Modelling wintertime sea-spray aerosols under Arctic Haze conditions

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Abstract. Anthropogenic and natural emissions contribute to enhanced concentrations of aerosols in the Arctic winter and early spring with most attention focusing on anthropogenic aerosols contributing to so-called Arctic Haze. Less well-studied wintertime sea-spray aerosols (SSA) under Arctic Haze conditions are the focus of this study since they can make an important contribution to wintertime Arctic aerosol abundances. Analysis of field campaign data shows evidence for enhanced local

- 5 sources of SSA including marine organics at Utqiaġvik (formerly known as Barrow, in northern Alaska, United States) during winter 2014. Models tend to underestimate sub-micron, and overestimate super-micron, SSA in the Arctic during winter, including the base version of the Weather Research Forecast, coupled with chemistry, (WRF-Chem) model used here, which includes a widely used SSA source function based on *Gong et al. (1997)*. Quasi-hemispheric simulations for winter 2014 including updated wind-speed and sea-surface temperature (SST) SSA emission dependencies, and sources of marine sea-salt
- 10 organics and sea-salt-sulphate lead to significantly improved model performance compared to observations at remote Arctic sites, notably for coarse mode sodium and chloride which are reduced. The improved model also simulates more realistic contributions of SSA to inorganic aerosols at different sites, ranging from 20-93% in the observations. Two thirds of the improved model performance is from inclusion of the dependence on SSTs. Simulation of nitrate aerosols is also improved due to less heterogeneous uptake of nitric acid on SSA in the coarse mode and related increases in fine mode nitrate. This highlights the
- 15 importance of interactions between natural SSA and inorganic anthropogenic aerosols contributing to Arctic Haze. Simulation of organic aerosols and the fraction of sea-salt sulphate are also improved compared to observations. However, the model underestimates episodes with elevated observed concentrations of SSA components, and sub-micron non-sea-salt sulphate at some Arctic sites, notably at Utqiaġvik. Possible reasons are explored in higher resolution runs over northern Alaska for periods corresponding to the Utqiaġvik field campaign in January and February 2014. The addition of a local source of sea-salt

- 20 marine organics, based on the campaign data, increases modelled organic aerosols over northern Alaska. However, comparison with previous available data suggests that local natural sources from open leads as well as local anthropogenic sources are underestimated in the model. Missing local anthropogenic sources may also explain low modelled (sub-micron) non-sea-sulphate at Utqiaġvik. The introduction of a higher wind speed dependence for sub-micron SSA emissions, also based on Arctic data, reduces biases in modelled sub-micron SSA while sea-ice fractions, including open leads, are shown to be an important factor
- 25 controlling modelled super-micron, rather than sub-micron, SSA over the north coast of Alaska. The regional results presented here show that modelled SSA are more sensitive to wind speed dependence, but that realistic modelling of sea-ice distributions is needed for simulation of local SSA, including marine organics. This study supports findings from the Utqiaġvik field campaign that open leads are the primary source of fresh and aged SSA, including marine organic aerosols, during wintertime at Utqiaġvik and do not suggest an influence from blowing snow and frost flowers. To improve model simulations of Arctic
- 30 wintertime aerosols, new field data on processes influencing wintertime SSA production, in particular for fine mode aerosols, are needed as well as improved understanding about possible local anthropogenic sources.

1 Introduction

The Arctic region is warming faster than any other region on Earth due to carbon dioxide, in particular, and also due to short-lived climate forcers like methane, tropospheric ozone (O₃) and aerosols (*AMAP*, 2015; Allan, 2021). During winter and early
spring, aerosols also affect clouds (aerosol-cloud indirect effects) and, more specifically, cloud droplet number concentrations and size by increasing the long-wave emissivity of clouds (long-wave warming effect) (*Zhao and Garrett, 2015; Horowitz et al., 2020*). At this time of year, elevated aerosol concentrations of black carbon (BC), nitrate (NO₃⁻), non-sea salt (nss) sulphate (SO₄²⁻) and organic aerosols (OA) are observed in the Arctic, a phenomenon known as Arctic Haze (*Rahn and McCaffrey, 1980; Barrie et al., 1994; Quinn et al., 2002*), due to transport of anthropogenic aerosols and precursors from mid-latitude

- 40 sources as well as within-Arctic sources (*Heidam et al., 2004; Quinn et al., 2007; Law et al., 2014, 2017; Schmale et al., 2018)*. Natural aerosol sources such as dust, volcanic emissions and sea-spray aerosols (SSA) also contribute to wintertime Arctic aerosol burdens (*Barrie and Barrie, 1990; Quinn et al., 2002; Zwaaftink et al., 2016; Kirpes et al., 2018)*, with SSA also peaking in the wintertime (*Schmale et al., 2022*). It is important to quantify wintertime natural aerosols, as well as anthropogenic components, since they can affect the Arctic radiative budget via indirect effects (*Schmale et al., 2018*). There are also
- 45 important interactions between SSA and anthropogenic components via heterogeneous uptake on surfaces of SSA, leading to inorganic aerosol formation (*Su et al., 2022*) which can influence the ability of models to simulate Arctic Haze. In this study, we focus on SSA under wintertime Arctic Haze conditions.

SSA are produced by bubble bursting (jet-drop and film-drop formation) on the sea surface due to wind stress during whitecap formation (*Monahan et al.*, 1986). SSA emissions also depend on sea surface temperatures (SSTs) and salinity

50 (*Jaeglé et al., 2011; Sofiev et al., 2011; Revell et al., 2019*). Frost flowers and blowing snow have also been proposed as a source in polar regions during wintertime (*Xu et al., 2013; Huang and Jaeglé, 2017*). SSA are composed primarily of sodium (Na⁺), chloride (Cl⁻), organics and sea-salt (ss) SO₄²⁻. SSA may influence cloud formation, including Arctic mixed-phase

clouds (*Adachi et al., 2022*), since they can act as cloud condensation nuclei (CCN) (*Quinn et al., 2017*), or organics may contribute to ice nucleating particles (INPs) (*Burrows et al., 2013*). Arctic warming is leading to a decrease in summer sea-ice

55 and thinner sea-ice during wintertime (*Stroeve et al., 2012*). Increases in the area of the open ocean or more open leads in seaice may increase winter SSA over Arctic coastal regions, potentially influencing radiative forcing (*Ma et al., 2008; Eidhammer et al., 2010; Partanen et al., 2014; Schmale et al., 2022*).

The ability of models to capture wintertime Arctic aerosols has largely focused on the evaluation of anthropogenic Arctic Haze components, in particular BC and SO₄²⁻ (e.g. *Eckhardt et al. (2015)*, *Whaley et al. (2022)*). *Whaley et al. (2022)* showed
that, in general, models underestimate SO₄²⁻ and BC in winter. Very few model studies have assessed both anthropogenic and natural aerosols, like SSA, and these models do not generally make the distinction between nss and ss components of SO₄²⁻ or OA. A recent study by *Moschos et al. (2022a)* estimated that wintertime Arctic OA have largely anthropogenic origins but a possible contribution from local marine organics was not considered. An analysis of single-particle data from a field campaign near Utqiaġvik, northern Alaska, in winter 2014, showed that, in addition to organic-sulphate Haze aerosols, there
were abundant fresh SSA, based on the presence of Na⁺, Cl⁻, magnesium and sulphur in ratios similar to seawater, that were

- produced locally from open leads and included marine OA originating from secretions from sea ice algae and bacteria (*Kirpes et al., 2018, 2019*). It can be noted that during winter, fresh SSA can be a significant fraction of particulate matter, contributing up to 40% of super-micron (1 to 10 μ m particle diameter) and 25% of sub-micron (up to 1 μ m particle diameter) aerosol mass (*Quinn et al., 2002*). *Kirpes et al. (2018*) also observed aged SSA, which made up the majority of the sub-micron number
- fraction during their campaign at Utqiaġvik. These aged SSA were internally mixed with secondary SO_4^{2-} , or both SO_4^{2-} and NO_3^{-} , and Cl^- was depleted, indicating that multi-phase reactions had occurred during transport. The aged SSA were sampled in air masses influenced by background Arctic Haze and regional northern Alaskan oil field emissions. Their findings support an earlier analysis of Ny-Alesund (Svalbard) data showing that aged SSA were always internally mixed with NO_3^{-} , SO_4^{2-} , and organics (*Chi et al.*, 2015). In fact, heterogeneous reactions occurring on the surface of SSA, involving uptake of
- ⁷⁵ sulphuric, nitric or organic acids, and associated Cl⁻ displacement, are more evident in aged SSA (*Chi et al., 2015*). *Chen et al.* (2016) showed that uncertainties in modelled SSA can have a significant impact on sub-micron and super-micron $NO_3^$ due to heterogeneous uptake of nitric acid (HNO₃) on SSA which produces NO_3^- . Uptake of sulphuric acid on SSA in the marine boundary layer can result in SO_4^{2-} production (*Alexander et al., 2005*). *Li et al.* (2018) and *Wu et al.* (2019) emphasised the importance of heterogeneous reactions occurring on SSA for improved simulation of SO_4^{2-} and NO_3^- size distributions.
- 80 Since nitric and sulphuric acid have largely anthropogenic origins this highlights important links between natural SSA and anthropogenic inorganic aerosols.

This study is motivated by the findings of *Kirpes et al. (2018, 2019)* on wintertime Arctic aerosols, including SSA coated with marine organics, in northern coastal Alaska. Our main objectives are to assess the ability of the regional atmospheric Weather Research Forecast, coupled with chemistry, (WRF-Chem) model to simulate wintertime aerosols, in particular SSA,

85 under Arctic Haze conditions, and to examine model sensitivity to processes which may be influencing SSA over northern Alaska, in particular. We also assess the observed and modelled contribution of SSA to total inorganic aerosols during Arctic wintertime. Firstly, we focus on improving the model SSA emission scheme over the wider Arctic during winter. This includes updating the wind speed dependence, including a dependence on SSTs, and adding sources of marine OA and ss- SO_4^{2-} . Due to the links

- 90 between SSA and other inorganic aerosols, contributing to Arctic Haze, model results are evaluated against observations of all aerosol components at remote Arctic sites. To our knowledge, these aspects have not been considered in previous studies using either WRF-Chem or other models (e.g. *Whaley et al. (2022)*). Secondly, we investigate the sensitivity of modelled SSA to processes influencing SSA, including organics and other inorganic aerosols, at Utqiaġvik using the improved model run at higher resolution over northern Alaska for periods corresponding to the *Kirpes et al. (2018, 2019)* campaign. The sensitivity
- 95 of the model results to the addition of a local source of marine organics, wind speed dependence and sea-ice fractions are investigated based on the findings of this field campaign and other data collected in the Arctic. A possible contribution from frost flowers or blowing snow is also considered.

The model setup, including anthropogenic and natural emissions, is described in Section 2. The aerosol observations used to evaluate the model are introduced in Section 3. Details about the SSA emission scheme, in the base model version, together

100 with improvements to this scheme, are presented in Section 4. Evaluation of simulated SSA, as well as other inorganic and organic aerosols, against Arctic observations, are presented in Section 5, together with an estimation of the contribution of SSA to total inorganic aerosols. Results from the regional study over northern Alaska are presented in Section 6. The implications of our findings for the simulation of SSA under wintertime Arctic Haze conditions are presented in Section 7 (Conclusions).

2 WRF-Chem

105 2.1 Model Setup

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WRF-Chem model version 3.9.1.1 is used in this study. It is a fully coupled, online meteorological and chemical transport mesoscale model (*Grell et al., 2005; Fast et al., 2006*). Recent improvements to the WRF-Chem model over the Arctic are included in the version used here (*Marelle et al., 2017*). The model setup, including meteorological and chemical schemes, is shown in Table 1. Briefly, Yonsei University (YSU - boundary layer), Model Version 5 similarity (MM5 - surface layer) and the NOAH-Multiparameterization Land Surface Model (NOAH MP LSM) are used. More details about the NOAH MP scheme

are given in APPENDIX A.

The well-known processes for aerosols in the atmosphere, like nucleation, evaporation, coagulation, condensation, dry deposition, aerosol/cloud interactions and aqueous chemistry, are included in the Model for Simulating Aerosol Interactions and Chemistry (MOSAIC, *Zaveri et al. (2008)*) aerosol scheme used here. MOSAIC treats all the major aerosol species, such

- 115 as SO_4^{2-} , NO_3^{-} , CI^{-} , ammonium (NH_4^+), Na^+ , calcium (Ca^{2+}), BC, and OA. Reactive inorganic species such as potassium (K^+) and magnesium (Mg^{2+}) are not modelled in MOSAIC. The size distribution of each aerosol species is represented by eight bins, from 0.0391 micrometres (μ m) to 10 μ m: [0.0391 to 0.0781], [0.0781 to 0.1562], [0.1562 to 0.3125], [0.3125 to 0.625], [0.625 to 1.25], [1.25 to 2.5], [2.5 to 5.0] and [5.0 to 10.0] in μ m. Each bin is assumed to be internally mixed, and both mass and number are simulated. As a result, aerosols are aged when emitted (coagulated with other species). The MO-
- 120 SAIC version used in this study also includes 18 irreversible heterogeneous reactions (see Table 1 in Zaveri et al. (2008)),

Table 1. WRF-Chem model setup. The source functions for SSA emissions and their main updates are summarised below. CONTROL includes only *Gong et al. (1997)*, while HEM_NEW includes updates to the SSA emission scheme. See text for details.

Parameterisation scheme	Options					
	Physics (WRF)					
Planetary boundary layer	Yonsei University (YSU) (Hong et al., 2006)					
Surface layer	Pennsylvania State / NCAR Mesoscale					
	Model Version 5 (MM5) similarity					
	(Grell et al., 1994; Jiménez et al., 2012)					
Land surface	NOAH MP (Niu et al., 2011)					
Microphysics	Morrison (Morrison et al., 2009)					
Shortwave & longwave radiation	Rapid Radiative Transfer Model (RRTMG) (Iacono et al., 2008)					
Cumulus parameterization	Kain-Fritsch with cumulus potential (KF-CuP)					
(Berg et al., 2013)						
Cher	Chemistry (WRF-Chem)					
Aerosols	MOSAIC 8-bins (Zaveri et al., 2008)					
Gas-phase chemistry	Statewide Air Pollution Research Center (SAPRC)-99					
	Modified dimethyl sulphide chemistry					
	(Carter, 2000; Marelle et al., 2017)					
Photolysis	Fast-J (Wild et al., 2000)					
	CONTROL					
Sea-salt aerosol emissions	Gong et al. (1997)					
HEM_NEW (u	updates to Gong et al. (1997))					
Sea-spray aerosol emissions (marine organics)	Fuentes et al. (2010, 2011)					
Satellite wind speed dependence, SST dependence	Salisbury et al. (2014), Jaeglé et al. (2011)					
$ss-SO_4^{2-}$ source	Kelly et al. (2010)					

such as the reaction of HNO₃ on sodium chloride (NaCl) to form sodium nitrate (NaNO₃), with depletion of Cl⁻. The reaction between NaCl and sulphuric acid (H₂SO₄) to produce sodium sulphate (Na₂SO₄), with associated Cl⁻ depletion, is also included. Nighttime chemistry, notably heterogeneous hydrolysis of dinitrogen pentoxide leading to HNO₃ formation, is also included (*Archer-Nicholls et al., 2014*). The applied MOSAIC version includes secondary organic aerosol (SOA) formation
from the oxidation of anthropogenic and biogenic species (*Shrivastava et al., 2011; Marelle et al., 2017*) and is combined with SAPRC-99 gas-phase chemistry. In the base model, OA is the sum of SOA and anthropogenic emissions of organic matter (OM). Aqueous chemistry in grid-scale (*Morrison et al., 2009*) and subgrid-scale clouds (*Berg et al., 2015*) is also included. Aerosol sedimentation in MOSAIC is calculated throughout the atmospheric column based on the Stokes velocity scheme, as described in *Marelle et al. (2017*). Wet removal of aerosols by grid-resolved stratiform clouds (precipitation) includes in-cloud

130 and below-cloud removal by rain, snow, and graupel by Brownian diffusion, interception, and impaction mechanisms following

Easter et al. (2004) and *Chapman et al.* (2009). Wet-removal due to subgrid-scale convective clouds (*Berg et al.*, 2015) is also included in this MOSAIC version and described in previous studies (*Marelle et al.*, 2017; *Raut et al.*, 2017).



Figure 1. WRF-Chem simulation domains: (a) d01 is the 100km domain and (b) d02 is the 20km domain. d02 shows sea-ice fractions interpolated at 20km obtained from NCEP FNL at $1^{\circ}x1^{\circ}$ resolution (*NCEP*, 2000). See text for details.

2.2 Anthropogenic and natural emissions

Anthropogenic emissions are from the Evaluating the Climate and Air Quality Impacts of Short-Lived Pollutants version 6
(ECLIPSE v6b) inventory, with a resolution of 0.5° x 0.5° (*Whaley et al., 2022*). Emissions of dimethyl sulphide (DMS) and lightning nitrogen oxides (NO_x) are calculated online in the model (see *Marelle et al. (2017)* and references therein). Dust emissions in MOSAIC are calculated following *Shaw et al. (2008)*. Biogenic emissions for 2014 are calculated online using the Model of Emissions of Gases and Aerosol from Nature (MEGAN) model (*Guenther et al., 2012*). Details about the treatment of SSA emissions and their improvement in the model, are provided in Section 4 and summarised in Table 1.

140 2.3 Simulations

Two simulation domains on a polar stereo-graphic projection are used in this study, as shown in Figure 1. The first (parent) domain (d01) covers a large part of the Northern Hemisphere with 100×100 km horizontal resolution. The boundary and initial conditions are derived from National Centres for Environmental Prediction Final meteorological reanalysis data (NCEP)

FNL) and Model for OZone And Related chemical Tracers (MOZART, *Emmons et al. (2010)*) for atmospheric trace gases
and aerosols. The nested domain (d02), run at a horizontal resolution of 20 × 20km, covers continental Alaska, a small area of northwest Canada, and the Chukchi and Beaufort Seas (see Fig. 1). 50 vertical levels and grid nudging are used for the 100km resolution domain, while spectral nudging, following *Hodnebrog et al. (2019)*, is implemented in the nested domain. WRF-Chem temperatures and winds are nudged at each dynamical step to the reanalysis, and updated every 6 hours, above the atmospheric boundary layer.

- Two simulations at 100km are performed, one using the base model version (CONTROL) and one using CONTROL plus improvements to the SSA emissions (HEM_NEW) (see Section 4). They are run for 4 months from November 2013 until the end of February 2014, with the first two months considered as spin-up. The results are evaluated over the wider Arctic in Section 5. Results from HEM_NEW are then used as boundary conditions for regional runs over northern Alaska at 20 km for two different periods (23–28 January 2014 and 24–28 February 2014) corresponding to the Utqiagvik campaign described
- 155 earlier (*Kirpes et al., 2018, 2019*), KRP18 and KRP19 from now on (see also sub-section 3.2). In this case, the model is run for 4 days prior to the beginning of each campaign and considered as spin-up. Results from a series of sensitivity runs to examine processes affecting SSA on a regional scale are discussed in Section 6. In all runs, model results are output every 3h.

3 Aerosol Observations

The sites discussed in this section are shown in Figure 1.

160 3.1 Routine monitoring sites

Surface mass concentration data (for aerodynamic diameters (defined as d_a) < 10 μ m), from EMEP (European Monitoring and Evaluation Programme), available via EBAS, for Zeppelin, Ny-Ålesund, Norway (78.9N, 11.9W) and Alert, Canada (82.5N, -63.3W), are used to evaluate the 100km model simulations together with total suspended particulate (TSP) (cutoff at 20 μ m) data from Villum Research Station, Station Nord, Greenland (81.6N, -16.7W), referred to as Villum from now on. The data are collected on a daily (Zeppelin) and weekly (Villum, Alert) basis. At Alert, observations for Na⁺, Cl⁻, NO₃⁻ and total SO₄²⁻

- 165 collected on a daily (Zeppelin) and weekly (Villum, Alert) basis. At Alert, observations for Na⁺, Cl⁻, NO₃⁻ and total SO₄²⁻ measured with ion chromatography are used (*Sharma et al., 2019*). This is also the case at Zeppelin (*Aas et al., 2021*). At Villum, the same observations are collected using a filter-pack over a week and analysed using ion-chromatography (Cl⁻, total SO₄²⁻), and cation ion-chromatography (Na⁺). For all the EBAS stations, observed inorganic aerosols (total SO₄²⁻, NO₃⁻) are converted to mass concentrations (μ gm⁻³), to compare to model results, using the ratio of molar weights of NO₃⁻ and SO₄²⁻
- 170 to molar weights of nitrogen and sulphur, respectively. With regard to measurement uncertainties, EBAS documentation notes that uncertainties range between 33% and 36% for Na⁺, total SO₄²⁻, NO₃⁻ and Cl⁻ at Alert. These high uncertainties may be related to uncertainties in the size cut-off of sub-micron filters. Uncertainties in coarse particle observations (d_a < 10 μ m) are based on the difference between high-volume (TSP) filters collected outside and sub-micron filters collected inside.

Fine mode ($d_a \le 2.5 \ \mu m$) mass concentration data from the Interagency Monitoring for Protected Visual Environments 175 (IMPROVE) database is also used for model evaluation for a sub-Arctic site (Simeonof (55.3N, -160.5W)) on the Aleutian Islands, south of Alaska and an inland site, Gates of the Arctic (66.9N, -151.5W), GoA from now on, which is located 391 km south-east of Utqiaġvik town in northern Alaska (see Fig. 1). The samples are collected on-site over 24 hours every three days (*Malm et al., 1994*). At these two sites observations of Na⁺, Cl⁻, organic carbon (OC), NO₃⁻ and total SO₄²⁻ are used. To compare with the OC observations at the two Alaskan sites, modelled OA is divided by 1.8, the reported ratio of OM/OC in

- 180 the documentation for these two stations (*Malm et al., 1994*). Sub-micron ($d_a < 1.0 \mu m$) and super-micron ($1.0 < d_a < 10 \mu m$) surface mass concentration data from the National Oceanic and Atmospheric Administration (NOAA) Barrow Observatory (71.3N, -156.8W), near Utqiaġvik town (Utqiaġvik from now on), are also used in this study, with daily and weekly temporal coverage, respectively. The sampling site is located 8 km northeast of Utqiaġvik, 3km southwest of the Arctic Ocean, covered with snow during winter and 20m above mean sea level (msl), with a prevailing, east-northeast wind off the Beaufort Sea. Na⁺,
- Cl⁻, NO₃⁻ and total SO₄²⁻ mass concentrations are determined by ion chromatography (*Quinn et al., 1998*) and are sampled only for wind directions between 0 and 130 degrees (with 0 degrees indicating north). According to *Quinn et al. (2002)*, measurement uncertainties in sub-micron components of SSA and nss-SO₄²⁻ are below 1.0 ± 6.1% µgm⁻³ (concentration ± 95% uncertainty). The uncertainties in sub-micron NH₄⁺ are below 0.2 ± 7.8% µgm⁻³. The uncertainties in super-micron aerosols can be up to 7 times higher than for sub-micron aerosols, since 7 daily sub-micron samples are collected for every (weekly) super-micron sample. Measurement uncertainties are also due to sampling losses in the inlets. At Utqiaġvik, for particles with a diameter up to 10 µm, losses in the inlet system, from all loss mechanisms, are estimated to be less than 10%,

and losses of particles between the diameters of 0.01 and 1 μ m are below 5% (Sheridan et al., 2001).

At each site observed ss-SO₄²⁻ is calculated from observed Na⁺ concentrations and the mass ratio of SO₄²⁻ to Na⁺ in seawater of 0.252 (*Bowen et al., 1979; Calhoun et al., 1991*) and nss-SO₄²⁻ is the difference between total SO₄²⁻ and ss-SO₄²⁻. Note

that, in some cases, observed nss-SO₄²⁻ has small negative concentrations, due to depletion of ss-SO₄²⁻ through fractionation processes (*Quinn et al., 2002*). We note that, apart from the sub-micron observations at Utqiaġvik and fine mode observations from the IMPROVE database, there are no other sub-micron or fine-mode observations collected routinely in the Arctic, as also reported recently by *Schmale et al. (2022)*. Finally, data from a scanning mobility particle sizer (SMPS), located at Utqiaġvik, are used, providing measurements of particle number distributions at high temporal resolution, with a size range from 8.6 to
985 nm (0.0086 to 0.985 µm). No wind speed criteria have been applied to exclude local (Utqiaġvik town) or regional (e.g. North Slope of Alaska oil fields) pollution. Here, only SMPS observations at Utqiaġvik are used since observations at the other Arctic sites used in this study are not available for winter 2014, as discussed by *Freud et al. (2017*).

The model Stokes aerosol diameters equivalent to d_a , estimated using the *Seinfeld and Pandis (1998)* formula are used. Thus, the diameter of modelled sub-micron particles is up to 0.73 μ m (including the first four MOSAIC bins and a fraction of the 5th

- bin), and super-micron particle diameters are between 0.73 to 7.3 μ m (fraction 5th bin, 6th and 7th bins and fraction 8th bin). Seven MOSAIC bins and a fraction of the 8th bin are used (modelled Stokes $r_a \le 7.3 \mu$ m) to compare with Alert and Zeppelin observations ($d_a < 10 \mu$ m). All model aerosol bins are used to compare with observations at Villum, where the observations are reported as TSP. For each site, modelled aerosols are estimated at the same conditions (temperature, pressure) as the reported observations. Also, observed total OC is assumed to include SOA, anthropogenic OA emissions and marine organics. Thus,
- 210 from now on it will be referred to as tOC, to distinguish from OA defined earlier and OM.

3.2 Campaign data

Details about the field campaign (January 23–27 and February 24–28, 2014) measurements near Utqiaġvik, Alaska can be found in KRP18 and KRP19. Briefly, atmospheric particles were collected using a rotating micro-orifice uniform deposition impactor located 2m above the snow surface at a site located 5km across the tundra from the NOAA Barrow Observatory

215 and inland from the Arctic Ocean. The sampled particles were analysed by computer-controlled scanning electron microscopy with energy scattering X-ray spectroscopy (CCSEM-EDX) to determine the individual particle morphology and elemental composition. The analysed samples were collected either during daytime or nighttime, and only when wind directions were between 75 and 225 degrees, to minimise local pollution influence. Data analysis provided information about the different chemical components as a fraction of the total number of particles sampled during the campaign.

220 4 Model SSA emission treatments and updates

This section introduces the treatment of SSA emissions in the base model version of WRF-Chem using the MOSAIC aerosol scheme, followed by a description of the updates to the SSA emissions implemented in the model. The model is run with the original scheme (CONTROL run) and with the updates (HEM_NEW run). Results from both runs are evaluated against observations in the Arctic, in Section 5.

225 4.1 SSA emissions - CONTROL run

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SSA emission fluxes (F) in MOSAIC are calculated per particle radius r, with 1000 sub-bins per MOSAIC bin, assuming that sea-salt is a simple mix of pure NaCl and using the density function dF/dr (in particles $m^{-2}s^{-1}\mu m^{-1}$) based on *Gong et al.* (1997) (G97 from now on). The G97 source function represents the rate that seawater droplets form per unit area (sea surface) and as a function of particle radius. The fraction of Na⁺ is calculated using the molar weight of Na⁺ and Cl⁻ and then the fraction of Cl⁻ is estimated, with the total being equal to 1. The G97 density function derived from the source function is based on laboratory experiments described in *Monahan et al.* (1986) (MO86 from now on):

$$\frac{dF}{dr} = 1.373 \times U_{10}^{3.41} \times r^{-3} (1 + 0.057 \times r^{1.05}) \times 10^{1.19e^{-B^2}}$$
(1)

where F is a function of U, the 10m-elevation wind speed, r at relative humidity (RH) equal to 80%, and $B = \frac{(0.380 - logr)}{0.650}$. The source function is applied for particles with dry diameters of 0.45 μ m or more (equivalent to model particle diameters). 235 For particles with dry diameters less than 0.45 μ m, a correction is applied based on reported data in *O'Dowd et al. (1997)*, since G97 overestimates the production of small particles (*Gong, 2003; De Leeuw et al., 2011*). G97 is based on the whitecap method, where the emission flux scales linearly with the fraction of the ocean area covered by whitecaps. Over open ocean, the whitecap fraction, W(U), is determined as a function of wind speed (*Monahan and Muircheartaigh (1980)*; MO80 from now on):

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$$W(U) = 3.84 \times 10^{-6} \times U_{10}^{3.41}$$
 (2)

This expression for W(U) is included implicitly in Equation (1) following details provided in MO80. In the base version, SSA emissions are calculated for every grid cell, which is open ocean or salt-water lakes. In this study grid cells which are covered by sea-ice are considered and then the fraction of ice-free grids is used. In this way, SSA emissions from open leads are taken into account. It can be noted that this SSA scheme, based on *Gong et al. (1997)*, is still being used in global and regional models (e.g. the Community Multiscale Air Quality Modeling System (CMAQ), (*Gantt et al., 2015*), Goddard Earth

Observing System (GEOS)-Chem, (*Huang and Jaeglé, 2017*), or in other models (e.g. LOTUS-EUROS) (*Barthel et al., 2019*) to simulate SSA, despite being relatively old. However, modelling studies have shown that G97 overestimates super-micron SSA (e.g. *Jaeglé et al. (2011)*, JA11 from now on) or underestimates sub-micron SSA (e.g. *Archer-Nicholls et al. (2014); Gantt et al. (2015*)).

250 4.2 Updates to SSA emissions - HEM_NEW run

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Here, updates to the model treatment of SSA emissions are described. They are included in the run HEM_NEW, which is also used as boundary conditions for the higher resolution runs over northern Alaska.

4.2.1 Sub-micron SSA emissions including marine organics

- Previous studies have shown that there are large numbers of SSA down to 10 nm (*Ovadnevaite et al., 2014; Cravigan et al., 2015; Xu et al., 2022*). Also, data-based studies in the Arctic (*Kirpes et al., 2019*), and over the Atlantic Ocean (*O'Dowd et al., 2004; Ovadnevaite et al., 2011; Saliba et al., 2019*), suggest that marine organics associated with SSA contribute significantly to natural aerosol composition. Marine organics are not included in G97. However, *Archer-Nicholls et al. (2014)* implemented a scheme in the SSA emission module of MOSAIC based on *Fuentes et al. (2010, 2011)* (F10 and F11 from now on) to include a source flux for marine organics with dry diameters from 0.003 to 0.45 μm, that is coupled to G97 for larger particles.
 260 This scheme is activated in HEM NEW simulations. F10 is applied from the lowest aerosol bin, namely 39 nm. The scheme
- is based on an analysis of data from a mid-latitude cruise investigating the influence of dissolved OM on the production of sub-micron SSA. The F10 SSA source function also depends on MO80 whitecap coverage and high wind speed dependence. Organic fractions equal to 0.2 for the first and second MOSAIC bins, 0.1 for the third bin and 0.01 for the remaining bins are used following the high biogenic activity scenario which assumes high carbon (C):Chlorophyll-a (Chl-a) ratios (see *Lee*
- 265 *et al.* (2010)). F11 found that higher particle organic fractions are expected in algal bloom regions with high C:Chl-a ratios and Chl-a varying between 0.4-10 μ gL⁻¹. The use of the F11 high biogenic activity option is justified since MODIS-Aqua satellite data (https://neo.sci.gsfc.nasa.gov/view.php?datasetId=MY1DMW_CHLORA&date=2014-12-01) for January-February 2014 show that Chl-a south of Alaska, and along the west coast of the United States, varied between 0.3 and 3.0 μ gL⁻¹. *Fujiki et al.* (2009) also found that Chl-a varied between 0.4 and 1.0 μ gL⁻¹ at six stations south of the Aleutian Islands, Alaska,
- 270 during a sub-Arctic cruise in autumn 2005. Details about the F10 SSA source function are given in APPENDIX B. Thus, in the HEM_NEW run, OA also includes marine organics. Possible regional sources of marine organics over northern Alaska are discussed further in Section 6.

4.2.2 Wind speed dependence

As noted earlier, SSA emissions are highly dependent on wind speed and sea state (presence of whitecaps). The G97 scheme,

- and the related parameterisation by *Gong (2003)*, depend on the whitecap method and thus have a high wind speed dependence (see Eq. 1). Several studies have tried to improve upon the whitecap method. *Callaghan et al. (2008)* used an automated whitecap extraction technique to derive two whitecap expressions that differ from MO80, and are based on cubed relationships for U_{10} . For sub-micron SSA, *Ovadnevaite et al. (2012)* showed that source functions, such as *Gong (2003)*, based on the MO80 wind speed dependence, are responsible for an overestimation of the SSA emission flux. They found a lower wind
- speed dependence for small particles, based on an autumn field study off the west coast of Ireland. Other factors, such as the wave field (*Salisbury et al., 2013*) or fetch-dependent threshold for breaking waves (*Revell et al., 2019; Hartery et al., 2020*), have also been shown to affect whitecap lifetime, with implications for SSA production. In a study by *Salisbury et al. (2014)* (SALI14 from now on), satellite data from Quick Scatterometer (QuikSCAT) were used to derive an expression with a lower wind speed dependence compared to MO80. Here, the SALI14 parameterisation is implemented, instead of the MO80
 whitecap fraction expression, since it is based on satellite data analysis providing information with global coverage including the Arctic (e.g. Chukchi Sea and Barents Sea during autumn) and south of Alaska:

$$W(U) = 4.60 \times 10^{-5} \times U_{10}^{-2.26}$$
(3)

Based on Figure 2 in SALI14, the seasonal mean of W(U) using Eq. 3 is lower at latitudes above 40N and 40S compared to MO80 during autumn and winter.

290 4.2.3 SST dependence

Wind speed alone cannot predict SSA variability, and it is important to also include a dependence on SSTs as pointed out by, for example, data-based studies in the Arctic (*Saliba et al., 2019; Liu et al., 2021b*) and mid-latitudes, such as *Ovadnevaite et al. (2014)*. Modelling studies also showed that the application of a SST dependence improves simulated SSA concentrations compared to observations (*Jaeglé et al., 2011; Sofiev et al., 2011; Spada et al., 2013; Barthel et al., 2019*), but not yet imple-

- 295 mented in WRF-Chem. More specifically, previous studies tested different SSA source functions and reported that including a SST dependence improves model results, regardless of the wind speed dependence employed (*Spada et al., 2013; Grythe et al., 2014; Barthel et al., 2019*). However, uncertainties still remain about the influence of SSTs on SSA production (*Revell et al., 2019*), including the role of other factors, such as seawater composition (*Callaghan et al., 2014*) or wave characteristics (e.g. wave speed and breaking wave type, *Callaghan et al. (2012*)), which might be more important than SSTs alone. Here,
- 300 the JA11 SST correction factor is applied when SSTs are between -2°C and 30°C to evaluate the effect of SSTs on sub- and super-micron SSA emissions. SSTs are provided by the reanalyses data, in this case, FNL, and in the presence of sea-ice, SSTs are set equal to -1.75°C. In that case, the SST correction factor is set to the minimum value based on *Barthel et al. (2019)*.

4.2.4 Sea-salt sulphate

A source of ss-SO₄²⁻ is included in the MOSAIC SSA emission scheme (HEM_NEW), since it was not included in the base model version (CONTROL). The mass fraction of ss-SO₄²⁻ is estimated to be 0.252 of the Na⁺ mass fraction based on *Kelly et al.* (2010) and *Neumann et al.* (2016). The fraction of ss-SO₄²⁻ is subtracted from the fraction of Na⁺, Cl⁻, and marine OA. Note that the total fraction of Na⁺, Cl⁻, marine OA, and ss-SO₄²⁻ is equal to 1.0, and additional emissions are not added. We find that, on average, the mass fraction of ss-SO₄²⁻ emissions in our simulations is around 9.9% of the total SSA emissions. This can be compared with the CMAQ model where the ss-SO₄²⁻ emissions are estimated to be 7% of the total SSA emissions 310 (*Kelly et al.*, 2010).



Figure 2. Average absolute differences in super-micron aerosol mass concentrations (in μ gm⁻³) between HEM_NEW and CONTROL during January and February 2014 at the surface. The black x in northern Alaska shows where Utqiagvik is located. The black circle shows Villum in Greenland, the black diamond shows Alert, Canada, while the black pentagon shows Zeppelin, Svalbard. Total SO₄²⁻ is shown. All the results are shown north of 50N. Note the different scales.

5 Evaluation of simulated wintertime SSA and other inorganic aerosols over the Arctic

First, absolute differences in simulated aerosol concentrations between the HEM_NEW and CONTROL results, averaged over January and February 2014, are presented. Model results from the two runs are then evaluated against available observations of, not only Na⁺ and Cl⁻, but also OA and SO₄²⁻ which now include a sea-salt component, and NO₃⁻ which is affected by heterogeneous reactions on SSA. We also show NH₄⁺ for completeness. Lastly, we compare observation-based and modelled contributions of SSA to total wintertime inorganic aerosol concentrations during winter 2014.

5.1 SSA emission updates: HEM_NEW versus CONTROL

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Average absolute differences in super-micron and sub-micron inorganic aerosols between the HEM NEW and CONTROL are shown in Figures 2 and 3, respectively. HEM_NEW simulates less super-micron Na⁺ by up to 20 μ gm⁻³, and Cl⁻ by up to 30 320 μ gm³, especially south of Alaska and the northern Atlantic Ocean. This is due to the combined effect of using a lower wind speed dependence and including the SST dependence (Fig. 2). Inclusion of a SST dependence leads to a larger decrease in locally produced super-micron Na⁺ and Cl⁻ over the Arctic and sub-Arctic ice-free regions, due to lower temperatures north of 50N, compared to using the lower wind speed dependence, based on SAL14, which has a smaller effect. Overall, one-third of the super-micron reductions can be attributed to the lower wind speed dependence and two-thirds to the SST dependence. Super-micron NO₃⁻ is also lower (by up to 1.0 μ gm⁻³) due to less formation of NO₃⁻ via heterogeneous uptake of HNO₃ on 325 SSA. These reactions involving heterogeneous uptake of acid gases also produce HCl, thus depleting Cl^- relative to Na⁺ (Su et al., 2022). The presence of sea-ice also plays a role. Smaller decreases in Na^+ and Cl^- are found north of Alaska (Beaufort Sea) compared to ice-free regions such as the northern Atlantic Ocean. The local influence of sea-ice fraction and open leads on SSA production over northern Alaska is examined further in Section 6. Furthermore, due to the addition of marine organics and ss-SO₄²⁻ in HEM_NEW, there is more super-micron SO₄²⁻, by up to 2 μ gm⁻³, and super-micron OA, by up to 0.6 μ gm⁻³, 330 over marine regions. Super-micron NH₄⁺ slightly increases up to 0.15 μ gm⁻³ over regions where NO₃⁻ increases.

There are smaller decreases in HEM_NEW sub-micron Na⁺ compared to CONTROL, by up to 0.25 μgm⁻³, south of Alaska and in the North Atlantic (Fig. 3). Again, this is due primarily to the introduction of the SST dependence. When using SALI14 lower wind speed dependence alone, there is a small decrease in sub-micron Cl⁻ and a small increase in sub-micron Na⁺
over the Arctic. Sub-micron Cl⁻ also decreases over continental areas, where NO₃⁻ and HNO₃ are higher due to anthropogenic sources (Fig. 3). Heterogeneous uptake on SSA reduces Cl⁻ and increases sub-micron NO₃⁻ by up to 6.0 µgm⁻³ in HEM_NEW over continental regions while the increases over the Arctic Ocean are smaller. This is in contrast to super-micron NO₃⁻ decreases. These results are consistent with the study of *Chen et al.* (2016), also using WRF-Chem with MOSAIC, who noted that, since SSA are primarily present in the coarse (super-micron) mode, this favours the formation of NaNO₃ which is
thermodynamically stable, and limits the formation of NH₄NO₃ which is semi-volatile (*Chen et al.*, 2020). Therefore, lower super-micron SSA in HEM_NEW, results in less super-micron NO₃⁻ and more sub-micron NO₃⁻. We also note that, for these reasons, sub-micron NH₄⁺ also increases, by up to 1.5 µgm⁻³, especially over continental areas, and displays similar regional



Figure 3. The same as Figure 2, but for sub-micron aerosol mass concentrations (in μ gm₋₃.

patterns to sub-micron NO₃⁻. Inclusion of marine organics linked to SSA, leads to increases in sub-micron OA, by up to 1.5 μ gm⁻³, and total SO₄²⁻ increases due to the addition of ss-SO₄²⁻.

345 5.2 Evaluation against observations

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Model results are evaluated against available observations of aerosols at different sites as shown in Figures 4, 5 and 6. These figures are grouped according to the size ranges of the measurements at the different sites as discussed in Section 3.1. Average biases and root mean square errors (RMSEs) between the observations and the model results for January and February 2014 are given in Table 2 and Table C.1 (APPENDIX C), respectively. In the following the main findings are discussed by aerosol component.

SSA (Na⁺ and Cl⁻): Updates to the treatment of SSA emissions in HEM_NEW greatly improves modelled SSA over the Arctic with notable reductions in in Na⁺ and Cl⁻ biases and RMSEs compared to observations at Alert, Zeppelin (d_a < 10 μ m), Villum (TSP), Gates of the Arctic (GoA) (fine mode), and the sub-Arctic site, Simeonof (fine mode). Overall,

Table 2. Biases in aerosol mass concentrations, in μ gm⁻³, averaged over January and February 2014 for the CONTROL and HEM_NEW simulations compared to the observations. NA stands for not available.

	CONTROL	HEM_NEW	CONTROL	HEM_NEW	CONTROL	HEM_NEW	CONTROL	HEM_NEW	CONTROL	HEM_NEW	CONTROL	HEM_NEW
	N	a ⁺		1-	N	$NO_3^ nss-SO_4^{2-}$ $nss-SO_4^{2-/}$ NH_4^+ OA		NO_3^-		\mathbf{NH}_4^+)A
Alert	0.81	0.12	1.05	-0.03	0.28	0.25	0.06	-0.02/ 0.04	0.011	0.01	NA	NA
Villum	1.3	0.27	1.9	0.27	0.25	0.14	0.05	0.04/ 0.07	0.01	0.01	NA	NA
Zeppelin	3.3	0.2	4.9	0.1	0.13	0.2	0.2	0.24/ 0.12	0.01	-0.01	NA	NA
Utqiagvik super-micron	0.3	-0.07	0.27	-0.26	0.26	0.13	0.005	0.006/ -0.02	0.004	-0.001	NA	NA
Utqiaģvik sub-micron	-0.485	-0.489	-0.116	-0.124	-0.065	-0.054	-0.621	-0.47/ -0.12	0.11	-0.06	NA	NA
GoA	0.6	0.2	0.7	0.1	0.3	0.2	-0.04	-0.07/ 0.04	NA	NA	-0.24	-0.21
Simeonof	1.4	0.3	2.0	0.1	0.12	0.08	-0.2	0.05/ 0.09	NA	NA	-0.08	-0.05

HEM_NEW captures the spatial variability between observed Na^+ and Cl^- at the different sites, in particular, the lower observed concentrations at Villum, which is surrounded by sea-ice at this time of year, and higher concentrations at Simeonof and Zeppelin. The extent to which sea-ice is present near different sites is an important factor. For example, the high variability in modelled SSA at Villum at the end of January and the middle of February 2014 is likely to be due to fluctuations in sea-ice fractions around the site (0.9-1.0 in the FNL analyses). At Utqiaġvik, the model captures better super-micron Na^+ , whereas Cl^- is now underestimated due to Cl^- depletion. Sub-micron Na^+ and Cl^- are still underestimated in HEM NEW at this site

- with average biases of about -0.5 µgm⁻³ for Na⁺ and -0.12 µgm⁻³ for Cl⁻ with higher biases during episodes with elevated observed SSA. Here, sub-micron SSA may have been transported to the Arctic from the Pacific Ocean (*Quinn et al., 2002; May et al., 2016*), and thus model underestimations may point to deficiencies in the SSA source function further south or issues related either to long-range transport or to wet and dry deposition treatments in the model. However, the fact that the model agrees better with observations over the wider Arctic, as well as at sub-Arctic Simeonof, provides confidence in the modelled
 long-range transport as a source of Arctic (sub-micron) SSA. HEM_NEW Na⁺ also compares well with reported weekly averaged sub-micron Na⁺ mass concentrations collected during January and February 2014 at Alert (0.1 µgm⁻³ observed, up to 0.08 µgm⁻³ modelled) (*Leaitch et al., 2018*). We also note that, at Utqiaġvik, while *May et al. (2016)* attributed sub-micron SSA to long-range transport, KRP18 estimated that in the sub-micron range, for their analysed samples, 42% were fresh SSA
- N) and depleted Cl⁻, and the remainder included organics and sulphate particles. Thus, model discrepancies may also be due to local processes influencing SSA over northern Alaska. This is investigated further in Section 6.

Nitrate: Improved SSA treatments in HEM_NEW also lead to improved simulation of NO_3^- at some sites, notably Simeonof, GoA, Alert, Villum, and biases are reduced (see Table 2). While modelled super-micron NO_3^- at Utqiagvik is improved, the

with chemical signatures similar to sea-water, and 18% classed as partially aged with enhanced anthropogenic components (S,



Figure 4. Evaluation of modelled aerosol composition against in-situ fine mode aerosol observations in μ gm⁻³, at (a) Simeonof, Aleutians Islands, Alaska and (b) GoA, north of Alaska, in local Alaskan time (AKST). Observations are shown as blue crosses, only when they are available. For observed SO₄²⁻: dark-green circles show ss-SO₄²⁻, while blue crosses are total SO₄²⁻. The black line shows model results from the CONTROL run; the red line shows results from the HEM_NEW run. Model daily averages are shown as black diamonds for CONTROL and red pentagons for HEM_NEW, while grey circles show ss-SO₄²⁻ from the HEM_NEW simulation. See the text for more details. Note the different scales.

model still underestimates certain periods when elevated sub-micron NO₃⁻ is observed, also the case at GoA and Simeonof.
The improved behaviour of modelled NO₃⁻ is, in general, due to reductions in Na⁺ and Cl⁻, leading to less NO₃⁻ production in the coarse mode, especially close to or just downwind of major anthropogenic emission regions at mid-latitudes, and a shift to more NO₃⁻ in the fine mode, as discussed previously. These effects are most evident at Utqiaġvik, where the model can be compared to sub- and super-micron data. Comparison with data from other sites is with either total, coarse plus fine mode, or fine mode aerosol observations, and therefore includes both increases and decreases in simulated NO₃⁻. Overall, these
results illustrate the importance of correctly simulating SSA and its effects on anthropogenic aerosols. While observed NO₃⁻ concentrations are generally lower than other aerosol components, such as Na⁺, Cl⁻ or nss-SO₄²⁻, during Arctic winter, a recent trend analysis study showed that NO₃⁻ is clearly increasing at Alert, especially during the winter months (*Schmale et al.*,

2022). Such increases in NO_3^- may be due to increased NO_3^- formation due to lower acidity following sulphur dioxide (SO₂)



Figure 5. Evaluation of modelled aerosol composition against observations at Utqiagvik in μ gm⁻³ for (a) super-micron and (b) sub-micron in Coordinated Universal Time (UTC) and standard temperature pressure (STP) conditions. Observations are shown only when available. The lines and the symbols are the same as in Figure 4. See the text for more details. Note the different scales.

reductions, that outweigh reductions in NO_x emissions at mid-latitudes (*Sharma et al., 2019*). However, increases in SSA over
the Arctic Ocean, due to reductions in ice-covered waters, may also explain these changes (e.g. *Browse et al. (2014)*) although no significant trends in Na⁺ have yet been detected (*Schmale et al., 2022*).

Sulphate: Figures 4, 5 and 6 show observed ss-SO₄²⁻ and total-SO₄²⁻ together with results from CONTROL and HEM_NEW. With regard to total simulated SO₄²⁻, the addition of ss-SO₄²⁻ improves the model results, for example, at Simeonof where observed fine mode ss-SO₄²⁻ makes a significant contribution (30-80%, up to 0.3 µgm⁻³) to total SO₄²⁻. ss-SO₄²⁻ also contributes between 10-40% of total SO₄²⁻ at Alert and Villum and modelled ss-SO₄²⁻ agrees better with the observations. The remainder is nss-SO₄²⁻, a dominant component of Arctic Haze resulting from long-range transport from sources in Russia and Europe at these sites (*Leaitch et al., 2018; Lange et al., 2018*). Model results are at the lower end (up to 0.3 µgm⁻³) of reported submicron nss-SO₄²⁻ mass concentrations (0.3-1.1 µgm⁻³) at Alert during winter 2014 (*Leaitch et al., 2018*). On the other hand, HEM_NEW further overestimates total observed SO₄²⁻ at Zeppelin due to the inclusion of ss-SO₄²⁻, especially during certain episodes with elevated concentrations. We note that Zeppelin is a mountain site at 471m, and thus discrepancies with the observations may also be due to issues simulating the vertical distribution and transport of nss-SO₄²⁻ from Eurasian source regions

(Hirdman et al., 2010). At Utqiagvik, on the northern coast of Alaska, most of total observed super-micron SO_4^{2-} is ss- SO_4^{2-} (up to 0.18 μ gm⁻³, around 80%), and the inclusion of ss-SO₄²⁻ in HEM_NEW improves agreement with the observations. With regard to total sub-micron SO_4^{2-} , high mass concentrations are observed at Utqiagvik compared to other Arctic sites, consisting

regional emissions (Ioannidis et al. 2023 in prep.). Indeed, at GoA, 391km inland from the coast, and south of the NSA oil

- mostly of nss-SO₄²⁻, peaking at 2.4 μ gm⁻³, much higher than total super-micron SO₄²⁻ (peaking at 0.5 μ gm⁻³), as also reported 400 by Quinn et al. (2002). However, the model underestimates nss-SO₄²⁻ at this site. As noted by KRP18 and KRP19, there is a local influence from the North Slope of Alaska (NSA) oil fields to the east, and these emissions may be underestimated in the model. In a companion paper investigating BC at Utgiagvik, it is estimated that up to 30-50% of BC originates from these
 - fields, the contribution of nss- SO_4^{2-} is more important and ss- SO_4^{2-} is negligible. Here, the model captures total fine mode 405 SO_4^{2-} (peaking up to 0.64 μ gm⁻³) and the addition of ss- SO_4^{2-} does not affect the results. In addition to local sources, difficulties simulating sub-micron nss- SO_4^{2-} at Utqiagvik may be due to an underestimation in the transport of nss- SO_4^{2-} from mid-latitudes to the Arctic or issues related to deposition (as noted earlier for SSA), as also discussed in previous studies such as Whaley et al. (2022). SO_4^{2-} formation mechanisms under dark, cold winter conditions may also be lacking in the model. For
 - example, high concentrations of hydroxymethane sulphonate (HMS) have been measured recently during winter in Fairbanks, 410 Alaskan Interior (*Campbell et al.*, 2022) contributing to secondary SO_4^{2-} formation during Arctic winter, although only a small contribution from HMS to SO_4^{2-} was found in observations at Oliktok Point, situated within NSA oil fields (*Liu et al., 2021a*). Oxidation of SO₂ by O₃ in alkaline SSA could also contribute up to 9% to SO_4^{2-} formation (*Alexander et al.*, 2005). However, the version of WRF-Chem used here does not include such reactions, in common with many chemistry-aerosol models run
 - 415 over the Arctic (Whaley et al., 2022).

Ammonium: NH₄⁺ observations are available at all sites except for Simeonof and GoA. Observed NH₄⁺ concentrations are very low (below 0.2 μ gm⁻³) at Alert and Villum, with higher concentrations observed at Zeppelin. Overall there is a good agreement between the model and measurements, with very low biases and RMSEs in both runs, apart from an underestimation of elevated NH_4^+ at Zeppelin. At Utqiagvik, there is good agreement with super-micron NH_4^+ , except for periods with higher

- observed NH_4^+ (up to 0.1 μ gm⁻³). However, the model underestimates periods with elevated sub-micron NH_4^+ , of up to 0.4 420 μ gm⁻³, which is higher compared to the other sites. Temporal variations in NH₄⁺ during January and February 2014, generally, follow nss- SO_4^{2-} as NH_4^+ preferentially forms ammonium bisulfate and, to a lesser extent, ammonium sulfate in the particle phase (Schmale et al., 2022), and they have common anthropogenic origins. Previous studies also noted that NH₄⁺ is two times higher at Utqiagvik than at Alert, Zeppelin and Villum, while SO_4^{2-} is similar at all the sites (Schmale et al., 2022), possibly
- suggesting differences in aerosol acidity at different sites. This is also found in this study based on the observations and model 425 results (HEM NEW). It is therefore interesting to investigate the effect of the improved SSA emissions on modelled aerosol acidity. For this, we estimate the neutralisation factor f, following *Fisher et al.* (2011). The results are discussed in APPENDIX D. CONTROL tends to predict more acidic aerosols than observed. Based on the observations, most acidic aerosols are found at Alert, Zeppelin and Utgiagvik (super-micron), with somewhat less acidic aerosols at Villum and Utgiagvik (sub-micron).
- This is improved to some degree in HEM_NEW with aerosols becoming less acidic at some sites, notably at Alert and Villum, 430 due to decreases in simulated NO₃⁻. However, modelled sub-micron aerosols at Utqiagvik are less acidic than the observations,



Figure 6. Evaluation of modelled aerosol composition against in-situ aerosol observations with $d_a < 10 \ \mu m$ in μgm^{-3} at (a) Alert, Canada (STP conditions), (b) Zeppelin, Svalbard, and (c) TSP aerosols at Villum, Greenland in UTC. Observations are shown only when available. The lines and the symbols are the same as in Figure 4. See the text for more details. Note the different scales.

Table 3. Calculated fractions of observed and modelled (HEM_NEW) SSA to total inorganic aerosol mass concentrations. For each site, SSA are defined as the sum of Na⁺, Cl⁻ and ss-SO₄²⁻. Total is defined as the sum of SSA and inorganic aerosols. Inorganic is the sum of nss-SO₄²⁻, NH₄⁺ and NO₃⁻ for each site except for Simeonof and GoA where inorganic is the sum of nss-SO₄²⁻ and NO₃⁻. Total_all below is defined as the sum of SSA, nss-SO₄²⁻, NH₄⁺, NO₃⁻, BC, OA and dust (model only). The aerosol size range for SSA, Total and Total_all varies per site and corresponds to observed aerosol sizes as described in Section 3.

Sites	SSA/Total [obs]	SSA/Total [HEM_NEW]	SSA/Total_all [HEM_NEW]
Simeonof (fine mode)	0.73	0.84	0.74
GoA (fine mode)	0.20	0.44	0.33
Utqiagvik-sub-micron	0.60	0.22	0.13
Utqiagvik-super-micron	0.93	0.57	0.54
Alert (coarse mode)	0.59	0.54	0.45
Villum (TSP)	0.32	0.63	0.52
Zeppelin (coarse mode)	0.56	0.75	0.62

due to the underestimation of $nss-SO_4^{2-}$. Overall, the updates to SSA emissions lead to somewhat less acidic anthropogenic aerosols over the Arctic, again highlighting the importance of interactions between SSA and other inorganic aerosols.

Organic aerosols: Only two sites provide tOC fine-mode observations ranging from between 0.15-0.3 μgm⁻³ at Simeonof
and 0.15-0.5 μgm⁻³ at GoA during January and February 2014. The inclusion of marine organics in HEM_NEW improves modelled OA, especially at the coastal Simeonof site. Since observations at other sites are not available for winter 2014, results are compared with other reported measurements. *Shaw et al. (2010)* reported sub-micron OA at Utqiaġvik around 0.3 μgm⁻³ during winter 2008 (November to February). However, a more recent study by *Moschos et al. (2022a)* reported lower wintertime OA concentrations (d_a < 10 μm) at this site (around 0.1 μgm⁻³), attributed mostly to primary-anthropogenic or
Haze OA originating from Eurasia. However, modelled OA for the same size range is only up to 0.05 μgm⁻³. At Villum, *Nielsen et al. (2019)* also reported higher sub-micron OA observations, peaking at 2.2 μgm⁻³ in February 2015, attributed mostly to Arctic Haze influence (up to 1.1 μgm⁻³) with secondary influences from hydrocarbon-like organics (up to 1.0 μgm⁻³) and marine sources (up to 0.2 μgm⁻³). Modelled OA in HEM_NEW at this site does not exceed 0.1 μgm⁻³. Overall, the model underestimates Arctic OA in common with many other models (*Whaley et al., 2022*). These discrepancies may be due missing

445 or underestimated anthropogenic or natural sources. For example, it is known that there are large uncertainties in anthropogenic OA emissions (*Marelle et al., 2017*). The possibility of a wintertime marine OA source over northern Alaska is explored further in Section 6.

5.3 Contribution of SSA to total inorganic aerosols

Lastly, we assess the contribution of SSA to total inorganic aerosols in the Arctic during wintertime since previous studies noted
that they can make an important contribution to total sub-micron and super-micron mass fractions at this time of year (*Quinn* et al., 2002; May et al., 2016; Kirpes et al., 2018, 2019). Moschos et al. (2022b) also showed SSA dominate wintertime PM₁₀

(particulate matter with $d_a \le 10 \ \mu m$) mass concentrations at remote Arctic sites, including Alert (56%), Baranova (northern Russia) (41%), Utqiaġvik (66%), Villum (32%), and Zeppelin (65%). In contrast, at sites such as Tiksi (northern Russia) and Pallas (Finland), SO_4^{2-} and OA dominate (70% and 55%, respectively). To investigate the contribution of SSA to total mass

- 455 concentrations during the period of this study, observed and modelled fractions of SSA to "total" (SSA plus inorganic) aerosols are estimated (see Table 3). It should be noted that this fraction varies between sites since not all components were measured. Taking into account the observations available at each site, the fraction of SSA to total SSA plus inorganics is higher at all the coastal sites (Utqiagvik, Alert, Simeonof, Villum) and Zeppelin ranging from 54 to 93%. Only at the GoA and Villum is the fraction of SSA smaller (20% and 32%, respectively). SSA fractions, calculated using the HEM_NEW results, show similar
- 460 patterns compared to the observations with fractions ranging between 44% and 84%. An exception, is Utqiaġvik where the modelled fraction is lower than in the observations due to low simulated sub-micron SSA concentrations. When taking into account all aerosol components in the model, including OA, BC and dust, SSA is dominant at Simeonof, Utqiaġvik (super-micron), Zeppelin and Villum (more than 54%), whereas at Alert, SSA contribute about 45%. This analysis shows that SSA are an important fraction of total inorganic aerosols at most Arctic coastal sites during wintertime.
- 465 Overall, the results presented here show that the simulation of Arctic SSA, and other inorganic and organic aerosols, is improved as a result of the updated SSA emission treatments. In particular, simulated aerosols, including the coarse mode or super-micron fraction, are improved compared to the observations. The results also show that it is important to include natural SSA emissions of ss-SO₄²⁻ and marine organics, although the latter are highly uncertain. Missing anthropogenic sources could also be contributing underestimation of OA and nss-SO₄²⁻. Many models in a recent Arctic Monitoring and Assessment
- 470 Programme (AMAP) model evaluation of Arctic composition also showed similar discrepancies, attributed to issues with anthropogenic emissions, or model transport, deposition and aerosol formation (*Whaley et al., 2022*). The results presented here also confirm the importance of interactions between SSA and other inorganic aerosols via heterogeneous uptake, affecting mass concentrations and size distributions, notably NO_3^- , and thus model ability to capture wintertime Arctic Haze.

6 Regional processes influencing SSA over northern Alaska

- 475 Possible processes affecting emissions of SSA on a regional scale over northern Alaska are now examined in more detail, and in particular, those which may explain low modelled sub-micron SSA at Utqiaġvik. Model simulations are run at 20km over northern Alaska for shorter periods in January and February 2014 corresponding to the KRP18 measurement campaign. The boundary and initial conditions are taken from HEM_NEW. The sensitivity of modelled SSA to a local source of marine organic aerosols, wind speed dependence, and the representation of sea-ice fraction is investigated (see Table 4 for details
- 480 about the simulations). Differences between runs with and without specific sensitivity tests are examined sequentially for the February period, before evaluating a run including the main changes against observations at Utqiaġvik during the January and February periods. The possible role of blowing snow and frost flowers is also addressed.

Table 4. Description of the regional-scale WRF-Chem model simulations at 20km resolution over northern Alaska. See text for details.

Simulation Name Description				
Regional simulations [20km]				
ALASKA_CONTROL_JAN	HEM_NEW run at 20km, 23-28 January 2014			
NEW_ALASKA_JAN	including regional updates as in NEW_ALASKA_FEB			
ALASKA_CONTROL_FEB	HEM_NEW run at 20km, 24-28 February 2014			
LOC_ORG_FEB	+ Local source marine organics (Kirpes et al., 2019)			
SSA_WS_DEP_FEB	+ Sub-micron SSA wind-speed dependence (Russell et al., 2010)			
NEW_ALASKA_FEB	+ ERA5 sea-ice fraction (all regional updates)			

6.1 Local source of marine organics

The F10 parametrisation used in the 100km HEM_NEW run is based on C:Chl-a from a cruise at mid-latitudes. Whilst phytoplankton blooms may not be expected in the high Arctic winter, previous studies have shown evidence of sea ice biological 485 activity under low light conditions coupled with decreased sea ice in the Arctic (Krembs et al., 2002; Lovejoy et al., 2007; Hancke et al., 2018). Analysis of data collected over the Arctic and North Atlantic during winter, and the winter-spring transition, also showed that the majority of sub-micron OM is highly correlated with Na⁺ concentrations (Russell et al., 2010: Shaw et al., 2010; Frossard et al., 2011; Leaitch et al., 2018). More specifically, Russell et al. (2010) (RUS10 from now on) analysed samples from the International Chemistry Experiment in the Arctic Lower Troposphere (ICEALOT) cruise and found 490 that most OM in the North Atlantic and the Arctic is composed of carbohydrate-like compounds containing organic hydroxyl groups from primary ocean emissions. Frossard et al. (2014) (FRSS14 from now on) investigated the sources and composition of atmospheric marine aerosol particles based on the analysis of various samples, including ICEALOT, reporting that ocean-derived organic particles include primary marine OA. In particular, they calculated the ratio of OC:Na⁺ as a metric for comparing the composition of model-generated primary marine aerosol and seawater, and reported OC:Na⁺ ratios of 0.45 in 495 atmospheric marine aerosol particles. KRP19 also reported that, during their campaign in 2014, almost all individual SSA had thick organic coatings made up of marine saccharides with average C:Na mole ratios of 0.5 and 0.3 for sub-micron and supermicron SSA, respectively. They also identified open sea-ice leads, enriched with exopolymeric substances, as contributing to OA in winter SSA. Here, elemental fractions for sub- and super-micron aerosols, sampled during the KRP19 campaign, are used to better constrain modelled marine OC emissions. The ratio of sub- and super-micron OC:Na⁺ is calculated, following 500

- FRSS14, and using the elemental fractions from KRP19, as an indicator of the presence of a local source of marine organics. The organic fraction of SSA emissions in WRF-Chem is increased from 0.2 to 0.4 for sub-micron (1st and 2nd MOSAIC bins), from 0.1 to 0.4 for the 3rd MOSAIC bin and from 0.01 to 0.11 for the remaining MOSAIC bins. Note again that no additional SSA mass is added.
- Figure 7 shows the sensitivity of the model results to including a larger marine organic fraction over the regional domain. Sub-micron OA concentrations increase by a small amount, by up to 0.009 μ gm⁻³, especially south-west of Alaska and



Figure 7. Average absolute differences in aerosol mass concentrations between LOC_ORG_FEB and ALASKA_CONTROL_FEB during the February campaign for sub-micron Na⁺, OA, NO₃⁻ (in μ gm⁻³). All the results are shown at the surface. See text and Table 4 for more details. Utqiagvik is shown by the black dot. Note the different scales.

along coastal areas, including around Utqiaġvik. There are only two available daily observations at GoA during the February simulation period to evaluate the model results. The model captures better observed tOC at the end of February in the run (LOC_ORG_FEB) with higher organic fractions (not shown here). However, it underestimates tOC on 25 February when the observed tOC reached 0.33 μ gm⁻³. As mentioned previously, this discrepancy could also be due to missing local anthropogenic OA sources. Higher OA fractions in the super-micron leads to lower Na⁺ and, as result lower NO₃⁻. As indicated above in Section 5, a decrease in super-micron NO₃⁻ results in an increase in sub-micron NO₃⁻. Sub-micron Na⁺ increases probably due to the formation of NaNO₃ in the model. In the following runs, higher organic fractions are used instead of those from F10.

6.2 Wind-speed dependence

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- 515 In the 100km HEM_NEW run, a lower wind speed dependence based on satellite data is used since it improves modelled SSA compared to observations at many sites over the Arctic as discussed in Section 5. However, RUS10 found evidence for higher wind speed dependence in the Arctic based on data collected during the Arctic leg of the ICEALOT cruise. They found that wind speed is a good predictor of a marine factor, calculated using positive matrix factorization, for sub-micron organic matter (OM1_{sea}). Their analysis showed a high correlation between OM1_{sea}, sub-micron sodium (Na⁺1) and wind speed at
- 520 18m (correlation r equal to 0.90 for the North Atlantic and Arctic region, see Table S3 Supplementary Material in RUS10). Average $OM1_{sea}$ concentrations (0.2 μ gm⁻³) reported by RUS10 for the eastern Arctic Ocean are about half those reported at Utqiaġvik, for example, by *Shaw et al. (2010)* during wintertime.

In a sensitivity run, results from RUS10 are used to include a higher wind speed dependence for sub-micron SSA. This linear dependence differs from the power dependencies included in G97, SAL14, and other studies, but is based on empirical relationships determined from analysis of data collected in the Arctic. Equations (5) and (6) from the RUS10 analysis for the

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Figure 8. Average absolute differences in aerosol mass concentrations of (a) sub-micron Na⁺, OA, NO₃⁻, in μ gm⁻³ between SSA_WS_DEP_FEB and LOC_ORG_FEB. (b) The map on the left shows the average value of SSA emission fluxes in μ gm⁻²s⁻¹ during the February campaign and the map on the right shows the average differences between SSA_WS_DEP_FEB and LOC_ORG_FEB emission fluxes in μ gm⁻²s⁻¹. All the results are shown at the surface. Utqiagvik is shown by the black dot. Note the different scales.

Arctic legs of their cruise are applied to the model as a correction factor:

$$Na^{+}1 = 0.022 \times U_{18} - 0.012 \tag{4}$$

$$OM1_{sea} = 0.025 \times U_{18} - 0.049 \tag{5}$$

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where U_{18} is wind speed at 18m in ms⁻¹, for wind speeds between 2 and 14 ms⁻¹ (Figure 2, RUS10). RUS10 used Na⁺¹ as a proxy for sub-micron NaCl, and subsequently SSA, because Na⁺¹ equalled sub-micron Cl⁻¹ on a molar basis for the North Atlantic and Arctic sampling regions. Thus, Equation (5) is also used to estimate a correction factor for Cl⁻. Here, wind speeds in the first model layer are used, i.e. around 26m. Differences in U_{18m} and U_{26m} reach a maximum of 1 ms⁻¹ (see Fig. E1 in APPENDIX E). Comparison with radiosonde data at Utqiaġvik shows that the model performs well in terms of winds and temperatures (see APPENDIX E) and the role of meteorology on aerosols is not discussed further here. The correction factors

are only applied to simulated number and mass of the SSA emissions when modelled wind speeds are between 2 and 14 ms⁻¹, and when RUS10-calculated sub-micron SSA emissions are greater than model calculated SSA. In this way, SSA emissions are enhanced during periods with higher wind speeds.

To illustrate the sensitivity of the results to applying this correction, Fig. 8 shows differences in sub-micron aerosol mass concentrations compared to the run including local marine organics, as well as model SSA emission fluxes, the latter being

- the sum of dry mass emissions calculated in the model. The SSA emission flux is affected over ice-free model grids leading to increased SSA production east and west of Utqiaġvik (by up to $0.015 \ \mu gm^{-2}s^{-1}$) while the highest increases are southwest of Alaska (by up to $0.035 \ \mu gm^{-2}$). This results in an increase of 0.25, 0.19 and 0.11 μgm^{-3} in sub-micron Na⁺, NO₃⁻ and OA, respectively, over the Utqiaġvik region and southwest Alaska during the February campaign. These results further illustrate the sensitivity of SSA emissions to wind speeds, in this case affecting fine mode aerosols. These results are in contrast to previous
- 545 studies finding stronger wind speed dependencies for larger SSA particles, such as *Liu et al. (2021b)* who analysed aircraft data, including over the Arctic. However, size dependent source functions need to be developed for the Arctic region.

6.3 Sea-ice fractions

The sensitivity of modelled SSA to prescribed sea-ice fractions during wintertime and the role of leads, is also investigated since KRP19 already pointed out the importance of using realistic sea-ice distributions to simulate marine aerosols. High

- 550 spatial resolution images of sea-ice cover are available, including during the polar night, from a radar operating on top of a building in Utqiaġvik town (71°17'13" N, 156°47'17" W), 22.5m above sea level, with a range of up to 11km to the northwest (*Druckenmiller et al., 2009; Eicken et al., 2011*). *May et al. (2016)* previously showed increased super-micron Na⁺ mass concentrations during periods of elevated wind speeds and lead presence, in a multi-year study using the sea ice radar data at Utqiaġvik. Between 23-28 January 2014, when the winds at Utqiaġvik were easterly, the radar showed that the coastal area
- 555 east of Utqiaġvik featured open leads (KRP19). From 24-28 February 2014, the west coastal area also featured leads as also shown by Moderate Resolution Imaging Spectroradiometer (MODIS) satellite images (KRP19). To examine the sensitivity of modelled SSA emissions to sea-ice cover, ERA5 sea-ice fractions with a resolution of 0.25° x 0.25° are used instead of FNL fraction at 1.0° x 1.0° resolution. Note that only the sea-ice fraction field is different, while the rest of the meteorological fields are from FNL.
- Results for February are shown in Fig. 9. The SSA emission flux (Fig. 9a) increases over a small region west of Utqiaġvik and across the North Slope of Alaska due to decreased sea-ice fractions, but decreases just to the east of Utqiaġvik and southwest of Alaska due to increased sea-ice fractions. Sub-micron Na⁺ slightly increases along the north coast of Alaska and around Utqiaġvik, by up to 0.1 μ gm⁻³ (see Fig. 9b). Larger super-micron Na⁺ are simulated by up to 0.4 μ gm⁻³ around Utqiaġvik, and decreases by up to 0.4 μ gm⁻³ southwest of Alaska (Fig. 9c). SSA emission fluxes increased more during February (0.035
- 565 μ gm⁻²s⁻¹) compared to January (0.015 μ gm⁻²s⁻¹), since there is more sea-ice in the region east of Utqiaġvik and south west of Alaska in the January simulation (not shown here). Two further simulations are performed to explore model sensitivity to seaice fractions. First, ERA5 sea-ice fractions are set equal to zero to the north, west, and east of Utqiaġvik to examine the effect of having ice-free conditions and the presence of open leads locally (as seen by the radar). Second, ERA5 sea-ice fractions



Figure 9. Average differences between ALASKA_NEW_FEB and SSA_WS_DEP_FEB showing the effect of switching from FNL to ERA5 sea-ice fractions during the February campaign for (a) SSA emission flux (μ gm⁻²s⁻¹), (b) sub-micron Na⁺ and (c) super-micron Na⁺ mass concentrations in μ gm⁻³. All the results are shown at the surface. Utgiagvik is shown by the black dot. Note the different scales.

are set equal to 0.75 north, west, east of Utqiaġvik and northwest of Alaska. In both cases, the model is run on a windy day (28 February 2014). The first sensitivity test leads to an increase in SSA emission fluxes by up to 0.2 μ gm⁻²s⁻¹ when sea-ice fractions equal zero and to an increase of up to 1.2 μ gm⁻³ and 0.05 μ gm⁻³ in super-micron and sub-micron Na⁺, respectively. The second sensitivity test yields similar results. This is because ERA5 sea-ice fractions are higher (more sea-ice) than the test case (0.75) leading to increases in the SSA emission fluxes especially east of Utqiaġvik. Again, super-micron SSA (increases of up to 1.5 μ gm⁻³) are affected more than sub-micron SSA. These results illustrate the regional sensitivity of super-micron SSA rather than sub-micron SSA to prescribed sea-ice fractions. Missing mechanisms influencing sub-micron SSA emissions may need to be included in the model such as SSA production of, in particular ultrafine particles, from breaking waves in the surf zone (*Clarke et al., 2006*). However, information about wave-breaking activity in the surf zone during winter along the

6.4 Evaluation against observations in northern Alaska

northern Alaskan coast is needed to address this.

Results from runs at 20km with and without the main changes included in the sensitivity tests (local source of marine organics, higher wind speed dependence and ERA-5 sea-ice fractions) are compared to sub-micron aerosol observations at Utqiaġvik for both the January and February 2014 periods (see Fig. 10). We note that there are no super-micron observations available due to the weekly sampling frequency. It is interesting to compare to these periods since the observations show different behaviours. While observed sub-micron Na⁺ and Cl⁻ concentrations reached up to 2.5 µgm⁻³ in February, they did not exceed 1 µgm⁻³ during January. As noted earlier such high concentrations were not observed at Alert and Villum during January and February 2014. The January run, including all the updates (ALASKA NEW JAN), captures better sub-micron Na⁺ and Cl⁻ (reduced



Figure 10. Evaluation of daily averaged modelled aerosol composition against in-situ daily observations at Utqiaġvik during the (a) January and (b) February 2014 campaign periods in UTC and STP conditions. The black lines show model results from the ALASKA_CONTROL_JAN/FEB; the red lines show the ALASKA_NEW_JAN/FEB, while the daily average observations are shown as blue crosses. The corresponding model daily averages are shown as black diamonds for the ALASKA_CONTROL_JAN/FEB runs and as red pentagons for ALASKA_NEW_JAN/FEB runs. Green circles show observed ss-SO₄²⁻. The grey lines and pentagons show modelled ss-SO₄²⁻ for ALASKA_CONTROL_JAN/FEB and dark turquoise pentagons show modelled daily average ss-SO₄²⁻ for ALASKA_CONTROL_JAN/FEB. Note the different scales. See the text for more details.

biases and RMSEs - see APPENDIX F, Table F1) although it underestimates observations by up to 0.6 and 0.8 μ gm⁻³ (Fig. 10a), respectively, while sub-micron NO₃⁻ is slightly overestimated. Biases and RMSEs (see APPENDIX F, Table F2) are also slightly improved for February although sub-micron Na⁺, Cl⁻ are still underestimated by up to 1.5 μ gm⁻³. Slightly more NO₃⁻ is simulated in January, even if the model still underestimates the elevated observations. Up to six times more OA is simulated during both periods in better agreement with reported observations (*Moschos et al., 2022b*), as discussed in section 5.2. The sensitivity tests discussed in this section do not directly address the model underestimation of elevated episodes of sub-micron total SO₄²⁻, since it is mostly nss-SO₄²⁻, and thus the changes are small (during both simulation periods). We note that these

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runs at 20km are for periods including elevated aerosols and are thus more challenging for the model to reproduce. Runs at higher resolution may be needed to better resolve, for example, sea-ice distributions.



Figure 11. Averaged modelled (black) and measured (blue) particle number size distributions (y-axis), in particle number(#)/cm⁻³, during (a) January and (b) February 2014 campaign periods at Utqiagvik as a function of diameter (D_p) in nm. The model results are from the ALASKA_NEW_JAN and ALASKA_NEW_FEB simulations. The black dot is plotted in the middle of each MOSAIC bin. The line left and right of each dot indicates the minimum and the maximum of range each bin.

Observations of particle number concentration are also used to validate the regional model results at Utgiagvik (see Fig. 11). High number concentrations are observed during both periods, up to 10^3 particles per cm³, especially for particle sizes less than 20 nm. Freud et al. (2017) reported similar wintertime magnitudes in the accumulation mode (diameter range 100-150 nm) at Utgiagvik, averaging between 1×10^2 and 2×10^2 particles per cm³, whereas smaller number concentrations were reported for particles less than 50nm than shown here. These differences can be explained by the fact that the SMPS data used here does not 600 exclude local/regional pollution based on wind speed criteria unlike Freud et al. (2017). Local or regional pollution influenced by Utgiagvik town or NSA oil fields could lead to new particle growth events (small particles) in the absence of sunlight, as discussed by Kolesar et al. (2017). Freud et al. (2017) also noted that particle number concentrations are higher at Utgiagvik and Tiksi (sites in proximity to local sources) compared to other Arctic sites (Alert, Villum and Zeppelin). The model tends to underestimate observed number concentrations, especially in the 4^{th} (312.5 to 625.0 nm) and 5^{th} (625.0 to 1250 nm) MOSAIC 605 bins, even if the model compares better in January when measured number concentrations are lower. This is consistent with the evaluation of sub-micron SSA and other aerosol components, particularly for episodes when observed aerosols were enhanced (Fig. 10). Note that the model results cannot be compared to measurements smaller than 39 nm because MOSAIC does not represent these aerosols explicitly and nucleation is parameterised. Inclusion of a source function to account for ultrafine SSA

emissions, for example, from breaking waves at the surf zone, could also lead to improved model results (*Clarke et al., 2006*).

Overall the results the results presented so far in this Section show that modelled sub-micron SSA (Na⁺, Cl⁻), and as a consequence NO_3^- , are more sensitive to using a higher wind speed dependence than sea-ice fractions, over northern Alaska, based on estimated biases and RMSEs for each test simulation (not shown here). Sea-ice fractions have a greater effect on super-micron SSA mass concentrations. Modelled sub-micron OA are more sensitive to a higher wind speed dependence and, to a lesser extent, the introduction of an additional source of local marine organics. However, the latter is highly uncertain.

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6.5 Are blowing snow and/or frost flowers a source of sub-micron SSA during wintertime at Utgiagvik?

Lastly, we consider whether enhanced SSA, in particular in the sub-micron size range, at Utgiagvik could be due to blowing snow or frost flower sources. As noted earlier, KRP19 found no evidence of blowing snow or frost flowers at this site but that SSA originated from open leads during wintertime. Their findings are supported by the earlier laboratory study of Roscoe et al. (2011) who reported that frost flowers are not an efficient source of SSA. However, Shaw et al. (2010) found that during winter

- 620 at Utgiagvik surface frost flowers forming on sea and lake ice are a source of marine-derived OM. Modelling studies including a source of blowing snow and frost flowers suggest that they are contributing to SSA at this time of year at Utqiagvik, Alert and Zeppelin (Xu et al., 2013, 2016; Huang and Jaeglé, 2017; Rhodes et al., 2017).
- To investigate whether blowing snow or frost flowers could also be a source of SSA during the campaign at Utgiagvik, depletion factors are estimated following Frev et al. (2020) (FR20 from now on). FR20 reported that blowing snow was the 625 main source of SSA rather than frost flowers and open-leads in Antarctic wintertime, based on SO_4^{2-} and bromide (Br⁻) depletion in SSA being indicative of a blowing snow origin, and not seawater. Other studies also suggested that blowing snow and frost flowers near Utqiagvik are characterised by SO_4^{2-} depletion compared to seawater (Douglas et al., 2012; Jacobi et al., 2012).
- Here, depletion factors are calculated using modelled and observed sub-micron aerosol mass concentrations during the 630 campaign periods. More specifically, total SO_4^{2-} depletion relative to Na⁺ (DF_{SO_4^{2-}}), Na⁺ depletion relative to Cl⁻ (DF_{Na⁺}) and Br^- depletion relative to Na^+ (DF_{Br^-} only for observations in this case) are calculated using the following equation:

$$DF_{\rm x} = 1 - \frac{R_{\rm smpl}}{R_{\rm RSW}} \tag{6}$$

where R is the mass ratio (x:y), of species x,y in the model or in the sample (smpl), and in reference seawater (RSW) 635 (*Millero et al.*, 2008). A depletion factor (DF_x) between zero (small) and 1 (strong) indicates 0–100% depletion, whereas DF_x less than zero indicates enrichment. FR20 suggested, based on depletion of SO_4^{2-} relative to Na⁺, that most SSA originates from blowing snow on sea-ice with minor contributions from frost flowers, and not from open leads.

The average values of modelled and observed DFs are shown in Table 5. Total SO_4^{2-} is enriched relative to Na⁺ in both the observations and the model results during both campaign periods, in contrast to FR20 who reported substantial depletion. In

February, observed and model results both indicate SO_4^{2-} enrichment relative to seawater, whereas in January, model results 640 are less enriched compared to the observations, possibly due to underestimation of nss-SO₄²⁻. In January, observed total SO₄²⁻ concentrations are 7.56 times higher than in reference seawater, possibly due to internal mixing with anthropogenic nss- SO_4^{2-} ,

Table 5. Average sub-micron modelled and observed depletion factors, following *Frey et al. (2020)*, during the campaign periods in January and February 2014 at Utqiagvik. Model results for ALASKA_NEW_JAN and ALASKA_NEW_FEB simulations are shown, respectively. Observations refer to sub-micron data from NOAA.

Depletion Factors	Model	Observations						
January campaign								
DF _{SO4} ²⁻	-0.94	-7.56						
DF _{Na} +	-0.95	-0.09						
Februa	February campaign							
$DF_{SO_4^2}$	-2.2	-2.15						
DF _{Na} +	-1.2	-0.19						
DF _{Br} -	-	0.063						

as noted by KRP18. Modelled total SO₄²⁻ is less enriched than the observations (0.94 times higher than in reference seawater), likely due the model underestimation of nss-SO₄²⁻. FR20 did report a case of enrichment due to possible contamination from the ship, an anthropogenic source. The Na⁺ depletion factor also shows enrichment during both campaigns, albeit more negligible in the observations than in the model. Observed Na⁺ depletion relative to Cl⁻ is 1.09 or 1.19 times more than in reference seawater, during January and February, respectively. Our analysis suggests that blowing snow and frost flowers are not a significant source of SSA, at least during this campaign in winter 2014.

SSA can also play an important role in polar tropospheric O_3 and halogen chemistry through the release of active bromine during spring (Fan and Jacob, 1992; Simpson et al., 2007; Peterson et al., 2017). Reactions involving bromine are an important 650 sink of O₃ (e.g. (Barrie, 1986; Marelle et al., 2021)). Br⁻ depletion relative to Na⁺ is calculated only during February since observed Br⁻ was zero during the January campaign period. The results for February show a small enrichment indicating a seawater origin. The observed mass ratio of Br⁻ to Na⁺ ranges between 0.0057 and 0.0059, while the mass ratio of Br⁻ to Na⁺ in reference seawater is equal to 0.006. FR20 reported no or little Br⁻ depletion relative to Na⁺ due to Br⁻ losses at 655 the surface and small depletion further aloft (in Antarctica). For a more comprehensive analysis, observations are required at different locations and altitudes across northern Alaska. We note that the version of WRF-Chem used in this study does not include halogen chemistry. It has since been implemented in a later version by Marelle et al. (2021) to examine springtime O₃ depletion events at Utgiagvik. Heterogeneous reactions on SSA from the sublimation of lofted blowing snow were also included. Their results suggested that blowing snow could be a source of SSA during spring although it should be noted that this model version, including blowing snow as a source of SSA, overestimated SSA ($d_a < 10 \mu m$) at Arctic sites, such as Alert 660 and Villum during spring and wintertime conditions were not examined.

Finally, modelled and observed molar ratios of sub-micron Cl⁻:Na⁺ and SO₄²⁻:Na⁺ are estimated to further examine the origins of SSA and compare our findings with KPR19 (see Table 6). The averaged molar ratios of sub-micron Cl⁻:Na⁺ and SO₄²⁻:Na⁺ derived here for the campaign periods (Table 6) agree with KRP19 (Cl⁻:Na⁻ equal to 1.08, see KRP19 supplement
- Table S3, text and references within). They indicate a seawater origin, and confirm the findings of KRP19 that there was no

Table 6. Average modelled and observed molar ratios for sub-micron SSA, following *Kirpes et al. (2019)*, during the campaign periods in January and February 2014 at Utqiaġvik. Model results from ALASKA_NEW_JAN and ALASKA_NEW_FEB simulations are used. Observations refer to sub-micron data from NOAA.

Molar ratios	Model	Observations					
January campaign							
total-SO ₄ ²⁻ :Na ⁺	0.12	0.55					
Cl ⁻ :Na ⁺	0.71	1.1					
Februa	February campaign						
total-SO ₄ ²⁻ :Na ⁺	1.5	0.2					
Cl ⁻ :Na ⁺	0.8	1.08					

evidence for blowing snow and frost flowers as a source of SSA during the Utgiagvik campaign, also in agreement with previous studies (May et al., 2016). Model averaged molar ratios are smaller in magnitude than the observations. These discrepancies could be due to the fact that the model underestimates sub-micron SSA and SO_4^{2-} , for the reasons noted earlier. Differences could also be due to issues with modelled SSA lifetime and chemical processing during long-range transport. May et al. (2016) used Cl⁻:Na⁺ molar ratio enrichment factors as an indicator of long-range transport influence on SSA at Utqiagvik. They 670 reported that Cl⁻ depletion was larger for aged sub-micron than for aged super-micron SSA due to a longer lifetime, in line with other studies (Leck et al., 2002; Hara et al., 2002). The regional results (ALASKA NEW JAN, ALASKA NEW FEB) indicate that modelled sub-micron Cl^- has undergone significant atmospheric processing. Thus, the regional model results, influenced by the simulation at 100km, have too much aged sub-micron SSA, while the contribution from locally produced 675 sub-micron SSA may be too low in the model (modelled enrichment factors equal to 0.5 and 0.4 during January and February simulation periods, respectively, lower than the threshold (0.75) defined by May et al. (2016)). On the other hand, modelled enrichment factors for super-micron are equal to 0.6 and 0.85 during January and February, respectively, indicating that there is a possibly background influence on super-micron SSA during January, while they are locally produced during February. KRP18 reported the presence of both fresh (locally-produced) SSA and aged (partially Cl⁻-depleted) SSA for sub-micron 680 SSA, while super-micron were mostly fresh (KRP18, Figure 2). Based on the analysis above (including observations), there is little evidence suggesting that blowing snow or frost flowers are a significant source of SSA at Utqiagvik during winter, and open leads are an important primary source, in agreement with KRP19.

6.6 Conclusions

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In this study, the ability of the WRF-Chem model to simulate wintertime Arctic aerosols is assessed with a particular focus on SSA under Arctic Haze conditions. The inclusion of updated treatments of SSA emissions leads to improved simulation of SSA over the wider Arctic compared to the still widely used *Gong et al. (1997)*-based source function included in the base model. Na⁺ and Cl⁻ biases are reduced by a factor of 7 to 16 compared to observations at Alert, Villum and Zeppelin, and by a factor of 4 compared to super-micron Na⁺ and Cl⁻ data at Utqiaġvik. The addition of a SST dependence has a larger effect on modelled SSA compared to updating the wind speed dependence, and is responsible for two-thirds of the reductions

- 690 in super-micron/coarse mode SSA, due to low SSTs in the Arctic. The use of a more realistic lower wind speed dependence, based on satellite data, also results in lower super-micron SSA, but up to 5 times less compared to the SST dependence. In addition to uncertainties in wind speed and SST dependencies influencing production of SSA, other factors such as seawater composition, wave characteristics, fetch and salinity may also be playing a role and should be considered in future versions of WRF-Chem. Other SST dependencies could also be tested which could increase sub-micron SSA at low temperatures (*Sofiev*
- 695 *et al.*, 2011; Salter et al., 2015; Barthel et al., 2019). In addition, missing sources of ultrafine SSA particles, for example, due to breaking waves at the coast, could also be included by defining a surf zone in the model (*Clarke et al.*, 2006). In all cases, more field data is needed to understand and develop source functions for SSA specific to the Arctic during winter.

Results from this study also highlight the importance of interactions between SSA and other inorganic aerosols, notably NO_3^- , which have largely anthropogenic origins, and contribute to wintertime Arctic Haze. Improved simulation of Na⁺ and

- 700 Cl⁻ leads to less coarse mode and more fine mode NO_3^- in the model, in better agreement with the observations. This is due to less formation of NO_3^- via heterogeneous uptake of HNO₃, primarily in the coarse mode, and more NO_3^- in the fine mode, in line with previous mid-latitude studies. As a result, simulated aerosols in the updated model are slightly less acidic in the Arctic, improving agreement at some Arctic sites, even if the model tends to simulate aerosols, which are too acidic (at some sites).
- Marine organic aerosols are also activated in the model since they are an important component of SSA in the Arctic, and globally, and a source of ss-SO₄²⁻ is also added. Simulated OA is improved at the Simeonof sub-Arctic site with reduced biases, by up to a factor of 4, although, in general, OA is underestimated at sites over the wider Arctic. The addition of ss-SO₄²⁻ agrees well with ss-SO₄²⁻ derived from the observations at most Arctic sites and leads to improved modelled total SO₄²⁻. However, at Zeppelin and Villum, which are dominated by nss-SO₄²⁻, this additional source results in further overestimation. While super-
- 710 micron SO_4^{2-} , primarily of sea-salt origin, is captured better at Utqiaġvik on the northern Alaskan coast, sub-micron SO_4^{2-} , which is primarily nss- SO_4^{2-} , is underestimated at this site during episodes with elevated concentrations, and also at GoA further inland. Model discrepancies in OA and nss- SO_4^{2-} may be due to missing local anthropogenic emissions, coupled with missing heterogeneous or dark reactions leading to secondary aerosol formation. In the case of OA, primary marine emissions may also be underestimated. It can be noted that such underestimations are a common feature in other models (*Whaley et al.*,
- 715 2022). The above, combined with uncertainties in model transport and wet and dry deposition processes contribute to model deficiencies in simulating wintertime Arctic aerosols (*Whaley et al.*, 2022).

Model sensitivity to different processes affecting wintertime SSA over northern Alaska is explored further with the aim to understand, in particular, model underestimation of sub-micron SSA at Utqiagvik during winter 2014 when field data analysis showed that marine emissions from open leads were an important source of SSA, including marine organics (KRP18, KRP19).

720 Based on observed ratios of OC:Na⁺ from the Utqiaġvik campaign, a local source of marine organics is included in model runs at 20km over northern Alaska. This results in higher modelled OA, in better agreement with previous measurements at this site, and other sites such as Alert, although the model still tends to underestimate reported data. The sensitivity of modelled SSA over northern Alaska to using a higher wind speed dependence, based on Arctic data, is also investigated. This leads to an increase in modelled sub-micron SSA, with the model performing better during January

- 725 than in February. Model sensitivity to prescribed sea-ice fractions is also explored. In a run with ERA5 instead of FNL sea-ice fractions, modelled super-micron SSA are more sensitive to sea-ice treatments than sub-micron SSA. In general, modelled sub-micron SSA are more sensitive to the use of a higher wind speed dependence rather than the distribution of sea-ice. To improve model simulations in this region, field campaigns are needed to study processes influencing wintertime production of SSA and to determine more realistic sea-ice fractions which vary on at least a daily basis. The use of satellite sea-ice data,
- combined with higher resolution simulations over Utqiagvik, will also help to gain further insights into the influence of open

leads on production of SSA, including marine organics, during wintertime.

Missing local anthropogenic sources could also explain some of the discrepancies in modelled sub-micron SSA. For example, anthropogenic sources of Cl^- may need to be considered, such as coal combustion, waste incineration, and other industrial activities (*Wang et al., 2019*) which are not included in current emission inventories. WRF-Chem, and models in general, also

735 lack anthropogenic emissions of Na⁺, which could possibly account for up to 30% of Na⁺, as noted by *Barrie and Barrie (1990)*. However, the analysis of depletion factors and molar ratios, presented here for Utqiaġvik, shows that the main source of fresh SSA is from marine sources including open ocean or leads. We also find no evidence for frost flowers or blowing snow as a source of SSA at Utqiaġvik, in agreement with the findings of KRP19 and previous studies (*May et al., 2016*). Further insights into wintertime marine SSA sources, including organics are needed, as well as improved quantification of local anthropogenic emissions.

Overall, we find that wintertime SSA at remote Arctic sites contribute between 54% and 84% to total inorganic SSA (observations and improved model results), in agreement with previous findings, that SSA are important contributors to super-micron (coarse mode, TSP) mass concentrations. Ice fractures and the area of open ocean are likely to become more important with decreasing sea-ice cover in the Arctic as a result of climate warming. This may lead to more SSA which can act as CCN

- or INPs with implications for Arctic aerosol-cloud indirect feedbacks, notably long-wave radiative effects which dominate in winter (*Eidhammer et al., 2010; Partanen et al., 2014*). As well as ground-based measurements, vertical profiles of SSA components are also needed to better understand SSA sources and their impacts on clouds. Such studies will ultimately help to reduce uncertainties in estimates of aerosol-cloud indirect radiative effects and the magnitude of the associated radiative cooling (*Horowitz et al., 2020*) or warming (*Zhao and Garrett, 2015*).
- 750 *Code availability.* The code used to calculate SSA emissions in this study is available on Zenodo as: Ioannidis et al.(2023) https://doi.org/10.5281/zenodo.7502210

Data availability. ECLIPSE v6b emissions are available online at https://previous.iiasa.ac.at/web/home/research/rese

- Norwegian Institute for Air research: http://ebas.nilu.no/. Observations for Villum are obtained after personal communication with co-

- 755 author Henrik Skov. Observations from the IMPROVE database can be obtained from: http://views.cira.colostate.edu/fed/QueryWizard/. Sub- and super-micron aerosol mass concentrations at Utqiaġvik, Alaska can be obtained from the follow link: https://saga.pmel.noaa.gov/ data/stations/. The SMPS data are obtained from ftp://ftp.cmdl.noaa.gov/aerosol/brw/smps/. High spatial resolution images of sea-ice cover are obtained from (http://seaice.alaska.edu/gi/data/barrow_radar). Meteorological data used in this study at Utqiaġvik are from National Oceanic and Atmospheric Administration / Earth System Research Laboratory / Global Monitoring Division (NOAA/ESRL/GMD) Baseline
- 760 Observatories (https://esrl.noaa.gov/). Radiosonde data (every 12h) are derived from Integrated Global Radiosonde Archive version 2 (IGRA 2) (*Durre et al. (2018*)).

Table A1. WRF-Chem Land surface (NOAH MP) parametrisation scheme. "Opt_" indicates the namelist option for NOAH MP.

NOAH MP Parametrization					
Dynamic Vegetation (DVEG)	On				
Stomatal Resistance	Ball-Berry Ball et al. (1987),				
	<i>Collatz et al. (1991),</i>				
	Collatz et al. (1992), Bonan (1996), Sellers et al. (1996)				
Surface layer drag	Original NOAH Chen et al. (1997)				
coefficient (opt_sfc)					
Soil moisture for	NOAH (soil moisture)				
stomatal resistance (opt_btr)					
Runoff (opt_run)	TOPMODEL with groundwater Niu et al. (2007)				
Supercooled liquid					
water (opt_frz)	no iteration Niu and Yang (2004)				
Soil permeability (opt_inf)	linear effects, more permeable (Niu and Yang, 2006)				
Radiative transfer (opt_rad)	modified two-stream				
	(gap = F(solar angle, 3D structure)<1-FVEG)				
	Yang and Friedl (2003), Niu and Yang (2004)				
Ground surface albedo (opt_alb)	BATS Yang ZL. and Vinnikov. (1997)				
Precipitation (snow/rain)	Jordan (1991)				
partitioning (opt_snf)					
Soil temperature lower	TBOT at ZBOT (8m) read from a file				
<pre>boundary (opt_tbot)</pre>	(original NOAH)				
Soil/snow temperature	semi-implicit; flux top boundary condition				
time scheme (opt_sfc)					
Surface resistance to					
evaporation/sublimation (opt_rsf)	Sakaguchi and Zeng (2009)				
Glacier treatment (opt_gla)	include phase change of ice				

Appendix A

Following *Monaghan et al. (2018)*, NOAH-MP parameter file MPTABLE.TBL has been modified, and it can be used for simulations over Alaska. These modifications improved the model's capability to capture cold surface temperatures and meteorological profiles (e.g. wind speed, relative humidity, temperature) over Alaska during winter. See Table A1.

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Appendix B

Fuentes size-resolved sea-spray source flux $(dF_0/dlogD_{p0})$ in m⁻²s⁻¹:

$$\frac{dF_0}{dlogD_{p0}} = \frac{dF_p}{dlogD_{p0}} \times W(U) = \frac{Q}{A_b} \times \frac{dN_T}{dlogD_{p0}} \times W(U)$$
(B1)

770

where W(U) is Monahan and O'Muircheartaigh whitecap coverage (*Monahan and Muircheartaigh, 1980*), $dF_p/dlogD_{p0}$, in m⁻²s⁻¹, is the size-resolved particle flux (F_p) per unit time and water surface covered by bubbles, D_{p0} (in nm) is the dry diameter, Q is the sweep air flow, A_b is the total surface area covered by bubbles, $dN_T/dlogD_{p0}$ is the sub-micron particle size distribution (the sum of four log-normal modes) and is equal to:

$$\frac{dN_{\rm T}}{dlogD_{\rm p0}} = \sum_{i=1}^{4} \frac{dN_{\rm T,i}}{dlogD_{\rm p0}} = \sum_{i=1}^{4} \frac{N_{\rm T,i}}{\sqrt{2\pi} \times \log\sigma_{\rm i}} \times exp[-\frac{1}{2} \times (\frac{\log \frac{D_{\rm p0}}{D_{\rm p0g,\rm i}}}{\log\sigma_{\rm i}})^2 \tag{B2}$$

where i is the sub-index for the mode number (1 to 4) and N_i, $D_{p0g,i}$ and σ_i are the total particle number, geometric mean diameter and geometric standard deviation for each log-normal mode . N_{T,i} and $D_{p0g,i}$ depend on parameters a_i and β_i derived from polynomial and exponential regressions and can be found in Table 5 in *Fuentes et al. (2010)*. N_{T,i} and $D_{p0g,i}$ of the log-normal modes comprising the size distributions as a function of the diatomaceous OC<0.2 μ m content of seawater.

Table C1. Root-mean-square errors (RMSEs), in μ gm⁻³, averaged over January and February 2014 for CONTROL and HEM_NEW simulations compared to the observations. NA stands for not available.

	CONTROL	HEM_NEW	CONTROL	HEM_NEW	CONTROL	HEM_NEW	CONTROL	HEM_NEW	CONTROL	HEM_NEW	CONTROL	HEM_NEW
	N	a ⁺	с	1–	N	O_3^-	$nss-SO_4^{2-}$	nss-SO ₄ ²⁻ / ss-SO ₄ ²⁻	N	H_4^+	C	DA
Alert	0.91	0.18	1.2	0.19	0.3	0.19	0.1	0.09/ 0.03	0.011	0.01	NA	NA
Villum	1.4	0.3	2.1	0.3	0.26	0.15	0.1	0.06/ 0.07	0.01	0.01	NA	NA
Zeppelin	4.4	0.3	6.5	0.3	0.4	0.3	0.3	0.3/ 0.2	0.08	0.07	NA	NA
Utqiagvik super-micron	0.37	0.25	0.48	0.51	0.3	0.17	0.05	0.06/	0.004	0.004	NA	NA
Utqiaģvik sub-micron	0.66	0.67	0.361	0.364	0.162	0.158	0.88	0.73/ 0.18	0.11	0.09	NA	NA
GoA	0.9	0.3	1.2	0.3	0.3	0.3	0.2	0.16/ 0.08	NA	NA	0.28	0.27
Simeonof	2.5	0.6	3.7	0.7	0.23	0.2	0.25	0.25/ 0.19	NA	NA	0.1	0.08

Appendix C

The RMSEs are shown for each site and aerosol component in Table C1.

Table D1. Estimated mean neutralized factor, f, using the observations and the results from the two quasi-hemispheric simulations (100km), CONTROL and HEM_NEW at the different sites. f is not estimated for Simeonof and GoA sites as there are no observations of NH_4^+ .

	Observations	CONTROL	HEM_NEW
Alert	0.26	0.14	0.26
Villum	0.46	0.12	0.26
Zeppelin	0.19	0.12	0.11
Utqiaģvik	0.18	0.01	0.08
super-micron	0.10	0.01	0.00
Utqiaģvik	0.4	0.6	0.7
sub-micron	0.4	0.0	0.7

780 Appendix D

To investigate aerosol acidity, the mean neutralized factor (f) is calculated as the ratio of NH_4^+ to the sum of $(2nss-SO_4^{2-})$ + NO_3^-), in molar concentrations, following Fisher et al. (2011), for sites in the Arctic with available observations of these aerosols. When f is equal to 1 then aerosols are assumed to be neutralized, while when f < 1 then aerosols are acidic, and more acidic when f is closer to zero (Fisher et al., 2011). In general, higher molar concentrations were observed for nss- SO_4^{2-} compared to NO₃⁻ and NH₄⁺. Table D1 shows f for observations and the two 100km simulations at the different sites. Overall, 785 modelled f increases due to the improved treatment of SSA and the associated influence on NO_3^- via heterogeneous reactions. Since aerosols are assumed to be internally mixed in the model, NH_4^+ and $nss-SO_4^{2-}$ mass concentrations also vary between the two simulations. Thus, aerosols in HEM_NEW tend to be less acidic (e.g. at Alert and Villum), due to NO₃⁻ decreases in the coarse-mode/TSP size range. This leads to better agreement with the observed f at Alert, in particular. At Villum, observed aerosols are less acidic than in the model. This could be due to the fact that the model has more NH_4^+ compared 790 to the observations. Only small changes are found at Utgiagvik between the two runs, and the model tends to have aerosols which are slightly more acidic (super-micron) and less acidic (sub-micron) compared to the observations. The small increase in model sub-micron f at Utqiagvik could be due to the increase in sub-micron NO_3^- and insignificant changes in NH_4^+ and nss-SO₄²⁻. Differences with the observed values could be explained by underestimation of nss-SO₄²⁻ at this site. The calculated f for observations could also be biased low (too acidic), since some of the NO_3^- and SO_4^{2-} are present as Na_2SO_4 and $NaNO_3$ 795 in the atmosphere, which are not measured.

Appendix E

Surface observations are used to validate the meteorological conditions at Utqiagvik in winter 2014. The model is validated against the surface (hourly) observations obtained from NOAA. Radiosondes data are also used to evaluate model performance at different altitudes. Utgiagvik site is located at latitude: 71.2N and longitude: -156.7W. Figure E1 below shows the comparison

800



Figure E1. Average temperatures, in degrees C, and wind speeds (WS), in ms⁻¹, as a function of altitude (m), up to 4km, during (a,b) January and (c,d) February campaign in 2014, at Utqiagvik, Alaska. The radiosonde observations are shown as black circles. The blue pentagons show the model results for the CONTROL simulation (at 100km) and the red diamonds show the model results for the NEW_ALASKA_JAN and NEW_ALASKA_FEB. Observations are from IGRA2 and are available every 12h (0Z and 12Z, UTC). For the comparison, model output at 0 and 12Z UTC are used. The corresponding horizontal lines show the standard deviations.

between the average observed and model temperatures and winds as a function of altitude, for January and February 2014. Figure E2 shows the time series of observed and modelled 2m and 10m temperatures and winds at Utqiaġvik during January and February 2014. In both figures, the model results are from the CONTROL run (100km) and NEW_ALASKA_JAN and NEW_ALASKA_FEB simulations at 20km.



Figure E2. Time series of observed and modelled 2m and 10m temperatures, and 10m wind speeds (WS), at Utqiagvik, Alaska, in UTC. The observations shown in red are from the NOAA Observatory. The blue line shows the results for the CONTROL simulation at 100km, while the black lines show results for ALASKA_NEW_JAN and ALASKA_NEW_FEB simulations at 20km. The observations are hourly, while the model output is every 3h.

Table E1. Biases and RMSEs, in temperature (°C), wind speed (ms⁻¹) and wind directions (degrees), are calculated between ALASKA_NEW_JAN, ALASKA_NEW_FEB and in-situ meteorological observations from the NOAA Observatory during the campaign periods in January and February 2014. Absolute biases are calculated as the difference between model simulations and the observations.

	Janua	ry 2014	February 2014		
	Bias	RMSE	Bias	RMSE	
2m Temperature	0.1	1.9	-1.0	3.2	
10m Temperature	-0.03	1.8	-0.66	2.7	
10m Wind speed	0.08	1,4	-0.33	1.7	
10m Wind direction	-11.2	13.2	-11.2	39.0	

Table F1. Absolute biases and RMSEs, in aerosol mass concentrations (in μ gm⁻³,) at Utqiaġvik, north of Alaska, during January 2014 for ALASKA_CONTROL_JAN and ALASKA_NEW_JAN simulations at 20km, compared to observations.

	ALASK	A_CONTROL_JAN	ALASI	KA_NEW_JAN
	Bias	RMSE	Bias	RMSE
Na ⁺	-0.31	0.38	-0.22	0.27
Cl-	-0.50	0.58	-0.35	0.44
NO_3^-	-0.040	0.07	0.039	0.06
SO_4^{2-}	-0.4	0.41	-0.39	0.41

Table F2. Absolute biases and RMSEs, in aerosol mass concentrations (in μ gm⁻³) at Utqiagvik, north of Alaska, during February 2014 for ALASKA_CONTROL_FEB and ALASKA_NEW_FEB simulations at 20km, compared to observations.

	ALAS	KA_CONTROL_FEB	ALASKA_NEW_FEB			
	Bias	RMSE	Bias	RMSE		
Na ⁺	-1.30	1.40	-1.29	1.39		
Cl-	-1.91	1.92	-1.90	1.90		
NO_3^-	-0.20	0.40	-0.18	0.38		
\mathbf{SO}_4^{2-}	-1.03	1.33	-1.01	1.31		

805 Appendix F

Biases and RMSEs calculated between ALASKA_NEW_JAN and ALASKA_NEW_FEB and the observations of aerosol composition for Utqiagvik at 20km. See Tables F1 and F2. See main text for details.

Author contributions. The first author (EI) implemented the updates, performed the simulations and the analysis, and drafted the paper. KSL designed the study, contributed to the interpretation of results and the analysis, and to writing the paper. JCR, LM and TO contributed to

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References

Aas, W., Eckhardt, S., Fiebig, M., Platt, S. M., Solberg, S., Yttri, K. E., and Zwaaftink, C. G.: Monitoring of long-range transported air

- pollutants in Norway. Annual Report 2020., NILU rapport, 2021.
 - Adachi, K., Tobo, Y., Koike, M., Freitas, G., Zieger, P., and Krejci, R.: Composition and mixing state of Arctic aerosol and cloud residual particles from long-term single-particle observations at Zeppelin Observatory, Svalbard, Atmospheric Chemistry and Physics, 22, 14421–14439, 2022.

Alexander, B., Park, R. J., Jacob, D. J., Li, Q., Yantosca, R. M., Savarino, J., Lee, C., and Thiemens, M.: Sulfate formation in sea-salt aerosols:

- 845 Constraints from oxygen isotopes, Journal of Geophysical Research: Atmospheres, 110, 2005.
 - Allan, R. P.: Climate Change 2021: The Physical Science Basis: Working Group I Contribution to the Sixth Assessment Report of the Intergovernmental Panel on Climate Change, WMO, IPCC Secretariat, 2021.

AMAP: AMAP Assessment 2015: Black carbon and ozone as Arctic climate forcers, 2015.

- Archer-Nicholls, S., Lowe, D., Utembe, S., Allan, J., Zaveri, R. A., Fast, J. D., Hodnebrog, Ø., Denier van der Gon, H., and McFiggans,
- 850 G.: Gaseous chemistry and aerosol mechanism developments for version 3.5.1 of the online regional model, WRF-Chem, Geoscientific Model Development, 7, 2557–2579, https://doi.org/10.5194/gmd-7-2557-2014, 2014.

Ball, J. T., Woodrow, I. E., and Berry, J. A.: A model predicting stomatal conductance and its contribution to the control of photosynthesis under different environmental conditions, in: Progress in photosynthesis research, pp. 221–224, Springer, 1987.

Barrie, L. and Barrie, M.: Chemical components of lower tropospheric aerosols in the high Arctic: Six years of observations, Journal of

855 Atmospheric Chemistry, 11, 211–226, 1990.

Barrie, L., Staebler, R., Toom, D., Georgi, B., Den Hartog, G., Landsberger, S., and Wu, D.: Arctic aerosol size-segregated chemical observations in relation to ozone depletion during Polar Sunrise Experiment 1992., Journal of Geophysical Research: Atmospheres, 99, 25439–25451, 1994.

Barrie, L. A.: Arctic air pollution: An overview of current knowledge, Atmospheric Environment (1967), 20, 643-663, 1986.

- 860 Barthel, S., Tegen, I., and Wolke, R.: Do new sea spray aerosol source functions improve the results of a regional aerosol model?, Atmospheric Environment, 198, 265–278, 2019.
 - Berg, L. K., Gustafson, W. I., Kassianov, E. I., and Deng, L.: Evaluation of a modified scheme for shallow convection: Implementation of CuP and case studies, Monthly weather review, 141, 134–147, 2013.

Berg, L. K., Shrivastava, M., Easter, R. C., Fast, J. D., Chapman, E. G., Liu, Y., and Ferrare, R.: A new WRF-Chem treatment for studying

865 regional-scale impacts of cloud processes on aerosol and trace gases in parameterized cumuli, Geoscientific Model Development, 8, 409–429, 2015.

Bonan, G. B.: Land surface model (LSM version 1.0) for ecological, hydrological, and atmospheric studies: Technical description and users guide. Technical note, Tech. rep., National Center for Atmospheric Research, Boulder, CO (United States ..., 1996.
Bowen, H. J. M. et al.: Environmental chemistry of the elements., Academic Press., 1979.

870 Browse, J., Carslaw, K., Mann, G., Birch, C., Arnold, S., and Leck, C.: The complex response of Arctic aerosol to sea-ice retreat, Atmos. Chem. Phys, 14, 7543–7557, 2014.

Burrows, S. M., Hoose, C., Pöschl, U., and Lawrence, M. G.: Ice nuclei in marine air: biogenic particles or dust?, Atmospheric Chemistry and Physics, 13, 245–267, 2013.

Calhoun, J. A., Bates, T. S., and Charlson, R. J.: Sulfur isotope measurements of submicrometer sulfate aerosol particles over the Pacific

- Ocean, Geophysical Research Letters, 18, 1877–1880, 1991.
 - Callaghan, A., de Leeuw, G., Cohen, L., and O'Dowd, C. D.: Relationship of oceanic whitecap coverage to wind speed and wind history, Geophysical Research Letters, 35, 2008.
 - Callaghan, A. H., Deane, G. B., Stokes, M. D., and Ward, B.: Observed variation in the decay time of oceanic whitecap foam, Journal of Geophysical Research: Oceans, 117, 2012.
- 880 Callaghan, A. H., Stokes, M., and Deane, G.: The effect of water temperature on air entrainment, bubble plumes, and surface foam in a laboratory breaking-wave analog, Journal of Geophysical Research: Oceans, 119, 7463–7482, 2014.
 - Campbell, J. R., Battaglia Jr, M., Dingilian, K., Cesler-Maloney, M., St Clair, J. M., Hanisco, T. F., Robinson, E., DeCarlo, P., Simpson, W., Nenes, A., et al.: Source and Chemistry of Hydroxymethanesulfonate (HMS) in Fairbanks, Alaska, Environmental Science & Technology, 2022.
- 885 Carter, W. P.: Documentation of the SAPRC-99 chemical mechanism for VOC reactivity assessment, Contract, 92, 95–308, 2000. Chapman, E. G., Gustafson Jr, W., Easter, R. C., Barnard, J. C., Ghan, S. J., Pekour, M. S., and Fast, J. D.: Coupling aerosol-cloud-radiative processes in the WRF-Chem model: Investigating the radiative impact of elevated point sources, Atmospheric Chemistry and Physics, 9, 945–964, 2009.

- Chen, Y., Cheng, Y., Ma, N., Wolke, R., Nordmann, S., Schüttauf, S., Ran, L., Wehner, B., Birmili, W., Gon, H. A., et al.: Sea salt emission, transport and influence on size-segregated nitrate simulation: a case study in northwestern Europe by WRF-Chem, Atmospheric Chemistry and Physics, 16, 12 081–12 097, 2016.
- Chen, Y., Cheng, Y., Ma, N., Wei, C., Ran, L., Wolke, R., Größ, J., Wang, Q., Pozzer, A., Denier van der Gon, H. A., et al.: Natural sea-salt
 emissions moderate the climate forcing of anthropogenic nitrate, Atmospheric Chemistry and Physics, 20, 771–786, 2020.
 - Chi, J., Li, W., Zhang, D., Zhang, J., Lin, Y., Shen, X., Sun, J., Chen, J., Zhang, X., Zhang, Y., et al.: Sea salt aerosols as a reactive surface for inorganic and organic acidic gases in the Arctic troposphere, Atmospheric Chemistry and Physics, 15, 11 341–11 353, 2015.
 - Clarke, A. D., Owens, S. R., and Zhou, J.: An ultrafine sea-salt flux from breaking waves: Implications for cloud condensation nuclei in the remote marine atmosphere, Journal of Geophysical Research: Atmospheres, 111, 2006.
- 900 Collatz, G. J., Ball, J. T., Grivet, C., and Berry, J. A.: Physiological and environmental regulation of stomatal conductance, photosynthesis and transpiration: a model that includes a laminar boundary layer, Agricultural and Forest meteorology, 54, 107–136, 1991.
 - Collatz, G. J., Ribas-Carbo, M., and Berry, J.: Coupled photosynthesis-stomatal conductance model for leaves of C4 plants, Functional Plant Biology, 19, 519–538, 1992.
- Cravigan, L. T., Ristovski, Z., Modini, R. L., Keywood, M. D., and Gras, J. L.: Observation of sea-salt fraction in sub-100 nm diameter
 particles at Cape Grim, Journal of Geophysical Research: Atmospheres, 120, 1848–1864, 2015.
 - De Leeuw, G., Andreas, E. L., Anguelova, M. D., Fairall, C., Lewis, E. R., O'Dowd, C., Schulz, M., and Schwartz, S. E.: Production flux of sea spray aerosol, Reviews of Geophysics, 49, 2011.
 - Douglas, T. A., Domine, F., Barret, M., Anastasio, C., Beine, H. J., Bottenheim, J., Grannas, A., Houdier, S., Netcheva, S., Rowland, G., et al.: Frost flowers growing in the Arctic ocean-atmosphere-sea ice-snow interface: 1. Chemical composition, Journal of Geophysical
- 910 Research: Atmospheres, 117, 2012.

890

Chen, F., Janjić, Z., and Mitchell, K.: Impact of atmospheric surface-layer parameterizations in the new land-surface scheme of the NCEP mesoscale Eta model, Boundary-Layer Meteorology, 85, 391–421, 1997.

- Druckenmiller, M. L., Eicken, H., Johnson, M. A., Pringle, D. J., and Williams, C. C.: Toward an integrated coastal sea-ice observatory: System components and a case study at Barrow, Alaska, Cold Regions Science and Technology, 56, 61–72, 2009.
- Durre, I., Yin, X., Vose, R. S., Applequist, S., and Arnfield, J.: Enhancing the data coverage in the Integrated Global Radiosonde Archive, Journal of Atmospheric and Oceanic Technology, 35, 1753–1770, 2018.
- Easter, R. C., Ghan, S. J., Zhang, Y., Saylor, R. D., Chapman, E. G., Laulainen, N. S., Abdul-Razzak, H., Leung, L. R., Bian, X., and Zaveri,
 R. A.: MIRAGE: Model description and evaluation of aerosols and trace gases, Journal of Geophysical Research: Atmospheres, 109, 2004.
 - Eckhardt, S., Quennehen, B., Oliviè, D. J. L., Berntsen, T. K., Cherian, R., Christensen, J., Collins, W., Crepinsek, S., Daskalakis, N., Flanner, M., et al.: Current model capabilities for simulating black carbon and sulfate concentrations in the Arctic atmosphere: a multi-model evaluation using a comprehensive measurement data set. Atmospheric Chemistry and Physics, 15, 9413–9433, 2015.
 - Eicken, H., Jones, J., Meyer, F., Mahoney, A., Druckenmiller, M. L., Rohith, M., and Kambhamettu, C.: Environmental security in Arctic ice-covered seas: From strategy to tactics of hazard identification and emergency response, Marine Technology Society Journal, 45, 37–48, 2011.

920

945

- Eidhammer, T., DeMott, P., Prenni, A., Petters, M., Twohy, C., Rogers, D., Stith, J., Heymsfield, A., Wang, Z., Pratt, K., et al.: Ice initiation
- 925 by aerosol particles: Measured and predicted ice nuclei concentrations versus measured ice crystal concentrations in an orographic wave cloud, Journal of the Atmospheric Sciences, 67, 2417–2436, 2010.
 - Emmons, L. K., Walters, S., Hess, P. G., Lamarque, J.-F., Pfister, G. G., Fillmore, D., Granier, C., Guenther, A., Kinnison, D., Laepple, T., et al.: Description and evaluation of the Model for Ozone and Related chemical Tracers, version 4 (MOZART-4), Geoscientific Model Development, 3, 43–67, 2010.
- 930 Fan, S.-M. and Jacob, D. J.: Surface ozone depletion in Arctic spring sustained by bromine reactions on aerosols, Nature, 359, 522–524, 1992.
 - Fast, J. D., Gustafson Jr, W. I., Easter, R. C., Zaveri, R. A., Barnard, J. C., Chapman, E. G., Grell, G. A., and Peckham, S. E.: Evolution of ozone, particulates, and aerosol direct radiative forcing in the vicinity of Houston using a fully coupled meteorology-chemistry-aerosol model, Journal of Geophysical Research: Atmospheres, 111, 2006.
- 935 Fisher, J. A., Jacob, D. J., Wang, Q., Bahreini, R., Carouge, C. C., Cubison, M. J., Dibb, J. E., Diehl, T., Jimenez, J. L., Leibensperger, E. M., et al.: Sources, distribution, and acidity of sulfate–ammonium aerosol in the Arctic in winter–spring, Atmospheric Environment, 45, 7301–7318, 2011.
 - Freud, E., Krejci, R., Tunved, P., Leaitch, R., Nguyen, Q. T., Massling, A., Skov, H., and Barrie, L.: Pan-Arctic aerosol number size distributions: seasonality and transport patterns, Atmospheric Chemistry and Physics, 17, 8101–8128, 2017.
- 940 Frey, M. M., Norris, S. J., Brooks, I. M., Anderson, P. S., Nishimura, K., Yang, X., Jones, A. E., Nerentorp Mastromonaco, M. G., Jones, D. H., and Wolff, E. W.: First direct observation of sea salt aerosol production from blowing snow above sea ice, Atmospheric Chemistry and Physics, 20, 2549–2578, 2020.
 - Frossard, A. A., Shaw, P. M., Russell, L. M., Kroll, J. H., Canagaratna, M. R., Worsnop, D. R., Quinn, P. K., and Bates, T. S.: Springtime Arctic haze contributions of submicron organic particles from European and Asian combustion sources, Journal of Geophysical Research: Atmospheres, 116, 2011.
 - Frossard, A. A., Russell, L. M., Burrows, S. M., Elliott, S. M., Bates, T. S., and Quinn, P. K.: Sources and composition of submicron organic mass in marine aerosol particles, Journal of Geophysical Research: Atmospheres, 119, 12,977–13,003, https://doi.org/https://doi.org/10.1002/2014JD021913, 2014.

Fuentes, E., Coe, H., Green, D., Leeuw, G. d., and McFiggans, G.: On the impacts of phytoplankton-derived organic matter on the properties

- 950 of the primary marine aerosol–Part 1: Source fluxes, Atmospheric Chemistry and Physics, 10, 9295–9317, 2010.
 - Fuentes, E., Coe, H., Green, D., and McFiggans, G.: On the impacts of phytoplankton-derived organic matter on the properties of the primary marine aerosol–Part 2: Composition, hygroscopicity and cloud condensation activity, Atmospheric Chemistry and Physics, 11, 2585–2602, 2011.
- Fujiki, T., Matsumoto, K., Honda, M. C., Kawakami, H., and Watanabe, S.: Phytoplankton composition in the subarctic North Pacific during
 autumn 2005, Journal of plankton research, 31, 179–191, 2009.
 - Gantt, B., Kelly, J., and Bash, J.: Updating sea spray aerosol emissions in the Community Multiscale Air Quality (CMAQ) model version 5.0. 2, Geoscientific Model Development, 8, 3733–3746, 2015.
 - Gong, S.: A parameterization of sea-salt aerosol source function for sub-and super-micron particles, Global biogeochemical cycles, 17, 2003.
- Gong, S., Barrie, L., and Blanchet, J.-P.: Modeling sea-salt aerosols in the atmosphere: 1. Model development, Journal of Geophysical
 Research: Atmospheres, 102, 3805–3818, 1997.
 - Grell, G. A., Dudhia, J., Stauffer, D. R., et al.: A description of the fifth-generation Penn State/NCAR Mesoscale Model (MM5), 1994.
 - Grell, G. A., Peckham, S. E., Schmitz, R., McKeen, S. A., Frost, G., Skamarock, W. C., and Eder, B.: Fully coupled "online" chemistry within the WRF model, Atmospheric Environment, 39, 6957–6975, 2005.
- Grythe, H., Ström, J., Krejci, R., Quinn, P., and Stohl, A.: A review of sea-spray aerosol source functions using a large global set of sea salt
 aerosol concentration measurements, Atmospheric Chemistry and Physics, 14, 1277–1297, 2014.
 - Guenther, A., Jiang, X., Heald, C. L., Sakulyanontvittaya, T., Duhl, T., Emmons, L., and Wang, X.: The Model of Emissions of Gases and Aerosols from Nature version 2.1 (MEGAN2. 1): an extended and updated framework for modeling biogenic emissions, Geoscientific Model Development, 5, 1471–1492, 2012.
- Hancke, K., Lund-Hansen, L. C., Lamare, M. L., Højlund Pedersen, S., King, M. D., Andersen, P., and Sorrell, B. K.: Extreme low light
 requirement for algae growth underneath sea ice: A case study from Station Nord, NE Greenland, Journal of Geophysical Research:
 Oceans, 123, 985–1000, 2018.
 - Hara, K., Osada, K., Matsunaga, K., Iwasaka, Y., Shibata, T., and Furuya, K.: Atmospheric inorganic chlorine and bromine species in Arctic boundary layer of the winter/spring, Journal of Geophysical Research: Atmospheres, 107, AAC–4, 2002.
- Hartery, S., Toohey, D., Revell, L., Sellegri, K., Kuma, P., Harvey, M., and McDonald, A. J.: Constraining the surface flux of sea spray
 particles from the Southern Ocean, Journal of Geophysical Research: Atmospheres, 125, e2019JD032 026, 2020.
 - Heidam, N., Christensen, J., Skov, H., and Wåhlin, P.: Monitoring and modelling of the atmospheric environment in Greenland. A review, Sci Total Environ, 331, 5–28, 2004.
 - Hirdman, D., Sodemann, H., Eckhardt, S., Burkhart, J. F., Jefferson, A., Mefford, T., Quinn, P. K., Sharma, S., Ström, J., and Stohl, A.: Source identification of short-lived air pollutants in the Arctic using statistical analysis of measurement data and particle dispersion model
- 980 output, Atmospheric Chemistry and Physics, 10, 669–693, 2010.
 - Hodnebrog, Ø., Marelle, L., Alterskjær, K., Wood, R. R., Ludwig, R., Fischer, E. M., Richardson, T., Forster, P., Sillmann, J., and Myhre, G.: Intensification of summer precipitation with shorter time-scales in Europe, Environmental Research Letters, 14, 124 050, 2019.
 - Hong, S.-Y., Noh, Y., and Dudhia, J.: A new vertical diffusion package with an explicit treatment of entrainment processes, Monthly weather review, 134, 2318–2341, 2006.
- 985 Horowitz, H. M., Holmes, C., Wright, A., Sherwen, T., Wang, X., Evans, M., Huang, J., Jaeglé, L., Chen, Q., Zhai, S., and Alexander, B.: Effects of Sea Salt Aerosol Emissions for Marine Cloud Brightening on Atmospheric Chemistry: Implications for Radiative

Forcing, Geophysical Research Letters, 47, e2019GL085838, https://doi.org/https://doi.org/10.1029/2019GL085838, e2019GL085838 2019GL085838, 2020.

Huang, J. and Jaeglé, L.: Wintertime enhancements of sea salt aerosol in polar regions consistent with a sea ice source from blowing snow,

Atmospheric Chemistry and Physics, 17, 3699–3712, https://doi.org/10.5194/acp-17-3699-2017, 2017.

990

1005

- Iacono, M. J., Delamere, J. S., Mlawer, E. J., Shephard, M. W., Clough, S. A., and Collins, W. D.: Radiative forcing by long-lived greenhouse gases: Calculations with the AER radiative transfer models, Journal of Geophysical Research: Atmospheres, 113, 2008.
 - Jacobi, H., Voisin, D., Jaffrezo, J., Cozic, J., and Douglas, T.: Chemical composition of the snowpack during the OASIS spring campaign 2009 at Barrow, Alaska, Journal of Geophysical Research: Atmospheres, 117, 2012.
- 995 Jaeglé, L., Quinn, P. K., Bates, T. S., Alexander, B., and Lin, J.-T.: Global distribution of sea salt aerosols: new constraints from in situ and remote sensing observations, Atmospheric Chemistry and Physics, 11, 3137–3157, https://doi.org/10.5194/acp-11-3137-2011, 2011.
 - Jiménez, P. A., Dudhia, J., González-Rouco, J. F., Navarro, J., Montávez, J. P., and García-Bustamante, E.: A revised scheme for the WRF surface layer formulation, Monthly weather review, 140, 898–918, 2012.

Jordan, R.: A One-dimensional temperature model for a snow cover : technical documentation for SNTHERM.89, 1991.

- 1000 Kelly, J. T., Bhave, P. V., Nolte, C. G., Shankar, U., and Foley, K. M.: Simulating emission and chemical evolution of coarse sea-salt particles in the Community Multiscale Air Quality (CMAQ) model, Geoscientific Model Development, 3, 257–273, https://doi.org/10.5194/gmd-3-257-2010, 2010.
 - Kirpes, R. M., Bondy, A. L., Bonanno, D., Moffet, R. C., Wang, B., Laskin, A., Ault, A. P., and Pratt, K. A.: Secondary sulfate is internally mixed with sea spray aerosol and organic aerosol in the winter Arctic, Atmospheric Chemistry and Physics, 18, 3937–3949, https://doi.org/10.5194/acp-18-3937-2018, 2018.
- Kirpes, R. M., Bonanno, D., May, N. W., Fraund, M., Barget, A. J., Moffet, R. C., Ault, A. P., and Pratt, K. A.: Winter
 - time Arctic Sea Spray Aerosol Composition Controlled by Sea Ice Lead Microbiology, ACS Central Science, 5, 1760–1767, https://doi.org/10.1021/acscentsci.9b00541, 2019.

Kolesar, K. R., Cellini, J., Peterson, P. K., Jefferson, A., Tuch, T., Birmili, W., Wiedensohler, A., and Pratt, K. A.: Effect of Prudhoe Bay

- 1010 emissions on atmospheric aerosol growth events observed in Utqiaġvik (Barrow), Alaska, Atmospheric Environment, 152, 146–155, https://doi.org/10.1016/j.atmosenv.2016.12.019, 2017.
 - Krembs, C., Eicken, H., Junge, K., and Deming, J.: High concentrations of exopolymeric substances in Arctic winter sea ice: implications for the polar ocean carbon cycle and cryoprotection of diatoms, Deep Sea Research Part I: Oceanographic Research Papers, 49, 2163–2181, https://doi.org/https://doi.org/10.1016/S0967-0637(02)00122-X, 2002.
- 1015 Lange, R., Dall'Osto, M., Skov, H., Nøjgaard, J., Nielsen, I., Beddows, D., Simó, R., Harrison, R. M., and Massling, A.: Characterization of distinct Arctic aerosol accumulation modes and their sources, Atmospheric Environment, 183, 1–10, 2018.
 - Law, K. S., Stohl, A., Quinn, P. K., Brock, C. A., Burkhart, J. F., Paris, J.-D., Ancellet, G., Singh, H. B., Roiger, A., Schlager, H., et al.: Arctic air pollution: New insights from POLARCAT-IPY, Bulletin of the American Meteorological Society, 95, 1873–1895, 2014.
- Law, K. S., Roiger, A., Thomas, J. L., Marelle, L., Raut, J.-C., Dalsøren, S., Fuglestvedt, J., Tuccella, P., Weinzierl, B., and Schlager, H.:
 Local Arctic air pollution: Sources and impacts, Ambio, 46, 453–463, 2017.
- Leaitch, W. R., Russell, L. M., Liu, J., Kolonjari, F., Toom, D., Huang, L., Sharma, S., Chivulescu, A., Veber, D., and Zhang, W.: Organic functional groups in the submicron aerosol at 82.5° N, 62.5° W from 2012 to 2014, Atmospheric Chemistry and Physics, 18, 3269–3287, https://doi.org/10.5194/acp-18-3269-2018, 2018.

Leck, C., Norman, M., Bigg, E. K., and Hillamo, R.: Chemical composition and sources of the high Arctic aerosol relevant for cloud forma-

- 1025 tion, Journal of Geophysical Research: Atmospheres, 107, AAC 1–1–AAC 1–17, https://doi.org/https://doi.org/10.1029/2001JD001463, 2002.
 - Lee, J., McFiggans, G., Allan, J., Baker, A., Ball, S., Benton, A., Carpenter, L., Commane, R., Finley, B., Evans, M., et al.: Reactive halogens in the marine boundary layer (RHaMBLe): the tropical North Atlantic experiments, Atmospheric Chemistry and Physics, 10, 1031–1055, 2010.
- 1030 Li, J., Han, Z., and Yao, X.: A modeling study of the influence of sea salt on inorganic aerosol concentration, size distribution, and deposition in the western Pacific Ocean, Atmospheric environment, 188, 157–173, 2018.
 - Liu, J., Gunsch, M. J., Moffett, C. E., Xu, L., El Asmar, R., Zhang, Q., Watson, T. B., Allen, H. M., Crounse, J. D., St. Clair, J., et al.: Hydroxymethanesulfonate (HMS) formation during summertime fog in an Arctic oil field, Environmental Science & Technology Letters, 8, 511–518, 2021a.
- 1035 Liu, S., Liu, C.-C., Froyd, K. D., Schill, G. P., Murphy, D. M., Bui, T. P., Dean-Day, J. M., Weinzierl, B., Dollner, M., Diskin, G. S., et al.: Sea spray aerosol concentration modulated by sea surface temperature, Proceedings of the National Academy of Sciences, 118, 2021b.
 - Lovejoy, C., Vincent, W. F., Bonilla, S., Roy, S., Martineau, M.-J., Terrado, R., Potvin, M., Massana, R., and Pedrós-Alió, C.: DISTRIBU-TION, PHYLOGENY, AND GROWTH OF COLD-ADAPTED PICOPRASINOPHYTES IN ARCTIC SEAS 1, Journal of Phycology, 43, 78–89, 2007.
- 1040 Ma, X., von Salzen, K., and Li, J.: Modelling sea salt aerosol and its direct and indirect effects on climate, Atmospheric Chemistry and Physics, 8, 1311–1327, https://doi.org/10.5194/acp-8-1311-2008, 2008.
 - Malm, W. C., Sisler, J. F., Huffman, D., Eldred, R. A., and Cahill, T. A.: Spatial and seasonal trends in particle concentration and optical extinction in the United States, Journal of Geophysical Research: Atmospheres, 99, 1347–1370, https://doi.org/https://doi.org/10.1029/93JD02916, 1994.
- 1045 Marelle, L., Raut, J.-C., Law, K. S., Berg, L. K., Fast, J. D., Easter, R. C., Shrivastava, M., and Thomas, J. L.: Improvements to the WRF-Chem 3.5.1 model for quasi-hemispheric simulations of aerosols and ozone in the Arctic, Geoscientific Model Development, 10, 3661–3677, https://doi.org/10.5194/gmd-10-3661-2017, 2017.
 - Marelle, L., Thomas, J. L., Ahmed, S., Tuite, K., Stutz, J., Dommergue, A., Simpson, W. R., Frey, M. M., and Baladima, F.: Implementation and Impacts of Surface and Blowing Snow Sources of Arctic Bromine Activation Within WRF-Chem 4.1.1, Journal of Advances in Mod-
- 1050 eling Earth Systems, 13, e2020MS002 391, https://doi.org/https://doi.org/10.1029/2020MS002391, e2020MS002391 2020MS002391, 2021.
 - May, N. W., Quinn, P. K., McNamara, S. M., and Pratt, K. A.: Multiyear study of the dependence of sea salt aerosol on wind speed and sea ice conditions in the coastal Arctic, Journal of Geophysical Research: Atmospheres, 121, 9208–9219, https://doi.org/10.1002/2016JD025273, 2016.
- 1055 Millero, F. J., Feistel, R., Wright, D. G., and McDougall, T. J.: The composition of Standard Seawater and the definition of the Reference-Composition Salinity Scale, Deep Sea Research Part I: Oceanographic Research Papers, 55, 50–72, https://doi.org/10.1016/j.dsr.2007.10.001, 2008.
- Monaghan, A. J., Clark, M. P., Barlage, M. P., Newman, A. J., Xue, L., Arnold, J. R., and Rasmussen, R. M.: High-Resolution Historical Climate Simulations over Alaska, Journal of Applied Meteorology and Climatology, 57, 709–731, https://www.jstor.org/stable/26501015, 2018.

Monahan, E. and Muircheartaigh, I.: Optimal Power-Law Description of Oceanic Whitecap Coverage Dependence on Wind Speed, Journal of Physical Oceanography, 10, 2094–2099, 1980.

- Moschos, V., Dzepina, K., Bhattu, D., Lamkaddam, H., Casotto, R., Daellenbach, K. R., Canonaco, F., Rai, P., Aas, W., Becagli, S., et al.: Equal abundance of summertime natural and wintertime anthropogenic Arctic organic aerosols, Nature geoscience, 15, 196–202, 2022a.
 - Moschos, V., Schmale, J., Aas, W., Becagli, S., Calzolai, G., Eleftheriadis, K., Moffett, C. E., Schnelle-Kreis, J., Severi, M., Sharma, S., et al.: Elucidating the present-day chemical composition, seasonality and source regions of climate-relevant aerosols across the Arctic
- 1070 land surface, Environmental Research Letters, 17, 034 032, 2022b.

1085

- NCEP: National Centers for Environmental Prediction, National Weather Service, NOAA, U.S. Department of Commerce, NCEP FNL Operational Model Global Tropospheric Analyses, continuing from July 1999, https://doi.org/10.5065/D6M043C6, 2000.
- Neumann, D., Matthias, V., Bieser, J., Aulinger, A., and Quante, M.: Sensitivity of modeled atmospheric nitrogen species and nitrogen deposition to variations in sea salt emissions in the North Sea
- and Baltic Sea regions, Atmospheric Chemistry and Physics, 16, 2921–2942, https://doi.org/10.5194/acp-16-2921-2016, 2016.
 - Nielsen, I. E., Skov, H., Massling, A., Eriksson, A. C., Dall'Osto, M., Junninen, H., Sarnela, N., Lange, R., Collier, S., Zhang, Q., et al.: Biogenic and anthropogenic sources of aerosols at the High Arctic site Villum Research Station, Atmospheric Chemistry and Physics, 19, 10239–10256, 2019.
- Niu, G.-Y. and Yang, Z.-L.: Effects of vegetation canopy processes on snow surface energy and mass balances, Journal of Geophysical
 Research: Atmospheres, 109, https://doi.org/https://doi.org/10.1029/2004JD004884, 2004.
 - Niu, G.-Y. and Yang, Z.-L.: Effects of frozen soil on snowmelt runoff and soil water storage at a continental scale, Journal of Hydrometeorology, 7, 937–952, 2006.
 - Niu, G.-Y., Yang, Z.-L., Dickinson, R. E., Gulden, L. E., and Su, H.: Development of a simple groundwater model for use in climate models and evaluation with Gravity Recovery and Climate Experiment data, Journal of Geophysical Research: Atmospheres, 112, https://doi.org/https://doi.org/10.1029/2006JD007522, 2007.
 - Niu, G.-Y., Yang, Z.-L., Mitchell, K. E., Chen, F., Ek, M. B., Barlage, M., Kumar, A., Manning, K., Niyogi, D., Rosero, E., Tewari, M., and Xia, Y.: The community Noah land surface model with multiparameterization options (Noah-MP):
 1. Model description and evaluation with local-scale measurements, Journal of Geophysical Research: Atmospheres, 116, https://doi.org/https://doi.org/10.1029/2010JD015139, 2011.
- 1090 O'Dowd, C. D., Smith, M. H., Consterdine, I. E., and Lowe, J. A.: Marine aerosol, sea-salt, and the marine sulphur cycle: a short review, Atmospheric Environment, 31, 73–80, https://doi.org/https://doi.org/10.1016/S1352-2310(96)00106-9, 1997.
 - O'Dowd, C. D., Facchini, M. C., Cavalli, F., Ceburnis, D., Mircea, M., Decesari, S., Fuzzi, S., Yoon, Y. J., and Putaud, J.-P.: Biogenically driven organic contribution to marine aerosol, Nature, 431, 676–680, 2004.
 - Ovadnevaite, J., Ceburnis, D., Martucci, G., Bialek, J., Monahan, C., Rinaldi, M., Facchini, M. C., Berresheim, H., Worsnop, D. R., and
- O'Dowd, C.: Primary marine organic aerosol: A dichotomy of low hygroscopicity and high CCN activity, Geophysical Research Letters,
 38, https://doi.org/https://doi.org/10.1029/2011GL048869, 2011.
 - Ovadnevaite, J., Ceburnis, D., Canagaratna, M., Berresheim, H., Bialek, J., Martucci, G., Worsnop, D. R., and O'Dowd, C.: On the effect of wind speed on submicron sea salt mass concentrations and source fluxes, Journal of Geophysical Research: Atmospheres, 117, 2012.

Monahan, E., Spiel, D. E., and Davidson, K.: A Model of Marine Aerosol Generation Via Whitecaps and Wave Disruption, 1986.

Morrison, H., Thompson, G., and Tatarskii, V.: Impact of cloud microphysics on the development of trailing stratiform precipitation in a simulated squall line: Comparison of one-and two-moment schemes, Monthly weather review, 137, 991–1007, 2009.

Ovadnevaite, J., Manders, A., de Leeuw, G., Ceburnis, D., Monahan, C., Partanen, A.-I., Korhonen, H., and O'Dowd, C. D.: A sea spray

- 1100 aerosol flux parameterization encapsulating wave state, Atmospheric Chemistry and Physics, 14, 1837–1852, https://doi.org/10.5194/acp-14-1837-2014, 2014.
 - Partanen, A.-I., Dunne, E., Bergman, T., Laakso, A., Kokkola, H., Ovadnevaite, J., Sogacheva, L., Baisnée, D., Sciare, J., Manders, A., et al.: Global modelling of direct and indirect effects of sea spray aerosol using a source function encapsulating wave state, Atmospheric Chemistry and Physics, 14, 11731–11752, 2014.
- 1105 Peterson, P. K., Pöhler, D., Sihler, H., Zielcke, J., General, S., Frieß, U., Platt, U., Simpson, W. R., Nghiem, S. V., Shepson, P. B., et al.: Observations of bromine monoxide transport in the Arctic sustained on aerosol particles, Atmospheric Chemistry and Physics, 17, 7567– 7579, 2017.
 - Quinn, P., Coffman, D., Johnson, J., Upchurch, L., and Bates, T.: Small fraction of marine cloud condensation nuclei made up of sea spray aerosol, Nature Geoscience, 10, 674–679, 2017.
- 1110 Quinn, P. K., Coffman, D. J., Kapustin, V. N., Bates, T. S., and Covert, D. S.: Aerosol optical properties in the marine boundary layer during the First Aerosol Characterization Experiment (ACE 1) and the underlying chemical and physical aerosol properties, Journal of Geophysical Research: Atmospheres, 103, 16 547–16 563, https://doi.org/https://doi.org/10.1029/97JD02345, 1998.
 - Quinn, P. K., Miller, T. L., Bates, T. S., Ogren, J. A., Andrews, E., and Shaw, G. E.: A 3-year record of simultaneously measured aerosol chemical and optical properties at Barrow, Alaska, Journal of Geophysical Research: Atmospheres, 107, AAC 8–1–AAC 8–15,

1115 https://doi.org/https://doi.org/10.1029/2001JD001248, 2002.

- Quinn, P. K., SHAW, G., ANDREWS, E., DUTTON, E. G., RUOHO-AIROLA, T., and GONG, S. L.: Arctic haze: current trends and knowledge gaps, Tellus B, 59, 99–114, https://doi.org/https://doi.org/10.1111/j.1600-0889.2006.00238.x, 2007.
 - Rahn, K. A. and McCaffrey, R. J.: On the origin and transport of the winter Arctic aerosol, Annals of the New York Academy of Sciences, 338, 486–503, 1980.
- 1120 Raut, J.-C., Marelle, L., Fast, J. D., Thomas, J. L., Weinzierl, B., Law, K. S., Berg, L. K., Roiger, A., Easter, R. C., Heimerl, K., et al.: Crosspolar transport and scavenging of Siberian aerosols containing black carbon during the 2012 ACCESS summer campaign, Atmospheric Chemistry and Physics, 17, 10969–10995, 2017.
 - Revell, L. E., Kremser, S., Hartery, S., Harvey, M., Mulcahy, J. P., Williams, J., Morgenstern, O., McDonald, A. J., Varma, V., Bird, L., and Schuddeboom, A.: The sensitivity of Southern Ocean aerosols and cloud microphysics to sea spray and sulfate aerosol production
- 1125 in the HadGEM3-GA7.1 chemistry-climate model, Atmospheric Chemistry and Physics, 19, 15447–15466, https://doi.org/10.5194/acp-19-15447-2019, 2019.
 - Rhodes, R. H., Yang, X., Wolff, E. W., McConnell, J. R., and Frey, M. M.: Sea ice as a source of sea salt aerosol to Greenland ice cores: a model-based study, Atmospheric Chemistry and Physics, 17, 9417–9433, https://doi.org/10.5194/acp-17-9417-2017, 2017.
- Roscoe, H. K., Brooks, B., Jackson, A., Smith, M., Walker, S., Obbard, R. W., and Wolff, E. W.: Frost flowers in the laboratory: Growth,
 characteristics, aerosol, and the underlying sea ice, Journal of Geophysical Research: Atmospheres, 116, 2011.
- Russell, L. M., Hawkins, L. N., Frossard, A. A., Quinn, P. K., and Bates, T. S.: Carbohydrate-like composition of submicron atmospheric particles and their production from ocean bubble bursting, Proceedings of the National Academy of Sciences, 107, 6652–6657, https://doi.org/10.1073/pnas.0908905107, 2010.
- Sakaguchi, K. and Zeng, X.: Effects of soil wetness, plant litter, and under-canopy atmospheric stability on ground evaporation in the Commu-
- 1135 nity Land Model (CLM3.5), Journal of Geophysical Research: Atmospheres, 114, https://doi.org/https://doi.org/10.1029/2008JD010834, 2009.

- Saliba, G., Chen, C.-L., Lewis, S., Russell, L. M., Rivellini, L.-H., Lee, A. K. Y., Quinn, P. K., Bates, T. S., Haëntjens, N., Boss, E. S., Karp-Boss, L., Baetge, N., Carlson, C. A., and Behrenfeld, M. J.: Factors driving the seasonal and hourly variability of sea-spray aerosol number in the North Atlantic, Proceedings of the National Academy of Sciences, 116, 20309–20314, https://doi.org/10.1073/pnas.1907574116,
- 1140

2019.

- Salisbury, D. J., Anguelova, M. D., and Brooks, I. M.: On the variability of whitecap fraction using satellite-based observations, Journal of Geophysical Research: Oceans, 118, 6201–6222, 2013.
- Salisbury, D. J., Anguelova, M. D., and Brooks, I. M.: Global distribution and seasonal dependence of satellite-based whitecap fraction, Geophysical Research Letters, 41, 1616–1623, https://doi.org/https://doi.org/10.1002/2014GL059246, 2014.
- 1145 Salter, M. E., Zieger, P., Acosta Navarro, J. C., Grythe, H., Kirkevåg, A., Rosati, B., Riipinen, I., and Nilsson, E. D.: An empirically derived inorganic sea spray source function incorporating sea surface temperature, Atmospheric Chemistry and Physics, 15, 11047– 11066, https://doi.org/10.5194/acp-15-11047-2015, 2015.
 - Schmale, J., Arnold, S., Law, K. S., Thorp, T., Anenberg, S., Simpson, W., Mao, J., and Pratt, K.: Local Arctic air pollution: A neglected but serious problem, Earth's Future, 6, 1385–1412, 2018.
- 1150 Schmale, J., Sharma, S., Decesari, S., Pernov, J., Massling, A., Hansson, H.-C., Von Salzen, K., Skov, H., Andrews, E., Quinn, P. K., et al.: Pan-Arctic seasonal cycles and long-term trends of aerosol properties from 10 observatories, Atmospheric Chemistry and Physics, 22, 3067–3096, 2022.

Seinfeld, J. and Pandis, S.: Atmospheric Chemistry and Physics: From Air Pollution to Climate Change, 1998.

- Sellers, P., Randall, D., Collatz, G., Berry, J., Field, C., Dazlich, D., Zhang, C., Collelo, G., and Bounoua, L.: A revised Land Surface
 parameterization (SiB2) for atmospheric GCMs. Part I: Model Formulation, Journal of Climate, 9, 676–705, https://doi.org/10.1175/1520-0442(1996)009<0676:ARLSPF>2.0.CO:2, 1996.
 - Sharma, S., Barrie, L. A., Magnusson, E., Brattström, G., Leaitch, W., Steffen, A., and Landsberger, S.: A factor and trends analysis of multidecadal lower tropospheric observations of Arctic aerosol composition, black carbon, ozone, and mercury at Alert, Canada, Journal of Geophysical Research: Atmospheres, 124, 14 133–14 161, 2019.
- 1160 Shaw, P. M., Russell, L. M., Jefferson, A., and Quinn, P. K.: Arctic organic aerosol measurements show particles from mixed combustion in spring haze and from frost flowers in winter, Geophysical Research Letters, 37, https://doi.org/https://doi.org/10.1029/2010GL042831, 2010.
 - Shaw, W. J., Jerry Allwine, K., Fritz, B. G., Rutz, F. C., Rishel, J. P., and Chapman, E. G.: An evaluation of the wind erosion module in DUSTRAN, Atmospheric Environment, 42, 1907–1921, https://doi.org/https://doi.org/10.1016/j.atmosenv.2007.11.022, 2008.
- 1165 Sheridan, P., Delene, D., and Ogren, J.: Four years of continuous surface aerosol measurements from the Department of Energy's Atmospheric Radiation measurement Program Southern Great Plains Cloud and Radiation Testbed site, Journal of Geophysical Research: Atmospheres, 106, 20735–20747, 2001.
 - Shrivastava, M., Fast, J., Easter, R., Gustafson Jr., W. I., Zaveri, R. A., Jimenez, J. L., Saide, P., and Hodzic, A.: Modeling organic aerosols in a megacity: comparison of simple and complex representations of the volatility basis set approach, Atmospheric Chemistry and Physics,
- 1170 11, 6639–6662, https://doi.org/10.5194/acp-11-6639-2011, 2011.
- Simpson, W., Carlson, D., Hönninger, G., Douglas, T., Sturm, M., Perovich, D., and Platt, U.: First-year sea-ice contact predicts bromine monoxide (BrO) levels at Barrow, Alaska better than potential frost flower contact, Atmospheric Chemistry and Physics, 7, 621–627, 2007.

Sofiev, M., Soares, J., Prank, M., de Leeuw, G., and Kukkonen, J.: A regional-to-global model of emission and transport of sea salt particles

- 1175 in the atmosphere, Journal of Geophysical Research: Atmospheres, 116, https://doi.org/https://doi.org/10.1029/2010JD014713, 2011. Spada, M., Jorba, O., Pérez García-Pando, C., Janjic, Z., and Baldasano, J. M.: Modeling and evaluation of the global sea-salt aerosol distribution: sensitivity to size-resolved and sea-surface temperature dependent emission schemes, Atmospheric Chemistry and Physics, 13, 11735–11755, https://doi.org/10.5194/acp-13-11735-2013, 2013.
- Stroeve, J., Serreze, M., Holland, M., Kay, J., Malanik, J., and Barrett, A.: The Arctic's rapidly shrinking sea ice cover: a research synthesis,
 Climatic Change, 110, 1005–1027, https://doi.org/10.1007/s10584-011-0101-1, 2012.
 - Su, B., Wang, T., Zhang, G., Liang, Y., Lv, C., Hu, Y., Li, L., Zhou, Z., Wang, X., and Bi, X.: A review of atmospheric aging of sea spray aerosols: Potential factors affecting chloride depletion, Atmospheric Environment, p. 119365, 2022.
 - Wang, X., Jacob, D. J., Eastham, S. D., Sulprizio, M. P., Zhu, L., Chen, Q., Alexander, B., Sherwen, T., Evans, M. J., Lee, B. H., et al.: The role of chlorine in global tropospheric chemistry, Atmospheric Chemistry and Physics, 19, 3981–4003, 2019.
- 1185 Whaley, C. H., Mahmood, R., von Salzen, K., Winter, B., Eckhardt, S., Arnold, S., Beagley, S., Becagli, S., Chien, R.-Y., Christensen, J., Damani, S. M., Eleftheriadis, K., Evangeliou, N., Faluvegi, G. S., Flanner, M., Fu, J. S., Gauss, M., Giardi, F., Gong, W., Hjorth, J. L., Huang, L., Im, U., Kanaya, Y., Krishnan, S., Klimont, Z., Kühn, T., Langner, J., Law, K. S., Marelle, L., Massling, A., Olivié, D., Onishi, T., Oshima, N., Peng, Y., Plummer, D. A., Popovicheva, O., Pozzoli, L., Raut, J.-C., Sand, M., Saunders, L. N., Schmale, J., Sharma, S., Skov, H., Taketani, F., Thomas, M. A., Traversi, R., Tsigaridis, K., Tsyro, S., Turnock, S., Vitale, V., Walker, K. A., Wang, M., Watson-
- 1190 Parris, D., and Weiss-Gibbons, T.: Model evaluation of short-lived climate forcers for the Arctic Monitoring and Assessment Programme: a multi-species, multi-model study, Atmospheric Chemistry and Physics Discussions, 2022, 1–88, https://doi.org/10.5194/acp-2021-975, 2022.

Wild, O., Zhu, X., and Prather, M.: Fast-J: Accurate Simulation of In- and Below-Cloud Photolysis in Tropospheric Chemical Models, Journal of Atmospheric Chemistry, 37, 245–282, 2000.

- 1195 Wu, S.-P., Dai, L.-H., Zhu, H., Zhang, N., Yan, J.-P., Schwab, J. J., and Yuan, C.-S.: The impact of sea-salt aerosols on particulate inorganic nitrogen deposition in the western Taiwan Strait region, China, Atmospheric Research, 228, 68–76, 2019.
 - Xu, L., Russell, L. M., Somerville, R. C. J., and Quinn, P. K.: Frost flower aerosol effects on Arctic wintertime longwave cloud radiative forcing, Journal of Geophysical Research: Atmospheres, 118, 13,282–13,291, https://doi.org/https://doi.org/10.1002/2013JD020554, 2013.
- 1200 Xu, L., Russell, L. M., and Burrows, S. M.: Potential sea salt aerosol sources from frost flowers in the pan-Arctic region, Journal of Geophysical Research: Atmospheres, 121, 10,840–10,856, https://doi.org/https://doi.org/10.1002/2015JD024713, 2016.
 - Xu, W., Ovadnevaite, J., Fossum, K. N., Lin, C., Huang, R.-J., Ceburnis, D., and O'Dowd, C.: Sea spray as an obscured source for marine cloud nuclei, Nature Geoscience, 15, 282–286, 2022.

Yang, R. and Friedl, M. A.: Modeling the effects of three-dimensional vegetation structure on surface radiation and energy balance in boreal

- forests, Journal of Geophysical Research: Atmospheres, 108, https://doi.org/https://doi.org/10.1029/2002JD003109, 2003.
 Yang Z.-L., R. E. Dickinson, A. R. and Vinnikov., K. Y.: Validation of the snow sub-model of the Biosphere-Atmosphere Transfer Scheme with Russian snow cover and meteorological observational data, J. Clim., 10, 353—373, https://doi.org/10.1175/1520-0442(1997)010<0353:VOTSSO>2.0.CO;2, 1997.
- Zaveri, R. A., Easter, R. C., Fast, J. D., and Peters, L. K.: Model for Simulating Aerosol Interactions and Chemistry (MOSAIC), Journal of
 Geophysical Research: Atmospheres, 113, https://doi.org/10.1029/2007JD008782, 2008.
 - Zhao, C. and Garrett, T. J.: Effects of Arctic haze on surface cloud radiative forcing, Geophysical Research Letters, 42, 557–564, 2015.

Zwaaftink, C. G., Grythe, H., Skov, H., and Stohl, A.: Substantial contribution of northern high-latitude sources to mineral dust in the Arctic, Journal of Geophysical Research: Atmospheres, 121, 13–678, 2016.