



Characteristics of particulate-bound *n*-alkanes indicating sources of PM2.5 in Beijing, China

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- 11 Abstract. The characteristics of *n*-alkanes and the contributions of various sources of fine particulate matter (PM2.5) in the
- 12 atmosphere in Beijing were investigated. PM2.5 samples were collected at Minzu University of China between November
- 13 2020 and October 2021, and n-alkanes in the samples were analyzed by gas chromatography mass spectrometry. A positive
- 14 matrix factorization analysis model and source indices (the main carbon peaks, carbon preference indices, and plant wax
- 15 contribution ratios) were used to identify the sources of n-alkanes, determine the contributions of different sources, and
- explain the differences. The *n*-alkane concentrations were 4.51–153 ng/m, (mean 32.7 ng/m³), and the particulate-bound *n*-
- 17 alkane and PM2.5 concentrations varied in parallel. There were marked seasonal and diurnal differences in the n-alkane
- 18 concentrations (p<0.01). The *n*-alkane concentrations in the different seasons decreased in the order
- 19 winter>spring>summer>fall. The mean concentration of each homolog was higher at night than in the day in all seasons.
- 20 Particulate-bound n-alkanes were supplied by common anthropogenic and biogenic sources, and fossil fuel combustion was
- 21 the dominant contributor. The positive matrix factorization model results indicated five sources of *n*-alkanes in PM2.5, which
- 22 were coal combustion, diesel vehicle emissions, gasoline vehicle emissions, higher plants, and dust. Vehicle emissions were
- 23 the main sources of n-alkanes, contributing 57.6%. The sources of PM2.5 can be indicated by n-alkanes (i.e., using n-alkanes
- 24 as organic tracers). Air quality in Beijing needs to be improved. Vehicle exhausts strongly affect PM2.5 pollution.
- 25 Controlling vehicle exhaust emissions is key to controlling *n*-alkane and PM2.5 pollution in Beijing.

1 Introduction

- 27 Serious air pollution in China is currently caused by a combination of haze and photochemical smog. The effects of haze on
- air quality are more obvious than the effects of photochemical pollution, which is relatively invisible. Haze is frequent in
- 29 urban areas with relatively dense populations and high traffic loads. Fine particulate matter is the main pollutant involved in
- 30 haze. Fine particulate matter has small particle sizes, a long atmospheric retention time, and a complex chemical composition.
- 31 Fine particulate matter is also a good substrate for chemical reactions, about which there is great concern because the
- 32 products can negatively affect the environment and human health (Wang et al., 2016; Zhu et al., 2005; Zhang, et al., 2015).
- 33 In recent years, measures such as energy structure adjustments, pollutant emission controls, and air pollution prevention have
- 34 markedly decreased atmospheric pollution and improved air quality in China. For example, the PM2.5 concentration in
- Beijing, a typical large city in China, has recently decreased markedly. The annual mean PM2.5 concentration decreased
- 36 from 73 $\mu g/m^3$ in 2016 to 33 $\mu g/m^3$ (meeting the requirement of the secondary ambient air quality standard for China, 35
- $37 \quad \mu g/m^3$) in 2021. Sources of fine particulate matter need to be better understood and controlled to decrease PM2.5 pollution,
- 38 improve air quality, and meet the primary ambient air quality standard for China (15 μg/m³) and even the World Health





Organization standard (5 µg/m³). 40 It has been found that n-alkanes are important components of organic pollutants in particulate matter and are mainly supplied 41 through anthropogenic emissions such as vehicle exhausts, fossil fuel combustion, and biomass combustion (Liu et al., 2013) 42 or through biogenic emissions such as from microorganisms and terrestrial plants (Simoneit et al., 1989; Rogge et al., 1993). 43 n-Alkanes are non-polar saturated hydrocarbons that are stable and found at high concentrations in the atmosphere. n-44 Alkanes readily adsorb to particles and can affect the environment and human health. n-Alkanes can participate in 45 atmospheric chemical reactions, and n-alkane volatility and reactivity decrease as the carbon chain length increases. The 46 products of reactions involving short-chain n-alkanes in the environment strongly contribute to secondary organic aerosol 47 formation (Michoud et al., 2012). Long-chain n-alkanes are relatively stable in the environment and generally accumulate in 48 particulate matter. The carbon number ranges, molecular compositions, and distributions of n-alkane mixtures in particulate 49 matter can be used to assess aerosol migration and particulate matter sources. The characteristics and sources of n-alkanes in 50 fine particulate matter are important parameters for developing pollutant control strategies to sustainably decrease haze 51 pollution and improve air quality. 52 Previous studies of n-alkanes in atmospheric particulate matter have mainly been focused on concentrations (Wang et al., 53 2005; Wang et al., 2006; Chen et al., 2014; Ren et al., 2017), characteristics (Simoneit et al., 2004; Li et al., 2013; Kang et al., 54 2016), and sources (Kavouras et al., 2001; Bi et al., 2003; Fu et al., 2010; Sun et al., 2021). A wide range of n-alkanes are 55 present in the atmosphere, including highly and poorly volatile n-alkanes with carbon chain lengths between eight and 40 56 (Kang et al., 2016; Aumont et al., 2012). n-Alkane concentrations between tens and hundreds of nanograms per cubic meter 57 have been found in fine particles (Ren et al., 2016; Lyu et al., 2019). The n-alkane concentration is affected by factors such 58 as meteorological conditions and contributing sources and is related to the particulate matter concentration and particle size 59 distribution. The total n-alkane concentration in particulate matter markedly varies by season, usually being higher in winter 60 and lowest in summer and fall (Lyu et al., 2016; Chen et al., 2019). n-Alkanes from different sources have different 61 molecular compositions and distributions that can be used to indicate the relative contributions of different sources of 62 particulate matter (Han et al., 2018). 63 In the past few decades, researchers in Zhengzhou (Wang et al., 2017), Guangzhou (Bi et al., 2003; Wang et al., 2016), 64 Shanghai (Lyu et al., 2016; Xu et al., 2015), Beijing (Ren et al., 2016; Lyu et al., 2019), Seoul (Kang et al., 2020), and Spain 65 (Caumo et al., 2020) have studied n-alkanes in atmospheric aerosols and determined total n-alkane concentrations, particle 66 size distributions, and the contributions of different sources. However, n-alkanes with different carbon number ranges were 67 analyzed in the different studies. Most studies were focused on n-alkanes containing <30 carbon atoms, but these do not fully 68 reflect the sources of n-alkanes in particulate matter. Air quality in Beijing is gradually improving, and exploring strategies 69 for controlling sources of fine particulate matter further requires more information about n-alkane homolog distributions and 70 variability in fine particulate matter and the relative contributions of different sources. Beijing is a large city with a dense 71 population and high traffic volumes. The sources of n-alkanes and particulate matter in Beijing require attention because of 72 the large number of volatile organic pollutants present, the high levels of vehicle exhaust emissions, and relatively severe 73 particulate matter pollution. Secondary aerosols have been found to make strong contributions to particulate pollution during 74 haze episodes in urban areas (Presto et al., 2009; Huang et al., 2014). n-Alkanes only contribute a proportion of the total 75 organic matter in particulate matter but are important contributors to particulate pollution by being important precursors of 76 secondary organic aerosols (Yang et al., 2019). n-Alkanes are also important indicators of the sources of particulate matter. 77 In this study, the concentrations of C₁₃-C₄₀ n-alkanes in atmospheric fine particulate matter in Beijing between 2020 and 78 2021 were determined. Diurnal and seasonal variations in n-alkane homolog concentrations were assessed by performing 79 diurnal and cross-seasonal sampling. The sources of n-alkanes were identified and the contributions of these sources to the 80 total n-alkane concentrations were determined using source indices and correlation models. The aim was to use n-alkanes to 81 indicate the sources of particulate matter to allow strategies for controlling particulate matter concentrations in urban areas to



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82 be developed.

2 Materials and methods

2.1 Sampling site and time

Fine particulate matter samples were collected between November 2021 and October 2022 on the roof (about 20 m above the ground) of the College of Pharmacy at the Minzu University of China (116.19° E, 39.57° N). Beijing is a typical heavily population and traffic-intensive Chinese city, with high emission intensities of nitrogen oxides and volatile organic pollutants and relatively serious fine particulate matter pollution. Haidian District is a prosperous urban area in Beijing with intense human activities and busy traffic. The sampling point in Haidian District reflected the influences of human activities and vehicle emissions on fine particulate matter concentrations. Samples were collected between the 23rd and 29th of each month during the study, but the exact sampling periods were adjusted to take into account pollution levels and the weather. Samples were collected in two periods on a sampling day. Daytime samples were collected between 07:00 and 20:00 and nighttime samples were collected between 20:30 and 06:30 the next morning. Diurnal and seasonal variations in n-alkane concentrations in fine particulate matter were investigated by collecting separate day and night samples and collecting samples in different seasons. The effects of n-alkanes on PM2.5 concentrations were assessed by comparing n-alkane concentrations on clear and hazy days.

2.2 Sample collection and pretreatment

- 98 Each fine particulate sample was collected using a TH-16A low-flow sampler (Wuhan Tianhong, Wuhan, China) containing 99 a Whatman QMA quartz fiber filter (Ø 47 mm; GE Healthcare Bio-Sciences, Pittsburgh, PA, USA) using a flow rate of 16.7 100 L/min. Before use, the quartz fiber filters were baked at 550 °C for 5 h to remove organic matter. Each filter was loosely
- 101 wrapped in aluminum foil and equilibrated for 24 h at 20 °C and 40% relative humidity and then weighed using a precision 102 electronic balance before being used to collect a sample. Once used, a filter was equilibrated for 24 h at 20 °C and 40%
- 103 relative humidity, weighed again, and then stored wrapped in aluminum foil at -20 °C.
- 104 The details for ultrasonic extraction methods used to analyze the samples of n-alkanes in PM2.5 are reported in previous 105 studies (Yang et al., 2019; Kang et al., 2020; Caumo et al., 2020). Each filter was cut into pieces and extracted by
- 106 ultrasonicating it with 15 mL of dichloromethane for 15 min. The extraction step was repeated five times, and the extracts

were combined and evaporated to 2 mL using a rotary evaporator. The extract was then transferred to a 15 mL centrifuge

- 108 tube and centrifuged at 3000 rpm for 5 min. The supernatant was evaporated just to dryness under a gentle flow of high
- 109 purity nitrogen and then redissolved in 100 µL of toluene for instrumental analysis. In the spiked recovery experiments, the
- 110 extraction recovery for *n*-alkanes range from 43.6% to 128%, the RSD for the concentrations of *n*-alkanes is 3.51%.

2.3 Instrumental analysis

- The n-alkanes (C13-C40) were analyzed qualitatively and quantitatively by gas chromatography mass spectrometry using an 112 113 Agilent 6890N-5975 system (Agilent Technologies, Santa Clara, CA, USA). n-Alkane standards (C8-C40) were purchased
- 114 from AccuStandard (New Haven, CT, USA). Separation was achieved using an Agilent J&W Scientific DB-5M column (30
- 115 m long, 0.25 mm inner diameter, 0.1 µm film thickness; Agilent Technologies). Splitless injection mode was used, and the
- 116 injection volume was 1.0 μL. The carrier gas was helium and the constant flow rate was 1.0 mL/min. The oven temperature
- 117 program started at 80 °C, which was held for 2 min, then increased at 10 °C/min to 200 °C, and then increased at 15 °C/min
- 118 to 300 °C, which was held for 30 min. The mass spectrometer was used in electron impact ionization mode and selected ion
- 119 detection mode. Ions with mass-to-charge ratios of 85 and 113 (characteristic of n-alkanes) were used to identify and



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120 quantify *n*-alkanes. The data were quantified using ChemStation software (Agilent Technologies).

2.4 Quality assurance and control

- When extracting *n*-alkanes from the fine particulate samples, blank samples were extracted with each batch of samples. The
- 123 concentration of an analyte substance in the blanks was subtracted from the concentration of the analyte in a sample during
- 124 data processing. The detection and quantification limits of the instrument were defined as three and 10 times the signal-to-
- noise ratio, respectively. The instrument detection limits for the *n*-alkanes were 1–10 pg.

2.5 Data analysis

- 127 PM2.5 data were provided by the China Meteorological Administration (cma.gov,cn/). Data analysis (statistical and other
- 128 analyses of the n-alkane data) was performed using SPSS 26.0 software (IBM, Armonk, NY, USA). Differences in the
- 129 concentrations of an *n*-alkane homolog in different groups of samples and differences in the overall *n*-alkane compositions in
- 130 different groups of samples were assessed by performing independent sample t-tests. Spearman correlations and Pearson
- correlations (two-tailed tests) were used to identify correlations between groups of data.
- 132 Source indices (the carbon maximum number (Cmax), carbon preference index (CPI), and plant wax n-alkane ratio
- 133 (WNA%)) were used to assess the n-alkane sources from the n-alkane molecular compositions and concentration
- 134 distributions. The Cmax is the homolog with the highest relative concentration in the n-alkane mixture, it is commonly used
- 135 to distinguish between the contributions of anthropogenic and natural sources of n-alkanes and is related to the degree of
- 136 thermal evolution that has affected the organic matter supplying *n*-alkanes. The CPI defined as the ratio of total odd carbon
- 137 n-alkanes to even carbon n-alkanes and was developed by Bray and Evans in 1961 (Bray et al., 1961), it can be used to
- 138 assess the contributions of anthropogenic and biogenic sources of n-alkanes and is the most commonly used empirical
- parameter for distinguishing between sources of *n*-alkanes (Marzi et al., 1993). WNA% and PNA% can be used to assess the
- relative contributions of biological and anthropogenic sources of *n*-alkanes in particulate matter (Simoneit, 1985), WNA%
- are calculated by subtraction of the average of the next higher and lower even carbon numbered homologues, while the
- 142 petrogenic n-alkane ratio (PNA%) was defined as the WNA% subtracted from 100% (Lyu et al., 2019). The source indices
- were calculated using Eqs. (1)–(3):

145 WNA% =
$$\frac{\sum (C_n - (\frac{C_{n-1} + C_{n+1}}{2}))}{\sum C_n} \times 100\%$$
 (2)

$$146 \quad PNA\% = 100\% - WNA\% \tag{3}$$

- 147 In Eq. (1), C_{2i+1} was the concentration of the *n*-alkane with odd carbon atoms range from 13-39, while C_{2i} was the
- 148 concentration of the *n*-alkane with even carbon atoms range from 14-40. In Eq. (2), C_n was the concentration of *n*-alkanes,
- taking as zero the negative value of $(C_n (\frac{C_{n-1} + C_{n+1}}{2}))$.
- 150 A positive matrix factorization (PMF) model was used to identify specific n-alkane sources and the contribution of each
- source through EPA PMF 5.0 software (USEPA). The PMF model is a factor analysis technique using multivariate statistical
- 152 methods. The PMF model is a receptor model, so can identify and determine the contributions of components of unknown
- 153 mixtures. The PMF model is one of the source resolution methods recommended by the US Environmental Protection
- 154 Agency. The PMF model does not require the complex pollutant sources to be determined and the treatment process can be
- optimized while limiting the decomposition matrix elements and sharing the rates of nonnegative matrices. The model can
- 156 use the chemical composition of particulate matter to identify the sources of particulate matter and calculate the contributions





- 157 of the different sources, so is widely used to investigate the sources of atmospheric particulate matter (Moeinaddini et al.,
- 158 2014; Liao et al., 2021; Li et al., 2021). The details of PMF have been described in the PMF 5.0 User Guide (USEPA, 2014).

159 **3 Results**

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3.1 Concentrations of *n*-alkanes

- 161 A total of 28 n-alkane homologs with carbon chain lengths of C₁₃-C₄₀ were analyzed. C₁₃-C₄₀ n-alkanes were detected in the
- 162 diurnal fine particulate matter samples collected in all seasons. Among them, C₂₁-C₃₅ n-alkanes were detected in all PM2.5
- samples, other *n*-alkanes were detected in more than half of the samples.
- 164 The n-alkane and PM2.5 concentrations in the different seasons are shown in Table 1 and temporal variations in the
- 165 concentrations are shown in Figure 1. The PM2.5 concentrations throughout the sampling period were 0–134 μg/m³, and the
- 166 mean was 32.0 μg/m³. The *n*-alkane concentrations throughout the sampling period were 4.51–153 ng/m³, and the mean was
- 167 32.7 ng/m³. Correlation analysis indicated that the n-alkane and PM2.5 concentrations significantly positively correlated
- 168 (p<0.01, r=0.313).

169 3.2 *n*-Alkane component distributions

- 170 The contributions of the individual C₁₃-C₄₀ n-alkane homologs to the total n-alkane concentrations are shown in Figure 2.
- 171 The C₁₆-C₂₅ n-alkanes were dominant in winter and the C₂₆-C₃₁ n-alkane contributions increased markedly spring, summer,
- 172 and fall.
- 173 The *n*-alkane homologs can be classed as low molecular weight (LMW), meaning *n*-alkanes with carbon chain lengths \leq 25,
- and high molecular weight (HMW), meaning n-alkanes with carbon chain lengths >25. As shown in Figure 3, LMW n-
- alkanes contributed ~60% of the total n-alkane concentrations in winter but only ~40% in spring, summer, and fall,
- indicating that there were marked differences between the compositions in winter and the other seasons.

3.3 Seasonal and diurnal differences in *n*-alkane concentrations

- 178 The C₁₃-C₄₀ n-alkane concentration distributions in the different seasons are shown in Figure 4. There were significant
- 179 differences (p<0.01) between the concentrations of various homologs in the different seasons. The mean n-alkane
- 180 concentrations for the different seasons decreased in the order winter>spring>summer>fall. The seasonal differences were
- 181 more marked for LMW than HMW n-alkanes. The concentrations of relatively short-chain n-alkanes (C16-C25) were
- markedly higher in winter than in the other seasons. The concentrations of C27, C29, C31, and C33 n-alkanes were higher than
- the concentrations of C₂₆, C₂₈, C₃₀, C₃₂, and C₃₄ n-alkanes (i.e., odd-carbon-number dominance occurred) in all of the seasons.
- 184 The C₁₃-C₄₀ n-alkane concentrations in the day and night samples are shown in Figure 5. The mean n-alkane homolog
- 185 concentrations were higher at night than in the day in all four seasons. The concentrations in the day and night were
- significantly different (p<0.01).

187 3.4 Source indices and PMF model

- 188 Source indices (Cmax, CPI, and WNA%) determined from the C13-C40 n-alkane data were used to assess the n-alkane
- 189 sources. The PMF model was used to quantify the amounts of *n*-alkanes in fine particles supplied by the different sources
- 190 and the relative contributions of the sources. The source index data for n-alkanes in the day and night samples in the different
- seasons are shown in Table 2.



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3.4.1 Source indices for *n*-alkanes

The Cmax for winter was C₂₃ but the Cmax for spring, summer, and fall was C₂₉. The mean CPI for the year the samples were collected were 1.66. The CPI was lowest in winter but higher in the day than the night in spring, summer, and fall. The mean contribution of plant wax *n*-alkanes to the total *n*-alkane concentration during the sampling period was 30.6% and the mean contribution of anthropogenic *n*-alkanes to the total *n*-alkane concentration was 69.4%. The plant wax *n*-alkane contribution was lowest in winter and markedly higher in spring, summer, and fall.

3.4.2 Results of the PMF model

- 199 According to the PMF 5.0 User Guide, the daily mean n-alkane concentrations during the sampling period and the 200 corresponding uncertainties were inputted into the PMF model to analyze the sources of n-alkanes in fine particulate matter. 201 Various numbers of factors were tested, and the optimal correlation coefficient for the relationship between the simulated and 202 observed values was found when five factors were used, the average correlation coefficient of n-alkane homologues is 0.832. 203 Q (robust) is a important parameter obtained after PMF run, it is the goodness-of-fit parameter calculated excluding points 204 not fit by the model (USEPA, 2014). In the process of running the PMF model, we got the lowest Q (robust) values when 205 selected five factors. This met the requirements to use the PMF model, EPA PMF 5.0 User Guide have stated that the lowest 206 Q (robust) value represents the most optimal solution from the multiple runs and it can be a critical parameter for choosing 207 optimal number of factors. Each factor indicated a source, and the factors could be used to identify the corresponding sources. 208 The *n*-alkane factor data given by the PMF model are shown in Figure 6.
- The PMF model indicated that the contributions of factors 1, 2, 3, 4, and 5 to the *n*-alkane concentrations were 14.8%, 26.1%, 31.5%, 18.6%, and 9.01%, respectively. The sources corresponding to the factors identified by the PMF model needed to be identified from the proportions of the different *n*-alkane homologs present, the sources corresponding to factors 2 and 3 were
- the main contributors of *n*-alkanes in particulate matter.

213 4 Discussion

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4.1 Sources and contributions of *n*-alkanes

215 n-Alkanes in PM2.5 have relatively complex sources, but different n-alkane compositions and distributions indicate different 216 sources. As shown in Figure 4, marked odd-carbon-number dominance was found in all seasons for the HMW n-alkanes, with n-alkanes with carbon chain lengths C27, C29, C31, and C33 being dominant. No odd-carbon-number dominance was 217 218 found for the LMW n-alkanes. It has previously been found that LMW n-alkanes in urban areas are mainly anthropogenic 219 (e.g., emitted during fossil fuel combustion and in vehicle exhaust gases) (Simoneit et al., 2004; Kang et al., 2016) but HMW 220 n-alkanes reflect sources such as biomass combustion and waxes in higher plants (Kawamura et al., 2003). LMW and HMW 221 n-alkane patterns can be used to identify the main sources of n-alkanes in urban areas. The n-alkane patterns in the different 222 seasons indicated that particulate-bound n-alkanes in the atmosphere in Beijing have both anthropogenic and biological 223 sources. The source indices and PMF model results further explained the sources and contributions of *n*-alkanes. 224 n-Alkane source indices are often used to identify the origins of n-alkanes. The n-alkane source indices shown in Table 2 225 indicated that anthropogenic emissions were the main contributors of particulate-bound n-alkanes in Beijing during the study 226 but that there were also biogenic emissions of particulate-bound n-alkanes. The CPI and WNA% data explained this. During 227 the sampling period, the mean CPI was 1.66, indicating that the main sources of particulate-bound n-alkanes were fossil fuel 228 combustion, plants, and biomass combustion. The mean WNA% and PNA% were 30.63% and 69.37%, respectively,

indicating that anthropogenic emissions contributed more than emissions from biota.





230 The PMF model can quantify the contributions of specific sources of n-alkanes relatively accurately. The n-alkane homolog 231 contributions to each factor identified by the PMF model were used to analyze and identify the corresponding source. The n-232 alkanes with carbon chain lengths of C₁₃-C₁₈ were dominant in factor 1. This was consistent with the n-alkane homolog 233 pattern for emissions during coal combustion found by Oros and Simoneit and Niu et al. (Oros et al., 2000; Niu et al., 2005), 234 so we concluded that factor 1 indicated n-alkanes emitted through coal combustion. Vehicle emissions are important sources 235 of n-alkanes in particulate matter in urban areas. n-Alkanes emitted by vehicles mainly have carbon-chain lengths <30 236 (Schauer et al., 2002). However, there are marked differences between the patterns of n-alkanes emitted in particulates in 237 gasoline vehicle and diesel vehicle exhaust gases. Cmax for n-alkanes is lower and the proportion of low-carbon-chain 238 length n-alkanes is higher for particulates in diesel vehicle exhaust gases than gasoline vehicle exhaust gases. This feature 239 can be used to distinguish between n-alkanes emitted by diesel and gasoline vehicles in fine particulate matter (Fujitani et al., 240 2012; Yuan et al., 2016). Factors 2 and 3 indicated n-alkanes emitted by diesel and gasoline vehicles, respectively. C27-C38 241 (i.e., high-carbon-chain-length) n-alkanes made large contributions and low-carbon-chain-length n-alkanes made small 242 contributions to the pattern for factor 4. Alves et al. (Alves et al., 2001) found that C₂₆-C₃₆ n-alkanes are mainly emitted 243 from cuticular waxes in higher plants, so factor 4 indicated n-alkanes emitted by higher plants. There was no clear n-alkane 244 homolog pattern for factor 5, but long-chain n-alkanes with carbon chain lengths >34 were dominant, so we concluded that 245 factor 5 indicated n-alkanes from mixed secondary sources, such as road dust. The contributions of the different sources to 246 the *n*-alkane concentrations are shown in Figure 7. 247 In summary, n-alkanes in airborne particulate matter in Beijing are both anthropogenic and biogenic. Vehicle exhaust 248 emissions are the main sources of n-alkanes, consistent with the current energy consumption structure in Beijing, and 249 gasoline and diesel vehicles accounted for a relatively large proportion of n-alkanes in airborne particulate matter.

4.2 Characteristics of PM2.5 and *n*-alkanes

- The mean *n*-alkane concentration during the sampling period was 32.7 ng/m³, which was lower than the C₁₉-C₃₆ *n*-alkane concentration of 282 ng/m³ found in Beijing in 2006 (Li et al., 2013) and the C₈-C₄₀ *n*-alkane concentration of 228 ng/m³ found in Shanghai in 2013 (Lyu et al., 2016). The temporal trends in the *n*-alkane concentrations were similar to the trends found in previous studies of *n*-alkanes in Beijing (Rogge et al., 1993; Li et al., 2013; Ren et al., 2019), the overall *n*-alkane
- 255 concentration being highest in winter. The seasonal pattern we found for *n*-alkanes in Beijing was similar to the pattern
- found in a previous study of C₁₆–C₃₅ *n*-alkanes in 14 Chinese cities (Wang et al., 2006).
- The *n*-alkane pattern varied by season, LMW *n*-alkanes being dominant in winter and HMW *n*-alkanes being more abundant
- in the other seasons. Cmax and WNA% explained the seasonal differences in the n-alkane patterns. In previous studies,
- 259 lower Cmax values were found for n-alkanes emitted from very mature organic matter such as coal and petroleum than for n-
- alkanes emitted from immature organic matter such as plants (Simoneit et al., 1989; Duan et al., 2010). The Cmax for n-
- alkanes in winter was C23, indicating that LMW n-alkanes were the main n-alkanes. Similar results were found by Lyu et al.
- 262 for Beijing in winter (Lyu et al., 2019). The Cmax for n-alkanes in spring, summer, and fall was C29. Ficken et al. (Ficken et
- 263 al., 2000) and Yadav et al. (Yadav et al., 2013) found that C29 n-alkanes are markers for n-alkanes emitted from the wax
- 264 layers of higher plants. Stronger n-alkane contributions will be made by plants in spring, summer, and fall than in winter
- (Rogge et al., 1993; Yadav et al., 2013). This is consistent with the results found in a study performed in Shanghai (Lyu et al.,
- 266 2016; Wang et al., 2016).
- 267 There were significant seasonal differences (p<0.01) in the concentrations of the C₁₃-C₄₀ n-alkane homologs, but the
- 268 seasonal differences were stronger for LMW n-alkanes than HMW n-alkanes. Similar results were found by Li et al. in
- Tianjin in 2010 (Li et al., 2010). The LMW *n*-alkane concentrations were markedly higher in winter than in the other seasons,
- 270 similar to the results of a study performed by Li et al. in Beijing in 2013 (Li et al., 2013). This indicated that there were





271 seasonal differences in n-alkane sources. The PMF model results shown in Figure 7 indicated that anthropogenic n-alkanes 272 strongly contributed to the total n-alkane concentration in winter. The CPI also indicated that different sources were 273 dominant in winter and in the other seasons. The lowest CPI was found for winter, indicating that particulate-bound n-274 alkanes made stronger contributions to the total n-alkane concentrations in winter than the other seasons. This may be related 275 to n-alkane emissions caused by fossil fuel combustion for heating in winter. Similar results have been found in Shanghai 276 (Lyu et al., 2016), Zhengzhou (Wang et al., 2017), southeastern Chinese cities (Chen et al., 2019), and Beijing (Kang et al., 2016). 277 278 The mean C₁₃-C₄₀ n-alkane homolog concentrations were higher at night than in the day in each season, and the differences 279 were significant (p<0.01). This would have been because the mean wind speed is lower, the boundary layer is lower, and 280 atmospheric diffusion conditions are worse at night than in the day (Yao et al., 2009). Similar results were found in 281 Liaocheng, Shandong Province (Liu et al., 2019). The differences in the n-alkane concentrations in the night and day may 282 also have been caused by differences in pollutant emissions in the night and day. Simoneit et al. found that LMW n-alkanes 283 dominate the n-alkanes emitted in diesel vehicle exhaust gases (Simoneit et al., 2004). We found markedly higher 284 concentrations of some homologs with carbon chain lengths <25 at night than during the day. This would be consistent with 285 short-chain alkane emissions from diesel vehicles in Beijing being higher at night than in the day.

4.3 PM2.5 sources in Beijing and strategies for controlling PM2.5 concentrations

286 287 During the sampling period, the mean daily PM2.5 concentration in Beijing was 32.0 µg/m³, which met the requirement of 288 the secondary ambient air quality standard for China (35.0 µg/m³). According to the Ecology and Environment Statement 289 from the Beijing Municipal Ecology and Environment Bureau (sthjj.beijing.gov.cn), the annual mean PM2.5 concentration in 290 Beijing has gradually decreased in the last five years. However, little research on n-alkanes in Beijing has been performed in 291 this period. We compared our results with the results of a previous study (Lyu et al., 2019) and found that the n-alkane 292 concentrations decreased in parallel with the PM2.5 concentrations. n-Alkanes are important molecular markers for 293 identifying the sources of PM2.5. Excluding when the PM2.5 concentration increased sharply because of meteorological 294 conditions, the PM2.5 and n-alkane concentrations varied in the same ways. A significant positive correlation was found 295 between the PM2.5 and n-alkane concentrations (p<0.01), so n-alkanes could be used as indicators of the sources of PM2.5 296 in the atmosphere. This method has been widely used to analyze sources of particulate matter (Kavouras et al., 2001; Bi et al., 297 2003). We therefore used the PMF model results for n-alkanes to identify the sources of PM2.5 and explain variations in the 298 sources. 299 The PMF model results for the contributions of the different sources shown in Figure 7 indicated that emissions in vehicle 300 exhaust gases and through coal combustion contributed up to 72.4% of PM2.5 in the sampling area throughout the sampling 301 period. This indicated that anthropogenic PM2.5 emissions are the main sources of PM2.5 in the urban study area. Emissions 302 from gasoline and diesel vehicles were the dominant anthropogenic sources, contributing 57.6% of total anthropogenic 303 PM2.5 emissions. Vehicles are the main sources of PM2.5 in urban areas and make important contributions to particulate 304 matter in the atmosphere in Beijing. Similar results were found in a previous study of PM2.5 sources in Beijing (Lv et al., 305 2021) and the results were consistent with the current energy consumption structure in Beijing (gasoline and diesel fuel make 306 large contributions to total fuel consumption). Human activities make larger contributions to PM2.5 emissions in winter than 307 the other seasons, indicating that more attention should be paid to emissions caused by fossil fuel combustion in winter than 308 the other seasons. 309 It is necessary to improve air quality in Beijing, and vehicle exhausts are key sources of PM2.5. Further improvements in 310 ambient air quality to meet stricter ambient air quality standards will require vehicle emissions to be controlled to decrease

particulate matter pollution. The number of vehicles using fossil fuels in Beijing needs to be decreased. Achieving this will

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require policies for restricting the use of vehicles using fossil fuels and the use of cleaner energy vehicles to be promoted. In summary, controlling and decreasing emissions caused by fossil fuel combustion will decrease PM2.5 emissions and improve ambient air quality in Beijing.

5 Conclusions

316 The PM2.5 concentrations and C13-C40 n-alkane concentrations in fine particulate matter between November 2020 and 317 October 2021 were determined and the concentrations were compared with concentrations found in previous studies. The 318 PM2.5 and n-alkane concentrations in Beijing have decreased in similar ways in the last five years. The mean PM2.5 319 concentration was 32.0 µg/m³, which met the secondary ambient air quality standard for China. The PM2.5 and C₁₃-C₄₀ n-320 alkane concentrations varied in similar ways and positively correlated (p<0.01), so long chain n-alkanes in particulate matter 321 can be used to assess the sources of particulate matter pollution in urban areas and to develop strategies for controlling 322 particulate matter pollution. 323 The n-alkane concentrations in the different seasons decreased in the order winter>spring>summer>fall. There were marked 324

The n-alkane concentrations in the different seasons decreased in the order winter>spring>summer>fall. There were marked seasonal and diurnal differences in the n-alkane homolog patterns and distributions. The source indices and PMF model results explained these variations in patterns and allowed the sources of n-alkanes to be identified. The source indices indicated that n-alkane concentrations in particulate matter in Beijing are affected by both anthropogenic and biogenic emissions but that anthropogenic emissions are dominant. The PMF model allowed the contributions of the sources of n-alkanes to be quantified and indicated that emissions from vehicles are currently the main sources of PM2.5 and n-alkanes in particulate matter in urban areas.

Controlling PM2.5 and n-alkane emissions from vehicles is key to decreasing PM2.5 and n-alkane pollution and improving air quality in urban areas. n-Alkanes in particulate matter can be used as organic tracers, and PMF model results can indicate the sources of PM2.5 pollution. Further research into the use of this method is required.

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337 Data availability

The date presented in this article are available from the authors upon request (junjin3799@126.com).

Author contribution

JJ conceived and designed the study, provided direct funding and helped with manuscript revision. JYY mainly conducted the sampling, sample analysis work, as well as manuscript writing and revision. Other authors helped this work by sampling and analysis. All authors read and approved the final manuscript.

343 Competing interests

344 The authors declare that they have no conflict of interest.





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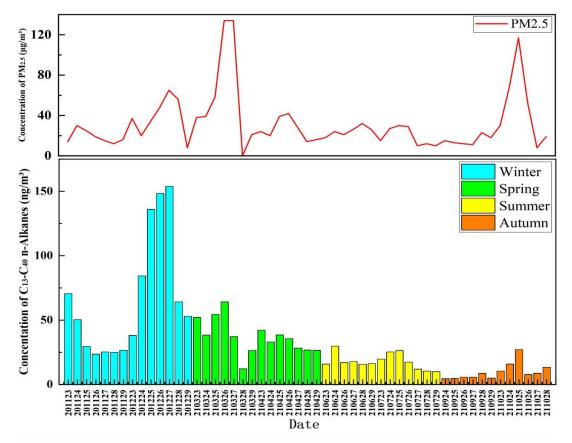


Figure 1. Temporal variations in PM2.5 and particulate-bound n-alkane concentrations during the sampling period in Beijing.



513



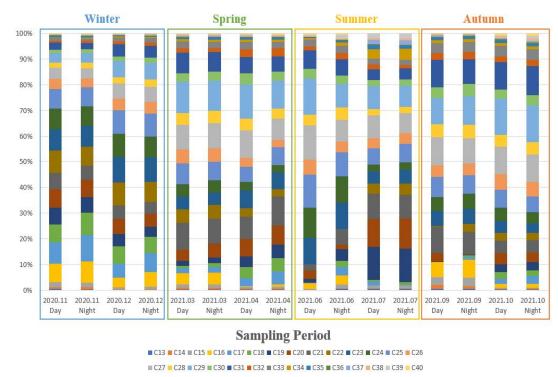


Figure 2. Contributions of particulate-bound n-alkane homologs to the total n-alkane concentrations in the day and night samples in the different seasons of Beijing.





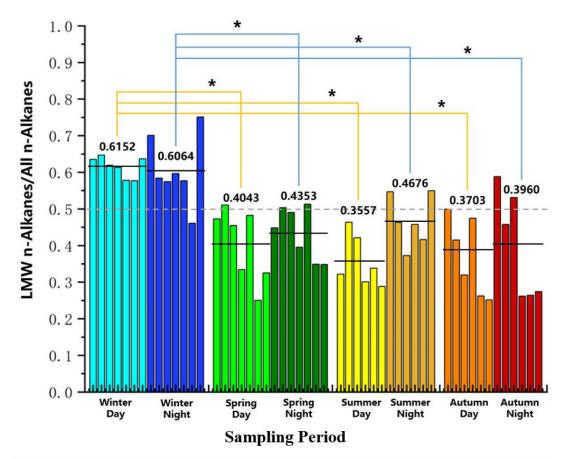
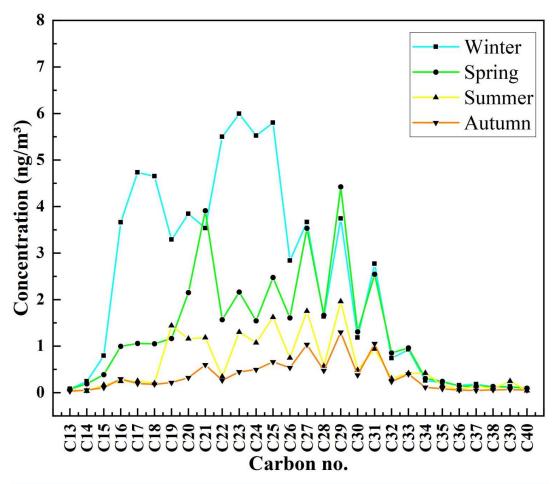


Figure 3. Contributions of low molecular weight *n*-alkanes in the day and night samples in the different seasons of Beijing (* indicates a significant difference).







518 Figure 4. Concentration distributions of the particulate-bound n-alkane homologs in the different seasons of Beijing.





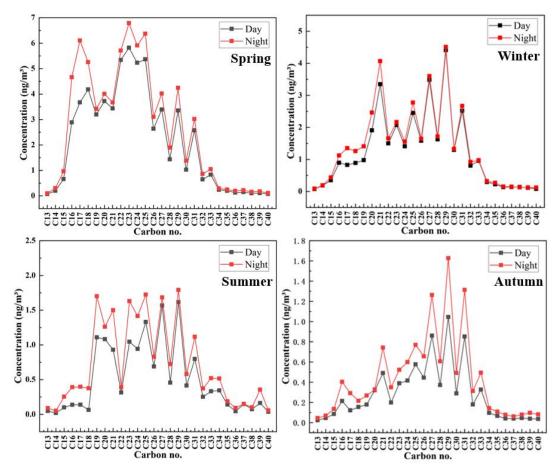


Figure 5. Concentration distributions of the particulate-bound *n*-alkane homologs in the day and night in the different seasons of Beijing.





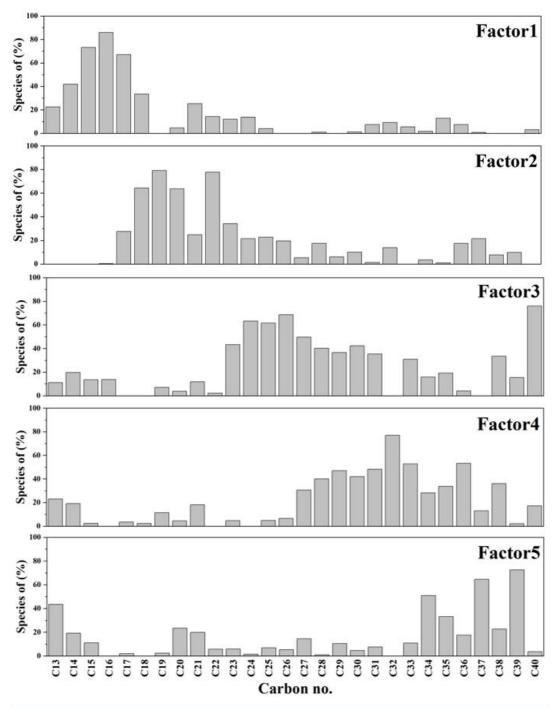


Figure 6. Proportions of the different n-alkane homologs in the factors identified by the positive matrix factorization model.





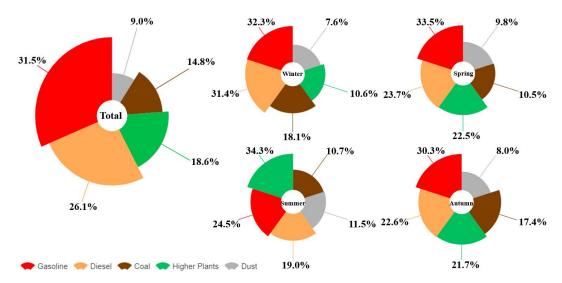


Figure 7. Sources and contributions of particulate-bound *n*-alkanes in Beijing.

Table 1. PM2.5 and particulate-bound *n*-alkane concentrations in different seasons in Beijing.

Species	Winter ^a		Spring ^b		Summer ^c		Fall ^d	
	Mean	Range	Mean	Range	Mean	Range	Mean	Range
PM2.5 ($\mu g/m^3$)	28.5	8.00-65.0	43.5	0–134	21.5	10.0-32.0	32.2	8.00-117
<i>n</i> -Alkanes (ng/m³)	66.3	17.1–89.9	36.8	12.2-64.1	18.0	9.92–29.7	9.78	4.51–27.1

⁵²⁷ a Winter: November and December in 2020;

Table 2. Source indices for particulate-bound *n*-alkane in Beijing.

Source Index	Winter		Spring		Summer		Fall	
	Day	Night	Day	Night	Day	Night	Day	Night
Cmax ^a	C23	C23	C29	C29	C29	C29	C29	C29
CPI ^b	1.16	1.18	1.85	1.76	2.15	1.87	1.90	1.78
WNA%c	17.4	18.5	35.0	33.1	43.0	39.2	39.6	35.1
PNA%d	82.6	81.5	65.0	66.9	57.0	60.8	60.5	64.9

⁵³² a Cmax: Carbon maximum number;

⁵²⁸ b Spring: March and April in 2021;

^{529 °}Summer: June and July in 2021;

 $^{^{}m d}$ Fall: September and October in 2021.

^{533 &}lt;sup>b</sup> CPI: Carbon preference index;

^{534 °} WNA%: Plant wax n-alkane ratio;

⁵³⁵ d PNA%: Petrogenic n-alkane ratio.