<u>Dissolved carbon flow to particulate organic carbon</u> Formation of particulate organic carbon from dissolved substrate input enhances soil carbon sequestration

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Abstract. Particulate organic carbon (POC) and mineral-associated organic carbon (MAOC), which are two primary components of the soil carbon (C) reservoir, have different physical and chemical properties and biochemical turnover rates. Microbial necromass entombment is a primary mechanism for MAOC formation from fast-decaying plant substrates, whereas POC is typically considered as the product of structural litter via physical fragmentation. However, emerging evidence shows that microbial by-products derived from labile C substrates can enter the POC pool. To date, it is still unclear to what extent dissolved C can enter the POC pool and how it affects the subsequent long-term SOC storage. To date, it is still unclear to what extent labile substrates contribute to the POC formation and the subsequent long term SOC stock. Our study here, through a ¹³C-labeling experiment in 10 soils from 5 grassland sites as well as a modeling analysis, showed that up to 12.29% of isotopelabeled glucose-C (i.e., dissolved C) was detected in the POC pool. In addition, the glucose-derived POC was correlated withdependent upon ¹³C-MBC and the fraction of clay and silt, suggesting that the flow of dissolved C to POC the POC formation from newly added labile C is dependent on interactions between soil physical and microbial processes. The modeling analysis showed that ignoring the C flow from MBC to POC significantly underestimated soil C sequestration by up to 53.52% by 7.79% 49.51% across the 10 soils. The results emphasize that the soil texture-regulated microbial process, besides the plant structural residues, is a significant contributor to POC, acting as a vital component in SOC dynamics.

Keywords: soil carbon sequestration, soil carbon modeling, particulate organic carbon, dissolved organic carbon, soil carbon input, glucose, grassland, fencing

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1 Introduction

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As the largest terrestrial carbon (C) pool, soil organic C (SOC) plays a vital role in regulating global climate change through C emissions and sequestration (White et al., 2000; Chapin et al., 2011; Wiesmeier et al., 2019; Basile-Doelsch et al., 2020; Bai and Cotrufo, 2022). Carbon from root exudates can be stabilized in the form of mineral-associated organic C (MAOC), while plant residues can enter the soil as particulate organic C (POC), which has different features from MAOCCarbon from root exudates and plant residues can be stabilized in the form of mineral associated organic C (MAOC) or particulate organic C (POC), which have different features (Cotrufo et al., 2019; Sokol et al., 2019b; Lavallee et al., 2020). MAOC is generally small molecular organo-mineral complexes with relatively low C/nitrogen (N) ratios (C/N ratios). Being associated with soil minerals and occluded in the silt- and clay-sized aggregates, MAOC has a longer mean residence time than POC and thus is considered the key to long-term soil sequestration (Baldock and Skjemstad, 2000). In contrast, POC is usually considered As opposed to this, POC is usually considered as the product of physically fragmented structural residues and is more susceptible to external environmental changes (Benbi et al., 2014; Lugato et al., 2021). Although physically roughly-dividing SOC into POC and MAOC is relatively easy to operate, the microbe-mediated SOC dynamics is a continuous process, and it is difficult to separate its biochemical and physical processes completely, making it is difficult to completely separate its biochemical and physical processes (Lehmann et al., 2020). During the gradual decomposition of plant residues, POC encapsulated by microbial by-products can bind silt- and clay-sized soil minerals, forming heavy-POC (or course-MAOC, >53 μm and >1.6 – 1.85 g cm⁻³) (Samson et al., 2020). Heavy-POC is a complex rich in plant residues, microbial products, and soil minerals. With the gradual decomposition of plant residues in the complex center, heavy-POC gradually fragmented as well, becoming a precursor of MAOCSince heavy POC is the hotspot of microbial activities and microaggregates formation, it could also be a precursor of MAOC (Prater et al., 2020; Witzgall et al., 2021). This decomposition step is also included in the model of SOM formation and persistence (Robertson et al., 2019).

Dissolved C input from living root_and the rhizodeposits, which has a_dominant effect on the_net formation of SOC, is considered approximately 2 to 13 times more efficient than litter inputs in forming SOC (Sokol et al., 2019b). The Microbial Efficiency-Matrix Stabilization (MEMS) framework also suggests that labile plant C inputs are a major source of microbial products, which are more efficiently utilized by microorganisms than recalcitrant ones (Cotrufo et al., 2013). However, the labile C input also plays a critical role in destabilizing SOC as well (Kuzyakov et al., 2000; Keiluweit et al., 2015). Most of the literature emphasizesIt is commonly believed that small-molecular labile plant substrates with low molecular weight – such as glucose and other dissolved C – are primary sources of MAOC through physical absorption and microbial *in vivo* turnover via cell uptake-biosynthesis-growth-death (Bai and Cotrufo, 2022; Mikutta et al., 2019; Sokol et al., 2019a; Liang et al., 2017). However, the potential for microbial products derived from labile C to stick to semi-decomposed plant residues and connect with minerals to become POC has received much less attention. However, it has been paid much less attention how much

microbial products derived from labile C may stick to semi decomposed plant residues and connect with soil minerals to become part of POC.

As an important component of SOC, POC is pivotal <u>in predictingto predict</u> SOC sequestration <u>as well</u>. <u>A few mechanistic models propose POC formation from microbial metabolism, but there is a limited understanding of the factors controlling POC <u>formation Although a few mechanistic models propose the POC formation from microbial metabolism, understanding of factors that control the POC formation is limited</u> (Li et al., 2014; Robertson et al., 2019; Cotrufo and Lavallee, 2022). Specifically, direct evidence is still lacking to what extent dissolved substrate (e.g., glucose) contributes to POC <u>formation</u>. Additionally, how the dissolved substrates-originated POC <u>formation</u> affects SOC sequestration is rarely studied.</u>

Meanwhile, the soil C dynamics are sensitive to land use changes (Del Galdo et al., 2003; Grandy and Robertson, 2007). Overgrazing and conversion of grasslands to farmlands have resulted in significant ecosystem degradation in the grasslands of northern China (Wang et al., 2023; Buisson et al., 2022). Fencing is a widely used strategy in order to retard and reverse the grassland degradation. To date, it has been well-studied that fencing can improve the plant community structure of degraded grasslands, increase species diversity, improve soil structure, promote soil microbial biomass and enzyme activity (Lu et al., 2018; Bardgett et al., 2021). However, how differently dissolved substrates affect POC and MAOC dynamics in fencing and grazing grasslands is still unclear.

In this study, we first collected soil samples from fencing and grazing grasslands from 5 sites (Table 1). Then, we conducted an incubation experiment by adding ¹³C-labeled glucose solution to the 10 soils from 5 grassland sites. At the end of the experiment, glucose-derived ¹³C in dissolved organic C (DOC), microbial biomass C (MBC), POC and MAOC were assessed. Then, we conducted a modeling experiment to simulate SOC dynamics at different C addition scenarios with and without a dissolved C flow from MBC to POC. This study was to answer the following three questions: (i) to what extent the added glucose contributes to the formation of POC? (ii) what factors control the dissolved C flow to POC in the fencing and grazing grasslands across sitesthe POC formation from glucose? (iii) how does dissolved substrates-originated POC formation-affect SOC sequestration? To answer the questions, we had three hypotheses. First, dissolved C can get into the POC pool in addition to the MAOC pool due to interactions between soil physical and biochemical processes. Second, the rate of POC conversion from glucose is dependent upon microbial activity due to the land use change across sites. Finally, adding the pathway from dissolved C input to the POC pool can promote microbial C use efficiency, further enhancing SOC sequestration.

2 Materials and Methods

2.1 Soil sampling

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In August 2021, 10 soils were sampled from 5 temperate grasslands of the Inner Mongolian Plateau, China (Table 1). <u>Before</u> sampling, we measured the plant aboveground biomass using the dry weighing method. At each site, soils of the top 20 cm

layer were sampled from continuous grazing grassland and grazing excluded (i.e., fencing) grassland, respectively. Before incubation, all soil samples were passed through a 2 mm sieve to remove visible stones, roots, and other plant debris. After homogenization, soil texture, pH, SOC, and MBC and DOC content were measured (See methods below; Table 1 and Fig. S2). All soil samples were stored at -20 °C until the incubation experiment started.

2.2 Incubation experiment

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For each soil, ¹³C-labeled glucose addition treatments were performed and four replicates were conducted. Soil samples equivalent to 20 g air-dried soil were added to 250 ml mason jars. All soils were incubated in the dark at 25 °C and a relative humidity of 60% for 102 days. To maintain soil moisture at 60% water holding capacity (WHC), we added distilled water regularly by measuring the weight changes of the jars which were covered by a sealing film passable for gases but not water molecules. After a 7-day pre-incubation, ¹³C-labeled glucose (99 atom% ¹³C, Shanghai Engineering Research Center of Stable Isotope) was added at a dose of 0.4 mg C g⁻¹ soil was added. The glucose solution was prepared by dissolving 0.5 g of glucose in 50 ml of water to make a 10 mg ml-1 solution. Further, 2 ml of glucose solution was slowly dripped into the soil using a pipette gun to keep the solution as uniformly distributed in the soil as possible. Correspondingly, 2 ml of water was added to the control. On days 1, 3, 6, 12, 19, 34, 47, 78, and 102 of the incubation, each jar was flushed by CO₂-free air for 3 minutes. After that, the CO₂ efflux emission rate was measured using an infrared gas analyzer (Li-8100A; Li-COR, USA) within 3 minutes from the headspace. Subsequently, we used the soil CO₂ emission data for model calibration and validation. After the last gas measurement, soils were destructively harvested and stored at -80°C for the subsequent measurements.

2.3 Measurements of DOC, MBC, POC and MAOC

The chloroform-fumigation-extraction method was used to determine DOC and MBC contents (Vance et al., 1987). One subsample of 5 g fresh soil was fumigated by chloroform in the dark for 24h, and a second subsample (5g) was unfumigated as the control. Soil microbes died after 24 hours of chloroform fumigation, and their cells lysed and released microbial biomass C. The soil was extracted with 0.5M K₂SO₄ solution subsequently. The dissolved C in the extracting solution was determined by a rapid CS analyzer (Multi N/C 3100, Analytik jena, Germany). The DOC content was calculated according to the organic C content of unfumigated soil. The MBC content was the difference of DOC between fumigated and unfumigated soils multiplying by the proportionality coefficient of 0.45.

The POC and MAOC content were assessed through the particle size fractionation method, which separates SOC into these two pools. Soil samples (10g) were shaken with 30 mL of sodium hexametaphosphate solution (NaHMP, 50 g L⁻¹) at 200 rpm. After 18h, samples were washed with deionized water over a 53 μm sieve in a vibratory shaker (AS 200 control, Retch, Germany) (Sokol et al., 2019b). Both fractions were dried at 65 °C, weighed, and fumigated with hydrochloric acid for 8h to remove inorganic C. Organic C content was determined by an elemental analyzer (rapid CS cube, elementar, Germany). The C from less than 53 μm fraction was considered MAOC, and the >53 μm fraction was considered POC the other was POC.

2.4 ¹³C partitioning

To analyze the MBC-¹³C concentration, an 8-ml extracting solution from each fumigated and unfumigated soil was freezedried, and approximately 8 mg of K₂SO₄-C was analyzed using an Isotope Ratio Mass Spectrometer (Delta V Advantage, ThermoFisher Scientific, America). The atom% of MBC in control and treated soils was determined using a two-pool mixing model (Fang et al., 2018):

$$at\%_{MBC} = \frac{at\%_{fumigated} \cdot C_{fumigated} - at\%_{unfumigated} \cdot C_{unfumigated}}{C_{fumigated} - C_{unfumigated}}, \qquad (1)$$

where $C_{fumigated}$ and $C_{unfumigated}$ are the C mass in fumigated and unfumigated samples, and $at\%_{fumigated}$ and $at\%_{unfumigated}$ are the C isotope abundance (in atom% 13 C) of the fumigated and non-fumigated samples, respectively.

To analyze the content of POC-¹³C and MAOC-¹³C, approximately 2 mg of wet-sieved and oven-dried soil samples were determined by the Isotope Ratio Mass Spectrometer. Further, the contributions of glucose-derived C to the DOC, MBC, POC, and MAOC pools were estimated following the isotopic mixing model:

$$C_{glucose-derived} = C_{total} \cdot \frac{at\%_{treatment} - at\%_{soil}}{at\%_{glucose} - at\%_{soil}}, \tag{2}$$

$$C_{soil} = C_{total} - C_{glucose-derived} , (3)$$

Where $at\%_{treatment}$, $at\%_{soil}$, $at\%_{glucose}$ are the C isotope compositions (in atom% ¹³C) of the glucose-treated soil, original soil, and added glucose, respectively; $C_{glucose-derived}$, C_{soil} and C_{total} are the glucose-derived, soil-derived C and total SOC content (mg C g⁻¹ soil) in the glucose-treated soil, respectively.

2.5 Modeling analysis

The SOC dynamics was were simulated using two mechanistic models. Most parts of the two models were identical except that Model I did not include the C flow from MBC to heavy-POC, but Model II did (Fig. 1). Model I assumed that plant structural residues were the only POC source, whereas Model II assumed that heavy-POC could be from both plant and microbial residues. Thus, dissolved C can be transformed to heavy-POC via microbial metabolism in Model II. The two models shared a similar structure:

$$\frac{dX(t)}{dt} = AKX(t) , \qquad (4)$$

where

$$X(t) = \begin{bmatrix} x_D \\ x_B \\ x_H \\ x_L \\ x_M \end{bmatrix} X(t) = \begin{bmatrix} \frac{x_P}{x_B} \\ \frac{x_P}{x_M} \\ \frac{x_{PP}}{x_{PM}} \end{bmatrix},$$
(5)

and

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$$K = \begin{bmatrix} k_D & & & & \\ & k_B & & & \\ & & k_H & & \\ & & & k_L & \\ & & & & k_M \end{bmatrix} K = \begin{bmatrix} k_D & - & - & - \\ - & k_B & - & - \\ - & - & k_D & - \\ - & - & - & k_M \end{bmatrix},$$

In Model I,

$$A = \begin{bmatrix} -1 & f_{DH} & f_{DL} & f_{DM} \\ f_{BD} & -1 & & & \\ & & -1 & & \\ & & & -1 & \\ & & & & -1 \end{bmatrix} \underbrace{A = \begin{bmatrix} -1 & - & f_{DP} & f_{DM} \\ f_{BD} & -1 & - & - \\ - & - & -1 & - \\ - & f_{MB} & f_{MP} & -1 \end{bmatrix}}_{(7)},$$

whereas in Model II

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$$A = \begin{bmatrix} -1 & f_{DH} & f_{DL} & f_{DM} \\ f_{BD} & -1 & & & \\ & f_{HB} & -1 & & \\ & & & -1 \\ & & f_{MB} & f_{MH} & & -1 \end{bmatrix} \underbrace{A = \begin{bmatrix} -1 & - & f_{DP} & f_{DM} \\ f_{BD} & -1 & - & - \\ - & f_{PB} & -1 & - \\ - & f_{MB} & f_{MP} & -1 \end{bmatrix}}_{(8)},$$

In matrix $X_{,}$ X_{D} , X_{B} , X_{H} , X_{L} , X_{M} , X_{B} , X_{B} , X_{B} , X_{M} are the pool sizes of DOC, MBC, heavy-POC, light-POC POC and MAOC, and k_{D} , k_{B} , k_{H} , k_{L} , k_{P} , k_{M} in matrix K are their turnover rates, respectively. In matrix A, f_{BD} means the fraction transfer from the DOC pool to the MBC pool, other transfer coefficients f represent in the same way (See details in Table 2). The measured DOC and MBC before incubation were used as their respective initial pool sizes, whereas a to-be-determined parameter $f_{heavy-POC}$ was used to represent the initial fraction of heavy-POC – i.e., initial POC = $(SOC - DOC - MBC) \times f_{heavy-POC}$. Correspondingly, the initial light-POC pool size was calculated as $(SOC - DOC - MBC) \times f_{light-POC}$.

and the initial MAOC pool size was calculated as $(SOC - DOC - MBC) \times (1 - f_{heavy-POC} - f_{light-POC})_2$ fp was used to represent the initial fraction of POC i.e., initial POC = $(SOC - DOC - MBC) \times f_p$. Correspondingly, the initial MAOC pool size was calculated as $(SOC - DOC - MBC) \times (1 - f_p)$. Overall, Model I had 10 - 13 and Model II had 11 - 14 to-bedetermined parameters (Table 2). Because the glucose addition was ¹³C-labelled, each C pool was further divided into soil-derived and glucose-derived pools. We considered all glucose addition entered the glucose-derived DOC pool in the beginning.

The models were calibrated <u>using soil C pools and CO₂ emission rate datausing the incubation experiment</u> through the adaptive Metropolis algorithm (Haario et al., 2001; Hararuk et al., 2014). <u>The CO₂ emission data were divided into two groups: 7 out of the 9 flux measurements for each soil were randomly selected for the model calibration, while the other 2 measurements were used for the model validation. The prior probability density functions (PDFs) were assumed as uniform distributions over parameter ranges based on previous studies (Li et al., 2014; Liang et al., 2015). The parameters' posterior PDFs were proportional to the prior PDFs and a cost function from data. The cost function was calculated as:</u>

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$$P(Z \mid \theta) \propto \exp\left\{-\sum_{t \in \text{obs}(Z)} \frac{\left[O_f(t) - M_f(t)\right]^2}{2\sigma_f^2(t)} - \sum_{i \in \text{obs}(Z)} \frac{\left[O_p(i) - M_p(i)\right]^2}{2\sigma_p^2(i)}\right\}, \tag{9}$$

where t denotes the measurement time of fluxes and t denotes C pools. σ^2 is the standard deviation of measurements. O_f and M_f are the observed and modelled respiration CO₂ emission fluxes. O_p and M_p are the observed and modelled values of C pools. After the model calibration and validation, we randomly select 100 sets of parameters for further modeling experiments. For each model, we set up two C input scenarios, DOC input only and DOC+POC input. The amount of C input was approximately equivalent to local annual C influxes (Table S1). The calibrated models were run to the steady states to compare the modelled SOC change under different scenarios. After that, the models were run along a gradient of C input increase from 1% to 20% with a 1% interval to reach another steady state. Then the impact of C flow from MBC to heavy-POC (i.e., $f_{HB}f_{PB}$) on long-term SOC sequestration was assessed by comparing the behaviors of SOC dynamics between Model I and Model II.

2.6 Statistical analysis

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The two-way analysis of variance (ANOVA) was used to reveal the effects of effects of sites, fencing, and their interaction on plant aboveground biomass, initial MBC, SOC, soil texture (Table S2), and glucose-derived SOC, MAOC, POC, MBC, and DOC, and cumulative respiration (Table \$2\$3). The differences between fencing and grazing treatment and the The differences caused by f_{PHB} between Model I and Model II were tested using the one-way ANOVA. All data were separately tested for normality using the Shapiro–Wilk test and for homoscedasticity using the Bartlett's test in advance. In cases where the assumptions of normality or homoscedasticity were not met, a reciprocal transformation was applied to the original data, and analyses were carried out on the transformed data. In cases where the reciprocal transformed data did not meet the test

requirements, the Kruskal-Wallis test was applied. The difference was considered statistically significant at the level of P < 0.05. The statistical was analyses were performed in R 4.1.2. The model was performed in Matlab 2021a.

190 **3 Results**

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3.1 Effects of fencing and sites on C sequestration

Analysis of different soils and plant investigation data showed that fencing and sites significantly affect plant aboveground biomass, MBC, SOC, and soil texture (Table S2). Generally, plant aboveground biomass, MBC, and SOC were significantly increased after fencing (Fig. S1, S2). For the new C sequestration, sites had significant effects on the sequestration of each C pool and respiration, in which glucose-derived MAOC and POC at HL site was significantly higher than that at other sites (Table S3, Fig. S3). Fencing also significantly affected the amount of glucose C entering MAOC as well as the cumulative soil respiration, in which fencing soils show a lower amount of MAOC sequestration and higher soil respiration (Table S3, Fig. S3, S4).

3.2 Effects of dissolved carbon inputs on C sequestration

Across the 10 soils, 84.28 –175.80 mg kg⁻¹ soil of the glucose C (equivalent to 21.07% – 43.95% of the initial glucose addition) retained in the soil after the 102-day incubation, among which 1.58% – 28.00%, 48.73% – 75.51%, 20.34% – 35.80% of retained glucose 13C distributed in POC, MAOC and MBC, respectively (Fig. 2). At the end of incubation, the proportion of total POC that is from glucose C was 0.16% – 0.67%. Across the 10 soils, 84.28 – 175.80 mg kg⁻¹ soil of the glucose C stayed in the soil after 102 days' incubation, with significant effects of site and fencing (Table S2, Fig. S1). Specifically, glucosederived MBC, MAOC and POC were 18.68 – 51.44, 59.96 – 100.11 and 1.33 – 49.14 mg kg⁻¹ soil, respectively (Fig. 2). Additionally, glucose-derived MAOC and POC were dependent uponcorrelated with glucose-derived MBC (Fig. 3a). Furthermore, glucose-derived MAOC and POC increased with the fraction of clay and silt (*R*² = 0.62 and 0.92, respectively, Fig. 3b).

The estimated C pool turnover rates were lower but the transfer coefficients among different C pools were greater in Model I than Model II (Table S3). On average, Model I underestimated k_D by 2.15%, k_B by 11.9%, k_D by 7.13%, k_M by 5.08%, whereas overestimated f_{MB} by 2.36%, f_{MP} by 4.47%, f_{DP} by 1.23%, f_{DM} by 2.77%, f_{BD} by 0.67%. Although both models fitted respiration flux data well (Fig. S2), Model I, without the C flow from MBC to POC, was not able to reproduce the observed glucose derived POC (Fig. S3). The absence of the f_{HB} affects other parameters differently (Table S5). On average, compared to Model II, Model I showed greater k_L , k_M , f_{BD} , f_{MB} , f_{DM} , but smaller k_D , k_B , k_H , f_{DL} , f_{DH} , f_{MH} . Although both models fitted respiration flux data well (Fig. S5), Model I, without the dissolved C flow from MBC to POC, was not able to reproduce the observed glucose-derived POC (Fig. S6). The estimated C pool turnover rates were lower but the transfer coefficients among different C pools were greater in Model I than Model II (Table S3). On average, Model I

underestimated k_D by 2.15%, k_B by 11.9%, k_P by 7.13%, k_M by 5.08%, whereas overestimated f_{MB} by 2.36%, f_{MP} by 4.47%, f_{DP} by 1.23%, f_{DM} by 2.77%, f_{BD} by 0.67%. Although both models fitted respiration flux data well (Fig. S2), Model I, without the C flow from MBC to POC, was not able to reproduce the observed glucose derived POC (Fig. S3).

At the steady state, when C input only included DOC (dissolved C input only), SOC content in Model I was 10.04% - 53.52%19.54% - 49.51% less than that in Model II (P < 0.05; Fig. 4). When C input was from both DOC and POC (dissolved and structural C input), excluding dissolved the C flow from MBC to POC in Model I decreased SOC content by 7.79%—up to 44.2448.02% compared to Model II by 7.79%—44.24% (P < 0.05; Fig. 4). The effect of microbe-derived POC on SOC sequestration still existed when C input increased. Along with the C input gradient, the SOC difference between the two models was enlarged (Fig. S4S7). When DOC input increased by 20%, the SOC increases (normalized to their respective steady state) were 0.08 - 4.40 Mg C ha⁻¹ soil0.23 mg g⁺ 3.68 mg g⁺ soil in Model I and 0.13 - 9.53 - Mg C ha⁻¹ soil0.32 mg g⁺ 5.13 mg g⁺ soil and Model II (P < 0.05, Fig. S5S8). Similarly, when both DOC and POC input increased by 20%, Model II produced a significantly greater SOC content than Model I (0.31 - 18.47 Mg C ha⁻¹ soil0.96 mg g⁺ 11.33 mg g⁺ soil by Model II vs. 0.21 - 12.55 Mg C ha⁻¹ soil0.84 mg g⁺ 9.18 mg g⁺ soil by Model I; P < 0.05, Fig. S5S8).

4 Discussion

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4.1 Microbe-mediated dissolved C flow to the POC poolformation of POC from dissolved C inputs

This study showed that most of labile C preferentially entered the MAOC pool, but still up to 12.29% of the glucose-C has become part of POC after the 102-day incubation, which is equivalent to 36.49% of the total POC and MAOC sequestration. The results indicate that dissolved labile plant compounds (glucose in our case), in addition to structural litter, could be a significant contributor to POC. Linear regression analyses indicate that glucose C can enter the POC pool via multiple pathways (Craig et al., 2022). The result is supported by Sokol et al. (2019b), which finds that living root inputs are efficiency in forming both MAOC and POC. Specifically, glucose-derived POC is positively addition, the POC formation positively correlated with the glucose-derived MBC (Fig. 3a), suggesting that the transformation of glucose to POC could be dependent on the microbe-mediated biochemical pathway. Meanwhile, glucose-derived POC is positively correlated withthe formation of POC is positively dependent on the fraction of clay and silt as well ($R^2 = 0.92$, Fig. 3b), further indicating that dissolved C entering into POCPOC formation is an interaction of physical and biochemical processes. These results are consistent with previous studies, which showed the formation of heavy-POC (or coarse-MAOC) from microbial by-products binding with the silt- and clay-sized soil minerals (Samson et al., 2020). From a microscopic perspective, ω 0 ur results is are supported by previous studies with images from scanning electron microscopy (SEM) and nano-scale secondary ion mass spectrometry (NanoSIMS), which show that microorganisms could absorb to the surface of particulate organic matter (POM) and bind it with mineral (Kopittke et al., 2020; Witzgall et al., 2021). Meanwhile, the higher clay and silt content means the more microaggregates and

more POC protected from decomposition (Wang et al., 2003). These results were used to support the model structure that we next use for prediction, whereby dissolved C inputs enter the heavy-POC pool under the processes of microbes.

The result that labile C can enter the POC pool are inconsistent with the two-pathway framework, which proposes that low-molecular-weight, water-soluble inputs contribute primarily to MAOC formation via the microbe-mediated biochemical pathway, whereas POC is formed primarily from the polymeric structural inputs via the physical transfer pathway (Cotrufo et al., 2015). Our results, combined with previous studies, demonstrate that the biochemical and physical pathways in SOC formation may not be independent with each other. Rather, the formations of POC and MAOC and POC are continuous through close interactions of physical and microbial processes, during which POC originated formation from dissolved substrates is a critical component in SOC dynamics.

4.2 Effect of dissolved substrates-originated POC microbe-mediated POC formation on SOC sequestration

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As POM surfaces are considered the hotspots of microbial activities and the cores of aggregate formation (Tisdall and Oades, 1982; Witzgall et al., 2021), our modeling analyses indicated that dissolved substrates-originated POC formation—can significantly influence long-term SOC sequestration. Although both Model I and Model II fitted the C flux data well, Model I, which does not include the dissolved C flow from MBC to POC, was not able to reproduce the observed POC changes (Fig. S3S6). The results emphasize the necessity of including the process of dissolved C flow to POC the microbe mediated POC formation in SOC dynamic models.

During the model calibration, including or not the <u>dissolved</u> C flow from MBC to POC significantly affected the estimations of turnover and transfer parameters. Specifically, the absence of the f_{HB} enabled more C flow into the MAOC pool, the algorithm tended to mistakenly elevate the turnover rate of MAOC by 12.28% in order to fit the C pool data in short-term incubation. While this does not have a great impact on the short-term data fitting process, it can significantly affect the long-term SOC predictions.—the turnover rates of C pools were underestimated and transfer coefficients were overestimated by Model I compared with Model II (Table S3). This is because the absence of C flow from MBC to POC in Model I would allow more C to be allocated to respiration. To alleviate this phenomenon, the algorithm tended to mistakenly decrease their turnover and increase C allocation to other C pools to fit respiration flux data. As a result, the absence of the mechanism of microbe-mediated dissolved C flow to POC leadsPOC formation can have a significant impact on the long term prediction of SOC, leading to an underestimation of SOC sequestration in Model I (Fig. 4). In addition, the underestimation of SOC sequestration would be proportionally exacerbated as the magnitude of C input increases (Fig. \$57, 584). These results indicate that the process of microbe-mediated dissolved C flow to POCPOC formation is critical for long-term SOC sequestration and should be considered in soil C dynamic models.

4.3 Fencing effect on C sequestration and soil respiration from incubation experiment

An additional goal of our study was to explore the mechanisms of soil C sequestration after the fencing management in grassland ecosystems. Many research suggests that appropriate grazing exclusion by fencing in degraded grassland can increase soil C storage, promoting restoration (Bardgett et al., 2021; Lu et al., 2018). Our field results showed that fencing sites had greater SOC and MBC contents (Table S2, Fig. S2). This can be attributed to the increased C input, which stimulates microbial growth and allows more C to stabilize in the SOC pool (Table S2, Fig. S1). However, in the incubation experiment, fencing soils showed greater cumulative respiration and lower MAOC sequestration (Fig. S3 and S4). These inconsistent results between the field observations and the incubation experiment suggest that the increased SOC sequestration by fencing could be primarily due to the C input instead of the C transformation in the soil. Specifically, the observed increases in soil C stocks of fencing grasslands were closely related to the increased plant production and C inputs from grazing exclusion (Fig. S1). Once the C input kept consistent between fencing and grazing soils, multiple linear regression showed that the predictor variable of clay and silt content explained 91.85% of the variance in new SOC sequestration (Table S4). Additionally, the clay and silt content also dominated the magnitude of soil C sequestration across sites (Fig. 3b). Meanwhile, higher cumulative respiration in fencing soils can be explained by initial SOC and soil texture, presenting a positive effect of higher SOC content but negative effect of clay and silt content (Table S4). Moreover, no significant difference of glucose-derived MBC was observed between fencing and grazing soils (Fig. S3c), which further validates that C input is the dominant factor influencing soil microorganisms.

5 Conclusions

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This study provides direct evidence that dissolved labile C input can not only enter MAOC, but also POC ean not only enter MAOC, but also POC enter M

Author contributions

Junyi Liang and <u>Qintana Si Siqintana</u> designed the study. Yaowen Zhang, Xun Sun conducted the soil sampling. <u>Qintana Si Siqintana</u>, Kangli Chen, and Bin Wei conducted the incubation experiment. Junyi Liang and <u>Qintana Si Siqintana</u> developed the modeling framework. Junyi Liang and <u>Qintana Si Siqintana</u> performed the analyses. All the authors contributed to writing the manuscript.

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310 Data availability statement

All data are freely available at https://doi.org/10.6084/m9.figshare.24773205.v1. All data will be freely available upon acceptance.

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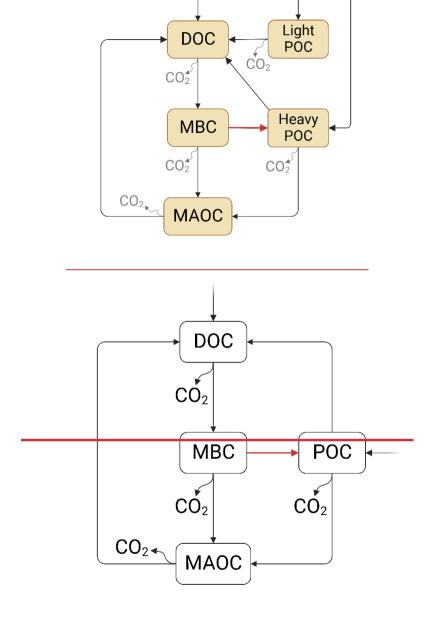
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Table 1: Information of the sampling sites and soil physical and chemical properties (mean±standard error).

Site	Fencing treatment	Abbreviation	Longitude (°E)	Latitude (°N)	Altitude (m)	Mean annual precipitation (mm)	Mean annual temperature (°C)	Clay (%)	Silt (%)	Sand (%)	рН	SOC (g kg ⁻¹)
DL	fencing	<u>DL</u> _{fencing} DL _{in}						9.4	17.7	71.3	7.52 ± 0.07	57.08±3.53
DL	grazing	DL _{grazing} DL _{ou}	116.27	42.06	1306.22	378.00	3.30	10.3	18.3	67.2	7.27 ± 0.20	59.70±1.62
GY	fencing	GY _{fencing} GY _i	115.59	41.78	1391.95	398.40	-1.40	9.2	20.1	70.6	8.02 ± 0.21	40.13±2.67
GY	grazing	GY _{grazing} GY _e	113.37	11.70	1371.75	370.10	1.10	10.0	14.9	74.1	7.57 ± 0.16	31.88±0.76
HL	fencing	<u>HL</u> fencingHLin						5.5	40.3	53.0	6.29 ± 0.12	37.01 ± 1.95
HL	grazing	HLgrazing HLou	120.16	49.44	673.95	352.00	-0.10	14.2	24.6	59.2	6.43 ± 0.08	29.04±1.81
XL	fencing	$\underline{XL}_{\underline{fencing}}\underline{XL}_{\underline{in}}$						5.9	8.7	83.4	6.65 ± 0.16	12.27 ± 1.03
XL	grazing	$\frac{\mathrm{XL}_{\mathrm{grazing}}}{\mathrm{t}}$	116.74	43.60	1198.22	263.50	3.50	8.6	5.0	84.5	6.60 ± 0.19	9.00 ± 0.60
XH	fencing	XH _{fencing} XH _i	114.09	42.37	1224.96	270.60	4.20	4.1	5.8	75.9	7.40 ± 0.15	7.84 ± 0.65
XH	grazing	XH _{grazing} XH _e	114.07	12.37	1224.90	270.00	1.20	4.8	10.9	75.0	7.65 ± 0.13	7.02 ± 0.80

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Parameter	Description	Unit
f _{heavy-POC}	Initial fraction of the heavy-POC pool	=
$f_{light-POC}$	Initial fraction of the light-POC pool	<u>=</u>
f_	Initial fraction of the POC pool	_
k_D	Turnover rate of the DOC pool	mg C g-1 soil h-1
k_B	Turnover rate of the MBC pool	mg C g-1 soil h-1
$k_H \frac{k_P}{k_P}$	Turnover rate of the heavy-POC pool Turnover rate of the POC pool	mg C g ⁻¹ soil h ⁻¹ mg C g ⁻¹ soil h ⁻¹
k_L	Turnover rate of the light-POC pool	mg C g ⁻¹ soil h ⁻¹
k_{M}	Turnover rate of the MAOC pool	mg C g-1 soil h-1
∫_{MB}	MBC to MAOC transfer coefficient	-
$f_{BD}f_{\overline{MP}}$	DOC to MBC transfer coefficient POC to MAOC transfer coefficient	=
f _{MB} f₽₿	MBC to MAOC transfer coefficient MBC to POC transfer coefficient (only exist in model II)	==
$f_{DM}f_{\overline{DP}}$	MAOC to DOC transfer coefficient POC to DOC transfer coefficient	==
$f_{DL}f_{\overline{DM}}$	Light-POC to DOC transfer coefficient MAOC to DOC transfer coefficient	-
f _{DH} f⊎D	Heavy-POC to DOC transfer coefficient DOC to MBC transfer coefficient	-
f_{MH}	Heavy-POC to MAOC transfer coefficient	Ξ
f_{HB}	MBC to heavy-POC transfer coefficient (Only exist in model II)	Ξ





Dissolved

carbon

input

Structural

carbon

input

Figure 1: The model scheme of soil carbon (C) dynamics. Model I and Model II share similar structure except that Model II includes a C flow from MBC to heavy-POC (red arrow) but Model I does not.

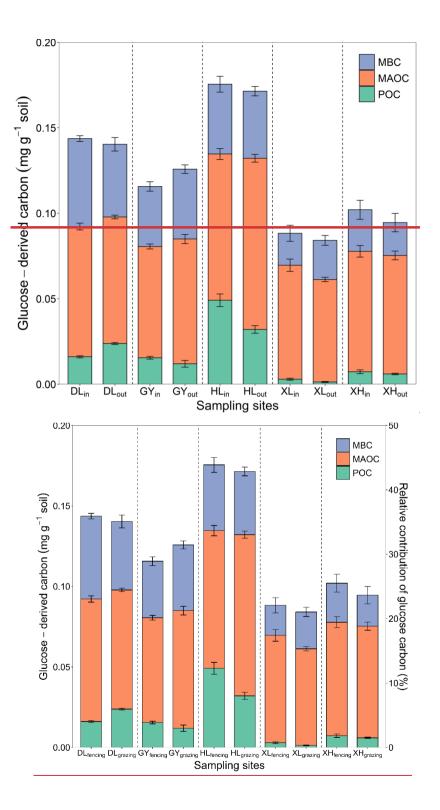


Figure 2: Distributions of glucose-derived C in soil C pools. microbial biomass C: MBC, mineral-associated organic C: MAOC, particulate organic C: POC. The left y-axis is absolute amounts of glucose C into MAOC, POC and MBC pools. The right y-axis is relative contribution of newly stabilized C to total glucose C input. The error bars represent the standard errors of four replicates. The vertical dashed line divides the x-axis into five sampling sites, each with fencing treatment in the first column and grazing treatment in the second column. The error bars represent the standard errors of four replicates. microbial biomass C: MBC, mineral associated organic C: MAOC, particulate organic C: POC.

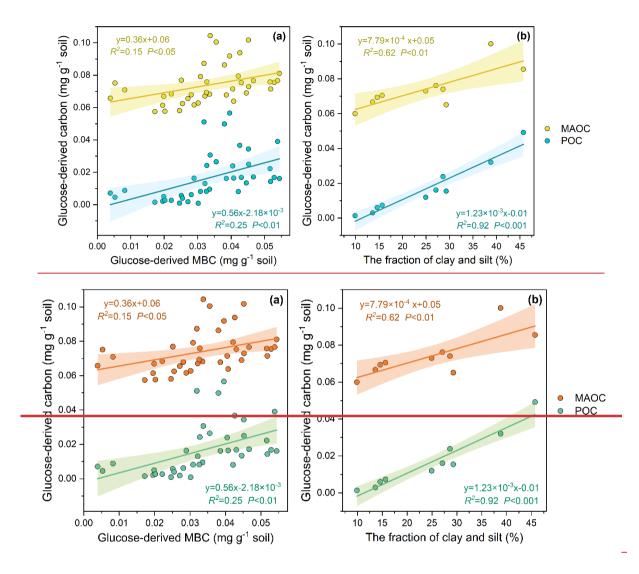


Figure 3: <u>Correlation Dependence</u> of glucose-derived POC and MAOC on MBC (a) and soil texture (b). Shaded areas represent the 95% confidence intervals for the regression lines.

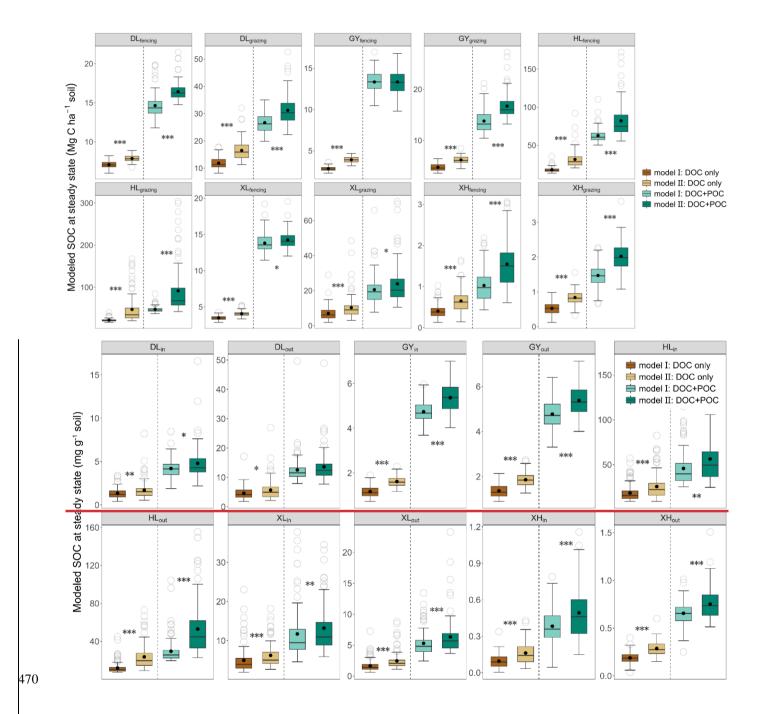


Figure 4: Modeled SOC content at steady state under two types of C input conditions. The two different C input scenarios for each site are separated by a dotted line. The upper and lower ends of boxes denote the 0.25 and 0.75 percentiles, respectively. The solid line and solid dots in the box mark the median and mean of each dataset. Hollow dot The open circles denotes outliers. Asterisks represent significant differences between Model I and Model II (*P < 0.05, **P < 0.01, ***P < 0.001).