



1 **Environmental Controls on Observed Spatial Variability of Soil**
2 **Pore Water Geochemistry in Small Headwater Catchments**
3 **Underlain with Permafrost**

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18 **Abstract.** Soil pore water (SPW) chemistry can vary substantially across multiple scales in Arctic permafrost
19 landscapes. The magnitude of these variations and their relationship to scale are critical considerations for
20 understanding current controls on geochemical cycling and for predicting future changes. These aspects are especially
21 important for Arctic change modelling where accurate representation of sub-grid variability may be necessary to
22 predict watershed scale behaviours. Our research goal was to characterize intra- and inter-watershed soil water
23 geochemical variations at two contrasting locations in the Seward Peninsula of Alaska, USA. We then attempt to
24 establish which environmental factors were important for controlling concentrations of important pore water solutes
25 in these systems. The SPW geochemistry of 18 locations spanning two small Arctic catchments were examined for
26 spatial variability and its dominant environmental controls. The primary environmental controls considered were
27 vegetation, soil moisture/redox condition, water/soil interactions and hydrologic transport, and mineral solubility. The
28 sampling locations varied in terms of vegetation type and canopy height, presence or absence of near-surface
29 permafrost, soil moisture, and hillslope position. Vegetation was found to have a significant impact on SPW NO₃
30 concentrations, associated with the localized presence of nitrogen-fixing alders and mineralization and nitrification of
31 leaf litter from tall willow shrubs. The elevated NO₃ concentrations were however, frequently equivoiced by increased
32 microbial denitrification in regions with sufficient moisture to support it. Vegetation also had an observable impact
33 on soil moisture sensitive constituents, but the effect was less significant. The redox conditions in both catchments
34 were generally limited by Fe reduction, seemingly well-buffered by a cache of amorphous Fe hydroxides, with the
35 most reducing conditions found at sampling locations with the highest soil moisture content. Non-redox-sensitive
36 cations were affected by a wide variety of water-soil interactions that affect mineral solubility and transport.
37 Identification of the dominant controls on current SPW hydrogeochemistry allows for qualitative prediction of future
38 geochemical trends in small Arctic catchments that are likely to experience warming and permafrost thaw. As source



39 areas for geochemical fluxes to the broader Arctic hydrologic system, geochemical processes occurring in these
40 environments are particularly important to understand and predict with regards to such environmental changes.

41 **1. Introduction**

42 Permafrost thaw in the Arctic is causing significant changes to landscape structure (Kokelj and Jorgenson, 2013;
43 Rowland et al., 2010), hydrology (Hiyama et al., 2021; Kurylyk et al., 2021; Liljedahl et al., 2016; Vonk, Tank, and
44 Walvoord, 2019; Walvoord and Kurylyk, 2016), vegetation (Lara, Nitze, Grosse, and McGuire, 2018; Myers-Smith
45 et al., 2011; Sturm, Racine, and Tape, 2001; K. D. Tape, Hallinger, Welker, and Ruess, 2012; K. Tape, Sturm, and
46 Racine, 2006;), and biogeochemistry (O'Donnell et al., 2021; Frey and McClelland, 2009; Salmon et al., 2019; Vonk,
47 Tank, and Walvoord, 2019). The integrated hydrogeochemical effects of these environmental changes are already
48 apparent in the chemistry of the large Arctic rivers, where fluxes of carbon and nutrients are increasing, leading to
49 enhanced nutrient loadings, with strong implications for the global carbon cycle (Bring et al., 2016; Fuchs et al., 2020;
50 McClelland et al., 2016). While the watershed areas of large Arctic rivers are vast, recent studies suggest that solute
51 concentrations in these large rivers are likely controlled by solute generation processes occurring at much smaller
52 scales (Harms and Ludwig, 2016; Koch, Runkel, Striegl, and McKnight, 2013; Shogren et al., 2019; Vonk et al.,
53 2015).

54 While there is a rapidly growing body of literature focused on observing and understanding environmental changes
55 over time with further Arctic warming, relatively few studies directly address the existing spatial variability, within
56 catchments or across catchments, and we are not aware of any studies that have combined field observations with
57 thermodynamic modelling in an effort to understand the causes of the existing spatial variability. Therefore, we have
58 a limited understanding of the key environmental controls on the spatial distribution of soil pore water solute
59 concentrations. In this study, we quantitatively evaluate the spatial variability of soil pore water (SPW) geochemistry
60 within and between two distinct catchments underlain with permafrost, and then seek to identify the source of the
61 observed spatial variability.

62 This study takes advantage of a scientifically diverse array of observations and datasets made available by the Next
63 Generation Ecosystem Experiment (NGEE) Arctic project, sponsored by the US Department of Energy Office of
64 Science. Most of the locations studied herein were selected by the NGEE Arctic project to provide co-located
65 measurements in a wide range of vegetation types, nested within representative hillslopes and catchments. Although
66 selected largely to represent a range of vegetation structure, such as shrub abundance and canopy height, these
67 locations also have considerable variability in other environmental parameters including, but not limited to: soil
68 moisture and temperature, presence or absence of near-surface permafrost, and maximum observed thaw depth (**Table**
69 **1** and **Table 2**). The vegetation-delineated sampling approach presented here provides an opportunity to not only
70 quantify the biogeochemical variability of SPW in Arctic environments, but also to investigate the root causes of that
71 observed variability. Data from additional sampling locations, available from a collaborative study, were also utilized
72 when possible.

73 Our overarching hypothesis is that vegetation-type and hillslope position are the dominant controls on spatial
74 variability of SPW geochemistry. Vegetation-type seems likely to have a significant effect on SPW geochemistry both



75 directly and indirectly. Directly, through the increased cycling of some solutes (e.g. increased nitrogen concentrations
76 in the vicinity of alders, which add nitrogen to soils via a symbiotic relationship with nitrogen-fixing bacterium;
77 Salmon et al., 2019), and indirectly, through effects on soil moisture (i.e. evapotranspiration and trapping of snow).
78 Soil moisture will also affect SPW geochemistry, particularly of redox sensitive species, by limiting oxygen diffusion
79 and thus controlling which regions develop anoxic/reducing geochemical conditions. Soil moisture impacts will likely
80 be correlated with vegetation-type as well as hillslope position, and the presence or absence of perching layers,
81 including permafrost, all of which impact the vertical and horizontal drainage characteristics of a watershed. Chemical
82 species that are not redox-sensitive or controlled by biogeochemical reactions are likely to be effected by transport,
83 solubility, and water/sediment/organic matter interactions, and therefore largely controlled by hillslope position as
84 well as soil moisture.

85 Identifying the dominant controls on solute concentration variability within each catchment and across catchments
86 will facilitate better projections of future soil pore hydrogeochemistry in permafrost landscapes, and how these
87 signatures are related to changing soil moisture and increasing in tundra shrub abundance in a changing Arctic (Bring
88 et al., 2016; Myers-Smith et al., 2011; Prowse, Bring, Mård, and Carmack, 2015; Salmon et al., 2019; Sturm et al.,
89 2001; K. D. Tape et al., 2012; K. Tape et al., 2006; Wrona et al., 2016, 2016). Arctic warming and associated
90 permafrost thaw will increase hydrological connectedness between terrestrial and aquatic environments through
91 deepening of the active layer and the formation of deeper, more coherent groundwater flow paths (Bring et al., 2016;
92 Harms and Jones, 2012; Prowse, Bring, Mård, Carmack, et al., 2015; Prowse, Bring, Mård, and Carmack, 2015).
93 Meanwhile, changes in hydrogeochemical signatures in larger Arctic rivers are likely to originate in smaller
94 catchments (McClelland et al., 2016; Prowse, Bring, Mård, and Carmack, 2015; Shogren et al., 2019; Spence, Kokelj,
95 McCluskie, and Hedstrom, 2015). In this sense, changes in hydrogeochemistry in small Arctic catchments not only
96 impact hydrogeochemistry at much larger scales, but also prognosticate the future hydrogeochemistry of larger Arctic
97 rivers.

98 **2. Methods**

99 **2.1 Site Descriptions**

100 This study focuses on two sites with permafrost on the Seward Peninsula of western Alaska, the Teller-27 Catchment
101 and the Kougarok-64 Hillslope (**Figure 1**). The Teller-27 Catchment, henceforth “Teller,” is a small (~2.25 km²)
102 headwater catchment located west of mile marker 27 along the Nome-Teller Highway northwest of Nome, Alaska.
103 The Kougarok-64 Hillslope, henceforth “Kougarok,” is a hillslope (~2.0 km²) located west of mile marker 64 along
104 the Nome-Taylor Highway northeast of Nome, Alaska. We utilized data from “intensive stations” at both Teller and
105 Kougarok where concentrated, multi-year, co-located observations of soil water chemistry, vegetation characteristics,
106 soil moisture and temperature, and other measurements have been collected as part of the NGEE Arctic Research
107 Project. These are identified as TL# (Teller Station #) or KG# (Kougarok Station #) in **Figure 2** and **Figure 3**,
108 respectively. Teller and Kougarok are not paired watersheds in the classical sense, differing in only one major
109 characteristic, which provides the basis for comparison. Instead, Teller and Kougarok differ in many respects and are



110 both representative of the broad range of hillslope conditions common on the Seward Peninsula. Detailed descriptions
111 of Teller and Kougarok have been published previously (Jafarov et al., 2018; Léger et al., 2019; Philben et al., 2019,
112 2020; Salmon et al., 2019; Yang et al., 2020), therefore, only the catchment characteristics that are probable sources
113 of variability in SPW chemistry will be highlighted here.

114 Teller is a discrete catchment with a well-defined central drainage, a vertical declivity of approximately 200 m, and a
115 catchment area of approximately 2.25 km². Temperature probes, soil pits, coring activities, and geophysical
116 interpretations at Teller have confirmed the catchment is underlain with discontinuous permafrost (Léger et al., 2019).
117 The upper shoulder of Teller (near Station 5, **Figure 2** and **Figure 5**) is underlain with near-surface permafrost and
118 appears to be a degraded peat plateau. The resultant microtopography of the degraded peat and the shallow perching
119 horizon caused by the permafrost creates a landscape of unsaturated peat mounds surrounded by ponds and saturated
120 soils. Downslope of the peat plateau, the Teller hillslope has highly variable soil moisture and vegetation (**Table 1**).
121 The microtopography within the lower footslope looks similar to the upper shoulder, but the peat appears more
122 severely degraded and the cause of the perched water table is less clear. Léger et al. (2019) suggest the presence of
123 permafrost at a depth of 1 – 2 m at Teller Station 9 (**Figure 2**), but the perching could also be caused by a layer of silt,
124 at a depth of approximately 30 cm (Graham et al., 2018). The full extent of permafrost and silt in this region of the
125 catchment remains unknown, but the thaw depth in July 2018 was greater than 1 m and maintained a perched water
126 table (Philben et al., 2020), suggesting perching could be the result of silt rather than permafrost. Vegetation type,
127 moisture content, permafrost extent, and hillslope position for all Teller Stations are summarized in **Table 1**.

128 Kougarok differs in many ways from Teller, although both have characteristics that are typical of hillslopes on the
129 Seward Peninsula. Kougarok is a convex hillslope, with a vertical declivity of approximately 70 meters. The study
130 area at Kougarok is approximately 2.0 km². Soil temperature measurements at Kougarok suggest that the vast majority
131 of the site is underlain by shallow continuous permafrost (Romanovsky, Cable, and Dolgikh, 2020a); Kougarok
132 Station 5 is an exception, where the permafrost is deeper (Romanovsky et al., 2020a). The upper shoulder of Kougarok
133 is a well-drained rocky outcrop composed of metagranitic rock (Hopkins et al., 1955; Till, Dumoulin, Werdon, and
134 Bleick, 2011). Saturated soils are not prevalent until the footslope and the lower backslope, where Kougarok Stations
135 2, 11, 10, 1, and 6 are situated (**Figure 3**). The lower backslope is characterized by persistent saturation between
136 ubiquitous tussocks, formed by the tussock cotton grass *Eriophorum vaginatum*. The tussock-lichen tundra at
137 Kougarok introduces microtopography and spatially variable saturation; in this sense, the Kougarok tussocks are
138 analogous to the peat mounds and hummocks at Teller, but on different spatial scales and formed by different
139 processes. Kougarok has numerous patches of alder shrubland in an altitudinal band within the upper backslope; it
140 should be emphasized that Teller lacks tussock-lichen tundra and alder (*Alnus viridis* ssp. *fruticosa*) shrubs that are a
141 feature of Kougarok. While continuous permafrost largely remains, the Kougarok site appears to be undergoing
142 environmental changes as evidenced by an increase in alder coverage over the past decades (Salmon et al., 2019). Soil
143 profiles underneath the alder patches are rocky with shallow bedrock and warmer permafrost (**Table 2**). Shrub tundra
144 (alder savanna in tussock tundra and willow-birch tundra) dominates the lower backslope, where the annual active
145 layer thickness is typically less than 100 cm. Vegetation type, moisture content, permafrost extent, and hillslope
146 position at all Kougarok stations are summarized in **Table 2**.



147 2.2 Sampling & Analytical Approach

148 SPWs were sampled using two complimentary techniques. Fiberglass wicks (Frisbee, Phillips, Campbell, and
149 Hendrickx, 2010) were deployed in the upper 30 cm of soils at stations where shallow soils were unsaturated. These
150 wicks were left in place from year-to-year and only replaced if damage was observed or suspected. The sample
151 reservoirs from the wicks were collected whenever possible, usually a few times each summer. MacroRhizons
152 (Rhizosphere Research Products; Netherlands) were used at stations that were more saturated, also targeting the upper
153 30 cm of soils. Both techniques were used at stations of intermediate saturation, where both could be deployed
154 effectively. MacroRhizons represent a relatively discrete temporal sampling event (minutes to hours), whereas wicks
155 represent a cumulative water collected over longer periods (weeks to months). It is in this sense, that the two techniques
156 are complimentary. Unfortunately, due to saturation variability both techniques could not be used at all stations and
157 conditions at some Kougarok stations were sometimes too dry to collect meaningful volumes of SPW using either
158 method. Additional SPW data from Kougarok were supplemented from a separate study focused on alder-related
159 nutrient dynamics (McCaully et al., In Review.). These data were collected by MacroRhizons and are captured as
160 Kougarok Stations 10 – 13, which were not part of the original stations established by the NGEE Arctic Program. A
161 total of 309 SPW samples from Kougarok were collected and analysed, whereas a total of 89 SPW samples from
162 Teller were collected and analysed.

163 After collection, SPW cation concentrations were measured in triplicate by inductively coupled plasma optical
164 emission spectroscopy (Optima 2100 DV; PerkinElmer, USA) following US EPA Method 200.7. Inorganic anion
165 concentrations were measured by ion chromatography (DX-600; Dionex, USA) following US EPA Method 300.0. B,
166 F, K, Na, and Si concentrations collected by wicks were excluded from the dataset due to known issues with these
167 ions leeching from fiberglass wick samplers (Perdrial et al., 2014; Wallenberger and Bingham, 2009). This effect is
168 illustrated in Supplementary **Figure 1** and the lack of such an effect for divalent cations is shown in Supplementary
169 **Figure 2**. Comparison of data from wicks and MacroRhizons, along with the observations from (Perdrial et al., 2014),
170 demonstrates that remaining constituents discussed herein were not affected by collection with fiberglass wicks.
171 Alkalinity, pH, and E_H are all critical geochemical parameters that are susceptible to change during storage (Petrone,
172 Hinzman, Shibata, Jones, and Boone, 2007); because of the large amount of data from wicks these parameters were
173 not considered further, except in the context of thermodynamic modelling.

174 Observations related to vegetation, soil moisture, and permafrost extend were compiled from datasets made available
175 by the NGEE Arctic project and are given for Teller in **Table 1** and for Kougarok in **Table 2**. The reported soil
176 moisture contents were derived from an average of gravimetric measurements (2017 and 2018) and time domain
177 reflectometry measurements (2017 and 2019), and from remotely-sensed P-band Synthetic Aperture Radar (2017).
178 End-of-winter snow depths were measured in March and April of 2016, 2017, and 2018. The annual average ground
179 temperature was measured using in-situ temperature sensors (HOBO U30 DataLogger) at a depth of 1.5 meters below
180 the ground surface (Romanovsky et al., 2020a; Romanovsky, Cable, and Dolgikh, 2020b) and the active layer
181 thicknesses were determined by frost probe in September 2019 at the end of the growing season.



182 2.3 Statistical Analysis

183 Principal Components Analysis (PCA) and the Mann-Whitney U-Test (MWUT) were both used to investigate
184 dominant environmental controls on solute concentrations in SPWs at Teller and Kougarok. PCA is an exploratory
185 data analysis tool that reduces the dimensionality of large complex datasets and considers how components (i.e. solute
186 concentrations) vary together. Because PCA was predominately used as a screening tool to reveal geochemical
187 correlations that may not have been evident by traditional geochemical causations or inference, a detailed discussion
188 of the PCA results is reserved to the Supplementary Materials. The MWUT was used to test for significant differences
189 in solute concentrations between Teller and Kougarok (inter-site variability) and between stations at each site (intra-
190 site variability). The MWUT is a non-parametric method of challenging a null hypothesis, which in this case is the
191 assumption that the concentrations of a given solute are not systematically greater at either site nor at any particular
192 station. Water chemistry data are typically not normally distributed and thus, non-parametric difference tests such as
193 the MWUT are preferred. The MWUT challenges the distribution of values, not the means. In this work, the level of
194 significance associated with the null hypothesis was operationally defined as 0.05, which equates to a 95 % chance
195 that an observed statistical difference is real and not coincidental. This error rate is operationally defined per contrast
196 (i.e. a 95 % chance that the observed statistical difference in nitrate concentrations between Teller Station 9 and Teller
197 Station 7 is real or that the observed statistical difference in sulphate concentrations between Teller and Kougarok is
198 real) as opposed to familywise (i.e. a 95 % chance that all of the observed/reported statistical differences are real and
199 not coincidental). MWUTs were completed using the methods described in Corder and Foreman (2009) and PCA was
200 completed using packages available in R statistical software, version 3.3.6 (Corder and Foreman, 2009; R Core Team,
201 2020). For all analyses, concentrations below the method detection limit were operationally defined as half the
202 detection limit, in agreeance with (Helsel, 2005, p. 43). While the emphasis of this study was on site/station (i.e.
203 spatial) variability, it should be recognized that seasonal and inter-annual variability could also be significant. To
204 minimize seasonal forcing on the variability observed, all SPW geochemical data presented were collected during the
205 thaw season between June and September.

206 2.4 Thermodynamic Modelling

207 To investigate thermodynamic controls on solute behaviour, particularly solubility limitations, thermodynamic
208 modelling exercises were undertaken using PHREEQC, a thermodynamic geochemical modelling code, and
209 PhreePlot, which facilitates repetitive PHREEQC calculations through looping (Kinniburgh and Cooper, 2011;
210 Parkhurst and Appelo, 2013). Because this study was focused on elucidating the primary geochemical controls on
211 solute concentrations in SPWs and not on developing a rigorous transport model, representative concentrations were
212 used instead of station specific concentrations. Representative “low”, “median”, and “high” concentration conditions
213 were proxied from the 25th, 50th, and 100th concentration percentiles, respectively, taken from both Teller and
214 Kougarok (Supplementary **Table 4**). Meanwhile, representative pH and E_H ranges were determined either through
215 direct measurement (pH), or indirectly by correlating dissolved Fe^{2+} concentrations and pH with a redox condition
216 through geochemical models and the Nernst equation. Modelling exercises were performed at 25 °C utilizing the
217 phreeqc.dat database, with the only modification being the suppression of methane production by inorganic carbonate



218 reduction. Methane production was “turned-off” to maintain carbonate availability under reducing conditions to help
219 identify any possible carbonate minerals that could be precipitating. Because alkalinity was only measured in a small
220 number of samples, carbonate concentration percentiles were estimated from charge imbalances. Alkalinity and charge
221 imbalance were very well correlated in samples where alkalinity was measured (Supplementary **Figure 3**). Although
222 not a particularly rigorous modelling exercise, this approach was sufficient to identify mineral phases that could be
223 controlling solute generation processes through solubility limitations.

224 **3. Results**

225 **3.1 Inter-site Variability: Teller versus Kougarak**

226 Mann-Whitney U-Testing revealed that the concentrations of 14 of the 23 constituents analysed were significantly
227 different ($1.96 < |z|$) between Teller and Kougarak (**Table 3**). The effect size, a measure of how significantly different
228 the concentrations were, were large for Na and F; medium-large for K and Si; medium for Al, Oxalate, B, Zn, SO₄,
229 Fe, Ba, Ti, and NO₂; and small-medium for Li. The terminology and thresholds for these semi-quantitative differences
230 in correlation were taken from Corder and Foreman (2009). When concentrations were significantly different between
231 the sites, Kougarak generally exhibited the higher concentrations of the two. Only SO₄ concentrations were
232 significantly greater at Teller. The concentrations of Br, NO₃, Sr, PO₄, Mg, Cr, Mn, Cl and Ca were not significantly
233 different between Teller and Kougarak. A summary of the inter-site MWUT results are given in **Table 3** with the
234 constituents that exhibited significant differences between the sites displayed over a darkened background.

235 **3.2 Intra-site Variability: Teller and Kougarak Stations**

236 Mann-Whitney U-Testing was also used to test for intra-site differences between stations at both Teller and Kougarak.
237 Boxplots and compact letter displays are used to visualize the within-site variability of a select group of constituents
238 of interest (COIs), which are given in **Figure 4**. Tables of the results of the intra-site MWUTs for all constituents that
239 were monitored, including those that did not demonstrate some systematic inter-station variability or were not
240 otherwise of interest, are given in the Supplementary Materials.

241 **3.3 Physical Characteristics of Stations (Co-Located Studies)**

242 Controls on the observed spatial variability of SWP solute concentrations at Teller and Kougarak stations were
243 deduced, in part, from differences in physical features and conditions of each station. Quantitative measures of many
244 of these physical characteristics were available from the interdisciplinary studies co-located at the Teller and Kougarak
245 stations. The extent of permafrost, ground temperature, active layer depth, soil moisture content, snow depth,
246 vegetation type, vegetation canopy height, dominant plant functional type, and hillslope position were all compiled
247 from these co-located studies. Using these measures, the physical characteristics of each station are summarized in
248 **Table 1** and **Table 2**, grouped by vegetation type.



249 4. Discussion

250 4.1 Inter-site Variability: Teller versus Kougarak

251 Mann-Whitney U-testing revealed that SPW concentrations of many constituents were significantly different between
252 Teller and Kougarak (**Table 3**). SPW concentrations of Na, F, K, Si, Al, oxalate, B, Zn, Fe, Ba, Ti, NO₂, and Li were
253 all significantly greater at Kougarak than Teller, while only SO₄ concentrations were significantly greater at Teller.
254 Meanwhile, the concentrations of Br, NO₃, Sr, PO₄, Mg, Cr, Mn, Cl, and Ca were not significantly different between
255 the two sites. Overall, the more frequent instance of significantly greater constituent concentrations at Kougarak
256 suggests a systematic cause. The extensive low-gradient toeslope (**Figure 2**) and lack of a well-defined drainage
257 channel at Kougarak, are likely causes of the systematically higher SPW solute concentrations at Kougarak. Water
258 perching, the result of near-surface permafrost in the lower-backslope and toeslope, increases evapotranspiration and,
259 thus, SPW solute concentrations. Meanwhile, the lack of a drainage channel at Kougarak suggests that runoff (and
260 therefore solute exports) is more limited than at Teller. Without a relatively rapid export mechanism such as a stream
261 channel, solute transport is likely limited to interflow within the Kougarak hillslope over much of the thaw season,
262 allowing weathering products to increase to significantly greater concentrations than those observed at Teller, where
263 a well-defined drainage/export mechanism does exist. Field observations from pits at Kougarak confirm the present
264 of interflow at the site. The exception to the general observation of elevated concentrations at Kougarak versus Teller
265 was SO₄. Although the cause of consistently higher SO₄ concentrations at Teller is unclear from the limited scope of
266 this study, it seems likely to be due to a greater abundance of sulfidic bedrock material. The presence of sulfidic
267 bedrock in the vicinity of Teller has been reported by mineral prospecting efforts (Brobst, Pinckney, and Sainsbury,
268 1971; Herreid, 1966; Mulligan, 1965); we are unaware of any such reports near Kougarak.

269 4.2 Intra-site Variability: Teller and Kougarak Stations

270 Our interpretation of the major environmental controls on the observed spatial variability of SPW solute concentrations
271 between stations are shown in **Table 4**. Each of these controls, including vegetation effects, soil moisture and redox
272 effects, weathering, water/soil interactions and hydrological transport effects, and mineral solubility effects, is
273 discussed in detail in the following sections.

274 4.3 Vegetation Effects

275 Vegetation can influence hydrogeochemical variability directly via vegetation-induced changes to elemental cycling
276 and soil moisture contents, or indirectly via the secondary impacts changes in soil moisture can have on mineral
277 solubility or on the soil redox condition. The geochemical consequences of solubility and redox conditions are the
278 focus of sections to follow, thus, this section will focus on direct vegetation effects via influences on elemental cycling
279 and soil moisture via evapotranspiration and preferential trapping of snow.

280 NO₃ was the only COI that showed a distinct effect from vegetation via elemental cycling. Elevated NO₃
281 concentrations were associated with the presence of alder shrubs and, in some cases, willow shrubs. NO₃
282 concentrations at both sites were generally low, with the exception of Kougarak Stations 3, 5, and 12, and Teller
283 Station 7 (**Figure 4**). Kougarak Stations 3, 5, and 12 all have a significant alder presence. Alders increase soil nitrogen



284 through a symbiotic relationship with nitrogen-fixing bacteria that reside in their root nodules, thus, an association
285 between NO_3 concentrations and alder vegetation is expected (Salmon et al., 2019).

286 Perhaps more noteworthy was the elevated NO_3 at Kougarak Station 5 and the lack of elevated NO_3 concentrations at
287 Kougarak Stations 1, 2, 6, 10, and 11. The vegetation type at Kougarak Stations 1, 2, 6, 10, and 11 is alder savanna
288 in tussock tundra, which is a mixed graminoid-shrub tundra with shorter stature and lower density of alder shrubs, yet
289 nonetheless nitrogen input via alder derived nitrogen-fixation is anticipated to occur. The lack of elevated NO_3
290 suggests either that 1) nitrogen-fixation in alder savanna in tussock tundra is insufficient to result in an increase in
291 NO_3 concentrations, 2) that the Kougarak footslope and lower backslope is very nitrogen-limited, and thus, that NO_3
292 is largely consumed by vegetation as it is fixed, or 3) that microbes in the Kougarak footslope and lower backslope
293 rapidly denitrify the available NO_3 as a substitute for oxygen in their metabolisms. The smaller shrub size and density
294 in the alder savanna in tussock tundra certainly results in less accumulated leaf litter relative to the denser and larger
295 alder shrubland intensive stations, as such, it seems reasonable that less nitrogen would be available at stations in alder
296 savanna in tussock tundra. Meanwhile, isotopic measurements of nitrogen downslope of alder patches at Kougarak
297 Stations 12 and 3 also support the occurrence of denitrification (McCaully et al., In Review). Therefore, we believe
298 the lack of elevated NO_3 concentrations at Kougarak Stations 1, 2, 6, 10, and 11 is a combination of less alder leaf
299 litter and greater denitrification, than at Kougarak Stations 3, 5, or 12.

300 At Teller, only Station 7 exhibited elevated NO_3 concentrations relative to the rest of the catchment. Teller Station 7
301 is dominated by tall willow shrubs and is relatively dry. Mineralization and nitrification of willow leaf litter coupled
302 with limited microbial denitrification is the presumed cause of elevated NO_3 concentrations at Teller Station 7. Teller
303 Station 2 also has tall willow shrubs but did not exhibit elevated NO_3 concentrations. From the limited scope of this
304 study, it is unclear why Teller Station 2 did not exhibit elevated NO_3 while Station 7 did, but we suspect that higher
305 seasonal moisture content and greater microbial denitrification at Teller Station 2 likely played a role. Also of note
306 was that despite significant intra-site NO_3 concentration differences, inter-site differences were not significant ($|z| =$
307 1.59) and that relatively few Kougarak stations showed elevated NO_3 concentrations, despite a widespread alder
308 presence. Increased microbial denitrification is suspected to balance increased nitrogen-fixation at these stations. This
309 is consistent with previous studies that have noted higher nitrogen mineralization rates in acidic tundra than non-acidic
310 tundra (Weiss et al., 2005); Kougarak is predominantly acidic tundra and Teller is non-acidic tundra.

311 The effect of vegetation on spatial variability of soil moisture was not readily observed in the volumetric moisture
312 content of soil (**Table 1** and **Table 2**) but was somewhat apparent in the spatial variability of moisture sensitive
313 constituents, such as Cl (**Figure 4**). The lack of a clear correlation between vegetation and soil moisture by TDR or
314 P-band SAR observations is perhaps do to the coarseness of the P-band SAR observations and the strong seasonality
315 associated with both methods. Moisture sensitive constituents, such as Cl, may provide a more seasonally averaged
316 tracer of soil moisture content at the stations. An increase in Cl concentrations with vegetation canopy height was
317 apparent at Teller stations suggesting an evapotranspiration effect. This trend was also apparent at Kougarak, but the
318 differences were rarely significant. Overall, the spatial variability of soil moisture sensitive constituents, like Cl, was
319 far less correlated with vegetation-type than expected; perhaps due to preferential trapping of snow, which may offset



320 the increased evapotranspiration of tall shrubs more than previously realized. Overall, Cl concentrations at Kougarok
321 appeared to be more correlated with hillslope position than with vegetation canopy height (**Figure 4**).

322 **4.4 Soil Moisture and Redox Effects**

323 Soil moisture content can have a profound effect on redox sensitive elements. Saturation limits oxygen diffusion into
324 soil, and thus, forces microorganisms to utilize less energetic electron acceptors to metabolize organic matter. In an
325 ideal system, soil microorganisms will use the strongest electron acceptor available, until it is exhausted. Although
326 natural environments are not ideal systems, redox conditions in soils follow a more or less stepwise progression.
327 Therefore, by evaluating the dissolved concentrations of NO₃, Mn, Fe, and SO₄ in SPWs, it is possible to qualitatively
328 assess soil redox conditions and their impact on hydrogeochemical variability.

329 The redox conditions at both Teller and Kougarok are generally limited by Fe reduction, with the most reducing
330 conditions found at stations with the highest soil moisture content. As such, NO₃ concentrations are generally low
331 (**Table 3**), SO₄ concentrations are relatively consistent (**Figure 4**), and Mn and Fe concentrations increase with
332 increasing soil moisture (**Figure 4**). NO₃ concentrations were generally low, except for drier stations in the proximity
333 of tall alders or willows. While NO₃ inputs are discussed in the vegetation effects section, the lack of high NO₃
334 concentrations at wetter stations that contain alders suggests that soil moisture coupled with microbial denitrification
335 bares a strong control on SPW NO₃ concentrations. Meanwhile, SO₄ concentrations at both sites are relatively constant
336 across clear moisture and redox gradients (**Figure 4**). This suggests that SO₄ reduction is not pervasive at either site.
337 Dissolved Fe concentrations were higher at stations with higher soil moisture content, consistent with Fe reduction.
338 Similarly, Mn concentrations were slightly elevated at wetter stations. The concentrations of Mn, however, rarely rose
339 above 0.05 mg·L⁻¹, suggesting either Mn solubility limitations or a lack of a significant Mn weathering source. Low
340 Mn concentrations at Teller Station 5, a wetter station on the upper shoulder of the Teller watershed (**Table 1; Figure**
341 **2**) seems to support the latter conclusion, as do geochemical modelling exercises (Section 4.5). Together, these results
342 suggest that the most reducing condition at both sites is typically limited to Fe reduction and that this only occurs at
343 stations with the highest soil moisture contents.

344 **4.4 Weathering, Water/Soil Interaction, and Hydrological Transport Effects**

345 A combination of weathering, water/soil interactions, and hydrological transport were clear drivers of
346 hydrogeochemical variability for some solutes. As noted by Philben et al. (2020), soil derived solutes tend to
347 accumulate in low-lying areas within watersheds. This is clearly seen at Teller, where the concentrations of Ca, Sr,
348 and Mg all increase dramatically at the transition from lower backslope to footslope (**Figure 5**). Both Teller and
349 Kougarok are underlain by carbonate-rich metamorphic facies, and Ca, Sr, and Mg are probable carbonate counter-
350 cations. Weathering of Ca, Sr, and Mg carbonates and subsequent transport of these cations downslope explains this
351 pattern of spatial variability. At Kougarok, concentrations of Ca, Mg, and Sr similarly increase from upper backslope
352 to footslope, but concentrations of Ca and Sr decrease further down the lower backslope (Stations 10 and 1), while
353 Mg concentrations continue to increase. A possible explanation for this behaviour is the greater affinity of cation



354 exchange surfaces for Ca and Sr compared to Mg, thus, Ca and Sr are preferentially retained in the footslope whilst
355 Mg is transported further down the lower backslope (Sparks, 2003, p. 189).

356 4.5 Mineral Solubility Effects

357 Although redox reactions are rarely at equilibrium in natural environments, comparison of field data with equilibrium
358 models provides valuable semi-quantitative insight into the redox condition of natural environments. Because Fe
359 appeared to be limiting the development of more reducing conditions (Section 4.3), select samples from both sites
360 were measured for soluble Fe^{2+} following methods presented in Viollier et al. (2000). These concentrations of aqueous
361 Fe^{2+} were then compared with model-predicted concentrations of Fe^{2+} , when coupled with an infinite $\text{Fe}(\text{OH})_{3(\text{am})}$
362 phase, across a range of pH values (2 – 10) and fixed E_{H} values of 400 mV, 200 mV, 0 mV, and -200 mV; activity
363 coefficients were assumed to be equal to 1. The measured and modelled Fe^{2+} concentrations are compared in **Figure**
364 **6**, where concentrations that were below the method detection limit (0.05 mg·L⁻¹) are set equal to 0.025 mg·L⁻¹ (half
365 the detection limit). Comparison of model predicted Fe^{2+} concentrations with field data suggests that while Teller
366 exhibits a narrower range of pH conditions than Kougarok, it exhibits a broader range of redox conditions (**Figure 6**).
367 Although several Fe^{2+} measurements were below the detection limit, suggesting oxidizing conditions, high Fe^{2+}
368 concentrations in some samples suggested E_{H} values below 0 mV. Therefore, Fe redox conditions at Teller ranged
369 from mildly reducing to oxic and Fe redox conditions at Kougarok ranged from mildly oxic to oxic. Oxidation-
370 reduction potentials (ORPs), calculated from pH, Fe^{2+} concentrations, and the Nernst equation suggest that ORPs at
371 Teller were as low as - 69 mV, while the lowest ORP at Kougarok was + 134 mV (**Figure 6**). Maximum ORP values
372 could not be determined quantitatively as some Fe^{2+} concentrations were below Fe^{2+} detection limits, at both sites.

373 E_{H} /pH predominance diagrams were created from the 25th, 50th, and 100th concentration percentiles and are shown
374 in **Figure 7** for the COIs where precipitation of mineral phases were predicted under some conditions. The
375 concentrations for these diagrams were taken from filtered aqueous concentration data, thus, predicted mineral
376 precipitation is an indication of nearly saturated or over-saturated conditions. The range of E_{H} and pH conditions
377 observed at Teller and Kougarok are overlaid as solid yellow and solid blue lines, respectively. Only the predominance
378 diagrams that indicated possible mineral formation under the E_{H} /pH conditions present at either site are shown in
379 **Figure 7**. These phases included $\text{Fe}(\text{OH})_{3(\text{am})}$ (Fe), siderite (Fe), $\text{Al}(\text{OH})_{3(\text{am})}$ (Al), chalcedony (Si), barite (Ba and
380 SO_4), calcite (Ca), dolomite (Ca and Mg), and rhodochrosite (Mn). Predominance diagrams for the remaining key
381 COIs that were not predicted to form any mineral phases under any site conditions are given in Supplementary **Figure**
382 **4**.

383 To further examine which mineral phases could be controlling solute generation processes, saturated conditions for
384 the mineral phases identified in **Figure 7** were modelled using sweeps of pH values from 2 – 10 at various fixed E_{H}
385 values (400mV, 200mV, 0mV, and -200mV). Predicted solute concentrations under the modelled saturated conditions
386 were then compared with field data to find common trends. In general, if solute concentrations were frequently
387 measured near the saturation of a mineral, or were identified to have similar dependence on pH or E_{H} , it was inferred
388 that the mineral phase could be controlling the generation of that solute. The mineral phases that were identified to
389 possibly be controlling solute concentrations were $\text{Al}(\text{OH})_{3(\text{am})}$, $\text{Fe}(\text{OH})_{3(\text{am})}$, chalcedony, and barite. This does not



390 preclude the presence of significant concentrations of other mineral phases, it only identifies these as likely controlling
391 the dissolved concentrations of Al, Fe, Si, and Ba, respectively.

392 Aluminium concentrations in SPWs at both Teller and Kougarok appear to be controlled by the
393 dissolution/precipitation of amorphous Al hydroxide ($\text{Al}(\text{OH})_{3(\text{am})}$) (**Figure 8**). The solubility limit of $\text{Al}(\text{OH})_{3(\text{am})}$ has
394 no redox dependence, but is highly pH dependent. Aluminium concentrations were generally clustered near the
395 solubility limit of $\text{Al}(\text{OH})_{3(\text{am})}$; $\text{Al}(\text{OH})_{3(\text{am})} + 3\text{H}^+ \leftrightarrow \text{Al}^{3+} + 3\text{H}_2\text{O}$; $\log k = 10.8$. This suggests that Al SPW
396 concentrations at both sites are controlled by wetting/drying (dissolution/precipitation) processes. It also suggests that
397 there could be a significant amount of $\text{Al}(\text{OH})_{3(\text{am})}$ in the soils at both sites. While organic matter may also sorb to
398 alumina surfaces, the adherence to the solubility of $\text{Al}(\text{OH})_{3(\text{am})}$ suggests that significant concentrations of Al are not
399 complexed with dissolved organic matter. The predominance diagrams highlight 1) the strong pH dependence on the
400 stability of $\text{Al}(\text{OH})_{3(\text{am})}$, 2) the influence of dissolved F can have on Al speciation when Al concentrations are low,
401 and 3) that Al is a cation at low pH and an anion at high pH (**Figure 7**). Despite being a weathering product, Al
402 concentrations show a dissimilar downslope trend to other weathering products, especially at Teller (Supplementary
403 **Figure 5**). While the concentrations of weathering products generally increase with distance downslope, Al
404 concentrations decrease. We suspect this can be attributed to increasing pH with distance downslope. Philben et al.
405 (2020) reported a 1 pH unit increase in pH in organic soils along the Teller transect (**Figure 2**), increasing from 5.6 at
406 Station 5 to 6.7 at Station 9. Such an increase would decrease the solubility of $\text{Al}(\text{OH})_{3(\text{am})}$, and thus, decrease the
407 concentration of dissolved Al (**Figure 8**).

408 Similar to Al, Fe concentrations in SPWs at both Teller and Kougarok appear to be controlled by the
409 dissolution/precipitation of amorphous Fe hydroxide ($\text{Fe}(\text{OH})_{3(\text{am})}$). Fe concentrations were generally clustered near
410 the solubility limit of $\text{Fe}(\text{OH})_{3(\text{am})}$ (**Figure 8**). Unlike $\text{Al}(\text{OH})_{3(\text{am})}$ however, $\text{Fe}(\text{OH})_{3(\text{am})}$ solubility is dependent on the
411 redox condition in addition to the pH; $\text{Fe}(\text{OH})_{3(\text{am})} + 3\text{H}^+ + \text{e}^- \leftrightarrow \text{Fe}^{2+} + 3\text{H}_2\text{O}$; $\log k = 16.0$ (**Figure 8**). Fe(III) is only
412 sparingly soluble in aqueous solutions and reduction to Fe(II) significantly increases the solubility of Fe, thus, at a
413 given pH value higher aqueous concentrations are predicted and observed under more reducing conditions (**Figure 8**).
414 Iron concentrations in SPWs at both sites generally follow the pH dependence of $\text{Fe}(\text{OH})_{3(\text{am})}$ solubility (**Figure 8**).
415 This suggests that SPW concentrations of Fe at both sites are controlled by wetting/drying (dissolution/precipitation)
416 processes, coupled with the redox condition.

417 Si concentrations are frequently limited by the solubility of chalcedony, a very finely grained form of SiO_2 , which is
418 much more soluble than quartz; $\text{SiO}_2 + 2\text{H}_2\text{O} \leftrightarrow \text{H}_4\text{SiO}_4$; $\log k = -3.55$. Particularly at Kougarok, the dissolved Si
419 concentrations, coupled with a lack of a strong pH or E_{H} dependence, suggest a controlling influence of chalcedony.

420 Ba concentrations also appear to be controlled by solubility, but rather than by the solubility of an oxide or a hydroxide
421 phase, by the solubility of barite [$\text{Ba}^{2+} + \text{SO}_4^{2-} \leftrightarrow \text{BaSO}_4(\text{s})$; $\log k = 9.97$]. Unlike Al hydroxide or Fe hydroxide, barite
422 solubility lacks a strong pH dependence and instead is dependent solely on the activities of Ba^{2+} and SO_4^{2-} . Unlike Ba,
423 SO_4 concentrations are not limited by the solubility limit of barite and are generally higher and not well correlated
424 with Ba concentrations. Together, these suggest that SO_4 from another source (likely, atmospheric deposition or
425 sulfidic mineral oxidation), is suppressing barite dissolution, and thus, is reducing dissolved Ba concentrations. Barite
426 solubility can exhibit a redox dependence if conditions are sufficiently reducing to reduce SO_4 to sulphide. This shifts



427 the equilibrium to greater dissolution of barite, and therefore higher conditions of Ba. The lack of E_H dependence in
428 observational data further suggests that neither site exhibits significant SO_4 reduction.

429 5. Conclusions

430 5.1 The Dominant Environmental Controls on Spatial Variability of SPW Solute Concentrations

431 The 18 stations examined herein (8 at Teller and 10 at Kougarak) were selected to represent a wide range of vegetation
432 types, soil moisture contents, permafrost extents, and hillslope positions. Coupling the spatial variability of these
433 landscape characteristics with the spatial variability of SPW solute concentrations provides valuable insight into the
434 dominant environmental controls on observed spatial variability of SPW geochemistry.

435 With regard to our initial hypotheses, our major findings are that:

436 Vegetation influences on elemental cycles were only readily apparent for nitrogen and although vegetation induced
437 changes to soil moisture content were discernible, they were far less significant than anticipated. NO_3 was the only
438 COI that exhibited a clear vegetation effect; elevated concentrations were associated with the presence of alder shrubs
439 and, in some cases, tall willow shrubs. These increases in NO_3 concentrations associated with alder nitrogen-fixation
440 and the mineralization and nitrification of willow leaf litter were frequently equivoiced by increased microbial
441 denitrification in regions sufficiently moist to support it. Although both Kougarak and Teller exhibited some
442 indications of increased Cl concentrations in the presence of tall shrubs, the net vegetation effect on soil moisture was
443 far less than hypothesized. Redox sensitivity was also less than hypothesized and most stations seemed well-buffered
444 at Fe redox conditions. The result of this buffering was generally low NO_3 concentrations (except where vegetation
445 effects dominated), consistent SO_4 concentrations across clear redox gradients, and variable Mn and Fe concentrations.
446 Mn concentrations were generally low, likely due to a limited source. Fe concentrations were higher at stations with
447 higher soil moisture content, consistent with Fe reduction. Weathering, water/soil interactions, and hydrological
448 transport were clear drivers of variability for Ca, Sr, and Mg. Ca, Sr, and Mg all tended to accumulate in low-lying
449 areas, although Ca and Sr demonstrated greater accumulation potential than Mg, likely via greater affinity of cation
450 exchange surfaces for Ca and Sr compared to Mg. Mineral solubility limitations were the primary controls on Al
451 ($Al(OH)_{3(am)}$), Fe ($Fe(OH)_{3(am)}$), Ba (barite), and Si (chalcedony) concentrations. This suggests that the SPW
452 concentrations of these constituents will remain stable until those mineral phases are exhausted or soil pore
453 hydrochemistry changes sufficiently to alter the solubility of those mineral phases. Changes in redox condition would
454 significantly alter $Fe(OH)_{3(am)}$ solubility, whereas changes in pH conditions would significantly alter $Al(OH)_{3(am)}$ and
455 $Fe(OH)_{3(am)}$ solubility.

456 Although discerning the environmental controls on spatial variability of SPW solute concentrations provides some
457 high-level insight into the effects changes in landscape character may have on soil pore hydrochemistry, our scope
458 was limited and leveraged on previously available datasets. The significance of SPW in small Arctic headwater
459 catchments as a key initial component in the freshwater hydrologic continuum is under recognized, and such
460 catchments warrant more detailed and systematic investigations.



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468 **7. Data availability statement**

469 The data that support the findings of this study are made openly available in the NGEE-Arctic data repository at (DOI:
470 10.5440/1735757).

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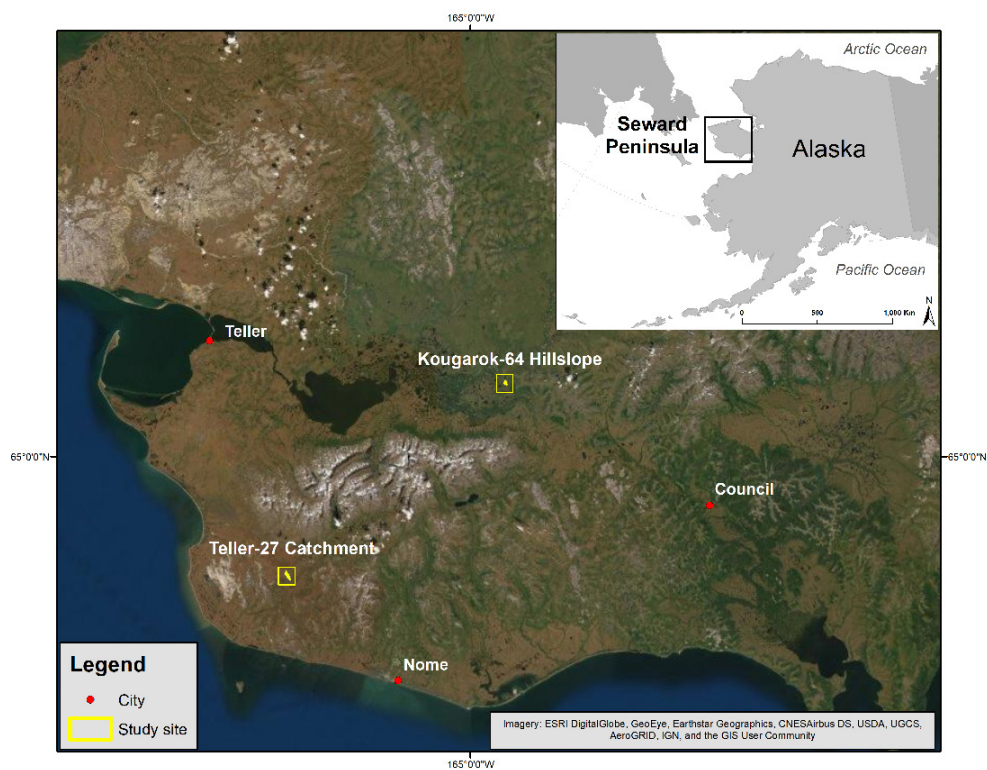
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633 9. Figures



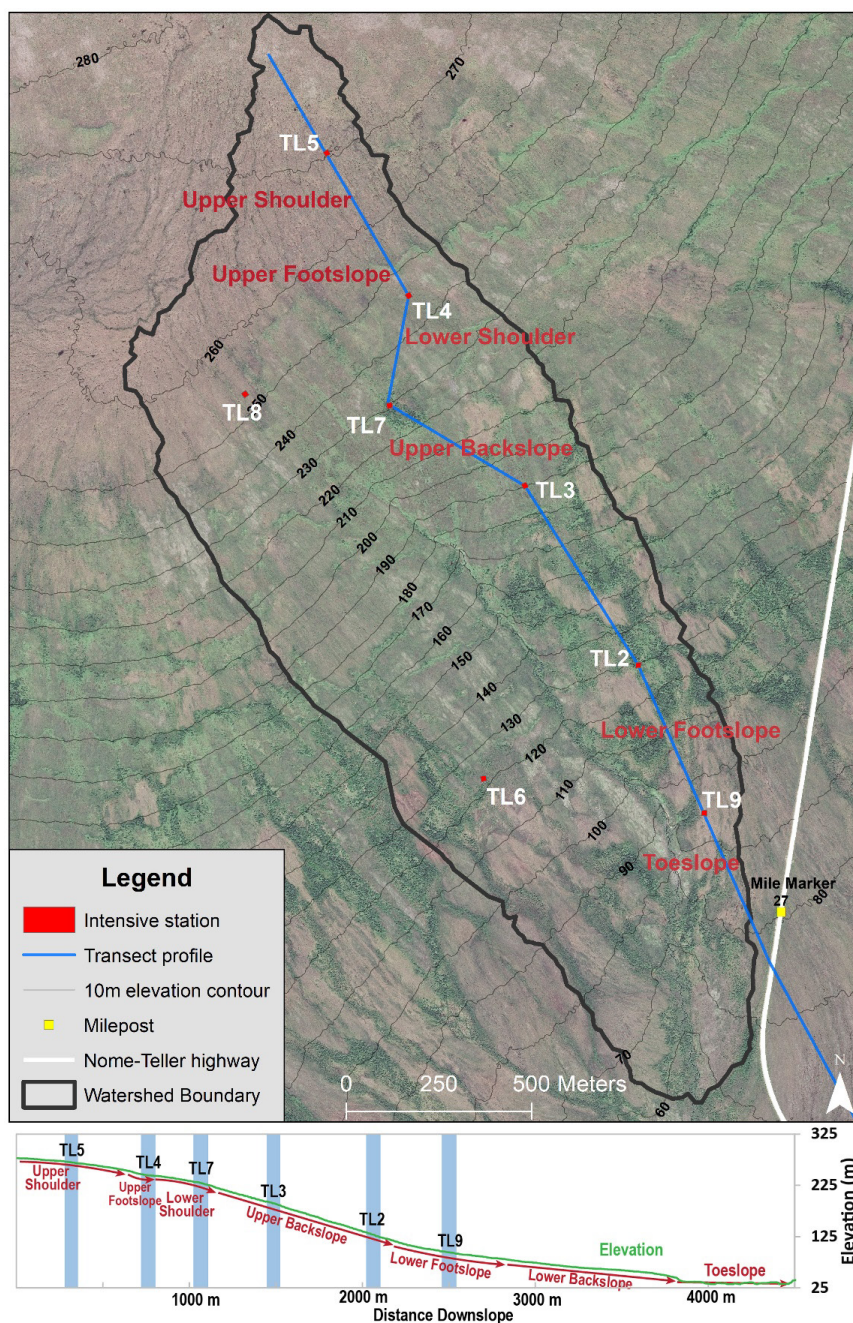
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Figure 1. Location of the Teller and Kougarak field sites with respect to the municipalities of Teller, Nome, and Council. All are located on the Seward Peninsula in northwestern Alaska.

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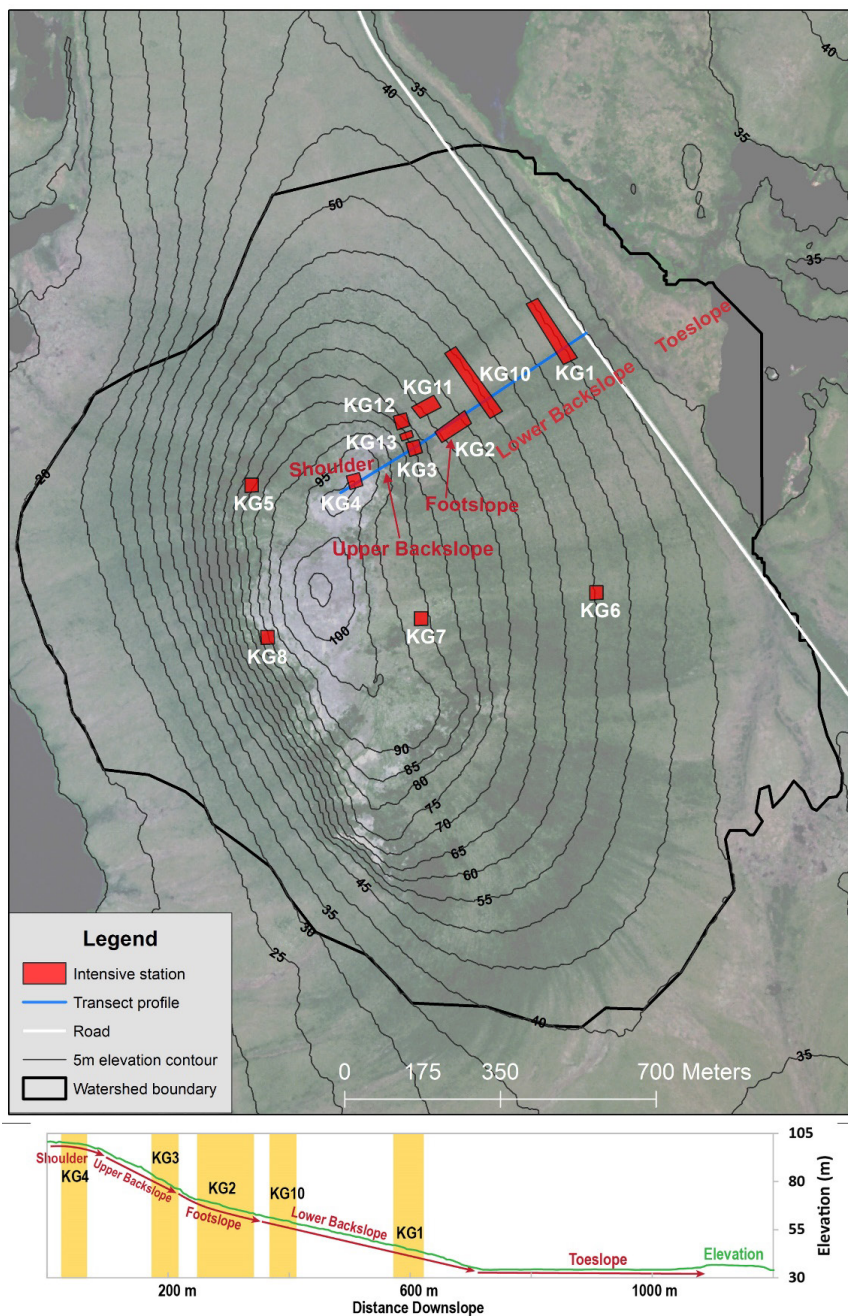
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Figure 2. Topographic map of Teller. Station areas are shown as red polygons and the topographic station transect is given as a solid blue line. The hillslope transect elevation profile is given below the map in green, with stations along the transect in blue and hillslope positions noted with red arrows and text. RGB composite imagery from the 8-band WorldView-2 imagery obtained on July 27, 2011 at 1.5 m resolution downloaded from the DigitalGlobe website (<https://www.digitalglobe.com/>).



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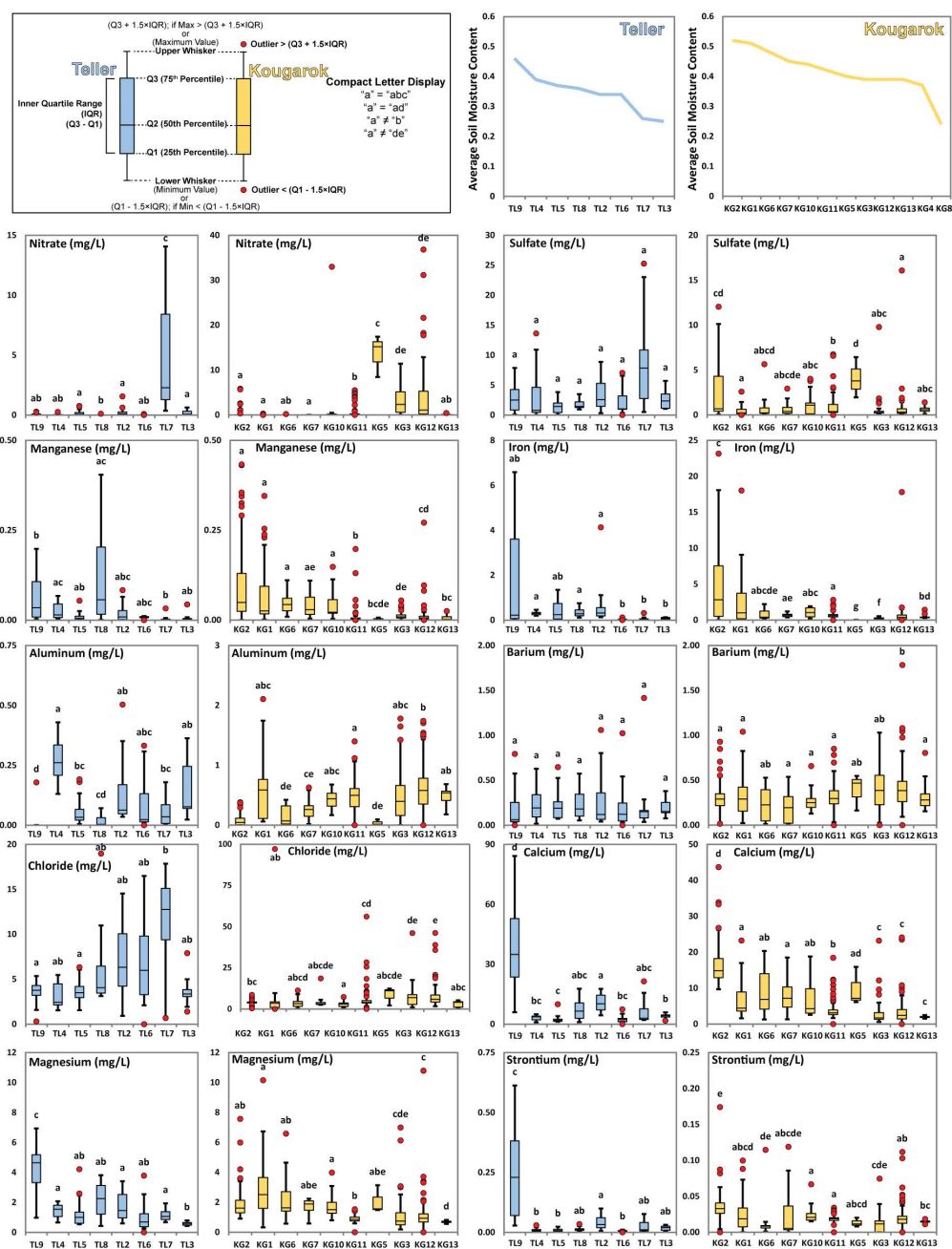
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Figure 3. Topographic map of Kougarak. Station areas are shown as red polygons and the station transect is given as a solid blue line. The transect elevation profile is given below the map in green, with stations along the transect in yellow and hillslope positions noted with red arrows and text. RGB composite imagery from the 8-band WorldView-2 imagery obtained on July 14, 2017 at 1.5 m resolution downloaded from the DigitalGlobe website (<https://www.digitalglobe.com/>).



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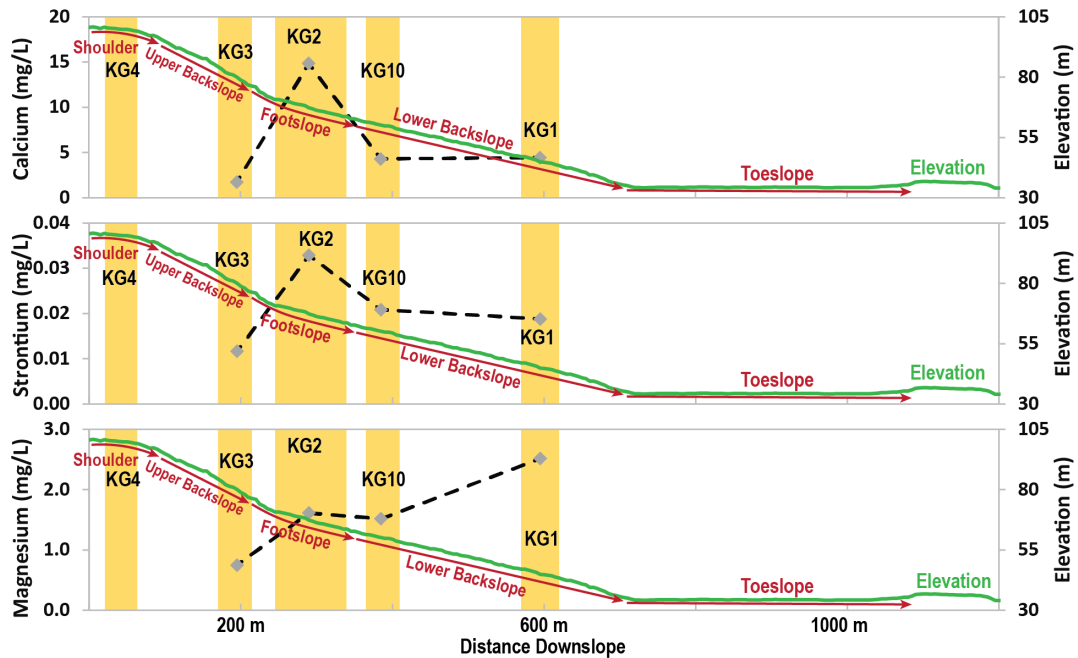
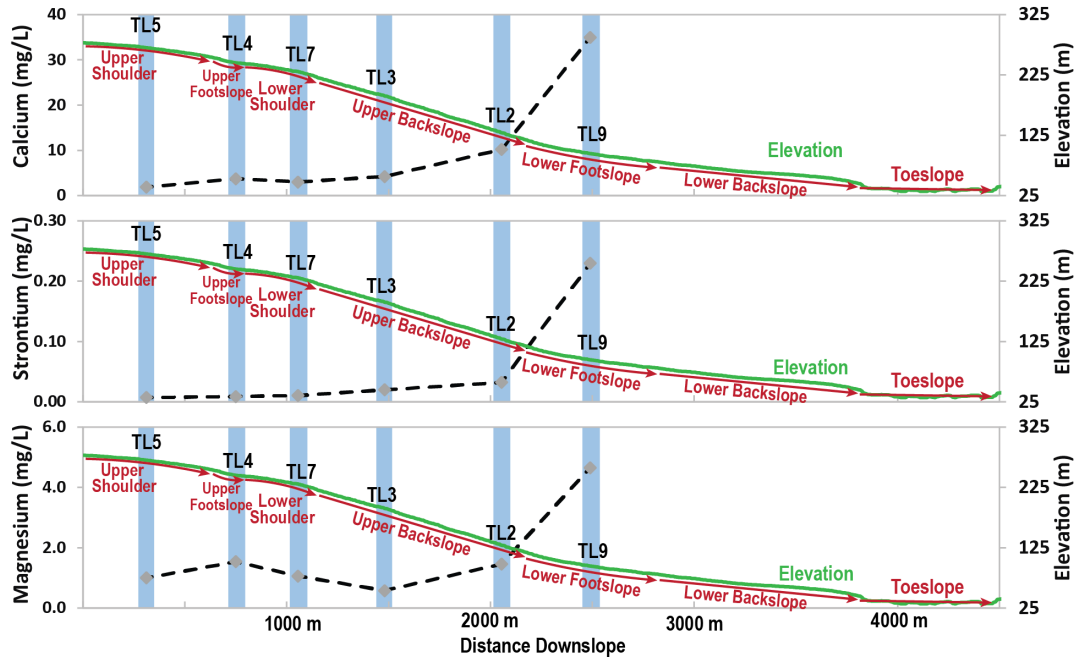


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652 **Figure 4.** Mean COI concentrations at Teller (blue) and Kougarak (yellow) stations. Stations are arranged (left to right) by
 653 soil moisture content determined by P-Band SAR (top right). Boxplots show the first, second, and third data quartiles, with
 654 box whiskers representing either 150% of the inner quartile range (IQR), or the maximum or minimum value, when that
 655 value was less than 1.5×IQR. Red circles represent data points outside of the 1.5×IQR whiskers (i.e. outliers). Note that the
 656 concentration scales on the Teller and Kougarak plots often differ.



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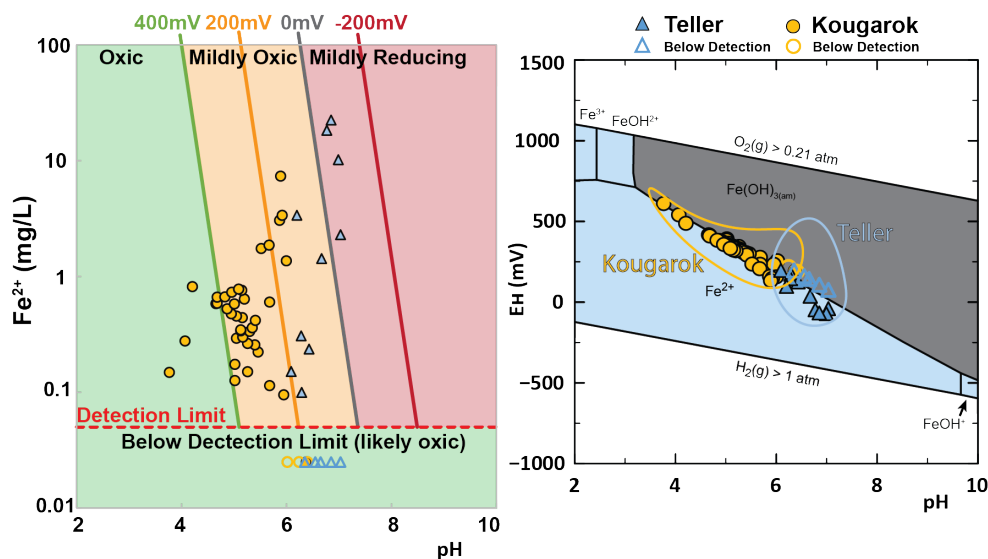


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659 **Figure 5.** Median (50th percentile) concentrations (grey diamonds with dashed black lines) of Ca, Sr, and Mg, with distance
 660 downslope at Teller (blue) and Kougarak (yellow) along topographic transects; areas of stations are indicated by blue and
 661 yellow colouring, respectively. The elevation profiles of the hillslopes are plotted in green, on separate y-axes (right axes).
 662 Topographic regions of both catchments are indicated by red arrows along the elevation gradient.



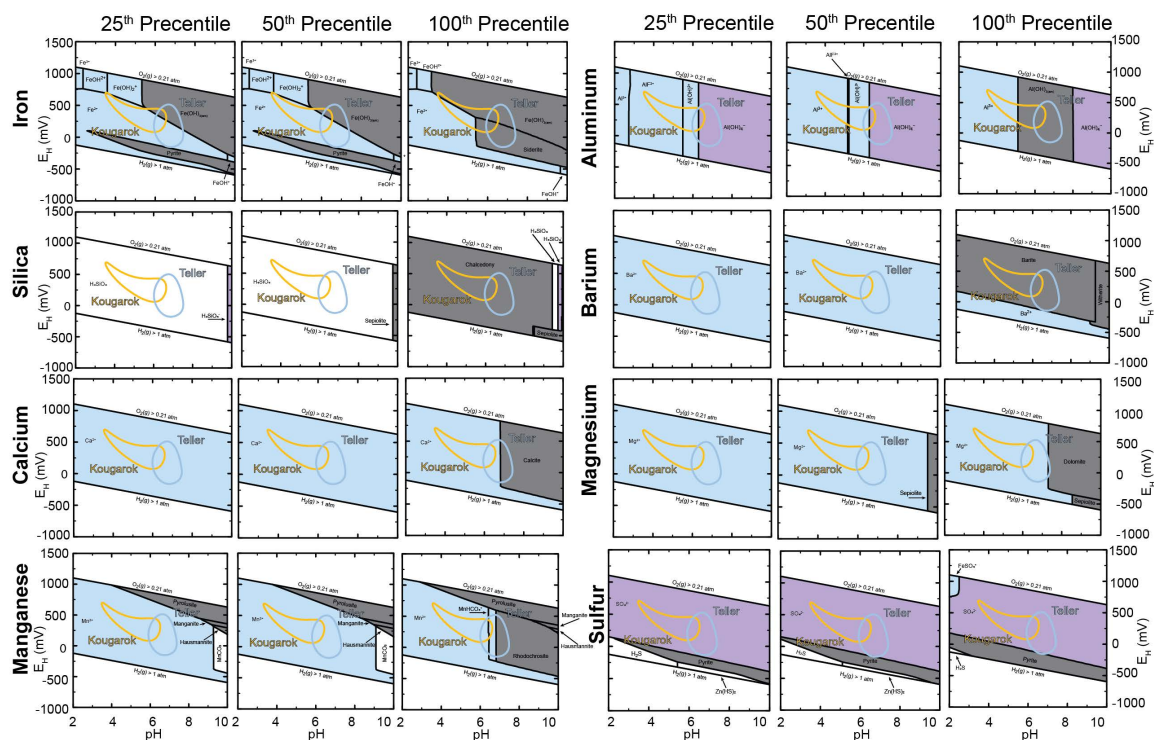
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665 **Figure 6. Left: Model-predicted Fe²⁺ concentrations in saturated solutions of Fe(OH)_{3(am)} at fixed E_H conditions of 400 mV**
 666 **(green), 200 mV (orange), 0 mV (grey), and -200 mV (red), compared with field concentrations of Fe²⁺ at Teller (red circles)**
 667 **and Kougarak (yellow circles). Right: Fe predominance diagram, showing the dominant specie of Fe under a range of**
 668 **E_H/pH conditions. E_H/pH regions relevant to Teller and Kougarak are outlined in blue and yellow, respectively. Samples**
 669 **with Fe²⁺ concentrations below the detection limit are given as colour coordinated open circles set at 0.025 mg·L⁻¹ (half the**
 670 **detection limit) in both sides of the figure.**

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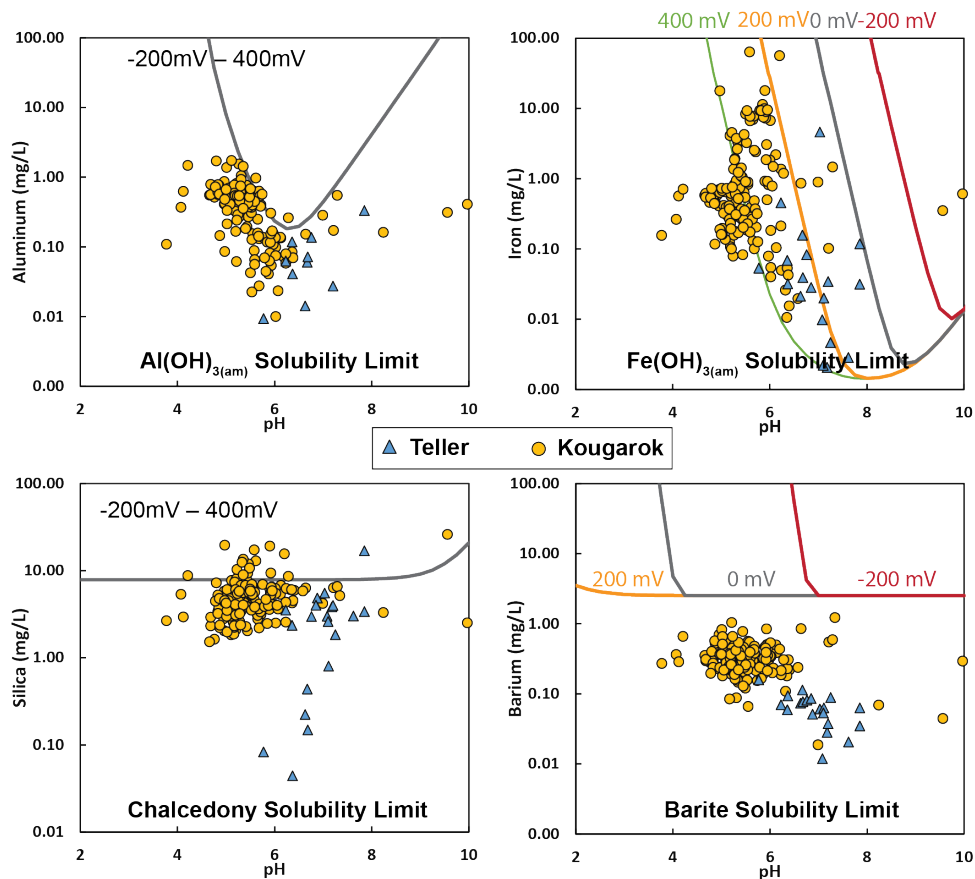
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Figure 7. E_H /pH diagrams for key species that indicated possible mineral formation under the E_H /pH conditions present at either Teller or Kougarok. The E_H and pH conditions observed at Teller and Kougarok are overlaid as blue and yellow lines, respectively. Mineral species (solids) are shown in grey, cations are shown in blue, anions are shown in purple, neutral species are shown in white. Predominance diagrams were created in PhreePlot using the phreeqc.dat database, with inorganic carbonate reduction to methane “turned off.”



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Figure 8. Modelled solute concentrations in solutions saturated with $\text{Al(OH)}_{3(\text{am})}$, $\text{Fe(OH)}_{3(\text{am})}$, chalcedony, and barite, with respect to pH (x-axis) and E_{H} (model lines), overlaid with observed solute concentrations.

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683 **10. Tables**

684 **Table 1. Teller Station Physical Characteristics**

	<i>Hillslope Position</i>	<i>Vegetation</i>				<i>Relative wetness</i>			<i>Permafrost</i>		
		Vegetation type	Average (maximum canopy height (cm))	Dominant PFT	Low to tall shrub cover	Average TDR soil moisture (VMC)	Average P-band SAR (VMC)	Average snow depth (cm)	Average Ground Temperature (°C)	Permafrost Extent	Average (maximum) thaw depth (cm)
<i>TL9</i>	Lower Footslope	Wetland complex	34 (41)	Graminoid	10	NA	0.46	68.4	0	Marginal	101 (>120)
<i>TL5</i>	Upper Shoulder		31 (45)		7	0.55	0.37	103.3	-0.45	Near-surface	97 ^r (>114 ^r)
<i>TL8</i>	Upper Footslope		17 (34)		20	0.55	0.36	77.7	-0.6		69 ^r (>120)
<i>TL3</i>	Upper Backslope	Cassiope dwarf shrub tundra	13 (23)	Evergreen dwarf shrub	4	NA	0.25	62.1	2.2	None/ deep	72 ^r (82 ^r)
<i>TL4</i>	Upper Footslope		10 (14)		12	0.35	0.39	89.5	0.5	Marginal	40 ^r (70 ^r)
<i>TL2</i>	Upper Backslope	Mesic willow shrubland	132 (141)	Deciduous low to tall shrub (willow)	44	0.4	0.34	124	2.4	None/ deep	75 ^r (>120)
<i>TL7</i>	Lower Shoulder		168 (189)		37	0.46	0.26	128.8	2.4		51 ^r (66 ^r)
<i>TL6</i>	Upper Backslope	Willow-birch tundra	74 (115)	Deciduous low shrub (willow & birch)	32	0.38	0.34	86.4	1.2	None/ deep	67 ^r (102 ^r)

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PFT – plant functional type; dwarf shrub (height <40 cm), low shrub (height 40–200 cm), low to tall shrub (height 40 to >200 cm tall).

¹Single point soil moisture measurements. Data are more accurate than P-band SAR but represent a much smaller spatial scale.

²P-band SAR has 30m resolution.

^rResistive layer was rock; all others are permafrost. A temperature probe was used to determine if the resistive layer was permafrost (≤ 0 °C) or rock (> 2 °C). Thaw depth is an average of 4 measurements from the vegetation plot corners within the IS and was measured at the end of the growing season.



693 **Table 2. Kougarok Station Physical Characteristics**

	<i>Hillslope Position</i>	<i>Vegetation</i>				<i>Relative wetness</i>			<i>Permafrost</i>		
		Vegetation type	Average (maximum) canopy height (cm)	Dominant PFT	Low to tall shrub cover	Average TDR soil moisture (VMC)	Average P-band SAR (VMC)	Average snow depth (cm)	Average Ground Temperature (°C)	Permafrost Extent	Average (maximum) thaw depth (cm)
<i>KG3</i>	Upper Backslope	Alder shrubland	204 (265)	Deciduous low to tall shrub (alder)	30	0.19	0.39	131.3	-0.01	Near-surface	48 ^r (53r)
<i>KG12</i>	Footslope		NA		NA	0.30*	0.39	NA	NA		NA
<i>KG1</i>	Lower Backslope	Alder savanna in tussock tundra	60 (90)	Deciduous low shrub (alder, willow & birch)	13	NA	0.51	83.4	-2.5	Near-surface	61 (68)
<i>KG2</i>	Footslope		48 (73)		20	0.63	0.52	102.3	-1.2		75 (89)
<i>KG6</i>	Lower Backslope		24 (61)		0	0.36	0.48	66.2	-2.2		58 (62)
<i>KG10</i>	Lower Backslope		NA		NA	NA*	0.44	71.4	NA		NA
<i>KG11</i>	Footslope		NA		NA	0.59*	0.42	NA	NA		NA
<i>KG7</i>	Upper Backslope	Tussock-lichen tundra	20 (22)	Graminoid	0	0.51	0.45	54.7	-2.1	Near-surface	76 (100)
<i>KG4</i>	Shoulder	Dryas-lichen shrub tundra	6 (12)	Evergreen dwarf shrub	0	NA	0.37	NA	-1.9	Near-surface	0 ^r (0 ^r)
<i>KG13</i>	Upper Backslope		NA		NA	0.41*	0.39	92.1	NA		NA
<i>KG5</i>	Upper Backslope	Willow-birch tundra	62 (137)	Deciduous low shrub (willow & birch)	3	NA	0.4	178.4	> 0	Deep	88 (96)
<i>KG8</i>	Upper Backslope		45 (120)		42	0.23	0.24	85.5	-0.04	Near-surface	44 ^r (55 ^r)

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Note: PFT – plant functional type.

¹Single point soil moisture measurements. Data are more accurate than P-band SAR but represent a much smaller spatial scale.

²P-band SAR has 30m resolution.

*Average gravimetric water content measurements, corrected to VMC by bulk density.

^rResistive layer was rock; all others are permafrost. A temperature probe was used to determine if the resistive layer was permafrost (≤ 0 °C) or rock (> 2 °C). Thaw depth is an average of 4 measurements from the vegetation plot corners within the IS and was measured at the end of the growing season.



Table 3. Inter-Site Mann-Whitney U-Test Results

	<u>Teller</u>			<u>Kougarok</u>			z	Site with Higher Median	Effect Size	Difference in Correlation
	n	$\sum R_i$	U_i	n	$\sum R_i$	U_i				
Na	59	3184	14811.5	275	52761.5	1413.5	9.95	Kougarok	0.54	large
F	59	3502	14375.5	273	51776.5	1731.5	9.46	Kougarok	0.52	large
K	59	3882	14113	275	52063	2112	8.92	Kougarok	0.49	medium-large
Si	59	4119	13876.5	275	51826.5	2348.5	8.56	Kougarok	0.47	medium-large
Al	58	4952	12709	275	50659	3241	7.11	Kougarok	0.39	medium
Oxalate	57	4996	12161.5	272	49289.5	3342.5	6.75	Kougarok	0.37	medium
B	59	5429	12566.5	275	50516.5	3658.5	6.62	Kougarok	0.36	medium
Zn	58	5605	12056	275	50006	3894	6.12	Kougarok	0.34	medium
SO₄	58	13653	3892.5	273	41293.5	11941.5	6.08	Teller	0.33	medium
Fe	58	5958	11703	275	49653	4247	5.60	Kougarok	0.31	medium
Ba	58	6256	11405.5	275	49355.5	4544.5	5.15	Kougarok	0.28	medium
Ti	58	6266	11395.5	275	49345.5	4554.5	5.13	Kougarok	0.28	medium
NO₂	54	5588	10585.5	272	47713.5	4102.5	5.12	Kougarok	0.28	medium
Li	58	7778	9883	275	47833	6067	2.86	Kougarok	0.16	small-medium
Br	58	8485	9060.5	273	46461.5	6773.5	1.73	Equal	0.09	small
NO ₃	58	8576	8969	273	46370	6865	1.59	Kougarok	0.09	small
Sr	58	8683	8978	275	46928	6972	1.51	Kougarok	0.08	small
PO ₄	54	9659	6460.5	271	43316.5	8173.5	1.36	Equal	0.08	small
Mg	58	10495	7166	275	45116	8784	1.21	Teller	0.07	small
Cr	58	8884	8777	275	46727	7173	1.20	Kougarok	0.07	small
Mn	58	9164	8497	275	46447	7453	0.78	Teller	0.04	small
Cl	58	9221	8266.5	272	45394.5	7509.5	0.57	Kougarok	0.03	small
Ca	58	10016	7645	275	45595	8305	0.50	Teller	0.03	small

705 **Table 4. Dominant Environmental Controls on SPW Geochemistry at Teller and Kougarok**

Environmental Control	Analytes Affected
Vegetation	NO ₃
Soil Moisture/Redox	NO ₃ , Mn, Fe, SO ₄ (occasionally)
Water/Soil Interactions & Hydrologic Transport	Ca, Mg, Sr
Mineral Solubility	Al, Ba, Si, Fe