1 **Fine particle chemistry under a special dust transport event:**

2 **impacts from unusually enhanced ozone and air mass**

3 **backflows over the ocean**

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22 **Abstract**

23   A five-days long-lasting dust event was observed with a synergy of field

24 measurements techniques in Shanghai in the autumn of 2019. Different from most dust

25   events, this dust was an unusual one characterized of low wind speed, high relative

26   humidity, high concentrations of gaseous precursors, and contrasting wind vectors

27   between low and high altitudes. Three dust stages were identified and the first stage

28   was a normal dust invasion with high particulate concentrations and short duration. In

29   contrast, unusual enhancement of ozone was observed in the second stage, due to

30   compound causes of weak synoptic system, transport from the ocean, and subsidence

31   of high-altitude O_3 down drafted by dust. As a result, sulfate and nitrate moderately

32   correlated with O_3 while had almost no correlation with aerosol liquid water content,

33   indicating the dominant role of gas phase oxidations. During the third stage of dust, a

34   special phenomenon of dust backflow was observed that the dust plume drifted from

35   the Shandong Peninsula and travelled slowly over the Yellow Sea and the East China

36   Sea, finally returning to Shanghai. The dust backflow was evidenced by the enrichment

37   of marine vessel emissions (V and Ni) and increased solubility of calcium. Under the

38   humid oceanic breezes, the formation of nitrate was dominated by aqueous processing,

39   while the strong correlation between SO_4^{2-} and Na^+ suggested that a considerable part

40   of sulfate was aged and directly transported. Based on the thermodynamic modeling,

41   sea salts probably involved more in the secondary aerosol formation than the dust

42   heterogeneous reactions. By developing an upstream-receptor relationship method, the

43   amounts of transported and secondarily formed aerosol species were separated. This

44   study highlights that the transport pathway of dust and environmental conditions could

45   significantly modify the aerosol properties, especially at the complex land-sea interface.

46

47   **1. Introduction**

48   As an important source of natural aerosols, dust accounted for about half of the

49   tropospheric aerosols (Zheng et al., 2016). Dust aerosols played important roles in

50   environmental and climatic changes by affecting the radiation balance (Feng et al., 2020;

51   Nagashima et al., 2016; Goodman et al., 2019). Furthermore, dust aerosols had

52   important influences on tropospheric chemistry by participating in heterogeneous and

53   photolysis reactions in the atmosphere (Wang et al., 2014; Xu et al., 2018). For instance,

54   dust could mix with gaseous pollutants, toxic metals, and soot during transport, thus

55   affecting air quality immediately and causing potential public health hazards (Liu et al.,

56   2021; Wang et al., 2021). Moreover, Barkley et al. (2021) found that iron-containing

57   aerosols transported from Africa to the equatorial North Atlantic Ocean provided

58   plentiful nutrients to algae in the ocean and accumulated inside algae.

59   Many studies focused on the emissions and transport of Asian dust, which

60   accounted for ~20% of the global dust budget (Ginoux et al., 2004). Asian dust mostly

61   originated from the deserts in western China, the Gobi, and the Loess Plateau

62   (Nagashima et al., 2016). Dust particles can be lifted to an altitude of several kilometers

63   due to strong winds and low soil moistures. During this process, most of the coarse dust

64   particles would settle near the dust source areas, while relatively fine particles could be

65   transported to further downstream regions such as eastern and southern China, and even

66   across the Pacific Ocean to the western America coast (Fu et al., 2010; Vicars and

67   Sickman, 2011).

68   The irregular shapes of dust particles provided efficient medium for

69   heterogeneous reactions with NO_2, O_3, SO_2, and NH_3, thus changing the particle size

70   spectrum, hygroscopicity, and radiative properties (Hsu et al., 2014; Tian et al., 2021;

71   Wang et al., 2018a). Wang et al. (2018a) observed that the concentrations of nitrate and

72   sulfate were significantly elevated due to the moderate/high levels of relative humidity

73   and gaseous precursors during a dust period in March, 2010 at Shanghai, implying that

74   dust can efficiently promoted the formation of sulfate and nitrate. The formation of

75   nitrate included two major pathways. During daytime, the OH radicals produced by the

76   photolysis of O_3 and HONO oxidized NO_2 to produce HNO_3 (Hertel et al., 2012), which

77   subsequently neutralized the alkaline substances to form nitrate in the particles. During

78   nighttime with low temperature and high humidity, NO_2 can be oxidized by O_3 to form

79   NO_3 radical (Mentel et al., 1996), which then reacted with NO_2 to form N_2O_5 radical

80   (Dall'osto et al., 2009; Petetin et al., 2016). Previous studies have revealed that HNO_3

81   formed through the reactions of NO_2 with hydroxyl radical or N_2O_5 hydrolysis would

82   preferentially react with mineral dust particles and produce nitrate, which was the main

83   source of nitrate during the dust period (Tang et al., 2016; Wu et al., 2020). Wang et al.

84   (2018b) found that the heterogeneous reactions on the dust surface were the main

85   sources of nitrate, and the dust surface nitrate observed in Japan mainly formed over

86   the Yellow Sea and the East China Sea during the long-range transport.

87   However, there were some controversies on the mixing of dust and anthropogenic

88   aerosols. Zhang et al. (2005) found that anthropogenic aerosols separated with dust

89   during a dust event in Qingdao, China. Coincidentally, there existed a time-lag between

90   dust and anthropogenic aerosols in Japan and South Korea downstream of the dust

91   transport. Single particle analysis confirmed that sulfate in fine particles appeared 12

92   hours before the dust arrival in Japan. Wang et al. (2013) also observed a 10 - 12 hours

93   lag between dust and anthropogenic aerosols in a dust day in Shanghai (Wang et al.,

94   2013). Furthermore, Huang et al. (2019) observed that there were vertical differences

95   of the long-transported aerosols during a pollution event in Taiwan. Dust from the Gobi

96   Desert in Inner Mongolia and China existed at the altitudes of 0.8km and 1.90km,

97   respectively, while biomass burning aerosols from South Asia existed at higher altitudes

98   of 3.5km.

99   In the coastal regions, the regional pollution was always a mix of inland

100   anthropogenic emissions and ocean-sourced releases. Due to the active human activities

101   and special weather conditions such as monsoon and sea-land breezes, the atmospheric

102   compound pollution was usually more complex in the coastal areas (Wang et al.; Hilario

103   et al., 2020; Patel and Rastogi, 2020; Perez et al., 2016; Wang et al., 2017). The eastern

104   coast of China is bordering the East China Sea and the Yellow Sea and is strongly

105   influenced by the Asian monsoon and high emissions from inland industries. The

106   meteorological conditions and pollution conditions in this region were among the most

107   complex in the world (Hilario et al., 2020). Due to the complex meteorological

108   conditions, the relative humidity and temperature of the marine boundary layer showed

109   significant seasonal and diurnal changes, further affecting the photochemical processes

110   and the heterogeneous reactions on the aerosol surface (Zhao et al., 2021). Under the

111   influence of sea and land breezes, land breezes at night blew the land pollutants to the

112   sea. In the next day, land breezes would evolve as sea breezes and return the pollutants

113   over the sea to the land, resulting in the increase of air pollutants over the land (Zhao et

114   al., 2021). High relative humidity caused by the sea breeze favored the participation of

115   gaseous precursors in heterogeneous reactions, hygroscopic growth of particles, and

116   secondary aerosol formation. Sun et al. (2020) found that nitrate was the main species

117   of aerosol in Shanghai during pollution events from 2017-2018, while most of these

118   events in winter were caused by the long-transport of air pollutants from the North

119   China Plain. Sea salt, as an important component of aerosol in coastal areas, has been

120   found in various studies that complex multiphase reactions can occur on the surface of

121   sea salts (Fu et al., 2010; Patel and Rastogi, 2020; Wang et al., 2022a). Wang et al.

122   (2022a) found that the addition of Na^+ in the ISORROPIA model improved the

123   simulation performance of aerosol and gaseous species at a coastal site in the South

124   China Sea, indicating that sea salts participated in the heterogeneous reactions with

125   other aerosol species. Liang et al. (2018) found that the rise of O_3 in Shanghai could be

126   affected by both local secondary formation and marine transport based on observation

127   and simulation. Meanwhile, Wang et al. (2022b) found that during the ozone pollution

128   in Shanghai in 2018, the presence of O_3 at high altitudes at night was transported

129   vertically downward during the daytime and high O_3 over the ocean was transported

130   horizontally to the land, thus jointly causing regional O_3 pollution in Shanghai.

131   Previous studies have shown that about 70% of Asian dust would pass through the

132   eastern coast of China and then moved out over Korean Peninsula and the Sea of Japan,

133   finally ending at the Pacific Ocean. The eastern coast of China is considered as the

134   essential route of the Asian dust transport to the Pacific Ocean (Arimoto et al., 1997;

135   Huang et al., 2010). In this study, an atypical dust event was observed in Shanghai, a

136   coastal mega-city in Eastern China. The unusualness of the meteorological conditions

137   and air pollutants during the dust was explicitly described. Three stages of the dust were

138   sorted and aerosol chemical compositions were compared. By focusing on the second

139   and third stages, the different formation mechanisms of nitrate and sulfate were

140   investigated. The amounts of major aerosol species from transport and secondary

141   formation were estimated based on a simple method of relating the upstream and

142   receptor simultaneous measurements.

143

144   **2. Methodology**

2.1. Observational sites

Measurements of various atmospheric parameters were conducted at Shanghai

Pudong Environmental Monitoring Station (31°13′ N, 121°32′E). All the

instruments were set up on the top floor of the building, about 25m above the

ground. As shown in Figure 1, the sampling site was located at the eastern tip of

Shanghai, close to the coastal line. The mean temperature and relative humidity

were 17.3°C and 72% in Shanghai during November, respectively. In autumn and

winter, air pollutants from upstream urban regions could be frequently transported

to Shanghai via the high-pressure system. Furthermore, air pollutants in Shanghai

tended to linger at the sea/land boundary regions due to the sea-land breeze (Shen

et al., 2019).

In addition to the measurements in Shanghai, data from environmental

monitoring stations in Qingdao and Lianyungang are also used in this study.
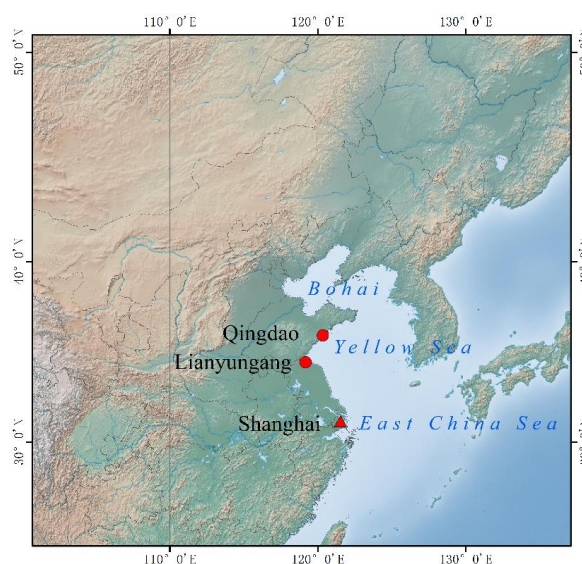
Figure 1. The observational sites in this study, including Shanghai, Qingdao, and

7

160   Lianyungang.

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162   **2.2. Instrumentation**

163   A set of online instruments was set up at the Pudong observational site. Inorganic

164   ions (NO_3^-, SO_4^{2-}, Cl^-, Na^+, NH_4^+, K^+, Mg^{2+}, Ca^{2+}) in $PM_{2.5}$ and soluble gases (NH_3,

165   HNO_3, HCl, HONO) were measured by an online ion chromatography (IC, MARGA-

166   1S, Metrohm). It operated at a flow rate of 16.7 L/min with a time resolution of one

167   hour. Briefly, air was drawn into a $PM_{2.5}$ cyclone inlet and passed through a wet

168   rotating denuder (gases) and a steam jet aerosol collector (aerosols). Subsequently, the

169   aqueous samples were analyzed with ion chromatography. More details can be found

170   in (Xu et al., 2020). Hourly trace metals (Si, Ca, Cu, Fe, K, Co, Mn, Cr, Zn, Pb, As,

171   Cd, V, Ni) in $PM_{2.5}$ were measured by using the Xact 625 multi-metals monitor

172   (Cooper Environmental, Beaverton, OR, USA). Particles were collected onto a Teflon

173   filter tape at a flow rate of 16.7 L/min, and then transported into the spectrometer

174   where the particles were analyzed with an X-ray fluorescence. Organic carbon and

175   elemental carbon were measured by an in situ Semi-Continuous Organic Carbon and

176   Elemental Carbon aerosol analyzer (RT-4, Sunset Laboratory, Beaverton, Oregon,

177   USA).

178   The concentrations of particles and gaseous pollutants were measured by a set of

179   Thermo Fisher Scientific instruments, including $PM_{2.5}$ (Thermo 5030i), PM_{10} (Thermo

180   5030i), SO_2 (Thermo Fisher 43i), NO_x (Thermo Fisher 42i), O_3 (Thermo Fisher 49i),

181   and CO (Thermo Fisher 48i-TLE). Meteorological parameters (ambient temperature,

182   relative humidity, wind speed, and wind direction) were obtained by a Vaisala Weather

183   transmitter (WXT520). Other supplementary parameters such as the height of planetary

184 boundary layer (PBL), vertical profiles of ozone and aerosol extinction were obtained

185 by a ceilometer (CL31, Vaisala), ozone lidar (LIDAR-G-2000, WUXIZHONGKE), and

186 aerosol lidar (AGJ, AIOFM), respectively.

187

188 **2.3. Thermodynamic simulation of aerosol pH and aerosol liquid water content**

189 The ISORROPIA II model is subject to the principle of minimizing the Gibbs energy

190 of the multi-phase aerosol system, leading to a computationally intensive optimization

191 problem (Song et al., 2018). ISORROPIA II calculates the aerosol pH, ALWC (aerosol

192 liquid water content) and compositions of ammonia-sulfate-nitrate-chloride-sodium-

193 calcium-potassium-magnesium in the thermodynamic equilibrium with gas-phase

194 precursors. The performances and advantages of ISORROPIA over the usage of other

195 thermodynamic equilibrium codes has been assessed in numerous studies (Nenes et al.,

196 1998; West et al., 1999; Ansari and Pandis, 1999; Yu et al., 2005). The ISORROPIA

197 running in the forward mode at the metastable state was applied in this study.

198

199 **2.4. Hybrid Single-Particle Lagrangian Integrated Trajectory Model**

200 The HYSPLIT (Hybrid Single-Particle Lagrangian Integrated Trajectory) was

201 used to compute the backward trajectories of the air parcels during the dust events. In

202 this study, the HYSPLIT model was driven by meteorological data outputs from the

203 Global Data Assimilation System (GDAS) (Su et al., 2015), which is available at

204 ftp://arlftp.arlhq.noaa.gov/pub/archives/gdas1. Air mass trajectories were launched at

205 different heights from the ground and a total duration of 48 hours simulation was

206 conducted.

207

3. Results and Discussion

3.1. Characteristics of an unusual dust event

Figure 2 shows the time series of PM_{10}, $PM_{2.5}$, and meteorological parameters as well as the vertical profiles of aerosol extinction coefficient and depolarization ratio observed at the Shanghai sampling site from October 25 to November 6, 2019. During October 25 to 28, the mean wind speed was relatively low of 0.9±0.72m/s with the peak value of 3.1m/s and winds prevailed from the northwest. The mean concentration of $PM_{2.5}$ and PM_{10} was 34.7 and 44.2 $\mu g/m^3$, respectively. Starting at 4:00 LST on October 29, the concentration of PM_{10} increased sharply and lasted till November 2 (Figure 2d). Based on the aerosol lidar observation, both aerosol extinction coefficient and depolarization ratio extended from the ground to around 2km during the same period. Specifically, the depolarization ratio was obviously enhanced (>0.1), indicating that Shanghai encountered a long-lasting dust event. During the whole dust period, the mean concentrations of $PM_{2.5}$ and PM_{10} reached $53.3 \pm 20.5\mu g/m^3$ and $172.4 \pm 70.2\mu g/m^3$, respectively, yielding a low $PM_{2.5}/PM_{10}$ ratio of 0.34 ± 0.15. As a comparison, $PM_{2.5}$ and PM_{10} during the non-dust period was $38.9\mu g/m^3$ and $49.8\mu g/m^3$, respectively, with a relatively high $PM_{2.5}/PM_{10}$ ratio of 0.62 ± 0.20.
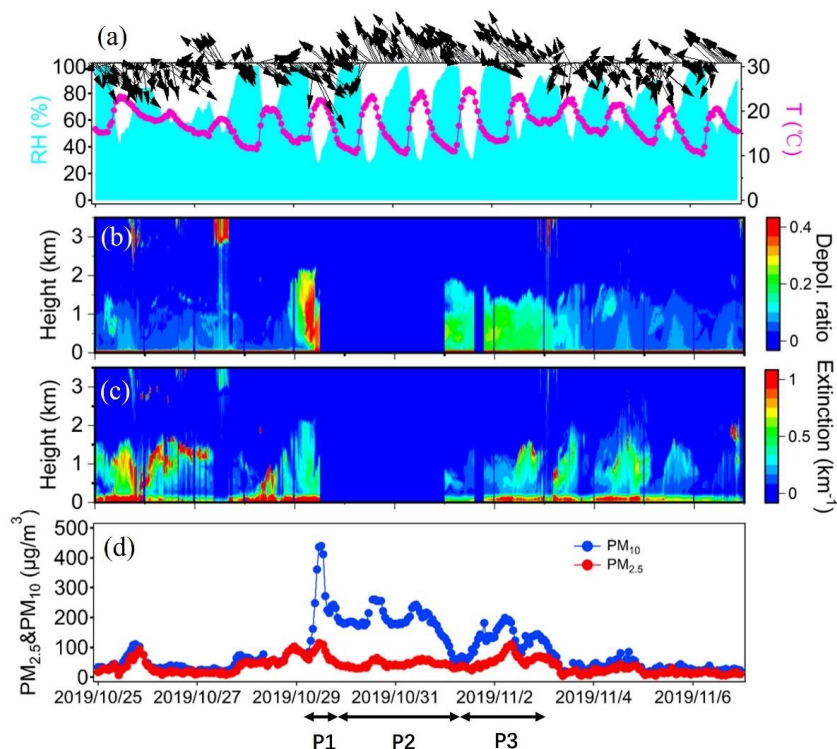
Figure 2. Time series of (a) relative humidity, temperature, wind vectors, (b) aerosol depolarization ratio, (c) aerosol extinction coefficient, (d) mass concentrations of $PM_{2.5}$ and PM_{10} during the study period. Three dust stages, i.e., P1, P2, and P3 are also marked. The missing aerosol lidar data were due to instrument malfunction.

The occurrences of dust were usually accompanied by low relative humidity and strong winds due to the passage of cold fronts (Fu et al., 2010; Huang et al., 2010; Wang et al., 2013; Wang et al., 2018b). In this study, relative humidity was exceptionally high with the mean value of 71±26%. It showed strong diurnal variation with minimum in the daytime and even close to 100% in the nighttime (Figure 2a). Also, wind speed was low of 0.54±0.59m/s with a maximum of 2.6m/s. Due to this stagnant synoptic condition, the mean concentrations of main gaseous pollutants such as O_3, SO_2, and

238   NO_2 reached $86.0 \pm 47.8 \mu g/m^3$, $11.8 \pm 3.4 \mu g/m^3$, and $63.3 \pm 27.9 \mu g/m^3$, respectively, even

239   higher than those during the non-dust period.

240   We further divided the dust event into three stages based on the temporal

241   characteristics of PM_{10} and the air masses transport patterns. As shown in Figure 2d,

242   PM_{10} quickly climbed from 4:00 on October 29 and reached a maximum of $436 \mu g/m^3$

243   after 8 hours. The air masses mainly originated from the semi-arid regions of northwest

244   China (Figure 3d), and this was consistent with both near surface wind observation

245   (Figure 2a) and wind lidar observation (Figure 3a). The wind profiles showed dominant

246   northwest winds from the ground to the altitudes of around 2km before the noon of

247   October 29, indicating the presence of a strong synoptic system. Afterwards, PM_{10}

248   quickly decreased to $199 \ \mu g/m^3$ at 20:00, October 29 within 8 hours. This was mainly

249   due to the shift of wind directions. As shown in Figure 3a, although the winds at

250   altitudes of higher than 700m kept blowing from the northwest, the near surface winds

251   had turned from the southeast. As Shanghai is a coastal city and adjacent to the East

252   China Sea, the relatively clean southeasterlies diluted the local air pollutants and thus

253   explained the quick decrease of PM_{10} concentrations. This short dust episode from 4:00

254   - 13:00, October 29 was defined as Stage P1.

255   However, the prevailing southeast winds didn't fully terminate the dust event.

256   Even under the persistent southeasterlies, hourly PM_{10} concentrations stayed above 150

257    $\mu g/m^3$ until November 1 and then decreased to $65 \ \mu g/m^3$ at 03:00, November 1 (Figure

258   2d). Compared to P1, wind speed during this stage was as low as $0.4 \pm 0.5 m/s$ while

259   RH was moderately high of $70 \pm 26\%$. Although the daytime RH stayed low between

260   30% and 50%, it frequently reached over 90% at nighttime. Figure 3e shows that

261   although the air masses originated from the Gobi Desert, they also passed over

12

262   considerable coastal regions. The wind profiles also showed that although northwest

263   winds prevailed at altitudes higher than 500m, the east and northeast winds were

264   dominant below 500m (Figure 3b). This explained the relatively high relative humidity

265   during this period due to the mixing between dust plumes and coastal sea breezes. This

266   dust episode from 14:00, October 29 to 3:00, November 1 was defined as Stage P2.

267   After P2, PM_{10} and $PM_{2.5}$ rose again and peaked at 5:00 and 9:00, November 2

268   with the hourly concentration of 199 and 117μg/m^3, respectively. Different from P1 and

269   P2, the air masses originated from the Shandong Peninsula and the northern region of

270   Jiangsu province, and then migrated over the Yellow Sea and the East China Sea (Figure

271   3f). Usually, the dust plumes would transport further eastwards and impact the western

272   Pacific region and even faraway oceanic regions (Wang et al., 2018b; Nagashima et al.,

273   2016). However, the air masses evidently deflected and pushed the dust back to the

274   mainland. The wind profiles on November 2 also showed winds at the detected range

275   of altitudes all came from the eastern and southeastern oceanic regions (Figure 3c). This

276   probably indicated the mixing between dust plumes and humid oceanic air masses was

277   quite sufficient, which was also reflected by the highest average RH of $76 \pm 24\%$ among

278   the three stages of the dust event. In addition, the concentrations of O_3 and NO_2 at this

279   stage were higher than those of P1 and P2, which could be conducive to the formation

280   of secondary aerosol components and will be discussed later. This rarely observed dust

281   backflow transport episode from 4:00, November 1 to 23:00, November 2 was defined

282   as Stage P3.
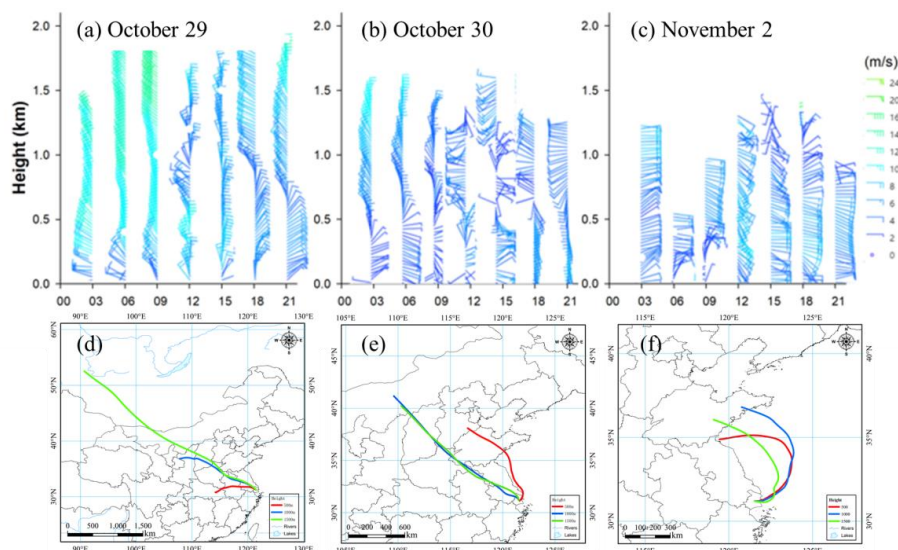
Figure 3. Wind profiles observed by a wind profiler radar on (a) October 29, (b) October 30, and (c) November 2. 48-hour backward trajectories simulated at the sampling site starting from (d) 4:00 AM, October 29, (e) 9:00 AM, October 30, and (f) 13:00 PM, November 2. The red, blue, and green trajectories represented starting altitudes of 100, 500, and 1500m, respectively.

3.2. Comparisons of aerosol chemical compositions among the three dust stages

Figure 4a shows the time-series of hourly aerosol chemical components, including SNA (NO_3^-, SO_4^{2-}, and NH_4^+), OM (organic matters = 2*OC), EC, and mineral aerosols ([Minerals]= (2.2*Al+2.49*Si+1.63*Ca+2.42*Fe+1.94*Ti), (Malm et al., 1994)) in $PM_{2.5}$. During P1, the mean concentration of SNA was 49.9 ± 31.6 μg/m^3. The mineral aerosols reached 16.4 ± 14.6 μg/m^3, accounting for 19% in $PM_{2.5}$. The contribution of OM to $PM_{2.5}$ was almost identical to that of mineral aerosols (Figure 4b).
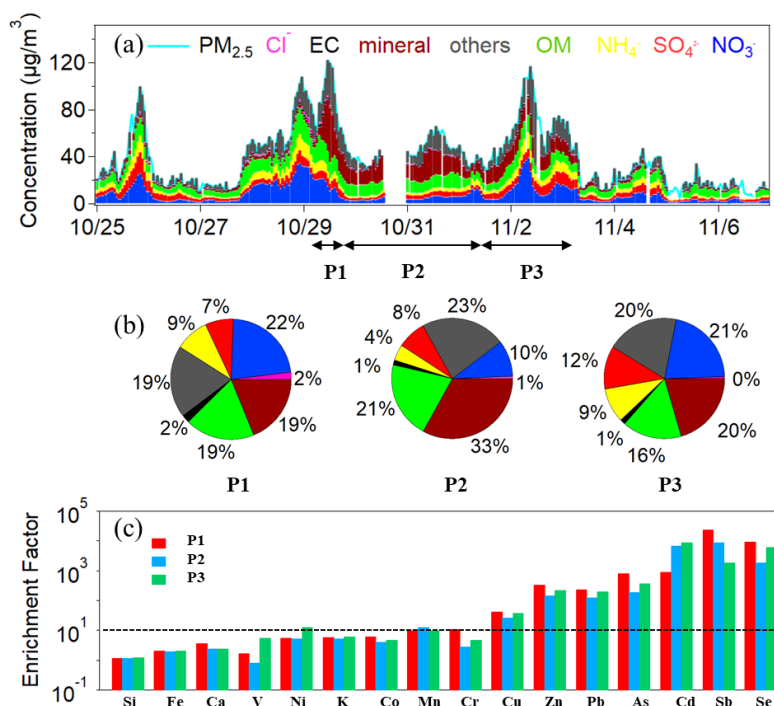
Figure 4. (a) Times-series of major chemical components in $PM_{2.5}$ during the study
period. (b) The mean proportion of major chemical components in $PM_{2.5}$ during the
three dust stages. (c) Enrichment factors of elements in $PM_{2.5}$ during the three dust
stages.

During P2, mineral aerosols increased to 23.4±54.1μg/m^3 and accounted for 33%
in $PM_{2.5}$, the highest among all three stages (Figure 4b). Due to the continuous dilution
effect of dust on local anthropogenic pollutants, the concentrations and proportions of
SNA in $PM_{2.5}$ were the lowest during this stage. For instance, NO_3^- only accounted for
10% in $PM_{2.5}$, indicating the formation of nitrate was suppressed to a certain extent.
The level of OM didn't show obvious changes and averaged 10.1±2.1μg/m^3, accounting
for 21% in $PM_{2.5}$.

During P3, mineral aerosols averaged 11.9±2.7μg/m^3, ranking the lowest among

15

312   all three stages. The proportion of mineral aerosols in $PM_{2.5}$ decreased to 20%,

313   suggesting the dust backflow from the ocean was less enriched in mineral components.

314   Compared to P2, SNA showed significant increases and much stronger diurnal

315   variations during P3. SO_4^{2-}, NO_3^-, and NH_4^+ averaged 6.7 ± 2.4, 12.4 ± 8.9, and 5.4 ±

316   2.7μg/m^3, respectively. As shown in Figure 4b, the contribution of nitrate to $PM_{2.5}$

317   increased to 21% while that of sulfate rose to 12%, the highest among all three stages.

318   The concentration (9.3±3.2μg/m^3) and proportion (16%) of OM during P3 were lower

319   than the other two stages, which was probably due to the unusual dust backflow

320   transport pathway.

321   Enrichment factors (EFs) of the measured elements in $PM_{2.5}$ were calculated by

322   using Al as a reference element, i.e., $EF_x=(X/Al)_{aerosol}/(X/Al)_{crust}$, where X was the

323   element of interest. As shown in Figure 4c, elements such as Si, Fe, and Ca were less

324   enriched as they mainly derived from the crust. While for anthropogenic elements

325   including Cu, Zn, Pb, As, Cd, Sb, and Se, they were enriched by different extents with

326   EFs between 10 and 10,000. In addition, these elements above were more enriched

327   during P1 than P2 and P3. The dust transport pathway via inland areas during P1 should

328   be the main cause as anthropogenic sources such as metallurgical industries, coal-fired

329   plants, and smelters were widely located in inland regions. Exceptions were found for

330   Ni and V, which were often used as tracers for heavy oil combustion. EFs of Ni and V

331   exhibited higher values during P3 than P1 and P2. Since the dust backflow transport

332   pathway during P3 had almost two-days travelling durations over the East China Sea,

333   which was on the one of the busiest international shipping trade routes (Fan et al., 2016),

334   the enrichments of Ni and Vi were probably ascribed to the mixing between dust and

335   marine shipping emissions.

336

3.3. Effect of unusually high O_3 on the formation of secondary aerosols during P2

338   Figure 5 shows the hourly near surface ozone concentrations and vertical profiles

339 of ozone during the study period. Interestingly, a few high O_3 peaks occurred during

340 the dust event (Figure 5a). O_3 averaged $92.8 \pm 52.8\mu g/m^3$ during the dust, about 50%

341 higher than the non-dust days. Among the three dust stages, O_3 substantially increased

342 from $35.9 \pm 36.4\mu g/m^3$ during P1 to $80.7 \pm 41.2\mu g/m^3$ during P2 and $104.0 \pm$

343 $48.7\mu g/m^3$ during P3. The low O_3 during P1 was due to the cleansing effect of strong

344 dust associated with the strong cold front, which was as similar as previous studies

345 that low oxidants concentrations were usually observed during strong dust events

346 (Benas et al., 2013). As for the relatively high O_3 during P2 and P3, several causes

347 may be responsible. Firstly, the mean wind speed was low of 0.4 and 0.6 m/s during

348 P2 and P3, respectively. Thus, this weak synoptic system exerted weak dilution effect

349 on the local air pollutants. Secondly, since the dust plume travelled mostly over the

350 coastal and oceanic areas, part of O_3 could be transported from the high ozone oceanic

351 areas (Wang et al., 2022b). Thirdly, the ozone lidar also observed high O_3 stripes

352 during P2 and P3. As shown in Figure 5b, the high O_3 profiles extended from the

353 surface to around 1km and the profile structure was similar to that of aerosol

354 depolarization ratio. The subsidence of dust particles probably down drafted high-

355 altitude O_3 and also contributed to the high O_3 near the ground.

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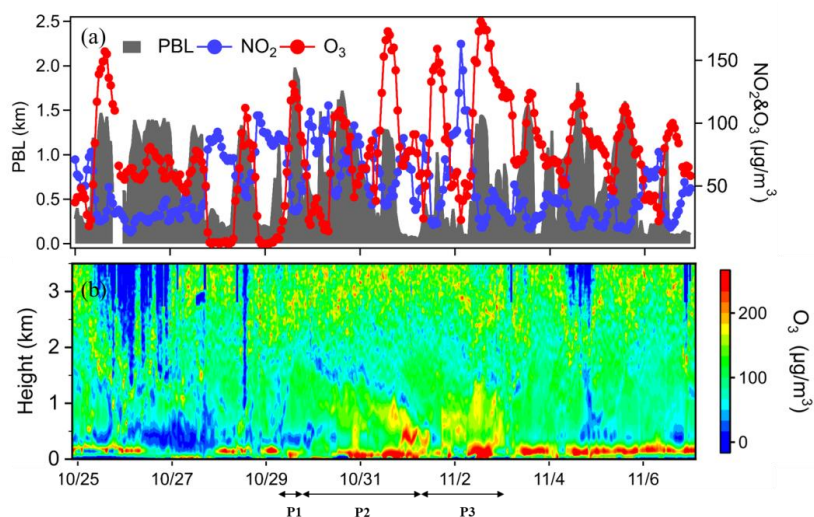
Figure 5. (a) Time-series of near surface O_3, NO_2 and planetary boundary layer height

(b) Vertical profiles of ozone observed by the ozone Lidar.

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360   To investigate the effect of the relatively high O_3 on the formation of secondary

361 aerosols, Figure 6 displays the relationship between sulfate/nitrate and ammonium. The

362 strong correlations of sulfate-ammonium ($R^2 = 0.91$) and nitrate-ammonium ($R^2 = 0.93$)

363 were expected. The scatters were further color coded by O_3 and it could be seen that

364 higher sulfate, nitrate, and ammonium tended to be formed under higher O_3

365 concentrations, suggesting the potential role of photochemistry in promoting the

366 formation of secondary aerosols.
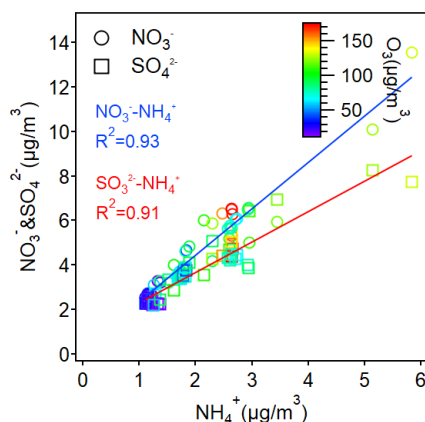
Figure 6. Linear relationship between sulfate/nitrate and ammonium with scatters colored by the concentrations of O_3 during P2

Figure 7 displays the linear relationship between sulfate and PM_{10}, ALWC, O_3, and HONO, respectively. Obviously, sulfate had no correlation with PM_{10} (Figure 7a), indicating that dust contributed negligibly to sulfate. Also, sulfate had almost no or even a weakly negative correlation with ALWC (Figure 7b), suggesting aqueous-phase processing was not a major reaction pathway of secondary aerosols during P2. In contrast, we found that sulfate correlated positively with O_3 (R^2=0.38, Figure 7c) and negatively correlated with HONO (R^2=0.25, Figure 7d). It was well recognized that HONO was an important precursor to OH radical and it was estimated that HONO accounted for 30%~60% of the OH budget in Shanghai (Bernard et al., 2016). The moderate correlations between sulfate and O_3/HONO corroborated the discussion above that the formation of secondary aerosols during P2 should be mainly promoted via the gas-phase oxidations.
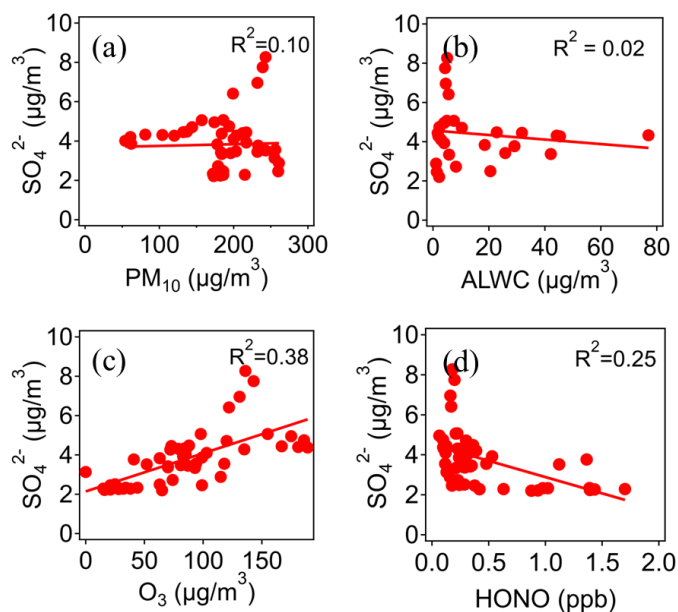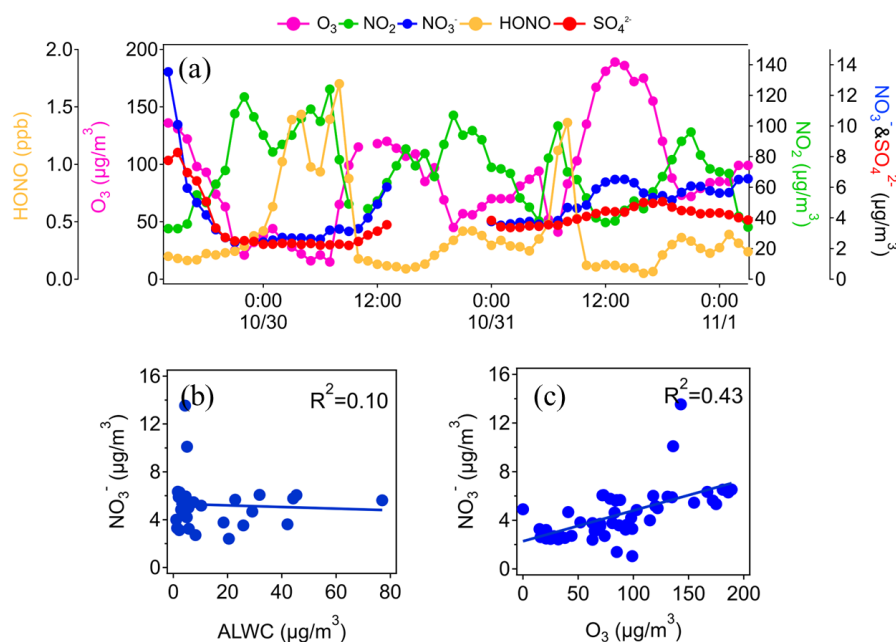
Figure 7. Linear relationship between sulfate and (a) PM_{10}, (b) ALWC, (c) O_3, and (d) HONO during P2

As similar as sulfate, nitrate didn't show correlation with ALWC (Figure 8b) but moderately correlated with O_3 (R^2=0.43, Figure 8c). Figure 8a displays the temporal variation of NO_2, O_3, HONO, and NO_3^- during P2. It can be found that although NO_2 was high at night as well as for HONO, the concentrations of NO_3^- were relatively low, indicating a low formation potential of secondary aerosols. Figure 2a shows that RH during P2 was relatively low at nighttime with a minimum of 31%, suggesting the aqueous reactions could be suppressed to some extents. During the daytime, O_3 reached its maximum while HONO reached its minimum, both contributing to the high levels of oxidants. The co-variation of nitrate and sulfate with O_3 (Figure 8a) confirmed that although dust often suppressed the formation of atmospheric oxidants, the photochemistry reactions under unusually high O_3 concentrations dominated the

398   formation of sulfate and nitrate during P2 in this study.

400   Figure 8. (a) Temporal variations of O_3, NO_2, HONO, nitrate, and sulfate during P2.
401   Linear relationship between nitrate and (b) ALWC and (c) O_3.

402

403   **3.4. Aerosol chemistry under dust backflows during P3**

404   **3.4.1. Evidence of dust backflows**

405   The dust during P3 was diagnosed as a backflow transport pathway from

406   mainland to Shanghai through the Yellow Sea and the East China Sea according to the

407   backward trajectory analysis (Figure 3f). This atypical dust transport pathway observed

408   in this study was defined as the dust backflow. In this section, we have provided more

409   evidences of dust backflow from various aspects.

410   Figure 3f shows the dust drifted away from the Shandong Peninsula, thus we

411   selected two coastal sites in Shandong province for supplementary analysis. Figure 9

412   compares the time-series of hourly air pollutants at Qingdao, Lianyungang, and

21

413   Shanghai. At Qingdao and Lianyungang, high PM_{10} concentrations were observed

414   during October 30 – 31, indicating the invasion of dust. After about two days, PM_{10}

415   showed a peak at early November 2 at Shanghai. This was consistent with simulation

416   duration of the backward trajectories, i.e., around 48 hours (Figure 3f). In Figure 9, it

417   could be seen that in the dust upstream regions (i.e., Qingdao and Lianyungang), PM_{10}

418   varied negatively with NO_2 and CO (the highlighted period in the figure). While in

419   Shanghai, positive correlations between PM_{10} and NO_2 (R^2=0.32) as well as between

420   PM_{10} and CO (R^2=0.55) indicated that the dust during P3 had acted as a carrier of the

421   gaseous pollutants but not a diluter.
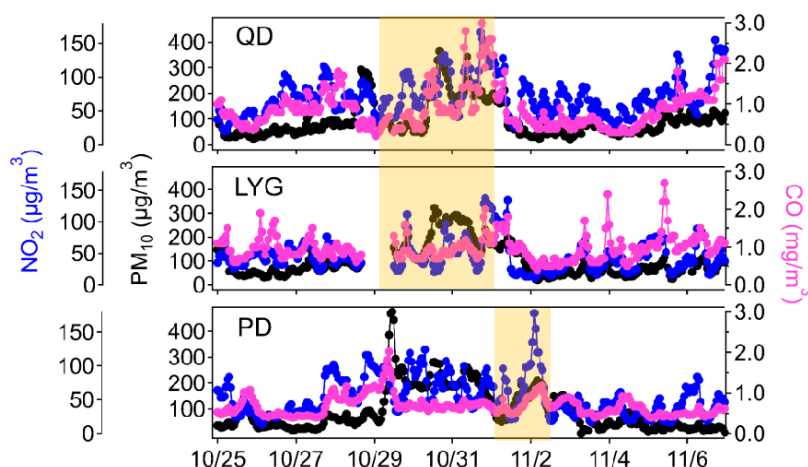
422

423   Figure 9. Time-series of PM_{10}, NO_2, and CO at Qingdao, Lianyungang, and Pudong.
424   The dust periods at these three sites are highlighted.

425

426   More evidence of dust backflows was provided from the perspective of aerosol

427   chemical tracers. As discussed in Section 3.2, enrichment factors of V and Ni were the

428   highest during P3. Figure 10a further displays the time-series of V and Ni and they

429   varied significantly during the study period. The mass concentrations of V and Ni

430   increased 4 and 1.8 times during P3 compared to P2, respectively. This indicated that

431   the dust had mixed with pollutants from marine vessel emissions and transported back

432   to Shanghai.
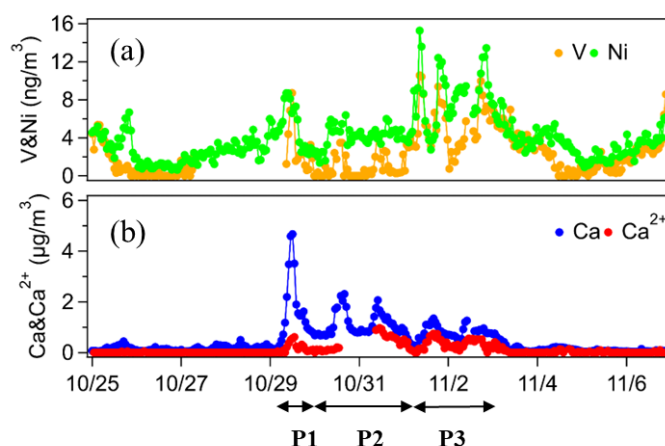
433

434   Figure 10. Time-series of (a) V, Ni, (b) Ca, and Ca^{2+} during the study period

435

436   Figure 10b plots the time-series of Ca and Ca^{2+}, which represented the total

437   calcium and the soluble part of calcium, respectively. It was observed that Ca and Ca^{2+}

438   didn't exhibit quite proportional variation trend, which should be related to the

439   solubility of calcium during different dust stages. During P1, the mean concentration of

440   Ca reached the highest of $1.63 \pm 1.53\mu g/m^3$ while Ca^{2+} was the lowest of $0.21 \pm$

441    $0.20\mu g/m^3$, thus resulting in the lowest Ca^{2+}/Ca ratio of 0.10 ± 0.08. As discussed in

442   Section 3.1, dust during P1 was the strongest and thus it contained higher fractions of

443   minerals, which were mainly in the form of insoluble metal oxides. The average

444   concentrations of Ca^{2+} and Ca during P2 were $0.33 \pm 0.28\mu g/m^3$ and $1.11 \pm 0.46\mu g/m^3$

445   with the higher Ca^{2+}/Ca ratio of 0.27 ± 0.20. As a comparison, the average

23

concentrations of Ca^{2+} and Ca during P3 reached $0.34 \pm 0.20 \mu g/m^3$ and $0.78 \pm$

$0.27 \mu g/m^3$, yielding the highest Ca^{2+}/Ca ratio of 0.38 ± 0.19. The much higher solubility

of calcium during P3 should be directly related to the lingerer of dust plumes over the

open ocean. The abundant water vapor over the ocean could accelerate the dissolution

of the insoluble components in particles during the mixing between continental dust

and oceanic air masses.

### 3.4.2. Formation processes of secondary aerosols during P3
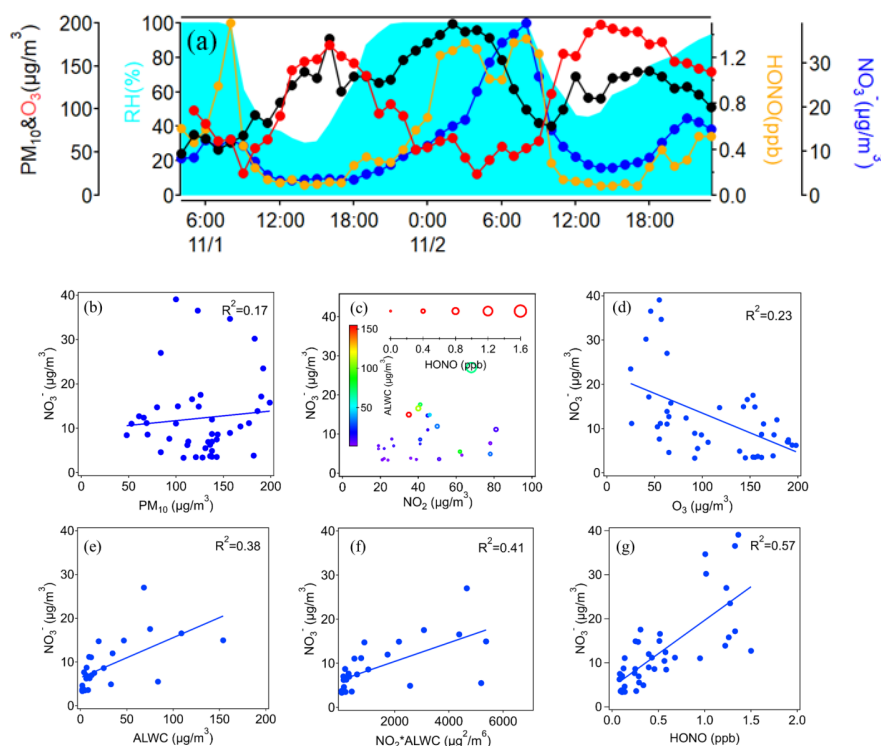
Figure 11. (a) Time series of PM_{10}, NO_3^-, O_3, HONO, and RH during P3. Linear
relationship between nitrate and (b) PM_{10}, (c) NO_2, (d) O_3, (e) ALWC, (f)
NO_2*AWLC, and (g) HONO.

460   In this section, the main processes that dominated the formation of secondary

461   aerosols during P3 were probed and compared to P2. Figure 11a shows the time-series

462   of hourly PM_{10}, NO_3^-, O_3, HONO, and RH during P3. It could be visualized that most

463   parameters co-varied relatively consistently to some extent. Different from P2, nitrate

464   showed weak correlation with PM_{10} (Figure b), indicating that part of nitrate could be

465   directly transported via the dust transport. Also different from P2, nitrate even showed

466   negative correlation with O_3 (Figure d). From Figure 11a, nitrate concentrations were

467   at its troughs during daytime when O_3 peaked. Thus, photochemical reactions didn't

468   play an important role in the formation of nitrate during this stage.

469   Figure c investigates the conjoint impact of multiple parameters on the formation

470   of nitrate. In general, NO_3^- was more favored under higher NO_2, which was obviously

471   expected as NO_3^- could be either formed by the photochemical oxidation of NO_2 by OH

472   radicals at daytime (Hertel et al., 2012) or produced by hydrolysis of N_2O_5 from the

473   oxidation of NO_2 by O_3 at nighttime (Ge et al., 2017). In addition, the formation of

474   higher NO_3^- was accompanied with higher ALWC and HONO, implying the role of

475   aqueous phase reactions rather than the photochemical reactions. Figure e-11g

476   separately investigate the relationship between NO_3^- and various parameters. NO_3^-

477   moderately correlated with ALWC ($R^2 = 0.38$). By relating NO_3^- and the multiplication

478   of ALWC and NO_2, the correlation coefficient ($R^2 = 0.41$) was further improved (Figure

479   f), indicating the reaction pathway of NO_2 to nitrate in the aqueous phase. Figure 11g

480   also observed strong correlation between NO_3^- and HONO ($R^2 = 0.57$). Alicke et al.

481   (2002) proposed that the heterogeneous reactions of NO_2 on the surface of moist

482   particles produced both nitrate and HONO, i.e.,

483    $2NO_2 + H_2O \rightarrow HONO + HNO_3$
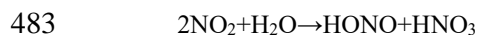
484   Compared to the mean ALWC ($11.8 \pm 17.1\mu g/m^3$) during P2, ALWC during P3

485   was much higher of $29.1 \pm 38.0\mu g/m^3$. This was mainly ascribed to the higher

486   atmospheric water vapor during P3, which was evidently caused by the backflows of

487   oceanic air masses. The different levels of ALWC between P2 and P3 caused divergent

488   role of aqueous processing in the secondary aerosol formation.

489   As for sulfate, its temporal variation during P3 was quite different from NO_3^- that

490   it showed no diurnal pattern while presented a gradually increasing trend (Figure a).

491   This suggested sulfate had gone through a different formation mechanism from nitrate.

492   SO_4^{2-} correlated strongly with Na^+ (Figure c), which can be regarded as the tracer of

493   sea salts. The co-variation of SO_4^{2-} and Na^+ probably suggested that a portion of sulfate

494   was aged and directly transported by the oceanic air masses. Additionally, SO_4^{2-} had

495   weakly positive correlation with O_3 (Figure d) and ALWC (Figure e), indicating that

496   both gas and aqueous phase processes contributed to the secondary formation of sulfate

497   to some extent. Figure 12a also adds the time-series of Ca^{2+}/Ca, which didn't co-vary

498   with either sulfate or nitrate. This probably indicated that the chemical reactions

499   between acidic gaseous precursors and dust mineral components were negligible.
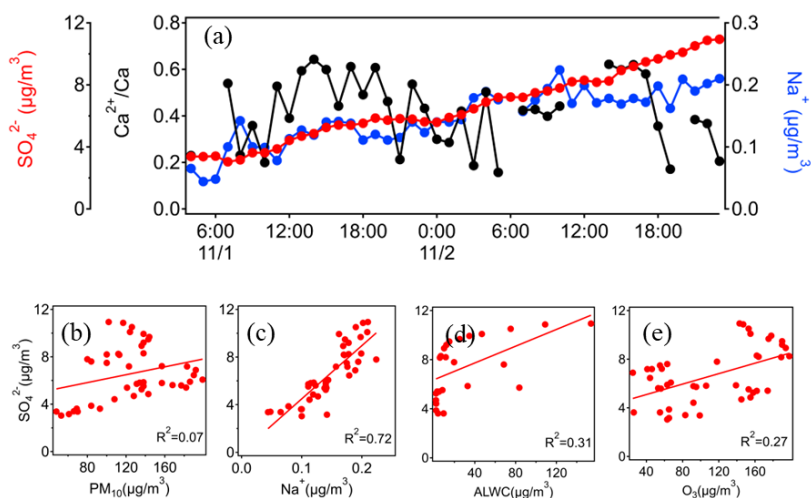
Figure 12. (a) Time-series of sulfate, Na^+, and Ca^{2+}/Ca during P3. Linear between sulfate and (b) PM_{10}, (c) Na^+, (d) O_3, and (e) ALWC.

To assess whether dust or sea salts participated in the heterogeneous reactions of secondary aerosol during P3, the ISORROPIA II model was run with different scenarios. Figure S1 shows the model performance for SO_4^{2-}, NO_3^-, NH_4^+, and NH_3 based on the $SO_4^{2-}-NO_3^--NH_4^+-Cl^--NH_3-HCl-HNO_3$ system. After adding Ca^{2+} into this thermodynamic equilibrium system, the correlations between the simulations vs observations for all four species were lowered with different extents (Figure S2). This indicated that Ca^{2+} was not internally mixed with sulfate and nitrate and probably suggested that the heterogeneous reactions on dust were very limited.

Then Na^+ was added into the thermodynamic equilibrium system (Figure S3). It could be seen the model performance was slightly improved. The correlation coefficients of the four species were closer to the unity and the regression slopes were also more parallel to the y=x line. This suggested that sea salts were involved in the formation of secondary inorganic aerosols during the dust backflow. This also explained

27

517 the strong correlation between SO_4^{2-} and Na^+ during P3.

518

519 **3.4.3. Estimation of transported and secondarily formed particles during P3**

520 As discussed in the previous sections, the sources of aerosols during P3 could be

521 derived from both aged aerosols via the dust backflows and secondary formation. In

522 this section, we aimed to estimate the contribution of transport and secondary formation

523 to the main aerosol species, respectively, based on the simultaneous measurements at

524 the Pudong site and the Lianyungang site. As discussed in Section 3.4.1, Lianyungang

525 acted as an upstream region of dust transport drifting from the mainland. The dust

526 duration observed at Lianyungang was from about 5:00, October 30 to 16:00, October

527 31, about 46 hours ahead of the dust invasion observed at Pudong (Figure 9).

528 To assess the extents of transported pollutants, black carbon (BC) was used as a

529 reference aerosol component. As shown in Figure S4, one BC pollution episode on

530 October 30 at Lianyungang was observed. Consistently, another BC pollution episode

531 on November 2 at Pudong emerged after about 46 hours. Since the air mass trajectory

532 from Lianyungang to Pudong was mostly over the ocean and BC had no secondary

533 sources, it could be reasonably assumed that the difference of BC between these two

534 sites was ascribed to the removal processes of particles.

535 We further defined the average concentrations of various aerosol components

536 during the previous five hours of the dust at Pudong as their background concentrations.

537 Then, a coefficient k was derived to calculate the removal fractions of aerosols during

538 the dust transport as below.

539

540 $$k = \frac{AV_{LYG,BC} - (AV_{PD,BC} - BKG_{PD,BC})}{AV_{LYG,BC}}$$

541

28

542 $AV_{LYG,BC}$ and $AV_{PD,BC}$ represent the average concentration of BC at Lianyungang

543 and Pudong during their respective dust period. $BKG_{PD,BC}$ represents the background

544 concentration of BC at Pudong. By assuming that other aerosol species were removed

545 at a similar efficiency as BC, the amounts of transported aerosol species from

546 Lianyungang to Pudong can be estimated as below.

547 $TP_{PD,i} = AV_{LYG,i} \times (1 - k)$

548 $TP_{PD,i}$ represents the transported amounts for aerosol species i. Then, the

549 secondarily formed aerosol species i at Pudong can be calculated as below.

550 $SF_{PD,i} = AV_{PD,i} - BKG_{PD,i} - TP_{PD,i}$

551 Figure 13 shows the results of the transported and the secondarily formed aerosol

552 species during P3. It was calculated that the secondarily formed and transported NO_3^-

553 averaged 6.8μg/m^3 and 4.7μg/m^3, accounting for about 45% and 31% of its total mass

554 concentration, respectively. This was consistent with the analysis above that a

555 considerable portion of nitrate was formed through the aqueous phase secondary

556 formation. In contrast, it was calculated that the transported SO_4^{2-} accounted for about

557 42% of its total mass concentration while the secondarily formed SO_4^{2-} was almost

558 negligible. This was also consistent with the phenomenon that SO_4^{2-} correlated

559 significantly with Na^+ (Figure 12c), suggesting most sulfate was aged and transported.

560 As for NH_4^+, its origin apportionment was similar to NO_3^- that the secondarily formed

561 and transported NH_4^+ accounted for about 35% and 28% of its total mass concentration,

562 respectively. Compared to NO_3^- and NH_4^+, OM was more dominated by transport (57%)

563 while its secondary formation only accounted for about 13%.

564

565
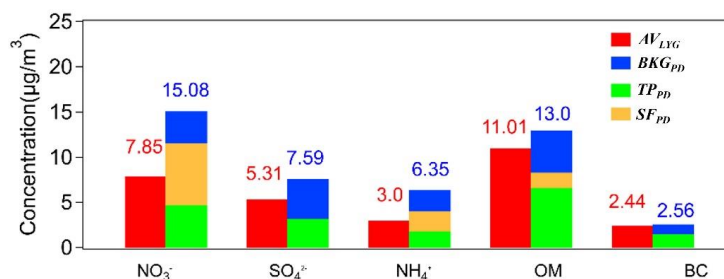
Figure 13. The apportioned concentrations of the major aerosol species during

P3.

4. Conclusion

During October 29 to November 2, 2019, a long-lasting dust event was observed in Shanghai based on a synergy measurement of near surface air pollutants, aerosol lidar, wind profiling lidar, and air masses trajectory modeling. During the whole dust period, the mean concentrations of $PM_{2.5}$ and PM_{10} reached $53.3 \pm 20.5\mu g/m^3$ and $172.4 \pm 70.2\mu g/m^3$. Different from most dust events, this dust event was characterized of exceptionally high relative humidity ($71 \pm 26\%$) and low wind speed ($0.54 \pm 0.59m/s$). Due to this stagnant synoptic condition, the mean concentrations of main gaseous pollutants such as O_3, SO_2, and NO_2 reached $86.0 \pm 47.8\mu g/m^3$, $11.8 \pm 3.4\mu g/m^3$, and $63.3 \pm 27.9\mu g/m^3$, respectively, even higher than those during the non-dust period.

The dust event was divided into three stages from P1 – P3. P1 was a short dust episode from 4:00 - 13:00, October 29, when wind profiles showed dominant northwest winds from the ground to the altitudes of around 2km, indicating the presence of a strong synoptic system. P2 was a dust episode from 14:00, October 29 to 3:00, November 1, when RH was moderately high of $70 \pm 26\%$ and the southeasterlies

585   prevailed with partial air masses from coastal regions. P3 was a rarely observed dust

586   backflow transport episode from 4:00, November 1 to 23:00, November 2. The air

587   masses originated from the Shandong Peninsula and the northern region of Jiangsu

588   province, and then migrated over the Yellow Sea and the East China Sea. RH reached

589   the highest of $76 \pm 24\%$ among the three stages of the dust event.

590   During P2, mineral aerosols accounted for 33% in $PM_{2.5}$, the highest among all

591   three stages. Abnormally high O_3 concentrations were observed, much higher than the

592   non-dust days. This was partially due to the weak synoptic system that exerted weak

593   dilution effect on the local air pollutants. Also, part of O_3 could be transported from the

594   high ozone oceanic areas. The ozone lidar observed that the subsidence of dust particles

595   probably down drafted high-altitude O_3 and also contributed to the high O_3 near the

596   ground. As a result, sulfate and nitrate moderately correlated with O_3 while had almost

597   no correlation with ALWC, indicating that the formation of secondary aerosols during

598   P2 should be mainly promoted via the gas-phase oxidations.

599   During P3, a special phenomenon of dust backflow was observed and confirmed

600   by various evidences. Two upstream sites (Qingdao and Lianyungang) showed dust

601   occurrences about 48 hours ahead that of Shanghai, consistent with the transport

602   duration of the dust backflow from the Shandong Peninsula to Shanghai over the Yellow

603   Sea and the East China Sea. In addition, the mass concentrations of V and Ni

604   significantly increased, indicating the mixing between dust and marine vessel emissions.

605   The highest Ca^{2+}/Ca ratio of 0.38 ± 0.19 was observed during P3, which should be due

606   to that the lingerer of dust plumes over the open ocean facilitated efficient solubility of

607   calcium. Different from P2, nitrate didn't correlate with O_3, while it was favored under

608   high NO_2 and ALWC and strongly correlated with HONO, indicating the reaction

609 pathway of NO_2 to nitrate in the aqueous phase. As for sulfate, the strong correlation

610 between SO_4^{2-} and Na^+ suggested that a portion of sulfate was aged and directly

611 transported by the oceanic air masses. The ISORROPIA II modeling added Na^+ into the

612 $SO_4^{2-}-NO_3^--NH_4^+-Cl^--NH_3-HCl-HNO_3$ system and found the models performances

613 of simulating the major aerosol and gaseous species could be improved. As a

614 comparison, the models performances became worse after adding Ca^{2+}. This suggested

615 that sea salts participated in the secondary aerosol formation while dust heterogeneous

616 reactions were limited during P3. Based on a simple method, the amounts of transported

617 and secondarily formed particles during P3 were quantified. It was calculated that about

618 45% and 31% of NO_3^- was contributed by secondary formation and transport,

619 respectively. In contrast, the transported SO_4^{2-} accounted for about 42% of its total mass

620 concentration while the rest was from its background concentration with negligible

621 secondary formation. OM was dominated by transport (57%) while its secondary

622 formation only accounted for about 13%.

623

624 **Data Availability Statement**

625 All data used in this study can be requested upon the corresponding author

626 (huangkan@fudan.edu.cn).

627

628 **Author contributions**

629 KH, QF, and YD designed this study. JH, FY, YL, and JC performed data

630 collection. DL and KH performed data analysis and wrote the paper. All have

631 commented on and reviewed the paper.

632

633 **Competing interests**

634   The authors declare that they have no conflict of interest.

635

636 **Acknowledgments**

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