



# Epidote dissolution—precipitation during viscous granular flow: a micro-chemical and isotope study

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Abstract. Deformation of polymineralic aggregates can be accommodated by viscous granular flow, a process mediated by the interplay among intracrystalline plasticity and dissolution-precipitation, each active in specific minerals at given P-T conditions. Common rock-forming minerals like quartz, feldspars and sheet silicates have been intensively studied in terms of deformation processes. Instead, the deformation behavior of epidote and its role during viscous granular flow is not well investigated, although this mineral is ubiquitous in granitic rocks deforming at greenschist-facies conditions. In this contribution, we provide microstructural and geochemical evidence for the occurrence of dissolution-precipitation of epidote during deformation of an epidote-quartz vein. The main part of the vein is deformed producing a fold, which is visible due to relicts of primary-growth layering inside the vein. The deformation mechanisms active during deformation include dynamic recrystallization of quartz by subgrain rotation recrystallization, producing grain-size reduction of the primary vein quartz. This occurs contemporaneously with dissolution and (re)precipitation of epidote, and grain-boundary sliding, leading to a combined process described as viscous granular flow. The combination of intracrystalline plasticity, grain boundary sliding and dissolution locally and repeatedly produce creep cavities. These represent not only loci for nucleation of new epidote grains at the expenses of dissolved one, but they also allow fluid-mediated transport of elements. The same trace element patterns between old epidote relicts and newly formed grains, with much narrower variability, indicate a process of chemical homogenization. The nature of the fluid mediating deformation is investigated using Pb-Sr isotope data of epidote, which suggest that deformation is assisted by internally recycled fluids with the addition of a syn-kinematic external fluid component.

#### 25 1 Introduction

Deformation microstructures preserved in rocks are the result of the interplay of multiple microscale deformational processes, which may be identified and appreciated through careful petrographic studies (e.g., Passchier, 2005). These processes – which may be coupled to metamorphic reactions and diffusion (e.g., Wintsch and Yeh, 2013; Lanari and Duesterhoeft, 2019) – are the response of the grains to varying physico-chemical conditions in order to minimize the system's internal energy (e.g., Evans et al., 2001; Herwegh and Berger, 2004; Passchier, 2005; Karato, 2008; Herwegh et al.,

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2011; Hobbs et al., 2010). Microstructures of monomineralic aggregates are determined by the deformation behavior of the constituent mineral and by the conditions (e.g., temperature, stress, availability of fluids) existing during deformation. For example, Stipp et al. (2002) showed that the microstructures resulting from dynamic recrystallization of quartz can be linked to different recrystallization mechanisms active at increasing temperatures. However, this type of interpretations is more 35 complicated when dealing with polymineralic aggregates, where the relative abundances and specific deformation mechanisms of each mineral have an effect on the deformation behavior of the bulk rock (e.g., Handy, 1990; 1994; Olgaard, 1990; Stünitz and Fitz Gerald, 1993; Kruse and Stünitz, 1999; Tullis, 2002; Herwegh and Berger, 2004; Passchier, 2005; Herwegh et al., 2011; Wehrens et al., 2017). As an example, it has been shown that the grain size of calcite in mylonitic carbonates is determined by the grain size and volume abundance of second phases, such as sheet silicates (Olgaard, 1990; Herwegh and Berger, 2004; Herwegh et al., 2011). In this context, mass transfer processes like dissolution-precipitation play a fundamental role in that they enable the redistribution of material within the deforming system (e.g., Herwegh and Jenni, 2001; Konrad-Schmolke et al., 2018). A circulating fluid may precipitate dissolved material into intragranular voids of an initially monomineralic aggregate during deformation. In this scenario, the process of dissolution-precipitation in combination with the presence of a second phase keeps the grain sizes of the deforming aggregate small in a process called viscous granular flow (Olgaard, 1990; Fitz Gerald and Stünitz, 1993; Stünitz and Fitz Gerald, 1993; Paterson, 1995; Herwegh and Berger, 2004; Herwegh et al., 2011). Viscous granular flow entails dissolution-precipitation that dissolves mineral asperities, thus facilitating grain boundary sliding in polymineralic aggregates (e.g., Fitz Gerald and Stünitz, 1993; Stünitz and Fitz Gerald, 1993; Paterson, 1995). Grain boundary sliding promotes the formation of creep cavities, hence allowing the nucleation of second-phase minerals (e.g., Herwegh and Jenni, 2001; Fusseis et al., 2009; Gilgannon et al., 2021). Despite the important role of fluids involved in deformation processes, their direct characterization is rare (e.g., Marquer and Burkhard, 1992; Marquer and Peucat, 1994), and it is often unclear whether the fluids are newly added to the system during deformation (i.e., of external origin) or if they are recycled (e.g., via dissolution of hydrous minerals). Hence, the interplay of recycled and newly added fluids, mass-transfer processes and deformation mechanisms to produce complex microstructures remains to be fully appreciated.

In this contribution, via combined microstructural observations and geochemical and Pb–Sr isotope geochemistry, we investigate the deformational processes affecting epidote (i.e., Ca<sub>2</sub>Al<sub>2</sub>(Al,Fe<sup>3+</sup>)Si<sub>3</sub>O<sub>12</sub>(OH)) and quartz defining a microfold in a deformed epidote-quartz hydrothermal vein. The data demonstrate that the present-day microstructure is the result of the interplay between dissolution–precipitation of epidote and dynamic recrystallization of quartz, both of which ensure grain size reduction in viscous granular flow. The nature of the fluid mediating dissolution–precipitation of epidote and the dynamic recrystallization of quartz is investigated using Pb–Sr isotope data. Because epidote is a widespread rock-forming, hydrothermal and alteration mineral in granitic rocks (e.g., Bird and Spieler, 2004; Enami et al., 2004; Franz and Liebscher, 2004; Grapes and Hoskin, 2004; Schmidt and Poli, 2004; Morad et al., 2010; Hentschel et al., 2020), the occurrence of epidote dissolution–precipitation and its control on the deformation mechanisms of other rock-forming minerals (e.g., dislocation creep in quartz) has important implications for the structural evolution of the granitic continental crust.





#### 65 2 Geological Setting

The Grimsel Pass area (Fig. 1; central Swiss Alps, Switzerland) is in the southern part of the Aar Massif, one of the External Crystalline Massifs of the Alps (e.g. Schneeberger et al., 2019; Berger et al., 2017). The Aar Massif consists of a polycyclic basement and Permian intrusives, including the post-Variscan Central Aar Granite. This granitoid was emplaced 299 ± 2 Ma (Schaltegger & Corfu, 1992) and mainly records Alpine deformation (e.g., Choukroune and Gapais, 1983; Bambauer et al., 2009; and references therein). This Alpine deformation is expressed by a large number of ductile shear zones, and in the southern Aar Massif it can be subdivided into two major phases: 1) a reverse faulting phase (ca. 22-17 Ma; Challandes et al., 2008; Rolland et al., 2009; Wehrens et al., 2017) and 2) a strike-slip phase (from ca. 14 Ma onwards; Rolland et al., 2009; Wehrens et al., 2017; Herwegh et al., 2020). In the shear zones, green biotite is stable in the first phase at  $\geq 400$  °C, and it is progressively replaced

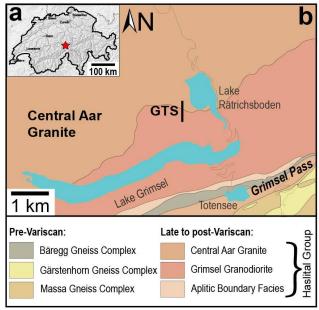


Figure 1: (a) Geological map of the Grimsel Pass area (modified from Wehrens et al., 2016). (b) Geographic location of the Grimsel Pass area (red star) in Switzerland (modified from map.geo.admin.ch).

by chlorite in the second one at lower temperatures (Rolland et al., 2009; Wehrens et al., 2017). The Alpine metamorphic overprint reached greenschist facies conditions in the area, with maximum temperatures and pressures of 450 ± 30 °C and 6 ± 1 kbar, respectively (Challandes et al., 2008; Goncalves et al., 2012; Villa and Hanchar, 2013). Hydrothermal activity upon exhumation in Alpine times is recorded in the area primarily by the Grimsel Breccia Fault hydrothermal system (e.g., Hofmann et al., 2004; Belgrano et al., 2016; Diamond et al., 2018; Egli et al., 2018), and by Alpine cleft mineralization (e.g., Mullis et al., 1994; Janots et al., 2012; Berger et al., 2013; 2022; Rossi and Rolland, 2014; Bergemann et al., 2017; Ricchi et al., 2019). U–Pb geochronology of hydrothermal epidote in veins returned Miocene and Permian ages (Peverelli et al., 2021; accepted).

#### 3 Field relations and sample description

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The investigated sample Grimsel-1 is an epidote-quartz (±biotite) vein that was collected inside Nagra's Grimsel Test Site (GTS; Figs. 1–2a), which is an underground rock laboratory in the Grimsel Pass area (see Schneeberger et al., 2019). In the GTS, the characteristic pattern of shear zones (e.g., Goncalves et al., 2012) is well exposed on the tunnel walls. The studied epidote vein runs oblique with respect to a steeply dipping shear zone of Alpine age hosted by the Central Aar Granite (Fig. 2b). The shear zone is ca. 10 cm wide, whereas the epidote vein reaches up to a couple of centimeters in width. Three other



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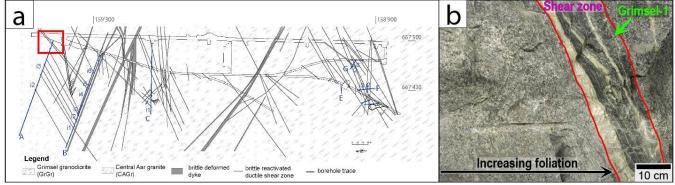


Figure 2 (a) Geological map of the Grimsel Test Site (GTS) of Nagra with the location of (b) shown in the red rectangle (modified from Schneeberger et al., 2019). (b) Field photograph of the location of the studied epidote-quartz vein (Grimsel-1) and the Alpine shear zone with which it is associated; both are in the Central Aar Granite, which shows increasing foliation towards the shear zone (modified from Peverelli et al., accepted).

epidote veins are associated with this shear zone (Fig. A1; not visible in Fig. 2b): (1) sample P2 is a weakly deformed epidote-quartz vein in a slightly deformed portion of the host; (2) sample P3 consists of a number of sheared epidote-quartz veins in a mylonitic portion; and (3) sample Gr0-c is a highly sheared epidote-quartz vein which subdivides the host rock into a mylonitic side and a slightly deformed one. Epidote in veins P2–3 returned Permian U–Pb ages (Peverelli et al., accepted). The host Central Aar Granite displays a gradation from slightly to highly deformed: this is expressed by an increasingly intense foliation and by a decreasing grain size moving towards the shear zone (Fig. 2b). In thin section (Fig. 3), the host rock mostly preserves its magmatic texture and it is made of ca. 55 vol. % of altered feldspar and 35 vol. % quartz. Feldspar grains are sometimes fractured, and quartz is mostly dynamically recrystallized by subgrain rotation recrystallization (SGR). Relict plagioclase and K-feldspar grains can be recognized by the different type and extent of

alteration. Plagioclase (ca. 20 vol. %) is highly altered into epidote and white mica, to a larger extent in the cores than in the rims of the grains. Exsolution lamellae are preserved in K-feldspar grains (ca. 35 vol. %), which are moderately altered into sericite. Green biotite, epidote and minor chlorite define a weak foliation and make up ca. 10 vol. %, with a few accessory titanite grains completing the host rock's mineral assemblage. The sharp boundary between host and vein is marked by a change in modal

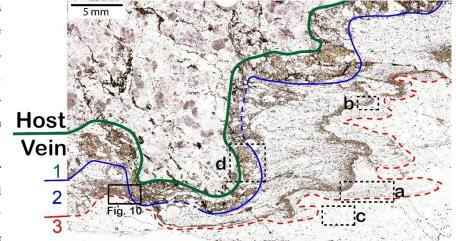


Figure 3: Transmitted-light microscope scan of the studied epidote-quartz (±biotite) vein and the host Central Aar Granite. The numbers 1–3 refer to the vein layers described in Sect. 5.1. Dashed rectangles A–D indicate the locations of the microstructural domains shown in Fig. 4a–d. The black rectangle indicates the location of Fig. 10 (rotated by 90° clockwise). Plane-polarized light.





#### 120 abundances.

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The object of the present investigation is Grimsel-1 epidote-quartz vein (Fig. 3). This vein is subdivided into three layers: (1) layer with coarse-grained epidote, quartz and minor green biotite, (2) heavily deformed and finer-grained epidote-quartz layer, and (3) nearly pure quartz layer. Grain sizes, mineral proportions and dominant deformation mechanisms in each layer are given in Table 1. Layer 1 is the least deformed one, whilst deformation is most intense in layers 2-3, where a fold is defined by the spatial distribution of epidote and quartz (Fig. 3). The reasons why layer 1 is less deformed than layers 2-3 may be linked to the closer proximity of layer 1 to the host rock, to a role of mechanically strong epidote clusters forming a load-bearing network, or to a combination of both (Handy, 1990; 1994; Masuda, 1990; 1995; Tullis, 2002; Passchier, 2005). The modal abundance of green biotite varies greatly among the three layers, with a sharp decrease from layer 1 to layer 2, and layer 3 being devoid of biotite (Table 1). The transition from layer 2 to layer 3 is marked by a change in the epidote/quartz ratio. This is up to ca. 40/60 in layer 2, whereas layer 3 is characterized by the near absence of epidote. Peverelli et al. (2021) performed U-Pb dating by LA-ICP-MS of epidote grains in layer 1, obtaining a Tera-Wasserburg age of  $19.2 \pm 4.3$  Ma. The Tera-Wasserburg regression revealed a single epidote generation at the current analytical precision (i.e., MSWD of 0.79), and the age is interpreted as the timing of epidote crystallization upon vein opening (Peverelli et al., 2021). The focus of this manuscript are the deformation mechanisms in layers 2-3.

Table 1: Main characteristics of the vein layers. Ep = epidote; SGR = subgrain rotation recrystallization; Qz = quartz.

| Layer | Minerals other than Ep + Qz  | Ep [vol. %] | Ep grain size | Qz deformation          | Qz grain size | Qz relicts |
|-------|------------------------------|-------------|---------------|-------------------------|---------------|------------|
| 1     | Green biotite (ca. 5 vol. %) | ca. 70      | 0.025–1.6 mm  | Dislocation glide + SGR | 80–400 μm     | 0.6–1.2 mm |
| 2     | Green biotite (ca. 1 vol. %) | ca. 5–40    | 5–90 µm       | SGR + dislocation glide | 20–170 μm     | 1.4–2.5 mm |
| 3     | -                            | < 1         | 5–200 µm      | SGR + dislocation glide | 30–400 μm     | 0.6–2 cm   |

#### 4 Methods

Analyses were performed at the Institute of Geological Sciences of University of Bern (Switzerland) unless specified otherwise. The petrographic characterization of the studied sample was done in a ca. 60 µm thick section on a Zeiss Axioplan petrographic microscope. For backscattered electron (BSE), forescattered electron (FSE) and cathodoluminescence (CL) images, and for electron backscatter diffraction (EBSD), a Zeiss EVO50 scanning electron microscope (SEM) was used with a beam current of ca. 1 nA and accelerating voltage of 20 kV.

#### 4.1 Grain-size analysis

Microstructural analysis was performed combining transmission light and scanning electron microscopy to determine average grain sizes of epidote and quartz, as well as their volume fractions in selected microstructural domains. Grain size https://doi.org/10.5194/egusphere-2022-311 Preprint. Discussion started: 20 May 2022 © Author(s) 2022. CC BY 4.0 License.





analysis was carried out in ImageJ (IJ 1.46r; Ferreira and Rasband, 2012) using sketches of the microstructures drawn on transmitted light, BSE and FSE images. In domains where the microstructure is not suitable for an automated grain size analysis by ImageJ (Figs. 4b and 7c), epidote and quartz grain sizes were directly measured on a transmitted light microscope.

#### 150 **4.2 Chemical maps**

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X-ray compositional maps in wave-length dispersive mode of Si, Fe, Al, Ca, Mn and Sr were obtained by electron probe micro-analyzer (EPMA) on a JEOL-8200 microprobe. Accelerating voltage was 15 keV, specimen current 100 nA, the step size 4 μm, and dwell time was 190 msec. For calibration of the X-ray maps, spot analyses were acquired with a specimen current of 10 nA, and calibrated using the following standards: wollastonite (SiO<sub>2</sub>), olivine (MgO), anorthite (CaO, Al<sub>2</sub>O<sub>3</sub>), magnetite (FeO), pyrolusite (MnO), tugtupite (Cl), rutile (TiO<sub>2</sub>), and celestite (SrO). The processing of the X-ray compositional maps, including map calibration, was done by using XMapTools (Lanari et al., 2014; 2019). Minerals were identified based on the concentrations of specific elements (i.e. Ca, Fe and Al for epidote; Si for quartz; K for micas) and classified. Maps of the structural formula of epidote were calculated on 12.5 oxygen basis.

#### 4.3 Trace elements

For trace element measurements, a RESOlutionSE 193 nm excimer laser system (Applied Spectra, USA) equipped with an S-155 large-volume constant-geometry chamber (Laurin Technic, Australia) coupled with an Agilent 7900 ICP-QMS was employed. During ablation, a He atmosphere was used, and Ar was admixed to the carrier gas before reaching the plasma of the ICP-MS. NIST SRM612 was used for optimization of the analytical conditions, ensuring that the ThO production rate was < 0.2 % and the Th/U sensitivity ratio > 97 %. On-sample fluence was 5 J cm<sup>-2</sup> and repetition rate 5 Hz. The size of the analysis spots ranged between 20–30 μm, and BSE images were used to plan the analyses to avoid locating any spots across zones with heterogeneous composition (e.g., chemical zoning) or on inclusions. External standardization was done with USGS GSD-1G standard, and the SRM612 standard was measured as an unknown for quality control in absence of a well characterized epidote standard. Bracketing standardization enabled a true-time linear drift correction. Data reduction was carried out with the software SILLS (Guillong et al., 2008), using the sum of total oxides minus H<sub>2</sub>O (98.3 % for epidote and 100 % for SRM612) as internal standard (see Halter et al., 2002). The formulation of Pettke et al. (2012) was employed to calculate limits of detection for each element in every analysis.

#### 4.4 U-Pb isotope data

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Measurements of <sup>238</sup>U/<sup>206</sup>Pb, <sup>207</sup>Pb/<sup>206</sup>Pb and <sup>87</sup>Sr/<sup>86</sup>Sr were made in epidote micro-separates following the procedure for sample digestion in acids and for column chemistry detailed by Peverelli et al. (2021; modified from Nägler and Kamber, 1996). Two epidote micro-separates – each mixing Epidote-A and Epidote-B (see Sect. 5.1) in unknown and different proportions – were prepared (Ep\_A+B\_1 and Ep\_A+B\_2). After hand-picking, the material was finely ground and washed

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with MilliQTM water. Two aliquots of each (Ep A+B 1a and Ep A+B 1b, and Ep A+B 2a and Ep A+B 2b) were weighed in as replicates for each micro-separate ensuring ca. 300 ng of Pb in each aliquot. During column chemistry with a Sr-specTM resin (Horwitz et al., 1992), the Sr and Pb fractions were collected in sequence (Haeusler et al., 2016). One large Epidote-A grain was also handpicked and ground, and an amount of the powder corresponding to 250 ng of Sr was digested in acids before the extraction of the Sr fraction by column chemistry. The Pb fraction of this Epidote-A grain was not collected since Pb isotopic data are available from in-situ U-Pb isotope measurements by LA-ICP-MS (Peverelli et al., 2021). Procedural blank samples were used to assess contamination during work in the laboratories. Measurements of <sup>238</sup>U and <sup>206</sup>Pb concentrations for calculation of <sup>238</sup>U/<sup>206</sup>Pb ratios were run at the Department of Geography of University of Bern on a 7700x Agilent quadrupole ICP-MS. For these analyses, two aliquots of each digested epidote microseparate were separated and diluted to different volumes. Three aliquots were separated from the digested AGV-2 standard and diluted in 0.05, 0.25 and 0.7 ml HNO<sub>3</sub>. The intensities in counts per second (cps) of <sup>238</sup>U and <sup>206</sup>Pb measured in the three aliquots of the AGV-2 standard were plotted against the real concentrations of these isotopes in ppb calculated for each aliquot using the reference data of Weis et al. (2006). The calibration curves thus obtained were used to calculate the real concentrations of <sup>238</sup>U and <sup>206</sup>Pb in the digested epidote material. The calculation of 2 standard errors (2 S.E.) on <sup>238</sup>U/<sup>206</sup>Pb ratios – which are not returned automatically by the measurement software -is detailed in Peverelli et al. (2021). Pb isotope ratios were measured on a Thermo Fisher Neptune Plus MC-ICP-MS in desolvated plasma mode equipped with a CETAC Aridus 2 desolvating system. Instrumental mass fractionation was corrected within-run by means of a Tl spike. External reproducibility of the measurements was quantified by measuring the NIST NBS 981 standard. The measured Pb isotope ratios were identical to those obtained by Rehkämper and Mezger (2000; their Table 4).

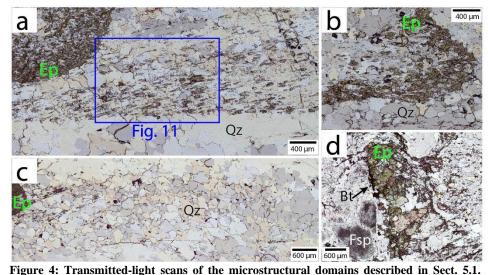
#### 4.5 Strontium isotope data

Strontium isotope ratios were measured on a ThermoFisher TritonTM thermal ionization mass spectrometer (TIMS) after loading 250 ng Sr diluted in 6.4 M HCl on Re filaments using 1.5  $\mu$ l Ta-oxide activator. The SRM 987 standard (200 ppm; Weis et al., 2006) was measured for quality control. The detected masses were 84, 85, 86 (center cup), 87 and 88. The interference of <sup>87</sup>Rb and within-run mass fractionation were corrected for by using the IUPAC <sup>87</sup>Rb/<sup>85</sup>Rb and <sup>88</sup>Sr/<sup>86</sup>Sr values of, respectively, 0.385617 and 8.735209. The SRM 987 standard returned a weighted average <sup>87</sup>Sr/<sup>86</sup>Sr ratio of 0.710279  $\pm$  0.000020 (2 standard deviation, S.D.; number of replicates, n = 12), which is higher than the reference preferred value of 0.710248 (see Weis et al., 2006). Standards AGV-2 (n = 1) and GSP-2 (n = 2) of USGS were measured as unknowns for quality control and the returned, respectively, <sup>87</sup>Sr/<sup>86</sup>Sr ratios of 0.704041  $\pm$  0.000018 (2 standard error, 2 S.E.), and 0.765396  $\pm$  0.000010 and 0.765202  $\pm$  0.0000008. These values are also higher than the reference preferred values of 0.703981  $\pm$  0.00009 (2 S.D.) and 0.765144  $\pm$  0.000075 (2 S.D.) for AGV-2 and GSP-2, respectively (see Weis et al., 2006). A correction based on the reference materials returning higher <sup>87</sup>Sr/<sup>86</sup>Sr values than their reference values would produce the same shift in all measured <sup>87</sup>Sr/<sup>86</sup>Sr ratios. Hence, we did not correct our data because only the variability among the samples is relevant in this study, while the interpretation of the absolute Sr isotope ratios is beyond the scope of this work. Rubidium





210 concentrations were not measured, as a correction for 87Rb-derived 87Sr is not necessary. This is valid approach because the 215 incompatibility of Rb in the epidote crystal structure results in negligible Rb concentrations (see Frei et al., 2004; Feineman et al., 2007).



## Microstructures (a-c) are used for microstructural analysis (Fig. 7; Sect. 5.1.2). The blue rectangle in (a) indicates the location of the cathodoluminescence image of Fig. 11. Bt =

5 Results

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#### **5.1** Microstructural analysis

The different characteristics of epidote and its microstructures in layers 1–3 (Figs. 3–4) allow the distinction of epidote into Epidote-A and Epidote-B as illustrated below. All microstructural domains indicate that vein opening with crystallization of epidote, quartz and minor biotite was followed by vein deformation.

biotite; Ep = epidote; Fsp = feldspar; Qz = quartz. Plane-polarized light.

#### 225 5.1.1 Layer 1: veining and Epidote-A

This epidote layer is characterized by coarse (ca.  $0.2{\text -}1.6$  mm) epidote grains associated with smaller angular ones (ca.  $20{\text -}200~\mu\text{m}$ ). The coarser epidote grains (Figs. 4d and 5) form clusters with random shape orientations or are found as isolated crystals. Larger epidote grains are often surrounded by the smaller angular epidote grains (Fig. 5; red arrow) as a result of brittle grain-size reduction upon deformation with brittle deformation behavior of epidote. Epidote is euhedral to anhedral, and smaller epidote crystals are found in the gaps among the larger ones as well (see Fig. 2b of Peverelli et al., 2021). Large quartz relicts are present displaying undulose extinction. More commonly, quartz is dynamically recrystallized, forming core-mantle structures with subgrain rotation around quartz relicts. Hereafter, we refer to epidote in layer 1 as Epidote-A.

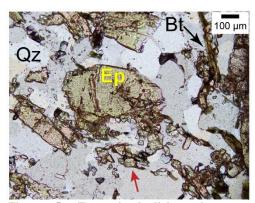


Figure 5: Transmitted light microscope photograph of one Epidote-A grain in layer 1 surrounded by small angular grains (red arrow). Bt = biotite; Ep = epidote; Qz = quartz. Plane-polarized light.



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#### 5.1.2 Layers 2–3: microfold and Epidote-B

In layers 2–3, the spatial distribution and the variable modal abundance of epidote define a fold (Figs. 3, 4b, 6 and A1). Smaller epidote grains with a shape-preferred orientation and quartz define axial planes and limbs (rectangle a in Fig. 3; Figs. 4a and 6). Epidote grain boundaries are mostly curved and irregular, but a few euhedral epidote grains are also observed (Fig. 6). Epidote grains vary between 5–90 µm in size, and the modal abundance of epidote varies between ca. 5 to ca. 40 vol. % within the layer (Table 1), defining a quartz-supported microstructure. Where epidote is most abundant, quartz and epidote define a "sponge-like" microstructure (Figs. 4b and 6). Small (ca. 1–20 µm in size) oval epidote grains and fluid inclusions are observed within

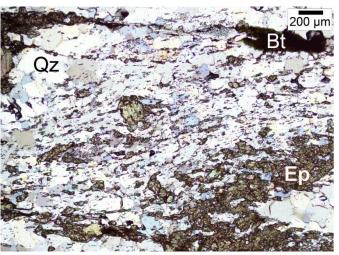


Figure 6: Transmitted light microphotograph of layer 2 where oval epidote grains are hosted by dynamically recrystallized quartz with their long axes oriented consistently with the microfold axial planes. Bt = biotite; Ep = epidote; Qz = quartz. Overlapped plane-polarized and cross-polarized light images.

quartz grains (Figs. 4a and 6). The size of dynamically recrystallized quartz grains correlates with grain size and abundance of epidote (Fig. 7; Zener relation; Herwegh et al., 2010). Dynamic recrystallization of quartz by subgrain rotation prevails creating a crystallographic preferred orientation (Fig. 8), and quartz relicts with undulose extinction are minor (Fig. 4a–c). However, a ca. 2 cm long relict quartz grain, whose orientation acquired during crystallization may have favored dislocation glide, marks the transition from layer 2 to layer 3 (Fig. A1, black arrow). Where minor epidote is present, anhedral epidote

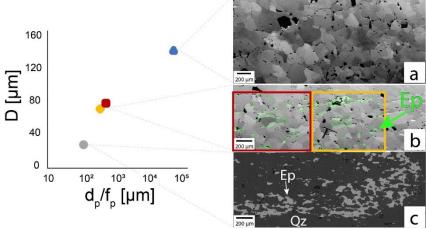


Figure 7: Correlation between quartz grain size (D) and Zener parameter  $(d_p/f_p)$  of epidote as the second phase determined in the FS and BSE images of the microstructures in the microfold of layers 2–3. The green lines in the central BSE image contour epidote grains. Ep = epidote; Qz = quartz. Panels a, b and c correspond, respectively, to panels b, a and c of Fig. 4.

grains of few to ca. 10 µm in size are mostly interstitial and found at triple junctions among quartz subgrains (Fig. 9). We refer to subhedral to anhedral epidote grains defining the microfold in layers 2-3 as Epidote-B. While U-Pb ages by LA-ICP-MS of Epidote-A (Sect. 3) are available, no U-Pb dating is possible in Epidote-B because the small grain size of epidote entails contamination from epidote-epidote or epidote-quartz grain boundaries upon measurements.





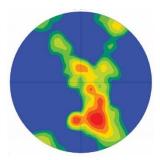


Figure 8: Orientation of quartz C axis in layer 3 relative to the long side of the thin section; measured by electron backscatter diffraction.

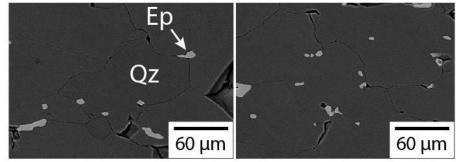


Figure 9: Backscattered electron images showing epidote (Ep) grains along quartz (Qz) grain boundaries and at triple junctions among quartz grains.

#### 5.2 Major and minor elements

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The compositional map in Fig. 10 (location shown in Fig. 3) covers layers 1–3. The compositions of FeO and Mn range between ca. 12.5–14 wt. % and ca. 2500–5000  $\mu g$  g<sup>-1</sup>, respectively, across all analyzed epidote grains. Larger Epidote-A grains in layer 1 are zoned, with FeO and Mn concentrations increasing from core to rim. Epidote in layers 2–3, on the other end, is characterized by uniform concentrations of FeO and Mn. It should be noted that, because the step size of the compositional maps is 4  $\mu$ m, the chemical variability of epidote in layers 2–3 is better assessed among different grains across the overall microstructure rather than within each crystal (i.e., the majority of grains do not contain enough 4×4  $\mu$ m pixels).

#### 5.2 Trace elements

The minimum spot size used for measurements by LA-ICP-MS is  $20~\mu m$ , which is ca. four times larger than the smallest epidote grains. Therefore, chemical variability is assessed throughout each microstructural layer by relying on measurements in large-enough epidote grains to avoid contamination from grain boundaries.

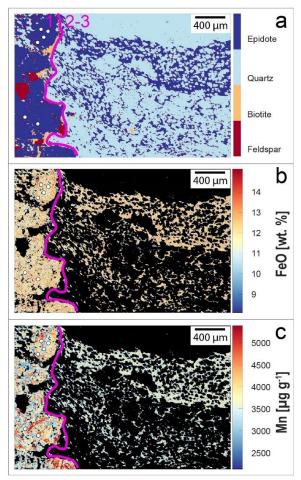


Figure 10: X-ray compositional maps across layers 1–3 (Fig. 3; Sect. 5.1). White circles indicate spots for U-Pb dating by LA-ICP-MS of Peverelli et al. (2021). The pink line separates layer 1 from layers 2–3.





Epidote-A is also addressed here with no specific reference to intra-grain zoning but only across the overall Epidote-A microstructure (i.e., layer 1). Data of Epidote-B are collected in anhedral epidote grains in layer 2. The different extents of chemical variability between Epidote-A and Epidote-B noted in the compositional maps is reflected by trace element data (Table 2).

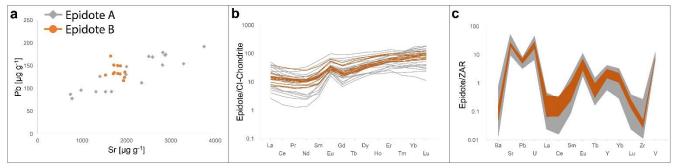


Figure 11: Epidote LA-ICP-MS data of (a) Pb and Sr, (b) CI-chondrite normalized (McDonough and Sun, 1995) rare earth element patterns (REE), and (c) trace elements normalized to the Central Aar Granite (ZAR; Schaltegger & Krähenbühl, 1990).

Table 2: Trace element composition of epidote in µg g<sup>-1</sup>measured by LA-ICP-MS.

|      |      |      |     |     |      |      |      |         |           | Epi        | dote-A    |          |            |     |     |     |      |     |      |     |     |
|------|------|------|-----|-----|------|------|------|---------|-----------|------------|-----------|----------|------------|-----|-----|-----|------|-----|------|-----|-----|
| Spot | Ва   | Sr   | Pb  | U   | La   | Ce   | Pr   | Nd      | Sm        | Eu         | Gd        | Tb       | Dy         | Υ   | Но  | Er  | Tm   | Yb  | Lu   | Zr  | V   |
| 1    | 0.99 | 1400 | 130 | 79  | 5.6  | 14   | 2.2  | 12      | 6.3       | 3.9        | 14        | 2.8      | 22         | 200 | 5.5 | 19  | 2.7  | 19  | 2.7  | 3.6 | 140 |
| 2    | 1.2  | 1990 | 130 | 130 | 7.7  | 18   | 2.7  | 15      | 8.5       | 5.6        | 18        | 3.6      | 27         | 230 | 6.4 | 22  | 3.1  | 21  | 3.4  | 5.1 | 140 |
| 3    | 22   | 3290 | 150 | 350 | 4.8  | 8.8  | 1.2  | 5.6     | 2.5       | 2.0        | 5.7       | 1.2      | 12         | 120 | 3.6 | 14  | 2.5  | 21  | 3.8  | 22  | 110 |
| 4    | 21   | 2500 | 170 | 270 | 0.99 | 1.9  | 0.24 | 1.1     | 0.66      | 1.1        | 2.8       | 0.86     | 9.8        | 110 | 3.1 | 14  | 2.7  | 23  | 4.4  | 24  | 110 |
| 5    | 1.9  | 970  | 97  | 60  | 5.2  | 9.3  | 1.1  | 5.0     | 1.8       | 1.3        | 2.9       | 0.62     | 5.1        | 32  | 1.0 | 3.5 | 0.53 | 3.7 | 0.57 | 4.2 | 160 |
| 6    | 23   | 2590 | 170 | 130 | 10   | 15   | 1.6  | 7.0     | 2.8       | 1.9        | 5.6       | 1.3      | 9.9        | 71  | 2.3 | 9.2 | 1.6  | 15  | 3.1  | 51  | 130 |
| 7    | 0.98 | 740  | 87  | 140 | 5.0  | 9.3  | 1.2  | 5.3     | 2.7       | 1.6        | 5.5       | 1.1      | 8.4        | 53  | 1.8 | 6.0 | 0.88 | 5.8 | 0.77 | 4.5 | 200 |
| 8    | 21   | 2870 | 170 | 240 | 12   | 17   | 1.8  | 8.3     | 4.6       | 3.7        | 11        | 2.5      | 19         | 130 | 4.4 | 17  | 2.8  | 25  | 4.4  | 17  | 150 |
| 9    | 9.7  | 2000 | 150 | 100 | 8.5  | 12   | 1.4  | 5.9     | 3.2       | 2.7        | 8.0       | 1.8      | 14         | 82  | 2.9 | 8.7 | 1.1  | 7.8 | 1.3  | 9.6 | 140 |
| 10   | 24   | 2890 | 170 | 220 | 6.6  | 9.7  | 1.1  | 4.6     | 2.0       | 2.2        | 5.9       | 1.4      | 12         | 90  | 2.9 | 11  | 2.1  | 18  | 3.2  | 31  | 140 |
| 11   | 0.92 | 1660 | 93  | 100 | 3.4  | 7.2  | 0.90 | 3.3     | 1.2       | 1.4        | 2.7       | 0.59     | 5.1        | 41  | 1.3 | 4.1 | 0.42 | 2.5 | 0.28 | 2.2 | 110 |
| 12   | 34   | 3750 | 190 | 140 | 5.3  | 8.5  | 1.0  | 4.6     | 1.9       | 2.3        | 4.2       | 1.0      | 8.8        | 67  | 2.2 | 6.7 | 0.80 | 4.8 | 0.64 | 25  | 120 |
| 13   | 0.59 | 770  | 78  | 160 | 1.6  | 3.0  | 0.39 | 1.6     | 1.0       | 1.5        | 3.2       | 1.0      | 9.1        | 65  | 2.2 | 7.9 | 0.97 | 5.9 | 0.72 | 3.2 | 180 |
| 14   | 27   | 2820 | 180 | 260 | 1.3  | 2.5  | 0.29 | 1.2     | 1.2       | 1.5        | 4.6       | 1.3      | 12         | 120 | 3.5 | 14  | 2.8  | 28  | 4.6  | 42  | 130 |
| 15   | 5.8  | 1530 | 93  | 54  | 0.62 | 0.97 | 0.12 | 0.62    | 0.45      | 0.94       | 1.4       | 0.40     | 4.1        | 42  | 1.2 | 4.7 | 0.77 | 6.2 | 1.0  | 5.0 | 140 |
| 16   | 17   | 2810 | 150 | 140 | 1.9  | 3.0  | 0.35 | 1.4     | 0.55      | 1.3        | 1.7       | 0.66     | 7.8        | 77  | 2.3 | 9.3 | 1.6  | 14  | 2.3  | 18  | 130 |
| 17   | 5.2  | 1310 | 93  | 120 | 3.5  | 6.3  | 0.78 | 3.7     | 1.3       | 1.4        | 2.0       | 0.57     | 4.9        | 47  | 1.3 | 6.1 | 0.88 | 7.1 | 0.93 | 5.2 | 120 |
| 18   | 3.8  | 2340 | 110 | 110 | 1.5  | 2.9  | 0.43 | 1.6     | 0.76      | 1.2        | 1.6       | 0.47     | 4.9        | 47  | 5.5 | 5.5 | 0.78 | 5.8 | 0.90 | 3.1 | 120 |
|      |      |      |     |     |      |      |      |         |           | Epi        | dote-B    |          |            |     |     |     |      |     |      |     |     |
| Spot | Ва   | Sr   | Pb  | U   | La   | Ce   | Pr   | Nd      | Sm        | Eu         | Gd        | Tb       | Dy         | Υ   | Но  | Er  | Tm   | Yb  | Lu   | Zr  | V   |
| 1    | 2.1  | 1640 | 171 | 130 | 4.7  | 10   | 1.4  | 7.1     | 3.4       | 3.0        | 7.3       | 1.7      | 16         | 150 | 4.4 | 14  | 2.1  | 14  | 2.3  | 5.2 | 130 |
| 2    | 2.9  | 1860 | 150 | 140 | 3.7  | 8.1  | 1.1  | 5.7     | 3.1       | 3.1        | 7.6       | 2.1      | 17         | 170 | 4.6 | 16  | 2.4  | 19  | 3.0  | 5.9 | 140 |
| 3    | 6.6  | 1950 | 120 | 190 | 1.7  | 4.2  | 0.55 | 2.9     | 1.2       | 1.6        | 3.1       | 0.81     | 8.1        | 85  | 2.1 | 8.6 | 1.5  | 9.7 | 1.7  | 8.2 | 120 |
| 4    | 5.0  | 1720 | 140 | 190 | 3.3  | 7.6  | 1.1  | 5.2     | 2.4       | 2.0        | 4.8       | 1.1      | 10         | 110 | 3.1 | 11  | 1.8  | 13  | 2.1  | 7.0 | 120 |
| 5    | 4.2  | 1700 | 130 | 170 | 3.0  | 7.4  | 0.97 | 5.0     | 2.3       | 2.1        | 4.2       | 1.2      | 11         | 120 | 3.0 | 11  | 2.0  | 14  | 2.5  | 7.7 | 120 |
| 6    | 5.1  | 1930 | 120 | 160 | 1.9  | 4.4  | 0.60 | 2.9     | 1.4       | 1.5        | 3.3       | 0.80     | 7.4        | 85  | 2.3 | 8.8 | 1.3  | 10  | 1.6  | 7.2 | 130 |
| 7    | 3.7  | 1530 | 130 | 160 | 2.9  | 6.6  | 0.88 | 4.5     | 2.1       | 1.7        | 4.1       | 0.73     | 7.9        | 84  | 2.3 | 8.3 | 1.3  | 9.0 | 1.6  | 5.2 | 130 |
| 8    | 6.5  | 1960 | 140 | 190 | 3.3  | 7.6  | 0.99 | 4.5     | 2.3       | 1.9        | 3.9       | 0.99     | 9.1        | 100 | 2.6 | 10  | 1.5  | 13  | 2.1  | 9.3 | 130 |
| 9    | 5.3  | 1850 | 130 | 190 | 3.4  | 8.2  | 1.1  | 4.7     | 2.2       | 1.9        | 3.7       | 0.91     | 8.4        | 100 | 2.7 | 9.9 | 1.5  | 13  | 2.1  | 8.6 | 130 |
| 10   | 5.4  | 1800 | 130 | 190 | 2.3  | 5.2  | 0.71 | 3.5     | 1.7       | 1.7        | 3.6       | 0.85     | 8.5        | 90  | 2.4 | 10  | 1.7  | 12  | 2.2  | 8.1 | 130 |
| 11   | 5.5  | 1800 | 150 | 180 | 3.9  | 8.5  | 1.2  | 5.6     | 2.5       | 1.9        | 4.9       | 0.99     | 9.8        | 100 | 2.7 | 10  | 1.6  | 12  | 1.9  | 8.0 | 130 |
| 12   | 4.8  | 1710 | 151 | 230 | 9.4  | 20   | 2.6  | 12      | 5.3       | 3.5        | 9.6       | 2.3      | 19         | 160 | 4.8 | 16  | 2.4  | 16  | 2.6  | 7.6 | 130 |
|      |      |      |     |     |      |      |      | Central | Aar Grani | ite (Schal | tegger ar | d Krähen | bühl, 1990 | 0)  |     |     |      |     |      |     |     |
| KAW  | Ва   | Sr   | Pb  | U   | La   | Ce   | Pr   | Nd      | Sm        | Eu         | Gd        | Tb       | Dy         | Y   | Но  | Er  | Tm   | Yb  | Lu   | Zr  | V   |
| 2219 | 430  | 75   | 24  | 8   | 26   | 61   | -    | -       | 5.1       | 0.51       | -         | 2.4      | -          | 56  | -   | -   | -    | 8.6 | 12   | 192 | 18  |



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The concentrations of Sr and Pb in Epidote-B overlap with the trend defined by the same elements measured Epidote-A, but they cover a more limited range values (Fig. 11a). The CI chondrite-normalized rare (REE) elements earth

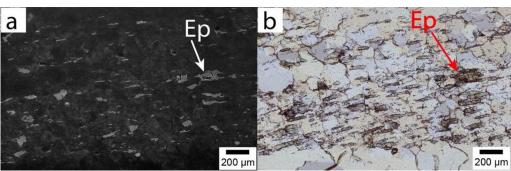


Figure 12: Overlapped cathodoluminescence CL) and backscattered electron images. The only minerals in the image are epidote (Ep) and quartz (unlabeled grains). The different CL contrasts in quartz are due to variable trace element contents.

patterns (Fig. 11b) of Epidote-A and Epidote-B have similar trends, characterized by positive slopes (La<sub>N</sub>/Yb<sub>N</sub> of 0.1–0.4 in Epidote-A and 0.03–0.5 in Epidote-B) and variably positive Eu anomalies (1.5–2.4 in Epidote-A and 1.2–2.3 in Epidote-B).

The REE trends of Epidote-B fall within the range of Epidote-A and confirm the lesser extent of chemical variability of Epidote-B relative to Epidote-A. Selected elements are plotted as values normalized to the concentrations in the Central Aar Granite in Fig. 11d reinforce chemical affinity between Epidote-A and Epidote-B, as well as the lesser chemical variability of the latter relative to the former. Cathodoluminescence images of the recrystallized quartz grains (Fig. 12) qualitatively indicate trace element variability also in quartz grains.

#### **5.3 Isotope data**

Table 3: U-Pb and Sr isotopic data by solution ICP-MS and TIMS. Uncertainties are 2 standard errors (2 S.E.).

|           | <sup>238</sup> U/ <sup>206</sup> Pb | 2 S.E.     | <sup>207</sup> Pb/ <sup>206</sup> Pb | 2 S.E.       | <sup>87</sup> Sr/ <sup>86</sup> Sr | 2 S.E.   |
|-----------|-------------------------------------|------------|--------------------------------------|--------------|------------------------------------|----------|
| Ep_A+B_1a | 6.49                                | 0.11       | 0.79159                              | 0.00001      | 0.727803                           | 0.000011 |
| Ep_A+B_1b | 6.54                                | 0.07       | 0.79108                              | 0.00001      | 0.727807                           | 0.000007 |
| Ep_A+B_2a | 6.44                                | 0.16       | 0.79428                              | 0.00001      | 0.726952                           | 0.000010 |
| Ep_A+B_2b | 6.44                                | 0.10       | 0.79391                              | 0.00001      | 0.726830                           | 0.000015 |
| Epidote-A | $4.42^{1}$                          | $0.09^{1}$ | $0.7867^{1}$                         | $0.0058^{1}$ | 0.726552                           | 0.000007 |
|           |                                     |            |                                      |              |                                    |          |

<sup>&</sup>lt;sup>1</sup> Datum-point #10 of Peverelli et al. (2021) by LA-ICP-MS.

The total <sup>238</sup>U/<sup>206</sup>Pb and <sup>207</sup>Pb/<sup>206</sup>Pb ratios measured by solution ICP-MS in the epidote microseparates (Table 3) are plotted in a Tera–Wasserburg diagram (Fig. 13a) together with the LA-ICP-MS data measured in Epidote-A. The <sup>238</sup>U/<sup>206</sup>Pb ratios of the dry aliquots of each microseparate are within uncertainty of each other, whereas their <sup>207</sup>Pb/<sup>206</sup>Pb ratios are slightly different. This is attributed to geological heterogeneity between the single aliquots of each microseparate: perfect



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homogenization the microseparates not possible because total <sup>207</sup>Pb/<sup>206</sup>Pb ratios include uranogenic lead in addition to initial Pb, and are therefore influenced by the variability in U concentrations. The

ICP-MS

data-

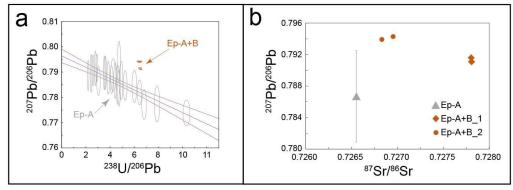


Figure 13: U–Pb and Sr isotope data. (a) Tera–Wasserburg diagram with the LA-ICP-MS data measured in Epidote-A (Ep-A) by Peverelli et al. (2021) and those measured in epidote microseparates (Ep-A+B) by solution ICP-MS; LA-ICP-MS error ellipses are 2σ. (b) Total <sup>207</sup>Pb/<sup>206</sup>Pb and initial <sup>87</sup>Sr/<sup>86</sup>Sr data of Epidote-A and the epidote microseparates; the <sup>207</sup>Pb/<sup>206</sup>Pb ratio of Epidote-A is by LA-ICP-MS (data-point #10 of Peverelli et al., 2021); all other <sup>207</sup>Pb/<sup>206</sup>Pb ratios are by solution ICP-MS; <sup>87</sup>Sr/<sup>86</sup>Sr ratios are by TIMS; error bars are smaller than the symbols where not shown.

points plot above the regression through the LA-ICP-MS data in the diagram, and the isotopic ratios measured by the different techniques cannot be combined in one single regression.

Strontium (Table 3) and Pb isotopic data of the epidote microseparates and of pure epidote (Epidote-A) are plotted in Fig. 13b. To compare the Pb and Sr isotopic characteristics of Epidote-A with those of the microseparates, data obtained from the different techniques have to be combined in a <sup>207</sup>Pb/<sup>206</sup>Pb versus <sup>87</sup>Sr/<sup>86</sup>Sr plot. While Sr in epidote is entirely non-radiogenic, the <sup>207</sup>Pb/<sup>206</sup>Pb ratios measured by solution ICP-MS contain both initial and radiogenic (i.e., U-derived) Pb. Therefore, to discuss Pb isotopic characteristics of Epidote-A (i.e., analyzed by LA-ICP-MS) against those of the epidote microseparates mixing Epidote-A and Epidote-B (i.e., analyzed by solution ICP-MS), we cannot simply use the initial <sup>207</sup>Pb/<sup>206</sup>Pb ratio obtained from a Tera–Wasserburg plot because this value excludes the ingrown radiogenic Pb component of Epidote-A. Therefore, in a <sup>207</sup>Pb/<sup>206</sup>Pb versus <sup>87</sup>Sr/<sup>86</sup>Sr plot (Fig. 13b), the total <sup>207</sup>Pb/<sup>206</sup>Pb ratio of Epidote-A is represented by analysis #10 of Peverelli et al. (2021; see their Table 5). This is the datum-point closest to the average (<sup>238</sup>U/<sup>206</sup>Pb = 4.62 and <sup>207</sup>Pb/<sup>206</sup>Pb = 0.786) and to the median (<sup>238</sup>U/<sup>206</sup>Pb = 4.35 and <sup>207</sup>Pb/<sup>206</sup>Pb = 0.787) values of the U–Pb isotopic ratios measured in Epidote-A by LA-ICP-MS.





### 345 6 Discussion of the formation mechanisms for the epidote-quartz microfold

#### 6.1 Interplay of epidote dissolution-precipitation and quartz dynamic recrystallization

The formation of a hydrothermal vein entails the crystallization of a mineral assemblage from a mineralizing fluid 350 that fills a fracture (Bons et al., 2012). Although the original morphology of the studied epidote-quartz vein is obliterated bv deformation. euhedral shapes of Epidote-A grains 355 (Fig. 4d) suggest that the formation of Epidote-A in layer 1 is related to vein-filling mineralization, with crystallization occurring in equilibrium with a fluid. While the formation of Epidote-A is ascribed to 360 veining, in the following we address the mechanisms affecting Epidote-B. The microstructures in layers 2-3 differing from those in layer 1 (Sect. 365 5.1), along with the lesser extent of chemical variability across epidote grains in layers 2-3 compared to Epidote-A in layer 1 (Figs. 10–11), suggest that the mechanism of

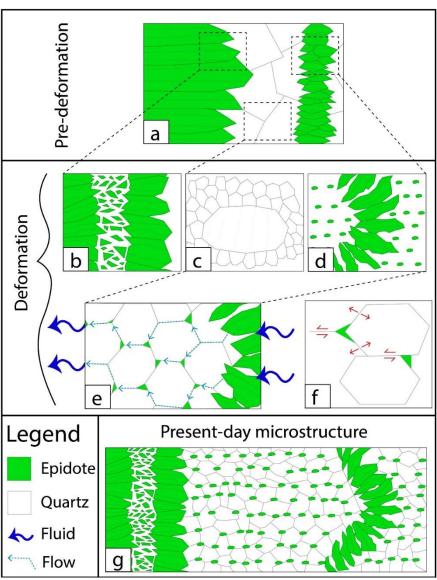


Figure 14: Sketch of the sequence of events affecting the studied epidote vein. (a) Original microstructure formed upon veining. (b) Detail of the fracturing occurring in Epidote-A in layer 1. (c) Dynamic recrystallization of quartz by subgrain rotation. (d) Folding of the epidote band in layer 2 and dissolution—precipitation of Epidote-B. (e) Detail of the viscous granular flow process and dynamic granular fluid pump allowing Epidote-B to precipitate in creep cavities along quartz grain boundaries. (f) Detail of quartz grain boundary sliding allowing dynamic granular fluid pump (modified from Fusseis et al., 2009). (g) Present-day microstructure. Not to scale.

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370 formation of Epidote-B is different than that of Epidote-A. The alignment of Epidote-B grains along the microfold axial

planes (Figs. 4a and 6), and the presence of minute Epidote-B along dynamically recrystallized quartz grains (Fig. 9) indicate

that Epidote-B formed during deformation. The overlap in major and trace element compositions of Epidote-B and Epidote-

A (Figs. 10-11), though, demonstrates that these epidote generations are chemically related. One way to reconcile this

geochemical affinity with different formation mechanisms is a scenario in which Epidote-B formed via deformation-induced

dissolution of Epidote-A grains and (re)precipitation. In fact, a few euhedral epidote grains can still be recognized in layer 2,

which are interpreted as Epidote-A relicts inherited from the original vein morphology. The gradation from epidote-

rich/quartz-poor domains to virtually epidote-free/quartz dominated ones in layers 2-3 (Fig. 14g) suggests that the

microstructure that formed upon veining was similar to that shown in Fig. 14a. In fact, shifts between epidote-rich and

quartz-rich domains are frequent in epidote-quartz veins (e.g., Peverelli et al., 2021; their Figs. 1a and 1c). Further evidence

for fluid-mediated mass transfer is also given by CL images of quartz (Fig. 12), which indicate that the grain boundaries of

quartz interacted with an aqueous fluid. The presence of a fluid during deformation is also supported by fluid inclusions in

dynamically recrystallized quartz.

The transport of dissolved epidote-forming material towards the microfold axial planes (Fig. 14d) calls for a mechanism

allowing the fluid to move throughout the deforming microstructural domain. The presence of tiny Epidote-B grains along

quartz grain boundaries and at triple junctions (Fig. 9) implies that dissolved epidote-forming material is transported and

precipitated into nucleation loci. One mechanism enabling this mass transfer process is dynamic granular fluid pump (Fig.

14e-f; see Fusseis et al., 2009) among dynamically recrystallized quartz. A first deformation step in which quartz grain size

is reduced by dynamic recrystallization via subgrain rotation (Fig. 14c) is suggested by a strong crystallographic preferred

orientation in quartz-dominated domains (Fig. 8). Once the new quartz grains are formed, grain boundary sliding produces

creep cavities (Fig. 14f; see Herwegh and Jenni, 2001; Fusseis et al., 2009; Gilgannon et al., 2017; 2021), thus creating

nucleation loci for Epidote-B. A similar process is described by Gottardi and Hughes (2022) in quartzites deformed at the

brittle-ductile transition in the crust, in which fluid inclusions are redistributed in the quartz matrix by dynamic

recrystallization of quartz by subgrain rotation and grain-boundary migration. A similar microstructure and a similar

interplay of processes are also discussed by Kruse and Stünitz (1999) for in mafic high-temperature mylonites. This first step

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Stünitz, 1999).

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is followed by one in which the grain size and volume abundance of the newly precipitated Epidote-B grains control the grain size of quartz (Fig. 7), hence exerting control on quartz deformation mechanisms by pinning the migrating quartz boundaries (see Olgaard, 1990; Herwegh and Berger, 2004; Herwegh et al., 2011). This allows to suggest a feedback process: (1) quartz grain boundary sliding creates creep cavities in which Epidote-B crystallizes, (2) Epidote-B keeps quartz grain size small, (3) more creep cavities are formed, and (4) more Epidote-B grains are formed. This process is referred to as viscous granular flow (e.g., Fitz Gerald and Stünitz, 1993; Stünitz and Fitz Gerald, 1993; Paterson, 1995; see also Kruse and

6.2 Open-system conditions and external fluids

Epidote dissolution-precipitation and the transport of dissolved epidote material to the loci of Epidote-B crystallization implies the presence of a fluid, whose nature can be assessed by Pb-Sr isotope data. We have mentioned that the epidote microseparates used for solution ICP-MS measurements mix Epidote-A and Epidote-B to unknown proportions, since the mechanical separation of pure Epidote-B is not feasible due to its small grain size. The observation that <sup>238</sup>U/<sup>206</sup>Pb and <sup>207</sup>Pb/<sup>206</sup>Pb ratios of the microseparates do not lie on the same regression as those measured by LA-ICP-MS in Epidote-A (Fig. 13a) suggests that the epidote microseparates include different epidote generation, thus that Epidote-A and Epidote-B crystallized from fluids with different Pb isotope compositions. This is supported by the 87Sr/86Sr ratios of the microseparates being different than that measured in an Epidote-A grain (Fig. 13b; Table 2). There are two principal processes that can induce changes in the Pb and Sr isotope compositions: (1) radioactive decay of <sup>235,238</sup>U into <sup>207,206</sup>Pb and <sup>87</sup>Rb into <sup>87</sup>Sr, or (2) open-system conditions allowing for advection of extraneous Pb and Sr with different isotopic compositions. Process (1) is unlikely in this case because ingrown radiogenic Pb and Sr of such young epidote are negligible. Consequently, the U-Pb-Sr isotope heterogeneity of the two epidote generations infers fluid-mediated addition of extraneous Sr and Pb during crystallization of Epidote-B. Because the vein microstructure carries evidence of a single deformation event (Fig. 14g), it is likely that a second fluid type entered the system during vein deformation and formation of Epidote-B (Fig. 14e-f). Thus, the differences in U-Pb and Sr isotopic data between LA-ICP-MS and solution ICP-MS/TIMS data are interpreted to document mixing between different contributions: (1) one reflecting the isotopic composition of Epidote-A incorporated upon vein formation, and (2) one reflecting the mixing of Epidote-A material with extraneous Pb and Sr brought to the site of Epidote-

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420 B crystallization. The existence of open-system conditions bears the potential for fluid-mediated transfer of chemical

constituents (Figs. 14e-f). The fact that the chemical composition of Epidote-B is identical to, and less variable than, that of

Epidote-A (Figs. 10-11) suggests rock-buffered conditions during fluid mediated crystallization of Epidote-A and Epidote-B

at closely comparable physico-chemical (e.g., P, T, fO<sub>2</sub>) conditions and thus fluid composition. The initial compositional

variability of Epidote-A got thereby homogenized upon dissolution and crystallization into Epidote-B. Considering that the

299 ± 2 Ma old host granitoid (Schaltegger and Corfu, 1992) contains minerals with high concentrations of U, Th and Rb

(e.g., allanite, biotite), even slightly variable contributions of radiogenic Pb and Sr leached from these minerals may produce

a measurable shift in Pb-Sr isotope ratios without a resolvable effect on Sr and Pb fluid concentrations. Also, crystallization

of Epidote-A and Epidote-B most likely occurred at comparable T conditions (ca. 400-450 °C as inferred from the presence

of biotite in the vein; see Goncalves et al., 2012). Hence, fluid/epidote partition coefficients of all elements remained similar

between the two epidote crystallization events. Such a scenario can well account for the compositional uniformity but Pb and

Sr isotopic heterogeneity between Epidote-A and Epidote-B, hence allowing that at least a fraction of the fluid involved in

the deformation of layers 2-3 is of external origin. Peverelli et al. (accepted) measured the H isotope composition of epidote

in this sample, obtaining values that can only be explained as a mixture of end-member waters (e.g., meteoric, seawater, etc.)

as source for the epidote-forming fluids. In light of the present data, it is possible that their measured  $\delta D$  value reflects the

mixing of internal and external fluids upon deformation of the epidote-quartz vein as described above.

6.3 Epidote geochemistry as a result of veining vs. granular fluid pump

The present major and trace element data show that Epidote-B is geochemically similar to, and less variable than, Epidote-A

(Figs. 10-11), and this result bears important implications on epidote crystallization mechanisms. In a rock-dominated

system, the larger geochemical variability of Epidote-A may be due to fluid distillation during epidote crystallization, slight

variations in physico-chemical conditions (e.g., P, T) throughout the veining processes, or pulsating fluid fluxes (i.e.,

multiple Epidote-A generations). The latter is unlikely because all laser spots for U-Pb dating define one statistically robust

regression with MSWD of 0.79, providing no evidence for multiple epidote generations at the current analytical precision.

Such a hypothesis is also inconsistent with trace element data (Fig. 11a-b) defining trends and not distinct epidote

populations. These observations suggest that the geochemical variability of Epidote-A is due to an evolving epidote-forming

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445 fluid chemistry along with more and more epidote precipitating, thus readily removing compatible elements from the fluid

(see also Anenburg et al., 2015). The lesser extent of chemical variation in Epidote-B indicates homogenization of the whole

trace element budget dissolved into the Epidote-B-forming fluid. Viscous granular flow is a dynamic process in which

continuous feedback exists among dissolution of minerals, grain boundary sliding, creep cavitation, mass transfer and

mineral precipitation (e.g., Fitz Gerald and Stünitz, 1993; Stünitz and Fitz Gerald, 1993; Paterson, 1995). This implies that

repeated dissolution and reprecipitation of the same material is likely to occur, and the chemical budget gets compositionally

homogenized with ongoing deformation. Consequently, in the case of the epidote-quartz microfold in layers 2-3, such a

mechanism may account for the chemical homogeneity of Epidote-B. If this respect, granular fluid pump promotes recycling

and homogenization of fluids in deforming polymineralic aggregates.

7 Consequences for epidote U-Pb ages

The inevitable question arising from the scenario developed above is what is the significance of the U-Pb age measured in

Epidote-A. Temperature-driven resetting of the U-Pb system is excluded based on the peak temperature reached in the area

(i.e.  $450 \pm 30$  °C; Challandes et al., 2008; Goncalves et al., 2012) never exceeding the closure temperature for Pb diffusion in

epidote (i.e. > 685 °C; Dahl, 1997). As a consequence, diffusional processes do not have the potential to affect Epidote-A U-

Pb ages. However, the disturbance of the U-Pb isotope system may be affected by dissolution-precipitation mechanisms, as

has been reported for many minerals (e.g., monazite; Tartèse et al., 2011; Williams et al., 2011; Seydoux-Guillaume et al.,

2012; Grand'Homme et al., 2018). Fluid-rock interaction has been shown to affect other isotopic systems as well (e.g., K-

Ar, B; Halama et al., 2014). The microstructural relationships between Epidote-A and biotite in layer 1 are consistent with

both minerals forming together in Alpine times and before vein deformation. Also, the U-Pb isotopic data presented in

Peverelli et al. (2021) resolve a single generation of Epidote-A when plotted in a Tera-Wasserburg diagram. Moreover, the

time-resolved <sup>206</sup>Pb/<sup>238</sup>U ratios corrected for downhole fractionation display flat trends when corrected for zoning in initial

lead by applying a <sup>208</sup>Pb correction (see Fig. 5 of Peverelli et al., 2021). This means that there is no resolvable isotopic

zoning across the ca. 10-12 µm crater depth of the LA-ICP-MS measurements. The trace element data (Figs. 10c and 11;

Table 2) reveal prominent chemical zoning. If any isotopic heterogeneity had been caused by interaction with the

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deformation-related fluid, the crystal rims would be affected to a greater extent than the cores. Consequently, if resolvable isotopic zoning existed in the dated Epidote-A grains, plotting data-points from different domains of isotopic zoning would create scatter of the data-points in a Tera-Wasserburg diagram. However, U-Pb isotope measurements cover all geochemically variable zones (Fig. 10; white circles) and they define a single Tera-Wasserburg regression regardless of their proximity to cores/rims of the analyzed epidote grains. This supports that the dissolution-precipitation processes during Epidote-B formation did not appreciably disturb the U-Pb isotope system in Epidote-A relicts.

#### 8 Conclusions and outlook

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This study combines microstructural and geochemical methods to investigate the formation mechanisms of an epidote-quartz microfold within an epidote-quartz (±biotite) vein. For the first time, we have demonstrated the occurrence of epidote dissolution-precipitation processes, and that this mechanism coexists with quartz dynamic recrystallization, both contributing to grain-size reduction of the system. The genetic link between epidote grains being dissolved and those being reprecipitated is revealed by the overlap of the major and trace element compositions of both generations. We have shown that fluid circulation and the formation of epidote nucleation loci in the deforming system are enabled by dynamic granular fluid pump. Repeated dissolution and (re)precipitation processes of epidote in creep cavities formed among sliding quartz grains also cause chemical homogenization of the epidote-dissolving/forming fluid, and hence of the new epidote generation forming during deformation. This demonstrates the importance of fluid recycling in deformational processes, although the application of Pb-Sr isotope geochemistry recognizes a role for the addition of externally derived fluids mediating mass transfer processes in the studied epidote-quartz vein. Epidote plays an active role in determining the deformation behavior of other minerals, hence in the deformation style of polymineralic aggregates and in producing the resulting microstructures. To our knowledge, before this study, only brittle deformation behavior of epidote had been proven (Masuda et al., 1990; 1995), but no other detailed studies had addressed epidote deformation mechanisms. In this respect, epidote dynamic recrystallization has never been either ruled out or demonstrated. However, since this mineral is widespread in crustal rocks (e.g., Bird and Spieler, 2004; Enami et al., 2004; Franz and Liebscher, 2004; Grapes and Hoskin, 2004; Schmidt and Poli,





2004; Morad et al., 2010), the occurrence of epidote ductile deformation may have effects on the deformation of the continental crust and it calls for a better understanding of this mineral.

#### Appendix A

495 Transmitted light microscope images of the other epidote-quartz veins (samples P2, P3 and Gr0-c) associated with the studied epidote-quartz vein.

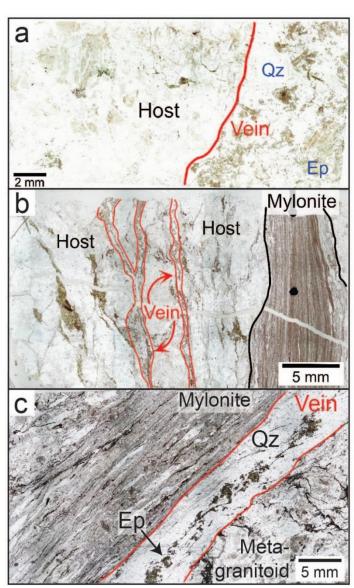


Figure A1: Transmitted light scans of (a) epidote vein P2, (b) epidote vein P3 and (c) epidote vein Gr0-c. Plane-polarized light.





#### Appendix B

#### 515 Transmitted light photograph of the microfold in layers 2–3.

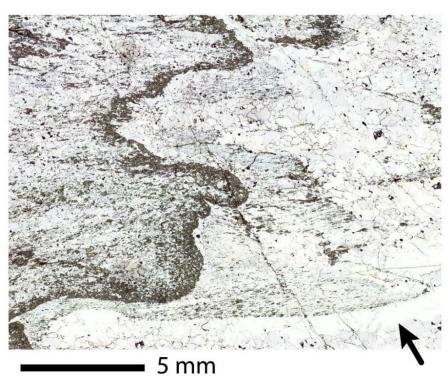


Figure B1: Transmitted light scan of the microfold in layers 2-3. Plane-polarized light.

#### 525 Appendix C

Electron backscatter diffraction (EBSD) maps of dynamically recrystallized quartz by subgrain rotation in layer 3.



Figure C1: Electron backscatter diffraction (EBSD) map showing the orientation of quartz C axis in layer

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**Author contribution** 

VP prepared the samples, carried out grain-size analysis and petrographic description, planned electron microprobe work,

ran trace element measurements by LA-ICP-MS, measured Sr isotopes by TIMS, and prepared the manuscript. AB and MH

supervised the work and greatly contributed to structuring the manuscript. MW supervised clean lab work and TIMS

analyses, and measured Pb isotope data with IMV. TP granted access to the LA-ICP-MS laboratory and was closely involved

in structuring the manuscript and in data evaluation. PL performed work at the electron microprobe and processed the X-ray

maps. All authors read the manuscript and contributed to its improvement.

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**Competing interests** 

The authors declare that they have no conflict of interest.

Code/data availability

All data are included in the manuscript (see tables).

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