- 1 Measurement report: Molecular-level investigation of
- 2 atmospheric cluster ions at the tropical high-altitude research
- 3 station Chacaltaya (5240 m a.s.l.) in the Bolivian Andes
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29 Abstract

30 Air ions are the key components for a series of atmospheric physicochemical 31 interactions, such as ion-catalyzed reactions, ion-molecule reactions, and ion-induced 32 new particle formation. They also control atmospheric electrical properties with effects 33 on global climate. We performed molecular-level measurements of cluster ions at the high-altitude research station Chacaltaya (CHC; 5240 m a.s.l.), located in the Bolivian 34 35 Andes, from January to May 2018 using an atmospheric pressure interface time-of-36 flight mass spectrometer. The negative ions mainly consisted of (H2SO₄)₀₋₃·HSO₄, 37 (HNO₃)₀₋₂·NO₃, SO₅, (NH₃)₁₋₆·(H₂SO₄)₃₋₇·HSO₄, malonic acid-derived, and CHO/CHON•(HSO4-/NO3-) cluster ions. Their temporal variability exhibited distinct 38 39 diurnal and seasonal patterns due to the changes in the corresponding neutral species' 40 molecular properties (such as electron affinity and proton affinity) and concentrations 41 resulting from the air masses arriving at CHC from different source regions. The 42 positive ions were mainly composed of protonated amines and organic cluster ions, but 43 exhibited no clear diurnal variation. H₂SO₄-NH₃ cluster ions likely contributed to the 44 new particle formation process, particularly during wet-to-dry transition period and dry 45 season when CHC was more impacted by air masses originating from source regions 46 with elevated SO₂ emissions. Our study provides new insights into the chemical 47 composition of atmospheric cluster ions and their role in new particle formation in the 48 high-altitude mountain environment of the Bolivian Andes.

1 Introduction

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Air ions regulate the electrical properties of the atmosphere by serving as carriers of 50 51 electrical charges (Williams, 2009). They also play an important role in atmospheric 52 chemistry by participating/catalyzing ion-molecule reactions and ion-induced new 53 particle formation (NPF, Hirsikko et al., 2011). The formation of tropospheric ions is initiated through simple-structured ions, such as O⁺, N₂⁺, O⁻, and O₂⁻, mainly from 54 55 radioactive decay in the soil (e.g., radon and gamma radiation), thunderstorm activity 56 (lightning), and galactic cosmic rays (GCR). These ions can transfer their charges to 57 other compounds, leading to the subsequent production of an assortment of ions, such 58 as the bisulfate ion (HSO₄⁻), nitrate ion (NO₃⁻), hydronium ion (H₃O⁺), and ammonium 59 ion (NH₄⁺; Smith and Spanel, 1995; Hirsikko et al., 2011). Depending on their sizes, air ions are usually classified into cluster ions (diameter ≤ 1.6 nm) that are charged 60 61 molecules or molecular clusters, and charged particles (diameter > 1.6 nm; Hirsikko et 62 al., 2005, 2011; Komppula et al., 2007).

Cluster ions exist almost always in the troposphere and can undergo frequent ionmolecule reactions during their lifetime (~100 seconds; Manninen et al., 2010; Hirsikko et al., 2011). Their chemical composition, in addition to the initial ionization, also depends on the concentrations of the parent neutral species (Eisele, 1986). Bianchi et al. (2017) showed that the diurnal cycle of negative organic ions followed the variations of their neutral molecules' concentrations in a boreal forest, since the higher 69 concentrations of neutral molecules would result in a larger probability of them being charged. Moreover, molecular properties of the neutral species, such as electron affinity 70 71 (EA) and proton affinity (PA), are also important for determining cluster ion 72 composition. Cluster ions derived from molecules with higher EA (e.g., HSO₄ and NO₃) 73 or PA (e.g., trimethylamine (C₃H₉N) and pyridine (C₅H₅N)) tend to obtain the ambient 74 negative or positive charge, respectively (Ferguson and Arnold, 1981; Hirsikko et al., 75 2011). Because of the strong EA or PA, it is almost unlikely that the ions derived from 76 those molecules will further transfer their charges to other neutral compounds via ion-77 molecule reactions. Thus, these negative (HSO₄⁻ and NO₃⁻) and positive (C₃H₁₀N⁺ and 78 $C_5H_6N^+$) ions are usually more abundant than other ions in the atmosphere (Eisele, 1986; 79 Ehn et al., 2010; Bianchi et al., 2017; Frege et al., 2017). In contrast, charge transfer 80 occurs more easily for ions derived from neutral species of lower EA or PA.

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Atmospheric cluster ions can contribute to new particle formation (NPF) via ion-82 induced nucleation (Yu, 2010). Since the discovery of this mechanism in the first cloud 83 chamber study in the early 1900s (Wilson, 1911), ion-induced nucleation has been 84 known as an important source of atmospheric aerosol particles. Recently, a series of 85 chamber studies conducted at the CLOUD (Cosmics Leaving Outdoor Droplets) facility 86 at CERN (the European Centre for Nuclear Research) have shown that aerosol 87 nucleation rates are substantially enhanced in the presence of some specific cluster ions, 88 such as sulfuric acid – ammonia (H₂SO₄ – NH₃) cluster ions (Kirkby et al., 2011; 89 Schobesberger et al., 2015), pure H₂SO₄ cluster ions (Kirkby et al., 2011), and organic 90 cluster ions (Kirkby et al., 2016). Field measurements have also suggested the important role of atmospheric ions in ion-induced nucleation (Manninen et al., 2010; Hirsikko et 92 al., 2011; Rose et al., 2018; Jokinen et al., 2018; Yan et al., 2018; Beck et al., 2021). 93 Among them, the onsets of high-altitude NPF events, compared to those occurring in 94 the lower troposphere, are often associated with more abundant cluster ions (Lee et al., 95 2003; Venzac et al., 2008; Boulon et al., 2010; Sellegri et al., 2019). Such increases are 96 due to the higher GCR intensity and lower condensation sink (CS) in the high-altitude 97 regions. As a result, potentially larger contributions of cluster ions to aerosol formation 98 would be expected (Smith and Spanel, 1995; Hirsikko et al., 2011).

However, the molecular-level understanding of ambient cluster ions and their influence on NPF in high-altitude environments (in the troposphere) is still very limited. Two mountaintop studies in the Alps show that, depending on the air mass origins, NPF could be triggered by sulfuric acid-ammonia clusters, or nitrate (or sulfuric acid) clustering with highly oxygenated organic molecules (Bianchi et al., 2016; Frege et al., 2017). Another study in the Himalayas found that NPF was mainly driven by organic vapors of biogenic origin (Bianchi et al., 2021). Recently, frequent and intensive NPF events were observed at the high-altitude research station Chacaltaya (CHC; 16.3505° S, 68.1314° W; 5240 m a.s.l.) located in the Bolivian Andes (Rose et al., 2015a), but the exact mechanism and the role of cluster ions in aerosol nucleation process remain unclear. Therefore, a detailed investigation of cluster ions at CHC, including their

- molecular composition, temporal variation (diurnal and seasonal), and source regions,
- is needed in order to understand their role in atmospheric processes such as NPF in the
- study regions.
- Here we present measurements of atmospheric ions from January to May 2018 at CHC.
- 114 The dataset is part of the Southern hemisphere high ALTitude Experiment on particle
- Nucleation And growth (SALTENA) field experiment campaign (Bianchi et al., 2022).
- During the study period, the sampled air masses originated from various source regions,
- such as the Amazon Basin to the east and the Altiplano and the Pacific Ocean to the
- west (Fig. 1a; Aliaga et al., 2021). Temporal evolution (diurnal and/or seasonal
- variations) of both negative and positive ion composition are investigated, and their
- potential connections with source regions and NPF are discussed. Our study thus adds
- important observational information on a better understanding of atmospheric ions and
- provides new insights into their role in high-altitude NPF in the troposphere of the
- 123 Bolivian Andes.

124 **2 Methods**

2.1 Measurement site description

- 126 The high-altitude research station CHC is ~140 m below the summit of Mount
- 127 Chacaltaya (5380 m a.s.l.) with an open view to the south and west (Andrade et al.,
- 128 2015). The La Paz El Alto metropolitan area (with 1.7 million inhabitants) is $\sim 1 1.6$
- 129 km lower (in altitude) and ~15 km south of CHC (Fig. 1b). The seasonal meteorological
- 130 conditions at CHC depend on the cycle between the wet (November to March; wet-to-
- dry transition period in April) and dry (May to September; dry-to-wet transition period
- in October) seasons driven by large-scale tropical circulation (Rose et al., 2015a;
- Bianchi et al., 2022). This pattern also affects the source regions of air masses arriving
- at CHC (Aliaga et al., 2021). Additionally, due to the strong diurnal cycle of the
- planetary boundary layer (PBL) height and the thermally-induced winds in the
- mountainous terrain, CHC is often affected by polluted PBL transported from the La
- 137 Paz El Alto metropolitan area during daytime (Wiedensohler et al., 2018) whereas at
- night CHC is located in the residual layer or tropical free troposphere (Rose et al., 2015b;
- 139 Chauvigné et al., 2019).

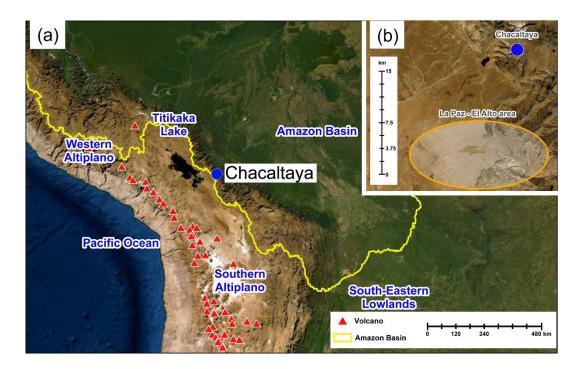


Figure 1 (a) True-color satellite image showing the location of CHC (blue circle) and its surrounding area. The yellow line presents the boundary of the Amazon Basin. Red triangles denote the volcanoes in this area. (b) A zoomed-in true-color satellite image showing the distance between CHC and the La Paz – El Alto metropolitan area (orange circle). Image sources: Esri, DigitalGlobe, GeoEye, i-cubed, USDA FSA, USGS, AEX, Getmapping, Aerogrid, IGN, IGP, swisstopo, and the GIS User Community.

2.2 Instrumentation

All the instruments involved in this study were installed in a temperature-controlled measurement room (at ~25 °C). All data are reported in local time (UTC-4).

2.2.1 Measurements of atmospheric cluster ions

The composition of cluster ions was measured by an atmospheric pressure interface time-of-flight mass spectrometer (APi-TOF, Aerodyne Research Inc. & Tofwerk AG). The APi-TOF consists of an atmospheric pressure interface (APi) module and a time-of-flight (TOF) mass spectrometer. The APi module allows the instrument to sample ions from ambient air directly. The positive or negative ions within the sampled airflow are focused and guided by two quadrupoles and an ion lens, with a gradually decreasing pressure (from atmospheric pressure to $\sim 10^{-4}$ mbar), before entering the TOF mass spectrometer ($\sim 10^{-6}$ mbar). A more detailed description of this instrument is given in Junninen et al. (2010). In this study, ambient air was sampled through a ~ 1.5 m stainless steel tube with a total sample flow of 14 standard liters per minute (SLPM) to ensure laminar flow during sampling, and 0.8 SLPM of the total flow entered the APi-TOF.

During the wet season, the APi-TOF was first operated in negative mode to measure negative cluster ions (January), and then switched to positive mode to measure positive

- 164 cluster ions (February to March). During the wet-to-dry transition period (April) and
- dry season (May), the instrument was changed back to negative mode to investigate the
- potential seasonality of negative ion composition. It is important to note that, similar to
- 167 Frege et al. (2017), for better characterization of the connection between cluster ions
- and NPF, we only included the ion data observed under the cloud-free condition in this
- study (to avoid influence from, e.g., lightning activity). CHC was considered to be
- affected by clouds when relative humidity (RH) exceeded 95 %, as suggested by a
- previous study at the same location (Rose et al., 2015a). From January to May 2018,
- the proportions of cloud-free hours to the total measurement time were 72 %, 78 %,
- 173 79 %, 98 %, and 98 %, respectively (Fig. S1).

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2.2.2 Measurements of H₂SO₄ and oxidized organic molecules

- 175 Concentrations of H₂SO₄ and oxidized organic molecules (OOM) were measured using
- a nitrate ion (NO₃⁻) based chemical ionization atmospheric pressure interface time-of-
- 177 flight mass spectrometer (CI-APi-TOF, Aerodyne Research Inc. & Tofwerk AG;
- Jokinen et al., 2012). The instrument is a combination of the APi-TOF and a chemical
- ionization (CI) unit, which has been widely used to measure H₂SO₄ in the atmosphere
- 180 (Jokinen et al., 2012; Bianchi et al., 2016; Zha et al., 2018). In this study, a soft X-ray
- source (L9490, Hamamatsu) was used to charge nitric acid (HNO₃) in a sheath flow of
- 20 SLPM to produce the reagent ion, NO₃⁻. H₂SO₄ and OOM in the sample flow (10
- SLPM) were then charged by either proton transfer or the formation of an adduct with
- the reagent ion during the ~200 ms residence time in the CI unit. A calibration factor of
- to the state of th
- 185 1.5×10^{10} cm⁻³ for H₂SO₄ was determined (with sampling loss corrected) according to
- the approach by Kürten et al. (2012). The same calibration coefficient was adopted for
- 187 determining OOM concentrations in this study, which could result in an
- underestimation of their concentrations due to a lower charging efficiency of OOM than
- 189 H₂SO₄ by NO₃ (Hyttinen et al., 2015).

2.2.3 Auxiliary measurements

- 191 The number concentration and size distribution of atmospheric ions and neutral
- 192 particles were measured with a neutral cluster and air ion spectrometer (NAIS, Airel
- 193 Ltd., Mirme and Mirme, 2013). The instrument can detect naturally charged air ions
- and total particles with mobility diameters from 1.4 50 nm and 3 50 nm, respectively.
- 195 It is important to note that the size range of the NAIS was corrected based on a side-
- by-side comparison with an updated version of NAIS (designed for measurements
- under low pressure environments; Mirme et al., (2010)) at CHC. Thus, the size range
- of detection was slightly different from the traditionally reported ranges (i.e., 0.8 42
- 199 nm and 2-42 nm, respectively; Manninen et al., 2010). The details of the instrument
- used can be found in Rose et al. (2017).
- 201 Particle number size distributions between 10 to 500 nm were measured by a Mobility
- 202 Particle Size Spectrometer (MPSS; Wiedensohler et al., 2012), and the data was used

- for calculating the CS, which represents the loss rate of condensing vapors and cluster
- ions on pre-existing particles (Kulmala et al., 2001).
- Meteorological parameters, such as temperature, RH, and global radiation, were also
- 206 measured simultaneously at CHC. Detailed descriptions can be found in Bianchi et al.
- 207 (2022).

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2.3 Simulation of air mass origin and history

- 209 To understand the source regions and transport pathways of the air masses arriving at
- 210 CHC, we used the results of air mass history analysis obtained from FLEXPART-WRF
- simulations described in Aliaga et al. (2021). In brief, a Lagrangian transport and
- dispersion model (FLEXPART-WRF; version 3.3.2; Brioude et al., 2013) was used to
- 213 calculate the air mass history during the campaign period. The backward simulation
- was driven by the high spatial and temporal resolution meteorological output from the
- Weather Research and Forecasting model (WRF; version 4.0.3; Skamarock et al., 2019).
- 216 In the simulation, twenty thousand particles were continuously released every hour
- from a 10 m deep layer (0 10 m a.g.l.) at CHC and traced back in the atmosphere for
- 218 96 hours. The output of the FLEXPART-WRF is the source-receptor relationship (SRR,
- 219 in seconds), which is calculated for each geographical grid cell included in the
- simulation. The SRR value depends on the particle's residence time and the number of
- particles in the output grid cells. Clustering analysis was conducted by applying a series
- of pretreatments (e.g., log-polar grid transformation and grid cell pre-processing) and a
- k-means clustering algorithm (Lloyd, 1982) to the calculated SSR dataset (see Aliaga
- et al. (2021) for more details).

2.3.1 Major air mass pathways

- 226 Six air mass pathways (PW) representing air masses arriving at CHC were determined
- from the clustering analysis. They are named based on their clock positions from CHC
- 228 (e.g., 03 PW indicates the pathway with its centroid located at the 3 o'clock direction
- (east, 90°) of CHC, Fig. 2a). Characteristics of these air mass pathways, such as source
- region, transport distance, and transport time, were distinct from each other (Table 1).
- A detailed description of the air mass pathways and their characteristics can be found
- 232 in Aliaga et al. (2021).
- 233 The influence of each air mass pathway on CHC varied with time, and was estimated
- by its SRR percentage (SRR[%]_{pathway}) as in equation (1):

$$SRR[\%]_{pathway} = \frac{SRR_{pathway}}{SRR_{total}} \times 100 \tag{1}$$

- where SRR_{pathway} and SRR_{total} are the residence time of a specific air mass pathway and
- in total (96 hours = 345600 seconds) in the simulation, respectively.
- Table 1. Overview of the six air mass pathways extracted from Aliaga et al. (2021).

Pathway	Direction to CHC	Representative source region	Transport distance (km) ¹	Transport time (hour)
			Median (25 – 75 %)	Median (25 – 75 %)
03_PW	East	Amazon Basin and Eastern/South-Eastern Lowlands	518 (413–608)	51 (45-57)
05_PW	South and Southeast	South-Eastern Lowlands and Southern Altiplano	428 (303-567)	45 (36-52)
07_PW	Southwest	The Pacific Ocean, coastal area, Western Altiplano, and La Paz – El Alto	721 (577-896)	54 (45-61)
08_PW	West	Western Altiplano and Titicaca lake, coastal area	238 (198-279)	36 (29-43)
11_PW	North and Northwest	Amazon Basin, Western Altiplano, coastal area	465 (326-563)	53 (46-59)
12_PW	North	Amazon Basin	76 (49 -95)	27 (21-33)

239 ¹Distance between CHC and each pathway's center point (see Fig. 2a).

2.3.2 Identification of representative periods for each air mass pathway

Air mass history analysis shows that the air sampled at CHC was typically a mixture of multiple pathways. Thus, the cluster ion composition observed during the study period was often influenced by multiple source regions concurrently. To characterize the influence of every single pathway on cluster ion composition, periods when an air mass pathway exerted its largest impact on CHC (the highest 10% of its SRR[%]_{pathway} values; Fig. 2b) during the whole study period are identified as the representative periods of the specific pathway. For instance, the representative periods of 03_PW (covering, e.g., the Amazon Basin) are more frequently seen during wet season (highest in January), whereas 08_PW (covering, e.g., Altiplano region) has most of its representative periods in dry season (highest in May). Note that SRR[%]_{pathway} of any individual pathway rarely reached 40 % due to the relatively short study period (see Fig. S2), and thus the representative periods cannot be directly identified via SRR[%]_{pathway} values. In contrast, such representative periods were determined by using a certain threshold (e.g., > 70 %) in a previous study at CHC, which was based on a more than 6-year dataset (Koenig et al., 2021).

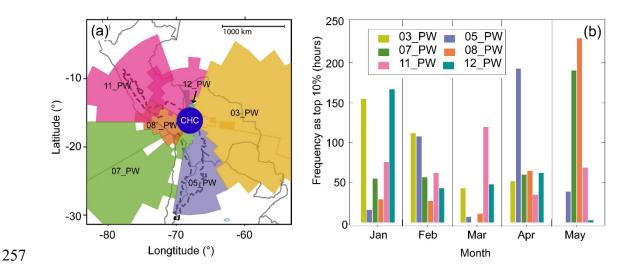


Figure 2 Influence of the six air pathways on CHC from January to May 2018. (a) Horizontal profile of the air mass pathways, adapted from Aliaga et al. (2021). (b) Frequency of the representative periods for each pathway (the highest 10% of their corresponding SRR[%]_{pathway}) in different months.

3 Results and discussion

3.1 Variation in total ion count

During the study period, the total ion count (TIC) observed by APi-TOF at CHC varied between <10 and 100 counts per second (cps; Fig. 3 and Fig. S3). Similar variations in the TIC were also observed in the long-term cluster ion measurements at the high-altitude station Jungfraujoch in Switzerland (JFJ; 3454 m a.s.l.; Frege et al., 2017) and were attributed to the seasonal changes of ion precursor and sink. Like the winter-summer seasonality of the JFJ, CHC and its adjacent mountain areas are frequently covered by snow in wet season and mostly free of snow in dry season (Bianchi et al., 2022; Koenig et al., 2021). Thus, in contrast with the generally stable GCR flux (primarily controlled by the decadal scale solar cycle; Shuman et al., 2015), a reduced radioactive decay from the soil and a lower ion production rate could be expected at CHC in wet season than in dry season.

However, the TIC measured by APi-TOF was significantly higher in wet season (41 \pm 23 cps, mean \pm standard deviation) and the wet-to-dry transition period (56 \pm 32 cps) than in dry season (14 \pm 11 cps; Fig. S3). Considering the slight negative correlation (Pearson's correlation coefficient (R): -0.41; Fig. 3) between the TIC and CS (representing the loss rate of condensing vapors and cluster ions on pre-existing particles; Kulmala et al., 2001), the observed TIC fluctuation may be related (at least partially) to the varying CS (\sim 1 \times 10⁻⁴ s⁻¹ in wet season to \sim 5 \times 10⁻² s⁻¹ in the dry season). Moreover, the cluster ions measured by APi-TOF usually account for only a small fraction of the total atmospheric ions (Rose et al., 2018). Changes in the fraction of small ions in total atmospheric ions can potentially lead to a fluctuation in TIC (Frege

et al., 2017). This is illustrated in Figure 3 that a smaller TIC determined from APi-TOF is associated with a lower fraction of smaller ions (< 2 nm) observed by NAIS (mostly cluster ions). However, for better characterization of the influences of different ion composition on CHC and their diurnal and seasonal relative changes, we normalized the observed ion signal to the TIC for APi-TOF measurements.

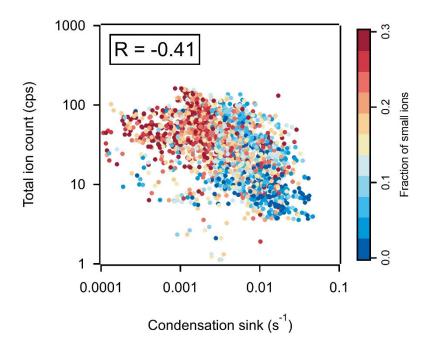


Figure 3 Correlation between the TIC measured by APi-TOF and condensation sink, colored by the fraction of small ions (defined as concentrations of ions with diameter <2 nm to the total ion concentrations) determined from NAIS data. Data are shown in the time resolution of 1 hour.

3.2 Negative ions

3.2.1 Main negative ions and their diurnal variation

A number of negative ions were consistently observed at CHC throughout the study period (i.e., in the wet, wet-to-dry transition, and dry seasons). Based on the chemical composition of the observed negative ions, we classified them into eight groups as follows: sulfuric acid ((H₂SO₄)₀₋₃·HSO₄-), nitric acid ((HNO₃)₀₋₂·NO₃-), SO₅-, sulfuric acid-ammonia ((NH₃)₁₋₆·(H₂SO₄)₃₋₇·HSO₄-), malonic acid-derived (MA-derived; including C₃H₃O₄-, C₃H₄O₄·NO₃-, and C₃H₄O₄·HSO₄-), oxidized organic molecules (CHO/CHON·(HSO₄-/NO₃-)), others (other identified negative ions, such as IO₃-), and unidentified ions. The campaign-average diurnal variations of these eight negative ion groups are shown in Figure 4. Different diurnal patterns of each negative ion group were observed, mainly due to changes in concentrations of the parent neutral species and their unique physicochemical properties (e.g., the EA and PA of molecules; Ferguson and Arnold, 1981; Bianchi et al., 2017; Hirsikko et al., 2011).

309 (HNO₃)₀₋₁·NO₃ and (H₂SO₄)₀₋₃·HSO₄ were among the highest in the signal of all negative ion groups in all seasons, making up 37 % (whole day) and 20 % (daytime; 310 311 07:00 - 19:00, and hereafter) of negative ions at CHC during the study period, 312 respectively. The EA of their neutral molecules (HSO₄ and NO₃) is higher than that of 313 most of the neutral species in the atmosphere, and thus hinders the direct electron 314 transfer from HSO₄⁻ and NO₃⁻ to other molecules through ion-molecule reactions 315 (Ferguson and Arnold, 1981). As a result, these ion groups were found to dominate 316 negative cluster ions at CHC and also other locations, such as a number of remote sites in the United States (Eisele, 1986), a boreal forest site in Finland (Ehn et al., 2010; 317 318 Bianchi et al., 2017), and the JFJ in Switzerland (Frege et al., 2017).

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Distinct diurnal patterns were observed for the (HNO₃)₀₋₁·NO₃⁻ and (H₂SO₄)₀₋₃·HSO₄⁻ ion groups. (HNO₃)₀₋₁•NO₃⁻ exhibited a relatively flat diurnal pattern (see Fig. 4) with similar fractions at daytime (35 %) and nighttime (19:00 – 07:00; 39 %). Such a diurnal pattern could result from the high EA of the NO₃ molecule (4.01 eV and an additional ~1 eV per HNO₃; Ferguson and Arnold, 1981), and its relatively abundant (usually several ppbv) parent neutral species (e.g., HNO₃ and N₂O₅) with multiple sources in the atmosphere (e.g., anthropogenic emission and lightning; Martin et al., 2007). In contrast, (H₂SO₄)₀₋₃ HSO₄ exhibited a strong diurnal variation. While the fraction of (H₂SO₄)₀₋ 3'HSO₄ remained low (2 %) during nighttime, it started to increase after sunrise (shortly after 07:00) and reached a maximum (30 %) at around 10:00. Despite an EA comparable to that of the NO₃ molecule (4.75 eV for HSO₄; Wang et al., 2000), the strong diurnal variation of (H₂SO₄)₀₋₃·HSO₄ is a result of the photochemical production of neutral H₂SO₄. The influence of neutral H₂SO₄ on (H₂SO₄)₀₋₃·HSO₄⁻ is indicated by their similar diurnal patterns (R: 0.52; see Fig. S4a). Similarly, a higher level of (NH₃)₁₋ 6'(H₂SO₄)₃₋₇'HSO₄' was only observed with the presence of abundant (H₂SO₄)₀₋₃'HSO₄' during daytime. It is also important to note that the decreases of (H₂SO₄)₀₋₃ HSO₄ and (NH₃)₁₋₆ (H₂SO₄)₃₋₇ HSO₄ at around noontime (12:00; see Fig. 4) coincided with an enhanced CS, indicating the influence of a higher ion sink in addition to the decrease in neutral H₂SO₄ concentration (Boulon et al., 2010; Frege et al., 2017).

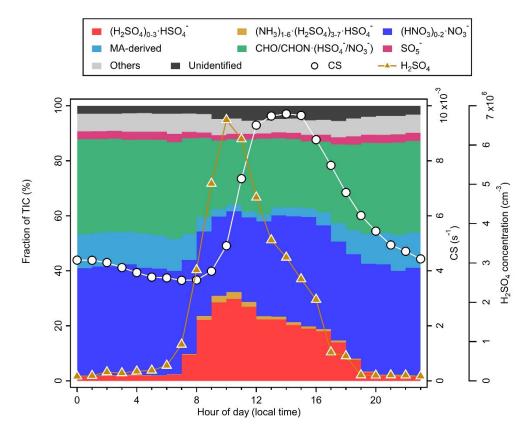


Figure 4 Diurnal variation of the negative ions (fraction of TIC), CS, and neutral H₂SO₄ concentrations at CHC, averaged over the periods when negative ions were measured (i.e., January, April, and May 2018).

The MA-derived ion group is distinguished from the CHO/CHON•(HSO4⁻/NO3⁻) ion group due to its abundance in total organic ions. This group of ions, mainly composed of C₃H₃O₄-, is formed from the deprotonation of malonic acid, with a higher EA (~4.60 eV) than that of the NO₃ molecule (4.01 eV; Ravi Kumar et al., 2005). Malonic acid, similar to HNO₃, has multiple origins in the atmosphere, such as primary and secondary anthropogenic sources, biogenic sources, and the degradation of larger organic compounds (Braban et al., 2003). It is also one of the main dicarboxylic acids which make up a substantial fraction of total carbon in aerosol particles (Kawamura and Bikkina, 2016). However, the concentration of gas-phase malonic acid is less well documented and is estimated to be in the range of 10⁷ to 10⁹ cm⁻³, up to three orders of magnitude higher than the typically reported ambient H₂SO₄ concentration (Fang et al., 2020). Thus, the fraction of the MA-derived ion group was high during nighttime (12 %; see Fig. 4). In contrast, its fraction decreased significantly (to 5 %) during daytime due to the increase of (H₂SO₄)₀₋₃•HSO₄-, which has an even higher EA (4.75 eV for HSO₄; Wang et al., 2000).

The CHO/CHON•(HSO₄-/NO₃-) ion group, with an overall molecular formula of C₂-15H₂-26O₂-13N₀-2•NO₃-/HSO₄-, constituted a significant fraction of negative ions (31 %)

359 at CHC. These organic ions are formed through the adduction between primary charge carriers, such as HSO₄ and NO₃, and neutral OOM. These OOM are likely the 360 oxidation products of the volatile organic compounds (VOC) from the Amazon, the 361 362 Altiplano, and the adjacent La Paz – El Alto metropolitan area. Their chemical composition is potentially affected by the changing air pathways covering different 363 364 VOC source regions (Aliaga et al., 2021), and by the different conditions during daytime/nighttime due to the evolution of different atmospheric layers (Beck et al., 365 366 2022). While the diurnal variation was relatively small (34 % for the nighttime and 27 % for the daytime), the ion composition of CHO/CHON•(HSO4⁻/NO3⁻) could also be 367 significantly different between daytime and nighttime due to the availability of the 368 369 charging ions (see more discussions in Section 3.2.2). A previous study from a boreal 370 forest shows that organic ions are mainly composed of CHO/CHON·NO₃- during 371 nighttime, and that the fraction of CHO/CHON·HSO₄ increases with the HSO₄ signal during daytime (Bianchi et al., 2017). This is also shown by the slightly positive 372 373 correlation between the CHO/CHON·HSO4 signal fraction and the total neutral OOM 374 concentration during daytime (R: 0.25; see Fig. S4b), whereas no clear dependence was 375 found between CHO/CHON•(HSO4-/NO3-) and the total neutral OOM concentration.

376 The SO₅ ion group, consisting of SO₅ ions and/or O₂ SO₃ cluster ions (Bork et al., 2013; Frege et al., 2017), exhibited a lower fraction (<5 %) than the aforementioned 377 ion groups during the study period. Similar to that of the (HNO₃)₀₋₁ NO₃ ion group, no 378 379 diurnal pattern was evident for the SO₅ ion group. This may be the result of its different 380 major formation pathways during daytime and nighttime (Bork et al., 2013; Frege et al., 2017). Daytime production of SO₅ ions is likely associated with photo-oxidation of 381 382 SO₂ (similar to the formation pathway of H₂SO₄; Ehn et al., 2010; Schobesberger et al., 383 2015). This is shown in the positive correlation (R: 0.46 for daytime data in Fig. S4c) 384 between the neutral H₂SO₄ concentration and the signal fraction of the SO₅ ion group. During nighttime, however, the SO₅ ion group is mainly composed of O₂ ·SO₃ cluster 385 386 ions, which are possibly formed via the oxidation of SO₂ with O₃ (producing SO₃), 387 and subsequent addition of O₂ (Bork et al., 2013).

3.2.2 Seasonalities of negative ions

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For better seasonality comparison at high-altitude CHC, we calculated the average mass 390 spectra of the negative ion groups for each season (Fig. 5 for daytime and Fig. S5 for nighttime). Distinct seasonalities (wet season, wet-to-dry transition period, and dry season) were found for the majority of the negative ion groups at CHC, including (H₂SO₄)₀₋₃ HSO₄, (NH₃)₁₋₆ (H₂SO₄)₃₋₇ HSO₄, SO₅, and organic cluster ions, as shown 394 in the averaged daytime mass spectra (Fig. 5; more detailed reason will be discussed 395 below). However, the signals of some other negative ion groups, e.g., MA-derived ions 396 and (HNO₃)₀₋₂•NO₃-, were generally stable (with differences ≤ 20 %) across the seasons. Such unclear seasonalities can be attributed to the high EA (Ferguson and Arnold, 1981; 398 Ravi Kumar et al., 2005) and/or the stability of the parent neutral species (Martin et al.,

2007; Kerminen et al., 2000; Bikkina et al., 2021). Similar patterns can also be found in the average nighttime mass spectra among the seasons (Fig. S5).

(H₂SO₄)₀₋₃•HSO₄⁻ group exhibited much higher contribution in dry season (May) than in wet season (January) and wet-to-dry transition period (April). The daytime fraction of (H₂SO₄)₀₋₃•HSO₄⁻ increased continuously from 16 % in wet, 20 % in wet-to-dry transition period, to 30 % in dry season. The maximum number of H₂SO₄ molecules increased concurrently from 2 to 4 in the cluster ions (i.e., from (H₂SO₄)₂•HSO₄⁻ to (H₂SO₄)₄•HSO₄⁻). Similar trends were also found for other H₂SO₄-related ions, such as the (NH₃)₁₋₆•(H₂SO₄)₃₋₇•HSO₄⁻ and SO₅⁻ during daytime (Fig. 5).

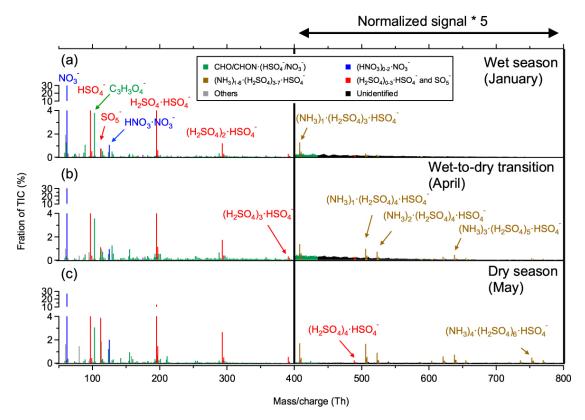


Figure 5 Mass spectra of negative ions at CHC averaged between 07:00 - 19:00 in (a) wet season (January), (b) wet-to-dry transition period (April), and (c) dry season (May). The normalized signal intensities from 400 Th to 800 Th are multiplied by a factor of 5 for better visualization.

The seasonal variations of the aforementioned H₂SO₄-related ion groups are likely due to the changes in neutral H₂SO₄ (Fig. S6) linked to the changing synoptic-scale wind patterns carrying different air masses with varying SO₂ (Bianchi et al., 2022). The air mass pathways 07_PW and 08_PW, covering the Western and Northern Altiplano plateau (see Table 1 and Fig. 2a), where active volcanic degassing of SO₂ has been reported (Moussallam et al., 2017; Carn et al., 2017), had their largest influence on CHC in dry season (i.e., May; see Fig. 2). The corresponding daytime fractions of (H₂SO₄)₀₋₃·HSO₄⁻ from these two pathways (Fig. 6a) were also the highest (27 % and 32 %, respectively). In contrast, air mass pathways 03_PW and 12_PW, originating in

the Amazon Basin and Eastern/South-Eastern Lowlands, exerted their most significant 422 423 impact on CHC in wet season (i.e., January) with lower daytime fractions of (H₂SO₄)₀-424 3 HSO₄ (13 % and 14 %, respectively). The low fractions of H₂SO₄-related cluster ions 425 in wet season are also consistent with the lower SO2 level in the Amazon Basin compared to the Altiplano plateau (Andreae et al., 1990). As for the wet-to-dry 426 427 transition period (i.e., April), 05 PW covering both the South-Eastern Lowlands and 428 Southern Altiplano plateau (where volcanic degassing is also significant; Carn et al., 429 2017) had an evident influence on CHC, resulting in a substantial level of H₂SO₄-430 related cluster ions (21 % for daytime). It is also noted that, because of the much lower nocturnal neutral H₂SO₄ concentrations, the nighttime fractions of H₂SO₄-related 431 432 cluster ions in all air mass pathways (Fig. 6b) were generally low (< 3 %) and no clear 433 seasonality was found.

- The organic cluster ion group exhibited a distinct seasonal variation than the (H₂SO₄)₀₋₃·HSO₄. The signal fraction of organic cluster ions was higher in wet season (31 % for daytime and 32 % for nighttime) than in dry season (23 % for daytime and 27 % for nighttime; Fig. 5 and Fig. S5), but it was highest for the wet-to-dry transition period (46 % for daytime and 52 % for nighttime; see Fig. 6).
- 439 The seasonal changes of organic cluster ions could be due to the combined effect of 440 different meteorological conditions and VOC from different air mass origins (see Fig. 441 2). The air masses that originated from the Amazon Basin and Lowlands (03 PW and 12 PW) showed their largest impact on CHC in wet season (i.e., January). They 442 443 contained higher fractions of organic cluster ions, which were 35 % and 34 % for 444 daytime, and 50 % and 45 % for nighttime, respectively (Fig. 6). In dry season (i.e., 445 May), however, the changes in air mass origin towards the Altiplano plateau and the 446 Pacific Ocean led to a lower content of organic cluster ions. The organic cluster ion 447 fractions for 07 PW and 08 PW (largest influence on CHC in dry season) in May were 448 23 % and 19 % for daytime, and 29 % and 27 % for nighttime, respectively. As for the 449 wet-to-dry transition period (i.e., April), due to the combined influences of biogenic 450 and anthropogenic VOC sources from 05 PW (evident impact on CHC in April), 451 covering the South-Eastern Lowlands and the Southern Altiplano plateau, the 452 corresponding organic cluster ion fractions from this air mass pathway were also the 453 highest (41 % for daytime and 53 % for nighttime; see Fig. 6).

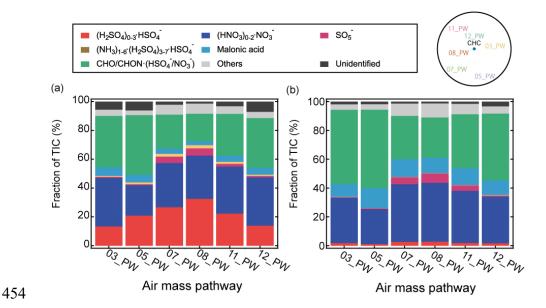


Figure 6 The fractions of the negative ion groups observed at CHC determined during the representative periods of each air mass pathway (described in Section 2.3.2) for (a) daytime (07:00 - 19:00) and (b) nighttime (19:00 - 07:00). A sketch of the horizontal profile of the air mass pathways (Fig. 2a) is shown in the upper right corner for clarity.

A further investigation of the organic ion group shows that the seasonal trends of the individual organic ions also varied (Fig. 7 for daytime and Fig. S7 for nighttime). Whereas the majority of the organic cluster ions at CHC were more abundant during wet season (Fig. 7a and Fig. S7a), fractions of CHO/CHON•HSO4⁻ increased during the dry season. The observed increases of CHO/CHON•HSO4⁻ cluster ions could be associated with the increased HSO4⁻/NO3⁻ ratios in dry season (Fig. 5 and Fig. S5). Similar increases of CHO/CHON•HSO4⁻ cluster ions were also found to relate to the ratio of HSO4⁻/NO3⁻ in a boreal forest environment (Bianchi et al., 2017). In addition, changes in OOM composition between wet and dry seasons may also play a role (Fig. 7b and Fig. S7b), as NO3⁻ tends to cluster with OOM containing hydroxyl and hydroperoxyl functional groups (Hyttinen et al., 2015) while some other observed OOM may be more efficiently charged by HSO4⁻.

The seasonal variations of the individual organic cluster ions are likely caused by different air masses (Fig. 7b and Fig. S7b). The air masses influenced by tropical rainforest vegetation from the Amazon Basin are dominated by isoprene (C₅H₈) emissions and isoprene oxidation products (Bianchi et al., 2022). This region corresponds to 03_PW and 12_PW (largest impact on CHC in wet season in January) consisting of relatively higher fractions of organic cluster ions with OOM containing 4-5 carbon atoms (50 % and 46 % for nighttime, and 29 % and 32 % for daytime, respectively). In contrast, when the air masses were more influenced by the Altiplano plateau (i.e., 05_PW, 07_PW, and 08_PW, with more anthropogenic emissions and less vegetation) in wet-to-dry transition period and dry season, organic cluster ions with 6-8 carbon atoms, potentially originating from anthropogenic sources (e.g., toluene

(C₇H₈); Huang et al., 2019; Cai et al., 2022), were of higher contributions. The signal fractions of these organic ions were thus the highest in these air mass pathways, accounting for 36-39 % for nighttime and 37-39 % for daytime. For all the air mass pathways, fractions of organic ions with more than 9 carbon atoms were relatively low (<10 %). This might be due to their lower volatilities compared to OOM with smaller carbon numbers (Donahue et al., 2012), resulting in a larger probability of them being removed during their transport to CHC (e.g., condensing on pre-existing particles).

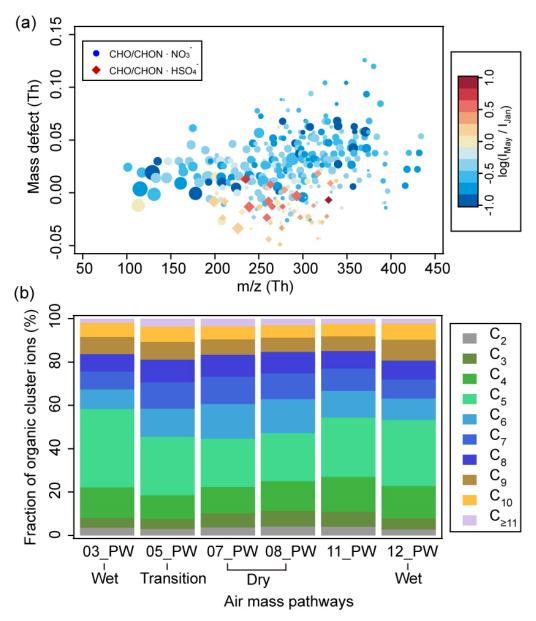


Figure 7 (a) Mass defect plot of organic cluster ions during nighttime (19:00-07:00). The color code indicates ratios (in log scale) between median signals of each ion detected in May (IMay) of dry season and January (IJan) of wet season. The marker size is proportional to the log-transformed median signals of ions in May. (b) Fraction of organic cluster ions from different air mass pathways as a function of carbon atom numbers during nighttime (19:00-07:00). A similar figure based on daytime data (07:00-19:00) is in the supplementary information (Figure S7). Note that MA-derived

ions were not included in this figure.

3.3 Positive ions

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Several positive cluster ion groups were consistently observed in February and March (i.e., wet season) during the study period. Based on their chemical composition, the positive cluster ions measured at CHC are classified into four groups (Fig. 8): (1) a series of protonated amines, including trimethylamine (C₃H₉N·H⁺), pyridine (C₅H₇N•H⁺), aniline (C₆H₇N•H⁺), and benzylamine (C₇H₉ N•H⁺); (2) organic cluster ions consisting of OOM (identified as C₃₋₂₄H₆₋₃₉O₂₋₁₂N₀₋₂) clustered with positive charge carriers such as protons (H⁺), ammonium (NH₄⁺), and aminium (NH⁺) ions; (3) contamination ions; and (4) unidentified ions (likely organic ions in higher masses; Bianchi et al., 2021). Contamination in the positive cluster ions includes ethylhexylglycerin (e.g., C₁₁H₂₄O₃•NH⁺), which is widely used in cosmetics (Aerts et al., 2016), and polydimethylsiloxane (e.g., (C₂H₆OSi)₇·NH₄⁺) possibly from instrument tubing (Bianchi et al., 2014). In contrast to the negative cluster ions, the four positive cluster ion groups were generally stable with smaller diurnal variability over the study period (Fig. 9). This is similar to the diurnal patterns determined in previous studies in a boreal forest environment (Ehn et al., 2010) and at the JFJ (Frege et al., 2017). However, due to the unavailable measurements of the corresponding neutral species (e.g., amines), the exact reason for such weak diurnal variations observed in different locations remains unclear.

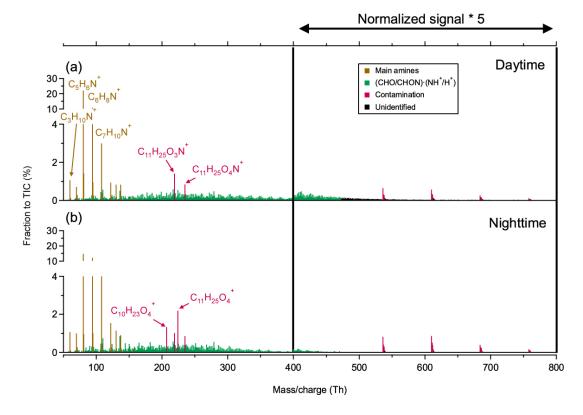


Figure 8 Averaged mass spectra of positive ions at CHC in February and March 2018 when APi-TOF was operating in positive ion mode (see section 2.2.1), during (a)

daytime (07:00 - 19:00) and **(b)** nighttime (19:00 - 07:00). The normalized signal intensities from 400 Th to 800 Th are multiplied by a factor of 5 for better visualization.

The protonated amines were the most abundant positive ion group (46 %), with no significant diurnal variations. Nighttime contributions of this ion group (47 %) were similar to its daytime contributions (45 %; Fig. 9). They also dominated the positive ion spectra observed in different environments, such as a boreal forest (Ehn et al., 2010), the JFJ (Frege et al., 2017), and the free troposphere (Schulte and Arnold, 1990). Their sources have not been fully identified (Kosyakov et al., 2020), but they are widely used as solvents and dyes (Sims et al., 1989), which may be potential sources of these ions observed at CHC.

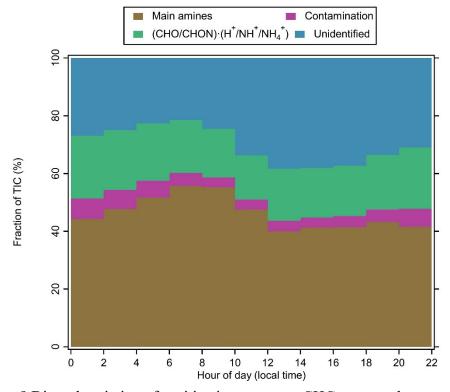


Figure 9 Diurnal variation of positive ion groups at CHC, averaged over measurements in February and March 2018 (when APi-TOF was operating in positive ion mode, see section 2.2.1).

Positive organic cluster ions were also relatively abundant (19 %) at CHC during the wet season. Similar to the negative organic ions in wet season (Fig. 6), this reflects the influence of air masses originating from the Amazon Basin and Eastern/South-Eastern Lowlands (e.g., 03_PW and 11_PW). Differences in the positive organic ion signals between nighttime (21 %) and daytime (18 %) were small, which is similar to the negative organic cluster ions (see Fig. 3). A further investigation of the relationship between these positive ions and their neutral species is, unfortunately, not possible due to the unavailability of CI-APi-TOF data in February and March caused by instrumental issues.

3.4 Potential connections between atmospheric ions and new particle formation events

During the SALTENA campaign from January to May 2018, NPF events were frequently observed at CHC (Fig. S8). While most of them occurred from April (the wet-to-dry transition period, 21 events) to May (dry season, 26 events), NPF events seldom occurred during wet season from January to March (8 events in total). This is consistent with a previous study performed at CHC (Rose et al., 2015a), which also found that NPF events mainly occurred during dry season.

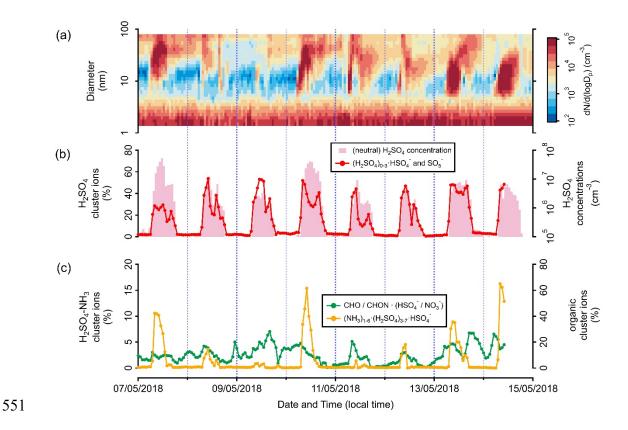


Figure 10 Time series of the **(a)** size distribution of aerosol particles (measured with NAIS and MPSS), **(b)** signal fraction of the $(H_2SO_4)_{0-3}$ •HSO₄ ion group and neutral H₂SO₄ concentration, and **(c)** signal fractions of the $(NH_3)_{1-6}$ •(H₂SO₄)₃₋₇•HSO₄ and negative organic cluster ion groups, observed at CHC from 7 to 14 May 2018 when NPF occurred frequently.

Previous field studies at high-altitude mountain sites have shown that NPF events can be triggered by different compounds, such as low-volatile neutral OOM (Bianchi et al., 2021), neutral H₂SO₄ and OOM (Bianchi et al., 2016), and H₂SO₄-NH₃ cluster ions (Frege et al., 2017). While the signal fractions of the negative organic cluster ions did not seem to have a strong correlation with the onset of the NPF events, the fractions of the (NH₃)₁₋₆•(H₂SO₄)₃₋₇•HSO₄⁻ (associated with (H₂SO₄)₀₋₃•HSO₄⁻) always increased concurrently with the number concentration of small particles (see example NPF events on, e.g., 7, 10, 13, and 14 May 2018; Fig. 10). The number of NPF days increased when

more H₂SO₄ molecules were present in the (NH₃)₁₋₆ (H₂SO₄)₃₋₇ HSO₄ cluster ion (Fig. 11a). In particular, more than half of the NPF events (28 out of the total 55 events) were observed in the presence of (NH₃)₄₋₆ (H₂SO₄)₇·HSO₄. The majority (35 events) of all the NPF events exhibited clear nucleation and growth processes (i.e., Class 1 events; the classification is defined following the approach by Yli-Juuti et al. (2009); Fig. 11b). In contrast, only Class 2 (similar to Class 1 but with less clarity) and bump events (early growth of the newly formed particles is interrupted) were observed when only (H₂SO₄)₂·HSO₄ was observed.

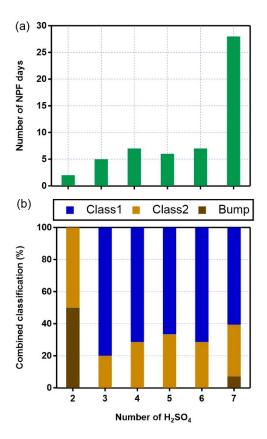


Figure 11 Connection between the maximum number of H₂SO₄ (in addition to HSO₄⁻) observed in the (H₂SO₄)₂·HSO₄⁻ and (NH₃)₁₋₆·(H₂SO₄)₃₋₇·HSO₄⁻ cluster ions during NPF events and (a) the number of NPF days; (b) proportions of different NPF classes.

Moreover, higher levels (up to an order of magnitude) of (NH₃)₁₋₆•(H₂SO₄)₃₋₇•HSO₄ and (H₂SO₄)₀₋₃•HSO₄ ions as well as negative organic cluster ions charged by HSO₄ were also observed during the NPF days (Fig. 12). In contrast, other negative ion groups (e.g., the majority of the negative organic cluster ions charged by NO₃⁻) were more abundant during the non-NPF days. Our observations indicate a potentially important role of (NH₃)₁₋₆•(H₂SO₄)₃₋₇•HSO₄⁻ cluster ions in NPF events at CHC from January to May 2018, particularly in wet-to-dry transition period and dry season.

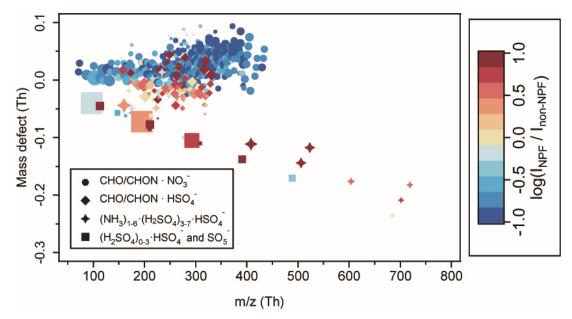


Figure 12 Mass defect plot of differences in negative cluster ion composition between NPF and non-NPF days. The negative ion composition of NPF events was averaged over all NPF days from 08:00 to 12:00 in January, April, and May 2018 (when APi-TOF was operating in negative ion mode, see section 2.2.1) at CHC. The ion composition of non-NPF days was averaged over non-NPF days from 08:00 to 12:00 for the same period. The x-axis is the exact mass of cluster ions, and the y-axis is the mass defect. The color code indicates ratios (in log scale) between median signals of each ion determined in NPF events (I_{NPF}) and non-NPF periods (I_{non-NPF}). The marker size is proportional to the log-transformed median signals of ions observed in the NPF events that occurred in January, April and May. Note that (HNO₃)₀₋₂·NO₃- cluster ions were not included here.

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The majority of the observed NPF events occurred when CHC was more impacted by air masses originating from source regions with elevated SO₂ emissions (05 PW, 07 PW, and 08 PW). This is similar to the observations from the high-altitude station JFJ (Frege et al., 2017). Moreover, consistent with the previous findings from Rose et al. (2015a), CHC was affected by the frequent arrival of air masses from the Pacific Ocean during the dry season (see also Fig. 2). And the NPF during the dry season at CHC seems to be triggered once the air masses arrive on the continent (Rose et al., 2015). However, these air masses of marine origin may not directly contain the nucleating species (also reflected in the poor connection between NPF events and the levels of methanesulfonic acid ion (CH₃SO₃⁻) or IO₃⁻; data not shown). In addition, inconsistent time series of CH₃SO₃⁻ and IO₃⁻ prohibit us from drawing further conclusions on the role of marine-derived compounds on NPF at Chacaltaya during the dry season. In addition, the fraction of large positive organic cluster ions (mass range from 500 to 800 Th; Fig. S9) was found to increase during the NPF events in the wet season. These organic cluster ions usually contained at least ten carbon atoms (Fig. S10). Such large positive organic ions have been found to contribute to NPF in the Himalayas (Bianchi et al., 2021), and thus, their contribution to NPF can not be completely ruled

out at CHC as well.

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4. Conclusion

In this study, both negative and positive atmospheric ions were measured at a highaltitude research station (CHC) in the Bolivian Andes for five months, from January to May 2018, using an APi-TOF mass spectrometer. Negative ions were mainly composed of (H₂SO₄)₀₋₃·HSO₄-, (HNO₃)₀₋₂·NO₃-, SO₅-, (NH₃)₁₋₆·(H₂SO₄)₃₋₇·HSO₄-, MA-derived, and CHO/CHON•(HSO₄-/NO₃-) ion groups. Positive ions mainly consisted of a series $(C_{3-7}H_{7-9}N^{\bullet}H^{+})$ organic protonated amines and CHO/CHON•(H⁺/NH₄⁺/NH⁺). Distinct diurnal variation was observed for the negative ions, and attributed mainly to the changes in the corresponding neutral species' concentrations and/or their EA / PA. An example is H₂SO₄-related cluster ions, the diel temporal variation of which was mainly due to the photochemical production of neutral H₂SO₄ during daytime. Strong seasonality of negative ions was also found, such as for H₂SO₄-related cluster ions owing to changes in SO₂ and the resulting neutral H₂SO₄ concentrations. The seasonal variation was mainly because of the differences in source regions of air masses arriving at CHC from wet to dry seasons. In contrast, no significant diurnal variation was observed for the positive ions. The comparison between NPF and non-NPF days infers that H₂SO₄-NH₃ cluster ions contribute to the aerosol nucleation process at CHC, particularly in wet-to-dry transition period and dry season when CHC was more impacted by air masses originating from source regions with elevated SO₂ emissions. The results further indicate that atmospheric ion composition at CHC is directly affected by air masses from different source regions.

Measurements of atmospheric ions in the field will improve understanding of atmospheric physical and chemical processes in the study regions, as the ions play important roles in atmospheric chemistry through participation in or catalysis of ion-molecule reactions and ion-induced new particle formation. Our study thus provides new insights into the chemical composition of atmospheric ions and their potential role in high-altitude NPF in the Bolivian Andes where both natural (e.g., biogenic and volcanic) and anthropogenic emissions are important.

- Data availability: The data that are involved in the figures can be found in
- doi.org/10.5281/zenodo.7271286 (Zha et al., 2022).
- Author contributions: Q. Z., W.H., and F.B. analysed the data; D.A. conducted the air
- mass history analysis; Q.Z., W.H., F.B., D.A., O.P., L.H., A.M.K., C.W., J.E., Y.G.,
- M.A., C.M., and F.B. collected the data and operated the instruments during the
- measurement campaign. Q.Z. and W.H. wrote the manuscript with contributions from
- J.C., V.S., S.C., D.W., R.K., M.A., C.M., and F.B. All authors commented on the
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