

Seasonal study of the Small-Scale Variability of Dissolved Methane in the western Kiel Bight (Baltic Sea) during the European Heatwave in 2018

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Abstract

15 Methane (CH₄) is a climate-relevant atmospheric trace gas which is emitted to the atmosphere from coastal areas such as the Baltic Sea. The oceanic CH₄ emission estimates are still associated with a high degree of uncertainty partly because the temporal and spatial variability of the CH₄ distribution in the ocean surface layer is usually not known. In order to determine the small-scale variability of dissolved CH₄ we set up a purge-and-trap system
20 with a significantly improved precision for the CH₄ concentration measurements compared to static headspace equilibration measurements. We measured the distribution of dissolved CH₄ in the water column of the western Kiel Bight and Eckernförde Bay in June and September 2018. The top 1 m was sampled in high-resolution to determine potential small-scale CH₄ concentration gradients within the mixed layer. CH₄ concentrations throughout the
25 water column of the western Kiel Bight and Eckernförde Bay were generally higher in September than in June. The increase of the CH₄ concentrations in the bottom water was accompanied by a strong decrease in O₂ concentrations which led to anoxic conditions favorable for microbial CH₄ production in September. In summer 2018, northwestern Europe experienced a pronounced heatwave. However, we found no relationship between the
30 anomalies of water temperature and excess CH₄ in both the surface and the bottom layer at the site of the Boknis Eck Time-Series Station (Eckernförde Bay). Therefore, the 2018 European heatwave most likely did not affect the observed increase of the CH₄ concentrations in the western Kiel Bight from June to September 2018. The high-resolution

measurements of the CH₄ concentrations in the upper 1 m of the water column were highly variable and showed no uniform decreasing or increasing gradients with water depth. Overall, our results show that the CH₄ distribution in the water column of the western Kiel Bight and Eckernförde Bay is strongly affected by both large-scale temporal (i.e. seasonal) and small-scale spatial variabilities which need to be considered when quantifying the exchange of CH₄ across the ocean/atmosphere interface.

1. Introduction

Methane (CH₄) is an important atmospheric greenhouse gas that is produced in open ocean and coastal environments (see e.g. Reeburgh, 2007; Wilson et al., 2020). Oceanic CH₄ emissions depend on the interplay of various biogeochemical, oceanographic and biological factors that drive production, consumption and transport processes of CH₄ (see e.g. Bakker et al., 2014). Oceanic CH₄ can be either of geologic or biological origin (see e.g. Bakker et al., 2014; Reeburgh, 2007; Wilson et al., 2020). A photochemical production of CH₄ in oxic surface layers of the coastal and open oceans was suggested only recently as an alternative, non-biological, production pathway (Li et al., 2020). Generally, open ocean surface waters are at atmospheric equilibrium or slightly oversaturated (Weber et al., 2019). Oceanic emissions (incl. open and coastal areas) contribute only ~1-3% to the global CH₄ budget (Saunois et al., 2020). Coastal areas including shelves and estuaries account for up to 75% of total CH₄ oceanic emissions to the atmosphere (Weber et al., 2019). However, large uncertainties remain regarding the temporal and spatial variability of CH₄ concentrations and the CH₄ emissions to the atmosphere (see e.g. Weber et al. 2019). Moreover, temporary extreme events such as warming of the upper ocean due to heatwaves can affect the dissolved CH₄ concentrations and its emissions (Humborg et al., 2019; Borges et al., 2019). To this end, we present here a seasonal study of dissolved CH₄ gradients in the western Kiel Bight (incl. Eckernförde Bay) during the European heatwave in 2018. The overarching objectives of this study were to (i) to set up a CH₄ measurement system with a precision that allows detection of small-scale variability of CH₄ concentrations, (ii) decipher the small-scale variability of dissolved CH₄ in the upper water column on a seasonal basis, (iii) assess how extreme events such as the European heatwave in 2018 might affect the CH₄ concentrations and (iv) determine the consequences of CH₄ concentration gradients for the CH₄ emissions to the atmosphere.

1.1 Study Site

The western Kiel Bight and the Eckernförde Bay are affected by the inflow of water along the bottom, from the North Sea through the Kattegat and the Great Belt, and also the surface

outflow of brackish water (Bange et al., 2010; Bange et al., 2011; Lennartz et al., 2014; Steinle et al., 2017); this results in strong fluctuations in bottom water salinity, between 17 and 24 (Lennartz et al., 2014). Complete vertical mixing of the water column is prevented from March to September, as a strong pycnocline develops due to the surface warming and the distinct salinity gradient between inflowing and outflowing water masses. During the winter months, the whole water column is mixed as a consequence of storms and surface-water cooling (Bange et al., 2010). Strong phytoplankton blooms in early spring (February–March) and autumn (September–November) are followed by high rates of organic matter sedimentation and microbial respiration, and thus also the consumption of O₂ (Bange et al., 2010; Dale et al., 2011). Consequently, pronounced hypoxia and sporadic anoxia occur in the bottom waters during late summer (Bange et al., 2010; Lennartz et al., 2014; Steinle et al., 2017). The occurrence of anoxic events has been continuously increasing in frequency since the 1970s (Lennartz et al., 2014). The high sedimentation rates of organic matter favour methanogenesis in the muddy sediment, and CH₄ release to the water column, resulting in high CH₄ concentrations in the overlying water (Bange et al., 2010; Ma et al., 2019).

2. Methods

Here, we present CH₄ measurements from two research cruises with the R/V Alkor as part of the Baltic GasEx experiment in 2018. The cruises AL510 (Booge, 2018a) and AL516 (Booge, 2018b) took place in June 2018 and September 2018, respectively. CH₄ samples were taken from 9 (AL510, June) and 10 (AL516, September) CTD rosette casts (Figure 1). The Baltic GasEx experiment was as a dual-tracer experiment to investigate the air-sea gas exchange in the western Kiel Bight (Ho et al., 2019). To this end, the cruise tracks emerged from following the patch of a surface water mass marked with a pair of tracers (³He/SF₆) which were released at the start of each campaign and led to a high spatial resolution coverage of the western Kiel Bight. To study the vertical CH₄ distribution within the water column, samples were taken from the mixed layer, from within the pycnocline, and from the water below the pycnocline (Booge, 2018a, b). The CTD was mounted to the rosette water sampler with twelve 10 L Niskin bottles that were closed during the upcast at the requested sampling depths.

To examine potential CH₄ concentration gradients in the very near-surface waters, additional samples from a zodiac were taken at selected stations. To avoid turbulence distributions caused by the ship, the sea water samples were taken at some distance from the ship. Using a self-built sampling device that consisted of an aquarium pump attached to a floating board, water from 0.1, 0.5 and 1 m depth below the surface was pumped on board the zodiac as

described in detail by Fischer et al. (2019). In addition, discrete underway (UW) samples were taken from the ship's continuous seawater supply system with a water inlet at ~2 m depth.

Triplicate water samples were taken through a silicon hose connected to the Niskin bottle (CTD samples), aquarium pump (zodiac samples), or the ship's underway system (UW samples). At a low flow-rate, 20 mL amber glass vials were filled bubble-free by overflowing the approximate threefold volume of seawater. The vials were closed with butyl rubber stoppers and crimp-sealed with aluminium caps. To inhibit microbial activity, 50 μ L of saturated mercury chloride solution (HgCl_2 (aq)) were added to each sample. To compensate for the added volume of HgCl_2 solution, a needle with a 3 mL syringe body was inserted into each sample before HgCl_2 injection. The samples were stored at room temperature in the dark until the measurements were carried out.

2.1 Purge and Trap System

We used a self-built purge and trap (PT) system coupled to a gas chromatograph equipped with a flame ionization detector (GC-FID) to measure CH_4 in the surface water and the water column. The set up (Figure 2) of the PT measurement system (Figure 2A) can be divided into three sections that describe the purge unit (Figure 2B), the trapping unit (Figure 2C), and the GC-FID system (Figure 2D). The materials that were used and all performance tests that were carried out are described in detail elsewhere (Gindorf, 2020). Helium (He) is used as the purge gas and as the carrier gas for the (GC). The gas stream is split and directed through the purge unit while a continuous gas stream through the GC is maintained. A digital thermometer is installed next to the system to monitor the temperature.

The purge unit contains a 2-position 4-way-valve that can be switched to enable the purging of the sample ('purge' mode) or the emptying of the purge chamber ('waste' mode). In the purge mode, the He gas stream is directed through the sample vial and consecutively through the purge chamber. The sample water is pushed into the purge chamber when the He gas stream is turned on. A long needle that reaches the bottom of the sample vial is used to completely empty the vial. Backflushing of the water into the sample vial is restricted by two check valves at the hose between the purge chamber and the sample. The purge chamber consists of a 50 mL sample vial that is placed upside down to minimize leakage through the stopper. The purge flow is directed into the purge chamber through a short needle with the needle tip placed close to the stopper to ensure the purging of the entire water sample. When the He gas is bubbling through the sample, the dissolved gases are stripped from the water phase. Due to its low solubility in seawater at room temperature and normal pressure (Duan et al., 1992), CH_4 is stripped out within 4 minutes using a purge flow

of approximately 0.03 L min^{-1} . The gas is extracted from the purge chamber via a long needle with the needle tip placed close to the bottom of the upside-down vial. The gas is then dried over a Nafion™ dryer with a counterflow of dry compressed air at a flow rate of approximately 200 mL min^{-1} . Additionally, two glass tubes filled with phosphorus pentoxide (Sicapent®; P_2O_5) are used to further dry the gas stream. To ensure a continuous and uniform flow rate during the measurements, a flowmeter is installed before the gas is transferred to the trapping unit.

The He gas stream from the purge unit as well as the carrier gas stream of the GC are connected to a two-position-six-port valve that enables the switching of the gas stream through the CH_4 trap. In the 'trap' position, the purge gas is conducted through the trap, which consists of a $20 \text{ cm} \times 1/8''$ stainless steel column filled with Spherocarb (100-200 mesh). The trap is put into liquid nitrogen during the trapping procedure. In the 'desorb' position, the GC carrier gas flow is conducted through the trap, which is subsequently removed from the liquid nitrogen and put into a water bath of $\sim 90^\circ \text{C}$ to desorb the trapped gases from the column. The gas flow is then directed through the GC-FID system (D).

(D) The GC-FID used in this setup is an HP 5890 II GC equipped with an FID detector. The gases are separated over a $1/8'' \text{ } 6'$ stainless steel column, with a carrier gas flow of $\sim 30 \text{ mL min}^{-1}$. A similar flow rate of the carrier gas streams during the trapping and the desorption steps was chosen to avoid baseline shifts when switching the Valco valve.

2.2 Calibration

On each measurement day, a set of standard gas mixtures and blank measurements with and without the injection of 20 mL He was measured prior to sample measurements. The gas standards have been calibrated against two National Oceanic and Atmospheric Administration (NOAA) primary standards that were provided by the SCOR Working Group for the intercomparison of oceanic CH_4 and N_2O measurements (Wilson et al., 2018). Prior to standard measurements, one sea water sample was purged on every measurement day. This water was left inside the purge chamber and used for the blank and standard measurements. For the standard and He blank injection, 20 mL plastic syringes were used. A Safeflow® infusion valve was attached to the check valve at the standard injection port to reduce the dead volume of the injection port. After each standard injection, 3 mL of He was injected with a 3 mL plastic syringe through the port to ensure that all injected volume of the standard is injected into the purge chamber. The He flow through the purge unit is switched off during the standard injection. Different volumes of the standard were injected into the purge system to create a calibration curve that covered the full concentration range of the samples. The amount of the injected standards was calculated from the respective mole

fraction and the injected volume using the ideal gas law. The chromatography software Chromstar 6.3 (SCPA GmbH, Weyhe-Leeste, Germany) was used for data acquisition and manual integration of the CH₄ peaks.

2.3 Comparison of Static Headspace Equilibrium and Purge and Trap

To ensure that the PT measurements are comparable with measurements with the previously used static headspace equilibration (HS) method (see e.g. Ma et al., 2020), triplicates of seawater samples from the six standard depths were taken for PT and HS during a Boknis Eck cruise. The CH₄ concentrations ranged from 5 to 222 nmol L⁻¹ allowing a comparison over a broad concentration range (Figure 3). Over all depths, the PT measured concentrations were slightly lower and showed significantly less variation among the triplicates, thereby reflecting a better precision of the PT measurements over the HS method (Table 1). The direct comparison of both techniques shows that the measurements agree well with the HS measurements (Figure 4). Other studies have proven higher precision and sensitivity as well as handling benefits of PT over HS (e.g. Capelle et al., 2015).

2.4 Calculation of Dissolved CH₄ Concentrations

The CH₄ concentration in the sample was calculated from the linear calibration fit using equation (1). The mean area of the blank measurements was subtracted from the sample peak area to account for the background contamination of the system.

$$n = \frac{PA_{sample} - PA_{blank}}{\delta} (1),$$

where

n is the amount of CH₄ [nmol] in the sample,

PA_{Sample} is the peak area of the measured sample,

PA_{Blank} is the mean peak area of the measured blanks,

δ is the slope of the calibration curve [nmol⁻¹].

The CH₄ concentration c [nmol L⁻¹] was calculated as the ratio between n and the sample volume V [L]. V was determined experimentally to be $0.0203 \pm .0002$ L.

We estimated the standard deviation for triplicates or duplicates according to the statistical analysis of David (1951). The mean analytical error of the CH₄ concentration was +/- 5.7 and 3.1 % during the June and ALSeptember cruise, respectively.

2.5 Δ CH₄ and CH₄ Saturations

Temperature and salinity data from the ship's thermosalinograph were used to compute the excess of CH₄ (Δ CH₄) as the difference between the measured CH₄ concentration (c , see

above) and the CH₄ equilibrium concentration (c_{eq}). The solubility equation for CH₄ in
 205 seawater (Wiesenburg and Guinasso, 1979) was applied to calculate c_{eq} (in nmol L⁻¹). The
 atmospheric dry mole fraction of CH₄ was taken from records of the Mace Head observatory
 in Ireland, which is part of the Advanced Global Atmospheric Gases Experiment (AGAGE,
<https://agage.mit.edu/>): The monthly means of atmospheric CH₄ were 1918.24 ppb ± 0.2%
 for June and 1925.83 ppb ± 0.2% for September 2018.

210 CH₄ saturations (CH₄sat in %) were computed as

$$CH_4sat = 100 * c/c_{eq} \text{ (2).}$$

2.6 Oxygen Measurements

During both cruises, a CTD-mounted altimeter (sn#453) oxygen sensor was used to obtain
 CTD-O₂ profiles from the surface to 1 m above the bottom. Additionally, 112 and 105 discrete
 215 oxygen samples were Winkler titrated during the June and September cruise, respectively,
 and were used to calibrate the sensor data.

2.7 Temperature and CH₄ Anomalies

The measurements at the Boknis Eck Time-Series Site performed during this study allows us
 determine the effect of the European heatwave in 2018 in the context of the monthly time-
 220 series measurements of water temperature and dissolved CH₄ at Boknis Eck (Lennartz et al.,
 2014; Ma et al. 2020). To this end, we computed the anomalies of water temperature and
 ΔCH_4 in 1 m and 25 m water depth for the period of January 2006 to December 2018. The
 anomalies were defined as

$$\Delta T = T - T_{i, avg} \text{ (3) and}$$

$$\Delta (\Delta CH_4) = (\Delta CH_4) - (\Delta CH_4)_{i, avg} \text{ (4),}$$

where T is the measured monthly water temperature (T) in the period January 2006 to
 December 2018 in 1 (25) m depth at Boknis Eck (Lennartz et al., 2014). $T_{i, avg}$ is the mean
 water temperature in 1 (25) m depth of the respective month i over this period at Boknis Eck.
 The resulting ΔT is the anomaly of the water temperature which is cleaned from seasonal
 230 differences throughout each year. $\Delta (\Delta CH_4)$ is calculated similarly to ΔT in the same time
 period using ΔCH_4 which is the monthly excess CH₄ (ΔCH_4 , see above) in 1 (25) m depth.
 $(\Delta CH_4)_{i, avg}$ is the mean excess CH₄ in 1 (25) m depth of the respective month i over this
 period at Boknis Eck. ΔCH_4 was computed as the difference of the monthly measurements of
 dissolved CH₄ in 1 (25) m water depth (Ma et al., 2020) and the monthly c_{eq} (see Section 2.5)
 235 in 1 (25) m water depth. C_{eq} was calculated with the monthly water temperature and salinity
 at Boknis Eck in 1 (25) m depth (Lennartz et al., 2014) and the monthly atmospheric CH₄ dry

mole fractions measured at Mace Head (see Section 2.5) from January 2006 to December 2018. Extremely high CH₄ surface concentrations with concentrations in the range from 87 to 689 nmol L⁻¹ have been measured in four months (November 2013, February/March 2014 and December 2014) and were, thus, omitted from the data set to avoid a statistical bias of the data set.

3. Results and Discussion

3.1 June 2018

The hydrographic conditions during the June cruise revealed a strong stratification of the water column: while the upper ~10 m were comparably uniform, a strong gradient in temperature prevailed between 10 and 15 m, and the lower water column (15-20 m) showed a strong salinity gradient (Figure 4). This is in line with former studies in the respective area (e.g. Bange et al., 2010; Dale et al., 2011; Lennartz et al., 2014; Ma et al., 2019, 2020).

Between June 9 and 13, surface water temperatures exceeded 20 °C, which is higher than the average surface temperature measured at the Boknis Eck station in June (Lennartz et al., 2014).

The highest O₂ concentrations were measured between approximately 7 m and 18 m depth and between the 1009 and the 1012 kg m⁻³ isopycnals. At the beginning of the measurements, the O₂ maximum had a vertical extension of more than 10 m, which decreased continuously in its thickness and intensity until it vanished after the 15th of June. The whole water column was oxygenated throughout the cruise, with oxygen concentrations decreasing to ~120 µmol L⁻¹ in the bottom waters. At this time of the year, the oxygen depletion in the deep water starts to evolve (Lennartz et al., 2014). The water column CH₄ concentrations ranged from 2.8 nmol L⁻¹ (99 % saturation) in the surface waters to 28.3 nmol L⁻¹ (750 %) in the bottom waters.

3.2 September 2018

During the September campaign, the water column also showed a strong stratification below 10 m with pronounced gradients in temperature, salinity and O₂ concentrations (Figure 5). In contrast to the cruise in June, the stratification below 10 m seemed to be primarily driven by the salinity gradient. The surface water showed a comparably homogeneous distribution of about 17 °C down to the 1014 kg m⁻³ isopycnal at ~15 m. A strong temperature decrease was observed below 20 m, with temperatures ~12 °C in the bottom waters. An increased surface density at the beginning and the end of the cruise (before Sept. 14, after Sept. 22)

indicated the upwelling of waters from 5-10 m to the surface. A strong storm event on the 21st could have induced this upwelling (mean wind speed on 21st ~12 m s⁻¹). The bottom water salinity was lower in the beginning than in the end of the cruise and the higher salinities were shifted upward in the water column in agreement with the upshift of the 1016 kg m⁻³ isopycnal.

During the whole cruise, the highest O₂ concentrations were measured in the surface waters above the 1013 kg m⁻³ isopycnal. Between the 1014 and 1015 kg m⁻³ isopycnals a strong oxycline could be observed as O₂ decreased by approximately 100 µmol L⁻¹ within a few meters. With increasing depth, the O₂ concentration decreased to suboxic (O₂ < 5 µmol L⁻¹) and almost anoxic (O₂ ~ 0 µmol L⁻¹) conditions in the bottom water. However, during the September cruise no indication of sulfidic conditions (e.g. the smell of hydrogen sulphide) was observed. Along with the upward shifting isopycnals, hypoxia (O₂ < 60 µmol L⁻¹) characterized almost half of the water column at the end of the cruise. Intense hypoxic and even anoxic conditions in the bottom water have been frequently observed at the Boknis Eck Time Series Station (Eckernförde Bay) between late summer and autumn (e.g. Bange et al., 2010; Dale et al., 2011; Lennartz et al., 2014; Ma et al., 2019, 2020).

CH₄ concentrations ranged from 4.7 nmol L⁻¹ at the surface to 104.0 nmol L⁻¹ in the bottom waters (Figure 5B). In contrast to the O₂, temperature and salinity profiles, the CH₄ distribution in the bottom and intermediate waters showed a larger variability. Strongest accumulation of CH₄ was confined to the bottom waters below 20 m. Although the surface samples revealed a stronger oversaturation of CH₄ than during the June campaign (see Figure 6), the stratification of the water column seemed to be an effective barrier for the CH₄ from the bottom waters reaching the atmosphere.

3.3 CH₄ in the Surface Layer

During both cruises, the surface layer was always oversaturated or close to equilibrium with the atmosphere (Figure 6). ΔCH₄ ranged between ~0 and 6 nmol L⁻¹ during the June cruise and between 2 and 8 nmol L⁻¹ during the September cruise, corresponding to saturations of 103-292 % and 201-366 %, respectively. The near-surface samples of CH₄ from the zodiac and underway measurements revealed that within the top 1 m of the water column, CH₄ concentration gradients existed (see Figure 7), with larger concentration differences between 0.1 and 1 m (0.2-2.7 nmol L⁻¹) than between 1 and 2 m (0.1 -1.8 nmol L⁻¹, mean difference between CH₄ gradients: ~1 nmol L⁻¹) during the September cruise. The direction of the gradients was highly variable, with some stations showing higher CH₄ concentrations in the topmost sample, while others displayed increasing concentrations with depth or intermediate maxima.

Interestingly, the CH₄ concentrations measured from the shallowest Niskin bottles (1-2 m) were generally higher than from the surface samples (average difference between CH₄ concentrations from Niskin bottle and from zodiac samples: $1.2 \pm 0.4 \text{ nmol L}^{-1}$). This could reflect the different sampling conditions, but it may also be a sign of mixing or carry-over effects from the CTD profiling. This can result from closing the Niskin bottles during the upcast, so that deeper waters might be brought up. A comparably large variation between the triplicate samples may result from the challenging sampling conditions on board of the zodiac. However, the majority of the observed gradients is larger than the cumulative uncertainty of the replicate measurements.

The sampling depth for surface water is not uniformly defined in oceanic measurements. While for open ocean CTD sampling depths down to 10 m are recognised as surface samples, for coastal areas often 1 m is considered as the surface depth (e.g. Ma et al., 2020). For continuous UW measurements the sampling depth depends on the vessel's hull and water intake depth which can range between 1 and 10 m depth (e.g. 2-5 m, Becker, 2016; Karlson et al., 2016; Kitidis et al., 2010; Rhee et al., 2009; Zhang et al., 2014). Although the near-surface gradients found in our study do not show a clear direction, our results indicate that at least in coastal areas with elevated CH₄ concentrations, a sampling depth of several meters may not correctly represent the surface CH₄ concentration.

3.4 Seasonal Variability

Please note that when comparing June and September data from figures 4 and 5, different colour schemes were used to emphasize gradients within each cruise. Between June and September, a strong increase in the salinity, along with changes in the vertical temperature distribution, indicated an exchange of the waters in the western Kiel Bight over the entire water column (Figure 8). With the change of the water masses, a clear shift in the CH₄-O₂ relationship from June to September was observed (Figure 9). While only a slight increase in the surface CH₄ concentrations was observed between June and September, much higher bottom CH₄ concentrations and much lower O₂ concentrations were found in September. The co-occurrence of O₂ depletion and CH₄ enrichment in the bottom water agrees with the observations from previous studies (e.g. Bange et al., 2010; Steinle et al., 2017).

The high CH₄ concentration in the bottom water most likely results from methanogenesis in the anoxic sediments (Bange et al., 2010) producing CH₄ that is partly released into the water column (Donis et al., 2017; Reindl and Bolalek, 2014). The summer stratification inhibits the CH₄ from reaching the surface and, thus, CH₄ accumulates below the pycnocline. Within the water column, CH₄ is efficiently oxidized and only a small fraction reaches the surface layer (Steinle et al., 2017). The salinity change between June and September also indicates that

the high CH₄ concentration in the bottom water in September does not result from long-term accumulation of CH₄ in the bottom waters. It is rather the result of recent local CH₄ release either at the BE site itself or advected to the BE site by the bottom waters.

3.5 The 2018 European Heatwave Impact on CH₄

Figure 10 shows the anomalies of T and ΔCH₄ in 1 and 25 m water depth from January 2006 to December 2018. A pronounced temperature anomaly is visible in 1 m depth in August 2018 reflecting the heatwave which occurred from mid-July to August 2018 across northwestern Europe (Kueh and Lin, 2020). However, a signal of the 2018 heatwave is not visible in the temperature anomalies in 25 m depth. The maximum anomaly of ΔCH₄ in 1 m water depth is visible in May 2018 and thus not associated with the heatwave signal of the temperature anomaly in 1 m. The maximum temperature anomaly is found in 1 m water depth for July 2006 and reflects another European heatwave which was experienced by large parts of western and central Europe during July 2006 (Chiriaco et al., 2014). Again, the signal of the 2006 heatwave is not visible in the anomalies of ΔCH₄. Overall, there is no relationship between the water temperature anomalies and ΔCH₄ anomalies in both 1 and 25 m depth.

This finding is in contrast to the results by Borges et al. (2019) who reported significantly enhanced CH₄ surface concentrations in coastal waters of the North Sea off Belgium in July 2018 (Figure 1). They hypothesized that the high dissolved CH₄ surface concentrations might have been caused by a temperature-driven enhancement of both methanogenesis and sedimentary release of CH₄. Humborg et al. (2019) measured dissolved CH₄ surface concentrations in the coastal waters of southern Finland after the heatwave in September 2018 (Figure 1). They concluded that the heatwave caused higher CH₄ emissions to the atmosphere from near shore sites which, in turn, might have been fueled by temperature-driven sedimentary release of CH₄. However, our data do not support a heatwave-driven enhancement of CH₄ concentrations at Boknis Eck (Eckernförde Bay) (see Figure 10). Thus, CH₄ emissions to the atmosphere at Boknis Eck do not seem to be affected by the heatwaves.

The shallow coastal waters off the Belgian coast (water depth < 30 m) are characterized by strong tidal currents which result in a well-mixed water column throughout the year (Borges et al., 2019). This implies that the temperature signal of the 2018 heatwave was most likely conveyed to the sediments and might have led to enhanced release of CH₄ due to enhanced methanogenesis in combination with ebullition/flaring of CH₄ from CH₄-enriched (gassy) sediments (Borges et al., 2019). In contrast, the water column at Boknis Eck was stratified during summer 2018. Therefore, the temperature signal of the heatwave was only detectable

in the surface layer but not in the bottom layer and a potential heatwave-triggered CH₄ release from the sediments was not detected (Figure 10).

In contrast to the measurements at Boknis Eck, the shallow bottom waters (water depth = 31 m) at the Tvärminne Zoological Station (TZS) coastal monitoring station (northeastern Baltic Proper) showed a significant temperature signal in 2018 (Humborg et al., 2019). The enhanced bottom water temperatures, in turn, might have led to an enhanced outgassing of sedimentary CH₄ via gas plumes (Humborg et al., 2019).

On the one hand, a temperature increase should lead to enhanced microbial CH₄ production but, on the other hand, its microbial consumption by aerobic and anaerobic CH₄ oxidation should increase as well in both, the water column and the sediments. At Boknis Eck, for example, methanogenesis and CH₄ oxidation indeed show the same seasonal (i.e. temperature) dependencies (Maltby et al., 2018; Steinle et al., 2017; Treude et al., 2005). This implies that a temperature increase most probably results only in a small increase of the net CH₄ production (= methanogenesis minus CH₄ oxidation). Any significantly enhanced CH₄ concentrations resulting from an increase in temperature might be, therefore, dominated by enhanced ebullition from gassy sediments in shallow coastal waters which are (i) well-mixed (as observed off the Belgian coast; Borges et al., 2019) or which (ii) show a deepening of the mixed layer to allow gas plumes to reach the mixed layer (as observed off southern Finland; Humborg et al., 2019). Despite the fact that CH₄-enriched (gassy) sediments are also found in Eckernförde Bay (Lohrberg et al., 2020), the prerequisites (bottom water temperature anomaly and ebullition) for a heatwave induced enhancement of CH₄ surface concentrations have not been observed at Boknis Eck during our study in 2006 and 2018. Boknis Eck experienced heatwaves after 2018. However, CH₄ concentrations measurements from Boknis Eck after 2018 were not available at the time of writing of this article.

The frequency of higher Δ CH₄ anomalies in 25 m seems to have increased since 2013 (Figure 10). We may, thus, speculate that sedimentary release of CH₄ to the overlying water column may have increased as well which in turn might be caused by the long-term warming trend observed at Boknis Eck (Lennartz et al., 2014).

4. Summary and Conclusions

Our measurements revealed higher CH₄ concentrations in the bottom waters and comparably lower CH₄ concentrations in the surface waters which was caused by a stratification of the water column. This, in turn, prevented upward mixing of CH₄-enriched waters during both campaigns. CH₄ concentrations in the bottom waters were significantly higher in September compared to June and were accompanied by a strong decrease in O₂

concentrations in the bottom water which led to anoxic conditions favorable for microbial CH₄ production in September 2018. The overall setting of the CH₄ water column distribution and the comparably rapid seasonal change in the CH₄ concentrations is in line with the time-series measurements of dissolved CH₄ concentrations at the Boknis Eck Time Series Station in Eckernförde Bay (Ma et al., 2020; Maltby et al., 2018; Steinle et al., 2017).

In summer 2018, northwestern Europe was experiencing a pronounced heatwave which led to significantly enhanced water temperatures in the Baltic and the North Seas. This, in turn, might have triggered enhanced CH₄ production and consequently might have led to enhanced CH₄ concentrations (Borges et al., 2019; Humborg et al., 2019). However, we found no relationship between the anomalies of water temperature and excess CH₄ in both the surface and the bottom layer at Boknis Eck (Eckernförde Bay) for the period 2006 to 2018. We conclude that pronounced European heatwaves which, for example, occurred in 2006 and 2018 did not affect CH₄ concentrations in the Eckernförde Bay. Therefore, the 2018 European heatwave most likely had no effect on the observed increase of the CH₄ concentrations in the western Kiel Bight from June to September 2018.

CH₄ saturation in the surface layer were always >100 % and, thus, the western Kiel Bight and Eckernförde Bay were sources of CH₄ to the atmosphere during both, June and September 2018. This agrees with the fact that the Baltic Sea is a source of atmospheric CH₄ throughout the year (see e.g. Gülzow et al., 2013; Gutiérrez-Loza et al., 2019; Ma et al., 2020). The high-resolution measurements of the CH₄ concentrations in the upper 1 m of the water column were highly variable and showed no uniform decreasing or increasing gradients with water depth. Surface CH₄ concentration measurements used for flux calculations are usually from one depth in the surface layer assuming that there are no concentration gradients and that the CH₄ concentration in the surface layer is uniform. Our results imply that the assumption of a uniform distribution of CH₄ concentrations in the upper surface layer is not justified. Thus, CH₄ flux calculations on the basis of the concentration difference across the ocean/atmosphere interface is associated with a degree of uncertainty when ignoring the CH₄ variability in the upper surface layer (Fischer et al., 2019; Calleja et al., 2013). However, since there were no uniform increasing or decreasing CH₄ gradients, we cannot assess whether CH₄ flux calculations would have been generally under- or overestimated.

Overall, our results show that the CH₄ distribution in the water column of the western Kiel Bight and Eckernförde Bay is strongly affected by both large-scale (i.e. seasonal) and small-scale variabilities. In order to reduce the uncertainties associated with concentration difference-based CH₄ emission estimates, we suggest high-resolution measurements in the upper surface layer on a regular (at least seasonal) basis. This is also in line with a study by

Roth et al. (2022) that recently showed that a high sampling frequency of at least 50 samples per day is needed to resolve the high variability of CH₄ emissions from coastal habitats. This emphasizes the need of a higher resolution of the CH₄ dynamics in coastal environments for future studies.

Author Contributions

AK, DB and HWB designed the study. SG set up the purge-and-trap system and performed the measurements. AK, HWB and SG wrote the manuscript. DB carried out the sampling during both campaigns and contributed to the manuscript.

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Data Availability

The data is available from the MEMENTO database (<https://memento.geomar.de>) and from the PANGAEA database under <https://doi.org/10.1594/PANGAEA.923992> and <https://doi.org/10.1594/PANGAEA.924372> for AL510 (June) and AL516 (September), respectively. Data from the Boknis Eck Time-Series Station are available from www.bokniseck.de.

Competing Interests

The authors declare no competing interests.

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Tables

Table 1: Boknis Eck April 2020 concentrations and deviations measured with PT and HS.

Depth [m]	Purge and Trap			Headspace		
	Mean CH ₄ [nmol L ⁻¹]	Std dev [nmol L ⁻¹]	Std dev [%]	Mean CH ₄ [nmol L ⁻¹]	Std dev [nmol L ⁻¹]	Std dev [%]
1	5.89	0.12	1.99	9.26	0.84	9.06
5	12.97	0.25	1.89	14.99	2.16	14.42
10	24.61	0.56	2.27	26.55	0.63	2.36
15	35.24	0.09	0.25	37.46	0.30	0.80
20	164.17	1.08	0.66	167.57	2.34	1.11
25	202.50	0.37	0.18	212.93	8.91	3.80

Figures

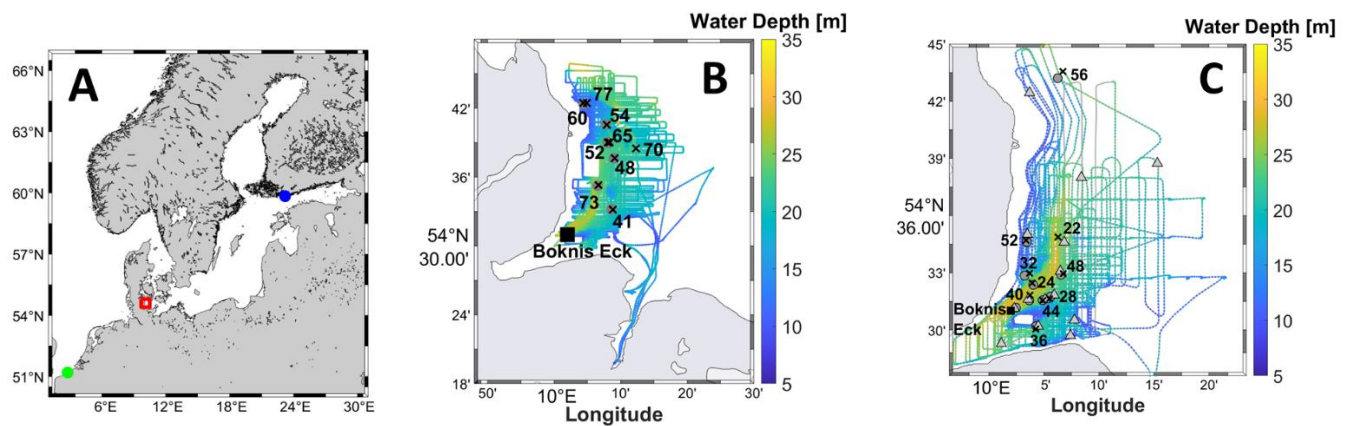


Figure 1: A) Overview map of the Baltic Sea and the southeastern North Sea. The red square marks the research area of this study in the Kiel Bight. The green and blue circles mark the sampling sites of Borges et al. (2019) and Humborg et al. (2019), respectively. B) Cruise track of cruise AL510 in June 2018. C) Cruise track of the cruise AL516 in September 2018. The black crosses mark CTD stations, dark grey dots mark zodiac sampling sites and light grey triangles mark UW sampling sites. The maps were computed using the m_map toolbox in Matlab (Pawlowicz, 2020).

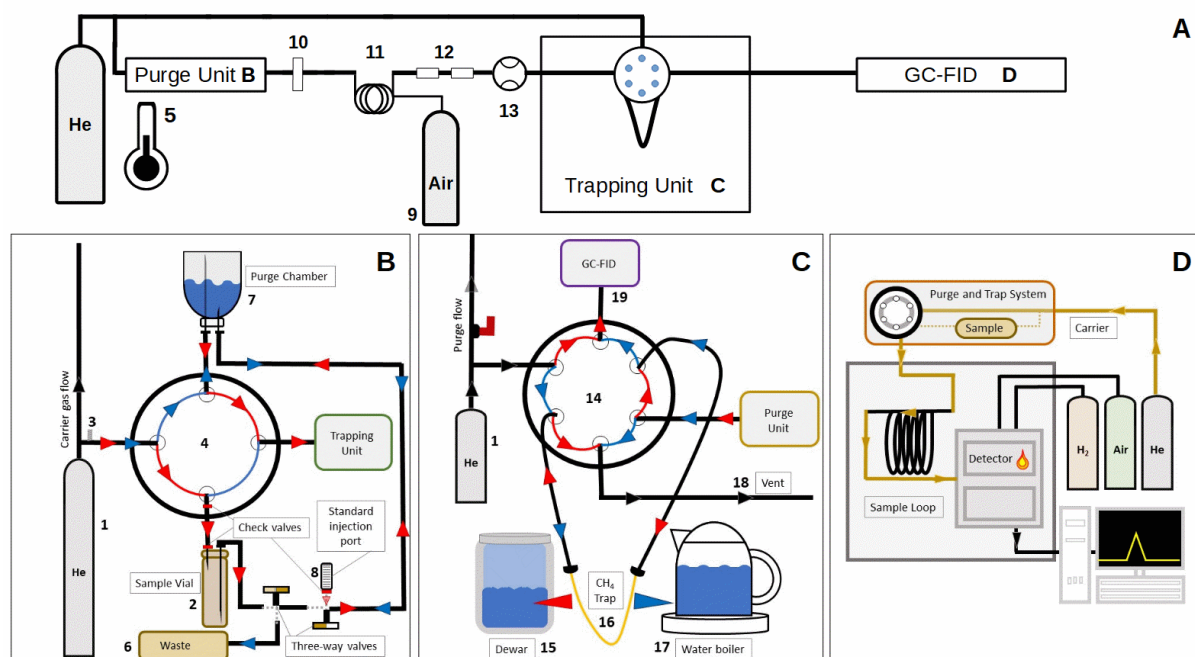


Figure 2: Schematic illustration of the PT system set-up: A) general set-up of the PT system, the components of the purge unit, the trapping unit and the GC-FID are shown in detail in (A-C); B) set-up of the purge unit showing the direction of the He gas stream when samples are purged (red) and when the purge chamber is emptied (blue); (C) set-up of the trapping unit showing the direction of the He gas stream when the gas is trapped (red) and desorbed (blue); (D) set-up of the GC-FID system. The individual components of the PT system are: 1) He gas cylinder with pressure regulator; 2) sample vial; 3) needle valve; 4) four-port valve; 5) thermometer; 6) double-walled wastewater pipe and wastewater canister; 7) purge chamber; 8) Luer Lock injection port with check valve and Safeflow® infusion valve; 9) compressed air cylinder with pressure regulator; 10) filter; 11) Nafion® counterflow drying tube; 12) two glass dry traps filled with P₂O₅; 13) flowmeter; 14) VICI Valco® six-port valve; 15) Dewar tank filled with liquid nitrogen; 16) CH₄ trap filled with molecular sieve (5Å); 17) water boiler; 18) vent; 19) connection to GC-FID.

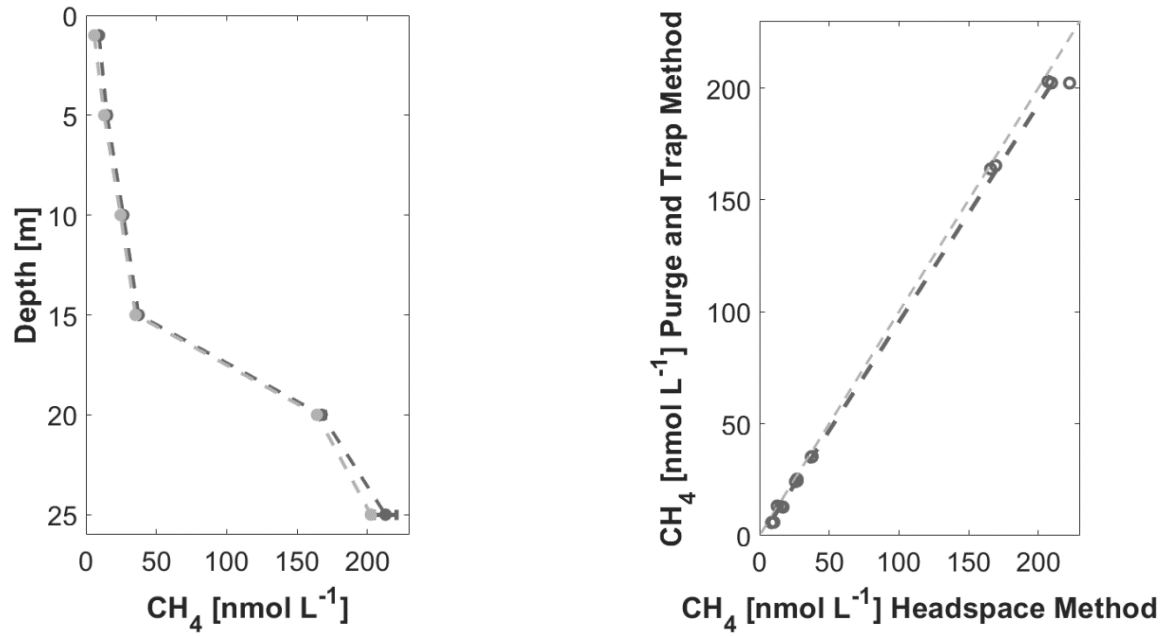


Figure 3: Boknis Eck depth profile of CH₄ measured with PT (light grey) and HS (dark grey) in April 2020 (left). Means are shown as filled dots and dashed line and standard deviation is displayed as error bars in the respective colors. Linear regression of CH₄ concentrations measured with PT against HS for samples from Boknis Eck in April 2020 (right). The light grey dashed line indicates the 1:1 relation. $Y = -1.59 + 0.97 \cdot X$. $R^2 = 0.999$; $p < 0.0001$.

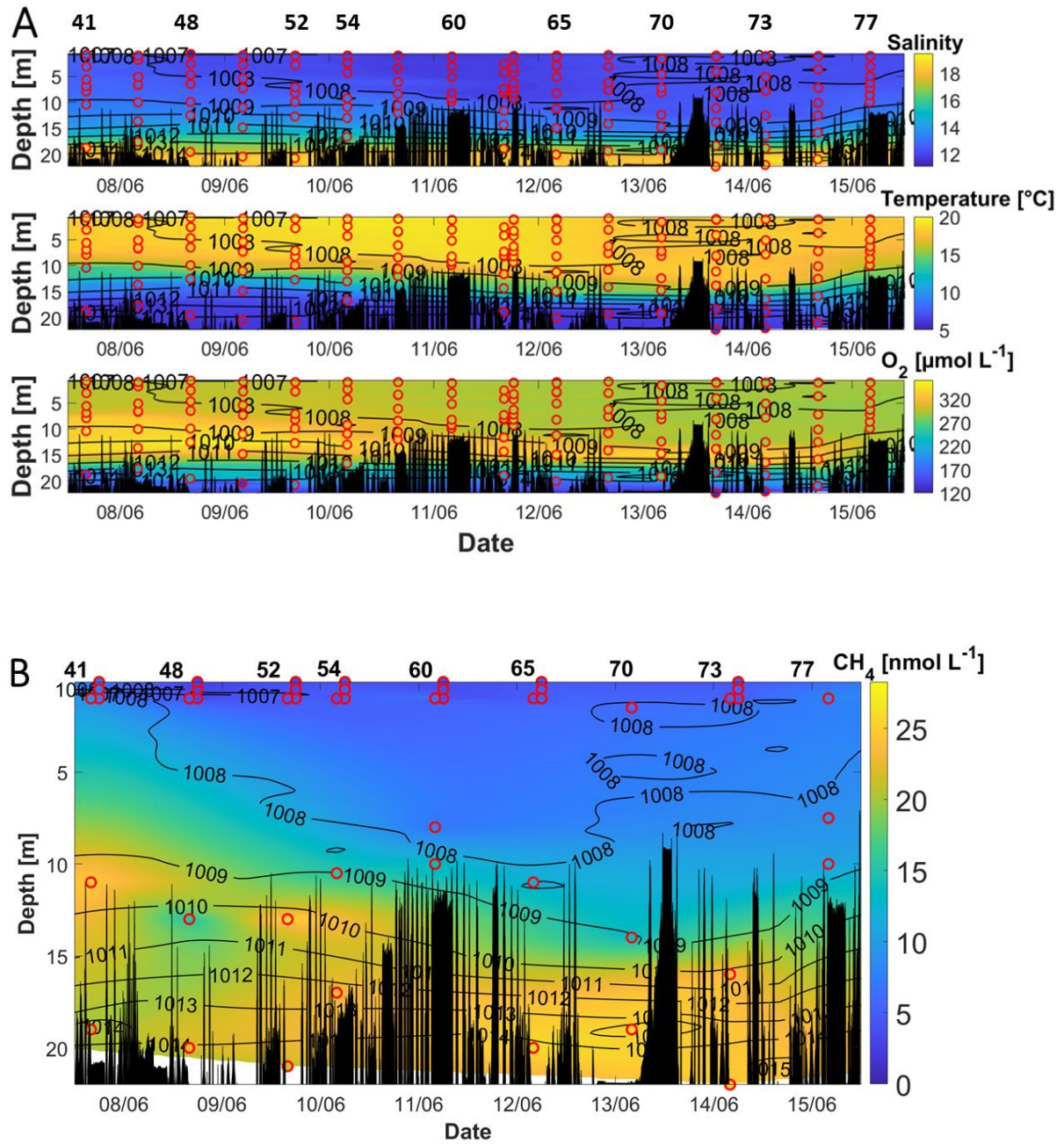


Figure 4: A) Salinity (upper panel), temperature (middle panel), and O₂ (lower panel) during AL510 in June 2018. B) CH₄ concentrations during AL510. Red circles mark the location of the discrete measurements. Black peaks show the topography along the cruise track. Contour lines represent the density. Data for CH₄ was not available for the beginning of the cruise because the measurements started on the 7th June 2018. The station numbers in B refer to those in figure 1.

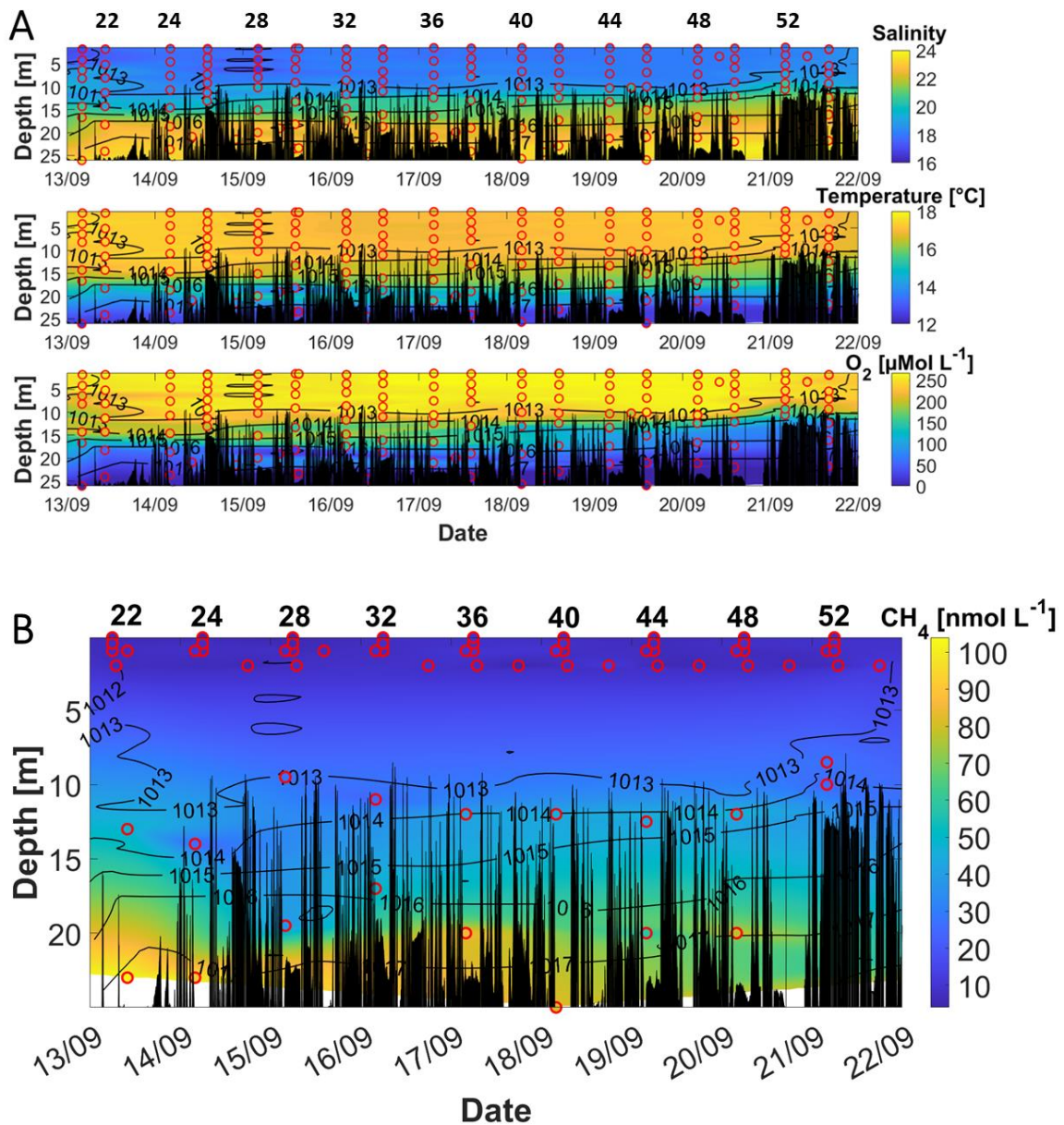


Figure 5: A) Salinity (upper panel), temperature (middle panel), and O_2 (lower panel) during September 2018. B) CH_4 concentrations in September. Red circles mark the location of the discrete measurements. Black peaks show the topography along the cruise track. Contour lines represent the density. The station numbers refer to those in figure 1.

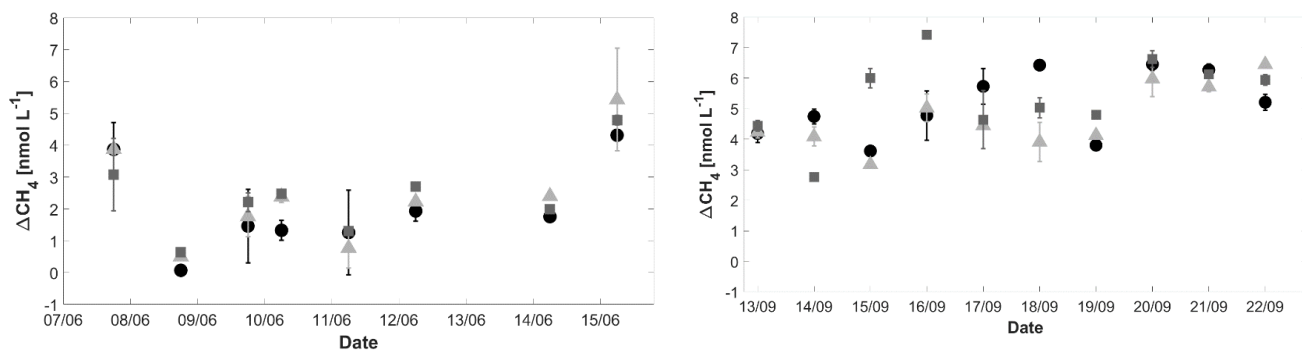


Figure 6: ΔCH_4 in June (left) and September (right) for surface water at 0.1 m depth (dots), 0.5 m depth (triangles), and 1 m depth (squares). Error bars were calculated as described in the method section.

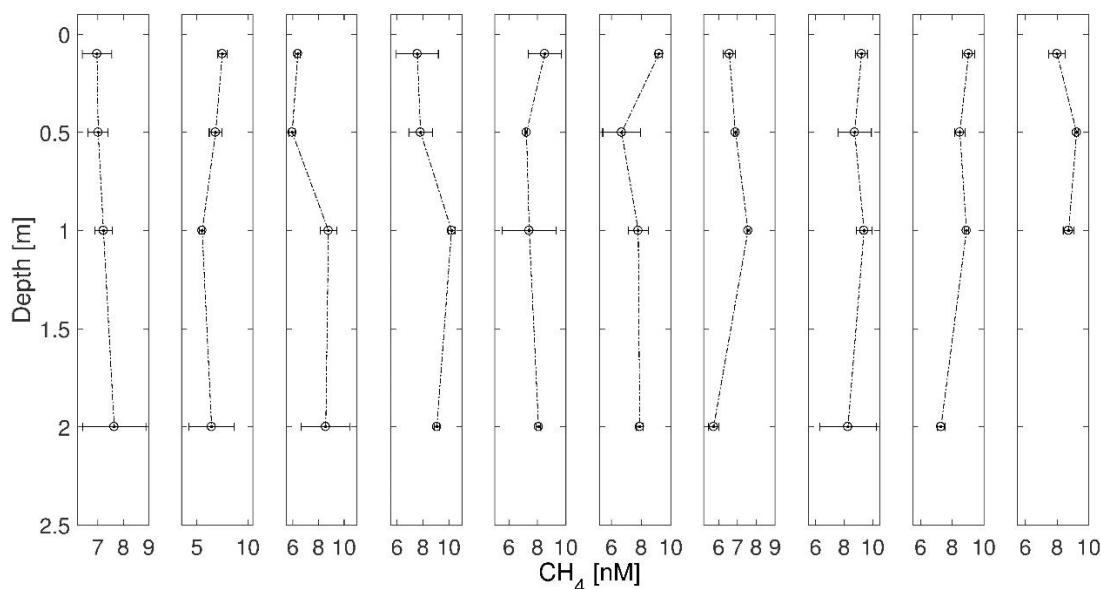


Figure 7: Near-surface profiles of CH_4 from sampling from the zodiac and UW sampling from the ship (~2 m) during AL 516 in September 2018. The time lag between zodiac sampling and UW sampling was max.1h. All samplings were conducted in the morning (~6:00 local time). Error bars indicate 95% CI, calculated from triplicate measurements.

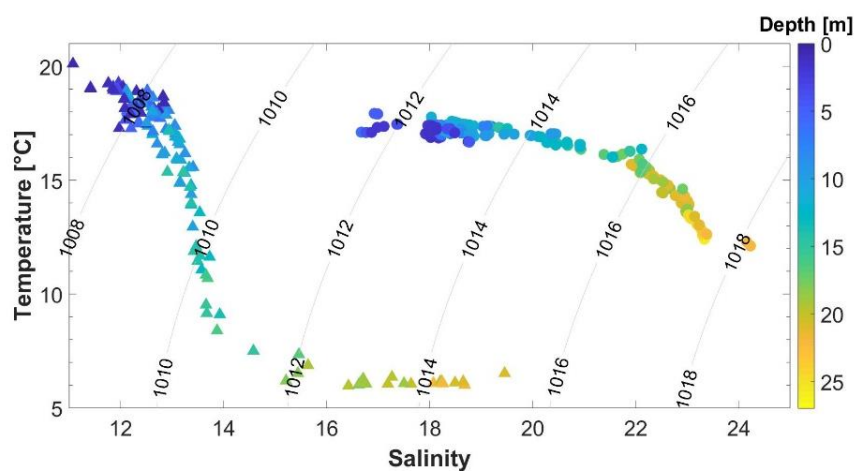


Figure 8: Temperature/salinity diagram of CTD bottle data from June 2018 (triangles) and September 2018 (circles). Grey lines represent the corresponding isopycnals in kg m^{-3} .

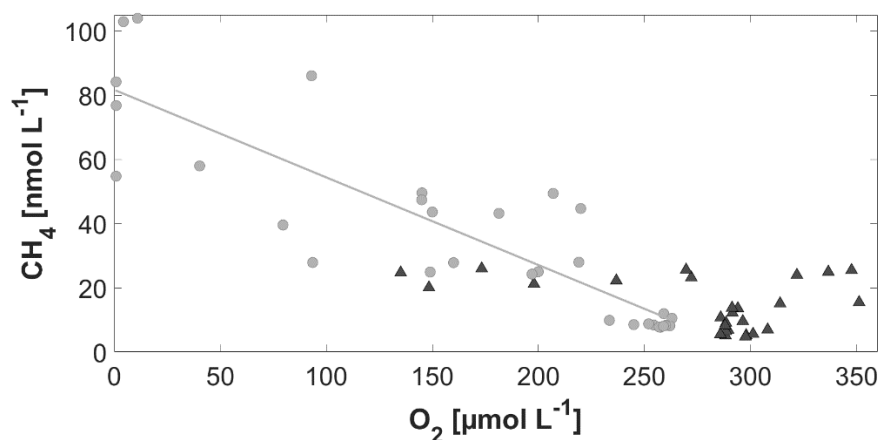


Figure 9: Comparison of the relationship of CH_4 vs. O_2 in June (black triangles; $p=0.07$) and September (filled grey circles, $y = -0.2729x + 81.65$; $R^2 = 0.764$; $P < 0.0001$).

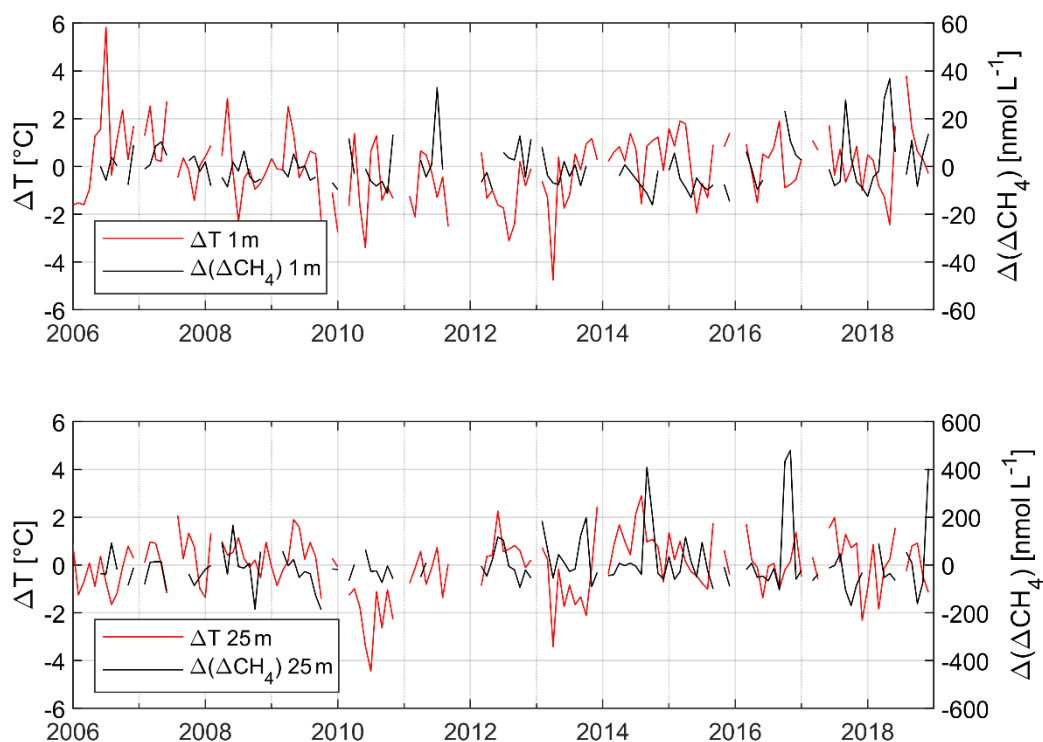


Figure 10. Monthly anomalies of temperature (ΔT , red solid line, left y-axis) and ΔCH_4 ($\Delta(\Delta \text{CH}_4)$, black solid line, right y-axis) in 1 m (upper panel) and 25 m (lower panel) from 2006 to 2019. Please note that gaps in the data sets are caused by missing data.