

Environmental Controls on Observed Spatial Variability of Soil Pore Water Geochemistry in Small Headwater Catchments Underlain with Permafrost

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Abstract. Soil pore water (SPW) chemistry can vary substantially across multiple scales in Arctic permafrost landscapes. The magnitude of these variations and their relationship to scale are critical considerations for understanding current controls on geochemical cycling and for predicting future changes. These aspects are especially important for Arctic change modelling where accurate representation of sub-grid variability may be necessary to predict watershed scale behaviours. Our research goal was to characterize intra- and inter-watershed soil water geochemical variations at two contrasting locations in the Seward Peninsula of Alaska, USA. We then attempt to establish which environmental factors were important for controlling concentrations of important pore water solutes in these systems. The SPW geochemistry of 18 locations spanning two small Arctic catchments were examined for spatial variability and its dominant environmental controls. The primary environmental controls considered were vegetation, soil moisture/redox condition, water/soil interactions and hydrologic transport, and mineral solubility. The sampling locations varied in terms of vegetation type and canopy height, presence or absence of near-surface permafrost, soil moisture, and hillslope position. Vegetation was found to have a significant impact on SPW NO_3^- concentrations, associated with the localized presence of nitrogen-fixing alders and mineralization and nitrification of leaf litter from tall willow shrubs. The elevated NO_3^- concentrations were however, frequently equivoiced by increased microbial denitrification in regions with sufficient moisture to support it. Vegetation also had an observable impact on soil moisture sensitive constituents, but the effect was less significant. The redox conditions in both catchments were generally limited by Fe reduction, seemingly well-buffered by a cache of amorphous Fe hydroxides, with the most reducing conditions found at sampling locations with the highest soil moisture content. Non-redox-sensitive cations were affected by a wide variety of water-soil interactions that affect mineral solubility and transport. Identification of the dominant controls on current SPW hydrogeochemistry allows for qualitative prediction of future geochemical trends in small Arctic catchments that are likely to experience warming and permafrost thaw. As source

39 areas for geochemical fluxes to the broader Arctic hydrologic system, geochemical processes occurring in these
40 environments are particularly important to understand and predict with regards to such environmental changes.

41 **1. Introduction**

42 Permafrost thaw in the Arctic is causing significant changes to landscape structure (Kokelj and Jorgenson, 2013;
43 Rowland et al., 2010), hydrology (Hiyama et al., 2021; Kurylyk et al., 2021; Liljedahl et al., 2016; Vonk, Tank, and
44 Walvoord, 2019; Walvoord and Kurylyk, 2016), vegetation (Lara et al., 2018; Myers-Smith et al., 2011; Sturm,
45 Racine, and Tape, 2001; K. D. Tape et al., 2012; Tape, Sturm, and Racine, 2006;), and biogeochemistry (O'Donnell
46 et al., 2021; Frey and McClelland, 2009; Salmon et al., 2019; Vonk, Tank, and Walvoord, 2019). The integrated
47 hydrogeochemical effects of these environmental changes are already apparent in the chemistry of the large Arctic
48 rivers, where fluxes of carbon and nutrients are increasing, leading to enhanced nutrient loadings, with strong
49 implications for the global carbon cycle (Bring et al., 2016; Fuchs et al., 2020; McClelland et al., 2016). While the
50 watershed areas of large Arctic rivers are vast, recent studies suggest that solute concentrations in these large rivers
51 are likely controlled by solute generation processes occurring at much smaller scales (Harms and Ludwig, 2016; Koch
52 et al., 2013; Shogren et al., 2019; Vonk et al., 2015).

53 While there is a rapidly growing body of literature focused on observing and understanding environmental changes
54 over time with further Arctic warming, relatively few studies directly address the existing spatial variability, within
55 catchments or across catchments, and we are not aware of any studies that have combined field observations with
56 thermodynamic modelling in an effort to understand the causes of the existing spatial variability. Therefore, we have
57 a limited understanding of the key environmental controls on the spatial distribution of soil pore water solute
58 concentrations. In this study, we quantitatively evaluate the spatial variability of soil pore water (SPW) geochemistry
59 within and between two distinct catchments underlain with permafrost, and then seek to identify the source of the
60 observed spatial variability.

61 This study takes advantage of a scientifically diverse array of observations and datasets made available by the Next
62 Generation Ecosystem Experiment (NGEE) Arctic project, sponsored by the US Department of Energy Office of
63 Science. Most of the locations studied herein were selected by the NGEE Arctic project to provide co-located
64 measurements in a wide range of vegetation types, nested within representative hillslopes and catchments. Although
65 selected largely to represent a range of vegetation structure, such as shrub abundance and canopy height, these
66 locations also have considerable variability in other environmental parameters including, but not limited to: soil
67 moisture and temperature, presence or absence of near-surface permafrost, and maximum observed thaw depth (**Table**
68 **1** and **Table 2**). The vegetation-delineated sampling approach provides an opportunity to not only quantify the
69 biogeochemical variability of SPW in Arctic environments, but also to investigate the root causes of that observed
70 variability. Data from additional sampling locations, available from a co-located study, were also utilized when
71 possible.

72 Our overarching hypothesis is that vegetation-type and hillslope position are the dominant controls on spatial
73 variability of SPW geochemistry at the NGEE Arctic field sites located on the Seward Peninsula. Vegetation-type
74 seems likely to have a significant effect on SPW geochemistry both directly and indirectly. Indirect effects would

75 include vegetation canopy impacts on soil moisture (through evapotranspiration and snow trapping). Direct effects of
76 vegetation would include nutrient cycle changes resulting from the annual deposition of plant litter. Such a direct
77 effect can be augmented at sites populated by alder shrubs due to this genus of deciduous shrubs ability to form a
78 symbiotic relationship with nitrogen-fixing *Frankia*, which they host in underground root nodules. Nitrogen fixation
79 associated with alders has previously been shown to accelerate local nitrogen cycling (Binkley et al., 1992; Clein and
80 Schimel, 1995; Bühlmann et al., 2014). Soil moisture will also affect SPW geochemistry, particularly of redox
81 sensitive species, by limiting oxygen diffusion and thus controlling which regions develop anoxic/reducing
82 geochemical conditions. Soil moisture impacts will likely be correlated with vegetation-type as well as hillslope
83 position, and the presence or absence of perching layers, including permafrost, all of which impact the vertical and
84 horizontal drainage characteristics of a watershed. Chemical species that are not redox-sensitive or controlled by
85 biogeochemical reactions are likely to be affected by transport, solubility, and water/sediment/organic matter
86 interactions, and therefore largely controlled by hillslope position as well as soil moisture.

87 Identifying the dominant controls on solute concentration variability within each catchment and across catchments
88 will facilitate better projections of future soil pore hydrogeochemistry in permafrost landscapes, and how these
89 signatures are related to changing soil moisture and increasing tundra shrub abundance in a changing Arctic (Bring et
90 al., 2016; Myers-Smith et al., 2011; Prowse et al., 2015; Salmon et al., 2019; Sturm et al., 2001; Tape et al., 2012;
91 Tape et al., 2006; Wrona et al., 2016, 2016). Arctic warming and associated permafrost thaw will increase hydrological
92 connectedness between terrestrial and aquatic environments through deepening of the active layer and the formation
93 of deeper, more coherent groundwater flow paths (Bring et al., 2016; Harms and Jones, 2012; Prowse et al., 2015a;
94 Prowse et al., 2015b). Meanwhile, changes in hydrogeochemical signatures in larger Arctic rivers are likely to
95 originate in smaller catchments (McClelland et al., 2016; Prowse et al., 2015; Shogren et al., 2019; Spence et al.,
96 2015). In this sense, changes in hydrogeochemistry in small Arctic catchments not only impact hydrogeochemistry at
97 much larger scales, but also prognosticate the future hydrogeochemistry of larger Arctic rivers.

98 **2. Methods**

99 **2.1 Site Descriptions**

100 This study focuses on two sites with permafrost on the Seward Peninsula of western Alaska, the Teller-27 Catchment
101 and the Kougarok-64 Hillslope (**Figure 1**). The Teller-27 Catchment, henceforth “Teller,” is a small (~2.25 km²)
102 headwater catchment located west of mile marker 27 along the Nome-Teller Highway northwest of Nome, Alaska.
103 The Kougarok-64 Hillslope, henceforth “Kougarok,” is a hillslope (~2.0 km²) located west of mile marker 64 along
104 the Nome-Taylor Highway northeast of Nome, Alaska. We utilized data from “intensive stations” at both Teller and
105 Kougarok where concentrated, multi-year, co-located observations of soil water chemistry, vegetation characteristics,
106 soil moisture and temperature, and other measurements have been collected as part of the NGEE Arctic Research
107 Project. These are identified as TL# (Teller Station #) or KG# (Kougarok Station #) in **Figure 2** and **Figure 3**,
108 respectively. It should be noted that Teller and Kougarok are not “paired watersheds” in the classical sense, differing
109 in only one major characteristic, which provides the basis for comparison. Instead, Teller and Kougarok differ in many

110 respects and are both representative of the broad range of hillslope conditions common on the Seward Peninsula.
111 Detailed descriptions of Teller and Kougarok have been published previously (Jafarov et al., 2018; Léger et al., 2019;
112 Philben et al., 2019, 2020; Salmon et al., 2019; Yang et al., 2020), therefore, only the catchment characteristics that
113 are probable sources of variability in SPW chemistry will be highlighted here.

114 Teller is a discrete catchment with a well-defined central drainage, a vertical declivity of approximately 200 m, and a
115 catchment area of approximately 2.25 km². Temperature probes, soil pits, coring activities, and geophysical
116 interpretations at Teller have confirmed the catchment is underlain with discontinuous permafrost (Léger et al., 2019).
117 The upper shoulder of Teller (near Station 5, **Figure 2** and **Figure 5**) is underlain with near-surface permafrost and
118 appears to be a degraded peat plateau. The resultant microtopography of the degraded peat and the shallow perching
119 horizon caused by the permafrost creates a landscape of unsaturated peat mounds surrounded by ponds and saturated
120 soils. Downslope of the peat plateau, the Teller hillslope has highly variable soil moisture and vegetation (**Table 1**).
121 The microtopography within the lower footslope looks similar to the upper shoulder, but the peat appears more
122 severely degraded and the cause of the perched water table is less clear. Léger et al. (2019) suggest the presence of
123 permafrost at a depth of 1 – 2 m at Teller Station 9 (**Figure 2**), but the perching could also be caused by a layer of silt,
124 at a depth of approximately 30 cm (Graham et al., 2018). The full extent of permafrost and silt in this region of the
125 catchment remains unknown, but the thaw depth in July 2018 was greater than 1 m and maintained a perched water
126 table (Philben et al., 2020), suggesting perching could be the result of silt rather than permafrost. Vegetation type,
127 moisture content, permafrost extent, and hillslope position for all Teller Stations are summarized in **Table 1**.

128 Kougarok differs in many ways from Teller, although both have characteristics that are typical of hillslopes on the
129 Seward Peninsula. Kougarok is a convex hillslope, with a vertical declivity of approximately 70 meters. The study
130 area at Kougarok is approximately 2.0 km². Soil temperature measurements at Kougarok suggest that the vast majority
131 of the site is underlain by shallow continuous permafrost (Romanovsky, Cable, and Dolgikh, 2020a); Kougarok
132 Station 5 is an exception, where the permafrost is deeper (Romanovsky et al., 2020a). The upper shoulder of Kougarok
133 is a well-drained rocky outcrop composed of metagranitic rock (Hopkins et al., 1955; Till, Dumoulin, Werdon, and
134 Bleick, 2011). Saturated soils are not prevalent until the footslope and the lower backslope, where Kougarok Stations
135 2, 11, 10, 1, and 6 are situated (**Figure 3**). The lower backslope is characterized by persistent saturation between
136 ubiquitous tussocks, formed by the tussock cotton grass *Eriophorum vaginatum*. The tussock-lichen tundra at
137 Kougarok introduces microtopography and spatially variable saturation; in this sense, the Kougarok tussocks are
138 analogous to the peat mounds and hummocks at Teller, but on different spatial scales and formed by different
139 processes. Kougarok has numerous patches of alder shrubland in an altitudinal band within the upper backslope; it
140 should be emphasized that Teller lacks tussock-lichen tundra and alder (*Alnus viridis* ssp. *fruticosa*) shrubs that are a
141 feature of Kougarok. While continuous permafrost largely remains, the Kougarok site appears to be undergoing
142 environmental changes as evidenced by an increase in alder coverage over the past decades (Salmon et al., 2019). Soil
143 profiles underneath the alder patches are rocky with shallow bedrock and warmer permafrost (**Table 2**). Shrub tundra
144 (alder savanna in tussock tundra and willow-birch tundra) dominates the lower backslope, where the annual active
145 layer thickness is typically less than 100 cm. Vegetation type, moisture content, permafrost extent, and hillslope
146 position at all Kougarok stations are summarized in **Table 2**.

147 2.2 Sampling & Analytical Approach

148 SPWs were sampled using two complimentary techniques. Fiberglass wicks (Frisbee et al., 2010) were deployed in
149 the upper 30 cm of soils at stations where shallow soils were unsaturated. These wicks were left in place from year-
150 to-year and only replaced if damage was observed or suspected. The sample reservoirs from the wicks were collected
151 whenever possible, usually a few times each summer. MacroRhizons (Rhizosphere Research Products; Netherlands)
152 were used at stations that were more saturated, also targeting the upper 30 cm of soils. Both techniques were used at
153 stations of intermediate saturation, where both could be deployed effectively. MacroRhizons represent a relatively
154 discrete temporal sampling event (minutes to hours), whereas wicks represent a cumulative water collected over longer
155 periods (weeks to months). It is in this sense, that the two techniques are complimentary. Unfortunately, due to
156 saturation variability both techniques could not be used at all stations and conditions at some Kougarok stations were
157 sometimes too dry to collect meaningful volumes of SPW using either method. Additional SPW data from Kougarok
158 were supplemented from a separate study focused on alder-related nutrient dynamics (McCaully et al., 2022). These
159 data were collected by MacroRhizons and are captured as Kougarok Stations 10 – 13, which were not part of the
160 original stations established by the NGEE Arctic Program. A total of 309 SPW samples from Kougarok were collected
161 and analysed, whereas a total of 89 SPW samples from Teller were collected and analysed.

162 After collection, SPW cation concentrations were measured in triplicate by inductively coupled plasma optical
163 emission spectroscopy (Optima 2100 DV; PerkinElmer, USA) following US EPA Method 200.7. Inorganic anion
164 concentrations were measured by ion chromatography (DX-600; Dionex, USA) following US EPA Method 300.0. B,
165 F, K, Na, and Si concentrations collected by wicks were excluded from the dataset due to known issues with these
166 ions leeching from fiberglass wick samplers (Perdrial et al., 2014; Wallenberger and Bingham, 2009). This effect is
167 illustrated in **Supplementary Figure 1** and the lack of such an effect for divalent cations is shown in **Supplementary**
168 **Figure 2**. Comparison of data from wicks and MacroRhizons, along with the observations from (Perdrial et al., 2014),
169 demonstrates that remaining constituents discussed herein were not affected by collection with fiberglass wicks.
170 Alkalinity, pH, and E_H are all critical geochemical parameters that are susceptible to change during storage (Petrone
171 et al., 2007); because of the large amount of data from wicks these parameters were not considered further, except in
172 the context of thermodynamic modelling.

173 Observations related to vegetation, soil moisture, and permafrost extend were compiled from datasets made available
174 by the NGEE Arctic project and are given for Teller in **Table 1** and for Kougarok in **Table 2**. The reported soil
175 moisture contents were derived from an average of gravimetric measurements (2017 and 2018) and time domain
176 reflectometry measurements (2017 and 2019), and from remotely-sensed P-band Synthetic Aperture Radar (2017).
177 End-of-winter snow depths were measured in March and April of 2016, 2017, and 2018. The annual average ground
178 temperature was measured using in-situ temperature sensors (HOBO U30 DataLogger) at a depth of 1.5 meters below
179 the ground surface (Romanovsky et al., 2020a; Romanovsky, Cable, and Dolgikh, 2020b) and the active layer
180 thicknesses were determined by frost probe in September 2019 at the end of the growing season. Vegetation data were
181 collected at the peak of the growing season in mid to late July 2016 and 2017 at the NGEE Arctic Kougarok and Teller
182 field sites, respectively. The distribution of plant communities in the Arctic is primarily controlled by landscape,
183 topography, soil chemistry, soil moisture, and the plants that historically colonized an area (Raynolds et al., 2019).

184 Soil available rooting depth, which can be limited by shallow depths to bedrock, permafrost, or the water table, can
185 also restrict plant growth and survival of certain species by reducing access to water and nutrients. We surveyed the
186 dominant plant communities along each hillslope, which varied in their shrub abundance, canopy height, and structure,
187 to characterize the vegetation composition at the sites following the recommended protocol of Walker et al. (2016).
188 Extensive field site details and vegetation sampling methods are more thoroughly described in previous studies
189 (Salmon et al., 2019; Langford et al., 2019; Yang et al., 2020; Sulman et al., 2021; Yang et al. 2021).
190 For this study, we provide summary statistics for vegetation plots associated with intensive stations. Vegetation
191 composition plots within each intensive station were chosen subjectively in areas of homogeneous and representative
192 vegetation varying in size from 1 to 25 m² depending on canopy structure and height. The surveyed plot area was 1 ×
193 1 m for all plant communities except for the taller stature willow-birch tundra, mesic willow shrubland (2.5 × 2.5 m),
194 and alder shrubland (5 × 5 m). For each plot, all plant species (vascular plants, lichens, and bryophytes) were recorded
195 along with visual estimates of their percent cover. For plots with multiple canopies, field cover estimates were recorded
196 as absolute cover, meaning that the total cover per plot can be >100%. We calculated relative cover values (adding to
197 100%) from the field data and use these for all subsequent analyses.
198 Plant species were further aggregated into nine plant functional types (PFTs), groupings of plant species that share
199 similar growth forms and roles in ecosystem function (Wullschleger et al., 2014), based on growth patterns and plant
200 traits. PFTs in this study include: (1) nonvascular mosses and lichens, (2) deciduous and evergreen shrubs of various
201 height classes, including an alder PFT, (3) graminoids, and (4) forbs. Photos of representative PFTs from both sites
202 are given in **Supplementary Figures 9-17**. Canopy height was estimated within each plot for each PFT as the average
203 of 4 measurements, including a maximum canopy height. Active layer depth was measured at the end of the growing
204 season for all plots in September 2018 using a frost probe. A temperature probe was used to determine if the resistive
205 layer was permafrost (≤ 0 °C) or rock (> 2 °C). Thaw depth is an average of 4 measurements from the vegetation plot
206 corners.

207 **2.3 Statistical Analysis**

208 Principal Components Analysis (PCA) and the Mann-Whitney U-Test (MWUT) were both used to investigate
209 dominant environmental controls on solute concentrations in SPWs at Teller and Kougarak. PCA is an exploratory
210 data analysis tool that reduces the dimensionality of large complex datasets and considers how components (i.e. solute
211 concentrations) vary together. Because PCA was predominately used as a screening tool to reveal geochemical
212 correlations that may not have been evident by traditional geochemical causations or inference, a detailed discussion
213 of the PCA results is reserved to the Supplementary Materials. The MWUT was used to test for significant differences
214 in solute concentrations between Teller and Kougarak (inter-site variability) and between stations at each site (intra-
215 site variability). The MWUT is a non-parametric method of challenging a null hypothesis, which in this case is the
216 assumption that the concentrations of a given solute are not systematically greater at either site nor at any particular
217 station. Water chemistry data are typically not normally distributed and thus, non-parametric difference tests such as
218 the MWUT are preferred. The MWUT challenges the distribution of values, not the means. In this work, the level of
219 significance associated with the null hypothesis was operationally defined as 0.05, which equates to a 95 % chance

220 that an observed statistical difference is real and not coincidental. This error rate is operationally defined per contrast
221 (i.e. a 95 % chance that the observed statistical difference in nitrate concentrations between Teller Station 9 and Teller
222 Station 7 is real or that the observed statistical difference in sulphate concentrations between Teller and Kougarok is
223 real) as opposed to familywise (i.e. a 95 % chance that all of the observed/reported statistical differences are real and
224 not coincidental). MWUTs were completed using the methods described in Corder and Foreman (2009) and PCA was
225 completed using packages available in R statistical software, version 3.3.6 (Corder and Foreman, 2009; R Core Team,
226 2020). For all analyses, concentrations below the method detection limit were operationally defined as half the
227 detection limit, in agreeance with (Helsel, 2005, p. 43). While the emphasis of this study was on site/station (i.e.
228 spatial) variability, it should be recognized that seasonal and inter-annual variability could also be significant. To
229 minimize seasonal forcing on the variability observed, all SPW geochemical data presented were collected during the
230 thaw season between June and September.

231 **2.4 Thermodynamic Modelling**

232 To investigate thermodynamic controls on solute behaviour, particularly solubility limitations, thermodynamic
233 modelling exercises were undertaken using PHREEQC, a thermodynamic geochemical modelling code, and
234 PhreePlot, which facilitates repetitive PHREEQC calculations through looping (Kinniburgh and Cooper, 2011;
235 Parkhurst and Appelo, 2013). Because this study was focused on elucidating the primary geochemical controls on
236 solute concentrations in SPWs and not on developing a rigorous transport model, representative concentrations were
237 used instead of station specific concentrations. Representative “low”, “median”, and “high” concentration conditions
238 were proxied from the 25th, 50th, and 100th concentration percentiles, respectively, taken from both Teller and
239 Kougarok (Supplementary **Table 4**). Meanwhile, representative pH and E_H ranges were determined either through
240 direct measurement (pH), or indirectly by correlating dissolved Fe^{2+} concentrations and pH with a redox condition
241 through geochemical models and the Nernst equation. Modelling exercises were performed utilizing the phreeqc.dat
242 database, with the only modification being the suppression of methane production by inorganic carbonate reduction.
243 Modelling exercises were performed at the default PHREEQC modelling temperature (25 °C), as the selection of an
244 alternative defensible temperature was non-trivial; temperatures on the Seward Peninsula span a very wide range and
245 its unclear what temperature would be most suitable for mineral solubility limitation modelling. Ultimately, because
246 the thermodynamic models were used as a tool understand what could be controlling soil pore water solute
247 concentrations and were not intended to model the system or to predict future concentrations, the default temperature
248 was decided to be the most suitable. While there is some temperature dependence of mineral solubility, the differences
249 in predicted solubility between 4 °C and 25 °C did not impact the interpretation of our results (**Supplementary Figure**
250 **8**). Methane production was “turned-off” to maintain carbonate availability under reducing conditions to help identify
251 any possible carbonate minerals that could be precipitating. Because alkalinity was only measured in a small number
252 of samples, carbonate concentration percentiles were estimated from charge imbalances. Alkalinity and charge
253 imbalance were very well correlated in samples where alkalinity was measured (**Supplementary Figure 3**). Although
254 not a particularly rigorous modelling exercise, this approach was sufficient to identify mineral phases that could be
255 controlling solute generation processes through solubility limitations.

256 **3. Results**

257 **3.1 Physical Characteristics of Stations (Co-Located Studies)**

258 Controls on the observed spatial variability of SWP solute concentrations at Teller and Kougarok stations were
259 deduced, in part, from differences in physical features and conditions of each station. Quantitative measures of many
260 of these physical characteristics were available from the interdisciplinary studies co-located at the Teller and Kougarok
261 stations. The extent of permafrost, ground temperature, active layer depth, soil moisture content, snow depth,
262 vegetation type, vegetation canopy height, dominant plant functional type, and hillslope position were all compiled
263 from these co-located studies. Using these measures, the physical characteristics of each station are summarized in
264 **Table 1** and **Table 2**, grouped by vegetation type.

265 **3.2 Inter-site Variability: Teller versus Kougarok**

266 Mann-Whitney U-Testing revealed that the concentrations of 14 of the 23 constituents analysed were significantly
267 different ($1.96 < |z|$) between Teller and Kougarok (**Table 3**). The effect size, a measure of how significantly different
268 the concentrations were, were large for Na and F; medium-large for K and Si; medium for Al, Oxalate, B, Zn, SO_4^{2-} ,
269 Fe, Ba, Ti, and NO_2^- ; and small-medium for Li. The terminology and thresholds for these semi-quantitative differences
270 in correlation were taken from Corder and Foreman (2009). Mann-Whitney U-testing revealed that SPW
271 concentrations of many constituents were significantly different between Teller and Kougarok (**Table 3**). When
272 concentrations were significantly different between the sites, Kougarok generally exhibited the higher concentrations
273 of the two. SPW concentrations of Na, F, K, Si, Al, oxalate, B, Zn, Fe, Ba, Ti, NO_2 , and Li were all significantly
274 greater at Kougarok than Teller, while only SO_4^{2-} concentrations were significantly greater at Teller. Meanwhile, the
275 concentrations of Br, NO_3^- , Sr, PO_4 , Mg, Cr, Mn, Cl, and Ca were not significantly different between Teller and
276 Kougarok. A summary of the inter-site MWUT results are given in **Table 3** with the constituents that exhibited
277 significant differences between the sites displayed over a darkened background.

278 **3.3 Intra-site Variability: Teller and Kougarok Stations**

279 Mann-Whitney U-Testing was also used to test for intra-site differences between stations at both Teller and Kougarok.
280 Boxplots and compact letter displays are used to visualize the within-site variability of a select group of constituents
281 of interest (COIs), which are given in **Figure 4**. Tables of the results of the intra-site MWUTs for all constituents that
282 were monitored, including those that did not demonstrate some systematic inter-station variability or were not
283 otherwise of interest, are given in the Supplementary Materials. Our interpretation of the major environmental controls
284 on the observed spatial variability of SPW solute concentrations between stations are shown in **Table 4**. Each of these
285 controls, including vegetation effects, soil moisture and redox effects, weathering, water/soil interactions and
286 hydrological transport effects, and mineral solubility effects, is considered in detail in the following sections.

287 **3.3.1 Vegetation Effects**

288 Vegetation can influence hydrogeochemical variability directly via vegetation-induced changes to elemental cycling
289 and soil moisture contents, or indirectly via the secondary impacts changes in soil moisture can have on mineral

290 solubility or on the soil redox condition. The geochemical consequences of solubility and redox conditions are the
291 focus of sections to follow, thus, this section will focus on direct vegetation effects via influences on elemental cycling
292 and soil moisture via evapotranspiration and preferential trapping of snow.

293 NO_3^- was the only COI that showed a distinct effect from vegetation via elemental cycling. Elevated NO_3^-
294 concentrations were associated with the presence of alder shrubs and, in some cases, willow shrubs. NO_3^-
295 concentrations at both sites were generally low, with the exception of Kougarok Stations 3, 5, and 12, and Teller
296 Station 7 (**Figure 4**). Low to tall alder shrubs are the dominant vegetation type at Kougarok Stations 3 and 12 (**Table**
297 **2**). Meanwhile, alders are present at Kougarok Station 5 despite the dominant vegetation type being low willow and
298 birch shrubs. Alders increase soil nitrogen through a symbiotic relationship with nitrogen-fixing bacteria that reside
299 in their root nodules, thus, an association between NO_3^- concentrations and alder vegetation is expected (Salmon et
300 al., 2019).

301 Perhaps more noteworthy was the lack of elevated NO_3^- concentrations at Kougarok Stations 1, 2, 6, 10, and 11. The
302 vegetation type at Kougarok Stations 1, 2, 6, 10, and 11 is alder savanna in tussock tundra, which is a mixed graminoid-
303 shrub tundra with shorter stature and lower density of alder shrubs, yet nonetheless nitrogen input via alder derived
304 nitrogen-fixation is anticipated to occur. The lack of elevated NO_3^- suggests either that 1) nitrogen-fixation in alder
305 savanna in tussock tundra is insufficient to result in an increase in NO_3^- concentrations, 2) that the Kougarok footslope
306 and lower backslope is very nitrogen-limited, and thus, that NO_3^- is largely consumed by vegetation as it is fixed, or
307 3) that microbes in the Kougarok footslope and lower backslope rapidly denitrify the available NO_3^- as a substitute
308 for oxygen in their metabolisms. The smaller shrub size and density in the alder savanna in tussock tundra certainly
309 results in less accumulated leaf litter relative to the denser and larger alder shrubland intensive stations, as such, it
310 seems reasonable that less nitrogen would be available at stations in alder savanna in tussock tundra. Meanwhile,
311 isotopic measurements of nitrogen downslope of alder patches at Kougarok Stations 12 and 3 also support the
312 occurrence of denitrification (McCaully et al., 2022). Therefore, we believe the lack of elevated NO_3^- concentrations
313 at Kougarok Stations 1, 2, 6, 10, and 11 is a combination of less alder leaf litter and greater denitrification, than at
314 Kougarok Stations 3, 5, or 12.

315 At Teller, only Station 7 exhibited elevated NO_3^- concentrations relative to the rest of the catchment (**Figure 4**). Teller
316 Station 7 is dominated by tall willow shrubs and is relatively dry. Mineralization and nitrification of willow leaf litter
317 coupled with limited microbial denitrification is the presumed cause of elevated NO_3^- concentrations at Teller Station
318 7. Teller Station 2 also has tall willow shrubs but did not exhibit elevated NO_3^- concentrations. From the limited scope
319 of this study, it is unclear why Teller Station 2 did not exhibit elevated NO_3^- while Station 7 did, but we suspect that
320 higher seasonal moisture content and greater microbial denitrification at Teller Station 2 likely played a role. Also of
321 note was that despite significant intra-site NO_3^- concentration differences, inter-site differences were not significant
322 ($|z| = 1.59$) and that relatively few Kougarok stations showed elevated NO_3^- concentrations, despite a widespread alder
323 presence. Increased microbial denitrification is suspected to balance increased nitrogen-fixation at these stations. This
324 is consistent with previous studies that have noted higher nitrogen mineralization rates in acidic tundra than non-acidic
325 tundra (Weiss et al., 2005); Kougarok is predominantly acidic tundra and Teller is non-acidic tundra.

326 The effect of vegetation on spatial variability of soil moisture was not readily observed in the volumetric moisture
327 content of soil (**Table 1** and **Table 2**) but was somewhat apparent in the spatial variability of moisture sensitive
328 constituents, such as Cl (**Figure 4**). The lack of a clear correlation between vegetation and soil moisture by TDR or
329 P-band SAR observations is perhaps do to the coarseness of the P-band SAR observations and the strong seasonality
330 associated with both methods. Moisture sensitive constituents, such as Cl, may provide a more seasonally averaged
331 tracer of soil moisture content at the stations. An increase in Cl concentrations with vegetation canopy height was
332 apparent at Teller stations suggesting an evapotranspiration effect. This trend was also apparent at Kougarok, but the
333 differences were rarely significant. Overall, the spatial variability of soil moisture sensitive constituents, like Cl, was
334 far less correlated with vegetation-type than expected; perhaps due to preferential trapping of snow, which may offset
335 the increased evapotranspiration of tall shrubs more than previously realized. Overall, Cl concentrations at Kougarok
336 appeared to be more correlated with hillslope position than with vegetation canopy height (**Figure 4**).

337 **3.3.2 Soil Moisture and Redox Effects**

338 Soil moisture content can have a profound effect on redox sensitive elements. Saturation limits oxygen diffusion into
339 soil, and thus, forces microorganisms to utilize less energetic electron acceptors to metabolize organic matter. In an
340 ideal system, soil microorganisms will use the strongest electron acceptor available, until it is exhausted. Although
341 natural environments are not ideal systems, redox conditions in soils follow a more or less stepwise progression.
342 Therefore, by evaluating the dissolved concentrations of NO_3^- , Mn, Fe, and SO_4^{2-} in SPWs, it is possible to
343 qualitatively assess soil redox conditions and their impact on hydrogeochemical variability.

344 The redox conditions at both Teller and Kougarok are generally limited by Fe reduction, with the most reducing
345 conditions found at stations with the highest soil moisture content. As such, NO_3^- concentrations are generally low
346 (**Table 3**), SO_4^{2-} concentrations are relatively consistent (**Figure 4**), and Mn and Fe concentrations increase with
347 increasing soil moisture (**Figure 4**). NO_3^- concentrations were generally low, except for drier stations in the proximity
348 of tall alders or willows. While NO_3^- inputs are discussed in the vegetation effects section, the lack of high NO_3^-
349 concentrations at wetter stations that contain alders suggests that soil moisture coupled with microbial denitrification
350 bares a strong control on SPW NO_3^- concentrations. Meanwhile, SO_4^{2-} concentrations at both sites are relatively
351 constant across clear moisture and redox gradients (**Figure 4**). This suggests that SO_4^{2-} reduction is not pervasive at
352 either site. Dissolved Fe concentrations were higher at stations with higher soil moisture content, consistent with Fe
353 reduction. Similarly, Mn concentrations were slightly elevated at wetter stations. The concentrations of Mn, however,
354 rarely rose above $0.05 \text{ mg}\cdot\text{L}^{-1}$, suggesting either Mn solubility limitations or a lack of a significant Mn weathering
355 source. Low Mn concentrations at Teller Station 5, a wetter station on the upper shoulder of the Teller watershed
356 (**Table 1; Figure 2**) seems to support the latter conclusion, as do geochemical modelling exercises (Section 4.5).
357 Together, these results suggest that the most reducing condition at both sites is typically limited to Fe reduction and
358 that this only occurs at stations with the highest soil moisture contents.

3.3.3 Weathering, Water/Soil Interaction, and Hydrological Transport Effects

A combination of weathering, water/soil interactions, and hydrological transport were identified as probable drivers of hydrogeochemical variability for some solutes. As noted by Philben et al. (2020), soil derived solutes tend to accumulate in low-lying areas within watersheds. This is observed at Teller, where the concentrations of Ca, Sr, and Mg all increase dramatically at the transition from lower backslope to footslope (**Figure 5**). Both Teller and Kougarok are underlain by carbonate-rich metamorphic facies, and Ca, Sr, and Mg are probable carbonate counter-cations. Weathering of Ca, Sr, and Mg carbonates and subsequent transport of these cations downslope explains this pattern of spatial variability. At Kougarok, concentrations of Ca, Mg, and Sr similarly increase from upper backslope to footslope, but concentrations of Ca and Sr decrease further down the lower backslope (Stations 10 and 1), while Mg concentrations continue to increase. A possible explanation for this behaviour is the greater affinity of cation exchange surfaces for Ca and Sr compared to Mg, thus, Ca and Sr are preferentially retained in the footslope whilst Mg is transported further down the lower backslope (Sparks, 2003, p. 189).

3.3.4 Mineral Solubility Effects

Although redox reactions are rarely at equilibrium in natural environments, comparison of field data with equilibrium models provides valuable semi-quantitative insight into the redox condition of natural environments. Because Fe appeared to be limiting the development of more reducing conditions (Section 4.3), select samples from both sites were measured for soluble Fe^{2+} following methods presented in Viollier et al. (2000). These concentrations of aqueous Fe^{2+} were then compared with model-predicted concentrations of Fe^{2+} , when coupled with an infinite $\text{Fe}(\text{OH})_{3(\text{am})}$ phase, across a range of pH values (2 – 10) and fixed E_{H} values of 400 mV, 200 mV, 0 mV, and -200 mV; activity coefficients were assumed to be equal to 1. The measured and modelled Fe^{2+} concentrations are compared in **Figure 6**, where concentrations that were below the method detection limit ($0.05 \text{ mg}\cdot\text{L}^{-1}$) are set equal to $0.025 \text{ mg}\cdot\text{L}^{-1}$ (half the detection limit). Comparison of model predicted Fe^{2+} concentrations with field data suggests that while Teller exhibits a narrower range of pH conditions than Kougarok, it exhibits a broader range of redox conditions (**Figure 6**). Although several Fe^{2+} measurements were below the detection limit, suggesting oxidizing conditions, high Fe^{2+} concentrations in some samples suggested E_{H} values below 0 mV. Therefore, Fe redox conditions at Teller ranged from mildly reducing to oxic and Fe redox conditions at Kougarok ranged from mildly oxic to oxic. Oxidation-reduction potentials (ORPs), calculated from pH, Fe^{2+} concentrations, and the Nernst equation suggest that ORPs at Teller were as low as -69 mV, while the lowest ORP at Kougarok was +134 mV (**Figure 6**). Maximum ORP values could not be determined quantitatively as some Fe^{2+} concentrations were below Fe^{2+} detection limits, at both sites. E_{H} /pH predominance diagrams were created from the 25th, 50th, and 100th concentration percentiles and are shown in **Figure 7** for the COIs where precipitation of mineral phases were predicted under some conditions. The concentrations for these diagrams were taken from filtered aqueous concentration data, thus, predicted mineral precipitation is an indication of nearly saturated or over-saturated conditions. The range of E_{H} and pH conditions observed at Teller and Kougarok are overlaid as solid yellow and solid blue lines, respectively. Only the predominance diagrams that indicated possible mineral formation under the E_{H} /pH conditions present at either site are shown in **Figure 7**. These phases included $\text{Fe}(\text{OH})_{3(\text{am})}$ (Fe), siderite (Fe), $\text{Al}(\text{OH})_{3(\text{am})}$ (Al), chalcedony (Si), barite (Ba and

395 SO₄), calcite (Ca), dolomite (Ca and Mg), and rhodochrosite (Mn). Predominance diagrams for the remaining key
396 COIs that were not predicted to form any mineral phases under any site conditions are given in **Supplementary Figure**
397 **4**.

398 To further examine which mineral phases could be controlling SPW solute concentrations, saturated conditions for
399 the mineral phases identified in **Figure 7** were modelled using sweeps of pH values from 2 – 10 at various fixed E_H
400 values (400mV, 200mV, 0mV, and -200mV). Predicted solute concentrations under the modelled saturated conditions
401 were then compared with field data to find common trends. In general, if solute concentrations were frequently
402 measured near the saturation of a mineral, or were identified to have similar dependence on pH or E_H, it was inferred
403 that the mineral phase could be controlling the generation of that solute. The mineral phases that were identified to
404 possibly be controlling solute concentrations were Al(OH)_{3(am)}, Fe(OH)_{3(am)}, chalcedony, and barite. This does not
405 preclude the presence of significant concentrations of other mineral phases, it only identifies these as possibly
406 controlling the dissolved concentrations of Al, Fe, Si, and Ba, respectively. Although it does not provide mineralogical
407 information, X-ray fluorescence (XRF) data reported by another study at Teller confirmed high concentrations of Al,
408 Fe, Si, and Ba in the organic and mineral soil layers at that site (Graham et al., 2018). We are unaware of any similar
409 studies at Kougarok, nor are we aware of any studies that provide would provide confirmatory mineralogical
410 information, for example by X-ray diffraction (XRD).

411 Aluminium concentrations in SPWs at both Teller and Kougarok appear to be controlled by the
412 dissolution/precipitation of amorphous Al hydroxide (Al(OH)_{3(am)}) (**Figure 8**). The solubility limit of Al(OH)_{3(am)} has
413 no redox dependence, but is highly pH dependent. Aluminium concentrations were generally clustered near the
414 solubility limit of Al(OH)_{3(am)}; $\text{Al(OH)}_{3(\text{am})} + 3\text{H}^+ \leftrightarrow \text{Al}^{3+} + 3\text{H}_2\text{O}$; $\log k = 10.8$. This suggests that Al SPW
415 concentrations at both sites are controlled by wetting/drying (dissolution/precipitation) processes. It also suggests that
416 there could be a significant amount of Al(OH)_{3(am)} in the soils at both sites. While organic matter may also sorb to
417 alumina surfaces, the adherence to the solubility of Al(OH)_{3(am)} suggests that significant concentrations of Al are not
418 complexed with dissolved organic matter. The predominance diagrams highlight 1) the strong pH dependence on the
419 stability of Al(OH)_{3(am)}, 2) the influence of dissolved F can have on Al speciation when Al concentrations are low,
420 and 3) that Al is a cation at low pH and an anion at high pH (**Figure 7**). Despite being a weathering product, Al
421 concentrations show a dissimilar downslope trend to other weathering products, especially at Teller (**Supplementary**
422 **Figure 5**). While the concentrations of weathering products generally increase with distance downslope, Al
423 concentrations decrease. We suspect this can be attributed to increasing pH with distance downslope. Philben et al.
424 (2020) reported a 1 pH unit increase in pH in organic soils along the Teller transect (**Figure 2**), increasing from 5.6 at
425 Station 5 to 6.7 at Station 9. Such an increase would decrease the solubility of Al(OH)_{3(am)}, and thus, decrease the
426 concentration of dissolved Al (**Figure 8**).

427 Similar to Al, Fe concentrations in SPWs at both Teller and Kougarok appear to be controlled by the
428 dissolution/precipitation of amorphous Fe hydroxide (Fe(OH)_{3(am)}). Fe concentrations were generally clustered near
429 the solubility limit of Fe(OH)_{3(am)} (**Figure 8**). Unlike Al(OH)_{3(am)} however, Fe(OH)_{3(am)} solubility is dependent on the
430 redox condition in addition to the pH; $\text{Fe(OH)}_{3(\text{am})} + 3\text{H}^+ + \text{e}^- \leftrightarrow \text{Fe}^{2+} + 3\text{H}_2\text{O}$; $\log k = 16.0$ (**Figure 8**). Fe(III) is only
431 sparingly soluble in aqueous solutions and reduction to Fe(II) significantly increases the solubility of Fe, thus, at a

432 given pH value higher aqueous concentrations are predicted and observed under more reducing conditions (**Figure 8**).
433 Iron concentrations in SPWs at both sites generally follow the pH dependence of $\text{Fe}(\text{OH})_{3(\text{am})}$ solubility (**Figure 8**).
434 This suggests that SPW concentrations of Fe at both sites are controlled by wetting/drying (dissolution/precipitation)
435 processes, coupled with the redox condition.
436 Si concentrations are frequently limited by the solubility of chalcedony, a very finely grained form of SiO_2 , which is
437 much more soluble than quartz; $\text{SiO}_2 + 2\text{H}_2\text{O} \leftrightarrow \text{H}_4\text{SiO}_4$; $\log k = -3.55$. Particularly at Kougarok, the dissolved Si
438 concentrations, coupled with a lack of a strong pH or E_{H} dependence, suggest a controlling influence of chalcedony.
439 Ba concentrations also appear to be controlled by solubility, but rather than by the solubility of an oxide or a hydroxide
440 phase, by the solubility of barite [$\text{Ba}^{2+} + \text{SO}_4^{2-} \leftrightarrow \text{BaSO}_4(\text{s})$; $\log k = 9.97$]. Unlike Al hydroxide or Fe hydroxide, barite
441 solubility lacks a strong pH dependence and instead is dependent solely on the activities of Ba^{2+} and SO_4^{2-} . Unlike Ba,
442 SO_4^{2-} concentrations are not limited by the solubility limit of barite and are generally higher and not well correlated
443 with Ba concentrations. Together, these suggest that SO_4^{2-} from another source (likely, atmospheric deposition or
444 sulfidic mineral oxidation), is suppressing barite dissolution, and thus, is reducing dissolved Ba concentrations. Barite
445 solubility can exhibit a redox dependence if conditions are sufficiently reducing to reduce SO_4^{2-} to sulphide (Neff,
446 2002). This shifts the equilibrium to greater dissolution of barite, and therefore higher conditions of Ba. The lack of
447 E_{H} dependence in observational data further suggests that neither site exhibits significant SO_4^{2-} reduction.

448 **4. Discussion**

449 The 18 stations examined herein (8 at Teller and 10 at Kougarok) represent a wide range of vegetation types, soil
450 moisture contents, permafrost extents, and hillslope positions. Coupling the spatial variability of these landscape
451 characteristics with the spatial variability of SPW solute concentrations provides valuable insight into the dominant
452 environmental controls on observed spatial variability of SPW geochemistry. It is our hope that correlating SPW
453 geochemistry with readily observable and scalable landscape features will inform earth system modelling efforts in
454 permafrost regions and provide fast and easy methods to determine if earth system models are working properly (i.e.
455 predicting the correct trends). The inferred dominant environmental controls on the observed inter-site and intra-site
456 variability of SPW solute concentrations are discussed in the following sections.

457 **4.1 The Dominant Environmental Controls on Inter-site Variability of SPW Solute Concentrations**

458 Overall, the more frequent instance of significantly greater constituent concentrations at Kougarok suggests a
459 systematic cause. The extensive low-gradient toeslope (**Figure 2**) and lack of a well-defined drainage channel at
460 Kougarok, are likely causes of the systematically higher SPW solute concentrations at Kougarok. Water perching, the
461 result of near-surface permafrost in the lower-backslope and toeslope, is likely to increase evapotranspiration and,
462 thus, SPW solute concentrations. Significant evapotranspiration caused by supra-permafrost water table perching has
463 been noted in several previous studies (Huang et al., 2022; Park et al., 2021; Sjöberg et al., 2021). Meanwhile, the
464 lack of a drainage channel at Kougarok suggests that runoff (and therefore solute exports) is more limited than at
465 Teller. Without a relatively rapid export mechanism such as a stream channel, solute transport is likely limited to

466 interflow within the Kougarok hillslope over much of the thaw season, allowing weathering products to increase to
467 significantly greater concentrations than those observed at Teller, where a well-defined drainage/export mechanism
468 does exist. Field observations from pits at Kougarok confirm observable interflow at the site. Overall, our study
469 suggests that evaporative concentration could be a significant control on SPW solute concentrations in permafrost
470 catchments, especially in those with limited drainage and therefore a perched near-surface water table. This effect has
471 been reported previously (Raudina et al., 2017), but does not appear to be widely considered, perhaps due to the
472 generally few studies of SPW solutes in permafrost regions. We suggest future efforts to predict future SPW solute
473 and nutrient dynamics directly address the impacts of evaporative concentration on permafrost catchments, especially
474 with future permafrost thaw.

475 The exception to the general observation of elevated concentrations at Kougarok versus Teller was SO_4^{2-} . Although
476 the cause of consistently higher SO_4^{2-} concentrations at Teller is unclear from the limited scope of this study, it seems
477 likely to be due to a greater abundance of sulfidic bedrock material. The presence of sulfidic bedrock in the vicinity
478 of Teller has been reported by mineral prospecting efforts (Brobst, Pinckney, and Sainsbury, 1971; Herreid, 1966;
479 Mulligan, 1965); we are unaware of any such reports near Kougarok. It should be recognized SO_4^{2-} concentrations at
480 both Kougarok and Teller are relatively low.

481 **4.2 The Dominant Environmental Controls on Intra-site Spatial Variability of SPW Solute Concentrations**

482 Vegetation influences on elemental cycles were only readily apparent for nitrogen and although vegetation induced
483 changes to soil moisture content were discernible, they were far less significant than anticipated. NO_3^- was the only
484 COI that exhibited a clear vegetation effect; elevated concentrations were associated with the presence of alder shrubs
485 and, in some cases, tall willow shrubs. These increases in NO_3^- concentrations associated with alder nitrogen-fixation
486 and the mineralization and nitrification of willow leaf litter were frequently equivoiced by increased microbial
487 denitrification in regions sufficiently moist to support it, this is perhaps one of the most significant findings of this
488 work. Although both Kougarok and Teller exhibited some indications of increased Cl concentrations in the presence
489 of tall shrubs, the net vegetation effect on soil moisture was far less than hypothesized. Redox sensitivity was also less
490 than hypothesized, and most stations seemed well-buffered at Fe redox conditions. The result of this buffering was
491 generally low NO_3^- concentrations (except where vegetation effects dominated), consistent SO_4^{2-} concentrations across
492 clear redox gradients, and variable Mn and Fe concentrations. Mn concentrations were generally low, likely due to a
493 limited source. Fe concentrations were higher at stations with higher soil moisture content, consistent with Fe
494 reduction. Similar Fe redox cycling between soluble Fe(II) species and precipitated Fe oxyhydroxides in permafrost
495 catchments has been reported recently (Patzner et al., 2022), which suggests that Fe redox buffering in permafrost
496 landscapes is widespread. Weathering, water/soil interactions, and hydrological transport were probable drivers of
497 variability for Ca, Sr, and Mg. Ca, Sr, and Mg all tended to accumulate in low-lying areas, although Ca and Sr
498 demonstrated greater accumulation potential than Mg, likely via greater affinity of cation exchange surfaces for Ca
499 and Sr compared to Mg. Mineral solubility limitations were the primary controls on Al ($\text{Al}(\text{OH})_{3(\text{am})}$), Fe ($\text{Fe}(\text{OH})_{3(\text{am})}$),
500 Ba (barite), and Si (chalcedony) concentrations. This suggests that the SPW concentrations of these constituents will
501 remain stable until those mineral phases are exhausted or soil pore hydrochemistry changes sufficiently to alter the

502 solubility of those mineral phases. Supersaturation of Al with respect to gibbsite (crystalline $\text{Al}(\text{OH})_3$) and Si with
503 respect to chalcedony in a permafrost wetland has been reported previously (Jesson et al., 2014). The solubility curves
504 for gibbsite and $\text{Al}(\text{OH})_{3(\text{am})}$ are similar, with $\text{Al}(\text{OH})_{3(\text{am})}$ being slightly more soluble at all pH values due to the
505 increased thermodynamic stability of the crystalline Al hydroxide mineral, gibbsite. Meanwhile, seasonal precipitation
506 of Fe oxyhydroxides in permafrost peatlands and their effect of carbon cycling was the subject of an excellent paper
507 by Patzner et al. (2022). Our study is the first observation we are aware of that reports the saturation controls of barite
508 on Ba in permafrost SPWs, although that could be because relatively few studies consider barium concentrations; it is
509 worthwhile emphasizing that Ba was not supersaturated with respect to barite but approached a saturated condition.
510 Future studies should also note that changes in redox condition would significantly alter $\text{Fe}(\text{OH})_{3(\text{am})}$ solubility,
511 whereas changes in pH conditions would significantly alter $\text{Al}(\text{OH})_{3(\text{am})}$ and $\text{Fe}(\text{OH})_{3(\text{am})}$ solubility.
512 Although discerning the environmental controls on spatial variability of SPW solute concentrations provides some
513 high-level insight into the effects changes in landscape character may have on soil pore hydrochemistry, our scope
514 was limited and leveraged on previously available datasets. The significance of SPW in small Arctic headwater
515 catchments as a key initial component in the freshwater hydrologic continuum is under recognized, and such
516 catchments warrant more detailed and systematic investigations.

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524 **6. Data availability statement**

525 The data that support the findings of this study are made openly available in the NGEE-Arctic data repository at (DOI:
526 10.5440/1735757).

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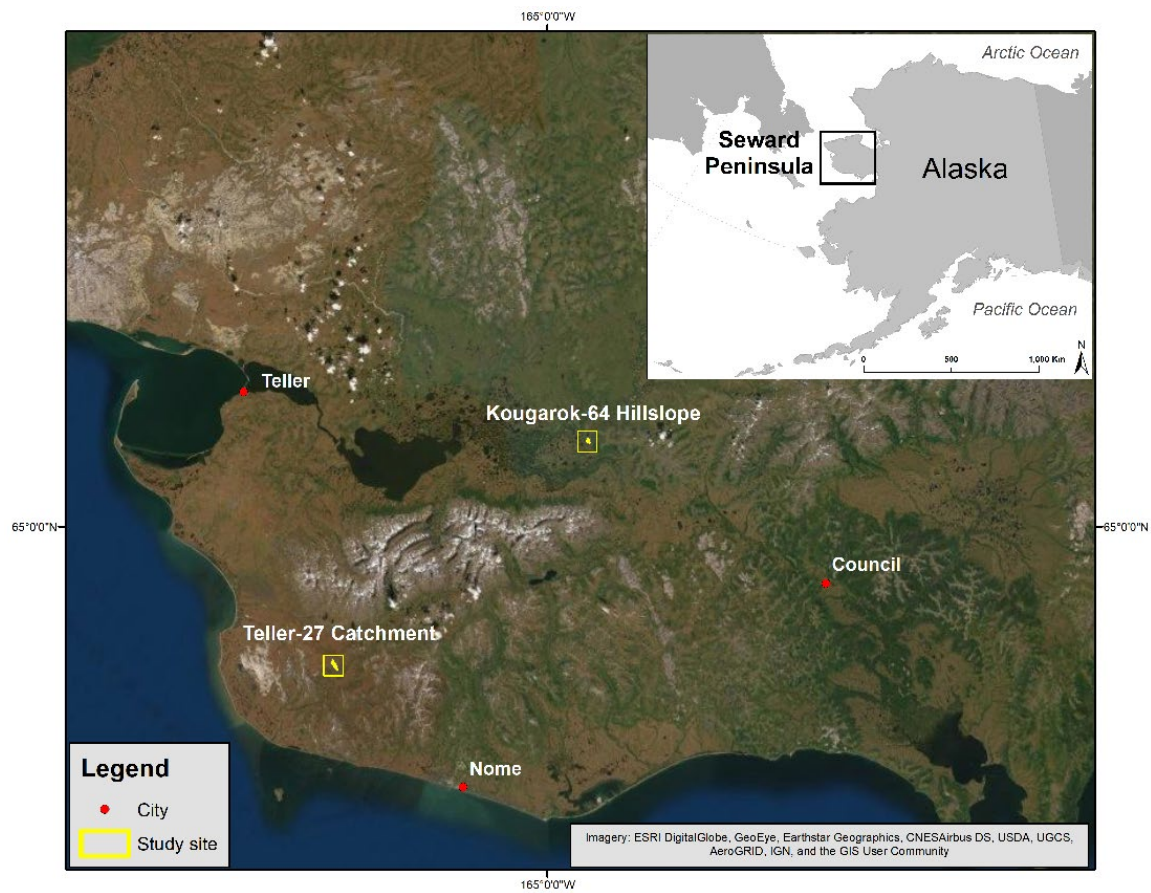
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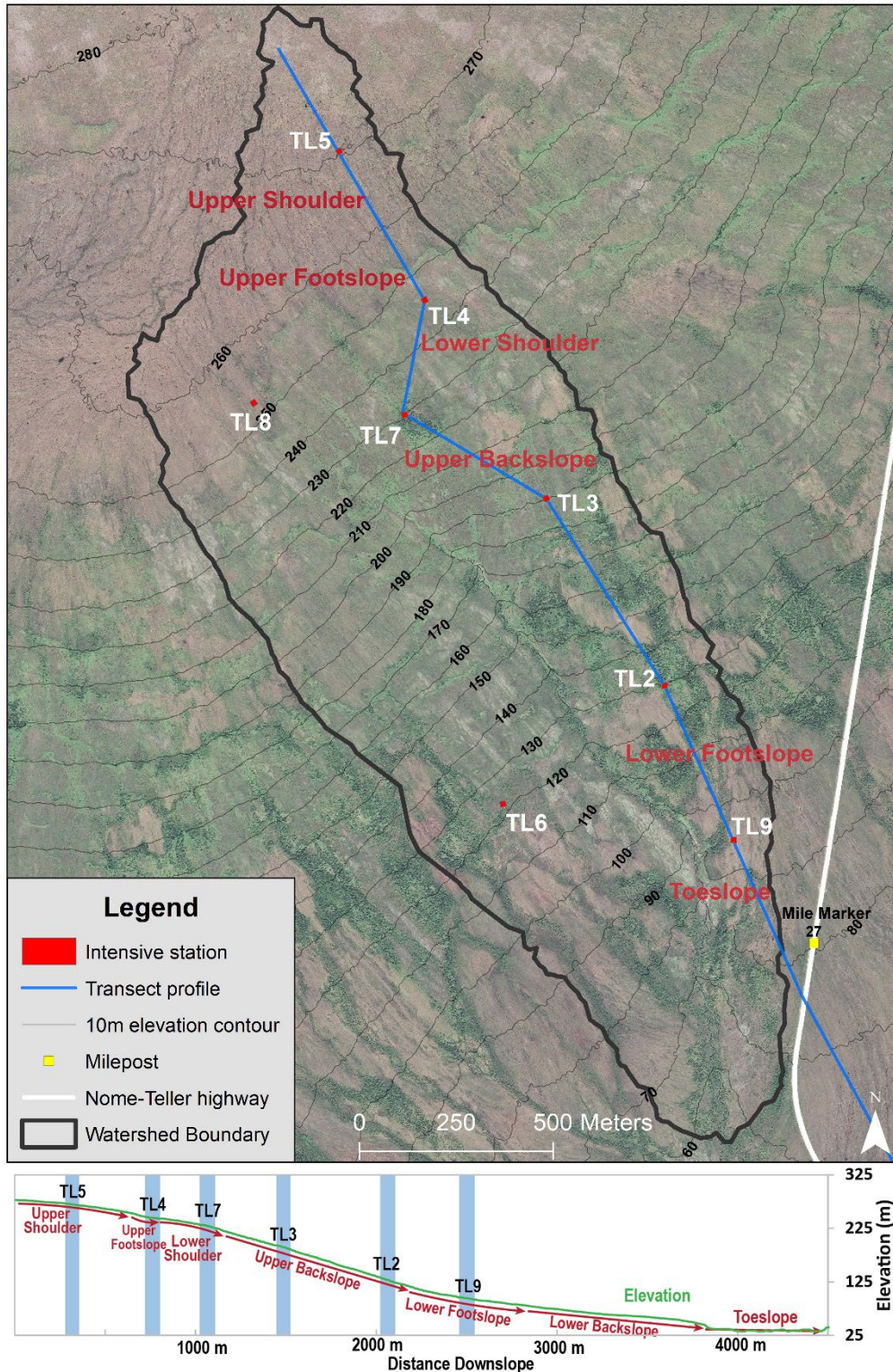
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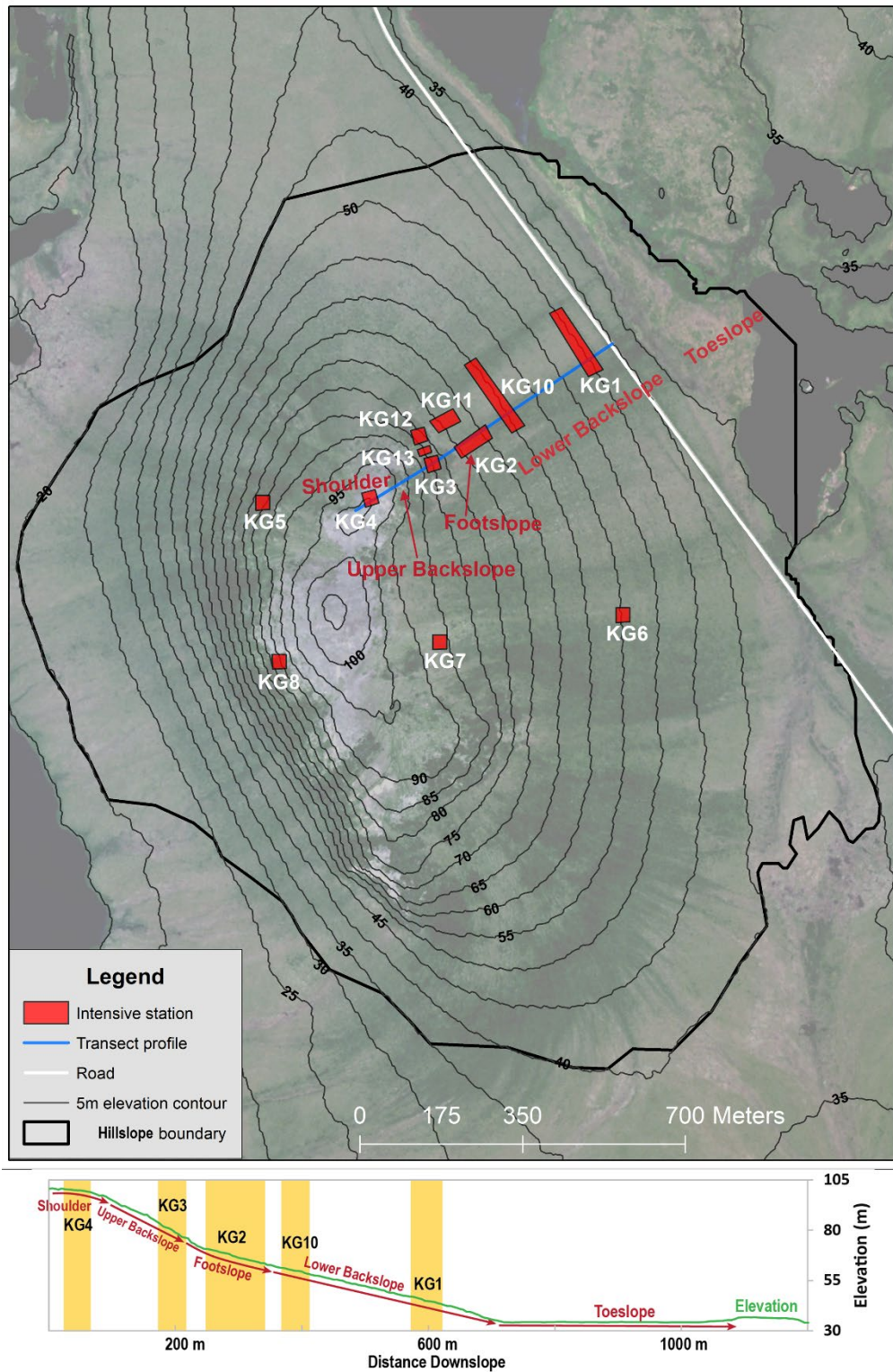
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758 **Figure 1. Location of the Teller and Kougarak field sites with respect to the municipalities of Teller, Nome, and Council.**
759 **All are located on the Seward Peninsula in northwestern Alaska. RGB composite imagery from the 8-band WorldView-2**
760 **imagery obtained on July 14, 2017 at 1.5 m resolution downloaded from the DigitalGlobe website**
761 **(<https://www.digitalglobe.com/>).**

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Figure 2. Topographic map of Teller. Station areas are shown as red polygons and the topographic station transect is given as a solid blue line. The hillslope transect elevation profile is given below the map in green, with stations along the transect in blue and hillslope positions noted with red arrows and text. RGB composite imagery from the 8-band WorldView-2 imagery obtained on July 27, 2011 at 1.5 m resolution downloaded from the DigitalGlobe website (<https://www.digitalglobe.com/>).



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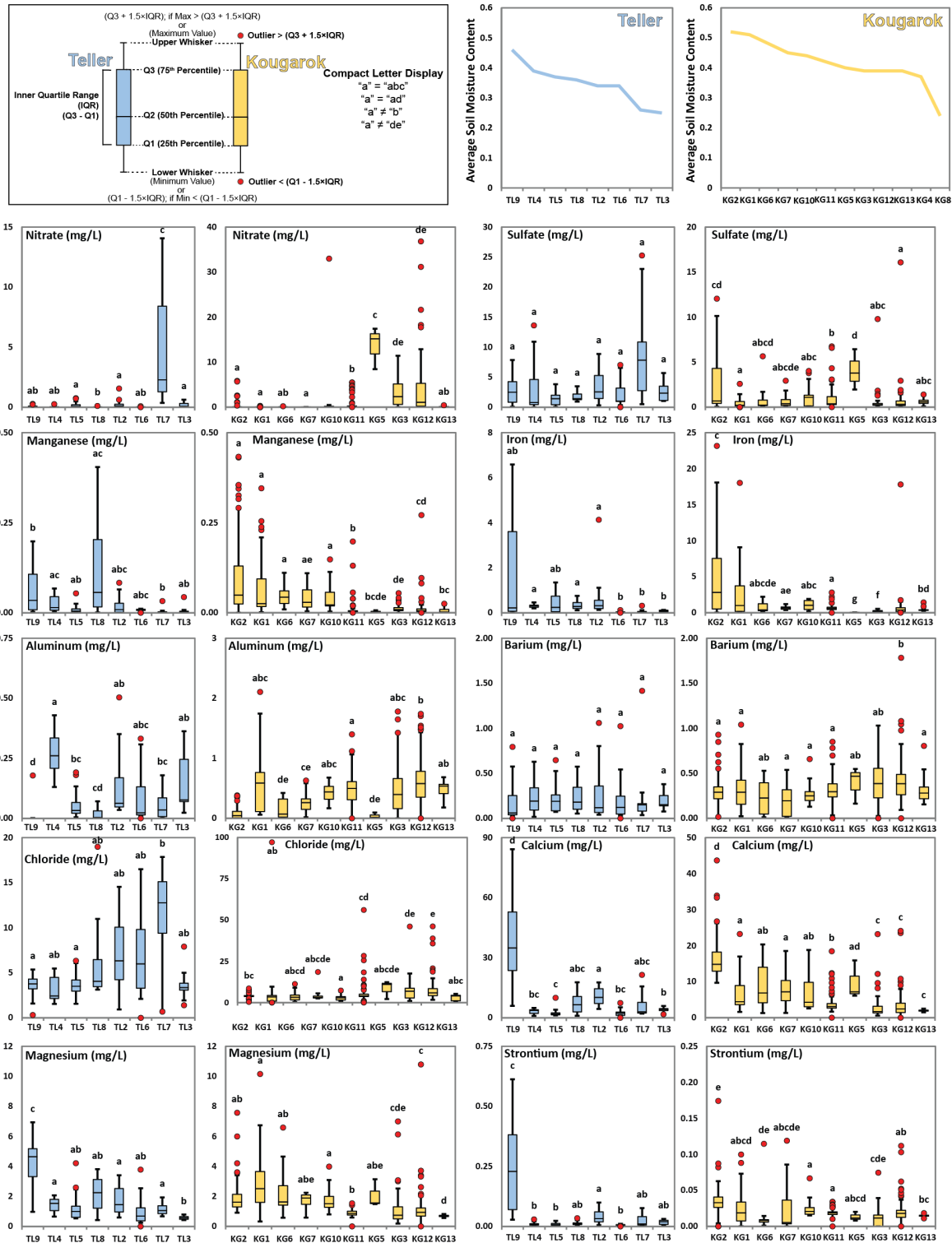
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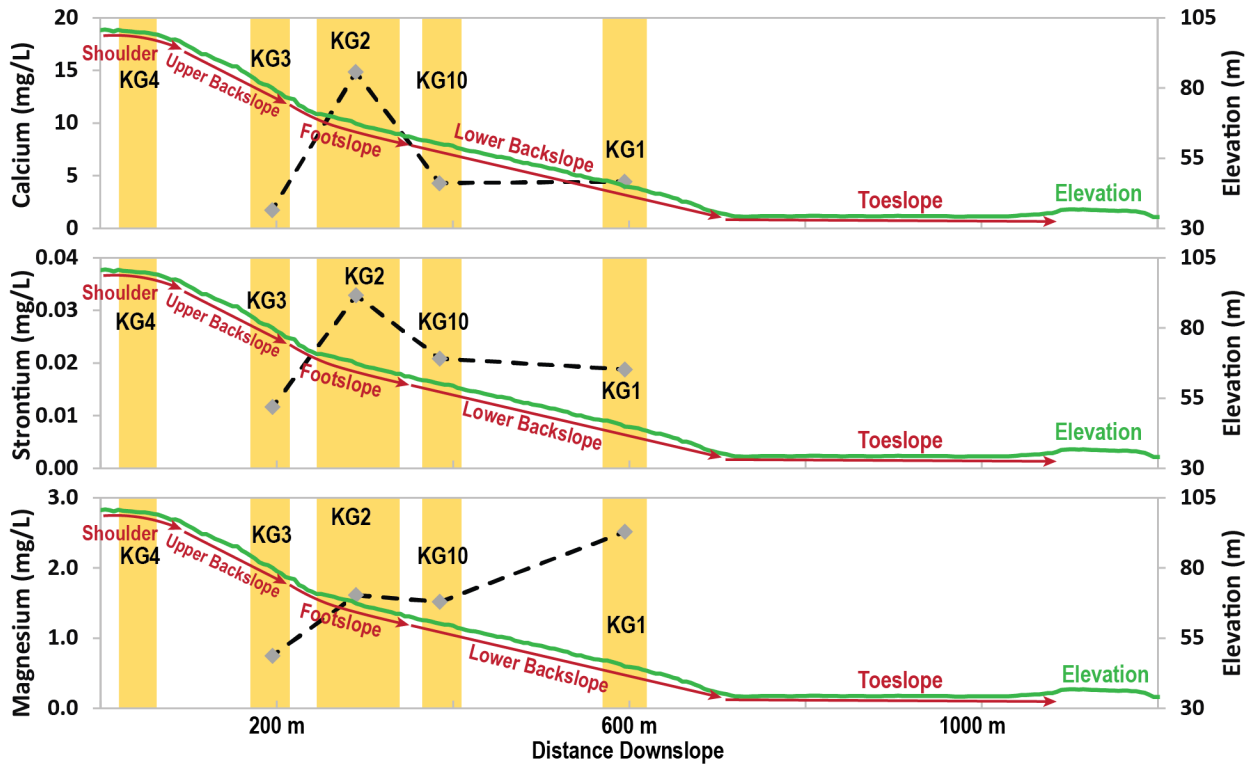
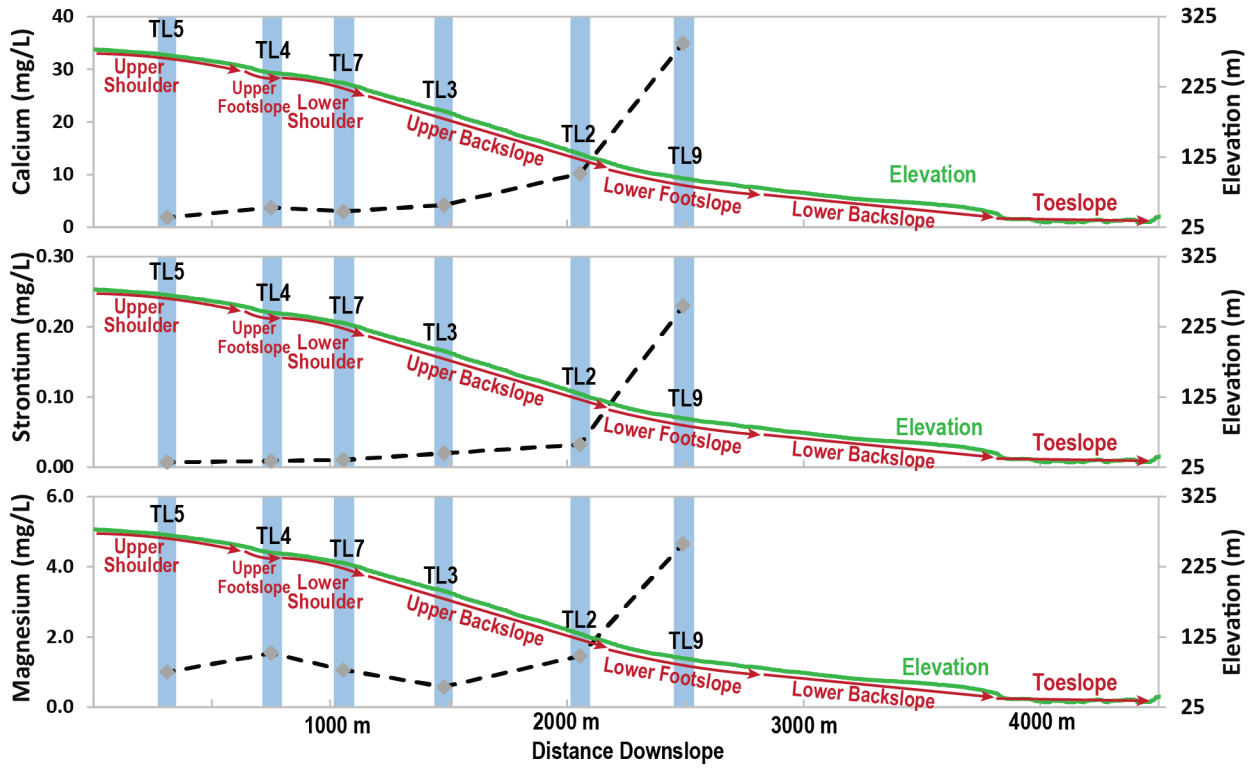
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Figure 3. Topographic map of Kougarok. Station areas are shown as red polygons and the station transect is given as a solid blue line. The transect elevation profile is given below the map in green, with stations along the transect in yellow and hillslope positions noted with red arrows and text. RGB composite imagery from the 8-band WorldView-2 imagery obtained on July 14, 2017 at 1.5 m resolution downloaded from the DigitalGlobe website (<https://www.digitalglobe.com/>).



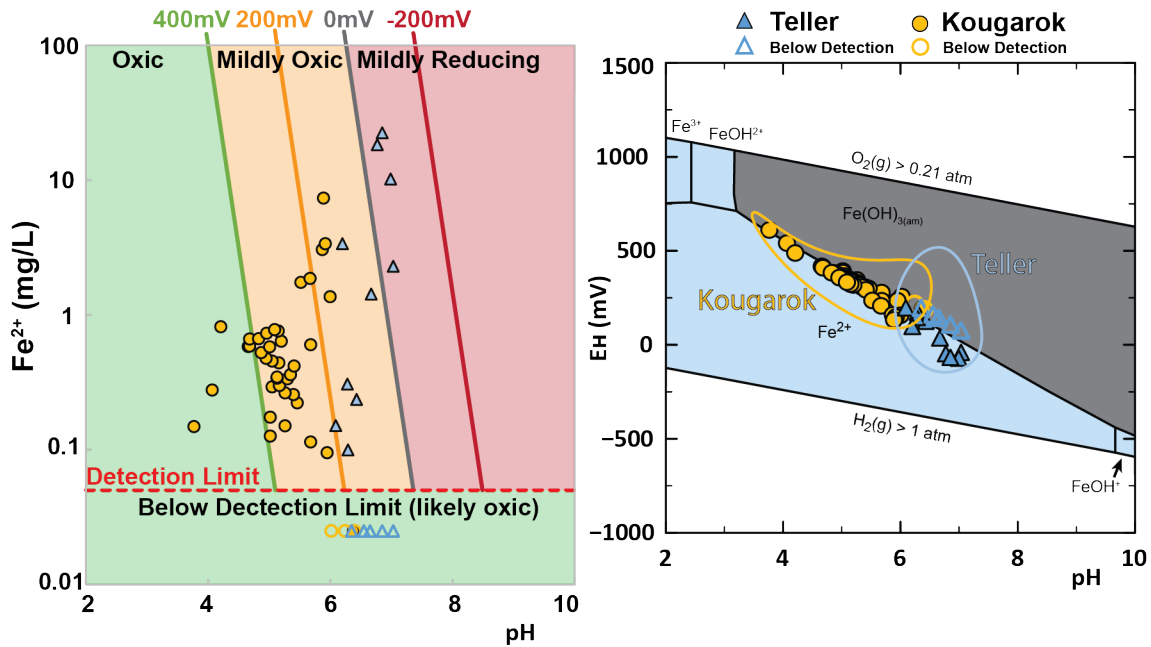
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777 **Figure 4. Mean COI concentrations at Teller (blue) and Kougarak (yellow) stations. Stations are arranged (left to right) by**
 778 **soil moisture content determined by P-Band SAR (top right). Boxplots show the first, second, and third data quartiles, with**
 779 **box whiskers representing either 150% of the inner quartile range (IQR), or the maximum or minimum value, when that**
 780 **value was less than 1.5×IQR. Red circles represent data points outside of the 1.5×IQR whiskers (i.e. outliers). Note that the**
 781 **concentration scales on the Teller and Kougarak plots often differ.**



784 Figure 5. Median (50th percentile) concentrations (grey diamonds with dashed black lines) of Ca, Sr, and Mg, with distance
 785 downslope at Teller (blue) and Kougarak (yellow) along topographic transects; areas of stations are indicated by blue and
 786 yellow colouring, respectively. The elevation profiles of the hillslopes are plotted in green, on separate y-axes (right axes).
 787 Topographic regions of both catchments are indicated by red arrows along the elevation gradient.

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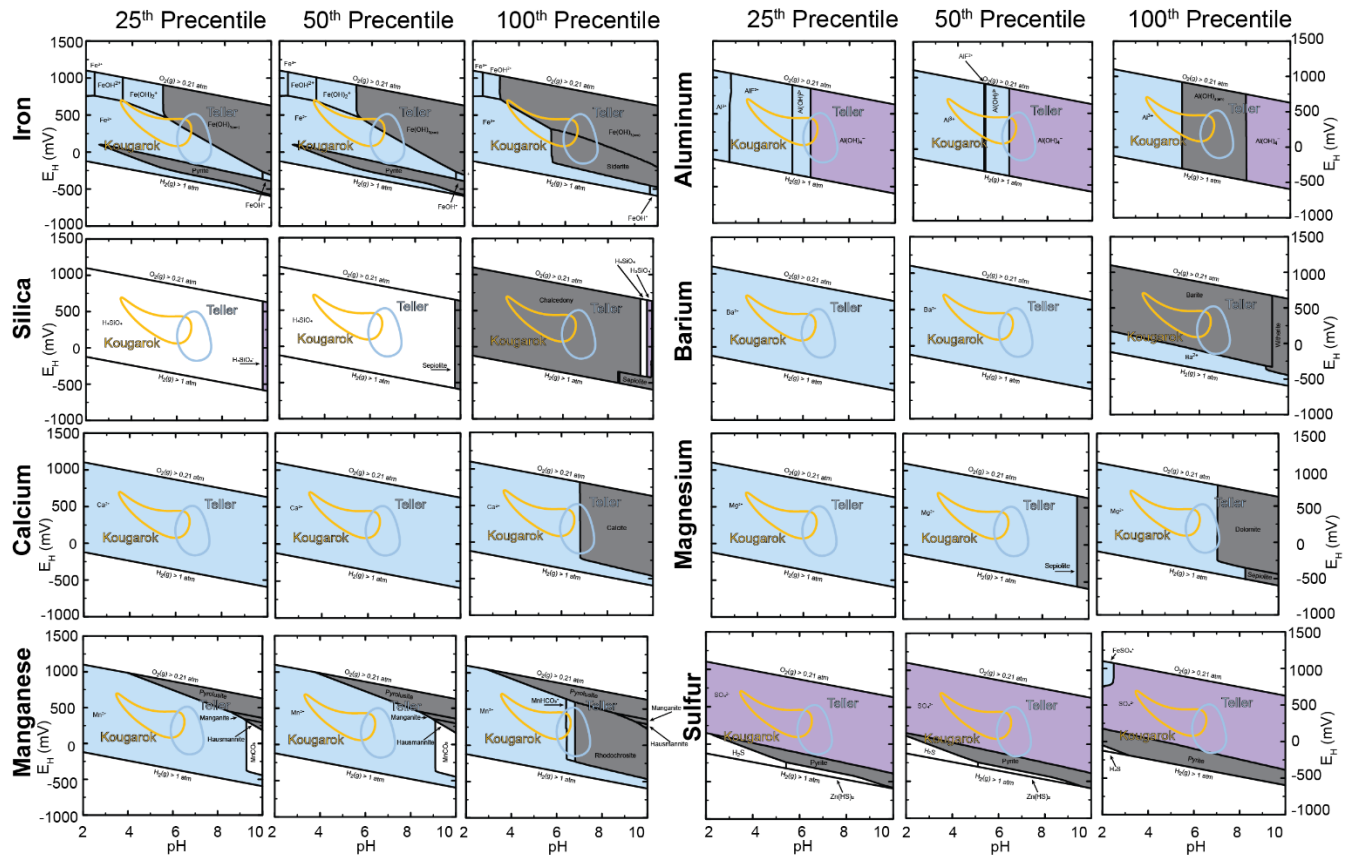
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Figure 6. Left: Model-predicted Fe²⁺ concentrations in saturated solutions of Fe(OH)_{3(am)} at fixed E_H conditions of 400 mV (green), 200 mV (orange), 0 mV (grey), and -200 mV (red), compared with field concentrations of Fe²⁺ at Teller (red circles) and Kougarak (yellow circles). Right: Fe predominance diagram, showing the dominant specie of Fe under a range of E_H/pH conditions. E_H/pH regions relevant to Teller and Kougarak are outlined in blue and yellow, respectively. Samples with Fe²⁺ concentrations below the detection limit are given as colour coordinated open circles set at 0.025 mg·L⁻¹ (half the detection limit) in both sides of the figure.

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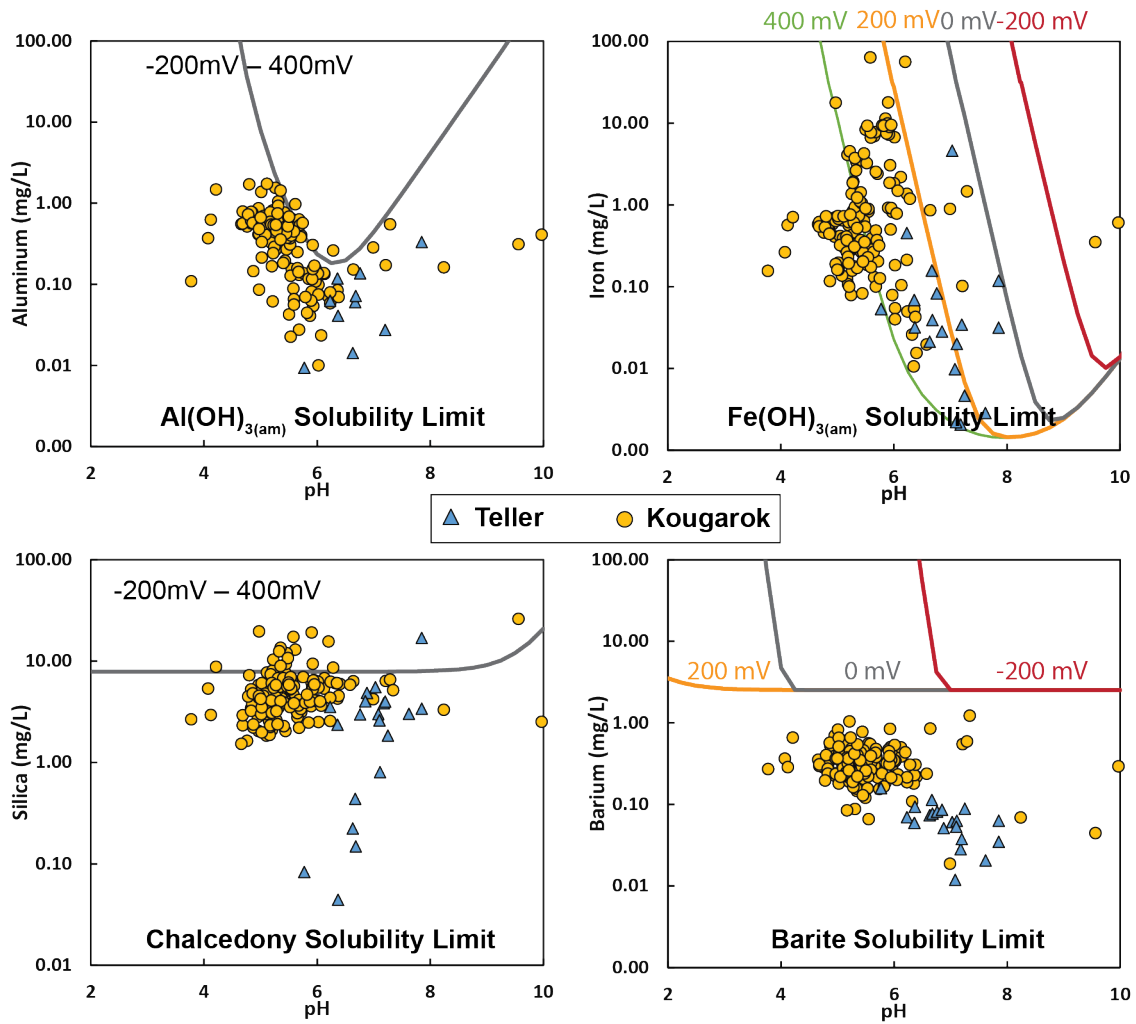
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Figure 7. E_H /pH diagrams for key species that indicated possible mineral formation under the E_H /pH conditions present at either Teller or Kougarok. The E_H and pH conditions observed at Teller and Kougarok are overlaid as blue and yellow lines, respectively. Mineral species (solids) are shown in grey, cations are shown in blue, anions are shown in purple, neutral species are shown in white. Predominance diagrams were created in PhreePlot using the phreeqc.dat database, with inorganic carbonate reduction to methane “turned off.”

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805 **Figure 8. Modelled solute concentrations in solutions saturated with $\text{Al(OH)}_{3(\text{am})}$, $\text{Fe(OH)}_{3(\text{am})}$, chalcedony, and barite, with**
806 **respect to pH (x-axis) and E_H (model lines), overlaid with observed solute concentrations.**

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809 Table 1. Teller Station Physical Characteristics

| | Hillslope Position | Vegetation | | | | | Relative wetness | | | Permafrost | | |
|-----|--------------------|-----------------------------|--------------------------------------|--------------------------------------|------|-------------------------|---------------------------------|--------------------------|-------------------------|---------------------------------|-------------------|--------------------------------------|
| | | Vegetation type | Average (maximum) canopy height (cm) | Dominant PFT | | Low to tall shrub cover | Average TDR soil moisture (VMC) | Average P-band SAR (VMC) | Average snow depth (cm) | Average Ground Temperature (°C) | Permafrost Extent | Average (maximum) thaw depth (cm) |
| TL9 | Lower Footslope | Wetland complex | 28 (41) | Bryophyte | 44 % | 10 % | NA | 0.46 | 68.4 | 0 | Marginal | 101 (>120) |
| TL5 | Upper Shoulder | | 12 (45) | Graminoid | 45 % | 7 % | 0.55 | 0.37 | 103.3 | -0.45 | Near-surface | 97 ^r (>114 ^f) |
| TL8 | Upper Footslope | | 7 (34) | Bryophyte | 33 % | 20 % | 0.55 | 0.36 | 77.7 | -0.6 | | 69 ^r (>120) |
| TL3 | Upper Backslope | Cassiope dwarf shrub tundra | 9 (23) | Evergreen dwarf shrub | 47 % | 12 % | NA | 0.25 | 62.1 | 2.2 | None/ deep | 72 ^r (82 ^r) |
| TL4 | Upper Footslope | | 8 (14) | | 58 % | 4 % | 0.35 | 0.39 | 89.5 | 0.5 | Marginal | 40 ^r (70 ^r) |
| TL2 | Upper Backslope | Mesic willow shrubland | 84 (141) | Deciduous low to tall shrub (willow) | 44 % | 44 % | 0.4 | 0.34 | 124 | 2.4 | None/ deep | 75 ^r (>120) |
| TL7 | Lower Shoulder | | 151 (189) | | 37 % | 37 % | 0.46 | 0.26 | 128.8 | 2.4 | | 51 ^r (66 ^r) |
| TL6 | Upper Backslope | Willow-birch tundra | 64 (115) | Forb | 23 % | 32 % | 0.38 | 0.34 | 86.4 | 1.2 | None/ deep | 67 ^r (102 ^r) |

810

811 PFT – plant functional type; dwarf shrub (height <40 cm), low shrub (height 40-200 cm), low to tall shrub (height 40 to >200 cm tall).
812 Deciduous shrub PFT classes identify the dominant species in the plant community as either willow or willow and birch. There is no alder at
813 the Teller site. Low to tall shrub cover represents the sum of deciduous low shrubs, deciduous low to potentially tall willow and birch, and
814 deciduous low to tall alder.

815 ¹Single point soil moisture measurements. Data are more accurate than P-band SAR but represent a much smaller spatial scale.

816 ²P-band SAR has 30m resolution.

817 ^rResistive layer was rock; all others are permafrost. A temperature probe was used to determine if the resistive layer was permafrost (≤ 0 °C)
818 or rock (>2 °C). Thaw depth is an average of 4 measurements from the vegetation plot corners within the IS and was measured at the end of
819 the growing season.

820

821 **Table 2. Kougarok Station Physical Characteristics**

| | <i>Hillslope Position</i> | <i>Vegetation</i> | | | | | <i>Relative wetness</i> | | | <i>Permafrost</i> | | |
|-------------|---------------------------|---------------------------------|--------------------------------------|-------------------------------------|------|-------------------------|---------------------------------|--------------------------|-------------------------|---------------------------------|-------------------|-----------------------------------|
| | | Vegetation type | Average (maximum) canopy height (cm) | Dominant PFT | | Low to tall shrub cover | Average TDR soil moisture (VMC) | Average P-band SAR (VMC) | Average snow depth (cm) | Average Ground Temperature (°C) | Permafrost Extent | Average (maximum) thaw depth (cm) |
| <i>KG3</i> | Upper Backslope | Alder shrubland | 204 (265) | Deciduous low to tall shrub (alder) | 30 % | 53 % | 0.19 | 0.39 | 131.3 | -0.01 | Near-surface | 48° (53r) |
| <i>KG12</i> | Footslope | | NA | NA | NA | NA | 0.30* | 0.39 | NA | NA | | NA |
| <i>KG1</i> | Lower Backslope | Alder savanna in tussock tundra | 60 (90) | Deciduous low shrub | 31 % | 44 % | NA | 0.51 | 83.4 | -2.5 | Near-surface | 61 (68) |
| <i>KG2</i> | Footslope | | 48 (73) | Graminoid | 30 % | 42 % | 0.63 | 0.52 | 102.3 | -1.2 | | 75 (89) |
| <i>KG6</i> | Lower Backslope | | 24 (61) | Graminoid | 46 % | 17 % | 0.36 | 0.48 | 66.2 | -2.2 | | 58 (62) |
| <i>KG10</i> | Lower Backslope | | NA | NA | NA | NA | NA* | 0.44 | 71.4 | NA | | NA |
| <i>KG11</i> | Footslope | | NA | NA | NA | NA | 0.59* | 0.42 | NA | NA | | NA |
| <i>KG7</i> | Upper Backslope | Tussock-lichen tundra | 20 (22) | Graminoid | 34 % | 14 % | 0.51 | 0.45 | 54.7 | -2.1 | Near-surface | 76 (100) |
| <i>KG4</i> | Shoulder | Dryas-lichen shrub tundra | 6 (12) | Evergreen dwarf shrub | 62 % | 1 % | NA | 0.37 | NA | -1.9 | Near-surface | 0° (0°) |
| <i>KG13</i> | Upper Backslope | | NA | NA | NA | NA | 0.41* | 0.39 | 92.1 | NA | | NA |
| <i>KG5</i> | Upper Backslope | Willow-birch tundra | 62 (137) | Deciduous low shrub | 60 % | 62 % | NA | 0.4 | 178.4 | > 0 | Deep | 88 (96) |
| <i>KG8</i> | Upper Backslope | | 45 (120) | Evergreen dwarf shrub | 52 % | 42 % | 0.23 | 0.24 | 85.5 | -0.04 | Near-surface | 44° (55°) |

822
823 Note: PFT – plant functional type. Deciduous shrub PFT classes identify the dominant species in the community as either willow, alder,
824 willow and birch, or alder, willow, and birch. Low to tall shrub cover represents the sum of deciduous low shrubs, deciduous low to potentially
825 tall willow and birch, and deciduous low to tall alder.
826 ¹Single point soil moisture measurements. Data are more accurate than P-band SAR but represent a much smaller spatial scale.
827 ²P-band SAR has 30m resolution.
828 *Average gravimetric water content measurements, corrected to VMC by bulk density.
829 [†]Resistive layer was rock; all others are permafrost. A temperature probe was used to determine if the resistive layer was permafrost (≤ 0 °C)
830 or rock (>2 °C). Thaw depth is an average of 4 measurements from the vegetation plot corners within the IS and was measured at the end of
831 the growing season.

Table 3. Inter-Site Mann-Whitney U-Test Results

| | <u>Teller</u> | | | <u>Kougarok</u> | | | z | Site with Higher Median | Effect Size | Difference in Correlation |
|-----------------------|---------------|------------|---------|-----------------|------------|---------|-------------|-------------------------|-------------|---------------------------|
| | n | $\sum R_i$ | U_i | n | $\sum R_i$ | U_i | | | | |
| Na | 59 | 3184 | 14811.5 | 275 | 52761.5 | 1413.5 | 9.95 | Kougarok | 0.54 | large |
| F | 59 | 3502 | 14375.5 | 273 | 51776.5 | 1731.5 | 9.46 | Kougarok | 0.52 | large |
| K | 59 | 3882 | 14113 | 275 | 52063 | 2112 | 8.92 | Kougarok | 0.49 | medium-large |
| Si | 59 | 4119 | 13876.5 | 275 | 51826.5 | 2348.5 | 8.56 | Kougarok | 0.47 | medium-large |
| Al | 58 | 4952 | 12709 | 275 | 50659 | 3241 | 7.11 | Kougarok | 0.39 | medium |
| Oxalate | 57 | 4996 | 12161.5 | 272 | 49289.5 | 3342.5 | 6.75 | Kougarok | 0.37 | medium |
| B | 59 | 5429 | 12566.5 | 275 | 50516.5 | 3658.5 | 6.62 | Kougarok | 0.36 | medium |
| Zn | 58 | 5605 | 12056 | 275 | 50006 | 3894 | 6.12 | Kougarok | 0.34 | medium |
| SO₄ | 58 | 13653 | 3892.5 | 273 | 41293.5 | 11941.5 | 6.08 | Teller | 0.33 | medium |
| Fe | 58 | 5958 | 11703 | 275 | 49653 | 4247 | 5.60 | Kougarok | 0.31 | medium |
| Ba | 58 | 6256 | 11405.5 | 275 | 49355.5 | 4544.5 | 5.15 | Kougarok | 0.28 | medium |
| Ti | 58 | 6266 | 11395.5 | 275 | 49345.5 | 4554.5 | 5.13 | Kougarok | 0.28 | medium |
| NO₂ | 54 | 5588 | 10585.5 | 272 | 47713.5 | 4102.5 | 5.12 | Kougarok | 0.28 | medium |
| Li | 58 | 7778 | 9883 | 275 | 47833 | 6067 | 2.86 | Kougarok | 0.16 | small-medium |
| Br | 58 | 8485 | 9060.5 | 273 | 46461.5 | 6773.5 | 1.73 | Equal | 0.09 | small |
| NO ₃ | 58 | 8576 | 8969 | 273 | 46370 | 6865 | 1.59 | Kougarok | 0.09 | small |
| Sr | 58 | 8683 | 8978 | 275 | 46928 | 6972 | 1.51 | Kougarok | 0.08 | small |
| PO ₄ | 54 | 9659 | 6460.5 | 271 | 43316.5 | 8173.5 | 1.36 | Equal | 0.08 | small |
| Mg | 58 | 10495 | 7166 | 275 | 45116 | 8784 | 1.21 | Teller | 0.07 | small |
| Cr | 58 | 8884 | 8777 | 275 | 46727 | 7173 | 1.20 | Kougarok | 0.07 | small |
| Mn | 58 | 9164 | 8497 | 275 | 46447 | 7453 | 0.78 | Teller | 0.04 | small |
| Cl | 58 | 9221 | 8266.5 | 272 | 45394.5 | 7509.5 | 0.57 | Kougarok | 0.03 | small |
| Ca | 58 | 10016 | 7645 | 275 | 45595 | 8305 | 0.50 | Teller | 0.03 | small |

835 **Table 4. Dominant Environmental Controls on SPW Geochemistry at Teller and Kougarok**

| Environmental Control | Analytes Affected |
|--|---|
| Vegetation | NO ₃ ⁻ |
| Soil Moisture/Redox | NO ₃ ⁻ , Mn, Fe, SO ₄ (occasionally) |
| Water/Soil Interactions & Hydrologic Transport | Ca, Mg, Sr |
| Mineral Solubility | Al, Ba, Si, Fe |