1 Measurement report: Molecular-level investigation of

atmospheric cluster ions at the tropical high-altitude research

3 station Chacaltaya (5240 m a.s.l.) in the Bolivian Andes

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29 Abstract

30 Air ions are the key components for a series of atmospheric physicochemical 31 interactions, such as ion-catalyzed reactions, ion-molecule reactions, and ion-induced 32 new particle formation. They also control atmospheric electrical properties with effects 33 on global climate. We performed molecular-level measurements of cluster ions at the high-altitude research station Chacaltaya (CHC; 5240 m a.s.l.), located in the Bolivian 34 35 Andes, from January to May 2018 using an atmospheric pressure interface time-of-36 flight mass spectrometer. The negative ions mainly consisted of (H₂SO₄)₀₋₃ HSO₄, 37 (HNO3)0-2 NO3, SO5, (NH3)1-6 (H2SO4)3-7 HSO4, malonic acid-derived, and CHO/CHON (HSO4-/NO3-) cluster ions. Their temporal variability exhibited distinct 38 39 diurnal and seasonal patterns due to the changes in the corresponding neutral species' 40 molecular properties (such as electron affinity and proton affinity) and concentrations 41 resulting from the air masses arriving at CHC from different source regions. The 42 positive ions were mainly composed of protonated amines and organic cluster ions, but 43 exhibited no clear diurnal variation. H2SO4-NH3 cluster ions likely contributed to the 44 new particle formation process, particularly during wet-to-dry transition period and dry 45 season when CHC was more impacted by air masses originating from source regions 46 with elevated SO₂ emissions. Our study provides new insights into the chemical 47 composition of atmospheric cluster ions and their role in new particle formation in the 48 high-altitude mountain environment of the Bolivian Andes.

49 **1 Introduction**

Air ions regulate the electrical properties of the atmosphere by serving as carriers of 50 51 electrical charges (Williams, 2009). They also play an important role in atmospheric 52 chemistry by participating/catalyzing ion-molecule reactions and ion-induced new 53 particle formation (NPF, Hirsikko et al., 2011). The formation of tropospheric ions is initiated through simple-structured ions, such as O⁺, N₂⁺, O⁻, and O₂⁻, mainly from 54 55 radioactive decay in the soil (e.g., radon and gamma radiation), thunderstorm activity 56 (lightning), and galactic cosmic rays (GCR). These ions can transfer their charges to 57 other compounds, leading to the subsequent production of an assortment of ions, such 58 as the bisulfate ion (HSO₄⁻), nitrate ion (NO₃⁻), hydronium ion (H₃O⁺), and ammonium 59 ion (NH4⁺; Smith and Spanel, 1995; Hirsikko et al., 2011). Depending on their sizes, air ions are usually classified into cluster ions (diameter ≤ 1.6 nm) that are charged 60 61 molecules or molecular clusters, and charged particles (diameter > 1.6 nm; Hirsikko et 62 al., 2005, 2011; Komppula et al., 2007).

63 Cluster ions exist almost always in the troposphere and can undergo frequent ion-64 molecule reactions during their lifetime (~100 seconds; Manninen et al., 2010; Hirsikko 65 et al., 2011). Their chemical composition, in addition to the initial ionization, also 66 depends on the concentrations of the parent neutral species (Eisele, 1986). Bianchi et 67 al. (2017) showed that the diurnal cycle of negative organic ions followed the variations 68 of their neutral molecules' concentrations in a boreal forest, since the higher

charged. Moreover, molecular properties of the neutral species, such as electron affinity 70 71 (EA) and proton affinity (PA), are also important for determining cluster ion 72 composition. Cluster ions derived from molecules with higher EA (e.g., HSO₄ and NO₃) 73 or PA (e.g., trimethylamine (C₃H₉N) and pyridine (C₅H₅N)) tend to obtain the ambient 74 negative or positive charge, respectively (Ferguson and Arnold, 1981; Hirsikko et al., 75 2011). Because of the strong EA or PA, it is almost unlikely that the ions derived from 76 those molecules will further transfer their charges to other neutral compounds via ion-77 molecule reactions. Thus, these negative (HSO_{4⁻} and NO_{3⁻}) and positive (C₃H₁₀N⁺ and

concentrations of neutral molecules would result in a larger probability of them being

- 78 $C_5H_6N^+$) ions are usually more abundant than other ions in the atmosphere (Eisele, 1986;
- Ehn et al., 2010; Bianchi et al., 2017; Frege et al., 2017). In contrast, charge transfer
- 80 occurs more easily for ions derived from neutral species of lower EA or PA.

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81 Atmospheric cluster ions can contribute to new particle formation (NPF) via ion-82 induced nucleation (Yu, 2010). Since the discovery of this mechanism in the first cloud 83 chamber study in the early 1900s (Wilson, 1911), ion-induced nucleation has been 84 known as an important source of atmospheric aerosol particles. Recently, a series of 85 chamber studies conducted at the CLOUD (Cosmics Leaving Outdoor Droplets) facility 86 at CERN (the European Centre for Nuclear Research) have shown that aerosol 87 nucleation rates are substantially enhanced in the presence of some specific cluster ions, 88 such as sulfuric acid – ammonia (H₂SO₄ – NH₃) cluster ions (Kirkby et al., 2011; 89 Schobesberger et al., 2015), pure H₂SO₄ cluster ions (Kirkby et al., 2011), and organic 90 cluster ions (Kirkby et al., 2016). Field measurements have also suggested the important 91 role of atmospheric ions in ion-induced nucleation (Manninen et al., 2010; Hirsikko et 92 al., 2011; Rose et al., 2018; Jokinen et al., 2018; Yan et al., 2018; Beck et al., 2021). 93 Among them, the onsets of high-altitude NPF events, compared to those occurring in 94 the lower troposphere, are often associated with more abundant cluster ions (Lee et al., 95 2003; Venzac et al., 2008; Boulon et al., 2010; Sellegri et al., 2019). Such increases are 96 due to the higher GCR intensity and lower condensation sink (CS) in the high-altitude 97 regions. As a result, potentially larger contributions of cluster ions to aerosol formation 98 would be expected (Smith and Spanel, 1995; Hirsikko et al., 2011).

99 However, the molecular-level understanding of ambient cluster ions and their influence on NPF in high-altitude environments (in the troposphere) is still very limited. Two 100 mountaintop studies in the Alps show that, depending on the air mass origins, NPF 101 102 could be triggered by sulfuric acid-ammonia clusters, or nitrate (or sulfuric acid) 103 clustering with highly oxygenated organic molecules (Bianchi et al., 2016; Frege et al., 104 2017). Another study in the Himalayas found that NPF was mainly driven by organic vapors of biogenic origin (Bianchi et al., 2021). Recently, frequent and intensive NPF 105 106 events were observed at the high-altitude research station Chacaltaya (CHC; 16.3505° 107 S, 68.1314° W; 5240 m a.s.l.) located in the Bolivian Andes (Rose et al., 2015a), but the exact mechanism and the role of cluster ions in aerosol nucleation process remain 108 109 unclear. Therefore, a detailed investigation of cluster ions at CHC, including their

- 110 molecular composition, temporal variation (diurnal and seasonal), and source regions,
- 111 is needed in order to understand their role in atmospheric processes such as NPF in the
- 112 study regions.

113 Here we present measurements of atmospheric ions from January to May 2018 at CHC. The dataset is part of the Southern hemisphere high ALTitude Experiment on particle 114 Nucleation And growth (SALTENA) field experiment campaign (Bianchi et al., 2022). 115 116 During the study period, the sampled air masses originated from various source regions, such as the Amazon Basin to the east and the Altiplano and the Pacific Ocean to the 117 west (Fig. 1a; Aliaga et al., 2021). Temporal evolution (diurnal and/or seasonal 118 119 variations) of both negative and positive ion composition are investigated, and their 120 potential connections with source regions and NPF are discussed. Our study thus adds 121 important observational information on a better understanding of atmospheric ions and 122 provides new insights into their role in high-altitude NPF in the troposphere of the 123 Bolivian Andes.

123 Bolivian Andes

124 **2** Methods

125 **2.1 Measurement site description**

The high-altitude research station CHC is ~140 m below the summit of Mount 126 127 Chacaltaya (5380 m a.s.l.) with an open view to the south and west (Andrade et al., 2015). The La Paz – El Alto metropolitan area (with 1.7 million inhabitants) is $\sim 1 - 1.6$ 128 km lower (in altitude) and ~15 km south of CHC (Fig. 1b). The seasonal meteorological 129 130 conditions at CHC depend on the cycle between the wet (November to March; wet-todry transition period in April) and dry (May to September; dry-to-wet transition period 131 in October) seasons driven by large-scale tropical circulation (Rose et al., 2015a; 132 Bianchi et al., 2022). This pattern also affects the source regions of air masses arriving 133 134 at CHC (Aliaga et al., 2021). Additionally, due to the strong diurnal cycle of the 135 planetary boundary layer (PBL) height and the thermally-induced winds in the mountainous terrain, CHC is often affected by polluted PBL transported from the La 136 Paz – El Alto metropolitan area during daytime (Wiedensohler et al., 2018) whereas at 137 138 night CHC is located in the residual layer or tropical free troposphere (Rose et al., 2015b; 139 Chauvigné et al., 2019).



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Figure 1 (a) True-color satellite image showing the location of CHC (blue circle) and
its surrounding area. The yellow line presents the boundary of the Amazon Basin. Red
triangles denote the volcanoes in this area. (b) A zoomed-in true-color satellite image
showing the distance between CHC and the La Paz – El Alto metropolitan area (orange
circle). Image sources: Esri, DigitalGlobe, GeoEye, i-cubed, USDA FSA, USGS, AEX,
Getmapping, Aerogrid, IGN, IGP, swisstopo, and the GIS User Community.

147 **2.2 Instrumentation**

148 All the instruments involved in this study were installed in a temperature-controlled 149 measurement room (at \sim 25 °C). All data are reported in local time (UTC-4).

150 2.2.1 Measurements of atmospheric cluster ions

151 The composition of cluster ions was measured by an atmospheric pressure interface time-of-flight mass spectrometer (APi-TOF, Aerodyne Research Inc. & Tofwerk AG). 152 153 The APi-TOF consists of an atmospheric pressure interface (APi) module and a timeof-flight (TOF) mass spectrometer. The APi module allows the instrument to sample 154 ions from ambient air directly. The positive or negative ions within the sampled airflow 155 are focused and guided by two quadrupoles and an ion lens, with a gradually decreasing 156 pressure (from atmospheric pressure to $\sim 10^{-4}$ mbar), before entering the TOF mass 157 spectrometer ($\sim 10^{-6}$ mbar). A more detailed description of this instrument is given in 158 159 Junninen et al. (2010). In this study, ambient air was sampled through a \sim 1.5 m stainless 160 steel tube with a total sample flow of 14 standard liters per minute (SLPM) to ensure 161 laminar flow during sampling, and 0.8 SLPM of the total flow entered the APi-TOF.

162 During the wet season, the APi-TOF was first operated in negative mode to measure 163 negative cluster ions (January), and then switched to positive mode to measure positive

cluster ions (February to March). During the wet-to-dry transition period (April) and 164 dry season (May), the instrument was changed back to negative mode to investigate the 165 166 potential seasonality of negative ion composition. It is important to note that, similar to Frege et al. (2017), for better characterization of the connection between cluster ions 167 and NPF, we only included the ion data observed under the cloud-free condition in this 168 169 study (to avoid influence from, e.g., lightning activity). CHC was considered to be affected by clouds when relative humidity (RH) exceeded 95 %, as suggested by a 170 171 previous study at the same location (Rose et al., 2015a). From January to May 2018, the proportions of cloud-free hours to the total measurement time were 72 %, 78 %, 172 173 79 %, 98 %, and 98 %, respectively (Fig. S1).

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2.2.2 Measurements of H₂SO₄ and oxidized organic molecules

Concentrations of H₂SO₄ and oxidized organic molecules (OOM) were measured using 175 176 a nitrate ion (NO3⁻) based chemical ionization atmospheric pressure interface time-of-177 flight mass spectrometer (CI-APi-TOF, Aerodyne Research Inc. & Tofwerk AG; Jokinen et al., 2012). The instrument is a combination of the APi-TOF and a chemical 178 ionization (CI) unit, which has been widely used to measure H₂SO₄ in the atmosphere 179 (Jokinen et al., 2012; Bianchi et al., 2016; Zha et al., 2018). In this study, a soft X-ray 180 181 source (L9490, Hamamatsu) was used to charge nitric acid (HNO₃) in a sheath flow of 182 20 SLPM to produce the reagent ion, NO3⁻. H₂SO₄ and OOM in the sample flow (10 SLPM) were then charged by either proton transfer or the formation of an adduct with 183 the reagent ion during the ~200 ms residence time in the CI unit. A calibration factor of 184 185 1.5×10^{10} cm⁻³ for H₂SO₄ was determined (with sampling loss corrected) according to the approach by Kürten et al. (2012). The same calibration coefficient was adopted for 186 187 determining OOM concentrations in this study, which could result in an 188 underestimation of their concentrations due to a lower charging efficiency of OOM than 189 H_2SO_4 by NO_3^- (Hyttinen et al., 2015).

190 2.2.3 Auxiliary measurements

The number concentration and size distribution of atmospheric ions and neutral 191 192 particles were measured with a neutral cluster and air ion spectrometer (NAIS, Airel Ltd., Mirme and Mirme, 2013). The instrument can detect naturally charged air ions 193 and total particles with mobility diameters from 1.4 - 50 nm and 3 - 50 nm, respectively. 194 195 It is important to note that the size range of the NAIS was corrected based on a side-196 by-side comparison with an updated version of NAIS (designed for measurements 197 under low pressure environments; Mirme et al., (2010)) at CHC. Thus, the size range of detection was slightly different from the traditionally reported ranges (i.e., 0.8 - 42198 199 nm and 2 - 42 nm, respectively; Manninen et al., 2010). The details of the instrument 200 used can be found in Rose et al. (2017).

Particle number size distributions between 10 to 500 nm were measured by a Mobility
Particle Size Spectrometer (MPSS; Wiedensohler et al., 2012), and the data was used

for calculating the CS, which represents the loss rate of condensing vapors and clusterions on pre-existing particles (Kulmala et al., 2001).

Meteorological parameters, such as temperature, RH, and global radiation, were also measured simultaneously at CHC. Detailed descriptions can be found in Bianchi et al. (2022).

208 **2.3 Simulation of air mass origin and history**

209 To understand the source regions and transport pathways of the air masses arriving at CHC, we used the results of air mass history analysis obtained from FLEXPART-WRF 210 simulations described in Aliaga et al. (2021). In brief, a Lagrangian transport and 211 dispersion model (FLEXPART-WRF; version 3.3.2; Brioude et al., 2013) was used to 212 213 calculate the air mass history during the campaign period. The backward simulation 214 was driven by the high spatial and temporal resolution meteorological output from the 215 Weather Research and Forecasting model (WRF; version 4.0.3; Skamarock et al., 2019). 216 In the simulation, twenty thousand particles were continuously released every hour from a 10 m deep layer (0 - 10 m a.g.l.) at CHC and traced back in the atmosphere for 217 218 96 hours. The output of the FLEXPART-WRF is the source-receptor relationship (SRR, 219 in seconds), which is calculated for each geographical grid cell included in the 220 simulation. The SRR value depends on the particle's residence time and the number of 221 particles in the output grid cells. Clustering analysis was conducted by applying a series 222 of pretreatments (e.g., log-polar grid transformation and grid cell pre-processing) and a 223 k-means clustering algorithm (Lloyd, 1982) to the calculated SSR dataset (see Aliaga 224 et al. (2021) for more details).

225 2.3.1 Major air mass pathways

Six air mass pathways (PW) representing air masses arriving at CHC were determined
from the clustering analysis. They are named based on their clock positions from CHC
(e.g., 03_PW indicates the pathway with its centroid located at the 3 o'clock direction
(east, 90°) of CHC, Fig. 2a). Characteristics of these air mass pathways, such as source
region, transport distance, and transport time, were distinct from each other (Table 1).
A detailed description of the air mass pathways and their characteristics can be found
in Aliaga et al. (2021).

The influence of each air mass pathway on CHC varied with time, and was estimated by its SRR percentage (SRR[%]_{pathway}) as in equation (1):

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$$SRR[\%]_{pathway} = \frac{SRR_{pathway}}{SRR_{total}} \times 100$$
(1)

where SRR_{pathway} and SRR_{total} are the residence time of a specific air mass pathway and in total (96 hours = 345600 seconds) in the simulation, respectively.

Table 1. Overview of the six air mass pathways extracted from Aliaga et al. (2021).

Pathway	Direction to CHC	Representative source region	Transport distance (km) ¹	Transport time (hour)
			Median (25 – 75 %)	Median (25 – 75 %)
03_PW	East	Amazon Basin and Eastern/South-Eastern Lowlands	518 (413–608)	51 (45-57)
05_PW	South and Southeast	South-Eastern Lowlands and Southern Altiplano	428 (303-567)	45 (36-52)
07_PW	Southwest	The Pacific Ocean, coastal area, Western Altiplano, and La Paz – El Alto	721 (577-896)	54 (45-61)
08_PW	West	Western Altiplano and Titicaca lake, coastal area	238 (198-279)	36 (29-43)
11_PW	North and Northwest	Amazon Basin, Western Altiplano, coastal area	465 (326-563)	53 (46-59)
12_PW	North	Amazon Basin	76 (49 -95)	27 (21-33)

¹Distance between CHC and each pathway's center point (see Fig. 2a).

240 2.3.2 Identification of representative periods for each air mass 241 pathway

242 Air mass history analysis shows that the air sampled at CHC was typically a mixture of 243 multiple pathways. Thus, the cluster ion composition observed during the study period 244 was often influenced by multiple source regions concurrently. To characterize the influence of every single pathway on cluster ion composition, periods when an air mass 245 246 pathway exerted its largest impact on CHC (the highest 10% of its SRR[%]_{pathway} values; Fig. 2b) during the whole study period are identified as the representative periods of 247 248 the specific pathway. For instance, the representative periods of 03 PW (covering, e.g., the Amazon Basin) are more frequently seen during wet season (highest in January), 249 250 whereas 08 PW (covering, e.g., Altiplano region) has most of its representative periods in dry season (highest in May). Note that SRR[%]_{pathway} of any individual pathway 251 252 rarely reached 40 % due to the relatively short study period (see Fig. S2), and thus the 253 representative periods cannot be directly identified via SRR[%]_{pathway} values. In contrast, 254 such representative periods were determined by using a certain threshold (e.g., > 70 %) in a previous study at CHC, which was based on a more than 6-year dataset (Koenig et 255 256 al., 2021).



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Figure 2 Influence of the six air pathways on CHC from January to May 2018. (a)
Horizontal profile of the air mass pathways, adapted from Aliaga et al. (2021). (b)
Frequency of the representative periods for each pathway (the highest 10% of their corresponding SRR[%]_{pathway}) in different months.

262 **3 Results and discussion**

3.1 Variation in total ion count

264 During the study period, the total ion count (TIC) observed by APi-TOF at CHC varied 265 between <10 and 100 counts per second (cps; Fig. 3 and Fig. S3). Similar variations in 266 the TIC were also observed in the long-term cluster ion measurements at the high-267 altitude station Jungfraujoch in Switzerland (JFJ; 3454 m a.s.l.; Frege et al., 2017) and were attributed to the seasonal changes of ion precursor and sink. Like the winter-268 269 summer seasonality of the JFJ, CHC and its adjacent mountain areas are frequently 270 covered by snow in wet season and mostly free of snow in dry season (Bianchi et al., 2022; Koenig et al., 2021). Thus, in contrast with the generally stable GCR flux 271 (primarily controlled by the decadal scale solar cycle; Shuman et al., 2015), a reduced 272 radioactive decay from the soil and a lower ion production rate could be expected at 273 274 CHC in wet season than in dry season.

However, the TIC measured by APi-TOF was significantly higher in wet season (41 \pm 275 23 cps, mean \pm standard deviation) and the wet-to-dry transition period (56 \pm 32 cps) 276 277 than in dry season (14 ± 11 cps; Fig. S3). Considering the slight negative correlation (Pearson's correlation coefficient (R): -0.41; Fig. 3) between the TIC and CS 278 279 (representing the loss rate of condensing vapors and cluster ions on pre-existing 280 particles; Kulmala et al., 2001), the observed TIC fluctuation may be related (at least partially) to the varying CS ($\sim 1 \times 10^{-4} \text{ s}^{-1}$ in wet season to $\sim 5 \times 10^{-2} \text{ s}^{-1}$ in the dry season). 281 282 Moreover, the cluster ions measured by APi-TOF usually account for only a small 283 fraction of the total atmospheric ions (Rose et al., 2018). Changes in the fraction of 284 small ions in total atmospheric ions can potentially lead to a fluctuation in TIC (Frege

et al., 2017). This is illustrated in Figure 3 that a smaller TIC determined from APi-TOF
is associated with a lower fraction of smaller ions (< 2 nm) observed by NAIS (mostly
cluster ions). However, for better characterization of the influences of different ion
composition on CHC and their diurnal and seasonal relative changes, we normalized
the observed ion signal to the TIC for APi-TOF measurements.



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Figure 3 Correlation between the TIC measured by APi-TOF and condensation sink, colored by the fraction of small ions (defined as concentrations of ions with diameter <2 nm to the total ion concentrations) determined from NAIS data. Data are shown in the time resolution of 1 hour.

295 **3.2 Negative ions**

3.2.1 Main negative ions and their diurnal variation

297 A number of negative ions were consistently observed at CHC throughout the study 298 period (i.e., in the wet, wet-to-dry transition, and dry seasons). Based on the chemical 299 composition of the observed negative ions, we classified them into eight groups as 300 follows: sulfuric acid ((H₂SO₄)₀₋₃·HSO₄⁻), nitric acid ((HNO₃)₀₋₂·NO₃⁻), SO₅⁻, sulfuric 301 acid-ammonia ((NH₃)₁₋₆·(H₂SO₄)₃₋₇·HSO₄⁻), malonic acid-derived (MA-derived; 302 including C₃H₃O₄⁻, C₃H₄O₄⁻NO₃⁻, and C₃H₄O₄⁻HSO₄⁻), oxidized organic molecules (CHO/CHON (HSO_4^{-}/NO_3^{-})), others (other identified negative ions, such as IO_3^{-}), and 303 304 unidentified ions. The campaign-average diurnal variations of these eight negative ion 305 groups are shown in Figure 4. Different diurnal patterns of each negative ion group 306 were observed, mainly due to changes in concentrations of the parent neutral species and their unique physicochemical properties (e.g., the EA and PA of molecules; 307 Ferguson and Arnold, 1981; Bianchi et al., 2017; Hirsikko et al., 2011). 308

309 (HNO₃)₀₋₁·NO₃⁻ and (H₂SO₄)₀₋₃·HSO₄⁻ were among the highest in the signal of all negative ion groups in all seasons, making up 37 % (whole day) and 20 % (daytime; 310 311 07:00 - 19:00, and hereafter) of negative ions at CHC during the study period, 312 respectively. The EA of their neutral molecules (HSO4 and NO3) is higher than that of 313 most of the neutral species in the atmosphere, and thus hinders the direct electron 314 transfer from HSO4⁻ and NO3⁻ to other molecules through ion-molecule reactions 315 (Ferguson and Arnold, 1981). As a result, these ion groups were found to dominate 316 negative cluster ions at CHC and also other locations, such as a number of remote sites in the United States (Eisele, 1986), a boreal forest site in Finland (Ehn et al., 2010; 317 318 Bianchi et al., 2017), and the JFJ in Switzerland (Frege et al., 2017).

319 Distinct diurnal patterns were observed for the (HNO₃)₀₋₁·NO₃⁻ and (H₂SO₄)₀₋₃·HSO₄⁻ 320 ion groups. (HNO₃)₀₋₁·NO₃⁻ exhibited a relatively flat diurnal pattern (see Fig. 4) with 321 similar fractions at daytime (35 %) and nighttime (19:00 - 07:00; 39 %). Such a diurnal 322 pattern could result from the high EA of the NO₃ molecule (4.01 eV and an additional 323 ~1 eV per HNO₃; Ferguson and Arnold, 1981), and its relatively abundant (usually several ppbv) parent neutral species (e.g., HNO₃ and N₂O₅) with multiple sources in the 324 325 atmosphere (e.g., anthropogenic emission and lightning; Martin et al., 2007). In contrast, 326 $(H_2SO_4)_{0-3}$ HSO₄ exhibited a strong diurnal variation. While the fraction of $(H_2SO_4)_{0-3}$ 327 3'HSO4⁻ remained low (2%) during nighttime, it started to increase after sunrise (shortly 328 after 07:00) and reached a maximum (30%) at around 10:00. Despite an EA comparable 329 to that of the NO₃ molecule (4.75 eV for HSO₄; Wang et al., 2000), the strong diurnal 330 variation of $(H_2SO_4)_{0-3}$ HSO₄ is a result of the photochemical production of neutral H2SO4. The influence of neutral H2SO4 on (H2SO4)0-3 HSO4 is indicated by their 331 332 similar diurnal patterns (R: 0.52; see Fig. S4a). Similarly, a higher level of (NH₃)₁₋ 333 6'(H2SO4)3-7'HSO4' was only observed with the presence of abundant (H2SO4)0-3'HSO4' 334 during daytime. It is also important to note that the decreases of $(H_2SO_4)_{0-3}$ HSO₄ and (NH₃)₁₋₆ (H₂SO₄)₃₋₇ HSO₄ at around noontime (12:00; see Fig. 4) coincided with an 335 enhanced CS, indicating the influence of a higher ion sink in addition to the decrease 336 337 in neutral H₂SO₄ concentration (Boulon et al., 2010; Frege et al., 2017).



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Figure 4 Diurnal variation of the negative ions (fraction of TIC), CS, and neutral H₂SO₄
concentrations at CHC, averaged over the periods when negative ions were measured
(i.e., January, April, and May 2018).

The MA-derived ion group is distinguished from the CHO/CHON (HSO4⁻/NO3⁻) ion 342 group due to its abundance in total organic ions. This group of ions, mainly composed 343 344 of $C_3H_3O_4$, is formed from the deprotonation of malonic acid, with a higher EA (~4.60 eV) than that of the NO₃ molecule (4.01 eV; Ravi Kumar et al., 2005). Malonic acid, 345 similar to HNO₃, has multiple origins in the atmosphere, such as primary and secondary 346 347 anthropogenic sources, biogenic sources, and the degradation of larger organic 348 compounds (Braban et al., 2003). It is also one of the main dicarboxylic acids which 349 make up a substantial fraction of total carbon in aerosol particles (Kawamura and Bikkina, 2016). However, the concentration of gas-phase malonic acid is less well 350 documented and is estimated to be in the range of 10^7 to 10^9 cm⁻³, up to three orders of 351 magnitude higher than the typically reported ambient H₂SO₄ concentration (Fang et al., 352 353 2020). Thus, the fraction of the MA-derived ion group was high during nighttime (12%; see Fig. 4). In contrast, its fraction decreased significantly (to 5 %) during daytime due 354 355 to the increase of (H₂SO₄)₀₋₃ HSO₄, which has an even higher EA (4.75 eV for HSO₄; 356 Wang et al., 2000).

The CHO/CHON•(HSO4⁻/NO3⁻) ion group, with an overall molecular formula of C₂₋ 15H₂₋₂₆O₂₋₁₃N₀₋₂•NO3⁻/HSO4⁻, constituted a significant fraction of negative ions (31 %) at CHC. These organic ions are formed through the adduction between primary charge 360 carriers, such as HSO₄ and NO₃, and neutral OOM. These OOM are likely the oxidation products of the volatile organic compounds (VOC) from the Amazon, the 361 362 Altiplano, and the adjacent La Paz - El Alto metropolitan area. Their chemical 363 composition is potentially affected by the changing air pathways covering different VOC source regions (Aliaga et al., 2021), and by the different conditions during 364 365 davtime/nighttime due to the evolution of different atmospheric layers (Beck et al., 2022). While the diurnal variation was relatively small (34 % for the nighttime and 27 % 366 367 for the daytime), the ion composition of CHO/CHON (HSO4⁻/NO3⁻) could also be significantly different between daytime and nighttime due to the availability of the 368 charging ions (see more discussions in Section 3.2.2). A previous study from a boreal 369 370 forest shows that organic ions are mainly composed of CHO/CHON·NO3⁻ during 371 nighttime, and that the fraction of CHO/CHON·HSO4⁻ increases with the HSO4⁻ signal 372 during daytime (Bianchi et al., 2017). This is also shown by the slightly positive correlation between the CHO/CHON·HSO4⁻ signal fraction and the total neutral OOM 373 374 concentration during daytime (R: 0.25; see Fig. S4b), whereas no clear dependence was 375 found between CHO/CHON (HSO4-/NO3-) and the total neutral OOM concentration.

376 The SO₅⁻ ion group, consisting of SO₅⁻ ions and/or O₂·SO₃⁻ cluster ions (Bork et al., 377 2013; Frege et al., 2017), exhibited a lower fraction (<5 %) than the aforementioned ion groups during the study period. Similar to that of the (HNO₃)₀₋₁·NO₃⁻ ion group, no 378 379 diurnal pattern was evident for the SO⁵ ion group. This may be the result of its different major formation pathways during daytime and nighttime (Bork et al., 2013; Frege et al., 380 381 2017). Daytime production of SO₅⁻ ions is likely associated with photo-oxidation of SO₂ (similar to the formation pathway of H₂SO₄; Ehn et al., 2010; Schobesberger et al., 382 383 2015). This is shown in the positive correlation (R: 0.46 for daytime data in Fig. S4c) 384 between the neutral H₂SO₄ concentration and the signal fraction of the SO₅⁻ ion group. 385 During nighttime, however, the SO_5^- ion group is mainly composed of $O_2^-SO_3^-$ cluster ions, which are possibly formed via the oxidation of SO_2 with O_3^- (producing SO_3^-), 386 387 and subsequent addition of O₂ (Bork et al., 2013).

388 3.2.2 Seasonalities of negative ions

389 For better seasonality comparison at high-altitude CHC, we calculated the average mass 390 spectra of the negative ion groups for each season (Fig. 5 for daytime and Fig. S5 for 391 nighttime). Distinct seasonalities (wet season, wet-to-dry transition period, and dry 392 season) were found for the majority of the negative ion groups at CHC, including 393 (H2SO4)0-3 HSO4, (NH3)1-6 (H2SO4)3-7 HSO4, SO5, and organic cluster ions, as shown 394 in the averaged daytime mass spectra (Fig. 5; more detailed reason will be discussed below). However, the signals of some other negative ion groups, e.g., MA-derived ions 395 396 and $(HNO_3)_{0-2}$, were generally stable (with differences ≤ 20 %) across the seasons. 397 Such unclear seasonalities can be attributed to the high EA (Ferguson and Arnold, 1981; 398 Ravi Kumar et al., 2005) and/or the stability of the parent neutral species (Martin et al., 399 2007; Kerminen et al., 2000; Bikkina et al., 2021). Similar patterns can also be found 400 in the average nighttime mass spectra among the seasons (Fig. S5).

401 $(H_2SO_4)_{0-3}$ ·HSO₄⁻ group exhibited much higher contribution in dry season (May) than 402 in wet season (January) and wet-to-dry transition period (April). The daytime fraction 403 of $(H_2SO_4)_{0-3}$ ·HSO₄⁻ increased continuously from 16 % in wet, 20 % in wet-to-dry 404 transition period, to 30 % in dry season. The maximum number of H₂SO₄ molecules 405 increased concurrently from 2 to 4 in the cluster ions (i.e., from $(H_2SO_4)_2$ ·HSO₄⁻ to 406 $(H_2SO_4)_4$ ·HSO₄⁻). Similar trends were also found for other H₂SO₄-related ions, such as 407 the $(NH_3)_{1-6}$ · $(H_2SO_4)_{3-7}$ ·HSO₄⁻ and SO₅⁻ during daytime (Fig. 5).



408

Figure 5 Mass spectra of negative ions at CHC averaged between 07:00 – 19:00 in (a)
wet season (January), (b) wet-to-dry transition period (April), and (c) dry season (May).
The normalized signal intensities from 400 Th to 800 Th are multiplied by a factor of 5
for better visualization.

413 The seasonal variations of the aforementioned H₂SO₄-related ion groups are likely due 414 to the changes in neutral H₂SO₄ (Fig. S6) linked to the changing synoptic-scale wind 415 patterns carrying different air masses with varying SO₂ (Bianchi et al., 2022). The air mass pathways 07 PW and 08 PW, covering the Western and Northern Altiplano 416 417 plateau (see Table 1 and Fig. 2a), where active volcanic degassing of SO₂ has been 418 reported (Moussallam et al., 2017; Carn et al., 2017), had their largest influence on 419 CHC in dry season (i.e., May; see Fig. 2). The corresponding daytime fractions of 420 (H₂SO₄)₀₋₃·HSO₄⁻ from these two pathways (Fig. 6a) were also the highest (27 % and 421 32 %, respectively). In contrast, air mass pathways 03 PW and 12 PW, originating in the Amazon Basin and Eastern/South-Eastern Lowlands, exerted their most significant 422

impact on CHC in wet season (i.e., January) with lower daytime fractions of (H₂SO₄)₀₋ 423 3⁻HSO₄⁻ (13 % and 14 %, respectively). The low fractions of H₂SO₄-related cluster ions 424 425 in wet season are also consistent with the lower SO2 level in the Amazon Basin compared to the Altiplano plateau (Andreae et al., 1990). As for the wet-to-dry 426 427 transition period (i.e., April), 05 PW covering both the South-Eastern Lowlands and 428 Southern Altiplano plateau (where volcanic degassing is also significant; Carn et al., 429 2017) had an evident influence on CHC, resulting in a substantial level of H₂SO₄-430 related cluster ions (21 % for daytime). It is also noted that, because of the much lower 431 nocturnal neutral H₂SO₄ concentrations, the nighttime fractions of H₂SO₄-related cluster ions in all air mass pathways (Fig. 6b) were generally low (< 3 %) and no clear 432 433 seasonality was found.

The organic cluster ion group exhibited a distinct seasonal variation than the $(H_2SO_4)_{0-3}$. 435 $_{3}$ ·HSO₄⁻. The signal fraction of organic cluster ions was higher in wet season (31 % for 436 daytime and 32 % for nighttime) than in dry season (23 % for daytime and 27 % for 437 nighttime; Fig. 5 and Fig. S5), but it was highest for the wet-to-dry transition period 438 (46 % for daytime and 52 % for nighttime; see Fig. 6).

439 The seasonal changes of organic cluster ions could be due to the combined effect of 440 different meteorological conditions and VOC from different air mass origins (see Fig. 441 2). The air masses that originated from the Amazon Basin and Lowlands (03 PW and 442 12 PW) showed their largest impact on CHC in wet season (i.e., January). They contained higher fractions of organic cluster ions, which were 35 % and 34 % for 443 444 daytime, and 50 % and 45 % for nighttime, respectively (Fig. 6). In dry season (i.e., 445 May), however, the changes in air mass origin towards the Altiplano plateau and the 446 Pacific Ocean led to a lower content of organic cluster ions. The organic cluster ion 447 fractions for 07 PW and 08 PW (largest influence on CHC in dry season) in May were 23 % and 19 % for daytime, and 29 % and 27 % for nighttime, respectively. As for the 448 449 wet-to-dry transition period (i.e., April), due to the combined influences of biogenic 450 and anthropogenic VOC sources from 05 PW (evident impact on CHC in April), 451 covering the South-Eastern Lowlands and the Southern Altiplano plateau, the 452 corresponding organic cluster ion fractions from this air mass pathway were also the 453 highest (41 % for daytime and 53 % for nighttime; see Fig. 6).



454

Figure 6 The fractions of the negative ion groups observed at CHC determined during the representative periods of each air mass pathway (described in Section 2.3.2) for (a) daytime (07:00 - 19:00) and (b) nighttime (19:00 - 07:00). A sketch of the horizontal profile of the air mass pathways (Fig. 2a) is shown in the upper right corner for clarity.

459 A further investigation of the organic ion group shows that the seasonal trends of the 460 individual organic ions also varied (Fig. 7 for daytime and Fig. S7 for nighttime). Whereas the majority of the organic cluster ions at CHC were more abundant during 461 wet season (Fig. 7a and Fig. S7a), fractions of CHO/CHON·HSO4⁻ increased during the 462 dry season. The observed increases of CHO/CHON·HSO4⁻ cluster ions could be 463 464 associated with the increased HSO₄⁻/NO₃⁻ ratios in dry season (Fig. 5 and Fig. S5). 465 Similar increases of CHO/CHON·HSO4⁻ cluster ions were also found to relate to the 466 ratio of HSO₄/NO₃⁻ in a boreal forest environment (Bianchi et al., 2017). In addition, 467 changes in OOM composition between wet and dry seasons may also play a role (Fig. 7b and Fig. S7b), as NO₃⁻ tends to cluster with OOM containing hydroxyl and 468 469 hydroperoxyl functional groups (Hyttinen et al., 2015) while some other observed 470 OOM may be more efficiently charged by HSO4⁻.

471 The seasonal variations of the individual organic cluster ions are likely caused by different air masses (Fig. 7b and Fig. S7b). The air masses influenced by tropical 472 473 rainforest vegetation from the Amazon Basin are dominated by isoprene (C_5H_8) 474 emissions and isoprene oxidation products (Bianchi et al., 2022). This region corresponds to 03 PW and 12 PW (largest impact on CHC in wet season in January) 475 476 consisting of relatively higher fractions of organic cluster ions with OOM containing 4-5 carbon atoms (50 % and 46 % for nighttime, and 29 % and 32 % for daytime, 477 478 respectively). In contrast, when the air masses were more influenced by the Altiplano plateau (i.e., 05 PW, 07 PW, and 08 PW, with more anthropogenic emissions and less 479 480 vegetation) in wet-to-dry transition period and dry season, organic cluster ions with 6-8 carbon atoms, potentially originating from anthropogenic sources (e.g., toluene 481

482 (C₇H₈); Huang et al., 2019; Cai et al., 2022), were of higher contributions. The signal 483 fractions of these organic ions were thus the highest in these air mass pathways, 484 accounting for 36-39 % for nighttime and 37-39 % for daytime. For all the air mass 485 pathways, fractions of organic ions with more than 9 carbon atoms were relatively low 486 (<10 %). This might be due to their lower volatilities compared to OOM with smaller 487 carbon numbers (Donahue et al., 2012), resulting in a larger probability of them being 488 removed during their transport to CHC (e.g., condensing on pre-existing particles).





Figure 7 (a) Mass defect plot of organic cluster ions during nighttime (19:00-07:00). The color code indicates ratios (in log scale) between median signals of each ion detected in May (IMay) of dry season and January (IJan) of wet season. The marker size is proportional to the log-transformed median signals of ions in May. (b) Fraction of organic cluster ions from different air mass pathways as a function of carbon atom numbers during nighttime (19:00-07:00). A similar figure based on daytime data (07:00-19:00) is in the supplementary information (Figure S7). Note that MA-derived

497 ions were not included in this figure.

498 **3.3 Positive ions**

Several positive cluster ion groups were consistently observed in February and March 499 (i.e., wet season) during the study period. Based on their chemical composition, the 500 positive cluster ions measured at CHC are classified into four groups (Fig. 8): (1) a 501 502 series of protonated amines, including trimethylamine (C₃H₉N·H⁺), pyridine 503 (C₅H₇N·H⁺), aniline (C₆H₇N·H⁺), and benzylamine (C₇H₉ N·H⁺); (2) organic cluster 504 ions consisting of OOM (identified as C3-24H6-39O2-12N0-2) clustered with positive 505 charge carriers such as protons (H^+) , ammonium (NH_4^+) , and aminium (NH^+) ions; (3) 506 contamination ions; and (4) unidentified ions (likely organic ions in higher masses; 507 Bianchi et al., 2021). Contamination in the positive cluster ions includes ethylhexylglycerin (e.g., C₁₁H₂₄O₃·NH⁺), which is widely used in cosmetics (Aerts et 508 al., 2016), and polydimethylsiloxane (e.g., (C₂H₆OSi)₇·NH₄⁺) possibly from instrument 509 510 tubing (Bianchi et al., 2014). In contrast to the negative cluster ions, the four positive cluster ion groups were generally stable with smaller diurnal variability over the study 511 period (Fig. 9). This is similar to the diurnal patterns determined in previous studies in 512 a boreal forest environment (Ehn et al., 2010) and at the JFJ (Frege et al., 2017). 513 514 However, due to the unavailable measurements of the corresponding neutral species 515 (e.g., amines), the exact reason for such weak diurnal variations observed in different locations remains unclear. 516



517

518 Figure 8 Averaged mass spectra of positive ions at CHC in February and March 2018 519 when APi-TOF was operating in positive ion mode (see section 2.2.1), during (a)

520 daytime (07:00 - 19:00) and **(b)** nighttime (19:00 - 07:00). The normalized signal 521 intensities from 400 Th to 800 Th are multiplied by a factor of 5 for better visualization.

522 The protonated amines were the most abundant positive ion group (46 %), with no 523 significant diurnal variations. Nighttime contributions of this ion group (47 %) were

524 similar to its daytime contributions (45 %; Fig. 9). They also dominated the positive

- 525 ion spectra observed in different environments, such as a boreal forest (Ehn et al., 2010),
- 526 the JFJ (Frege et al., 2017), and the free troposphere (Schulte and Arnold, 1990). Their
- 527 sources have not been fully identified (Kosyakov et al., 2020), but they are widely used
- 528 as solvents and dyes (Sims et al., 1989), which may be potential sources of these ions

529 observed at CHC.



530

Figure 9 Diurnal variation of positive ion groups at CHC, averaged over measurements
in February and March 2018 (when APi-TOF was operating in positive ion mode, see

533 section 2.2.1).

Positive organic cluster ions were also relatively abundant (19 %) at CHC during the 534 535 wet season. Similar to the negative organic ions in wet season (Fig. 6), this reflects the 536 influence of air masses originating from the Amazon Basin and Eastern/South-Eastern Lowlands (e.g., 03 PW and 11 PW). Differences in the positive organic ion signals 537 538 between nighttime (21 %) and daytime (18 %) were small, which is similar to the 539 negative organic cluster ions (see Fig. 3). A further investigation of the relationship 540 between these positive ions and their neutral species is, unfortunately, not possible due 541 to the unavailability of CI-APi-TOF data in February and March caused by instrumental 542 issues.

543 3.4 Potential connections between atmospheric ions and new particle 544 formation events

545 During the SALTENA campaign from January to May 2018, NPF events were 546 frequently observed at CHC (Fig. S8). While most of them occurred from April (the 547 wet-to-dry transition period, 21 events) to May (dry season, 26 events), NPF events 548 seldom occurred during wet season from January to March (8 events in total). This is 549 consistent with a previous study performed at CHC (Rose et al., 2015a), which also 550 found that NPF events mainly occurred during dry season.



551

Figure 10 Time series of the (a) size distribution of aerosol particles (measured with NAIS and MPSS), (b) signal fraction of the $(H_2SO_4)_{0-3}$ ·HSO4⁻ ion group and neutral H₂SO₄ concentration, and (c) signal fractions of the $(NH_3)_{1-6}$ · $(H_2SO_4)_{3-7}$ ·HSO4⁻ and negative organic cluster ion groups, observed at CHC from 7 to 14 May 2018 when NPF occurred frequently.

Previous field studies at high-altitude mountain sites have shown that NPF events can 557 be triggered by different compounds, such as low-volatile neutral OOM (Bianchi et al., 558 2021), neutral H₂SO₄ and OOM (Bianchi et al., 2016), and H₂SO₄-NH₃ cluster ions 559 560 (Frege et al., 2017). While the signal fractions of the negative organic cluster ions did 561 not seem to have a strong correlation with the onset of the NPF events, the fractions of 562 the (NH₃)₁₋₆ (H₂SO₄)₃₋₇ HSO₄ (associated with (H₂SO₄)₀₋₃ HSO₄) always increased concurrently with the number concentration of small particles (see example NPF events 563 on, e.g., 7, 10, 13, and 14 May 2018; Fig. 10). The number of NPF days increased when 564

565 more H₂SO₄ molecules were present in the (NH₃)₁₋₆ (H₂SO₄)₃₋₇ HSO₄ cluster ion (Fig. 11a). In particular, more than half of the NPF events (28 out of the total 55 events) were 566 observed in the presence of (NH₃)₄₋₆·(H₂SO₄)₇·HSO₄⁻. The majority (35 events) of all 567 568 the NPF events exhibited clear nucleation and growth processes (i.e., Class 1 events; 569 the classification is defined following the approach by Yli-Juuti et al. (2009); Fig. 11b). 570 In contrast, only Class 2 (similar to Class 1 but with less clarity) and bump events (early growth of the newly formed particles is interrupted) were observed when only 571 572 (H₂SO₄)₂·HSO₄⁻ was observed.



573

Figure 11 Connection between the maximum number of H_2SO_4 (in addition to HSO_4^-) observed in the $(H_2SO_4)_2 \cdot HSO_4^-$ and $(NH_3)_{1-6} \cdot (H_2SO_4)_{3-7} \cdot HSO_4^-$ cluster ions during

576 NPF events and (a) the number of NPF days; (b) proportions of different NPF classes.

577 Moreover, higher levels (up to an order of magnitude) of $(NH_3)_{1-6} \cdot (H_2SO_4)_{3-7} \cdot HSO_4^{-1}$ 578 and $(H_2SO_4)_{0-3} \cdot HSO_4^{-1}$ ions as well as negative organic cluster ions charged by HSO_4^{-1} 579 were also observed during the NPF days (Fig. 12). In contrast, other negative ion groups 580 (e.g., the majority of the negative organic cluster ions charged by NO_3^{-1}) were more 581 abundant during the non-NPF days. Our observations indicate a potentially important 582 here $f(2HL) = f(1, 2O_3) = HSO_3^{-1}$ days in the formula of t

⁵⁸² role of (NH₃)₁₋₆•(H₂SO₄)₃₋₇•HSO₄⁻ cluster ions in NPF events at CHC from January to

583 May 2018, particularly in wet-to-dry transition period and dry season.



584

585 Figure 12 Mass defect plot of differences in negative cluster ion composition between NPF and non-NPF days. The negative ion composition of NPF events was averaged 586 587 over all NPF days from 08:00 to 12:00 in January, April, and May 2018 (when APi-588 TOF was operating in negative ion mode, see section 2.2.1) at CHC. The ion composition of non-NPF days was averaged over non-NPF days from 08:00 to 12:00 589 for the same period. The x-axis is the exact mass of cluster ions, and the y-axis is the 590 591 mass defect. The color code indicates ratios (in log scale) between median signals of 592 each ion determined in NPF events (I_{NPF}) and non-NPF periods (I_{non-NPF}). The marker 593 size is proportional to the log-transformed median signals of ions observed in the NPF 594 events that occurred in January, April and May. Note that (HNO₃)₀₋₂·NO₃⁻ cluster ions 595 were not included here.

596 The majority of the observed NPF events occurred when CHC was more impacted by 597 air masses originating from source regions with elevated SO₂ emissions (05 PW, 598 07 PW, and 08 PW). This is similar to the observations from the high-altitude station 599 JFJ (Frege et al., 2017). Moreover, consistent with the previous findings from Rose et 600 al. (2015a), CHC was affected by the frequent arrival of air masses from the Pacific 601 Ocean during the dry season (see also Fig. 2). And the NPF during the dry season at CHC seems to be triggered once the air masses arrive on the continent (Rose et al., 602 603 2015). However, these air masses of marine origin may not directly contain the 604 nucleating species (also reflected in the poor connection between NPF events and the levels of methanesulfonic acid ion (CH₃SO₃⁻) or IO₃⁻; data not shown). In addition, 605 inconsistent time series of CH₃SO₃⁻ and IO₃⁻ prohibit us from drawing further 606 607 conclusions on the role of marine-derived compounds on NPF at Chacaltaya during the 608 dry season. In addition, the fraction of large positive organic cluster ions (mass range 609 from 500 to 800 Th; Fig. S9) was found to increase during the NPF events in the wet 610 season. These organic cluster ions usually contained at least ten carbon atoms (Fig. S10). 611 Such large positive organic ions have been found to contribute to NPF in the Himalayas 612 (Bianchi et al., 2021), and thus, their contribution to NPF can not be completely ruled

613 out at CHC as well.

614 **4. Conclusion**

In this study, both negative and positive atmospheric ions were measured at a high-615 altitude research station (CHC) in the Bolivian Andes for five months, from January to 616 617 May 2018, using an APi-TOF mass spectrometer. Negative ions were mainly composed of (H2SO4)0-3'HSO4, (HNO3)0-2'NO3, SO5, (NH3)1-6'(H2SO4)3-7'HSO4, MA-derived, 618 619 and CHO/CHON (HSO4⁻/NO3⁻) ion groups. Positive ions mainly consisted of a series 620 $(C_{3-7}H_{7-9}N^{\bullet}H^{+})$ organic of protonated amines and cluster ions 621 CHO/CHON•(H⁺/NH⁺/NH⁺). Distinct diurnal variation was observed for the negative 622 ions, and attributed mainly to the changes in the corresponding neutral species' concentrations and/or their EA / PA. An example is H₂SO₄-related cluster ions, the diel 623 624 temporal variation of which was mainly due to the photochemical production of neutral 625 H₂SO₄ during daytime. Strong seasonality of negative ions was also found, such as for H2SO4-related cluster ions owing to changes in SO2 and the resulting neutral H2SO4 626 627 concentrations. The seasonal variation was mainly because of the differences in source 628 regions of air masses arriving at CHC from wet to dry seasons. In contrast, no 629 significant diurnal variation was observed for the positive ions. The comparison between NPF and non-NPF days infers that H₂SO₄-NH₃ cluster ions contribute to the 630 631 aerosol nucleation process at CHC, particularly in wet-to-dry transition period and dry 632 season when CHC was more impacted by air masses originating from source regions 633 with elevated SO₂ emissions. The results further indicate that atmospheric ion 634 composition at CHC is directly affected by air masses from different source regions.

635 Measurements of atmospheric ions in the field will improve understanding of 636 atmospheric physical and chemical processes in the study regions, as the ions play 637 important roles in atmospheric chemistry through participation in or catalysis of ion-638 molecule reactions and ion-induced new particle formation. Our study thus provides 639 new insights into the chemical composition of atmospheric ions and their potential role 640 in high-altitude NPF in the Bolivian Andes where both natural (e.g., biogenic and 641 volcanic) and anthropogenic emissions are important.

643 Data availability: The data that are involved in the figures can be found in
644 doi.org/10.5281/zenodo.7271286 (Zha et al., 2022).

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mass history analysis; Q.Z., W.H., F.B., D.A., O.P., L.H., A.M.K., C.W., J.E., Y.G.,
M.A., C.M., and F.B. collected the data and operated the instruments during the
measurement campaign. Q.Z. and W.H. wrote the manuscript with contributions from
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