# **Environmental Controls on Observed Spatial Variability of Soil**

# Pore Water Geochemistry in Small Headwater Catchments

#### **Underlain with Permafrost** 3

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  - Abstract. Soil pore water (SPW) chemistry can vary substantially across multiple scales in Arctic permafrost landscapes. The magnitude of these variations and their relationship to scale are critical considerations for understanding current controls on geochemical cycling and for predicting future changes. These aspects are especially important for Arctic change modelling where accurate representation of sub-grid variability may be necessary to predict watershed scale behaviours. Our research goal was to characterize intra- and inter-watershed soil water geochemical variations at two contrasting locations in the Seward Peninsula of Alaska, USA. We then attempt to establish which environmental factors were important for controlling concentrations of important pore water solutes in these systems. The SPW geochemistry of 18 locations spanning two small Arctic catchments were examined for spatial variability and its dominant environmental controls. The primary environmental controls considered were vegetation, soil moisture/redox condition, water/soil interactions and hydrologic transport, and mineral solubility. The sampling locations varied in terms of vegetation type and canopy height, presence or absence of near-surface
- 30 concentrations, associated with the localized presence of nitrogen-fixing alders and mineralization and nitrification of

permafrost, soil moisture, and hillslope position. Vegetation was found to have a significant impact on SPW NO3NOs-

- 31 leaf litter from tall willow shrubs. The elevated NO3NO<sub>6</sub> concentrations were however, frequently equipoised by
- 32 increased microbial denitrification in regions with sufficient moisture to support it. Vegetation also had an observable
- 33 impact on soil moisture sensitive constituents, but the effect was less significant. The redox conditions in both
- 34 catchments were generally limited by Fe reduction, seemingly well-buffered by a cache of amorphous Fe hydroxides,
- 35 with the most reducing conditions found at sampling locations with the highest soil moisture content. Non-redox-
- 36 sensitive cations were affected by a wide variety of water-soil interactions that affect mineral solubility and transport.
- 37 Identification of the dominant controls on current SPW hydrogeochemistry allows for qualitative prediction of future
- 38 geochemical trends in small Arctic catchments that are likely to experience warming and permafrost thaw. As source

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- areas for geochemical fluxes to the broader Arctic hydrologic system, geochemical processes occurring in these 39
- 40 environments are particularly important to understand and predict with regards to such environmental changes.

#### 1. Introduction

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- 42 Permafrost thaw in the Arctic is causing significant changes to landscape structure (Kokelj and Jorgenson, 2013;
- Rowland et al., 2010), hydrology (Hiyama et al., 2021; Kurylyk et al., 2021; Liljedahl et al., 2016; Vonk, Tank, and 43
- 44 Walvoord, 2019; Walvoord and Kurylyk, 2016), vegetation (Lara et al., Nitze, Grosse, and McGuire, 2018; Myers-
- 45 Smith et al., 2011; Sturm, Racine, and Tape, 2001; K. D. Tape et al., Hallinger, Welker, and Ruess, 2012; K. Tape,
- 46 Sturm, and Racine, 2006;), and biogeochemistry (O'Donnell et al., 2021; Frey and McClelland, 2009; Salmon et al.,
- 47 2019; Vonk, Tank, and Walvoord, 2019). The integrated hydrogeochemical effects of these environmental changes
- 48 are already apparent in the chemistry of the large Arctic rivers, where fluxes of carbon and nutrients are increasing,
- 49 leading to enhanced nutrient loadings, with strong implications for the global carbon cycle (Bring et al., 2016; Fuchs
- 50 et al., 2020; McClelland et al., 2016). While the watershed areas of large Arctic rivers are vast, recent studies suggest
- 51 that solute concentrations in these large rivers are likely controlled by solute generation processes occurring at much 52
  - smaller scales (Harms and Ludwig, 2016; Koch et al., Runkel, Striegl, and McKnight, 2013; Shogren et al., 2019;
- 53 Vonk et al., 2015).
- 54 While there is a rapidly growing body of literature focused on observing and understanding environmental changes
  - over time with further Arctic warming, relatively few studies directly address the existing spatial variability, within
- 56 catchments or across catchments, and we are not aware of any studies that have combined field observations with
- 57 thermodynamic modelling in an effort to understand the causes of the existing spatial variability. Therefore, we have
- 58 a limited understanding of the key environmental controls on the spatial distribution of soil pore water solute
- 59 concentrations. In this study, we quantitatively evaluate the spatial variability of soil pore water (SPW) geochemistry
  - within and between two distinct catchments underlain with permafrost, and then seek to identify the source of the
- 61 observed spatial variability.
- 62 This study takes advantage of a scientifically diverse array of observations and datasets made available by the Next
- 63 Generation Ecosystem Experiment (NGEE) Arctic project, sponsored by the US Department of Energy Office of
- Science. Most of the locations studied herein were selected by the NGEE Arctic project to provide co-located 64
- 65 measurements in a wide range of vegetation types, nested within representative hillslopes and catchments. Although
- selected largely to represent a range of vegetation structure, such as shrub abundance and canopy height, these 66
- 67 locations also have considerable variability in other environmental parameters including, but not limited to: soil
- 68 moisture and temperature, presence or absence of near-surface permafrost, and maximum observed thaw depth (Table
- 69 1 and Table 2). The vegetation-delineated sampling approach presented here-provides an opportunity to not only
- 70 quantify the biogeochemical variability of SPW in Arctic environments, but also to investigate the root causes of that
- 71 observed variability. Data from additional sampling locations, available from a collaborative co-located study, were
- 72 also utilized when possible.
- 73 Our overarching hypothesis is that vegetation-type and hillslope position are the dominant controls on spatial
- variability of SPW geochemistry at the NGEE Arctic field sites located on the Seward Peninsula. Vegetation-type 74

seems likely to have a significant effect on SPW geochemistry both directly and indirectly. Indirect effects would include vegetation canopy impacts on soil moisture (through evapotranspiration and snow trapping). Direct effects of vegetation would include nutrient cycle changes resulting from the annual deposition of plant litter. Such a direct effect can be augmented at sites populated by alder shrubs due to this genus of deciduous shrubs ability to form a symbiotic relationship with nitrogen-fixing Frankia, which they host in underground root nodules. Nitrogen fixation associated with alders has previously been shown to accelerate local nitrogen cycling (Binkley et al., 1992; Clein and Schimel, 1995; Bühlmann et al., 2014). Directly, through the increased cycling of some solutes (e.g. increased nitrogen concentrations in the vicinity of alders, which add nitrogen to soils via a symbiotic relationship with nitrogen-fixing bacterium; Salmon et al., 2019), and indirectly, through effects on soil moisture (i.e. evapotranspiration and trapping of snow). Soil moisture will also affect SPW geochemistry, particularly of redox sensitive species, by limiting oxygen diffusion and thus controlling which regions develop anoxic/reducing geochemical conditions. Soil moisture impacts will likely be correlated with vegetation-type as well as hillslope position, and the presence or absence of perching layers, including permafrost, all of which impact the vertical and horizontal drainage characteristics of a watershed. Chemical species that are not redox-sensitive or controlled by biogeochemical reactions are likely to be effected affected by transport, solubility, and water/sediment/organic matter interactions, and therefore largely controlled by hillslope position as well as soil moisture. Identifying the dominant controls on solute concentration variability within each catchment and across catchments will facilitate better projections of future soil pore hydrogeochemistry in permafrost landscapes, and how these signatures are related to changing soil moisture and increasing in tundra shrub abundance in a changing Arctic (Bring et al., 2016; Myers-Smith et al., 2011; Prowse et al.e, Bring, Mård, and Carmack, 2015; Salmon et al., 2019; Sturm et al., 2001; K. D. Tape et al., 2012; K. Tape et al., 2006; Wrona et al., 2016, 2016). Arctic warming and associated permafrost thaw will increase hydrological connectedness between terrestrial and aquatic environments through deepening of the active layer and the formation of deeper, more coherent groundwater flow paths (Bring et al., 2016; Harms and Jones, 2012; Prowse, Bring, Mård, Carmack, et al., 2015a; Prowse, Bring, Mård, and Carmack, et al., 2015b). Meanwhile, changes in hydrogeochemical signatures in larger Arctic rivers are likely to originate in smaller catchments (McClelland et al., 2016; Prowse et al., Bring, Mård, and Carmack, 2015; Shogren et al., 2019; Spence, et al., Kokelj, McCluskie, and Hedstrom, 2015). In this sense, changes in hydrogeochemistry in small Arctic catchments not only impact hydrogeochemistry at much larger scales, but also prognosticate the future hydrogeochemistry of larger Arctic rivers.

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## 2. Methods

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# 2.1 Site Descriptions

- 106 This study focuses on two sites with permafrost on the Seward Peninsula of western Alaska, the Teller-27 Catchment
- and the Kougarok-64 Hillslope (Figure 1). The Teller-27 Catchment, henceforth "Teller," is a small (~2.25 km²)
- 108 headwater catchment located west of mile marker 27 along the Nome-Teller Highway northwest of Nome, Alaska.
- 109 The Kougarok-64 Hillslope, henceforth "Kougarok," is a hillslope (~2.0 km²) located west of mile marker 64 along

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111 Kougarok where concentrated, multi-year, co-located observations of soil water chemistry, vegetation characteristics, 112 soil moisture and temperature, and other measurements have been collected as part of the NGEE Arctic Research 113 Project. -These are identified as TL# (Teller Station #) or KG# (Kougarok Station #) in Figure 2 and Figure 3, 114 respectively. It should be noted that Teller and Kougarok are not "paired watersheds" in the classical sense, differing 115 in only one major characteristic, which provides the basis for comparison. Instead, Teller and Kougarok differ in many 116 respects and are both representative of the broad range of hillslope conditions common on the Seward Peninsula. 117 Detailed descriptions of Teller and Kougarok have been published previously (Jafarov et al., 2018; Léger et al., 2019; 118 Philben et al., 2019, 2020; Salmon et al., 2019; Yang et al., 2020), therefore, only the catchment characteristics that 119 are probable sources of variability in SPW chemistry will be highlighted here. 120 Teller is a discrete catchment with a well-defined central drainage, a vertical declivity of approximately 200 m, and a 121 catchment area of approximately 2.25 km<sup>2</sup>. Temperature probes, soil pits, coring activities, and geophysical 122 interpretations at Teller have confirmed the catchment is underlain with discontinuous permafrost (Léger et al., 2019). 123 The upper shoulder of Teller (near Station 5, Figure 2 and Figure 5) is underlain with near-surface permafrost and appears to be a degraded peat plateau. The resultant microtopography of the degraded peat and the shallow perching 124 125 horizon caused by the permafrost creates a landscape of unsaturated peat mounds surrounded by ponds and saturated 126 soils. Downslope of the peat plateau, the Teller hillslope has highly variable soil moisture and vegetation (Table 1). 127 The microtopography within the lower footslope looks similar to the upper shoulder, but the peat appears more 128 severely degraded and the cause of the perched water table is less clear. Léger et al. (2019) suggest the presence of 129 permafrost at a depth of 1-2 m at Teller Station 9 (Figure 2), but the perching could also be caused by a layer of silt, 130 at a depth of approximately 30 cm (Graham et al., 2018). The full extent of permafrost and silt in this region of the 131 catchment remains unknown, but the thaw depth in July 2018 was greater than 1 m and maintained a perched water 132 table (Philben et al., 2020), suggesting perching could be the result of silt rather than permafrost. Vegetation type, 133 moisture content, permafrost extent, and hillslope position for all Teller Stations are summarized in Table 1. 134 Kougarok differs in many ways from Teller, although both have characteristics that are typical of hillslopes on the 135 Seward Peninsula. Kougarok is a convex hillslope, with a vertical declivity of approximately 70 meters. The study 136 area at Kougarok is approximately 2.0 km<sup>2</sup>. Soil temperature measurements at Kougarok suggest that the vast majority 137 of the site is underlain by shallow continuous permafrost (Romanovsky, Cable, and Dolgikh, 2020a); Kougarok 138 Station 5 is an exception, where the permafrost is deeper (Romanovsky et al., 2020a). The upper shoulder of Kougarok is a well-drained rocky outcrop composed of metagranitic rock (Hopkins et al., 1955; Till, Dumoulin, Werdon, and 139 140 Bleick, 2011). Saturated soils are not prevalent until the footslope and the lower backslope, where Kougarok Stations 141 2, 11, 10, 1, and 6 are situated (Figure 3). The lower backslope is characterized by persistent saturation between ubiquitous tussocks, formed by the tussock cotton grass Eriophorum vaginatum. The tussock-lichen tundra at 142 143 Kougarok introduces microtopography and spatially variable saturation; in this sense, the Kougarok tussocks are 144 analogous to the peat mounds and hummocks at Teller, but on different spatial scales and formed by different 145 processes. Kougarok has numerous patches of alder shrubland in an altitudinal band within the upper backslope; it 146 should be emphasized that Teller lacks tussock-lichen tundra and alder (Alnus viridis ssp. fruticosa) shrubs that are a

the Nome-Taylor Highway northeast of Nome, Alaska. We utilized data from "intensive stations" at both Teller and

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147 feature of Kougarok. While continuous permafrost largely remains, the Kougarok site appears to be undergoing 148 environmental changes as evidenced by an increase in alder coverage over the past decades (Salmon et al., 2019). Soil 149 profiles underneath the alder patches are rocky with shallow bedrock and warmer permafrost (Table 2). Shrub tundra 150 (alder savanna in tussock tundra and willow-birch tundra) dominates the lower backslope, where the annual active layer thickness is typically less than 100 cm. Vegetation type, moisture content, permafrost extent, and hillslope 151 152 position at all Kougarok stations are summarized in Table 2.

#### 2.2 Sampling & Analytical Approach

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SPWs were sampled using two complimentary techniques. Fiberglass wicks (Frisbee et al., Phillips, Campbell, and Hendrickx, 2010) were deployed in the upper 30 cm of soils at stations where shallow soils were unsaturated. These wicks were left in place from year-to-year and only replaced if damage was observed or suspected. The sample reservoirs from the wicks were collected whenever possible, usually a few times each summer. MacroRhizons (Rhizosphere Research Products; Netherlands) were used at stations that were more saturated, also targeting the upper 30 cm of soils. Both techniques were used at stations of intermediate saturation, where both could be deployed effectively. MacroRhizons represent a relatively discrete temporal sampling event (minutes to hours), whereas wicks represent a cumulative water collected over longer periods (weeks to months). It is in this sense, that the two techniques are complimentary. Unfortunately, due to saturation variability both techniques could not be used at all stations and conditions at some Kougarok stations were sometimes too dry to collect meaningful volumes of SPW using either method. Additional SPW data from Kougarok were supplemented from a separate study focused on alder-related nutrient dynamics (McCaully et al., 2022In Review.). These data were collected by MacroRhizons and are captured as Kougarok Stations 10 - 13, which were not part of the original stations established by the NGEE Arctic Program. A total of 309 SPW samples from Kougarok were collected and analysed, whereas a total of 89 SPW samples from Teller were collected and analysed.

After collection, SPW cation concentrations were measured in triplicate by inductively coupled plasma optical emission spectroscopy (Optima 2100 DV; PerkinElmer, USA) following US EPA Method 200.7. Inorganic anion concentrations were measured by ion chromatography (DX-600; Dionex, USA) following US EPA Method 300.0. B, F, K, Na, and Si concentrations collected by wicks were excluded from the dataset due to known issues with these

173 ions leeching from fiberglass wick samplers (Perdrial et al., 2014; Wallenberger and Bingham, 2009). This effect is 174 illustrated in Supplementary Figure 1 and the lack of such an effect for divalent cations is shown in Supplementary

Figure 2. Comparison of data from wicks and MacroRhizons, along with the observations from (Perdrial et al., 2014),

176 demonstrates that remaining constituents discussed herein were not affected by collection with fiberglass wicks. 177

Alkalinity, pH, and EH are all critical geochemical parameters that are susceptible to change during storage (Petrone

178 ct al., , Hinzman, Shibata, Jones, and Boone, 2007); because of the large amount of data from wicks these parameters

179 were not considered further, except in the context of thermodynamic modelling.

Observations related to vegetation, soil moisture, and permafrost extend were compiled from datasets made available

by the NGEE Arctic project and are given for Teller in Table 1 and for Kougarok in Table 2. The reported soil

182 moisture contents were derived from an average of gravimetric measurements (2017 and 2018) and time domain Formatted: Font: Bold

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reflectometry measurements (2017 and 2019), and from remotely-sensed P-band Synthetic Aperture Radar (2017). 183 184 End-of-winter snow depths were measured in March and April of 2016, 2017, and 2018. The annual average ground temperature was measured using in-situ temperature sensors (HOBO U30 DataLogger) at a depth of 1.5 meters below 185 the ground surface (Romanovsky et al., 2020a; Romanovsky, Cable, and Dolgikh, 2020b) and the active layer 186 187 thicknesses were determined by frost probe in September 2019 at the end of the growing season. Vegetation data were 188 collected at the peak of the growing season in mid to late July 2016 and 2017 at the NGEE Arctic Kougarok and Teller 189 field sites, respectively. The distribution of plant communities in the Arctic is primarily controlled by landscape, 190 topography, soil chemistry, soil moisture, and the plants that historically colonized an area (Raynolds et al., 2019). 191 Soil available rooting depth, which can be limited by shallow depths to bedrock, permafrost, or the water table, can 192 also restrict plant growth and survival of certain species by reducing access to water and nutrients. We surveyed the 193 dominant plant communities along each hillslope, which varied in their shrub abundance, canopy height, and structure, 194 to characterize the vegetation composition at the sites following the recommended protocol of Walker et al. (2016). 195 Extensive field site details and vegetation sampling methods are more thoroughly described in previous studies 196 (Salmon et al., 2019; Langford et al., 2019; Yang et al., 2020; Sulman et al., 2021; Yang et al. 2021). 197 For this study, we provide summary statistics for vegetation plots associated with intensive stations. Vegetation 198 composition plots within each intensive station were chosen subjectively in areas of homogeneous and representative 199 vegetation varying in size from 1 to 25 m<sup>2</sup> depending on canopy structure and height. The surveyed plot area was 1 × 200 1 m for all plant communities except for the taller stature willow-birch tundra, mesic willow shrubland (2.5 × 2.5 m), 201 and alder shrubland (5 × 5 m). For each plot, all plant species (vascular plants, lichens, and bryophytes) were recorded 202 along with visual estimates of their percent cover. For plots with multiple canopies, field cover estimates were recorded 203 as absolute cover, meaning that the total cover per plot can be >100%. We calculated relative cover values (adding to 204 100%) from the field data and use these for all subsequent analyses. 205 Plant species were further aggregated into nine plant functional types (PFTs), groupings of plant species that share 206 similar growth forms and roles in ecosystem function (Wullschleger et al., 2014), based on growth patterns and plant 207 traits. PFTs in this study include: (1) nonvascular mosses and lichens, (2) deciduous and evergreen shrubs of various 208 height classes, including an alder PFT, (3) graminoids, and (4) forbs. Photos of representative PFTs from both sites 209 are given in Supplementary Figures 9-17. Canopy height was estimated within each plot for each PFT as the average 210 of 4 measurements, including a maximum canopy height. Active layer depth was measured at the end of the growing 211 season for all plots in September 2018 using a frost probe. A temperature probe was used to determine if the resistive 212 layer was permafrost (<0 °C) or rock (>2 °C). Thaw depth is an average of 4 measurements from the vegetation plot 213 corners.

### 2.3 Statistical Analysis

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Principal Components Analysis (PCA) and the Mann-Whitney U-Test (MWUT) were both used to investigate dominant environmental controls on solute concentrations in SPWs at Teller and Kougarok. PCA is an exploratory data analysis tool that reduces the dimensionality of large complex datasets and considers how components (i.e. solute concentrations) vary together. Because PCA was predominately used as a screening tool to reveal geochemical

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correlations that may not have been evident by traditional geochemical causations or inference, a detailed discussion of the PCA results is reserved to the Supplementary Materials. The MWUT was used to test for significant differences in solute concentrations between Teller and Kougarok (inter-site variability) and between stations at each site (intrasite variability). The MWUT is a non-parametric method of challenging a null hypothesis, which in this case is the assumption that the concentrations of a given solute are not systematically greater at either site nor at any particular station. Water chemistry data are typically not normally distributed and thus, non-parametric difference tests such as the MWUT are preferred. The MWUT challenges the distribution of values, not the means. In this work, the level of significance associated with the null hypothesis was operationally defined as 0.05, which equates to a 95 % chance that an observed statistical difference is real and not coincidental. This error rate is operationally defined per contrast (i.e. a 95 % chance that the observed statistical difference in nitrate concentrations between Teller Station 9 and Teller Station 7 is real or that the observed statistical difference in sulphate concentrations between Teller and Kougarok is real) as opposed to familywise (i.e. a 95 % chance that all of the observed/reported statistical differences are real and not coincidental). MWUTs were completed using the methods described in Corder and Foreman (2009) and PCA was completed using packages available in R statistical software, version 3.3.6 (Corder and Foreman, 2009; R Core Team, 2020). For all analyses, concentrations below the method detection limit were operationally defined as half the detection limit, in agreeance with (Helsel, 2005, p. 43). While the emphasis of this study was on site/station (i.e. spatial) variability, it should be recognized that seasonal and inter-annual variability could also be significant. To minimize seasonal forcing on the variability observed, all SPW geochemical data presented were collected during the thaw season between June and September.

#### 2.4 Thermodynamic Modelling

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To investigate thermodynamic controls on solute behaviour, particularly solubility limitations, thermodynamic modelling exercises were undertaken using PHREEQC, a thermodynamic geochemical modelling code, and PhreePlot, which facilitates repetitive PHREEQC calculations through looping (Kinniburgh and Cooper, 2011; Parkhurst and Appelo, 2013). Because this study was focused on elucidating the primary geochemical controls on solute concentrations in SPWs and not on developing a rigorous transport model, representative concentrations were used instead of station specific concentrations. Representative "low", "median", and "high" concentration conditions were proxied from the 25th, 50th, and 100th concentration percentiles, respectively, taken from both Teller and Kougarok (Supplementary Table 4). Meanwhile, representative pH and E<sub>H</sub> ranges were determined either through direct measurement (pH), or indirectly by correlating dissolved Fe2+ concentrations and pH with a redox condition through geochemical models and the Nernst equation. Modelling exercises were performed at 25 °C utilizing the phreeqc.dat database, with the only modification being the suppression of methane production by inorganic carbonate reduction. Modelling exercises were performed at the default PHREEQC modelling temperature (25 °C), as the selection of an alternative defensible temperature was non-trivial; temperatures on the Seward Peninsula span a very wide range and its unclear what temperature would be most suitable for mineral solubility limitation modelling. Ultimately, because the thermodynamic models were used as a tool understand what could be controlling soil pore water solute concentrations and were not intended to model the system or to predict future concentrations, the default

256 the differences in predicted solubility between 4 °C and 25 °C did not impact the interpretation of our results 257 (Supplementary Figure 8). Methane production was "turned-off" to maintain carbonate availability under reducing Formatted: Font: Bold 258 conditions to help identify any possible carbonate minerals that could be precipitating. Because alkalinity was only 259 measured in a small number of samples, carbonate concentration percentiles were estimated from charge imbalances. 260 Alkalinity and charge imbalance were very well correlated in samples where alkalinity was measured 261 (Supplementary Figure 3). Although not a particularly rigorous modelling exercise, this approach was sufficient to Formatted: Font: Bold 262 identify mineral phases that could be controlling solute generation processes through solubility limitations. 263 3. Results 264 3.1 Physical Characteristics of Stations (Co-Located Studies) 265 Controls on the observed spatial variability of SWP solute concentrations at Teller and Kougarok stations were 266 deduced, in part, from differences in physical features and conditions of each station. Quantitative measures of many 267 of these physical characteristics were available from the interdisciplinary studies co-located at the Teller and Kougarok 268 stations. The extent of permafrost, ground temperature, active layer depth, soil moisture content, snow depth, 269 vegetation type, vegetation canopy height, dominant plant functional type, and hillslope position were all complied 270 from these co-located studies. Using these measures, the physical characteristics of each station are summarized in 271 Table 1 and Table 2, grouped by vegetation type. 272 3.12 Inter-site Variability: Teller versus Kougarok 273 Mann-Whitney U-Testing revealed that the concentrations of 14 of the 23 constituents analysed were significantly 274 different (1.96 < |z|) between Teller and Kougarok (Table 3). The effect size, a measure of how significantly different 275 the concentrations were, were large for Na and F; medium-large for K and Si; medium for Al, Oxalate, B, Zn, SO4SO42-276 , Fe, Ba, Ti, and NO<sub>2</sub>; and small-medium for Li. The terminology and thresholds for these semi-quantitative Formatted: Subscript 277 differences in correlation were taken from Corder and Foreman (2009). Mann-Whitney U-testing revealed that SPW Formatted: Superscript 278 concentrations of many constituents were significantly different between Teller and Kougarok (Table 3). When 279 concentrations were significantly different between the sites, Kougarok generally exhibited the higher concentrations 280 of the two. SPW concentrations of Na, F, K, Si, Al, oxalate, B, Zn, Fe, Ba, Ti, NO2, and Li were all significantly 281 greater at Kougarok than Teller, while only SO<sub>4</sub><sup>2</sup> concentrations were significantly greater at Teller. Meanwhile, the 282 concentrations of Br, NO3, Sr, PO4, Mg, Cr, Mn, Cl, and Ca were not significantly different between Teller and 283 Kougarok. When concentrations were significantly different between the sites, Kougarok generally exhibited the 284 higher concentrations of the two. Only SO4 concentrations were significantly greater at Teller. The concentrations of 285 Br, NO<sub>2</sub>, Sr, PO<sub>4</sub>, Mg, Cr, Mn, Cl and Ca were not significantly different between Teller and Kougarok. A summary 286 of the inter-site MWUT results are given in Table 3 with the constituents that exhibited significant differences between 287 the sites displayed over a darkened background.

temperature was decided to be the most suitable. While there is some temperature dependence of mineral solubility,

## 3.23 Intra-site Variability: Teller and Kougarok Stations

- 289 Mann-Whitney U-Testing was also used to test for intra-site differences between stations at both Teller and Kougarok. 290 Boxplots and compact letter displays are used to visualize the within-site variability of a select group of constituents of interest (COIs), which are given in Figure 4. Tables of the results of the intra-site MWUTs for all constituents that 291
- 292 were monitored, including those that did not demonstrate some systematic inter-station variability or were not
- 293 otherwise of interest, are given in the Supplementary Materials.

#### 294 3.3 Physical Characteristics of Stations (Co-Located Studies)

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- 298 stations. The extent of permafrost, ground temperature, active layer depth, soil moisture content, snow depth,
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#### 4. Discussion

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#### 4.1 Inter-site Variability: Teller versus Kougarok

Mann-Whitney U-testing revealed that SPW concentrations of many constituents were significantly different between Teller and Kougarok (Table 3). SPW concentrations of Na, F, K, Si, Al, oxalate, B, Zn, Fe, Ba, Ti, NO2, and Li were all significantly greater at Kougarok than Teller, while only SO4-concentrations were significantly greater at Teller. Meanwhile, the concentrations of Br, NO<sub>3</sub>, Sr, PO<sub>4</sub>, Mg, Cr, Mn, Cl, and Ca were not significantly different between the two sites. Overall, the more frequent instance of significantly greater constituent concentrations at Kougarok suggests a systematic cause. The extensive low-gradient toeslope (Figure 2) and lack of a well-defined drainage channel at Kougarok, are likely causes of the systematically higher SPW solute concentrations at Kougarok. Water perching, the result of near-surface permafrost in the lower-backslope and toeslope, increases evapotranspiration and, thus, SPW solute concentrations. Meanwhile, the lack of a drainage channel at Kougarok suggests that runoff (and therefore solute exports) is more limited than at Teller. Without a relatively rapid export mechanism such as a stream channel, solute transport is likely limited to interflow within the Kougarok hillslope over much of the thaw season, allowing weathering products to increase to significantly greater concentrations than those observed at Teller, where a well-defined drainage/export mechanism does exist. Field observations from pits at Kougarok confirm the present of interflow at the site. The exception to the general observation of elevated concentrations at Kougarok versus Teller was SO<sub>4</sub>. Although the cause of consistently higher SO<sub>4</sub> concentrations at Teller is unclear from the limited scope of this study, it seems likely to be due to a greater abundance of sulfidic bedrock material. The presence of sulfidic bedrock in the vicinity of Teller has been reported by mineral prospecting efforts (Brobst, Pinckney, and Sainsbury, 1971; Herreid, 1966; Mulligan, 1965); we are unaware of any such reports near Kougarok.

322 4.2 Intra-site Variability: Teller and Kougarok Stations Formatted: Heading 2.Sub Heading 323 Our interpretation of the major environmental controls on the observed spatial variability of SPW solute concentrations 324 between stations are shown in Table 4. Each of these controls, including vegetation effects, soil moisture and redox effects, weathering, water/soil interactions and hydrological transport effects, and mineral solubility effects, is 325 326 discussed considered in detail in the following sections. 327 34.3.1 Vegetation Effects Vegetation can influence hydrogeochemical variability directly via vegetation-induced changes to elemental cycling 328 329 and soil moisture contents, or indirectly via the secondary impacts changes in soil moisture can have on mineral 330 solubility or on the soil redox condition. The geochemical consequences of solubility and redox conditions are the 331 focus of sections to follow, thus, this section will focus on direct vegetation effects via influences on elemental cycling 332 and soil moisture via evapotranspiration and preferential trapping of snow. 333 NO<sub>3</sub>NO<sub>6</sub>: was the only COI that showed a distinct effect from vegetation via elemental cycling. - Elevated NO<sub>3</sub>NO<sub>6</sub>: Formatted: Subscript 334 Formatted: Subscript Formatted: Subscript 335 concentrations at both sites were generally low, with the exception of Kougarok Stations 3, 5, and 12, and Teller 336 Station 7 (Figure 4). Low to tall alder shrubs are the dominant vegetation type at Kougarok Stations 3 and 12 (Table Formatted: Font: Bold 337 2). Meanwhile, alders are present at Kougarok Station 5 despite the dominant vegetation type being low willow and 338 birch shrubs. Kougarok Stations 3, 5, and 12 all have a significant alder presence. Alders increase soil nitrogen through 339 a symbiotic relationship with nitrogen-fixing bacteria that reside in their root nodules, thus, an association between 340 NO<sub>3</sub>NO<sub>3</sub>= concentrations and alder vegetation is expected (Salmon et al., 2019). 341 Perhaps more noteworthy was the elevated NO<sub>3</sub> at Kougarok Station 5 and the lack of elevated NO<sub>3</sub> NO<sub>3</sub>: 342 concentrations at Kougarok Stations 1, 2, 6, 10, and 11. The vegetation type at Kougarok Stations 1, 2, 6, 10, and 11 343 is alder savanna in tussock tundra, which is a mixed graminoid-shrub tundra with shorter stature and lower density of 344 alder shrubs, yet nonetheless nitrogen input via alder derived nitrogen-fixation is anticipated to occur. The lack of 345 elevated NO3NO3 suggests either that 1) nitrogen\_fixation in alder savanna in tussock tundra is insufficient to result Formatted: Not Superscript/ Subscript 346 in an increase in NO<sub>3</sub>NO<sub>3</sub>- concentrations, 2) that the Kougarok footslope and lower backslope is very nitrogen-Formatted: Not Superscript/ Subscript 347 limited, and thus, that NO2NO3 is largely consumed by vegetation as it is fixed, or 3) that microbes in the Kougarok Formatted: Superscript 348 footslope and lower backslope rapidly denitrify the available NO<sub>3</sub>NO<sub>3</sub> as a substitute for oxygen in their metabolisms. 349 The smaller shrub size and density in the alder savanna in tussock tundra certainly results in less accumulated leaf 350 litter relative to the denser and larger alder shrubland intensive stations, as such, it seems reasonable that less nitrogen 351 would be available at stations in alder savanna in tussock tundra. Meanwhile, isotopic measurements of nitrogen

downslope of alder patches at Kougarok Stations 12 and 3 also support the occurrence of denitrification (McCaully et

al., In Review2022). Therefore, we believe the lack of elevated NO<sub>3</sub>NO<sub>3</sub>- concentrations at Kougarok Stations 1, 2, 6,

10, and 11 is a combination of less alder leaf litter and greater denitrification, than at Kougarok Stations 3, 5, or 12.

At Teller, only Station 7 exhibited elevated NO<sub>3</sub>NO<sub>3</sub>- concentrations relative to the rest of the catchment (Figure 4).

Teller Station 7 is dominated by tall willow shrubs and is relatively dry. Mineralization and nitrification of willow

leaf litter coupled with limited microbial denitrification is the presumed cause of elevated NO3NO3 concentrations at

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Teller Station 7. Teller Station 2 also has tall willow shrubs but did not exhibit elevated NO<sub>3</sub>NO<sub>3</sub>- concentrations. From the limited scope of this study, it is unclear why Teller Station 2 did not exhibit elevated NO2NO3- while Station 7 did, but we suspect that higher seasonal moisture content and greater microbial denitrification at Teller Station 2 likely played a role. Also of note was that despite significant intra\_site NO<sub>2</sub>NO<sub>3</sub> concentration differences, inter-site differences were not significant (|z| = 1.59) and that relatively few Kougarok stations showed elevated NO<sub>3</sub>NO<sub>3</sub>-concentrations, despite a widespread alder presence. Increased microbial denitrification is suspected to balance increased nitrogen-fixation at these stations. This is consistent with previous studies that have noted higher nitrogen mineralization rates in acidic tundra than non-acidic tundra (Weiss et al., 2005); Kougarok is predominantly acidic tundra and Teller is non-acidic tundra. The effect of vegetation on spatial variability of soil moisture was not readily observed in the volumetric moisture content of soil (Table 1 and Table 2) but was somewhat apparent in the spatial variability of moisture sensitive

The effect of vegetation on spatial variability of soil moisture was not readily observed in the volumetric moisture content of soil (**Table 1** and **Table 2**) but was somewhat apparent in the spatial variability of moisture sensitive constituents, such as Cl (**Figure 4**). The lack of a clear correlation between vegetation and soil moisture by TDR or P-band SAR observations is perhaps do to the coarseness of the P-band SAR observations and the strong seasonality associated with both methods. Moisture sensitive constituents, such as Cl, may provide a more seasonally averaged tracer of soil moisture content at the stations. An increase in Cl concentrations with vegetation canopy height was apparent at Teller stations suggesting an evapotranspiration effect. This trend was also apparent at Kougarok, but the differences were rarely significant. Overall, the spatial variability of soil moisture sensitive constituents, like Cl, was far less correlated with vegetation-type than expected; perhaps due to preferential trapping of snow, which may offset the increased evapotranspiration of tall shrubs more than previously realized. Overall, Cl concentrations at Kougarok appeared to be more correlated with hillslope position than with vegetation canopy height (**Figure 4**).

## 3.3.24.4 Soil Moisture and Redox Effects

Soil moisture content can have a profound effect on redox sensitive elements. Saturation limits oxygen diffusion into soil, and thus, forces microorganisms to utilize less energetic electron acceptors to metabolize organic matter. In an ideal system, soil microorganisms will use the strongest electron acceptor available, until it is exhausted. Although natural environments are not ideal systems, redox conditions in soils follow a more or less stepwise progression.

Therefore, by evaluating the dissolved concentrations of NO<sub>3</sub>NO<sub>3</sub>, Mn, Fe, and SO<sub>4</sub>SO<sub>4</sub> in SPWs, it is possible to

qualitatively assess soil redox conditions and their impact on hydrogeochemical variability.

The redox conditions at both Teller and Kougarok are generally limited by Fe reduction, with the most reducing conditions found at stations with the highest soil moisture content. As such, NO<sub>3</sub>NO<sub>3</sub><sup>2</sup> concentrations are generally low (Table 3), SO<sub>4</sub>SO<sub>4</sub><sup>2</sup> concentrations are relatively consistent (Figure 4), and Mn and Fe concentrations increase with increasing soil moisture (Figure 4). NO<sub>3</sub>NO<sub>3</sub><sup>2</sup> concentrations were generally low, except for drier stations in the proximity of tall alders or willows. While NO<sub>3</sub>NO<sub>3</sub><sup>2</sup> inputs are discussed in the vegetation effects section, the lack of high NO<sub>3</sub>NO<sub>3</sub><sup>2</sup> concentrations at wetter stations that contain alders suggests that soil moisture coupled with microbial denitrification bares a strong control on SPW NO<sub>3</sub>NO<sub>3</sub><sup>2</sup> concentrations. Meanwhile, SO<sub>4</sub>SO<sub>4</sub><sup>2</sup> concentrations at both sites are relatively constant across clear moisture and redox gradients (Figure 4). This suggests that SO<sub>4</sub>SO<sub>4</sub><sup>2</sup> reduction is not pervasive at either site. Dissolved Fe concentrations were higher at stations with higher soil moisture

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content, consistent with Fe reduction. Similarly, Mn concentrations were slightly elevated at wetter stations. The concentrations of Mn, however, rarely rose above 0.05 mg·L 1, suggesting either Mn solubility limitations or a lack of a significant Mn weathering source. Low Mn concentrations at Teller Station 5, a wetter station on the upper shoulder of the Teller watershed (**Table 1**; **Figure 2**) seems to support the latter conclusion, as do geochemical modelling exercises (Section 4.5). Together, these results suggest that the most reducing condition at both sites is typically limited to Fe reduction and that this only occurs at stations with the highest soil moisture contents.

#### 4.43.3.3 Weathering, Water/Soil Interaction, and Hydrological Transport Effects

A combination of weathering, water/soil interactions, and hydrological transport were elear-identified as probable drivers of hydrogeochemical variability for some solutes. As noted by Philben et al. (2020), soil derived solutes tend to accumulate in low-lying areas within watersheds. This is elearly seenobscrved at Teller, where the concentrations of Ca, Sr, and Mg all increase dramatically at the transition from lower backslope to footslope (Figure 5). Both Teller and Kougarok are underlain by carbonate-rich metamorphic facies, and Ca, Sr, and Mg are probable carbonate countercations. Weathering of Ca, Sr, and Mg carbonates and subsequent transport of these cations downslope explains this pattern of spatial variability. At Kougarok, concentrations of Ca, Mg, and Sr similarly increase from upper backslope to footslope, but concentrations of Ca and Sr decrease further down the lower backslope (Stations 10 and 1), while Mg concentrations continue to increase. A possible explanation for this behaviour is the greater affinity of cation exchange surfaces for Ca and Sr compared to Mg, thus, Ca and Sr are preferentially retained in the footslope whilst Mg is transported further down the lower backslope (Sparks, 2003, p. 189).

# 4.53.3.4 Mineral Solubility Effects

Although redox reactions are rarely at equilibrium in natural environments, comparison of field data with equilibrium models provides valuable semi-quantitative insight into the redox condition of natural environments. Because Fe appeared to be limiting the development of more reducing conditions (Section 4.3), select samples from both sites were measured for soluble Fe<sup>2+</sup> following methods presented in Viollier et al. (2000). These concentrations of aqueous Fe2+ were then compared with model-predicted concentrations of Fe2+, when coupled with an infinite Fe(OH)3(am) phase, across a range of pH values (2 - 10) and fixed E<sub>H</sub> values of 400 mV, 200 mV, 0 mV, and -200 mV; activity coefficients were assumed to be equal to 1. The measured and modelled Fe2+ concentrations are compared in Figure 6, where concentrations that were below the method detection limit (0.05 mg·L-1) are set equal to 0.025 mg·L-1 (half the detection limit). Comparison of model predicted Fe<sup>2+</sup> concentrations with field data suggests that while Teller exhibits a narrower range of pH conditions than Kougarok, it exhibits a broader range of redox conditions (Figure 6). Although several Fe<sup>2+</sup> measurements were below the detection limit, suggesting oxidizing conditions, high Fe<sup>2+</sup> concentrations in some samples suggested EH values below 0 mV. Therefore, Fe redox conditions at Teller ranged from mildly reducing to oxic and Fe redox conditions at Kougarok ranged from mildly oxic to oxic. Oxidationreduction potentials (ORPs), calculated from pH, Fe2+ concentrations, and the Nernst equation suggest that ORPs at Teller were as low as - 69 mV, while the lowest ORP at Kougarok was + 134 mV (Figure 6). Maximum ORP values could not be determined quantitatively as some Fe<sup>2+</sup> concentrations were below Fe<sup>2+</sup> detection limits, at both sites.

E<sub>H</sub>/pH predominance diagrams were created from the 25th, 50th, and 100th concentration percentiles and are shown in Figure 7 for the COIs where precipitation of mineral phases were predicted under some conditions. The concentrations for these diagrams were taken from filtered aqueous concentration data, thus, predicted mineral precipitation is an indication of nearly saturated or over-saturated conditions. The range of EH and pH conditions observed at Teller and Kougarok are overlaid as solid yellow and solid blue lines, respectively. Only the predominance diagrams that indicated possible mineral formation under the E<sub>H</sub>/pH conditions present at either site are shown in Figure 7. These phases included Fe(OH)3(am) (Fe), siderite (Fe), Al(OH)3(am) (Al), chalcedony (Si), barite (Ba and SO<sub>4</sub>), calcite (Ca), dolomite (Ca and Mg), and rhodochrosite (Mn). Predominance diagrams for the remaining key COIs that were not predicted to form any mineral phases under any site conditions are given in Supplementary Figure -To further examine which mineral phases could be controlling solute generation processes SPW solute concentrations, saturated conditions for the mineral phases identified in Figure 7 were modelled using sweeps of pH values from 2 -10 at various fixed E<sub>H</sub> values (400mV, 200mV, 0mV, and -200mV). Predicted solute concentrations under the modelled saturated conditions were then compared with field data to find common trends. In general, if solute concentrations were frequently measured near the saturation of a mineral, or were identified to have similar dependence on pH or EH, it was inferred that the mineral phase could be controlling the generation of that solute. The mineral phases that were identified to possibly be controlling solute concentrations were Al(OH)3(am), Fe(OH)3(am), chalcedony, and barite. This does not preclude the presence of significant concentrations of other mineral phases, it only identifies these as likely possibly controlling the dissolved concentrations of Al, Fe, Si, and Ba, respectively. Although it does not provide mineralogical information, X-ray fluorescence (XRF) data reported by another study at Teller confirmed high concentrations of Al, Fe, Si, and Ba in the organic and mineral soil layers at that site (Graham et al., 2018). We are unaware of any similar studies at Kougarok, nor are we aware of any studies that provide would provide confirmatory mineralogical information, for example by X-ray diffraction (XRD). Aluminium concentrations in SPWs at both Teller and Kougarok appear to be controlled by the dissolution/precipitation of amorphous Al hydroxide (Al(OH)3(am)) (Figure 8). The solubility limit of Al(OH)3(am) has no redox dependence, but is highly pH dependent. Aluminium concentrations were generally clustered near the solubility limit of Al(OH)<sub>3(am)</sub>; Al(OH)<sub>3(am)</sub> +  $3H^+ \leftrightarrow Al^{3+} + 3H_2O$ ; log k = 10.8. This suggests that Al SPW concentrations at both sites are controlled by wetting/drying (dissolution/precipitation) processes. It also suggests that there could be a significant amount of Al(OH)3(am) in the soils at both sites. While organic matter may also sorb to alumina surfaces, the adherence to the solubility of Al(OH)3(am) suggests that significant concentrations of Al are not complexed with dissolved organic matter. The predominance diagrams highlight 1) the strong pH dependence on the stability of Al(OH)3(am), 2) the influence of dissolved F can have on Al speciation when Al concentrations are low, and 3) that Al is a cation at low pH and an anion at high pH (Figure 7). Despite being a weathering product, Al concentrations show a dissimilar downslope trend to other weathering products, especially at Teller (Supplementary Figure 5). While the concentrations of weathering products generally increase with distance downslope, Al

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concentrations decrease. We suspect this can be attributed to increasing pH with distance downslope. Philben et al.

(2020) reported a 1 pH unit increase in pH in organic soils along the Teller transect (Figure 2), increasing from 5.6 at

Station 5 to 6.7 at Station 9. Such an increase would decrease the solubility of Al(OH)<sub>3(am)</sub>, and thus, decrease the concentration of dissolved Al (**Figure 8**).

Similar to Al, Fe concentrations in SPWs at both Teller and Kougarok appear to be controlled by the dissolution/precipitation of amorphous Fe hydroxide (Fe(OH)<sub>3(am)</sub>). Fe concentrations were generally clustered near the solubility limit of Fe(OH)<sub>3(am)</sub> (**Figure 8**). Unlike Al(OH)<sub>3(am)</sub> however, Fe(OH)<sub>3(am)</sub> solubility is dependent on the

the solubility limit of Fe(OH)<sub>3(am)</sub> (**Figure 8**). Unlike Al(OH)<sub>3(am)</sub> however, Fe(OH)<sub>3(am)</sub> solubility is dependent on the redox condition in addition to the pH; Fe(OH)<sub>3(am)</sub> + 3H<sup>+</sup> + e-  $\leftrightarrow$  Fe<sup>2+</sup> + 3H<sub>2</sub>O; log k = 16.0 (**Figure 8**). Fe(III) is only sparingly soluble in aqueous solutions and reduction to Fe(II) significantly increases the solubility of Fe, thus, at a

given pH value higher aqueous concentrations are predicted and observed under more reducing conditions (**Figure 8**).

Iron concentrations in SPWs at both sites generally follow the pH dependence of Fe(OH)3(am) solubility (Figure 8).

This suggests that SPW concentrations of Fe at both sites are controlled by wetting/drying (dissolution/precipitation)

476 processes, coupled with the redox condition.

Si concentrations are frequently limited by the solubility of chalcedony, a very finely grained form of SiO<sub>2</sub>, which is much more soluble than quartz; SiO<sub>2</sub> + 2H<sub>2</sub>O  $\leftrightarrow$  H<sub>4</sub>SiO<sub>4</sub>; log k = - 3.55. Particularly at Kougarok, the dissolved Si concentrations, coupled with a lack of a strong pH or E<sub>H</sub> dependence, suggest a controlling influence of chalcedony. Ba concentrations also appear to be controlled by solubility, but rather than by the solubility of an oxide or a hydroxide phase, by the solubility of barite [Ba<sup>2+</sup> + SO<sub>4</sub><sup>2-</sup>  $\leftrightarrow$  BaSO<sub>4</sub>(s); log k = 9.97]. Unlike Al hydroxide or Fe hydroxide, barite solubility lacks a strong pH dependence and instead is dependent solely on the activities of Ba<sup>2+</sup> and SO<sub>4</sub><sup>2-</sup>. Unlike Ba, SO<sub>4</sub>SO<sub>4</sub><sup>2-</sup> concentrations are not limited by the solubility limit of barite and are generally higher and not well correlated with Ba concentrations. Together, these suggest that SO<sub>4</sub>SO<sub>4</sub><sup>2-</sup> from another source (likely, atmospheric deposition or sulfidic mineral oxidation), is suppressing barite dissolution, and thus, is reducing dissolved Ba concentrations. Barite solubility can exhibit a redox dependence if conditions are sufficiently reducing to reduce SO<sub>4</sub>SO<sub>4</sub><sup>2-</sup> to sulphide. This shifts the equilibrium to greater dissolution of barite, and therefore higher conditions of Ba. The lack of E<sub>H</sub> dependence in observational data further suggests that neither site exhibits significant SO<sub>4</sub>SO<sub>4</sub><sup>2-</sup> reduction.

# 54. Conclusions Discussion

The 18 stations examined herein (8 at Teller and 10 at Kougarok) represent a wide range of vegetation types, soil moisture contents, permafrost extents, and hillslope positions. Coupling the spatial variability of these landscape characteristics with the spatial variability of SPW solute concentrations provides valuable insight into the dominant environmental controls on observed spatial variability of SPW geochemistry. It is our hope that correlating SPW geochemistry with readily observable and scalable landscape features will inform earth system modelling efforts in permafrost regions and provide fast and easy methods to determine if earth system models are working properly (i.e. predicting the correct trends). The inferred dominant environmental controls on the observed inter-site and intra-site variability of SPW solute concentrations are discussion in the following sections.

#### 4.1 The Dominant Environmental Controls on Inter-site Variability of SPW Solute Concentrations

Overall, the more frequent instance of significantly greater constituent concentrations at Kougarok suggests a systematic cause. The extensive low-gradient toeslope (Figure 2) and lack of a well-defined drainage channel at Kougarok, are likely causes of the systematically higher SPW solute concentrations at Kougarok. Water perching, the result of near-surface permafrost in the lower-backslope and toeslope, increases evapotranspiration and, thus, SPW solute concentrations. Meanwhile, the lack of a drainage channel at Kougarok suggests that runoff (and therefore solute exports) is more limited than at Teller. Without a relatively rapid export mechanism such as a stream channel, solute transport is likely limited to interflow within the Kougarok hillslope over much of the thaw season, allowing weathering products to increase to significantly greater concentrations than those observed at Teller, where a well-defined drainage/export mechanism does exist. Field observations from pits at Kougarok confirm observable interflow at the site. Overall, our study suggests that evaporative concentration could be a significant control on SPW solute concentrations in permafrost catchments, especially in those with limited drainage and therefore a perched near-surface water table. This effect has been reported previously (Raudina et al., 2017), but does not appear to be widely considered, perhaps due to the generally few studies of SPW solutes in permafrost regions. We suggest future efforts to predict future SPW solute and nutrient dynamics directly address the impacts of evaporative concentration on permafrost catchments, especially with future permafrost thaw.

The exception to the general observation of elevated concentrations at Kougarok versus Teller was SO<sub>4</sub><sup>2</sup>. Although the cause of consistently higher SO<sub>4</sub><sup>2</sup> concentrations at Teller is unclear from the limited scope of this study, it seems likely to be due to a greater abundance of sulfidic bedrock material. The presence of sulfidic bedrock in the vicinity of Teller has been reported by mineral prospecting efforts (Brobst, Pinckney, and Sainsbury, 1971; Herreid, 1966;

518 Mulligan, 1965); we are unaware of any such reports near Kougarok.

45.12 The Dominant Environmental Controls on Intra-site Spatial Variability of SPW Solute Concentrations

The 18 stations examined herein (8 at Teller and 10 at Kougarok) were selected to represent a wide range of vegetation types, soil moisture contents, permafrost extents, and hillslope positions. Coupling the spatial variability of these landscape characteristics with the spatial variability of SPW solute concentrations provides valuable insight into the dominant environmental controls on observed spatial variability of SPW geochemistry.

With regard to our initial hypotheses, our major findings are that:

Vegetation influences on elemental cycles were only readily apparent for nitrogen and although vegetation induced changes to soil moisture content were discernible, they were far less significant than anticipated. NO<sub>4</sub>NO<sub>2</sub>: was the only COI that exhibited a clear vegetation effect; elevated concentrations were associated with the presence of alder shrubs and, in some cases, tall willow shrubs. These increases in NO<sub>4</sub>NO<sub>2</sub>: concentrations associated with alder nitrogen-fixation and the mineralization and nitrification of willow leaf litter were frequently equipoised by increased

microbial denitrification in regions sufficiently moist to support it, this is perhaps one of the most significant finding of this work. -Although both Kougarok and Teller exhibited some indications of increased Cl concentrations in the

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presence of tall shrubs, the net vegetation effect on soil moisture was far less than hypothesized. Redox sensitivity was also less then hypothesized and most stations seemed well-buffered at Fe redox conditions. The result of this buffering was generally low NO3NO3 concentrations (except where vegetation effects dominated), consistent SO<sub>4</sub>SO<sub>4</sub>2 concentrations across clear redox gradients, and variable Mn and Fe concentrations. Mn concentrations were generally low, likely due to a limited source. Fe concentrations were higher at stations with higher soil moisture content, consistent with Fe reduction. Similar Fe redox cycling between soluble Fe(II) species and precipitated Fe oxyhydroxides in permafrost catchments has been reported recently (Patzner et al., 2022), which suggests that Fe redox buffering in permafrost landscapes is widespread. Weathering, water/soil interactions, and hydrological transport were elear-probable drivers of variability for Ca, Sr, and Mg. Ca, Sr, and Mg all tended to accumulate in low-lying areas, although Ca and Sr demonstrated greater accumulation potential than Mg, likely via greater affinity of cation exchange surfaces for Ca and Sr compared to Mg. Mineral solubility limitations were the primary controls on Al (Al(OH)<sub>3(am)</sub>), Fe (Fe(OH)<sub>3(am)</sub>), Ba (barite), and Si (chalcedony) concentrations. This suggests that the SPW concentrations of these constituents will remain stable until those mineral phases are exhausted or soil pore hydrochemistry changes sufficiently to alter the solubility of those mineral phases. Supersaturation of Al with respect to gibbsite (crystalline Al(OHa) and Si with respect to chalcedony in a permafrost wetland has been reported previously (Jesson et al., 2014). The solubility curves for gibbsite and Al(OH)<sub>3(am)</sub> are similar, with Al(OH)<sub>3(am)</sub> being slightly more soluble at all pH values do to the increased thermodynamic stability of the crystalline Al hydroxide

Fe(OH)3(am) solubility.

Although discerning the environmental controls on spatial variability of SPW solute concentrations provides some high-level insight into the effects changes in landscape character may have on soil pore hydrochemistry, our scope was limited and leveraged on previously available datasets. The significance of SPW in small Arctic headwater catchments as a key initial component in the freshwater hydrologic continuum is under recognized, and such

mineral, gibbsite. Meanwhile, seasonal precipitation of Fe oxyhydroxides in permafrost peatlands and their effect of

carbon cycling was the subject of an excellent paper by Patzner et al. (2022). Our study is the first observation we are

aware of the saturation controls of barite on Ba in permafrost SPWs, although that could be because few relatively

few studies consider barium concentrations; it is worthwhile noting the Ba was not supersaturated with respect to

barite but approached a saturated condition. Future studies should also note that Cchanges in redox condition would

significantly alter Fe(OH)3(am) solubility, whereas changes in pH conditions would significantly alter Al(OH)3(am) and

catchments warrant more detailed and systematic investigations.

# 563 **56.** Acknowledgements

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- their assistance with laboratory analyses.

## 570 **76**. Data availability statement

- 571 The data that support the findings of this study are made openly available in the NGEE-Arctic data repository at (DOI:
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# **98**. Figures

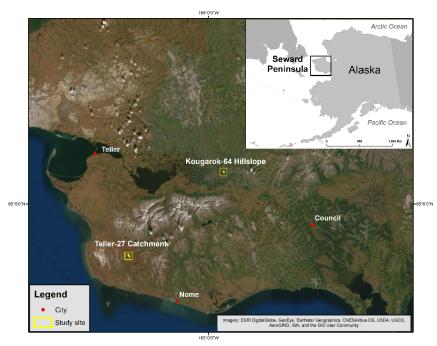


Figure 1. Location of the Teller and Kougarok field sites with respect to the municipalities of Teller, Nome, and Council. All are located on the Seward Peninsula in northwestern Alaska. RGB composite imagery from the 8-band WorldView-2 imagery obtained on July 14, 2017 at 1.5 m resolution downloaded from the DigitalGlobe website (https://www.digitalglobe.com/).

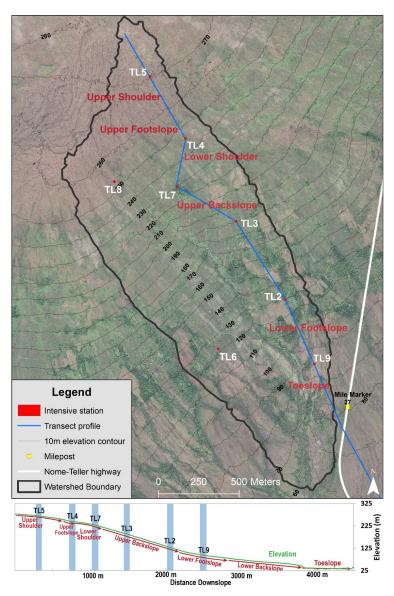
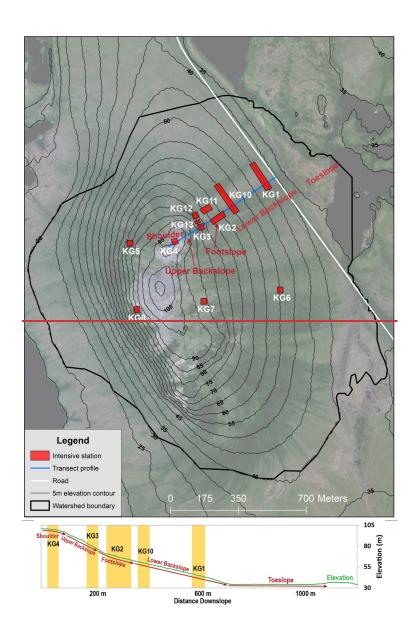


Figure 2. Topographic map of Teller. Station areas are shown as red polygons and the topographic station transect is given as a solid blue line. The hillslope transect elevation profile is given below the map in green, with stations along the transect in blue and hillslope positions noted with red arrows and text. RGB composite imagery from the 8-band WorldView-2 imagery obtained on July 27, 2011 at 1.5 m resolution downloaded from the DigitalGlobe website (https://www.digitalglobe.com/).



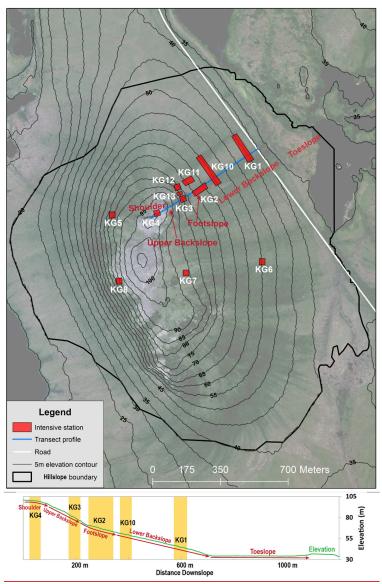


Figure 3. Topographic map of Kougarok. Station areas are shown as red polygons and the station transect is given as a solid blue line. The transect elevation profile is given below the map in green, with stations along the transect in yellow and hillslope positions noted with red arrows and text. RGB composite imagery from the 8-band WorldView-2 imagery obtained on July 14, 2017 at 1.5 m resolution downloaded from the DigitalGlobe website (https://www.digitalglobe.com/).

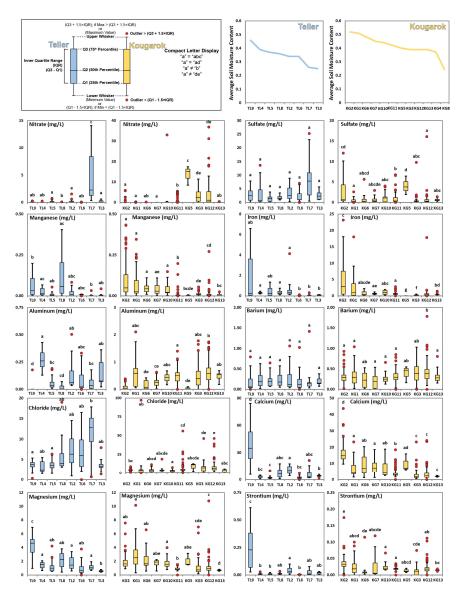


Figure 4. Mean COI concentrations at Teller (blue) and Kougarok (yellow) stations. Stations are arranged (left to right) by soil moisture content determined by P-Band SAR (top right). Boxplots show the first, second, and third data quartiles, with box whiskers representing either 150% of the inner quartile range (IQR), or the maximum or minimum value, when that value was less than  $1.5 \times IQR$ . Red circles represent data points outside of the  $1.5 \times IQR$  whiskers (i.e. outliers). Note that the concentration scales on the Teller and Kougarok plots often differ.

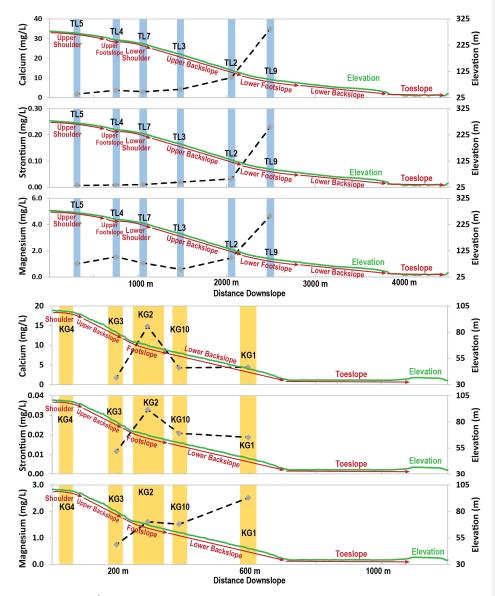


Figure 5. Median (50th percentile) concentrations (grey diamonds with dashed black lines) of Ca, Sr, and Mg, with distance downslope at Teller (blue) and Kougarok (yellow) along topographic transects; areas of stations are indicated by blue and yellow colouring, respectively. The elevation profiles of the hillslopes are plotted in green, on separate y-axes (right axes). Topographic regions of both catchments are indicated by red arrows along the elevation gradient.



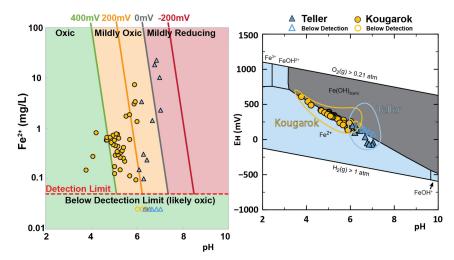


Figure 6. Left: Model-predicted  $Fe^{2+}$  concentrations in saturated solutions of  $Fe(OH)_{3(am)}$  at fixed  $E_H$  conditions of 400 mV (green), 200 mV (orange), 0 mV (grey), and -200 mV (red), compared with field concentrations of  $Fe^{2+}$  at Teller (red circles) and Kougarok (yellow circles). Right: Fe predominance diagram, showing the dominant specie of Fe under a range of  $E_H/pH$  conditions.  $E_H/pH$  regions relevant to Teller and Kougarok are outlined in blue and yellow, respectively. Samples with  $Fe^{2+}$  concentrations below the detection limit are given as colour coordinated open circles set at 0.025 mg·L<sup>-1</sup> (half the detection limit) in both sides of the figure.

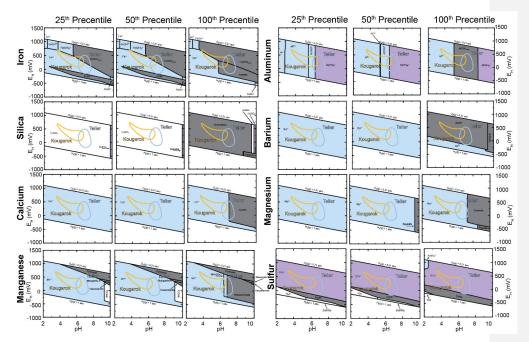


Figure 7.  $E_{\rm H}/pH$  diagrams for key species that indicated possible mineral formation under the  $E_{\rm H}/pH$  conditions present at either Teller or Kougarok. The  $E_{\rm H}$  and pH conditions observed at Teller and Kougarok are overlaid as blue and yellow lines, respectively. Mineral species (solids) are shown in grey, cations on are shown in blue, anions are shown in purple, neutral species are shown in white. Predominance diagrams were created in PhreePlot using the phreeqc.dat database, with inorganic carbonate reduction to methane "turned off."

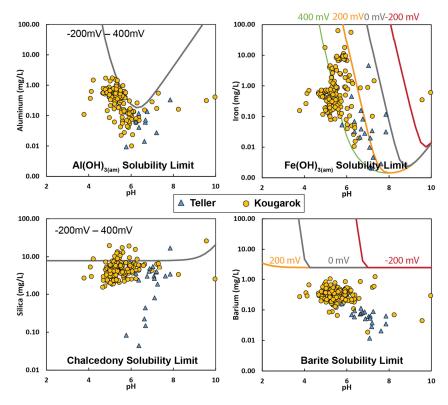


Figure 8. Modelled solute concentrations in solutions saturated with  $Al(OH)_{3(am)}$ ,  $Fe(OH)_{3(am)}$ , chalcedony, and barite, with respect to pH (x-axis) and  $E_H$  (model lines), overlaid with observed solute concentrations.

#### 850 910. Tables

#### 851 **Table 1. Teller Station Physical Characteristics**

				Veg	getation			R	elative wetne	ess	Permafrost			
		Hillslope Position	Vegetation type	Average (maximum) canopy height (cm)	Dominant PFT		Low to tall shrub cover	Average TDR soil moisture (VMC)	Average P-band SAR (VMC)	Average snow depth (cm)	Average Ground Temperature (°C)	Permafrost Extent	Average (maximum) thaw depth (cm)	
	TL9	Lower Footslope		<u>28</u> 34 (41)	Graminoid Bryophyte	44 %	10 <u>%</u>	NA	0.46	68.4	0	Marginal	101 (> <del>12</del> 0)-	
	TL5	Upper Shoulder	Wetland complex	<u>12</u> 31 (45)	Graminoid	45 %	7 <u>%</u>	0.55	0.37	103.3	-0.45	Near- surface	97° (>114°)	
ĺ	TL8	Upper Footslope		<del>1</del> 7 (34)	Bryophyte	33 %	20 <u>%</u>	0.55	0.36	77.7	-0.6		69° (>120)	
	TL3	Upper Backslope	Cassiope dwarf shrub	<del>913</del> (23)	Evergreen	<u>47 %</u>	<u>12</u> 4 <u>%</u>	NA	0.25	62.1	2.2	None/ deep	72 <sup>r</sup> (82 <sup>r</sup> )	
	TL4	Upper Footslope	tundra	<del>10</del> - <u>8</u> (14)	dwarf shrub	<u>58 %</u>	4 <u>%</u> 12	0.35	0.39	89.5	0.5	Marginal	40 <sup>r</sup> (70 <sup>r</sup> )	
İ	TL2	Upper Backslope	Mesic	84132 (141)	Deciduous low to tall	44 %	44 <u>%</u>	0.4	0.34	124	2.4	None/ deep	75° (>120)	
İ	TL7	Lower Shoulder			shrub (willow)	<u>37 %</u>	37 <u>%</u>	0.46	0.26	128.8	2.4		51 <sup>r</sup> (66 <sup>r</sup> )	
-	TL6	Upper Backslope	Willow- birch tundra	<del>74 <u>64</u> (</del> 115)	ForbDecidu ous low shrub (willow & birch)	23 %	32 <u>%</u>	0.38	0.34	86.4	1.2	None/ deep	67r <sup>r</sup> (102 <sup>r</sup> )	

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PFT – plant functional type; dwarf shrub (height <40 cm), low shrub (height 40-200 cm), low to tall shrub (height 40 to >200 cm tall). Deciduous shrub PFT classes identify the dominant species in the plant community as either willow or willow and birch. There is no alder at the T-eller site. Low to tall shrub cover represents the sum of deciduous low shrubs, deciduous low to potentially tall willow and birch, and

deciduous low to tall alder.

Single point soil moisture measurements. Data are more accurate than P-band SAR but represent a much smaller spatial scale.

P-band SAR has 30m resolution.

Resistive layer was rock; all others are permafrost. A temperature probe was used to determine if the resistive layer was permafrost (≤0 °C) or rock (>2 °C). Thaw depth is an average of 4 measurements from the vegetation plot corners within the IS and was measured at the end of the growing season.

## **Table 2. Kougarok Station Physical Characteristics**

863

			Vegetation					Relative wetness			Permafrost			
		Hillslope Position	Vegetation type	Average (maximum) canopy height (cm)	Dominant	PFT	Low to tall shrub cover	Average TDR soil moisture (VMC)	Average P-band SAR (VMC)	Average snow depth (cm)	Average Ground Temperature (°C)	Permafrost Extent	Average (maximum) thaw depth (cm)	
	KG3	Upper Backslope	Alder shrubland	204 (265)	Deciduous low to tall shrub (alder)	<u>30 %</u>	53 <u>%</u> 30	0.19	0.39	131.3	-0.01	Near- surface	48 <sup>r</sup> (53r)	
	KG12	Footslope		NA	<u>NA</u>	NA	NA	0.30*	0.39	NA	NA		NA -	Formatted: Centered
	KG1	Lower Backslope		60 (90)	Deciduous low shrub (alder, willow & birch)	<u>31 %</u>	44 <u>%</u> 13	NA	0.51	83.4	-2.5		61 (68)	Formatted Table
	KG2	Footslope	Alder savanna in	48 (73)	Graminoid	30 %	42 %20	0.63	0.52	102.3	-1.2	Near-	75 (89)	
	KG6	Lower Backslope	tussock tundra	24 (61)	Graminoid	<u>46 %</u>	17 <u>%</u> 0	0.36	0.48	66.2	-2.2	surface	58 (62)	
	KG10	Lower Backslope		NA	NA	<u>NA</u>	NA	NA*	0.44	71.4	NA		NA -	Formatted: Centered
1	KG11	Footslope		NA	<u>NA</u>	<u>NA</u>	NA	0.59*	0.42	NA	NA		NA•	Formatted Table
Ì	KG7	Upper Backslope	Tussock- lichen tundra	20 (22)	Graminoid	<u>34 %</u>	<u>14</u> <u>%</u> 0	0.51	0.45	54.7	-2.1	Near- surface	76 (100)	Formatted: Centered
I	KG4	Shoulder	Dryas-	6 (12)	Evergreen dwarf shrub	<u>62 %</u>	<u>1 %</u> 0	NA	0.37	NA	-1.9	27	0r (0r)	
	KG13	Upper Backslope	lichen shrub tundra	NA	<u>NA</u>	<u>NA</u>	NA	0.41*	0.39	92.1	NA	Near- surface	NA-	Formatted: Centered
	KG5	Upper Backslope	Willow- birch tundra	62 (137)	Deciduous low shrub (willow & birch)	<u>60 %</u>	62 <u>%</u> 3	NA	0.4	178.4	> 0	Deep	88 (96)	Formatted Table
	KG8	Upper Backslope	onen tandra	45 (120)	Evergreen dwarf shrub	<u>52 %</u>	42 <u>%</u>	0.23	0.24	85.5	-0.04	Near- surface	44 <sup>r</sup> (55 <sup>r</sup> ).	Formatted: Centered Formatted Table
86 86 86 86 87 87	864 865 Note: PFT – plant functional type. Deciduous shrub PFT classes identify the dominant species in the community as either willow, alder, willow and birch, or alder, willow, and birch. Low to tall shrub cover represents the sum of deciduous low shrubs, deciduous low to potentially tall willow and birch, and deciduous low to tall alder. 868 'Isingle point soil moisture measurements. Data are more accurate than P-band SAR but represent a much smaller spatial scale. 869 'P-band SAR has 30m resolution. 870 *Average gravimetric water content measurements, corrected to VMC by bulk density. 871 'Resistive layer was rock; all others are permafrost. A temperature probe was used to determine if the resistive layer was permafrost (≤0 °C) or rock (>2 °C). Thaw depth is an average of 4 measurements from the vegetation plot corners within the IS and was measured at the end of the growing season.										Tornatted Table			

P-band SAR has 30m resolution.

\*Average gravimetric water content measurements, corrected to VMC by bulk density.

Resistive layer was rock; all others are permafrost. A temperature probe was used to determine if the resistive layer was permafrost (≤0 °C) or rock (>2 °C). Thaw depth is an average of 4 measurements from the vegetation plot corners within the IS and was measured at the end of the growing season.

875 Table 3. Inter-Site Mann-Whitney U-Test Results

	<u>Teller</u>				Kougaro	<u>k</u>		Site with	Effect	Difference in
	n	$\textstyle\sum R_i$	$\mathbf{U}_{\mathbf{i}}$	n	$\textstyle\sum R_i$	$U_{i}$	<b>z</b>	Higher Median	Size	Correlation
Na	59	3184	14811.5	275	52761.5	1413.5	9.95	Kougarok	0.54	large
F	59	3502	14375.5	273	51776.5	1731.5	9.46	Kougarok	0.52	large
K	59	3882	14113	275	52063	2112	8.92	Kougarok	0.49	medium-large
Si	59	4119	13876.5	275	51826.5	2348.5	8.56	Kougarok	0.47	medium-large
Al	58	4952	12709	275	50659	3241	7.11	Kougarok	0.39	medium
Oxalate	57	4996	12161.5	272	49289.5	3342.5	6.75	Kougarok	0.37	medium
В	59	5429	12566.5	275	50516.5	3658.5	6.62	Kougarok	0.36	medium
Zn	58	5605	12056	275	50006	3894	6.12	Kougarok	0.34	medium
SO <sub>4</sub>	58	13653	3892.5	273	41293.5	11941.5	6.08	Teller	0.33	medium
Fe	58	5958	11703	275	49653	4247	5.60	Kougarok	0.31	medium
Ba	58	6256	11405.5	275	49355.5	4544.5	5.15	Kougarok	0.28	medium
Ti	58	6266	11395.5	275	49345.5	4554.5	5.13	Kougarok	0.28	medium
NO <sub>2</sub>	54	5588	10585.5	272	47713.5	4102.5	5.12	Kougarok	0.28	medium
Li	58	7778	9883	275	47833	6067	2.86	Kougarok	0.16	small-medium
Br	58	8485	9060.5	273	46461.5	6773.5	1.73	Equal	0.09	small
$NO_3$	58	8576	8969	273	46370	6865	1.59	Kougarok	0.09	small
Sr	58	8683	8978	275	46928	6972	1.51	Kougarok	0.08	small
PO4	54	9659	6460.5	271	43316.5	8173.5	1.36	Equal	0.08	small
Mg	58	10495	7166	275	45116	8784	1.21	Teller	0.07	small
Cr	58	8884	8777	275	46727	7173	1.20	Kougarok	0.07	small
Mn	58	9164	8497	275	46447	7453	0.78	Teller	0.04	small
Cl	58	9221	8266.5	272	45394.5	7509.5	0.57	Kougarok	0.03	small
Ca	58	10016	7645	275	45595	8305	0.50	Teller	0.03	small

Table 4. Dominant Environmental Controls on SPW Geochemistry at Teller and Kougarok

Environmental Control	Analytes Affected	
Vegetation	<u>NO₃NO₃</u> -	 Formatted: Subscript
Soil Moisture/Redox	NO <sub>3</sub> NO <sub>3</sub> , Mn, Fe, SO <sub>4</sub> (occasionally)	 Formatted: Subscript
Water/Soil Interactions & Hydrologic Transport	Ca, Mg, Sr	
Mineral Solubility	Al, Ba, Si, Fe	