

Characteristics of particulate-bound *n*-alkanes indicating sources of PM_{2.5} in Beijing, China

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Abstract. The characteristics of *n*-alkanes and the contributions of various sources of fine particulate matter (PM_{2.5}) in the atmosphere in Beijing were investigated. PM_{2.5} samples were collected at Minzu University of China between November 2020 and October 2021, and *n*-alkanes in the samples were analyzed by gas chromatography mass spectrometry. A positive matrix factorization analysis model and source indices (the main carbon peaks, carbon preference indices, and plant wax contribution ratios) were used to identify the sources of *n*-alkanes, determine the contributions of different sources, and explain the differences. The *n*-alkane concentrations were 4.51–153 ng/m³, (mean 32.7 ng/m³), and the particulate-bound *n*-alkane and PM_{2.5} concentrations varied in parallel. There were marked seasonal and diurnal differences in the *n*-alkane concentrations ($p < 0.01$). The *n*-alkane concentrations in the different seasons decreased in the order winter > spring > summer > fall. The mean concentration of each homolog was higher at night than in the day in all seasons. Particulate-bound *n*-alkanes were supplied by common anthropogenic and biogenic sources, and fossil fuel combustion was the dominant contributor. The positive matrix factorization model results indicated five sources of *n*-alkanes in PM_{2.5}, which were coal combustion, diesel vehicle emissions, gasoline vehicle emissions, terrestrial plants, and mixed source. Vehicle emissions were the main sources of *n*-alkanes, contributing 57.6%. The sources of PM_{2.5} can be indicated by *n*-alkanes (i.e., using *n*-alkanes as organic tracers). Vehicle exhausts strongly affect PM_{2.5} pollution. Controlling vehicle exhaust emissions is key to controlling *n*-alkane and PM_{2.5} pollution in Beijing.

1 Introduction

Serious air pollution in China is currently caused by a combination of haze and photochemical smog (Ma et al., 2012). The effects of haze on air quality are more obvious than the effects of photochemical pollution, which is relatively invisible. Haze is frequent in urban areas with relatively dense populations and high traffic loads. Fine particulate matter is the main pollutant involved in haze. Fine particulate matter has small particle sizes, a long atmospheric retention time, and a complex chemical composition. Fine particulate matter is also a good substrate for chemical reactions, about which there is great concern because the products can negatively affect the environment and human health (Wang et al., 2016; Zhu et al., 2005; Zhang, et al., 2015). In recent years, measures such as energy structure adjustments, pollutant emission controls, and air pollution prevention have markedly decreased atmospheric pollution and improved air quality in China. For example, the PM_{2.5} concentration in Beijing, a typical large city in China, has recently decreased markedly. The annual mean PM_{2.5} concentration decreased from 73 µg/m³ in 2016 to 33 µg/m³ (meeting the requirement of the secondary ambient air quality standard for China, 35 µg/m³) in 2021 (Beijing Ecology and Environment Statement, 2016-2021). Sources of fine particulate matter need to be better understood and controlled to decrease PM_{2.5} pollution, improve air quality, and meet the primary

39 ambient air quality standard for China ($15 \mu\text{g}/\text{m}^3$) and even the World Health Organization standard ($5 \mu\text{g}/\text{m}^3$).

40 It has been found that *n*-alkanes are important components of organic pollutants in particulate matter and are mainly supplied

41 through anthropogenic emissions such as vehicle exhausts, fossil fuel combustion, and biomass combustion (Liu et al., 2013)

42 or through biogenic emissions such as from microorganisms and terrestrial plants (Simoneit et al., 1989; Rogge et al., 1993).

43 *n*-Alkanes are non-polar saturated hydrocarbons that are stable and found at high concentrations in the atmosphere. *n*-

44 Alkanes readily adsorb to particles and can affect the environment and human health (Chen et al., 2019). *n*-Alkanes can

45 participate in atmospheric chemical reactions, and *n*-alkane volatility and reactivity decrease as the carbon chain length

46 increases (Aumont et al., 2013). The products of reactions involving short-chain *n*-alkanes ($C \leq 16$) in the environment

47 strongly contribute to secondary organic aerosol formation (Michoud et al., 2012). Long-chain *n*-alkanes ($C > 16$) are

48 relatively stable in the environment and generally accumulate in particulate matter (Chrysikou et al., 2009). The carbon

49 number ranges, molecular compositions, and distributions of *n*-alkane mixtures in particulate matter can be used to assess

50 aerosol migration and particulate matter sources. Particulate-bound *n*-alkanes play an important role in studying organic

51 aerosols and the sources of the PM_{2.5}. The characteristics and sources of *n*-alkanes in fine particulate matter are important

52 parameters for developing pollutant control strategies to sustainably decrease haze pollution and improve air quality.

53 Previous studies of *n*-alkanes in atmospheric particulate matter have mainly been focused on concentrations (Wang et al.,

54 2005; Wang et al., 2006; Chen et al., 2014; Ren et al., 2017), characteristics (Simoneit et al., 2004; Li et al., 2013; Kang et al.,

55 2016), and sources (Kavouras et al., 2001; Bi et al., 2003; Fu et al., 2010; Sun et al., 2021). A wide range of *n*-alkanes are

56 present in the atmosphere, including highly and poorly volatile *n*-alkanes with carbon chain lengths between 8 and 40 (Kang

57 et al., 2016; Aumont et al., 2012). *n*-Alkane concentrations between tens and hundreds of nanograms per cubic meter have

58 been found in fine particles (Ren et al., 2016; Lyu et al., 2019). The *n*-alkane concentration is affected by factors such as

59 meteorological conditions and contributing sources and is related to the particulate matter concentration and particle size

60 distribution. The total *n*-alkane concentration in particulate matter markedly varies by season, usually being higher in winter

61 and lowest in summer and fall (Lyu et al., 2016; Chen et al., 2019). *n*-Alkanes from different sources have different

62 molecular compositions and distributions that can be used to indicate the relative contributions of different sources of

63 particulate matter (Han et al., 2018).

64 In the past few decades, researchers in Zhengzhou (Wang et al., 2017), Guangzhou (Bi et al., 2003; Wang et al., 2016),

65 Shanghai (Lyu et al., 2016; Xu et al., 2015), Beijing (Ren et al., 2016; Lyu et al., 2019), Seoul (Kang et al., 2020), and Spain

66 (Caumo et al., 2020) have studied *n*-alkanes in atmospheric aerosols and determined total *n*-alkane concentrations, particle

67 size distributions, and the contributions of different sources. However, *n*-alkanes with different carbon number ranges were

68 analyzed in the different studies. Most studies were focused on *n*-alkanes containing < 30 carbon atoms, but these do not fully

69 reflect the sources of *n*-alkanes in particulate matter. Air quality in Beijing is gradually improving, and exploring strategies

70 for controlling sources of fine particulate matter further requires more information about *n*-alkane homolog distributions and

71 variability in fine particulate matter and the relative contributions of different sources. Beijing is a large city with a dense

72 population and high traffic volumes. The sources of *n*-alkanes and particulate matter in Beijing require attention because of

73 the large number of volatile organic pollutants present, the high levels of vehicle exhaust emissions, and relatively severe

74 particulate matter pollution. Secondary aerosols have been found to make strong contributions to particulate pollution during

75 haze episodes in urban areas (Presto et al., 2009; Huang et al., 2014). *n*-Alkanes only contribute a proportion of the total

76 organic matter in particulate matter but are important contributors to particulate pollution by being important precursors of

77 secondary organic aerosols (Yang et al., 2019). *n*-Alkanes are also important indicators of the sources of particulate matter.

78 In this study, the concentrations of C₁₃–C₄₀ *n*-alkanes in atmospheric fine particulate matter in Beijing between 2020 and

79 2021 were determined. Diurnal and seasonal variations in *n*-alkane homolog concentrations were assessed by performing

80 diurnal and cross-seasonal sampling. The sources of *n*-alkanes were identified and the contributions of these sources to the

81 total *n*-alkane concentrations were determined using source indices and correlation models. The aim was to use *n*-alkanes to

82 indicate the sources of particulate matter to allow strategies for controlling particulate matter concentrations in urban areas to
83 be developed.

84 **2 Materials and methods**

85 **2.1 Sampling site and time**

86 Fine particulate matter samples were collected between November 2020 and October 2021 on the roof (about 20 m above the
87 ground) of the College of Pharmacy at the Minzu University of China (116.19° E, 39.57° N). Beijing is a typical heavily
88 populated and traffic-intensive Chinese city, with high emission intensities of nitrogen oxides and volatile organic pollutants
89 and relatively serious fine particulate matter pollution. Haidian District is a prosperous urban area in Beijing with intense
90 human activities and busy traffic. The sampling point in Haidian District reflected the influences of human activities and
91 vehicle emissions on fine particulate matter concentrations. Samples were collected between the 23rd and 29th of each
92 month during the study, but the exact sampling periods were adjusted to take into account pollution levels and the weather.
93 Samples were collected in two periods on a sampling day. Daytime samples were collected between 07:00 and 20:00 and
94 nighttime samples were collected between 20:30 and 06:30 the next morning. Diurnal and seasonal variations in *n*-alkane
95 concentrations in fine particulate matter were investigated by collecting separate day and night samples and collecting
96 samples in different seasons. The effects of *n*-alkanes on PM_{2.5} concentrations were assessed by analyzing the correlation
97 between their concentrations.

98 **2.2 Sample collection and pretreatment**

99 Each fine particulate sample was collected using a TH-16A low-flow sampler (Wuhan Tianhong, Wuhan, China) containing
100 a Whatman QMA quartz fiber filter (Ø 47 mm; GE Healthcare Bio-Sciences, Pittsburgh, PA, USA) using a flow rate of 16.7
101 L/min. Before use, the quartz fiber filters were baked at 550 °C for 5 h to remove organic matter. Each filter was loosely
102 wrapped in aluminum foil and equilibrated for 24 h at 20 °C and 40% relative humidity and then weighed using a precision
103 electronic balance before being used to collect a sample. Once used, a filter was equilibrated for 24 h at 20 °C and 40%
104 relative humidity, weighed again, and then stored wrapped in aluminum foil at -20 °C.

105 The details for ultrasonic extraction methods used to analyze the samples of *n*-alkanes in PM_{2.5} are reported in previous
106 studies (Yang et al., 2019; Kang et al., 2020; Caumo et al., 2020). Each filter was cut into pieces and extracted by
107 ultrasonication with 15 mL of dichloromethane for 15 min. The extraction step was repeated five times, and the extracts
108 were combined and evaporated to 2 mL using a rotary evaporator. The extract was then transferred to a 15 mL centrifuge
109 tube and centrifuged at 3000 rpm for 5 min. The supernatant was evaporated just to dryness under a gentle flow of high
110 purity nitrogen and then redissolved in 100 µL of toluene for instrumental analysis.

111 **2.3 Instrumental analysis**

112 The *n*-alkanes (C₁₃–C₄₀) were analyzed qualitatively and quantitatively by gas chromatography mass spectrometry using an
113 Agilent 6890N-5975 system (Agilent Technologies, Santa Clara, CA, USA). *n*-Alkane standards (C₈–C₄₀) were purchased
114 from AccuStandard (New Haven, CT, USA). Separation was achieved using an Agilent J&W Scientific DB-5M column (30
115 m long, 0.25 mm inner diameter, 0.1 µm film thickness; Agilent Technologies). Temperature of the GC inlet was 290°C,
116 splitless injection mode was used, and the injection volume was 1.0 µL. The carrier gas was helium and the constant flow
117 rate was 1.0 mL/min. The oven temperature program started at 80 °C, which was held for 2 min, then increased at 10 °C/min
118 to 200 °C, and then increased at 15 °C/min to 300 °C, which was held for 30 min. The mass spectrometer was used in
119 electron impact ionization mode and selected ion detection mode. Ions with mass-to-charge ratios of 85 and 113

120 (characteristic of *n*-alkanes) were used to identify and quantify *n*-alkanes. The data were quantified using ChemStation
121 software (Agilent Technologies).

122 **2.4 Quantitative analysis**

123 Particulate-bound *n*-alkanes were quantified by external standard method. We prepared standard solutions of C₈-C₄₀ *n*-
124 alkanes with concentration gradients of 10 ppm, 1 ppm, 500 ppb, 100 ppb, 50 ppb and 10 ppb. The calibration curves is
125 plotted with the concentrations of the standard solution as the abscissa axis and the corresponding chromatographic response
126 obtained by GC-MS as the ordinate axis, the correlation coefficient of each individual calibration curve is greater than 0.99.
127 The concentrations of particulate-bound *n*-alkanes were finally quantified by the calibration curves.

128 **2.5 Quality assurance and control**

129 When extracting *n*-alkanes from the fine particulate samples, blank samples were extracted with each batch of samples. The
130 concentration of an analyte substance in the blanks was subtracted from the concentration of the analyte in a sample during
131 data processing. The detection and quantification limits of the instrument were defined as three and 10 times the signal-to-
132 noise ratio, respectively. The instrument detection limits for the *n*-alkanes were 1–10 pg.

133 Spiked recovery experiment was used to evaluate the recovery efficiency of particulate-bound *n*-alkanes. Mixed standard
134 solution of C₈-C₄₀ *n*-alkanes (20 μL, 1 ppm) was added to the blank samples, then the blank samples was pre-treat according
135 to the same methods and the concentrations of *n*-alkanes was detected by GC-MS. The recovery was calculated based on the
136 theoretical concentrations of *n*-alkanes standard solution and the measured concentrations of *n*-alkanes in the blank spiked
137 samples. The blank spiked recovery experiments were repeated three times and the final recovery was averaged over the
138 three experiments, the extraction recovery for *n*-alkanes range from 43.6% to 128%, the RSD for the concentrations of *n*-
139 alkanes is 3.51%.

140 **2.6 Data analysis**

141 PM_{2.5} data were provided by the China Meteorological Administration (cma.gov.cn). Data analysis (statistical and other
142 analyses of the *n*-alkane data) was performed using SPSS 26.0 software (IBM, Armonk, NY, USA). Differences in the
143 concentrations of an *n*-alkane homolog in different groups of samples and differences in the overall *n*-alkane compositions in
144 different groups of samples were assessed by performing independent sample t-tests. Spearman correlations and Pearson
145 correlations (two-tailed tests) were used to identify correlations between groups of data.

146 Source indices (the carbon maximum number (C_{max}), carbon preference index (CPI), and plant wax *n*-alkane ratio
147 (WNA%)) were used to assess the *n*-alkane sources from the *n*-alkane molecular compositions and concentration
148 distributions. The C_{max} is the homolog with the highest relative concentration in the *n*-alkane mixture, it is commonly used
149 to distinguish between the contributions of anthropogenic and natural sources of *n*-alkanes and is related to the degree of
150 thermal evolution that has affected the organic matter supplying *n*-alkanes. The CPI defined as the ratio of total odd carbon
151 *n*-alkanes to even carbon *n*-alkanes and was developed by Bray and Evans in 1961 (Bray et al., 1961), it can be used to
152 assess the contributions of anthropogenic and biogenic sources of *n*-alkanes and is the most commonly used empirical
153 parameter for distinguishing between sources of *n*-alkanes (Marzi et al., 1993). WNA% and PNA% (petrogenic *n*-alkane
154 ratio) can be used to assess the relative contributions of biological and anthropogenic sources of *n*-alkanes in particulate
155 matter (Simoneit, 1985), WNA% are calculated by subtraction of the average of the next higher and lower even carbon
156 numbered homologues, while PNA% was defined as the WNA% subtracted from 100% (Lyu et al., 2019). The source
157 indices were calculated using Eqs. (1)–(3):

$$158 \quad \text{CPI} = \frac{\sum_{i=6}^{19} C_{2i+1}}{\sum_{i=7}^{20} C_{2i}} \quad (1)$$

$$159 \quad \text{WNA}\% = \frac{\sum (C_n - (\frac{C_{n-1} + C_{n+1}}{2}))}{\sum C_n} \times 100\% \quad (\text{"n" is an odd number}) \quad (2)$$

$$160 \quad \text{PNA}\% = 100\% - \text{WNA}\% \quad (3)$$

161 In Eq. (1), C_{2i+1} was the concentration of the n -alkane with odd carbon atoms range from 13-39, while C_{2i} was the
 162 concentration of the n -alkane with even carbon atoms range from 14-40. In Eq. (2), C_n was the concentration of n -alkanes,
 163 taking as zero the negative value of $(C_n - (\frac{C_{n-1} + C_{n+1}}{2}))$.

164 A positive matrix factorization (PMF) model was used to identify specific n -alkane sources and the contribution of each
 165 source through EPA PMF 5.0 software (USEPA). The PMF model is a factor analysis technique using multivariate statistical
 166 methods. The PMF model is a receptor model, so can identify and determine the contributions of components of unknown
 167 mixtures. The PMF model is one of the source resolution methods recommended by the US Environmental Protection
 168 Agency. The PMF model does not require the complex pollutant sources to be determined and the treatment process can be
 169 optimized while limiting the decomposition matrix elements and sharing the rates of nonnegative matrices. The model can
 170 use the chemical composition of particulate matter to identify the sources of particulate matter and calculate the contributions
 171 of the different sources, so is widely used to investigate the sources of atmospheric particulate matter (Moeinaddini et al.,
 172 2014; Liao et al., 2021; Li et al., 2021). The details of PMF have been described in the PMF 5.0 User Guide (USEPA, 2014).

173 **3 Results**

174 **3.1 Concentrations of n -alkanes**

175 A total of 28 n -alkane homologs with carbon chain lengths of C_{13} – C_{40} were analyzed. C_{13} – C_{40} n -alkanes were detected in the
 176 diurnal fine particulate matter samples collected in all seasons. Among them, C_{21} – C_{35} n -alkanes were detected in all PM2.5
 177 samples, other n -alkanes were detected in more than half of the samples.

178 The n -alkane and PM2.5 concentrations in the different seasons are shown in Table 1 and temporal variations in the average
 179 concentrations between day and night are shown in Figure 1. The PM2.5 concentrations throughout the sampling period were
 180 0–134 $\mu\text{g}/\text{m}^3$, and the mean was 32.0 $\mu\text{g}/\text{m}^3$. The n -alkane concentrations throughout the sampling period were 4.51–153
 181 ng/m^3 , and the mean was 32.7 ng/m^3 . As shown in Figure 8, under the condition of excluding the influence of the sharp rise
 182 of PM2.5 concentration in heavy haze days, correlation analysis indicated that the n -alkane and PM2.5 concentrations
 183 significantly positively correlated ($p < 0.01$, $r = 0.618$).

184 **3.2 n -Alkane component distributions**

185 The contributions of the individual C_{13} – C_{40} n -alkane homologs to the total n -alkane concentrations are shown in Figure 2.
 186 The C_{16} – C_{25} n -alkanes were dominant in winter and the C_{26} – C_{31} n -alkane contributions increased markedly spring, summer,
 187 and fall.

188 The n -alkane homologs can be classed as low molecular weight (LMW), meaning n -alkanes with carbon chain lengths ≤ 25 ,
 189 and high molecular weight (HMW), meaning n -alkanes with carbon chain lengths > 25 . As shown in Figure 3, LMW n -
 190 alkanes contributed $\sim 60\%$ of the total n -alkane concentrations in winter but only $\sim 40\%$ in spring, summer, and fall,
 191 indicating that there were marked differences between the compositions in winter and the other seasons.

192 3.3 Seasonal and diurnal differences in *n*-alkane concentrations

193 The average concentration distributions of C₁₃–C₄₀ *n*-alkanes in the different seasons are shown in Figure 4. There were
194 significant differences ($p < 0.01$) between the concentrations of various homologs in the different seasons. The mean *n*-alkane
195 concentrations for the different seasons decreased in the order winter > spring > summer > fall. The seasonal differences were
196 more marked for LMW than HMW *n*-alkanes. The concentrations of relatively short-chain *n*-alkanes (C₁₆–C₂₅) were
197 markedly higher in winter than in the other seasons. The concentrations of C₂₇, C₂₉, C₃₁, and C₃₃ *n*-alkanes were higher than
198 the concentrations of C₂₆, C₂₈, C₃₀, C₃₂, and C₃₄ *n*-alkanes (i.e., odd-carbon-number dominance occurred) in all of the seasons.
199 The C₁₃–C₄₀ *n*-alkane concentrations in the day and night samples are shown in Figure 5. The mean *n*-alkane homolog
200 concentrations were higher at night than in the day in all four seasons. The concentrations in the day and night were
201 significantly different ($p < 0.01$). Statistical tests on the differences in concentration of individual homolog of *n*-alkanes
202 between day and night in different seasons showed that fewer *n*-alkane homologs with significant differences in winter (C₁₆,
203 C₁₇) and spring (C₂₁) while more *n*-alkane homologs (C > 21) with significant differences in summer and autumn.

204 3.4 Source indices and PMF model

205 Source indices (C_{max}, CPI, and WNA%) determined from the C₁₃–C₄₀ *n*-alkane data were used to assess the *n*-alkane
206 sources. The PMF model was used to quantify the amounts of *n*-alkanes in fine particles supplied by the different sources
207 and the relative contributions of the sources. The source index data for *n*-alkanes in the day and night samples in the different
208 seasons are shown in Table 2.

209 3.4.1 Source indices for *n*-alkanes

210 The C_{max} for winter was C₂₃ but the C_{max} for spring, summer, and fall was C₂₉. The mean CPI for the year the samples
211 were collected were 1.66. The CPI was lowest in winter but higher in the day than the night in spring, summer, and fall. The
212 mean contribution of plant wax *n*-alkanes to the total *n*-alkane concentration during the sampling period was 30.6% and the
213 mean contribution of anthropogenic *n*-alkanes to the total *n*-alkane concentration was 69.4%. The plant wax *n*-alkane
214 contribution was lowest in winter and markedly higher in spring, summer, and fall.

215 3.4.2 Results of the PMF model

216 According to the PMF 5.0 User Guide (USEPA, 2014), the daily mean *n*-alkane concentrations during the sampling period
217 and the corresponding uncertainties were inputted into the PMF model to analyze the sources of *n*-alkanes in fine particulate
218 matter. Various numbers of factors were tested, and the optimal correlation coefficient for the relationship between the
219 simulated and observed values was found when five factors were used, the average correlation coefficient of *n*-alkane
220 homologues is 0.832. Q (robust) is a important parameter obtained after PMF run, it is the goodness-of-fit parameter
221 calculated excluding points not fit by the model (USEPA, 2014). In the process of running the PMF model, we got the lowest
222 Q (robust) values when selected five factors. This met the requirements to use the PMF model, EPA PMF 5.0 User Guide
223 (USEPA, 2014) have stated that the lowest Q (robust) value represents the most optimal solution from the multiple runs and
224 it can be a critical parameter for choosing optimal number of factors. Each factor indicated a source, and the factors could be
225 used to identify the corresponding sources. The *n*-alkane factor data given by the PMF model are shown in Figure 6.
226 The PMF model indicated that the contributions of factors 1, 2, 3, 4, and 5 to the *n*-alkane concentrations were 14.8%, 26.1%,
227 31.5%, 18.6%, and 9.01%, respectively. The sources corresponding to the factors identified by the PMF model needed to be
228 identified from the proportions of the different *n*-alkane homologs present, the sources corresponding to factors 2 and 3 were
229 the main contributors of *n*-alkanes in particulate matter.

231 **4.1 Sources and contributions of *n*-alkanes**

232 *n*-Alkanes in PM_{2.5} have relatively complex sources, but different *n*-alkane compositions and distributions indicate different
233 sources. As shown in Figure 4, marked odd-carbon-number dominance was found in all seasons for the HMW *n*-alkanes,
234 with *n*-alkanes with carbon chain lengths C₂₇, C₂₉, C₃₁, and C₃₃ being dominant. No odd-carbon-number dominance was
235 found for the LMW *n*-alkanes. It has previously been found that LMW *n*-alkanes in urban areas are mainly anthropogenic
236 (e.g., emitted during fossil fuel combustion and in vehicle exhaust gases) (Simoneit et al., 2004; Kang et al., 2016) but HMW
237 *n*-alkanes reflect sources such as biomass combustion and waxes in terrestrial plants (Kawamura et al., 2003). LMW and
238 HMW *n*-alkane patterns can be used to identify the main sources of *n*-alkanes in urban areas. The *n*-alkane patterns in the
239 different seasons indicated that particulate-bound *n*-alkanes in the atmosphere in Beijing have both anthropogenic and
240 biological sources. The source indices and PMF model results further explained the sources and contributions of *n*-alkanes.

241 *n*-Alkane source indices are often used to identify the origins of *n*-alkanes. The *n*-alkane source indices shown in Table 2
242 indicated that anthropogenic emissions were the main contributors of particulate-bound *n*-alkanes in Beijing during the study
243 but that there were also biogenic emissions of particulate-bound *n*-alkanes. The CPI and WNA% data explained this. During
244 the sampling period, the mean CPI was 1.66, indicating that the main sources of particulate-bound *n*-alkanes were fossil fuel
245 combustion, plants, and biomass combustion. The mean WNA% and PNA% were 30.63% and 69.37%, respectively,
246 indicating that anthropogenic emissions contributed more than emissions from biota.

247 The PMF model can quantify the contributions of specific sources of *n*-alkanes relatively accurately. The *n*-alkane homolog
248 contributions to each factor identified by the PMF model were used to analyze and identify the corresponding source. As
249 shown in factor 1 of Figure 6, the *n*-alkanes with carbon chain lengths of C₁₃–C₁₈ were dominant, which similar to the *n*-
250 alkane homolog (C<20) pattern for emissions during coal combustion found by Oros and Simoneit and Niu et al. (Oros et al.,
251 2000; Niu et al., 2005). Therefore, we concluded that factor 1 indicated *n*-alkanes emitted through coal combustion. Vehicle
252 emissions are important sources of *n*-alkanes in particulate matter in urban areas (Lyu et al., 2019). *n*-Alkanes emitted by
253 vehicles mainly have carbon-chain lengths <30 (Wang et al., 2017). However, there are marked differences between the
254 patterns of *n*-alkanes emitted in particulates in gasoline vehicle and diesel vehicle exhaust gases. C_{max} for *n*-alkanes is lower
255 and the proportion of low-carbon-chain length *n*-alkanes is higher for particulates in diesel vehicle exhaust gases than
256 gasoline vehicle exhaust gases. This feature can be used to distinguish between *n*-alkanes emitted by diesel and gasoline
257 vehicles in fine particulate matter (Fujitani et al., 2012; Yuan et al., 2016). As shown in Figure 6, the homologs with a higher
258 proportion of *n*-alkane species in factor 2 are concentrated around C₂₀, while in factor 3 are concentrated around C₂₇.
259 According to studies of Schauer et al. for gasoline and diesel vehicle emissions (Schauer et al., 1999; Schauer et al., 2002),
260 we determine that factor 2 and factor 3 indicated diesel and gasoline vehicle emission sources, respectively. C₂₇–C₃₈ (i.e.,
261 high-carbon-chain-length) *n*-alkanes made large contributions and low-carbon-chain-length *n*-alkanes made small
262 contributions to the pattern for factor 4. Studies have shown that C₂₆–C₃₆ *n*-alkanes are mainly emitted from cuticular waxes
263 in terrestrial plants (Alves et al., 2001; Lyu et al., 2016), so we inferred that factor 4 indicated *n*-alkanes emitted by terrestrial
264 plants. *n*-Alkanes do not have an obvious regularity in composition and there was no clear *n*-alkane homologs pattern for
265 factor 5, but long-chain *n*-alkanes with carbon chain lengths ≥34 were dominant. We found that road dust is one of the
266 sources of particulate-bound *n*-alkanes (Anh et al., 2019), *n*-alkanes with ≥C₃₄ may come from road dust (Daher et al., 2013)
267 and biogenic source (Liebezeit et al., 2009). Therefore, we concluded that factor 5 may be a mixed source of *n*-alkanes from
268 road dust and biogenic emissions.

269 The contributions of the different sources to the *n*-alkane concentrations are shown in Figure 7. In summary, *n*-alkanes in
270 airborne particulate matter in Beijing are both anthropogenic and biogenic. Vehicle exhaust emissions are the main sources
271 of *n*-alkanes, consistent with the current energy consumption structure in Beijing, and gasoline and diesel vehicles accounted

272 for a relatively large proportion of *n*-alkanes in airborne particulate matter.

273 **4.2 Characteristics of PM_{2.5} and *n*-alkanes**

274 The mean *n*-alkane concentration during the sampling period was 32.7 ng/m³, which was lower than the C₁₉–C₃₆ *n*-alkane
275 concentration of 282 ng/m³ found in Beijing in 2006 (Li et al., 2013) and the C₈–C₄₀ *n*-alkane concentration of 228 ng/m³
276 found in Shanghai in 2013 (Lyu et al., 2016). The temporal trends in the *n*-alkane concentrations were similar to the trends
277 found in previous studies of *n*-alkanes in Beijing (Rogge et al., 1993; Li et al., 2013; Ren et al., 2019), the overall *n*-alkane
278 concentration being highest in winter. The seasonal pattern we found for *n*-alkanes in Beijing was similar to the pattern
279 found in a previous study of C₁₆–C₃₅ *n*-alkanes in 14 Chinese cities (Wang et al., 2006).

280 The *n*-alkane pattern varied by season, LMW *n*-alkanes being dominant in winter and HMW *n*-alkanes being more abundant
281 in the other seasons. C_{max} and WNA% explained the seasonal differences in the *n*-alkane patterns. In previous studies,
282 lower C_{max} values were found for *n*-alkanes emitted from very mature organic matter such as coal and petroleum than for *n*-
283 alkanes emitted from immature organic matter such as plants (Simoneit et al., 1989; Duan et al., 2010). The C_{max} for *n*-
284 alkanes in winter was C₂₃, indicating that LMW *n*-alkanes were the main *n*-alkanes. Similar results were found by Lyu et al.
285 for Beijing in winter (Lyu et al., 2019). The C_{max} for *n*-alkanes in spring, summer, and fall was C₂₉. Ficken et al. (Ficken et
286 al., 2000) and Yadav et al. (Yadav et al., 2013) found that C₂₉ *n*-alkanes are markers for *n*-alkanes emitted from the wax
287 layers of terrestrial plants. Stronger *n*-alkane contributions will be made by plants in spring, summer, and fall than in winter
288 (Rogge et al., 1993; Yadav et al., 2013). This is consistent with the results found in a study performed in Shanghai (Lyu et al.,
289 2016; Wang et al., 2016). There were significant seasonal differences ($p < 0.01$) in the concentrations of the C₁₃–C₄₀ *n*-alkane
290 homologs, but the seasonal differences were stronger for LMW *n*-alkanes than HMW *n*-alkanes. Similar results were found
291 by Li et al. in Tianjin in 2010 (Li et al., 2010). The LMW *n*-alkane concentrations were markedly higher in winter than in the
292 other seasons, similar to the results of a study performed by Li et al. in Beijing in 2013 (Li et al., 2013). This indicated that
293 there were seasonal differences in *n*-alkane sources. The PMF model results shown in Figure 7 indicated that anthropogenic
294 *n*-alkanes strongly contributed to the total *n*-alkane concentration in winter. The CPI also indicated that different sources
295 were dominant in winter and in the other seasons. The lowest CPI was found for winter, indicating that LMW *n*-alkanes
296 made stronger contributions to the total *n*-alkane concentrations in winter than the other seasons. This may be related to *n*-
297 alkane emissions caused by fossil fuel combustion for heating in winter. Similar results have been found in Shanghai (Lyu et
298 al., 2016), Zhengzhou (Wang et al., 2017), southeastern Chinese cities (Chen et al., 2019), and Beijing (Kang et al., 2016).

299 Meteorological factors affect the concentrations and composition of *n*-alkanes in different seasons. The mixing layer height
300 influences the concentration of *n*-alkanes by affecting the particulate matter, it's shown that the mixing layer height is
301 correlated with the concentration of particulate matter and the peak concentration of particulate matter increases as the
302 mixing layer height decreases (Wagner et al., 2017). The atmospheric mixing layer height in Beijing has obvious seasonal
303 characteristics, showing low in winter and high in summer (Wang et al., 2020; Tang et al., 2016). Therefore, the increased
304 concentrations of PM_{2.5} and *n*-alkanes in winter were influenced by the mixing layer height. Wind direction is one of the
305 factors affecting the seasonal differences in particulate matter and *n*-alkanes, the northwest wind in winter brought the
306 polluted air masses from inland to Beijing, while the southeast wind in summer transported cleaner aerosols from oceans to
307 here (Wei et al., 2020). In addition, the seasonal distribution of *n*-alkanes is influenced by the temperature. The temperature
308 in Beijing is high in summer and low in winter, when the temperature is lower in winter, gaseous *n*-alkanes are more likely to
309 partition into particles with the higher partition coefficient of gas-particle partitioning (Lyu et al., 2016; Wick et al., 2002).
310 Therefore, the increase of LWM *n*-alkanes proportion in winter also affected by temperature.

311 The mean C₁₃–C₄₀ *n*-alkane homolog concentrations were higher at night than in the day in each season, and the differences
312 were significant ($p < 0.01$). According to the study by Yao et al. in 2009, lower average wind speeds, atmospheric mixing

313 layer height and poorer atmospheric diffusion conditions can lead to higher concentrations of *n*-alkanes at night than in the
314 day (Yao et al., 2009). Similar results were found in Liaocheng, Shandong Province (Liu et al., 2019). The differences in the
315 *n*-alkane concentrations in the night and day may also have been caused by differences in pollutant emissions in the night
316 and day. Particulate-bound *n*-alkanes from vehicular emissions usually of low molecular weight (Lyu et al., 2019), diesel
317 emissions have higher concentrations of particulate-bound *n*-alkanes with carbon chain lengths less than 25 (Schauer et al.,
318 1999). Differences in diurnal concentrations of LMW *n*-alkanes may reflect the differences in the contribution of
319 anthropogenic sources. We found markedly higher concentrations of some homologs with carbon chain lengths <25 at night
320 than during the day. This would be consistent with short-chain alkane emissions from diesel vehicles in Beijing being higher
321 at night than in the day.

322 **4.3 PM2.5 sources in Beijing and strategies for controlling PM2.5 concentrations**

323 During the sampling period, the mean daily PM2.5 concentration in Beijing was 32.0 $\mu\text{g}/\text{m}^3$, which met the requirement of
324 the secondary ambient air quality standard for China (35.0 $\mu\text{g}/\text{m}^3$). According to the Ecology and Environment Statement
325 from the Beijing Municipal Ecology and Environment Bureau (sthjj.beijing.gov.cn), the annual mean PM2.5 concentration in
326 Beijing has gradually decreased in the last five years. However, little research on *n*-alkanes in Beijing has been performed in
327 this period. We compared our results with the results of a previous study (Lyu et al., 2019) and found that the *n*-alkane
328 concentrations decreased in parallel with the PM2.5 concentrations. *n*-Alkanes are important molecular markers for
329 identifying the sources of PM2.5. Excluding when the PM2.5 concentration increased sharply because of meteorological
330 conditions, the PM2.5 and *n*-alkane concentrations varied in the same ways. As shown in Figure 8, a significant positive
331 correlation was found between the PM2.5 and *n*-alkane concentrations ($p < 0.01$), so *n*-alkanes could be used as indicators of
332 the sources of PM2.5 in the atmosphere. This method has been widely used to analyze sources of particulate matter (Cass,
333 1998; Kavouras et al., 2001; Bi et al., 2003; Xu et al., 2013; Zhao et al., 2016; Han et al., 2018). We therefore used the PMF
334 model results for *n*-alkanes to identify the sources of PM2.5 and explain variations in the sources.

335 The PMF model results for the contributions of the different sources shown in Figure 7 indicated that emissions in vehicle
336 exhaust gases and through coal combustion contributed up to 72.4% of PM2.5 in the sampling area throughout the sampling
337 period. This indicated that anthropogenic PM2.5 emissions are the main sources of PM2.5 in the urban study area. Emissions
338 from gasoline and diesel vehicles were the dominant anthropogenic sources, contributing 57.6% of total anthropogenic
339 PM2.5 emissions. Vehicles are the main sources of PM2.5 in urban areas and make important contributions to particulate
340 matter in the atmosphere in Beijing. Similar results were found in a previous study of PM2.5 sources in Beijing (Lv et al.,
341 2020; Qi et al., 2018) and the results were consistent with the current energy consumption structure in Beijing (gasoline and
342 diesel fuel make large contributions to total fuel consumption). Human activities make larger contributions to PM2.5
343 emissions in winter than the other seasons, indicating that more attention should be paid to emissions caused by fossil fuel
344 combustion in winter than the other seasons.

345 It is necessary to improve air quality in Beijing, and vehicle exhausts are key sources of PM2.5. Further improvements in
346 ambient air quality to meet stricter ambient air quality standards will require vehicle emissions to be controlled to decrease
347 particulate matter pollution. The number of vehicles using fossil fuels in Beijing needs to be decreased. Achieving this will
348 require policies for restricting the use of vehicles using fossil fuels and the use of cleaner energy vehicles to be promoted. In
349 summary, controlling and decreasing emissions caused by fossil fuel combustion will decrease PM2.5 emissions and
350 improve ambient air quality in Beijing.

351 **5 Conclusions**

352 The PM_{2.5} concentrations and C₁₃–C₄₀ *n*-alkane concentrations in fine particulate matter between November 2020 and
353 October 2021 were determined and the concentrations were compared with concentrations found in previous studies. The
354 PM_{2.5} and *n*-alkane concentrations in Beijing have decreased in similar ways in the last five years. The mean PM_{2.5}
355 concentration was 32.0 µg/m³, which met the secondary ambient air quality standard for China. The PM_{2.5} and C₁₃–C₄₀ *n*-
356 alkane concentrations varied in similar ways and positively correlated ($p < 0.01$), so long chain *n*-alkanes in particulate matter
357 can be used to assess the sources of particulate matter pollution in urban areas and to develop strategies for controlling
358 particulate matter pollution.

359 The *n*-alkane concentrations in the different seasons decreased in the order winter>spring>summer>fall. There were marked
360 seasonal and diurnal differences in the *n*-alkane homolog patterns and distributions. The source indices and PMF model
361 results explained these variations in patterns and allowed the sources of *n*-alkanes to be identified. The source indices
362 indicated that *n*-alkane concentrations in particulate matter in Beijing are affected by both anthropogenic and biogenic
363 emissions but that anthropogenic emissions are dominant. The PMF model allowed the contributions of the sources of *n*-
364 alkanes to be quantified and indicated that emissions from vehicles are currently the main sources of PM_{2.5} and *n*-alkanes in
365 particulate matter in urban areas.

366 Controlling PM_{2.5} and *n*-alkanes emissions from vehicles is key to decreasing PM_{2.5} and *n*-alkanes pollution and improving
367 air quality in urban areas. *n*-Alkanes in particulate matter can be used as organic tracers, and PMF model results can indicate
368 the sources of PM_{2.5} pollution. Further research into the use of this method is required.

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373 **Data availability**

374 The data presented in this article are available from the authors upon request (junjin3799@126.com).

375 **Author contribution**

376 JJ conceived and designed the study, provided direct funding and helped with manuscript revision. JYY mainly conducted
377 the sampling, sample analysis work, as well as manuscript writing and revision. Other authors helped this work by sampling
378 and analysis. All authors read and approved the final manuscript.

379 **Competing interests**

380 The authors declare that they have no conflict of interest.

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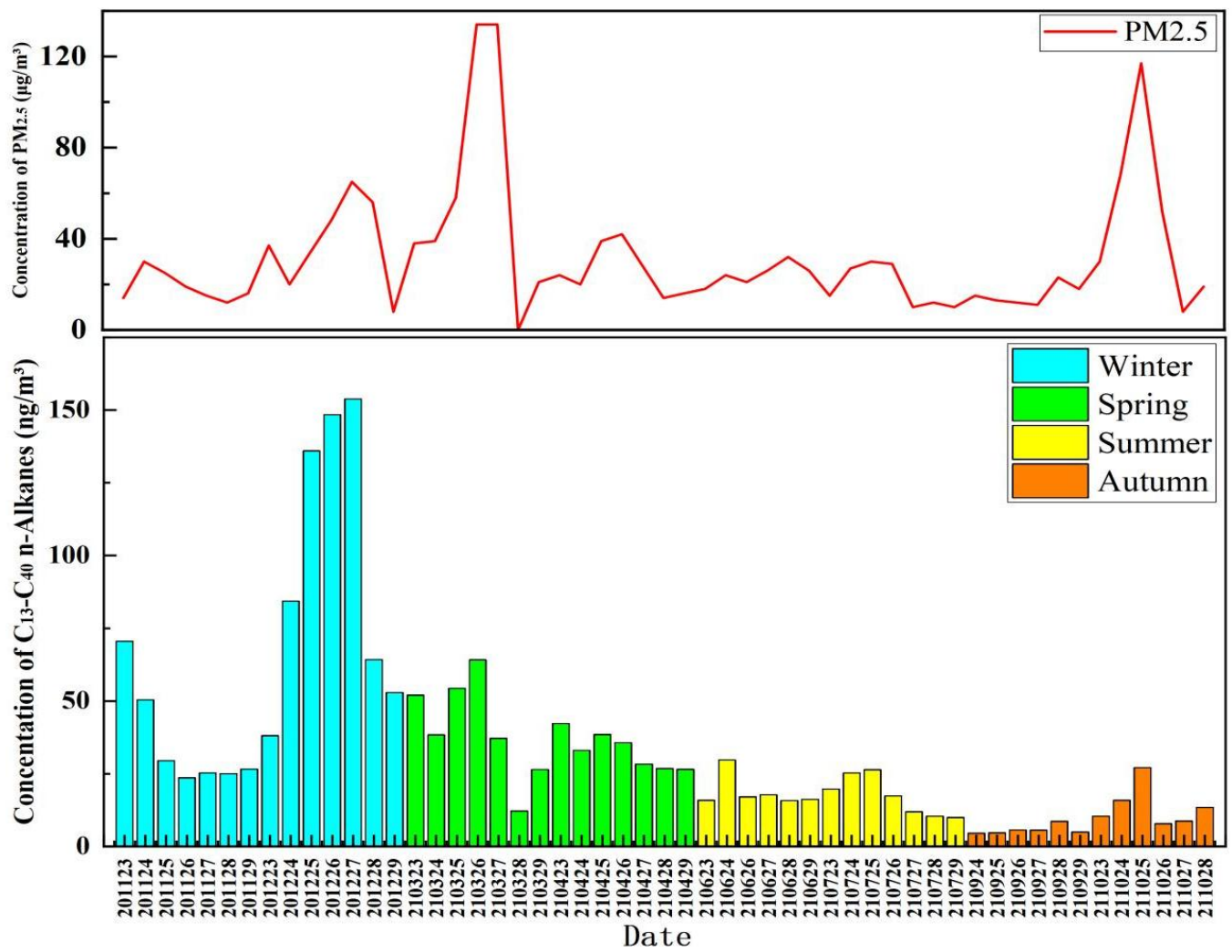
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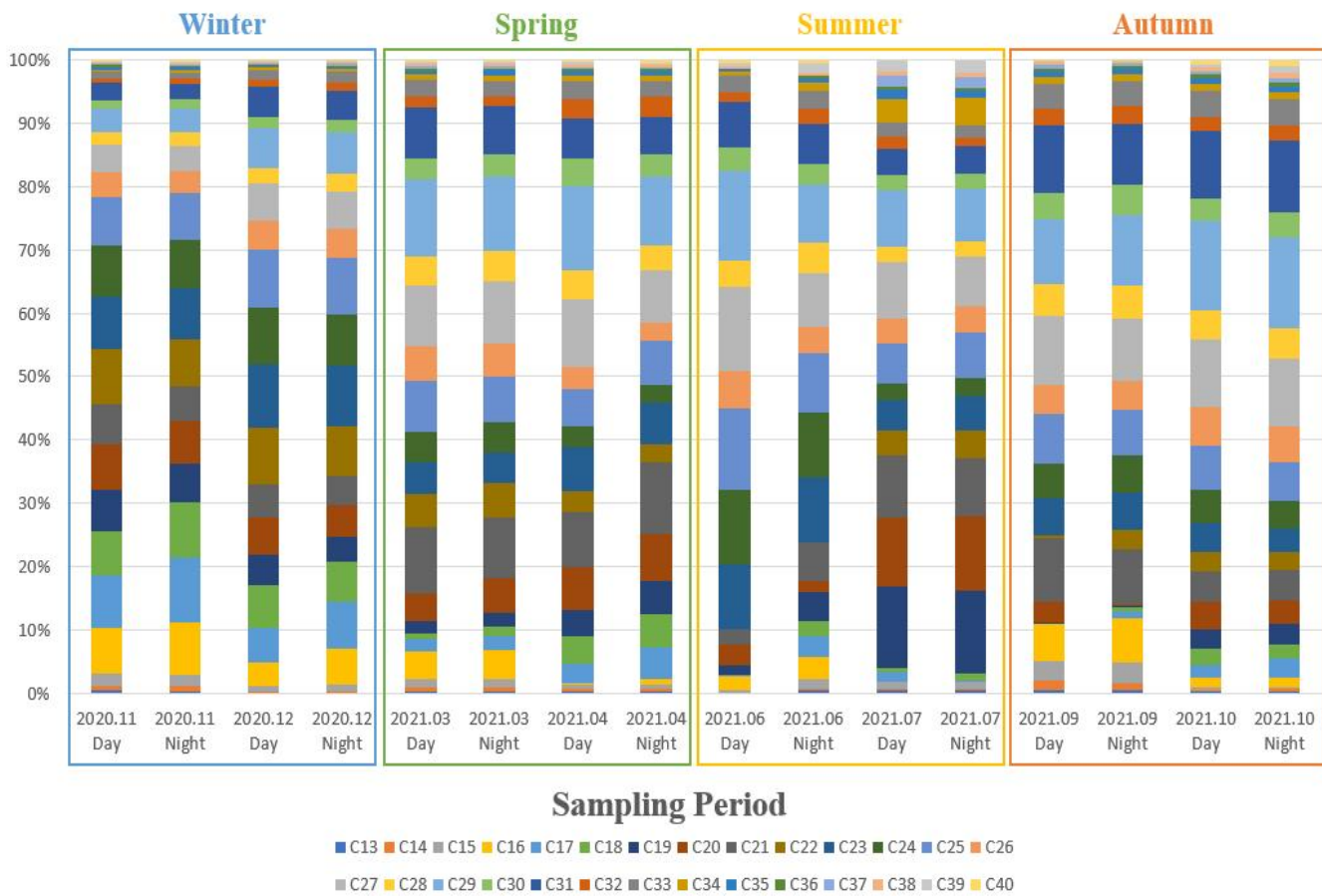
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Figure 1. Temporal variations in PM_{2.5} and particulate-bound *n*-alkane concentrations during the sampling period in Beijing.

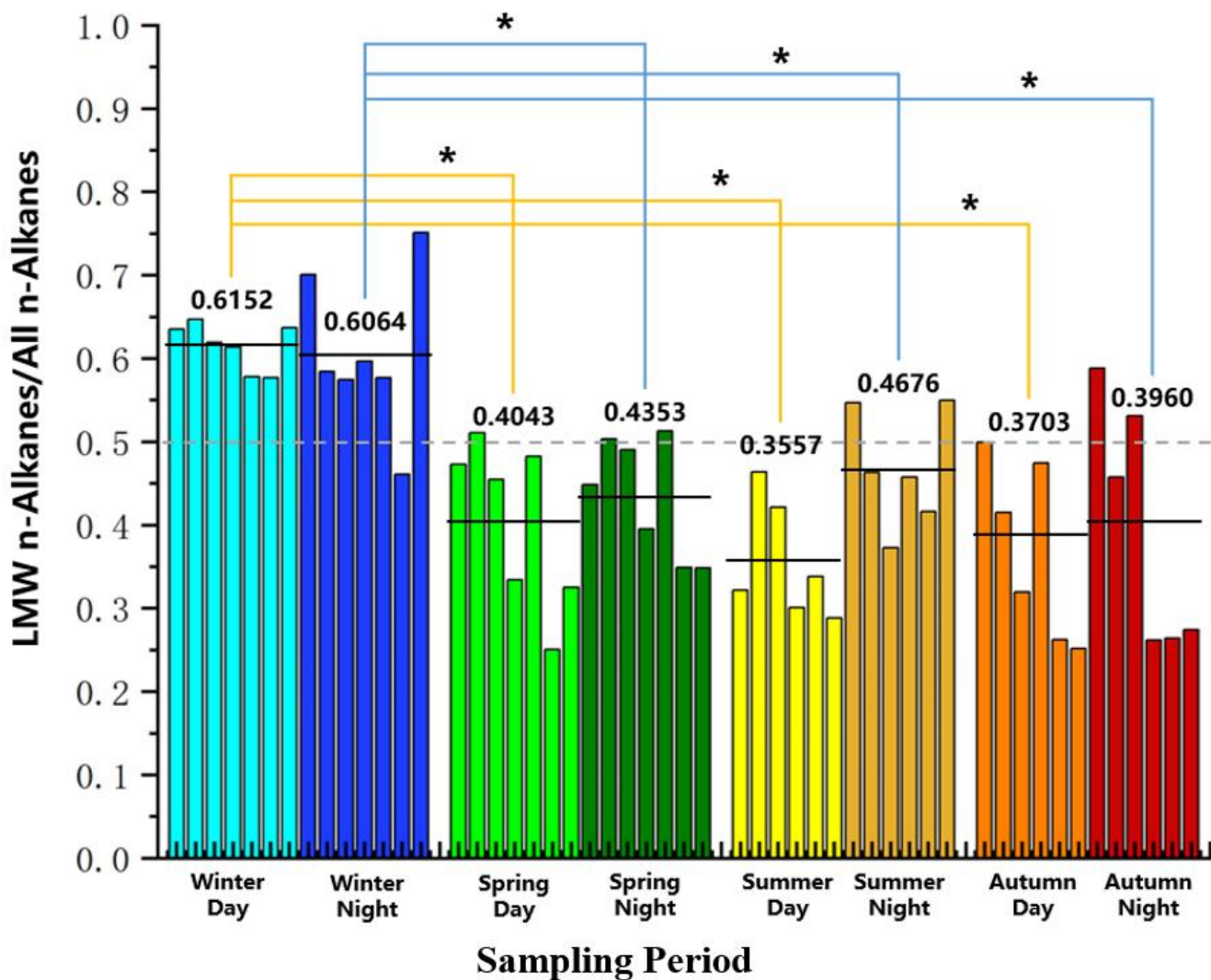
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(The concentrations of C₁₃-C₄₀ *n*-Alkanes and PM_{2.5} are the average of the day and night).



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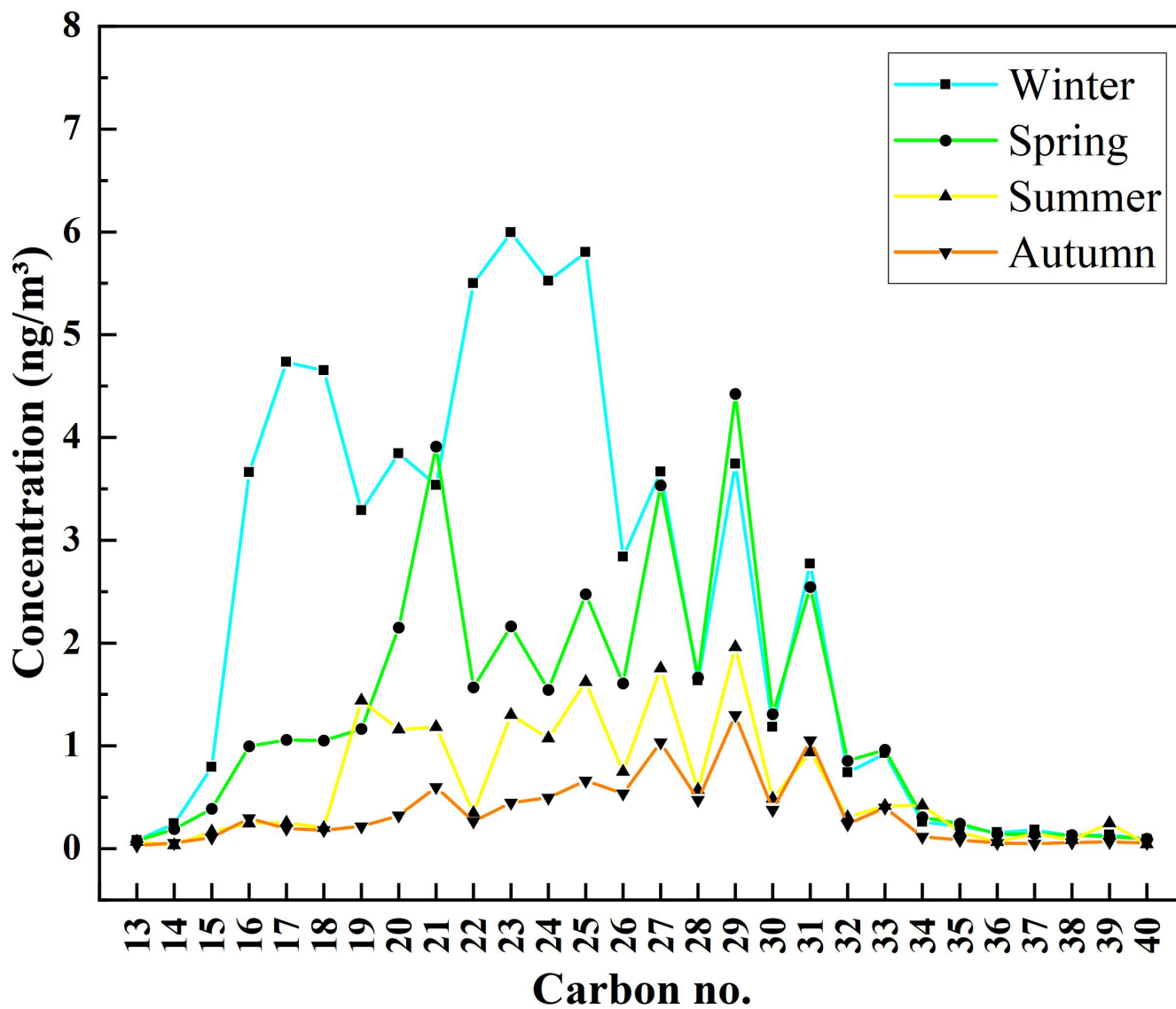
581 **Figure 2. Contributions of particulate-bound *n*-alkane homologs to the total *n*-alkane concentrations in the day and night samples**
 582 **in the different seasons of Beijing.**



583

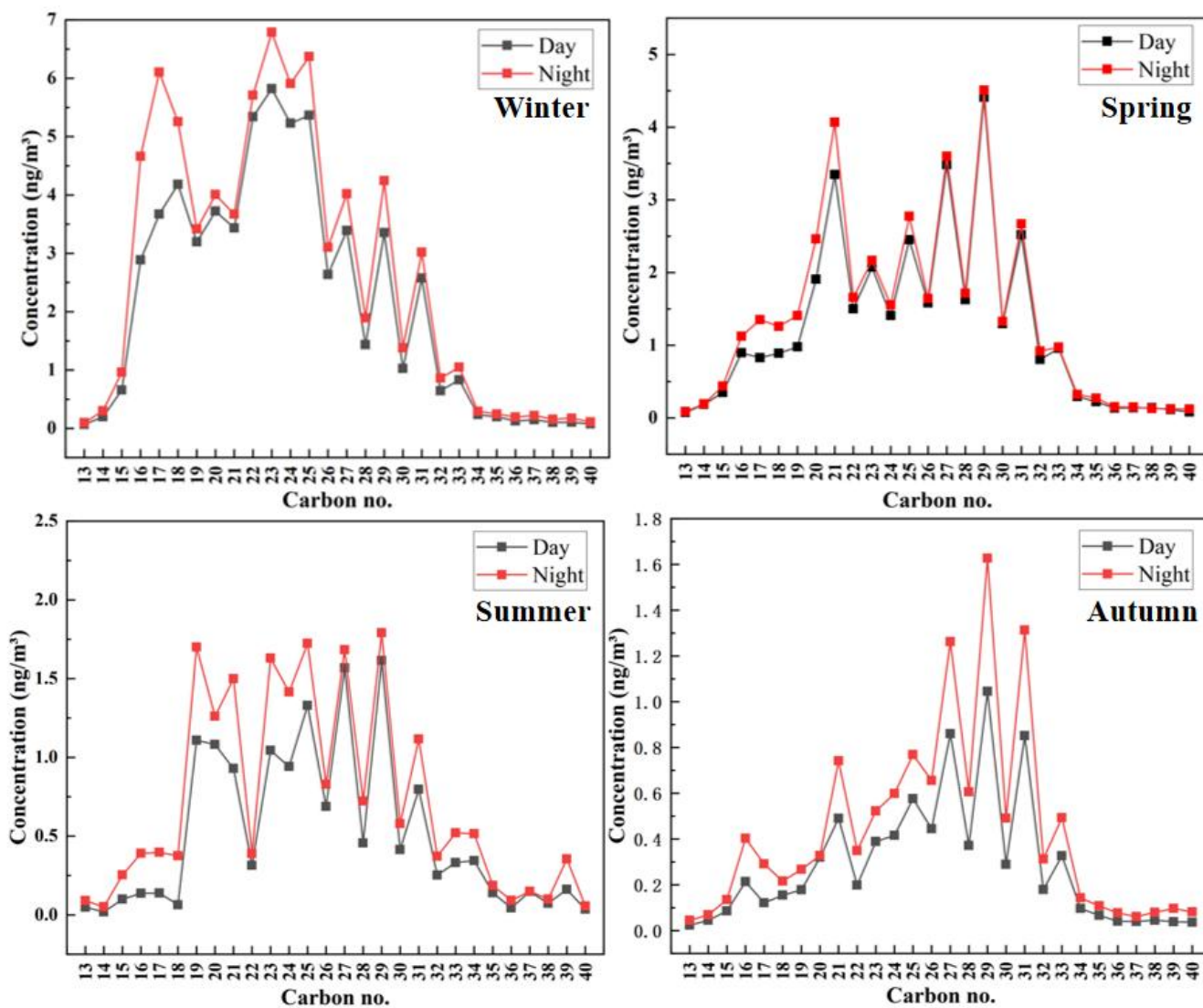
584 Figure 3. Contributions of low molecular weight *n*-alkanes in the day and night samples in the different seasons of Beijing.

585 (* indicates a significant difference, dashed line represents the 50% percentage, solid line shows the average proportion of LMW
 586 *n*-alkanes).



587

588 Figure 4. Average concentration distributions of the particulate-bound *n*-alkane homologs in the different seasons of Beijing.

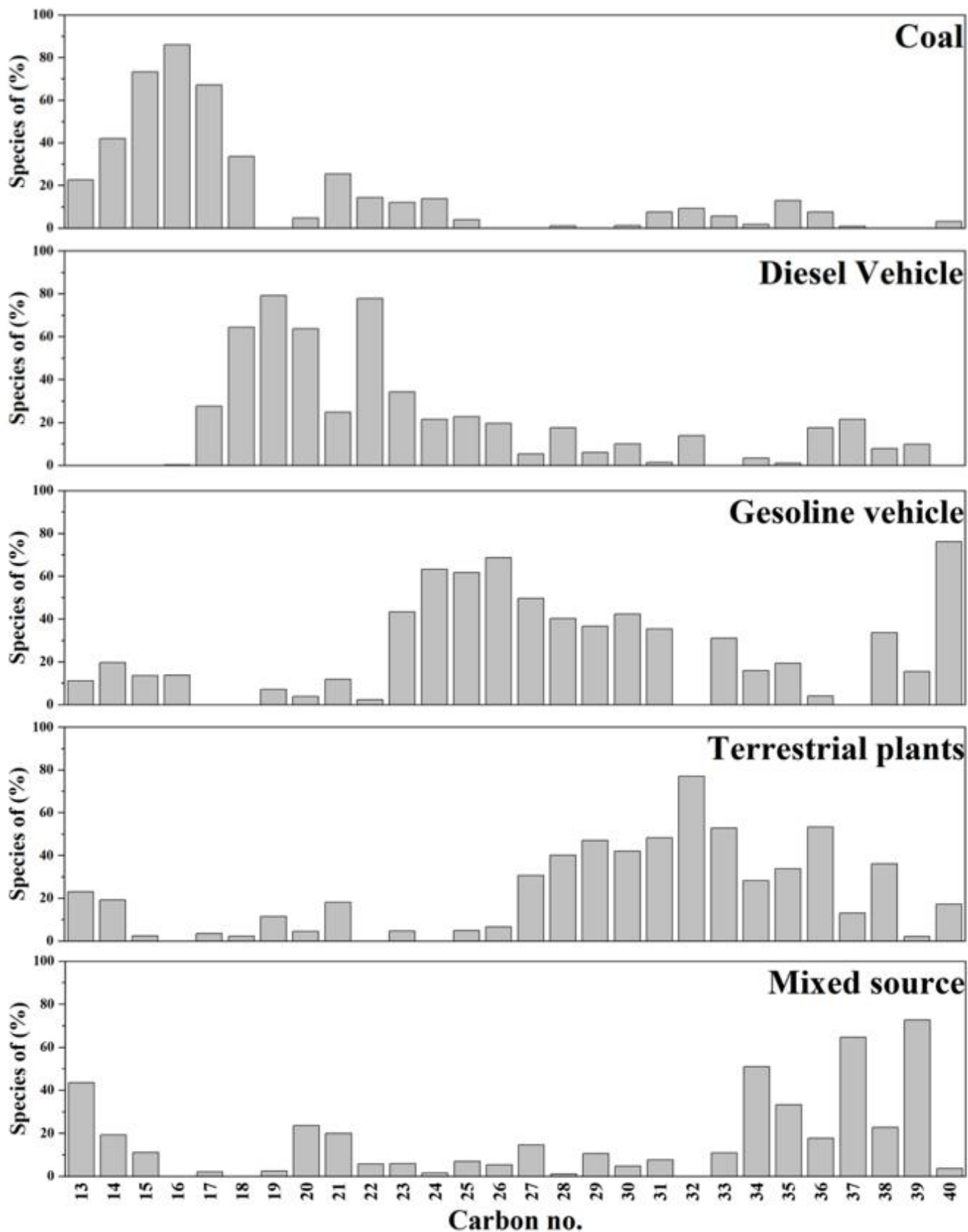


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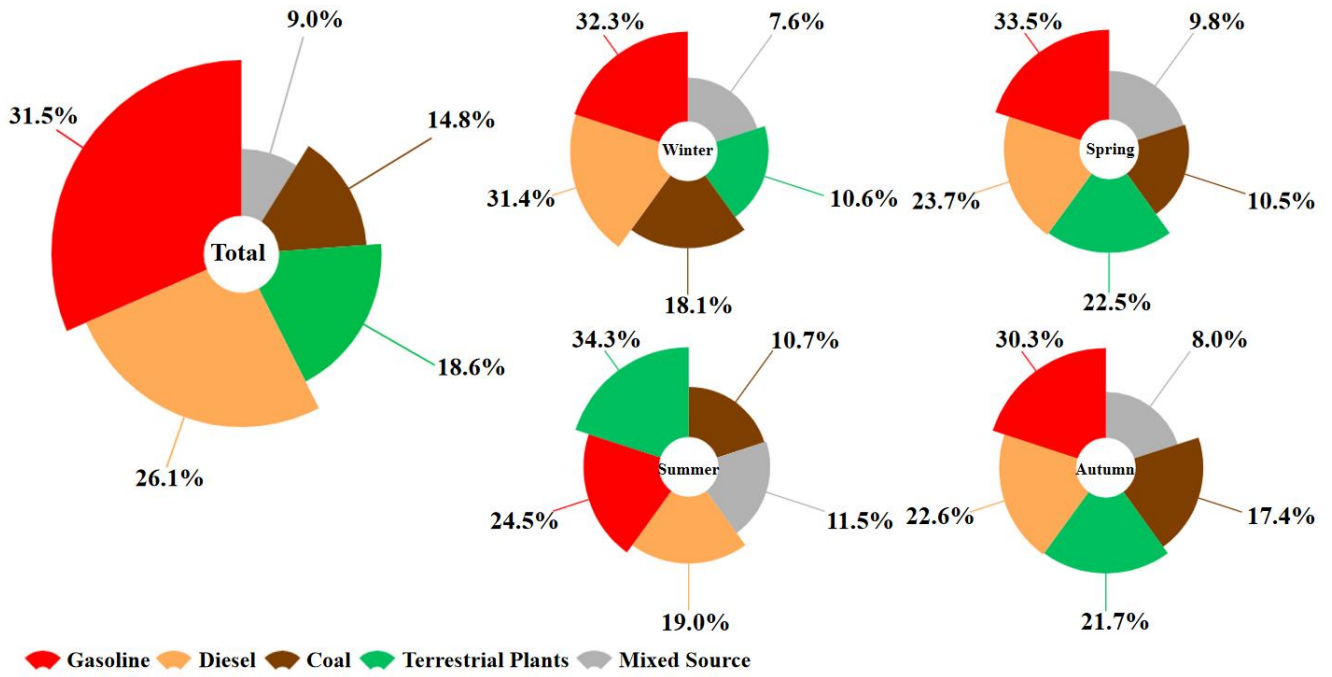
Figure 5. Concentration distributions of the particulate-bound *n*-alkane homologs in the day and night in the different seasons of Beijing.



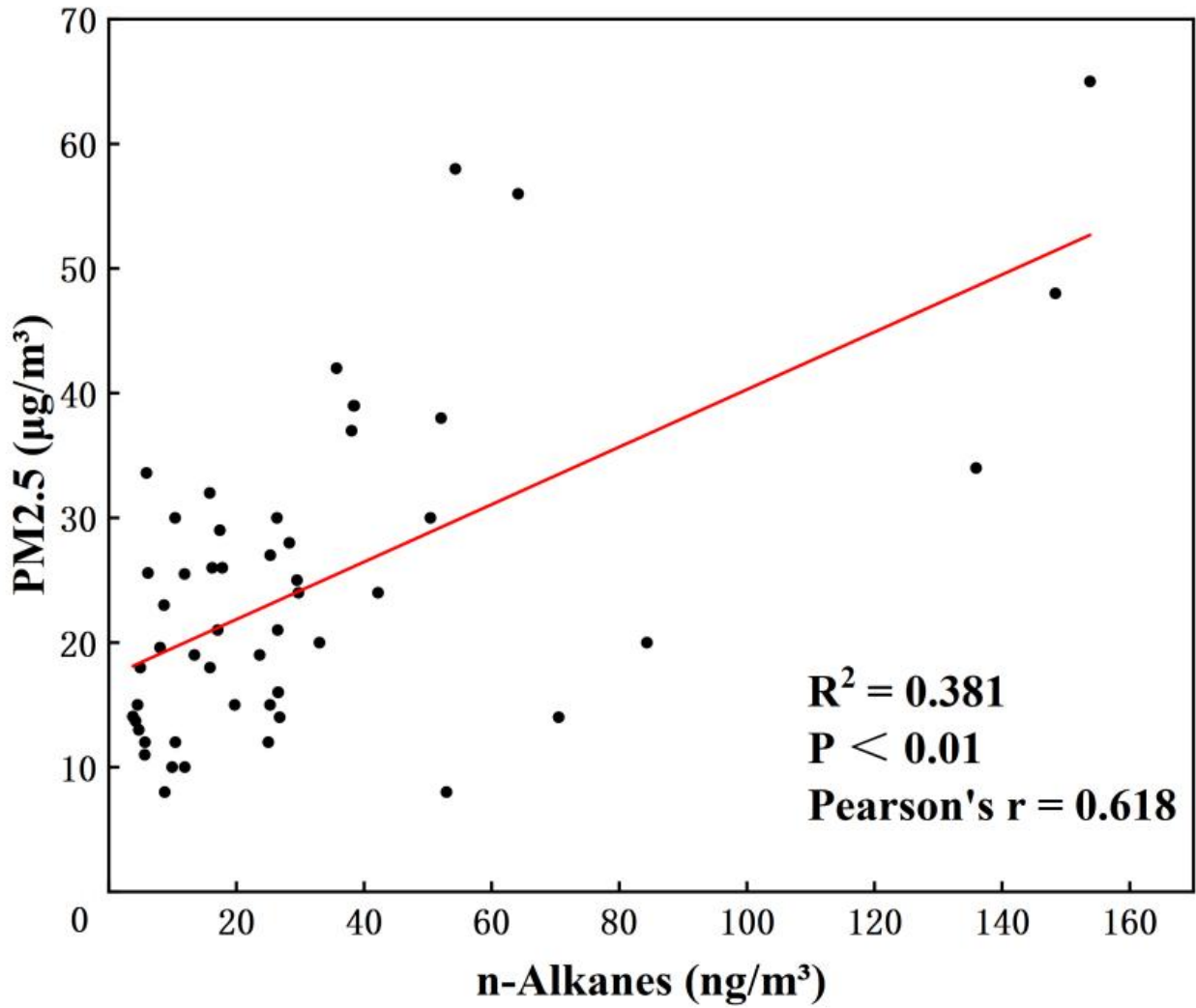
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Figure 6. Proportions of the different *n*-alkane homologs in the factors identified by the positive matrix factorization model.



594
 595 Figure 7. Sources and contributions of particulate-bound *n*-alkanes in Beijing.



596
 597 Figure 8. Association between particulate-bound *n*-alkanes and PM2.5 in Beijing.

598 **Table 1. PM2.5 and particulate-bound *n*-alkane concentrations in different seasons in Beijing.**

Species	Winter ^a		Spring ^b		Summer ^c		Fall ^d	
	Mean	Range	Mean	Range	Mean	Range	Mean	Range
PM2.5 ($\mu\text{g}/\text{m}^3$)	28.5	8.00–65.0	43.5	0–134	21.5	10.0–32.0	32.2	8.00–117
<i>n</i> -Alkanes (ng/m^3)	66.3	17.1–89.9	36.8	12.2–64.1	18.0	9.92–29.7	9.78	4.51–27.1

599 ^a Winter: November and December in 2020;600 ^b Spring: March and April in 2021;601 ^c Summer: June and July in 2021;602 ^d Fall: September and October in 2021.603 **Table 2. Source indices for particulate-bound *n*-alkane in Beijing.**

Source Index	Winter		Spring		Summer		Fall	
	Day	Night	Day	Night	Day	Night	Day	Night
Cmax^a	C23	C23	C29	C29	C29	C29	C29	C29
CPI^b	1.16	1.18	1.85	1.76	2.15	1.87	1.90	1.78
WNA%^c	17.4	18.5	35.0	33.1	43.0	39.2	39.6	35.1
PNA%^d	82.6	81.5	65.0	66.9	57.0	60.8	60.5	64.9

604 ^a Cmax: Carbon maximum number;605 ^b CPI: Carbon preference index;606 ^c WNA%: Plant wax *n*-alkane ratio;607 ^d PNA%: Petrogenic *n*-alkane ratio.