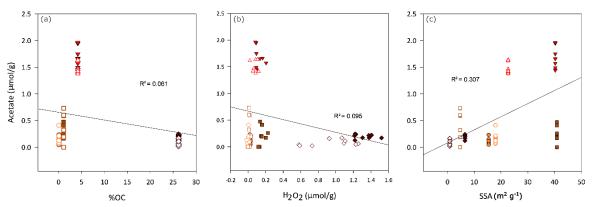
<u>Supplementary A: Acetate concentration in solution, normalized to grams of sample in incubation experiments plotted against potential contributing variables</u>



SSv (Coarse)

JS (Crushed)

SLM (Crushed) SLM (Coarse)

JS (Coarse) RMC (Crushed) RMC (Coarse)

Figure S1 Relationships between acetate concentration (μ mol) produced per gram of crushed and coarse samples relative to (a) organic carbon (OC), (b) H_2O_2 and (c) specific surface area (SSA). The concentration of H_2O_2 measured in solution was normalised to gram of sample.

<u>Supplementary B: Calculations to estimate potential NH₄⁺ contribution to Robertson Glacier and Longvearbreen subglacial systems</u>

Calculations to estimate daily NH₄⁺ contributions are based on Macdonald et al. (2018), using the following equation:

Daily
$$NH_4^+$$
 production (catchment) = $((\mu mol NH_4^+ g^{-1}) \times (flux suspended sediment (SS) g a^{-1}))/$ (glaciated area m²) x (365)

This is a first order estimate, and will likely overestimate the actual amount, since it assumes all the suspended sediment is of the rock type used for our incubation experiments.

No suspended sediment flux has been measured for Mercer Subglacial Lake (SLM) and thus these calculations were not extended to the SLM samples.

Catchment	Flux SS (g a ⁻¹)	Area (km²)	Daily NH ₄ ⁺ production umol m ⁻² day ⁻¹ for glaciated area of catchment	Daily NH ₄ ⁺ demand umol m ⁻² day ⁻¹ for glaciated area of catchment
Robertson Glacier	3x10 ⁸	1.4	3.5	0.2
Longyearbreen	15x10 ⁸	2.5	4.7	* 0.2

^{* =} Published estimates for NH_4^+ demand by subglacial microbes at Longyearbreen or nearby locations are not available in the literature to the best of the authors' knowledge, thus '0.2' was used for illustrative purposes.

Calculations:

<u>Robertson Muddy Carbonate RMC</u> - Suspended sediment flux (300 t km⁻² yr⁻¹) and area (1.4 km²) of Robertson Glacier taken from Macdonald et al. (2018)

$$NH_4^+ m^{-2} day = 6.0 \times 3 \times 10^8 / 1.4 \times 1000000 \times 365$$

RMC =
$$3.5 \mu mol NH_4^+ m^{-2} day$$

 NH_4^+ demand can be calculated using same equation, but substituting the N oxidation values from Boyd et al. (2011) = 0.28 μ mol N gdws⁻¹ day⁻¹. (gdws = grams dry weight sediment)

First convert N to NH_4^+ : $0.28 \ x \ (18.0383/14.0067) = 0.36 \ \mu mol \ NH_4^+ \ gdws^{-1} \ day^{-1}$.

Since the NH_4^+ demand is already a daily value, then it is only necessary to divide the annual suspended sediment flux by 365 to produce a daily flux, rather than multiply the area by 365.

=
$$(0.36 \text{ x} (3\text{x}10^8/365))/1.4 \text{ x} 1000000$$

= $0.2 \mu \text{mol NH}_4^+ \text{ m}^{-2} \text{ day}$

 \underline{SSv} - Longyearbreen suspended sediment (1500 t km⁻² yr⁻¹) and area (2.5 km² in 1990) values from Etzelmüller et al. (2000)

$$NH_4^+ m^{-2} day = 2.9 \times 15 \times 10^8 / 2.5 \times 1000000 \times 365$$

SSv =
$$4.7 \mu \text{mol NH}_4^+ \text{m}^{-2} \text{day}$$

Since the NH_4^+ demand is already a daily value, then it is only necessary to divide the annual suspended sediment flux by 365 to produce a daily flux, rather than multiply the area by 365.

$$= (0.36 \text{ x} (15 \text{x} 10^8/365))/2.5 \text{ x} 1000000$$

$$= 0.6 \mu mol NH_4^+ m^{-2} day$$

Supplementary References

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