An Improved Representation of Aerosol Mixing State for Air-Quality-Weather Interactions

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Abstract. We implement a detailed representation of aerosol mixing-state into the GEM-MACH air quality and weather forecast model. Our mixing-state representation includes three categories: one for more-hygroscopic aerosol, one for less-hygroscopic aerosol with a high black carbon (BC) mass fraction, and one for less-hygroscopic aerosol with a low BC mass fraction. ~~This is the first model with a mixing-state representation of this type simulating a continent-scale domain.~~ The more-

5 detailed representation allows us to better resolve two different aspects of aerosol mixing state: differences in hygroscopicity due to aerosol composition, and the amount of absorption enhancement of BC due to non-absorbing coatings. Notably, this three-category representation allows us to account for BC thickly coated with primary organic matter, which enhances the absorption of the BC but has a low hygroscopicity.

We compare the results of the three-category representation (1L2B) with a simulation that uses two categories, split by hy-

10 groscopicity (HYGRO), and a simulation using the original size-resolved internally mixed assumption (SRIM). We perform a case study focused on North America during July 2016, when there were intense wildfires over northwestern North America. We find that the more-detailed representation of the aerosol hygroscopicity in both 1L2B and HYGRO decreases wet deposition, which increases aerosol concentrations, particularly of less-hygroscopic species. The concentration of $PM_{2.5}$ increases by 23 % on average. We show that these increased aerosol concentrations increase cloud droplet number concentrations and cloud

15 reflectivity in the model, decreasing surface temperatures.

Using two categories based on hygroscopicity yields only a modest benefit in resolving the coating thickness on black carbon, however. The 1L2B representation resolves BC with thinner coatings than the HYGRO simulation, resulting in absorption aerosol optical depths that are 3 % less on average, with greater differences over strong anthropogenic source regions. We did not find strong subsequent effects of this decreased absorption on meteorology.

20 ## 1 Introduction

Aerosol chemical mixing state refers to the distribution of chemical species across a population of aerosol particles. An aerosol population is said to be fully externally mixed if each aerosol particle consists of a single chemical species. If all chemical species are distributed evenly amongst all aerosol particles, then the aerosol population is said to be fully internally mixed.

Aerosol populations in the real atmosphere are never fully externally mixed nor fully internally mixed, but instead exist somewhere between these two extremes. In general, particles emitted from different sources are initially externally mixed with respect to each other, and become more internally mixed with time through condensation, coagulation, and chemical reactions.

Internal mixing of hydrophilic and hydrophobic species can allow the hydrophobic species to act as cloud condensation nuclei (CCN) (e.g. McFiggans et al., 2006; Anttila, 2010; Kim et al., 2018; Dalirian et al., 2018). An increase in CCN concentrations will generally render clouds more reflective, and can also increase cloud lifetime. Internal mixing of hydrophilic and hydrophobic species also allows the hydrophobic species to be more efficiently removed from the atmosphere through wet deposition. Additionally, weakly absorbing species can form a coating on black carbon (BC), which strongly absorbs solar radiation. The weakly absorbing species then act as a lens, enhancing the absorption of solar radiation by the BC, compared to the case where the BC is uncoated (e.g. Lesins et al., 2002; Liu and Mishchenko, 2018; Schnaiter et al., 2005; Cui et al., 2016). Estimates of the factor by which the absorption of BC increases due to coatings (the absorption enhancement) vary from 1 (no enhancement) to 4, with the majority of the studies reporting values between 1 and 2.5 (Adachi et al., 2010; Zhang et al., 2008; Khalizov et al., 2009; Cappa et al., 2012; Lack et al., 2012; Liu et al., 2015; Peng et al., 2016; Schnaiter et al., 2005; Wang et al., 2014a; Xu et al., 2018; Zanatta et al., 2018; Zhang et al., 2018b). Differences in experimental methods and regional and seasonal variations in BC coating thickness both likely contribute to this diversity. This absorption enhancement leads to a local heating of the atmosphere and a cooling of the surface, potentially increasing stability and affecting cloud cover and precipitation (Bond et al., 2013; Boucher et al., 2013). We refer the reader to two recent reviews (Stevens and Dastoor, 2019; Riemer et al., 2019) for more details about aerosol mixing state.

Many previous representations of aerosol mixing-state have been implemented in models to predict CCN concentrations and aerosol optical properties, and we include a partial list of these in Table 1. These include representing each particle individually (PartMC-MOSAIC, Riemer et al., 2009; Zaveri et al., 2010); multiple mixing-state categories separated by BC mass fraction, including MADRID-BC (Oshima et al., 2009b, a), ATRAS (Matsui et al., 2014; Matsui, 2017), MADE-soot (Riemer et al., 2003; Vogel et al., 2009), MADE-in (Aquila et al., 2011) and MADE-3 (Kaiser et al., 2019, 2014); two categories for at least BC and organic carbon based on hygroscopicity, implemented in GEOS-Chem (Bey et al., 2001; Wang et al., 2018, 2014b), the GLObal Model of Aerosol Processes (GLOMAP) in both its bin (Manktelow et al., 2010; Spracklen et al., 2005, 2011) and modal (Mann et al., 2010; Bellouin et al., 2013) configurations, GMXe (Pringle et al., 2010), the M3+ module (Wilson et al., 2001), M7 (Vignati et al., 2010; Stier et al., 2005; Vignati et al., 2004; Zhang et al., 2012), MAM4 (Liu et al., 2016), MAM7 (Liu et al., 2012), the Model for Ozone and Related chemical Tracers (MOZART, Emmons et al., 2010) and the Sectional Aerosol module for Large Scale Applications (SALSA, Bergman et al., 2012; Andersson et al., 2015; Kokkola et al., 2008; Tonttila et al., 2017; Kokkola et al., 2018); and representing all aerosol within the same size bin or mode as internally-mixed, including the Canadian Aerosol Module (CanAM, Gong et al., 2006; Moran et al., 2012; Gong et al., 2003, 2015), CHIMERE (Menut et al., 2013), the Community Multiscale Air Quality (CMAQ, Binkowski and Roselle, 2003; Appel et al., 2013; Elleman and Covert, 2009; USEPA, 2017) model, the Modal Aerosol Dynamics module for Europe (MADE, Lauer et al., 2005)

and the Modal Aerosol Module with three lognormal modes (MAM3, Liu et al., 2012). We refer the reader to Stevens and Dastoor (2019) for more detail on previous model representations of aerosol mixing state, including mixing-state representations that did not specifically target resolving CCN concentrations and optical properties, such as detailed categorizations based on chemical composition and source-oriented approaches. Previous studies using the model approaches listed above have found that if all aerosol in the same size bin or mode is assumed to be internally-mixed, CCN concentrations will frequently be overestimated by 10-20 % and absorption coefficients of BC will be overestimated by 20-40 % (Stevens and Dastoor, 2019, and containing references).

However, it still remains unclear how best to efficiently represent aerosol mixing state in atmospheric models. In this study, we implement a detailed representation of aerosol mixing-state into the Global Environmental Multiscale - Modelling Air quality and CHemistry (GEM-MACH) (Moran et al., 2010) air quality model with online air-quality-weather interactions. We refer to this new configuration of GEM-MACH as GM-MixingState. Our approach was inspired by the results of Ching et al. (2016): We independently account for both changes in hygroscopicity and BC mass fraction, as aerosol hygroscopic properties and optical properties do not necessarily co-vary. The existing air-quality-weather interactions in GEM-MACH include aerosol-radiation interactions and changes in cloud droplet activation based on CCN concentrations (Gong et al., 2015; Majdzadeh et al., 2022). We perform a case study focused on biomass-burning over North America to evaluate GM-MixingState. We investigate the interactions between the representation of aerosol mixing state and air-quality-weather interactions.

The paper is structured as follows: In Sect. 2, we describe the GEM-MACH model and the GM-MixingState configuration, as well as the experiments performed. In Sect. 3, we present our results and analysis. In Sect. 4, we summarize our study and present our conclusions.

2   Model description and methods

GEM-MACH is an online chemical transport model embedded within the Environment and Climate Change Canada (ECCC) Numerical Weather Prediction (NWP) model GEM (Côté et al., 1998b, a; Charron et al., 2012). GEM-MACH has been in use as the ECCC operational air quality prediction model since 2009 (Moran et al., 2010). The representations of many atmospheric processes in GEM-MACH are the same as in the ECCC AURAMS (A Unified Regional Air-quality Modelling System) offline chemical transport model (Gong et al., 2006), including gas-phase, aqueous-phase, and heterogeneous chemistry (inorganic gas-particle partitioning); secondary organic aerosol (SOA) formation; aerosol microphysics (nucleation, condensation, coagulation, and activation); sedimentation of particles; and dry deposition and wet removal (in-cloud and below-cloud scavenging) of gases and particles. The gas-phase and aqueous-phase chemistry mechanisms in GEM-MACH are adapted from ADOM (Acid Deposition and Oxidant Model, Venkatram et al., 1988; Fung et al., 1991). The heterogeneous chemistry mechanism currently implemented in GEM-MACH is a bulk scheme based on ISOROPPIA (Makar et al., 2003). Eight dry aerosol chemical species are included in GEM-MACH: sulphate, nitrate, ammonium, sea-salt, dust and crustal material, SOA,

Table 1. Partial list of previous representations of mixing-state in aerosol modules, from Stevens and Dastoor (2019). See text for expansion of acronyms, where applicable.

Representation	Example Aerosol Modules	References
Particle-Resolving	PartMC-MOSAIC	(Riemer et al., 2009; Zaveri et al., 2010)
Multiple categories based on BC mass fraction	MADRID-BC	(Oshima et al., 2009b, a)
	ATRAS	(Matsui et al., 2014; Matsui, 2017)
	MADE-soot	(Riemer et al., 2003; Vogel et al., 2009)
	MADE-in	(Aquila et al., 2011)
	MADE-3	(Kaiser et al., 2019, 2014)
Two categories based on hygroscopicity	GEOS-Chem	(Bey et al., 2001; Wang et al., 2018, 2014b)
	GLOMAP	(Manktelow et al., 2010; Spracklen et al., 2005, 2011; Mann et al., 2010; Bellouir
	GMXe	(Pringle et al., 2010)
	M3+	(Wilson et al., 2001)
	M7	(Vignati et al., 2010; Stier et al., 2005; Vignati et al., 2004; Zhang et al., 2012)
	MAM4	(Liu et al., 2016)
	MAM7	(Liu et al., 2012)
	MOZART	(Emmons et al., 2010)
	SALSA	(Bergman et al., 2012; Andersson et al., 2015; Kokkola et al., 2008; Tonttila et al.
Size-Resolved Internally-Mixed	CanAm	(Gong et al., 2006; Moran et al., 2012; Gong et al., 2003, 2015)
	CHIMERE	(Menut et al., 2013)
	CMAQ	(Binkowski and Roselle, 2003; Appel et al., 2013; Elleman and Covert, 2009; US
	MADE	(Lauer et al., 2005)

primary organic aerosol (POA), and BC. We note that we will refer to dust and crustal material collectively as "dust" in the rest of this paper. GEM-MACH uses a single-moment aerosol scheme; It does not include a prognostic aerosol number tracer. When needed, diagnostic number concentrations are calculated assuming that aerosol particles within each size bin are monodisperse with diameter equal to the midpoint diameter of the size bin.

By default, GEM-MACH uses a size-resolved internally mixed representation of the aerosol population: the aerosol population within each size bin is internally mixed, but the population of aerosol in each size bin is externally mixed with respect to each other size bin. The operational version of GEM-MACH uses two size bins (aerosol dry diameters 0-2.5 μm and 2.5-10 μm, Moran et al. (2010)), but for this study we use twelve size bins spanning 10 nm to 10 μm. The 12-bin configuration has been shown to yield results that more closely resemble observations (Akingunola et al., 2018).

For this study, we implemented a more detailed representation of the aerosol mixing state into GEM-MACH. Within each size bin, we separate the aerosol into up to three mixing-state categories based on hygroscopicity and BC mass fraction: 1. high hygroscopicity (hi-κ); 2. low hygroscopicity, high BC mass fraction (lo-κ_hi-BC); and 3. low hygroscopicity, low BC mass fraction (lo-κ_lo-BC). This configuration is similar to the MADE-soot, MADE-in and MADE-3 aerosol modules, which include three categories: generally hydrophilic BC-free particles, hydrophilic BC-containing particles and hydrophobic BC-containing particles. We differ in that we ~~our BC-free category is also hydrophobic, and we have~~ use a single category for all hydrophilic particles (hi-κ) and we use two categories for hydrophobic particles (lo-κ_hi-BC and lo-κ_lo-BC). This allows us to resolve BC coated with organic material (weakly hygroscopic, but thickly-coated) from BC ~~coated with hydrophilic material, which is both hydrophilic and thickly-coated.~~ that has thin coatings or no coatings of other aerosol matter (weakly hygroscopic and thinly-coated). Each of the eight dry species are tracked in each mixing-state category, resulting in a total of 288 (8 species x 3 mixing-state categories x 12 size bins) aerosol tracers. Following the recommendations in Ching et al. (2016), we use a threshold value of the hygroscopicity parameter (κ; Petters and Kreidenweis (2007)) of 0.1 between hi-κ and lo-κ mixing-state categories, and a threshold BC mass fraction of 0.3 between lo-BC and hi-BC mixing-state categories. We will discuss these mixing-state categories further in Sect. 2.2.

Coagulation of two particles within the same mixing-state category is assumed to result in a particle of the same mixing-state category, as both BC mass fraction and volume-weighted hygroscopicity would be within the range spanned by the two original particles. For coagulation of particles from two different mixing-state categories, we calculate the hygroscopicity and BC mass fraction of the new particle, and add the mass to the mixing-state category that matches the new particle's properties. ~~Additionally~~No other process directly transfers mass between mixing-state categories. However, after all other aerosol processes, we calculate the hygroscopicity and BC mass fraction for each size bin and mixing-state category. If either the hygroscopicity or the BC mass fraction is outside of the bounds of the current mixing-state category, ~~the mass~~ all of the mass in the current combination of size bin and mixing-state category is moved to the mixing-state category that matches the hygroscopicity and BC mass fraction of the aerosol mass. Through this method, as condensation and other processes

change the volume-weighted hygroscopicity and BC mass fraction over time, aerosol particles will generally move from the lo-κ_hi-BC mixing-state category to the lo-κ_lo-BC mixing-state category, and from both lo-κ mixing-state categories to the hi-κ mixing-state category.

130

To calculate the hygroscopicity of aerosol in the model, we assume that sulphate, nitrate, ammonium, sea-salt, dust, SOA, POA, and BC have κ values of 0.65, 0.65, 0.65, 1.1, 0.03, 0.1, 0.001, and 0, respectively (Ching et al., 2016; Zieger et al., 2017; Koehler et al., 2009). Following the volume Zdanovskii-Stokes-Robinson (ZSR) mixing rule (Petters and Kreidenweis, 2007), we assume that the hygroscopicity of a particle is the volume-weighted average of the component species. We therefore

135 do not account explicitly for coating of insoluble components by soluble components, nor do we consider how particle size or shape may affect the mass fraction of coating material necessary for a particle to be rendered "hydrophilic". However, other studies have shown that neither CCN concentrations (Liu et al., 2016; Lee et al., 2013) nor aerosol effective radiative forcing, either through aerosol-cloud interactions or through aerosol-radiation interactions (Regayre et al., 2018), are sensitive to the threshold amount of soluble material needed to render a particle hydrophilic. However, global burdens of BC and POA, es-

140 pecially in remote regions, have been shown to be sensitive to this parameter (Liu et al., 2012, 2016). This volume-weighted hygroscopicity is only used to determine the proper mixing-state category for aerosol. It is not used to determine cloud droplet activation.

Instead, cloud droplet activation is calculated using the parameterization for sectional models described by Abdul-Razzak

145 and Ghan (2002). Particle hygroscopicity is calculated separately for each mixing-state category based on molecular weights and ion dissociation, as per eq. 7 from Abdul-Razzak and Ghan (2002). Properties of SOA are assumed to be those of adipic acid; BC, POA and dust are assumed to be insoluble. We assume that aerosol in the lo-κ mixing-state categories does not participate in aqueous chemistry, and is not removed by cloud-to-rain conversion and subsequent wet deposition. ~~It~~ We discuss this in more detail in the supplement. Aerosol in the lo-κ mixing-state categories is still removed from the atmosphere by

150 below-cloud impaction by rain, as this process is not expected to depend strongly on aerosol composition.

Aerosol-radiation interactions are calculated as described in Majdzadeh et al. (2022). For the radiation calculations, sea-spray and dust are always assumed to exist as pure particles, externally mixed from the other components. For each size bin within each mixing-state category, ~~the remaining components~~if the BC mass fraction is greater than 40 %, BC is also

155 assumed to be externally mixed from other components. For each size bin within each mixing-state category, sulphate, nitrate, ammonium and organic matter are assumed to be internally mixed. If the BC mass fraction is less than 40 % we assume that these same species (sulphate, nitrate, ammonium, POA and SOA) form a spherical shell ~~around~~ over a spherical BC core ~~. The~~ in order to calculate an absorption enhancement factor. The total wet radius of the core-shell particle is calculated using the volume-weighted hygroscopic growth factor of the components in the core-shell particle. The absorption enhancement of the

160 BC cores is ~~calculated following Bond et al. (2006)~~then calculated following the parameterization of Bond et al. (2006) with the observationally constrained maximum threshold of 1.93 (Bond et al., 2006). However, we assume that there is no absorp-

tion ~~enhancment~~ enhancement for BC cores that comprise more than 40 % of the particle by mass, in agreement with more recent observations of thinly-coated BC particles (Liu et al., 2017; Peng et al., 2016). ~~Therefore, there is never any absorption enhancement for particles in the hi-BC~~ This process is applied independently to each size bin and each mixing-state category.

We choose our domain and time period to be consistent with the 2016 North Americain domain used in the fourth phase of the Air Quality Model Evaluation International Initiative (AQMEII4; Galmarini et al., 2021). To reduce computational expenses while still providing sufficient data for analysis, we perform simulations only from June 15th to July 31st, 2016, and we restrict our analysis to output from the month of July to provide sufficient time for spin-up. We use a 10 km horizontal resolution and 84 hybrid vertical levels up to 0.1 hPa, consistent with the ECCC contribution to AQMEII4 multi-model experiment. Chemical boundary conditions are sourced from a climatology from the global chemical transport model MOZART-4 (Model for Ozone and Related chemical Tracers, version 4; Emmons et al., 2010).

2.1 Emissions

The emissions inventories used in study are the same as those described in Majdzadeh et al. (2022), and very similar to the protocol for contributions to AQMEII4 (Galmarini et al., 2021). Anthropogenic emissions from Canada and the United States were sourced from the Canadian Air Pollutant Emissions Inventory and the US Environmental Protection Agency (EPA) 2011 Air Emissions Modelling Platform, respectively. We use forest fire emissions from the Canadian Forest Fire Emissions Production System (CFFEPS; Chen et al., 2019). The Sparse Matrix Operator Kernel Emissions (SMOKE) emissions processing system (https://www.cmascenter.org/smoke; Bieser et al., 2011; Hogrefe et al., 2003; Houyoux et al., 2000) is used to speciate emissions prior to input within GEM-MACH (Zhang et al., 2018a). Bulk aerosol mass emissions are associated with one of the 91 composite particulate matter speciation profiles compiled from the EPA's SPECIATE4.5 database (https://www.epa.gov/air-emissions-modeling/speciate-2; Reff et al., 2009). Each composite particulate matter speciation profile gives relative fractions of sulphate, nitrate, ammonium, BC, and POA. ~~Sea salt and SOA are assumed to make no contribution, and dust~~ Dust is defined to be the ~~residual after the other~~ sum of all remaining species in the profile after these components are accounted for. Sea salt and SOA are assumed to make no contribution. As an example, particulate emissions from wildfires are speciated as 78.5 % POA, 9.7 % dust, 9.5 % BC, 1.3 % sulphate, 0.9 % ammonium, and 0.1 % nitrate. Sea-spray emissions are parameterized according to Gong (2003). GEM-MACH does not currently include natural dust emissions; the only natural dust included in our simulations is that included in the boundary conditions.

We differ from ~~these previous studies~~ previous studies using GEM-MACH in that we allocate aerosol emissions across the different mixing-state categories, as follows: Major stationary point-source emissions, such as emissions from smelters or fossil-fuel power plants, are assumed to be size-resolved internally mixed with other particulate mass from the same point source. Area emissions, including sea-spray, dust, and disperse anthropogenic emissions, including traffic emissions, are assumed to be as close to fully externally mixed as possible within the limits of the mixing-state representation used. If all three mixing-state categories were used, sulphate, ammonium, nitrate, and sea-spray would be emitted in the hi-κ category; dust and

POA would be emitted into the lo-κ_lo-BC category; and BC would be emitted into the lo-κ_hi-BC category. There are no primary emissions of SOA. We note that these emissions are not truly fully externally mixed in reality, and that this assumption will provide the maximum sensitivity to the mixing-state configuration used. However, observations have shown that particles emitted from traffic sources are either primarily BC or primarily organic, rather than being fully internally mixed at emission (Willis et al., 2016). Wildfire emissions are treated separately from other area emissions in GEM-MACH, and we assume that wildfire emissions are emitted into the lo-κ_lo-BC category category, as BC-containing particles within wildfire emissions have been observed to frequently be thickly coated with low-hygroscopicity organic material (Perring et al., 2017; Kondo et al., 2011).

2.2 Sensitivity studies

An important question remains regarding the minimum level of complexity required to well represent aerosol-weather feedbacks in air quality models. We therefore perform several simulations with diverse representations of the aerosol mixing state. We consider a configuration with two mixing-state categories, split based on particle hygroscopicity (denoted as representation HYGRO; categories hi-κ and lo-κ). We also consider a mixing-state representation with three mixing-state categories: we use one mixing-state category for all high-hygroscopicity particles and two mixing-state categories for low-hygroscopicity particles, split based on BC mass fraction (high-κ, low-κ_hi-BC, and low-κ_lo-BC). We refer to this representation as 1L2B (one hydrophilic, two hydrophobic). We would not expect any improvement over HYGRO in the representation of the radiative properties of hydrophilic particles, but we would expect that 1L2B would better represent the radiative properties of low-hygroscopicity BC-containing particles. In particular, this representation should better distinguish BC thickly coated with POA from BC that is bare or only thinly coated with POA.

In addition to performing simulations with different representations of the aerosol mixing state, we also perform simulations with aerosol effects on meteorology (feedbacks) either permitted or disabled. When feedbacks are disabled, cloud droplet nucleation is independent of aerosol concentrations and aerosol interactions with radiation have no effect on atmospheric temperatures or any other meteorological variables. Cloud droplet nucleation is instead determined following Cohard and Pinty (2010) as a function of updraft velocity, temperature, and pressure assuming a pre-specified CCN concentration that does not vary with space or time. The meteorology in these simulations is independent of the aerosol and gas-phase concentrations. This allows us to directly attribute any differences in results solely to differences in aerosol processes caused by the differences in the representation of mixing state. We designate the simulations where aerosol effects on meteorology are permitted with the suffix "_feedbacks", as these simulations include feedbacks of changes in aerosol concentrations and properties on the meteorology.

Table 2. Temporally and spatially averaged results for each simulation.

Simulation	SRIM	HYGRO	1L2B	SRIM_feedbacks	HYGRO_feedbacks	1L2B_feedbacks
PM$_{2.5}$ [μg kg^{-1}]	4.64	5.71	5.72	4.67	5.70	5.70
PM$_{10}$ [μg kg^{-1}]	5.67	7.98	7.99	5.69	7.97	7.98
AQHI$_{2.5}$	1.36	1.40	1.40	1.34	1.38	1.38
AQHI$_{10}$	1.27	1.33	1.33	1.27	1.33	1.33
NH$_4$ [μg kg^{-1}]	0.1572	0.1624	0.1627	0.1598	0.1653	0.1657
NO$_3$ [μg kg^{-1}]	0.0251	0.0255	0.0257	0.0239	0.0244	0.0246
SO$_4$ [μg kg^{-1}]	0.537	0.551	0.552	0.554	0.570	0.572
SOA [μg kg^{-1}]	1.75	2.11	2.12	1.76	2.11	2.12
POA [μg kg^{-1}]	0.875	1.013	1.014	0.870	0.993	0.994
Sea salt [μg kg^{-1}]	6.89	6.93	6.93	6.89	6.99	6.99
Dust [μg kg^{-1}]	1.72	3.31	3.31	1.72	3.30	3.30
BC [μg kg^{-1}]	0.150	0.175	0.175	0.150	0.173	0.172
BC mass fraction*	2.12 %	4.29 %	13.63 %	2.09 %	4.21 %	13.39 %
AOD**	0.0720	0.0962	0.0959	0.0732	0.0966	0.0964
AAOD**	0.0074	0.0103	0.0100	0.0076	0.0104	0.0101
precipitation [mm day^{-1}]	0.001607	0.001607	0.001607	0.001621	0.001611	0.001612

*Average BC mass fraction within BC-containing particles, see Sect. 3.1.4

**Averaged over 1300-2100 UTC daily.

3   Results and analysis

3.1   Non-Feedbacks Simulations

We present a summary of the domain-averaged, temporally averaged results from all simulations in Table 2. We will start by discussing differences between simulations with aerosol effects on weather disabled, in order to simplify the analysis. We remind the reader that because the meteorology is identical in these simulations, any differences in results can be attributed solely to differences in aerosol processes caused by the differences in the representation of mixing state.

3.1.1   Aerosol Concentrations

We show the mean concentrations of particulate matter with a diameter smaller than 2.5 microns (PM$_{2.5}$) in Fig. 1, along with the absolute and relative differences in PM$_{2.5}$ concentrations between the HYGRO and SRIM simulations, and we show a similar figure for PM$_{10}$ concentrations as Fig. S1. We note that PM$_{2.5}$ and PM$_{10}$ concentrations are ~~nearly identical~~ very similar in the HYGRO and 1L2B simulations ~~.~~ (Fig. S3). We find that spatially and temporally averaged surface PM$_{2.5}$ concentrations and PM$_{10}$ concentrations increase by 23 % and 41 %, respectively, from the SRIM simulation to either the HYGRO or 1L2B
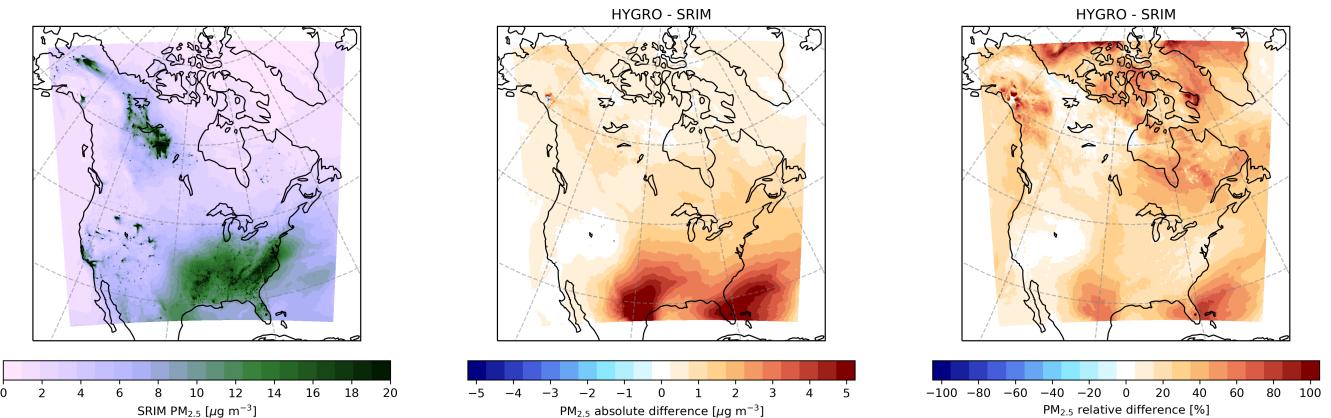
Figure 1. left: mean PM$_{2.5}$ concentrations from the SRIM simulation; ~~top right~~centre: mean difference in PM$_{2.5}$ concentrations between the HYGRO and SRIM simulations; ~~bottom~~ right: relative difference in mean PM$_{2.5}$ concentrations between the HYGRO and SRIM simulations.~~Note that PM$_{2.5}$ concentrations are nearly identical in the HYGRO and 1L2B simulations.~~

240   simulations. These differences are due mostly to increases in less-hygroscopic species, with concentrations of BC, POA, SOA, and dust being increased in the HYGRO and 1L2B simulations by 16 %, 16 %, 21 %, and 93 %. The concentrations of more-hygroscopic species (NH$_4$, NO$_3$, SO$_4$, and sea-spray aerosol) were increased by 3 % or less.

     These changes in aerosol concentrations are due primarily to changes in aerosol wet deposition. In the HYGRO and 1L2B

245   simulations, all aerosol in the low-κ categories are excluded from ~~wet deposition~~ in-cloud scavenging processes. However, direct comparison of wet deposition fluxes between simulations is complicated because of the greater aerosol mass concentrations in the HYGRO and 1L2B simulations than in the SRIM simulation. Even though the wet deposition process is less efficient for the same air parcel under the same conditions, local wet deposition fluxes can be greater in the HYGRO and 1L2B simulations due to the greater mass concentrations of aerosol in these simulations. For example, a reduced wet deposition flux close to

250   an emissions source can yield an increased wet deposition flux further downwind, as more aerosol mass will be transported further downwind. We attempt to isolate for these effects by dividing the daily wet deposition flux by the daily mean surface aerosol concentrations, to approximate the wet deposition efficiency. This approach is limited in that cloud uptake of gases also contributes to the wet deposition fluxes, and cloud uptake of aerosol and subsequent wet deposition are not necessarily co-located in space and time with surface aerosol concentrations. However, we expect that the relationships between surface

255   concentrations and wet deposition fluxes are similar enough across simulations for the comparison between simulations to be informative.

     We show the temporal means of the wet deposition fluxes normalized by the surface aerosol concentrations in Fig. 2, along with the absolute and relative differences between the HYGRO and SRIM simulations. Both aerosol concentrations and wet

260   deposition fluxes are ~~nearly identical~~ very similar in the HYGRO and 1L2B simulations~~.~~, therefore differences in normalized
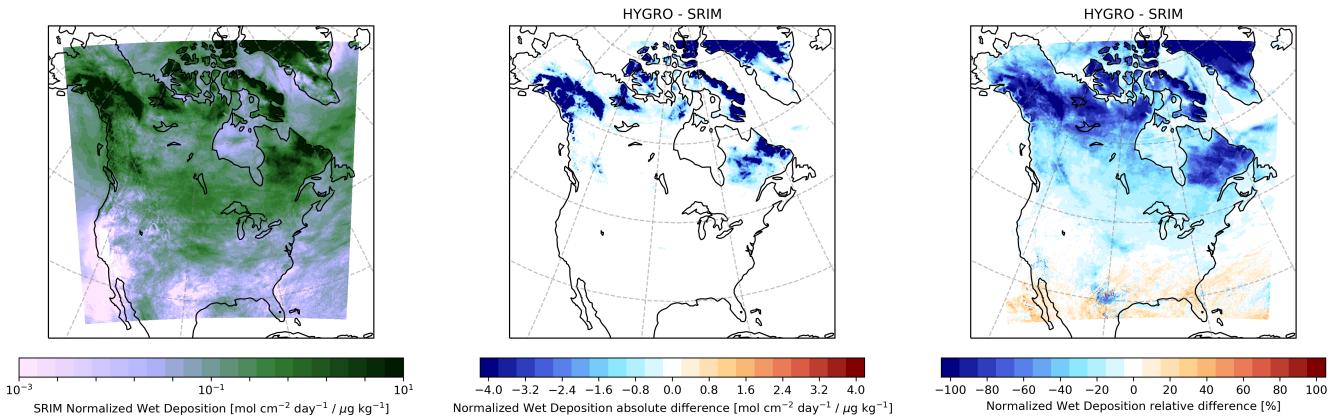
Figure 2. left: temporal means of wet deposition fluxes normalized by surface total aerosol concentrations from the SRIM simulation; ~~top right~~centre: mean difference in normalized wet deposition fluxes between the HYGRO and SRIM simulations; ~~bottom~~ right: relative difference in mean normalized wet deposition fluxes between the HYGRO and SRIM simulations.~~Note that wet deposition fluxes and total aerosol concentrations are nearly identical in the HYGRO and 1L2B simulations.~~

wet deposition between these simulations are also small (Fig. S4). We note that the normalized wet deposition shows some similar patterns to surface precipitation, as shown in Fig. 10, and the greatest values of normalized wet deposition are in the northern part of the domain where $PM_{2.5}$ concentrations are low (Fig. 1). In the HYGRO simulation, normalized wet deposition fluxes decrease over most of the domain, except for some regions in the south of the model domain. Deeper clouds would be expected in this part of the domain, which may decouple wet deposition fluxes from surface aerosol concentrations. In the far north of the domain, there are large decreases in normalized wet deposition fluxes, in some cases approaching 100 %. These overlap with regions of low aerosol concentrations and large relative increases in surface aerosol concentrations, as shown in Fig. 1. However, other regions with greater mean surface aerosol concentrations and smaller differences in surface aerosol concentrations between the SRIM and HYGRO simulations also show large relative differences in normalized wet deposition fluxes. Normalized wet deposition fluxes over most of Canada are reduced by 20-80 %. A large part of this region is influenced by the forest fires that took place in Alaska and northern Canada during this period. As these forest fire emissions are composed primarily of POA, they are particularly sensitive to the changes in the representation of the aerosol mixing state.

We can further control for differences in location and timing between wet deposition and surface concentrations by temporally and spatially averaging both the wet deposition fluxes and the surface concentrations before we divide the former by the latter. We therefore show the spatially and temporally averaged wet deposition fluxes normalized by the spatially and temporally averaged surface concentrations of each species in Table 3. After normalizing by the surface concentrations of aerosol, wet deposition rates of BC, POA, SOA, and dust were reduced in the HYGRO and 1L2B simulations by 27 %, 40 %, 12 %, and 10 %. The normalized wet deposition rates of more-hygroscopic species were reduced by less than 5 %.

Table 3. Temporally and spatially averaged wet deposition fluxes normalized by temporally and spatially averaged surface concentrations for each simulation. All units are (mol cm^{-2} day^{-1}) / (μg kg^{-1}).

Simulation	SRIM	HYGRO	1L2B	SRIM_feedbacks	HYGRO_feedbacks	1L2B_feedbacks
NH$_4$	0.765	0.744	0.742	0.750	0.726	0.724
NO$_3$	1.547	1.533	1.525	1.618	1.588	1.577
SO$_4$	0.0413	0.0394	0.0394	0.0381	0.0362	0.0361
SOA	0.0314	0.0276	0.0275	0.0312	0.0277	0.0277
POA	0.0214	0.0129	0.0129	0.0211	0.0134	0.0133
Sea salt	0.0727	0.0724	0.0724	0.0735	0.0723	0.0721
Dust	0.0526	0.0473	0.0472	0.0525	0.0472	0.0471
BC	0.267	0.195	0.196	0.265	0.199	0.201

In the HYGRO and 1L2B simulations, all aerosol in the low-κ categories are excluded from ~~wet deposition~~ in-cloud scavenging processes. Since the low-κ category is defined as having a κ less than 0.1, this excludes large aerosol with low hygroscopicities from participating in ~~wet deposition~~ in-cloud scavenging, even if their large size would allow them activate as droplets despite their low hygroscopicity. In particular, this may cause the wet deposition of dust particles to be underestimated in the HYGRO and 1L2B simulations, while it is likely overestimated in the SRIM simulation. A more detailed treatment of cloud uptake of aerosol is beyond the scope of this study, but will be revisited in a future version of GEM-MACH.

3.1.2 Comparison with Observations

We compare the results of our non-feedbacks simulations against data from the Interagency Monitoring of Protected Visual Environments (IMPROVE; http://vista.cira.colostate.edu/Improve/, last access 3 March 2022), the US EPA Chemical Speciation Network (CSN), and hourly measurements of PM$_{2.5}$ and PM$_{10}$ from the US EPA Air Quality System (AQS; https://www.epa.gov/aqs, last access: 3 March 2022). IMPROVE is a collaborative association of state, tribal, and federal agencies, and international partners. US Environmental Protection Agency is the primary funding source, with contracting and research support from the National Park Service. The Air Quality Group at the University of California, Davis is the central analytical laboratory, with ion analysis provided by Research Triangle Institute, and carbon analysis provided by Desert Research Institute. We convert observed organic carbon to organic matter assuming a mass-to-carbon ratio of 1.8, ~~we calculate~~ consistent with the assumption routinely used to calculate aerosol extinction based on the IMPROVE data (Pitchford et al., 2007). We calculate a concentration of mineral dust as 2.49 Si + 2.20 Al + 1.63 Ca + 2.42 Fe + 1.94 Ti, and we calculate a concentration of sea salt as 3.25 Na for comparison to the model results.

We evaluate the SRIM and 1L2B simulations against the IMPROVE, CSN, and AQS data by calculating the correlation coefficient (R), the normalized Mean Bias (NMB), the Root Mean Square Error (RMSE), and the fraction of simulated data within a factor of 2 of the observations (Fac2), shown in Table 4. As noted previously, the concentrations of aerosol species in

the HYGRO and 1L2B simulations are very similar. We note that R, NMB, and Fac2 differed by ≤0.03 and RMSE differed by <0.01 μg m^{-3} between the SRIM and 1L2B results for sulphate, nitrate and ammonium from the IMPROVE and CSN networks. The SRIM and 1L2B simulations therefore compare similarly well to observations for these species. This is expected, as these species are more weakly affected by the difference in mixing-state representation. There is an existing high bias in the SRIM-predicted concentrations of BC, organic aerosol, and dust. This high bias is worsened in the 1L2B simulation, due to the slower removal of these species by wet deposition in the 1L2B simulation, and this affects the calculated NMB, RMSE, and Fac2 values for these species. The correlation coefficients for EC and organic aerosol are not strongly affected. This suggests that the variability in BC and organic aerosol concentrations is not primarily controlled by wet deposition at these sites during the case study time period. As discussed, the wet deposition of dust is likely reduced too much in the 1L2B simulation, which may be responsible for the lower correlation between the observed and simulated dust concentrations in the 1L2B simulation. There is a slight shift of the sea salt size distribution to larger sizes in the 1L2B simulation, perhaps due to more coagulation with the larger concentrations EC, organic aerosol, and dust. This reduces the fine sea salt aerosol mass, even while total sea salt aerosol concentrations slightly increase. The NMB and RMSE for sea salt is therefore reduced in the 1L2B simulation compared to the SRIM simulation. The increased concentrations of $PM_{2.5}$ and PM_{10} in the 1L2B simulation increase the already high bias in $PM_{2.5}$ and reduces the underprediction of PM_{10}, as compared to the SRIM simulation. However, in both cases the RMSE is reduced, and the R and Fac2 values are either unchanged or slightly improved.

3.1.3 Air Quality Health Index

The Air Quality Health Index (AQHI; Stieb et al., 2008) is used by Environment and Climate Change Canada to communicate adverse health risks due to poor air quality to Canadians. It is formulated as a scale that ranges from 0 (excellent air quality) to 10 (very poor air quality), and is calculated based on the concentrations of $PM_{2.5}$ or PM_{10}, ozone (O_3), and nitrogen dioxide (NO_2). While the equations for calculating the AQHI permit values greater than 10 under exceptionally high concentrations of $PM_{2.5}$, PM_{10}, O_3, or NO_2, we restrict the values of AQHI to a maximum of 10, both because this is the intended range of the AQHI, and to reduce the influence of exceptional, highly concentrated plumes in uninhabited areas (such as those from forest fires) on our results.

The concentration of O_3 was, on average, 0.05 % less in the HYGRO or 1L2B cases than the SRIM case, and the concentration of NO_2 was, on average, 0.2 % greater in the HYGRO or 1L2B cases than the SRIM case. We can therefore attribute differences in the AQHI primarily to differences in $PM_{2.5}$ and PM_{10}. The $PM_{2.5}$ AQHI values in the HYGRO and 1L2B cases were nearly identical; the differences between them are shown in Fig. S5. The $PM_{2.5}$ AQHI was, on average, 0.04 units greater in the HYGRO and 1L2B simulations than in the SRIM simulation, and the PM_{10} AQHI was 0.06 units greater in the HYGRO and 1L2B simulations than in the SRIM simulation. This can be seen in the spatial patterns of the differences in AQHI, shown in Fig. 3 for $PM_{2.5}$ AQHI and in Fig. S2 for PM_{10} AQHI.

Table 4. Evaluation of SRIM and 1L2B simulations against observations. N = number of model/observation pairs, R = correlation coefficient, NMB = normalized Mean Bias, RMSE = Root Mean Square Error, Fac2 = fraction within a factor of two. Better performance between SRIM and 1L2B (larger values of R and Fac2 and smaller values of NMB and RMSE) are in bold font.

		N	R	NMB	RMSE	Fac2
IMPROVE Daily Fine Sulfate [μg m^{-3}]	SRIM	1543	0.38	-0.10	0.88	0.72
	1L2B	1543	**0.39**	**-0.09**	0.88	0.72
CSN Daily Fine Sulfate [μg m^{-3}]	SRIM	1050	0.28	-0.22	1.24	0.69
	1L2B	1050	0.28	**-0.21**	1.24	**0.70**
IMPROVE Daily Fine Nitrate [μg m^{-3}]	SRIM	1543	0.79	-0.72	0.34	0.10
	1L2B	1543	0.79	**-0.71**	0.34	0.10
CSN Daily Fine Nitrate [μg m^{-3}]	SRIM	1045	0.52	-0.20	0.96	0.20
	1L2B	1045	0.52	**-0.19**	0.96	**0.21**
CSN Daily Fine Ammonium [μg m^{-3}]	SRIM	1005	0.41	**1.08**	0.48	0.31
	1L2B	1005	0.41	1.11	0.48	0.31
IMPROVE Daily Fine EC [μg m^{-3}]	SRIM	1573	0.26	**0.64**	0.51	**0.57**
	1L2B	1573	0.26	0.74	0.51	0.54
CSN Daily Fine EC [μg m^{-3}]	SRIM	956	0.34	**0.27**	**0.62**	0.70
	1L2B	956	0.34	0.31	0.63	**0.71**
IMPROVE Daily Fine Organic Matter [μg m^{-3}]	SRIM	1573	0.26	**1.01**	**5.73**	**0.52**
	1L2B	1573	0.26	1.19	5.94	0.44
CSN Daily Fine Organic Matter [μg m^{-3}]	SRIM	956	**0.33**	**1.03**	**7.04**	**0.56**
	1L2B	956	0.32	1.19	7.42	0.50
IMPROVE Daily Fine Dust [μg m^{-3}]	SRIM	1539	**0.64**	**0.76**	**2.15**	**0.39**
	1L2B	1539	0.61	1.50	3.21	0.32
CSN Daily Fine Dust [μg m^{-3}]	SRIM	1060	**0.68**	**1.71**	**3.18**	**0.29**
	1L2B	1060	0.64	2.59	4.53	0.24
IMPROVE Daily Fine Sea Salt [μg m^{-3}]	SRIM	1534	0.63	4.18	2.16	0.10
	1L2B	1534	**0.64**	**4.04**	**2.10**	**0.11**
CSN Daily Fine Sea Salt [μg m^{-3}]	SRIM	1007	0.64	4.51	2.13	0.12
	1L2B	1007	0.64	**4.45**	**2.08**	0.12
AQS Hourly PM2.5 [μg m^{-3}]	SRIM	290614	0.06	**0.54**	67.45	0.53
	1L2B	290614	**0.07**	0.70	**66.86**	0.53
AQS Hourly PM10 [μg m^{-3}]	SRIM	238699	0.02	-0.40	97.98	0.44
	1L2B	238699	0.02	**-0.31**	**97.96**	**0.45**
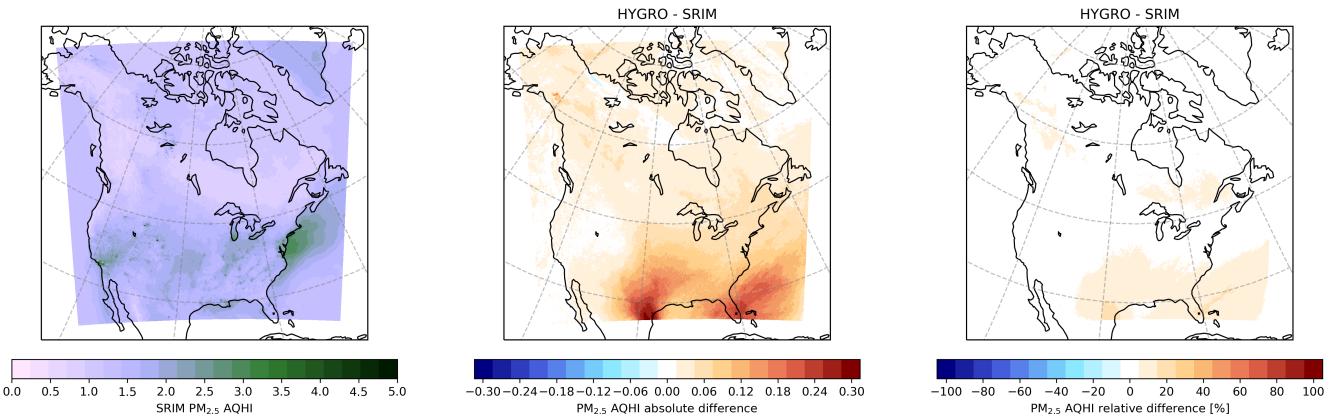
Figure 3. left: mean PM$_{2.5}$ AQHI values from the SRIM simulation; ~~top right~~centre: mean difference in PM$_{2.5}$ AQHI values between the HYGRO and SRIM simulations; ~~bottom~~ right: relative difference in mean PM$_{2.5}$ AQHI values between the HYGRO and SRIM simulations. Note that PM$_{2.5}$ AQHI values are nearly identical in the HYGRO and 1L2B simulations.

3.1.4 BC mass fraction in BC-containing particles

Before discussing the BC mass fractions, we will discuss the concentrations of BC in more detail. We show in Fig. 4 the mean BC ~~concentrations~~ mixing ratios at the surface from the SRIM simulation, as well as the absolute and relative differences in the BC mixing ratio between the HYGRO and SRIM simulations. The BC mixing ratios in the HYGRO and 1L2B simulations were very similar; the differences are shown in Fig. S6. We note that during the time period simulated, several large forest fires burned in Alaska and northern Canada (see BC emissions in Fig. S9), and the influence of these fires on PM$_{2.5}$, AQHI$_{2.5}$, and BC ~~concentrations~~ mixing ratios are clearly visible in Figures 1, 3, and 4, respectively.

As discussed in Sect. 3.1.1, the ~~concentrations~~ mixing ratios of BC typically increase in the HYGRO and 1L2B simulations, as less BC is removed through wet deposition. However, there are notable locations downwind of forest fires in Alaska and northern Canada where the ~~concentrations~~ mixing ratios of BC at the surface decrease. If we examine the differences in BC concentrations at a higher altitude typically above clouds, as shown at about 185 hPa above the surface in Fig. S7, we do not see these decreases, but we do see that BC from these wildfires has been lofted to this altitude. Aerosol in the high-κ category and all aerosol in the SRIM simulation that is sufficiently large can be ingested into cloud droplets. These cloud droplets can grow to drizzle sizes, and would then be subject to gravitational settling. If the drizzle droplets evaporate before reaching the surface, they will transport any aerosol mass in the droplets to lower altitudes. However, aerosol in the low-κ categories in the HYGRO and 1L2B simulations are not subject to this process, and therefore BC that is lofted to higher altitudes takes longer to reach the surface in these simulations, reducing the surface ~~concentrations~~ mixing ratios close to the forest fires and increasing them further downwind. For this reason, the peaks in absolute differences are further downwind at the surface, in Ontario and Quebec (Fig. 4), than at about 185 hPa above the surface, where the peaks in absolute differences are in Nunavut and above
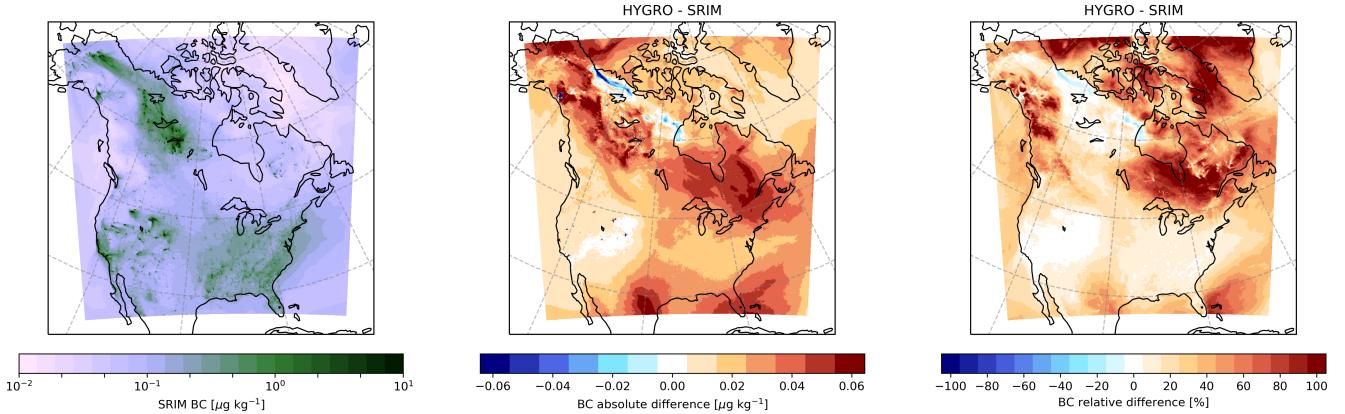
Figure 4. left: mean ~~surface~~ BC ~~concentrations~~ mixing ratios from the SRIM simulation; ~~top right~~centre: mean difference in BC ~~concentrations~~ mixing ratios between the HYGRO and SRIM simulations; ~~bottom~~ right: relative difference in mean BC ~~concentrations~~ mixing ratios between the HYGRO and SRIM simulations.~~Note that BC concentrations are nearly identical in the HYGRO and 1L2B simulations.~~

Hudson's Bay (Fig. S7).

In order to explain the effects of the mixing-state representation on aerosol-radiation interactions (discussed further in Sect. 3.1.5), we calculate the BC mass fraction within particles that contain BC. To do this, we calculate the BC mass fraction
360   in each combination of size bin and mixing-state category separately, then report the average value weighted by the mass of BC in the same combination of size bin and mixing-state category. We show these results in Fig. 5. The SRIM configuration assumes that BC is internally mixed with all other aerosol mass in particles of the same size, thus the BC mass fraction is similar to the total BC mass divided by the total aerosol mass. For 99 % of the grid cells, the mean BC mass fraction in the SRIM simulation is less than 5 %. The HYGRO configuration shows modest improvements over the SRIM configuration, while
365   the 1L2B configuration is able to resolve many regions close to emission sources where BC is thinly coated (high BC mass fractions). The high spatial resolution of GEM-MACH is an asset in resolving these regions close to emission sources.

We note that the 1L2B simulation is better able to capture regions with large BC mass fractions than the HYGRO simulation because the HYGRO configuration assumes that all low-hygroscopicity species within the same size bin are internally mixed, including BC, dust, and POA. Most dust mass exists in larger size bins than BC. Therefore, even the SRIM simulation does not
370   assume much internal mixing of BC and dust. However, BC and POA are emitted into the same size bins and from the same source regions. When the BC is assumed to be internally mixed with other low-hygroscopicity species, the resulting particles frequently consist of BC thickly coated with POA. The 1L2B simulation is able to distinguish BC thinly coated with POA from BC thickly coated with POA, and it predicts that a large proportion of BC near source regions has only a thin coating of
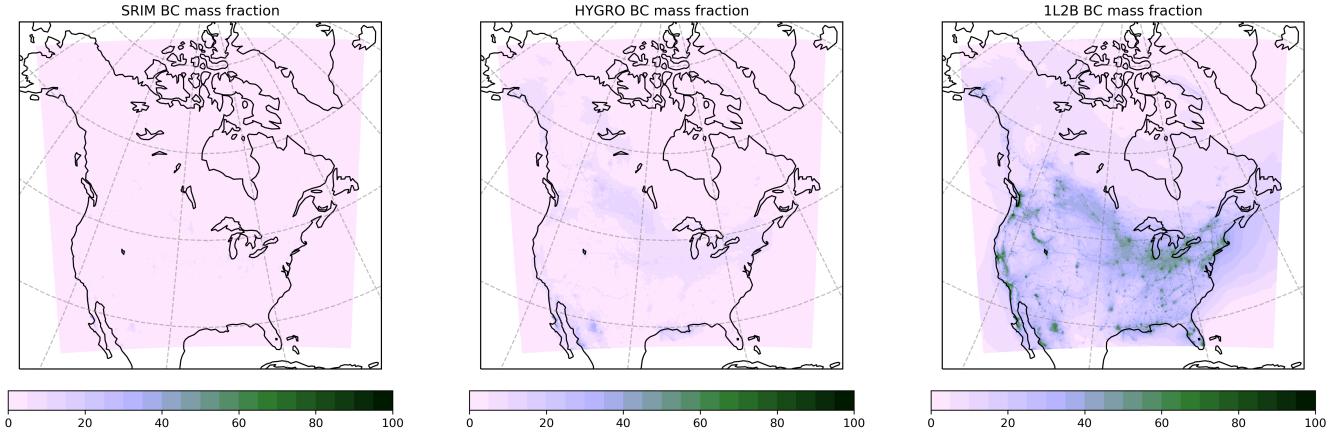
Figure 5. Black carbon mass fractions in BC-containing particles. Values are given at the surface, and weighted by the BC mass in each size bin and mixing-state category. Values are shown for the following simulations: ~~top~~ left: SRIM; ~~top right~~centre: HYGRO; ~~Bottom left~~right: 1L2B.

375 non-BC species.

3.1.5 Aerosol-radiation interactions

We show the monthly mean AOD from the SRIM simulation and the difference between the HYGRO and SRIM simulations in Fig. 6. We remind the reader that the calculations of aerosol optical properties are restricted to daylight hours in GEM-MACH.
380 As such, we include only data from between 1300 and 2100 UTC in Fig. 6, in order to exclude times of day when the AOD was not calculated for some part of the domain shown. We also note that the mean AOD in the 1L2B and HYGRO simulations differs by no more than ~~0.0011~~ 0.9 % for any grid cell in the domain, as shown in Fig. S8. When using the HYGRO configuration, the AOD is 34 % larger than in the SRIM case. A comparison of the AOD with the absorption aerosol optical depth (AAOD) (see Table 2 and Fig. 7) reveals that the AOD is dominated by aerosol scattering, rather than aerosol absorption. Previous
385 studies have found that the optical properties of non-absorbing aerosol ~~is~~ are not strongly sensitive to the mixing-state of the aerosol (e.g. Zaveri et al., 2010; Klingmüller et al., 2014), and that because AOD is dominated by the scattering component, ambient AOD is not strongly sensitive to mixing-state (e.g. Matsui et al., 2013, 2014; Klingmüller et al., 2014; Han et al., 2013), although a recent study has shown that aerosol scattering can be very sensitive to aerosol mixing-state under certain conditions (Yao et al., 2022). We therefore do not expect our more-detailed representation of the BC mass to yield strong
390 changes in aerosol scattering, but we do expect a decrease in aerosol absorption. We therefore conclude that the differences are due predominantly to the increases in aerosol mass, in turn due to the decrease in aerosol wet deposition. This is supported by the fact that the aerosol AOD and differences in AOD are visibly well-correlated with $PM_{2.5}$ and the differences in $PM_{2.5}$
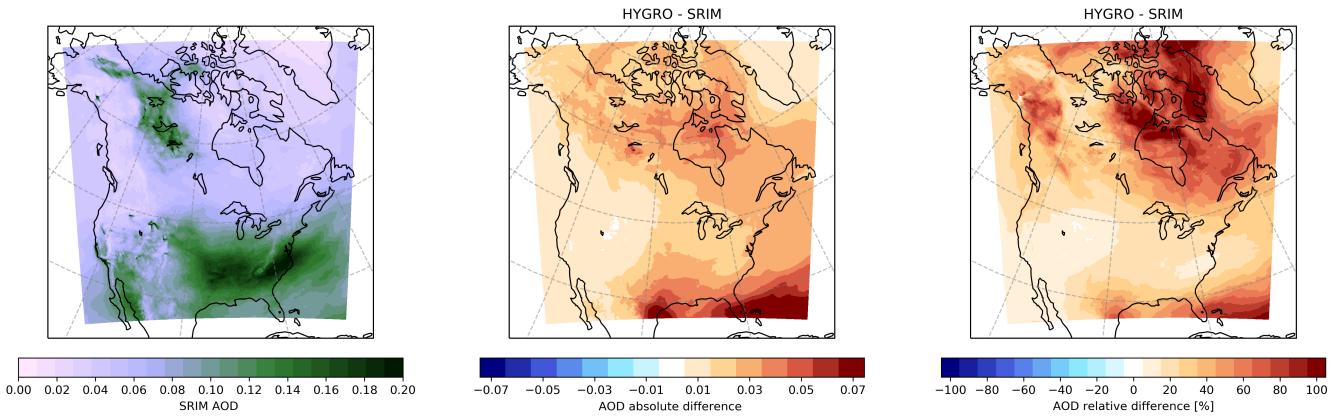
Figure 6. left: mean AOD from the SRIM simulation; ~~top right~~centre: mean difference in AOD between the HYGRO and SRIM simulations; ~~bottom~~ right: relative difference in mean AOD between the HYGRO and SRIM simulations. Only results from the hours of 1300-2100 UTC are included as the AOD is only calculated during local daylight hours.

shown in Fig. 1.

We show the monthly mean AAOD from the SRIM simulation and the differences between the 1L2B, HYGRO and SRIM simulations in Fig. 7. The AAOD is 39 % higher in the HYGRO case than in the SRIM case. As shown in Fig. 5, the BC mass fraction in BC-containing particles is only slightly larger in the HYGRO case than the SRIM cases. If the mass concentrations of all aerosol species were equal in both cases, higher BC mass fractions would imply thinner coatings and smaller absorption enhancements for the BC-containing particles. This effect would be expected to reduce the AAOD in the HYGRO case as compared to the SRIM case. The simulated increase in AAOD is due primarily to the increased concentrations of BC in the HYGRO case compared to the SRIM case.

When using the 1L2B configuration, the AAOD is on average 3 % less than in the HYGRO case, with these decreases being primarily over the eastern United States and around the Gulf of California. These are similar to the regions where the 1L2B case has higher BC mass fractions than the HYGRO case, as shown in Fig. 5, and are typically downwind of large anthropogenic sources of BC. We note that there are smaller differences in the plumes of the large northern forest fires; this is because emissions of BC from forest fires are assumed to be thickly coated as observations (Perring et al., 2017; Kondo et al., 2011) have shown that this is typically the case. Around the Gulf of ~~Mexico~~ California and just south the Great Lakes region, the decreases in AAOD between 1L2B and HYGRO (due to better-resolving the coating thickness on BC) are larger in magnitude than the increases in AAOD between HYGRO and SRIM (due to reduced wet deposition of low-hygroscopicity aerosol, including BC). As seen in Fig. 2, wet deposition was relatively low during our simulations in both of these regions, especially around the Gulf of California. This is due largely to less cloudiness and to less precipitation, which is shown in Fig. 10. Also, both of these regions have greater emissions of anthropogenic BC, as can be seen in Fig. S9. Therefore in these
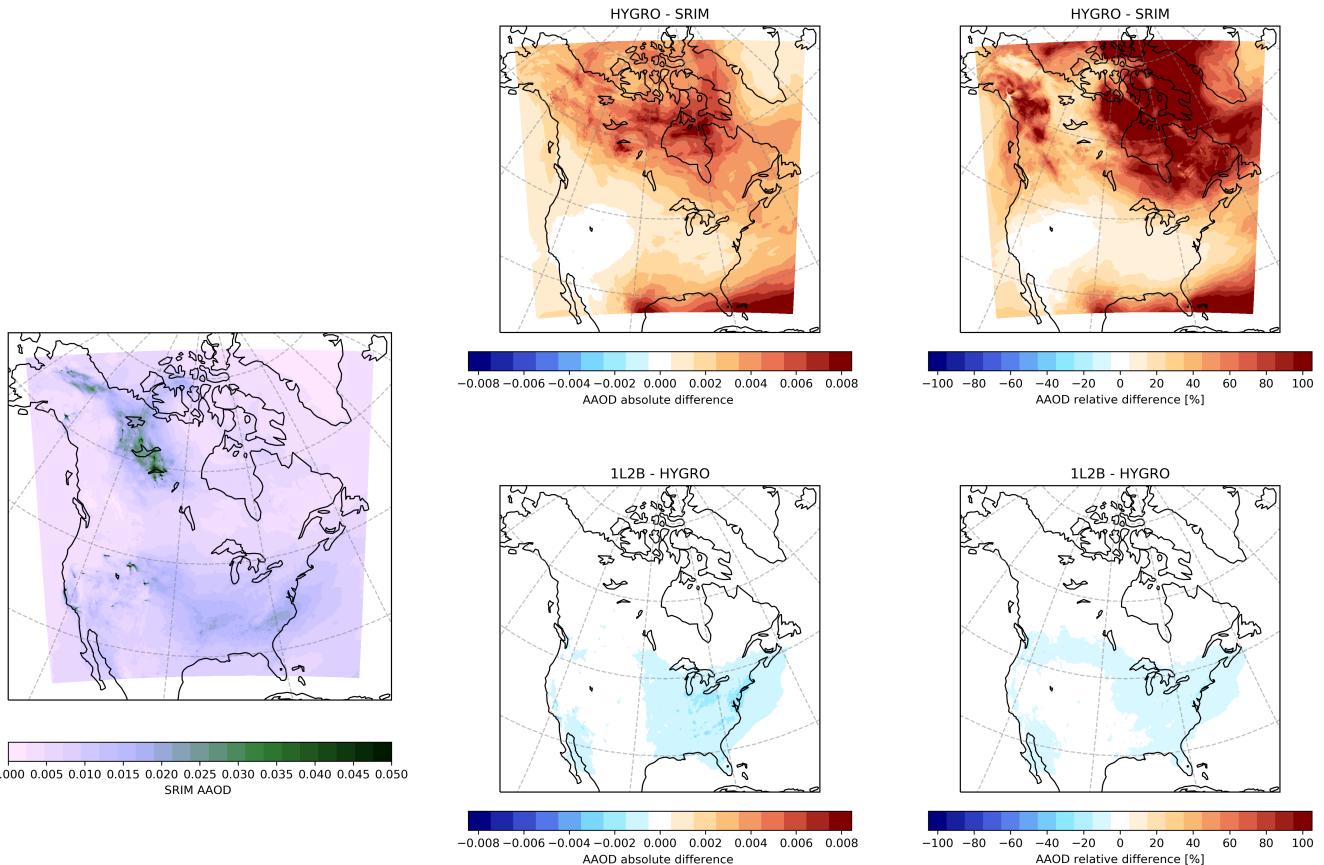
Figure 7. left: mean AAOD from the SRIM simulation; top centre: mean difference in AAOD between the HYGRO and SRIM simulations; top right: ~~mean~~ relative difference in mean AAOD between the ~~1L2B and~~ HYGRO and SRIM simulations; bottom centre: ~~relative~~ mean difference in ~~mean~~ AAOD between the ~~HYGRO~~ 1L2B and ~~SRIM~~ HYGRO simulations; bottom right: relative difference in mean AAOD between the 1L2B and HYGRO simulations. Only results from the hours of 1300-2100 UTC are included as the AOD is only calculated during local daylight hours.

regions, there is a net decrease in AAOD from the SRIM to 1L2B, when both the effects of mixing-state on wet deposition and absorption enhancement is considered, while for most of the rest of the North American domain, AAOD increases as the effect due to decreases in wet deposition is more important that the effect due to decreased absorption enhancement.

3.2 Aerosol-Meteorology Feedbacks

In order to examine the interactions between aerosol mixing-state representation and meteorology, we will now describe the results of the aerosol-meteorology feedbacks simulations. In these simulations, the cloud droplet number concentration is parameterized based on the aerosol size distribution using Abdul-Razzak and Ghan (2002), as described in Sect. 2 and the supplement. In the case of multiple mixing-state categories, the distinct composition of aerosol in each mixing-state category is considered, so aerosol in different mixing-state categories will have different critical radii for activation under the same atmospheric conditions. Additionally, aerosol and trace gas concentrations are permitted to reduce incoming radiation, which would subsequently alter atmospheric and surface energy balances.

As our focus is on the effects of differences in the aerosol mixing-state representation, we only compare cases with aerosol-meteorology feedbacks to other cases with aerosol-meteorology feedbacks. For comparisons of GEM-MACH results with and without aerosol-meteorology feedbacks, we refer the reader to Gong et al. (2015) and Makar et al. (2015a, b).

3.2.1 Aerosol-cloud interactions

In order to target low clouds most likely to be affected by aerosol emitted from the surface, we restrict our analysis to the clouds with model hybrid levels between 0.807 and 0.962, approximately 35-185 hPa below surface pressure. As all cloud variables were saved as 3-hourly means, which will include transitions between cloudy and cloud-free periods, our reported cloud properties will have smaller values than if we had analyzed instantaneous model output. This includes, most notably, the cloud droplet and raindrop number mixing ratios. However, as our interest is in the comparison between simulations, which are all treated identically, this would not alter our conclusions. Additionally, in order to provide more physically meaningful values, when calculating temporally and horizontally averaged cloud properties we define "cloudy" grid cells as those with 3-hourly cloud water mixing ratios (Q_C) >0.005g kg^{-1}, and we filter out grid cells with lower 3-hourly Q_C values. The mean number of cloudy grid cells differs by less than 0.7 % between simulations (not shown), with the HYGRO_feedbacks and 1L2B_feedbacks simulations having slightly more cloudy grid cells than the SRIM_feedbacks simulation. Therefore differences between simulations are better explained as changes in in-cloud properties, rather than as changes in the spatial extent of clouds. We note that the cloud fraction over the western United States was low during July of 2016, as evidenced in the MODIS satellite retrievals (NASA Earth Observations, https://neo.gsfc.nasa.gov/view.php?datasetId=MODAL2_M_CLD_FR&date=2016-07-28, last access November 19th, 2021).

We show in Fig. 8 the vertical distributions of the temporally and horizontally meaned in-cloud cloud droplet number mixing ratios (N_C), cloud water mixing ratios (Q_C), rain drop number mixing ratios (N_R), and rain water mixing ratios (Q_R) in

the SRIM_feedbacks, HYGRO_feedbacks, and 1L2B_feedbacks simulations. The HYGRO_feedbacks and 1L2B_feedbacks model simulations predict N_C values that are approximately 15 % larger than in the SRIM_feedbacks simulation. The difference in N_C is approximately constant with altitude. This difference is due to increased aerosol number concentrations, in turn due to both greater aerosol mass concentrations and smaller aerosol diameter, as shown in Fig. 9. These increased N_C values lead to mean Q_C values that are about 7 % greater than in the SRIM_feedbacks simulation. As N_C increases more than Q_C, the mean cloud droplet size will be decreased in the HYGRO_feedbacks and 1L2B_feedbacks simulations. These reduced cloud droplet sizes would be expected to result in reduced autoconversion and slower drizzle formation. Indeed, both N_R and Q_R are reduced in the HYGRO_feedbacks and 1L2B_feedbacks simulations relative to the SRIM_feedbacks simulation, by about 20 % for N_R and 9 % for Q_R. The difference in N_C is approximately constant with altitude, while the differences in Q_C, N_R and Q_R increase with altitude. For all cloud variables, the differences are slightly larger in the 1L2B_feedbacks simulation compared to the HYGRO_feedbacks simulation.

The decreases in in-cloud Q_R discussed above would be expected to result in decreases in precipitation at the surface. We show the mean precipitation from the SRIM simulation and the effects on precipitation of the HYGRO and 1L2B mixing-state representations in Fig. 10. Many of the differences shown in Fig. 10 include large decreases near large increases. These are due in part to small changes in advection patterns, which subsequently alter the locations of precipitation. We can determine the net effect of the difference in mixing-state representation on surface precipitation by averaging across the domain. When spatially and temporally averaged, the ~~effects~~ effect of mixing-state representation on precipitation is modest: In the HYGRO_feedbacks and 1L2B_feedbacks simulations, the precipitation is reduced by 0.6 % relative to the SRIM_feedbacks simulation, much smaller than the differences in in-cloud Q_R discussed above. As the decreases in N_R are greater than those in Q_R, the HYGRO_feedbacks and 1L2B_feedbacks simulations would have larger rain drops than the SRIM_feedbacks simulation, and these larger rain drops would settle to the surface more efficiently, thereby partially offsetting the reduction in Q_R.

The increases in N_C and Q_C shown above, along with the small increases in AOD shown in Sect. 3.1.5, would be expected to reduce the shortwave radiation reaching the surface and to potentially reduce surface temperatures. We show in Fig. 11 the differences in mean surface temperatures between the HYGRO_feedbacks and SRIM_feedbacks simulations and between the 1L2B_feedbacks and HYGRO_feedbacks simulations. Between HYGRO_feedbacks and SRIM_feedbacks, eastern and southern North America shows either small differences or noisy differences that would be consistent with slight changes in the locations of clouds. However, there is a clear increase of about 0.01 K over large areas of the oceans and a clear decrease of about 0.06 K over northern Quebec and eastern Nunavut. We note that this region encompasses the outflow of forest fires that occurred in north-eastern Canada during the simulation, as is visible in the differences in surface BC ~~concentrations~~ mixing ratios (Fig. 4). In the HYGRO_feedbacks simulation, the emissions from these forest fires are removed more slowly by wet deposition. Therefore, more aerosol particles remain to act as CCN further downwind from the source. The greater CCN concentration increases both N_C and Q_C within the cloud, reducing the solar radiation reaching the surface, and reducing surface temperatures. The differences in surface temperatures between 1L2B_feedbacks and HYGRO_feedbacks are noisy throughout
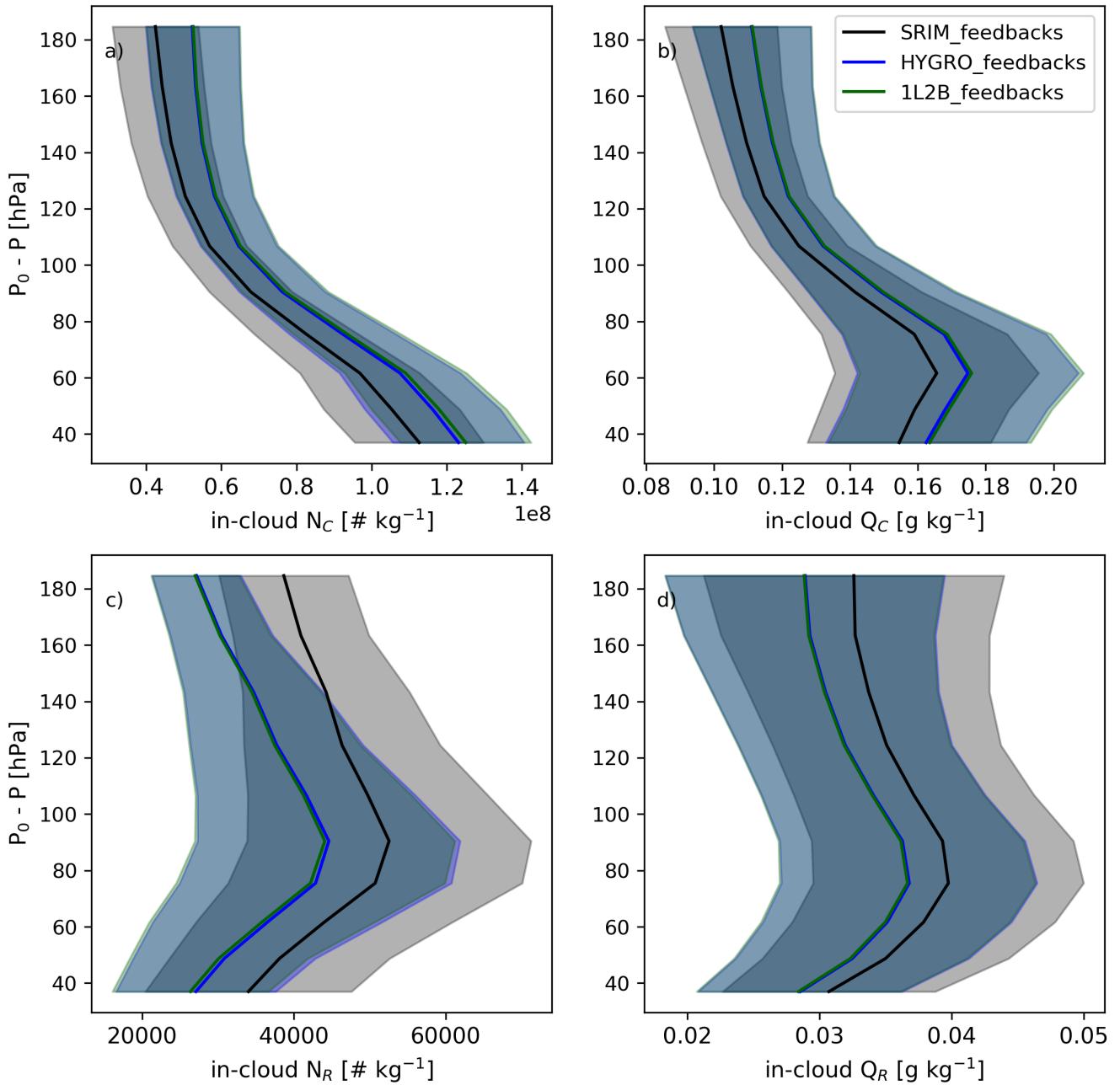
Figure 8. Vertical distribution of temporally and horizontally averaged cloud droplet number mixing ratios (N_C, a), cloud water mixing ratios (Q_C, b), rain drop number mixing ratios (N_R, c), and rain water mixing ratios (Q_R, d) in the SRIM_feedbacks, HYGRO_feedbacks, and 1L2B_feedbacks simulations. The shading indicates one temporal standard deviation about the mean.
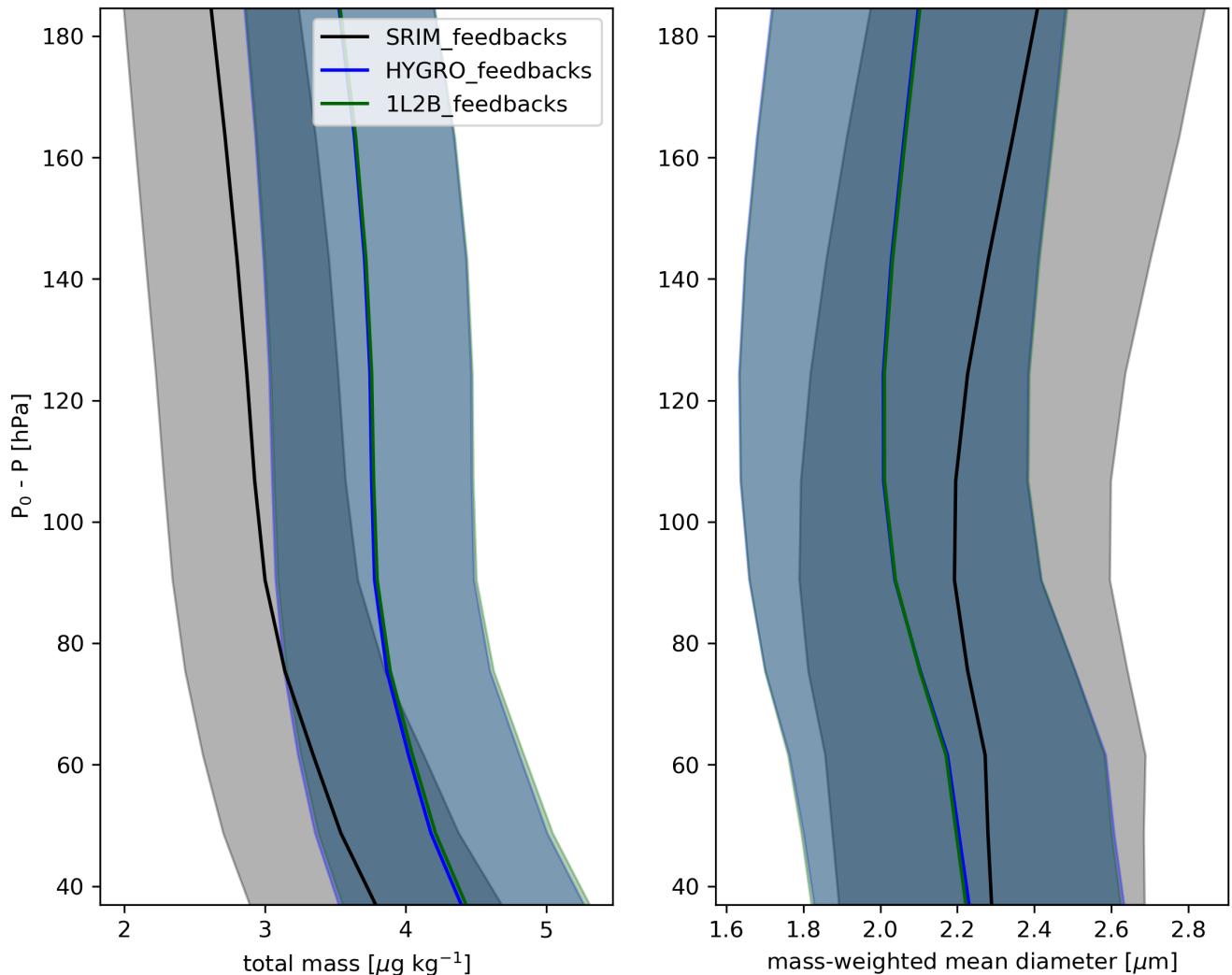
Figure 9. Vertical distribution of temporally and horizontally averaged aerosol properties in cloudy grid cells in the SRIM_feedbacks, HYGRO_feedbacks, and 1L2B_feedbacks simulations. Left: total mass mixing ratios. Right: mass-mean aerosol diameter. The shading indicates one temporal standard deviation about the mean.
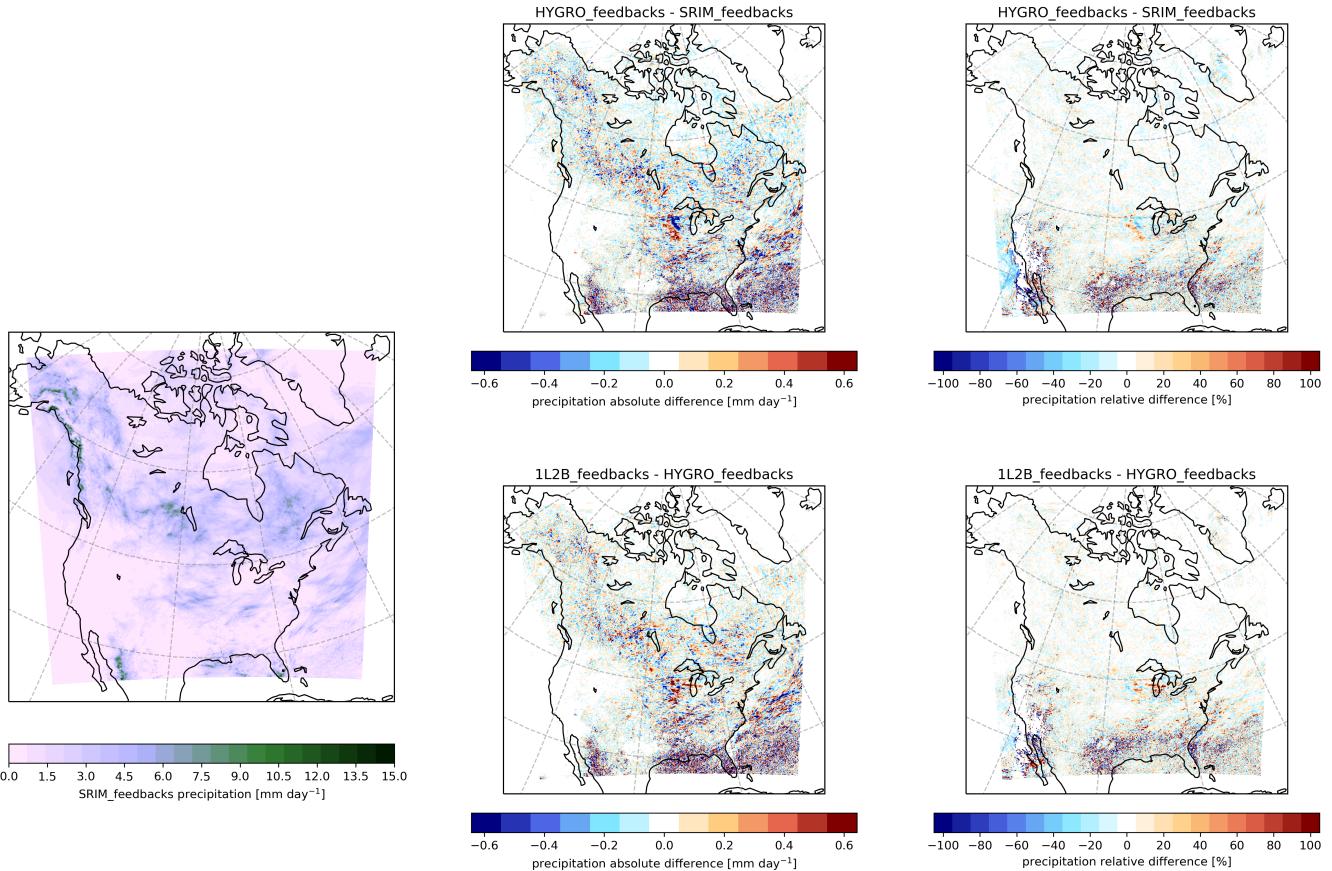
Figure 10. Left: mean precipitation from the SRIM_feedbacks simulation; top centre: mean difference in precipitation between the HYGRO_feedbacks and SRIM_feedbacks simulations; top right: ~~mean~~ relative difference in mean precipitation between the ~~1L2B~~HYGRO_feedbacks and ~~HYGRO~~SRIM_feedbacks simulations; bottom centre: ~~relative~~ mean difference in ~~mean~~ precipitation between the ~~HYGRO~~1L2B_feedbacks and ~~SRIM~~HYGRO_feedbacks simulations; bottom right: relative difference in mean precipitation between the 1L2B_feedbacks and HYGRO_feedbacks simulations.
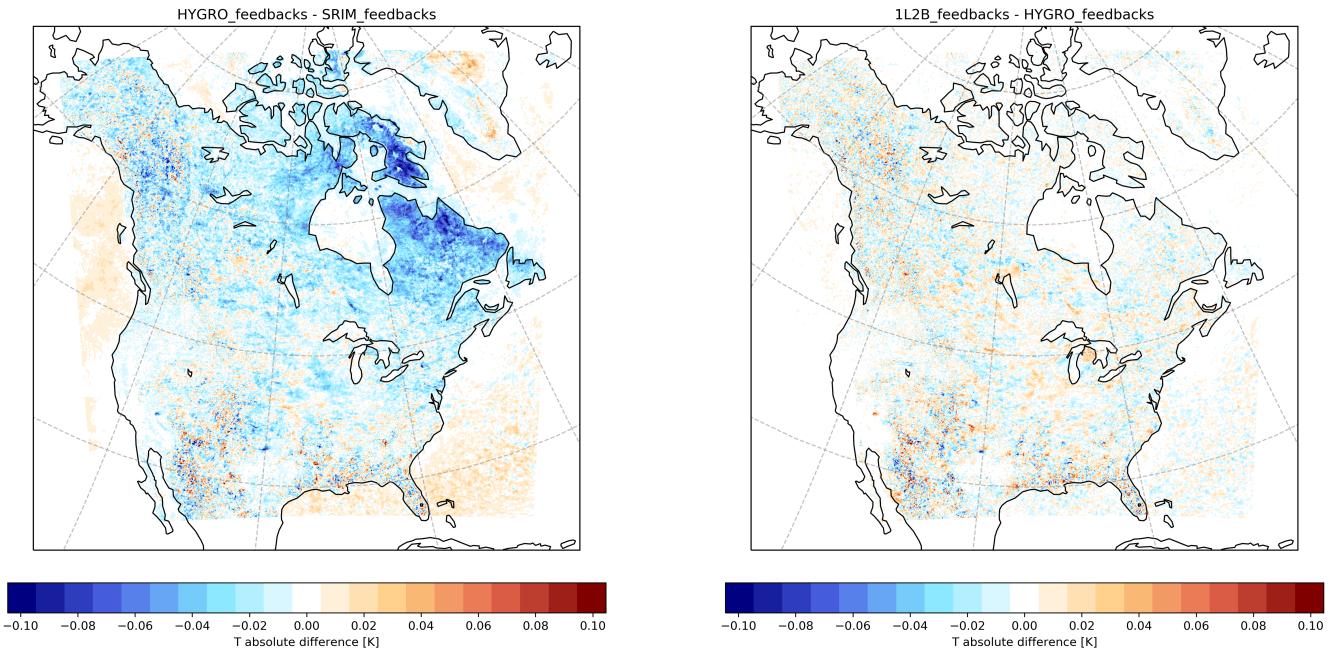
Figure 11. Left: mean difference in surface temperature between the HYGRO_feedbacks and SRIM_feedbacks simulations; right: mean difference in surface temperature between the 1L2B_feedbacks and HYGRO_feedbacks simulations.

the domain, consistent with slight changes in the locations of clouds. We therefore cannot determine any clear effect on surface temperatures due to differences in mixing-state representation between these two simulations.

485   We note that for all cloud properties, there are only small differences between 1L2B_feedbacks and HYGRO_feedbacks. We remind the reader that the differences in mixing-state representation between 1L2B and HYGRO were designed to capture the effects of correctly resolving the thickness of non-absorbing shells on BC and the subsequent enhancement in aerosol absorption. The effects of these differences in absorption enhancement would be permitted to affect atmospheric temperatures in our simulations, with potential subsequent effects on atmospheric stability. However, we do not find a strong effect on cloud 490   properties, surface precipitation or surface temperatures in this study. This may be, in part, due to our choice of case study.

4   Conclusions

In this study, we have implemented a detailed representation of aerosol mixing-state into the GEM-MACH air quality and weather forecast model. Our mixing-state representation includes three categories: one for more-hygroscopic aerosol, one for 495   less-hygroscopic aerosol with a high BC mass fraction, and one for less-hygroscopic aerosol with a low BC mass fraction. ~~This is the first model with a mixing-state representation of this type simulating a continent-scale domain.~~ Currently, the HYGRO

and 1L2B configurations require approximately 70 % and 150 % more running-time, respectively, than the SRIM configuration. We expect to reduce this additional cost through improvements to the efficiency of the model tracer transport scheme in the near future. The more-detailed representation allowed us to better resolve two different aspects of aerosol mixing state: First, differences in hygroscopicity due to differences in aerosol composition, including the change in hygroscopicity with time as less-hygroscopic aerosol becomes coated with hydrophilic material. Second, the thickness of non-absorbing coatings on BC aerosol which enhance the absorption of the BC aerosol.

We compared the results of the three-category representation (1L2B) with a simulation that uses two categories, split by hygroscopicity (HYGRO), and a simulation using the original size-resolved internally-mixed assumption (SRIM). We showed that when we included one or two categories of less-hygroscopic aerosol, wet deposition of BC, POA, SOA and dust was reduced, yielding increases in the mean concentrations of these species of 16-93 %, and an increase in the mean $PM_{2.5}$ concentration by 23 %. The effect on dust concentrations is likely overestimated, as the current implementation prevents ~~wet deposition~~ in-cloud scavenging of aerosol in the hydrophobic category, even if the aerosol is large. We intend to improve on this in a future version of GEM-MACH. As BC, POA, and SOA mass is more concentrated in smaller aerosol particles, we believe that the reductions of wet deposition in these species is realistic. The increased $PM_{2.5}$ concentrations led to an increase in the $AQHI_{2.5}$ by 0.05 units on average. The increases in aerosol concentrations also led to increases in both AOD and AAOD.

We briefly compared the results of the SRIM and 1L2B simulations and observations from the IMPROVE, CSN and AQS networks. However, we did not find significant improvement in model-observation agreement with the more-detailed mixing-state representation. The reduced wet deposition worsened an existing high bias in BC, organic matter, and dust concentrations, and we saw only small changes in correlation with the observations. It is likely that a more thorough assessment will require observations from sites that are strongly affected by long-range transport of BC and organic aerosol. The CSN network sites in particular are located in urban centres, and would therefore be expected to be weakly affected by changes in wet deposition. We will investigate this further in future work.

However, using two categories to resolve more-hygroscopic and less-hygroscopic aerosol only yielded modest improvements in resolving the amount of coating material on BC particles, which alters their absorption of solar radiation. We found that using three mixing-state categories (more-hygroscopic, less-hygroscopic high BC mass fraction, less-hygroscopic low BC mass fraction) allowed us to distinguish thinly coated BC from BC that was thickly coated with POA. This yielded a mean AAOD that was 3 % less than when separating the aerosol by hygroscopicity alone. Many sources of BC are also sources of POA, and observations indicate that the BC-containing particles frequently also contain POA, even close to emission sources (Perring et al., 2017; Kondo et al., 2011). We note that we assumed that particles from area sources were externally mixed at emission. This assumption will yield a maximum difference between our sensitivity simulations. Nonetheless, as thinly coated BC particles have been observed in the ambient atmosphere, even far from emission sources (Zanatta et al., 2018; Sharma et al., 2017), it is clear that POA and BC are not evenly distributed across particles in the same size range. The proportion of

POA that is emitted as BC-containing particles vs. BC-free particles is currently poorly constrained. We therefore suggest that future observation campaigns record not only the coating thickness on BC-containing particles, but also, when possible, the proportion of organic matter that exists as BC-free particles vs. BC-containing particles.

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We then performed simulations that included aerosol feedbacks on meteorology in order to determine the effects of mixing-state representation on the forecast meteorology. We found a clear effect due to including two categories of aerosol hygroscopicity: the increased aerosol concentrations due to the decreases in wet deposition increased cloud droplet mixing ratios by approximately 15 %. This led to a reduction in the mean precipitation by 0.6 %. The increased cloud reflectivity resulted

540 in a decrease in surface temperatures by about 0.06 K over northeastern Canada, in the outflow of large forest fires. When we compared the results of the HYGRO simulation with those of the ~~IL2B~~ 1L2B simulation, which better resolves BC mass fraction and aerosol absorption, we did not find a strong effect on forecast meteorology.

Code availability. GEM-MACH, the atmospheric chemistry library for the GEM numerical atmospheric model (© 2007–2013, Air Quality

545 Research Division and National Prediction Operations Division, Environment and Climate Change Canada), is a free software which can be redistributed and/or modified under the terms of the GNU Lesser General Public License as published by the Free Software Foundation – either version 2.1 of the license or any later version. The GEM (meteorology) code (CMC, 2021) is available to download from https://github.com/mfvalin?tab=repositories (last access: 27 April 2022). The executable for GEM-MACH is obtained by providing the chemistry library to GEM when generating its executable.

550 *Author contributions.* RS designed and implemented the novel mixing-state representations in GEM-MACH. RS led the writing and preparation of the manuscript. RS and AR performed the computational experiments and led the analysis. MM designed and implemented improved aerosol-radiation interactions in GEM-MACH. AD provided project oversight and guidance.

Competing interests. We declare that no competing interests are present.

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References

Abdul-Razzak, H. and Ghan, S. J.: A parameterization of aerosol activation 3. Sectional representation, Journal of Geophysical Research, 107, 4026, https://doi.org/10.1029/2001JD000483, 2002.

Adachi, K., Chung, S. H., and Buseck, P. R.: Shapes of soot aerosol particles and implications for their effects on climate, Journal of Geophysical Research, 115, D15 206, https://doi.org/10.1029/2009JD012868, 2010.

Akingunola, A., Makar, P. A., Zhang, J., Darlington, A., Li, S. M., Gordon, M., Moran, M. D., and Zheng, Q.: A chemical transport model study of plume-rise and particle size distribution for the Athabasca oil sands, Atmospheric Chemistry and Physics, 18, 8667–8688, https://doi.org/10.5194/acp-18-8667-2018, 2018.

Andersson, C., Bergström, R., Bennet, C., Robertson, L., Thomas, M., Korhonen, H., Lehtinen, K. E. J., and Kokkola, H.: MATCH-SALSA – Multi-scale Atmospheric Transport and CHemistry model coupled to the SALSA aerosol microphysics model – Part 1: Model description and evaluation, Geoscientific Model Development, 8, 171–189, https://doi.org/10.5194/gmd-8-171-2015, 2015.

Anttila, T.: Sensitivity of cloud droplet formation to the numerical treatment of the particle mixing state, Journal of Geophysical Research, 115, https://doi.org/10.1029/2010jd013995, 2010.

Appel, K. W., Pouliot, G. A., Simon, H., Sarwar, G., Pye, H. O. T., Napelenok, S. L., Akhtar, F., and Roselle, S. J.: Evaluation of dust and trace metal estimates from the Community Multiscale Air Quality (CMAQ) model version 5.0, Geoscientific Model Development, 6, 883–899, https://doi.org/10.5194/gmd-6-883-2013, 2013.

Aquila, V., Hendricks, J., Lauer, A., Riemer, N., Vogel, H., Baumgardner, D., Minikin, A., Petzold, A., Schwarz, J. P., Spackman, J. R., Weinzierl, B., Righi, M., and Dall'Amico, M.: MADE-in: a new aerosol microphysics submodel for global simulation of insoluble particles and their mixing state, Geoscientific Model Development, 4, 325–355, https://doi.org/10.5194/gmd-4-325-2011, 2011.

Bellouin, N., Mann, G. W., Woodhouse, M. T., Johnson, C., Carslaw, K. S., and Dalvi, M.: Impact of the modal aerosol scheme GLOMAP-mode on aerosol forcing in the Hadley Centre Global Environmental Model, Atmospheric Chemistry and Physics, 13, 3027–3044, https://doi.org/10.5194/acp-13-3027-2013, 2013.

Bergman, T., Kerminen, V.-M., Korhonen, H., Lehtinen, K. J., Makkonen, R., Arola, A., Mielonen, T., Romakkaniemi, S., Kulmala, M., and Kokkola, H.: Evaluation of the sectional aerosol microphysics module SALSA implementation in ECHAM5-HAM aerosol-climate model, Geoscientific Model Development, 5, 845–868, https://doi.org/10.5194/gmd-5-845-2012, 2012.

Bey, I., Jacob, D. J., Yantosca, R. M., Logan, J. A., Field, B. D., Fiore, A. M., Li, Q., Liu, H. Y., Mickley, L. J., and Schultz, M. G.: Global modeling of tropospheric chemistry with assimilated meteorology: Model description and evaluation, Journal of Geophysical Research: Atmospheres, 106, 23 073–23 095, https://doi.org/10.1029/2001JD000807, 2001.

Bieser, J., Aulinger, A., Matthias, V., Quante, M., and Builtjes, P.: SMOKE for Europe – adaptation, modification and evaluation of a comprehensive emission model for Europe, Geoscientific Model Development, 4, 47–68, https://doi.org/10.5194/gmd-4-47-2011, 2011.

Binkowski, F. S. and Roselle, S. J.: Models-3 Community Multiscale Air Quality (CMAQ) model aerosol component 1. Model description, Journal of Geophysical Research, 108, 4183, https://doi.org/10.1029/2001JD001409, 2003.

Bond, T. C., Habib, G., and Bergstrom, R. W.: Limitations in the enhancement of visible light absorption due to mixing state, Journal of Geophysical Research, 111, D20 211, https://doi.org/10.1029/2006JD007315, 2006.

Bond, T. C., Doherty, S. J., Fahey, D. W., Forster, P. M., Berntsen, T., DeAngelo, B. J., Flanner, M. G., Ghan, S., Kärcher, B., Koch, D., Kinne, S., Kondo, Y., Quinn, P. K., Sarofim, M. C., Schultz, M. G., Schulz, M., Venkataraman, C., Zhang, H., Zhang, S., Bellouin, N., Guttikunda, S. K., Hopke, P. K., Jacobson, M. Z., Kaiser, J. W., Klimont, Z., Lohmann, U., Schwarz, J. P., Shindell, D., Storelvmo, T.,

Warren, S. G., and Zender, C. S.: Bounding the role of black carbon in the climate system: A scientific assessment, Journal of Geophysical Research: Atmospheres, 118, 5380–5552, https://doi.org/10.1002/jgrd.50171, 2013.

595 Boucher, O., Randall, D., Artaxo, P., Bretherton, C., Feingold, G., Forster, P., Kerminen, V.-M. M., Kondo, Y., Liao, H., Lohmann, U., Rasch, P., Satheesh, S. K., Sherwood, S., Stevens, B., and Zhang, X. Y.: Clouds and Aerosols, in: Climate Change 2013: The Physical Science Basis. Contribution of Working Group I to the Fifth Assessment Report of the Intergovernmental Panel on Climate Change, edited by Stocker, T., Qin, D., Plattner, G.-K., Tignor, M., Allen, S., Boschung, J., Nauels, A., Xia, Y., Bex, V., and Midgley, P., pp. 571–657, Cambridge University Press, Cambridge, United Kingdom and New York, NY, USA, 2013.

600 Cappa, C. D., Onasch, T. B., Massoli, P., Worsnop, D. R., Bates, T. S., Cross, E. S., Davidovits, P., Hakala, J., Hayden, K. L., Jobson, B. T., Kolesar, K. R., Lack, D. A., Lerner, B. M., Li, S. M., Mellon, D., Nuaaman, I., Olfert, J. S., Petaja, T., Quinn, P. K., Song, C., Subramanian, R., Williams, E. J., and Zaveri, R. A.: Radiative Absorption Enhancements Due to the Mixing State of Atmospheric Black Carbon, Science, 337, 1078–1081, https://doi.org/10.1126/science.1223447, 2012.

Charron, M., Polavarapu, S., Buehner, M., Vaillancourt, P. A., Charette, C., Roch, M., Morneau, J., Garand, L., Aparicio, J. M., MacPherson, S., Pellerin, S., St-James, J., and Heilliette, S.: The Stratospheric Extension of the Canadian Global Deterministic Medium-Range Weather Forecasting System and Its Impact on Tropospheric Forecasts, Monthly Weather Review, 140, 1924–1944, https://doi.org/10.1175/MWR-D-11-00097.1, 2012.

Chen, J., Anderson, K., Pavlovic, R., Moran, M. D., Englefield, P., Thompson, D. K., Munoz-Alpizar, R., and Landry, H.: The FireWork v2.0 air quality forecast system with biomass burning emissions from the Canadian Forest Fire Emissions Prediction System v2.03, Geoscientific Model Development, 12, 3283–3310, https://doi.org/10.5194/gmd-12-3283-2019, 2019.

Ching, J., Zaveri, R. A., Easter, R. C., Riemer, N., and Fast, J. D.: A three-dimensional sectional representation of aerosol mixing state for simulating optical properties and cloud condensation nuclei, Journal of Geophysical Research: Atmospheres, 121, 5912–5929, https://doi.org/10.1002/2015jd024323, 2016.

Cohard, J.-M. and Pinty, J.-P.: A comprehensive two-moment warm microphysical bulk scheme. I: Description and tests, Quarterly Journal of the Royal Meteorological Society, 126, 1815–1842, https://doi.org/10.1002/qj.49712656613, 2010.

Côté, J., Desmarais, J.-G., Gravel, S., Méthot, A., Patoine, A., Roch, M., and Staniforth, A.: The Operational CMC–MRB Global Environmental Multiscale (GEM) Model. Part II: Results, Monthly Weather Review, 126, 1397–1418, https://doi.org/10.1175/1520-0493(1998)126<1397:TOCMGE>2.0.CO;2, 1998a.

Côté, J., Gravel, S., Méthot, A., Patoine, A., Roch, M., and Staniforth, A.: The Operational CMC-MRB Global Environmental Multiscale (GEM) Model. Part I: Design Considerations and Formulation, Monthly Weather Review, 126, 1373–1395, https://doi.org/10.1175/1520-0493(1998)126<1373:TOCMGE>2.0.CO;2, 1998b.

Cui, X., Wang, X., Yang, L., Chen, B., Chen, J., Andersson, A., and Gustafsson, O.: Radiative absorption enhancement from coatings on black carbon aerosols, Science of The Total Environment, 551-552, 51–56, https://doi.org/10.1016/j.scitotenv.2016.02.026, 2016.

Dalirian, M., Ylisirniö, A., Buchholz, A., Schlesinger, D., Ström, J., Virtanen, A., and Riipinen, I.: Cloud droplet activation of black carbon particles coated with organic compounds of varying solubility, Atmospheric Chemistry and Physics, 18, 12477–12489, https://doi.org/10.5194/acp-18-12477-2018, 2018.

Elleman, R. A. and Covert, D. S.: Aerosol size distribution modeling with the Community Multiscale Air Quality modeling system in the Pacific Northwest: 1. Model comparison to observations, Journal of Geophysical Research, 114, D11206, https://doi.org/10.1029/2008JD010791, 2009.

630 Emmons, L. K., Walters, S., Hess, P. G., Lamarque, J.-F., Pfister, G. G., Fillmore, D., Granier, C., Guenther, A., Kinnison, D., Laepple, T., Orlando, J., Tie, X., Tyndall, G., Wiedinmyer, C., Baughcum, S. L., and Kloster, S.: Description and evaluation of the Model for Ozone and Related chemical Tracers, version 4 (MOZART-4), Geoscientific Model Development, 3, 43–67, https://doi.org/10.5194/gmd-3-43-2010, 2010.

Fung, C., Misra, P., Bloxam, R., and Wong, S.: A numerical experiment on the relative importance of H2O2 O3 in aqueous conversion of
635 SO2 to SO42-, Atmospheric Environment. Part A. General Topics, 25, 411–423, https://doi.org/10.1016/0960-1686(91)90312-U, 1991.

Galmarini, S., Makar, P., Clifton, O. E., Hogrefe, C., Bash, J. O., Bellasio, R., Bianconi, R., Bieser, J., Butler, T., Ducker, J., Flemming, J., Hodzic, A., Holmes, C. D., Kioutsioukis, I., Kranenburg, R., Lupascu, A., Perez-Camanyo, J. L., Pleim, J., Ryu, Y.-H., San Jose, R., Schwede, D., Silva, S., and Wolke, R.: Technical note: AQMEII4 Activity 1: evaluation of wet and dry deposition schemes as an integral part of regional-scale air quality models, Atmospheric Chemistry and Physics, 21, 15 663–15 697, https://doi.org/10.5194/acp-21-15663-
640 2021, 2021.

Gong, S. L.: A parameterization of sea-salt aerosol source function for sub- and super-micron particles, Global Biogeochemical Cycles, 17, https://doi.org/10.1029/2003gb002079, 2003.

Gong, S. L., Barrie, L. A., and Lazare, M.: Canadian Aerosol Module: A size-segregated simulation of atmospheric aerosol processes for climate and air quality models 1. Module development, Journal of Geophysical Research, 108, https://doi.org/10.1029/2001JD002002,
645 2003.

Gong, W., Dastoor, A. P., Bouchet, V. S., Gong, S., Makar, P. A., Moran, M. D., Pabla, B., Ménard, S., Crevier, L.-P., Cousineau, S., and Venkatesh, S.: Cloud processing of gases and aerosols in a regional air quality model (AURAMS), Atmospheric Research, 82, 248–275, https://doi.org/10.1016/j.atmosres.2005.10.012, 2006.

Gong, W., Makar, P., Zhang, J., Milbrandt, J., Gravel, S., Hayden, K., Macdonald, A., and Leaitch, W.: Modelling aerosol–cloud–meteorology
650 interaction: A case study with a fully coupled air quality model (GEM-MACH), Atmospheric Environment, 115, 695–715, https://doi.org/10.1016/j.atmosenv.2015.05.062, 2015.

Han, X., Zhang, M., Zhu, L., and Xu, L.: Model analysis of influences of aerosol mixing state upon its optical properties in East Asia, Advances in Atmospheric Sciences, 30, 1201–1212, https://doi.org/10.1007/s00376-012-2150-4, 2013.

Hogrefe, C., Sistla, G., Zalewsky, E., Hao, W., and Ku, J.-Y.: An Assessment of the Emissions Inventory Processing Systems EMS-
655 2001 and SMOKE in Grid-Based Air Quality Models, Journal of the Air & Waste Management Association, 53, 1121–1129, https://doi.org/10.1080/10473289.2003.10466261, 2003.

Houyoux, M. R., Vukovich, J. M., Coats, C. J., Wheeler, N. J. M., and Kasibhatla, P. S.: Emission inventory development and processing for the Seasonal Model for Regional Air Quality (SMRAQ) project, Journal of Geophysical Research: Atmospheres, 105, 9079–9090, https://doi.org/10.1029/1999JD900975, 2000.

660 Kaiser, J. C., Hendricks, J., Righi, M., Riemer, N., Zaveri, R. A., Metzger, S., and Aquila, V.: The MESSy aerosol submodel MADE3 (v2.0b): Description and a box model test, Geoscientific Model Development, 7, 1137–1157, https://doi.org/10.5194/gmd-7-1137-2014, 2014.

Kaiser, J. C., Hendricks, J., Righi, M., Jöckel, P., Tost, H., Kandler, K., Weinzierl, B., Sauer, D., Heimerl, K., Schwarz, J. P., Perring, A. E., and Popp, T.: Global aerosol modeling with MADE3 (v3.0) in EMAC (based on v2.53): model description and evaluation, Geoscientific Model Development, 12, 541–579, https://doi.org/10.5194/gmd-12-541-2019, 2019.

665 Khalizov, A. F., Xue, H., Wang, L., Zheng, J., and Zhang, R.: Enhanced Light Absorption and Scattering by Carbon Soot Aerosol Internally Mixed with Sulfuric Acid, The Journal of Physical Chemistry A, 113, 1066–1074, https://doi.org/10.1021/jp807531n, 2009.

Kim, N., Park, M., Yum, S. S., Park, J. S., Shin, H. J., and Ahn, J. Y.: Impact of urban aerosol properties on cloud condensation nuclei (CCN) activity during the KORUS-AQ field campaign, Atmospheric Environment, 185, 221–236, https://doi.org/10.1016/j.atmosenv.2018.05.019, 2018.

670 Klingmüller, K., Steil, B., Brühl, C., Tost, H., and Lelieveld, J.: Sensitivity of aerosol radiative effects to different mixing assumptions in the AEROPT 1.0 submodel of the EMAC atmospheric-chemistry–climate model, Geoscientific Model Development, 7, 2503–2516, https://doi.org/10.5194/gmd-7-2503-2014, 2014.

Koehler, K. A., Kreidenweis, S. M., DeMott, P. J., Petters, M. D., Prenni, A. J., and Carrico, C. M.: Hygroscopicity and cloud droplet activation of mineral dust aerosol, Geophysical Research Letters, 36, https://doi.org/10.1029/2009GL037348, 2009.

675 Kokkola, H., Korhonen, H., Lehtinen, K. E. J., Makkonen, R., Asmi, A., Järvenoja, S., Anttila, T., Partanen, A.-I., Kulmala, M., Järvinen, H., Laaksonen, A., and Kerminen, V.-M.: SALSA - a sectional aerosol module for large scale applications, Atmospheric Chemistry and Physics, 8, 2469–2483, https://doi.org/10.5194/acp-8-2469-2008, 2008.

Kokkola, H., Kühn, T., Laakso, A., Bergman, T., Lehtinen, K. E. J., Mielonen, T., Arola, A., Stadtler, S., Korhonen, H., Ferrachat, S., Lohmann, U., Neubauer, D., Tegen, I., Siegenthaler-Le Drian, C., Schultz, M. G., Bey, I., Stier, P., Daskalakis, N., Heald, C. L., and

680 Romakkaniemi, S.: SALSA2.0: The sectional aerosol module of the aerosol–chemistry–climate model ECHAM6.3.0-HAM2.3-MOZ1.0, Geoscientific Model Development, 11, 3833–3863, https://doi.org/10.5194/gmd-11-3833-2018, 2018.

Kondo, Y., Matsui, H., Moteki, N., Sahu, L., Takegawa, N., Kajino, M., Zhao, Y., Cubison, M. J., Jimenez, J. L., Vay, S., Diskin, G. S., Anderson, B., Wisthaler, A., Mikoviny, T., Fuelberg, H. E., Blake, D. R., Huey, G., Weinheimer, A. J., Knapp, D. J., and Brune, W. H.: Emissions of black carbon, organic, and inorganic aerosols from biomass burning in North America and Asia in 2008, Journal of Geo-

685 physical Research, 116, D08 204, https://doi.org/10.1029/2010JD015152, 2011.

Lack, D. A., Langridge, J. M., Bahreini, R., Cappa, C. D., Middlebrook, A. M., and Schwarz, J. P.: Brown carbon and internal mixing in biomass burning particles, Proceedings of the National Academy of Sciences, 109, 14 802–14 807, https://doi.org/10.1073/pnas.1206575109, 2012.

Lauer, A., Hendricks, J., Ackermann, I., Schell, B., Hass, H., and Metzger, S.: Simulating aerosol microphysics with the ECHAM/-

690 MADE GCM - Part I: Model description and comparison with observations, Atmospheric Chemistry and Physics, 5, 3251–3276, https://doi.org/10.5194/acp-5-3251-2005, 2005.

Lee, L. a., Pringle, K. J., Reddington, C. L., Mann, G. W., Stier, P., Spracklen, D. V., Pierce, J. R., and Carslaw, K. S.: The magnitude and causes of uncertainty in global model simulations of cloud condensation nuclei, Atmospheric Chemistry and Physics, 13, 8879–8914, https://doi.org/10.5194/acp-13-8879-2013, 2013.

695 Lesins, G., Chylek, P., and Lohmann, U.: A study of internal and external mixing scenarios and its effect on aerosol optical properties and direct radiative forcing, Journal of Geophysical Research: Atmospheres, 107, AAC 5–1–AAC 5–12, https://doi.org/10.1029/2001jd000973, 2002.

Liu, D., Whitehead, J., Alfarra, M. R., Reyes-Villegas, E., Spracklen, D. V., Reddington, C. L., Kong, S., Williams, P. I., Ting, Y.-C., Haslett, S., Taylor, J. W., Flynn, M. J., Morgan, W. T., McFiggans, G., Coe, H., and Allan, J. D.: Black-carbon absorption enhancement in the

700 atmosphere determined by particle mixing state, Nature Geoscience, 10, 184–188, https://doi.org/10.1038/ngeo2901, 2017.

Liu, L. and Mishchenko, M.: Scattering and Radiative Properties of Morphologically Complex Carbonaceous Aerosols: A Systematic Modeling Study, Remote Sensing, 10, 1634+, https://doi.org/10.3390/rs10101634, 2018.

Liu, S., Aiken, A. C., Gorkowski, K., Dubey, M. K., Cappa, C. D., Williams, L. R., Herndon, S. C., Massoli, P., Fortner, E. C., Chhabra, P. S., Brooks, W. A., Onasch, T. B., Jayne, J. T., Worsnop, D. R., China, S., Sharma, N., Mazzoleni, C., Xu, L., Ng, N. L., Liu, D., Allan,

705   J. D., Lee, J. D., Fleming, Z. L., Mohr, C., Zotter, P., Szidat, S., and Prévôt, A. S. H.: Enhanced light absorption by mixed source black and brown carbon particles in UK winter, Nature Communications, 6, 8435, https://doi.org/10.1038/ncomms9435, 2015.

Liu, X., Easter, R. C., Ghan, S. J., Zaveri, R., Rasch, P., Shi, X., Lamarque, J. F., Gettelman, A., Morrison, H., Vitt, F., Conley, A., Park, S., Neale, R., Hannay, C., Ekman, A. M. L., Hess, P., Mahowald, N., Collins, W., Iacono, M. J., Bretherton, C. S., Flanner, M. G., and

710   Mitchell, D.: Toward a minimal representation of aerosols in climate models: description and evaluation in the Community Atmosphere Model CAM5, Geoscientific Model Development, 5, 709–739, https://doi.org/10.5194/gmd-5-709-2012, 2012.

Liu, X., Ma, P.-L. L., Wang, H., Tilmes, S., Singh, B., Easter, R. C., Ghan, S. J., and Rasch, P. J.: Description and evaluation of a new four-mode version of the Modal Aerosol Module (MAM4) within version 5.3 of the Community Atmosphere Model, Geoscientific Model Development, 9, 505–522, https://doi.org/10.5194/gmd-9-505-2016, 2016.

Majdzadeh, M., Stroud, C. A., Sioris, C., Makar, P. A., Akingunola, A., McLinden, C., Zhao, X., Moran, M. D., Abboud, I., and

715   Chen, J.: Development of aerosol optical properties for improving the MESSy photolysis module in the GEM-MACH v2.4 air quality model and application for calculating photolysis rates in a biomass burning plume, Geoscientific Model Development, 15, 219–249, https://doi.org/10.5194/gmd-15-219-2022, 2022.

Makar, P., Bouchet, V., and Nenes, A.: Inorganic chemistry calculations using HETV—a vectorized solver for the SO42-–NO3-–NH4+ system based on the ISORROPIA algorithms, Atmospheric Environment, 37, 2279–2294, https://doi.org/10.1016/S1352-2310(03)00074-

720   8, 2003.

Makar, P. A., Gong, W., Hogrefe, C., Zhang, Y., Curci, G., Žabkar, R., Milbrandt, J., Im, U., Balzarini, A., Baró, R., Bianconi, R., Cheung, P., Forkel, R., Gravel, S., Hirtl, M., Honzak, L., Hou, A., Jiménez-Guerrero, P., Langer, M., Moran, M. D., Pabla, B., Pérez, J. L., Pirovano, G., San José, R., Tuccella, P., Werhahn, J., Zhang, J., Galmarini, S., Hogrefe, C., Zhang, Y., Curci, G., Žabkar, R., Im, U., Balzarini, A., Baró, R., Bianconi, R., Cheung, P., Forkel, R., Gravel, S., Hirtl, M., Honzak, L., Hou, A., Jiménez-Guerrero, P., Langer, M., Moran, M. D.,

725   Pabla, B., Pérez, J. L., Pirovano, G., San José, R., Tuccella, P., Werhahn, J., Zhang, J., Galmarini, S., Milbrandt, J., Im, U., Balzarini, A., Baró, R., Bianconi, R., Cheung, P., Forkel, R., Gravel, S., Hirtl, M., Honzak, L., Hou, A., Jiménez-Guerrero, P., Langer, M., Moran, M. D., Pabla, B., Pérez, J. L., Pirovano, G., San José, R., Tuccella, P., Werhahn, J., Zhang, J., Galmarini, S., Hogrefe, C., Zhang, Y., Curci, G., Žabkar, R., Im, U., Balzarini, A., Baró, R., Bianconi, R., Cheung, P., Forkel, R., Gravel, S., Hirtl, M., Honzak, L., Hou, A., Jiménez-Guerrero, P., Langer, M., Moran, M. D., Pabla, B., Pérez, J. L., Pirovano, G., San José, R., Tuccella, P., Werhahn, J., Zhang, J.,

730   and Galmarini, S.: Feedbacks between air pollution and weather, Part 1: Effects on weather, Atmospheric Environment, 115, 442–469, https://doi.org/10.1016/j.atmosenv.2014.12.003, 2015a.

Makar, P. A., Gong, W., Hogrefe, C., Zhang, Y., Curci, G., Žabkar, R., Milbrandt, J., Im, U., Balzarini, A., Baró, R., Bianconi, R., Cheung, P., Forkel, R., Gravel, S., Hirtl, M., Honzak, L., Hou, A., Jiménez-Guerrero, P., Langer, M., Moran, M. D., Pabla, B., Pérez, J. L., Pirovano, G., San José, R., Tuccella, P., Werhahn, J., Zhang, J., Galmarini, S., Hogrefe, C., Zhang, Y., Curci, G., Žabkar, R., Im, U., Balzarini, A.,

735   Baró, R., Bianconi, R., Cheung, P., Forkel, R., Gravel, S., Hirtl, M., Honzak, L., Hou, A., Jiménez-Guerrero, P., Langer, M., Moran, M. D., Pabla, B., Pérez, J. L., Pirovano, G., San José, R., Tuccella, P., Werhahn, J., Zhang, J., Galmarini, S., Milbrandt, J., Im, U., Balzarini, A., Baró, R., Bianconi, R., Cheung, P., Forkel, R., Gravel, S., Hirtl, M., Honzak, L., Hou, A., Jiménez-Guerrero, P., Langer, M., Moran, M. D., Pabla, B., Pérez, J. L., Pirovano, G., San José, R., Tuccella, P., Werhahn, J., Zhang, J., Galmarini, S., Hogrefe, C., Zhang, Y., Curci, G., Žabkar, R., Im, U., Balzarini, A., Baró, R., Bianconi, R., Cheung, P., Forkel, R., Gravel, S., Hirtl, M., Honzak, L., Hou, A.,

740   Jiménez-Guerrero, P., Langer, M., Moran, M. D., Pabla, B., Pérez, J. L., Pirovano, G., San José, R., Tuccella, P., Werhahn, J., Zhang, J., and Galmarini, S.: Feedbacks between air pollution and weather, part 2: Effects on chemistry, Atmospheric Environment, 115, 442–469, https://doi.org/10.1016/j.atmosenv.2014.10.021, 2015b.

Manktelow, P. T., Carslaw, K. S., Mann, G. W., and Spracklen, D. V.: The impact of dust on sulfate aerosol, CN and CCN during an East Asian dust storm, Atmospheric Chemistry and Physics, 10, 365–382, https://doi.org/10.5194/acp-10-365-2010, 2010.

745 Mann, G. W., Carslaw, K. S., Spracklen, D. V., Ridley, D. A., Manktelow, P. T., Chipperfield, M. P., Pickering, S. J., and Johnson, C. E.: Description and evaluation of GLOMAP-mode: a modal global aerosol microphysics model for the UKCA composition-climate model, Geoscientific Model Development, 3, 519–551, https://doi.org/10.5194/gmd-3-519-2010, 2010.

Matsui, H.: Development of a global aerosol model using a two-dimensional sectional method: 1. Model design, Journal of Advances in Modeling Earth Systems, 9, 1921–1947, https://doi.org/10.1002/2017ms000936, 2017.

750 Matsui, H., Koike, M., Kondo, Y., Moteki, N., Fast, J. D., and Zaveri, R. A.: Development and validation of a black carbon mixing state resolved three-dimensional model: Aging processes and radiative impact, Journal of Geophysical Research: Atmospheres, 118, 2304–2326, https://doi.org/10.1029/2012JD018446, 2013.

Matsui, H., Koike, M., Kondo, Y., Fast, J. D., and Takigawa, M.: Development of an aerosol microphysical module: Aerosol Two-dimensional bin module for foRmation and Aging Simulation (ATRAS), Atmospheric Chemistry and Physics, 14, 10 315–10 331,
755 https://doi.org/10.5194/acp-14-10315-2014, 2014.

McFiggans, G., Artaxo, P., Baltensperger, U., Coe, H., Facchini, M. C., Feingold, G., Fuzzi, S., Gysel, M., Laaksonen, A., Lohmann, U., Mentel, T. F., Murphy, D. M., O'Dowd, C. D., Snider, J. R., and Weingartner, E.: The effect of physical and chemical aerosol properties on warm cloud droplet activation, Atmospheric Chemistry and Physics, 6, 2593–2649, https://doi.org/10.5194/acp-6-2593-2006, 2006.

Menut, L., Bessagnet, B., Khvorostyanov, D., Beekmann, M., Blond, N., Colette, A., Coll, I., Curci, G., Foret, G., Hodzic, A., Mailler, S.,
760 Meleux, F., Monge, J.-L., Pison, I., Siour, G., Turquety, S., Valari, M., Vautard, R., and Vivanco, M. G.: CHIMERE 2013: a model for regional atmospheric composition modelling, Geoscientific Model Development, 6, 981–1028, https://doi.org/10.5194/gmd-6-981-2013, 2013.

Moran, M. D., Ménard, S., Talbot, D., Huang, P., Makar, P. A., Gong, W., Landry, H., Gravel, S., Gong, S., Crevier, L.-P., Kallaur, A., and Sassi, M.: Particulate-matter forecasting with GEM-MACH15, a new Canadian air-quality forecast model, in: Air pollution modelling and
765 its application XX, pp. 289–292, Springer, Dordrecht, 2010.

Moran, M. D., Makar, P. A., Ménard, S., Pavlovic, R., Sassi, M., Beaulieu, P.-A., Anselmo, D., Mooney, C. J., Gong, W., Stroud, C., Gong, S., and Zhang, J.: Improvements to Wintertime Particulate-Matter Forecasting With GEM-MACH15, NATO Science for Peace and Security Series C: Environmental Security, pp. 591–597, Springer Netherlands, Dordrecht, https://doi.org/10.1007/978-94-007-1359-8_98, 2012.

Oshima, N., Koike, M., Zhang, Y., and Kondo, Y.: Aging of black carbon in outflow from anthropogenic sources using a mixing state
770 resolved model: 2. Aerosol optical properties and cloud condensation nuclei activities, Journal of Geophysical Research, 114, D18 202+, https://doi.org/10.1029/2008jd011681, 2009a.

Oshima, N., Koike, M., Zhang, Y., Kondo, Y., Moteki, N., Takegawa, N., and Miyazaki, Y.: Aging of black carbon in outflow from anthropogenic sources using a mixing state resolved model: Model development and evaluation, Journal of Geophysical Research, 114, https://doi.org/10.1029/2008jd010680, 2009b.

775 Peng, J., Hu, M., Guo, S., Du, Z., Zheng, J., Shang, D., Levy Zamora, M., Zeng, L., Shao, M., Wu, Y.-S., Zheng, J., Wang, Y., Glen, C. R., Collins, D. R., Molina, M. J., and Zhang, R.: Markedly enhanced absorption and direct radiative forcing of black carbon under polluted urban environments, Proceedings of the National Academy of Sciences, 113, 4266–4271, https://doi.org/10.1073/pnas.1602310113, 2016.

Perring, A. E., Schwarz, J. P., Markovic, M. Z., Fahey, D. W., Jimenez, J. L., Campuzano-Jost, P., Palm, B. D., Wisthaler, A., Mikoviny, T., Diskin, G., Sachse, G., Ziemba, L., Anderson, B., Shingler, T., Crosbie, E., Sorooshian, A., Yokelson, R., and Gao, R.-S.: In situ mea-

surements of water uptake by black carbon-containing aerosol in wildfire plumes: WATER UPTAKE BY BC IN WILDFIRE PLUMES, Journal of Geophysical Research: Atmospheres, 122, 1086–1097, https://doi.org/10.1002/2016JD025688, 2017.

Petters, M. D. and Kreidenweis, S. M.: A single parameter representation of hygroscopic growth and cloud condensation nucleus activity, Atmospheric Chemistry and Physics, 7, 1961–1971, https://doi.org/10.5194/acp-7-1961-2007, 2007.

Pitchford, M., Malm, W., Schichtel, B., Kumar, N., Lowenthal, D., and Hand, J.: Revised Algorithm for Estimating Light Extinction from IMPROVE Particle Speciation Data, Journal of the Air & Waste Management Association, 57, 1326–1336, https://doi.org/10.3155/1047-3289.57.11.1326, 2007.

Pringle, K. J., Tost, H., Message, S., Steil, B., Giannadaki, D., Nenes, A., Fountoukis, C., Stier, P., Vignati, E., and Lelieveld, J.: Description and evaluation of GMXe: a new aerosol submodel for global simulations (v1), Geoscientific Model Development, 3, 391–412, https://doi.org/10.5194/gmd-3-391-2010, 2010.

Reff, A., Bhave, P. V., Simon, H., Pace, T. G., Pouliot, G. A., Mobley, J. D., and Houyoux, M.: Emissions Inventory of PM$_{2.5}$ Trace Elements across the United States, Environmental Science & Technology, 43, 5790–5796, https://doi.org/10.1021/es802930x, 2009.

Regayre, L. A., Johnson, J. S., Yoshioka, M., Pringle, K. J., Sexton, D. M. H., Booth, B. B. B., Lee, L. A., Bellouin, N., and Carslaw, K. S.: Aerosol and physical atmosphere model parameters are both important sources of uncertainty in aerosol ERF, Atmospheric Chemistry and Physics, 18, 9975–10006, https://doi.org/10.5194/acp-18-9975-2018, 2018.

Riemer, N., Vogel, H., Vogel, B., and Fiedler, F.: Modeling aerosols on the mesoscale-γ: Treatment of soot aerosol and its radiative effects, Journal of Geophysical Research, 108, https://doi.org/10.1029/2003jd003448, 2003.

Riemer, N., West, M., Zaveri, R. A., and Easter, R. C.: Simulating the evolution of soot mixing state with a particle-resolved aerosol model, Journal of Geophysical Research, 114, https://doi.org/10.1029/2008jd011073, 2009.

Riemer, N., Ault, A. P., West, M., Craig, R. L., and Curtis, J. H.: Aerosol Mixing State: Measurements, Modeling, and Impacts, Reviews of Geophysics, 57, 187–249, https://doi.org/10.1029/2018RG000615, 2019.

Schnaiter, M., Linke, C., Möhler, O., Naumann, K.-H., Saathoff, H., Wagner, R., Schurath, U., and Wehner, B.: Absorption amplification of black carbon internally mixed with secondary organic aerosol, Journal of Geophysical Research, 110, D19204+, https://doi.org/10.1029/2005jd006046, 2005.

Sharma, S., Leaitch, W. R., Huang, L., Veber, D., Kolonjari, F., Zhang, W., Hanna, S. J., Bertram, A. K., and Ogren, J. A.: An evaluation of three methods for measuring black carbon in Alert, Canada, Atmospheric Chemistry and Physics, 17, 15225–15243, https://doi.org/10.5194/acp-17-15225-2017, 2017.

Spracklen, D. V., Pringle, K. J., Carslaw, K. S., Chipperfield, M. P., and Mann, G. W.: A global off-line model of size-resolved aerosol microphysics: II. Identification of key uncertainties, Atmospheric Chemistry and Physics, 5, 3233–3250, https://doi.org/10.5194/acp-5-3233-2005, 2005.

Spracklen, D. V., Carslaw, K. S., Pöschl, U., Rap, A., and Forster, P. M.: Global cloud condensation nuclei influenced by carbonaceous combustion aerosol, Atmospheric Chemistry and Physics, 11, 9067–9087, https://doi.org/10.5194/acp-11-9067-2011, 2011.

Stevens, R. and Dastoor, A.: A Review of the Representation of Aerosol Mixing State in Atmospheric Models, Atmosphere, 10, 168, https://doi.org/10.3390/atmos10040168, 2019.

Stieb, D. M., Burnett, R. T., Smith-Doiron, M., Brion, O., Shin, H. H., and Economou, V.: A New Multipollutant, No-Threshold Air Quality Health Index Based on Short-Term Associations Observed in Daily Time-Series Analyses, Journal of the Air & Waste Management Association, 58, 435–450, https://doi.org/10.3155/1047-3289.58.3.435, 2008.

Stier, P., Feichter, J., Kinne, S., Kloster, S., Vignati, E., Wilson, J., Ganzeveld, L., Tegen, I., Werner, M., Balkanski, Y., Schulz, M., Boucher, O., Minikin, A., and Petzold, A.: The aerosol-climate model ECHAM5-HAM, Atmospheric Chemistry and Physics, 5, 1125–1156, https://doi.org/10.5194/acp-5-1125-2005, 2005.

820 Tonttila, J., Maalick, Z., Raatikainen, T., Kokkola, H., Kühn, T., and Romakkaniemi, S.: UCLALES–SALSA v1.0: a large-eddy model with interactive sectional microphysics for aerosol, clouds and precipitation, Geoscientific Model Development, 10, 169–188, https://doi.org/10.5194/gmd-10-169-2017, 2017.

USEPA, U. S. E. P. A.: CMAQv5.2 Operational Guidance Document, pp. 1–224, https://github.com/USEPA/CMAQ/blob/5.2/DOCS/User_Manual/PDF/CMAQ_OGD_Full.06302017.pdf%0Apapers3://publication/uuid/2E2E71FA-F10A-4350-A526-34E61987E2DC, 2017.

825 Venkatram, A., Karamchandani, P., and Misra, P.: Testing a comprehensive acid deposition model, Atmospheric Environment (1967), 22, 737–747, https://doi.org/10.1016/0004-6981(88)90011-X, 1988.

Vignati, E., Wilson, J., and Stier, P.: M7: An efficient size-resolved aerosol microphysics module for large-scale aerosol transport models, Journal of Geophysical Research: Atmospheres, 109, n/a, https://doi.org/10.1029/2003jd004485, 2004.

Vignati, E., Karl, M., Krol, M., Wilson, J., Stier, P., and Cavalli, F.: Sources of uncertainties in modelling black carbon at the global scale, 830 Atmospheric Chemistry and Physics, 10, 2595–2611, https://doi.org/10.5194/acp-10-2595-2010, 2010.

Vogel, B., Vogel, H., Bäumer, D., Bangert, M., Lundgren, K., Rinke, R., and Stanelle, T.: The comprehensive model system COSMO-ART – Radiative impact of aerosol on the state of the atmosphere on the regional scale, Atmospheric Chemistry and Physics, 9, 8661–8680, https://doi.org/10.5194/acp-9-8661-2009, 2009.

Wang, Q., Huang, R. J., Cao, J., Han, Y., Wang, G., Li, G., Wang, Y., Dai, W., Zhang, R., and Zhou, Y.: Mixing State of Black Carbon 835 Aerosol in a Heavily Polluted Urban Area of China: Implications for Light Absorption Enhancement, Aerosol Science and Technology, 48, 689–697, https://doi.org/10.1080/02786826.2014.917758, 2014a.

Wang, X., Heald, C. L., Ridley, D. A., Schwarz, J. P., Spackman, J. R., Perring, A. E., Coe, H., Liu, D., and Clarke, A. D.: Exploiting simultaneous observational constraints on mass and absorption to estimate the global direct radiative forcing of black carbon and brown carbon, Atmospheric Chemistry and Physics, 14, 10 989–11 010, https://doi.org/10.5194/acp-14-10989-2014, 2014b.

840 Wang, X., Heald, C. L., Liu, J., Weber, R. J., Campuzano-Jost, P., Jimenez, J. L., Schwarz, J. P., and Perring, A. E.: Exploring the observational constraints on the simulation of brown carbon, Atmospheric Chemistry and Physics, 18, 635–653, https://doi.org/10.5194/acp-18-635-2018, 2018.

Willis, M. D., Healy, R. M., Riemer, N., West, M., Wang, J. M., Jeong, C.-H., Wenger, J. C., Evans, G. J., Abbatt, J. P. D., and Lee, A. K. Y.: Quantification of black carbon mixing state from traffic: implications for aerosol optical properties, Atmospheric Chemistry and Physics, 845 16, 4693–4706, https://doi.org/10.5194/acp-16-4693-2016, 2016.

Wilson, J., Cuvelier, C., and Raes, F.: A modeling study of global mixed aerosol fields, Journal of Geophysical Research: Atmospheres, 106, 34 081–34 108, https://doi.org/10.1029/2000jd000198, 2001.

Xu, X., Zhao, W., Qian, X., Wang, S., Fang, B., Zhang, Q., Zhang, W., Venables, D. S., Chen, W., Huang, Y., Deng, X., Wu, B., Lin, X., Zhao, S., and Tong, Y.: The influence of photochemical aging on light absorption of atmospheric black carbon and aerosol single-scattering 850 albedo, Atmospheric Chemistry and Physics, 18, 16 829–16 844, https://doi.org/10.5194/acp-18-16829-2018, 2018.

Yao, Y., Curtis, J. H., Ching, J., Zheng, Z., and Riemer, N.: Quantifying the effects of mixing state on aerosol optical properties, Atmospheric Chemistry and Physics, 22, 9265–9282, https://doi.org/10.5194/acp-22-9265-2022, 2022.

Zanatta, M., Laj, P., Gysel, M., Baltensperger, U., Vratolis, S., Eleftheriadis, K., Kondo, Y., Dubuisson, P., Winiarek, V., Kazadzis, S., Tunved, P., and Jacobi, H.-W.: Effects of mixing state on optical and radiative properties of black carbon in the European Arctic, Atmospheric Chemistry and Physics, 18, 14 037–14 057, https://doi.org/10.5194/acp-18-14037-2018, 2018.

Zaveri, R. A., Barnard, J. C., Easter, R. C., Riemer, N., and West, M.: Particle-resolved simulation of aerosol size, composition, mixing state, and the associated optical and cloud condensation nuclei activation properties in an evolving urban plume, Journal of Geophysical Research, 115, D17 210+, https://doi.org/10.1029/2009jd013616, 2010.

Zhang, J., Moran, M. D., Zheng, Q., Makar, P. A., Baratzadeh, P., Marson, G., Liu, P., and Li, S.-M.: Emissions preparation and analysis for multiscale air quality modeling over the Athabasca Oil Sands Region of Alberta, Canada, Atmospheric Chemistry and Physics, 18, 10 459–10 481, https://doi.org/10.5194/acp-18-10459-2018, 2018a.

Zhang, K., O'Donnell, D., Kazil, J., Stier, P., Kinne, S., Lohmann, U., Ferrachat, S., Croft, B., Quaas, J., Wan, H., Rast, S., and Feichter, J.: The global aerosol-climate model ECHAM-HAM, version 2: sensitivity to improvements in process representations, Atmospheric Chemistry and Physics, 12, 8911–8949, https://doi.org/10.5194/acp-12-8911-2012, 2012.

Zhang, R., Khalizov, A. F., Pagels, J., Zhang, D., Xue, H., and McMurry, P. H.: Variability in morphology, hygroscopicity, and optical properties of soot aerosols during atmospheric processing, Proceedings of the National Academy of Sciences, 105, 10 291–10 296, https://doi.org/10.1073/pnas.0804860105, 2008.

Zhang, Y., Favez, O., Canonaco, F., Liu, D., Močnik, G., Amodeo, T., Sciare, J., Prévôt, A. S. H., Gros, V., and Albinet, A.: Evidence of major secondary organic aerosol contribution to lensing effect black carbon absorption enhancement, npj Climate and Atmospheric Science, 1, 47, https://doi.org/10.1038/s41612-018-0056-2, 2018b.

Zieger, P., Väisänen, O., Corbin, J. C., Partridge, D. G., Bastelberger, S., Mousavi-Fard, M., Rosati, B., Gysel, M., Krieger, U. K., Leck, C., Nenes, A., Riipinen, I., Virtanen, A., and Salter, M. E.: Revising the hygroscopicity of inorganic sea salt particles, Nature Communications, 8, https://doi.org/10.1038/ncomms15883, 2017.

~~Supplemental figures~~ Supplement for "An Improved Representation of Aerosol Mixing State for Air-Quality-Weather Interactions"

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1 Detailed description of aerosol partitioning into cloud droplets

In GEM-MACH, the calculation of cloud droplet number concentrations (N_C) depends on whether or not aerosol feedbacks are enabled. In the simulations without aerosol-meteorology feedbacks, aerosol number concentrations have no effect on N_C for the purposes of determining cloud-radiation interactions and all cloud microphysics processes, including rain formation.

5 Instead, N_C for meteorological processes is calculated using a single pre-specified, constant cloud condensation nuclei type and concentration at all points in time and space.

For the purposes of determining aerosol processes, including aqueous chemistry and transport or removal of aerosol within cloud droplets, a diagnostic N_C is calculated based off of the total hydrophilic aerosol number concentration N_A, according to

10 the parameterisation of Jones et al. (1994):

$$N_C = 375(1 - exp(-2.5 \times 10^{-3} N_A)) \tag{1}$$

The N_C calculated in this way may differ from the N_C used for meteorological processes. Here, N_A is calculated by dividing the aerosol volume concentration in the hi-κ mixing-state category in each size bin by the volume of an aerosol particle with the midpoint diameter of the size bin and summing over the size bins. In the SRIM simulation, this reduces to the sum of the

15 total aerosol volume concentration in each size bin divided by the the volume of an aerosol particle with the midpoint diameter of the size bin. The largest N_C particles in the hi-κ mixing-state category are then selected to participate in in-cloud aerosol processes, including aqueous chemistry.

In the simulations with aerosol-meteorology feedbacks enabled, the algorithm described above differs in that N_C is parameterized

20 using Abdul-Razzak and Ghan (2002). Particle hygroscopicity is calculated separately for each mixing-state category based on molecular weights and ion dissociation, as per eq. 7 from Abdul-Razzak and Ghan (2002). Therefore, the aerosol mass in the lo-κ mixing-state categories is included when calculating N_C. The same value of N_C is used both for meteorological processes

and aerosol aqueous-phase processes.

25   However, the largest N_C particles in the hi-κ mixing-state category are still selected to participate in in-cloud aerosol processes, neglecting the aerosol mass in the lo-κ mixing-state category(ies). This may lead to some unphysical behaviour where additional mass in the lo-κ mixing-state category(ies) can lead to smaller aerosol particles in the hi-κ mixing-state category participating in in-cloud aerosol processes, due to the increase in N_C. Investigating this behaviour was beyond the scope of the current work. We intend to improve on the representation of this process in a future version of GEM-MACH, so

30   that the aerosol mass contributing to N_C (in both hi-κ and lo-κ mixing-state categories) is consistent with the aerosol mass participating in in-cloud aerosol processes. This will allow for the possibility of large aerosol particles in lo-κ mixing-state categories to participate in in-cloud aerosol processing and wet deposition.

      There are no cloud-borne aerosol tracers transported between chemistry time steps. Instead, aerosol mass is activated as

35   described above. For this portion of the aerosol mass only, aqueous chemistry is calculated, and cloud-to-rain conversion followed by either downwards transport by evaporating precipitation or wet deposition to the surface is accounted for. Afterwards, the new in-cloud aerosol mass is transferred back to the aerosol tracers.

2   Additional Figures
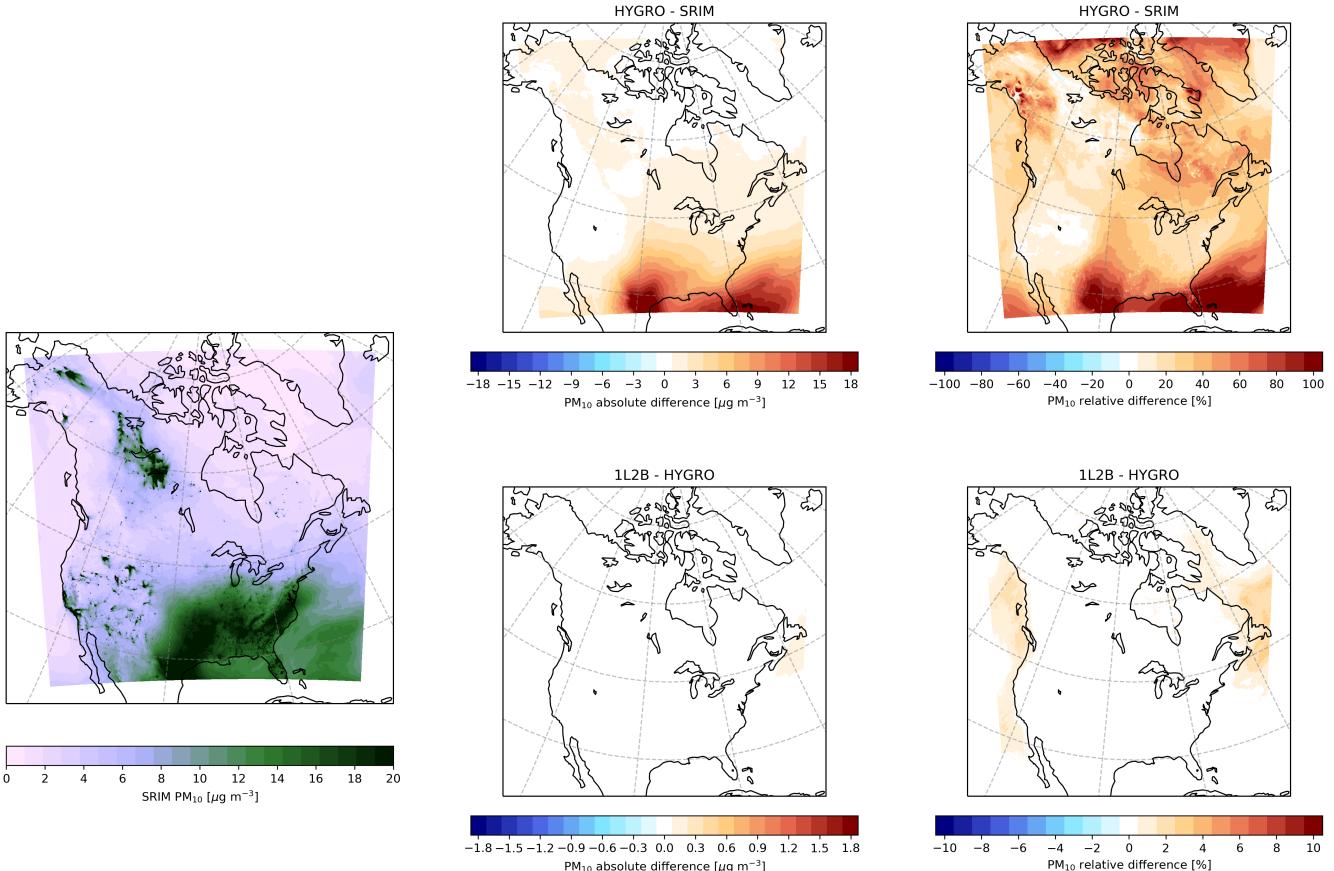
Figure S1. left: mean PM_{10} concentrations from the SRIM simulation; top ~~right~~centre: mean difference in PM_{10} concentrations between the HYGRO and SRIM simulations; ~~bottom~~ top right: relative difference in mean PM_{10} concentrations between the HYGRO and SRIM simulations~~. Note that~~; bottom centre: mean difference in PM_{10} concentrations ~~are nearly identical in~~ between the ~~HYGRO~~ 1L2B and HYGRO simulations; bottom right: relative difference in mean PM_{10} concentrations between the 1L2B and HYGRO simulations. Note that colour bar scales differ by a factor of ten between subplots in top row and subplots in bottom row.
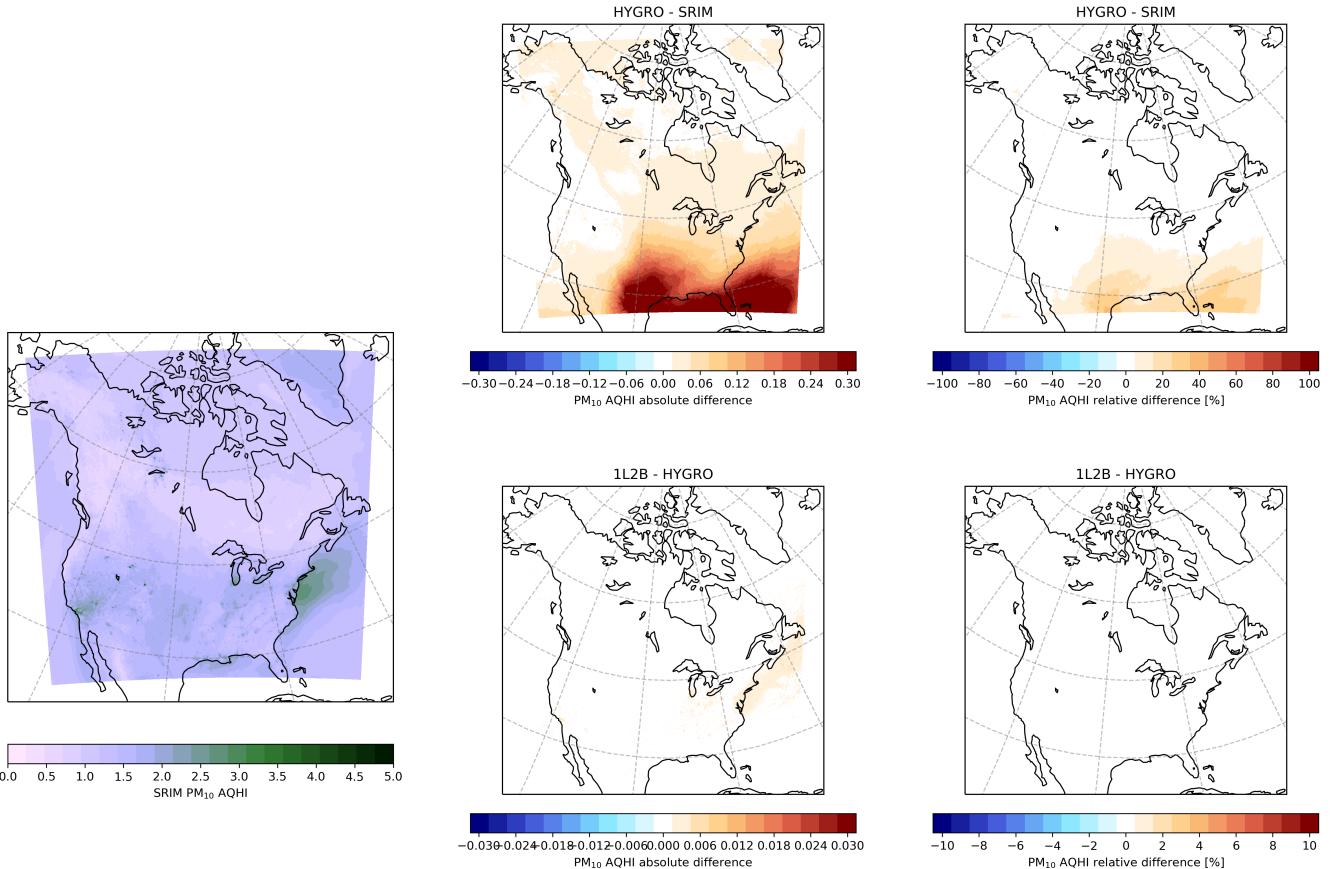
Figure S2. left: mean PM$_{10}$ AQHI values from the SRIM simulation; top ~~right~~centre: mean difference in PM$_{10}$ AQHI values between the HYGRO and SRIM simulations; ~~bottom~~ top right: relative difference in mean PM$_{10}$ AQHI values between the HYGRO and SRIM simulations~~. Note that~~; bottom centre: mean difference in PM$_{10}$ AQHI values ~~are nearly identical in~~ between the ~~HYGRO~~ 1L2B and HYGRO simulations; bottom right: relative difference in mean PM$_{10}$ AQHI values between the 1L2B and HYGRO simulations. Note that colour bar scales differ by a factor of ten between subplots in top row and subplots in bottom row.
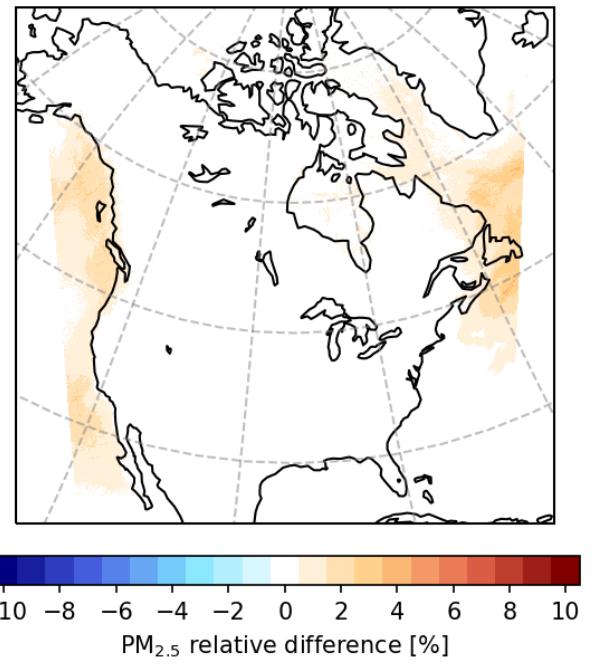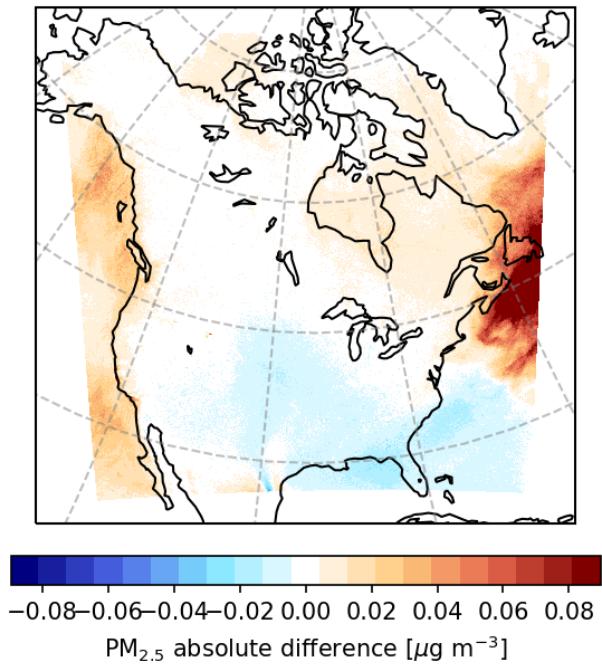
Figure S3. Mean differences in PM$_{2.5}$ values between the 1L2B and HYGRO simulations. left: absolute differences; right: relative differences. Note the finer colour bar scales compared to Fig. 1.
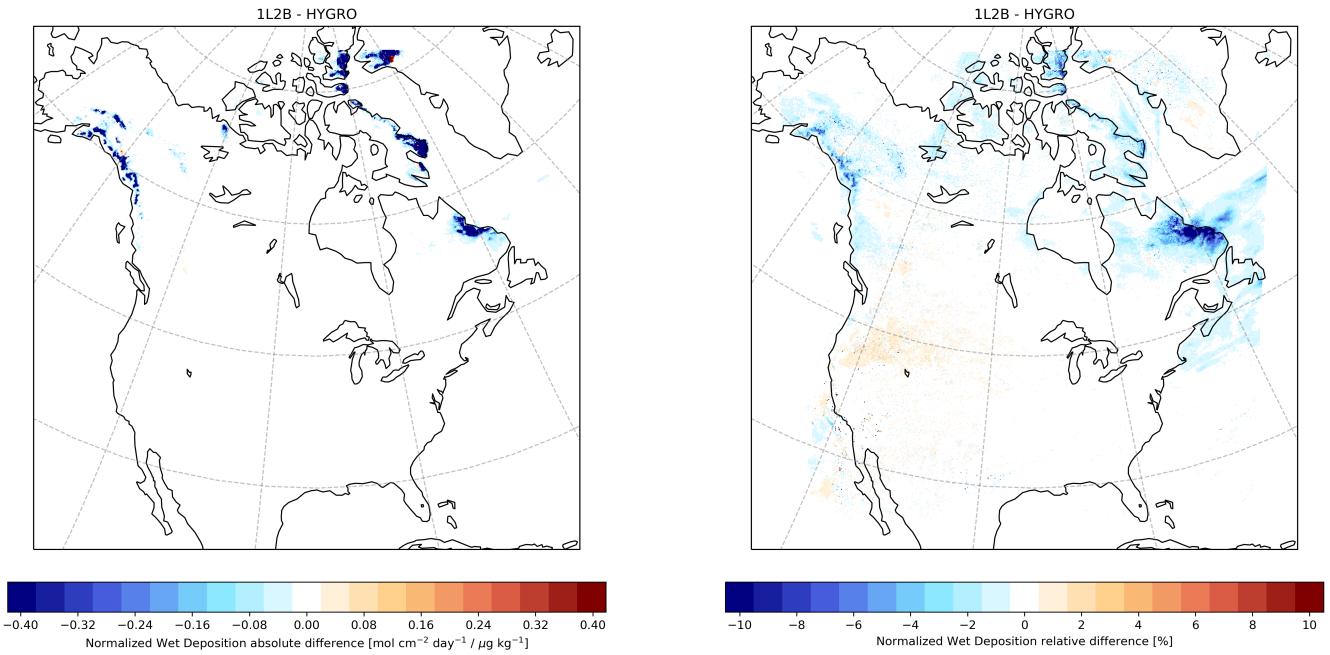
Figure S4. Mean differences in wet deposition fluxes normalized by surface total aerosol concentrations between the 1L2B and HYGRO simulations. left: absolute differences; right: relative differences. Note the finer colour bar scales compared to Fig. 2.
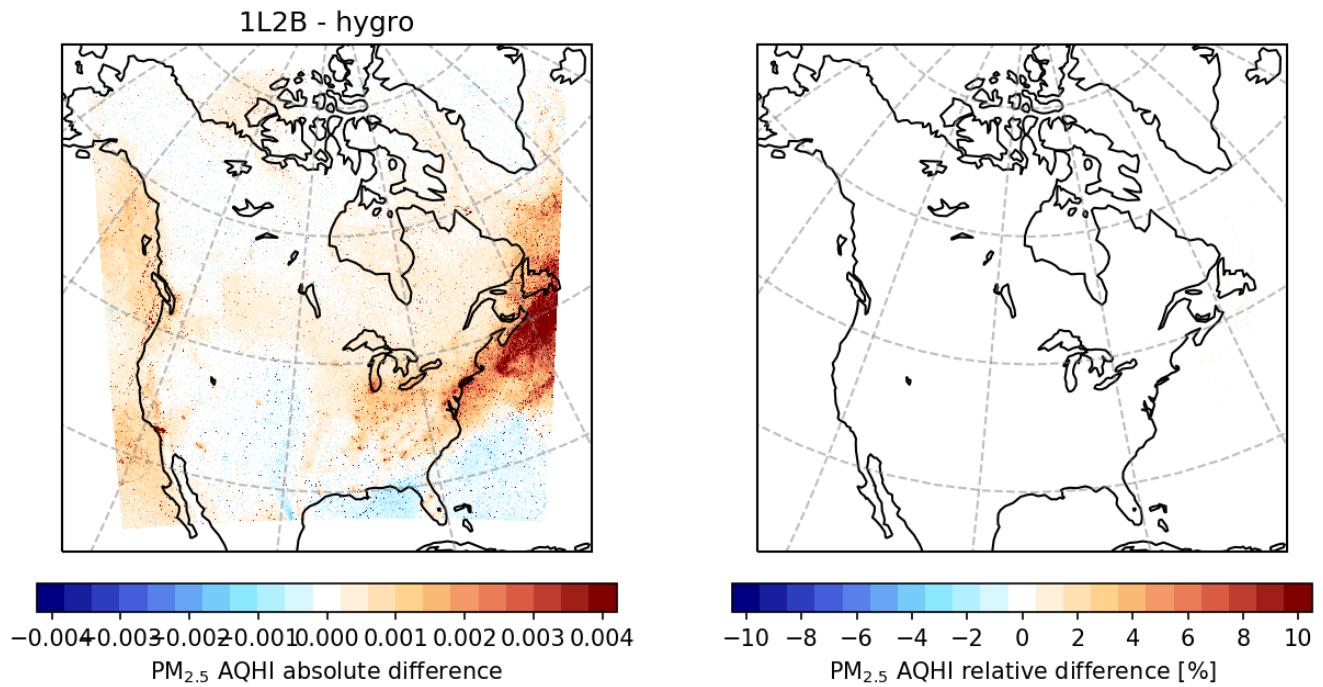
Figure S5. Mean differences in $PM_{2.5}$ AQHI values between the 1L2B and HYGRO simulations. left: absolute differences; right: relative differences. Note the finer colour bar scales compared to Fig. 3.
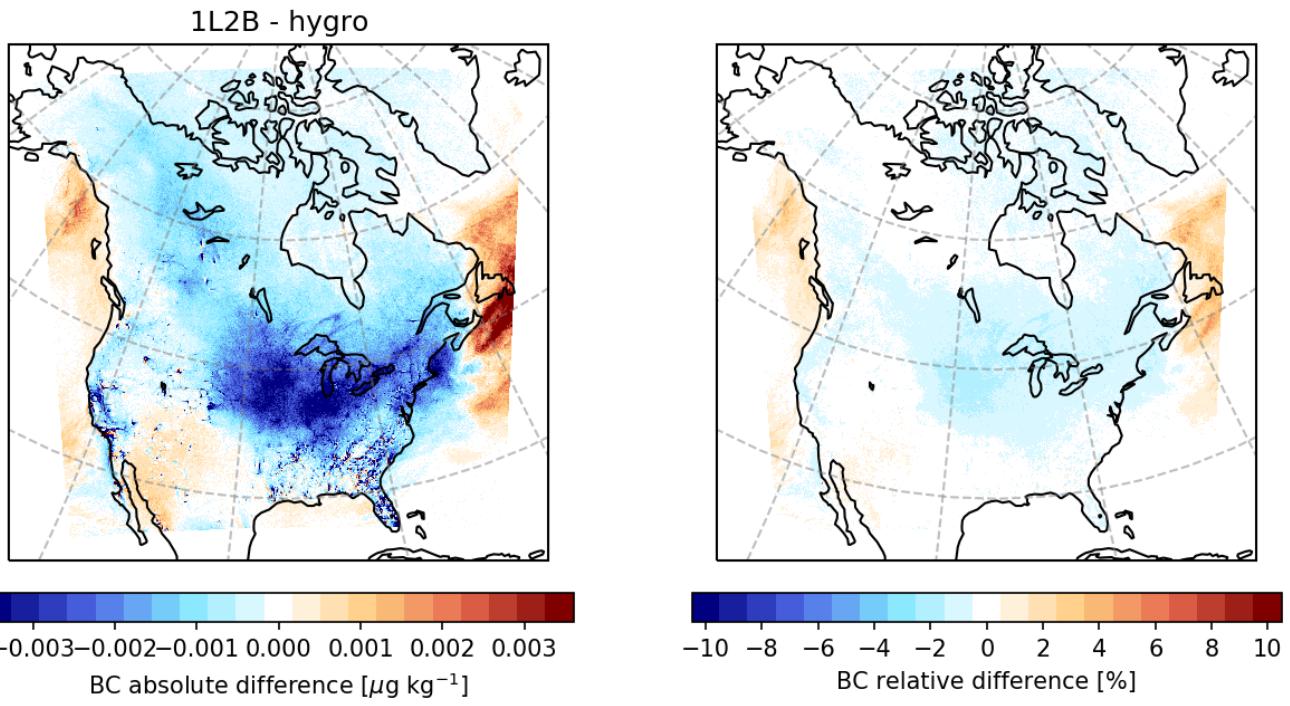
Figure S6. Mean differences in BC surface mixing ratios between the 1L2B and HYGRO simulations. left: absolute differences; right: relative differences. Note the finer colour bar scales compared to Fig. 4.
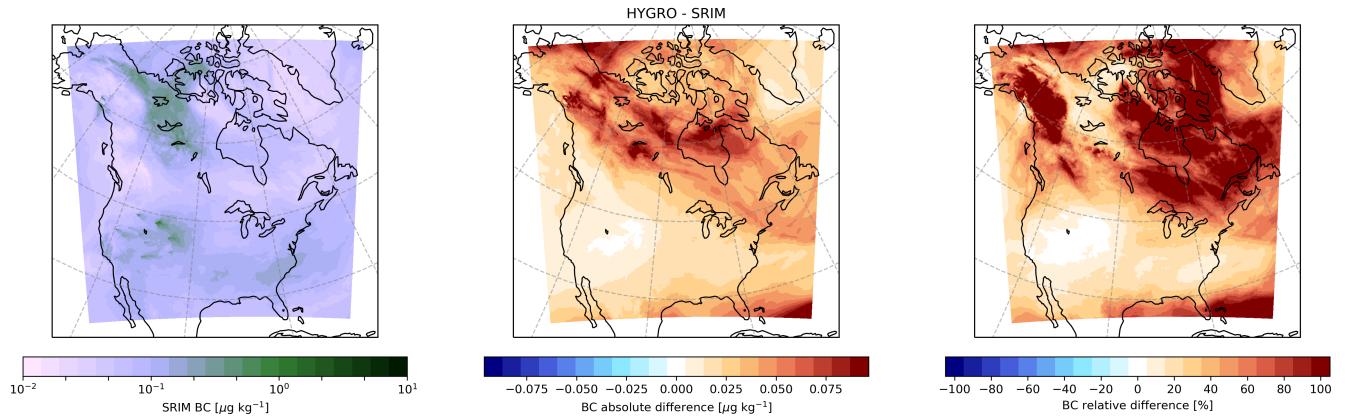
Figure S7. left: mean BC mixing ratios at about 185 hPa above the surface from the SRIM simulation; centre: mean difference in BC mixing ratios between the HYGRO and SRIM simulations; right: relative difference in mean BC mixing ratios between the HYGRO and SRIM simulations.
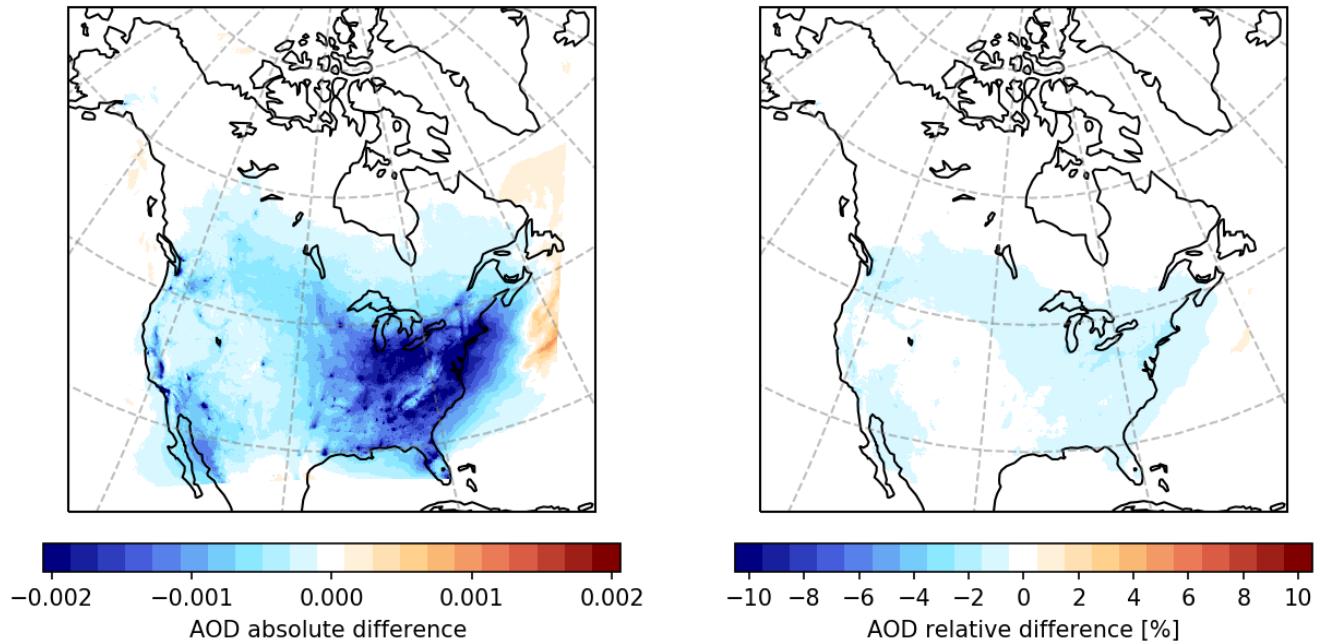
Figure S8. Mean differences in AOD values between the 1L2B and HYGRO simulations. left: absolute differences; right: relative differences. Note the finer colour bar scales compared to Fig. 5.
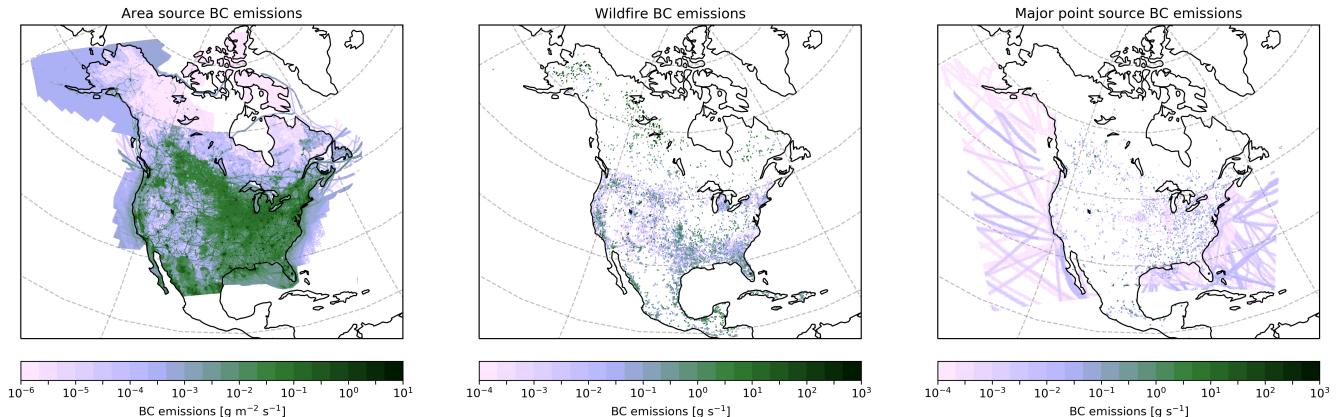
Figure S9. Mean emissions of BC during July 2016. Left: area emissions; centre: emissions from wildfires; right: other major point-source emissions. The wildfire and major point-source emissions are displayed on a 0.25° latitude by 0.25° longitude grid to allow the colours of isolated point sources to be visible. Note the difference in units between area and point sources. Note also that the same source sector (e.g. ocean shipping) can be classified as a major point source in some regions and an area source in others.

40 References

Abdul-Razzak, H. and Ghan, S. J.: A parameterization of aerosol activation 3. Sectional representation, Journal of Geophysical Research, 107, 4026, https://doi.org/10.1029/2001JD000483, 2002.

Jones, A., Roberts, D. L., and Slingo, A.: A climate model study of indirect radiative forcing by anthropogenic sulphate aerosols, Nature, 370, 450–453, https://doi.org/10.1038/370450a0, 1994.