



1 Measurement report: Molecular-level investigation of

2 atmospheric cluster ions at the tropical high-altitude research

3 station Chacaltaya (5240 m a.s.l.) in the Bolivian Andes

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30 Abstract

31 Air ions are the key components for a series of atmospheric physicochemical 32 interactions, such as ion-catalyzed reactions, ion-molecule reactions, and ion-induced 33 new particle formation. They also control atmospheric electrical properties with effects 34 on global climate. We performed molecular-level measurements of cluster ions at the 35 high-altitude research station Chacaltaya (CHC; 5240 m a.s.l.), located in the Bolivian Andes, from January to May 2018 using an atmospheric pressure interface time-of-36 37 flight mass spectrometer. The negative ions mainly consisted of (H₂SO₄)₀₋₃·HSO₄⁻, 38 $(HNO_3)_{0-2}$ ·NO₃, SO₅, $(NH_3)_{1-6}$ · $(H_2SO_4)_{3-7}$ ·HSO₄, malonic acid-derived, and 39 CHO/CHON (HSO4⁻/NO3⁻) cluster ions. Their temporal variability exhibited distinct 40 diurnal and seasonal patterns due to the changes in the corresponding neutral species' 41 molecular properties (such as electron affinity and proton affinity) and concentrations 42 resulting from the air masses arriving at CHC from different source regions. The 43 positive ions were mainly composed of protonated amines and organic cluster ions, but 44 exhibited no clear diurnal variation. H2SO4-NH3 cluster ions likely contributed to the 45 new particle formation process, particularly during wet-to-dry transition period and dry 46 season when CHC was more impacted by air masses originating from source regions 47 with elevated SO₂ emissions. Our study provides new insights into the chemical 48 composition of atmospheric cluster ions and their role in new particle formation in the 49 high-altitude mountain environment of the Bolivian Andes.

50 1 Introduction

51 Air ions regulate the electrical properties of the atmosphere by serving as carriers of 52 electrical charges (Williams, 2009). They also play an important role in atmospheric 53 chemistry by participating/catalyzing ion-molecule reactions and ion-induced new 54 particle formation (NPF, Hirsikko et al., 2011). The formation of tropospheric ions is initiated through simple-structured ions, such as O⁺, N₂⁺, O⁻, and O₂⁻, mainly from 55 56 radioactive decay in the soil (e.g., radon and gamma radiation), thunderstorm activity 57 (lightning), and galactic cosmic rays (GCR). These ions can transfer their charges to 58 other compounds, leading to the subsequent production of an assortment of ions, such 59 as the bisulfate ion (HSO₄⁻), nitrate ion (NO₃⁻), hydronium ion (H₃O⁺), and ammonium ion (NH4⁺; Smith and Spanel, 1995; Hirsikko et al., 2011). Depending on their sizes, 60 61 air ions are usually classified into cluster ions (diameter ≤ 1.6 nm) that are charged 62 molecules or molecular clusters, and charged particles (diameter > 1.6 nm; Hirsikko et 63 al., 2005, 2011; Komppula et al., 2007).

Cluster ions exist almost always in the troposphere and can undergo frequent ionmolecule reactions during their lifetime (~100 seconds; Manninen et al., 2010; Hirsikko et al., 2011). Their chemical composition, in addition to the initial ionization, also depends on the concentrations of the parent neutral species (Eisele, 1986). Bianchi et al. (2017) showed that the diurnal cycle of negative organic ions followed the variations of their neutral molecules' concentrations in a boreal forest, since the higher





70 concentrations of neutral molecules would result in a larger probability of them being 71 charged. Moreover, molecular properties of the neutral species, such as electron affinity 72 (EA) and proton affinity (PA), are also important for determining cluster ion 73 composition. Cluster ions derived from molecules with higher EA (e.g., HSO4 and NO3) 74 or PA (e.g., trimethylamine (C_3H_9N) and pyridine (C_5H_5N)) tend to obtain the ambient 75 negative or positive charge, respectively (Ferguson and Arnold, 1981; Hirsikko et al., 76 2011). Because of the strong EA or PA, it is almost unlikely that the ions derived from 77 those molecules will further transfer their charges to other neutral compounds via ion-78 molecule reactions. Thus, these negative (HSO₄⁻ and NO₃⁻) and positive (C₃H₁₀N⁺ and 79 $C_5H_6N^+$) ions are usually more abundant than other ions in the atmosphere (Eisele, 1986; 80 Ehn et al., 2010; Bianchi et al., 2017; Frege et al., 2017). In contrast, charge transfer 81 occurs more easily for ions derived from neutral species of lower EA or PA.

82 Atmospheric cluster ions can contribute to new particle formation (NPF) via ion-83 induced nucleation (Yu, 2010). Since the discovery of this mechanism in the first cloud 84 chamber study in the early 1900s (Wilson, 1911), ion-induced nucleation has been 85 known as an important source of atmospheric aerosol particles. Recently, a series of 86 chamber studies conducted at the CLOUD (Cosmics Leaving Outdoor Droplets) facility 87 at CERN (the European Centre for Nuclear Research) have shown that aerosol 88 nucleation rates are substantially enhanced in the presence of some specific cluster ions, 89 such as sulfuric acid – ammonia ($H_2SO_4 - NH_3$) cluster ions (Kirkby et al., 2011; 90 Schobesberger et al., 2015), pure H₂SO₄ cluster ions (Kirkby et al., 2011), and organic 91 cluster ions (Kirkby et al., 2016). Field measurements have also suggested the important 92 role of atmospheric ions in ion-induced nucleation (Manninen et al., 2010; Hirsikko et 93 al., 2011; Rose et al., 2018; Jokinen et al., 2018; Yan et al., 2018; Beck et al., 2021). 94 Among them, the onsets of high-altitude NPF events, compared to those occurring in 95 the lower troposphere, are often associated with more abundant cluster ions (Lee et al., 96 2003; Venzac et al., 2008; Boulon et al., 2010). Such increases are due to the higher 97 GCR intensity and lower condensation sink (CS) in the high-altitude regions. As a result, potentially larger contributions of cluster ions to aerosol formation would be expected 98 99 (Smith and Spanel, 1995; Hirsikko et al., 2011).

100 However, the molecular-level understanding of ambient cluster ions and their influence 101 on NPF in high-altitude environments (in the troposphere) is still very limited. Two 102 mountaintop studies in the Alps show that, depending on the air mass origins, NPF 103 could be triggered by sulfuric acid-ammonia clusters, or nitrate (or sulfuric acid) 104 clustering with highly oxygenated organic molecules (Bianchi et al., 2016; Frege et al., 105 2017). Another study in the Himalayas found that NPF was mainly driven by organic 106 vapors of biogenic origin (Bianchi et al., 2021). Recently, frequent and intensive NPF 107 events were observed at the high-altitude research station Chacaltaya (CHC; 16.3505° 108 S, 68.1314° W; 5240 m a.s.l.) located in the Bolivian Andes (Rose et al., 2015), but the 109 exact mechanism and the role of cluster ions in aerosol nucleation process remain 110 unclear. Therefore, a detailed investigation of cluster ions at CHC, including their





- 111 molecular composition, temporal variation (diurnal and seasonal), and source regions,
- 112 is needed in order to understand their role in atmospheric processes such as NPF in the 113 study regions.
- 114 Here we present measurements of atmospheric ions from January to May 2018 at CHC. 115 The dataset is part of the Southern hemisphere high ALTitude Experiment on particle 116 Nucleation And growth (SALTENA) field experiment campaign (Bianchi et al., 2022). 117 During the study period, the sampled air masses originated from various source regions, 118 such as the Amazon Basin to the east and the Altiplano and the Pacific Ocean to the 119 west (Fig. 1a; Aliaga et al., 2021). Temporal evolution (diurnal and/or seasonal 120 variations) of both negative and positive ion composition are investigated, and their 121 potential connections with source regions and NPF are discussed. Our study thus adds 122 important observational information on a better understanding of atmospheric ions and 123 provides new insights into their role in high-altitude NPF in the troposphere of the 124 Bolivian Andes.

125 2 Methods

126 **2.1 Measurement site description**

The high-altitude research station CHC is ~140 m below the summit of Mount 127 128 Chacaltaya (5380 m a.s.l.) with an open view to the south and west (Andrade et al., 129 2015). The La Paz – El Alto metropolitan area (with 1.7 million inhabitants) is $\sim 1 - 1.6$ 130 km lower (in altitude) and ~15 km south of CHC (Fig. 1b). The seasonal meteorological 131 conditions at CHC depend on the cycle between the wet (November to March; wet-to-132 dry transition period in April) and dry (May to September; dry-to-wet transition period 133 in October) seasons driven by large-scale tropical circulation (Rose et al., 2015; Bianchi 134 et al., 2022). This pattern also affects the source regions of air masses arriving at CHC 135 (Aliaga et al., 2021). Additionally, due to the strong diurnal cycle of the planetary boundary layer (PBL) height and the thermally-induced winds in the mountainous 136 137 terrain, CHC is often affected by polluted PBL transported from the La Paz - El Alto metropolitan area during daytime (Wiedensohler et al., 2018) whereas at night CHC is 138 139 located in the residual layer or tropical free troposphere (Coen et al., 2018).





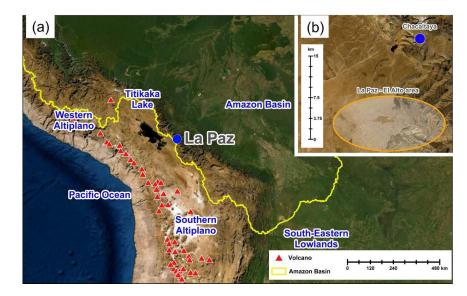




Figure 1 (a) True-color satellite image showing the location of CHC (blue circle) and its surrounding
area. The yellow line presents the boundary of the Amazon Basin. Red triangles denote the volcanoes in
this area. (b) A zoomed-in true-color satellite image showing the distance between CHC and the La Paz
- El Alto metropolitan area (orange circle). Image sources: Esri, DigitalGlobe, GeoEye, i-cubed, USDA
FSA, USGS, AEX, Getmapping, Aerogrid, IGN, IGP, swisstopo, and the GIS User Community.

146 2.2 Instrumentation

147 All the instruments involved in this study were installed in a temperature-controlled 148 measurement room (at ~ 25 °C). All data are reported in local time (UTC-4).

149 2.2.1 Measurements of atmospheric cluster ions

150 The composition of cluster ions was measured by an atmospheric pressure interface 151 time-of-flight mass spectrometer (APi-TOF, Aerodyne Research Inc. & Tofwerk AG). 152 The APi-TOF consists of an atmospheric pressure interface (APi) module and a time-153 of-flight (TOF) mass spectrometer. The APi module allows the instrument to sample 154 ions from ambient air directly. The positive or negative ions within the sampled airflow 155 are focused and guided by two quadrupoles and an ion lens, with a gradually decreasing 156 pressure (from atmospheric pressure to $\sim 10^{-4}$ mbar), before entering the TOF mass spectrometer ($\sim 10^{-6}$ mbar). A more detailed description of this instrument is given in 157 Junninen et al. (2010). In this study, ambient air was sampled through a ~1.5 m stainless 158 159 steel tube with a total sample flow of 14 standard liters per minute (SLPM) to ensure 160 laminar flow during sampling, and 0.8 SLPM of the total flow entered the APi-TOF. 161 During the wet season, the APi-TOF was first operated in negative mode to measure

negative cluster ions (January), and then switched to positive mode to measure positive

163 cluster ions (February to March). During the wet-to-dry transition period (April) and





164 dry season (May), the instrument was changed back to negative mode to investigate the 165 potential seasonality of negative ion composition. It is important to note that, similar to 166 Frege et al. (2017), for better characterization of the connection between cluster ions 167 and NPF, we only included the ion data from cloud-free days in this study (to avoid 168 influence from, e.g., lightning activity).

169 2.2.2 Measurements of H₂SO₄ and oxidized organic molecules

170 Concentrations of H2SO4 and oxidized organic molecules (OOM) were measured using 171 a nitrate ion (NO3⁻) based chemical ionization atmospheric pressure interface time-of-172 flight mass spectrometer (CI-APi-TOF, Aerodyne Research Inc. & Tofwerk AG; 173 Jokinen et al., 2012). The instrument is a combination of the APi-TOF and a chemical 174 ionization (CI) unit, which has been widely used to measure H2SO4 in the atmosphere 175 (Jokinen et al., 2012; Bianchi et al., 2016; Zha et al., 2018). In this study, a soft X-ray 176 source (L9490, Hamamatsu) was used to charge nitric acid (HNO3) in a sheath flow of 177 20 SLPM to produce the reagent ion, NO_3^- . H₂SO₄ and OOM in the sample flow (10 178 SLPM) were then charged by either proton transfer or the formation of an adduct with 179 the reagent ion during the ~ 200 ms residence time in the CI unit. A calibration factor of 1.5×10^{10} cm⁻³ for H₂SO₄ was determined (with sampling loss corrected) according to 180 the approach by Kürten et al. (2012). The same calibration coefficient was adopted for 181 182 determining OOM concentrations in this study, which could result in an underestimation of their concentrations due to a lower charging efficiency of OOM than 183 184 H_2SO_4 by NO_3^- (Hyttinen et al., 2015).

185 2.2.3 Auxiliary measurements

186 The number concentration and size distribution of atmospheric ions and neutral 187 particles were measured with a neutral cluster and air ion spectrometer (NAIS, Airel 188 Ltd., Mirme and Mirme, 2013). The instrument can detect air ions with a diameter from 189 1.4 to 50 nm, including both cluster ions and charged particles. The details of the 190 instrument used can be found in Rose et al. (2017).

191 Particle number size distributions between 10 to 500 nm were measured by a Mobility

192 Particle Size Spectrometer (MPSS; Wiedensohler et al., 2012), and the data was used

193 for calculating the CS, which represents the loss rate of condensing vapors and cluster

194 ions on pre-existing particles (Kulmala et al., 2001).

195 Meteorological parameters, such as temperature, relative humidity (RH), and global

radiation, were also measured simultaneously at CHC. Detailed descriptions can befound in Bianchi et al. (2022).

197 found in Bianchi et al. (2022).

2.3 Simulation of air mass origin and history

199 To understand the source regions and transport pathways of the air masses arriving at

200 CHC, we used the results of air mass history analysis obtained from FLEXPART-WRF

201 simulations described in Aliaga et al. (2021). In brief, a Lagrangian transport and





202 dispersion model (FLEXPART-WRF; version 3.3.2; Brioude et al., 2013) was used to 203 calculate the air mass history during the campaign period. The backward simulation 204 was driven by the high spatial and temporal resolution meteorological output from the 205 Weather Research and Forecasting model (WRF; version 4.0.3; Skamarock et al., 2019). 206 In the simulation, twenty thousand particles were continuously released every hour 207 from a 10 m deep layer (0 - 10 m a.g.l.) at CHC and traced back in the atmosphere for 96 hours. The output of the FLEXPART-WRF is the source-receptor relationship (SRR, 208 209 in seconds), which is calculated for each geographical grid cell included in the 210 simulation. The SRR value depends on the particle's residence time and the number of 211 particles in the output grid cells. Clustering analysis was conducted by applying a series 212 of pretreatments (e.g., log-polar grid transformation and grid cell pre-processing) and a 213 k-means clustering algorithm (Lloyd, 1982) to the calculated SSR dataset (see Aliaga 214 et al. (2021) for more details).

215 2.3.1 Major air mass pathways

Six air mass pathways (PW) representing air masses arriving at CHC were determined
from the clustering analysis. They are named based on their clock positions from CHC
(e.g., 03_PW indicates the pathway with its centroid located at the 3 o'clock direction
(east, 90°) of CHC, Fig. 2a). Characteristics of these air mass pathways, such as source
region, transport distance, and transport time, were distinct from each other (Table 1).
A detailed description of the air mass pathways and their characteristics can be found
in Aliaga et al. (2021).

The influence of each air mass pathway on CHC varied with time, and was estimated by its SRR percentage (SRR[%]_{pathway}) as in equation (1):

225
$$SRR[\%]_{pathway} = \frac{SRR_{pathway}}{SRR_{total}} \times 100$$
(1)

226 where SRR_{pathway} and SRR_{total} are the residence time of a specific air mass pathway and

227 in total (96 hours = 345600 seconds) in the simulation, respectively.

Table 1. Overview of the six air mass pathways extracted from Aliaga et al. (2021).

Pathway	Direction to CHC	Representative source region	Transport distance (km) ¹ Median (25 – 75 %)	Transport time (hour) Median (25 – 75 %)
03_PW	East	Amazon Basin and Eastern/South-Eastern Lowlands	518 (413–608)	51 (45-57)
05_PW	South and Southeast	South-Eastern Lowlands and Southern Altiplano	428 (303-567)	45 (36-52)
07_PW	Southwest	The Pacific Ocean, coastal area, Western Altiplano, and La Paz – El Alto	721 (577-896)	54 (45-61)
08_PW	West	Western Altiplano and	238 (198-279)	36 (29-43)





		Titicaca lake, coastal area		
11_PW	North and Northwest	Amazon Basin, Western Altiplano, coastal area	465 (326-563)	53 (46-59)
12_PW	North	Amazon Basin	76 (49 -95)	27 (21-33)

¹Distance between CHC and each pathway's center point (see Fig. 2a).

230 2.3.2 Identification of representative periods for each air mass231 pathway

232 Air mass history analysis shows that the air sampled at CHC was typically a mixture of 233 multiple pathways. Thus, the cluster ion composition observed during the study period 234 was often influenced by multiple source regions concurrently. To characterize the 235 influence of every single pathway on cluster ion composition, periods when an air mass 236 pathway exerted its largest impact on CHC (the highest 10% of its SRR[%]pathway values; 237 Fig. 2b) during the whole study period are identified as the representative periods of 238 the specific pathway. For instance, the representative periods of 03 PW (covering, e.g., 239 the Amazon Basin) are more frequently seen during wet season (highest in January), 240 whereas 08 PW (covering, e.g., Altiplano region) has most of its representative periods 241 in dry season (highest in May). Note that SRR[%]pathway of any individual pathway 242 rarely reached 40 % during the whole study period (see Fig. S1), and thus the 243 representative periods cannot be directly identified via SRR[%]pathway values (e.g., using 244 a certain threshold of the value) as in a previous study (Koenig et al., 2021).

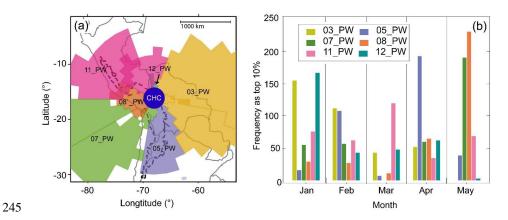


Figure 2 Influence of the six air pathways on CHC from January to May 2018. (a) Horizontal profile of
the air mass pathways, adapted from Aliaga et al. (2021). (b) Frequency of the representative periods for
each pathway (the highest 10% of their corresponding SRR[%]_{pathway}) in different months.

249 3 Results and discussion

250 **3.1 Variation in total ion count**

251 During the study period, the total ion count (TIC) observed by APi-TOF at CHC varied



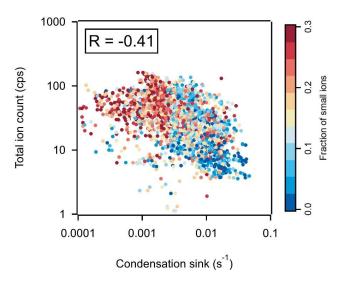


252 between <10 and 100 counts per second (cps; Fig. 3 and Fig. S2). Similar variations in 253 the TIC were also observed in the long-term cluster ion measurements at the high-254 altitude station Jungfraujoch in Switzerland (JFJ; 3454 m a.s.l.; Frege et al., 2017) and 255 were attributed to the seasonal changes of ion precursor and sink. Like the winter-256 summer seasonality of the JFJ, CHC and its adjacent mountain areas are frequently 257 covered by snow in wet season and mostly free of snow in dry season (Bianchi et al., 2022; Koenig et al., 2021). Thus, in contrast with the generally stable GCR flux 258 259 (primarily controlled by the decadal scale solar cycle; Shuman et al., 2015), a reduced 260 radioactive decay from the soil and a lower ion production rate could be expected at 261 CHC in wet season than in dry season.

262 However, the TIC measured by APi-TOF was significantly higher in wet season (41 \pm 263 23 cps, mean \pm standard deviation) and the wet-to-dry transition period (56 \pm 32 cps) 264 than in dry season (14 ± 11 cps; Fig. S2). Considering the slight negative correlation 265 (Pearson's correlation coefficient (R): -0.41; Fig. 3) between the TIC and CS (representing the loss rate of condensing vapors and cluster ions on pre-existing 266 particles; Kulmala et al., 2001), the observed TIC fluctuation may be related (at least 267 partially) to the varying CS ($\sim 1 \times 10^{-4} \text{ s}^{-1}$ in wet season to $\sim 5 \times 10^{-2} \text{ s}^{-1}$ in the dry season). 268 Moreover, the cluster ions measured by APi-TOF usually account for only a small 269 270 fraction of the total atmospheric ions (Rose et al., 2018). Changes in the fraction of 271 small ions in total atmospheric ions can potentially lead to a fluctuation in TIC (Frege 272 et al., 2017). This is illustrated in Figure 3 that a smaller TIC determined from APi-TOF 273 is associated with a lower fraction of smaller ions (< 2 nm) observed by NAIS (mostly 274 cluster ions). However, for better characterization of the influences of different ion 275 composition on CHC and their diurnal and seasonal relative changes, we normalized the observed ion signal to the TIC for APi-TOF measurements. 276







277

278 Figure 3 Correlation between the TIC measured by APi-TOF and condensation sink, colored by the

279 fraction of small ions (defined as concentrations of ions with diameter <2 nm to the total ion

280 concentrations) determined from NAIS data. Data are shown in the time resolution of 1 hour.

281 3.2 Negative ions

282 **3.2.1** Main negative ions and their diurnal variation

283 A number of negative ions were consistently observed at CHC throughout the study 284 period (i.e., in the wet, wet-to-dry transition, and dry seasons). Based on the chemical 285 composition of the observed negative ions, we classified them into eight groups as follows: sulfuric acid ((H2SO4)0-3 HSO4), nitric acid ((HNO3)0-2 NO3), SO5, sulfuric 286 287 acid-ammonia ((NH₃)₁₋₆·(H₂SO₄)₃₋₇·HSO₄⁻), malonic acid-derived (MA-derived; 288 including C₃H₃O₄, C₃H₄O₄·NO₃, and C₃H₄O₄·HSO₄), oxidized organic molecules (CHO/CHON (HSO4 /NO3)), others (other identified negative ions, such as IO3), and 289 290 unidentified ions. The campaign-average diurnal variations of these eight negative ion 291 groups are shown in Figure 4. Different diurnal patterns of each negative ion group 292 were observed, mainly due to changes in concentrations of the parent neutral species 293 and their unique physicochemical properties (e.g., the EA and PA of molecules; Ferguson and Arnold, 1981; Bianchi et al., 2017; Hirsikko et al., 2011). 294

(HNO₃)₀₋₁·NO₃⁻ and (H₂SO₄)₀₋₃·HSO₄⁻ were among the highest in the signal of all negative ion groups in all seasons, making up 37 % (whole day) and 20 % (daytime;
07:00 - 19:00, and hereafter) of negative ions at CHC during the study period,
respectively. The EA of their neutral molecules (HSO₄ and NO₃) is higher than that of
most of the neutral species in the atmosphere, and thus hinders the direct electron
transfer from HSO₄⁻ and NO₃⁻ to other molecules through ion-molecule reactions
(Ferguson and Arnold, 1981). As a result, these ion groups were found to dominate

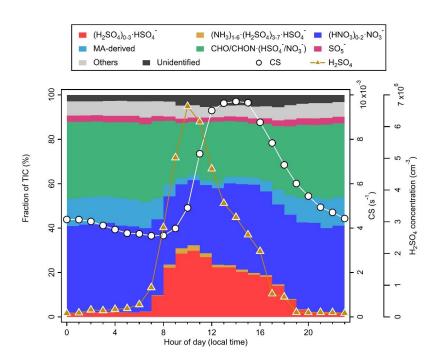




- negative cluster ions at CHC and also other locations, such as a number of remote sites
 in the United States (Eisele, 1986), a boreal forest site in Finland (Ehn et al., 2010;
- Bianchi et al., 2017), and the JFJ in Switzerland (Frege et al., 2017).
- 305 Distinct diurnal patterns were observed for the (HNO₃)₀₋₁·NO₃⁻ and (H₂SO₄)₀₋₃·HSO₄⁻ 306 ion groups. (HNO₃)₀₋₁·NO₃⁻ exhibited a relatively flat diurnal pattern (see Fig. 4) with 307 similar fractions at daytime (35 %) and nighttime (19:00 - 07:00; 39 %). Such a diurnal pattern could result from the high EA of the NO₃ molecule (4.01 eV and an additional 308 ~1 eV per HNO₃; Ferguson and Arnold, 1981), and its relatively abundant (usually 309 310 several ppby) parent neutral species (e.g., HNO_3 and N_2O_5) with multiple sources in the 311 atmosphere (e.g., anthropogenic emission and lightning; Martin et al., 2007). In contrast, 312 (H₂SO₄)₀₋₃·HSO₄⁻ exhibited a strong diurnal variation. While the fraction of (H₂SO₄)₀₋ 313 3. HSO4⁻ remained low (2 %) during nighttime, it started to increase after sunrise (shortly 314 after 07:00) and reached a maximum (30 %) at around 10:00. Despite an EA comparable 315 to that of the NO₃ molecule (4.75 eV for HSO₄; Wang et al., 2000), the strong diurnal 316 variation of (H2SO4)0-3 HSO4 is a result of the photochemical production of neutral 317 H2SO4. The influence of neutral H2SO4 on (H2SO4)0-3 HSO4 is indicated by their similar diurnal patterns (R: 0.52; see Fig. S3a). Similarly, a higher level of (NH₃)₁₋ 318 319 6'(H2SO4)3-7'HSO4' was only observed with the presence of abundant (H2SO4)0-3'HSO4' 320 during daytime. It is also important to note that the decreases of $(H_2SO_4)_{0-3}$ ·HSO₄⁻ and 321 (NH₃)₁₋₆•(H₂SO₄)₃₋₇•HSO₄⁻ at around noontime (12:00; see Fig. 4) coincided with an 322 enhanced CS, indicating the influence of a higher ion sink in addition to the decrease 323 in neutral H₂SO₄ concentration (Boulon et al., 2010; Frege et al., 2017).







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Figure 4 Diurnal variation of the negative ions (fraction of TIC), CS, and neutral H₂SO₄ concentrations
 at CHC, averaged over the periods when negative ions were measured (i.e., January, April, and May
 2018).

The MA-derived ion group is distinguished from the CHO/CHON•(HSO4⁻/NO3⁻) ion group due to its abundance in total organic ions. This group of ions, mainly composed of C₃H₃O_{4⁻}, is formed from the deprotonation of malonic acid, with a higher EA (~4.60 eV) than that of the NO₃ molecule (4.01 eV; Ravi Kumar et al., 2005). Thus, the fraction of the MA-derived ion group was high during nighttime (12 %; see Fig. 4). In contrast, its fraction decreased significantly (to 5 %) during daytime due to the increase of (H₂SO₄)₀₋₃•HSO_{4⁻}, which has an even higher EA (4.75 eV for HSO₄; Wang et al., 2000).

335 The CHO/CHON (HSO₄⁻/NO₃⁻) ion group, with an overall molecular formula of C₂-336 15H2-26O2-13N0-2 NO3⁻/HSO4⁻, constituted a significant fraction of negative ions (31 %) 337 at CHC. These organic ions are formed through the adduction between primary charge 338 carriers, such as HSO4⁻ and NO3⁻, and neutral OOM. These OOM are likely the 339 oxidation products of the volatile organic compounds (VOC) from the Amazon and the 340 adjacent La Paz - El Alto metropolitan area. While the diurnal variation was relatively 341 small (34 % for the nighttime and 27 % for the daytime), the ion composition of 342 CHO/CHON (HSO4/NO3) could be significantly different between daytime and 343 nighttime due to the availability of the charging ions (see more discussions in Section 344 3.2.2). A previous study from a boreal forest shows that organic ions are mainly





345 composed of CHO/CHON·NO₃⁻ during nighttime, and that the fraction of 346 CHO/CHON·HSO₄⁻ increases with the HSO₄⁻ signal during daytime (Bianchi et al., 347 2017). This is also shown by the slightly positive correlation between the 348 CHO/CHON·HSO₄⁻ signal fraction and the total neutral OOM concentration during 349 daytime (*R*: 0.25; see Fig. S3b), whereas no clear dependence was found between 350 CHO/CHON·(HSO₄⁻/NO₃⁻) and the total neutral OOM concentration.

351 The SO₅⁻ ion group, consisting of SO₅⁻ ions and/or O_2 ·SO₃⁻ cluster ions (Bork et al., 352 2013; Frege et al., 2017), exhibited a lower fraction (<5 %) than the aforementioned 353 ion groups during the study period. Similar to that of the $(HNO_3)_{0-1}$ ·NO₃⁻ ion group, no 354 diurnal pattern was evident for the SO₅ ion group. This may be the result of its different 355 major formation pathways during daytime and nighttime (Bork et al., 2013; Frege et al., 356 2017). Daytime production of SO_5^- ions is likely associated with photo-oxidation of 357 SO₂ (similar to the formation pathway of H₂SO₄; Ehn et al., 2010; Schobesberger et al., 358 2015). This is shown in the positive correlation (R: 0.46 for daytime data in Fig. S3c) 359 between the neutral H₂SO₄ concentration and the signal fraction of the SO₅⁻ ion group. During nighttime, however, the SO₅⁻ ion group is mainly composed of O₂ ·SO₃⁻ cluster 360 ions, which are possibly formed via the oxidation of SO_2 with O_3^- (producing SO_3^-), 361 362 and subsequent addition of O₂ (Bork et al., 2013).

363 3.2.2 Seasonalities of negative ions

364 For better seasonality comparison at high-altitude CHC, we calculated the average mass spectra of the negative ion groups for each season (Fig. 5 for daytime and Fig. S4 for 365 366 nighttime). Distinct seasonalities (wet season, wet-to-dry transition period, and dry season) were found for the majority of the negative ion groups at CHC, including 367 368 (H2SO4)0-3*HSO4⁻, (NH3)1-6*(H2SO4)3-7*HSO4⁻, SO5⁻, and organic cluster ions, as shown in the averaged daytime mass spectra (Fig. 5; more detailed reason will be discussed 369 370 below). However, the signals of some other negative ion groups, e.g., MA-derived ions 371 and $(HNO_3)_{0-2}$, were generally stable (with differences ≤ 20 %) across the seasons. 372 Such unclear seasonalities can be attributed to the high EA (Ferguson and Arnold, 1981; 373 Ravi Kumar et al., 2005) and/or the stability of the parent neutral species (Martin et al., 374 2007; Kerminen et al., 2000; Bikkina et al., 2021). Similar patterns can also be found 375 in the average nighttime mass spectra among the seasons (Fig. S4).

(H₂SO₄)₀₋₃•HSO₄⁻ group exhibited much higher contribution in dry season (May) than
in wet season (January) and wet-to-dry transition period (April). The daytime fraction
of (H₂SO₄)₀₋₃•HSO₄⁻ increased continuously from 16 % in wet, 20 % in wet-to-dry
transition period, to 30 % in dry season. The maximum number of H₂SO₄ molecules
increased concurrently from 2 to 4 in the cluster ions (i.e., from (H₂SO₄)₂•HSO₄⁻ to
(H₂SO₄)₄•HSO₄⁻). Similar trends were also found for other H₂SO₄-related ions, such as
the (NH₃)₁₋₆•(H₂SO₄)₃₋₇•HSO₄⁻ and SO₅⁻ during daytime (Fig. 5).





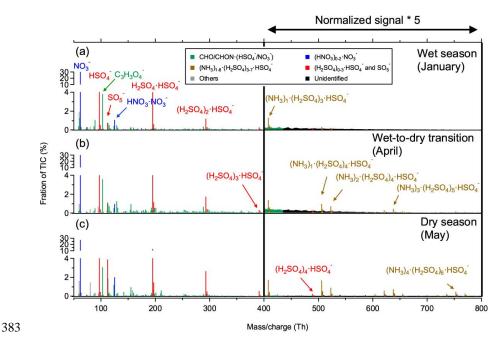


Figure 5 Mass spectra of negative ions at CHC averaged between 07:00 – 19:00 in (a) wet season
(January), (b) wet-to-dry transition period (April), and (c) dry season (May). The normalized signal
intensities from 400 Th to 800 Th are multiplied by a factor of 5 for better visualization.

387 The seasonal variations of the aforementioned H₂SO₄-related ion groups are likely due 388 to the changes in neutral H₂SO₄ (Fig. S5) linked to the changing synoptic-scale wind 389 patterns carrying different air masses with varying SO₂ (Bianchi et al., 2022). The air 390 mass pathways 07 PW and 08 PW, covering the Western and Northern Altiplano 391 plateau (see Table 1 and Fig. 2a), where active volcanic degassing of SO_2 has been 392 reported (Moussallam et al., 2017; Carn et al., 2017), had their largest influence on 393 CHC in dry season (i.e., May; see Fig. 2). The corresponding daytime fractions of 394 $(H_2SO_4)_{0.3}$ HSO₄ from these two pathways (Fig. 6a) were also the highest (27 % and 395 32 %, respectively). In contrast, air mass pathways 03 PW and 12 PW, originating in the Amazon Basin and Eastern/South-Eastern Lowlands, exerted their most significant 396 397 impact on CHC in wet season (i.e., January) with lower daytime fractions of (H2SO4)0-398 3⁺HSO₄⁻ (13 % and 14 %, respectively). The low fractions of H₂SO₄-related cluster ions 399 in wet season are also consistent with the lower SO2 level in the Amazon Basin 400 compared to the Altiplano plateau (Andreae et al., 1990). As for the wet-to-dry 401 transition period (i.e., April), 05 PW covering both the South-Eastern Lowlands and 402 Southern Altiplano plateau (where volcanic degassing is also significant; Carn et al., 403 2017) had an evident influence on CHC, resulting in a substantial level of H₂SO₄-404 related cluster ions (21 % for daytime). It is also noted that, because of the much lower 405 nocturnal neutral H₂SO₄ concentrations, the nighttime fractions of H₂SO₄-related cluster ions in all air mass pathways (Fig. 7b) were generally low (< 3 %) and no clear 406





407 seasonality was found.

408 The organic cluster ion group exhibited a distinct seasonal variation than the $(H_2SO_4)_{0-}$ 409 3'HSO4⁻. The signal fraction of organic cluster ions was higher in wet season (31 % for 410 daytime and 32 % for nighttime) than in dry season (23 % for daytime and 27 % for 411 nighttime; Fig. 5 and Fig. S4), but it was highest for the wet-to-dry transition period 412 (46 % for daytime and 52 % for nighttime; see Fig. 6).

413 The seasonal changes of organic cluster ions could be due to the combined effect of 414 different meteorological conditions and VOC from different air mass origins (see Fig. 415 2). The air masses that originated from the Amazon Basin and Lowlands (03 PW and 416 12 PW) showed their largest impact on CHC in wet season (i.e., January). They 417 contained higher fractions of organic cluster ions, which were 35 % and 34 % for 418 daytime, and 50 % and 45 % for nighttime, respectively (Fig. 6). In dry season (i.e., 419 May), however, the changes in air mass origin towards the Altiplano plateau and the 420 Pacific Ocean led to a lower content of organic cluster ions. The organic cluster ion 421 fractions for 07 PW and 08 PW (largest influence on CHC in dry season) in May were 422 23 % and 19 % for daytime, and 29 % and 27 % for nighttime, respectively. As for the 423 wet-to-dry transition period (i.e., April), due to the combined influences of biogenic 424 and anthropogenic VOC sources from 05 PW (evident impact on CHC in April), 425 covering the South-Eastern Lowlands and the Southern Altiplano plateau, the corresponding organic cluster ion fractions from this air mass pathway were also the 426 427 highest (41 % for daytime and 53 % for nighttime; see Fig. 6).

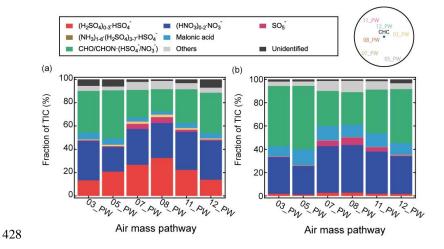


Figure 6 The fractions of the negative ion groups observed at CHC determined during the representative
periods of each air mass pathway (described in Section 2.3.2) for (a) daytime (07:00 – 19:00) and (b)
nighttime (19:00 – 07:00). A sketch of the horizontal profile of the air mass pathways (Fig. 2a) is shown

432 in the upper right corner for clarity.



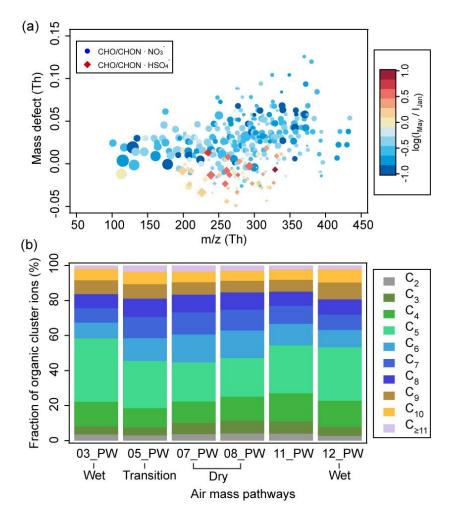


433 A further investigation of the organic ion group shows that the seasonal trends of the 434 individual organic ions also varied (Fig. 7 for daytime and Fig. S6 for nighttime). 435 Whereas the majority of the organic cluster ions at CHC were more abundant during 436 wet season (Fig. 7a and Fig. S6a), fractions of CHO/CHON HSO4 increased during the 437 dry season. The observed increases of CHO/CHON·HSO4⁻ cluster ions could be 438 associated with the increased HSO4⁻/NO3⁻ ratios in dry season (Fig. 5 and Fig. S4). 439 Similar increases of CHO/CHON·HSO4⁻ cluster ions were also found to relate to the 440 ratio of HSO₄⁻/NO₃⁻ in a boreal forest environment (Bianchi et al., 2017). In addition, 441 changes in OOM composition between wet and dry seasons may also play a role (Fig. 442 7b and Fig. S6b), as NO₃⁻ tends to cluster with OOM containing hydroxyl and 443 hydroperoxyl functional groups (Hyttinen et al., 2015) while some other observed 444 OOM may be more efficiently charged by HSO4⁻.

445 The seasonal variations of the individual organic cluster ions are likely caused by 446 different air masses (Fig. 7b and Fig. S6b). The air masses influenced by tropical 447 rainforest vegetation from the Amazon Basin are dominated by isoprene (C₅H₈) 448 emissions and isoprene oxidation products (Bianchi et al., 2022). This region 449 corresponds to 03 PW and 12 PW (largest impact on CHC in wet season in January) 450 consisting of relatively higher fractions of organic cluster ions with OOM containing 451 4-5 carbon atoms (50 % and 46 % for nighttime, and 29 % and 32 % for daytime, 452 respectively). In contrast, when the air masses were more influenced by the Altiplano 453 plateau (i.e., 05 PW, 07 PW, and 08 PW, with more anthropogenic emissions and less 454 vegetation) in wet-to-dry transition period and dry season, organic cluster ions with 6-455 8 carbon atoms, potentially originating from anthropogenic sources (e.g., toluene 456 (C7H₈); Huang et al., 2019; Cai et al., 2022), were of higher contributions. The signal 457 fractions of these organic ions were thus the highest in these air mass pathways, 458 accounting for 36-39 % for nighttime and 37-39 % for daytime. For all the air mass 459 pathways, fractions of organic ions with more than 9 carbon atoms were relatively low 460 (<10 %). This might be due to their lower volatilities compared to OOM with smaller carbon numbers (Donahue et al., 2012), resulting in a larger probability of them being 461 462 removed during their transport to CHC (e.g., condensing on pre-existing particles).







463

Figure 7 (a) Mass defect plot of organic cluster ions during nighttime (19:00-07:00). The color code indicates ratios (in log scale) between median signals of each ion detected in May (I_{May}) of dry season and January (I_{Jan}) of wet season. The marker size is proportional to the log-transformed median signals of ions in May. (b) Fraction of organic cluster ions from different air mass pathways as a function of carbon atom numbers during nighttime (19:00-07:00). A similar figure based on daytime data (07:00-19:00) is in the supplementary information (Figure S5). Note that MA-derived ions were not included in this figure.

471 **3.3 Positive ions**

472 Several positive cluster ion groups were consistently observed in February and March 473 (i.e., wet season) during the study period. Based on their chemical composition, the 474 positive cluster ions measured at CHC are classified into four groups (Fig. 8): (1) a 475 series of protonated amines, including trimethylamine ($C_3H_9N\cdot H^+$), pyridine 476 ($C_5H_7N\cdot H^+$), aniline ($C_6H_7N\cdot H^+$), and benzylamine ($C_7H_9N\cdot H^+$); (2) organic cluster



490



477 ions consisting of OOM (identified as C3-24H6-39O2-12N0-2) clustered with positive 478 charge carriers such as protons (H^+) , ammonium (NH_4^+) , and aminium (NH^+) ions; (3) 479 contamination ions; and (4) unidentified ions (likely organic ions in higher masses; 480 Bianchi et al., 2021). Contamination in the positive cluster ions includes 481 ethylhexylglycerin (e.g., $C_{11}H_{24}O_3 \cdot NH^+$), which is widely used in cosmetics (Aerts et 482 al., 2016), and polydimethylsiloxane (e.g., (C2H6OSi)7'NH4⁺) possibly from instrument tubing (Bianchi et al., 2014). In contrast to the negative cluster ions, the four positive 483 484 cluster ion groups were generally stable with smaller diurnal variability over the study 485 period (Fig. 9). This is similar to the diurnal patterns determined in previous studies in 486 a boreal forest environment (Ehn et al., 2010) and at the JFJ (Frege et al., 2017). 487 However, due to the unavailable measurements of the corresponding neutral species 488 (e.g., amines), the exact reason for such weak diurnal variations observed in different 489 locations remains unclear.

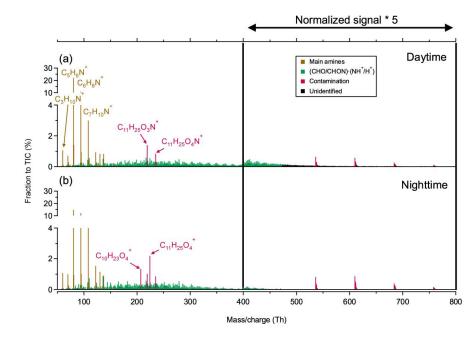


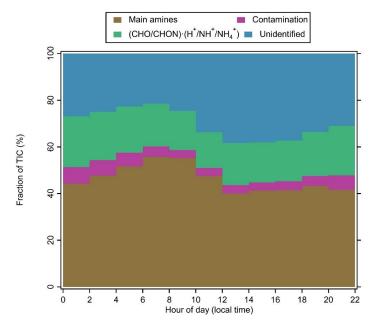
Figure 8 Averaged mass spectra of positive ions at CHC in February and March 2018 when APi-TOF
was operating in positive ion mode (see section 2.2.1), during (a) daytime (07:00 - 19:00) and (b)
nighttime (19:00 - 07:00). The normalized signal intensities from 400 Th to 800 Th are multiplied by a
factor of 5 for better visualization.

The protonated amines were the most abundant positive ion group (46 %), with no significant diurnal variations. Nighttime contributions of this ion group (47 %) were similar to its daytime contributions (45 %; Fig. 9). They also dominated the positive ion spectra observed in different environments, such as a boreal forest (Ehn et al., 2010), the JFJ (Frege et al., 2017), and the free troposphere (Schulte and Arnold, 1990). Their





- 500 sources have not been fully identified (Kosyakov et al., 2020), but they are widely used
- 501 as solvents and dyes (Sims et al., 1989), which may be potential sources of these ions
- 502 observed at CHC.



503

Figure 9 Diurnal variation of positive ion groups at CHC, averaged over measurements in February and
 March 2018 (when APi-TOF was operating in positive ion mode, see section 2.2.1).

506 Positive organic cluster ions were also relatively abundant (19%) at CHC during the 507 wet season. Similar to the negative organic ions in wet season (Fig. 6), this reflects the 508 influence of air masses originating from the Amazon Basin and Eastern/South-Eastern 509 Lowlands (e.g., 03 PW and 11 PW). Differences in the positive organic ion signals 510 between nighttime (21 %) and daytime (18 %) were small, which is similar to the 511 negative organic cluster ions (see Fig. 3). A further investigation of the relationship 512 between these positive ions and their neutral species is, unfortunately, not possible due 513 to the unavailability of CI-APi-TOF data in February and March caused by instrumental 514 issues.

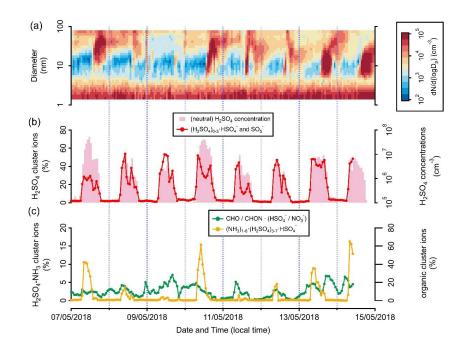
515 3.4 Potential connections between atmospheric ions and new particle 516 formation events

517 During the SALTENA campaign from January to May 2018, NPF events were 518 frequently observed at CHC (Fig. S7). While most of them occurred from April (the 519 wet-to-dry transition period, 21 events) to May (dry season, 26 events), NPF events 520 seldom occurred during wet season from January to March (8 events in total).





521



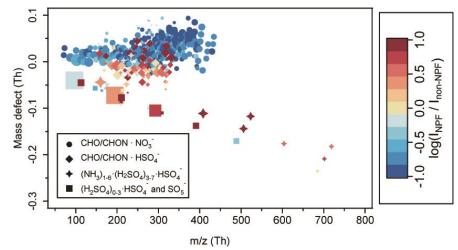
522Figure 10 Time series of the (a) size distribution of aerosol particles (measured with NAIS and MPSS),523(b) signal fraction of the $(H_2SO_4)_{0.3}$ ·HSO₄⁻ ion group and neutral H_2SO_4 concentration, and (c) signal524fractions of the $(NH_3)_{1-6}$ ·(H_2SO_4)₃₋₇·HSO₄⁻ and negative organic cluster ion groups, observed at CHC525from 7 to 14 May 2018 when NPF occurred frequently.

526 Previous field studies at high-altitude mountain sites have shown that NPF events can 527 be triggered by different compounds, such as low-volatile neutral OOM (Bianchi et al., 528 2021), neutral H₂SO₄ and OOM (Bianchi et al., 2016), and H₂SO₄-NH₃ cluster ions 529 (Frege et al., 2017). While the signal fractions of the negative organic cluster ions did 530 not seem to have a strong correlation with the onset of the NPF events, the fractions of 531 the (NH₃)₁₋₆·(H₂SO₄)₃₋₇·HSO₄⁻ (associated with (H₂SO₄)₀₋₃·HSO₄⁻) always peaked 532 before NPF events, and started to decrease when NPF started (see example NPF events 533 on, e.g., 7, 10, 13, and 14 May 2018; Fig. 10).

Moreover, higher levels (up to an order of magnitude) of (NH₃)₁₋₆•(H₂SO₄)₃₋₇•HSO₄⁻
and (H₂SO₄)₀₋₃•HSO₄⁻ ions as well as negative organic cluster ions charged by HSO₄⁻
were also observed during the NPF days (Fig. 11). In contrast, other negative ion groups
(e.g., the majority of the negative organic cluster ions charged by NO₃⁻) were more
abundant during the non-NPF days. Our observations indicate a potentially important
role of (NH₃)₁₋₆•(H₂SO₄)₃₋₇•HSO₄⁻ cluster ions in NPF events at CHC from January to
May 2018, particularly in wet-to-dry transition period and dry season.







541

542 Figure 11 Mass defect plot of differences in negative cluster ion composition between NPF and non-NPF 543 days. The negative ion composition of NPF events was averaged over all NPF days from 08:00 to 12:00 in January, April, and May 2018 (when APi-TOF was operating in negative ion mode, see section 2.2.1) 544 545 at CHC. The ion composition of non-NPF days was averaged over non-NPF days from 08:00 to 12:00 546 for the same period. The x-axis is the exact mass of cluster ions, and the y-axis is the mass defect. The 547 color code indicates ratios (in log scale) between median signals of each ion determined in NPF events 548 (I_{NPF}) and non-NPF periods (I_{non-NPF}). The marker size is proportional to the log-transformed median 549 signals of ions observed in the NPF events that occurred in January, April and May. Note that (HNO₃)₀₋ 550 2. NO3⁻ cluster ions were not included here.

551 The majority of the observed NPF events occurred when CHC was more impacted by air masses originating from source regions with elevated SO₂ emissions (05 PW, 552 553 07 PW, and 08 PW). This is similar to the observations from the high-altitude station 554 JFJ (Frege et al., 2017). In addition, the fraction of large positive organic cluster ions 555 (mass range from 500 to 800 Th; Fig. S8) was found to increase during the NPF events in the wet season. Such large positive organic ions have been found to contribute to 556 557 NPF in the Himalayas (Bianchi et al., 2021), and thus, their contribution to NPF can 558 not be completely ruled out at CHC as well.

559 4. Conclusion

560 In this study, both negative and positive atmospheric ions were measured at a highaltitude research station (CHC) in the Bolivian Andes for five months, from January to 561 562 May 2018, using an APi-TOF mass spectrometer. Negative ions were mainly composed 563 of (H2SO4)0-3[•]HSO4⁻, (HNO3)0-2[•]NO3⁻, SO5⁻, (NH3)1-6[•](H2SO4)3-7[•]HSO4⁻, MA-derived, 564 and CHO/CHON (HSO4⁻/NO3⁻) ion groups. Positive ions mainly consisted of a series 565 of protonated amines $(C_{3-7}H_{7-9}N \cdot H^{+})$ and organic cluster ions 566 $CHO/CHON \cdot (H^+/NH^+)$. Distinct diurnal variation was observed for the negative 567 ions, and attributed mainly to the changes in the corresponding neutral species'





568 concentrations and/or their EA / PA. An example is H₂SO₄-related cluster ions, the diel 569 temporal variation of which was mainly due to the photochemical production of neutral 570 H₂SO₄ during daytime. Strong seasonality of negative ions was also found, such as for 571 H₂SO₄-related cluster ions owing to changes in SO₂ and the resulting neutral H₂SO₄ 572 concentrations. The seasonal variation was mainly because of the differences in source 573 regions of air masses arriving at CHC from wet to dry seasons. In contrast, no 574 significant diurnal variation was observed for the positive ions. The comparison 575 between NPF and non-NPF days infers that H₂SO₄-NH₃ cluster ions contribute to the aerosol nucleation process at CHC, particularly in wet-to-dry transition period and dry 576 577 season when CHC was more impacted by air masses originating from source regions 578 with elevated SO₂ emissions. The results further indicate that atmospheric ion 579 composition at CHC is directly affected by air masses from different source regions.

580 Measurements of atmospheric ions in the field will improve understanding of 581 atmospheric physical and chemical processes in the study regions, as the ions play 582 important roles in atmospheric chemistry through participation in or catalysis of ion-583 molecule reactions and ion-induced new particle formation. Our study thus provides 584 new insights into the chemical composition of atmospheric ions and their potential role 585 in high-altitude NPF in the Bolivian Andes where both natural (e.g., biogenic and 586 volcanic) and anthropogenic emissions are important.

587





588 **Data availability:** The data that are involved in the figures can be found in 589 doi.org/10.5281/zenodo.7271286 (Zha et al., 2022).

Author contributions: Q. Z., W.H., and F.B. analysed the data; D.A. conducted the air
mass history analysis; Q.Z., W.H., F.B., D.A., O.P., L.H., A.M.K., C.W., J.E., Y.G.,
M.A., C.M., and F.B. collected the data and operated the instruments during the
measurement campaign. Q.Z. and W.H. wrote the manuscript with contributions from
J.C., V.S., S.C., D.W., R.K., M.A., C.M., and F.B. All authors commented on the
manuscript.

596 **Competing interests:** The authors declare no competing interests.

597 Acknowledgment: We thank the Bolivian staff of the IIF-UMSA (Physics Research 598 Institute, UMSA) working at CHC and the long-term observations performed within 599 the framework of GAW and ACTRIS. We thank the IRD (Institut de Recherche pour le 600 Développement) for the logistic and financial support during the campaign, including 601 shipping and customs concerns. We thank the CSC-IT Center for Science, Finland, for 602 the generous computational resources that allowed the WRF and FLEXPART-WRF 603 simulations to be conducted.

Grant information: This research has received support from European Union (EU)
H2020 program via the findings European Research Council (ERC; project CHAPAs
no. 850614 and ATM-GTP no. 742206), the Marie Skłodowska Curie (CLOUDMOTION no. 764991), the Finnish Centre of Excellence as well as the Academy of
Finland (project no. 311932, 315203 and 337549), and the Knut and Alice Wallenberg
Foundation (WAF project CLOUDFORM no. 2017.0165).

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