

Dear Editor,

We very much appreciate your diligent reading and spotting of typos! Thank you very much also for your efficient handling of the manuscript over the whole review stage.

We followed all your recommendations. Only those items, where we thought an answer would be helpful for you, are answered below.

In excess of the changes you requested, we had to change Figure A1 very slightly, as we realised that the figure version in the revised manuscript had a clipped axis label.

Best wishes,

Simon Unterstrasser (on behalf of the author team)

Abstract: Consider shortening the abstract. The length should be about 250 words (see ACP guidelines) and yours has currently 444 words.

We shortened it considerably from 444 to 357 words.

P5, L135: References should be embedded in the text -> change parenthesis -> Beer et al. (2020)

P8, L213: remaineder -> remainder

P11, L292: You refer here to the previous version of your manuscript? In that case you should not write it like this in the current version of the manuscript. Mentioning this is only important in the reply to the referees, but not in the current version of your manuscript.

This made a reference to our previous paper, not the previous version of the current manuscript. It is already explicitly written in this sentence.

P12, L338: Add here after equation 9 "for $A(t)$ " or for the "effective area $A(t)$ ".

P13, L367: Parenthesis around the reference not correct. It should read "Kulmala et al. (1993).

P13, L366: Eq -> equation (here it should not be abbreviated).

P14, L387: I guess something went wrong here. Who is O? I guess the rest of the last name got somehow lost.

The last name has indeed only one character!

P14, L393 and throughout the manuscript: Tab -> Table. Note usually Table is not abbreviated (see ACP guidelines).

P14, L398: Sec. -> Sect.

P15, L416: Why 3 and 5s? Can you give a motivation why this short time is sufficient?

We added an explanation.

P20, L526 and throughout the remainder of the manuscript: Add always also which figure you are refereeing to. The way to mention the respective panels should be e.g Fig 6a, Fig6b and so on.

P21, L534: appendix -> Appendix

P25, L623: add used -> of the cryoplanes used in our

P25, l624: study -> studies (?)

P27, Figure 6 caption, line 4: remove "the" -> sum of both co-existing ensembles

P31, L685: What is your case 4? Please provide more details on the respective case or give the reference to the respective section.

P35, L779: What is the abbreviation "SIP" standing for?

The abbreviation SIP was introduced in section "3.1 LCM box model"

P35, Figure A caption: Simulation Particles -> simulation particles.

We introduced the abbreviation SIP here again to help the reader.

P36, L796 and 797: check reference.

"O and Wood" is correct.

Contrail formation on ambient aerosol particles for aircraft with hydrogen combustion: A box model trajectory study

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Abstract.

Future air traffic using (green) hydrogen (H_2) promises zero carbon emissions but the effects of contrails from this new technology have hardly been investigated. We study contrail formation behind aircraft with H_2 combustion by means of the particle-based Lagrangian Cloud Module (LCM) box model. Assuming the absence of soot and ultrafine volatile particle formation, contrail ice crystals form solely on atmospheric background particles mixed into the plume. While a recent study extended the original LCM with regard to the contrail formation on soot particles, we further advance the LCM to cover the
5 formation, contrail ice crystals form solely on atmospheric background particles mixed into the plume. While a recent study extended the original LCM with regard to the contrail formation on soot particles, we further advance the LCM to cover the contrail formation on ambient particles. For each simulation, we perform an ensemble of box model runs using the dilution along 1000 different plume trajectories~~that are based on 3D Large Eddy Simulations using the FLUDILES solver.~~

The formation threshold temperature of H_2 contrails is by around 10 K higher than for conventional contrails (which form
10 behind aircraft with kerosene combustion)~~due to a factor of 2.6 higher energy-specific water vapor emission. Therefore, Then,~~ contrail formation becomes primarily limited by the homogeneous freezing temperature of the water droplets ~~formed on the ambient particles~~ such that contrails can form at temperatures down to around 234 K.

The number of formed ice crystals varies strongly with ambient temperature even far away from the contrail formation threshold. The ~~latter is because the water supersaturation in the plume lasts longer for colder conditions and, hence, more of the entrained aerosol particles can form droplets and ice crystals.~~ The contrail ice crystal number clearly increases ~~for a higher~~
15 ~~with~~ ambient aerosol number concentration ~~. The increase becomes weaker for higher number concentrations ($\gtrsim 200 \text{ cm}^{-3}$) and lower ambient temperatures ($< 230 \text{ K}$). The ice crystal number and decreases significantly for ambient particles with mean dry radii $\lesssim 10 \text{ nm}$ due to the Kelvin effect.~~

Besides simulations with one aerosol particle ensemble, we analyze contrail formation scenarios with two co-existing aerosol
20 particle ensembles ~~that differ either in their~~ with different mean dry size or hygroscopicity parameter. We compare them to scenarios with a single ensemble that is the average of the two aerosol ensembles. We find that the total ice crystal number can differ significantly between the two cases, in particular if nucleation mode particles are involved.

Due to the absence of soot particle emissions, the ice crystal number in H_2 contrails is typically reduced by more than 80–90% compared to conventional contrails. The contrail optical thickness is significantly reduced and H_2 contrails either become later
25 visible than kerosene contrails or are not visible at all for low ambient particle number concentrations. On the other hand, H_2 contrails can form at lower flight altitudes where conventional contrails would not form.

1 Introduction

The contribution of aviation to the total anthropogenic climate forcing is estimated to be around 3.5% (Lee et al., 2021). Besides the aircraft CO₂ emissions, contrail cirrus have a large contribution to the aviation radiative forcing (e.g., Boucher et al., 2013; Bock and Burkhardt, 2016b; Bier and Burkhardt, 2022). There are several measures to mitigate the climate impact due to contrail cirrus. One mitigation option is reducing the number of formed contrail ice crystals, which strongly impact the further contrail cirrus life cycle and the radiative forcing (e.g., Unterstrasser and Gierens, 2010; Bier et al., 2017; Burkhardt et al., 2018). This might be achieved by reducing soot particle number emissions, since contrail ice crystals form in particular on soot particles relative to co-emitted organic-sulfate particles for conventional passenger aircraft engines (e.g., Kärcher and Yu, 2009; Kleine et al., 2018). Several ground and flight measurement campaigns have shown significant reductions in engine soot number emissions by using alternative fuel blends with a lower aromatic content (e.g., Moore et al., 2017; Voigt et al., 2021; Bräuer et al., 2021). Switching from the reference Jet A-1 fuel to semisynthetic or biofuel blends, Voigt et al. (2021) and Bräuer et al. (2021) also find significant reductions in young contrail ice crystal numbers by around 20–70%. Burkhardt et al. (2018) and Bier and Burkhardt (2022) emphasize that there is a strong non-linearity between the global contrail cirrus radiative forcing and the young contrail ice crystal number. Hence, even larger reductions in the number of formed ice crystals are desirable to obtain a substantial mitigation effect.

(Green) hydrogen (H₂) combustion is a promising technology to reduce the overall aviation climate impact. It provides around 3 times more energy per fuel mass than kerosene fuel (Najjar, 2013), but it delivers much less energy by volume at typical atmospheric conditions. Hence, H₂ is typically brought to liquid phase at 20 K and stored in special tanks of the cryoplane. During H₂ combustion, the main emission product is water vapor and its emission is roughly a factor of 2.6 larger compared to kerosene for the same amount of released combustion energy and a similar propulsion efficiency (e.g., Schumann, 1996). Increased water vapor emissions in the stratosphere would cause a significant radiative warming (Pletzer et al., 2022), but this impact would be low as long as the aircraft fly at altitudes in the troposphere (e.g., Wilcox et al., 2012). While NO_x is still produced due to high flame temperatures, we expect neither direct CO₂ nor soot particle emissions during H₂ combustion. However, it was observed in laboratory studies that the emission of lubricant oil vapors can lead to the formation of ultrafine volatile particles (Ungeheuer et al., 2022). In ground field measurements, lubrication oil droplets with volumetric mean dry radii ranging between around 125–175 nm were observed by sampling directly from the breather vents (Yu et al., 2010). Furthermore, Yu et al. (2012) performed the first field study that investigates in-flight lubrication oil emissions behind a commercial aircraft. Thereby, they find a significant contribution of lubrication oil constituents in organic particulate matter emissions from the engine exhausts that are typically associated with high soot number emissions.

Up to now, measurements on H₂ contrails do not exist. Airbus and DLR are planning measurements behind a glider equipped with a small H₂ combustion engine within the Blue Condor campaign (Airbus, 2022). Moreover, Airbus aims at establishing the world's first commercial aircraft based on hydrogen propulsion by 2035 within the "ZEROe" project. Marquart et al. (2005) and Ponater et al. (2006) estimated the radiative forcing (RF) of line-shaped contrails for a hypothetical fleet of cryoplanes in comparison with a conventional fleet within a global climate model (GCM). They found similar RF values for both types of

fleets. The decrease in optical thickness for H₂ contrails (RF down) was roughly balanced by the larger contrail coverage (RF up). This estimate is based on a simple parameterization of line-shaped contrails (Ponater et al., 2002), where e. g. the contrail cover scales with the contrail formation frequency and the ice water content is simply diagnosed by the atmospheric water vapor available for deposition. Recent GCM contrail parameterizations are more advanced as they simulate the full contrail
65 (cirrus) life cycle, treat contrails as a separate cloud class to natural clouds and introduce contrail ice water content, coverage and ice crystal number as prognostic variables (Bock and Burkhardt, 2016a; Bier and Burkhardt, 2022).

The hot exhaust plume behind the aircraft engines continuously expands and cools due to entrainment of ambient air. Under certain atmospheric conditions and depending on specific engine/fuel parameters, the plume humidity temporally surpasses water-saturation in the early jet phase and enables the formation of contrails. This condition is described by the Schmidt-
70 Appleman (SA)-criterion (Schumann, 1996) which is based purely on the thermodynamics of the plume mixing process. If the SA-criterion is fulfilled, plume particles can activate into water droplets (e.g., Kärcher and Yu, 2009; Kärcher et al., 2015). They subsequently turn to ice crystals by homogeneous freezing if ambient temperature is below the homogeneous freezing temperature. Switching to H₂ combustion with expected soot-free emissions, contrail ice crystals can still form on upper tropospheric (UT) background particles that are entrained into the plume (e.g., Kärcher et al., 1996, 2015).

75 Some recent box model studies and analytical approaches (Kärcher and Yu, 2009; Kärcher et al., 2015; Bier and Burkhardt, 2019) have already included ice crystal formation on ambient particles mixed into the plume. They show that this process will become relevant if soot number emissions from conventional aircraft engines are reduced by at least 2 orders of magnitude (referred to as "soot-poor emissions"). Kärcher (2018) estimate a decrease in contrail ice crystal number by around 1-2 orders of magnitude when switching from conventional to soot-poor emissions at ambient temperatures for which ice crystals cannot
80 form on ultrafine volatile particles. While those studies in general assumed fixed ambient particle properties, Lewellen (2020) investigated contrail formation on ambient aerosol (besides soot and volatile particles) in a box model and Large Eddy Simulations (LES) and varied the ambient aerosol number concentration. Finally, we expect a high uncertainty in the estimated H₂ contrail ice crystal number due to a large variability in atmospheric particle properties (e.g., Minikin et al., 2003; Hermann et al., 2003; Brock et al., 2021; Voigt et al., 2022), which has not been examined in sufficient detail before. Moreover, it is
85 not clear whether ultrafine volatile particles originating from lubrication oils and NO_x emissions play a role in droplet and ice crystal formation.

The contrail formation studies mentioned in the preceding paragraph have been performed only for fuel/engine parameters that represent the kerosene case and considered the competition between ambient aerosol, soot and volatile particles. Ström and Gierens (2002) is the only contrail evolution study considering aircraft with H₂ combustion. They performed 2D simulations
90 of young contrails including the contrail formation process in the jet phase. They prescribe a bi-modal log-normal aerosol size distribution and vary the aerosol number concentration as the most relevant input parameter. They employ a bulk approach for the treatment of the ice microphysics and simulate the homogeneous freezing on wetted aerosol particles. They do not use any solubility model and simply assume that the background aerosol particles are composed of ammonium sulfate.

In the present study, we aim at providing a basic understanding of the processes regarding the contrail formation on ambient
95 particles ("H₂ contrails"). Moreover, we will highlight main differences compared to conventional contrails where ice crystals

mainly form on soot particles. We will also explain the impact of the increased water vapor emission due to H₂ combustion on the contrail formation criterion and the thermodynamic plume properties. Our main objective is to explore the variability in contrail properties (in particular ice crystal number) due to the variability in atmospheric parameters on the one hand and due to the variability in ambient aerosol particle properties on the other hand. While previous studies focused only on the variation of the aerosol number concentration, we also investigate the impact of the mean aerosol dry size and the solubility. Moreover, we will analyze the impact of the competition of two co-existing ambient aerosol particle ensembles instead of a single one on contrail ice nucleation. Finally, we will compare the number of formed ice crystals and optical thickness of H₂ contrails with conventional contrails.

2 Background and state of the art

This section provides a basic summary over the observed and modeled aerosol particle properties and then explains the impact of H₂ combustion on the thermodynamic contrail formation criterion.

2.1 Observed and modeled aerosol particle properties

The major source of UT aerosol particles are natural and anthropogenic emissions of gaseous aerosol precursors that are transported from lower altitudes by vertical updrafts like synoptic scale lifting or deep convection (e.g., Minikin et al., 2003), and form particles due to chemical ion nucleation (e.g., Lee et al., 2003). Another important source is the in-situ formation, caused by mixing processes and aircraft emissions (e.g., Hermann et al., 2003). Aviation contributes about 30–40% of the particle number concentration in the northern mid-latitudes' UT between 7 and 12 km (Righi et al., 2013). The major relevance of ambient aerosol particles for contrails is likely over the high density air traffic regions like Central Europe, the Eastern USA and North Atlantic where contrails frequently form. This relevance will increase in the near future when first hydrogen engines become available (Righi et al., 2016). On the other hand, future atmospheric conditions are likely to have a reduced aerosol content due to the long term pursuit of a cleaner atmosphere (Andreae et al., 2005).

Currently, there are still few observations of UT aerosol particle properties available and we here provide a short summary of some important measurement campaigns: Minikin et al. (2003) investigated within two flight campaigns spatial distributions and vertical profiles of aerosol number concentrations both over the northern hemisphere (NH) and over the mid-latitudes of the southern hemisphere (SH) in the UT. As displayed in their [Tab. Table 1](#), the measured number concentrations in the Aitken mode range from 130 to 400 cm⁻³ (290 to 9600 cm⁻³) in the SH (NH) those in the accumulation mode from 6 to 43 cm⁻³ (24 to 480 cm⁻³) in the NH (SH). In several measurement flights, Petzold et al. (2002) observed aerosol particle properties over eastern Germany in summer 1998 at altitudes from ground level to 11 km within the "Lindenberg Aerosol Characterization Experiment (LACE 98)". In addition to number concentrations, they derived aerosol particle size distributions at different altitudes (see their Fig. 5). In the considered UT and tropopause region, the smallest measured particle sizes (radius ≈ 50 nm) were the most abundant. Large data sets of aerosol particle number densities were acquired in the UT/lower stratosphere (LS) in the subtropics, the tropical tropopause and the mid-latitudes during the "SCOUT-O3", "SCOUT-AMMA"

and "TROCCINOX" campaigns (Borrmann et al., 2010). They reveal a large variability with number densities between 100 and more than 1000 cm^{-3} in the altitude range of 9 to 12 km. Brock et al. (2021) performed in-situ measurements of aerosol properties as part of the Atmospheric Tomography Mission (ATom) from 2016-2018 in particular over the Atlantic and Pacific Ocean. They show in their Fig. 12 vertical profiles of aerosol number concentration as well as fitted log-normal geometric diameter and geometric width of the size distribution for the Nucleation, Aitken, Accumulation and Coarse mode particles. Additionally, Cloud Condensation Nucleii (CCN)-concentrations were measured at different water-supersaturations (Fig. 14), which show a slight increase in the UT above 10 km and vary strongly with latitude. While these measurements are the most recent and comprehensive, they were mainly taken outside the main air traffic regions. (Beer et al., 2020) Beer et al. (2020) compared aerosol profiles above the North American continent and Europe to model data. They show (in their Supplement) altitude profiles of number concentrations with average values between 200 and 300 cm^{-3} for Condensation Nuclei (CN) with dry radii larger than 5 nm. Recently, long term aerosol measurements from a commercial aircraft platform within the "Civil Aircraft for Regular Investigation of the Atmospheric Based on an Instrument Container (CARIBIC)" project (Hermann et al., 2003) were compared with measurements over Europe during the Covid-19 pandemic (Voigt et al., 2022). Due to massive reductions in aviation and industrial emissions during the pandemic, significant reductions in aerosol number concentrations were observed in the UT potentially reflecting future low emission scenarios.

The chemical composition of aerosol particles is of great importance and impacts several microphysical processes like hygroscopic growth and activation into water droplets. Liu et al. (2014) investigated hygroscopic properties of CCN based on their chemical composition in the North China Plain. They derived the hygroscopicity parameter (κ), introduced in the solubility model from Petters and Kreidenweis (2007), of 16 relevant inorganic salts and sulfuric acid. Thereby, a higher κ -value is associated with a higher solubility of the aerosol species. Sulfuric acid and most of the inorganic salts have $\kappa > 0.5$. In other studies, the κ -value of water soluble organic carbon is estimated to be around 0.3 (e.g., Padro et al., 2010) and that of freshly emitted aviation soot is close to zero (e.g Petzold et al., 2005; Kärcher et al., 2015). Pre-activated soot particles (e. g., by contrail ice in their pores) can be more water-soluble but also serve as heterogeneous ice nucleii (e.g., Marcolli, 2017). Composition measurements using single particle mass spectrometry investigating the size resolved mixing state of aerosol have gained much attention and provide the source for estimates on the hygroscopicity of background aerosol in the UT/LS (Froyd et al., 2019; Tomsche et al., 2022; Schmale et al., 2010).

Besides observation campaigns, climate models with aerosol physics (e.g., Stier et al., 2005; Kaiser et al., 2019) have been developed to simulate chemical formation and microphysical processes of aerosol particles. These models have been evaluated with observations and can be used for the investigation of the global aerosol climatology (e.g., Beer et al., 2020). Among others, they highlight the large spatio-temporal variability of aerosol particle properties in terms of their number concentration, size distribution and chemical composition. In this work, we investigate the sensitivity to these parameters and their relevance for H_2 contrail properties.

160

2.2 Contrail formation criterion

Behind an aircraft engine, the hot and moist plume air mixes with the colder ambient air and the plume is continuously diluted. The so called "mixing line" describes the linear dependency between the partial vapor pressure and excess temperature in the plume. The Schmidt-Appleman (SA)-criterion is fulfilled for a sufficiently low ambient temperature such that the mixing line
 165 crosses the saturation vapor pressure over liquid water and hence the plume becomes water-supersaturated in a particular time period (Schumann, 1996). It is a necessary condition for contrail formation and has been empirically validated by several flight campaigns for kerosene combustion (e.g., Busen and Schumann, 1995; Schumann et al., 2002). The SA-threshold temperature (Θ_G) is the largest ambient temperature for which water saturation is still reached in the plume (Schumann, 1996). It depends on the ambient relative humidity over water and the slope of the mixing line

$$170 \quad G = \frac{EI_v c_p p_a}{0.622 Q (1 - \eta)}, \quad (1)$$

where c_p is the specific heat capacity, p_a the ambient pressure, EI_v is the exhaust water vapor (mass) emission index, Q the specific combustion heat, η the propulsion efficiency and EI_v/Q is called the energy-specific water vapor emission index. The calculation of Θ_G is described in Appendix of Schumann (1996).

fuel/engine parameters	kerosene	hydrogen	ratio
$EI_v / \text{kg kg}^{-1}$	1.26	8.94	7.10
$Q / \text{MJ kg}^{-1}$	43	120	2.79
$EI_v/Q / \text{kg MJ}^{-1}$	0.029	0.075	2.57
η	0.36	0.36	1

Table 1. Fuel and engine parameters for kerosene (2nd column), hydrogen propulsion (3rd column) and the ratio between both (last column). The water vapor mass emission index and specific combustion heat are based on [Tab. Table 1](#) of Schumann (1996). The propulsion efficiency is fixed for both fuel types to a value typical of an A340 aircraft according to Vancassel et al. (2014) and Bier et al. (2022).

Fig. 1 shows that the SA-threshold temperature generally increases with rising relative humidity over water (RH_{wat}) on the
 175 one hand and with increasing ambient pressure on the other hand. The parts of the curves lying above the solid black line depict the ice-supersaturated cases supporting persistent contrails. In the following, we compare Θ_G for hydrogen combustion (blue lines) with those for kerosene combustion (red lines). Using the parameters from [Tab. Table 1](#), EI_v is around 7.1 and Q 2.8 times higher for the hydrogen than for the kerosene case (see also [Tab. Table 1](#)). This leads to an overall increase in the slope of the mixing line (Eq. (1)) by a factor of $EI_v/Q \approx 2.6$ for fixed η and ambient pressure. As a consequence, Θ_G is by around
 180 10 K larger for the hydrogen than for the kerosene case (for otherwise fixed conditions). Considering the same atmospheric conditions and ensuring that $T_a < \Theta_G$, this will cause significantly higher (peak) plume water-supersaturation for the hydrogen case in the early jet phase because the difference $\Delta T = |T_a - \Theta_G|$ is accordingly higher (e.g., Kärcher et al., 2015; Bier et al.,

2022). Moreover, droplet formation on aerosol particles will be enabled at higher ambient temperatures as we will show in the results section.

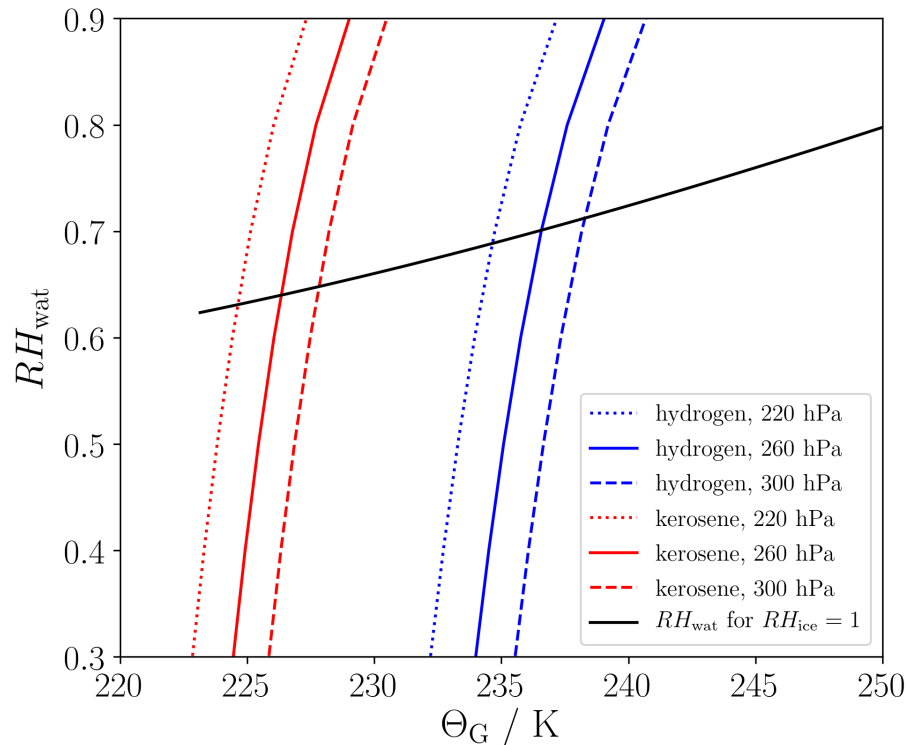


Figure 1. SA-threshold temperature (Θ_G) versus relative humidity over water (RH_{wat}) for three ambient pressures (differentiated by the line style) and for the kerosene (red) and hydrogen (blue) engine parameters as they are defined in [Tab. Table 1](#) each. The black solid line displays those RH_{wat} that would result in ice saturation assuming ambient temperature equal to Θ_G .

185 3 Methods

First, Sect. 3.1 gives an overview over the Lagrangian Cloud Module (LCM) and Sect. 3.2.1 describes the employed trajectory data and plume thermodynamics. Sect. 3.3 explains the basic contrail formation pathway on ambient particles and the associated extension of the LCM based box model. Finally, Sect. 3.4 gives an overview over the box model settings and the baseline conditions for the H_2 combustion scenario.

LCM is a particle-based microphysical model that includes aerosol, droplet and ice microphysics (Sölch and Kärcher, 2010). This particle-based approach has several numerical and physical advantages over common grid-based approaches, which are typically used in Computational Fluid Dynamics (CFD). It has been used for the simulation of natural cirrus clouds (e.g., Sölch and Kärcher, 2011), young contrails (e.g., Unterstrasser, 2014) and aged contrail-cirrus (e.g., Unterstrasser et al., 2017a). Recently, it has been extended by contrail formation microphysics on soot particles (Bier et al., 2022). Aerosol particles and hydrometeors are described by simulation particles (SIPs). Each SIP represents a certain number of aerosol particles/droplet-/ice crystals with the same properties and contains information about the liquid/ice water mass, radius, phase and particle type, among others. These properties may change due to microphysical processes like hygroscopic growth of aerosol particles and activation into water droplets, condensational droplet growth, homogeneous freezing of super-cooled droplets, depositional ice crystal growth, latent heat release, aggregation of ice crystals, sedimentation and radiative effects. In this study, we will consider only those processes that are relevant for contrail formation and exclude aggregation, sedimentation and radiative effects.

3.2 Plume evolution

3.2.1 Trajectory data

In a box model approach, fluid dynamics is not resolved and changes of thermodynamic properties inside the box are externally prescribed. We use the general plume dilution equations, which are described in Sect. 2.3.1 of Bier et al. (2022), to calculate the cooling and expansion of the plume as well as the evolution of the humidity. Based on 3D Large Eddy Simulations (LES) using the FLUDILES solver, Vancassel et al. (2014) sampled an aircraft plume with 25000 trajectories behind the engine of an A340-300 aircraft. Thereby, the temperature evolution $T_{3D,k}(t)$ has been tracked for each trajectory indexed by k . As in Bier et al. (2022), we use these data to infer the plume dilution factor by assuming that temperature is a passive tracer

$$\mathcal{D}_k(t) = \frac{T_{3D,k}(t) - T_{3D,a}}{T_{3D,E} - T_{3D,a}}, \quad (2)$$

where $T_{3D,E} = 580$ K and $T_{3D,a} = 220$ K are the plume exit and ambient temperature of the FLUDILES simulation. In the ~~remaineder~~ remainder of the text, the subscripts "E" and "a" denote conditions at the engine exit plane and in the atmospheric background, respectively.

In the following, we describe some modifications of the FLUDILES trajectory data set compared to the original one from Vancassel et al. (2014):

- As in Bier et al. (2022), we introduce a lower limit $T_{3D,k}(t) + \epsilon$ and all $T_{3D,k}(t)$ -values below this lower limit are set to this value. We choose $\epsilon = 0.2$ K, such that the implied dilution and plume area are consistent with the area enclosed by the trajectories.

- 220 – We have smoothed the time evolution of $T_{3D,k}$ for each trajectory such that $T_{3D,k}$ becomes a monotonically decreasing function with increasing plume age. This means we set $T_{3D,k}(t) := \text{MIN}(T_{3D,k}(t - \Delta t), T_{3D,k}(t))$.
- In our current model approach, the thermodynamic plume evolution and microphysics are calculated independently for each trajectory and Bier et al. (2022) find that such an ensemble approach without considering mixing effects among nearby trajectories is not perfect, in particular when the plume is sampled with many trajectories. Hence, we reduce the
- 225 number of trajectories to ntr_{sub} by merging $ntr_{\text{gr}} = 25000/ntr_{\text{sub}}$ trajectories that are initially close to each other into a single trajectory. Thereby, we apply a mass-conserving average (that is described more detailed in the supplement of Bier et al. (2022)) to obtain the temporal evolution of the new passive tracer temperature

$$\hat{T}_{3D}(t) = \sum_{k=1}^{ntr_{\text{gr}}} \frac{T_{3D,k}(t)}{T_{3D,k}(t) - T_{3D,a}}. \quad (3)$$

230 where $1/(T_{3D,k}(t) - T_{3D,a})$ is the weighting factor for a mass-conserving averaging and the equation is exemplarily written for one of the ntr_{sub} trajectories.

We have performed sensitivity studies for different ntr_{sub} -values. We find that $ntr_{\text{sub}} = 1000$ is a reasonable value which still sufficiently resolves the plume heterogeneity and will be used for the analysis in the present paper.

3.2.2 Plume cross-sectional area

The emitted air at the engine exit plane is a mixture of ambient air (going through the engine) and combustion products. The

235 initial plume dilution \mathcal{C}_E (i.e. the air-to-fuel ratio) is linked to the exit temperature T_E :

$$\mathcal{C}_E = \frac{Q(1 - \eta)}{\bar{c}_p(T_E - T_a)}, \quad (4)$$

This equation follows from the conservation of fuel mass and thermal energy flow. Moreover, $\bar{c}_p = 1020 \text{ J}(\text{kg K})^{-1}$ is assumed to be an average heat capacity for dry air over the temperature range of interest and the effects of the jet kinetic energy is neglected. The latter assumption is justified when (most) kinetic energy is converted into thermal energy well before the plume

240 has cooled down to temperatures where supersaturation is reached and formation of droplets commences. The overall dilution $\mathcal{C}(t)$ generally increases with plume age due to continuous entrainment of ambient air and is related to the dilution factor $\mathcal{D}(t)$ via

$$\mathcal{C}(t) = \mathcal{C}_E/\mathcal{D}(t), \quad (5)$$

implying $\mathcal{D}_E = 1$. Whereas \mathcal{C} denotes an air-to-fuel ratio, the dilution factor \mathcal{D} is defined as a fuel-to-air ratio (due to legacy

245 reasons) and decreases with plume age. The word "overall" in the name of $\mathcal{C}(t)$ refers to the fact that this quantity considers the combined effect of dilution occurring in the engine and the free atmosphere. Compared to Eq. (2) we drop the index k from

the dilution factor \mathcal{D}_k . With this we imply, that derivations in this subsection consider the total plume and not individual plume parts represented by a single trajectory.

The flow of the (total) plume mass is defined as

$$250 \quad \dot{m}_{\text{plume}}(t) := \rho(t)A(t)U_{\text{tot}}(t) = \dot{m}_{\text{F}}\mathcal{C}(t). \quad (6)$$

Here, $\rho(t)$ is the air density, $A(t)$ the plume cross-sectional area (perpendicular to the axial direction) and the total exhaust velocity $U_{\text{tot}}(t) = U_{\infty} + U_{\text{jet}}(t)$. The latter quantity $U_{\text{tot}}(t)$ is the sum of the absolute value of the aircraft speed, U_{∞} , and jet velocity $U_{\text{jet}}(t)$ relative to the ambient air. The fuel flow rate (units: kg/s) is denoted as \dot{m}_{F} .

Applying Eq. (6) at the engine exit plane and at any other time for fixed fuel flow rate, plugging in Eq. (5), using the ideal
255 gas law and assuming isobaricity, we obtain the following relation:

$$A(t) = A_{\text{E}} \frac{T(t)}{T_{\text{E}}\mathcal{D}(t)} \frac{U_{\text{tot,E}}}{U_{\text{tot}}(t)}. \quad (7)$$

Note that Eq. (16) in Bier et al. (2022) shows a similar relation, however misses the ratio of the velocities. The same holds for their Eq. (15). This term $U_{\text{tot,E}}/U_{\text{tot}}(t)$ accounts for the fact that the plume is compressed/stretched due to the axial divergence of the jet. This clearly affects how the cross-sectional plume area changes over time. However, this divergence effect does not
260 change the volume of the considered air parcel, it only distributes its mass over a segment with a changing extent along axial direction. The volume of the plume changes only due to the entrainment of ambient air (reflected by the dilution factor \mathcal{D}) and through associated density changes (reflected by the temperature ratio). Introducing an effective area

$$\hat{A}(t) = A(t) \frac{U_{\text{tot}}(t)}{U_{\infty}}, \quad (8)$$

Eq. (7) can be written as

$$265 \quad \hat{A}(t) = \hat{A}_{\text{E}} \frac{T(t)}{T_{\text{E}}\mathcal{D}(t)}. \quad (9)$$

The effective area $\hat{A}(t)$ is the cross-sectional area that is reached when the exhaust plume is (theoretically) decelerated from $U_{\text{tot}}(t)$ to U_{∞} in a volume-preserving manner. Since we provide all quantities in terms of 'per meter of flightpath', the effective area is the appropriate quantity that is representative for the volume of the plume at a certain age. Note that Eq. (9) has the same form as Eq. (16) in Bier et al. (2022). The only difference is that it uses the effective plume cross-sectional areas instead
270 of the real physical ones. In our case, $U_{\infty} = 250$ m/s and $U_{\text{jet,E}} = 230$ m/s. Hence, the ratio $\gamma_{\text{U}} = U_{\text{tot,E}} / U_{\infty}$ is 1.92.

The fuel consumption per meter of flight path m_{C} can be computed by

$$m_{\text{C}} = \dot{m}_{\text{F}}/U_{\infty} = \frac{p_{\text{a}}}{R_{\text{d}}T_{\text{E}}} \frac{A_{\text{E}}}{\mathcal{C}_{\text{E}}} \frac{U_{\text{tot,E}}}{U_{\infty}} \quad (10)$$

m_{C} is the amount of fuel that is burnt per unit length along flight direction. The amount of a corresponding combustion product per flight distance is then, e.g., $m_{\text{PT}} = EI_{\text{PT}} \times m_{\text{C}}$ (where EI_{PT} is the emission index of an assumed passive tracer).
275 However, we have to keep in mind that m_{PT} only specifies the amount of the tracer in a segment where the jet velocity approached 0. At the engine exit plane, the total tracer amount would be lower by a factor of γ_{U} .

In the preceding paper of Bier et al. (2022), formulas were expressed in terms of fuel consumption and never in terms of fuel flow rate. There, the symbol m_F was used for the fuel consumption. In the present paper, this is replaced by m_C in order to avoid confusion with the fuel flow rate \dot{m}_F .

280 Even though we made the comment that the derivations here consider the total plume, they are also valid for individual plume air parcels represented by a trajectory. The FLUDILES simulation data feature an initial plume temperature profile with a smooth radial transition between the plume and the environment instead of a step at the plume edge. Hence, trajectories in the outer plume regions start with $T_{E,k}(t_0) < T_E$, $U_{\text{tot},E,k} < U_{\text{tot},E}$ and $\mathcal{D}_k(t) < 1.0$. In principle, one could assign a trajectory-specific value of γ_U to each trajectory. The fraction of trajectories having a γ_U -value much smaller than the default is rather
285 small. Hence, we apply a bulk-correction factor $\gamma_U = 1.92$ to ensemble mean values (instead of applying individual correction factors to each trajectory and then perform a suitable mass-weighted averaging/summation to trajectory ensemble data). This simplification leads only to minor quantitative deviations.

As written above, Bier et al. (2022) missed to include the term γ_U in the computations of plume area and fuel consumption. However, the implications of this are not overly serious. The time evolution of intensive quantities in the box model are not
290 affected by this oversight. Only in the final step of translating ice crystal number concentrations into a total number of ice crystals N_{ice} per flight distance, one has to include the term γ_U . However, for quantities like the apparent ice emission index AEI or the activation fraction as shown in the figures of the previous publication the term γ_U cancels out and the displayed figures are all unaffected by our oversight. For the sake of clarity, we want to note that the simulations of kerosene contrails discussed in [See Sect. 4.4](#) of the present study are computed with the correct formulas.

295 3.3 Contrail formation pathway on entrained ambient particles

In our study, we assume H_2 combustion with soot-free emissions. Moreover, we exclude the potential formation of ultrafine particles due to lubricant oil vapor, which will be separately discussed in Sect. 5. Hence, contrails will solely form on ambient background particles at suitable atmospheric conditions (e.g., Kärcher et al., 1996, 2015). The microphysics of contrail formation on soot particles, as implemented in the LCM, has been described in detail by Bier et al. (2022). These microphysical
300 processes are mostly also relevant for contrail formation on ambient particles and will be summarized briefly in this section. We focus our description on additional aspects for H_2 contrails and the associated extensions in the LCM based box model. The major difference besides the chemical composition is the fact that ambient particles are continuously entrained instead of releasing a fixed number of emitted soot particles. Finally, we introduce an alternative activation criterion and an extended homogeneous freezing parameterization. This accounts for the better solubility of the majority of UT particles compared to
305 engine soot.

3.3.1 Ambient aerosol particle properties

We prescribe background particles as an ensemble that is characterized by a log-normal size distribution with a geometric mean dry radius (\bar{r}_d), geometric width (σ_{aer}) and a number concentration (n_{aer}). Moreover, we specify the hygroscopicity parameter (Petters and Kreidenweis, 2007). The parameters are varied mostly independently of each other representing different types of

310 ambient aerosol particles and accounting for their natural variability.

According to the classical microphysical pathway of contrail formation with the liquid transition phase (e.g., Kärcher et al., 2015; Bier et al., 2022), we consider only CCN or partially soluble mixed particles (i. e., an insoluble core and a hydrophylic coating), where we will assign the latter simply to "weakly soluble particles". We exclude heterogeneous ice nucleation on insoluble particles since measured and modeled number concentrations of ice nuclei (IN) in the UT are typically several orders
 315 of magnitudes lower than those of CCN (e.g., Rogers et al., 1998; Beer et al., 2022). Even though IN may have important effects on natural cirrus cloud properties (e.g., Hendricks et al., 2011), we expect a negligible contribution to the overall contrail ice crystal formation. Insoluble but still wettable particles like uncoated soot and oil droplets could be also treated by adsorption activation theory. It is based on standard Köhler theory but uses a specific description of the water activity that accounts for adsorption processes. In particular, the Frankel, Halsey and Hill (FHH) adsorption approach (Sorjamaa and Laaksonen, 2007;
 320 Kumar et al., 2009) is able to treat multilayer adsorption of water vapor onto insoluble particles and should be considered in future studies.

For simplicity, we assume that the entrained ambient particles are initially completely dry, i. e., without any environmental hygroscopic water uptake before. We allow for condensational growth of the entrained aerosol particles not only at water-supersaturated conditions but also at plume relative humidities lying between the deliquescence relative humidity (DRH) and
 325 water saturation. Thereby, DRH is the minimum threshold relative humidity that allows for hygroscopic water uptake by a given substance.

3.3.2 Entrainment of ambient particles

Aerosol particles from the environment are continuously entrained into the plume. The background aerosol abundance is specified in terms of an aerosol background number concentration n_{aer} .

330 As a next step, we quantify the number of aerosol particles $N_{\text{aer}}(t)$ (here per flight distance) being present in the expanding plume.

$$N_{\text{aer}}(t) = n_{\text{aer}} \left(\frac{\rho(t)}{\rho_a} \hat{A}(t) - \frac{\rho_E}{\rho_a} \hat{A}_E \right) \quad (11)$$

$$= n_{\text{aer}} T_a \left(\frac{\hat{A}(t)}{T(t)} - \frac{\hat{A}_E}{T_E} \right) \quad (12)$$

$$= n_{\text{aer}} \frac{T_a \hat{A}_E}{T_E} (\mathcal{D}(t)^{-1} - 1) := \alpha(t) n_{\text{aer}} \quad (13)$$

335 The ratio of densities in the first line accounts for the fact that during an adiabatic entrainment process the mixing ratio is conserved and not the concentration. Moreover, we assume that the initial plume is void of any aerosol particles, which implies that any aerosol particle sucked into the aircraft engine is destroyed. Hence, the plume area at engine exit is subtracted (second term in first line). The further re-formulations assume isobaric conditions and use the ideal gas law and [the definition of the effective area](#) (Eq. (9)).

340 If we prescribed aerosol particles also in the initial plume (with identical properties as in the environment), results would not change too much as their contribution to the total aerosol particle number gets smaller and smaller while the plume expands (this case is easily treated by dropping "-1" in the term " $\mathcal{D}(t)^{-1} - 1$ ").

The number of particles being entrained into the plume during one time step Δt is then given by

$$\Delta N_{\text{aer}}(t) = (\alpha(t) - \alpha(t - \Delta t)) n_{\text{aer}}. \quad (14)$$

345 In every time step, a new SIP ensemble representing these newly entrained particles is created. Only in the initial stages of the simulation when RH_{wat} is still below DRH no new SIPs have to be created as aerosol particles inside the plume are still dry. In this case, it is sufficient to only increase the SIP weight (i. e., the number denoting how many real particles are represented by a SIP).

Clearly, the continuous creation of new SIPs would cause huge values of N_{SIP} and lead to computationally expensive or even
350 unfeasible simulations. Hence, we apply a SIP merging algorithm if the overall SIP number N_{SIP} gets too large (see Appendix A).

Moreover, we found that T_{3D} happens to increase in certain (short) segments along several plume trajectories. This implies that the cross-sectional area represented by the trajectory shrinks (i.e. $\mathcal{C}(t) \leq \mathcal{C}(t - \Delta t)$) and a negative value of $\Delta N_{\text{aer}}(t)$ follows. To inhibit such an unwanted detrainment of particles and hydrometeors, we have smoothed our trajectory data such that the
355 dilution $C(t)$ is a monotonically increasing function with time (see Sect. 3.2.1).

3.3.3 Diffusional growth and freezing

For spherical droplets, the single droplet mass growth equation is given by [\(Kulmala, 1993\)](#) [Kulmala \(1993\)](#) as

$$\frac{dm_w}{dt} = \frac{4\pi r(e_v - e_{K,\text{wat}})}{\frac{R_v T}{D_v} \beta_m^{-1} + \frac{e_{K,\text{wat}} L_c^2}{R_v K T^2} \beta_t^{-1}}, \quad (15)$$

360 where r is the wet aerosol or droplet radius, e_v is the partial vapor pressure. L_c denotes the specific latent heat for condensation/evaporation, D_v the binary diffusion coefficient of air and water vapor, K the conductivity of air, R_v the specific gas constant of vapor and T the temperature. The transitional correction factors β_m and β_t are calculated according to Eqs. (A4) and (A5) of Bier et al. (2022) based on Fuchs and Sutugin (1971). Note that there is a transcription error in the previous study and the denominators of Eqs. (A4) and (A5) miss both the term "+ 1". With this correction, β_m and β_t tend to one for small
365 Knudsen numbers, as intended. The quantity $e_{K,\text{wat}}$ is the product of the saturation vapor pressure over a flat water surface $e_{\text{sat},\text{wat}}$ and the equilibrium saturation ratio over a solution droplet surface S_K . As in Bier et al. (2022), we calculate S_K using the κ -Köhler [Eq. equation](#) (Petters and Kreidenweis, 2007)

$$S_K = \frac{r^3 - r_d^3}{r^3 - r_d^3(1 - \kappa)} \cdot \exp\left(\frac{2\hat{\sigma}M_{\text{wat}}}{RT\rho_{\text{wat}}r}\right), \quad (16)$$

where the first term is the activity of water (a_{wat}) and the exponential expression is the Kelvin term. r_d is the particle dry
 370 radius, κ the hygroscopicity parameter, $\hat{\sigma}$ the surface tension of the solution droplet, ρ_{wat} the mass density of water, M_{wat} the
 molar mass of water and R the universal gas constant. The surface tension typically increases with decreasing a_{wat} for salt
 solutions due to negative adsorption and increases with decreasing a_{wat} for acidic solutions due to positive adsorption. Since
 we do not prescribe specific aerosol particle species but only the hygroscopicity parameter in the present study, we approximate
 $\hat{\sigma}$ with the surface tension of pure water droplets and use the polynomial expression by Hacker (1951) as in Bier et al. (2022).
 375 Hence, a slight error is caused in the Kelvin term for more concentrated solution droplets.

The entrained ambient aerosol grow by condensation if RH_{wat} is larger than the deliquescence relative humidity (DRH). As
 long as the entrained particle is completely dry, S_K is not applicable since $r = r_d$. Therefore, we set S_K to a value that is
 slightly lower than RH_{wat} for the first time step with condensation. This causes the particle to grow hygroscopically so that
 $r > r_d$ in the next time step and then S_K is calculated according to Eq. (16). During the subsequent plume evolution, the wetted
 380 particle/droplet grows further due to condensation if $RH_{\text{wat}} > S_K$ or shrink due to evaporation if $RH_{\text{wat}} < S_K$ according to
 Eq. (15).

In Bier et al. (2022), the soot particles have been considered to be activated into water droplets if the wet radius has exceeded
 the critical radius, which is the radius at the maximum of S_K . In the present study, we consider aerosol particles to be activated
 into water droplets if the activity of water has exceeded a critical value $a_{\text{wat},c} = 0.90$ to ensure a sufficient water uptake for
 385 freezing. Once an aerosol particle has been activated into a droplet, it can freeze to an ice crystal if the plume temperature
 drops below the homogeneous freezing temperature of that solution droplet (T_{frz}). For the calculation of T_{frz} , Bier et al. (2022)
 follow the approach of Kärcher et al. (2015) and Riechers et al. (2013) assuming pure water droplets. We extend this approach
 by including a simple correction term, based on the parameterization of O and Wood (2016), to account for the decrease in T_{frz}
 due to the solution effect (e.g., Koop et al., 2000). Further details are described in Appendix B.

390 The depositional growth of the formed ice crystals is calculated according to Eq. (7) of Bier et al. (2022) which is based on
 Mason (1971). Note that there is a transcription error in that equation, where the correction term β_v^{-1} occurs twice in the mass
 diffusion term and should be removed in the denominator.

3.4 Model settings and baseline parameters

[Tab:Table 2](#) summarizes our baseline initial and background conditions for H_2 combustion as well as the ambient particle
 395 properties and model set-up parameters. We prescribe an ambient temperature T_a of 225 K, ambient pressure p_a of 260 hPa and
 relative humidity over ice $RH_{\text{ice},a}$ of 120%. We define a water vapor mass emission index EI_v and specific combustion heat
 Q that are typical of hydrogen propulsion ([Tab:Table 1](#)). We set the propulsion efficiency, engine exit temperature and initial
 plume area to the same values as in Bier et al. (2022). The fuel and engine parameters EI_v , Q , η and T_E are kept constant
 for all sensitivity studies even though they can slightly change with ambient conditions, see Sec 5.2. We determine the initial
 400 plume dilution C_E and the fuel consumption m_C according to Eqs. (4) and (10). Since A_E is the area of one engine nozzle
 exit plane (based on the FLUDILES data for the 4-engine A340-300 aircraft and kept constant in this study), our m_C -value is
 representative of a single engine and would be 4 times larger for the whole aircraft. Thus, we simulate contrail formation behind

a single aircraft engine. The C_E and m_C -values listed in [Tab:Table 1](#) are given for the atmospheric baseline T_a and p_a -values. Note that C_E is larger and m_C lower by a factor of 2.8 compared to kerosene combustion because Q is accordingly higher (with the relations $C_E \sim Q$ and $m_C \sim Q^{-1}$). An analogous kerosene setup would have baseline values $C_E \approx 76$ and $m_C \approx 3.1 \text{ g m}^{-1}$. We prescribe ambient particles with a mono-modal log-normal size-distribution. We set the geometric-mean dry radius (\bar{r}_d) to 15 nm and geometric width of 1.6 representing a typical Aitken aerosol mode in the UT (e.g., Brock et al., 2021). We define an aerosol number concentration (n_{aer}) of 600 cm^{-3} , which is lying well in between the observed values for Aitken and accumulation mode particles (e.g., Minikin et al., 2003; Borrmann et al., 2010). We set the hygroscopicity parameter κ to 0.5, which is e. g. a typical value of ammonium sulfate particles (Liu et al., 2014). Peng et al. (2022) provide a large data set for DRH of different atmospheric compounds (e.g. see their Table 1). Even though many inorganic compounds have a DRH significantly below one, we set our baseline DRH very close to water saturation. The reason for that will be discussed in Sect. 5.2. For a comparison with conventional contrails, we also provide associated soot particle properties in [Tab:Table 2](#). For any ambient particle ensemble, we use around 110 SIPs to represent its log-normal size distribution. For this, we use the algorithm described in Unterstrasser and Sölch (2014), which has favorable numerical convergence properties (Unterstrasser et al., 2017b). Each simulation contains an ensemble of box model runs for 1000 different trajectory data as described in the previous section. The runs are performed independently of each other for each trajectory. The standard simulation time is 3 s; the numerical time step is 0.001 s. For some simulations with $T_a < 220 \text{ K}$, we extend the simulation time to 5 s since the time period where droplet and, hence, ice crystal formation occurs is longer than 3 s. Typically, droplet and ice formation comes to a halt well before a plume age of 5 s and hence, the ice crystal number does not increase anymore.

ambient conditions	H ₂ fuel/engine prop.	H ₂ engine exit conditions	ambient particle prop.	soot part. prop	set-up parameters
$T_a = 225 \text{ K}$	$EI_v = 8.94 \text{ kg kg}^{-1}$	$T_E = 580 \text{ K}$	$n_{\text{aer}} = 600 \text{ cm}^{-3}$	$N_s = 3.07 \cdot 10^{12} \text{ m}^{-1}$	$t_{\text{sim}} = 3\text{--}5 \text{ s}$
$p_a = 260 \text{ hPa}$	$Q = 1.2 \cdot 10^8 \text{ J kg}^{-1}$	$A_E = 0.25\pi \text{ m}^2$	$\bar{r}_d = 15 \text{ nm}$	$\bar{r}_d = 15 \text{ nm}$	$dt = 0.001 \text{ s}$
$RH_{\text{ice,a}} = 120\%$	$\eta = 0.36$	$C_E \approx 210$	$\sigma_s = 1.6$	$\sigma_{\text{aer}} = 1.6$	$dt_{\text{out}} = 0.01 \text{ s}$
—	—	$m_C \approx 1.1 \text{ g m}^{-1}$	$\kappa = 0.5$	$\kappa = 0.005$	$N_{\text{SIP},0} \approx 110$
—	—	—	$DRH = 0.99$	$DRH = 0.99$	$N_{\text{SIP},m} = 1600$

Table 2. Baseline parameters for our LCM box model studies. The fuel, engine properties and exit conditions refer to H₂ combustion. In addition to the entrained ambient particle properties, we provide the baseline soot particle properties for a comparison of H₂ with conventional contrails.

4 Results

In this section, we first analyze the temporal evolution of thermodynamic and microphysical H₂ contrail properties for our baseline case (Sect. 4.1). We then investigate the impact of atmospheric conditions on those properties in Sect. 4.2. In Sect. 4.3,

we analyze the influence of ambient aerosol particle properties on contrail ice crystal formation either prescribing one or two
425 co-existing aerosol particle ensembles. Finally, we compare our results with conventional kerosene contrails in terms of ice
crystal number and optical thickness in Sect. 4.4. The thermodynamic and microphysical properties are either averaged or
summed up over all box model trajectories obeying a mass-conserving weighting as each trajectory may represent a different
share of the plume at later times. In this study, we always display our microphysical properties in units (number or mass) per
flight distance.

430 4.1 Temporal evolution of contrail properties for the baseline case

Fig. 2 shows the temporal evolution of thermodynamic and microphysical properties for our baseline case defined in [Tab. Table 2](#).
The mean plume temperature ([panel \(a\)](#)) [Fig. 2a](#)) decreases with increasing plume age due to continuous mixing of the exhaust
with ambient air approaching the ambient temperature. Accordingly, the plume dilution and, therefore, the effective plume
area increase, the latter from around 1.5 to 540 m² after 3 s. The mean relative humidity over water RH_{wat} ([panel \(b\)](#)) [Fig. 2b](#))
435 surpasses the deliquescence relative humidity after around 0.2 s and reaches its maximum of around 220% after 0.4 s. Com-
pared to a plume behind a conventional aircraft (e.g., see Fig. 2 in Bier et al. (2022)), our maximum RH_{wat} and accordingly
 RH_{ice} values are substantially higher since the difference between the ambient and the SA-threshold temperature is larger (the
SA-threshold temperature is by around 10 K higher for H₂ than for kerosene combustion, see Sect. 2.2).

[Panel \(c\)](#) [Fig. 2c](#) shows the accumulated number of aerosol particles entrained into the plume (N_{aer}), the number of formed
440 droplets (N_{drp}) and the number of ice crystals (N_{ice}). N_{aer} increases nearly linearly with time reaching values of around
1.15·10¹¹ and 3.07·10¹¹ m⁻¹ after 1.5 and 3 s, respectively. This is different to exhaust species like soot particles since they
typically form right behind the engine exit (when plume relative humidity is still low and no activation occurs) and their emitted
number (per flight distance) is then assumed to be constant over time.

The N_{aer} -values are mainly controlled by the ambient aerosol number concentration (n_{aer}) and the plume area expansion. The
445 first aerosol particles activate into water droplets (dashed line) after RH_{wat} surpasses the DRH in the corresponding trajecto-
ries. A few tenth of seconds later, they freeze to ice crystals (solid line) once plume temperature falls below the homogeneous
freezing temperature of those droplets. Later on (at plume ages between around 0.6–1.1 s), N_{ice} is very close to N_{aer} . This
means that basically all entrained aerosol particles nearly instantaneously form droplets and freeze. After 1.5 s when the mean
 RH_{wat} falls below approximately 95%, no further droplets and ice crystals form. Therefore, N_{ice} stays constant afterwards
450 at 1.04·10¹¹ m⁻¹. Some of the droplets (small peak in N_{drp} at around 1.4 s) cannot freeze and evaporate afterwards. This is
because the homogeneous freezing temperature of those droplets, which are either too small or have a too low water activity,
is not reached. The ice crystals grow by deposition so that the ice water mass (m_{ice}) continuously increases ([panel \(d\)](#)) [Fig. 2d](#)).
The mean ice crystal radius \bar{r}_{ice} (red line) tends to increase over time and finally reaches a size of around 2.2 μm. The decrease
of \bar{r}_{ice} between 0.6 and 1.3 s is because more and more of the smaller droplets manage to freeze to ice crystals and, hence, the
455 mean ice crystal size drops in that time period. Since the plume is still ice-supersaturated (solid line in [panel \(b\)](#)) [Fig. 2b](#)) after
3 s, m_{ice} and \bar{r}_{ice} would increase even further.

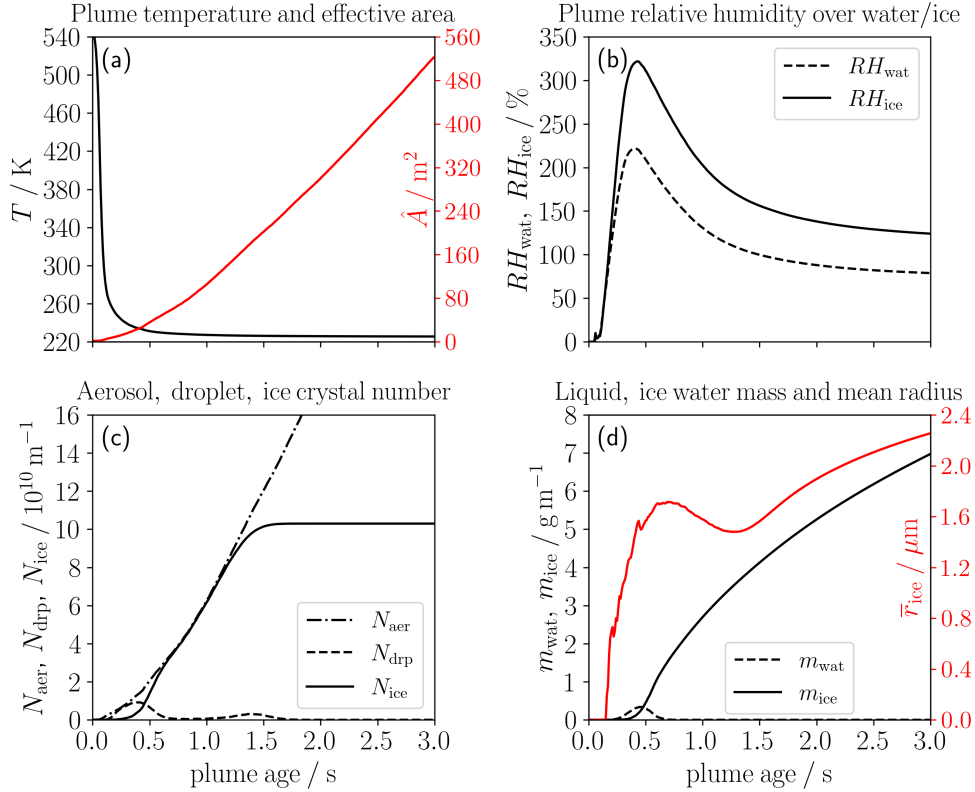


Figure 2. Temporal evolution of thermodynamic and microphysical properties in a single- engine plume for the baseline case: The panels show (a) temperature (black) and effective cross-sectional area (red and labels on the right axis), (b) relative humidity over water (dashed) and ice (solid), (c) number of aerosol particle entrained into the plume (dash-dotted), number of droplets (dashed) and ice crystals (solid) per flight distance and (d) liquid (black dashed) and ice water mass per flight distance (black solid) and mean radius of the ice crystals (red and labels on the right axis).

4.2 Impact of atmospheric properties

Here, we investigate in detail the impact of ambient temperature on the plume thermodynamical and microphysical contrail properties. Moreover, we analyze the influence of ambient pressure and relative humidity over ice. We prescribe our baseline ambient aerosol particle properties and keep them constant in this section. Note that a fixed aerosol number concentration for varying atmospheric parameters (in particular pressure) is an idealized assumption in the following sensitivity studies.

4.2.1 Influence of ambient temperature on temporal evolution of thermodynamic and contrail properties

Fig. 3 highlights the strong impact of ambient temperature T_a on contrail ice crystal formation. As shown in [panel \(a\) Fig. 3a](#), the plume temperature at a given plume age is certainly lower in a colder environment and finally approaches the corresponding

465 T_a value. The peak mean relative humidity over water (displayed in [panel \(b\)](#) [Fig. 3b](#)) increases with decreasing T_a (reaching values of around 350% and 550% for T_a of 220 K and 215 K, respectively). The large increase for low T_a -values is due to the non-linearity between saturation vapor pressure and temperature. The relative humidity over ice behaves accordingly (not shown).

The slight change in the evolution of N_{aer} is a consequence of our model set-up with a fixed aerosol number concentration and
470 the varying plume air density with temperature (at fixed ambient pressure). In general, the droplet formation is basically controlled by the time period where RH_{wat} is above DRH such that water can condense on the entrained aerosol particles. (Note that the evolution in RH_{wat} and, therefore, this time period varies with each trajectory and we here only display the ensemble mean quantity.) This mean time period for possible droplet and ice crystal formation substantially increases with decreasing ambient temperature (e.g. from 0.15 s up to 2.8 s for $T_a = 215$ K). This means that ice crystal formation is initiated earlier and
475 comes to a halt later (see red and blue solid lines in [panel \(e\)](#) [Fig. 3c](#)) compared to the baseline case. Moreover, nearly all formed droplets freeze very quickly to ice crystals so that N_{drp} approaches zero. For these reasons, the final ice crystal number strongly increases with decreasing T_a over the whole temperature range (see also Fig. 4). This is different to what we find for conventional soot contrails as all soot particles turn into ice crystals if T_a is several K below the SA-threshold temperature (e.g., Kärcher et al., 2015; Bier and Burkhardt, 2019; Bier et al., 2022). Yet, any further reduction in T_a does not lead to more
480 ice crystals in the conventional case. Moreover, the peak plume RH_{wat} -values are substantially higher for H_2 than for kerosene combustion at same ambient conditions. Therefore, droplet and ice crystal formation on ambient particles is controlled more strongly by the time period in which the plume is water-supersaturated than by the maximum water-supersaturation.

For higher ambient temperatures ($T_a \geq 230$ K), many droplets cannot freeze to ice crystals and evaporate thereafter (which is indicated by declining N_{drp} at nearly constant N_{ice}). This is because the homogeneous freezing temperature of the smaller
485 and/or more concentrated solution droplets is below the plume/ambient temperature. Hence, the final ice crystal numbers are decreased further in addition to the fact that the time period for possible droplet formation is lower. The decrease of N_{ice} with increasing T_a becomes stronger for $T_a \geq 232$ K (see also Fig. 4) and for $T_a = 233$ K, only a few large droplets can form ice crystals. For higher ambient temperatures no ice crystal formation occurs anymore. This means that for H_2 combustion, the freezing temperature is typically smaller than the SA-threshold temperature and becomes a more limiting criterion for contrail
490 formation. Yet, the SA-threshold temperature is still relevant, as its difference to the ambient temperature determines the peak and time period of water supersaturation in the plume. The ice water mass (shown in [panel \(Fig. 3d\)](#) in general increases with decreasing ambient temperature. The strong increase between T_a of 230 K and 233 K is mainly due to the increase in N_{ice} .

In the following sections, we will focus our analysis on the final number of formed ice crystals ($N_{\text{ice,f}}$) since the young
495 contrail ice number mostly impacts the further contrail (cirrus) properties and radiative forcing (e.g., Bier et al., 2017; Burkhardt et al., 2018; Bier and Burkhardt, 2022). In contrast, the initial ice water mass/mean ice crystal radius was shown to have a low impact on the contrail life cycle in the dispersion phase (e.g., Unterstrasser and Gierens, 2010), but the size distribution of the formed contrail ice crystals can strongly impact the sublimation loss of ice crystals during the vortex phase (e.g., Unterstrasser, 2014). The latter will be investigated in future studies.

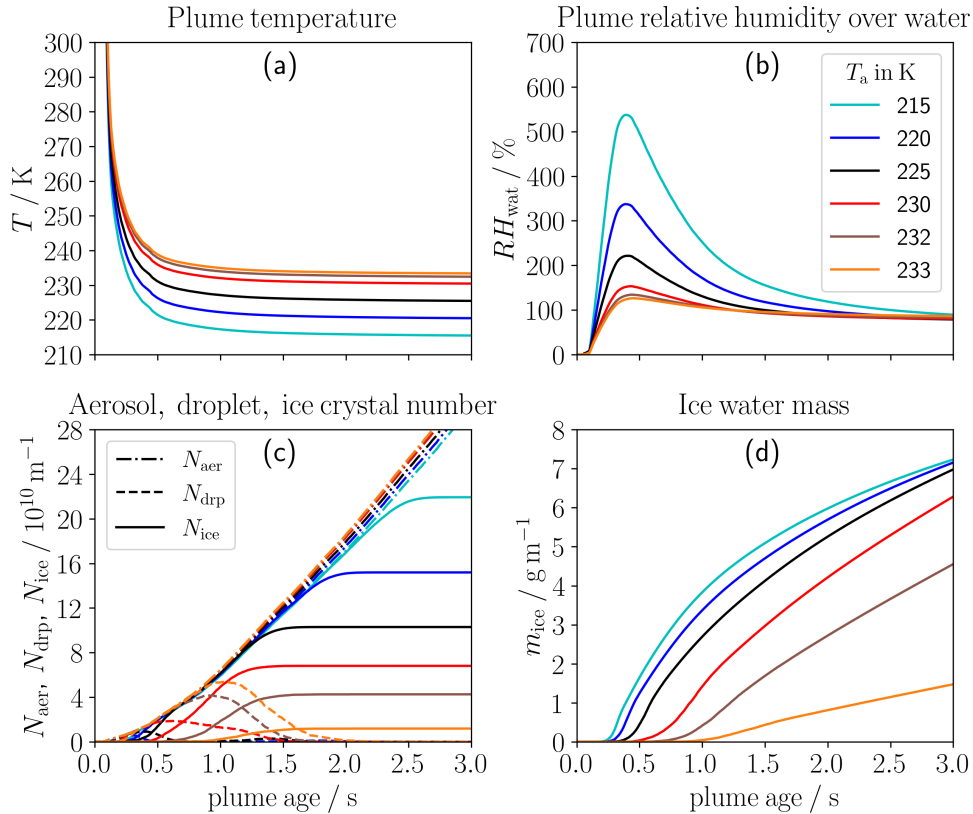


Figure 3. Impact of ambient temperature (T_a) on temporal evolution of thermodynamic and microphysical properties in a single-engine plume: The panels show (a) temperature, (b) relative humidity over ice, (c) number of aerosol particle entrained into the plume (dash-dotted), number of droplets (dashed) and ice crystals (solid) per flight distance and (d) ice water mass per flight distance. The colors represent the different T_a -values as defined in the legend.

500 4.2.2 Final ice crystal number

Fig. 4 displays $N_{ice,f}$ versus ambient temperature T_a for (a) 3 different pressure p_a and (b) 3 different ambient relative humidity over ice $RH_{ice,a}$ -values. Note that our parameter settings are simplified in the sense that some combinations of parameter values are not realistic for the atmosphere (e.g., lowest T_a value at the highest p_a value). Compared to the previous subsection, we now include a further case with $T_a = 210$ K (that requires a longer simulation time since the period where the plume is water-supersaturated is higher than 3 s). This case emphasizes the enhanced increase of $N_{ice,f}$ with decreasing T_a for very cold conditions and is consistent with the findings by Ström and Gierens (2002). $N_{ice,f}$ is increased for a higher ambient pressure because the slope of the mixing line G , defined by Eq. (1), is larger for a higher pressure. Moreover, [panel \(b\) Fig. 4b](#) shows an incline of $N_{ice,f}$ with increasing $RH_{ice,a}$. Both the increased G and the higher $RH_{ice,a}$ lead to higher peak plume relative humidities and enlarge the time period for possible droplet and subsequent ice crystal formation. Interestingly, this increase

510 is quite strongly pronounced for the highly ice-supersaturated case (red line) at ambient temperatures between 230 and 234 K
 (and for 234 K ice crystals can only form for this case at all). This is because the droplet freezing is mainly limited by the
 droplet size in that T_a -range and for the high $RH_{ice,a}$ -case, more larger droplets can form that turn into ice crystals.
 Finally, ambient temperature is the parameter that influences the number of formed contrail ice crystals most while the impact
 of ambient pressure and relative humidity is clearly smaller. This behavior is similar to the conventional case with kerosene
 515 combustion despite the very different temporal evolution in the exhaust particle number concentration.

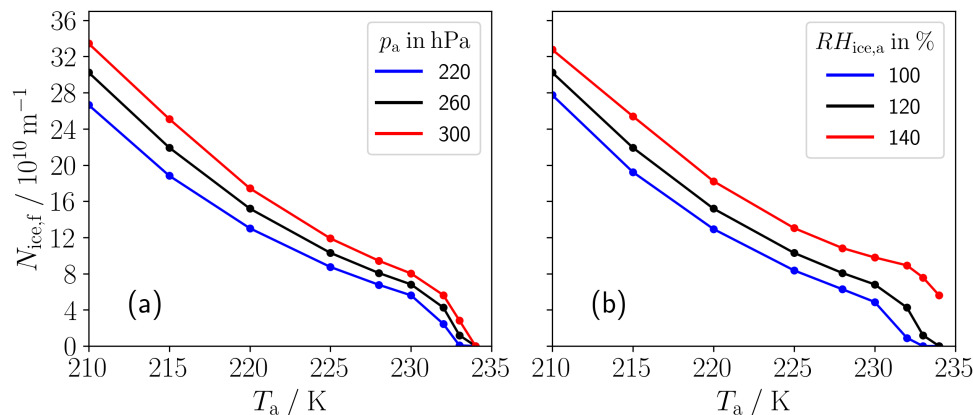


Figure 4. Final ice crystal number per flight distance ($N_{ice,f}$) in a single-engine plume versus ambient temperature for (a) three different ambient pressures (p_a) and (b) for three different ambient relative humidities over ice ($RH_{ice,a}$). The black line in both panels always refers to the baseline p_a and $RH_{ice,a}$ -values. $N_{ice,f}$ is given at a plume age of 3 s for $T_a \geq 215$ K, and at a plume age of 5 s for $T_a = 210$ K.

4.3 Sensitivity of ice crystal number to ambient aerosol particle properties

In this section, we investigate the impact of ambient aerosol particle properties on the (final) number of formed ice crystals in H_2 contrails. We prescribe one aerosol particle ensemble (single mode) in Sect. 4.3.1 (as in the previous analysis) and two co-existing aerosol particle ensembles in Sect. 4.3.2.

520 4.3.1 Studies with a single aerosol particle ensemble

Fig. 5 shows the variation of $N_{ice,f}$ with different aerosol particle properties. The sensitivities are always shown for three ambient temperatures T_a (differentiated by the color). In general, we see an increase in $N_{ice,f}$ with decreasing T_a for any particle property combination, being consistent with the findings in the previous section. **Panel (a) Fig. 5a** shows that $N_{ice,f}$ inclines with increasing aerosol number concentration n_{aer} . This increase becomes weaker for higher n_{aer} ($\gtrsim 200 cm^{-3}$) values.
 525 This is due to the enhanced competition for plume water vapor between the growing droplets/ice crystals for increased aerosol number concentrations.

Next, we analyse the importance of the mean dry radius of the aerosol size distribution \bar{r}_d and the hygroscopicity parameter κ .

Typically, the ice crystal number strongly increases with increasing mean dry size for $\bar{r}_d \lesssim 10$ nm and then stays nearly constant (panel (b)) see Fig. 5b). The increase is mainly due to the Kelvin effect, i. e., larger aerosol particles are easier to activate into water droplets since they require lower plume water-supersaturations and grow more quickly to water droplets (e.g., Bier et al., 2022).

Panel (c) Fig. 5c shows the dependence of $N_{ice,f}$ on κ . For $T_a = 230$ K, $N_{ice,f}$ slightly increases for a larger κ for all three \bar{r}_d as indicated in the legend. For the lower T_a cases, the variation of $N_{ice,f}$ with \bar{r}_d and κ is more complex. For the small sized particles (dotted-dashed lines), there are two counteracting effects: $N_{ice,f}$ increases with rising hygroscopicity parameter for $\kappa < 0.1$ since better soluble particles can easier form water droplets. On the other hand, $N_{ice,f}$ subsequently decreases. This is because some of the droplets cannot freeze to ice crystals since their water activity is lower due to the enhanced solution effect for higher κ and, therefore, the homogeneous freezing temperature is significantly decreased (see Fig. B1 in the appendix Appendix). For the other \bar{r}_d cases, $N_{ice,f}$ hardly changes with the solubility and mean aerosol particle size.

We also analyze in panel (d) Fig. 5d the impact of the geometric width of the aerosol size distribution for four different \bar{r}_d , κ -combinations as displayed in the legend. Our results imply a very low sensitivity of $N_{ice,f}$ to the geometric width.

In conclusions, the sensitivity of the ice crystal number to \bar{r}_d and κ is low for aerosol particles with a large mean dry size. For the small-sized particles, we find a quite complex variation of $N_{ice,f}$ with κ , mainly for low ambient temperatures, due to various counteracting effects.

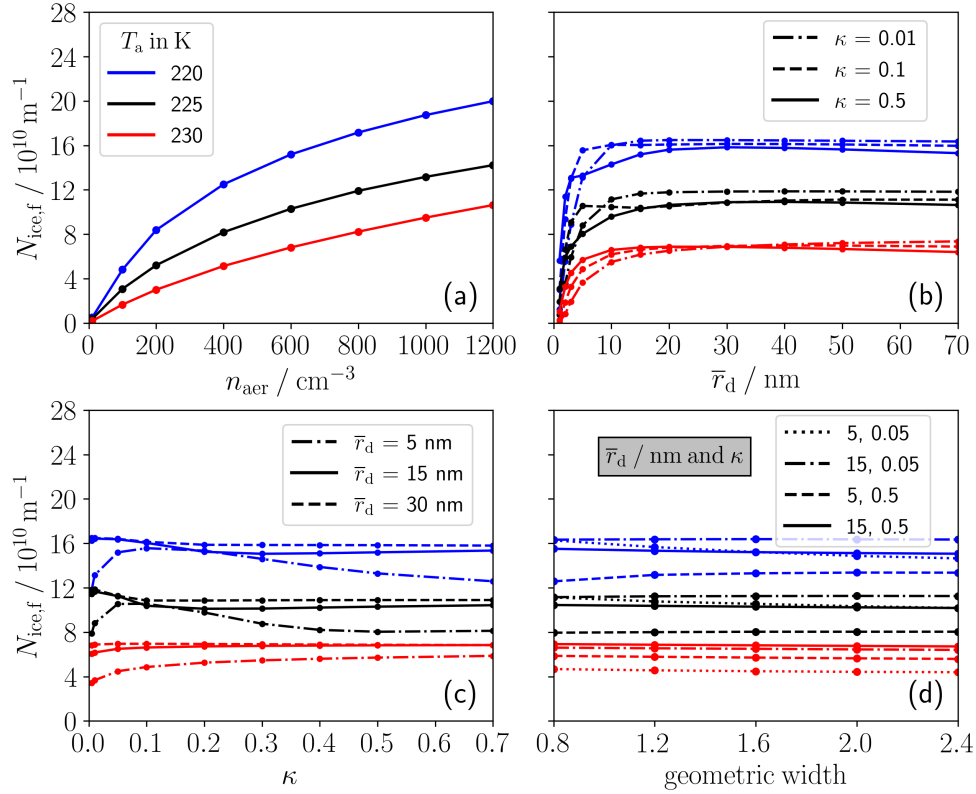


Figure 5. Final ice crystal number per flight distance ($N_{ice,f}$) in a single-engine plume depending on different aerosol particle properties assuming uni-modal size distributions for three ambient temperatures (T_a) (different colors defined in legend (a)): $N_{ice,f}$ is shown versus (a) ambient aerosol number concentration, (b) geometric mean dry radius (\bar{r}_d) for three solubility-values (see line style in legend (b)), (c) hygroscopicity parameter (κ) for three \bar{r}_d -values (see line style in legend (c)) and (d) the geometric width of the size distribution for 4 different \bar{r}_d, κ combinations as displayed in the legend.

4.3.2 Studies with two co-existing aerosol particle ensembles

545 So far, the aerosol particles were prescribed with a single log-normal size distribution and a fixed hygroscopicity value. In the present section, we prescribe two co-existing ambient aerosol particle ensembles and analyze contrail ice crystal formation for two ambient temperatures. The given n_{aer} value is the total number concentration of both aerosol ensembles. We restrict our analysis to cases where each ensemble has a number concentration of $0.5 \cdot n_{\text{aer}}$. We consider nucleation mode ($\bar{r}_d = 3$ nm), Aitken mode ($\bar{r}_d = 15$ nm as in our baseline case) and accumulation mode ($\bar{r}_d = 50$ nm) particles. The given mean dry sizes of the single modes are prescribed consistently with typical observed UT geometric mean diameters over the Atlantic and Pacific Ocean within the ATom campaign (see Fig. 12 of Brock et al. (2021)). Moreover, we consider well soluble ($\kappa = 0.5$) particles like inorganic salts and weakly soluble ambient particle ($\kappa = 0.05$) like organic species or aviation soot. We restrict our analysis to a scenario, where the two co-existing particle ensembles always differ either in their mode (in our case \bar{r}_d) or the solubility κ . Moreover, we compare the total N_{ice} of the two aerosol particle ensembles with a reference case, which uses a single aerosol population with the same value of n_{aer} and the average \bar{r}_d and κ -values of the two particle ensembles.

The first two rows of Fig. 6 show the temporal evolution of N_{ice} with a fixed $n_{\text{aer}} = 600 \text{ cm}^{-3}$. First, we analyze the contrail ice number evolution for bi-modal aerosol size distributions with same κ (first row). Panel (a) shows that N_{ice} for the Aitken mode particle is at the end by around 50% larger than for the nucleation mode particle for both temperatures. This is because, for a given plume relative humidity, the larger particles can better activate into water droplets (and freeze thereafter) than the very small nucleation mode particles due to the Kelvin effect. Moreover, the total ice crystal number is at the end by around 10% lower than the reference N_{ice} . Considering the Aitken and accumulation mode in panel (b), N_{ice} between the two particle ensembles is quite similar and for $T_a = 230$ K nearly identical. This is consistent with the findings for a single particle ensemble where the variation of the ice crystal number for $\bar{r}_d \gtrsim 10$ nm is low (see Fig. 5 (b)). For the same reason, the total and reference N_{ice} are close to each other.

565 Now, we investigate the ice crystal formation for particle ensembles with two different solubility characteristics but same \bar{r}_d (second row). For $T_a = 230$ K, N_{ice} for the weakly soluble particles is lower than for the well soluble particles. This is because the less hygroscopic particles are harder to activate and/or cannot grow to sufficiently large droplet sizes in order to freeze to ice crystals. Hence, the total ice crystal numbers are slightly reduced (by around 5–10%) relative to the reference ice numbers. For $T_a = 225$ K, N_{ice} is at the end lower for $\kappa = 0.5$ than for $\kappa = 0.05$ and the total ice number is lower than the reference ice number contrary to the high T_a cases. This is due to the decrease in T_{frz} for droplets with a higher solution effect (lower a_{wat}), as explained in Sect. 4.3.1 and shown in Fig. 5 (c).

The last two rows show the final ice crystal numbers ($N_{\text{ice,f}}$) for the same aerosol particle ensembles as in panels (a)–(d) but for different aerosol number concentrations. Basically, we see a similar trend for the ice crystal numbers of the single ensembles and the reference case as in the panels above throughout the whole n_{aer} -range. In general, the increase in $N_{\text{ice,f}}$ with increasing n_{aer} becomes weaker for higher n_{aer} , in particular for the low T_a cases. This is consistent with the findings for a single aerosol ensemble (as already shown in Fig. 5 (a)). The flattening in $N_{\text{ice,f}}$ is pronounced most strongly for the nucleation mode and the weakly hygroscopic particles. Again, the total $N_{\text{ice,f}}$ of the two-particle ensemble is nearly always reduced compared to the

reference $N_{\text{ice},f}$ of the single average particle ensemble except for the low T_a cases in panel (g) and (h).

580 Finally, we find largest differences between the total ice crystal number of the co-existing particle ensembles and the associated average single particle ensemble for the cases where nucleation mode particles are involved. This is mainly due to the non-linearity between (final) contrail ice crystal number and mean dry size for those very small aerosol particles.

4.4 Comparison of H₂ contrails with conventional contrails

In the present section, we study differences in microphysical and optical properties of H₂ contrails compared to conventional contrails formed behind aircraft with kerosene combustion. For the latter, we include in our set-up both ice crystal formation
585 on soot and on the entrained ambient particles. We analyze in Sect. 4.4.1 the number of formed contrail ice crystals as a first step to estimate the mitigation potential of H₂ combustion. In Sect. 4.4.2, we analyze the optical thickness, which can provide information about the visibility of young contrails. While first measurements like Blue Condor could use this information for their planning, they potentially also provide a first chance for the evaluation of our model. We use the respective engine and fuel parameters defined in [Tab. Table 1](#). We define the properties of ambient and soot particles according to [Tab. Table 2](#).

590 4.4.1 Mitigation potential

Fig. 7 compares ice crystal numbers of conventional and H₂ contrails. Panel (a) displays the ice crystal number $N_{\text{ice},f}$ behind a conventional aircraft as a function of ambient temperature and for three typical soot number emission levels. For our calculated fuel consumption (that accounts for the different Q -values of H₂ and kerosene), the displayed soot particle numbers ($N_s = 1.54 \cdot 10^{12} \text{ m}^{-3}$, $3.07 \cdot 10^{12} \text{ m}^{-3}$ and $6.14 \cdot 10^{12} \text{ m}^{-3}$) represent soot number emission indices of around $5 \cdot 10^{14}$, $1 \cdot 10^{15}$ and
595 $2 \cdot 10^{15} \text{ kg}^{-1}$, respectively. Prescribing our baseline ambient pressure and ambient relative humidity, the SA-threshold temperature Θ_G for kerosene is around 227 K (≈ 10 K lower than that for H₂). Very close to Θ_G , only a few soot and the entrained ambient aerosol particles can form ice crystals due to very low plume water-supersaturations. For ambient temperatures ($T_a \lesssim \Theta_G - 0.5$ K), ice crystals mainly form on soot particles since N_s is by around 2 orders of magnitude higher than N_{aer} during the contrail formation time (not shown). Consistent with previous studies (e.g., Kärcher et al., 2015; Bier and Burkhardt, 2019; Bier et al., 2022), $N_{\text{ice},f}$ strongly increases with decreasing T_a and then approaches the respective N_s -values for sufficiently low
600 T_a . For higher N_s , the number of formed ice crystals rises more steeply and approaches N_s at lower T_a .

Now, we investigate the ratio of the ice crystal numbers between H₂ and kerosene contrails ($N_{\text{ice},f,\text{rel}}$) shown in panels (b)–(d). We constrain our analysis to that temperature range in which kerosene contrails are able to form according to the SA-criterion. A mitigation is achieved for $N_{\text{ice},f,\text{rel}} < 1$, where a lower value is connected with a higher mitigation potential. Panel (b) shows
605 $N_{\text{ice},f,\text{rel}}$ versus n_{aer} for the three N_s (see line style) and for three T_a cases (different colors). In general, we see a clear decrease in $N_{\text{ice},f,\text{rel}}$ with increasing N_s and decreasing n_{aer} . Thereby, $N_{\text{ice},f,\text{rel}}$ is below 0.1 for $n_{\text{aer}} \leq 600 \text{ cm}^{-3}$ and below 0.15 for higher n_{aer} . Interestingly, we see for the higher N_s and n_{aer} cases a larger difference in $N_{\text{ice},f,\text{rel}}$ between 220 and 222 K than between 220 and 225 K.

Therefore, we investigate in more detail the temperature dependency of $N_{\text{ice},f,\text{rel}}$ in the second row of the Figure. Panel (c)
610 shows the relative change in the ice crystal number w. r. t. the three soot cases for our baseline n_{aer} . In connection with the

minimum at $T_a \approx 224$ K, there are two different dominating effects: Below 224 K, $N_{ice,f,rel}$ increases with decreasing T_a , in particular for the high soot case. This is because $N_{ice,f}$ for the kerosene contrails approaches N_s with decreasing T_a (well below the SA-threshold) while $N_{ice,f}$ for the H₂ contrails increases further for lower T_a . The latter is because the ambient aerosol is continuously entrained into the plume and the time period for droplet and ice crystal formation increases for colder ambient conditions due to a longer lasting water-supersaturation (see Sect. 4.2). The strong increase in $N_{ice,f,rel}$ above around 225 K results from the strong decrease of $N_{ice,f}$ for the conventional contrail. This is because ice crystal formation on the weakly soluble soot particles becomes more and more limited the closer the ambient temperature approaches the SA-threshold temperature. Thereby, $N_{ice,f,rel}$ is around 0.3 for $T_a = 226.5$ K and around 20 for $T_a = 227$ K (the latter not visible in the figure). Finally, we show the change in ice crystal number relative to our baseline soot case but for three different n_{aer} -values. The sensitivity of $N_{ice,f,rel}$ to n_{aer} is quite similar to that to N_s in panel (c). The most obvious difference is that an increase of n_{aer} by a factor of 2 (brown line in (d)) has a much lower impact on $N_{ice,f,rel}$ than a decrease in N_s by the same factor (dashed line in panel (c)). This is due to a weaker increase of the H₂ contrail ice crystal number with increasing n_{aer} for high number concentrations (shown in Fig. 5 a).

We can conclude that a switch to H₂ combustion indicates a high mitigation potential if the ambient temperature is more than 0.5 K lower than the SA-threshold temperature for kerosene. This is mainly because N_{aer} during contrail formation is around 2 orders of magnitude lower than N_s for typical n_{aer} -values. Another aspect is that the fuel consumption of ~~the cryoplane is in our study by a cryoplane is chosen to be~~ around a factor of 2.8 lower than that of ~~conventional aircraft~~ conventional aircraft (only based on the difference in the combustion heat of the fuels). If air traffic with H₂ combustion occurs at ambient temperatures between the kerosene SA-threshold temperature and the droplet freezing temperature, clearly additional contrails are produced, which would be absent in the case of kerosene combustion.

4.4.2 Contrail visibility

We investigate the young contrail optical thickness (τ) both for kerosene and for H₂ combustion. For the quantities analyzed and presented so far, we used a reduced trajectory data set after merging trajectories with similar radial coordinates (see Sect. 3.2.1). However, the column-wise computation of τ requires spatial information of the trajectories, namely the lateral and vertical Cartesian coordinates x and z . Hence, the results presented next were obtained by using the full trajectory data set from Vancassel et al. (2014). Fig. 8 (a)–(d) shows the contrail width (indicated by the x -coordinate)-plume age (t) distribution of τ : The first row displays the kerosene and the second row our H₂ baseline case, both for T_a of 220 K and 225 K, respectively. Observations suggest that the threshold τ for the visibility of contrails is around 0.05 (e.g., Kärcher et al., 2009). Panel (a) indicates that the kerosene contrail at $T_a = 220$ K becomes visible after around 0.25 s of plume age. Afterwards, the plume quickly spreads and τ tends to increase due to further formation and growth of ice crystals. Peak values of around two and slightly higher are reached for t between 1 and 1.5 s. For $T_a = 225$ K, ice crystals form later and the contrail is visible after around 0.5 s. Maximum τ is by around 50% lower than for the low T_a case since that contrail forms near the formation threshold and the nucleated ice crystal number is significantly reduced (see Fig. 7 (a), black line). For the H₂ case, the optical thickness is substantially decreased compared to kerosene contrails, being consistent with the findings by Ström and Gierens (2002).

645 Moreover, these contrails become visible later than those for kerosene at same T_a .

Finally, panel (e) shows the temporal evolution of the 90th percentile optical thickness over the contrail width τ_{90} . We juxtapose the kerosene contrail with three H_2 contrails with different n_{aer} -values. Consistent with the spatio-temporal distributions, τ_{90} of the H_2 contrails is significantly lower than for the kerosene contrails (in particular for $T_a = 220$ K). The optical thickness decreases for lower n_{aer} -values. We expect that the H_2 contrail for $n_{\text{aer}} = 100 \text{ cm}^{-3}$ would be hardly visible. A lower temperature
650 causes a slight increase in τ_{90} for the higher n_{aer} cases and for $t < 1.5$ s. This is mainly due to the higher ice water content for lower T_a (not shown). Instead, the much stronger increase of τ for the kerosene case is due to the increased ice crystal number already at the beginning of contrail formation.

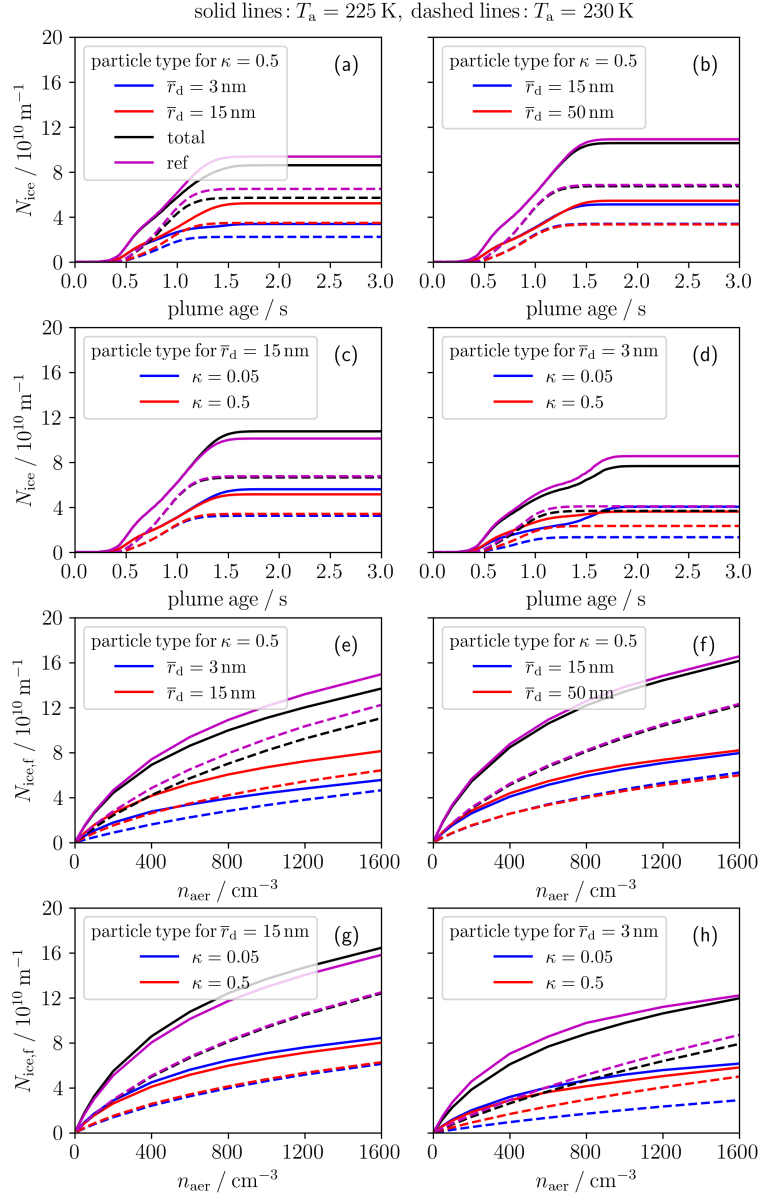


Figure 6. Ice crystal number per flight distance for two co-existing ambient aerosol particle ensembles: All results are shown for two ambient temperatures (225 K solid and 230 K dashed). The first two rows show the temporal evolution of ice crystal number N_{ice} and the last two rows show the final ice crystal number $N_{\text{ice},f}$ (at a plume age of 3 s) versus the total ambient aerosol number concentration of both ensembles. The blue and red lines depict N_{ice} of the single aerosol ensembles and the black lines the sum of the both co-existing ensembles. The magenta lines represent reference cases from simulated single aerosol particle ensembles prescribing average quantities of the two co-existing ensembles. Panels (a)/(e) and (b)/(f) show results for a bi-modal aerosol size distribution (with mean dry radii \bar{r}_d as displayed in the legends) for a fixed hygroscopicity parameter $\kappa = 0.5$. The other panels show results for two aerosol particle ensembles with a different solubility ($\kappa = 0.05$ and 0.5) but a fixed \bar{r}_d , namely 15 nm in panels (c)/(g) and 3 nm in panels (d)/(h).

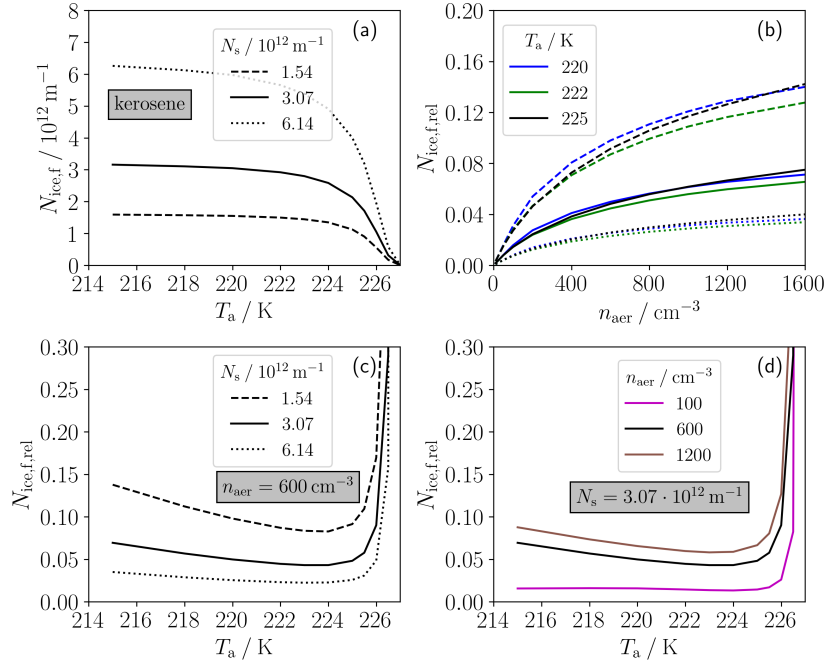


Figure 7. (a) Final ice crystal number of a conventional (kerosene) contrail versus ambient temperature (T_a) for 3 soot particle emission numbers per flight distance N_s (different line styles). The other panels show the final ice crystal number of H_2 contrails relative to the final ice crystal number of kerosene contrails at same atmospheric conditions. Panel (b) shows $N_{ice,f,rel}$ versus ambient aerosol number concentration (n_{aer}) for the different soot number emissions and three T_a cases (see legend (b)). Panel (c) shows the temperature variation of $N_{ice,f,rel}$ of one H_2 contrail (with $n_{aer} = 600 cm^{-3}$) with respect to three kerosene contrails with different N_s as in panel (b). Panel (d) shows the temperature variation of three H_2 contrails (with different n_{aer} as displayed in the legend) with respect to one kerosene contrail ($N_s = 3.07 \cdot 10^{12} m^{-1}$). Thereby, the black solid lines in panels (c) and (d) represent the same case. The different line styles always refer to the three soot emission cases. The displayed quantities are given at a plume age of 3 s. All absolute numbers refer to a single-engine plume.

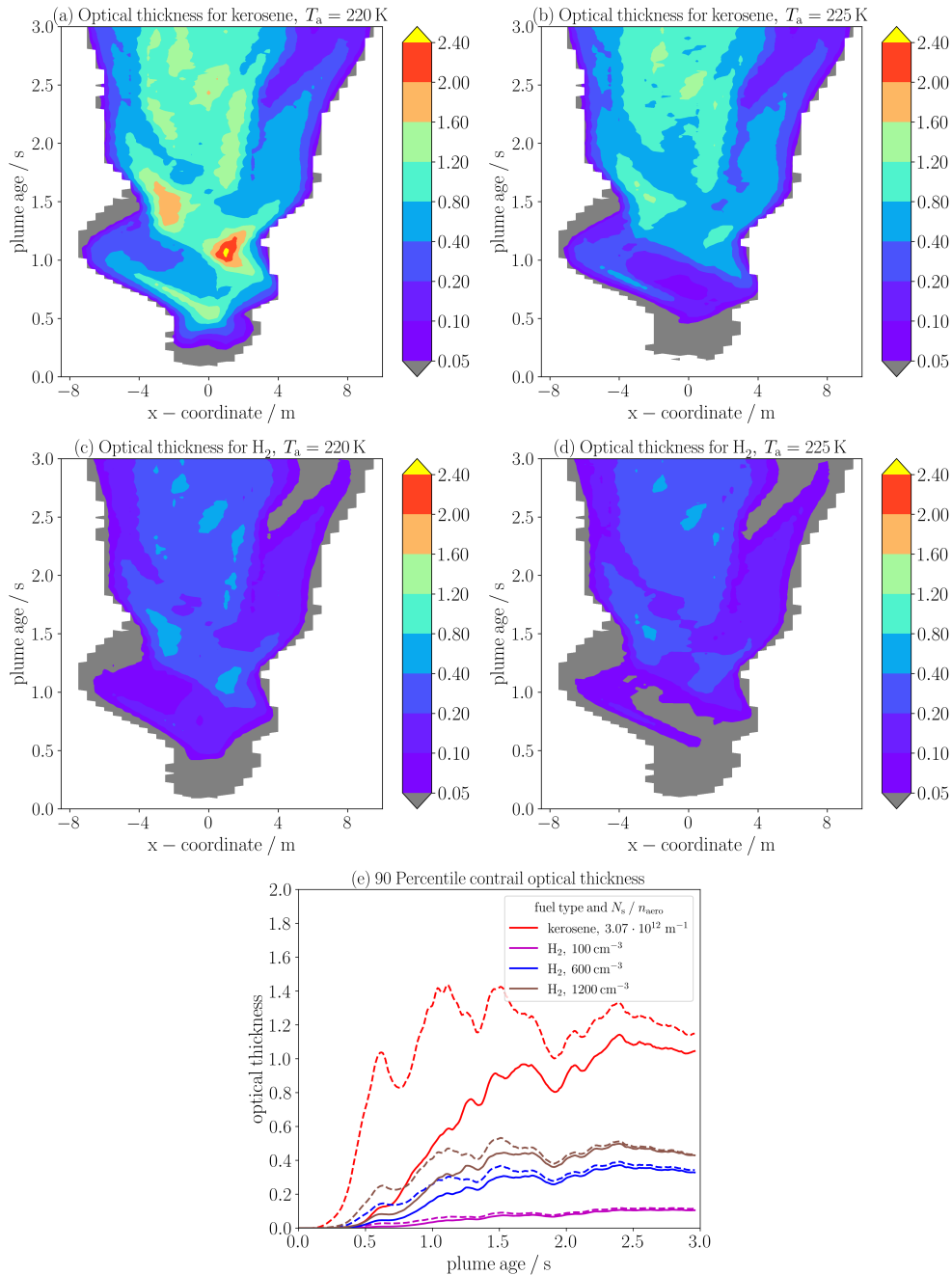


Figure 8. Contour plots (first two rows) of young single-engine contrail optical thickness τ over the trajectories' x-coordinate indicating the contrail width and the plume age. The first row show results for kerosene (with $N_s = 3.07 \cdot 10^{12} \text{ m}^{-1}$) and the second for the baseline H_2 case, both for ambient temperature (T_a) of 220 K on the lefthand and of 225 K on the righthand side. (e) shows the 90th percentile optical thickness over the width τ_{90} for kerosene (red) and for H_2 prescribing three different ambient aerosol number concentrations (other colors in the legend) for $T_a = 220$ K (dashed lines) and $T_a = 225$ K (solid lines). The temporal evolution of τ_{90} has been smoothed using a running average method. For this analysis, we use the full FLUIDILES trajectory ensemble of 25000 instead of the reduced ensemble of 1000.

5 Discussions

5.1 Potential sources for the formation of ultrafine volatile particles

655 In this study, we considered contrail ice crystals to form solely on ambient particles entrained in the plume. While H₂ combustion emissions are in general expected to be soot-free, the formation of ultrafine volatile particles (UFPs), which can also contribute to contrail ice crystal formation, is still possible. Two potential sources for the formation of UFPs behind H₂ combustion engines are discussed in the following:

5.1.1 Nitrogen compounds

660 Several recent studies considered chemi-ions that are mainly composed of sulfur species (e.g., Yu and Turco, 1997). While sulfur is likely not produced during H₂ combustion, NO_x is still emitted. The reaction of NO_x with H₂O and OH-species potentially leads to the formation of nitrogen compounds like nitric acid that were observed in conventional aircraft plumes (e.g., Tremmel et al., 1998). Moreover, Wang et al. (2020) have shown within cloud chamber experiments that nitric acid and ammonia can nucleate directly to form volatile ammonium nitrate particles at temperatures below 258 K. Finally, nitrogen
665 species might be a potential source for the nucleation of ultrafine volatile particle both in conventional and H₂ combustion plumes, but the formation process behind is not yet sufficiently understood.

5.1.2 Ultrafine oil particles

We expect that engine lubrication systems will be still used for H₂ engines causing emissions of oil vapors. Even though an air-oil separator recovers around 99% of the oil emissions, the residual may contaminate the engine plume. Ungeheuer et al.
670 (2022) have shown in laboratory experiments that "jet oil vapors reach gas-phase supersaturation in cooling emission plumes leading to rapid nucleation and formation of ultrafine volatile particles in the range of ~ 10–20 nm." These diameter ranges appear to be consistent with the ground-measured ambient UFPs downwind of the Frankfurt Airport, in which organic engine oil constituents have been identified (Ungeheuer et al., 2021).

The formed UFPs can contribute to droplet and ice crystal formation in addition to the background aerosol. Assuming a typical
675 oil consumption of around 1 lh⁻¹ with 1% volume fraction (residual) that enters the plume, our estimates supply that the UFP number per flight distance could be even larger than that of soot particles. Since plume water-supersaturations are much higher for H₂ than for kerosene at same ambient conditions, we expect that droplets and ice crystals would mainly form on those UFPs rather than on the entrained ambient particles. This means that the number of ice crystals could be similar or even increased compared to conventional contrails. One should still keep in mind that the experiments of Ungeheuer et al. (2022) refer to
680 conventional kerosene combustion and the properties of the oil particles (size and chemical composition) could significantly change for H₂. Moreover, associate measurements at cruise altitude conditions are necessary which confirm the occurrence of those UFPs.

Finally, a hermetic and clean sealing of the engines from the oil system aiming at a complete jet oil recovery could be a

technical means to achieve a valuable mitigation effort for contrail formation if future model studies and flight campaigns give
685 hints at abundant droplet formation on oil particles.

5.2 Scaling relations for different plume area evolution and fuel consumption

In the present study, we use the FLUDILES trajectory data (Vancassel et al., 2014) that were modified according to the descrip-
tion in Sect. 3.2.1. The presented results represent a single engine plume of an A340 aircraft. Extensive contrail quantities (like
ice crystal number and mass) for the whole aircraft may be scaled with the number of engines (in our case, [it is](#) 4). This scaling
690 is valid as long as we assume that there is no interference of the two exhaust plumes (on one side of the aircraft) during the
contrail formation stage. This scaling could also be interpreted as a scaling with the ratio of the total fuel consumption ($m_{C,tot}$)
and our reference value $m_C = 1.1 \text{ g m}^{-1}$ (for the baseline T_a and p_a -values). Basically, one could plug in any reasonable value
for $m_{C,tot}$. However, this flexible fuel consumption scaling approach is only valid with several underlying assumptions that are
usually not fulfilled. Firstly, one had to assume that the initial plume area A_E scales with the fuel consumption and, secondly,
695 that the plume dilution $\mathcal{C}(t)$ is independent of A_E and hence $A(t) \sim A_E$.

However, Lewellen (2020) showed that the evolution of \mathcal{C} itself depends on engine size. A different dilution does not only
impact the plume area evolution but also the thermodynamic plume and the microphysical contrail properties (Lewellen, 2020;
Bier et al., 2022). Neglecting those constraints and assuming fixed atmospheric conditions and aerosol particle properties,
we observe that $N_{ice} \sim N_{aer} \sim A(t) \sim A_E \sim m_{C,tot}$. This behavior is similar to kerosene contrails, where we find $N_{ice} \sim N_s \sim$
700 $m_{C,tot}$. Again, this is only a rough estimate with several underlying assumptions and for a fixed soot number emission index of
 $EI_s = N_s/m_{C,tot}$.

So far, we have stressed that the given reference value of m_C holds (only) for the baseline values of ambient temperature and
pressure. In our study, we use a constant A_E , which refers to a particular aircraft type with an engine nozzle radius of 0.5 m
(see [Tab. Table 2](#)). Moreover, we keep the plume exit temperature T_E fixed in our setup.

705 From Eqs. (4) and (10), it follows that the initial dilution and, correspondingly, the fuel consumption implicitly change when
 T_a takes a different value ($m_C \sim (T_E - T_a)$). However, this T_a -dependence of C_E and m_C is much less crucial for N_{ice} than the
impact of T_a on the plume relative humidity evolution. A change in ambient pressure p_a does not affect C_E , but m_C changes
linearly with p_a . Hence, the implied m_C -values in our p_a -sensitivity study differ non-negligibly. Clearly, these changes of the
 m_C -values are implications of our choices in study design. In reality, the fuel consumption depends on the thrust setting and
710 may change differently to how we prescribed it, when p_a and/or T_a change (see e.g. Lewellen, 2020; Volponi, 1998; Kurzke,
2003).

5.3 Deliquescence relative humidity

We prescribe a fixed deliquescence relative humidity of the aerosol particles $DRH = 0.99$ in our set-up, a value close to water
saturation. The definition of a lower baseline value (according to Peng et al. (2022)) with appropriate sensitivity variations
715 would have been more reasonable. However, we achieved in several test simulations not robust results for $DRH \lesssim 0.95$, in
particular for low mean aerosol particle dry size and hygroscopicity parameter. This is likely due to one technical reason: While

720 water saturation is reached first near the plume edge and later on in the plume center in the first tenths of s, higher relative humidities last longer in the plume center than at the edge towards the end of contrail formation (see e. g. Fig. 1 (d) of Bier et al. (2022)). The lower the DRH -value, the longer is the potential time period for droplet and subsequent ice crystal formation, in particular for lower ambient temperatures. The ambient particles are mainly entrained near the plume edge. Since we cannot resolve this heterogeneous entrainment but ambient particles are mixed in for each trajectory with equal share, droplet and ice crystal formation is likely overestimated and this overestimation increases with decreasing DRH .

6 Conclusions and outlook

In the recent past, several model studies investigated contrail formation behind commercial aircraft by means of analytical approaches (Kärcher et al., 2015; Bier and Burkhardt, 2019), 0D box models (e.g., Kärcher and Yu, 2009; Vancassel et al., 2014; Bier et al., 2022) and LES (e.g., Paoli et al., 2013; Khou et al., 2015; Lewellen, 2020). These studies focused on contrail formation on soot particles on which the majority of ice crystals form for conventional engines (e.g., Kärcher and Yu, 2009; Kleine et al., 2018). Switching to liquid hydrogen (H_2) propulsion, ice crystals are expected to form solely on background particles mixed into the plume. Even though some of those studies account for ice crystal formation on background particles (e.g., Kärcher and Yu, 2009; Kärcher et al., 2015; Lewellen, 2020), the implementation of this process is quite simplified (e. g. activation relaxation approach in Kärcher et al. (2015)) and ambient particle properties are mostly kept fixed.

While Bier et al. (2022) have extended the particle-based Lagrangian Cloud Module (LCM; (Sölch and Kärcher, 2010)) by contrail formation microphysics on soot particles, we here advance the LCM by specific contrail formation microphysics on entrained background aerosol particles. The most relevant feature is that ambient particles are continuously entrained into the plume instead of releasing a fixed number of soot particles. Moreover, we define an alternative droplet activation criterion and improve the homogeneous freezing parameterization by accounting for the impact of the solution effect on droplet freezing.

Given the same atmospheric conditions and propulsion efficiency, the Schmidt-Appleman (SA)-threshold temperature (Schumann, 1996) is by around 10 K higher for H_2 than for kerosene fueled aircraft due to around 2.6 times higher water vapor emissions for the same amount of released combustion heat. The homogeneous freezing temperature of water droplets is in general smaller than the SA-threshold temperature for H_2 contrails and, therefore, becomes a more limiting criterion for contrail formation as already pointed out by Gierens (2021).

The ice crystal number is strongly cut down for temperatures above around 230 K since smaller droplets do not freeze to ice crystals any longer. Contrails cannot form anymore at temperatures above around 233–234 K in our study. While for kerosene combustion the number of formed ice crystals approaches the emitted soot particle number for a sufficiently low ambient temperature (e.g., Kärcher et al., 2015), the ice crystal number of H_2 contrails increases further with decreasing temperature. The latter is because the water-supersaturation in the plume lasts longer for colder conditions and, hence, more of the entrained aerosol particles can form droplets and ice crystals.

Our results highlight a large variability in the number of formed contrail ice crystals with varying ambient aerosol properties. For a fixed particle size distribution and chemical composition, the ice crystal number clearly rises with increasing aerosol number concentration. This increase becomes weaker for higher number concentrations ($\gtrsim 200 \text{ cm}^{-3}$), in particular in a colder environment. The variation of contrail ice nucleation with aerosol mean dry size and water solubility is low for larger aerosol particles and high for small (mean radius $\lesssim 10 \text{ nm}$) particles. For these smaller particles, the sensitivity of the ice crystal number with the water solubility of the aerosol particles is quite complex for lower temperatures due to various counteracting microphysical processes.

In the real atmosphere, the background aerosol typically consists of multiple particle types with different mean dry sizes (modes) and chemical composition. Therefore, we analyze contrail formation prescribing two co-existing aerosol particle en-

sembles that either differ in the mean dry size or hygroscopicity parameter. If these co-existing particle ensembles contain only larger (mean dry radii more than around 10 nm) and well soluble aerosol particles, the ice crystal number for each of the ensemble can be estimated well from a simulation with appropriate single particle ensembles. The total ice crystal number of the co-existing particle ensembles can be also approximated from one single particle ensemble prescribing average properties of mean dry size and solubility and the total number concentration of the co-existing particle ensembles. This is because the ice crystal number is not so sensitive to changes in this large particle range as mentioned above. Conversely, such an approach is not meaningful if a substantial fraction of small and weakly soluble aerosol particles (in particular nucleation mode particles) is present. If these particles co-exist with larger and/or better soluble particles, droplet and ice crystal formation on these particles might be significantly suppressed due to the competition effects between the aerosol particles. Due to the non-linearity between ice crystal number and mean dry size for small particles, the total ice crystal number of the co-existing particle ensembles might be significantly different to the average single particle ensemble.

Finally, we compare ice crystal formation, as a first measure of the mitigation potential, and visibility of H₂ contrails with conventional contrails. Varying both aerosol number concentration and soot number emissions, the H₂ contrail ice crystal number is significantly reduced (by more than 80–90%) compared to conventional contrails implying a great mitigation potential. This is mainly because ambient aerosol number concentrations are at least 1-2 orders of magnitude lower than soot particle number concentrations in young exhaust plumes behind conventional aircraft. For ambient temperatures only slightly below ($\lesssim 0.5$ K) the SA-threshold temperature for kerosene combustion, the H₂ contrail ice crystal number can be higher since ice crystal formation on the weakly soluble soot particles becomes strongly limited by very low plume water-supersaturations (e.g., Kärcher et al., 2015). The optical thickness is significantly decreased and the H₂ contrails either become slightly later visible or they might not be visible at all for low ambient aerosol number concentrations. On the other hand, H₂ contrails can form at lower flight altitudes (connected with ambient temperatures lying between the SA-threshold temperature for kerosene and the homogeneous freezing temperature of the water droplets) than conventional contrails, as also mentioned by Ström and Gierens (2002). In case of persistent contrails, this would increase the contrail coverage and partially compensate the mitigation potential of hydrogen for the aviation climate impact.

Data availability. The presented data are available from the corresponding author upon request (Andreas.Bier@dlr.de).

Appendix A: Numerical convergence and SIP merging

Once RH_{wat} surpasses DRH for the first time, a new SIP ensemble that represents the newly entrained dry aerosol is created in every time step. In order to keep the total SIP number in an acceptable range, we employ a SIP merging technique where several similar-sized SIPs of the same category (aerosol, droplets and ice crystals) are merged into a single SIP. The merge operation is implemented such that the number and mass of the represented physical particles are conserved (Unterstrasser and Sölch, 2014). The SIP merging is executed when N_{SIP} exceeds a certain threshold value (fixed to 1600 in our study). Then the new N_{SIP} value is well below that threshold and starts to increase again. In the end, the SIP number follows a jigsaw pattern as exemplarily shown in Fig. A1. The merge operation has some more (internal) parameters, e.g. how many SIPs are at most merged and what is maximum relative difference between SIPs that are merged. We experimented with those parameters and found numerical convergence, that means our present configuration yields basically identical results with simulations with higher N_{SIP} .

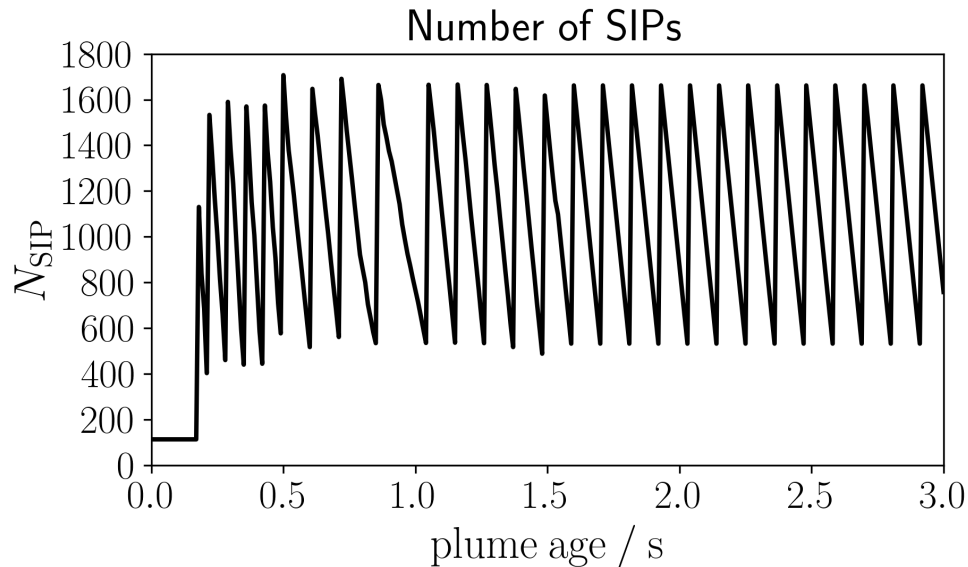


Figure A1. Temporal evolution of the number N_{SIP} of Simulation Particles (Simulation Particles) (N_{SIP}) for a contrail forming on ambient particles using one average FLUDILES trajectory at baseline conditions (see Tab. Table 2).

Appendix B: Parameterization of homogeneous freezing temperature

795 We calculate the homogeneous freezing temperature as

$$T_{\text{frz}} = T_{\text{frz},0}(V_{\text{wat}}, \dot{T}) - \Delta T(a_{\text{wat}}), \quad (\text{B1})$$

where $T_{\text{frz},0}$ is the freezing temperature assuming a pure super-cooled water droplet ($a_{\text{wat}} = 1$). It is determined from Eq. (6) of Bier et al. (2022) following the approach of Kärcher et al. (2015) and Riechers et al. (2013) and varies with droplet water volume V_{wat} and cooling rate \dot{T} .

800 The second term is a correction term based on the study from O and Wood (2016) approximating the decrease in T_{frz} with decreasing activity of water a_{wat} (e.g., Koop et al., 2000). O and Wood (2016) developed an approximation for the homogeneous freezing temperature of solution droplets ($T_{\text{frz},\text{owood}}$) as a function of water volume and activity, however neglecting the cooling rate. Thereby, that temperature is iteratively calculated for which the mean number of critical embryos becomes equal to one triggering the freezing process in the droplet (see their Eq. (1)). For our activated water droplets ($a_{\text{wat}} > 0.90$), we find that the
 805 parameterization yield robust results for droplet radii $r \geq 1 \mu\text{m}$. On the other hand, the solution effect is frequently important for smaller droplets ($r < 1 \mu\text{m}$) formed on freshly entrained aerosol particles.

For simplicity, we prescribe a fixed droplet radius $r = 1 \mu\text{m}$ for the estimation of our correction term. We evaluate $T_{\text{frz},\text{owood}}$ for different a_{wat} values that result from a variation of r_d . Setting $\Delta T = T_{\text{frz},\text{owood}}(a_{\text{wat}} = 1) - T_{\text{frz},\text{owood}}(a_{\text{wat}})$ and $\Delta a_{\text{wat}} = 1 - a_{\text{wat}}$, we define the following quadratic fit function

$$810 \quad \Delta T = a \cdot \Delta a_{\text{wat}}^2 + b \cdot \Delta a_{\text{wat}} + c, \quad (\text{B2})$$

which is valid for $a_{\text{wat}} \geq 0.90$ with the fit parameters $a = 345.746$, $b = 100.977$ and $c = 0.01687$ and a square root mean error $R^2 = 0.99999$. Even though the expression for ΔT is a simplified correction term, Eq. (B1) combines the sensitivity of the homogeneous freezing to all major effects.

Fig. B1 shows the variation of T_{frz} with droplet radius r and water activity for a plume cooling rate of -10 K s^{-1} . For $a_{\text{wat}} = 1$
 815 (representing pure water droplets), T_{frz} equals to $T_{\text{frz},0}$ and ranges between around 229 and 233 K. Thereby, T_{frz} increases with rising r . The strong decrease of T_{frz} with decreasing a_{wat} (for fixed r) down to around 215 K emphasizes the importance of accounting for the solution effect in droplets. In contrast, the impact of a varying cooling rate is low (not shown).

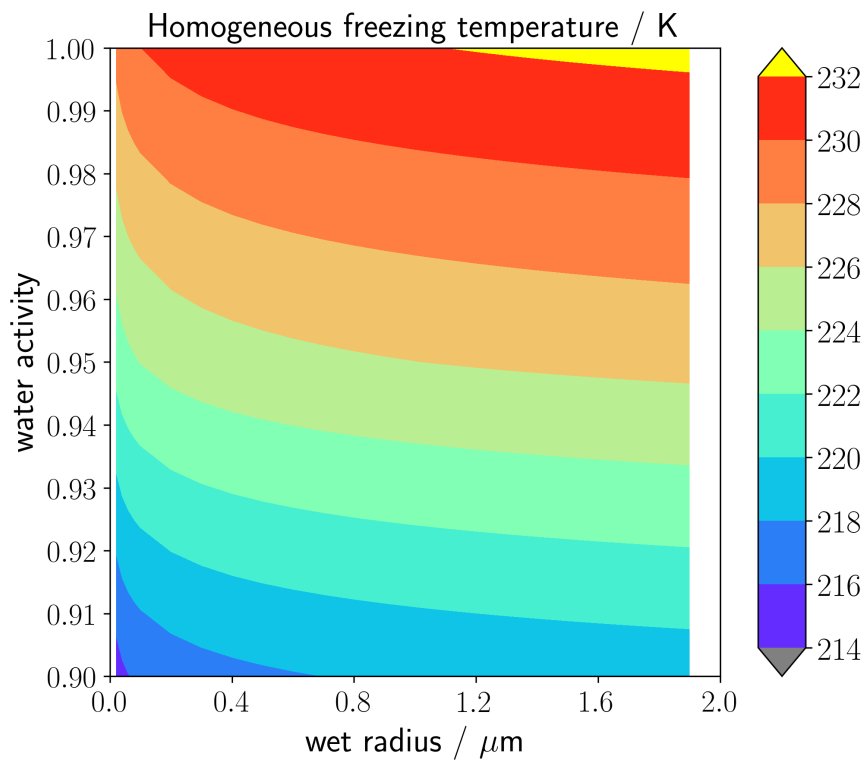


Figure B1. Contour plot showing the homogeneous freezing temperature of supercooled solution droplets over the wet radius and water activity. The cooling rate is set to -10 Ks^{-1} .

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