

Influences of downward transport and photochemistry on surface ozone over East Antarctica during austral summer: in situ observations and model simulations

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Abstract. Studies of atmospheric trace gases in remote, pristine environments are critical for
20 assessing the accuracy of climate models and advancing our understanding of natural processes
and global changes. We investigated the surface ozone (O₃) variability over East Antarctica during
the austral summer of 2015–2017 by combining surface and balloon-borne measurements at the
Indian station Bharati (69.4° S, 76.2° E, ~35 m above mean sea level) with EMAC atmospheric
chemistry-climate model simulations. The model reproduced the observed surface O₃ level (18.8
25 ± 2.3 nmol mol⁻¹) with negligible bias and captured much of the variability (R=0.5). Model
simulated tropospheric O₃ profiles were in reasonable agreement with balloon-borne
measurements (mean bias: 2–12 nmol mol⁻¹). Our analysis of a stratospheric tracer in the model
showed that about 41–51% of surface O₃ over the entire Antarctic region was of stratospheric
origin. Events of enhanced O₃ (~4–10 nmol mol⁻¹) were investigated by combining O₃ vertical
30 profiles and air mass back trajectories, which revealed the rapid descent of O₃-rich air towards the
surface. The photochemical loss of O₃ through its photolysis (followed by H₂O+O(¹D)) and
reaction with hydroperoxyl radicals (O₃+HO₂) dominated over production from precursor gases
(NO+HO₂ and NO+CH₃O₂) resulting in overall net O₃ loss during the austral summer.
Interestingly, the east coastal region, including the Bharati station, tends to act as a stronger
35 chemical sink of O₃ (~190 pmol mol⁻¹ d⁻¹) than adjacent land and ocean regions (by ~100 pmol

mol⁻¹ d⁻¹). This is attributed to reverse latitudinal gradients between H₂O and O(¹D), whereby O₃ loss through photolysis (H₂O+O(¹D)) reaches a maximum over the east coast. Further, the net photochemical loss at the surface is counterbalanced by downward O₃ fluxes, maintaining the observed O₃ levels. The O₃ diurnal variability of ~1.5 nmol mol⁻¹ was a manifestation of combined effects of mesoscale wind changes and up- and downdrafts, in addition to the net photochemical loss. The study provides valuable insights into the intertwined dynamical and chemical processes governing the O₃ levels and variability over East Antarctica.

1 Introduction

Tropospheric ozone (O₃) plays a pivotal role in governing the atmospheric oxidation capacity, and influences air quality and climate warming (Seinfeld and Pandis, 1998). The major source of O₃ in the troposphere is its photochemical formation involving precursors such as nitrogen oxides (NO_x), carbon monoxide (CO) and non-methane hydrocarbons (NMHCs; Lelieveld and Dentener, 2000). The contribution of downward transport from the stratosphere is generally minor near the surface, although it can be significant at middle to high latitudes (Stohl et al., 2003). Numerous studies have investigated the chemistry and dynamics of tropospheric O₃ and the roles of local to synoptic-scale processes (e.g., boundary layer height variation, horizontal and vertical transport, etc.; Nguyen et al., 2022; Young et al., 2018). Investigations of O₃ variations in remote pristine environments, isolated from major anthropogenic influences, are essential to understand the global changes in atmospheric composition, the role of natural processes including downward transport from the stratosphere, and photo-denitrification of the snowpack (Jones et al., 2001). In this regard, the observations over environments such as Antarctica are extremely valuable and can provide insights into the global background atmosphere, besides providing data to test the results of chemistry-climate models. The mean surface O₃ over the Antarctic region was observed to be lower by nearly 5 nmol mol⁻¹ than that over the Arctic polar region (Helmig et al., 2007). Surface O₃ shows a pronounced seasonality (~15–20 nmol mol⁻¹ amplitude) with a summer minimum and a winter maximum over Antarctica, accompanied by periodic fluctuations associated with long-range transport (Kumar et al., 2021; Legrand et al., 2016; Oltmans and Komhyr, 1976; Winkler et al., 1992). In line with global increases in tropospheric O₃ due to the enhanced anthropogenic emissions since the pre-industrial era and impacts of climate warming (Wang et al., 2022; Murazaki and Hess, 2006; Lelieveld et al., 2004), a positive trend (0.08–0.13

nmol mol⁻¹ y⁻¹ over Syowa, Arrival Heights, Neumayer, and South Pole) in surface O₃ has also been reported from Antarctica (Kumar et al., 2021).

Previous studies have investigated the long-term, inter-annual, seasonal and diurnal variations in surface O₃ over Antarctica (Legrand et al., 2009, 2016), as well as the role of horizontal transport (Tian et al., 2022) and chemistry including that of radicals (Preunkert et al., 2012), halogen driven O₃ depletion (Tarasick and Bottenheim, 2002; Jones et al., 2013), and stratospheric intrusions (Das et al., 2020). Antarctic observations have provided evidence of widespread O₃ production during austral spring and summertime, affecting all stations through horizontal mixing. This O₃ production contributes to a significant enhancement in annual mean O₃ over the Antarctic plateau (Helmig et al., 2007). While a weak coupling between stratospheric and tropospheric O₃ was inferred earlier (Oltmans and Komhyr, 1976), frequent stratospheric intrusions in this region were also reported (Cristofanelli et al., 2018; Das et al., 2020; Greenslade et al., 2017). There have been extensive studies on a range of species utilizing datasets from dedicated campaigns and projects over West Antarctica and South Pole (CHABLIS—Chemistry of the Antarctic Boundary Layer and the Interface with Snow, Jones et al., 2008; ISCAT—Investigation of Sulfur Chemistry in Antarctica, Davis et al., 2004; ANTCI—Antarctic Tropospheric Chemistry Investigation (Eisele et al., 2008); WAIS—West Antarctic Ice Sheet, Frey et al., 2005; Masclin et al., 2013). The variability of volatile organic compounds (VOCs), radicals, O₃ and its precursors has been investigated over the East Antarctic plateau and eastern coastal Antarctica (OPALE—Oxidant Production over Antarctic Land and its Export; Preunkert et al., 2012 and references therein). But the east coast of Antarctica remains a relatively less explored region as compared to west Antarctica and the South Pole.

The east coast of Antarctica is distinct from the west coast as well as the inland region of Antarctica. Relatively high levels of hydroxyl and peroxy radicals over eastern Antarctica (Dumont d'Urville; 66.67° S, 140.02° E, 40 m above mean sea level—amsl) during austral summer (Kukui et al., 2012) indicate chemical differences from west Antarctica (Palmer; 64.77° S; 64.05° W) where radical concentrations are lower. Short-term events of O₃ enhancements are observed over the coastal as well as inland regions with higher frequency during the summer season and they are associated with ultraviolet radiation reaching the surface, photochemical production and transport (Crawford et al., 2001; Frey et al., 2015; Cristofanelli et al., 2018; Legrand et al., 2016).

Net summertime O₃ production (4–5 nmol mol⁻¹ d⁻¹) has been observed at the eastern coastal Antarctica through NO_x emission from snow (Legrand et al., 2009, 2016). In contrast, surface or boundary layer O₃ depletion are also observed mainly due to halogen chemistry involving iodine, bromine and chlorine oxides, being more frequent in west Antarctica (Saiz-Lopez et al., 2007; Simpson et al., 2007). Weaker or less frequent surface O₃ depletion is observed over the east coast compared to the west coast of Antarctica (Jones et al., 2013; Legrand et al., 2016).

Most studies of East Antarctica have been based on in situ measurements of various trace gases including radical species (O₃, NO, HONO, OH, DMS, BrO, etc.) at Dumont d'Urville, Syowa (69.00° S; 39.58° E, ~29 m amsl) and Zhongshan (69.37° S, 76.36° E, 18.5 m amsl; Kukui et al., 2012; Legrand et al., 2016, 2009; Murayama et al., 1992; Tian et al., 2022) stations. These studies have shown the surface O₃ variability on different scales (i.e., diurnal: ~2 nmol mol⁻¹, seasonal: ~18 nmol mol⁻¹, and long-term trend: 0.07±0.07 nmol mol⁻¹ y⁻¹). Only few studies have analyzed the relevant larger-scale trace gas distributions and discussed the model performance of seasonal changes in surface O₃ or tropospheric O₃ (Wang et al., 2022; Griffiths et al., 2021), including halogen chemistry (Yang et al., 2005; Fernandez et al., 2019). Studies investigating the chemistry and dynamics of surface O₃ are scarce for Antarctica (Morgenstern et al., 2013). To the best of our knowledge, there are no comprehensive studies discussing the surface O₃ variability and associated processes based on the synergy of in situ measurements and chemistry-climate modeling over East Antarctica. It is timely to investigate the underlying processes since an increasing O₃ trend has been reported over this part of the world recently (Kumar et al., 2021).

Our study aims to contribute to the understanding of chemical and dynamical processes governing the surface O₃ variability over the east coast of Antarctica. We have conducted in-situ measurements during three different years and performed simulations using a global chemistry-climate model to unravel the atmospheric processes that control the summertime O₃ levels and variability. Details of the measurements and model simulations are given in the next section. Results of the O₃ variability and a comparison of model results with measurements, and an analysis of photochemical and dynamical contributions are presented in section 3. A summary, the main conclusions, and a future outlook are presented in section 4.

125 2 Measurements and model simulations

2.1 In situ measurements

Surface O₃ was measured at the Indian station Bharati (69.4° S, 76.2° E, ~35 m above mean sea level) at the Larsemann Hills in the east coast of Antarctica during the summer seasons of three years 2015–2017: 29 January–13 February 2015, 17 January–24 February 2016, and 11 December
130 2016–16 February 2017. The Bharati site experiences a surface pressure of ~980±10 hPa, cold temperatures (-0.1±3 °C; -11–8 °C), moderate humidity (60±13.5%; 34–98%) and mainly easterly winds with a number of blizzards during the summer season. A detailed overview of the meteorological conditions at Bharati station can be found in Soni et al., 2017.

Surface O₃ mixing ratios were measured using an online ultraviolet photometric ozone
135 analyzer manufactured by the Environnement S.A, France (model O342). The instrument derives O₃ mixing ratios using the Beer–Lambert law considering the absorption of ultraviolet radiation around 253.7 nm by O₃ molecules. The measurement noise, lower detection limits, linearity and minimum response time are 0.5 nmol mol⁻¹, 1 nmol mol⁻¹, ±1 % and 10 s, respectively. The instrument was operated on the auto-response mode (response time of 10–90 s) under permissible
140 range of temperature. O₃ mixing ratios were recorded continuously at 5 min averaging intervals. Air samples were drawn from a height of approximately 2 m above the ground level through a Teflon tube and filtered through 5 µm non-reactive polytetrafluoroethylene dust filter prior to injection into the analyzer. Prior to each expedition, the analyzer was calibrated for mixing ratios of 20 and 30 nmol mol⁻¹ using a multichannel calibrator. The measurement uncertainty is estimated
145 to be ~5 % (Tanimoto et al., 2007). In addition to measurements at Bharati, surface O₃ at Syowa and Arrival Heights (77.80° S; 166.67° E) available from <https://ebas-data.nilu.no/Default.aspx> (last access: 1 January 2024) for the study period are also used for the comparison of model results.

The vertical profiles of O₃ partial pressure were measured using balloon-borne electrochemical ozonesondes manufactured by the En-Sci Corporation, USA (Model: 2Z-V7). A
150 total of 12 profiles were measured during the study period. The O₃ partial pressure was converted to O₃ mixing ratios using the simultaneously measured atmospheric pressure by radiosonde (model: iMet-1-RSB). Air is passed through an electrochemical concentration cell (ECC) using a built-in non-reactive pump, and the current generated by the electrochemical reaction of O₃ (with

potassium iodide) is measured by an electronic interface board and converted into an O₃ partial
155 pressure. The detailed operation principle and performance evaluation of ozonesonde instrument
are described in Komhyr et al., 1995 and references therein. The accuracy of O₃ measurements is
reported to be 5–10% up to an altitude of 30 km (Smit et al., 2007). Additional details of the O₃
measurements and meteorological parameters using this technique can be found elsewhere
(Ajayakumar et al., 2019; Ojha et al., 2014). Besides our measurements at Bharati, we utilized
160 available O₃ vertical profiles measured using ECC ozonesondes at Davis station (68.58° S 77.97°
E; https://data.aad.gov.au/metadata/records/AAS_4293_Ozonesonde, last access: 5 December
2023) in this study.

The surface level wind speed and direction were measured using an automatic weather
station, which meets the standards of the World Meteorological Organization and was operated by
165 the India Meteorological Department. Wind direction measurements are used here to analyze the
changes in surface O₃ on a diurnal time scale. To understand the impacts of updraft and downdrafts,
the vertical wind at the surface was measured using a fast response ultrasonic anemometer (make:
METEK, GmbH, Germany; model: USA-1 Scientific). The factory calibrated sensor was mounted
at 3 m level above the ground and was operated at 25 Hz during January 2016. The measuring
170 resolution and accuracy of the vertical velocity are $\pm 0.01 \text{ m s}^{-1}$ and 0.2 m s^{-1} , respectively. Further
details on the instrument can be found in Reddy et al. (2021).

2.2 Model simulations

In this work the EMAC (ECHAM5/MESSy Atmospheric Chemistry) model (Jöckel et al.,
2010, 2006) has been used. This model is a numerical chemistry and climate simulation system
175 that includes sub-models describing tropospheric and middle atmospheric processes and their
interaction with oceans, land and human influences. It uses the second version of the Modular
Earth Submodel System (MESSy2) to link multi-institutional computer codes. The core
atmospheric model is the 5th generation European Centre Hamburg general circulation model
(ECHAM5, Roeckner et al., 2006). The physics subroutines of the original ECHAM code have
180 been modularized and reimplemented as MESSy submodels and have continuously been further
developed. Only the spectral transform core, the flux-form semi-Lagrangian large scale advection
scheme, and the nudging routines for Newtonian relaxation are remaining from ECHAM5. For the

present study we applied EMAC (MESSy version 2.55.0) in the T106L47MA-resolution, i.e. with a spherical truncation of T106 (corresponding to a quadratic Gaussian grid of approximately $1.1^\circ \times 1.1^\circ$ in latitude and longitude) with 47 vertical hybrid pressure levels up to 0.01 hPa. In this work we used the same set up as in Reifenberg et al. (2022), and the model results encompass the years 2014-2018 with a 3-hours output frequency. Global atmospheric chemistry models are known to overestimate tropospheric ozone (Young et al., 2018), and EMAC is no exception to this. Nevertheless, extensive ozone evaluation (Jöckel et al., 2016) shows that the EMAC model has a very low (less than 10%) or no bias in the troposphere against observations for latitudes below 60° S. Furthermore, the EMAC model has been extensively evaluated in the last years both for the gas phase (Jöckel et al., 2016; Taraborrelli et al., 2021) and for the aerosol phase (Pozzer et al., 2012; Brühl et al., 2018; Pozzer et al., 2022).

The model includes emissions of bromine from sea spray following the approach of Kerkweg et al. (2008), and important heterogeneous reactions involving bromine (e.g., liquid phase reactions of $\text{HOBr} + \text{HBr} \rightarrow \text{Br}_2 + \text{H}_2\text{O}$) are included via the AERCHEM subroutines (Rosanka et al., 2023) in the GmXe submodel (Pringle et al., 2010). With the ONLEM submodel, the air-snow subroutines are activated (Falk and Sinnhuber, 2018), which include the bromine release on sea-ice and snow-covered surface, based on the scheme of Toyota et al. (2011). Beside the bromine release, no NO_x release is included by the deposition of O_3 . Note that NO_x and HONO emissions from snowpack (Honrath et al., 2002; Bond et al., 2023) are not incorporated in the model.

To investigate the effects of transport, air-mass back trajectories have been computed using the HYSPLIT (HYbrid Single Particle Lagrangian Integrated Trajectory) model version-4 (Rolph et al., 2017; Stein et al., 2015) with the input of $1^\circ \times 1^\circ$ gridded GDAS (Global Data Assimilation System) meteorological data.

3 Results and Discussions

3.1 O_3 variability: comparison of observations with model simulations

Figure 1a shows the elevation map of Antarctica marked with the location of the Indian station Bharati (69.4° S, 76.2° E, ~ 35 m amsl), where surface-based and balloon-borne measurements of O_3 have been conducted during this study. The surface elevation is higher (up to

4 km) over the eastern part of Antarctica. Figure 1b shows the spatial distribution of surface O₃ during summer of 2015–2017 (29 January–13 February 2015, 17 January–24 February 2016, and 11 December 2016–16 February 2017) as simulated by the EMAC model, along with the mean observed value at Bharati station ($18.8 \pm 2.3 \text{ nmol mol}^{-1}$). The mean O₃ distribution shows increase
215 from the oceanic region (10–16 nmol mol⁻¹) to the landmass (15–24 nmol mol⁻¹), nearly following the topographical features of Antarctica. Overall, the model simulated spatial distribution of O₃ (Fig. 1b) is seen to be in agreement with the distribution based on measurements from different stations (Fig. S1). This is further consistent with previous studies showing higher O₃ mixing ratios over elevated site (South Pole; 2830 m amsl) as compared to the coastal/oceanic region (Helmig
220 et al., 2007). The balloon-borne observations (Fig. 3) also show increase in mean O₃ mixing ratios with altitude.

Figure 1c shows the stratospheric contribution (in percent) to the surface O₃ based on the stratospheric O₃ tracer in the model (O_{3s}). O_{3s} is seen to contribute by 41–51% over the Antarctic region with greater contribution (45–51%) over the continent with higher elevation than that over
225 the surrounding ocean (41–45%). The mean stratospheric contribution at the observation site Bharati is estimated to be ~48% (~10 nmol mol⁻¹), showing that nearly half of O₃ at the surface is of the stratospheric origin. Mihalikova and Kirkwood, 2013 have estimated a 6–7% occurrence rate of tropospheric folds (1-2 folds month⁻¹) during summer using radar observations at Troll station (72.0° S, 2.5° E, 1275 m amsl). In another recent study also, the enhancement by 20–30
230 nmol mol⁻¹ (67–100% as compared to the climatological mean) is seen in upper tropospheric O₃ above Bharati station due to stratospheric intrusions (Das et al., 2020). In addition, gradual subsidence through the tropopause also contribute to stratospheric ozone transport into the troposphere. Therefore, stratospheric intrusions are suggested to transport the O₃-rich airmasses to the troposphere, which subsequently descend to the surface and get redistributed across the region
235 through horizontal transport. Descent of O₃-rich airmasses is further discussed in section 3.2.

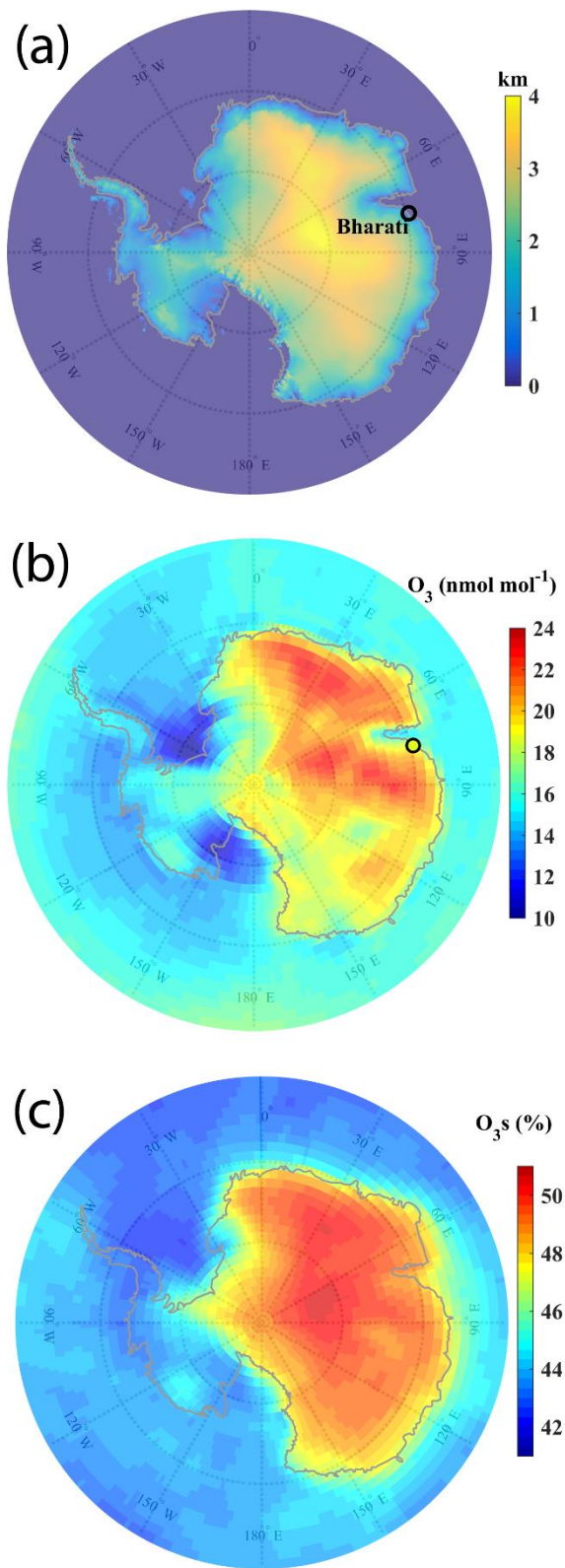


Figure 1. (a) Elevation map of Antarctica along with the location of the Indian station Bharati marked by a black circle. (b) Spatial distribution of surface O₃ simulated by the EMAC model,

averaged over the study period (29 January–13 February 2015, 17 January–24 February 2016, and
 240 11 December 2016–16 February 2017). Colour in the black circle in (b) represents the mean value
 from the in situ measurements at Bharati. (c) Percent contribution of stratospheric O₃ to the surface
 O₃ derived from the EMAC model during the study period.

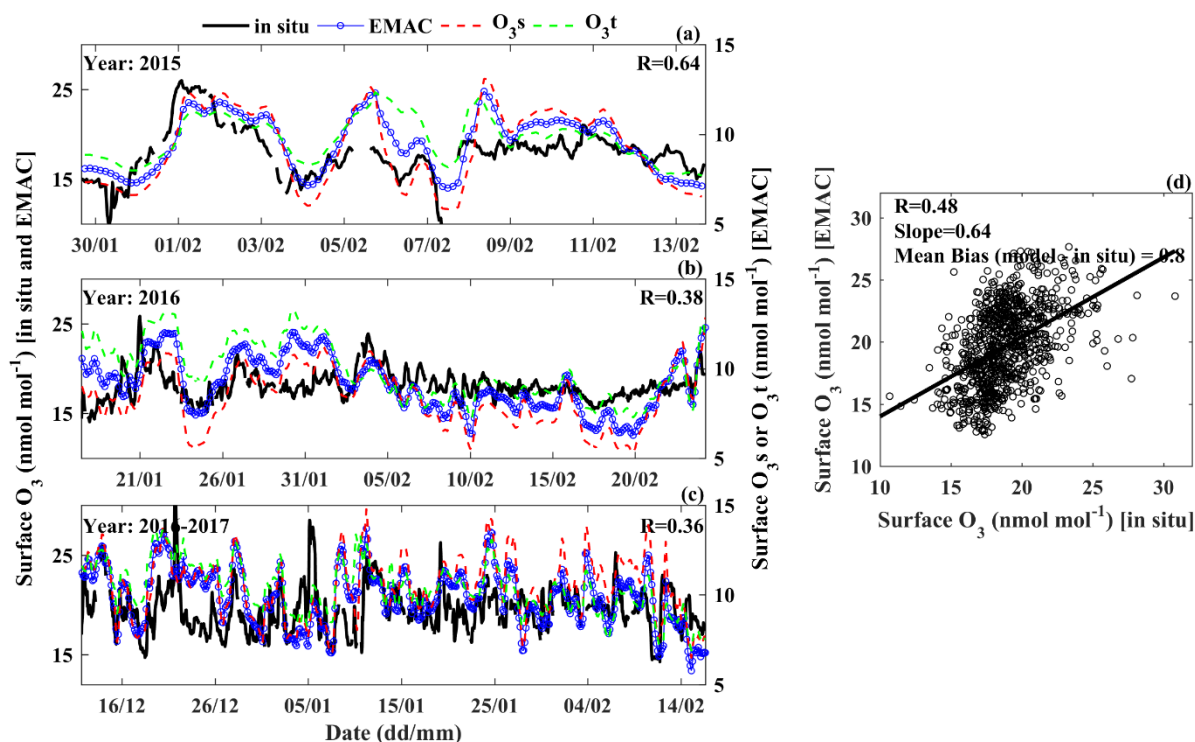


Figure 2. Variability in surface O₃ (a-c) at Bharati during austral summer of 2015–2017 based on
 245 in situ measurements (black) and EMAC simulations (blue). Green and red curves show the
 absolute stratospheric (O₃s) and tropospheric (O₃t) contributions to the surface O₃. A scatter plot
 between in situ measurements and model simulated O₃ is shown in (d). Note that O₃s and O₃t are
 on a different scale on right y-axis in (a-c).

Figure 2a–c shows the variations in surface O₃ at Bharati station from in situ measurements
 250 and model simulations during the summer seasons of 2015–2017. The mean O₃ levels estimated
 from the model simulations (19.7 ± 3.2 nmol mol⁻¹) are in very good agreement with the
 measurements (18.8 ± 2.3) with negligible bias (~ 1 nmol mol⁻¹) at this station. Further, the surface
 O₃ level at Bharati is observed to be similar to an earlier observation (~ 13 – 20 nmol mol⁻¹) at this
 station (Ali et al., 2017) and also to other stations in the coastal region of East Antarctica (Fig. S1).
 255 The model tends to successfully capture several features of the observed variability (Fig. 2a–c),
 nevertheless the overall correlation coefficient is 0.48 (Fig. 2d). The comparison for two other

coastal stations, Syowa and Arrival Heights during the same study period also shows that the model can reproduce the summertime O₃ levels with small bias and the temporal variability moderately well (Fig. S2–3). The blue and green curves in Fig. 2 show the individual contributions from stratospheric (O_{3s}) and tropospheric sources (O_{3t}=O₃ minus O_{3s}), respectively. Both stratospheric and tropospheric sources are estimated to be contributing nearly equally, 48% and 52%, respectively. Further, the stratospheric and tropospheric O₃ at the surface are seen to be strongly correlated (R=0.9; figure not shown) over most of the region mainly due to the mixing of stratospheric and tropospheric airmasses during the transport from the tropopause to the surface. Direct transport of stratospheric air or local O₃ production would decrease the correlation or perturb the variations in O_{3s} and O_{3t}. Overall, similar variability of comparable magnitude in O_{3s} and O_{3t} indicate the absence of strong “local” production or “direct” stratospheric transport to the surface. However, about 50% stratospheric contribution to surface O₃ points to significant stratospheric intrusions over the Antarctic region.

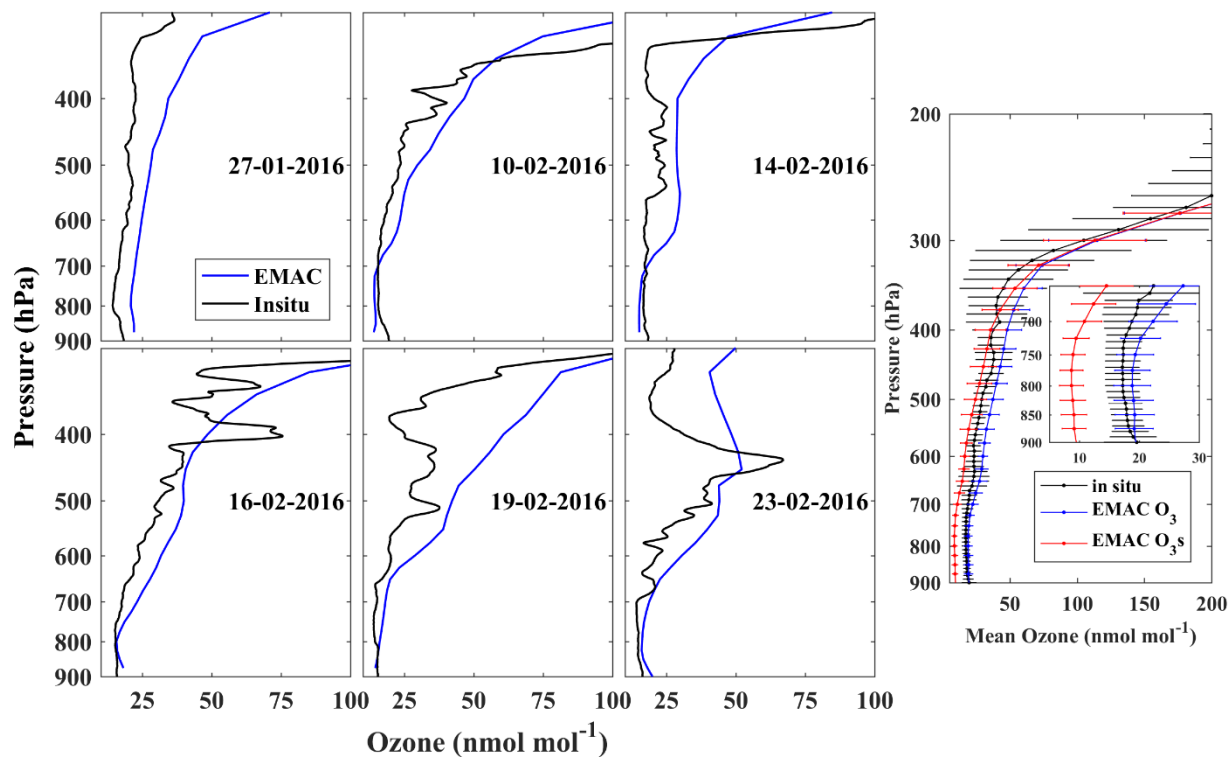


Figure 3. Vertical profiles of O₃ over Bharati station during a few representative days, based on the in situ measurements (black) and EMAC simulation (blue). The insert on the right shows the mean vertical distribution of O₃ and O_{3s} (red) corresponding to 12 profiles during the study period.

Figure 3 shows the comparison of balloon-borne observations of O₃ vertical profiles with
275 model simulation over Bharati station in 2016. Out of twelve, six individual representative profiles
are shown in the figure. O₃ mixing ratios gradually increase with altitude up to the tropopause
(~8.5 km; ~300 hPa), showing also O₃ peaks in the middle/upper troposphere during some days.
The model successfully captures the mean vertical distribution especially in the lower troposphere
(pressure > ~700 hPa) with a mean bias of less than 2 nmol mol⁻¹. There is an agreement between
280 the model and observations in the upper troposphere, however, the model overestimates O₃ levels
(by ~12 nmol mol⁻¹) at the tropopause. Ozone-sonde measurements from another station in the
region, Davis (68.6° S, 78.0° E), were also compared with the model results for the study period
(Fig. S4 and S5). The O₃ variability from model results (standard deviation: 3–13 nmol mol⁻¹) is
comparable or slightly lower than the observed variability in the vertical distribution (950–350
285 hPa). The O₃s contribution is ~45–50% in the lower troposphere (pressure > ~700 hPa) but
increases with altitude to 65% at 500 hPa up to 100% at and above the tropopause (~300 hPa). The
EMAC model captures both the mean vertical structure as well as some secondary O₃ peaks (e.g.,
23 February 2016; Fig. 3) in the upper troposphere (~6 km; 450 hPa). However, there are some
noticeable differences between model and observations on individual days (e.g., 19 February 2016;
290 Fig. 3). The model limitation in reproducing some features of secondary peaks have been
suggested to be due to coarser vertical resolution and the temporal differences (Ojha et al., 2017),
and confirmed recently in a study focusing on tropopause folding frequency (Bartusek et al., 2023).

Overall, the model reproduces the observed tropospheric O₃ distribution and most of the day-
to-day variability in the surface and tropospheric O₃. It is to be noted that the performance of global
295 chemistry-climate models is also limited by the parameterization schemes developed for such
pristine environments with extreme climatic conditions (e.g., frequent blizzards). Note that
depletion of surface O₃ was observed over Antarctica during blizzards as blowing snow, which is
a source of sea salt aerosols and subsequently bromine, could deplete O₃ (Jones et al., 2009; Ali et
al., 2017). Nevertheless, our study fills a gap with respect to the evaluation of the widely applied
300 EMAC model for the Antarctic region and the results may have implications to further improve
the model in future studies.

3.2 Influences of downward transport on surface O₃

Several events of surface O₃ enhancements were observed during the study period, as illustrated in Fig. 2. Two such events on 23 February 2016 and 01 February 2015 are investigated in detail to understand the mechanism driving such variability.

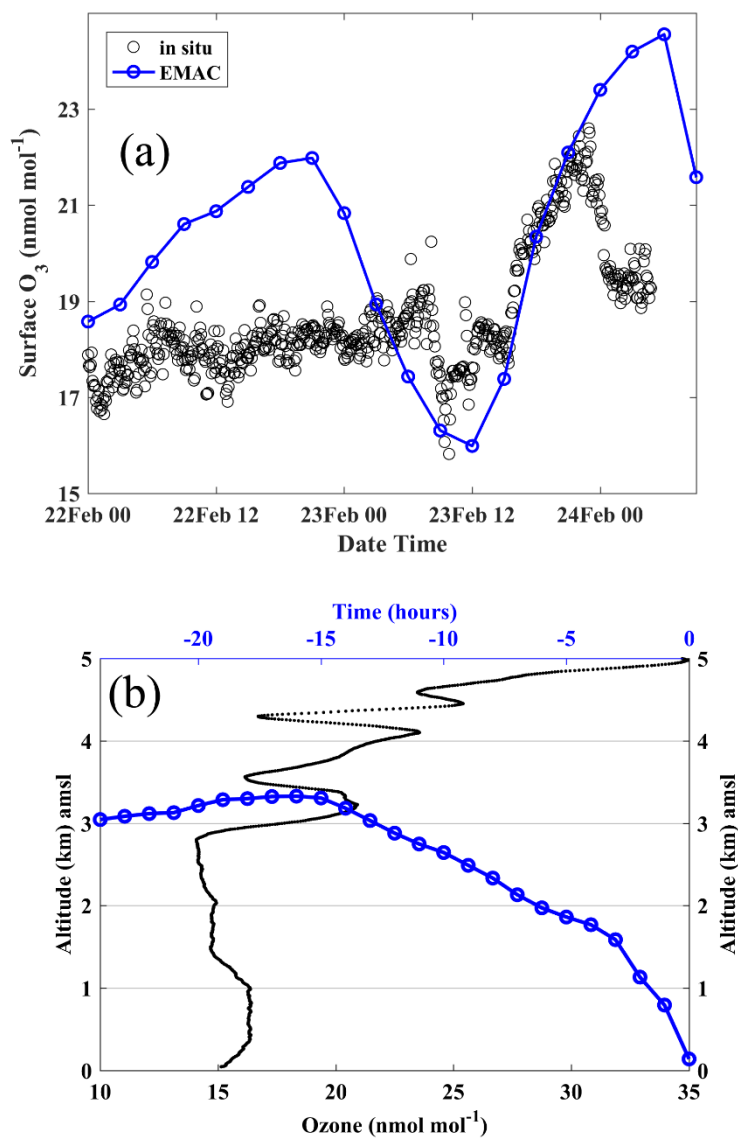
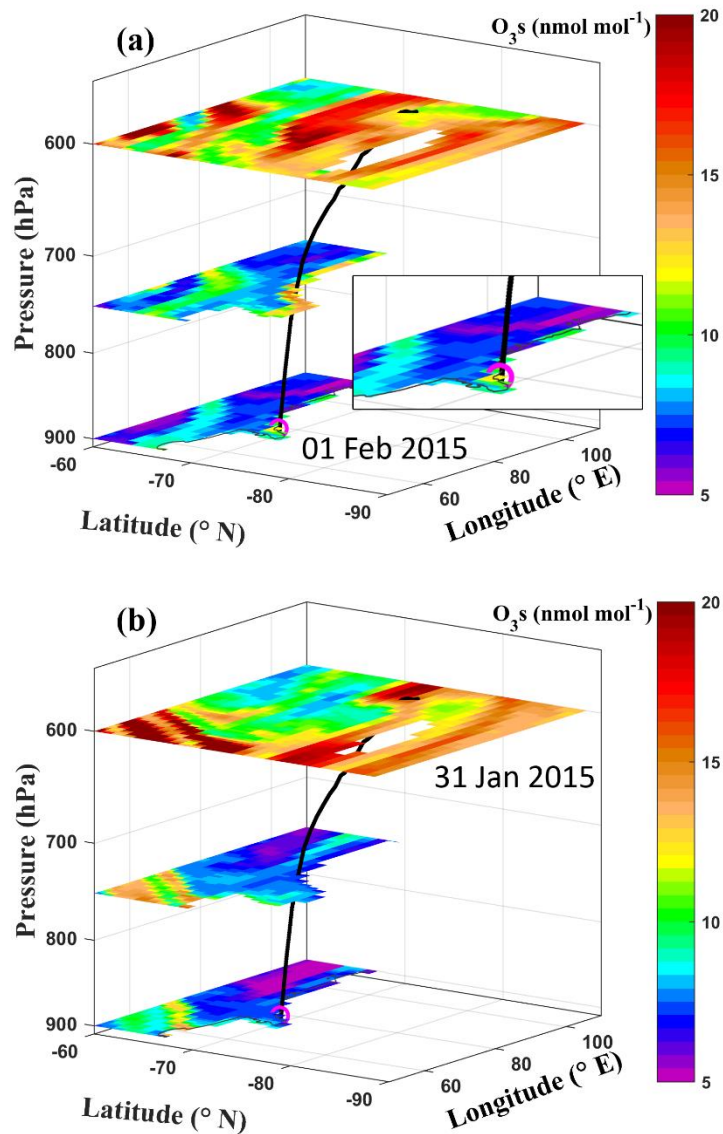


Figure 4. (a) Surface O₃ variations at Bharati station depicting an event of significant O₃ enhancement around 23:00 local time on 23 February 2016. (b) Variations in the altitude of airmass (blue) along the backward trajectory with respect to time from the O₃ enhancement event. Vertical profile of O₃ measured around 11:00 on 23 February 2016 (black).

Figure 4a shows that surface O₃ over Bharati was enhanced sharply by ~4 nmol mol⁻¹ around 23:00 on 23 February 2016. Backward air mass trajectories show that this air mass originated from ~3 km altitude about 12 hours before the event. The balloon-borne O₃ vertical profile obtained at that time (11:00 on 23 February 2016; Fig. 4b) shows the presence of a layer with enhanced O₃ (~22 nmol mol⁻¹) relative to lower altitudes (~15 nmol mol⁻¹). Based on these collocated observations and trajectory simulations, it is suggested that the O₃ rich air from this layer descended to the surface over Bharati in ~12 hours with a descent rate of >250 m h⁻¹ (0.07 m s⁻¹). The estimated descent velocity seems to be consistent with the in situ measured mean vertical wind speed (0.09±0.29 m s⁻¹) measured at this station during 18–29 January 2016. The O₃ enhancement observed in the upper troposphere (~6 km; see Fig. 3) on 23 February 2016 is associated with a stratospheric intrusion (Das et al., 2020). The presence of the jet-stream in the vicinity of the tropopause (~9 km altitude; ~300 hPa) can enhance the turbulence due to strong wind shear (squared wind shear=5 × 10⁻⁴ s⁻²). Along with this turbulence, tropopause oscillations led to the stratospheric intrusion during 22–23 February 2016 (Das et al., 2020). The presence of similar surface O₃ enhancement events on several other days also (Fig. 2) suggests that this is a periodic phenomenon that significantly contributes to tropospheric O₃ in the region.



330 **Figure 5.** Spatial distribution of the stratospheric O₃ tracer in the model at different pressure levels for (a) 01 February 2015 depicting an enhancement at surface, and (b) 31 January 2015 depicting an enhancement in the upper troposphere. The black curve represents a 24-hour backward air mass trajectory ending at Bharati station (magenta circle) on 01 February 2015 and originating around 600 hPa on 31 January 2015.

Surface O₃ shows a continuous enhancement from about 12–14 nmol mol⁻¹ on 31 January 2015 to about 25 nmol mol⁻¹ on 01 February 2015 (Fig. 2a). To analyze the influence of transport from the stratosphere, the spatial distribution of the stratospheric O₃ tracer at different pressure levels is combined with air mass trajectories (Fig. 5). The insert in Fig. 5a shows a zoomed view

335

of O_{3s} around Bharati station on 01 February 2015. The airmass backward trajectory ending at Bharati station at the time of the observed enhancement is shown by the black curve. The airmass is traced back to ~600 hPa (~4 km) one day prior to the observed enhancement (i.e., 31 January 2015) at a lower latitude, where a patch of stratospheric O₃ (20 nmol mol⁻¹) is simulated by the model. A clear descent of airmass with a descent rate of ~0.05 m s⁻¹ is seen, leading to the enhancement in surface O₃ on 01 February 2015.

The above analysis of two representative events shows that the intrusion of stratospheric O₃ followed by descent of O₃-rich air can cause a 4–10 nmol mol⁻¹ enhancement in surface O₃ during the study period. The result is in line with a continuous increase in O₃ and O_{3s} with altitude, as shown in Fig. 1c and 3. Similar variations of O_{3t} compared to O_{3s} (Fig. 2) indicate significant air mass mixing during the transport process. O₃ enhancement events with similar magnitude were also observed at the nearby station Zhongshan (69.37° S 76.36° E; Ding et al., 2020; Tian et al., 2022) and with larger magnitude at South Pole (8–20 nmol mol⁻¹; Oltmans et al., 2008) attributed to transport or NO_x driven cumulative photochemical production, assuming a marginal role of transport from the stratosphere or free-troposphere (Cristofanelli et al., 2018; Ding et al., 2020). The occurrence of such O₃ enhancement is less evident over the coastal regions compared to the Antarctic plateau (Jones, 2003). However, substantial contributions of stratosphere-troposphere exchange were associated with airmass fluxes up to 60 kg m⁻² d⁻¹ (Sanak et al., 1985) using in situ measurement of Beryllium isotope at Dumont d'Urville station. Based on long-term balloon-borne measurements and GOES-Chem (Goddard Earth Observing System coupled with Chemistry) model simulations, Greenslade et al. (2017) also reported large stratosphere to troposphere O₃ fluxes (0.50–0.75 × 10¹⁷ molecules cm⁻² month⁻¹) during summer, which exceed those during winter (0.25–0.50 × 10¹⁷ molecules cm⁻² month⁻¹).

360 **3.3 Influences of photochemistry on surface O₃**

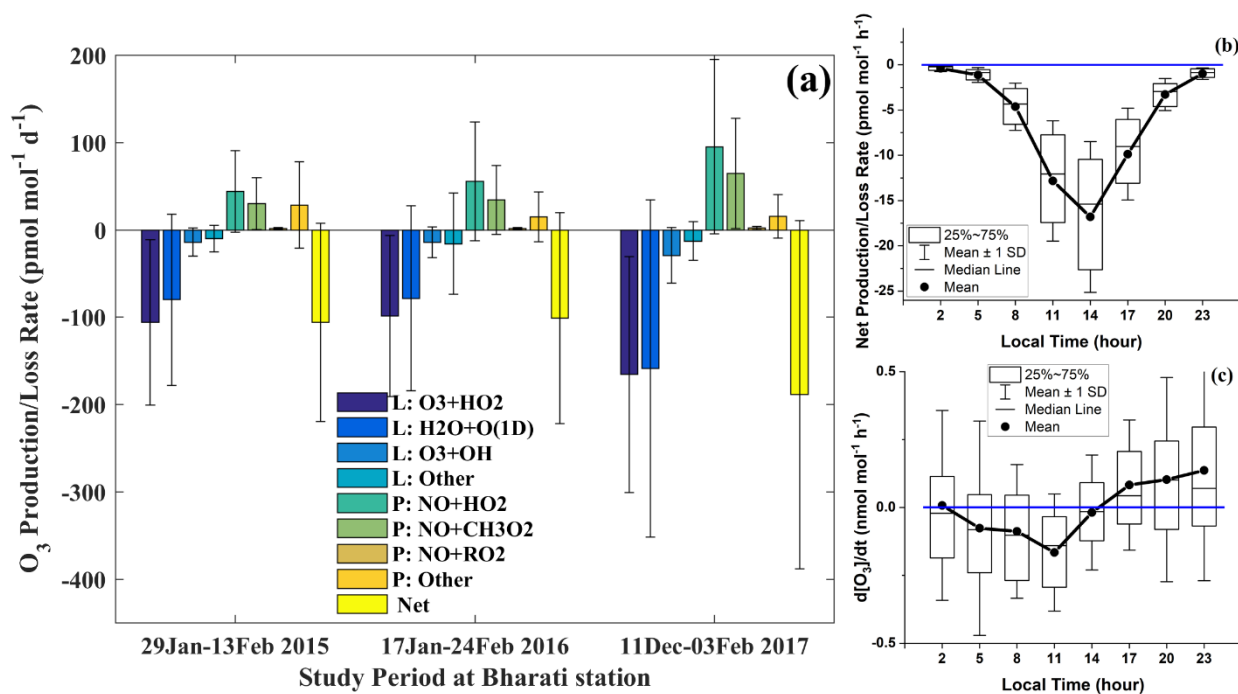


Figure 6. (a) Mean production and loss rates of surface O₃ through different chemical pathways at Bharati during the study period, (b) diurnal variation of net O₃ change due to photochemistry, derived from the EMAC model simulations, and (c) rate of change of surface O₃ (dO₃/dt) based on the in situ measurements at Bharati station.

The production and loss rates of O₃ through different chemical pathways have been estimated from the EMAC model simulation, and the mean values during the study period are shown in Fig. 6a. Among various production and loss reactions, O₃+HO₂ and H₂O+O(¹D) are found to be the dominant O₃ loss pathways, whereas NO+HO₂ and NO+CH₃O₂ are the major O₃ production reactions. Overall, the aforementioned chemical losses tend to dominate the production leading to a net photochemical loss in surface O₃ at Bharati. Effectively the study region acts as a net chemical sink of O₃. Note that loss through O₃+OH and other reactions, and production through NO+RO₂ and other reactions are relatively small in magnitude (Fig. 6). Dry deposition over ice and the surrounding ocean is a minor O₃ removal mechanism as well. The substantial variability (large error bars in Fig. 6a) in production and loss terms arises from the diurnal and day-to-day variations. Figure 6b-c shows the mean diurnal variation of net photochemical production or loss rates from the EMAC model and the rate of change of O₃ (i.e., dO₃/dt) from in situ measurements.

The net loss is relatively high during noontime (11–14 h) and negligibly small after 23:00 and prior to 5:00. In situ measured rate of change, dO_3/dt , is negative around 11:00 indicating overall loss which includes the influences of both photochemistry, dynamics and deposition losses. Since the mean amplitude of dO_3/dt in figure 6c is $0.3 \text{ nmol mol}^{-1} \text{ h}^{-1}$, which is comparable or smaller than the variability at any given hour of the day, diurnal patterns on different days might vary from the mean picture. The positive rate of change after 17:00 and prior to 5:00 represents an increase in O_3 mainly through horizontal or vertical transport as photochemistry is weak under conditions of low solar irradiance.

Despite of being a net photochemical sink of surface O_3 , it is observed that the levels of O_3 are relatively steady or continuous over time (Fig. 2c). We estimated surface O_3 fluxes by multiplying the model simulated vertical wind with the O_3 concentration at the model level just above the surface. Figure 7b shows the mean O_3 flux averaged over the study period. The negative flux represents the number of O_3 molecules moving downward (contributing to surface O_3) per unit area and per unit of time. A stronger downward flux along the east coast (Fig. 7b) counterbalances the net photochemical O_3 loss (Fig. 7a). Assuming a boundary layer height of 500 m, the loss rates integrated over boundary layer are estimated at $2.7 \times 10^{13} \text{ molecules m}^{-2} \text{ s}^{-1}$, which is of comparable magnitude to the modelled downward flux (Fig. 7b). O_3 and O_3 fluxes (i.e., fluxes at level just above the surface) correlate negatively ($R=-0.3$) at Bharati in the EMAC simulation, as shown in Fig. S6a. This is substantiated with a negative correlation of surface O_3 with the vertical wind (Fig. S6b), suggesting enhanced O_3 during conditions of descent. The results suggest that despite the net chemical sink of O_3 , the surface O_3 is maintained by a flux from above during the summer over the coastal region. The O_3 loss through chemistry is counterbalanced by the contribution from dynamics (or vice versa) over East Antarctica during austral summer.

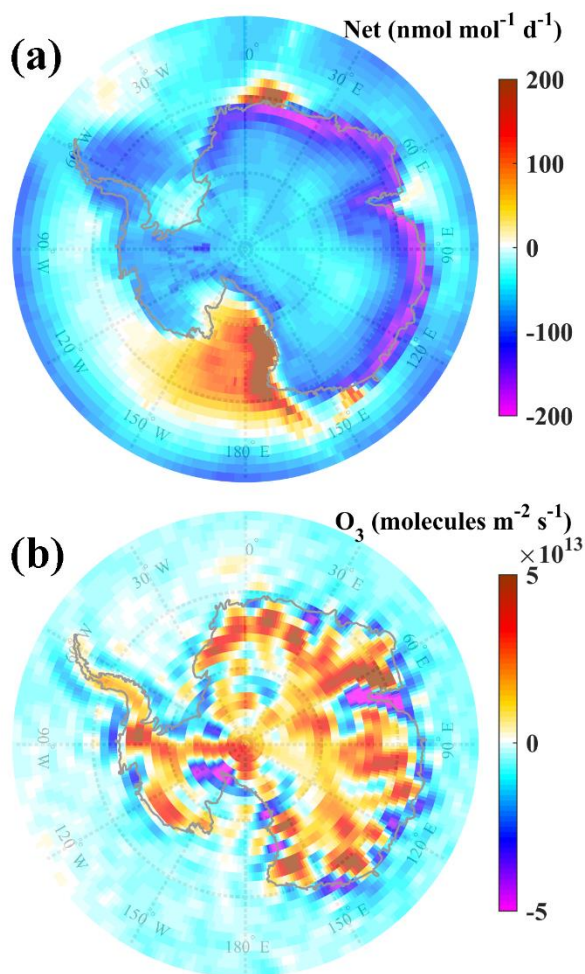
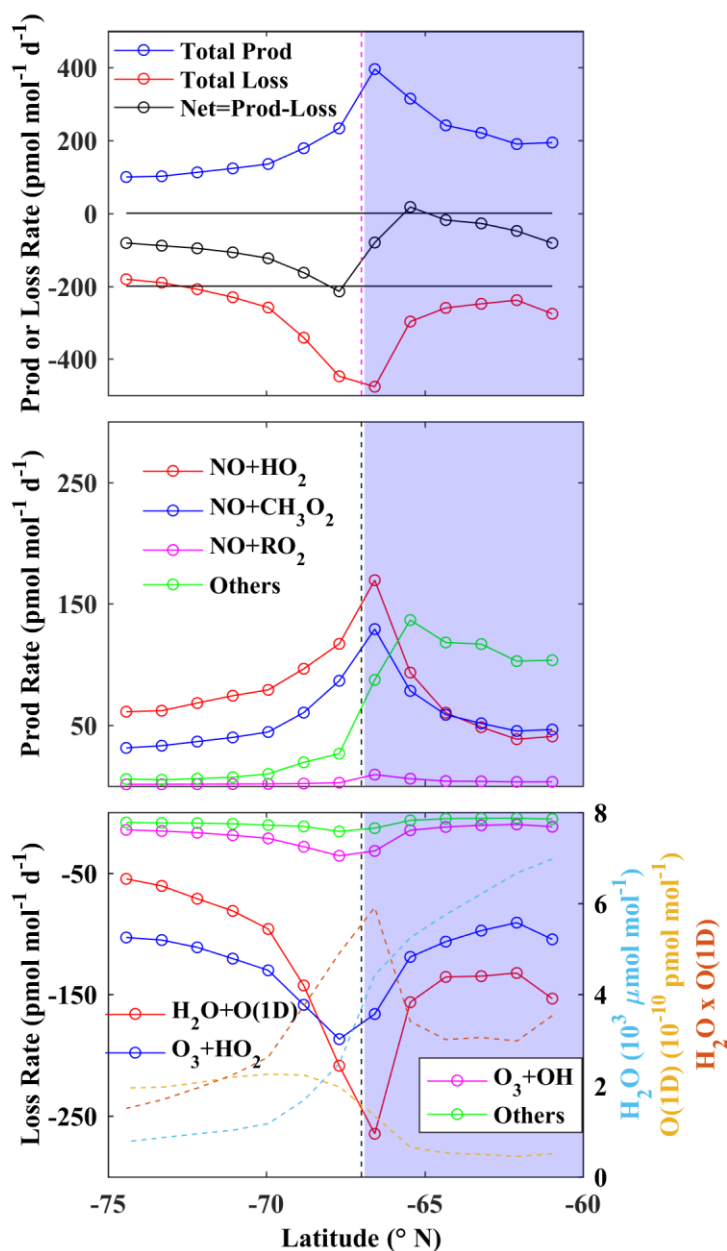


Figure 7. Spatial distribution of (a) net rate of change (production minus loss) of surface O_3 due to photochemistry and (b) O_3 flux at surface averaged over the study period.

In order to understand whether the O_3 photochemical loss over the Bharati station also
 405 prevails over larger regions in Antarctica, we analyze the spatial distribution of net production or
 loss rates averaged during the austral summer (Fig. 7a). It is important to note that our simulations
 show that the entire Antarctic continent act as sink of O_3 , in contrast to the previously reported net
 O_3 production through NO emission from snow (Legrand et al., 2016 and references therein). In
 the east coastal Antarctic region O_3 loss rates are significantly higher ($\sim 190 \text{ pmol mol}^{-1} \text{ d}^{-1}$)
 410 suggesting that it acts as a relatively strong chemical sink of surface O_3 . The loss rate is at peak
 ($\sim 190 \text{ pmol mol}^{-1} \text{ d}^{-1}$) over the east coast, higher by $\sim 100 \text{ pmol mol}^{-1} \text{ d}^{-1}$ compared to adjacent land
 and ocean, and it gets further lower ($\sim 50 \text{ pmol mol}^{-1} \text{ d}^{-1}$) much away from the coast. Note that
 model simulated mean OH and NO are in the range of $0.05\text{--}0.5 \times 10^6 \text{ molecules cm}^{-3}$ and $0.5\text{--}10$

415 pmol mol⁻¹, respectively, over entire the Antarctic region which is in line with earlier measurements at the west coast (OH mean: 0.11 × 10⁶ molecules cm⁻³, ranging <0.1–0.9 × 10⁶ molecules cm⁻³; NO: estimated value of 5 pmol mol⁻¹) by Jefferson et al., 1998 and Bloss et al., 2010, but lower (almost 5 times) than those measured during OPALE campaign (OH mean: 2.1 × 10⁶ molecules cm⁻³, ranging <0.8–6.2 × 10⁶ molecules cm⁻³; NO: 5–70 pmol mol⁻¹; Kukui et al., 2012).



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Figure 8. Latitudinal variation of production, loss and net rates of changes of surface O₃ averaged along the east coastal longitudinal band of 15–130° E during January 2017. The blue areas represent the ocean environment and vertical dashed line shows the approximate coast line.

The net O₃ loss rate (Fig. 7a) is found to be lower over land than over ocean and is highest
425 along the east coast. We further considered 6 grids at both sides of the coast line and averaged over the longitude range 15–130° E (i.e., East Antarctica). The variations in average production, loss and net rates with latitude are shown in Fig. 8. Since the latitude corresponding to different grids at 15–130° E are different, latitudes shown on the x-axis represent average latitudes. Thus, as we proceed from left to right (lower latitude to higher latitude) in Fig. 8, we move from land to ocean.

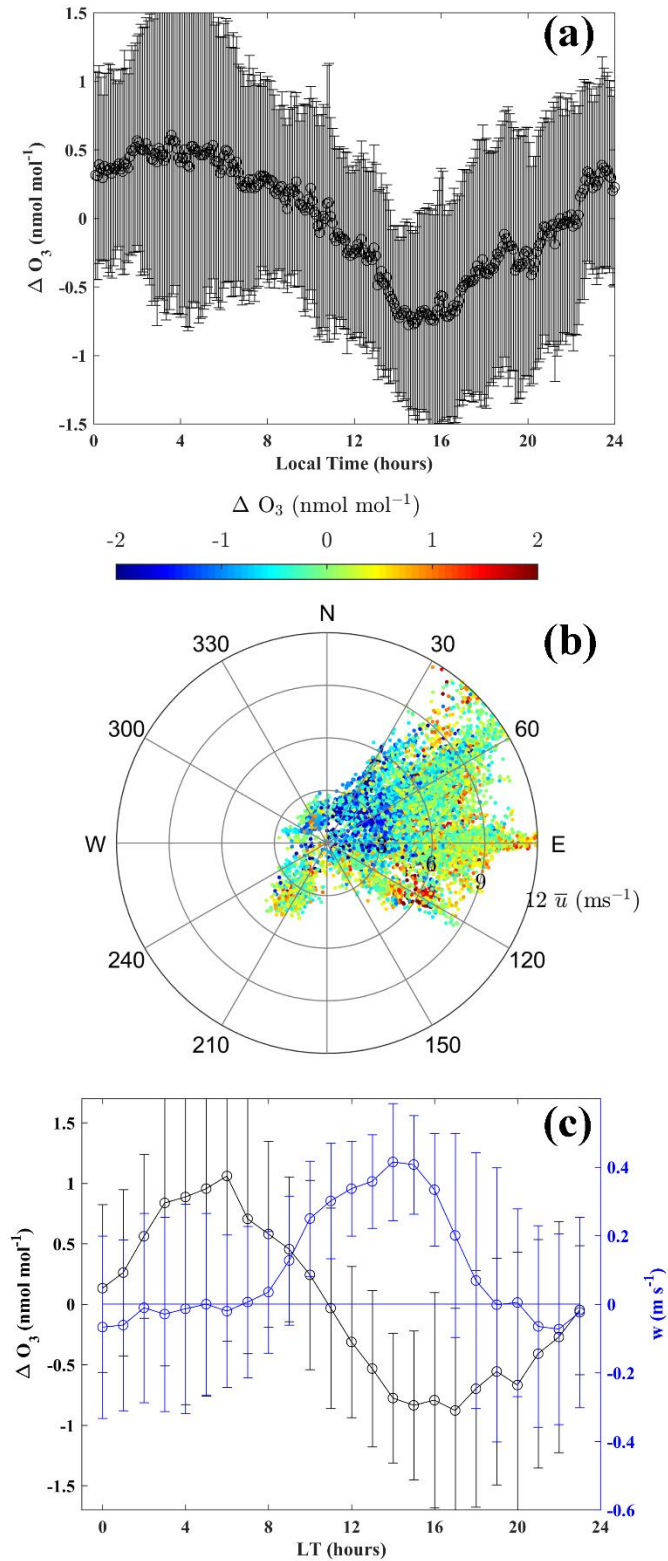
430 From Fig. 8a, it is clearly seen that the O₃ production as well as loss are maximum near the coast. Since loss dominates over the production, the net rate is negative with ~190 pmol mol⁻¹ d⁻¹. Figure 8b and c represent changes in different production and loss pathways across the coast. The photolytic O₃ loss, followed by H₂O+O(¹D), is found to be the dominating loss process peaking at ~300 pmol mol⁻¹ d⁻¹ along the coast. The reason for the peak loss rate at coast is related to the
435 opposite latitudinal gradients in H₂O and O(¹D) (see Fig. 8c; right axis). H₂O is substantially higher (~6000 μmol mol⁻¹) over ocean but much lower (1000 μmol mol⁻¹) in the drier atmosphere above continent. In contrast, O(¹D) is higher (2×10^{-10} pmol mol⁻¹) over the continent primarily due to intense solar insolation at higher elevation and over the bright ice surface. Therefore, latitudinally opposite variations of H₂O and O(¹D) lead to a relative maximum in H₂O+O(¹D) near the coast.
440 We also note that there is significant O₃ production over the ocean due to reactions other than the three primary reactions of peroxy radicals (HO₂, RO₂, CH₃O₂) with NO.

Thus, under the prevailing relatively strong O₃ sink along the east coast, the mean O₃ level during summer is maintained by the downward flux of O₃ from the stratosphere.

3.4 Diurnal variation of surface O₃ at Bharati

445 Considering day-to-day variability, including enhancement events governed by stratospheric influence, ΔO₃ is computed by subtracting the running mean O₃ (288 points; 5 min interval—daily running mean) from the observed O₃. Figure 9a shows the mean diurnal variation of ΔO₃ during the 18 January–23 February 2016 period for which measurements of horizontal wind at surface were also available. Surface O₃ exhibits a diurnal variation, being relatively low during the

450 afternoon (15:00 local time) and relatively high during nighttime (Fig. 9) with a diurnal amplitude
of $\sim 1.2 \text{ nmol mol}^{-1}$. Figure 9b shows the wind rose color coded with ΔO_3 mixing ratios. Sunlight
at Bharati is abundant during summer and the land-sea thermal contrast explains the typical diurnal
change in the wind direction under normal meteorological conditions, i.e., excluding blizzards and
snow storms. Figure S7 shows time series of the wind direction and surface ΔO_3 , depicting the
455 link between O_3 and the wind direction. Due to higher O_3 over the eastern Antarctic land regions,
winds from that sector transport the O_3 -rich air to the Bharati station causing enhanced O_3 mixing
ratios. O_3 is higher when wind is parallel to the coast (easterly; wind direction $\sim 90^\circ$) or from the
land (wind direction: $90\text{--}240^\circ$). Under calm wind condition, the influence of transport is minimal
and photochemical loss is more pronounced. When the wind is weak and from the ocean (wind
460 direction: $30\text{--}90^\circ \text{ N}$), O_3 levels are lower due to dilution by mixing with air from the oceanic
sector. The O_3 diurnal variation is also closely linked with the vertical wind. Based on limited in
situ measurements of the vertical wind at the surface during 18–29 January 2016, the mean diurnal
variation of vertical wind (w) along with ΔO_3 is shown in Fig. 9c. Downdrafts and stronger
updrafts (up to $\sim 0.4 \text{ ms}^{-1}$) are seen during nighttime (or lower solar zenith angle; 20:00–07:00) and
465 daytime (08:00–19:00), respectively. Higher O_3 during nighttime is associated with downdrafts
and O_3 mixing ratios are reduced with increasing updraft intensity. The EMAC model shows
limitations in reproducing the observed diurnal variation likely because of coarse resolution
averaging out the topography and mesoscale dynamics.



470 **Figure 9.** (a) Diurnal variation of ΔO_3 , (b) wind rose color coded with ΔO_3 , and (c) variation in collocated vertical wind and ΔO_3 at Bharati during the austral summer of 2016.

Diurnal patterns with an amplitude ranging from ~ 0.2 – 2 nmol mol^{-1} were reported at coastal (Syowa and McMurdo) and inland (Concordia; 75° S ; 123° E ; 3220 m above sea level) stations (Ghude et al., 2006; Legrand et al., 2009). However, such a pattern is absent over the South Pole (Oltmans, 1981). Interestingly, photochemical production during the morning hours (05:00–11:00) due to the NO_x released from snow was followed by a reduction due to an increase in boundary layer height ($200 \pm 100 \text{ m}$) at the inland station Concordia (Legrand et al., 2009; 2016). Shallow convective boundary layers (less than 300 m) were reported over the Antarctic Plateau region by Mastrantonio et al. (1999). Unlike these studies, we did not observe photochemical O_3 production nor a clear signature of changes in O_3 transport across the top of boundary layer from our ozonesonde measured O_3 profiles over Bharati station. Therefore, the diurnal patterns of O_3 over coastal Antarctica are found to be different than those over the inland region, mainly due to differences in meteorological conditions and the concentrations of precursor gases.

3.5 Absence of signature of halogen chemistry

Reactive halogens (e.g., iodine, bromine) have been shown to deplete O_3 in the boundary layer over the Antarctic region (Barrie et al., 1988; Oltmans and Komhyr, 1976). However, ground-based remote sensing observations found very low concentrations of iodine oxide ($\sim 0.3 \pm 0.1 \text{ pmol mol}^{-1}$) in the boundary layer over Bharati station during the study period (Mahajan et al., 2021) and no clear sign of O_3 depletion was observed.

Satellite (SCIAMACHY—SCanning Imaging Absorption spectroMeter for Atmospheric CartographY and OMI—Ozone Monitoring Instrument) observations also show lower monthly mean iodine monoxide (IO) columnar density (0 – $1 \times 10^{12} \text{ molecules cm}^{-2}$) over Bharati compared to west Antarctica (figure not shown). This is consistent with previous studies (e.g., Schönhardt et al., 2012) showing relatively low IO over east Antarctica and the adjacent ocean ($\leq 0.7 \times 10^{12} \text{ molecules cm}^{-2}$) compared to west Antarctica ($\sim 1.5 \times 10^{12} \text{ molecules cm}^{-2}$) during summer season (December–January–February 2004–2009).

Bromine (Br) driven O_3 depletion events, resulting into BrO, are less frequent over the Antarctic region compared to the Arctic region due to differences in springtime surface temperatures (Tarasick and Bottenheim, 2002). However, large O_3 depletion events were observed

at Neumayer (70.62° S, 8.37° W; 42 m amsl) during the late winter (July to September), likely due to stronger BrO episodes from the larger sea ice coverage around the site (Legrand et al., 2009). Analysis of BrO from OMI possibly indicates an O₃ depletion event on 7 February 2015 at Bharati where BrO was enhanced, $\sim 9.2 \times 10^{13}$ molecules cm⁻² with lower O₃ (~ 7 nmol mol⁻¹), marked by red rectangle in Fig. S8. Except for this event, BrO remained below 8×10^{13} molecules cm⁻² around Bharati station ($\pm 0.5^\circ$ latitude/longitude) during the study. O₃ depletion was also not seen at Syowa (Fig. S9) during the study period. The coastal region of east Antarctica exhibits slightly higher values of BrO ($\sim 7 \times 10^{13}$ molecules cm⁻²) compared to the ocean and land regions ($4\text{--}6 \times 10^{13}$ molecules cm⁻²). However, it is low ($4\text{--}8 \times 10^{13}$ molecules cm⁻²) during December–February (2004–2009) compared to the levels during September–November ($5\text{--}10 \times 10^{13}$ molecules cm⁻²) over the Antarctic region (Schönhardt et al., 2012). The impact of Br chemistry on surface O₃ is suggested to be weaker along the east coast of Antarctica (Dumont d'Urville and Syowa) in contrast to western coastal Antarctica as observed over the Neumayer and Halley (75.55° S, 26.53° W, 30 m amsl) stations (Legrand et al., 2016). Nevertheless, simultaneous measurements of O₃ and halogen species including BrO are desirable to quantify the role of halogen chemistry over eastern Antarctica.

3.6 Surface O₃ during winter

To take into account the seasonality of O₃ at the surface, the wintertime distribution is shown in Fig. S10. Mean surface O₃ level is higher during winter (20–32 nmol mol⁻¹) compared to the summer (11–23 nmol mol⁻¹), in line with the reported seasonality in the literature (Legrand et al., 2009). Figure S10 reveals three low-O₃ patches over the coastal oceanic region. One is close to Bharati station, however, we do not have observations during wintertime for comparison. Model simulations suggest that surface O₃ is composed of 63–67% O_{3s} of stratospheric origin during winter (Fig. S10b), which is significantly higher than during austral summer. The probability of downward transport from the stratosphere during winter, also associated with a lower altitude of the tropopause, is larger (Kumar et al., 2021). Comparison of surface O₃ at Syowa (69.00 °S; 39.58 °E; not shown here) shows that the model captures the variability with R=0.3 and a negative bias of ~ 5 nmol mol⁻¹. The model performance seems to be better during summer, indicative of limitations to reproduce the mean O₃ concentrations and the variability during winter. Further studies are needed to understand and rectify the factors causing greater bias in the model during

winter. Analysis of the O₃ budget suggests a small net loss of O₃ by 10–25 pmol mol⁻¹ d⁻¹ over the oceanic region and close to zero (<5 pmol mol⁻¹ d⁻¹) over the Antarctic continent (Fig. S11). To study this in greater detail, we highly recommend to conduct continuous wintertime measurements of O₃, its precursors including halogens over Bharati during winter season.

535 4 Summary

Ground- and balloon-borne O₃ measurements have been conducted over the Indian station Bharati at the east coast of Antarctica during the austral summers of 2015–2017. The observations have been used to evaluate the performance of the global chemistry-climate model EMAC over this part of the world. A comprehensive analysis of observations and model simulations provided
540 significant insights into the dynamical and photochemical processes affecting surface O₃ and its variability. The main results are:

1. Surface O₃ levels over the Indian station Bharati at the eastern coastal Antarctica have been observed to be ~19 nmol mol⁻¹ with a small variability of ~2 nmol mol⁻¹ during austral summer. While similar levels prevail over the east coast, O₃ is typically higher over land at higher
545 elevation. EMAC model successfully reproduced the observed mean levels with negligible bias over this unique environment and also captured the temporal variability (R=0.5). In particular, the model successfully reproduced some events during which O₃ was enhanced. Analysis of the stratospheric O₃ tracer in the model suggests that 41–51% of surface O₃ is of stratospheric origin with larger fractions over the higher elevation regions in Antarctica.
- 550 2. The model successfully reproduced the mean vertical distribution of O₃ over Bharati observed by balloon-borne soundings. Detailed analysis combining the balloon profiles, model tracers, and airmass trajectories shows that downward transport caused the observed events during which O₃ was enhanced.
3. Along the east coast of Antarctica, including Bharati station, photochemistry acts as a relatively
555 strong sink of surface O₃ (~190 pmol mol⁻¹ d⁻¹) when compared to adjacent land and ocean regions. Chemical loss through O₃ photolysis (followed by H₂O+O(¹D)) and O₃+HO₂ dominates over the major production (through NO+HO₂ and NO+CH₃O₂). Reverse latitudinal gradients between H₂O and O(¹D) lead to maximum O₃ loss at the coastal region. The continuous chemical loss is found to be counterbalanced by downward O₃ transport from above.
560 The findings show the intertwined roles of dynamics and photochemistry that govern the O₃

variability over east Antarctica, and maintaining significant O₃ levels despite the absence of local precursor sources.

4. In addition to the role of photochemistry, the diurnal variation of O₃ at Bharati was found to correlate with the diurnal wind changes. Surface O₃ varied with a diurnal amplitude of 1.2 nmol mol⁻¹, with the higher levels occurring when the wind blew parallel to the coast or from land regions. In addition, up- and downdrafts also play a role in the diurnal variation.

Our observations during austral summer over three years complement available data e.g. from eastern coastal Antarctica. The observations, besides revealing diurnal and day-to-day variability, helped in evaluating the performance of a global chemistry-climate model over this unique, pristine environment. The study provides valuable insights into the complementary roles of photochemistry and dynamics in governing O₃ and its variability over Antarctica. In view of increasing anthropogenic activities and the changing climate, monitoring of O₃ and related species (NO, NO₂, CO, VOCs and halogens) is needed.

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Code availability. The Modular Earth Submodel System (MESSy) is continuously further developed and applied by a consortium of institutions. The usage of MESSy and access to the source code is licensed to all affiliates of institutions that are members of the MESSy Consortium. Institutions can become a member of the MESSy Consortium by signing the MESSy Memorandum of Understanding. More information can be found on the MESSy Consortium Website (<http://www.messy-interface.org>, last access: 04 July 2023). The code presented here has been based on MESSy version 2.55 and is available as git commit #a5bd54d5b in the MESSy repository.

580

Data availability. Measured ozone and EMAC simulated fields shown in the figures can be obtained from the website of Space Physics Laboratory (<https://spl.gov.in/SPLv2/index.php/spl-metadata/104-spl/550-trace-gases-metadata.html>, last access: 1 January 2024) or from the direct link, [https://spl.gov.in/SPLv2/images/SPL-METADATA/Girach et al 2024 ACP Ozone Bharati Antarctica.xlsx](https://spl.gov.in/SPLv2/images/SPL-METADATA/Girach_et_al_2024_ACP_Ozone_Bharati_Antarctica.xlsx) (last access: 1 January 2024).

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Author contributions. I.A. Girach conceptualized and designed the study, performed measurements and analysed the datasets. K.V. Subrahmanyam, Koushik N., Mohammed Nazeer M., N.V.P. Kiran Kumar contributed in the measurements. A. Pozzer performed the model
595 simulations. N. Ojha, A. Pozzer, P.R. Nair, S.S. Babu and J. Lelieveld helped I. A. Girach in the analysis and interpretation of the results. I. A. Girach wrote the manuscript and all the co-authors contributed to the review and editing.

Declaration of competing interest. At least one of the (co-)authors is a member of the editorial
600 board of Atmospheric Chemistry and Physics.

Acknowledgements. We gratefully acknowledge the organiser, Centre for Polar and Ocean Research (NCPOR), Goa, Ministry of Earth Sciences, India for providing the opportunity to participate in the 34th, 35th and 36th Indian Scientific Expedition to Antarctica (ISEA). We also
605 acknowledge the leaders of Bharati Station and Voyage for providing necessary support for the smooth conduct of experiments at Bharati station. We are really thankful to Mr. Santosh Muralidharan, Space Physics Laboratory; Mr. Brijesh Desai, Laboratory in-charge of Bharati station during 35th expedition and other expedition members of 34th, 35th and 36th ISEA for their help during the field measurements. We also thank Mriganka Sekhar Biswas, Indian Institute of
610 Tropical Meteorology, India for fruitful discussion on halogen chemistry. We are also thankful to India Meteorological Department (IMD) for providing meteorological observations, hydrogen gas cylinders for balloon ascents and for the help during balloon launches. The EMAC model simulations have been performed at the German Climate Computing Centre (DKRZ). Surface ozone observations at Antarctic stations (South Pole, United States; Arrival Heights, New Zealand; Marambio, Argentina; Syowa, Japan) were obtained from the newly established World Data
615 Centre for Reactive Gases (WDCRG), WMO's GAW (Global Atmosphere Watch; World Meteorological Organization) programme (<https://ebas.nilu.no/>, last access: 1 January 2024 and <https://ebas-data.nilu.no/Default.aspx>, last access: 1 January 2024). Vertical O₃ profiles measured at Davis station were obtained from <https://woudc.org/data/explore.php> (last access: 1 January
620 2024). We highly acknowledge teams of researchers who made ozone measurements at various Antarctic stations and made them available publically. We also acknowledge the NOAA Air

Resources Laboratory (ARL) for providing air mass trajectory from the HYSPLIT transport and dispersion model from their READY website (<https://www.ready.noaa.gov/HYSPLIT.php>, last access: 1 January 2024).

625 **References**

Ali, K., Trivedi, D. K., and Sahu, S. K.: Surface ozone characterization at Larsemann Hills and Maitri, Antarctica, *Sci. Total Environ.*, 584–585, 1130–1137, <https://doi.org/10.1016/j.scitotenv.2017.01.173>, 2017.

630 Barrie, L. A., Bottenheim, J. W., Schnell, R. C., Crutzen, P. J., and Rasmussen, R. A.: Ozone destruction and photochemical reactions at polar sunrise in the lower Arctic atmosphere, *Nature*, 334, 138–141, <https://doi.org/10.1038/334138a0>, 1988.

Bartusek, S., Wu, Y., Ting, M., Zheng, C., Fiore, A., Sprenger, M., and Flemming, J.: Higher-Resolution Tropopause Folding Accounts for More Stratospheric Ozone Intrusions, *Geophys. Res. Lett.*, 50, e2022GL101690, <https://doi.org/10.1029/2022GL101690>, 2023.

635 Bloss, W. J., Camredon, M., Lee, J. D., Heard, D. E., Plane, J. M. C., Saiz-Lopez, A., J.-B. Bauguitte, S., Salmon, R. A., and Jones, A. E.: Coupling of HO_x, NO_x and halogen chemistry in the antarctic boundary layer, *Atmos. Chem. Phys.*, 10, 10187–10209, <https://doi.org/10.5194/acp-10-10187-2010>, 2010.

640 Bond, A. M. H., Frey, M. M., Kaiser, J., Kleffmann, J., Jones, A. E., and Squires, F. A.: Snowpack nitrate photolysis drives the summertime atmospheric nitrous acid (HONO) budget in coastal Antarctica, *Atmos. Chem. Phys.*, 23, 5533–5550, <https://doi.org/10.5194/acp-23-5533-2023>, 2023.

Brühl, C., Schallock, J., Klingmüller, K., Robert, C., Bingen, C., Clarisse, L., Heckel, A., North, P., and Rieger, L.: Stratospheric aerosol radiative forcing simulated by the chemistry climate
645 model EMAC using Aerosol CCI satellite data, *Atmos. Chem. Phys.*, 18, 12845–12857, <https://doi.org/10.5194/acp-18-12845-2018>, 2018.

Crawford, J. H., Davis, D. D., Chen, G., Buhr, M., Oltmans, S., Weller, R., Mauldin, L., Eisele, F., Shetter, R., Lefer, B., Arimoto, R., and Hogan, A.: Evidence for photochemical production of ozone at the South Pole surface, *Geophys. Res. Lett.*, 28, 3641–3644,

650 <https://doi.org/10.1029/2001GL013055>, 2001.

Cristofanelli, P., Putero, D., Bonasoni, P., Busetto, M., Calzolari, F., Camporeale, G., Grigioni, P., Lupi, A., Petkov, B., Traversi, R., Udisti, R., and Vitale, V.: Analysis of multi-year near-surface ozone observations at the WMO/GAW “Concordia” station (75°06'S, 123°20'E, 3280 m a.s.l. – Antarctica), *Atmos. Environ.*, 177, 54–63, <https://doi.org/10.1016/j.atmosenv.2018.01.007>, 2018.

655 Das, S. S., Ramkumar, G., Koushik, N., Murphy, D. J., Girach, I. A., Suneeth, K. V., Subrahmanyam, K. V., Soni, V. K., Kumar, V., and Nazeer, M.: Multiplatform observations of stratosphere-troposphere exchange over the Bharati (69.41° S, 76° E), Antarctica during ISEA-35, *J. Atmos. Solar-Terrestrial Phys.*, 211, <https://doi.org/10.1016/j.jastp.2020.105455>, 2020.

Davis, D. D., Eisele, F., Chen, G., Crawford, J., Huey, G., Tanner, D., Slusher, D., Mauldin, L.,
660 Oncley, S., Lenschow, D., Semmer, S., Shetter, R., Lefer, B., Arimoto, R., Hogan, A., Grube, P., Lazzara, M., Bandy, A., Thornton, D., Berresheim, H., Bingemer, H., Hutterli, M., McConnell, J., Bales, R., Dibb, J., Buhr, M., Park, J., McMurry, P., Swanson, A., Meinardi, S., and Blake, D.: An overview of ISCAT 2000, *Atmos. Environ.*, 38, 5363–5373, <https://doi.org/10.1016/j.atmosenv.2004.05.037>, 2004.

665 Ding, M., Tian, B., Ashley, M. C. B., Putero, D., Zhu, Z., Wang, L., Yang, S., Li, C., and Xiao, C.: Year-round record of near-surface ozone and O₃ enhancement events (OEEs) at Dome A, East Antarctica, *Earth Syst. Sci. Data*, 12, 3529–3544, <https://doi.org/10.5194/essd-12-3529-2020>, 2020.

Eisele, F., Davis, D., Helmig, D., Oltmans, S., Neff, W., Huey, G., Tanner, D., Chen, G., Crawford, J., and Arimoto, R.: Antarctic Tropospheric Chemistry Investigation (ANTCI) 2003 overview, *Atmos. Environ.*, 42, 2749–2761, <https://doi.org/10.1016/j.atmosenv.2007.04.013>, 2008.

Falk, S. and Sinnhuber, B.-M.: Polar boundary layer bromine explosion and ozone depletion events in the chemistry–climate model EMAC v2.52: implementation and evaluation of AirSnow algorithm, *Geosci. Model Dev.*, 11, 1115–1131, <https://doi.org/10.5194/gmd-11-1115-2018>, 2018.

675 Fernandez, R. P., Carmona-Balea, A., Cuevas, C. A., Barrera, J. A., Kinnison, D. E., Lamarque, J.-F., Blaszcak-Boxe, C., Kim, K., Choi, W., Hay, T., Blechschmidt, A.-M., Schönhardt, A., Burrows, J. P., and Saiz-Lopez, A.: Modeling the Sources and Chemistry of Polar Tropospheric

Halogens (Cl, Br, and I) Using the CAM-Chem Global Chemistry-Climate Model, *J. Adv. Model. Earth Syst.*, 11, 2259–2289, <https://doi.org/10.1029/2019MS001655>, 2019.

680 Frey, M. M., Stewart, R. W., McConnell, J. R., and Bales, R. C.: Atmospheric hydroperoxides in West Antarctica: Links to stratospheric ozone and atmospheric oxidation capacity, *J. Geophys. Res.*, 110, D23301, <https://doi.org/10.1029/2005JD006110>, 2005.

Frey, M. M., Roscoe, H. K., Kukui, A., Savarino, J., France, J. L., King, M. D., Legrand, M., and Preunkert, S.: Atmospheric nitrogen oxides (NO and NO₂) at Dome C, East Antarctica, during the
685 OPALE campaign, *Atmos. Chem. Phys.*, 15, 7859–7875, <https://doi.org/10.5194/acp-15-7859-2015>, 2015.

Ghude, S. D., Jain, S. L., Arya, B. C., Kulkarni, P. S., Kumar, A., and Ahmed, N.: Temporal and spatial variability of surface ozone at Delhi and Antarctica, *Int. J. Climatol.*, 26, 2227–2242, <https://doi.org/10.1002/joc.1367>, 2006.

690 Greenslade, J. W., Alexander, S. P., Schofield, R., Fisher, J. A., and Klekociuk, A. K.: Stratospheric ozone intrusion events and their impacts on tropospheric ozone in the Southern Hemisphere, *Atmos. Chem. Phys.*, 17, 10269–10290, <https://doi.org/10.5194/acp-17-10269-2017>, 2017.

Griffiths, P. T., Murray, L. T., Zeng, G., Shin, Y. M., Abraham, N. L., Archibald, A. T., Deushi, M., Emmons, L. K., Galbally, I. E., Hassler, B., Horowitz, L. W., Keeble, J., Liu, J., Moeini, O.,
695 Naik, V., O'Connor, F. M., Oshima, N., Tarasick, D., Tilmes, S., Turnock, S. T., Wild, O., Young, P. J., and Zanis, P.: Tropospheric ozone in CMIP6 simulations, *Atmos. Chem. Phys.*, 21, 4187–4218, <https://doi.org/10.5194/acp-21-4187-2021>, 2021.

Helmig, D., Ganzeveld, L., Butler, T., and Oltmans, S. J.: The role of ozone atmosphere-snow gas
700 exchange on polar, boundary-layer tropospheric ozone – a review and sensitivity analysis, *Atmos. Chem. Phys.*, 7, 15–30, <https://doi.org/10.5194/acp-7-15-2007>, 2007.

Honrath, R. E., Lu, Y., Peterson, M. C., Dibb, J. E., Arsenault, M. A., Cullen, N. J., and Steffen, K.: Vertical fluxes of NO_x, HONO, and HNO₃ above the snowpack at Summit, Greenland, *Atmos. Environ.*, 36, 2629–2640, [https://doi.org/10.1016/S1352-2310\(02\)00132-2](https://doi.org/10.1016/S1352-2310(02)00132-2), 2002.

705 Jefferson, A., Tanner, D. J., Eisele, F. L., Davis, D. D., Chen, G., Crawford, J., Huey, J. W., Torres, A. L., and Berresheim, H.: OH photochemistry and methane sulfonic acid formation in the coastal Antarctic boundary layer, *J. Geophys. Res. Atmos.*, 103, 1647–1656, <https://doi.org/10.1029/97JD02376>, 1998.

Jöckel, P., Tost, H., Pozzer, A., Brühl, C., Buchholz, J., Ganzeveld, L., Hoor, P., Kerkweg, A., 710 Lawrence, M. G., Sander, R., Steil, B., Stiller, G., Tanarhte, M., Taraborrelli, D., van Aardenne, J., and Lelieveld, J.: The atmospheric chemistry general circulation model ECHAM5/MESSy1: consistent simulation of ozone from the surface to the mesosphere, *Atmos. Chem. Phys.*, 6, 5067–5104, <https://doi.org/10.5194/acp-6-5067-2006>, 2006.

Jöckel, P., Kerkweg, A., Pozzer, A., Sander, R., Tost, H., Riede, H., Baumgaertner, A., Gromov, 715 S., and Kern, B.: Development cycle 2 of the Modular Earth Submodel System (MESSy2), *Geosci. Model Dev.*, 3, 717–752, <https://doi.org/10.5194/gmd-3-717-2010>, 2010.

Jöckel, P., Tost, H., Pozzer, A., Kunze, M., Kirner, O., Brenninkmeijer, C. A. M., Brinkop, S., Cai, D. S., Dyroff, C., Eckstein, J., Frank, F., Garny, H., Gottschaldt, K.-D., Graf, P., Grewe, V., Kerkweg, A., Kern, B., Matthes, S., Mertens, M., Meul, S., Neumaier, M., Nützel, M., Oberländer- 720 Hayn, S., Ruhnke, R., Runde, T., Sander, R., Scharffe, D., and Zahn, A.: Earth System Chemistry integrated Modelling (ESCiMo) with the Modular Earth Submodel System (MESSy) version 2.51, *Geosci. Model Dev.*, 9, 1153–1200, <https://doi.org/10.5194/gmd-9-1153-2016>, 2016.

Jones, A. E.: An analysis of the oxidation potential of the South Pole boundary layer and the influence of stratospheric ozone depletion, *J. Geophys. Res.*, 108, 4565, 725 <https://doi.org/10.1029/2003JD003379>, 2003.

Jones, A. E., Weller, R., Anderson, P. S., Jacobi, H.-W., Wolff, E. W., Schrems, O., and Miller, H.: Measurements of NO_x emissions from the Antarctic snowpack, *Geophys. Res. Lett.*, 28, 1499–1502, <https://doi.org/10.1029/2000GL011956>, 2001.

Jones, A. E., Wolff, E. W., Salmon, R. A., Bauguitte, S. J.-B., Roscoe, H. K., Anderson, P. S., 730 Ames, D., Clemitshaw, K. C., Fleming, Z. L., Bloss, W. J., Heard, D. E., Lee, J. D., Read, K. A., Hamer, P., Shallcross, D. E., Jackson, A. V., Walker, S. L., Lewis, A. C., Mills, G. P., Plane, J. M. C., Saiz-Lopez, A., Sturges, W. T., and Worton, D. R.: Chemistry of the Antarctic Boundary Layer

and the Interface with Snow: an overview of the CHABLIS campaign, *Atmos. Chem. Phys.*, 8, 3789–3803, <https://doi.org/10.5194/acp-8-3789-2008>, 2008.

735 Jones, A. E., Anderson, P. S., Begoin, M., Brough, N., Hutterli, M. A., Marshall, G. J., Richter, A., Roscoe, H. K., and Wolff, E. W.: BrO, blizzards, and drivers of polar tropospheric ozone depletion events, *Atmos. Chem. Phys.*, 9, 4639–4652, <https://doi.org/10.5194/acp-9-4639-2009>, 2009.

Jones, A. E., Wolff, E. W., Brough, N., Bauguitte, S. J.-B., Weller, R., Yela, M., Navarro-Comas, 740 M., Ochoa, H. A., and Theys, N.: The spatial scale of ozone depletion events derived from an autonomous surface ozone network in coastal Antarctica, *Atmos. Chem. Phys.*, 13, 1457–1467, <https://doi.org/10.5194/acp-13-1457-2013>, 2013.

Kerkweg, A., Jöckel, P., Pozzer, A., Tost, H., Sander, R., Schulz, M., Stier, P., Vignati, E., Wilson, J., and Lelieveld, J.: Consistent simulation of bromine chemistry from the marine boundary layer 745 to the stratosphere – Part 1: Model description, sea salt aerosols and pH, *Atmos. Chem. Phys.*, 8, 5899–5917, <https://doi.org/10.5194/acp-8-5899-2008>, 2008.

Kukui, A., Legrand, M., Ancellet, G., Gros, V., Bekki, S., Sarda-Estève, R., Loisil, R., and Preunkert, S.: Measurements of OH and RO₂ radicals at the coastal Antarctic site of Dumont d’Urville (East Antarctica) in summer 2010–2011, *J. Geophys. Res. Atmos.*, 117, 750 <https://doi.org/10.1029/2012JD017614>, 2012.

Kumar, P., Kuttippurath, J., von der Gathen, P., Petropavlovskikh, I., Johnson, B., McClure-Begley, A., Cristofanelli, P., Bonasoni, P., Barlasina, M. E., and Sánchez, R.: The Increasing Surface Ozone and Tropospheric Ozone in Antarctica and Their Possible Drivers, *Environ. Sci. Technol.*, 55, 8542–8553, <https://doi.org/10.1021/acs.est.0c08491>, 2021.

755 Legrand, M., Preunkert, S., Jourdain, B., Gallée, H., Goutail, F., Weller, R., and Savarino, J.: Year-round record of surface ozone at coastal (Dumont d’Urville) and inland (Concordia) sites in East Antarctica, *J. Geophys. Res. Atmos.*, 114, <https://doi.org/10.1029/2008JD011667>, 2009.

Legrand, M., Preunkert, S., Savarino, J., Frey, M. M., Kukui, A., Helmig, D., Jourdain, B., Jones, A. E., Weller, R., Brough, N., and Gallée, H.: Inter-annual variability of surface ozone at coastal 760 (Dumont d’Urville, 2004–2014) and inland (Concordia, 2007–2014) sites in East Antarctica,

- Atmos. Chem. Phys., 16, 8053–8069, <https://doi.org/10.5194/acp-16-8053-2016>, 2016.
- Lelieveld, J. and Dentener, F. J.: What controls tropospheric ozone?, *J. Geophys. Res. Atmos.*, 105, 3531–3551, <https://doi.org/10.1029/1999JD901011>, 2000.
- Lelieveld, J., van Aardenne, J., Fischer, H., de Reus, M., Williams, J., and Winkler, P.: Increasing
765 Ozone over the Atlantic Ocean, *Science* (80-.), 304, 1483–1487, <https://doi.org/10.1126/science.1096777>, 2004.
- Mahajan, A. S., Li, Q., Inamdar, S., Ram, K., Badia, A., and Saiz-Lopez, A.: Modelling the impacts of iodine chemistry on the northern Indian Ocean marine boundary layer, *Atmos. Chem. Phys.*, 21, 8437–8454, <https://doi.org/10.5194/acp-21-8437-2021>, 2021.
- 770 Masclin, S., Frey, M. M., Rogge, W. F., and Bales, R. C.: Atmospheric nitric oxide and ozone at the WAIS Divide deep coring site: a discussion of local sources and transport in West Antarctica, *Atmos. Chem. Phys.*, 13, 8857–8877, <https://doi.org/10.5194/acp-13-8857-2013>, 2013.
- Mastrantonio, G., Malvestuto, V., Argentini, S., Georgiadis, T., and Viola, A.: Evidence of a Convective Boundary Layer Developing on the Antarctic Plateau during the Summer, *Meteorol.*
775 *Atmos. Phys.*, 71, 127–132, <https://doi.org/10.1007/s007030050050>, 1999.
- Mihalikova, M. and Kirkwood, S.: Tropopause fold occurrence rates over the Antarctic station Troll (72° S, 2.5° E), *Ann. Geophys.*, 31, 591–598, <https://doi.org/10.5194/angeo-31-591-2013>, 2013.
- Morgenstern, O., Zeng, G., Luke Abraham, N., Telford, P. J., Braesicke, P., Pyle, J. A., Hardiman,
780 S. C., O'Connor, F. M., and Johnson, C. E.: Impacts of climate change, ozone recovery, and increasing methane on surface ozone and the tropospheric oxidizing capacity, *J. Geophys. Res. Atmos.*, 118, 1028–1041, <https://doi.org/10.1029/2012JD018382>, 2013.
- Murayama, S., Nakazawa, T., Tanaka, M., Aoki, S., and Kawaguchi, S.: Variations of tropospheric ozone concentration over Syowa Station, Antarctica, *Tellus B Chem. Phys. Meteorol.*,
785 <https://doi.org/10.3402/tellusb.v44i4.15454>, 1992.
- Murazaki, K. and Hess, P.: How does climate change contribute to surface ozone change over the United States?, *J. Geophys. Res. Atmos.*, 111, <https://doi.org/10.1029/2005JD005873>, 2006.

Nguyen, D.-H., Lin, C., Vu, C.-T., Cheruiyot, N. K., Nguyen, M. K., Le, T. H., Lukkhasorn, W.,
Vo, T.-D.-H., and Bui, X.-T.: Tropospheric ozone and NO_x: A review of worldwide variation and
790 meteorological influences, *Environ. Technol. Innov.*, 28, 102809,
<https://doi.org/10.1016/j.eti.2022.102809>, 2022.

Ojha, N., Pozzer, A., Akritidis, D., and Lelieveld, J.: Secondary ozone peaks in the troposphere
over the Himalayas, *Atmos. Chem. Phys.*, 17, 6743–6757, [https://doi.org/10.5194/acp-17-6743-](https://doi.org/10.5194/acp-17-6743-2017)
2017, 2017.

795 Oltmans, S. J.: Surface ozone measurements in clean air, *J. Geophys. Res. Ocean.*, 86, 1174–1180,
<https://doi.org/10.1029/JC086iC02p01174>, 1981.

Oltmans, S. J. and Komhyr, W. D.: Surface ozone in Antarctica, *J. Geophys. Res.*, 81, 5359–5364,
<https://doi.org/10.1029/JC081i030p05359>, 1976.

Oltmans, S. J., Johnson, B. J., and Helmig, D.: Episodes of high surface-ozone amounts at South
800 Pole during summer and their impact on the long-term surface-ozone variation, *Atmos. Environ.*,
42, 2804–2816, <https://doi.org/10.1016/j.atmosenv.2007.01.020>, 2008.

Pozzer, A., de Meij, A., Pringle, K. J., Tost, H., Doering, U. M., van Aardenne, J., and Lelieveld,
J.: Distributions and regional budgets of aerosols and their precursors simulated with the EMAC
chemistry-climate model, *Atmos. Chem. Phys.*, 12, 961–987, [https://doi.org/10.5194/acp-12-961-](https://doi.org/10.5194/acp-12-961-2012)
805 2012, 2012.

Pozzer, A., Reifenberg, S. F., Kumar, V., Franco, B., Kohl, M., Taraborrelli, D., Gromov, S.,
Ehrhart, S., Jöckel, P., Sander, R., Fall, V., Rosanka, S., Karydis, V., Akritidis, D., Emmerichs, T.,
Crippa, M., Guizzardi, D., Kaiser, J. W., Clarisse, L., Kiendler-Scharr, A., Tost, H., and Tsimpidi,
A.: Simulation of organics in the atmosphere: evaluation of EMACv2.54 with the Mainz Organic
810 Mechanism (MOM) coupled to the ORACLE (v1.0) submodel, *Geosci. Model Dev.*, 15, 2673–
2710, <https://doi.org/10.5194/gmd-15-2673-2022>, 2022.

Preunkert, S., Ancellet, G., Legrand, M., Kukui, A., Kerbrat, M., Sarda-Estève, R., Gros, V., and
Jourdain, B.: Oxidant Production over Antarctic Land and its Export (OPALE) project: An
overview of the 2010–2011 summer campaign, *J. Geophys. Res. Atmos.*, 117,
815 <https://doi.org/10.1029/2011JD017145>, 2012.

- Pringle, K. J., Tost, H., Message, S., Steil, B., Giannadaki, D., Nenes, A., Fountoukis, C., Stier, P., Vignati, E., and Lelieveld, J.: Description and evaluation of GMXe: a new aerosol submodel for global simulations (v1), *Geosci. Model Dev.*, 3, 391–412, <https://doi.org/10.5194/gmd-3-391-2010>, 2010.
- 820 Reddy, N. S. K., N.V.P., K., K., R. G., G., B., and K., R. R.: Characteristics of atmospheric surface layer during winter season over Anantapur (14.62° N, 77.65° E), a semi-arid location in peninsular India, *J. Atmos. Solar-Terrestrial Phys.*, 216, 105554, <https://doi.org/10.1016/j.jastp.2021.105554>, 2021.
- 825 Reifenberg, S. F., Martin, A., Kohl, M., Bacer, S., Hamryszczak, Z., Tadic, I., Röder, L., Crowley, D. J., Fischer, H., Kaiser, K., Schneider, J., Dörich, R., Crowley, J. N., Tomsche, L., Marsing, A., Voigt, C., Zahn, A., Pöhlker, C., Holanda, B. A., Krüger, O., Pöschl, U., Pöhlker, M., Jöckel, P., Dorf, M., Schumann, U., Williams, J., Bohn, B., Curtius, J., Harder, H., Schlager, H., Lelieveld, J., and Pozzer, A.: Numerical simulation of the impact of COVID-19 lockdown on tropospheric composition and aerosol radiative forcing in Europe, *Atmos. Chem. Phys.*, 22, 10901–10917, 830 <https://doi.org/10.5194/acp-22-10901-2022>, 2022.
- Roeckner, E., Brokopf, R., Esch, M., Giorgetta, M., Hagemann, S., Kornblueh, L., Manzini, E., Schlese, U., and Schulzweida, U.: Sensitivity of Simulated Climate to Horizontal and Vertical Resolution in the ECHAM5 Atmosphere Model, *J. Clim.*, 19, 3771–3791, <https://doi.org/10.1175/JCLI3824.1>, 2006.
- 835 Rolph, G., Stein, A., and Stunder, B.: Real-time Environmental Applications and Display sYstem: READY, *Environ. Model. Softw.*, 95, 210–228, <https://doi.org/10.1016/j.envsoft.2017.06.025>, 2017.
- 840 Rosanka, S., Tost, H., Sander, R., Jöckel, P., Kerkweg, A., and Taraborrelli, D.: How non-equilibrium aerosol chemistry impacts particle acidity: the GMXe AERosol CHEMistry (GMXe–AERCHEM, v1.0) sub-submodel of MESSy, *Egusph.* [preprint], <https://doi.org/10.5194/egusphere-2023-2587>, 2023.
- Saiz-Lopez, A., Mahajan, A. S., Salmon, R. A., Bauguitte, S. J.-B., Jones, A. E., Roscoe, H. K., and Plane, J. M. C.: Boundary Layer Halogens in Coastal Antarctica, *Science (80-.)*, 317, 348–

351, <https://doi.org/10.1126/science.1141408>, 2007.

845 Sanak, J., Lambert, G., and Ardouin, B.: Measurement of stratosphere-to-troposphere exchange in Antarctica by using short-lived cosmonuclides, *Tellus B Chem. Phys. Meteorol.*, <https://doi.org/10.3402/tellusb.v37i2.15005>, 1985.

Schönhardt, A., Begoin, M., Richter, A., Wittrock, F., Kaleschke, L., Gómez Martín, J. C., and Burrows, J. P.: Simultaneous satellite observations of IO and BrO over Antarctica, *Atmos. Chem. Phys.*, 12, 6565–6580, <https://doi.org/10.5194/acp-12-6565-2012>, 2012.

Seinfeld, J. H. and Pandis, S. N.: *Atmospheric Chemistry and Physics: From Air Pollution to Climate Change*, 88–90 pp., <https://doi.org/10.1063/1.882420>, 1998.

Simpson, W. R., von Glasow, R., Riedel, K., Anderson, P., Ariya, P., Bottenheim, J., Burrows, J., Carpenter, L. J., Frieß, U., Goodsite, M. E., Heard, D., Hutterli, M., Jacobi, H.-W., Kaleschke, L., 855 Neff, B., Plane, J., Platt, U., Richter, A., Roscoe, H., Sander, R., Shepson, P., Sodeau, J., Steffen, A., Wagner, T., and Wolff, E.: Halogens and their role in polar boundary-layer ozone depletion, *Atmos. Chem. Phys.*, 7, 4375–4418, <https://doi.org/10.5194/acp-7-4375-2007>, 2007.

Soni, V. K., Sateesh, M., Das, A. K., and Peshin, S. K.: Progress in meteorological studies around Indian stations in Antarctica, *Proc. Indian Natl. Sci. Acad.*, 83, 860 <https://doi.org/10.16943/ptinsa/2017/48954>, 2017.

Stein, A. F., Draxler, R. R., Rolph, G. D., Stunder, B. J. B., Cohen, M. D., and Ngan, F.: NOAA's HYSPLIT Atmospheric Transport and Dispersion Modeling System, *Bull. Am. Meteorol. Soc.*, 96, 2059–2077, <https://doi.org/10.1175/BAMS-D-14-00110.1>, 2015.

Stohl, A., Bonasoni, P., Cristofanelli, P., Collins, W., Feichter, J., Frank, A., Forster, C., 865 Gerasopoulos, E., Gäggeler, H., James, P., Kentarchos, T., Kromp-Kolb, H., Krüger, B., Land, C., Meloan, J., Papayannis, A., Priller, A., Seibert, P., Sprenger, M., Roelofs, G. J., Scheel, H. E., Schnabel, C., Siegmund, P., Tobler, L., Trickl, T., Wernli, H., Wirth, V., Zanis, P., and Zerefos, C.: Stratosphere-troposphere exchange: A review, and what we have learned from STACCATO, *J. Geophys. Res. Atmos.*, 108, <https://doi.org/10.1029/2002JD002490>, 2003.

870 Taraborrelli, D., Cabrera-Perez, D., Bacer, S., Gromov, S., Lelieveld, J., Sander, R., and Pozzer,

- A.: Influence of aromatics on tropospheric gas-phase composition, *Atmos. Chem. Phys.*, 21, 2615–2636, <https://doi.org/10.5194/acp-21-2615-2021>, 2021.
- Tarasick, D. W. and Bottenheim, J. W.: Surface ozone depletion episodes in the Arctic and Antarctic from historical ozonesonde records, *Atmos. Chem. Phys.*, 2, 197–205,
875 <https://doi.org/10.5194/acp-2-197-2002>, 2002.
- Tian, B., Ding, M., Putero, D., Li, C., Zhang, D., Tang, J., Zheng, X., Bian, L., and Xiao, C.: Multi-year variation of near-surface ozone at Zhongshan Station, Antarctica, *Environ. Res. Lett.*, 17, 044003, <https://doi.org/10.1088/1748-9326/ac583c>, 2022.
- Toyota, K., McConnell, J. C., Lupu, A., Neary, L., McLinden, C. A., Richter, A., Kwok, R.,
880 Semeniuk, K., Kaminski, J. W., Gong, S.-L., Jarosz, J., Chipperfield, M. P., and Sioris, C. E.: Analysis of reactive bromine production and ozone depletion in the Arctic boundary layer using 3-D simulations with GEM-AQ: inference from synoptic-scale patterns, *Atmos. Chem. Phys.*, 11, 3949–3979, <https://doi.org/10.5194/acp-11-3949-2011>, 2011.
- Wang, H., Lu, X., Jacob, D. J., Cooper, O. R., Chang, K.-L., Li, K., Gao, M., Liu, Y., Sheng, B.,
885 Wu, K., Wu, T., Zhang, J., Sauvage, B., Nédélec, P., Blot, R., and Fan, S.: Global tropospheric ozone trends, attributions, and radiative impacts in 1995–2017: an integrated analysis using aircraft (IAGOS) observations, ozonesonde, and multi-decadal chemical model simulations, *Atmos. Chem. Phys.*, 22, 13753–13782, <https://doi.org/10.5194/acp-22-13753-2022>, 2022.
- Winkler, P., Brylka, S., and Wagenbach, D.: Regular fluctuations of surface ozone at Georg-von-
890 Neumayer station, Antarctica, *Tellus B*, 44, 33–40, <https://doi.org/10.1034/j.1600-0889.1992.00003.x>, 1992.
- Yang, X., Cox, R. A., Warwick, N. J., Pyle, J. A., Carver, G. D., O'Connor, F. M., and Savage, N. H.: Tropospheric bromine chemistry and its impacts on ozone: A model study, *J. Geophys. Res. Atmos.*, 110, <https://doi.org/10.1029/2005JD006244>, 2005.
- 895 Young, P. J., Naik, V., Fiore, A. M., Gaudel, A., Guo, J., Lin, M. Y., Neu, J. L., Parrish, D. D., Rieder, H. E., Schnell, J. L., Tilmes, S., Wild, O., Zhang, L., Ziemke, J., Brandt, J., Delcloo, A., Doherty, R. M., Geels, C., Hegglin, M. I., Hu, L., Im, U., Kumar, R., Luhar, A., Murray, L., Plummer, D., Rodriguez, J., Saiz-Lopez, A., Schultz, M. G., Woodhouse, M. T., and Zeng, G.:

Tropospheric Ozone Assessment Report: Assessment of global-scale model performance for
900 global and regional ozone distributions, variability, and trends, *Elem. Sci. Anthr.*, 6, 10,
<https://doi.org/10.1525/elementa.265>, 2018.