

1 **The underappreciated impact of emission source profiles on the simulation of**
2 **PM_{2.5} components: New evidence from sensitivity analysis**

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19 **Abstract**

20 The chemical transport model (CTM) is an essential tool for air quality prediction
21 and management, widely used in air pollution control and health risk assessment.
22 However, the current models do not perform very well in simulating PM_{2.5} components.
23 Studies suggested that the uncertainties of model chemical mechanism, source emission
24 inventory and meteorological field can cause inaccurate simulation results. Still, the
25 emission source profile of PM_{2.5} has not been fully taken into account in current
26 numerical simulation. This study aims to answer (1) Whether the variation of source
27 profile adopted in ~~chemical transport models~~ (CTMs) has an impact on the simulation
28 of PM_{2.5} chemical components? (2) How much does it impact? (3) How does the impact
29 work? Based on the characteristics and variation rules of chemical components in
30 typical PM_{2.5} sources, different simulation scenarios were designed and the sensitivity
31 of components simulation results to PM_{2.5} sources profile was explored. Our findings
32 showed that the influence of source profile changes on simulated PM_{2.5} concentration
33 was insignificant, but its impact on PM_{2.5} components could not be ignored. The
34 variations of simulated components ranged from 8% to 167% under selected different
35 source profiles, and simulation results of some components were sensitive to the
36 adopted PM_{2.5} source profile in CTMs. These influences are connected to the chemical
37 mechanisms of the model since the variation of species allocations in emission sources
38 directly affected the thermodynamic equilibrium system. We also found that the
39 perturbation of the PM_{2.5} source profile caused the variation of simulated gaseous
40 pollutants, which indirectly indicated that the perturbation of the source profile affected
41 the simulation of secondary PM_{2.5} components. Given the vital role of air quality
42 simulation in environment management and health risk assessment, the
43 representativeness and timeliness of source profile should be considered.

44 **Keywords**

45 PM_{2.5}; source profile; component; numerical simulation; chemical transport model

46 1. Introduction

47 Ambient fine particulate matter (PM_{2.5}) pollution in some key regions of China
48 has attracted much attention (Liang et al., 2020; Huang et al., 2021). The chemical
49 components of PM_{2.5}, including elements (Al, Si, Fe, Mn, Ti, Cu, Zn, Pb, etc.), water-
50 soluble ions (SO₄²⁻, NO₃⁻, Cl⁻, F⁻, NH₄⁺, Na⁺, K⁺, Mg²⁺, Ca²⁺, etc.), and carbon-
51 containing components (Organic Carbon, OC; Elemental Carbon, EC) (Yang et al.,
52 2011; Li et al., 2013), have different physical and chemical properties, such as reactivity,
53 thermal stability, particle size distribution, residence time, optical properties, health
54 hazards, etc (Seinfeld and Pandis, 2006; Tang et al., 2006). According to long-term
55 monitoring results, in most regions of China, SO₄²⁻, NO₃⁻, NH₄⁺ and OC are the most
56 important species in ambient PM_{2.5} (Li et al., 2017a; Li et al., 2021), which has a certain
57 adverse impact on human health (Shi et al., 2018) and ecosystem (~~Han et al., 2019;~~
58 ~~Zhou et al., 2018~~), such as acid rain in southwest China (Han et al., 2019), food security
59 (Zhou et al., 2018), etc.

60 The chemical transport models (CTMs) play an important role in policy making
61 for regulatory purposes. Based on the scientific understanding of atmospheric physical
62 and chemical processes, CTMs are built to simulate the transport, reaction and removal
63 of pollutants on a certain scale in horizontal and vertical directions. With the
64 development of CTMs, the simulation accuracy of PM_{2.5} concentration has been
65 significantly improved. Higher requirements have been put forward for the precise
66 simulation of PM_{2.5} components so as to provide support for the use of CTMs in human
67 health risk assessment, climate effects, pollution sources apportionment, and so on
68 (Peterson et al., 2020; Lv et al., 2021). However, the current models perform not very
69 well in simulating some components (for example, PM_{2.5}-bound sulfate, nitrate,
70 ammonium, trace elements, etc.) (Zheng et al., 2015; Fu et al., 2016; Ying et al., 2018;
71 Cao et al., 2021). In the current literature, the correlation coefficient (R) and normalized
72 mean bias (NMB) are highly variable and inconsistent between the simulated and the
73 observed values (listed in Table S1). This is mainly attributable to the uncertainties of

74 model chemical mechanism, source emission inventory and meteorological field
75 simulation.

76 The chemical mechanisms involved in CTMs are derived from parameterized
77 assumptions based on laboratory simulation and field observations. The actual
78 atmospheric chemical processes are very complex, and some reaction mechanisms are
79 still limitedly understood. In addition, the integration of chemical reactions and
80 simplified treatment methods in the model cannot fully reflect the correlation among
81 atmospheric pollutants. For example, in some model mechanisms, other important
82 sulfate and nitrate formation pathways were added through new heterogeneous
83 chemistry, including the chemical reaction between SO₂ and aerosol, NO₂/NO₃/N₂O₃
84 and aerosol (Zheng et al., 2015), nitrous acid oxidized SO₂ to produce sulfate (Zheng
85 et al., 2020), dust particles promoted the oxidation of SO₂ (Yu et al., 2020), modified
86 the uptake coefficients for heterogeneous oxidation of SO₂ to sulfate (Zhang et al.,
87 2019), updated the heterogeneous N₂O₅ parameterization (Foley et al., 2010). Even
88 though the aforementioned processes can significantly improve the simulation of SO₄²⁻
89 and NO₃⁻, there is still a gap between the modeled and the actual atmospheric chemical
90 processes.

91 The uncertainty of source emission inventory also significantly affects the
92 simulation results of PM_{2.5} components (Shi et al., 2017; Sha et al., 2019). Due to
93 incomplete information or insufficient representativeness, pollutant emissions are
94 sometimes overestimated or underestimated, and the method for temporal and spatial
95 allocation also needs to be improved.

96 The uncertainty of meteorological field simulation is another crucial reason for the
97 simulation deviation, especially on heavy pollution days, the variation trends of PM_{2.5}
98 chemical components were not well-captured (Ying et al., 2018; Qi et al., 2019; Wang
99 et al., 2022). Precipitation is the key meteorological factor determining wet removal of
100 pollutants; boundary layer height and wind speed are the main factors affecting
101 convection and transport of pollutants; solar radiation, temperature and relative
102 humidity are the key factors affecting the formation of secondary particles (Huang et

103 al., 2019; Chen et al., 2020). Some literature reported that deviation from precipitation
104 and wind field simulation might lead to underestimation of SO_4^{2-} , NO_3^- and NH_4^+
105 (Cheng et al., 2015; Zhang et al., 2017). Devaluation of liquid water path and cloud
106 cover cause a decrease of sulfate formation in cloud, and ultimately results in
107 significantly underestimated components in simulation values (Sha et al., 2019; Foley
108 et al., 2010). Underestimation of temperature and relative humidity may also cause
109 adverse effects of temperature- and/or relative humidity-dependence chemical reaction
110 in the simulation (Sha et al., 2019).

111 In particular, the emission source profile of $\text{PM}_{2.5}$ (Hereinafter referred to as
112 "source profile"), [creating speciated emission inventories for CTMs \(Hsu et al., 2019\)](#),
113 has not been fully taken into account in the current numerical simulation ~~by CTMs~~. In
114 the reported literature, $\text{PM}_{2.5}$ species allocation coefficients of emission sources are
115 commonly treated in the following ways: (1) allocated $\text{PM}_{2.5}$ components of source
116 emissions by referring to source profile data in published literature or database like the
117 US SPECIATE (Fu et al., 2013; Wang et al., 2014; Ying et al., 2018); (2) chemical
118 profiles come from local measurement (Fu et al., 2013; Appel et al., 2013). However,
119 with the development of production technology and the innovation of pollution
120 treatment technology in recent years, some source profiles have changed dramatically
121 (Bi et al., 2019), such as SO_4^{2-} from coal burning, SO_4^{2-} content in $\text{PM}_{2.5}$ is generally
122 low in coal-fired power plant without desulfurizing facilities, while existing coal-fired
123 power plants using limestone/gypsum wet desulphurization, the contents of SO_4^{2-} in
124 $\text{PM}_{2.5}$ are significantly higher than that without desulfurization facilities (Zhang et al.,
125 2020). The timeliness of $\text{PM}_{2.5}$ species allocation coefficients in current CTMs also
126 needs to be considered.

127 This paper attempts to answer the following questions: (1) Whether the variation
128 of the source profile adopted in the air quality model has an impact on the simulated
129 results of $\text{PM}_{2.5}$ chemical components? (2) How much does it impact? (3) How does
130 the impact work? Aiming at these problems above, chemical composition and its
131 variation law for typical $\text{PM}_{2.5}$ emission sources are summarized, on this basis,

132 sensitivity tests are designed to identify whether PM_{2.5} source profiles and species
133 allocation in the model are important parameters that affect the simulation results of
134 chemical components in PM_{2.5}. We take CMAQ (one of the most widely used CTMs),
135 MEIC (a high-resolution inventory of anthropogenic air pollutants in China) as the
136 carriers. The same kind of experiment is also applicable to other CTMs and emission
137 inventories. The aim of this study is to provide support for the effective utilization of
138 source profiles in the CTMs and improvement of the simulation schemes.

139 **2. Model and Data**

140 **2.1 Model configuration**

141 Weather Research and Forecasting model (WRF-3.7.1), the widely used
142 Community Multiscale Air Quality model (CMAQv5.0.2), and Multi-resolution
143 Emission Inventory for China (MEICv1.3) have been used in this study. MEIC provided
144 the emission inventory which is developed by Tsinghua University, mainly tracked
145 anthropogenic emissions in China including coal-fired power plants, industry, vehicles,
146 residents and agriculture (http://meicmodel.org/?page_id=135) (Li et al., 2017b; Zheng
147 et al., 2018). The WRF model was used to generate meteorological inputs for the
148 CMAQ model. Three nested modeling domains consisting of 36 km×36 km (Dom1),
149 12 km×12km (Dom2), and 4 km×4km (Dom3) horizontal grid sizes were set, as shown
150 in Fig. 1. The initial and boundary conditions for WRF were based on the North
151 American Regional Reanalysis data archived at National Center for Atmospheric
152 Research (NCAR). In addition, surface and upper air observations obtained from
153 NCAR were used to further refine the analysis data. The major configurations we used
154 in CMAQ were illuminated as follows: Gas-phase chemistry was based on the CB05
155 mechanism and the aerosol dynamics/chemistry was based on the aero6 module
156 (cb05tucl_ae6_aq). The detailed model configurations were shown in Table S2,
157 regional distribution of PM_{2.5} emission sources were shown in Fig. S2.

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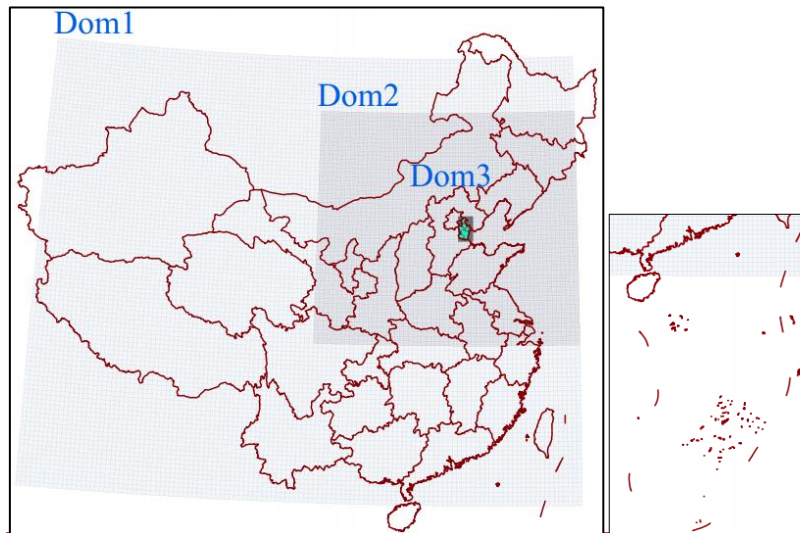
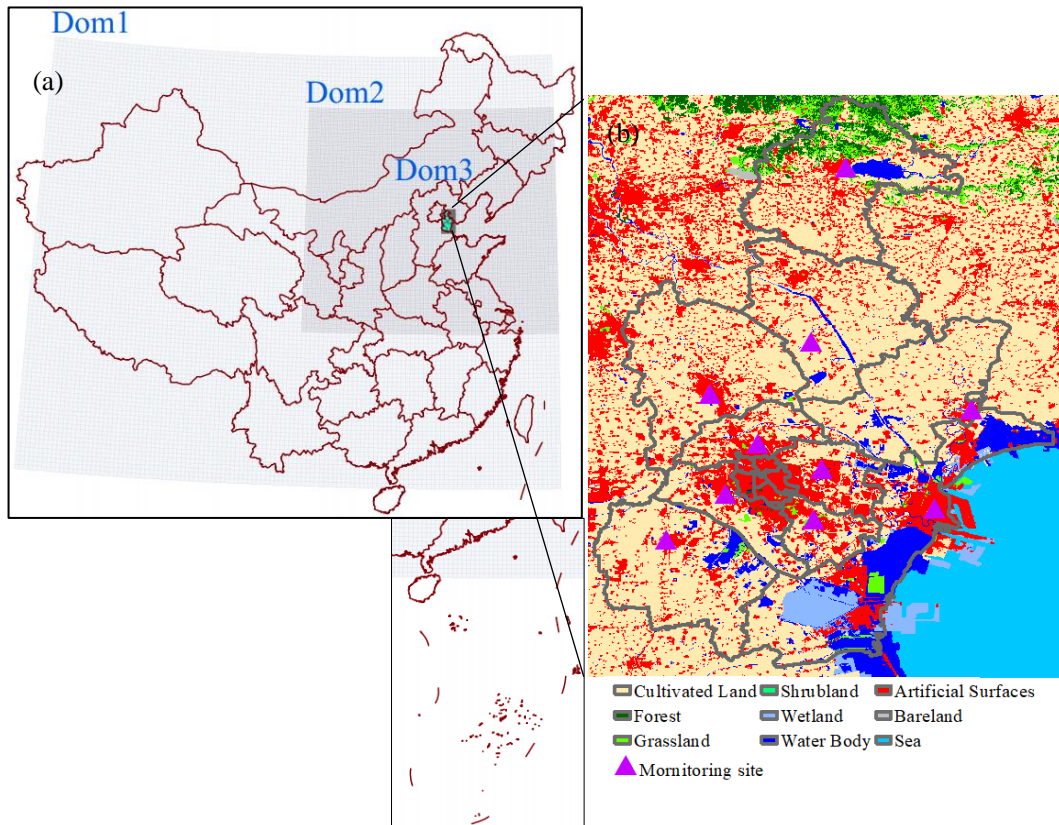


Fig.1 Modeling domains of the CMAQ model.



160

161 Fig.1 Modeling domains of the CMAQ model. (a) The three-domain nested CMAQ domains; (b)
162 Land use and observation sites of Dom3 (Data source of Land use: GLOBELAND30,
163 www.globeland30.org, National Geomatics Center of China).

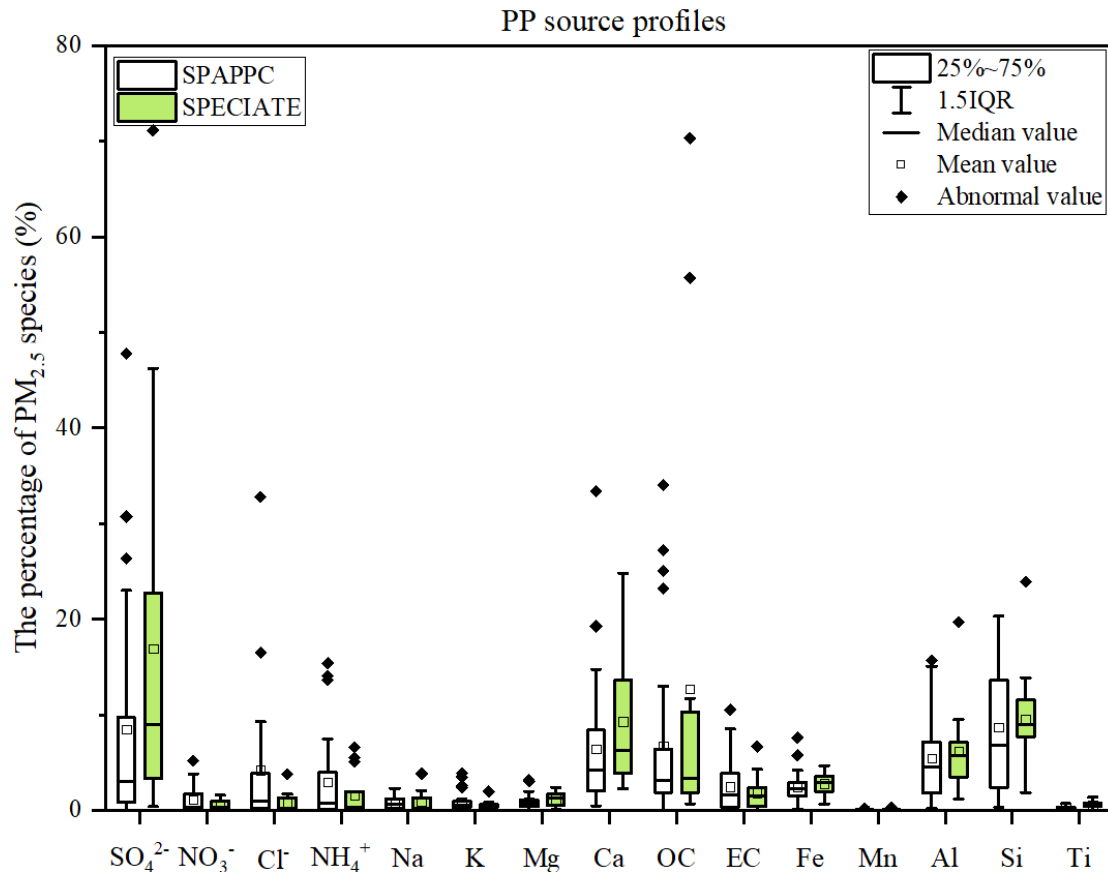
164 2.2 Selection and comparison of PM_{2.5} source profile

165 The PM_{2.5} emission source profiles from database of Source Profiles of Air
166 Pollution (SPAP) (<http://www.nkspap.com:9091/>), U.S. Environmental Protection
167 Agency's (EPA) SPECIATE database (<https://www.epa.gov/air-emissions->

168 modeling/speciate) as well as from published literature were selected, respectively. The
169 SPAP was developed by the State Environment Protection Key Laboratory of Urban
170 Particulate Air Pollution Prevention, Nankai University, China. This database contains
171 more than 3000 size-resolved source profiles of stationary combustion sources,
172 industrial processes, vehicle exhaust, biomass burning, dust and cooking emissions and
173 other sources, collected from more than 40 cities in China since 2001. In addition to
174 inorganic elements, water-soluble ions, OC, EC and other conventional components,
175 some source profiles also encompass a series of tracer information, such as organic
176 markers, isotopes, single particle mass spectrometry, VOCs and other gaseous
177 precursors. Based on species in the aerosol chemical mechanism (AERO6) (Appel et
178 al., 2013; Chapel Hill, 2012), we selected 15 components in PM_{2.5} source profiles
179 including Al, Ca, Cl, EC, Fe, K, Mg, Mn, Na, OC, Si, Ti, NH₄⁺, NO₃⁻ and SO₄²⁻, the
180 remaining components are classified as “other”. Emission sources are divided into four
181 main categories referred to the classification in MEIC: coal combustion by power plants
182 (PP), industrial processes (IN), residential emission (RE) and transportation sector (TR).

183 Coal-fired power plants remain the main coal consumers in China, which
184 accounted for 50.2% of total coal consumption in 2019 (NBS, 2021) and gained much
185 more attention (Wu et al., 2022), especially with the wide implementation of the
186 strictest ultralow emission standards, PM_{2.5} emission characteristics have changed
187 accordingly (Wu et al., 2020). There are obvious differences in PM_{2.5} source profiles
188 between SPAPPC (SPAP database and published source profiles in China) and
189 SPECIATE (SPECIATE database), detailed information is shown in Table S3. The
190 percentages of species in PP source profiles are plotted in Fig. 2. The main components
191 in SPAPPC are sorted by Si, SO₄²⁻, OC, Ca with average values of 8.7±6.8%, 8.5±11.5%,
192 6.8±9.1% and 6.5±6.9%, respectively; The SPECIATE are enriched in SO₄²⁻
193 (16.9%±20.0%), OC (12.7±21.8%), Si (9.6±5.0%) and Ca (9.3±7.3%), higher than
194 SPAPPC. Coal properties, burning conditions, pollution control measures and sampling
195 methods are the main reasons for those great percentage fluctuations. Different
196 treatment processes of flue gases, e.g. wet/dry limestone, ammonia and double-alkali

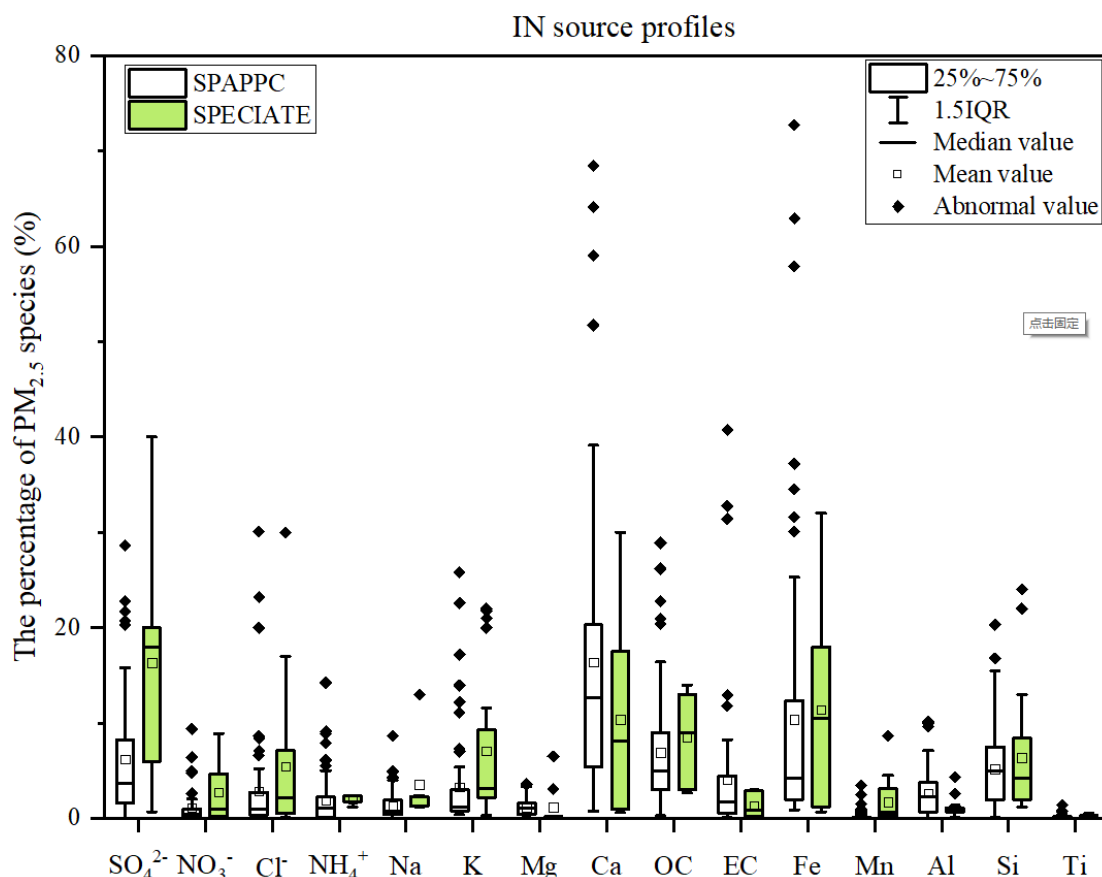
197 flue gas desulfurization, will affect the percentages of components in source profiles
 198 (Zhang et al., 2020). It has been reported that the percentage of Ca, Mg, SO_4^{2-} and Cl^-
 199 in PP profiles increased after the limestone-gypsum method was used in coal-fired
 200 power plants (Bi et al., 2019). Besides that, the percentage of Cl^- in SPAPPC is
 201 obviously higher than that in SPECIATE, which might attribute to the generally higher
 202 Cl^- content in raw coal in China (Guo et al., 2004).



203 SO_4^{2-} NO_3^- Cl^- NH_4^+ Na K Mg Ca OC EC Fe Mn Al Si Ti
 204 Fig. 2 Chemical profiles for $\text{PM}_{2.5}$ emitted from coal-fired power plant (PP). Data obtained from
 205 SPAPPC (SPAP database and published source profiles in China) and SPECIATE (U.S. EPA
 206 SPECIATE database)

207 Industrial emissions are one of the major sources of $\text{PM}_{2.5}$ (Hopke et al., 2020),
 208 the percentages of Ca, Fe, OC and SO_4^{2-} are relatively high both in SPAPPC and
 209 SPECIATE of industrial processes, but the shares in different source profile database
 210 varied (Detailed information were shown in Table S4~S7). In SPAPPC, these four
 211 components account for $16.4 \pm 14.9\%$, $10.4 \pm 14.4\%$, $6.9 \pm 6.1\%$, $6.2 \pm 6.4\%$, the
 212 proportions in SPECIATE are $10.4 \pm 9.8\%$, $11.4 \pm 10.6\%$, $8.5 \pm 4.9\%$, $16.3 \pm 13.3\%$,
 213 respectively (Fig. 3). Large variations of components and their percentages in industrial

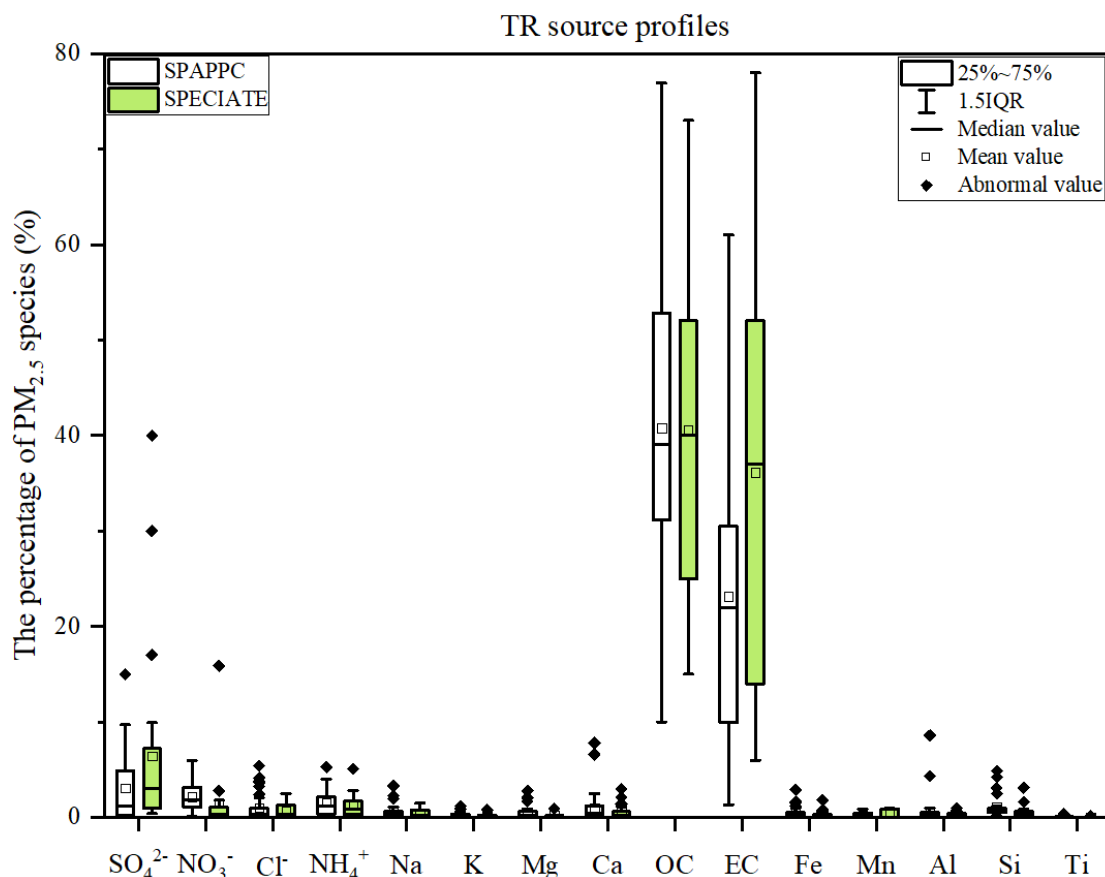
214 processes are attributed to the manufacturing processes, raw material, pollution control
 215 measures and so on (Ji et al., 2017; Bi et al., 2019; Gao et al., 2022). For example, Ca,
 216 Al, OC and SO_4^{2-} are found to have the highest percentage in cement sources (Guo et
 217 al., 2021); Fe, Si and SO_4^{2-} are the most abundant species in steel industry emission
 218 (Guo et al., 2017).



219 Fig. 3 Chemical profiles for $\text{PM}_{2.5}$ emitted from industry processes (IN). Data obtained from
 220 SPAPPC (SPAP database and published source profiles in China) and SPECIATE (U.S. EPA
 221 SPECIATE database)
 222

223 Traffic contributed a large fraction of $\text{PM}_{2.5}$ in many locations (Hopke et al., 2022).
 224 It is well-known that the transportation sector makes a dominant contribution of OC
 225 and EC. The main components of $\text{PM}_{2.5}$ emitted from traffic sources are OC, EC and
 226 SO_4^{2-} both in SPAPPC and SPECIATE, but still vary in wide range (Detailed
 227 information was given in Table S8~S10). In SPAPPC, the percentages of OC, EC and
 228 SO_4^{2-} are $40.8 \pm 15.0\%$, $23.1 \pm 13.8\%$, $3.1 \pm 3.7\%$, and in SPECIATE, the percentages are
 229 $40.6 \pm 16.4\%$, $36.1 \pm 21.5\%$, $6.4 \pm 9.9\%$, respectively (Fig. 4). These significant
 230 differences mainly attribute to the vehicle type, fuel quality, mixing ratio between oil

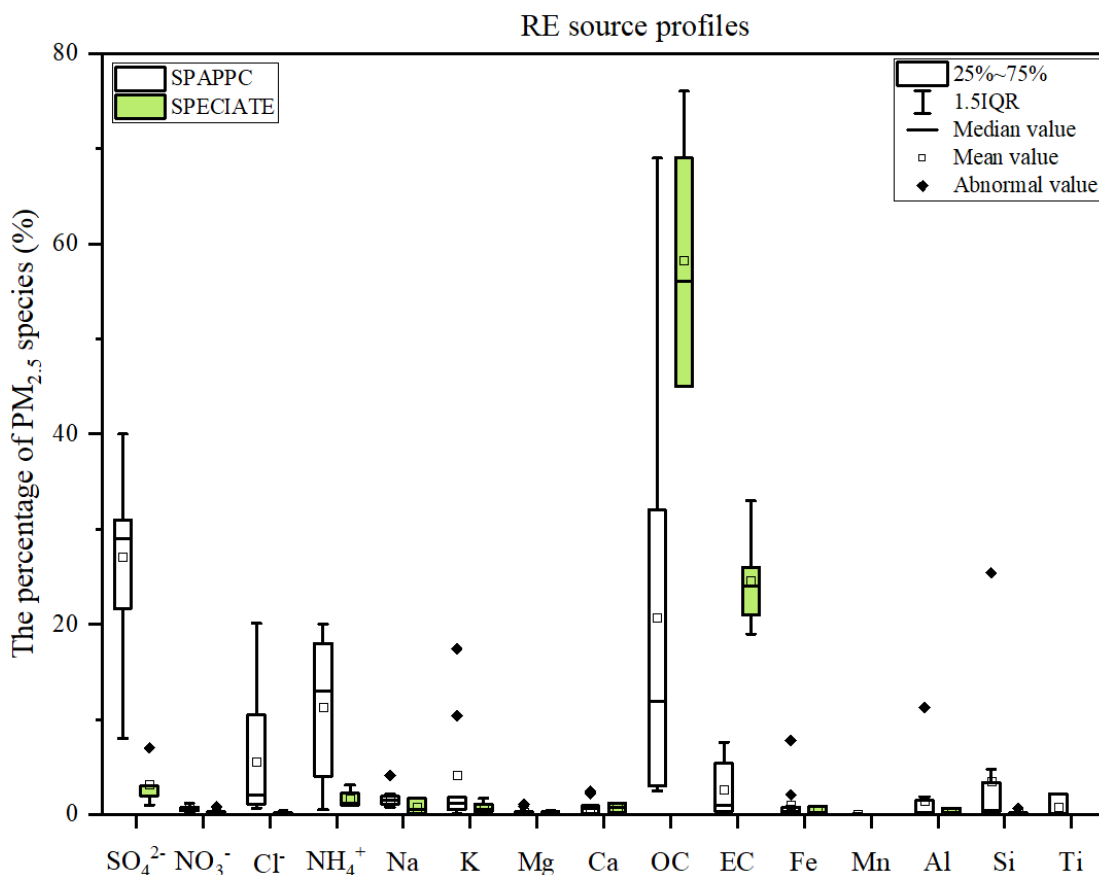
231 and gas and the combustion phase in vehicle engine and so on (Xia et al., 2017).



232
233 Fig. 4 Chemical profiles for PM_{2.5} emitted from transportation sector (TR). Data obtained from
234 SPAPPC (SPAP database and published source profiles in China) and SPECIATE (U.S. EPA
235 SPECIATE database)

236 Residential coal combustion, as the leading source of global PM_{2.5} emission
237 (Weagle et al., 2018), has a much higher emission factor than coal-fired power plant
238 (Wu et al., 2022). The fraction of components varied greatly in the profiles measured
239 from SPAPPC and SPECIATE (Detailed information was given in Table S11), SO₄²⁻,
240 OC, NH₄⁺ and EC make the main contribution to PM_{2.5} emitted from residential coal
241 combustion. In SPAPPC, the average percentages of SO₄²⁻, OC, NH₄⁺, EC are
242 27.1±10.1%, 20.7±20.6%, 11.3±7.7%, 2.6±2.8%, respectively. In SPECIATE, the
243 average percentages are OC (58.2±14.0%), EC (24.6±5.4%), SO₄²⁻ (3.2±2.3%) and
244 NH₄⁺ (1.6±1.0%) (Fig. 5). Total percentages of OC and EC in SPECIATE are over 80%,
245 obviously higher than that in SPAPPC, while a higher percentage of SO₄²⁻, Cl⁻, K and
246 Si are observed in SPAPPC. The coal type and properties, burning condition are the
247 main factors affecting the percentages of PM_{2.5} components, like the chunk coal burning

248 has relatively higher percentages of OC, EC, SO_4^{2-} , NO_3^- and NH_4^+ than honeycomb
 249 briquette (Wu et al., 2021; Song et al., 2021).



250
 251 Fig. 5 Chemical profiles for $\text{PM}_{2.5}$ emitted from residential coal combustion (RE). Data obtained
 252 from SPAPPC (SPAP database and published source profiles in China) and SPECIATE (U.S. EPA
 253 SPECIATE database)

254 Briefly, many factors can affect $\text{PM}_{2.5}$ source profiles, and with the innovation of
 255 manufacturing technique and pollution control technology, changes in fuel and raw and
 256 auxiliary materials, the main chemical components and their percentages would change
 257 dramatically. To explore whether the variations of source profile would be one of the
 258 important factors affecting the simulation results of $\text{PM}_{2.5}$ species in CTMs, we
 259 designed a series of simulation tests as follows.

260 **3 Whether the variation of source profile adopted in CTMs has an impact on the**
 261 **simulation of chemical components in $\text{PM}_{2.5}$?**

262 In this part, we separately selected source profiles from SPAPPC and SPECIATE
 263 databases and applied them in emission inventory for simulating $\text{PM}_{2.5}$ and its

264 components with other modeling conditions unchanged, corresponding to case
265 CMAQ_SPA and CMAQ_SPE. The detailed information ~~on~~of source profiles is shown
266 in Figure S1. To determine the similarity between the two groups of source profiles,
267 Coefficient ~~divergence~~Divergence (CD) is calculated using the following formula
268 (Wongphatarakul et al., 1998):

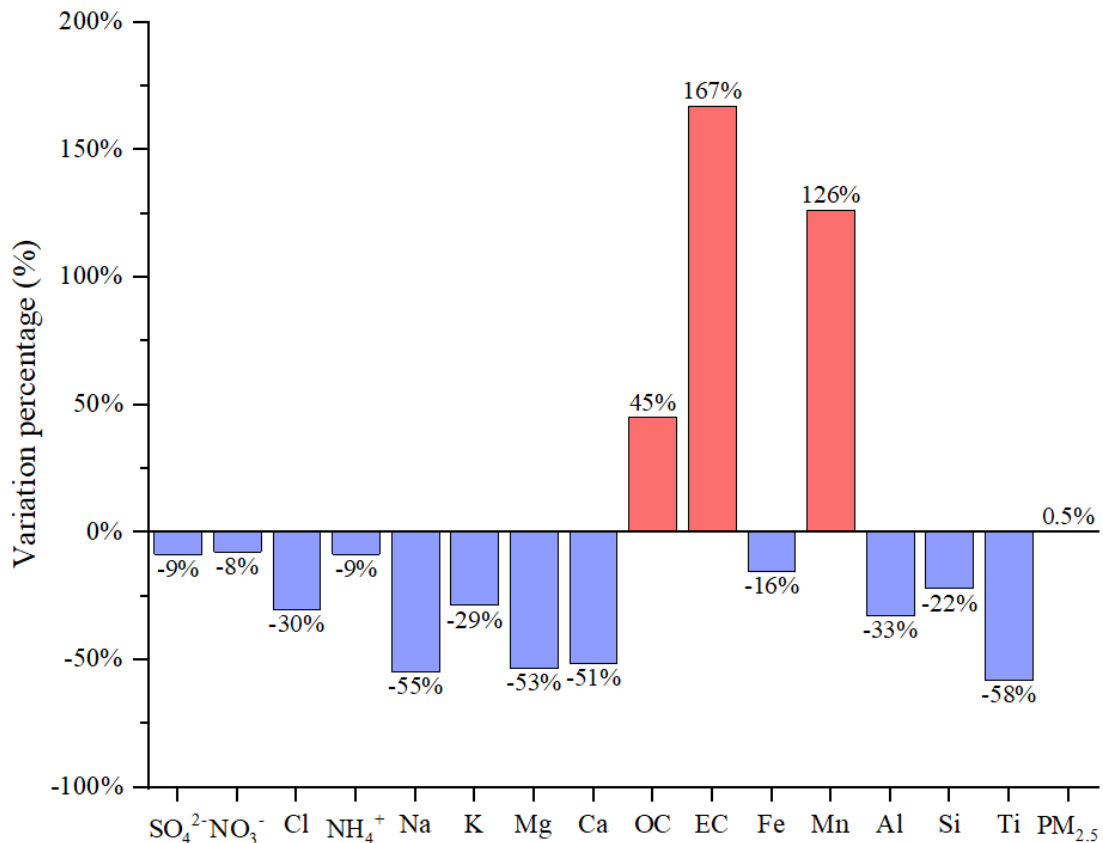
$$269 \quad CD_{jk} = \sqrt{\frac{1}{p} \sum_{i=1}^p \left(\frac{x_{ij} - x_{ik}}{x_{ij} + x_{ik}} \right)^2} \dots\dots\dots (1)$$

270 Where CD_{jk} is the coefficient of divergence of source profile j and k , p was the
271 number of chemical components in source profile, x_{ij} is the weight percentage for
272 chemical component i in source profile j , x_{ik} is the weight percentage for i in source
273 profile k (%). The CD value is in the range of 0 to 1, if the two source profiles are
274 similar, the value of CD is close to 0; if the two are very different, the value was close
275 to 1.

276 By comparing the selected SPAPPC source profiles with the selected SPECIATE
277 source profiles, the coefficient divergences for the four main source categories were
278 $CD_{PP}(0.67) > CD_{RE}(0.62) > CD_{TR}(0.60) > CD_{IN}(0.60)$, which meant the selected source
279 profiles in the two simulation cases were quite different. The simulated concentration
280 of $PM_{2.5}$ and its components (For this part and each test case in next section) at 10
281 ambient air quality monitoring stations (Table S12) were extracted from CMAQ outputs
282 of the innermost simulation domain. We selected one air quality monitoring station to
283 study the influence of $PM_{2.5}$ source profile on numerical simulation of $PM_{2.5}$ -bound
284 components and to explore the relevant laws in the atmosphere, then used the left 9 sites
285 to further illustrate the conclusions suggested.

286 The simulation results for $PM_{2.5}$ species under CMAQ_SPA and CMAQ_SPE
287 cases also showed big differences (as shown in Fig. 6 and Table S13), in which the
288 largest difference in simulated concentration was EC with CAMQ_SPE giving higher
289 by 167% than CMAQ_SPA; For OC and Mn, higher values were also given by
290 CMAQ_SPE than by CMAQ_SPA (45% and 126% on average, respectively); For the
291 remaining components, the simulated concentration by CMAQ_SPE was lower than

292 CMAQ_SPA with Ti (58%), Na (55%), Mg (53%), Ca (51%), Al (33%), Cl (30%), K
 293 (29%), Si (22%), Fe (16%), NH₄⁺ (9%), SO₄²⁻ (9%), NO₃⁻ (8%), separately. While the
 294 simulated PM_{2.5} concentrations under the two cases were quite close. The influence of
 295 source profile variation on the simulated PM_{2.5} concentration was not significant, but
 296 the influence on the simulation of chemical components in PM_{2.5} could not be ignored.



297
 298 Fig. 6 The percentage difference of simulated concentration (PM_{2.5} and its components) between
 299 CMAQ_SPE and CAMQ_SPA (relative to CAMQ_SPA); PM_{2.5} source profiles from SPAPP and
 300 SPECIATE database were applied in emission inventory for simulating PM_{2.5} and its components,
 301 corresponding to case CMAQ_SPA and CMAQ_SPE, respectively.

302 **4 How much did the variation of source profile adopted in CTMs impact on the**
 303 **simulation of chemical components in PM_{2.5}?**

304 In order to quantitatively characterize how much the source profiles affect the
 305 simulation results of PM_{2.5} and its components, we selected the chemical composition
 306 of code 000002.5 (Variety of different categories, used for the overall average

307 composite profiles (Hsu et al., 2019)) in the US EPA Speciate_5.0_0 database as species
 308 allocation of PM_{2.5} components. The corresponding percentages of EC, OC, Mn, Fe, Ti,
 309 Al, Si, Ca, Mg, K, Na, Cl, NH₄⁺, NO₃⁻ and SO₄²⁻ in PM_{2.5} were shown in Fig. 7 (SGL,
 310 base case simulation).

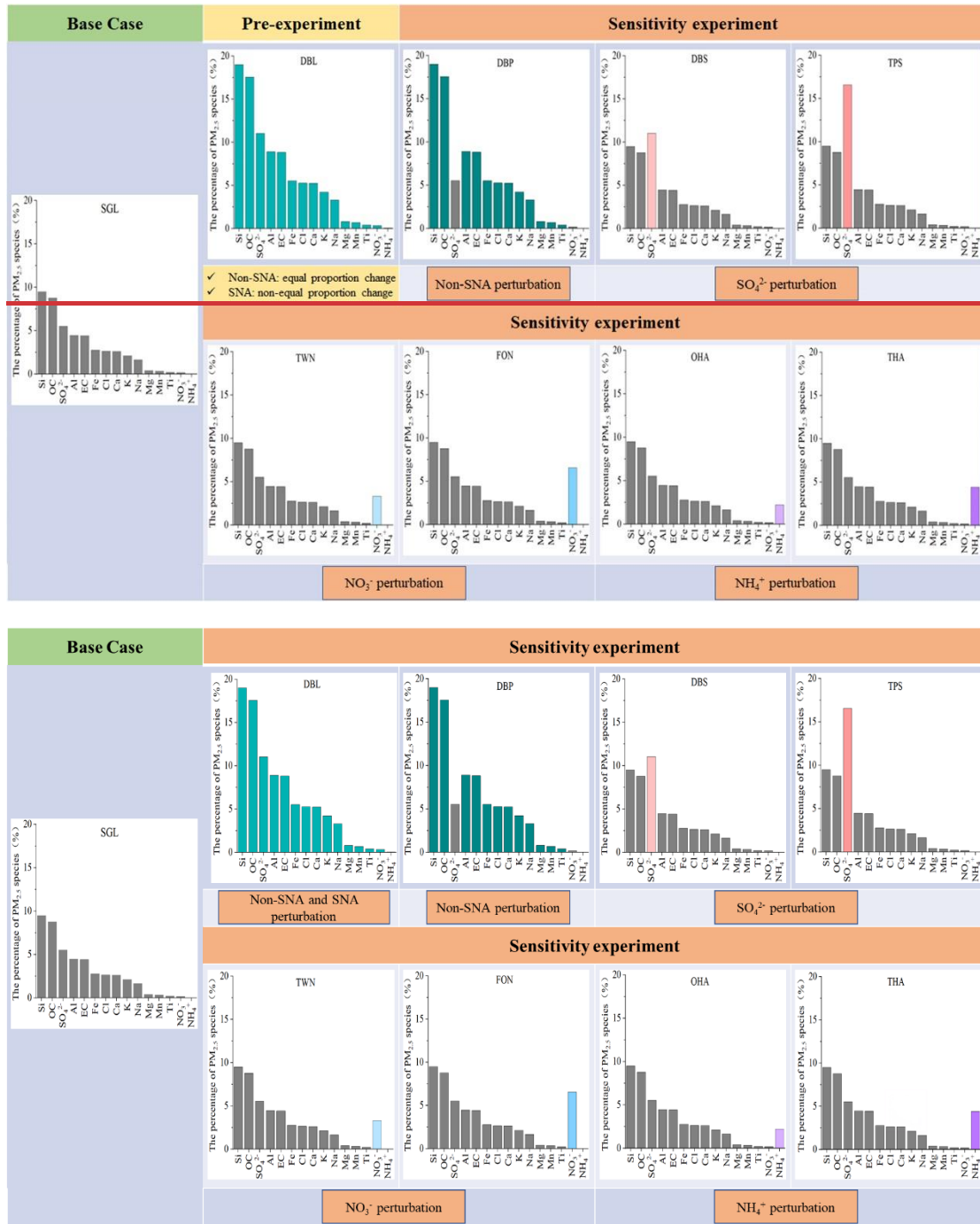


Fig. 7 The general roadmap of sensitivity tests
 (The histogram in each case were the speciation profile in CTMs)

Table 1 The content of sensitivity experiment cases

Cases	Description
Pre-experiment <u>Case S0</u> (DBL): add perturbation to Non-SNA and SNA	The percentage of all the listed components in the source profile of base case (SGL) were doubled, and the proportion of unlisted components (Other) decreased to 9%.
Case S1 (DBP): add perturbation to Non-SNA	The percentages of non-SNA were doubled and SNA(SO_4^{2-} , NO_3^- , NH_4^+) species stayed the same with that in SGL (the cumulative percentage of listed species was 85.3%), the proportion of unlisted components decreased to 14.7%.
Case S2 (DBS and TPS): add perturbation to SO_4^{2-}	The percentage of SO_4^{2-} was doubled (11%, DBS, represented Double Sulfate), tripled (16.5%, TPS, represented Triple Sulfate) and the other listed 14 species stayed the same with that in SGL (the cumulative percentage of listed species was 51% and 57%, respectively), the proportion of unlisted components decreased to 49% and 43%.
Case S3 (TWN and FON): add perturbation to NO_3^-	The NO_3^- content was raised up to 20 times (3.3%, TWN) and 40 times (6.6%, FON) of that in SGL (0.16%), the other 14 species stayed the same with SGL (the cumulative percentage of listed species was 48.6% and 51.9%, respectively), the proportion of unlisted components decreased to 51.4% and 48.1%.
Case S4 (OHA and THA): add perturbation to NH_4^+	The NH_4^+ content was raised up to 100 times (2.2%, OHA), 200 times (4.4%, THA) of that in SGL (0.02%), the other 14 species stayed the same with SGL (the cumulative percentage of listed species was 47.7% and 49.9%, respectively), the proportion of unlisted components decreased to 52.3% and 50.1%.

Note: The source profiles in all cases listed in the table were calculated based on the base case SGL. In the design of simulation cases, the reason why the disturbance amplitude of NH_4^+ and NO_3^- were significantly higher than that of other components such as SO_4^{2-} and Non-SNA, was because the percentages of NH_4^+ and NO_3^- in the base source profile (SGL, based on the chemical composition of code 000002.5 in the EPA Speciate_5.0_0 database) were very low, while the percentage of NH_4^+ and NO_3^- in SPAPPC exhibited in section 2.2 were orders of magnitude higher than those in SGL.

316 Given the large number and complex chemical composition of $\text{PM}_{2.5}$, it is
317 advisable to classify it reasonably before designing sensitivity experiments. The ~~pre-~~
318 ~~experiment~~ Case S0 was to double the percentage of the listed 15 components mentioned
319 above (SGL) in $\text{PM}_{2.5}$ species allocation for emission sources (DBL case, the
320 cumulative percentage was 91%, the details were shown in Fig. 7 and Table 1). As the
321 percentage of these components increased, the proportion of unlisted components

322 (represented by Other) decreased to 9% in order to meet the requirement that the total
323 percentage of all components is 100%. Then we compared the simulation results before
324 (SGL case) and after perturbation (DBL case) in species allocation of PM_{2.5} sources.

325 In the case DBL, when the percentage of all the components except “other” were
326 doubled in the source profile, the simulated concentrations of Al, Ca, Cl, EC, Fe, K,
327 Mg, Mn, Na, OC, Si and Ti doubled as well, while the simulated concentration of NO₃⁻,
328 SO₄²⁻ and NH₄⁺ only increased at about 3%, 10% and 4%, respectively, although the
329 simulated concentration of PM_{2.5} was not obviously changed (Detailed simulation
330 results were shown in Table S14). Through this ~~pre-experiment~~Case S0, we found that
331 the results for SNA (SO₄²⁻, NO₃⁻, and NH₄⁺) and Non-SNA were obviously different.
332 Therefore, we divided the components in the source profile into two groups (Non-SNA
333 and SNA) and designed a series of sensitivity tests listed in next section to further
334 explore how species allocation of PM_{2.5} in emission sources of CTMs would affect the
335 simulation results.

336 4.1 Sensitivity tests design

337 Based on the ~~pre-experiment~~Case S0 results, sensitivity tests were designed by
338 changing the percentages of the target components and related components in the base
339 case (SGL): perturbation on each component of Non-SNA, perturbation on SO₄²⁻,
340 perturbation on NO₃⁻, and perturbation on NH₄⁺. The general roadmap of sensitivity
341 tests was shown in Fig. 7, and the illustration of each case was summarized in Table 1.
342 The basic rules must be followed: a) perturbation on the percentage of each component
343 in source profile fell within the variation range of its measured value described in
344 section 2.2. b) The sum of the percentage of listed Non-SNA, SNA and Other
345 components in PM_{2.5} source profile was 100%.

346 4.2 Evaluation index for simulation result

347 In order to quantify the concentration changes of simulated PM_{2.5} components
348 caused by the perturbation in source profile, we proposed the sensitivity coefficient (δ)
349 as evaluation index. The calculation formula is as follows:

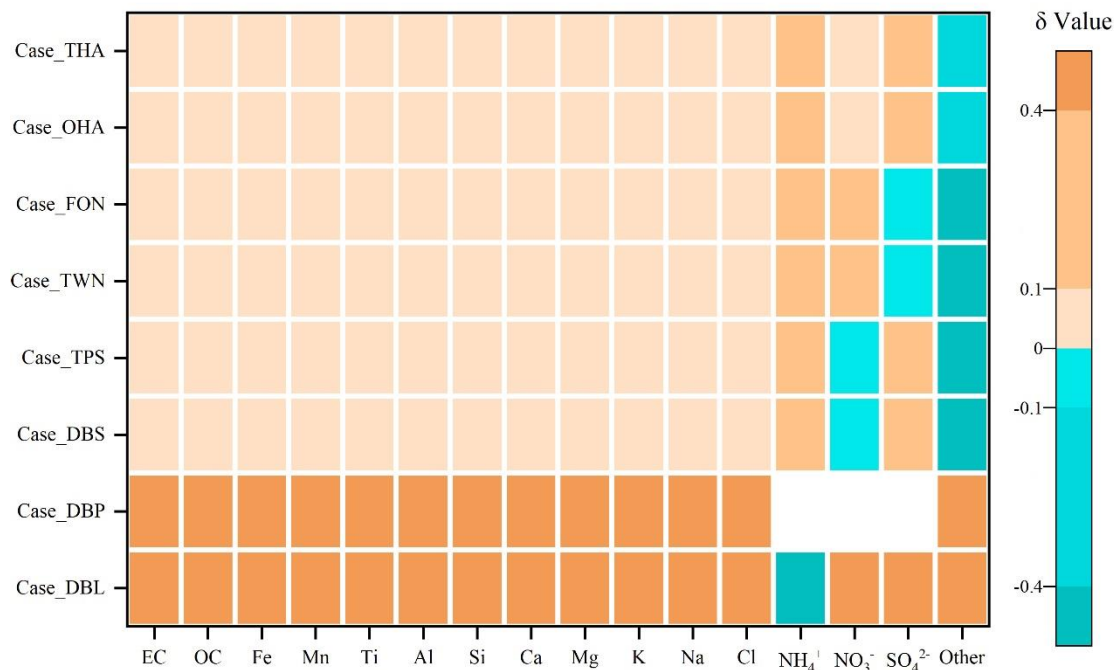
$$\delta_i = \begin{cases} \frac{\frac{C_{i_case}}{C_{PM_{2.5_case}}} \times 100\% - \frac{C_{i_base}}{C_{PM_{2.5_base}}} \times 100\%}{P_{i_case} - P_{i_base}} & \text{(For DBL and DBP)} \\ \frac{\frac{C_{i_case}}{C_{PM_{2.5_case}}} \times 100\% - \frac{C_{i_base}}{C_{PM_{2.5_base}}} \times 100\%}{P_{j_case} - P_{j_base}} & \text{(For other cases)} \end{cases} \dots\dots\dots (2)$$

351 Wherein, δ_i is the sensitivity coefficient of component i , representing the change
352 of the simulated value of its content in ambient $PM_{2.5}$ corresponded to 1% perturbation
353 in the source profiles. C_{i_case} is the simulation result of component i in different
354 sensitivity experiment cases, $\mu g/m^3$; C_{i_base} is the simulation result of components i in
355 base case, $\mu g/m^3$; $C_{PM_{2.5_case}}$ is the simulation result of $PM_{2.5}$ in different sensitivity
356 experiment cases, $\mu g/m^3$; $C_{PM_{2.5_base}}$ is the simulation result of $PM_{2.5}$ in base case, $\mu g/m^3$;
357 P_{i_case} is the percentage of component i in different source profile of sensitivity
358 experiment cases, %; P_{j_case} is the percentage of perturbed component j in different
359 source profile of sensitivity experiment cases, %; P_{i_base} is the percentage of component
360 i in base case source profile, %; ; P_{j_base} is the percentage of perturbed component j in
361 base case source profile, %.

362 The positive value of δ means the simulated concentration of $PM_{2.5}$ component
363 increases (decreases) with the increase (decrease) of the perturbation to the percentage
364 of components in source profile, while the meaning of negative δ is just the opposite. If
365 the absolute value of δ is less than or equal to 0.1, the simulated result of $PM_{2.5}$ chemical
366 component is considered to be insensitive to the corresponding variation of source
367 profile; If the absolute value of δ falls between 0.1 and 0.4 (included), the simulated
368 results of $PM_{2.5}$ chemical component is considered to be sensitive to the variation of
369 source profile; If the absolute value of δ is larger than 0.4, the simulated results of $PM_{2.5}$
370 chemical component is very sensitive to the variation of source profile. The greater the
371 absolute value of δ is, indicates the variation of source profile adopted in CMAQ has
372 more obvious impact on the simulated results of $PM_{2.5}$ chemical components.

373 **4.3 The response of simulated PM_{2.5} components**

374 Fig.8 listed the sensitivity coefficients of simulated ambient PM_{2.5} components to
 375 the perturbation of source profile under each test case. In case DBL (The percentage of
 376 all the listed components in the source profile of base case (SGL) was doubled), the
 377 sensitivity coefficient (δ) of NH₄⁺ was negative, and the absolute value was the highest,
 378 indicating that the simulated proportion of NH₄⁺ in ambient PM_{2.5} decreased, and it was
 379 very sensitive to the variation of source profile. Conversely, the sensitivity coefficient
 380 of NO₃⁻ was close to 1, which illustrated that the simulated proportion of NO₃⁻ in
 381 ambient PM_{2.5} increased proportionally with the change in source profile. The δ of SO₄²⁻
 382 also showed a very sensitive property. The simulated Non-SNA concentrations were
 383 doubled when compared to the base case (SGL).



384 Fig. 8 The sensitivity coefficients (δ) of simulated components to the perturbation of adopted source
 385 profile in different cases. Note: Each small color box in the figure represented the sensitivity level
 386 (indicated by the legend on the right) of PM_{2.5} components (the x-coordinate) in different cases (y-
 387 coordinate). The blank grids in DBP case indicated no perturbation to SNA in PM_{2.5} source profile
 388 under this case.
 389

390 In case DBP, when the percentages of listed Non-SNA in the source profile were
 391 doubled, the simulated proportions of Non-SNA (Al, Ca, Cl, EC, Fe, K, Mg, Mn, Na,
 392 OC, Si and Ti) in ambient PM_{2.5} synchronous increased, and were very sensitive to the

393 change in the adopted source profile with a sensitivity coefficient (δ) of 0.5.
394 Interestingly, the simulated concentration of SNA in ambient $\text{PM}_{2.5}$ also changed
395 although the SNA in source profile did not change, the concentration of NO_3^- and SO_4^{2-}
396 increased by 2% and 3%, respectively, NH_4^+ decreased by 10% (Detail simulation
397 results of different cases were shown on Table S15~S21).

398 Under SO_4^{2-} perturbation cases (Case DBS and Case TPS), we found the simulated
399 results of Non-SNA and NO_3^- had no obvious variation when compared with the base
400 case. Either in Case DBS or in Case TPS, the δ of Non-SNA and NO_3^- were always
401 between -0.1 to 0.1. But when the percentage of SO_4^{2-} was doubled in $\text{PM}_{2.5}$ source
402 profile (DBS), the simulated concentration of NH_4^+ and SO_4^{2-} increased by 6% and 8%,
403 respectively. In Case TPS (the percentage of SO_4^{2-} was tripled), the simulated
404 concentration of NH_4^+ and SO_4^{2-} were increased by 11% and 16%, respectively. The δ
405 of NH_4^+ and SO_4^{2-} were 0.12 and 0.36, sensitive toward to positive direction with the
406 increase of SO_4^{2-} in the source profile.

407 In the situation of NO_3^- perturbation (Case TWN and Case FON), the simulated
408 concentrations of Non-SNA hardly change when compared to the base case, while the
409 changing characteristics of SNA concentrations were different. In cases TWN and FON,
410 the simulation concentration of NH_4^+ increased by 2.6% and 5.4% when compared with
411 the base case, the simulated NO_3^- increased by 14% and 30%, the simulated SO_4^{2-}
412 decreased slightly, even could be neglected in some observation sites. The simulated
413 concentrations of Non-SNA and SO_4^{2-} were insensitive to the perturbation of NO_3^- in
414 $\text{PM}_{2.5}$ source profile; NH_4^+ was sensitive, and NO_3^- was very sensitive.

415 When we put perturbation to NH_4^+ in the source profile (Case OHA and Case
416 THA), the simulation results of Non-SNA were almost not changed, the simulated
417 concentration of SO_4^{2-} , NH_4^+ , NO_3^- increased in OHA and THA. The δ of SNA to the
418 variation of NH_4^+ in the source profile were positive and $\delta(\text{SO}_4^{2-}) > \delta(\text{NH}_4^+) > \delta(\text{NO}_3^-)$,
419 SO_4^{2-} and NH_4^+ were sensitive to the NH_4^+ perturbation in the source profile, but NO_3^-
420 was not so sensitive.

421 In general, the simulation results of components in ambient $\text{PM}_{2.5}$ were affected in

422 one way or another by the change of source profiles adopted by CMAQ. Both of the
423 simulated Non-SNA and SNA were very sensitive to the perturbation of Non-SNA in
424 source profile. When the percentage of SNA changed in the source profile, simulated
425 concentrations of Non-SNA generally have little change, but the simulation results of
426 SNA could change in different levels: the simulated SO_4^{2-} was very sensitive and NH_4^+
427 was sensitive to the perturbation of SO_4^{2-} in source profile, simulated NO_3^- was very
428 sensitive and NH_4^+ was sensitive to the perturbation of NO_3^- , SO_4^{2-} and NH_4^+ were
429 sensitive to the perturbation of NH_4^+ . The simulated component such as SO_4^{2-} was
430 influenced not only by the change of SO_4^{2-} itself but also by other components like
431 some Non-SNA and NH_4^+ in the source profile. In other words, there was a linkage
432 effect, variation of some components in the source profile would bring changes to the
433 simulated results of other components.

434 **5 How the variation of source profile adopted in CTMs impact on the simulation** 435 **of chemical components in $\text{PM}_{2.5}$?**

436 The variation of species allocation in emission sources directly affected the
437 composition of aerosol system in CTMs. In CMAQv5.0.2, the aerosol thermodynamic
438 equilibrium process was carried out according to ISORROPIA II, including a SO_4^{2-} -
439 NO_3^- - Cl^- - NH_4^+ - Na^+ - K^+ - Mg^{2+} - Ca^{2+} - H_2O system which was established on the basis of
440 ISORROPIA I by adding the effects of K^+ , Ca^{2+} and Mg^{2+} (Detailed equilibrium
441 relations were shown in Table S22). Some assumptions had been made in the
442 ISORROPIA model to simplify the simulation system (Fountoukis and Nenes, 2007):
443 (1) Because the vapor pressure of sulfuric acid and metal salts (such as Na^+ , Ca^{2+} , K^+ ,
444 Mg^{2+}) were very low, it was assumed that all the sulfuric acid and metal salts in the
445 system existed in the aerosol phase; (2) For ammonia in the system, it was preferred to
446 have an irreversible reaction with sulfuric acid to produce ammonium sulfate. Only
447 when there was still surplus NH_3 after the neutralization of H_2SO_4 , can it have a
448 reversible reaction with HNO_3 and HCl to produce NH_4NO_3 and NH_4Cl . (3) For sulfuric
449 acid in the system, if there were metal ions (such as Ca^{2+} , Mg^{2+} , K^+ , Na^+) in the system,
450 sulfuric acid would react with metal ions to produce metal salts. Only in the case of

451 insufficient sodium, sulfuric acid would react with ammonia. Based on these
 452 assumptions, the ISORROPIA model introduced the following three judgment
 453 parameters (R_1 , R_2 and R_3 were calculated by the following formulas) to determine the
 454 simulation subsystems. In this paper, R_1 , R_2 , R_3 and the corresponding solid phase
 455 species under different perturbation cases on source profiles were shown in Table 3.
 456 These components achieved thermodynamic equilibrium in the order of preference for
 457 more stable salts, obviously, the simulation processes of these components may
 458 influence each other.

459
$$R_1 = \frac{[\text{NH}_4^+] + [\text{Ca}^{2+}] + [\text{K}^+] + [\text{Mg}^{2+}] + [\text{Na}^+]}{[\text{SO}_4^{2-}]} \dots\dots\dots (3)$$

460
$$R_2 = \frac{[\text{Ca}^{2+}] + [\text{K}^+] + [\text{Mg}^{2+}] + [\text{Na}^+]}{[\text{SO}_4^{2-}]} \dots\dots\dots (4)$$

461
$$R_3 = \frac{[\text{Ca}^{2+}] + [\text{K}^+] + [\text{Mg}^{2+}]}{[\text{SO}_4^{2-}]} \dots\dots\dots (5)$$

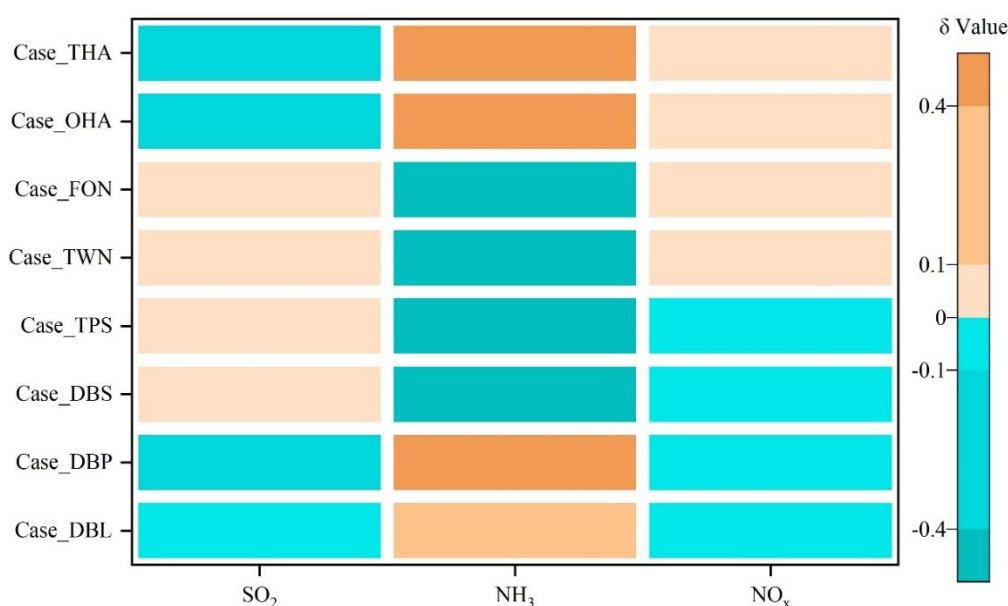
462 Table 2 Potential aerosol species in ISORROPIA II under different cases

Cases	R_1	R_2	R_3	Solid phase species*
SGL、DBL TWN、FON	2.53	2.52	1.9	CaSO ₄ , MgSO ₄ , K ₂ SO ₄ , Na ₂ SO ₄ , NaCl, NaNO ₃ , NH ₄ Cl, NH ₄ NO ₃
DBS	1.26	1.26	0.95	CaSO ₄ , MgSO ₄ , K ₂ SO ₄ , KHSO ₄ , Na ₂ SO ₄ , NaHSO ₄ , (NH ₄) ₂ SO ₄ , NH ₄ HSO ₄ , (NH ₄) ₃ H(SO ₄) ₂
TPS	0.84	0.84	0.63	CaSO ₄ , KHSO ₄ , NaHSO ₄ , NH ₄ HSO ₄
DBP	5.04	5.03	3.79	CaSO ₄ , MgSO ₄ , K ₂ SO ₄ , CaCl ₂ , Ca(NO ₃) ₂ , MgCl ₂ ,
OHA	3.58	2.52	2.95	Mg(NO ₃) ₂ , KCl, KNO ₃ , NaCl, NaNO ₃ , NH ₄ Cl,
THA	4.64	2.52	4.02	NH ₄ NO ₃

463 * The solid phase species were determined based on the research of (Fountoukis and Nenes, 2007)

464 In Non-SNA perturbation case, when the percentage of Non-SNA in source profile
 465 doubled (Case DBP), meant there were more Na, K, Mg, Ca, Cl participated in aerosol
 466 chemistry, the model system needed more SO₄²⁻ and NO₃⁻ on the basis of charge balance
 467 and the thermodynamic equilibrium shifted to the direction of consuming Ca Mg, K
 468 and Na, which resulted in the increase of the simulated concentration of SO₄²⁻ and NO₃⁻.
 469 Meanwhile, according to the rule of anions preferentially binding with nonvolatile
 470 cations in ISORROPIA, the increased cations Na⁺, K⁺, Mg²⁺, Ca²⁺ directly led to

471 the decrease of anions binding with NH_4^+ , there were less reaction dose between SO_4^{2-}
 472 and NH_4^+ to form $(\text{NH}_4)_2\text{SO}_4$ or NH_4HSO_4 , ultimately resulted in a decrease in
 473 simulated concentration of NH_4^+ when compared to the base case. Because in this case
 474 more anions such as SO_4^{2-} were passively needed, according to the principle of chemical
 475 equilibrium mentioned above, the chemical conversion of SO_2 to SO_4^{2-} was promoted,
 476 the simulated secondary SO_4^{2-} increased, this could be proved by that the δ of SO_2 in
 477 Case DBP was negative (shown in Fig. 9, details of other monitoring stations were
 478 shown Table S24).



479
 480 Fig.9 The sensitivity coefficients (δ) of simulated gas pollutants to the change of adopted source
 481 profile in different cases.

482 Similarly, with the increase of metal ions in the system to bond with anions, the
 483 number of anions which can bind to NH_4^+ decreased. The system needed less NH_4^+ and
 484 weakened the need for conversion from NH_3 to NH_4^+ , the simulated NH_4^+ concentration
 485 decreased while the δ of NH_3 was positive and very sensitive. Different trends of
 486 simulated concentration of gaseous pollutants mirrored the rules mentioned above from
 487 another aspect. The δ of SO_2 and NO_x was negative, NH_3 was positive. We could see
 488 the same phenomena in DBL case (Fig. 9). When the percentages of Non-SNA in source
 489 profile increased, they not only affected the simulated concentration of Non-SNA, but
 490 also the secondary SO_4^{2-} , NO_3^- and NH_4^+ .

491 In SO_4^{2-} perturbation cases (Case DBS and TPS), as the percentage of SO_4^{2-} in

492 source profile increased, for the chemical reactions of sulfate radical consuming (as
493 shown in Table S22), the chemical equilibrium would move toward the products when
494 compared to the base case. While for the chemical reactions of sulfate radical formation
495 (The equations were shown in Table S23), meant the product was added in, the chemical
496 equilibrium would be pushed toward the reactants. The chemical reactions between
497 SO_4^{2-} and NH_4^+ would shift to the direction of $(\text{NH}_4)_2\text{SO}_4$ generation, we could see the
498 simulated concentrations of NH_4^+ in DBS and TPS were both higher and NH_3 were
499 lower than those in the base case (SGL). In addition, when more SO_4^{2-} was added in,
500 the conversion of SO_2 to SO_4^{2-} was affected in some level and consumed less SO_2 than
501 the base case, simulated SO_2 showed insensitive but positive trend (Fig.9). And from
502 the potential solid phase species in ISORROPIA II under DBS and TPS cases (Table 3),
503 the solid phase species were mainly consisted of sulfate salts, so the simulated
504 concentration of NO_3^- did not change apparently.

505 As the percentage of NO_3^- in source profile increased (Case FON and TWN), the
506 associated chemical equilibrium shifted towards the consumption of NO_3^- , such as NH_4^+
507 $+ \text{NO}_3^- \rightarrow \text{NH}_4\text{NO}_3$, which would also consume more NH_4^+ and form more ammonium
508 salt, finally consumed more NH_3 because of $\text{NH}_3(\text{gas}) + \text{H}_2\text{O}(\text{aq}) \rightarrow \text{NH}_4^+(\text{aq}) + \text{OH}^-$
509 (aq). The simulation results also manifested that the concentration of NH_4^+ increased
510 while that of NH_3 decreased. Based on the assumption of ISORROPIA, the cations like
511 Na^+ , K^+ , Mg^{2+} , Ca^{2+} and NH_4^+ preferentially to react with SO_4^{2-} , only if there were
512 cations left after neutralized SO_4^{2-} , could they react with NO_3^- to form salts, so the
513 simulated concentration of SO_4^{2-} was not obviously changed. Accordingly, the
514 simulated concentration of NO_x and SO_2 almost unchanged (The δ of NO_x and SO_2 was
515 insensitive).

516 In the cases of NH_4^+ perturbation (Case OHA and THA), when the percentage of
517 NH_4^+ in source profile increased, the related chemical equilibrium shifted towards the
518 direction of NH_4^+ consumption, such as in $2\text{NH}_4^+ + \text{SO}_4^{2-} \rightarrow (\text{NH}_4)_2\text{SO}_4$, more SO_4^{2-}
519 was consumed at the same time, which further promoted the conversion of SO_2 to SO_4^{2-} .
520 The increased NH_4^+ in OHA and THA also would inhibit the conversion of NH_3 to NH_4^+

521 when compared to the base case. This, in turn appeared as the increase of the simulated
522 secondary SO_4^{2-} and NH_3 , and the decrease of the simulated SO_2 .

523 In summary, the effects of source profile variation on the simulation results of
524 different components were linked. When the percentages of Non-SNA, SO_4^{2-} , NO_3^- and
525 NH_4^+ in the source profile changed, they not only affected the simulated concentration
526 of themselves, but also affected the simulation results of some other components. Both
527 the simulation results of primary components and secondary components were affected
528 by the change of source profile, the secondary SO_4^{2-} and NH_4^+ were affected more than
529 the secondary NO_3^- .

530 **6 Conclusions**

531 Although the influence of source profile variation on the simulated concentration
532 of ambient $\text{PM}_{2.5}$ is not significant, its influence on the simulated chemical components
533 cannot be ignored. The variation of simulated components ranges from 8% to 167%
534 under selected different source profiles, and the simulation results of some components
535 are sensitive to the adopted $\text{PM}_{2.5}$ source profile in CTMs, e.g., both the simulated Non-
536 SNA and SNA are sensitive to the perturbation of Non-SNA in source profile, the
537 simulated SO_4^{2-} and NH_4^+ are sensitive to the perturbation of SO_4^{2-} , simulated NO_3^- and
538 NH_4^+ are sensitive to the perturbation of NO_3^- , SO_4^{2-} and NH_4^+ are sensitive to the
539 perturbation of NH_4^+ . These influences are not only specific to an individual component,
540 but also can be transmitted and linked among components, that is, the influence path is
541 connected to chemical mechanisms in the model since the variation of species allocation
542 in emission sources directly affect the thermodynamic equilibrium system
543 (ISORROPIA II, SO_4^{2-} - NO_3^- - Cl^- - NH_4^+ - Na^+ - K^+ - Mg^{2+} - Ca^{2+} - H_2O system).

544 Traditionally, the source profiles are regarded as a primary emission, but
545 interestingly, their variation could affect the simulation result of secondary components
546 as well in CTMs. We found the perturbation of $\text{PM}_{2.5}$ source profile caused the variation
547 of simulated gaseous pollutants, and related chemical reactions like gas-phase
548 chemistry of SO_2 , NO_x and NH_3 , which mirrored that the perturbation of source profile

549 had an effect on the simulation of secondary PM_{2.5} components. Overall, the emission
550 source profile used in CTMs is one of the important factors affecting the simulation
551 results of PM_{2.5} chemical components. Additionally, organic species are one of the most
552 important components in PM_{2.5} and gain much more attention on human health. While
553 the number of organic species in source profile is relatively scarce which brings a
554 challenge for simulation test designing, the variation of source profile adopted in CTMs
555 has an impact on the simulation of organic species is not taken into account in this study.

556 With the change of fuel and raw materials, the development of production
557 technology and the innovation of pollution treatment technology in recent years, some
558 components have changed significantly in the source profile. Given the important role
559 of air quality simulation in environment management and health risk assessment, the
560 representativeness and timeliness of the source profile should be considered.

561 Our study tentatively discussed the impact mechanism of emission source profiles
562 on PM_{2.5} components simulation results in CTMs. In the next work, we will use
563 different source profile for simulation, compare the simulation results with local
564 measured PM_{2.5} components and discuss the influence of sub-source profiles variation
565 on the simulation results. In addition, the size distribution, mixing state, aging and
566 solubility for different aerosol components might have something to do with source
567 profile, how much the influence of source profile changes on these physical and
568 chemical process, is deserved to do in the future.

569 **Data availability**

570 The input datasets for WRF simulation are available at
571 <https://rda.ucar.edu/datasets/ds351.0/index.html> (The National Center for Atmospheric
572 Research (NCAR)). The Multi-resolution Emission Inventory for China (MEICv1.3) is
573 available at http://meicmodel.org/?page_id=135. The PM_{2.5} emission source profiles
574 from database of Source Profiles of Air Pollution (SPAP)
575 (<http://www.nkspap.com:9091/>, Nankai university), SPECIATE database
576 (<https://www.epa.gov/air-emissions-modeling/speciate>, U.S. Environmental Protection

577 Agency's (EPA)), Mendeley data repository (<https://doi.org/10.17632/x8dfshjt9j.2>, Bi
578 et al., 2019).

579 **Code availability**

580 The source code for CMAQ version 5.0.2 is available at
581 <https://github.com/USEPA/CMAQ/tree/5.0.2> (last access: April 2014)
582 (<https://doi.org/10.5281/zenodo.1079898>, US EPA Office of Research and
583 Development, 2018). The source code for WRF version 3.7.1 is available at
584 ~~<https://github.com/NCAR/WRFV3> (last access: 14 August 2015, NCAR)~~
585 <https://www2.mmm.ucar.edu/wrf/src/WRFV3.7.1.TAR.gz>.

586 **Author contributions**

587 Zhongwei Luo: Data curation and collection, writing—original draft. Yan Han:
588 Modeling, writing—original draft. Kun Hua: Data collection. Yufen Zhang:
589 Supervision—Review & editing. Jianhui Wu: Supervision in source profile. Xiaohui Bi:
590 Supervision in source profile. Qili Dai: Resources. Baoshuang Liu: Resources. Yang
591 Chen: Modification and editing. Xin Long: Supervision in modeling. Yinchang Feng:
592 Supervision—Review & editing.

593 **Competing interests**

594 The authors declare that they have no known competing financial interests or
595 personal relationships that could have appeared to influence the work reported in this
596 paper.

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