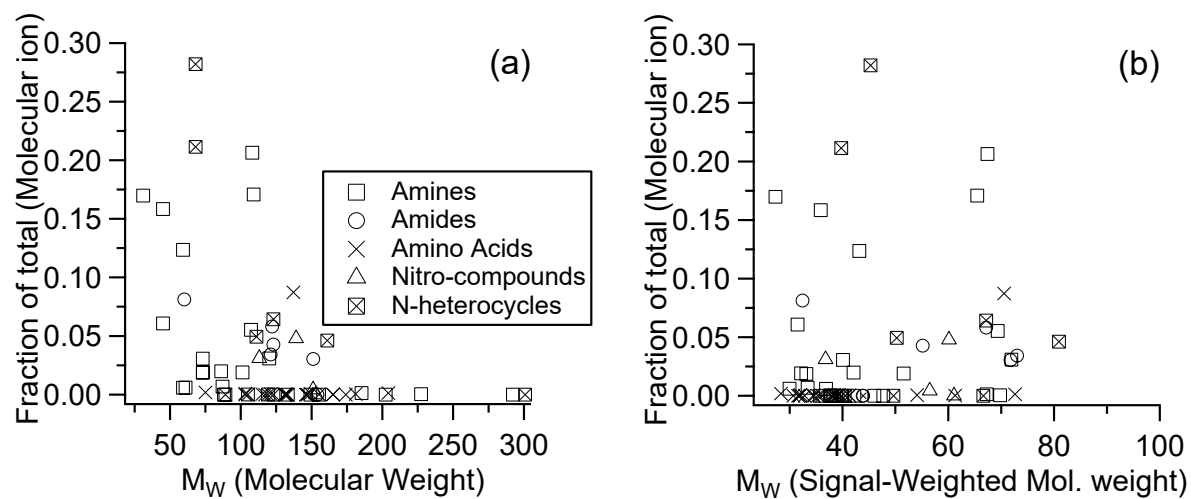


Figure S4. Mass fractions of the molecular ions in the HR-AMS spectra of the ON standards as a function of the molecular weights ( $M_w$ ) (a), and the signal-weighted molecular weights (b) (calculated as  $\sum f_i * i_{m/z}$ , where  $i_{m/z}$  is the exact mass of ion  $i$  and  $f_i$  is the mass fraction of that ion in the compound's spectrum).

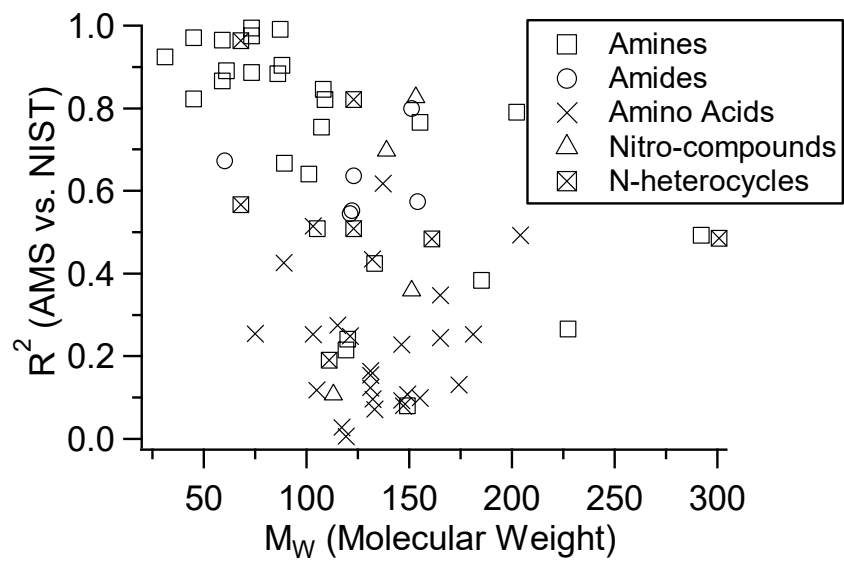


Figure S5. Correlation coefficients (Pearson's  $r^2$ ) between the HR-AMS spectra and the NIST spectra as a function of the molecular weights of the ON standards.

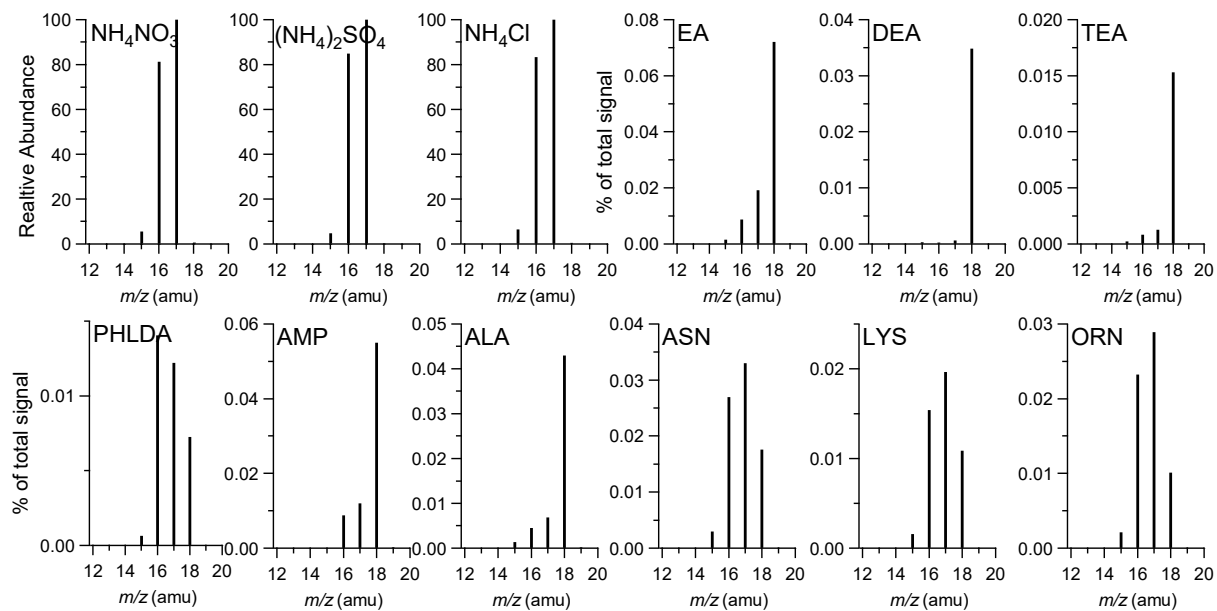


Figure S6. Mass fractional contributions of  $\text{NH}_x^+$  ion fragments in the HR-AMS spectra of ammonium nitrate, ammonium sulfate and ammonium chloride, and a few selected amines and amino acids.

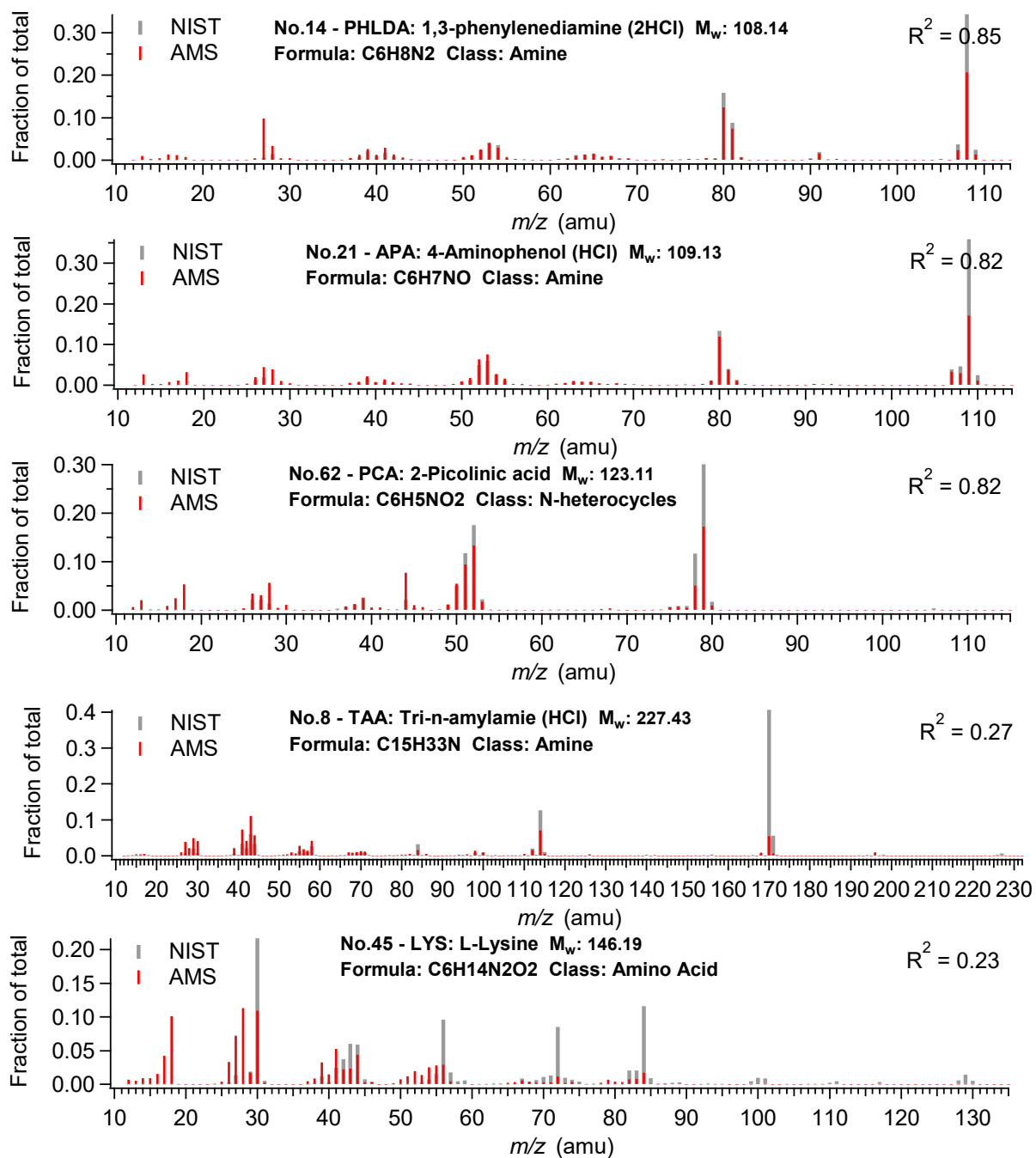


Figure S7. AMS spectra (summed in unit mass resolution from HR spectra) versus NIST spectra for a few selected ON standards.